

Research article

Production of wax by virgin queens of the stingless bee *Melipona bicolor* (Apidae, Meliponinae)

D. Koedam¹, H. Jungnickel², J. Tentschert², G.R. Jones² and E.D. Morgan^{2,*}

¹ Laboratório de Abelhas, Instituto de Biociências, Universidade de São Paulo, Rua de Matão, Traversa 14, 05508-900, São Paulo, SP, Brasil,

² Chemical Ecology group, Lennard-Jones Laboratory, Keele University, Staffordshire ST5 5BG, England, e-mail: e.d.morgan@keele.ac.uk

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Summary. Young virgin queens of the stingless bee *Melipona bicolor* have been shown to produce wax like workers. The small, white to transparent flakes of wax of the queens protrude from the intersegmental space and cover the anterior part of the tergite cuticle, in a way similar to that in workers. This points to the presence of wax glands in the queens. However, workers produce wax from glands located at the fourth to seventh tergites whereas queens secrete wax from the epidermal gland at the third tergite only. Analysis of the queen-produced wax showed that it contains the same substances as the worker-produced wax with minor differences in composition. The wax consists chiefly of the long-chain esters triacontanyl acetate and octacosanyl acetate, smaller amounts of linear C₂₁ to C₃₁ alkanes and alkenes, and still smaller quantities of linear aldehydes and isobutyrate esters. Analysis of wax from wax deposits and wax constructs showed the same composition. Wax from *M. bicolor* is similar to that of other stingless bees in containing the range of linear long-chain alkenes and alkanes and different from that of *Apis* bees which contains a more complex mixture, less hydrocarbons and more long chain esters.

Key words: *Melipona*, wax production, virgin queens, long-chain acetates, hydrocarbons.

Introduction

Melipona bicolor is a stingless bee occurring in the more southern parts of the Brazilian Atlantic rain forest. It is unusual in that its colonies can have more than one physogastric queen active in egg laying (Bego, 1989). Workers in stingless bees are known to produce wax with epidermal

glands located at the fourth to seventh tergites (Cruz-Landim, 1967). Wax is produced by workers in *M. bicolor* predominantly at about 6–30 days of age (Bego, 1983). The workers first gather it at specific places in the nest, so-called wax deposits (Hebling et al., 1964), and then wax nest structures are made using the wax from these deposits, mixing it with propolis, or plant resin, to increase its rigidity.

There are a few recorded cases of apparent wax production in stingless bees by individuals other than workers. Da Silva (1977; reviewed by Kerr, 1997) reported the production by males, there is a notice about wax secretion by a virgin queen of *Trigona* (*Scaptotrigona*) *bipunctata* (Kerr and de Lello, 1962), and in the case of *Melipona marginata*, wax glands were encountered in males as well as in queens (Cruz-Landim, 1967). In none of these cases was the original wax ever collected nor was the secreted substance tested for its characteristics. We have occasionally seen virgin queens of *M. bicolor* secreting small amounts of wax-like structures. We have collected and examined wax produced by the two castes of *Melipona bicolor* by picking the secreted scales directly from the abdomen of living individuals. We also show that the queen-produced wax and worker wax have essentially the same composition, and that the wax deposit and involucre wax of the nest are very similar with the addition of propolis.

Materials and methods

Wax scales were collected directly from the abdomen of workers and virgin queens of *M. bicolor* Lepelletier by carefully peeling them off the cuticle with the help of an insect pin. Since individual virgin queens always produced only very tiny bits of wax, five queens were used for the collection of the wax scales. These were immediately placed in soft glass capillaries and sealed as described by Morgan (1990). The wax scales of single workers were placed in several capillaries using in total three workers. About 100 mg of wax from wax deposits and a similar quantity of wax constructs were collected in small plastic vials (Low Temperature Freezer Vials; 1.2 ml; VWR, Scientific Products).

* Author for correspondence.

The sealed capillaries and plastic vials were transported to Keele where they were analysed by gas chromatography using the solid-sampling method described by Morgan (1990). Chromatography was carried out with a Hewlett-Packard 5890 gas chromatograph directly coupled to a 5970B Mass Selective Detector (quadrupole mass spectrometer using 70 eV electron impact ionisation). The system was controlled and data accumulated on a Hewlett-Packard series 300 computer with HP 5970C Chemstation. Mass spectra were scanned from m/z 35 to m/z 550. Scan time was about 2.4 sec. Chromatography was performed on polydimethylsiloxane phase in a fused silica column (15 m \times 0.2 mm, 0.33 μ m film thickness) (SGE, Milton Keynes, UK). Helium was used as carrier gas at 1 ml/min. The sample in the glass capillary was heated in the injection port (temp 250°C) for 2 min before crushing and starting the chromatography. The oven was programmed from 60°C (3 min) at 5°C/min to 320°C. The split valve was closed before crushing the sample and reopened 30 s later. Deposit wax and construct wax were analysed both by solid injection and dissolved in hexane for injection (1 μ l) in the usual way. The identification of compounds was confirmed by comparison of their mass spectra and retention times with those of standard alkanes and using the Wiley-NIST database, version 1994.

Double-bond positions in alkenes were determined in a sample of wax by the dimethyldisulphide method (Billen et al., 1986), except that the reaction was carried out at 46°C for 12 hours under an atmosphere of argon. The dimethyldisulphide adducts were identified in the gas chromatogram of the product mixture and the positions of attachment were deduced from the fragmentation pattern.

Transesterification experiments were performed with boron trifluoride-methanol complex and sodium methoxide in methanol. Portions of wax from the involucrum and wax deposits in the nest (100–200 mg) were added to boron trifluoride-methanol (100 ml of 14% BF_3 in methanol) and the mixture heated at 60°C for 1 h. Reaction was stopped by adding water (100 ml). The wax compounds were extracted twice with chloroform (200 ml), which was evaporated to a small volume and used for gas chromatography. Basic transesterification was performed by dissolving sodium (100 mg) in dry methanol (1 ml). When solution was complete, the methanol was cooled and a sample of wax from the involucrum or wax deposit was added. The mixture was then heated at 50°C for 6 h, the reaction was stopped by adding dilute aqueous hydrochloric acid and extracting the wax compounds with chloroform for gas chromatography as above.

Results

Occasionally, young virgin queens of *M. bicolor* were seen walking around the nest having scales of wax on the abdomen. In the queens the positioning of the wax flakes, being small, white to transparent bands protruding from the intersegmental space and covering the anterior part of the tergite cuticle, was similar to that for worker-produced wax. However, virgin queens were seen to produce wax scales from their epidermal glands at the third tergite only whereas workers produce them from the glands at the fourth to the seventh tergites.

Analysis of the white translucent scales of wax by gas chromatography and mass spectrometry showed, as the principal component, the ester triacontanyl acetate, with small amounts of acetates of other long-chain alcohols. There were still smaller amounts of the isobutyrate of the same alcohols, the corresponding aldehydes and a series of linear alkenes and alkanes from C_{20} to C_{31} (Fig. 1). It is clear from the relative retention times that the alkenes are mono-enes. The constituents were the same for both queen and worker wax, but there were differences in the proportions of alkenes to alkanes between them and while the quantity of hydrocarbons peaked at C_{25} for the queen wax, it peaked at C_{23} for the workers.

Formation of the dimethyldisulphide derivatives of the alkenes revealed that they had the double bond either at $\Delta 7$ or $\Delta 9$, with some contribution from $\Delta 8$ (Table 1). Because the proportions of these isomers could not be detected from the gas chromatograms, they were estimated from the single ion chromatograms for the specific ions m/z 145 (characteristic of the derivatives of $\Delta 7$ alkenes) and m/z 159 and 173 (for $\Delta 8$ and $\Delta 9$ alkene derivatives respectively). In the case of the minor components with even-numbered chain length

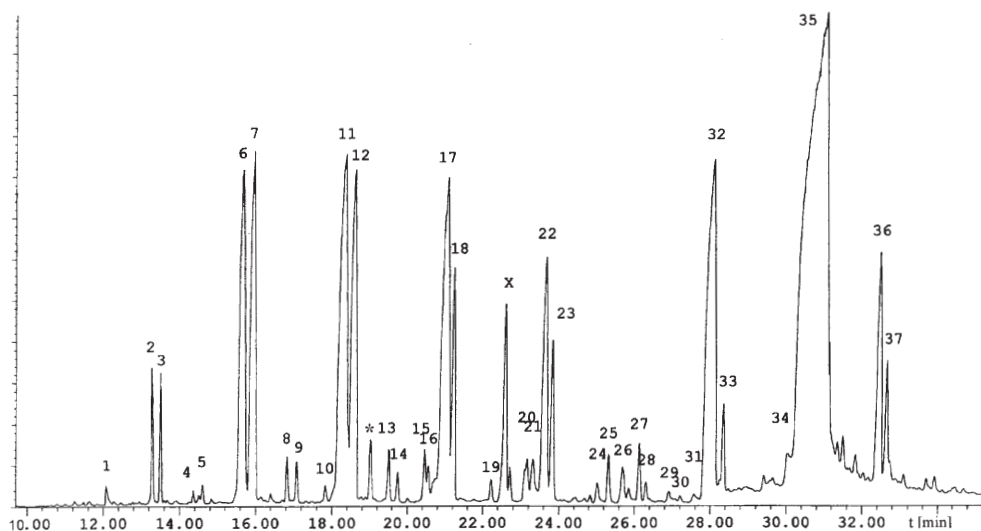


Figure 1. Total ion chromatogram of a tiny flake of wax taken from the third abdominal tergite of a virgin queen of *Melipona bicolor*. Numbered peaks correspond to the numbers in Table 1. The peak marked with * is phthalate plasticizer, a universal contaminant, and that marked X is squalene, a contaminant from human hands

Table 1. Quantification of compounds identified in the wax of queens ($n = 5$) and workers ($n = 3$) of *Melipona bicolor* (ratios in parentheses are the proportions of the positional isomers of the alkenes)

Peak Number in Figure 1	Compound	Queens		Workers	
		Mean (%) \pm SD		Mean (%) \pm SD	
1	Unknown	0.14	0.09	0.11	0.09
2	7-Heneicosene	0.15	0.27	0.28	0.31
3	Heneicosane	0.17	0.24	0.39	0.25
4	7-Docosene	0.05	0.10	0.02	0.03
5	Docosane	0.24	0.46	0.12	0.08
6	7-Tricosene and 8-tricosene (80:1)	0.96	2.10	3.06	2.50
7	Tricosane	1.76	1.90	4.95	1.81
8	7-Tetracosene	0.11	0.17	0.16	0.10
9	Tetracosane	0.20	0.10	0.32	0.06
10	Docosanal	0.14	0.11	0.10	0.07
11	7-Pentacosene, trace of 8-pentacosene and 9-pentacosene (110:1)	2.73	2.97	6.25	4.44
12	Pentacosane	6.67	1.51	6.81	1.21
13	7-Hexacosene	0.07	0.14	0.35	0.07
14	Hexacosane	0.51	0.31	0.38	0.23
15	Tetracosanal	0.36	0.57	0.29	0.17
16	Tetracosanal	0.62	1.36	0.23	0.13
17	7-Heptacosene and 9-heptacosene (2.3:1)	5.82	3.40	7.37	1.77
18	Heptacosane	4.05	2.87	3.01	0.83
19	7-Octacosene	0.18	0.12	0.23	0.11
20	Tetracosanyl acetate	1.31	0.82	0.56	0.23
21	Hexacosanal	0.84	0.81	0.40	0.14
22	7-Nonacosene and 9-nonacosene (2.3:1)	0.70	0.76	0.45	0.18
23	Nonacosane	2.81	2.44	2.70	1.68
24	7- and 9-Triacontene	3.97	2.61	2.48	1.10
25	Triacontane	0.69	1.75	0.19	0.21
26	Tetracosanyl isobutyrate	0.90	1.41	t	–
27	Hexacosanyl acetate	1.19	0.98	0.83	0.35
28	Octacosanal	0.64	0.70	0.34	0.04
29	7-Hentriacontene and 9-hentriacontene (0.8:1)	0.51	0.50	0.33	0.13
30	Hentriacontane	0.47	0.36	0.44	0.39
31	Hexacosyl isobutyrate	0.80	0.49	t	–
32	Octacosanyl acetate	18.08	5.34	12.01	2.23
33	Triacontanal	0.20	0.48	0.95	0.57
34	Octacosanyl isobutyrate	0.57	0.26	0.09	0.05
35	Triacontanyl acetate	38.84	7.68	40.50	10.16
36	Triacontanyl isobutyrate	0.63	1.41	2.24	1.14
37	Dotriacontanyl acetate	1.94	1.22	1.03	0.38

t = trace, less than 0.01 %.

(docosene, tetracosene, etc.) while $\Delta 7$ could be detected for all of them, the weakness of their spectra did not permit ruling out smaller amounts of the other isomers (Table 1).

To check if there were any higher mass esters too involatile for gas chromatography, samples of deposit wax and wax from the involucre were subjected to acidic and basic hydrolysis. In both cases essentially only a spectrum of the hydrocarbons remained. The amounts of long-chain alcohols (tetracosanol to triacontanol) released from their esters were smaller than expected, as these substances were partial-

ly adsorbed on the column during chromatography. Traces of C_{14} to C_{18} methyl esters were seen, but these may have come from surface triglycerides from the bodies of worker bees or from plant substances. No free fatty acids were found in the wax. It was noticed that involucre wax, construct wax from the walls of the hive, whether subjected to hydrolysis or simply making a solution injection showed amounts of terpenes (chiefly monoterpenes and sesquiterpenes) comparable with the amounts of wax hydrocarbons. Deposit wax contained only a little contamination with terpenes. *Trans*- β -caryophyllene was a major component of the terpenes, but this only reflects the plant from which the resin was collected (Patricio et al., 2002).

Discussion

Comparison of chromatograms of the waxes of queens and workers of *M. bicolor* showed the same pattern of composition, the differences were found chiefly in the proportions of the alkenes and alkanes (Table 1). These differences would have very little effect upon the physical properties of the wax. When virgin beeswax scales were compared with wax from wax deposits and involucre wax (i. e. the inner lining of the nest), the pattern was the same, except that the involucre had been strengthened by the addition of propolis, which was evident in the large amount of monoterpenes, sesquiterpenes and even triterpenes found in the chromatogram. The range of hydrocarbons and esters and their proportions were the same as in the pure wax flakes from workers and queens.

It is noteworthy that $\Delta 7$ -alkenes predominated in the shorter carbon chains, and tended towards more of the $\Delta 9$ -alkenes at longer chains. This suggests the shorter carbon chains are built up from palmitoleic acid by addition of 3, 4 or 5 acetate units before decarboxylation and the longer ones from oleic acid by similar chain lengthening (Tillman et al., 1999).

Up to now, the only wax from stingless bees that has been analysed has been from nest material, and that only from trigonine bees (Table 2) (Blomquist et al., 1985; Milborrow et al., 1987). The composition of *Melipona* wax found here is quite different from that of honeybee wax, and much closer to that of the trigonine wax, but still different from that. The wax of the Australian stingless bee *Trigona australis* consisted almost entirely of hydrocarbons (90%) in the range C_{23} to C_{35} , with odd chain lengths predominating (Milborrow et al., 1987). There was only 6% of esters compared with 64% esters in queen wax and 57% esters in pure worker wax found here for *M. bicolor*. The American *Trigona (Trigonisca) buyssoni* and *T. atomaria* waxes were much closer to those we found in *Melipona*. There was 26.8% and 25.5% esters in *T. buyssoni* and *T. atomaria* respectively, with the alcohols in the C_{24} to C_{30} range as found here (Blomquist et al., 1985). Of the 59.3% hydrocarbons in *T. buyssoni*, 90.2% were alkanes, 8.4% alkenes and 1.4% methyl-branched alkanes, while in *T. atomaria* of the 70.9% hydrocarbons, 87.1% were normal alkanes, 12.4% were alkenes and no branched chain hydro-

carbons were found (Blomquist et al., 1985). The hydrocarbons in *T. buyssoni* and *T. atomaria* are mostly saturated (about 90%) whereas in *T. australis* they are in general unsaturated. The hydrocarbons were in the range C₂₁–C₃₇. Our results showed approximately equal quantities of alkanes and alkenes with no evidence of branched chains. Free acids have a chain length between C₁₂–C₃₀+ in *T. buyssoni* and *T. atomaria*, and between C₁₀–C₂₀ in *T. australis* (Table 2). We found no free acids but should note that squalene was also found in the samples we analysed, but as this is an ubiquitous compound, found in human skin, and difficult to avoid transferring to instruments and samples, we discount its presence.

The chemistry of *Apis* wax has already been investigated in detail revealing more than 300 individual components (Tulloch, 1971, 1972, 1980). Hydrocarbons (14%), monoesters (35%), diesters (14%), free acids (12%) and hydroxy polyesters (8%) form the major part (Table 2). Three long-chain esters, with total carbon numbers C₄₀, C₄₆ and C₄₈, and one C₂₄ acid make up over 20% of the total. Because of their high molecular mass, not all these materials are directly analysable by gas chromatography. Analysis of honeybee wax under the same conditions as used here showed it to contain a mixture of C₁₆ to C₂₂ fatty acids, C₂₄ to C₃₂ linear alcohols and C₂₃ to C₂₇ alkanes and alkenes. A summary of an analysis of honeybee wax is given in Table 2. While both honeybee and stingless bee waxes contain high proportions of monoesters and hydrocarbons, the appearance of the chromatograms of the two types, under the same conditions look very different.

Because some high-mass components are not sufficiently volatile to be detected in gas chromatographic analysis of honeybee wax, we considered the possibility that similar high-mass components could be present in stingless bee wax also. We therefore carried out acidic and alkaline hydrolysis

experiments on the wax from the nest, but found no evidence of high mass esters.

The great diversity in composition of *Apis* beeswax is most probably the principal reason for its plasticity and relatively low melting point (Tulloch, 1980). A study of the conversion of virgin beeswax into finished comb showed that the crystalline virgin beeswax scales are masticated giving the wax a random crystallographic arrangement which lessens its stiffness (Kurstjens et al., 1985). However, the addition of protein during worker mastication converts it into a final product having twice the stiffness of the starting material.

The beeswax of stingless bees is simpler in composition than *Apis* wax, has a lower melting point and has lipids with average shorter chain lengths (Blomquist et al., 1985; Milborrow et al., 1987). Meliponine wax contains a higher percentage of hydrocarbons and a lower percentage of monoesters. Di- and tri-esters were not found here or in the *Trigona* species. With the exception of *T. buyssoni* and *T. atomaria* (Blomquist et al., 1985), meliponines generally use wax mixed with a variety of plant resins (Roubik, 1989; Patricio et al., 2002), probably to increase its strength. *T. australis* workers mix wax mainly with pollen of *Eucalyptus* sp. causing the solid residue to comprise between 12% and 30% by weight (Milborrow et al., 1987).

In the few recorded cases of apparent wax production by individuals, other than workers, mentioned for stingless bees (Kerr and de Lello, 1962; Cruz-Landim, 1967; da Silva, 1977; reviewed by Kerr, 1997), wax samples were never taken from the bees nor was the secreted material verified for its properties. Wax production by stingless bee males is one of the arguments which Kerr and da Cunha (1990) use to support their concept that in stingless bees the overall similarity

Table 2. Summary of analysis of comb wax from *Apis mellifera* colonies (Tulloch, 1971), and three species of *Trigona* stingless bees (Milborrow et al., 1987; Blomquist et al., 1985)

Component	<i>Apis mellifera</i>		<i>T. australis</i>		<i>T. atomaria</i>		<i>T. buyssoni</i>	
	Chain length	% of total	Chain length	% of total	Chain length	% of total	Chain length	% of total
Free fatty acids	24–34	12	10–20	–	12–30+	1.7	12–30+	5.4
Primary alcohols	–	–	–	–	16–32	1.0	16–32	7.0
Long chain esters								
Monoesters	38–52	36	a	6	b	25.5	–	27
Diesters	56–64	14	–	–	–	–	–	–
Triesters	48–54	3	–	–	–	–	–	–
Hydroxyesters	40–50	12	–	–	–	–	–	–
Polyesters	69–75	2	–	–	–	–	–	–
Hydrocarbons								
Saturated	19–31	14			23–37	62	23–37	59
Unsaturated	31–33	–	23–35	90 ^c	27–37	9	27–37	5
Branched	–	–	–	–	–	0	30 & 32	1.4
Unidentified (pigments, propolis, etc.)	–	7	–	16–30	–	0.3 ^d	–	1.5 ^d

^a Higher molecular mass esters beyond C₃₅ were found, but after hydrolysis, no acids higher than C₂₀ were present.

^b Fatty acids C₁₂ to C₃₀+, alcohols C₁₆ to C₃₀ with C₂₄ to C₂₈ the major portion.

^c Unstaurated and saturated hydrocarbons were not resolved in chromatography, but unsaturates represented a greater proportion of the C₃₁ to C₃₅ hydrocarbons.

^d These bees do not add plant resins to their wax.

between workers and males is higher than that between workers and queens when compared to *Apis*.

The resemblance in the positioning of the wax flakes on the abdomen of both virgin queens and workers in *M. bicolor* points to the presence of wax glands in the queens as well. Virgin queens were seen walking around the nest on a regular basis. However, the occasional presence of these queens with wax scales shows that they synthesize wax during only a very short period.

Here we report the first analysis of virgin beeswax scales in stingless bees and of queen-produced wax in particular. The wax production by virgin queens in *M. bicolor* does not tell us whether queens really deposit their wax scales at deposits as do the workers or use them in another way. It does reveal however that in as far as the production of wax in *M. bicolor* is concerned, workers are more similar to queens than to males, a view to some extent opposite to the idea proposed by Kerr and da Cunha (1990).

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