# Cytotoxic Triterpenes from Crataegus pinnatifida

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Bioassay-guided fractionation of *Crataegus pinnatifida* (Rosaceae) gave two cytotoxic ursane-type triterpenes which were identified as uvaol (1) and ursolic acid (2) by physicochemical and spectroscopic methods. 3-Oxo-ursolic acid (3) was synthesized from ursolic acid (2) by Jones method. The cytotoxic activities of these compounds were tested against murine L1210 and human cancer cell lines (A549, SK-OV-3, SK-MEL-2, XF498, and HCT15) *in vitro*. Compounds **1** and **2** showed moderate cytotoxicities against L1210, whereas they showed weak activities against human cancer cell lines. However, compound **3** exhibited potent cytotoxic activities both in murine and in human cancer cell lines.

Key words: Crataegus pinnatifida, Rosaceae, Uvaol, Ursolic acid, 3-Oxo-ursolic acid, Cytotoxicity

## INTRODUCTION

Crataegus pinnatifida Bunge (Rosaceae) is a perennial tree which grows widely in Korea. The fruits of C. pinnatifida have been used for gastric malignancy and diarrhea (Perry, 1980). The leaf of C. pinnatifida has been recognized to possess antihyperlipemic, antiarrhythmic, and antioxidative activity (Jeong et al., 1999). Several triterpenes (Oh et al., 1993; Park et al., 1994) and flavonoids (Kim et al., 1993; Oh et al., 1994) were reported from the fruits and leaves of C. pinnatifida var. psilosa. In continuation of our study on anticancer agents from natural products, we found that the dichloromethane soluble fraction of the leaves of C. pinnatifida displayed cytotoxic activity against L1210 murine leukemia cell. Repeated column chromatography of the cytotoxic fraction resulted in two ursane-type triterpenes. We report here the isolation, and structure identification of compounds 1 and 2, along with the synthesis of 3 from 2. Cytotoxic activities of these compounds 1-3 were also evaluated with several cancer cell lines in vitro.

## MATERIALS AND METHODS

#### General procedures

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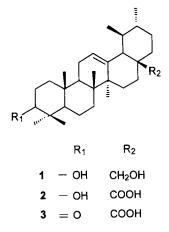


Fig. 1. Structures of Compounds 1-3

Melting points were obtained with an Electrothermal Series IA9100 apparatus and are uncorrected. Optical rotations were measured with a DIP-370 automatic polarimeter (JASCO Co.). IR spectra were recorded on a JASCO Infrared spectrophotometer IR Report-100 (JASCO Co.). <sup>1</sup>H and <sup>13</sup>C NMR experiments were run in CDCl<sub>3</sub> and DMSO- $d_6$  containing TMS as an internal standard using Bruker DRX 300 spectrometers. Si gel 60 (70-230 mesh; Merck) was used for chromatography. TLC was performed on pre-coated Si gel 60 (F254, Merck). Fischer's and RPMI medium were purchased from Hyclone and fetal calf serums from GIBCO-BRL (Grand Island, NY). Other supplement materials for cell culture were purchased from Sigma Chemical Co. (St. Louis, MO).

## **Plant material**

The leaves of *C. pinnatifida* were collected at Kongju, Chungnam province, Korea in October, 1996. The voucher specimens (CNU0954) are deposited at the herbarium in College of Pharmacy, Chungnam National University.

#### **Extraction and isolation**

The leaf of *C. pinnatifida* was extracted with MeOH three times to give MeOH extract (401 g) on removal of solvent *in vacuo*. The MeOH extract was fractionated with dichloromethane and H<sub>2</sub>O. The dichloromethane fraction was chromatographed on silica gel eluted with gradiently hexane-ethyl acetate (10:1 $\rightarrow$ 2:1). Repeated column chromatography with DOWEX-50W and recrystallization in MeOH gave two triterpenoids, compound 1 (307 mg) and compound 2 (977 mg).

**Compound 1 (uvaol)**: colorless needle (MeOH); mp. 214-216°C;  $[\alpha]_D^{25}$ : +56 (*c*=1.0, EtOH), IR  $\nu_{max}$ (KBr) cm<sup>-1</sup>: 3350, 2910, 1450, 1360; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>+ DMSO-*d*6): 5.09 (1H, *t*, *J*=3.5 Hz, H-12), 3.84 (2H, s, H-28), 3.09 (1H, dd, *J*=10.8, 4.8 Hz, H-3), 1.08, 0.98, 0.96, 0.93 (each 3H, s, H-23, 27, 26, 24), 0.92 (3H, d, *J*=5.3 Hz, H-30), 0.79 (3H, d, *J*=5.8 Hz, H-29) and 0.75 (3H, s, H-25); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>+DMSO-*d*6): see Table I.

**Compound 2 (ursolic acid)**: white amorphous powder (MeOH); mp. 261-263°C;  $[\alpha]_0^{25}$ : +63 (c=1.0, EtOH); IR v<sub>max</sub> (KBr) cm<sup>-1</sup>: 3400, 2920, 1680, 1450, 1380; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>+DMSO-d6): 5.17 (1H, t, *J*=3.5 Hz, H-12), 3.04 (1H, dd, *J*=10.9, 4.8 Hz, H-3), 1.07 (3H, s, H-23), 0.94 (6H, s, H-26, 27), 0.90 (3H, s, H-24), 0.84 (3H, d, *J*=6.4 Hz, H-30), 0.77 (3H, d, *J*=9.6 Hz, H-29) and 0.72 (3H, s, H-25); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>+DMSO-d6): see Table I.

#### Synthesis of 3-oxo-ursolic acid (Compound 3)

Jones reagent (1 ml) was added to a solution of compound **2** (100 mg, 0.22 mM) dissolved in dichloromethaneacetone (1:1, 1 ml), and reaction mixture was stirred for 45 min at 5°C (Fieser *et al.*, 1967). Then, *i*-PrOH (0.1 ml) and H<sub>2</sub>O (0.4 ml) were added to the reaction mixture and it was stirred for 10 min at room temperature. The reaction product was extracted with dichloromethane and washed with saturated NaCl solution (10 ml). The dichloromethane layer was evaporated to dryness and chromatographed on Si gel with *n*-hexane-acetone (5:2). Compound **3** (65 mg) was obtained by recrystallization in

**Table I.** <sup>13</sup>C NMR Spectral Data of Compounds 1-3 (CDCl<sub>3</sub> + DMSO-d<sub>6</sub>, 75 MHz)

Carbon No.		2	3	
	37.4	38.1	39.3	
2	25.4	26.9	34.2	
1 2 3 4 5 6 7	76.8	76.8	217.8	
4	37.4	36.5	47.4	
ς	54.6	54.7	55.3	
5	17.8	17.8	19.6	
7	32.3	32.6	32.5	
8	38.3	38.7	39.5	
9	47.0	46.7	46.6	
10	36.3	36.4	36.7	
11	22.8	22.7	23.4	
12	123.9	124.5	125.6	
13	138.5	137.9	138.1	
14	41.4	41.5	42.1	
15	26.8	27.4	28.0	
16	20.0	23.7	24.1	
17	47.0	46.9	48.0	
18	53.6	52.2	52.6	
19	38.2	38.4	39.1	
20	38.2	38.2	38.8	
21	30.2	30.1	30.6	
22	35.1	36.2	36.7	
23	28.0	28.1	26.6	
24	15.7	15.1	17.0	
25	15.3	15.8	15.2	
26	16.3	16.8	17.0	
27	22.5	23.1	23.5	
28	67.9	178.3	183.4	
29	17.1	1/0.3	21.5	
30	21.1	20.9	21.3	

MeOH.

White amorphous powder (MeOH); mp. 238-241°C;  $[\alpha]_{c}^{25}$  :+87 (*c*=1.0, EtOH); IR  $\nu_{max}$ (KBr) cm<sup>-1</sup>: 3400, 2920, 1695, 1670, 1450, 1380; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>+ DMSO-*d*<sub>6</sub>): 5.27 (1H, t, *J*=3.5 Hz, H-12), 1.09, 1.08, 1.05, 1.03 (each 3H, s, H-23, 27, 26, 24), 0.94 (3H, d, *J*=7.8 Hz, H-30), 0.87 (3H, d, *J*=6.4 Hz, H-29) and 0.84 (3H, s, H-25); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>+DMSO*d*<sub>6</sub>): see Table I.

#### In vitro cytotoxicity

In vitro cytotoxicity was measured with murine (L1210, leukemia) and human cancer cells. Five different human cancer lines, A549 (lung adenocarcinoma), SK-OV-3 (ovarian), SK-MEL-2 (skin melanoma), XF498 (CNS) and HCT15 (colon) were purchased from National Cancer Institute (NCI) in U. S. A. The cytotoxic activities were carried out previously before (Kwon *et al.*, 1998) using sulforhodamine B (SRB) method (Skehan *et al.*, 1990). ED<sub>50</sub> value was determined graphically by plotting the viability versus the concentration of the test sample.

#### **RESULTS AND DISCUSSION**

For the isolation of cytotoxic compounds from the leaves of *C. pinnatifida*, the methanol extract was

fractionated with dichloromethane, ethyl acetate and buthanol, successively. The dichloromethane-soluble fraction showed a moderate cytotoxic activity against L1210 cells with  $ED_{50}$  value of 9.5 µg/ml, while the ethyl acetate and buthanol-soluble fraction displayed no cytotoxic activity (>20 µg/ml). Bioassay-guided isolation of dichloromethane fraction afforded two triterpenoids as active principles. Their structures were determined by physico-chemical and spectral data (Nakanishi et al., 1983). They were identified as uvaol (1) and ursolic acid (2) in comparison with authentic samples. Compound 3 was synthesized from compound 2 by lones reagent with 65% yield. The spectral data of <sup>1</sup>H and <sup>13</sup>C NMR of compound 3 were similar to those of compound 2. However, compound 3 showed a carbonyl carbon at  $\delta$  216.2 instead of carbinol carbon at  $\delta$  76.8 and downfield shift of C-2 ( $\delta$ 26.9 $\rightarrow$  $\delta$ 34.2) and C-4 ( $\delta$ 36.5 $\rightarrow$   $\delta$ 47.4) in the <sup>13</sup>C NMR spectrum. Therefore, the structure of compound **3**, was confirmed to be 3-oxo-urs-12-en-28-oic acid (3-oxo-ursolic acid).

The cytotoxicities of 1-3 were evaluated in vitro against six cancer cells which were originated from murine (L1210) and human (A549, SK-OV-3, SK-MEL-2, XF498, and HCT15) cells. As indicated in Table II, 1 and 2 showed moderate cytotoxicities against L1210 with ED<sub>50</sub> values of 2.97 and 3.70 µg/ml, respectively, while they showed weak cytotoxicities against five human cancer cell lines. Compound 3 exhibited a significant cytotoxic activities against all tested cancer cells. ED<sub>50</sub> values were 3.15, 0.91, 3.27, 3.46, 3.89, and 0.56 µg/ml, against L1210, A549, SK-OV-3, SK-MEL-2, XF498, and HCT15, respectively. Especially, compound 3 showed more potent cytotoxic activity against HCT15 than cisplatin. The oxo-oleanolic acid, isolated from Pilea mongolica, also showed a significant potent cytotoxicity against HCT15 cancer cell lines with  $ED_{50}$  value of 0.72 µg/ml (Kwon et al., 1997). It is possible that carbonyl group at C-3 position in triterpene (ursane and oleanane type) might be a pharmacophore for cytotoxicity expression against HCT15. Ursolic acid (2) was previously reported to possess various biological activities such as antiviral (Kashiwada et al., 1998, Min et al., 1999, Tommasi et al., 1992), antitumor promotion (Trumbull et al., 1976), protein kinase C inhibition (Ahn et al., 1998), and antiangiogenic (Sohn et al., 1995) activities. Moreover, this compound have been reported from many kinds of

Table II. Cytotoxicities data of Compounds 1-3

Compound		ED <sub>50</sub> values (µg/ml)							
Compound	L1210	L1210 A549		SK-OV-3 SK-MEL-2 XF498					
1	2.97	11.46	13.17	>10	9.54	9.32			
2	3.70	>10	>10	9.27	>10	>10			
3	3.15	0.91	3.27	3.46	3.89	0.56			
Cisplatin <sup>a)</sup>	3.25	0.91	1.32	0.89	0.72	3.17			

a) Positive Control

plants with relatively low toxicity (Jeong et al., 1999). These facts suggested that compound **2** or its analogue would be an important target for developing new antitumor agents.

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