

ELSEVIER Magnetic Resonance Materials in Physics, Biology and Medicine 12 (2001) 114-120

www.elsevier.com/locate/magma

**Magnetic Resonance Materials in** Physics, Biology and Medicine

 $\mathbf{MAGM}$ 

# **Mixed micelles containing lipophilic gadolinium complexes as MRA contrast agents**

Pier Lucio Anelli<sup>a</sup>, Luciano Lattuada<sup>a,\*</sup>, Vito Lorusso<sup>a</sup>, Michael Schneider<sup>b</sup>, Hervé Tournier<sup>b</sup>, Fulvio Uggeri<sup>a</sup>

> *~' Bracco S.p.A., MiNno Research Centre, via E. Folli 50, 20134, Milan, Italy*  <sup>b</sup> Bracco Research S.A., 31, route de la Galaise, CH-1228 Plan-les-Ouates, Geneva, Switzerland

> > Received 9 January 2001; accepted 11 January 2001

#### **Abstract**

Mixed micelles for MRA are multicomponent systems containing a phospholipid, a biocompatible non-ionic surfactant (e.g. Synpcronic" F-108) and a lipophilic gadolinium complex. A variety of lipophilic gadolinium complexes were designed taking into account features such as: (i) nature of ligand (cyclic versus acyclic); (ii) lipophilic moiety: (iii) global charge of the complex; and  $(iv)$  nature of bond connecting the complex to the lipophilic moiety. All the lipophilic gadolinium complexes after formulation as mixed micelles show high relaxivities in water and in blood (rat). Mixed micelles containing gadolinium complexes bearing only one aliphatic chain cannot be used as MRA contrast agents because they have a high haemolytic effect. Furthermore, in rats they are quickly eliminated from the blood stream. These drawbacks are completely circumvented using gadolinium complexes bearing two aliphatic chains. Mixed micelles containing such complexes show high relaxivities, no haemolytic effect and long blood permanence. This makes them promising candidates as MRA contrast agents. However, elimination, which occurs exclusively through the liver, is not complete, even after 7 days. Complexes containing labile (e.g. ester) bonds between the lipophilic moieties and the chelate subunit are eliminated through both the liver and the kidneys. However, elimination is still not complete after 7 days.  $© 2001$  Elsevier Science B.V. All rights reserved.

*Kevwords:* Blood pool agents: Contrast agents; Lipophilic gadolinium complexes; Mixed micelles; MR angiography; MR imaging

## **1. Introduction**

In the last years, great effort has been devoted to the discovery of blood pool contrast agents for MRI. Indeed, the gadolinium complexes currently available on the market (e.g. Magnevist®) are all low molecular weight species that rapidly equilibrate between the intra and extravascular spaces after intravenous administration. For this reason, intravascular contrast agents are more desirable for coronary artery imaging [1,2] and for the assessment of other important features such as: relative blood volume of tissues, relative blood flow and endothelial permeability [3].

In order to obtain a long permanence in blood of a gadolinium complex, three main different approaches

*E-mail address:* llattuada@bracco.it (L. Lattuada).

have been proposed. The first approach is based on the use of a low molecular weight gadolinium complex (e.g. Angiomark®, Epix Medical or MP-2269, Mallinckrodt) [4,5] which is able to reversibly bind to endogenous blood components, e.g. serum albumin. The second approach takes advantage of gadolinium containing macromolecules like dendrimes (e.g. Gadomer17, Schering) [6] or polymers (e.g. Gd-DTPA-mer, Nycomed) [7] which, because of their large molecular size, remain confined in plasma. The third approach is based on the use of large supramolecular systems like liposomes [8] or micelles [9].

According to this third approach, we found mixed micelles very promising as MRA contrast agents. The term mixed micelles typically refers to three component aggregates of: (i) a non-ionic surfactant containing a polyoxyethylene chain: (ii) an amphipatic compound (e.g. a phospholipid), and (iii) a lipophilic gadolinium complex (Fig. 1).

<sup>\*</sup> Corresponding author. Tel.:  $+39-221-772621$ ; fax:  $+39-221-$ 772770.

<sup>1352-8661/01/\$ -</sup> see front matter © 2001 Elsevier Science B.V. All rights reserved. PII: S1352-8661(01)00107-7





**Phospholipid (e.g. 1,2-dipalmitoyl-3 phosphatidic acid sodium salt)** 

**Non ionic surfactant (e.g. Synperonic F108)** 

```
(Gd
Lipophilic gadolinium complex
```
Fig. 1. Typical composition of a mixed micelle.

Here we report on a series of lipophilic gadolinium complexes which were designed taking into account features such as: type of ligand (cyclic versus acyclic), number and kind of lipophilic moieties, charge of the complex and type of bond connecting the complex to the lipophilic moiety.

The physico-chemical and pharmacokinetic properties of the mixed micelles prepared with such lipophilic gadolinium complexes are also discussed.

## **2. Materials and methods**

#### *2. I. ChemicaLs"*

All organic and inorganic reagents were purchased from Fluka A.G. (Buchs, Switzerland) except for: gadolinium acetate (Alfa Aesar, Karlsruhe, Germany), 1,2-epoxyoctadecane (Aldrich, Milan, Italy), 1 propanephosphonic acid cyclic anhydride (Lancaster, Mijhlheim am Main, Germany), 1,2-dipalmitoyl-glycero-3-phosphatidic acid sodium salt and 1,2-distearoylglycero-3-phosphatidylethanolamine-polyethyleneglycol 2000 (Genzyme/Sygena, Liestal, Switzerland).

## 2.2. Synthesis of lipophilic gadolinium complexes

Building blocks easily synthetizable, like DO3A tris(t-butyl) ester 1 [10], DTPA tetra(t-butyl) ester 2 [11], DTPA dianhydride 3 [12] and DOTA tribenzyl ester 4 [13], were used as starting materials for the synthesis of the lipophilic gadolinium complexes, by reaction with synthons containing  $C_{16}-C_{18}$  aliphatic chains. The synthetic pathways followed to obtain the



Fig. 2. Synthesis of lipophilic gadolinium complexes (first generation). Conditions: (i) BrCH<sub>2</sub>CONHC<sub>18</sub>H<sub>37</sub> [14], EtOH, reflux; (ii) CF<sub>3</sub>COOH; (iii)  $Gd_2O_3$ ; (iv) 1,2-epoxyoctadecane, EtOH, reflux; (v) 5 M HCl; (vi) isobutyl chloroformate,  $C_{18}H_{37}NH_2$ , Et<sub>3</sub>N, THF; (vii) 0.5 M H<sub>2</sub>SO<sub>4</sub>, dioxane, 90°C; (viii) GdCl<sub>3</sub>, 1 M NaOH, H<sub>2</sub>O, CH<sub>3</sub>CN.



Fig. 3. Synthesis of lipophilic gadolinium complexes (second generation). Conditions: (i) BrCH<sub>2</sub>CON(C<sub>18</sub>H<sub>37</sub>)<sub>2</sub> [13], EtOH, reflux; (ii) 5 M HCl; (iii) GdCl<sub>3</sub>, 1 M NaOH, EtOH, reflux; (iv)  $C_{18}H_{37}NH_2$ , DMF, 80°C.

lipophilic complexes are shown in Figs. 2-4. Compounds 6a and 7b labelled with 153Gd were obtained by complexation with  $153GdCl$ <sub>3</sub> (NEN Life Science Products, Switzerland) of the corresponding ligands.

#### 2.3. Mixed micelles formulations

Two different formulations were used. The first formulation was prepared by mixing the lipophilic complex (20 mg ml<sup>-1</sup>), Synperonic<sup>®</sup> F-108 (20 mg ml<sup>-1</sup>) and 1,2-dipalmitoyl-glycero-3-phosphatidic acid sodium salt (DPPA-Na) (20 mg ml<sup>-1</sup>) in buffer at pH 7.3 (Tris 1 g  $1^{-1}$ , 0.3 M glycerol), sonication at 70 $^{\circ}$ C (Branson sonicator 250) for 30 min and filtration through 0.45 um membrane filter. The second formulation was prepared in the same way using the lipophilic complex (20 mg  $ml^{-1}$ ) and 1,2-distearoyl-glycero-3-phosphatidylethanolamine-polyethyleneglycol 2000 (PE-PEG) (20 mg ml<sup> $-1$ </sup>) [17]. In the latter case, PE-PEG acts both as the phospholipid and the surfactant.

## *2.4. Relaxivities*

The relaxation times  $T_1$  and  $T_2$  were measured at 39°C in buffer (Tris 1 g  $1^{-1}$ , 0.3 M glycerol) and rat blood using a Minispec PC120 (Bruker, 0.47T) and using, respectively, the 'inversion recovery' method for  $T_1$  and the Carr-Purcell-Meiboom-Gill (CPMG) technique for  $T_2$ .

#### *2.5. Pharmacokinetic and biodistribution studies in rats*

Blood kinetics were studied by measurement of relaxation times  $T_1$  and  $T_2$  in rat blood. The rats (Sprague-Dawley,  $\approx 250$  g) were weighed and the total amount of blood calculated (6% of body weight). The theoretical Gd concentration (mM) in blood at time  $t=0$ , corresponding to  $100\%$  of the injected dose (ID), was calculated for each rat. The rats  $(n = 2)$  were sacrificed at 10, 30, 60 and 90 min and 2, 4, 7 and 24 h after intravenous injection of 1 ml of the formulation/rat  $(z \approx 75 \text{ }\mu\text{mol} \text{Gd}^{-1} \text{ kg}^{-1})$ . At each time t, 5 ml of blood were collected and  $T_1(t)$  and  $T_2(t)$  were measured. The Gd concentration (mM) in the blood at time  $t$  was calculated using the following formula:

$$
[Gd] = [1/T_i(t) - 1/T_i(0)]/r_i \quad i = 1 \text{ or } 2
$$

where  $T_{\nu}(t)$  is the relaxation time of the blood measured at time t;  $T<sub>i</sub>(0)$  is the relaxation time measured in fresh blood; and  $r_i$  is the relaxivity of the tested formulation in blood. The results were then normalized (injected dose = 75 µmol Gd<sup>-1</sup> kg<sup>-1</sup>) and expressed as percent of ID.

Body distribution and urinary and faecal elimination in rats were obtained by  $153Gd$  radioactivity ( $\gamma$ -counter Packard Minaxi Autogamma<sup>®</sup> 5000 series) or by inductively coupled plasma atomic emission spectrometry (ICP-AES; ARL  $3410 +$ , Redox Laboratories, Monza, Italy) measurements.

The urinary and faecal eliminations of gadolinium were studied after single administration of 6a, 7a and **7b** to male Sprague–Dawley rats ( $\approx 250$  g) ( $n = 3$  per group) at a dose of 1 ml per rat of formulation. Urine and faeces were collected daily in metabolism cages for up to 7 days after the injection. Both urine and faeces were immediately frozen and maintained at  $-20^{\circ}$ C until assayed. Three animals were sacrificed at 24 and 72 h and 7 days after administration and blood, liver, spleen, femur and kidneys were collected to evaluate residual content of gadolinium in organs.



Fig. 4. Synthesis of lipophilic gadolinium complexes (third generation). Conditions: (i)  $(C_1, H_3, COOCH, CH_2)$ , NH [15], 1-propanephosphonic acid cyclic anhydride [16], Et<sub>3</sub>N, CH<sub>2</sub>Cl<sub>2</sub>; (ii) H<sub>2</sub>, Pd/C, EtOH; (iii) (CH<sub>3</sub>COO)<sub>3</sub>Gd, CHCl<sub>3</sub>, MeOH, H<sub>2</sub>O; (iv) C<sub>17</sub>H<sub>35</sub>COOCH<sub>2</sub>CH<sub>2</sub>OCH<sub>2</sub>CH<sub>2</sub>NH<sub>2</sub> [13], DMF,  $60^{\circ}$ C.

#### 2.6. Coronary imaging in micropigs

The potential of compounds to improve contrast-enhanced magnetic resonance coronary angiography (ceMRCA) was tested with an inversion-recovery 3D gradient-recalled echo sequence (IR 3D GRE). This technique uses a segmented 3D GRE sequence. In order to minimise the effect of cardiac motion, the data acquisition window is gated to mid-diastole, which in turn is determined by cine-imaging in a short axis plane through the left ventricle. The acquisition window is  $\approx 10-15\%$  of the RR interval in the ECG trace. For RR intervals of 600-700 ms, the data acquisition window is < 100 ms. Respiratory motion was controlled by acquiring navigator-echo 3D coronary angiography on the anaesthetised and artificially-ventilated animals. Depending on the extent of heart coverage and on spatial resolution, the total acquisition times varied between 8 and 15 min.

For all post-contrast imaging, an inversion pulse preceded the data acquisition segment to null the myocardium signal. This step helps to maximise the contrast-to-noise ratio (CNR) for the coronary arteries. The time interval (TI) between the inversion pulse and the beginning of the acquisition segment is ideally chosen in a way to null the myocardium signal with the blood having returned to the fully relaxed condition (maximum signal) at the same time point.

In order to assess the ability of mixed micelles formulations to cope with these requirements, the  $T_1$  in the blood of three domestic pigs ( $\approx$  20 kg body weight) after administration of cumulative doses of 6a (from 0.025 to 0.2 mmol Gd  $kg<sup>-1</sup>$ ) was measured in vivo, 10 min after injection, using the above techniques on a 1.5

T Siemens Symphony<sup>®</sup> scanner (TR/TE/TI/Flip angle: 3.35/1.81/200 ms/25°; Matrix:  $125 \times 256 \times 32$ ; F.O.V.:  $125 \times 256 \times 64$  mm<sup>3</sup>; Spatial resolution:  $1 \times 1 \times 2$  mm<sup>3</sup>) at Northwestern University (Department of Radiology, Chicago, IL).

## **3. Results**

The mixed micelles formulated with complexes  $5a-c$ , containing one aliphatic chain (first generation), all showed high relaxivities  $(r_1 \ 16-26 \ mM^{-1} \ s^{-1})$  but a very poor blood permanence. For example, after I0 min of injection only  $30\%$  of ID of 5a was present in plasma (Fig. 5). Moreover, for all complexes 5 a strong haemolytic effect was noticed both in vitro and in vivo.

The mixed micelles which were formulated with complexes 6a,b (containing two aliphatic chains, second generation) showed high relaxivities  $(r_1 \ 18-23 \ mM^{-1})$ 



Fig. 5. Blood elimination profiles of 5a, 6a and 6b as evaluated on the basis of  $T_1$  measurements (formulation: complex/DPPA-Na/Synperonic<sup> $\dot{\text{F}}$ </sup> F108 = 20/20/20 mg ml<sup>-1</sup>; injected dose = 1 ml per rat)

Table 1

Organ distribution (ID%) of PE-PEG formulation of $6a$ , $7a$ and $7b$ in	
rats 7 days after administration (dose $= 1$ ml per rat)	



<sup>a</sup> Assessed by radioactivity (<sup>153</sup>Gd).

 $b$  Values in parentheses refer to the DPPA/Synperonic<sup>®</sup> F108 formulation.

<sup>c</sup> Assessed by ICP-AES.

a Values in parentheses were assessed by ICP-AES.

 $s^{-1}$ ) and good blood permanence, without haemolytic effect. Compound 6a was still present in plasma at 70% of ID 30 min after the administration (Fig. 5), while its elimination from plasma was complete  $(< 1\%$  of ID) after 24 h, as evaluated by  $153Gd$  radioactivity studies. A 50-58% of ID was found in faeces and  $\langle 1\% \rangle$  in urine 7 days after injection. The biodistribution data for two different formulations of 6a are reported in Table 1.

The pig was chosen as an animal model for demonstration of the potential of 6a to improve ceMRCA for the size and anatomy of the coronaries that are similar to those in humans. Multiplanar reconstructions (MPR) from a typical 3D experiment showing the right coronary artery (RCA) are displayed in Fig. 6. High blood signal-to-noise ratio (SNR) and almost total myocardial signal suppression is maintained for almost 1 h after administration.

Compounds 7a,b (third generation) contain labile (e.g. ester) bonds which connect the gadolinium complex to the aliphatic chains. For mixed micelles prepared using 7a,b relaxivities  $(r_1 \ 18-22 \ mM^{-1} \ s^{-1})$ , blood permanence and haemolytic effect are very similar to those of mixed micelles containing compounds 6a,b. A significant difference was found in the elimination route and the case of 7b is a good example: 32% of ID was found in faeces and 36% in urine 7 days after injection. (In the case of 7a, a 40% of ID was found in faeces and 27% in urine 7 days after injection). A comparison of the biodistribution of the PE-PEG formulations of 6a, 7a and 7b is reported in Table 1.

## **4. Discussion**

Mixed micelles that contain gadolinium complexes bearing a single aliphatic chain cannot be used as MRA contrast agents because they are quickly eliminated from the blood stream. Moreover, they are strong haemolytic agents, independently of either the charge or the ligand structure (cyclic or acyclic). They apparently behave on erythrocyte membranes like surfactants do. Indeed, single chain surfactants are commonly used to disrupt biological membranes [18]. An empirical explanation of their disruptive power could be the mismatch between their intrinsic geometry and that of the phospholipids, the main components of biological membranes [19].

Mixed micelles containing gadolinium complexes with two aliphatic chains do not show any haemolytic effect. This feature, together with high relaxivities and, more importantly, a long permanence in blood, make these systems very efficient MRA contrast agents. The



Fig. 6. Inversion recovery Nav' Echo 3D MR coronary angiography in pigs after cumulative doses of 6a at 1.5 T. (a) Precontrast, blood  $T_1$ : 1200 ms; (b) 0.025 mmol Gd kg<sup>-1</sup>, blood T<sub>1</sub>: 147 ms; (c) 0.05 mmol Gd kg<sup>-1</sup>, blood T<sub>1</sub>: 82 ms; (d) 0.1 mmol Gd kg<sup>-1</sup>, blood T<sub>1</sub>: 48 ms: (e) 0.15 mmol Gd kg<sup>-1</sup>, blood  $T_1$ : 40 ms; (f) 0.2 mmol Gd kg<sup>-1</sup>, blood  $T_1$ : 30 ms.



Fig. 7. Rationale of the in vivo cleavable lipophilic complexes.

images taken from pigs (Fig. 6) using a mixed micelle formulation of 6a clearly show the usefulness of this product in the detection of blood vessels, such as the right coronary artery (RCA). Compound 6b [20] is also promising, although the relaxivities and the permanence in blood are a little inferior to those of 6a.

The main problem of compounds such as 6 is the elimination. Compound 6a is eliminated, although not completely through the hepatic route. After administration to rats of micelles labelled with 153Gd, a significant amount of radioactivity is found in some organs, even after 7 days (Table 1). Better results, especially for residual radioactivity in spleen and bone, are obtained with the PE-PEG formulation (Table 1). This is an important issue because even better results, in terms of elimination, could be achieved simply investigating different formulations of the mixed micelles.

The synthesis of compounds containing an in vivo labile bond between the gadolinium complex and the lipophilic moiety was taken into account to overcome the elimination issue. The idea was that, after degradation of the mixed micelle, the lipophilic gadolinium complex, exposed to the biological environment, should be converted into a small, hydrophilic, easily eliminable gadolinium complex and a lipophilic counterpart, preferably endogenous or with a known metabolic pathway (Fig. 7). The body of mammals is rich in enzymes capable of hydrolysing esters (i.e. esterases are practically ubiquitous all over the body) and for this reason, an ester was chosen as the cleavable bond [21], while fatty acids were chosen as lipophilic moieties.

Compounds 7a,b are eliminated, not only through liver but also through kidneys and this is an indirect evidence of a partial metabolisation of the products Unfortunately, significant amount of gadolinium is still found in liver and bone of rats 7 days after injection (Table 1).

The present study shows that mixed micelles containing lipophilic gadolinium complexes are promising candidates as MRA contrast agents. The synthesis of new classes of labile gadolinium complexes and the study of different formulations are in progress in order to improve the elimination.

## **Acknowledgements**

We thank Debiao Li, Finn P. and Jei Zheng at Northwestern University (Chicago) and Laub G. and Simonetti O. of Siemens AG for the imaging studies.

#### **References**

- [I] Li D, Zheng J, Bae KT, Woodard PK, Haacke EM. Contrastenhanced magnetic resonance imaging of the coronary arteries. Invest Radiol 1998;33:578-86.
- [2] Mühler A. The future of contrast-enhanced magnetic resonance angiography. Are blood pool agents needed? Invest Radiol 1998:33:709-- 14.
- [3] Mühler A. Assessment of myocardial perfusion using contrastenhanced MR imaging: Current status and future developments. MAGMA 1995:3:21-33.
- [4] Lauffer RB, Parmelee DJ, Dunham SU, Ouellet HS, Dolan RP, Witte S, McMurry TJ, Walovitch RC. MS-325: albumin-targeted contrast agent for MR angiography. Radiology 1998;207:529- 38.
- [5] Wallace RA, Haar JP, Miller DB, Woulfe SR, Polta JA, Galen KP, Hynes MR, Adzamli K. Synthesis and preliminary evaluation of MP-2269: a novel, nonaromatic small-molecule bloodpool MR contrast agent. Magn Reson Med 1998;40:733-9.
- [6] Dong Q, Hurst DR, Weinmann HJ, Chenevert TL, Londy FJ, Prince MR. Magnetic resonance angiography with Gadomer-17. An animal study original investigation. Invest Radiol 1998;33:699-708.
- [7] Ladd DL, Hollister R, Peng X, Wei D, Wu G, Delecki D, Snow RA, Toner JL, Kellar K, Eck J, Desai VC, Raymond G, Kinter LB, Desser TS. Rubin DL. Polymeric gadolinium chelate magnetic resonance imaging contrast agents: design, synthesis and properties. Bioconjugate Chem 1999;10:361-70.
- [8] Unger EC, Shen D-K, Fritz TA. Status of liposomes as MR contrast agents. J Magn Reson Imag 1993;3:195-8.
- [9] Tournier H, Hyacinthe R, Schneider M. Gadolinium-containing mixed micelles formulations: a new class of blood pool MRI/ MRA contrast agents, Acad Radiol (in press).
- [101 Ranganathan RS, Marinelli ER, Pillai R, Tweedle MF. Aromatic amide compounds and metal chelates thereof. PCT Int Appl WO95/27705, 1995; October 19.
- **[11]**  Anelli PL, Fedeli F, Gazzotti O, Lattuada L, Lux G, Rebasti F. L-Glutamic acid and L-lysine as useful building blocks for the preparation of bifunctional DTPA-like ligands. Bioconj Chem 1999;10:137-40.
- [12] Eckelman WC, Karesh SM, Reba RC. New compounds: fatty [16] acid and long chain hydrocarbon derivatives containing a strong chelating agent. J Pharm Sci  $1975;64:704-6$ .
- Anelli PL, Lattuada L, Uggeri F, Lux G, Serleti M, Gabellini M, Tournier H. Amphipatic polycarboxylic chelates and complexes [18] with paramagnetic metals as MRI contrast agents. PCT Int Appl WO00/30688, 2000; June 2.
- [14] Rudkevich DM, Verboom W, Brzozka Z, Palys MJ, Stauthamer WPRV, van Hummel GJ, Franken SM, Harkema S, Engbersen JFJ, Reinhoudt DN. Functionalized UO2 salenes: neutral receptors for anions. J Am Chem Soc 1994;116:4341-51.
- [15] Tundo P, Kippenberger DJ, Politi MJ, Klahn P, Fendler JH. Redox active functionally polymerized surfactant vesicles. Syntheses and characterization. J Am Chem Soc 1982;104:5352-8.
- [16] Wissmann H, Kleiner H-J. New peptide synthesis. Angew Chem Int Ed Engl 1980;19:133-4.
- Lasic L, Martin F. Stealth Liposomes. Boca Raton, FL: CRC Press, 1995.
- [18] Kondo T. Mechanisms of hemolysis by surface active agents. Adv Coll Interface Sci 1976;6:139-72.
- [19] Jayasuriya N, Bosak S, Regen SL. Design, synthesis and activity of membrane-disrupting bolaphiles. J Am Chem Soc 1990;112:5844-50.
- [20] Jasanada F, Nepveu F. Synthesis of amphiphilic chelating agents: bis(hexadecylamide) and bis(octadecylamide) of diethylenetriaminepentaacetic acid. Tetrahed Lett 1992;33:5745-8.
- [21] Bundgaard H. Novel chemical approaches in prodrug design. Drugs Future 1991;16:443-58.