

## Template-Catalyzed Oligomerization with an Atactic Glycerol-Based Polynucleotide Analog

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**Summary.** We have tested the effect of a nonstereoregular (atactic) template, based on the acyclic nucleotide analog pCp, on the oligomerization of the complementary bis-phosphoimidazolidone monomer, ImpGpIm. Although there is little catalysis when the template is added together with magnesium ions, the substitution of manganese for magnesium produces a substantial increase in the extent of oligomerization as well as increasing the effect of the template. These results are discussed in relation to the hypothesis that prochiral nucleotide analogs may have been precursors to the first RNA molecules.

**Key words:** Chemical evolution — Nucleic acid analogs — Oligonucleotides — Template-directed synthesis

### Introduction

The problems inherent in the synthesis of the enantiomerically pure nucleosides, which would appear to be necessary for the first self-replicating RNA molecules to arise, have led to suggestions that RNA may have been preceded by a more primitive set of

autocatalytic molecules (Joyce et al. 1987). These precursors, although retaining the base-pairing properties of nucleic acids, might have had simpler backbone structures. Prochiral nucleotide analogs of the kind shown in Fig. 1 have received considerable attention in this context. A central point is that although the monomers are not themselves chiral, a possibility for the generation of chirality exists as a result of polymerization, as each residue can assume a D-like or an L-like configuration (Spach 1984; Joyce et al. 1987). In earlier work, it has been demonstrated that the oligomerizations of bisphosphoimidazolidones of G̃ and Ã are catalyzed by the complementary polyribonucleotide templates, producing pyrophosphate-linked polynucleotide analogs (Visscher and Schwartz 1988). We have now synthesized a nonstereoregular (i.e., atactic) set of pyrophosphate-linked oligomers based on pCp and examined the effect of these products on the oligomerization of ImpGpIm.

### Materials and Methods

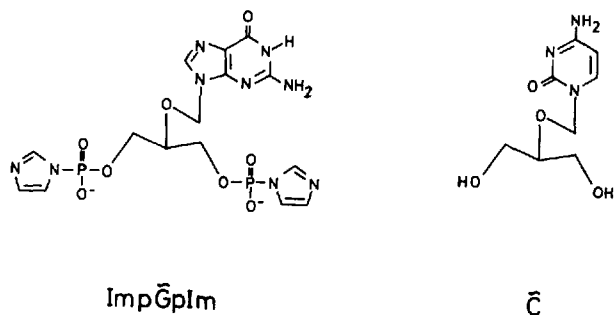
Alkaline phosphatase (type III from *Escherichia coli*) was purchased from Sigma Chemical Company. Phosphodiesterase I from *Crotalus adamanteus* venom was purchased from P-L Biochemicals. EDAC was purchased from Janssen Chimica.

The preparation of 1-[(1,3-dihydroxy-2-propoxy)methyl]cytosine (C̃) and the bis-phosphorylation of C̃ were accomplished as described in Visscher and Schwartz (1990). The synthesis of 9-[(1,3-dihydroxy-2-propoxy)methyl]guanine (G̃) and the bis-phosphorylation of G̃ were described in Visscher and Schwartz (1988).

The cyclic pyrophosphate of pCp was prepared by treating 0.1 mmol pCp with 1.0 mmol EDAC and 0.4 mmol MgCl<sub>2</sub> in 4.0 ml 0.4 M imidazole (pH 6.5 with HCl). After 1 h at 40°C the reaction was 95% complete. Purification was accomplished by chromatography on DEAE-Sephadex (A25, Pharmacia) with a

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**Abbreviations:** Ã, 9-[(1,3-dihydroxy-2-propoxy)methyl]adenine; G̃, 9-[(1,3-dihydroxy-2-propoxy)methyl]guanine; ImpGpIm, the bisphosphoimidazolidone of G̃; C̃, 1-[(1,3-dihydroxy-2-propoxy)methyl]cytosine; pCp, the bisphosphate of C̃; cpCp, the cyclic pyrophosphate formed from pCp; CTP, cytidine 5'-triphosphate; EDAC, 1-ethyl-3-(3-dimethylaminopropyl)carbodiimide hydrochloride; EDTA, ethylenediamine tetraacetic acid; TEAB, triethylammonium bicarbonate; Tris, tris(hydroxymethyl)aminomethane

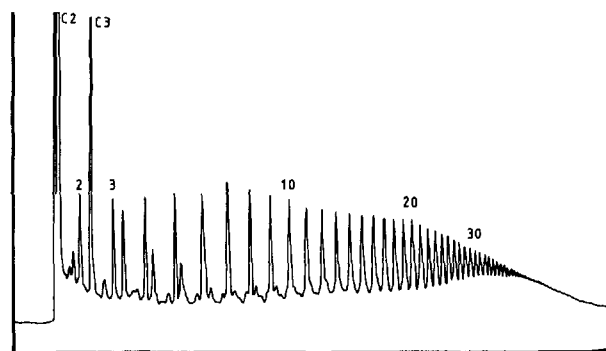


**Fig. 1.** Structures of the glycerol-based analogs  $\tilde{G}$  and Imp $\tilde{G}$ pIm (the bisphosphoimidazolide of  $\tilde{G}$ )

linear gradient of 0.1–0.4 M TEAB at pH 8. The cyclic product was identified by enzymatic analysis, as described by Schwartz et al. (1987), and by  $^{31}\text{P}$ -NMR ( $^{31}\text{P}$ -NMR shifts relative to phosphoric acid in  $\text{D}_2\text{O}$  were:  $\text{p}\tilde{\text{Cp}}$ , 2.8 ppm, singlet;  $\text{cp}\tilde{\text{Cp}}$ ,  $-9.8$  ppm, singlet). The cyclic pyrophosphate was resistant to the action of bacterial alkaline phosphatase. Treatment with venom phosphodiesterase resulted in the production of  $\text{p}\tilde{\text{Cp}}$ .

The polymerization of  $\text{cp}\tilde{\text{Cp}}$  was carried out following Tohidi and Orgel (1990). Samples of  $1\ \mu\text{mol}$   $\text{cp}\tilde{\text{Cp}}$  (as triethylamine salt), each dissolved in  $10\ \mu\text{l}$  0.1 M imidazole (pH 6.5 with HCl), were spotted on a glass plate and incubated over  $\text{P}_2\text{O}_5$  at  $83^\circ\text{C}$ . After 3 days more than 70% of the cyclic-pyrophosphate had reacted to form oligomers and polymers. Analysis by HPLC (see below) showed that the length of the polymers extended to about 60 monomer-units. A collection of 100 reactions was treated with 10 units of alkaline phosphatase in 3 ml 0.04 M tris-HCl (pH 8) containing 0.02 M  $\text{MgCl}_2$ , to remove terminal phosphate groups. After incubation for 4 h at  $37^\circ\text{C}$ , oligomers with chain lengths  $\geq 20$  monomer-units were isolated by fractionating on Q-Sepharose (Fast Flow, Pharmacia) in a linear gradient of 0.1 M to 1.5 M TEAB (pH 8). The identity of the products as oligomers of  $\text{p}\tilde{\text{Cp}}$  was confirmed by enzymic digestion with phosphodiesterase combined with HPLC, and by  $^{31}\text{P}$ -NMR ( $^{31}\text{P}$ -NMR shift relative to phosphoric acid in  $\text{D}_2\text{O}$  was  $-9.0$  ppm, singlet).

The in situ bis-imidazolization of  $\text{p}\tilde{\text{Cp}}$  was carried out by treating 0.1 M  $\text{p}\tilde{\text{Cp}}$  with 10 equivalents each of imidazole and EDAC at pH 6.5 and  $0^\circ\text{C}$  for 2 h as described in Visscher and Schwartz (1988). Reaction mixtures were prepared by adding  $2.5\ \mu\text{l}$  of the



**Fig. 2.** HPLC analysis on RPC-5 of the products of the polymerization of  $\text{cp}\tilde{\text{Cp}}$ . C2 and C3 show the positions of cyclic dimer and trimer, respectively.

imidazolating mixture to  $10\ \mu\text{mol}$  NaCl, 0.5 or  $2\ \mu\text{mol}$   $\text{MgCl}_2$  or  $\text{MnCl}_2$  and, when required, 0.25  $\mu\text{mol}$  of the template (monomer equivalents) in a total volume of  $10\ \mu\text{l}$ . The final reaction conditions are given in Table 1. After a reaction period of 7 days at  $0^\circ\text{C}$  the reaction mixtures were quenched by adding  $4\ \mu\text{l}$  of EDTA (1.0 M, pH 9) and water to a total volume of  $30\ \mu\text{l}$ . Storage was at  $-25^\circ\text{C}$ . Prior to analysis any surviving imidazolides were hydrolyzed by adding  $1\ \mu\text{l}$  of the quenched reaction mixture to  $50\ \mu\text{l}$  1.0 M NaOH and incubating at  $37^\circ\text{C}$  for 4 h. The sample was then diluted with water to 1 ml. Analyses were performed by HPLC on RPC-5 in 0.02 M NaOH with a linear gradient of  $\text{NaClO}_4$  (0–0.04 M over 60 min) at a flow rate of 1 ml/min. Peak detection was by absorbance monitoring at 254 nm. The cyclic-dimer, dimer, and tetramer of  $\text{p}\tilde{\text{Cp}}$  were identified by isolating these products from the RPC-5 column and carrying out sequential degradation with phosphodiesterase as described in Visscher and Schwartz (1988).

## Results and Discussion

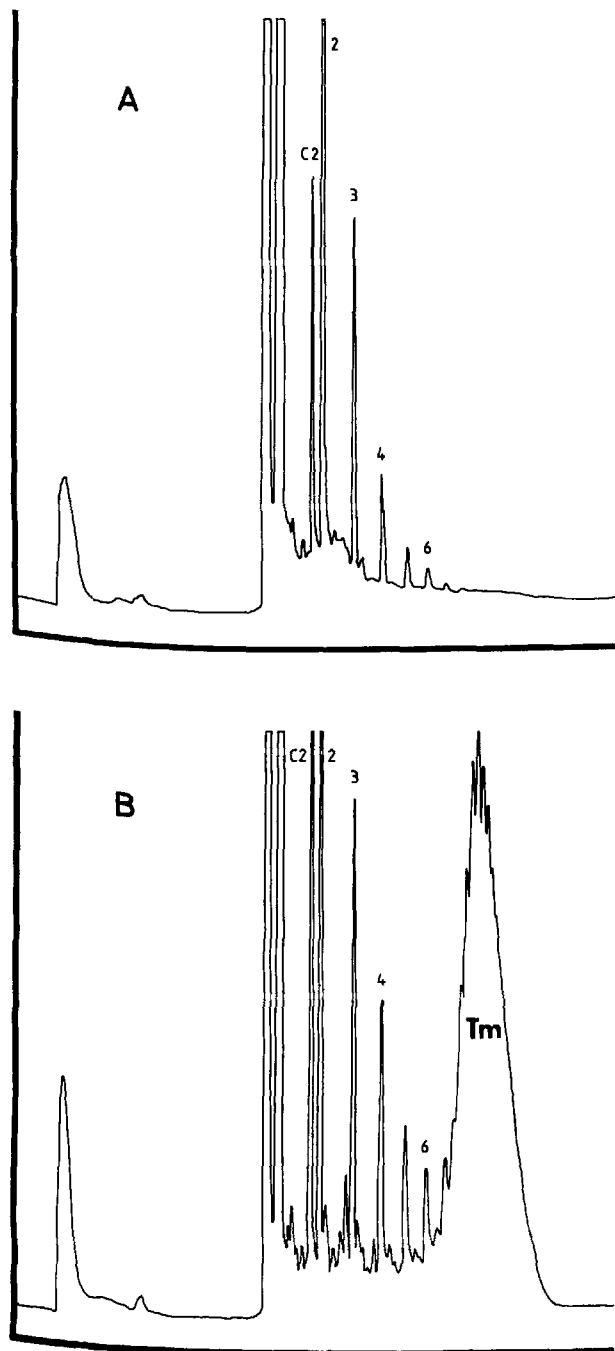
The range of oligomers obtained from the thermal polymerization of  $\text{cp}\tilde{\text{Cp}}$  is illustrated in Fig. 2. The products extended to at least the 60-mer. After removal of terminal phosphates and isolation of those oligomers with chain lengths of 20 and more, we carried out the experiments summarized in Table 1. In previous work on the catalysis of the oligomerization of nucleotide analogs based on deoxyribose bisphosphates, we demonstrated that the substitution of pyrophosphate for phosphodiester linkages between deoxyribose units does not destroy the ability of the polynucleotide analog to catalyze the synthesis of its complement (Visscher et al., 1989). In contrast to the earlier results, we observed only a marginal effect of the glycerol-based template on the oligomerization in the presence of magnesium (Table 1). Attempts to reduce the effects of charge repulsion between the pyrophosphate chains by the addition of spermine or spermidine to the system produced no improvement in the yields.

We also studied the effect of substituting manganese for magnesium in the reaction. We had pre-

**Table 1.** Product distributions in the oligomerization of Imp $\tilde{G}$ pIm

$\tilde{\text{Cp}}(\text{p}\tilde{\text{Cp}})_n\text{p}\tilde{\text{C}}$ (with $n \geq 18$ ) template (M)	Mag- nesium (M)	Man- ganes (M)	Incorporation of monomer into oligomers of length $n$ (%)		
			$n \geq 2$	$n \geq 3$	$n \geq 4$
0.025	0.2	—	6	1	Trace
—	0.2	—	3	—	—
0.025	0.05	—	3	—	—
—	0.05	—	1	—	—
0.025	—	0.2	27	13	6
—	—	0.2	18	6	2
0.025	—	0.05	16	8	4
—	—	0.05	9	3	1

Conditions: 0.025 M monomer, 1.0 M NaCl, metal chloride as shown and, when required, 0.025 M template [monomer equivalents of  $\tilde{\text{Cp}}(\text{p}\tilde{\text{Cp}})_n\text{p}\tilde{\text{C}}$ ]. Reactions were at pH 6.5 and  $0^\circ\text{C}$  for 7 days. For further details see text



**Fig. 3.** Oligomerization of ImpGpIm. **A** 0.025 M ImpGpIm, 0.05 M MnCl<sub>2</sub>, 1.0 M NaCl, pH 6.5. (Reaction was at 0°C for 7 days. For exact conditions see Table 1 and text.) **B** As A, but with the addition of 0.025 M template [monomer equivalents of Cp(pCp)<sub>n</sub>pC, n ≥ 18]. The position of the template mixture is indicated by Tm. Control experiments established that no spurious peaks were produced by degradation of the template under the conditions of the reaction and analysis.

viously reported that although manganese has a favorable effect on the oligomerization of ImpGpIm in the absence of a template, it appeared to reduce the catalytic effect of a polyribonucleotide template (Visscher and Schwartz 1989). Somewhat to our sur-

prise, therefore, we found that there was a substantial increase in oligomerization with manganese, both in the presence and absence of the template. It is possible that this difference may be due to an interaction of Mn(II) with the pyrophosphate groups of the analog template. Recent studies of coordination complexes between Mn(II) and CTP in water suggest that the metal coordinates with the phosphates of CTP rather than with the cytosine ring-system (Sigel et al. 1987). This type of interaction, which is stronger for Mn(II) than for Mg(II), may help to stabilize the catalytic complex between the pyrophosphate-linked template and monomer plus products, without interfering with base-pairing by cytosine.

The presence of an atactic template increased the total yield of oligomers from 18 to 27% in 0.2 M MnCl<sub>2</sub> and from 9 to 16% in 0.05 M MnCl<sub>2</sub> (Table 1). Under the more dilute conditions, the effect of the template was more obvious. As can be seen in Fig. 3, the longer oligomers (4–6-mers) are most noticeably affected. Although modest in terms of yields, these results essentially confirm the prediction by Joyce et al. (1987) that the flexibility of a glycerol-based pyrophosphate-linked polynucleotide analog might be sufficient to permit chain elongation even when the individual units of the polymer are partially atactic. The increase in the height of the cyclic dimer peak in the presence of the template, however, may indicate that chain elongation becomes more difficult with increasing lengths of products. This could reflect the statistics of distribution of isotactic segments of the template along the chain. It is therefore not yet clear whether chain extension is actually occurring on an atactic template or is dependent on the existence of short isotactic regions and chain sliding. It will be of interest to compare these results with those obtained with a fully isotactic template.

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