

Deuterium NMR study of the MP-2269: albumin interaction — a step forward to the dynamics of non-covalent binding

K. Adzamli ^{a,*}, L. Vander Elst ^b, S. Laurent ^b, R.N. Muller ^b

^a *Imaging Division, Mallinckrodt Inc., 675 McDonnell Blvd., P.O. Box 5840, St. Louis, MO 63134, USA*

^b *Department of Organic Chemistry and NMR Lab., University of Mons-Hainaut, B-7000 Mons, Belgium*

Received 7 December 2000; accepted 14 December 2000

Abstract

MP-2269, the Gd(III) complex of 4-pentylbicyclo[2.2.2]octane-1-carboxyl-di-L-aspartyl-lysine-derived-DTPA, is a small Gd-agent that binds non-covalently to serum albumin *in vivo* to assume the enhanced relaxivities associated with macromolecular agents, (due in part to increased rotational correlation time, τ_R). To further explore the fundamental parameters that govern the dynamics of water proton relaxation enhancement by this prototypical albumin-binding agent, the rotational correlation time (τ_R) for the deuterated La(III) analog of MP-2269 has been independently measured in the presence and absence of 4% albumin using ²H-NMR approaches. The diamagnetic La(III) analog of MP-2269 was deuterated at the α -position of the carbonyl groups. ²H-NMR studies were conducted at 7.05T (46 MHz) and 310°K on a Bruker NMR spectrometer. Spectral deconvolution permitted calculation of transverse relaxation rates, $1/T_2$, from the NMR linewidths and subsequently, τ_R . The results yielded a τ_R of the albumin bound complex of ~ 8 ns. This value is intermediate between those earlier estimated by ¹⁷O-NMR (~ 1 ns) and ¹H-NMRD (~ 20 – 50 ns) and significantly shorter than that of albumin. The ²H-NMR study results also indicate that the exchange between free and albumin-bound forms of the La(III) analog is slow (exchange lifetimes > 1 ms). This slow exchange does not affect the water residence lifetimes (τ_M 140–280 ns). © 2001 Elsevier Science B.V. All rights reserved.

Keywords: NMR; Non-covalent binding; MP-2269; Deuterium

1. Introduction

The proton NMR and NMRD studies of the prototype blood pool MRI contrast agent MP-2269 (the Gd(III) complex of 4-pentylbicyclo[2,2,2]octane-1-carboxyl-di-L-aspartyl-lysine derived DTPA) which binds to serum albumin through hydrophobic interaction between the non-aromatic cyclic side chain and the protein, have been the subject of recent studies [1,2]. In these, comparisons of the NMR, EPR, and NMRD profiles with relaxation theory were carried out in order to probe the utility of theory as a guide for the design of better paramagnetic agents for MRI as earlier described [3,4].

Briefly, inner sphere contributions tend to dominate T_1 relaxation enhancement from macromolecular Gd-complexes at typical MRI fields. Such inner sphere contribution to macromolecular $1/T_1$ is given by: $1/T_{1P}^{is} = q[\text{Gd}^{3+}]/55.5(T_{1M} + \tau_M)$; where, q is the number of coordinated inner sphere water molecules (typically $q = 1$), T_{1M} is the longitudinal relaxation time of the coordinated water protons, and τ_M is their lifetime on the metal ion (typically of the order of microseconds for Gd^{3+}). $1/T_{1M}$ is a complicated function of the strength of the interaction, the magnitude of the applied MRI field, and the overall correlation times, τ_{Ci} ($i = 1$ or 2) ($\tau_{Ci}^{-1} = \tau_R^{-1} + \tau_{Si}^{-1} + \tau_M^{-1}$), the inverse of the sums of the reciprocals of the rotational correlation time τ_R , the electronic relaxation times of the paramagnetic metal ion τ_{Si} , and τ_M . Attachment of Gd^{3+} -chelates to macromolecules generally leads to enhancement in $1/T_1$ in the 10–50 MHz field-range. This phenomenon is well understood and is associated with the long τ_R of macromolecular complexes. Hence,

* Corresponding author. Present address: 16212 Trade Winds Court, St. Louis, No, 63040-1814, USA. Tel.: +1-314-719-4609; fax: +1-314-6547212.

E-mail address: kadzamli00@aol.com (K. Adzamli).

formation of a rigid macromolecular paramagnetic complex has been the design goal for obtaining high-relaxivity contrast agents [5–7]. The peak relaxivity of the covalently bonded Gd^{3+} -macromolecules have been limited either by lack of rigidity or by a long value of τ_M . Recent innovations yielded relatively rigid paramagnetic macromolecular complexes in-vivo in the blood pool through reversible non-covalent hydrophobic interactions of small lipophilic Gd^{3+} -chelates with serum, leading to development of highly efficient intravascular contrast agents [8,9].

Combined 1H -NMRD, ^{17}O -NMR and EPR studies of MP-2269 bound to serum albumin showed that the water residence time (τ_M) is not markedly influenced by the protein binding. These studies, however, were unable to assess the time scale of the protein-to-ligand exchange interactions and also gave indirect (and disparate) values for the rotational correlation time (τ_R) of the albumin-bound complex [1,2,10]. 2H relaxation rates of diamagnetic compounds originate predominantly from quadrupolar interactions and are thus modulated by τ_R . 2H transverse relaxation rate of carbon-deuterium bond is described by the following equation:

$$\frac{1}{T_2} = \frac{3\pi^2}{20} \left(\frac{e^2Qq}{h} \right)^2 \left[3\tau_R + \frac{5\tau_R}{1 + \omega^2\tau_R^2} + \frac{2\tau_R}{1 + 4\omega^2\tau_R^2} \right]$$

where e^2qQ/h is the quadrupolar coupling constant and ω is the angular frequency of deuterium.

Hence, binding a deuterated sample to a macromolecule results in enhanced transverse relaxation rate ($1/T_2$, R_2) of the bound material. If the exchange between bound and unbound compound is fast on the relaxation rates scale, a decrease in the initial concentration of the ligand at constant macromolecule concentration results in an increase of the observed R_2 [11,12]. An 2H -NMR relaxation rate study of the diamagnetic La(III)- d_{15} -MP-2269 was, therefore, undertaken in order to obtain an independent and non-ambiguous value of τ_R for the albumin bound MP-2269 analog.

2. Material and methods

Details of the synthesis and analyses of MP-2269, the tetrasodium salt of 4-pentylbicyclo[2.2.2]octane-1-carboxyl-di-L-aspartyl-lysine-derived-Gd(DTPA), have been published [8] (Fig. 1).

The La(III) analog was prepared by mixing aqueous solutions of equimolar amounts of lanthanum chloride and ligand (4-pentylbicyclo[2.2.2]octane-1-carboxyl-di-L-aspartyl-lysine-derived-DTPA). The La-complex was deuterated on the α -position of the carbonyl groups following the procedure outlined by Wheeler and Legg [13]. The deuterated La complex was dialyzed and

isolated after lyophilization. The deuteration was confirmed by 1H -NMR spectroscopy; the product contains 15 deuterium atoms. The sodium and potassium contents were determined by flame photometry (Flame Photometer IL 943, Instrumentation Laboratory S.p.A., Milan, Italy). A solution of 4% HSA (Sigma, Bornem, Belgium) in deuterium depleted water was used for the binding study and for the 2H -NMR measurements performed at 7.05 T and 310 K on a Bruker AMX-300 spectrometer. 2H transverse relaxation rates ($1/T_2$) were derived from the linewidths ($\Delta\nu_{1/2} = 1/\pi T_2$) of the deconvoluted spectra using a computer algorithm (Linesim Bruker program). A quadrupolar coupling constant of 170 KHz was used for the calculations.

3. Results

Since 2H relaxation rates ($1/T_2$, R_2) of diamagnetic compounds originate predominantly from quadrupolar interactions modulated by τ_R , the enhanced transverse relaxation rate of the bound material is a direct source of τ_R . When the deuterated La-analog of MP-2269 was added to HSA, no marked change of the linewidth was noticed. The lack of effect on the linewidth can be explained either by an absence of binding interaction, (unreasonable since prior evidence points to association, and ongoing studies yield K_a of the order of 9000 M^{-1} , at a minimum of two equivalent binding sites) or by a slow exchange between bound and unbound material.

The presence of a broad component in the 2H -NMR spectra of La(III)- d_{15} -MP-2269 (0.25 and 0.75 mM), in aqueous albumin solutions at 310 K (Fig. 2) confirms the slow exchange hypothesis. The spectra were deconvoluted in their underlying components.

At low concentrations of the ligand (0.25 and 0.75 mM), the deuterium spectra show two narrow peaks—one from residual HDO (left hand peak), one from the unbound material (right hand peak) and a broad component due to the albumin bound ligand. Analysis of the linewidth of the broad component allows the evaluation of the τ_R of the bound ligand and the relative area

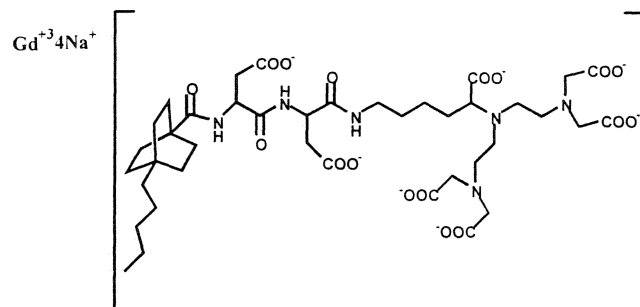


Fig. 1. Chemical structure of MP-2269.

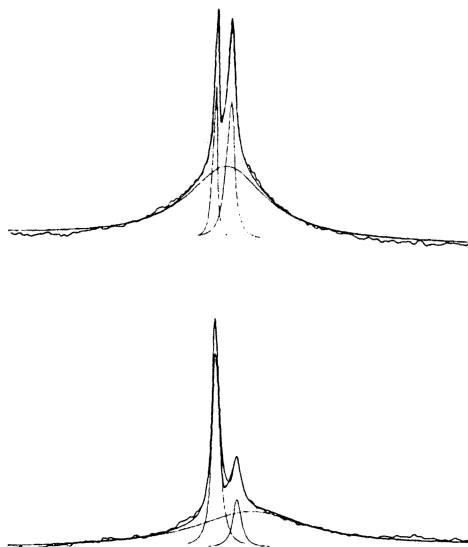


Fig. 2. ^2H -NMR spectra of solutions containing 4% of HSA and 0.75 mM (top spectrum) or 0.25 mM (bottom spectrum) of the La-analog of MP-2269.

of the narrow and broad peaks give the amounts of each species present (Fig. 2 and Table 1). Values of τ_R for the albumin bound contrast agent around 8 ns are found (compared with ~ 140 ps in albumin free solution). The percentages of bound compounds are in very good agreement with the expected values (Table 1).

4. Discussion

In earlier studies [1,2,10], the Gd^{3+} contributions to the measured NMR, EPR and NMRD profiles of albumin solutions were fit to theory, using values for the parameters for the analogous Gd-DTPA^{2-} as initial values [3]. The assumptions needed to simplify the theoretical fits and the extent to which these carry over to other Gd-analogs and their macromolecular adducts remain to be established. Using these approaches, studies of the title compound yielded disparate τ_R values for the albumin adduct despite the firm establishment of τ_M [2]. Similar controversies exist in the current literature [11,14,15]. We have, therefore, adopted the ^2H -NMR approach to reach τ_R values that are not subject to additional extrinsic parameters. This hopefully would form a basis for generating consistent τ_R values that

can be compared, (in addition to independently-derived τ_M) in assessment of the impact of various structural modifications on relaxation rates.

The τ_R value of the bound contrast agent found in this work ($\tau_R = 8$ ns) is shorter than the expected τ_R for a fully immobilized ligand ($\tau_R = 20\text{--}22$ ns) [16] indicating some segmental mobility of the contrast agent with respect to the protein one.

Further, the impact of the protein–ligand exchange rate on the water–metal exchange is not easily delineated from the earlier studies. The deuterium NMR approach in this study clearly demonstrated the various time regimes for this system, i.e. significantly slow protein–ligand exchanges that do not impact the metal–water exchange (τ_M).

5. Conclusion

As compared with the other albumin binding contrast media studied so far [11,12], MP-2269 exhibits slower exchange rate between the bound and free states. These kinetic aspects which are not reflected by water ^1H -NMR relaxometry are adequately addressed by ^2H studies. Clearly, no relationship exists between the residence time of the coordinated water molecule and the residence time of the lanthanide complex on the protein. In the present case, MP-2269 has a residence time on the protein of the order or longer than 1 ms, whereas the exchange of the water molecule ranges between 140 and 280 ns. This study also provided the first estimate of the exchange rate between free and bound forms of this molecule (the slowest amongst albumin-binding agents of this class).

Acknowledgements

The authors thank Patricia de Francisco for her help in preparing the manuscript. This work was sponsored by the Imaging Division, Mallinckrodt, Inc., and supported by the ARC Program 95/00-194 of the French Community of Belgium, the Biomed II MACE Program of the Commission of the European Communities (contract BMH4-CT-96-0051; DG12-SSMA) and the Fonds National de la Recherche Scientifique (FNRS) of Belgium.

Table 1

	$\Delta\nu_{1/2}$ of the broad component (Hz)	τ_R (ns)	Percentage of bound compound ^a
4% of HSA + 0.75 mM of La-analog of MP-2269	379	~ 7	84 (84)
4% of HSA + 0.25 mM of La-analog of MP 2269	400	8–9	90 (93)

^a Values in parentheses calculated for $K_d = 9000 \text{ M}^{-1}$ and two equivalent binding sites.

References

- [1] Toth E, Connac F, Helm L, Adzamli K, Merbach AE. Direct assessment of water exchange on a Gd(III) chelate bound to a protein. *J Biol Inorg Chem* 1998;3:606–13.
- [2] Adzamli K, Toth E, Periasamy MP, Koenig SH, Merbach AE, Adams MD. ^1H -NMRD and ^{17}O -NMR assessment of water exchange and rotational dynamics of two potential MRI agents: MP-1177 (an extracellular agent) and MP-2269 (a blood pool agent). *MAGMA* 1999;8:163–71.
- [3] Kellar KE, Henrichs PM, Spiller M, Koenig SH. Relaxation of solvent protons by solute Gd^{3+} -chelates, revisited. *Magn Reson Med* 1997;37:730–5.
- [4] Powell DH, Dhubhghaill OM, Pubanz D, Helm L, Lebedev YS, Schlaepfer W, Merbach AE. Structural and dynamic parameters obtained from ^{17}O -NMR, EPR and NMRD studies of monomeric and dimeric Gd^{3+} complexes of interest in magnetic resonance imaging: an integrated and theoretically self-consistent approach. *J Am Chem Soc* 1996;118:9333–46.
- [5] Ogan MD, Schmiedl U, Moseley ME, Grodd W, Paaajanen H, Brasch RC. Albumin labeled with Gd-DTPA: an intravascular contrast-enhancing agent for magnetic resonance blood pool imaging: preparation and characterization. *Invest Radiol* 1987;22:665–71.
- [6] Adam G, Neuerburg J, Spuntrup E, Muhler A, Scherer K. Gd-DTPA-cascade polymer: potential blood pool contrast agent for MRI. *J Magn Reson Imaging* 1994;4:462–6.
- [7] Lauffer RB, Brady TJ, Brown III RD, Baglin C, Koenig SH. $1/T_1$ NMRD profiles of solutions of Mn^{2+} and Gd^{3+} protein-chelate conjugates. *Magn Reson Med* 1986;3:541–8.
- [8] Wallace RA, Haar JP, Miller DB, et al. Synthesis and preliminary evaluation of a small-molecule blood pool MR contrast agent. *Magn Reson Med* 1998;39:733–9.
- [9] Lauffer RB, Parmelee DJ, Dunham SU, Ouellet HS, Dolan RP, Witte S, McMurry TJ, Walovitch RC. MS-325: albumin-targeted contrast agent for MR angiography. *Radiology* 1998;207:529–38.
- [10] Adzamli K, Spiller M, Koenig SH. Water-Proton Relaxation by a non-covalent albumin-binding Gadolinium-chelate: An NMRD study of a potential blood pool agent, *Acad Radiol* 2000, in press.
- [11] Muller RN, Radüchel B, Laurent S, Platzek J, Piérart C, Mareski P, Vander Elst L. Physicochemical characterization of MS-325, a new gadolinium complex, by multinuclear relaxometry, *Eur J Inorg Chem* 1999;1949–1955.
- [12] Vander Elst L, Maton F, Laurent S, Seghi F, Chapelle F, Muller RN. A multinuclear MR study of Gd-EOB-DTPA: comprehensive preclinical characterization of an organ specific MRI contrast agent. *Magn Reson Med* 1997;38:604–614.
- [13] Wheeler WD, Legg JI. Assignment of the deuterium NMR spectra of chromium (III) complexes with EDTA and related ligands. *Inorg Chem* 1985;24:1292–7.
- [14] Aime S, Chiaussa M, Digilio G, Gianolio E, Terreno E. Contrast agents for magnetic resonance angiographic applications: ^1H and ^{17}O -NMR relaxometric investigations on two gadolinium(III) DTPA-like chelates endowed with high binding affinity to human serum albumin. *J Biol Inorg Chem* 1999;4:766–74.
- [15] Caravan P, Bulte JWM, Dunham SU, Amedio J, Lauffer RB. T_1 and T_2 NMRD studies of ANGIOMARK™ (MS-325). In: *Proceedings of the International Society of Magnetic Resonance in Medicine, Seventh Annual Meeting, 1999*, p.724.
- [16] Vander Elst et al. submitted to *MAGMA*.