ORIGINAL RESEARCH

Novel quinolyl-thienyl chalcones and their 2-pyrazoline derivatives with diverse substitution pattern as antileishmanial agents against Leishmania major

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Abstract A series of twenty-two new pyrazoline derivatives was prepared from quinoline-based chalcones which in turn were synthesized by condensing formylquinolines with diverse acetylthiophenes. The titled compounds were characterized by spectroscopic techniques (NMR, IR and MS) and elemental analysis. All the compounds were screened for antileishmanial activities. Compounds 1e, 1f, 2a, 2c, 2d, 2g, 2k, and 4a were found potentially active antileishmanial agents. Bioassay results show that the type and positions of the substituents seem to be critical for their antileishmanial activities.

Keywords Chalcones · Pyrazolines · Antileishmanial activity · Formylquinolines · Acetylthiophenes

Introduction

1,3-Diaryl-2-propen-1-ones (chalcones) are one of the most important classes of natural products, and are widespread in the plant kingdom. Chalcones (natural or synthetic) possess a broad spectrum of biological activities including anti-inflammatory (Ballesteros et al.,[1995\)](#page-10-0), antifungal (Nowakowska, [2007\)](#page-10-0), antioxidant (Mukherjee et al., [2001](#page-10-0)), antimalarial (Wu et al., [2002\)](#page-11-0), antituberculosis (Sivakumar

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et al., [2007\)](#page-10-0), analgesic (Viana et al., [2003](#page-11-0)), antitumor (Shibata, [1994\)](#page-10-0), anticancer (Wattenberg et al., [1994](#page-11-0)), antiviral (Trivedi et al., [2007\)](#page-10-0), anti-AIDS (Wu et al., [2003\)](#page-11-0) and antileishmanial agents (Boeck et al., [2006](#page-10-0)).

Pyrazoline derivatives of chalcones have been reported to possess a widespread range of biological activities like antibacterial (Nauduri and Reddy, [1998](#page-10-0)), antifungal (Azarifar and Shaebanzadeh, [2002](#page-10-0)), antidepressant (Bilgin et al., [1993](#page-10-0)), antitumor (Taylor and Patel, [1992](#page-10-0)), antimicrobial (Ramalingham et al., [1977\)](#page-10-0), anti-inflammatory, molluscicidal activity (Barsoum et al., [2006](#page-10-0)), antiamoebic (Budakoti et al., [2006\)](#page-10-0), anticonvulsant activities (Ozdemir et al., [2007](#page-10-0)). One of the most famous pyrazole-based drugs used as a non-steroidal anti-inflammatory drug (NSAID) is celecoxib (Fig. [1](#page-1-0)) (Rezende, et al., [2010\)](#page-10-0). Considerable attention has been focused on the pyrazoline family in the last two decades. Among various pyrazoline derivatives, 2-pyrazolines seem to be the most frequently studied pyrazoline-type compounds (Lévai, 2005). After the pioneering work by Fischer and Knövenagel in the late nineteenth century, the reaction of α , β -unsaturated aldehydes and ketones with hydrazines became one of the most popular methods for the preparation of 2-pyrazolines (Lévai et al., [2004](#page-10-0)).

Quinolines and their derivatives, which represent a major class of heterocycles (Meth-Cohn and Narine, [1978\)](#page-10-0) and are widely found in natural products (Michael [2003,](#page-10-0) [2004](#page-10-0)) and drugs (Alhaider et al., [1985;](#page-10-0) Campbell et al., [1988](#page-10-0); Du, [2003\)](#page-10-0), exhibit significant role in medicinal chemistry. Several quinoline derivatives have been reported to exhibit bactericidal (Awad et al., [1991\)](#page-10-0), antimalarial (Ginsburg et al., [1999\)](#page-10-0), antiallergenic (Althuis et al., [1980\)](#page-10-0) and anti-inflammatory (Dillard et al., [1973\)](#page-10-0) properties. Some of the famous antimalarial drugs, containing quinoline ring system; available in the market are plasmoquine

Fig. 1 Structures of a quinoline and a pyrazole-based drugs

(Manske and Kulka, [1953](#page-10-0)), primaquine and chloroquine (Singh et al., [1978](#page-10-0)). Many quinoline derivatives are found to possess anticancer and antitumor activities (Loaiza et al., [2004\)](#page-10-0). Among the quinolines, 2-chloro-3-formylquinolines find an important place in synthetic organic chemistry, as these are key intermediates for further β -annelation of a wide variety of ring systems and for the inter-conversions of many functional groups (Meth-Cohn, [1993\)](#page-10-0).

In this study, the 2-chloro-7/8-methyl-3-formylquinoline nucleus and chalcone functionality have been incorporated in a single molecule $(1a-k$ and $2a-k$). Then each of the prepared chalcones was refluxed with hydrazine hydrate in ethanol to yield twenty-two novel 2-pyrazolines (3a–k and 4a–k) based on quinolyl-thienyl ring systems. Finally, all the title compounds were tested for their antileishmanial activities.

Results and discussion

Chemistry

The two precursors, 2-chloro-3-formyl-7- and 8-methylquinoline were prepared by reported method (Meth-Cohn et al., [1981\)](#page-10-0). Synthesis of the chalconess (1a–k and 2a–k) was based on Claisen-Schmidt condensation (Li et al., [1995\)](#page-10-0). For this purpose, the prepared formyl quinolines were condensed with commercially available acetyl thiophenes (Table 1), in the presence of sodium hydroxide. Finally, chalcones thus prepared, on cyclization with hydrazine hydrate, gave the corresponding 2-pyrazoline derivatives (3a–k and 4a–k) in a reasonably good yield (Scheme [1](#page-2-0)).

Spectral data $\left(\mathbb{IR}, \right.$ ¹H-NMR and MS) of all the newly synthesized compounds were found in good agreement with the proposed structures. IR spectra of the compounds 1a–k and 2a–k showed an absorption band at 1650 cm^{-1} , typical of the stretching vibrations of chalcone moiety. No

Table 1 Aryl moiety (Ar) in chalcones (1a–k) and (2a–k)

Ketones	Ar
a	$C_4H_3S-3-yl$
b	$3 - CH_3C_4H_2S - 2 - yl$
c	$4-CH_3C_4H_2S-2-yl$
d	5 -CH ₃ C ₄ H ₂ S-2-yl
e	$2,5$ -diCH ₃ C ₄ HS-3-yl
f	$3-CIC4H2S-2-yl$
g	$5-CIC4H2S-2-yl$
h	2,5-diClC ₄ HS-3-yl
i	$3-BrC4H2S-2-yl$
j	$5-BrC4H2S-2-yl$
k	$5-IC_4H_2S-2-yl$

peaks were found due to starting material aldehydic functionality as impurity. In the 1 H-NMR spectra of the chalcones (1a–k except 1e, 1f and 1i) and (2a–k except 2e, 2f and 2i), two very sharp doublets around δ 7.40 ppm for H_{α} and δ 8.20 ppm for H_b, with J-value 15–16 Hz for the *trans* chalcones were exhibited. Interestingly, in chalcones 1e and 2e, H_{α} and H_{β} showed a doublet in the upfield at δ 7.33–7.32 and 8.10–8.08 ppm, respectively. This may be attributed to an additional $+I$ effect induced by $CH₃$ group present in the close vicinity. Similarly, in chalcones (1f, 1i, 2f and 2i), H_{α} revealed a doublet relatively in the downfield at δ 7.82–7.81 ppm. This may be attributed to an additional -I effect due to Cl/Br in the vicinity. The molecular ion observed in the mass spectra for all the chalcones confirmed their molecular masses. The base peak, in the mass spectra of most of the chalcones, was obtained possibly by the cleavage of HC–CO bond in the chalcone moiety. While in bromo- and iodo-substituted chalcones (1i–1k and $2i-2k$, the base peak is due to the cleavage of two bonds, i.e. CO—thiophenyl and Br/I—thiophenyl bonds. The Econfiguration was confirmed by X-ray structure of two similar structured chalcones which were already reported (Rizvi et al., [2008](#page-10-0)).

Similarly, in case of 2-pyrazolines (3a–k and 4a–k), IR spectra of all the compounds did not show absorbance at 1650 cm^{-1} which confirmed the absence of the chalcone moiety. A new peak with absorption band at 3280 cm^{-1} was revealed due to NH of 2-pyrazoline ring. ¹H-NMR spectra of the pyrazolines (3a–k and 4a–k) ascertained the presence of two doublets of doublet signals due to $CH₂$ protons H_{α} (upfield H of CH₂) at δ 2.82–3.21 ppm region and H_{β} (downfield H of CH₂) at δ 3.64–4.10 ppm, respectively. The CH proton appeared as a triplet at δ 5.28–5.40 ppm region. The molecular ion M^+ , observed in the mass spectra for all the pyrazolines confirmed their molecular masses. The base peak, in almost all the mass spectra (except for 3f), was exhibited by M^+ itself.

Scheme 1 Reaction protocol for the synthesis of 2-pyralozine derivatives (3a–k) and $(4a-k)$ (i) AcOH, H_3PO_4 , reflux, 4–6 h, (ii) POCl₃, DMF, 80 $^{\circ}$ C, (iii) a–k, NaOH, rt, 2 h, (iv) Hydrazine, EtOH, reflux

Antileishmanial activity

According to the results obtained, structure–activity relationship among the two series of chalcones (1a–k and 2a–k) may be explained in terms of stereo- and electronic and/or steric properties (see Fig. 2).

For example, the unsubstituted thiophenyl derivatives (1a and 2a) have prominent difference in antileishmanial activities, i.e. 2a is more active than 1a (IC₅₀ = 0.88 \pm 0.20 µg/ml for **1a** and $IC_{50} = 0.61 \pm 0.81$ µg/ml for **2a**), while the activity decreased considerably by the introduction of CH_3 group at position 3 of thiophenyl ring (1b and 2b; Table [2\)](#page-3-0) perhaps due to steric effect, whereas, the activity is relatively increased on moving the methyl substituent to position 4 (1c and 2c) or 5 (1d and 2d) impairing the steric effect. Conversely, replacing the methyl group by chloro group at position 3 of thiophenyl ring (1f and 2f) results in enhanced activity. This may be attributed to the

Fig. 2 Proposed stereo-, electronic and/or steric properties (i) electronic effect (attractive forces), (ii) steric effect; (iii) electronic effect (dipolar repulsive forces)

greater electronic effect of chloro group, while the steric effect of bromo group overweighs the electronic effect at position 3 of thiophenyl ring (1i and 2i). Moreover, substituting the halogen atoms (Cl, Br and I) at position 5 of thiophenyl ring $(1g, 1j, 1k, 2g, 2j$ and $2k$) deactivates these compounds due to the absence of electronic effect stereochemically. Similarly, 2,5-disubstituted methyl derivatives (1e and 2e) displayed more activity than their dichloro analogs (1h and 2h). This may be attributed to the possibility of existence of electronic effect which is due to the dipolar repulsive forces as shown in Fig. 2d by curved arrow (iii). While, there is no such electronic or steric effect in dimethyl derivatives (1e and 2e).

It is quite obvious from Table [2](#page-3-0) that conversion of both series of chalcones (1a–k and 2a–k) to their corresponding 2-pyrazoline derivatives (3a–k and 4a–k) results in an overall decrease in the antileishmanial activity. This fact clearly indicates the significance of chalcone moiety

Table 2 Antileishmanial activity of series 1a–k, 2a–k, 3a–k and 4a– k (IC₅₀ values)

Compounds	IC_{50} (µg/ml)
1a	0.88 ± 0.20
1 _b	0.92 ± 0.11
1c	0.83 ± 0.05
1d	0.74 ± 0.31
1e	0.62 ± 0.24
1f	0.59 ± 0.20
1g	0.81 ± 0.09
1 _h	0.78 ± 0.14
1i	0.71 ± 0.18
1j	0.91 ± 0.31
1 _k	0.84 ± 0.22
2a	0.61 ± 0.81
2 _b	0.94 ± 0.10
2c	0.59 ± 0.09
2d	0.61 ± 1.25
2e 2f	0.73 ± 0.08
	0.78 ± 0.15 0.65 ± 0.14
2g 2 _h	0.85 ± 0.18
2i	0.93 ± 0.99
2j	0.78 ± 0.032
2k	0.67 ± 0.23
3a	0.94 ± 0.02
3 _b	0.76 ± 0.05
3c	0.87 ± 0.08
3d	0.89 ± 0.03
3e	0.76 ± 0.19
3f	0.78 ± 0.07
3g	0.83 ± 0.50
3 _h	0.71 ± 0.45
3i	0.85 ± 0.18
3j	0.93 ± 0.62
3k	0.88 ± 0.27
4a	0.67 ± 0.09
4b	0.78 ± 0.23
4c	0.74 ± 0.09
4d	0.89 ± 0.10
4e	0.93 ± 0.16
4f	0.84 ± 0.07
4g	0.75 ± 0.02
4h	0.79 ± 0.03
4i	0.94 ± 0.20
4j	0.93 ± 0.20
4k	0.77 ± 0.02
Standard drug MIC(μ g/ml \pm SD) (Amphotericin B)	0.56 ± 0.20

towards antileishmanial activity in the titled compounds. In Fig. [2](#page-2-0), this is proposed to be due to electronic effect shown by curved arrows, labelled as (i), which is vanished by the ring-closure at chalcones moiety.

Conclusion

It is evident from the above discussion that the chalcones (1a–k and 2a–k) exhibited more activity than their corresponding pyrazoline derivatives (3a–k and 4a–k). We divided the compounds into four categories for their antileishmanial activities and represrented in Table [3](#page-4-0), i.e. $IC_{50} = 0.59{\text -}0.56$ µg/ml or below as significantly active, 0.69–0.60 μ g/ml as good activity, 0.79–0.70 μ g/ml as moderately active and 0.95–0.80 μ g/ml as low activity. The compounds 1e, 1f, 2a, 2c, 2d, 2g, 2k and 4a were found potentially active antileishmanial agents.

Experimental

General

Melting points were taken on Gallenkamp melting point apparatus and remained uncorrected. IR spectra were recorded in KBr pellets on Perkin Elmer infrared spectrophotometer. ¹H NMR spectra were performed in CDCl₃ on Brücker/XWIN NMR (400 MHz) and TMS was used as internal standard (chemical shifts, δ in ppm) unless otherwise specified. Mass spectra were recorded on a Jeol MSRoute instrument. Thin layer chromatography (TLC) was performed with aluminium sheets-Silica gel 60 F254 purchased from Merck. Purification of synthesized compounds was made by recrystallization from appropriate solvents. Reagent grade chemiocals such as phosphoryl chloride, acetyl thiophenes, o-toluidine, m-toluidine, N,Ndimethylformamide and hydrazine hydrate (Aldrich and Alpha Aesar) were used as received. Elemental analyses were performed by C.S.I.C., Madrid Spain and were within ±0.4% of predicted values for all compounds.

General procedure for the preparation of (2E)-3-(2-chloro-7/8-methylquinolin-3-yl)-1-(Ary) prop-2-en-1-ones $(1a-k)$ and $(2a-k)$

The two precursors, 7/8-methyl-substituted 2-chloro-3 formylquinolines were synthesized following literature method (Meth-Cohn et al., [1981](#page-10-0)). A mixture of formylquinoline (10 mmol) and an aromatic ketone (10 mmol) in methanol (50 ml) was stirred at room temperature,

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Table 3 Proposed categories of	
antileishmanial agents	

followed by dropwise addition of aq. NaOH (4 ml, 10%). The stirring was continued for 2 h and the reaction mixture was then kept at 0° C for 24 h. Subsequently, it was poured onto ice-cold water (200 ml). The precipitates were collected by filtration, washed with cold water followed by cold MeOH. The resulting chalcones were recrystallized from $CHCl₃$ and dried in vacuo.

(2E)-3-(2-Chloro-7-methylquinolin-3-yl)-1-thien-3-ylprop- 2 -en-1-one ($1a$) Yield, 65% ; colourless solid. mp 180–182°C. IR (KBr, cm⁻¹): 1649 (C=O), 1594 (C=C), 1565 (C=N of quinoline ring). ¹H-NMR (CDCl₃) δ : 2.57 $(3H, s, Me), 7.39$ (1H, dd, H_4 ['], $J = 5.1$ Hz, 2.9 Hz), 7.42 (1H, dd, H₅, $J = 8.2$ Hz, 1.2 Hz), 7.45 (1H, d, H_a, $J = 15.7$ Hz), 7.69 (1H, dd, H₅', $J = 5.1$ Hz, 1.0 Hz), 7.76 (1H, d, H₆, $J = 8.3$ Hz), 7.79 (1H, s, H₈), 8.19 (1H, d, H_β, $J = 15.6$ Hz), 8.20 (1H, dd, H₂', $J = 2.9$ Hz, 1.0 Hz), 8.42 (1H, s, H₄). MS (m/z) : 313 $(M^+, 1.86\%)$, 111 $(M^+$ –C₁₂H₉NCl, 100%). Anal. Calcd for C₁₇H₁₂ClNOS: C, 65.07; H, 3.85; N, 4.46. Found: C, 65.03; H, 3.76; N, 4.43.

(2E)-3-(2-Chloro-7-methylquinolin-3-yl)-1-(3-methylthien-2-yl) prop-2-en-1-one $(1b)$ Yield, 51%; pale yellow solid. mp 208–210°C. IR (KBr, cm⁻¹): 1653 (C=O), 1594 (C=C), 1563 (C=N of quinoline ring). ¹H-NMR (CDCl₃) δ : 2.56–2.66 (s, 2 \times Me), 7.02 (1H, d, H₄', J = 4.9 Hz), 7.40 (1H, d, H_a, $J = 15.4$ Hz), 7.42 (1H, dd, H₅, $J = 8.2$ Hz, 1.3 Hz), 7.49 (1H, d, H_5' J = 4.9 Hz), 7.77 (1H, d, H_6 , $J = 8.6$ Hz), 7.78 (1H, s, H₈), 8.18 (1H, d, H_b, $J = 15.4$ Hz), 8.40 (1H, s, H₄). MS (m/z): 327 (M⁺, 10%), 125 $(M^{\dagger}-C_{12}H_9NCl, 100\%)$. Anal. Calcd for $C_{18}H_{14}CINOS$: C, 65.95; H, 4.30; N, 4.27. Found: C, 65.92; H, 4.25; N, 4.25.

(2E)-3-(2-Chloro-7-methylquinolin-3-yl)-1-(4-methylthien-2-yl)prop-2-en-1-one $(1c)$ Yield, 56%; yellow solid. mp 173-174°C. IR (KBr, cm⁻¹): 1655 (C=O), 1594 (C=C), 1564 (C=N of quinoline ring). 1 H-NMR (CDCl₃) δ : 2.33–2.56 (s, 2 \times Me), 7.31 (1H, s, H₅'), 7.42 (1H, dd, H₅, $J = 8.2$ Hz, 1.3 Hz), 7.44 (1H, d, H_a, $J = 15.5$ Hz), 7.71 $(1H, s, H_3')$, 7.76 $(1H, d, H_6, J = 8.3 \text{ Hz})$, 7.79 $(1H, s, H_8)$, 8.21 (1H, d, H_B, $J = 15.6$ Hz), 8.42 (1H, s, H₄). MS (m/z): 327 (M^+ , %), 125 (M^+ – $C_{12}H_9NCl$, 100%). Anal. Calcd for C18H14ClNOS: C, 65.95; H, 4.30; N, 4.27. Found: C, 65.85; H, 4.24; N, 4.23.

(2E)-3-(2-Chloro-7-methylquinolin-3-yl)-1-(5-methylthien-2-yl)prop-2-en-1-one (\mathbf{Id}) Yield, 52%; pale yellow solid. mp 173–175°C. IR (KBr, cm⁻¹): 1652 (C=O), 1595 (C=C), 1563 (C=N of quinoline ring). ¹H-NMR (CDCl₃) δ : 2.56–2.57 (s, 2 \times Me), 6.86 (1H, d, H₄', J = 3.3 Hz), 7.42 (1H, d, H_n, $J = 15.6$ Hz), 7.43 (1H, dd, H₅, $J = 8.2$ Hz, 1.2 Hz), 7.71 (1H, d, H_3' , $J = 3.7$ Hz), 7.75 (1H, d, H_6 , $J = 8.3$ Hz), 7.78 (1H, s, H₈), 8.18 (1H, d, H_B, $J = 15.6$ Hz), 8.40 (1H, s, H₄). MS (m/z): 327 (M⁺, 3.61%), 125 $(M⁺-C₁₂H₉NCl, 100%)$. Anal. Calcd for $C_{18}H_{14}CINOS$: C, 65.95; H, 4.30; N, 4.27. Found: C, 65.89; H, 4.26; N, 4.25.

(2E)-3-(2-Chloro-7-methylquinolin-3-yl)-1-(2,5-dimethylthien-3-yl)prop-2-en-1-one (1e) Yield, 70%; yellow solid. mp 183-185°C. IR (KBr, cm⁻¹): 1648 (C=O), 1590 (C=C), 1565 (C=N of quinoline ring). ¹H-NMR (CDCl₃) δ : 2.44– 2.72 (s, 3 \times Me), 7.10 (1H, s, H₄'), 7.32 (1H, d, H_α, J = 15.7 Hz), 7.41 (1H, dd, H₅, $J = 8.4$ Hz, 1.2 Hz), 7.74 (1H, d, H_6 , $J = 8.3$ Hz), 7.78 (1H, s, H₈), 8.08 (1H, d, H_b, $J =$ 15.7 Hz), 8.37 (1H, s, H₄). MS (m/z): 341 (M⁺, 10.31%), 306 $(M^+$ –Cl, 100%). Anal. Calcd for C₁₉H₁₆ClNOS: C, 66.75; H, 4.72; N, 4.10. Found: C, 66.65; H, 4.68; N, 4.08.

(2E)-3-(2-Chloro-7-methylquinolin-3-yl)-1-(3-chlorothien-2-yl)prop-2-en-1-one (1f) Yield, 66%; yellow solid. mp 160–162°C. IR (KBr, cm⁻¹): 1650 (C=O), 1591 (C=C), 1569 (C=N of quinoline ring). ¹H-NMR (CDCl₃) δ : 2.56 $(3H, s, Me), 7.07$ (1H, d, H_4 ['], $J = 5.2$ Hz), 7.28 (1H, dd, H_5 , $J = 8.3$ Hz, 1.1 Hz), 7.60 (1H, d, H_5' , $J = 5.2$ Hz), 7.77 (1H, d, H₆, $J = 8.6$ Hz), 7.82 (1H, d, H_α, $J = 15.5$ Hz), 7.96 (1H, s, H₈), 8.23 (1H, d, H_b, $J = 15.5$ Hz), 8.42 (1H, s, H₄). MS (m/z): 347 (M⁺, 1.8%), 145 $(M^{\dagger}-C_{12}H_9NCl, 100\%)$. Anal. Calcd. for $C_{17}H_{11}Cl_2NOS$: C, 58.63; H, 3.18; N, 4.02. Found: C, 58.53; H, 3.16; N, 3.97.

(2E)-3-(2-Chloro-7-methylquinolin-3-yl)-1-(5-chlorothien-2-yl)prop-2-en-1-one (Ig) Yield, 80%; pale yellow solid. mp 170–171°C. IR (KBr, cm⁻¹): 1656 (C=O), 1598 (C=C), 1570 (C=N of quinoline ring). ¹H-NMR (CDCl₃) δ : 2.52 $(3H, s, Me), 6.92$ (1H, d, H₄', $J = 4.1$ Hz), 7.35 (1H, dd, H_5 , $J = 8.3$ Hz, 1.2 Hz), 7.39 (1H, d, H_a, $J = 15.6$ Hz), 7.57 (1H, d, H_3' , $J = 4.1$ Hz), 7.64 (1H, d, H_6 , $J = 8.3 \text{ Hz}$), 7.72 (1H, s, H₈), 8.20 (1H, d, H_{*β*},

 $J = 15.6$ Hz), 8.39 (1H, s, H₄). MS (m/z): 347 (M⁺, 2.41%), 145 $(M⁺-C₁₂H₉NCl, 100%)$. Anal. Calcd for $C_{17}H_{11}Cl_2NOS$: C, 58.63; H, 3.18; N, 4.02. Found: C, 58.57; H, 3.14; N, 3.96.

(2E)-3-(2-Chloro-7-methylquinolin-3-yl)-1-(2,5-dichloroth $ien-3-yl$)prop-2-en-1-one (1h) Yield, 63%; off-white solid. mp 163°C. IR (KBr, cm⁻¹): 1662 (C=O), 1596 (C=C), 1572 (C=N of quinoline ring). ¹H-NMR (CDCl₃) δ : 2.56 (3H, s, Me), 7.23 (1H, s, H₄'), 7.42 (1H, dd, H₅, $J = 8.3$ Hz, 1.0 Hz), 7.45 (1H, d, H_a, $J = 15.7$ Hz), 7.76 (1H, d, H₆, $J = 8.3$ Hz), 7.79 (1H, s, H₈), 8.15 (1H, d, H_{*B*}, $J = 15.7$ Hz), 8.39 (1H, s, H₄). MS (m/z): 383 (M⁺, 1.7%), 346 (M⁺-Cl, 100%). Anal. Calcd for C₁₇H₁₀Cl₃NOS: C, 53.35; H, 2.63; N, 3.66. Found: C, 53.26; H, 2.58; N, 3.67.

(2E)-1-(3-Bromothien-2-yl)-3-(2-chloro-7-methylquinolin-3 y l)prop-2-en-1-one (1 i) Yield, 79%; yellow solid. mp 164–165°C. IR (KBr, cm⁻¹): 1652 (C=O), 1592 (C=C), 1568 (C=N of quinoline ring). ¹H-NMR (CDCl₃) δ : 2.57 $(3H, s, Me), 7.16$ $(1H, d, H₄, J = 5.16 Hz), 7.42$ $(1H, dd,$ H_5 , $J = 8.3$ Hz, 1.3 Hz), 7.58 (1H, d, H_5' , $J = 5.2$ Hz), 7.78 (1H, d, H₆, $J = 8.6$ Hz), 7.82 (1H, d, H_α, $J = 15.6$ Hz), 7.79 (1H, s, H₈), 8.23 (1H, d, H_b, $J = 15.5$ Hz), 8.43 (1H, s, H₄). MS (m/z): 393 (M⁺, 1.5%), 82 $(M^{\dagger}-C_{13}H_9NOClBr, 100\%)$. Anal. Calcd for C17H11BrClNOS: C, 51.99; H, 2.82; N, 3.57. Found: C, 51.94; H, 2.76; N, 3.56.

(2E)-1-(5-Bromothien-2-yl)-3-(2-chloro-7-methylquinolin-3 y l)prop-2-en-1-one (1j) Yield, 75%; yellow solid. mp 162-164°C. IR (KBr, cm⁻¹): 1653 (C=O), 1588 (C=C), 1566 (C=N of quinoline ring). ¹H-NMR (CDCl₃) δ : 2.56 $(3H, s, Me), 7.16$ (1H, d, H_4 ['], $J = 4.0$ Hz), 7.38 (1H, d, H_α , $J = 15.6$ Hz), 7.42 (1H, dd, H₅, $J = 8.3$ Hz, 1.0 Hz), 7.63 $(H, d, H_3', J = 4.0 \text{ Hz}),$ 7.76 $(H, d, H_6, J = 8.3 \text{ Hz}),$ 7.79 (1H, s, H₈), 8.21 (1H, d, H_{*β*}, $J = 15.6$ Hz), 8.40 (1H, s, H₄). MS (*m/z*): 393 (M⁺, 1.5%), 82 (M⁺-C₁₃H₉NOClBr, 100%). Anal. Calcd for $C_{17}H_{11}BrClNOS$: C, 51.99; H, 2.82; N, 3.57. Found: C, 51.95; H, 2.79; N, 3.56.

(2E)-3-(2-Chloro-7-methylquinolin-3-yl)-1-(5-iodothien-2 y l)prop-2-en-1-one (1k) Yield, 90%; deep yellow solid. mp 164–165°C. IR (KBr, cm⁻¹): 1650 (C=O), 1596 (C=C), 1565 (C=N of quinoline ring). ¹H-NMR (CDCl₃) δ : 2.56 $(3H, s, Me), 7.36$ (1H, d, H_4 ['], $J = 4.0$ Hz), 7.37 (1H, d, H_α , $J = 15.5$ Hz), 7.42 (1H, dd, H₅, $J = 8.4$ Hz, 1.2 Hz), 7.50 (1H, d, H₃', $J = 4.0$ Hz), 7.76 (1H, d, H₆, $J = 8.3$ Hz), 7.79 (1H, s, H₈), 8.21 (1H, d, H_B, $J = 15.6$ Hz), 8.40 (1H, s, H₄). MS (m/z): 439 (M⁺, 1.5%), 82 (M⁺-C₁₃H₉NOICl, 100%). Anal. Calcd for $C_{17}H_{11}$ ClINOS: C, 46.44; H, 2.52; N, 3.19. Found: C, 46.44; H, 2.43; N, 3.18.

(2E)-3-(2-Chloro-8-methylquinolin-3-yl)-1-thien-3-ylprop-2-en-1-one $(2a)$ Yield, 60% ; pale yellow solid. mp 128-130°C. IR (KBr, cm⁻¹): 1649 (C=O), 1591 (C=C), 1561 (C=N of quinoline ring). ¹H-NMR (CDCl₃) δ : 2.76 $(3H, s, Me), 7.39$ $(1H, dd, H₄, J = 5.1 Hz, 2.9 Hz), 7.46$ (1H, d, H_a, $J = 15.6$ Hz), 7.47 (1H, t, H₆, $J = 7.6$ Hz), 7.60 (1H, d, H₇, $J = 7.0$ Hz), 7.69 (1H, d, H₅', $J = 4.7$ Hz, 1.1 Hz), 7.70 (1H, d, H₅, $J = 6.7$ Hz), 8.20 (1H, d, H_b, $J = 15.7$ Hz), 8.20 (1H, dd, H₂', $J = 2.8$ Hz, 1.1 Hz), 8.42 (1H, s, H₄). MS (m/z) : 313 $(M^+, 1.9\%)$, 111 $(M^+$ - $C_{12}H_9NCl$, 100%). Anal. Calcd for $C_{17}H_{12}CINOS$: C, 65.07; H, 3.85; N, 4.46. Found: C, 65.04; H, 3.78; N, 4.44.

(2E)-3-(2-Chloro-8-methylquinolin-3-yl)-1-(3-methylthien-2-yl)prop-2-en-1-one $(2b)$ Yield, 56%; yellow solid. mp 174–175°C. IR (KBr, cm⁻¹): 1654 (C=O), 1594 (C=C), 1563 (C=N of quinoline ring). ¹H-NMR (CDCl₃) δ : 2.66–2.76 (s, $2 \times$ Me), 7.02 (1H, d, H₄', $J = 4.9$ Hz), 7.41 (1H, d, H_α, $J = 15.4$ Hz), 7.47 (1H, t, H₆, $J = 7.6$ Hz), 7.49 (1H, d, H₅', $J = 5.3$ Hz), 7.59 (1H, d, H₇, $J = 7.0$ Hz), 7.71 (1H, d, H₅, $J = 8.0$ Hz), 8.20 (1H, d, H_b, $J = 15.4$ Hz), 8.40 (1H, s, H₄). MS (m/z) : 327 (M⁺, 6.74%), 125 (M⁺-C₁₂H₉NCl, 100%). Anal. Calcd for C₁₈H₁₄ClNOS: C, 65.95; H, 4.30; N, 4.27. Found: C, 65.90; H, 4.27; N, 4.27.

(2E)-3-(2-Chloro-8-methylquinolin-3-yl)-1-(4-methylthien- $2-yl$)*prop-2-en-1-one* (2c) Yield, 49%; yellow solid. mp 146–147°C. IR (KBr, cm⁻¹): 1655 (C=O), 1593 (C=C), 1565 (C=N of quinoline ring). ¹H-NMR (CDCl₃) δ : 2.33–2.76 (s, 2 \times Me), 7.31 (1H, s, H₅'), 7.46 (1H, d, H_a, $J = 15.5$ Hz), 7.47 (1H, t, H₆, $J = 7.6$ Hz), 7.60 (1H, d, $H_7, J = 7.0$ Hz), 7.70 (1H, d, $H_5, J = 7.0$ Hz), 7.71 (1H, s, H_3'), 8.23 (1H, d, H_β , $J = 15.6$ Hz), 8.42 (1H, s, H₄). MS (m/z) : 327 (M⁺, 5.02%), 292 (M⁺-Cl, 100%). Anal. Calcd for $C_{18}H_{14}$ ClNOS: C, 65.95; H, 4.30; N, 4.27. Found: C, 65.85; H, 4.23; N, 4.22.

(2E)-3-(2-Chloro-8-methylquinolin-3-yl)-1-(5-methylthien- $2-yl$)*prop-2-en-1-one* (2*d*) Yield, 55%; yellow solid. mp 180–181°C. IR (KBr, cm⁻¹): 1652 (C=O), 1596 (C=C), 1563 (C=N of quinoline ring). ¹H-NMR (CDCl₃) δ : 2.57–2.75 (s, 2 \times Me), 6.86 (1H, d, H₄', J = 3.1 Hz), 7.43 (1H, d, H_a, $J = 15.6$ Hz), 7.46 (1H, t, H₆, $J = 7.7$ Hz), 7.59 (1H, d, H₇, $J = 7.0$ Hz), 7.69 (1H, d, H₅, $J = 8.2$ Hz), 7.72 (1H, d, H₃', $J = 3.8$ Hz), 8.19 (1H, d, H_β, $J = 15.6$ Hz), 8.40 (1H, s, H₄). MS (m/z): 327 (M⁺, 5.56%), 125 (M^+ – $C_{12}H_9NCl$, 100%). Anal. Calcd for C18H14ClNOS: C, 65.95; H, 4.30; N, 4.27. Found: C, 65.86; H, 4.25; N, 4.25.

(2E)-3-(2-Chloro-8-methylquinolin-3-yl)-1-(2,5-dimethylthien-3-yl)prop-2-en-1-one (2e) Yield, 67%; yellow solid. mp 138–140°C. IR (KBr, cm⁻¹): 1648 (C=O), 1585 (C=C),

1565 (C=N of quinoline ring). ¹H-NMR (CDCl₃) δ : 2.44–2.75 (s, $3 \times$ Me), 7.10 (1H, s, H₄'), 7.33 (1H, d, H_α, $J = 15.7$ Hz), 7.46 (1H, t, H₆, $J = 7.7$ Hz), 7.59 (1H, d, $H_7, J = 7.0$ Hz), 7.68 (1H, d, H₅, $J = 8.1$ Hz), 8.10 (1H, d, H_{β} , $J = 15.7$ Hz), 8.37 (1H, s, H₄). MS (m/z): 341 (M⁺, 7.71%), 139 (M^+ –C₁₂H₉NCl, 100%). Anal. Calcd for $C_{19}H_{16}CINOS$: C, 66.75; H, 4.72; N, 4.10. Found: C, 66.66; H, 4.62; N, 4.02.

(2E)-3-(2-Chloro-8-methylquinolin-3-yl)-1-(3-chlorothien- $2-yl$)*prop-2-en-1-one* (2*f*) Yield, 73%; yellow solid. mp 162-163°C. IR (KBr, cm⁻¹): 1650 (C=O), 1592 (C=C), 1570 (C=N of quinoline ring). ¹H-NMR (CDCl₃) δ : 2.76 $(3H, s, Me), 7.07$ (1H, d, H₄', $J = 5.3$ Hz), 7.47 (1H, t, H₆, $J = 7.7$ Hz), 7.59 (1H, d, H₇, $J = 7.0$ Hz), 7.60 (1H, d, H_5' , $J = 5.3$ Hz), 7.71 (1H, d, H_5 , $J = 8.1$ Hz), 7.82 (1H, d, H_a, $J = 15.5$ Hz), 8.24 (1H, d, H_b, $J = 15.6$ Hz), 8.42 (1H, s, H₄). MS (m/z): 347 (M⁺, 1.18%), 145 (M⁺ $-C_{12}H_9NCl$, 100%). Anal. Calcd. for $C_{17}H_{11}Cl_2NOS$: C, 58.63; H, 3.18; N, 4.02. Found: C, 58.59; H, 3.12; N, 3.98.

(2E)-3-(2-Chloro-8-methylquinolin-3-yl)-1-(5-chlorothien-2-yl) prop-2-en-1-one $(2g)$ Yield, 85%; pale yellow solid. mp 166–168°C. IR (KBr, cm⁻¹): 1656 (C=O), 1598 (C=C), 1572 (C=N of quinoline ring). ¹H-NMR (CDCl₃) δ : 2.76 $(3H, s, Me), 7.02$ (1H, d, H_4 ['], $J = 4.0$ Hz), 7.39 (1H, d, H_{α} , $J = 15.5$ Hz), 7.48 (1H, t, H₆, $J = 7.6$ Hz), 7.61 (1H, d, H_7 , $J = 7.0$ Hz), 7.68 (1H, d, H_3' , $J = 4.2$ Hz), 7.70 (1H, d, H₅, $J = 8.1$ Hz), 8.23 (1H, d, H_{β}, $J = 15.6$ Hz), 8.41 (1H, s, H₄). MS (m/z): 347 (M⁺-Cl, 1.24%), 145 (M⁺ $-C_{12}H_9NCl$, 100%). Anal. Calcd for $C_{17}H_{11}Cl_2NOS$: C, 58.63; H, 3.18; N, 4.02. Found: C, 58.55; H, 3.13; N, 3.97.

(2E)-3-(2-Chloro-8-methylquinolin-3-yl)-1-(2,5-dichloroth $ien-3-yl$)prop-2-en-1-one (2h) Yield, 69%; colourless solid. mp 120-121°C. IR (KBr, cm⁻¹): 1664 (C=O), 1596 (C=C), 1570 (C=N of quinoline ring). ¹H-NMR (CDCl₃) δ : 2.76 (3H, s, Me), 7.15 (1H, s, H₄'), 7.45 (1H, d, H_α, $J = 15.7$ Hz), 7.47 (1H, t, H₆, $J = 7.7$ Hz), 7.61 (1H, d, $H_7, J = 6.8$ Hz), 7.70 (1H, d, $H_5, J = 8.1$ Hz), 8.17 (1H, d, H_β , $J = 15.7$ Hz), 8.39 (1H, s, H₄). MS (m/z): 383 (M⁺, 1.8%), 179 $(M^{\dagger}-C_{12}H_9NCl, 100\%)$. Anal. Calcd for $C_{17}H_{10}Cl_3NOS$: C, 53.35; H, 2.63; N, 3.66. Found: C, 53.24; H, 2.55; N, 3.60.

(2E)-1-(3-Bromothien-2-yl)-3-(2-chloro-8-methylquinolin-3-yl) prop-2-en-1-one $(2i)$ Yield, 86%; yellow solid. mp 210–212°C. IR (KBr, cm⁻¹): 1652 (C=O), 1592 (C=C), 1568 (C=N of quinoline ring). ¹H-NMR (CDCl₃) δ : 2.76 $(3H, s, Me), 7.16$ (1H, d, H₄', $J = 5.2$ Hz), 7.47 (1H, t, H₆, $J = 7.6$ Hz), 7.58 (1H, d, H₅', $J = 5.2$ Hz), 7.60 (1H, d, $H_7, J = 7.1$ Hz), 7.71 (1H, d, $H_5, J = 8.0$ Hz), 7.81 (1H, d, H_{α} , $J = 15.6$ Hz), 8.25 (1H, d, H_b, $J = 15.5$ Hz), 8.44 (1H, s, H₄). MS (m/z) : 393 $(M^+$, 1.0%), 82 $(M^+$ $-C_{13}H_9NOCIBr$, 100%). Anal. Calcd. for $C_{17}H_{11}BrClNOS$: C, 51.99; H, 2.82; N, 3.57. Found: C, 51.98; H, 2.77; N, 3.59.

(2E)-1-(5-Bromothien-2-yl)-3-(2-chloro-8-methylquinolin- $3-yl$) prop-2-en-1-one (2j) Yield, 71%; off-white solid. mp 204–206°C. IR (KBr, cm⁻¹): 1653 (C=O), 1588 (C=C), 1566 (C=N of quinoline ring). ¹H-NMR (CDCl₃) δ : 2.76 $(3H, s, Me), 7.17$ (1H, d, H_4 ['], $J = 4.0$ Hz), 7.39 (1H, d, H_α , $J = 15.6$ Hz), 7.48 (1H, t, H₆, $J = 7.6$ Hz), 7.61 (1H, d, H_7 , $J = 7.1$ Hz), 7.63 (1H, d, H_3' , $J = 4.0$ Hz), 7.70 (1H, d, H₅, $J = 8.0$ Hz), 8.23 (1H, d, H_b, $J = 15.6$ Hz), 8.41 (1H, s, H₄). MS (m/z) : 393 $(M^+, 2\%)$, 82 $(M^+$ $-C_{13}H_9NOClBr$, 100%). Anal. Calcd for $C_{17}H_{11}BrClNOS$: C, 51.99; H, 2.82; N, 3.57. Found: C, 51.93; H, 2.75; N, 3.55.

(2E)-3-(2-Chloro-8-methylquinolin-3-yl)-1-(5-iodothien-2 y l)prop-2-en-1-one (2k) Yield, 86%; yellow solid. mp 196-198°C. IR (KBr, cm⁻¹): 1649 (C=O), 1596 (C=C), 1565 (C=N of quinoline ring). ¹H-NMR (CDCl₃) δ : 2.76 $(3H, s, Me), 7.36$ (1H, d, H₄', $J = 3.8$ Hz), 7.38 (1H, d, H_α, $J = 15.6$ Hz), 7.47 (1H, t, H₆, $J = 7.6$ Hz), 7.51 (1H, d, H_3 ['], $J = 3.9$ Hz), 7.61 (1H, d, H₇, $J = 7.0$ Hz), 7.70 (1H, d, H₅, $J = 8.1$ Hz), 8.23 (1H, d, H_{β}, $J = 15.6$ Hz), 8.41 (1H, s, H₄). MS (m/z) : 439 (M⁺, 1%), 82 (M⁺ $-C_{13}H_9NOICl$, 100%). Anal. Calcd for $C_{17}H_{11}ClINOS$: C, 46.44; H, 2.52; N, 3.19. Found: C, 46.39; H, 2.42; N, 3.13.

General procedure for the preparation of 2-chloro-3-[3- (aryl)-4,5-dihydro-1H-pyrazol-5-yl]-7/8-methylquinoline $(3a-k)$ and $(4a-k)$

A mixture of chalcone (1a–k or 2a–k, 1.0 mmol) and hydrazine hydrate (3.0 mmol) in ethanol (10 ml) was refluxed. The crude product got precipitated within 8–15 min which was poured onto ice-cold water (50 ml). The precipitates were collected by filtration, washed with cold water followed by cold EtOH to obtain 2-pyrazolines which were recrystallised from EtOH (95%) to obtain pure compounds (3a–k and 4a–k).

2-Chloro-7-methyl-3-(3-thiophen-3-yl-4,5-dihydro-1H-pyr $a zol-5-yl$)quinoline (3a) Yield, 80%; colourless solid. mp 180-181°C. IR (KBr, cm⁻¹): 3275 (NH), 1596 (C=N of pyrazoline ring), 1555 (C=N of quinoline ring). 1 H-NMR (CDCl₃) δ : 2.50 (3H, s, CH₃), 2.95 (1H, dd, $J = 16.3$, 9.4 Hz, 4-H_a), 3.74 (1H, dd, $J = 16.4$, 10.6 Hz, 4-H_b), 5.38 (1H, t, $J = 9.9$ Hz, 5-H), 7.31 (1H, dd, H₄', $J = 5.0$ Hz, 2.8 Hz), 7.40 (1H, d, H₅, $J = 8.1$ Hz), 7.62 (1H, dd, H₅', $J = 4.9$ Hz, 0.9 Hz), 7.73 (1H, d, H₆, $J = 8.3$ Hz), 7.75

 $(1H, s, H_8)$, 8.08 $(1H, dd, H_2', J = 2.8 Hz, 1.0 Hz)$, 8.40 (1H, s, H₄). MS (m/z): 327 (M⁺, 78.53%). Anal. Calcd for $C_{17}H_{14}CIN_3S$: C, 62.28; H, 4.30; N, 12.82. Found: C, 62.24; H, 4.25; N, 12.80.

2-Chloro-7-methyl-3-[3-(3-methylthiophen-2-yl)-4,5-dihydro-1H-pyrazol-5-yl]quinoline (3b) Yield, 78%; yellow solid. mp 160–161°C. IR (KBr, cm⁻¹): 3285 (NH), 1602 (C=N of pyrazoline ring), 1559 (C=N of quinoline ring). 1 H-NMR (CDCl₃) δ : 2.53–2.62 (s, 2× CH₃), 2.84 (1H, dd, J = 16.3, 9.4 Hz, 4-H_a), 3.64 (1H, dd, $J = 16.3$, 10.4 Hz, 4-H_b), 5.29 (1H, t, $J = 9.9$ Hz, 5-H), 6.82 (1H, d, H₄', $J = 5.0$ Hz), 7.35 (1H, d, H₅, $J = 8.1$ Hz), 7.35 (1H, d, H₅', $J = 4.9$ Hz), 7.74 (1H, d, H₆, $J = 8.5$ Hz), 7.76 (1H, s, H₈), 8.39 (1H, s, H₄). MS (m/z): 341 (M⁺, 84.01%). Anal. Calcd. for $C_{18}H_{16}CIN_3S$: C, 63.24; H, 4.72; N, 12.29. Found: C, 63.14; H, 4.65; N, 12.29.

2-Chloro-7-methyl-3-[3-(4-methylthiophen-2-yl)-4,5-dihydro- $1H$ -pyrazol-5-yl]quinoline $(3c)$ Yield, 83%; colourless solid. mp 200°C. IR (KBr, cm⁻¹): 3280 (NH), 1599 (C=N of pyrazoline ring), 1555 (C=N of quinoline ring). 1 H-NMR (CDCl₃) δ : 2.30–2.50 (s, 2× CH₃), 2.85 (1H, dd, $J = 16.3, 9.3$ Hz, 4-H_a), 3.66 (1H, dd, $J = 16.3, 10.5$ Hz, $4-H_b$), 5.30 (1H, t, $J = 9.9$ Hz, 5-H), 7.11 (1H, s, $H₅'$), 7.37 $(1H, d, H_5, J = 8.1 \text{ Hz})$, 7.51 $(1H, s, H_3')$, 7.74 $(1H, d, H_6,$ $J = 8.3$ Hz), 7.76 (1H, s, H₈), 8.41 (1H, s, H₄). MS (m/z): 341 (M^+ , 84.02%). Anal. Calcd. for C₁₈H₁₆ClN₃S: C, 63.24; H, 4.72; N, 12.29. Found: C, 63.21; H, 4.69; N, 12.25.

2-Chloro-7-methyl-3-[3-(5-methylthiophen-2-yl)-4,5-dihydro-1H-pyrazol-5-yl]quinoline (3d) Yield, 87%; pale yellow solid. mp 198°C. IR (KBr, cm⁻¹): 3275 (NH), 1595 (C=N of pyrazoline ring), 1558 (C=N of quinoline ring). ¹H-NMR (CDCl₃) δ : 2.52–2.54 (s, 2× CH₃), 2.85 (1H, dd, $J = 16.3, 9.3$ Hz, 4-H_a), 3.66 (1H, dd, $J = 16.3, 10.5$ Hz, 4-H_b), 5.31 (1H, t, $J = 9.9$ Hz, 5-H), 6.66 (1H, d, H₄', $J = 3.2$ Hz), 6.85 (1H, d, H₃', $J = 3.5$ Hz), 7.37 (1H, d, H_5 , $J = 8.1$ Hz), 7.74 (1H, d, H₆, $J = 8.3$ Hz), 7.76 (1H, s, H₈), 8.40 (1H, s, H₄). MS (m/z): 341 (M⁺, 84.04%). Anal. Calcd. for $C_{18}H_{16}CIN_3S$: C, 63.24; H, 4.72; N, 12.29. Found: C, 63.23; H, 4.69; N, 12.31.

2-Chloro-3-[3-(2,5-dimethylthiophen-3-yl)-4,5-dihydro-1Hpyrazol-5-yl]-7-methylquinoline (3e) Yield, 86%; offwhite solid. mp 116-117°C. IR (KBr, cm⁻¹): 3279 (NH), 1610 (C=N of pyrazoline ring), 1556 (C=N of quinoline ring). ¹H-NMR (CDCl₃) δ : 2.36–2.65 (s, 3× CH₃), 2.82 (1H, dd, $J = 16.2$, 9.7 Hz, 4-H_a), 3.65 (1H, dd, $J = 16.3$, 10.5 Hz, 4-H_b), 5.28 (1H, t, $J = 10.0$ Hz, 5-H), 6.86 (1H, s, H₄'), 7.41 (1H, d, H₅, $J = 8.3$ Hz), 7.72 (1H, d, H₆,

 $J = 8.3$ Hz), 7.76 (1H, s, H₈), 8.34 (1H, s, H₄). MS (m/z): 355 (M⁺, 100%). Anal. Calcd. for C₁₉H₁₈ClN₃S: C, 64.12; H, 5.10; N, 11.81. Found: C, 64.08; H, 5.05; N, 11.76.

2-Chloro-3-[3-(3-chlorothiophen-2-yl)-4,5-dihydro-1Hpyrazol-5-yl]-7-methylquinoline (3f) Yield, 79%; colourless solid. mp 166167°C. IR (KBr, cm⁻¹): 3288 (NH), 1608 (C=N of pyrazoline ring), 1560 (C=N of quinoline ring). ¹H-NMR (CDCl₃) δ : 2.52 (3H, s, CH₃), 3.15 (1H, dd, $J = 16.9, 10.1$ Hz, 4-H_a), 4.00 (1H, dd, $J = 16.9, 10.8$ Hz, 4-H_b), 5.39 (1H, t, $J = 10.4$ Hz, 5-H), 6.86 (1H, d, H₄', $J = 5.2$ Hz), 7.21 (1H, d, H₅, $J = 8.3$ Hz), 7.32 (1H, d, H_5 ['], $J = 5.2$ Hz), 7.76 (1H, d, H_6 , $J = 8.6$ Hz), 7.89 (1H, s, H₈), 8.37 (1H, s, H₄). MS (m/z): 361 (M⁺, 96.20%) 185 $(M^{\dagger}$ –C₁₀H₇NCl, 100%). Anal. Calcd. for C₁₇H₁₃Cl₂N₃S: C, 56.36; H, 3.62; N, 11.60. Found: C, 56.32; H, 3.61; N, 11.58.

2-Chloro-3-[3-(5-chlorothiophen-2-yl)-4,5-dihydro-1H-pyr $azol-5-yl$]-7-methylquinoline $(3g)$ Yield, 83%; off-white solid. mp 205-207°C. IR (KBr, cm⁻¹): 3284 (NH), 1605 (C=N of pyrazoline ring), 1560 (C=N of quinoline ring). ¹H-NMR (CDCl₃) δ : 2.50 (3H, s, CH₃), 2.87 (1H, dd, $J = 16.2, 10.0$ Hz, 4-H_a), 3.71 (1H, dd, $J = 16.2, 10.7$ Hz, 4-H_b), 5.36 (1H, t, $J = 10.3$ Hz, 5-H), 6.82 (1H, d, H₄', $J = 4.3$ Hz), 7.28 (1H, d, H₅, $J = 8.3$ Hz), 7.47 (1H, d, H_3 ['], $J = 4.3$ Hz), 7.65 (1H, d, H_6 , $J = 8.3$ Hz), 7.69 (1H, s, H₈), 8.34 (1H, s, H₄). MS (m/z): 361 (M⁺, 100%). Anal. Calcd for $C_{17}H_{13}Cl_2N_3S$: C, 56.36; H, 3.62; N, 11.60. Found: C, 56.34 H, 3.59; N, 11.52.

2-Chloro-3-[3-(2,5-dichlorothiophen-3-yl)-4,5-dihydro-1Hpyrazol-5-yl]-7-methylquinoline (3h) Yield, 75%; offwhite solid. mp 178-179°C. IR (KBr, cm⁻¹): 3282 (NH), 1612 (C=N of pyrazoline ring), 1561 (C=N of quinoline ring). ¹H-NMR (CDCl₃) δ : 2.52 (3H, s, CH₃), 3.11 (1H, dd, $J = 16.7, 9.9$ Hz, 4-H_a), 3.98 (1H, dd, $J = 16.7, 10.5$ Hz, 4-H_b), 5.39 (1H, t, $J = 10.2$ Hz, 5-H), 6.97 (1H, s, H₄'), 7.35 (1H, d, H₅, $J = 8.3$ Hz), 7.72 (1H, d, H₆, $J = 8.2$ Hz), 7.75 (1H, s, H₈), 8.33 (1H, s, H₄). MS (m/z): 397 (M⁺, 100%). Anal. Calcd. for $C_{17}H_{12}Cl_3N_3S$: C, 51.47; H, 3.05; N, 10.59. Found: C, 51.45; H, 3.02; N, 10.54.

3-[3-(3-Bromothiophen-2-yl)-4,5-dihydro-1H-pyrazol-5-yl]- 2-chloro-7-methylquinoline (3i) Yield, 92%; colourless solid. mp 170-171°C; IR (KBr, cm⁻¹): 3279 (NH), 1608 (C=N of pyrazoline ring), 1556 (C=N of quinoline ring). ¹H-NMR (CDCl₃) δ : 2.53 (3H, s, CH₃), 3.19 (1H, dd, $J = 16.8$, 10.1 Hz, 4-H_a), 4.06 (1H, dd, $J = 16.8$, 10.8 Hz, 4-H_b), 5.39 (1H, t, $J = 10.4$ Hz, 5-H), 6.95 (1H, d, H₄', $J = 5.5$ Hz), 7.35 (1H, d, H₅, $J = 8.2$ Hz), 7.21 (1H, d, H_5 ['], $J = 5.5$ Hz), 7.72 (1H, d, H_6 , $J = 8.6$ Hz), 7.74 (1H,

s, H₈), 8.38 (1H, s, H₄). MS (m/z): 407 (M⁺, 100%). Anal. Calcd. for $C_{17}H_{13}BrClN_3S$: C, 50.20; H, 3.22; N, 10.33. Found: C, 50.12; H, 3.14; N, 10.30.

3-[3-(5-Bromothiophen-2-yl)-4,5-dihydro-1H-pyrazol-5-yl]- 2-chloro-7-methylquinoline (3j) Yield, 76%; colourless solid. mp 195°C. IR (KBr, cm⁻¹): 3282 (NH), 1600 (C=N of pyrazoline ring), 1552 (C=N of quinoline ring). 1 H-NMR (CDCl₃) δ : 2.52 (3H, s, CH₃), 2.88 (1H, dd, $J = 16.2, 9.9$ Hz, 4-H_a), 3.71 (1H, dd, $J = 16.2, 10.7$ Hz, 4-H_b), 5.36 (1H, t, $J = 10.3$ Hz, 5-H), 6.86 (1H, d, H₄', $J = 4.2$ Hz), 7.36 (1H, d, H₅, $J = 8.3$ Hz), 7.43 (1H, d, H_3' , $J = 4.2$ Hz), 7.74 (1H, d, H_6 , $J = 8.3$ Hz), 7.75 (1H, s, H₈), 8.35 (1H, s, H₄). MS (m/z): 407 (M⁺, 100%). Anal. Calcd. for $C_{17}H_{13}BrClN_3S$: C, 50.20; H, 3.22; N, 10.33. Found: C, 50.15; H, 3.19; N, 10.25.

2-Chloro-3-[3-(5-iodothiophen-2-yl)-4,5-dihydro-1H-pyrazol-5-yl]-7-methylquinoline $(3k)$ Yield, 85%; colourless solid. mp 212°C. IR (KBr, cm⁻¹): 3281 (NH), 1610 (C=N of pyrazoline ring), 1550 (C=N of quinoline ring). 1 H-NMR (CDCl₃) δ : 2.52 (3H, s, CH₃), 2.87 (1H, dd, $J = 16.2$, 9.9 Hz, 4-H_a), 3.71 (1H, dd, $J = 16.2$, 10.7 Hz, 4-H_b), 5.36 $(H, t, J = 10.3 \text{ Hz}, 5\text{-H}), 6.71 (1H, d, H₄', J = 4.2 \text{ Hz}),$ 7.15 (1H, d, H₃', $J = 4.2$ Hz), 7.34 (1H, d, H₅, $J = 8.4$ Hz), 7.74 (1H, d, H₆, $J = 8.3$ Hz), 7.75 (1H, s, H₈), 8.35 (1H, s, H₄). MS (m/z): 453 (M⁺, 100%). Anal. Calcd. for $C_{17}H_{13}ClIN_3S$: C, 45.00; H, 2.89; N, 9.26. Found: C, 44.95; H, 2.85; N, 9.23.

2-Chloro-8-methyl-3-(3-thiophen-3-yl-4,5-dihydro-1H-pyr $a z o l$ -5-yl)quinoline (4a) Yield, 72%; colourless solid. mp 195–196°C. IR (KBr, cm⁻¹): 3274 (NH), 1595 (C=N of pyrazoline ring), 1550 (C=N of quinoline ring). 1 H-NMR (CDCl₃) δ : 2.70 (3H, s, CH₃), 2.97 (1H, dd, $J = 16.4$, 9.4 Hz, 4-H_a), 3.75 (1H, dd, $J = 16.4$, 10.7 Hz, 4-H_b), 5.39 (1H, t, $J = 9.9$ Hz, 5-H), 7.32 (1H, dd, H₄', $J = 5.0$ Hz, 2.8 Hz), 7.45 (1H, t, H₆, $J = 7.6$ Hz), 7.55 (1H, d, H₇, $J = 7.0$ Hz), 7.60 (1H, d, H₅', $J = 4.6$ Hz, 1.0 Hz), 7.67 (1H, d, H₅, $J = 6.6$ Hz), 8.08 (1H, dd, H₂', $J = 2.7$ Hz, 1.0 Hz), 8.39 (1H, s, H₄). MS (m/z) : 327 (M⁺, 70.30%). Anal. Calcd. for $C_{17}H_{14}CIN_3S$: C, 62.28; H, 4.30; N, 12.82. Found: C, 62.22; H, 4.22; N, 12.78.

2-Chloro-8-methyl-3-[3-(3-methylthiophen-2-yl)-4,5-dihydro- $1H$ -pyrazol-5-yl]quinoline (4b) Yield, 67%; colourless solid. mp 130–131°C. IR (KBr, cm⁻¹): 3277 (NH), 1605 (C=N of pyrazoline ring), 1552 (C=N of quinoline ring). ¹H-NMR (CDCl₃) δ : 2.63–2.73 (s, 2× CH₃), 2.85 (1H, dd, $J = 16.3$, 9.4 Hz, 4-H_a), 3.66 (1H, dd, $J = 16.3$, 10.5 Hz, 4-H_b), 5.31 (1H, t, $J = 9.9$ Hz, 5-H), 6.82 (1H, d, H₄', $J = 5.1$ Hz), 7.45 (1H, t, H₆, $J = 7.6$ Hz), 7.33 (1H, d, H₅',

 $J = 5.1$ Hz), 7.54 (1H, d, H₇, $J = 7.0$ Hz), 7.69 (1H, d, H₅, $J = 7.9$ Hz), 8.38 (1H, s, H₄). MS (m/z): 341 (M⁺, 59.47%). Anal. Calcd. for C₁₈H₁₆ClN₃S: C, 63.24; H, 4.72; N, 12.29. Found: C, 63.22; H, 4.75; N, 12.23.

2-Chloro-8-methyl-3-[3-(4-methylthiophen-2-yl)-4,5-dihydro-1H-pyrazol-5-yl]quinoline $(4c)$ Yield, 71%; colourless solid. mp 182°C. IR (KBr, cm⁻¹): 3281 (NH), 1595 (C=N of pyrazoline ring), 1555 (C=N of quinoline ring). ¹H-NMR (CDCl₃) δ : 2.30–2.72 (s, 2× CH₃), 2.87 (1H, dd, $J = 16.3, 9.3$ Hz, 4-H_a), 3.69 (1H, dd, $J = 16.3, 10.5$ Hz, 4-H_b), 5.31 (1H, t, $J = 9.9$ Hz, 5-H) 7.11 (1H, s, H₅'), 7.44 $(H, t, H_6, J = 7.5 \text{ Hz})$, 7.55 (1H, d, H₇, $J = 7.0 \text{ Hz}$), 7.69 $(1H, d, H_5, J = 7.0 \text{ Hz}), 7.52 (1H, s, H_3'), 8.37 (1H, s, H_4).$ MS (m/z) : 341 (M⁺, 60.02%). Anal. Calcd. for C₁₈H₁₆ ClN3S: C, 63.24; H, 4.72; N, 12.29. Found: C, 63.20; H, 4.69; N, 12.25.

2-Chloro-8-methyl-3-[3-(5-methylthiophen-2-yl)-4,5-dihydro-1H-pyrazol-5-yl]quinoline (4d) Yield, 80%; pale yellow solid. mp 209-210°C. IR (KBr, cm⁻¹): 3278 (NH), 1592 (C=N of pyrazoline ring), 1550 (C=N of quinoline ring). ¹H-NMR (CDCl₃) δ : 2.53–2.72 (s, 2× CH₃), 2.86 (1H, dd, $J = 16.3, 9.3$ Hz, 4-H_a), 3.67 (1H, dd, $J = 16.3, 10.5$ Hz, 4-H_b), 5.31 (1H, t, $J = 9.9$ Hz, 5-H), 6.66 (1H, d, H₄', $J = 3.0$ Hz), 6.86 (1H, d, H₃' $J = 3.4$ Hz), 7.44 (1H, t, H₆, $J = 7.6$ Hz), 7.54 (1H, d, H₇, $J = 7.0$ Hz), 7.62 (1H, d, H₅, $J = 8.1$ Hz), 8.33 (1H, s, H₄). MS (m/z): 341 (M⁺, 59.81%). Anal. Calcd. for C₁₈H₁₆ClN₃S: C, 63.24; H, 4.72; N, 12.29. Found: C, 63.23; H, 4.70; N, 12.27.

2-Chloro-3-[3-(2,5-dimethylthiophen-3-yl)-4,5-dihydro-1Hpyrazol-5-yl]-8-methylquinoline (4e) Yield, 88%; brown solid. mp 126-127°C. IR (KBr, cm⁻¹): 3282 (NH), 1609 (C=N of pyrazoline ring), 1553 (C=N of quinoline ring). ¹H-NMR (CDCl₃) δ : 2.42–2.73 (s, 3× Me), 2.84 (1H, dd, $J = 16.3, 9.7$ Hz, 4-H_a), 3.68 (1H, dd, $J = 16.3, 10.6$ Hz, 4-H_b), 5.30 (1H, t, $J = 10.0$ Hz, 5-H), 6.86 (1H, s, H₄'), 7.42 (1H, t, H_6 , $J = 7.7$ Hz), 7.54 (1H, d, H_7 , $J = 7.0$ Hz), 7.68 (1H, d, H₅, $J = 8.1$ Hz), 8.34 (1H, s, H₄). MS (m/z): 355 (M⁺, 100%). Anal. Calcd. for C₁₉H₁₈ClN₃S: C, 64.12; H, 5.10; N, 11.81. Found: C, 64.10; H, 5.08; N, 11.79.

2-Chloro-3-[3-(3-chlorothiophen-2-yl)-4,5-dihydro-1H-pyrazol-5-yl]-8-methylquinoline (4f) Yield, 81%; yellowish brown solid. mp 152°C. IR (KBr, cm⁻¹): 3277 (NH), 1610 (C=N of pyrazoline ring), 1559 (C=N of quinoline ring). ¹H-NMR (CDCl₃) δ : 2.72 (3H, s, CH₃), 3.16 (1H, dd, $J = 16.9, 10.2$ Hz, 4-H_a), 4.01 (1H, dd, $J = 16.9, 10.8$ Hz, 4-H_b), 5.39 (1H, t, $J = 10.4$ Hz, 5-H), 6.87 (1H, d, H₄', $J = 5.4$ Hz), 7.41 (1H, t, H₆, $J = 7.6$ Hz), 7.54 (1H, d, H₇, $J = 7.0$ Hz), 7.32 (1H, d, H₅', $J = 5.4$ Hz), 7.65 (1H, d,

H₅, $J = 8.1$ Hz), 8.37 (1H, s, H₄). MS (m/z): 361 (M⁺, 100%). Anal. Calcd. for C₁₇H₁₃Cl₂N₃S: C, 56.36; H, 3.62; N, 11.60. Found: C, 56.34; H, 3.58; N, 11.54.

2-Chloro-3-[3-(5-chlorothiophen-2-yl)-4,5-dihydro-1H-pyrazol-5-yl]-8-methylquinoline $(4g)$ Yield, 80%; pale yellow solid. mp 230–232°C. IR (KBr, cm^{-1}): 3285 (NH), 1603 (C=N of pyrazoline ring), 1560 (C=N of quinoline ring). ¹H-NMR (CDCl₃) δ : 2.72 (3H, s, CH₃), 2.89 (1H, dd, $J = 16.3, 10.1$ Hz, 4-H_a), 3.71 (1H, dd, $J = 16.3, 10.7$ Hz, 4-H_b), 5.39 (1H, t, $J = 10.3$ Hz, 5-H), 6.85 (1H, d, H₄', $J = 4.2$ Hz), 7.42 (1H, t, H₆, $J = 7.6$ Hz), 7.56 (1H, d, H₇, $J = 7.0$ Hz), 7.48 (1H, d, H₃', $J = 4.4$ Hz), 7.63 (1H, d, H₅, $J = 8.0$ Hz), 8.35 (1H, s, H₄). MS (m/z): 361 (M⁺, 100%). Anal. Calcd. for $C_{17}H_{13}Cl_2N_3S$: C, 56.36; H, 3.62; N, 11.60. Found: C, 56.31; H, 3.56; N, 11.55.

2-Chloro-3-[3-(2,5-dichlorothiophen-3-yl)-4,5-dihydro-1Hpyrazol-5-yl]-8-methylquinoline (4h) Yield, 75%; yellowish brown solid. mp 153°C. IR (KBr, cm^{-1}): 3280 (NH), 1615 (C=N of pyrazoline ring), 1560 (C=N of quinoline ring). ¹H-NMR (CDCl₃) δ : 2.72 (3H, s, CH₃), 3.15 (1H, dd, $J = 16.8$, 10.0 Hz, 4-H_a), 4.00 (1H, dd, $J = 16.8, 10.5$ Hz, 4-H_b), 5.40 (1H, t, $J = 10.3$ Hz, 5-H), 6.96 (1H, s, H₄'), 7.41 (1H, t, H₆, $J = 7.6$ Hz), 7.56 (1H, d, $H_7, J = 6.8$ Hz), 7.62 (1H, d, H₅, $J = 8.1$ Hz), 8.35 (1H, s, H₄). MS (m/z) : 397 $(M^+, 100\%)$. Anal. Calcd. for $C_{17}H_{12}Cl_3N_3S$: C, 51.47; H, 3.05; N, 10.59. Found: C, 51.41; H, 3.00; N, 10.56.

3-[3-(3-Bromothiophen-2-yl)-4,5-dihydro-1H-pyrazol-5-yl]- 2-chloro-8-methylquinoline $(4i)$ Yield, 77%; pale yellow solid. mp 166–168°C. IR (KBr, cm⁻¹): 3279 (NH), 1607 $(C=N \text{ of pyrazoline ring})$, 1555 $(C=N \text{ of quinoline ring})$. ¹H-NMR (CDCl₃) δ : 2.72 (3H, s, CH₃), 3.21 (1H, dd, $J = 16.9, 10.2$ Hz, 4-H_a), 4.10 (1H, dd, $J = 16.8, 10.8$ Hz, 4-H_b), 5.40 (1H, t, $J = 10.4$ Hz, 5-H), 6.86 (1H, d, H₄', $J = 5.5$ Hz), 7.43 (1H, t, H₆, $J = 7.6$ Hz), 7.21 (1H, d, H₅', $J = 5.5$ Hz), 7.54 (1H, d, H₇, $J = 7.1$ Hz), 7.65 (1H, d, H₅, $J = 8.0$ Hz), 8.39 (1H, s, H₄). MS (m/z): 407 (M⁺, 100%). Anal. Calcd. for $C_{17}H_{13}BrClN_3S$: C, 50.20; H, 3.22; N, 10.33. Found: C, 50.14; H, 3.18; N, 10.29.

3-[3-(5-Bromothiophen-2-yl)-4,5-dihydro-1H-pyrazol-5-yl]- 2-chloro-8-methylquinoline $(4j)$ Yield, 80%; pale yellow solid. mp 215-216°C. IR (KBr, cm⁻¹): 3282 (NH), 1597 (C=N of pyrazoline ring), 1552 (C=N of quinoline ring). ¹H-NMR (CDCl₃) δ : 2.72 (3H, s, CH₃), 2.89 (1H, dd, $J = 16.2, 9.9$ Hz, 4-H_a), 3.71 (1H, dd, $J = 16.2, 10.6$ Hz, 4-H_b), 5.36 (1H, t, $J = 10.3$ Hz, 5-H), 6.87 (1H, d, H₄', $J = 4.2$ Hz), 7.42 (1H, t, H₆, $J = 7.6$ Hz), 7.55 (1H, d, H₇, $J = 7.1$ Hz), 7.43 (1H, d, H₃', $J = 4.2$ Hz), 7.63 (1H, d, H₅, $J = 8.0$ Hz), 8.36 (1H, s, H₄). MS (m/z): 407 (M⁺, 100%). Anal. Calcd. for $C_{17}H_{13}BrCN_3S$: C, 50.20; H, 3.22; N, 10.33. Found: C, 50.19 H, 3.17; N, 10.28.

2-Chloro-3-[3-(5-iodothiophen-2-yl)-4,5-dihydro-1H-pyrazol-5-yl]-8-methylquinoline (4k) Yield, 82%; pale yellow solid. mp 178°C. IR (KBr, cm⁻¹): 3280 (NH), 1605 (C=N of pyrazoline ring), 1550 (C=N of quinoline ring). 1 H-NMR (CDCl₃) δ : 2.72 (3H, s, CH₃), 2.90 (1H, dd, $J = 16.3, 9.9$ Hz, 4-H_a), 3.73 (1H, dd, $J = 16.3, 10.6$ Hz, 4-H_b), 5.37 (1H, t, $J = 10.2$ Hz, 5-H), 6.71 (1H, d, H₄', $J = 4.0$ Hz), 7.42 (1H, t, H₆, $J = 7.6$ Hz), 7.16 (1H, d, H₃', $J = 4.1$ Hz), 7.56 (1H, d, H₇, $J = 7.0$ Hz), 7.63 (1H, d, H₅, $J = 8.1$ Hz), 8.36 (1H, s, H₄). MS (m/z): 453 (M⁺, 100%). Anal. Calcd. for C₁₇H₁₃ClIN₃S: C, 45.00; H, 2.89; N, 9.26. Found: C, 44.98; H, 2.81; N, 9.25.

In vitro antileishmanial assay

The title compounds $(1a-k$ and $2a-k$ and $(3a-k)$ and 4a–k) were tested for the antileishmanial activity using L. major promastigotes as parasites for in vitro screening. Parasites were cultured at 24° C in shaking incubator on M 199 medium containing foetal bovine serum (10%); HEPES (25 mM), and penicillin and streptomycin (0.22 μ g each) (Ali *et al.*, [1997](#page-10-0)).

Each compound (1 mg) was dissolved in DMSO (1 ml) and Amphotericin B (1 mg) taken in DMSO (1 ml) was used as a positive control. Parasites were taken from lag phase of their growth and were centrifuged at 3000 rpm for 3 min. The parasite density was maintained at 2×10^6 cells/ml by diluting with fresh culture medium. In 96-well plates, 180 µl of medium was added in different wells. The experimental compound $(20 \mu l)$ was added in medium and serially diluted. Parasite culture $(100 \mu l)$ was added in all wells. In negative controls, DMSO was serially diluted in medium; while the positive control, contained varying concentrations of standard antileishmanial compound, i.e. Amphotercin B. The plates were incubated for 72 h at 24° C. The culture was examined microscopically on an improved neubaur counting chamber and IC_{50} values of compounds possessing antileishmanial activity were calculated. All assays were run in duplicate. The results are summarized in Table [2.](#page-3-0) IC_{50} values of compounds were determined using prism windows-based software.

After running the samples, % of inhibition is calculated in serial dilution methods. It depends on the activity of the compounds; some of them show in 4-6-7 or 10 dilution the inhibitory concentration. Subsequently, we count the number of parasite in neubauer chamber (0.0025 mm^2) and implement the result manually in the prism windows-based software.

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