Influence of Long-Term Aging and Superimposed Creep Stress on the Microstructure of 2.25Cr-1Mo Steel

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To understand the influence of high-temperature aging and superimposed creep stress on the microstructural variations in a 2.25Cr-1Mo steel, the shoulder and gage portions of the specimens subjected to stress-rupture tests at 540 °C and 580 °C have been studied by transmission electron microscopy. In the normalized and tempered condition, the steel exhibited a tempered bainitic structure and the carbides were present as M_3C globules, M_2C platelets, and $M_{23}C_6$ rectangular parallelepipeds. Aging the steel at 540 °C for 7022 hours or 17,946 hours resulted in considerable coarsening of M_2C and caused precipitation of M_6C carbides. The superimposed creep stress enhanced the M_2C precipitation. The ferrite matrix exhibited some recovery in the specimens exposed for 17,946 hours. While M_2C platelets were observed in a few areas after 14,836 hours of aging at 580 °C, this carbide was virtually nonexistent when a stress of 78 MPa was superimposed. Amounts of $M_{23}C_6$ persisted throughout the tests at both 540 °C and 580 °C. The M_6C carbide became more predominant after long exposure at 580 °C for 14,836 hours.

I. INTRODUCTION

A 2.25Cr-1Mo steel has been used for hightemperature service applications in power generation and chemical industries for over five decades. Apart from possessing adequate creep strength, the corrosion resistance of this grade of steel is superior to that of 0.5Mo, 1Cr-0.50Mo, and 1.25Cr-0.50Mo steel's. Microstructural changes due to long-term tempering and the creep-deformed structures were investigated a few decades ago using carbon extraction replicas.^[1-4] Recently, the microstructure of the Cr-Mo and Cr-Mo-V steels exposed to 20 years of service life has been studied to understand the microstructural degradation.^[5-9] Some investigators used accelerated aging at 630 °C to produce microstructures that simulate 25,000 to 1,000,000 hours of service exposure at 540 °C.^[10] Though these studies established various structural changes due to prolonged service exposure, the influence of exposure time and temperature vis-à-vis that of superimposed creep stress in bringing about the observed microstructural changes is still not clear. Considering the recent interest in predicting the remaining life of the components exposed to the designed life,^[11-12] a simultaneous study of the two factors becomes important. We have recently reported the results of our investigations on the role of hightemperature exposure time and that of the superimposed creep stress on the microstructures of 1.25Cr-0.5Mo, 1Cr-0.5Mo, and 0.5Cr-0 · 5Mo-0.25V steels.[13,14,15]

Since creep-resistant steels are presently imported by India, Tata Steel embarked on a program of producing 2.25Cr-1Mo grade steel to international specifications.

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Stress-rupture tests have been carried out at various temperatures, and the results of these tests have been reported elsewhere.^[16–18] In the present investigation, the results of a detailed transmission electron microscopy study carried out on both the shoulder and gage portions of the stress-rupture tested specimens of a 2.25Cr-1Mo steel are presented in order to arrive at an understanding of the effects of high-temperature exposure and those of superimposed creep stress.

II. EXPERIMENTAL

A 2.25Cr-1Mo steel, with the chemistry shown in Table I, was made in a 6-ton electric arc furnace and teemed into a mold fitted with a hot top. The ingot was subsequently rolled to 250×250 mm square blooms, and these blooms were machined down to 210×210 mm square gothic bars and subsequently rolled to seamless tubes of 215-mm outside diameter and 40-mm wall thickness through a piercing elongator and pilgering process. Longitudinal blanks cut from the tubes were machined down to 25-mm-diameter rounds and austenized at 950 °C for 1 hour, air cooled, and subsequently tempered at 680 °C for 6 hours.

Stress-rupture tests were carried out at 540 °C, 560 °C, and 580 °C in multipoint creep-testing machines on cylindrical specimens prepared from the normalized-andtempered rounds. The loading was done with the aim of achieving rupture times of 1000, 3000, and 10,000 hours at each test temperature. The shoulder and gage portions of the specimens that ruptured at 540 °C and 580 °C were selected for a detailed microstructural study. In order to

 Table I.
 Chemical Composition of the Steel (Weight Percent) Investigated

С	Mn	Si	S	P	Cr	Mo	Fe
0.16	0.55	0.34	0.018	0.017	2.10	1.06	bal.

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Table II. Details of Stress-Rupture Specimens of 2.25Cr-1Mo Steel Used for the TEM Study*

Temperature (°C)	Stress (MPa)	Rupture Time (h)	Elongation (Pct)	Reduction in Area (Pct)	
540	165	7022	27	79	
540	132	17,946	50	87	
580	98	6526	41	82	
580	78	14,836	86	87	

*Normalized and tempered specimens were also examined.



Fig. 1-Optical microstructure of steel: (a) normalized and tempered and (b) stress rupture tested at 580 °C for 14,836 h.



Fig. 2—TEMs of the steel in normalized and tempered conditions: (a) low-magnification BF shows the preferential alignment of rod-shaped carbides; (b) reveals the three morphologies of carbide (1) globular, (2) fine, and (3) rectangular parallelepipeds; (c) DF shows the globular morphology carbide, (d) SAD and (e) schematic representation of SAD identifying the globular carbide as M_3C ; (f) DF shows the plate carbide; (g) SAD and (h) schematic representation of SAD identifying it as M_2C ; (i) BF and (j) DF revealing the rectangular parallelepiped morphology of carbide; (k) SAD and (l) schematic representation of SAD identifying it as $M_{22}C_6$.









Fig. 2 Cont. — TEMs of the steel in normalized and tempered conditions: (a) low-magnification BF shows the preferential alignment of rodshaped carbides; (b) reveals the three morphologies of carbide (1) globular, (2) fine, and (3) rectangular parallelepipeds; (c) DF shows the globular morphology carbide, (d) SAD and (e) schematic representation of SAD identifying the globular carbide as M_3C ; (f) DF shows the plate carbide; (g) SAD and (h) schematic representation of SAD identifying it as M_2C ; (i) BF and (j) DF revealing the rectangular parallelepiped morphology of carbide; (k) SAD and (l) schematic representation of SAD identifying it as $M_{23}C_6$.



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understand the effects of exposure time and the superimposed creep stress, the shoulder and gage portions of each specimen were studied. The details of the specimens used for transmission electron microscopic study are shown in Table II.

Heat-treated samples and creep-deformed specimens were mechanically polished and etched with 2 pct nital for optical microscopic examination. Slices of 0.50-mm thickness were cut from the heat-treated and creep-tested specimens at an extremely low speed using an ISOMET low-speed saw. These slices were mechanically ground to less than 0.10-mm thickness, and thin foils were prepared by electropolishing using an electrolyte of 10 pct perchloric acid and 90 pct glacial acetic acid. Thin foils were examined in a JEOL JEM 200 CX transmission electron microscope operating at 160 KV.

III. RESULTS

A. Optical Microscopy

The microstructure of the steel in the normalized and tempered condition, shown in Figure 1(a), reveals that it is fully bainitic. Figure 1(b) shows the microstructure

of the steel after holding at 580 °C for 14,836 hours superimposed with creep stress; the microstructure reveals some recrystallization of the ferrite matrix of bainite.

B. Transmission Electron Microscopy

The transmission electron micrographs (TEMs) of the steel in the normalized and tempered condition are shown in Figure 2. Figure 2(a) is a low-magnification TEM showing the tempered bainitic structure; even at this magnification, the preferential alignment of rod-shaped carbides along the lath boundaries of ferrite is seen. Figure 2(b) is a bright-field micrograph (BF) revealing three morphologies of carbide: (1) globular (0.15 μ m diameter), (2) fine plate (0.05 μ m long), and (3) rectangular parallelepiped shaped (0.5 μ m long).

The dark-field micrograph (DF) for the globular morphology carbide of Figure 2(b) is shown in Figure 2(c). The selected area diffraction (SAD) pattern taken from the carbide and its schematic representation are shown in Figures 2(d) and (e), respectively, which identify it as Fe₃C (this pattern contains reflections from the carbide $M_{23}C_6$ as well). The DF for plate-shaped carbide revealed in Figure 2(b) is shown in Figure 2(f) and the SAD, taken from this precipitate and its schematic representation, shown in Figures 2(g) and (h), respectively, identify it as Mo₂C. The following orientation relationship is obtained from this figure:

 $(101)_{\alpha} \sim 5 \deg (1\overline{1}.\overline{1})M_2C$ (233)_{α} ~ 5 deg (11.0)M₂C

Figures 2(i) and (j), which are the BF and DFs, respectively, reveal the carbides with a rectangular parallelepiped morphology. The SAD taken from this carbide and its schematic representation are shown in Figures 2(k) and (l), respectively, which identify it as $M_{23}C_6$. The following orientation relationship exists in this figure:

$$(001)_{\alpha} //(011) M_{23} C_6$$

 $(\bar{1}20)_{\alpha} //(21\bar{1}) M_{23} C_6$

This is in agreement with the Nishiyama-Wassermann relationship.^[19] The TEMs for the shoulder portion of the specimen creep tested at 540 °C for 7022 hours are shown in Figures 3(a) through (c). Figure 3(a) reveals numerous plate-shaped carbides, 0.50- μ m long. These are identified as M₂C by selected area diffraction. The coarse carbides (0.2- μ m diameter) revealed in this figure have been identified as M₆C. Figure 3(b) is the BF revealing the fine globular M₆C precipitates (0.1- μ m size) coexisting with plate-shaped precipitates. The grain boundary carbide shown in Figure 3(c) was identified as M₆C, whereas the rectangular parallelepiped carbides revealed in the same figure were identified as M₂C₆. The carbide M₃C has dissolved in the matrix after 7022 hours at 540 °C.

The TEMs for the gage portion of the specimen creep tested at 540 °C for 7022 hours are shown in Figures 4(a) through (e). Figure 4(a) shows numerous M_2C plateshaped precipitates, 0.4 μ m long, coexisting with fine globular carbides identified as M_6C . Figure 4(b) reveals the spheroidization tendency of coarse M_6C carbides, while Figure 4(c) illustrates the grain boundary carbide. The SAD pattern and its schematic representation shown in Figures 4(d) and (e), respectively, identify it as M_6C . The carbide M_3C has dissolved in the matrix after 7022 hours at 540 °C.

The TEMs for the shoulder portion of the creep specimen tested at 540 °C for 17,946 hours are shown in Figures 5(a) through (c). Figure 5(a) shows well-developed (maximum length ~ 1.5 μ m) M₂C precipitates coexisting with fine ellipsoidal M₆C precipitates. Figure 5(b) reveals the significant tendency of coarse M₆C carbide of 0.25 μ m size to spheroidize. The ferrite matrix adjacent to grain boundaries has become depleted of plateshaped carbides. Figure 5(c) demonstrates all the features: plate-shaped carbide M₂C, fine globular carbide M₆C, and rectangular parallelepiped carbide M₂₃C₆.

The TEMs for the gage portion of the specimen creep tested at 540 °C for 17,946 hours are shown in Figures 6(a) through (c). Figure 6(a) reveals numerous plate-shaped M₂C carbides and a rectangular parallelepiped carbide M₂₃C₆. Figures 6(b) and (c) show some well-developed M₂C precipitates (0.8- μ m long) and coarse



Fig. 3—TEMs for the shoulder portion of the specimen creep tested at 540 °C for 7022 h: (a) the numerous M_2C plate carbides, (b) the fine globular M_6C precipitates coexisting with M_2C precipitates, and (c) grain boundary carbide as well as rectangular $M_{23}C_6$ carbides.



Fig. 4—TEMs for the gage portion of the specimen creep tested at 540 °C for 7022 h: (a) M_2C precipitate, (b) the spheroidization tendency of coarse M_6C carbide, (c) the grain boundary carbide, (d) SAD, and (e) schematic representation of SAD identifying it as M_6C .

 M_6C (0.4- μ m diameter). Some recovery of ferrite matrix is also revealed in Figure 6(b).

The TEMs for the shoulder portion of the specimen creep tested at 580 °C for 6526 hours are given in Figures 7(a) through (d). Figure 7(a) is a low-magnification TEM revealing a partially recrystallized bainitic structure. The TEM shown in Figure 7(b) dem-

onstrates the two morphologies of carbides, rectangular parallelepiped $M_{23}C_6$ and globular (coarse as well as fine) M_6C . Figure 7(c) also shows fine globular carbides, which are identified as M_6C . The presence of numerous fine M_2C plates, 0.2- μ m long, is revealed in Figure 7(d).

The TEMs for the gage portion of the specimen creep tested at 580 °C for 6526 hours are shown in Figures 8(a)



Fig. 5—TEMs for the shoulder portion of the specimen creep tested at 540 °C for 17,946 h: (a) the well-developed M₂C plate precipitates coexisting with fine ellipsoidal precipitates, (b) M₂C plate carbide and coarse M₆C carbide, and (c) all of the features: plate-shaped carbide, fine globular carbide, and rectangular carbide.

Fig. 6—TEMs for the gage portion of the specimen creep tested at 540 °C for 17,946 h (a) M_2C carbide, (b) some M_2C plate precipitates, coarse M_6C precipitates, and recovery in the ferrite matrix, and (c) coarse M_6C and $M_{23}C_6$ parallelepipeds.



Fig. 7—TEMs for the shoulder portion of the specimen creep tested at 580 °C for 6526 h: (a) low-magnification TEM revealing the recovered bainitic structure, (b) rectangular parallelepiped-shaped $M_{23}C_6$ and globular (coarse as well as fine) M_6C carbide, (c) the globular morphology of M_6C carbides, and (d) the presence of numerous platelets of M_2C .

through (d). Figure 8(a) illustrates the coarse globular (0.3- μ m diameter) carbides as well as extremely fine plates of M₂C. Figure 8(b) reveals the M₂₃C₆ carbide in its rectangular parallelepiped morphology along with fine plates of M₂C. Figure 8(c) illustrates numerous M₂C precipitates of 0.1- μ m length. A coarse-grain boundary carbide, shown in Figure 8(d), was identified as M₆C.

The TEMs for the shoulder portion of specimen creep tested at 580 °C for 14,836 hours are shown in Figures 9(a) through (d). Figure 9(a) is a lowmagnification TEM revealing the recovered lath structure containing coarse M_6C carbides. Figure 9(b) shows a highly recovered ferrite matrix containing several subgrains and coarse M_6C carbides (0.4- μ m size). Figure 9(c) reveals the newly-formed ellipsoidal M_6C precipitates. Figure 9(d) demonstrates the existence of plate-shaped M_2C precipitates, 0.3- μ m long, in a few areas.

The TEMs for the gage portion of the specimen creep tested at 580 °C for 14,836 hours are shown in Figures 10(a) through (c). Figure 10(a) shows the recrystallized ferrite matrix and also coarse spheroids of M_6C carbides (0.6- μ m diameter), while Figure 10(b) re-

es (0.0- μ m diameter), while Figure 10(b

veals the size distribution of M_6C carbides. Figure 10(c) indicates coarse as well as fine M_6C carbides along with the $M_{23}C_6$ carbides in their rectangular parallelepiped morphology. The M_2C carbide has almost dissolved in this condition, and the ferrite matrix exhibits a recrystallized structure.

IV. DISCUSSION

A summary of the transmission electron microscopy results on 2.25Cr-1Mo steel is presented in Table III. The present investigation has clearly shown that in the normalized and tempered condition, the 2.25Cr-1Mo steel studied exhibits a fully bainitic structure and that the tempered bainite has three different types of carbides. The orthorhombic carbide, M_3C , is present in the nearly spheroidal form, the hexagonal M_2C as extremely fine and short platelets, and the face-centered cubic $M_{23}C_6$ in the shape of rectangular parallelepipeds. In addition, $M_{23}C_6$ was also present at grain boundaries.

Baker and Nutting^[1] were the first to investigate systematically the sequence of carbide precipitation during tempering of 2.25Cr-1Mo steel using extraction replicas. They also obtained similar microstructures but found that



Fig. 8—TEMs for the gage portion of the specimen creep tested at 580 °C for 6526 h: (a) coarse globular M_6C carbide as well as fine M_2C , (b) the rectangular parallelepiped $M_{23}C_6$ along with fine platelets of M_2C , (c) numerous plate precipitates and coarse M_6C , and (d) the grain boundary carbide, M_6C .

the carbides that would be present on subjecting the steel to tempering at 680 °C for 6 hours would be Fe₃C, Mo₂C, and Cr₇C₃. The morphologies of the first two carbides are similar to those observed in the present work, and that of Cr_7C_3 is similar to that of $M_{23}C_6$ observed in the present investigation.

Murphy and Branch⁽³⁾ also studied normalized and tempered 2.25Cr-1Mo steel and found evidence of M_3C and M_2C with an occasional presence of $M_{23}C_6$. They also observed that the normalized structures for different heats of 2.25Cr-1Mo steels exhibited different volume fractions of ferrite and bainite, and that as the bainite content increased, the size of cementite particles decreased. The fineness of the cementite spheroids observed in the present work is, thus, in accordance with the finding of Murphy and Branch.^[3]

Pilling and Ridley^[20] also observed rodlike precipitates which were similar in morphology to the $M_{23}C_6$ carbides observed in the present work and indicated that they could be either $M_{23}C_6$ or M_6C . They identified the fine needle carbide as Mo_2C . Baker and Nutting^[1] and Wada^[21] reported the presence of fibrous M_2C carbide in the normalized and tempered condition of 2.25Cr-1Mo steel in cases where ferrite was observed along with bainite. In the present case, the microstructure was entirely bainitic, and the absence of free ferrite is probably responsible for the absence of fibrous precipitates.

In cases where Cr₇C₃ was reported by earlier investigators, some detected its presence by streaking in electron diffraction patterns (not by obtaining reciprocal lattice reflections of this carbide).^[21] In the present investigation, we could obtain the selected area diffraction patterns from each of the morphologies of the carbides, and by this technique, we could establish the presence of M_3C , M₂C, and M₂₃C₆. Similarly, Yukitoshi and Nishida^[2] confirmed the presence of M₃C, M₂C, and M₂₃C₆ only in the normalized and tempered condition and could not get enough evidence for Cr₇C₃ and M₆C. The present results are in excellent agreement with this work. There has also been some evidence to suggest that while the carbide M_7C_3 is formed in chromium steels, the addition of molybdenum would result in an increased tendency to form $M_{23}C_6$ at the expense of M_7C_3 .^[22]

Although there have been earlier investigations on the microstructure of 2.25Cr-1Mo steel subjected to creep or long-term service exposure, no information is avail-



Fig. 9—TEMs for the shoulder portion of the sample creep tested at 580 °C for 14,836 h: (a) a low-magnification TEM revealing the recovered lath structure containing coarse carbide, (b) the mostly coarse M_6C carbides and recrystallization of ferrite matrix, (c) the newly formed ellipsoidal M_6C precipitates, and (d) the existence of M_2C plate precipitates.

able on the individual contribution of the high-temperature exposure time on the microstructure and that of the superimposed creep stress. The present investigation is aimed at isolating these effects to arrive at an understanding of the long-term tempering effects and the role of creep stress in altering these microstructural variations.

The present work shows that aging the 2.25Cr-1Mo steel at 540 °C for 7022 hours considerably coarsened the M₂C carbide, and it increased in length from about 0.05 μ m in the normalized and tempered condition to 1.5 μ m after 17,946 hours of exposure at 540 °C. The superimposed creep stress surprisingly restricted the length of these platelets to 0.8 μ m; however, these platelets are more profuse in the latter case. Another effect of 540 °C aging is that the M₃C carbide dissolved in the matrix, and in its place, M₆C carbide was observed as globular carbide. There was no noticeable change in M23C6 carbide occurring as rectangular parallelepipeds either in the shoulder or gage section of the specimens tested at 540 °C up to 17,946 hours. The ferrite matrix showed some signs of recovery only in specimens exposed for 17,946 hours, with or without the creep stress. Interestingly, the length of the M_2C platelets exposed at 580 °C is much less than the corresponding ones at 540 °C, both in the shoulder portion and the gage section of the steel.

While the M_2C plates grew up to 0.30 μ m long after 14,836 hours aging at 580 °C, this carbide is virtually absent when 78 MPa stress is superimposed on it. The M_6C carbide became more prominent, both within the ferrite matrix and at the grain boundaries, after long exposure with or without the creep stress. The onset of recrystallization is clearly visible, the extent of which has become more prominent in specimens subjected to creep stress at 580 °C for 14,836 hours.

Murphy and Branch^[3] carried out a detailed investigation of the microstructural changes taking place during the creep of 2.25Cr-1Mo steel. They reported that the M₃C carbide rapidly spheroidizes and begins to transform to $M_{23}C_6$ and M_6C . The present results are in agreement with this finding, as M₃C is observed only in the normalized and tempered condition and not in any other creep-tested sample, either in the shoulder or the gage section. Murphy and Branch^[3] also reported that M₂C carbides would dissolve gradually in the matrix during creep testing at 566 °C for 15,000 hours. The present investigation reveals that the M2C carbide grows to about 1.5 μ m at 540 °C after 17,946 hours, and at 580 °C, it would grow initially in size before the dissolution process starts. The size of the M₂C carbide came down to 0.3 μ m after 14,836 hours at 580 °C in the shoulder



Fig. 10—TEMs for the gage portion of the steel creep tested at 580 °C for 14,836 h: (a) the recrystallized ferrite matrix and the coarse spheroids of M_6C carbides, (b) the size distribution of M_6C carbides, and (c) the coarse as well as fine M_6C carbides along with $M_{23}C_6$ carbides in their rectangular parallelepiped morphology.

Temperature (°C)	Specimen		Ferrite			
(Rupture time, h)	Location	M ₃ C	M ₂ C	M ₂₃ C ₆	M ₆ C	Matrix
As heat treated	_	globular, 0.15 μm	fine platelets 0.05-µm long	rectangular parallelepiped, 0.50-µm long	absent	tempered bain tic structure
540 (7022)	S	absent	0.50 - μ m long	0.70 μm	globular, 0.20 µm	tempered bainitic structure
540 (7022)	G	absent	0.40-µm long	1.0 μm	0.30 μm	tempered bainitic structure
540 (17,946)	S	absent	1.50-µm long	0.80 µm	0.25 μm	α reveals ten- dency to recover
540 (17,946)	G	absent	0.80-µm long	0.80 µm	0.40 µm	α reveals ten- dency to recover
580 (6526)	S	absent	0.20-µm long	0.80 µm	0.30 µm	α reveals ten- dency to recover
580 (6526)	G	absent	0.10-µm long	0.80 µm	0.30 µm	α reveals ten- dency to recover
580 (14,836)	S	absent	0.30-µm long in a very few areas	0.80 µm	0.40 μm	α recovery is substantial
580 (14,836)	G	absent	absent	0.60 µm	0.60 µm	α recrystallize

 Table III.
 Summary of Transmission Electron Microscopy Results on 2.25Cr-Mo Steel

portion, and the corresponding gage section was totally free from this carbide.

The recovery and recrystallization process was also more prominent in the gage section than the shoulder when tested at 580 °C, suggesting that creep stress exerted a significant increase in the rate of recovery and recrystallization of bainitic ferrite.

Wada^[21] has reported that the only carbides observed in a 2.25Cr-1Mo steel after long-term exposure at 600 °C to 650 °C were M_6C and $M_{23}C_6$. He also reported that service exposure for 12 years in the lower temperature range of 500 °C to 570 °C resulted in the growth of needlelike carbides. Abdul-Latif et al.[10] used the accelerated aging at 630 °C for various times to produce microstructures which simulate 25,000 to 100,000 hours of service exposure at 540 °C. They reported a lower volume fraction of acicular carbide as compared with the as-received material. They found that the general trend was an increase in molybdenum content of the carbide as the service duration was increased. The majority of the carbides have face-centered cubic structure which was either $M_{23}C_6$ or M_6C , and they indicated that the former carbide has the alloying elements in the ratio of 65Fe-23Cr-12Mo, compared to 20-Fe-3Cr-77Mo for the M₆C precipitates. They reported that the acicular carbides were not M_2C but either $M_{23}C_6$ or M_6C . Clear evidence has been obtained in the present work by selected area diffraction that the acicular carbides are, in fact, M₂C of hexagonal symmetry. It should, however, be understood that carbide structures produced at 630 °C may not be identical to those produced by long-term aging at a lower temperature, like 540 °C or 580 °C used in the present work, even if the morphologies are similar.

IV. CONCLUSIONS

The transmission electron microscopy studies of the heat-treated and stress-rupture samples of 2.25Cr-1Mo steel, tested in the temperature range of 540 °C to 580 °C, have revealed the following:

- 1. In the normalized and tempered condition, carbides occur in three morphologies: (1) plate, M_2C , (2) globular, M₃C, and (3) rectangular parallelepiped, M₂₃C₆.
- 2. Aging the steel at 540 °C for 7022 and 17,946 hours resulted in a lengthening of the M₂C platelets and led to the dissolution of M_3C and precipitation of M_6C . The superimposed creep stress does not seem to have much of an effect in short-term (7022 hour) testing at 540 °C, but in the long-term (17,946 hour) test, it resulted in the reduction of the length of M₂C but increased its density.
- 3. Aging at 580 °C for 6526 hours resulted in a slight lengthening of M₂C, but holding at 580 °C for 14,836 hours resulted in the disappearance of the M_2C plates in most of the areas. The superimposed creep stress resulted in the dissolution of M₂C precipitates after 14,836 hours.
- 4. The superimposition of 78 MPa stress on the steel at 580 °C for 14,836 hours promoted the formation of

M₆C and resulted in significant recovery and recrystallization of the ferrite matrix.

5. The rectangular parallelepiped carbide, identified as M₂₃C₆, in the normalized and tempered condition, did not change morphology under any test conditions studied in the present work.

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