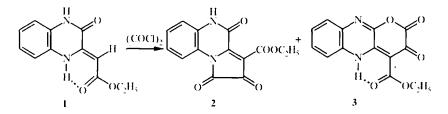
UNUSUAL REACTION OF A HETEROCYCLIC ENAMINE WITH OXALYL CHLORIDE

A. N. Maslivets, O. V. Golovnina, O. P. Krasnykh, and Z. G. Aliev

Keywords: heterocyclic enamine; 2,3-dihydro-2,3-pyrroledione; 2-quinoxalone; pyrrolo[1,2-*a*]-quinoxaline-1,2,4-trione; pyrano[2,3-*b*]quinoxaline-2,3-dione.

The reaction of primary enamines with oxalyl chloride is the most common method for synthesis of substituted 2,3-dihydro-2,3-pyrrolediones [1]. The corresponding substituted 2,3-dihydro-2,3-pyrrolediones, annelated with the azaheterocycles by side [*a*], have been obtained by this method in practically quantitative yield from heterocyclic enamines such as substituted 1-methylene-1,2,3,4-tetrahydroisoquinolines [2], 3-methylene-3,4-dihydro-2H-1,4-benzoxazin-2-ones [3], and 2-methylene-3,4-dihydro-2H-1,3-benzoxazin-4-ones [4]. The isolation and identification of other products was not reported.

Reaction of (Z)-3-ethoxycarbonylmethylene-1,2,3,4-tetrahydro-2-quinoxalone (1) with oxalyl chloride formed the expected 3-ethoxycarbonyl-1,2,4,5-tetrahydro-pyrrolo[1,2-a]quinoxaline-1,2,4-trione (2) and in addition the unexpected 4-ethoxycarbonyl-3,5-dihydro-2H-pyrano[2,3-b]quinoxaline-2,3-dione (3) was formed as a minor product. Compound 3 was identified by X-ray structural analysis.



Probably a small quantity (not identifiable from ¹H NMR spectral data) of the tautomeric hydroxyimine form of **1**, *viz*. 2-ethoxycarbonylmethylene-3-hydroxy-1,2-dihydroquinoxaline, is present in the solution of the initial quinoxalone **1**. This affords the possibility for acylation of the vinyl CH group and the hydroxy imine OH group. An alternative route of forming compound **3** is amide--hydroxy imine isomerization in the course of the reaction. Analogs of quinoxaline 1, *viz*. (*Z*)-3-phenacylidene-1,2,3,4-tetrahydro-2-quinoxalones, under analogous conditions readily form 3-aroyl-1,2,4,5-tetrahydropyrrolo[1,2-*a*]quinoxaline-1,2,4-triones [1], but the possibility of the alternative acylation described above does not occur. The given reaction is a new method for construction of the poorly available condensed pyrano[2,3-*b*]quinoxaline system.

EXPERIMENTAL

3-Ethoxycarbonyl-1,2,4,5-tetrahydropyrrolo[1,2-*a*]quinoxaline-1,2,4-trione (2) and 4-Ethoxycarbonyl-**3,5-dihydro-2H-pyrano**[2,3-*b*]quinoxaline-2,3-dione (3). A solution of oxalyl chloride (1.37 ml, 16.1 mmol) in absolute chloroform (2 ml) was poured into a solution of quinoxalone 1 (3.60 g, 15.5 mmol) in absolute chloroform

Perm State University, Perm 614000, Russia; e-mail: info@psu.ru. Translated from Khimiya Geterotsiklicheskikh Soedinenii, No. 1, pp. 113-114, January, 2000. Original article submitted November 18, 1999.

0009-3122/00/3601-0105\$25.00©2000 KluwerAcademic/Plenum Publishers

(50 ml). The mixture was boiled for 55 min, cooled, and the precipitated solid compound **3** was filtered off. Yield 0.67 g (15%); mp 194-196°C (MeCN). IR spectrum: 3060 (N–H), 1795 ($C_{(2)}$ =O), 1661 cm⁻¹ ($C_{(3)}$ =O, COOC₂H₃). ¹H NMR spectrum (MeCN-d₃): 1.32 (3H, t, *J* = 7.0 Hz, CH₃); 4.37 (2H, q, *J* = 7.0 Hz, CH₂O); 7.72 (4H, m, C₆H₄); 13.90 ppm (1H, s, NH). Mass spectrum, *m/z*: 286 [M]⁻¹.

The solvent (40 ml) was distilled from the mother liquor, and the precipitated solid compound **2** was filtered off. Yield 2.70 g (60%); mp 179-181°C (decomp., from MeCN). IR spectrum: 3122 (N–H), 178 (C_{c1} =O), 1740 (COO), 1720 (C_{c2} =O), 1672 cm⁻¹ (C_{c4} =O). ¹H NMR spectrum (MeCN-d₃): 1.34 (3H, t, *J* = 7.0 Hz, CH₃); 4.37 (2H, q, *J* = 7.0 Hz, CH₃O); 7.43 (4H, m, $C_{c}H_{4}$); 10.60 ppm (1H, s, NH).

The data of elemental analysis corresponded to the calculated values.

The work was carried out with the financial support of the Russian Fund for Fundamental Investigations (grant No. 98-03-32888a).

REFERENCES

- 1. Yu. S. Andreichikov (editor), *Chemistry of Five-membered 2,3-Dioxoheterocycles*, Izd. Perm. Univ., Perm (1994), p. 91.
- 2. T. Sano, J. Toda, N. Maehara, and Y. Tsuda, *Canad. J. Chem.*, 65, 94 (1987).
- 3. A. N. Maslivets, I. V. Mashevskaya, O. P. Krasnykh, S. N. Shurov, and Yu. S. Andreichikov, *Zh. Org. Khim.*, **28**, 2545 (1992).
- 4. G. Kollenz, R. Theuer, W. Ott, and E. Ziegler, *Liebigs Ann. Chem.*, 1964 (1977).