

Metallogenesis of cratonic oolitic ironstone deposits in the Bled el Mass, Azzel Matti, Ahnet and Mouydir basins, Central Sahara, Algeria

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With 12 figures and 8 tables

Zusammenfassung

Die Gebiete des Bled el Mass, des Azzel Matti, des Ahnet und des Mouydir, liegen im Nordwesten des Touaregschildes (Zentral Sahara, Algerien). Neun eisenoolithische Horizonte sind hier in die devonischen Sedimente eingeschaltet. Ihr Mineralinhalt kann durch vier verschiedene Paragenesen charakterisiert werden:

P1 = Chamosit + Magnetit + Maghemit + Goethit; P2 = Chamosit + Hämatit + Goethit + Kalzit; P3 = Chamosit + Hämatit + Goethit + Quarz; P4 = Chamosit + Hämatit + Goethit.

Vier vererzte Fazies Typen treten auf: FOD: (die Ooide sind in einer detritischen Matrix eingelagert); FOND: (die Ooide sind in einer nichtdetritischen Matrix eingelagert); FOC: (die Ooide sind verfestigt); FMC: mikrokonglomeratische Fazies.

Die Ooide entwickeln sich in ruhigen Bedingungen, in Lagunen oder Meerbusen, durch die Bildung von Konkretionen im Sediment aus einem silikatischen und eisenreichen Schlamm; sie werden dann wie detritische Komponenten aufgenommen und transportiert.

Die eisenoolithischen Ablagerungen scheinen durch mehrere Sedimentationsphasen während des Devons entstanden zu sein; so können sie als Zeichen von Sedimentmikrozuklen betrachtet werden. Diese oolithischen Sedimente sind charakteristisch für eine Kratonsedimentation am Rand eines breiten, epikontinentalen Meeres.

Der Ursprung des Eisens ist in einem südlichen Kontinent zu suchen, wahrscheinlich in dem mobilen panafrikanischen Gebirge von Nigeria, oder auf dem kongolesischen Schild.

Diese Erze der Zentralsahara können als ein wichtiger Zweig des Eisengürtels betrachtet werden, der sich in Nordafrika von Rio de Oro bis Libyen erstreckt.

Abstract

The Bled el Mass, Azzel Matti, Ahnet and Mouydir areas are located in the northwest of the Touareg Shield (Central Sahara, Algeria). Within the Devonian sedimentary forma-

tions, nine oolitic ironstone occurrences of EXID type (Extensive Ironstone Deposition) are interbedded.

Their mineralogical composition is characterized by four different paragenetic associations: P1 (Chamosite – magnetite – maghemite – goethite); P2 (chamosite – hematite – goethite – calcite); P3 (chamosite – hematite – goethite – quartz); and P4 (chamosite – hematite – goethite). Using textural analysis, four main ironstone facies are distinguished: FOD (ooliths scattered in a detrital groundmass); FOND (ooliths scattered in a non detrital groundmass); FOC (cemented ooliths) and FMC (microconglomeratic facies).

Primarily developed in calm conditions by intrasedimentary processes within an iron-rich silicated mud, in lagoons or embayments, ooliths subsequently acquired a detrital character.

The ironstone deposition seems to be induced by several pulses of sedimentation through the Devonian and is considered as indicator of sedimentary subcycles. Therefore, the oolitic ferriferous sediments indicate a cratonic sedimentation on the borders of a large epicontinental sea. The source of the iron could be a remote southern continent, probably the Pan-African mobile belt of Nigeria and the Congo Shield.

The ironstones of the Central Sahara can be considered as an important branch of the North-African Oolitic Ironstone Belt, extending from Rio de Oro to Libya.

Résumé

Les régions du Bled el Mass, de l'Azzel Matti, de l'Ahnet et du Mouydir sont situées au Nord-Ouest du Bouclier Touareg (Sahara Central, Algérie). Neuf niveaux de minerai de fer oolithique sont interstratifiés dans les formations sédimentaires du Dévonien.

Quatre différentes paragenèses caractérisent la composition minéralogique de ces minerais. P1 (chamosite – magnétite – maghémite – goéthite); P2 (chamosite – hématite – goéthite – calcite); P3 (chamosite – hématite – goéthite – quartz) et P4 (chamosite – hématite – goéthite).

Quatre faciès minéralisés ont été en évidence: FOD (oolithes dispersées dans une matrice détritique); FOND (oolithes dispersées dans une matrice non détritique); FOC (oolithes cimentées) et FMC (faciès microconglomératique).

Développées dans des conditions calmes par concrétionnement intrasédimentaire dans une boue silicatée riche en

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fer, dans des lagons ou des baies, les oolithes vont acquérir par la suite un comportement détritique.

Les dépôts de minéral oolithique semblent avoir été induits par plusieurs pulsations sédimentaires durant le Dévonien et peuvent être ainsi considérés comme des marqueurs de microcycles sédimentaires.

Ces sédiments oolithiques sont caractéristiques d'une sédimentation cratonique, sur les bords d'une mer épicontinentale étendue. La source du fer est à rechercher dans un continent situé au Sud, probablement dans la chaîne mobile Pan-Africaine du Nigeria et le Bouclier du Congo.

Ces minerais de fer oolithiques du Sahara Central peuvent être considérés comme une branche importante de la ceinture ferrifère Nord africaine, qui s'étend du Rio de Oro à la Libye.

Краткое содержание

На северо-западе туарегского щита (центральная Сахара, Алжир) располагаются области Bled el Mass, Azzel Mati, Ahnet и Mouydir. Здесь в девонских осадках установлено 9 горизонтов, содержащих железистые оолиты. Для них являются четыре следующих парагенеза:

П₁ = шамозит - магнетит - маггемит - гётит; П₂ = шамозит - гематит - гётит - кальцит; П₃ = шамозит - гематит - гётит - кварц; П₄ = шамозит - гематит - гётит.

Эти 4 типа фаций можно описать так: DFOD - оолиты заключены в матрице из детритного материала; FOND - оолиты заключены в матрице, составленной из не детритного материала; FOC - оолиты спечены; FMC - фаций микроконгломератов.

Оолиты образовались в условиях покоя, в лагунах, или морских бухтах, в результате появления конкреций в осадке из силикатного и богатого железом ила; они стали затем детритом. Вероятно, отложения железистых оолитов явилось результатом повторных стадий седиментации во время девонского периода. Их можно рассматривать, как отменки в циклах отложения. Эти оолитические осадки являются характерными для седиментации у кратонов на краю широкого эпиконтинентального моря. Железо, возможно, происходит из южной части материка, вероятно, из подвижных панафриканских гор Нигерии, или щита Конго - эти руды центральной Сахары могут представлять собой важное ответвление железистого пояса, простирающегося в северной Африке от Rio de Oro до Ливии.

Introduction and geological framework

The Algerian Sahara is located between 7°30' W and 10° E longitude and from 33° N to 19° N latitude. Two Precambrian shields outcrop in the southern part (Tuareg Shield) and western part (Reguibat Shield) of this desert area. Paleozoic sediments form

a narrow but important belt, 2000 km long, dipping gently northward, and containing fluvial and marine formations.

On the northern part of the Ahaggar (or Touareg) Shield, they constitute the Tassilian Belt (KILIAN, 1922; FOLLOT, 1952; BEUF et al., 1971) composed of the Ahnet, Mouydir and Tassilis N'Ajjer regions (Fig. 1). On the north-western border of the Ahnet and of the Mouydir, the Azzel Matti and the Bled el Mass structures occur.

The four areas: Bled el Mass, Azzel Matti, Ahnet and Mouydir contain numerous ironstone occurrences interbedded within the marine Devonian sediments and are spread over about 150000 sq. kilometers.

Overlying Cambro-Ordovician continental formations and Silurian marine sediments (both about 600 meters thick), the Devonian sediments present thickness and facies variations trending from the north-west to the south-east (Fig. 2).

In the north, the Bled el Mass area is dominated by marine sediments (250 meters thick) which are Lochkovian, Pragjan and Emsian in age, are referred to as the Zeimlet, Saheb el Djir and Dkhissa formations and are composed of argillaceous, calcareous and sandy materials, respectively (BOROLLOT et al., 1955; LEGRAND, 1965, 1967; FABRE, 1976).

In the south, the Lower Devonian outcrops in a sinuous cliff called the Outer Tassilis (Biju-Duval et al., 1968). The arenaceous sediments of Asedjrad and Oued Samène Formations (BEUF et al., 1971) indicate a sedimentary zonation in the elementary units from west to east. The unconformities described in the Lower Devonian Ajjers Tassilis (DUBOIS et al., 1967; BEUF et al., 1968, 1970; BENNACEF et al., 1971) have been found again in Ahnet and Mouydir Outer Tassilis.

Marine and continental conditions alternated during sedimentation with the major part of the Lower Devonian consisting of fluvial sandstones. The scattering of palaeocurrent directions suggests that stream discharge of meandering rivers has been irregular, probably seasonal, but always northwest trending. On the other hand, an increase of marine conditions is clearly visible in sediments deposited in the northwest, particularly during the Emsian which was the starting point of a new major transgressive cycle. At that time, the argillaceous-sandy-calcareous sediments of the Azzel Matti Formation were deposited in Azzel Matti and Bled el Mass.

The Middle Devonian commenced with the large Eifelian transgression during which 20-60 meters of shales and limestones of the Meredoua Formation were deposited.

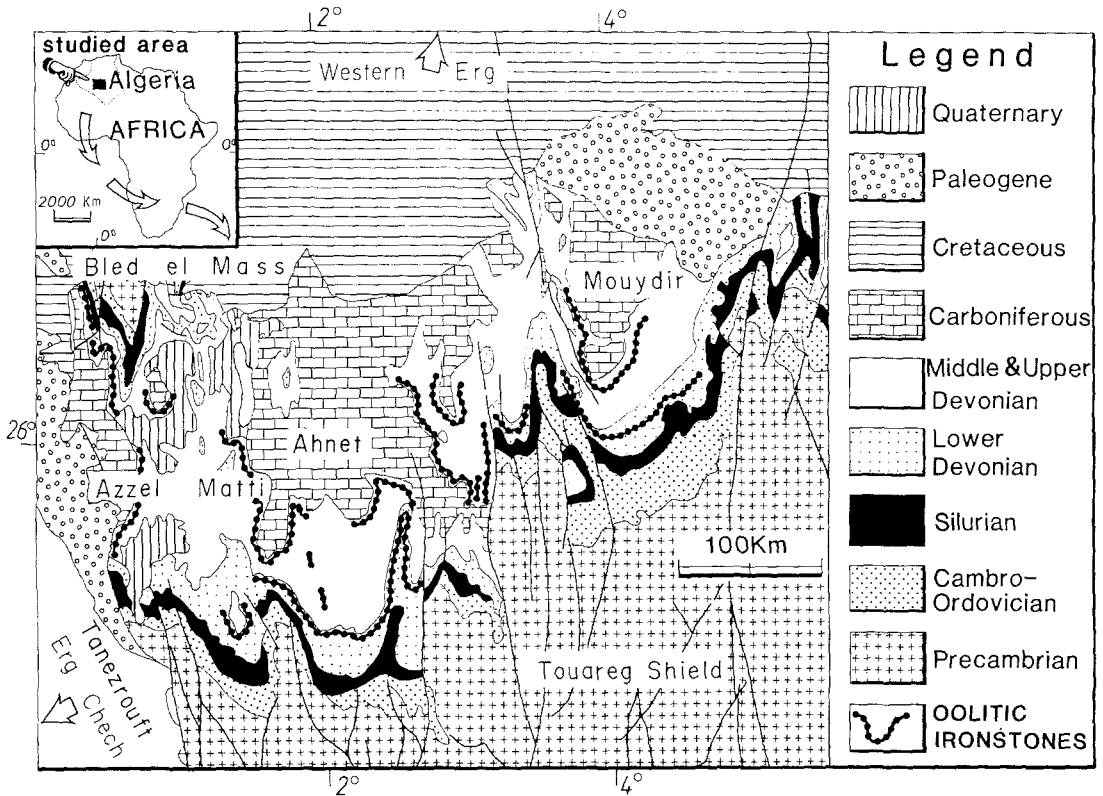


Fig. 1. Geological map of the study area.

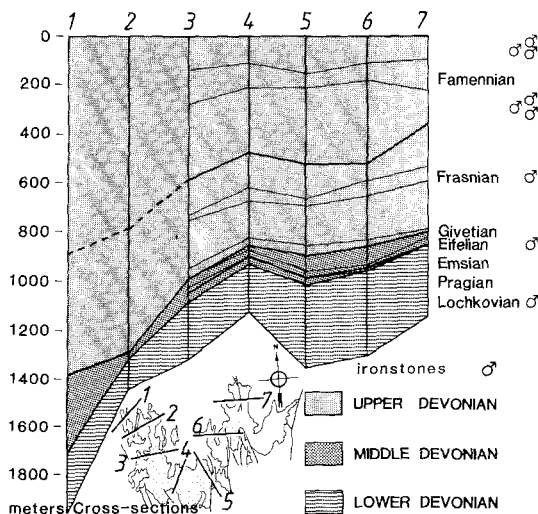


Fig. 2. Variation of Devonian thicknesses from Bled el Mass to Mouydir area (after unpublished data from A. MOUSSINE-POUCHKINE).

The subsequent transgression, Givetian in age, deposited 40–100 meters thick sediments constituting the Takoula and In Heguis Formations. Mainly calcareous to the north with numerous reefs (Azzel Matti), they become argillaceous and silty southeastward. These reefs, located directly above old basement faults (MOUSSINE-POUCHKINE, 1972) occur within an argillaceous environment.

The Frasnian (200–300 meters) is subdivided into three successive formations: Fom Agam, Imeragen and Meden Yahya (MOUSSINE-POUCHKINE, pers. comm.). From bottom to top, calcareous sedimentation decreases and arenaceous deposits increase.

The Early Famennian transgression saw the deposition of 600–700 meters thick of the argillaceous El Ouatiya Formation followed by the silty El Aricha Formation (MOUSSINE-POUCHKINE, pers. comm.).

The overlying sandy sediments constitute the Khenig Formation. The Devonian-Carboniferous boundary which separates the Khenig Lower sandstones of »Strunien« age from the Khenig Upper

sandstones, of Tournaisian age, is not well known (CONRAD & LEMOSQUET, 1984).

The Carboniferous (about 1800 meters) can be subdivided into several cycles resulting largely from eustatic variations (FABRE & MOUSSINE-POUCHKINE, 1971).

Finally, the northward trending Cretaceous and Pliocene »Hamada« cover overlies the Palaeozoic rocks. All these formations are covered by the extensive sand dune fields of the Western Erg and the Erg Chech.

The geological record, particularly for Devonian sediments, indicates a close relationship between sedimentation and epeirogenic movements, especially for those movements which were synchronous with the Acadian orogeny, which developed since late Silurian times and allowed a lateral sedimentary progradation northwestward. Previous to the Acadian orogeny, some basement movements highlighted major structural directions such as the Fom Belrem axis. Since the Cambro-Ordovician, tectonic movements have resulted in the isolation of several basins, e.g. Ahnet and Mouydir.

This process subsequently continued and a general northwestward slope initiated a multipulsed sedimentation. The Hercynian orogeny (Variscan Orogeny of DONZEAU et al., 1982) which followed the Acadian orogeny, resulted in folding of this part of the Saharan Platform.

The oolitic ironstones

Occurrence

Within the Devonian sediments, nine ironstone occurrences of different thickness are allocated to the Lochkovian, Eifelian, Frasnian and Famennian (Fig. 3). Detailed cross-sections through the study area, show the distribution in time of the ironstones. Table 1 indicates the thickness of the oolitic ironstones, indexed 1 to 9, from bottom to top in the stratigraphic column. The oolitic ironstones are of the EXID type (Extensive Ironstone Deposition) extending over several tens to hundreds of kilometers (Fig. 4.a. b. c.). This type contrasts with the LOID type (Local Ironstone Deposition) previously defined for the oolitic ironstones of the Saharan Platform (GUERRAK, 1987).

Ironstones of the study area are very similar to the »Oolitic Minette« type of ROUTHIER (1963) and to the »Clinton« type of GROSS (1965). In another classification, they are similar to the SCOS-IF type

AGE	IRONSTONES	THICKNESS (meters)
	9	3.5
	8	1.2
Famennian	7	2.0
	6	1.0
	5	4.5
	4	0.8
-----	3	5.0
Frasnian		
-----	2	4.0
Eifelian		
-----	1	0.8
Lochkovian		
Total		22.8

Tab. 1. Thickness of oolitic ironstone beds.

(sandy, clayey and oolitic, shallow-inland-sea iron formation: KIMBERLEY, 1978, 1981).

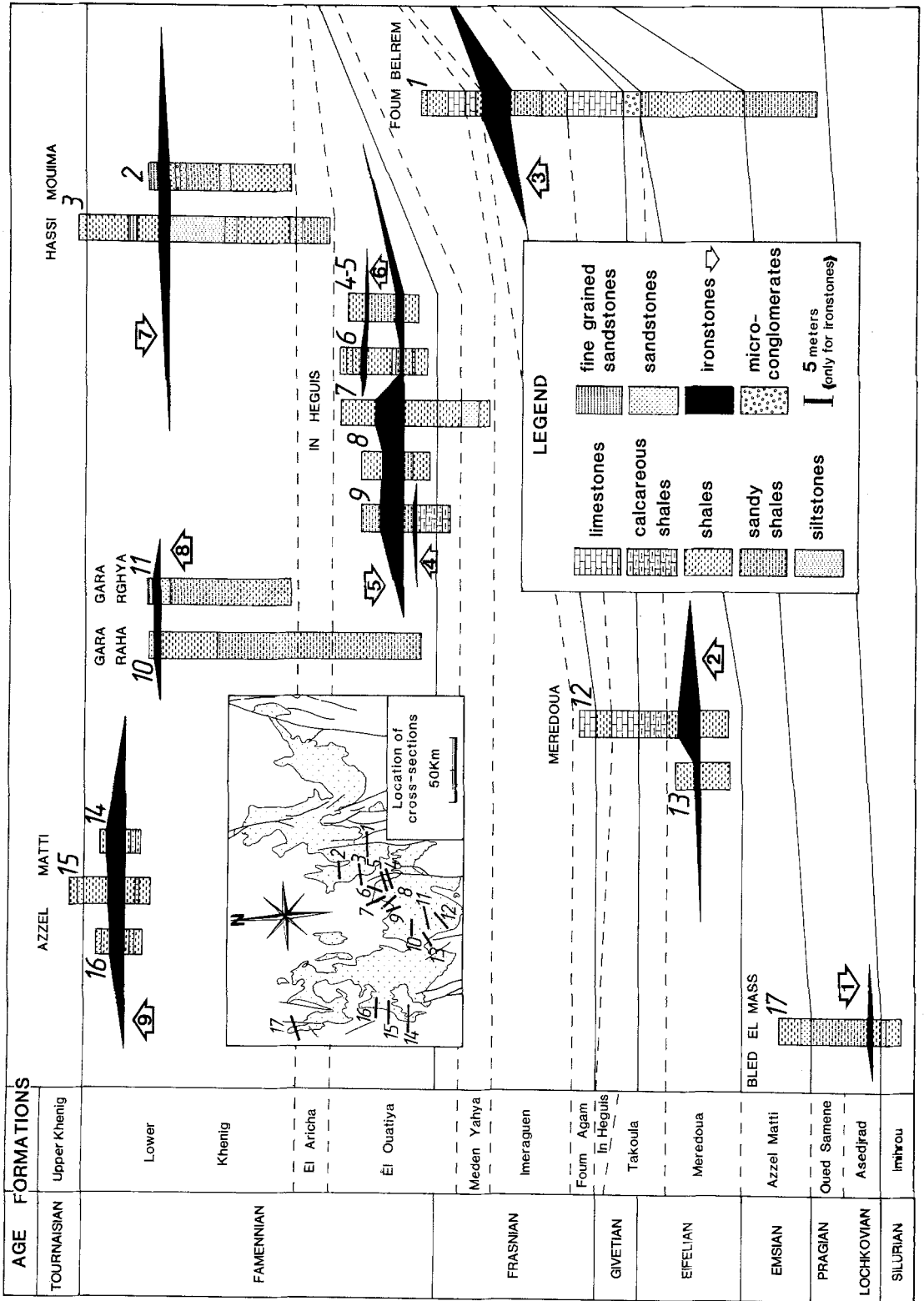
Mineralogical composition

The ironstones can be subdivided into four types of paragenetic associations P1, P2, P3, P4, as shown in Table 2, P3 and P4 being the most widespread.

Whereas hematite and goethite are common in Phanerozoic ironstones, chamosite which is a 14 Å trioctahedral chlorite (BAYLISS, 1975; BAILEY, 1980) is very frequent in Paleozoic ironstones of the Saharan Platform (GUERRAK, 1987), and especially in the deposits of the study area. It has been often confused with berthierine (7 Å trioctahedral serpentine).

Chamosite appears widespread within oololiths and groundmass, as nuclei, as intraclasts or as the major component of the whole oololith. Hematite, strictly related to chamosite, or concentrically alternating within the layers of oololiths, results in the centrifugal increase of the chamosite oxidation and occurs in all the ironstone: nucleus, matrix and layers.

To our knowledge, the Azzel Matti ironstone is the only EXID type on the Saharan Platform reported to contain magnetite. Here magnetite, overlying chamositic-hematitic ironstone, seems to be due to



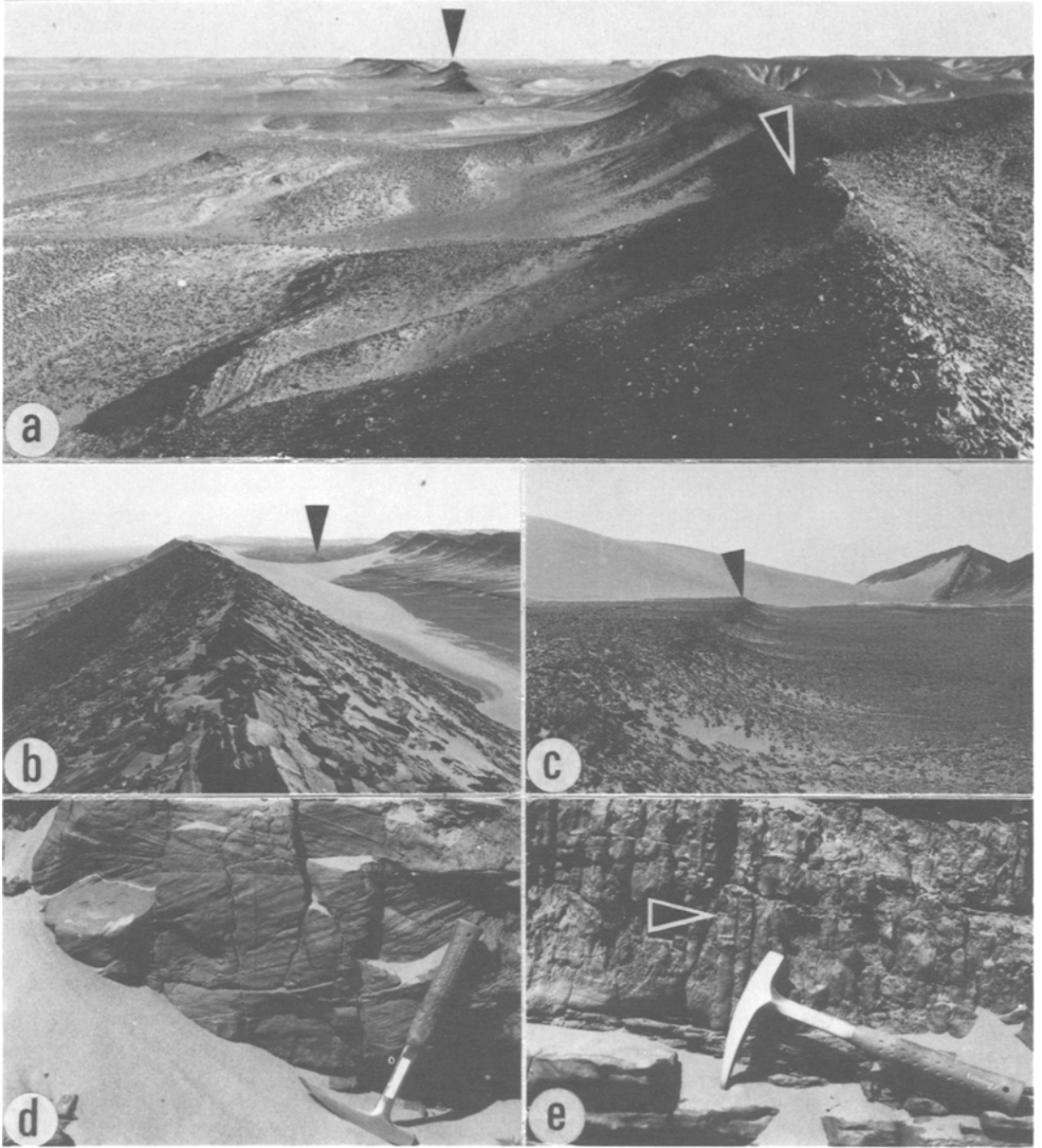


Fig. 4. (a-e). a: General view of a typical ironstone of EXID type (In Heguis, area). b-c: Hassi-Mouïma: the ironstone bed number 7. d: Gara Raha: herringbone structures in the ironstone. e: Skolithos burrows well preserved within the ironstone.

local chemical conditions of iron deposition, at the beginning of deepening of the microbasin where ironstone was deposited. It could be linked to an increase of organic matter within the sea water during the more advanced stage of diagenesis (named locomorphic by DAPPLES, 1967). According to HAN

(1986) this discharge of organic matter allows the decrease of oxidation-reduction potential and the increase of pH. The slow percolation of these reducing and basic waters within a porous ironstone could induce the transformation of hematite (previously resulting from differential oxidation of chamosite)

PARAGENESES	INDEX	IRONSTONE BEDS
chamosite-hematite goethite	P ₄	1-2-5-7-8-9
chamosite-hematite goethite-quartz	P ₃	4-5-6
chamosite-hematite goethite-calcite	P ₂	3
chamosite-magnetite maghemite-goethite	P ₁	9

Tab. 2. Distribution of paragenetic associations within the oolitic ironstone.

into magnetite. Maghemite occurs as the main weathering product of magnetite (in meteoric conditions), martite having been never detected. This is confirmed by chemical analyses (Table 6) showing a very low concentration of FeO in the magnetitic ironstone. Hematite can probably be considered as a syn-sedimentary mineral induced by the transformation of chamosite. We also described the same phenomenon for Mecheri and Gara Djebilet deposits (GUERAK & CHAUVEL, 1985). Goethite is the typical mineral of weathering processes, typically formed under oxidizing conditions at ordinary temperatures and pressures. It occurs with hematite and can completely replace it. So, goethite appears as the last oxidized mineral of the oolitic ironstone, and cannot be directly produced by chamosite.

Calcite, when it is present (e.g. Fom Belrem), is generally a secondary mineral. It appears to be replacing chamosite and hematite. It can also occur as the main constituent of bioclasts.

Quartz, as detrital grains or oolite nuclei, is not very widespread: it mainly occurs in the detrital and microconglomeratic ironstone facies. Very rare rutile grains have been found acting as nuclei but never within the matrix. Apatite is especially present as intraclasts within practically all the ironstone facies, and very rarely as nuclei. It could again occur as microinclusions within ferrous minerals.

Textural characters

Four main ironstone facies have been defined on the basis of systematic petrographical analyses:

(i) a facies containing oolites scattered in a detrital groundmass (FOD); (ii) a facies with oolites scattered in a non-detrital groundmass (FOND); (iii) a facies with cemented oolites and pore-filling struc-

ture (FOC); and (iv) a microconglomeratic facies (FMC).

These ironstone facies reflect the history of the oolitic sediment. After accretion mechanisms related to intrasedimentary processes, and several periods of suspension and deposition, took place a peculiar oolitic ironstone facies formed. The microconglomeratic facies (FMC) and the detrital facies (FOD) could be related to a final deposition within a high energy environment, whereas non-detrital (FOND) and cemented facies (FOC) could be related to quiet water conditions of the final period of sedimentation.

The variations in the mineralogical composition of oolites allow the distinction of sub-facies.

Fig. 5 shows the distribution of ferriferous facies within the different ironstone beds and emphasizes the detailed mineralogy of nuclei, cements and groundmasses.

The main ferriferous facies are vertically distributed as indicated in Table 3. The FOND facies is the widespread one, with a total thickness of about 12.3 meters. FOD and FOC facies, respectively contain 5.2 meters and 5.3 meters of ironstone deposited during Devonian times.

The FMC facies, only located at the top of Bed 5, is 0.5 meter thick. The textural characteristics are as follows:

- (i) generally, oolites are very abundant, particularly within FOD and FOC facies, occurring separately (Fig. 6.d) or as nuclei of other oolites (Fig. 6.e);
- (ii) in some examples, early stages of fragmentation where oolites are completely crushed by detrital grains, are well preserved, e.g. in the Meredoua area;
- (iii) »complex oolites« or »multiple oolites« (Fig. 6.a.

AGE	IRONSTONE BEDS	IRONSTONE FACIES	PARAGENESES
Famennian	9	FOC (FOND)	P ₁ - P ₄
	8	FOD (FOND-FOC)	P ₄
	7	FOND (FOC)	P ₄
	6	FOC	P ₃
	5	FOND (FMC-FOC)	P ₃ - P ₄
Frasnian	4	FOC	P ₃
	3	FOND	P ₂
Eifelian	2	FOD	P ₄
Lochkovian	1	FOND	P ₄

Tab. 3. Distribution in time of ironstone facies and parageneses (FMC = microconglomeratic facies, FOD = detrital facies, FOND = non detrital facies, FOC = cemented facies).

IRONSTONES		1	2	3	4	5						6	7	8	9						
FERRIFEROUS FACIES		FOND	FOD	FOND	FDC	FOND				FOC		FMC	FDC	FOND	FOC	FDC	FOND	FOD	FOC		FOND
						a	b	c	d	a	b								a	b	
MINERALOGY OF OOLITHS	HEMATITE																				
	MAGNETITE																				
	MAGHEMITE																				
	GOETHITE																				
	CHAMOSITE																				
	QUARTZ																				
COMPLEX OOLITHS																					
BROKEN OOLITHS																					
SPASTOLITHS																					
BIOCLASTS																					
INTRACLASTS																					
MINERALOGY OF NUCLEI	QUARTZ																				
	CHAMOSITE																				
	IRON OXIDE																				
	CALCITE																				
MINERALOGY OF CEMENT	QUARTZ																				
	CHAMOSITE																				
	IRON OXIDE																				
	CALCITE																				
MINERALOGY OF GROUNDMASS	QUARTZ																				
	CHAMOSITE																				
	IRON OXIDE																				
	CALCITE																				

Fig. 5. Petrographical and mineralogical composition of the oolitic ironstone.

b. c.) formed by accretion around several oolites, occur especially within the FOC facies, which appears to be the most evolved oolitic ironstone facies; (iv) many oolites are distorted (spastoliths, Fig. 6.f). Flattened or only slightly deformed, they can be associated with »normal oolites« or concentrated in some peculiar facies. Originally composed of chamosite, they probably have been deformed when they were still in a plastic stage. Within such spastoliths, nuclei could consist of chamosite, hematite or quartz. When the nucleus was chamositic, it was also deformed and various shapes appeared.

The size range of oolites is 160 μm to 1800 μm. The biggest oolites occur in Fom Belrem area (FOND facies with calcitic matrix), the smallest in In Heguis area (FOND facies with oxidized matrix).

These observations indicate the lack of direct relationships between the facies and the size of oolites. The coarseness of oolites seems to be dependent on the accretion rapidity and on the frequency of alternating accretion and reworking periods.

This emphasizes differences between the primary oolization environment and the final ironstone deposit.

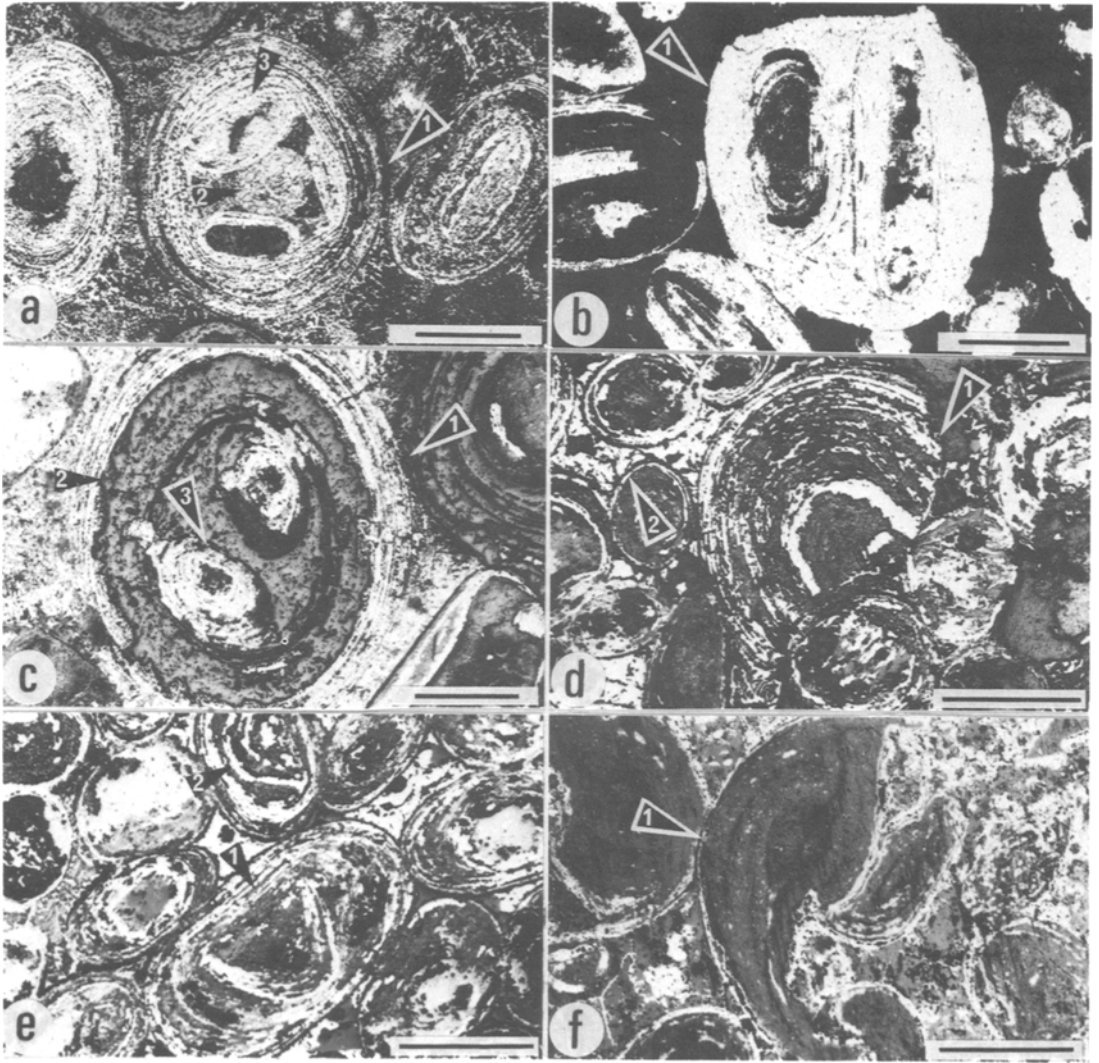


Fig. 6. (a-f). a: Ironstone bed 6 (In Heguis); reflected light. A complex oolite (in a FOC facies is composed by three hematitic-chamositic oolites (3) around a central goethitic intraclast (2). Layers are alternating chamositic-hematitic (1). Scale bar is 150 μm . b: Ironstone bed 6 (In Heguis); transmitted light. A complex oolite (FOC facies) is composed by two goethitic oolites replaced by amorphous silica (1). Scale bar is 250 μm . c: Ironstone bed 6 (In Heguis); reflected light. A complex oolite is composed by two hematitic-goethitic oolites (3) surrounded by chamositic (2) and hematitic layers (1). Scale bar is 150 μm . d: Ironstone bed 9 (Azzel Matti); reflected light. A chamositic-hematitic oolite (1) is included within a FOC facies cemented by chamosite (grey) and goethite (white). A pore filling structure is well developed in (2). Scale bar is 100 μm . e: Ironstone bed 9 (Azzel Matti); reflected light. Two broken chamositic-goethitic oolites (1) and (2) are used as nuclei of other oolites: FOC facies with pore-filling structure. Scale bar is 100 μm . f: Ironstone bed 1 (Bled el Mass); reflected light. A chamositic spastolite (1) oxidized in the cortex (white) is included in a chamositic-goethitic matrix (FOND facies). Scale bar is 100 μm .

Sedimentological paleoenvironment

Methodology

The sedimentological study, using Markov chain analysis (KRUMBEIN & DACEY, 1969; SELLEY, 1970;

ALLEN, 1974; HARMS et al., 1982), was successfully applied to ironstone sedimentation by GUERRAK & CHAUVEL (1985) in the case of a LOID type deposit (Local Ironstone Deposit). This method is used in the present paper because of the difficulty of making

coherent correlations over such a large area and of interpreting the sedimentological features of the ironstone deposits. The main interest of this analy-

tical study is to define the vertical arrangement of facies in sedimentary sequences, using relatively little data.

	A	B	C	D	E	F	G	H	I	TOTAL
A	-	-	1	-	-	-	1	-	1	3
B	1	-	1	2	-	5	10	-	-	19
C	1	2	-	-	2	1	9	-	-	15
D	-	-	-	-	-	-	1	-	-	1
E	1	1	-	-	-	-	2	-	-	4
F	-	3	4	-	-	-	4	-	-	11
G	-	12	13	1	2	2	-	1	2	33
H	-	1	-	-	-	-	-	-	1	2
I	-	-	-	-	-	-	3	-	-	3
TOTAL	3	19	19	3	4	8	30	1	4	91

OBSERVED NUMBER OF TRANSITIONS BETWEEN FACIES

	A	B	C	D	E	F	G	H	I	TOTAL
A	0.10	0.63	0.63	0.10	0.13	0.26	0.99	0.03	0.13	3
B	0.63	3.97	3.97	0.63	0.83	1.67	6.26	0.21	0.83	19
C	0.49	3.13	3.13	0.49	0.66	1.32	4.94	0.16	0.66	15
D	0.03	0.21	0.21	0.03	0.04	0.09	0.33	0.01	0.04	1
E	0.13	0.83	0.83	0.13	0.17	0.35	1.32	0.04	0.17	4
F	0.36	2.30	2.30	0.36	0.48	0.97	3.63	0.12	0.48	11
G	1.09	6.89	6.89	1.09	1.45	2.90	10.88	0.36	1.45	33
H	0.06	0.42	0.42	0.06	0.09	0.17	0.66	0.02	0.09	2
I	0.10	0.63	0.63	0.10	0.13	0.26	0.99	0.03	0.13	3
TOTAL	3	19	19	3	4	8	30	1	4	91

TRANSITION PROBABILITIES FOR RANDOM SEQUENCE

	A	B	C	D	E	F	G	H	I
A	-0.10	-0.63	0.37	-0.10	-0.13	-0.26	0.01	-0.03	0.87
B	0.37	-3.97	-2.97	1.37	-0.83	3.33	3.74	-0.21	-0.83
C	0.51	-1.13	-3.13	-0.49	1.34	-0.32	4.06	-0.16	-0.66
D	-0.03	-0.21	-0.21	-0.03	-0.04	-0.09	0.67	-0.01	-0.04
E	0.87	0.17	-0.83	-0.13	-0.17	-0.35	0.68	-0.04	-0.17
F	-0.36	0.70	1.70	-0.36	-0.48	-0.97	-0.37	-0.12	-0.48
G	-1.09	5.11	6.11	-0.09	0.55	-0.90	-10.88	0.64	0.55
H	-0.06	0.58	-0.42	-0.06	-0.09	-0.17	-0.66	-0.02	0.91
I	-0.10	-0.63	-0.63	-0.10	-0.13	-0.26	2.01	-0.03	-0.13

OBSERVED MINUS RANDOM TRANSITION PROBABILITIES

Tab. 4. Matrix analysis of sedimentary data.

tions. Three sedimentary sequences, SAH.1, SAH.2, SAH.3, (Fig. 7.b. c. d.) are deduced.

In both SAH.2 and SAH.3 sequences, facies are argillaceous to sandy, the SAH.1 being calcareous to argillaceous.

The ironstone is located:

(i) within SAH.1 and SAH.3: in a transition zone, between a coarsening and a fining-upwards sequence (top of a regressive cycle), and (ii) within SAH.2, at the bottom of a fining-upwards sequence (transgressive cycle).

The distribution of SAH.1, SAH.2 and SAH.3 through the geological formations (Fig. 8), suggests vertical successions with two main trends appearing:

- (i) Lochkovian → Eifelian → Frasnian
- (ii) Frasnian → Famennian, with internal pulses.

The comparison between this evolution and the ironstone facies distribution shows a parallel behaviour, with the exception of ironstone deposits 3 and 4. This seems to be related to the textural evolution and the nature of the sequence (transgressive or regressive).

Sedimentary structures and morphology

Numerous sedimentary structures have been recorded in the ironstone:

(i) tabular cross-beddings, sometimes of large amplitude, (ii) herringbone structures (Fig. 4.d) (1–1.5 m), (iii) channels, (iv) vertical burrows of different length (10–30 cm) related to *Skolithos ichnogenus* (GUERRAK, unpublished data; Fig. 4.e). They can be widespread (e.g. Djebel Hassi Mouïma) where the hanging wall of the ironstone is entirely perforated by numerous burrows.

Locally (e.g. Meredoua area), well preserved brachiopod shells are present (e.g. *Paraspirifer cultrijugatus*: MOUSSINE-POUCHKINE, unpublished data).

Although ironstone occurred as extensive beds, their thicknesses are not always constant. Variations have been observed between cross-sections and from one locality to another, showing a more or less lenticular aspect related to the topography of the paleoenvironment.

Interpretation

The oolitic ironstone sedimentation forms an important part of the sedimentary cycles of the Devonian. Each large sedimentary cycle was not continuous but appears to be made up of many small cycles. The evidence for such pulses is found in the oolitic ironstone deposition, similar to that described in Ordovician sediments of the Ougarta Ranges area in the western Sahara.

In the Ahnet, Azzel Matti and Bled Mass areas, the cycles were controlled by eustatic rise and fall of sea level or by epeirogenic movements.

BERTRAND-SARFATI et al. (1977) have suggested a link between oolitic sedimentation of the north-western border of the Touareg Shield, and marine transgressions. HALLAM & BRADSHAW (1979) related ironstones to falling sea level (regressions) for the NW European Jurassic. VAN HOUTEN & BHATTACHARYYA (1982) suggest that Phanerozoic ironstones were fostered by widespread transgressions and decrease of detrital influx into shallow seas flooding cratonic or intracratonic basins.

In accordance with suggestions of VAN HOUTEN & BHATTACHARYYA (1982), our opinion is that ironstones were mainly deposited during the transition between transgressive and regressive cycles, during a period of relative tectonic stability.

At the scale of the Saharan area, the marine Paleozoic sedimentation would have involved large, flat, quiet and constant epicontinental environments as described by BERTRAND-SARFATI et al. (1977), BUROLLET & BUSSON (1983), and LEGRAND (1985). This explains the large regional extension of EXID type deposits.

While the large Silurian transgression and the Permo-Carboniferous transgressions and regressions were probably caused largely by eustatic variations (FABRE & MOUSSINE-POUCHKINE, 1971), the Devonian regressions and transgressions of the Ahnet basin and limitroph areas were dominated by epeirogenic movements (BERTRAND-SARFATI et al., 1977). The main argument for this hypothesis is the development of the northwestward subsidence, in the direction of the Saoura (MENCHIKOFF, 1957). This subsidence is evident laterally and vertically in Devonian marine sediments which thicken towards the

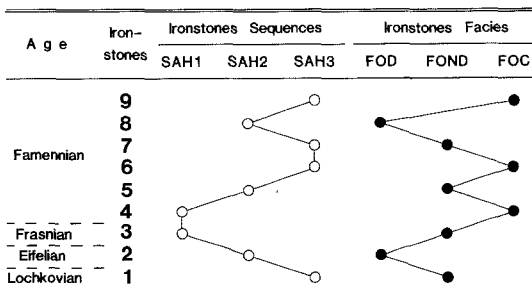


Fig. 8. Distribution in time of sedimentary sequences and ironstone facies.

Ougarta Ranges. This could explain the decrease of the oolitic ironstone deposition northwestward, where both rapid subsidence and rapid deposition gradually changed to a slow epicontinental sedimentation. Therein, we can note the lack of ironstones in the Devonian thick sediments of Ougarta Ranges. However, the extensive distribution of the oolitic ironstones (EXID type) in the Ahnet area could be the product of multiple processes. These processes which were likely of an eustatic origin, could have been disturbed by epeirogenic movements. The oscillations of sea level were probably supplemented by »local« tectonic effects, particularly revealed by the irregularity of the ironstone deposits (thickness and granulometry). VAN HOUTEN (1985) and more recently, MAYNARD (1986), related oolitic ironstones to periods of high sea level and dispersed cratonic areas.

In the area discussed by this paper, we have tried to quantify the oolitic ironstone sedimentation in terms of the duration through Devonian times. Fig. 9 shows the increase of frequency of the oolitic ironstone during Devonian times. Therefore we were able to define »Oolitic Ironstone Preservation Amounts« for each period, and to estimate the ironstone thickness deposited per million years. Nevertheless, one must realize that these amounts have been obtained without taking into account possible compaction or hiatuses of deposition, or erosion.

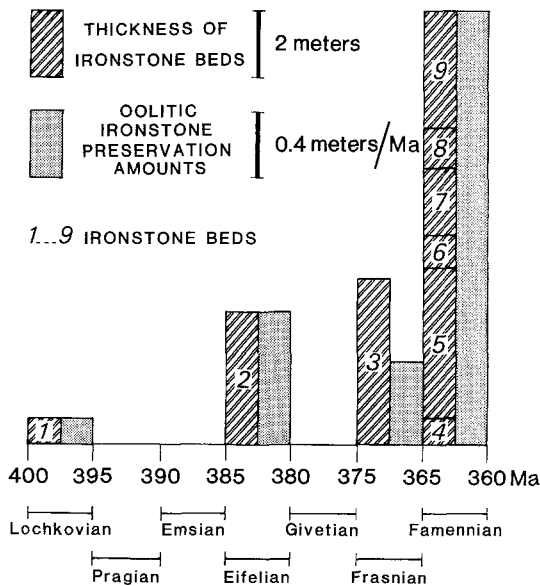


Fig. 9. Distribution of ironstone thicknesses and Oolitic Preservation Amounts through Devonian times (Ages after geochronological data from ODIN & GALE, 1982).

This oolitic ironstone sedimentation is proportional to the deposition of argillaceous and silty materials. Over the very flat shelves bordering the paleocontinent of Gondwanaland, ironstones began to develop as chamositic sediments in tranquil zones (JAMES & VAN HOUTEN, 1979; VAN HOUTEN & PURUCKER, 1984) such as lagoons or embayments. Flat barrier island systems or sand ridges could occur parallel to the coast, leaving some connections with the open sea. This tranquil but continuously submerged area of ironstone deposition, can be considered as a provisional source for the occurrence of the ooliths.

Broken ooliths, centrifugal and concentric oxidation of the layers and the formation of multiple ooliths, demonstrate the detrital nature of the ooliths and of the beds formed between successive accretionary environments. In some cases, the energy level could have been fairly important in the ironstone deposition (i.e. tabular cross-beddings of high amplitude).

Sometimes, herringbone structures indicate an intertidal environment for the ironstone. A tidal control of oolitic sedimentation has previously been suggested by TEYSSEN (1984) for the Jurassic minette of Luxembourg and Lorraine.

Geochemistry of oolitic ironstones

Analytical Procedure

Thirty-six samples of different types of oolitic ironstones (Table 5) were selected from the less weathered outcrops and analyzed for major and some trace elements (Bi, Co, Cr, Cu, F, Ni, Pb, Zn; Table 6).

About 100g of crushed material were ground to a fine powder in a tungsten carbide mill (for major elements) and in an agate ball mill (for trace elements).

The methods used and their detection limits are presented in Table 7. The Atomic Absorption Spectrophotometer used was a Perkin-Elmer 403. However, for some trace elements, the results have been rounded by the laboratory, indicating that these results are partly semi-quantitative.

The multi-element correlations and linear regressions were calculated using a microcomputer Hewlett-Packard 98-45 B.

Composition

The distribution of the elements, as shown in Fig. 10, enables the recognition of three geochemical types of oolitic ironstones (OIS), namely Fe-poor, Fe-rich and Ca-rich types.

location of cross-sections	Corresponding ironstone beds	Number of samples
1. Foug Belrem	3	2
2. Hassi Mouima	7	2
3. Hassi Mouima	7	2
4. In Heguis	6	1
5. In Heguis	6	1
6. In Heguis	6	2
4. In Heguis	5	1
5. In Heguis	5	1
6. In Heguis	5	1
7. In Heguis	5	2
8. In Heguis	5	2
9. In Heguis	5	2
10. Gara Raha	8	2
11. Gara Rghya	8	2
12. Meredoua	2	2
13. Meredoua	2	2
14. Azzel Matti	9 & 9'	3
15. Azzel Matti	9	2
16. Azzel Matti	9	2
17. Bled El Mass	1	2
Total of samples		36

Tab. 5. Chemical analyses: location of oolitic ironstone samples (according to Fig. 3).

(i) Fe-poor ironstones: this type is represented by the OIS2, located in the Meredoua area. It contains least amounts of iron with Total Fe being 18.77%; SiO₂ content is 42.50%. This can be considered as an oolitic sandstone rather than a pure oolitic ironstone. The relatively high value of CaO, when compared to Ca-rich ironstone, can be related to calcitic bioclasts, particularly of Brachiopods. Compared to the other geochemical types, phosphorus is very low, with P2O₅ < 1%.

(ii) Ca-rich ironstones: OIS.3 of Foug Belrem area is fairly rich in calcite (16.92% CaO). This is a function of its occurrence within an argillaceous to calcareous dominated sequence.

(iii) Fe-rich ironstones (OIS 1, 4, 5, 6, 7, 8, 9): these OIS reveal a range of Fe₂O₃ varying from 56.32% to 74.40%. FeO is very low with 0.05–0.37% and P2O₅ rather high: 1.85–3.58%. P2O₅ concentration is commonly proportional to Fe₂O₃ content in the Saharan ironstones. This is due to the presence of apatite as

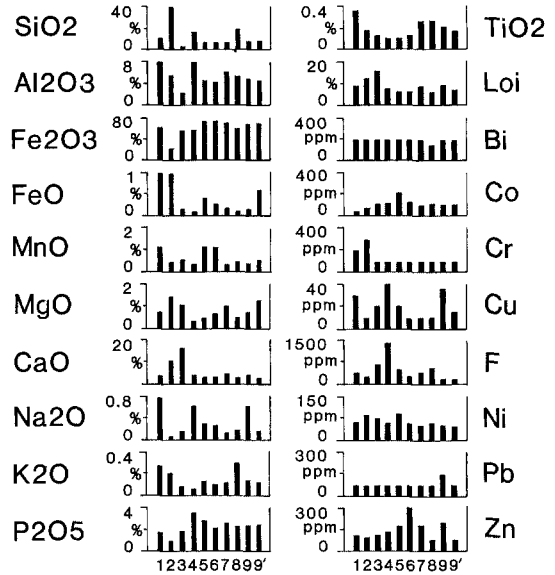


Fig. 10. Average distribution of major and minor elements for the oolitic ironstone beds number 1 to 9 (9' is the magnetitic ironstone).

microinclusions within chamosite, hematite and goethite.

Nevertheless, it must be noted for these three groups that the chemical composition has likely been modified by weathering. Consequently the enrichment in iron can be partly induced by secondary conditions developed after the deposition.

Linear correlations

Positive and negative multi-elements linear correlations have been calculated but only values of correlation coefficients *r* greater than 0.5 have been represented in Table 8. On the other hand, the most significant values of *r* are those greater than 0.75.

The main oxides of the ironstone are SiO₂ and Fe₂O₃. There is an inverse correlation between the two which is a common feature in the ironstones. However, the Meredoua ironstone (bed 3) is exception to this rule.

The positive correlation of FeO and Cr is explained by the similar behavior of Fe⁺⁺ and Cr⁺⁺⁺ which are commonly concentrated in clays (WEDEPOHL, 1974).

CaO after Loss on Ignition commonly shows a positive correlation, whereas the correlation between Na₂O and Cu cannot be interpreted because of the very low values of sodium. This also is the case for the negative correlation of K₂O/Bi.

%											CONTINENTAL CRUST			
	(1)	(2)	(3)	(4)	(5)	(6)	(7)	(8)	(9)	(9')	Average	Range		
SiO ₂	11.50	41.50	4.10	18.50	7.37	6.15	7.95	20.00	7.65	7.85	13.26	4.10	41.50	61.9
Al ₂ O ₃	8.12	5.72	2.43	7.93	4.73	4.41	6.04	5.78	5.02	4.66	5.48	2.43	8.12	15.6
Fe ₂ O ₃	62.50	24.50	56.40	56.32	73.27	74.40	69.15	60.10	68.85	72.00	61.75	24.50	72.00	2.6
FeO	1.22	2.12	0.11	0.05	0.37	0.18	0.14	0.05	0.08	0.66	0.50	0.05	2.12	3.9
MnO	1.23	0.58	0.71	0.39	1.24	1.19	0.35	0.58	0.45	0.68	0.74	0.35	1.24	0.1
MgO	0.69	1.36	1.05	0.30	0.44	0.63	0.83	0.50	0.74	1.26	0.78	0.30	1.36	3.1
CaO	3.84	11.32	16.92	3.89	3.04	3.21	4.55	3.19	5.15	2.89	5.80	2.89	16.92	5.7
Na ₂ O	0.79	0.06	0.18	0.64	0.31	0.29	0.12	0.19	0.60	0.19	0.34	0.06	0.79	3.1
K ₂ O	0.27	0.20	0.09	0.06	0.13	0.10	0.12	0.30	0.15	0.13	0.15	0.06	0.30	2.9
P ₂ O ₅	1.85	0.98	1.94	3.58	2.92	2.25	2.64	2.44	2.51	2.54	2.36	0.98	3.58	0.3
TiO ₂	0.36	0.18	0.12	0.10	0.10	0.12	0.28	0.28	0.21	0.18	0.19	0.10	0.36	0.8
LOI	8.32	12.23	15.98	8.21	6.40	6.77	8.03	6.24	8.33	7.23	8.77	6.24	15.98	
Total	100.69	100.75	100.03	99.97	100.32	99.70	100.20	99.65	99.74	100.27	100.13			100
ppm														
Bi	200	200	200	200	200	200	200	150	200	200	195	150	200	0.06
Co	40	80	120	120	220	130	100	125	115	115	116	40	220	29
Cr	200	300	100	100	100	100	100	100	100	100	130	100	300	185
Cu	30	10	20	40	20	10	10	10	35	15	20	10	40	75
F	600	400	1000	1500	775	450	600	800	300	300	672	300	1500	700
Ni	80	120	100	80	132	80	70	80	65	65	87	65	132	105
Pb	100	100	100	100	100	100	100	100	200	100	110	100	200	8
Zn	100	250	100	230	305	220	175	140	130	155	180	100	305	80

Tab. 6. Average chemical analysis of ironstone beds (9') is the magnetic oolitic ironstone of Azzel Matti.

Oxides & Elements	Analytical Method	Detection Limits in ppm
SiO ₂	Colorimetry	500
Al ₂ O ₃	Colorimetry	500
Fe ₂ O ₃	Colorimetry	500
FeO	Volumetry	500
MnO	Atomic absorption spectrophotometry	500
MgO	Atomic absorption spectrophotometry	500
CaO	Atomic absorption spectrophotometry	500
Na ₂ O	Flame photometry	100
K ₂ O	Flame photometry	100
TiO ₂	Colorimetry	100
P ₂ O ₅	Colorimetry	100
LOI	Gravimetry	500
Bi	Atomic absorption spectrophotometry	100
Co	Atomic absorption spectrophotometry	40
Cu	Atomic absorption spectrophotometry	20
Cr	Atomic absorption spectrophotometry	40
Ni	Atomic absorption spectrophotometry	100
Pb	Atomic absorption spectrophotometry	100
Zn	Atomic absorption spectrophotometry	20
F	Potentiometry	100

Tab. 7. Analytical methods and detection limits (after EREM Laboratory).

CORRELATION COEFFICIENTS (r > 0.5)	
SiO ₂	Fe ₂ O ₃ (-0.88), FeO (0.54), K ₂ O (0.52), Cr (0.63)
Al ₂ O ₃	Cu (0.58)
Fe ₂ O ₃	FeO (-0.52), CaO (-0.60), P ₂ O ₅ (0.54), LOI (-0.59), Cr (-0.66)
FeO	P ₂ O ₅ (-0.51), Cr (0.76)
MnO	P ₂ O ₅ (-0.53)
CaO	LOI (0.95)
Na ₂ O	Cu (0.75), Pb (0.51)
K ₂ O	TiO ₂ (0.68), Bi (-0.82)
TiO ₂	Zn (-0.61), Bi (-0.63)
P ₂ O ₅	Co (0.55), Cr (-0.52)
Ni	Co (0.81), Zn (0.80)
Co	Zn (0.71)
Cu	Pb (0.74)

Tab. 8. Main correlation coefficients.

Ni versus Co and Zn, presents a good correlation which can be related to their similar ionic radii.

Comparison with crustal abundance

The comparison of the mean chemical composition of the ironstones of the study area and the bulk composition of the crust (Table 6, Fig. 11) has been made on the basis of anhydrous analyses. The data of

the crustal abundance used in the present paper are those reported by RONOV & YAROSHEVSKY (1969) for major elements and by TAYLOR & MCLENNAN (1985) for trace elements. This comparison highlights the following aspects:

(i) the enrichment of Fe₂O₃ was probably increased by weathering processes and this is very common for ironstone deposits;

(ii) the behaviour of MnO is closely linked to Fe₂O₃, the geochemistry of iron and manganese being very similar. Therefore, with an increasing in iron, an enrichment in manganese can be expected. However, a high Fe/Mn ratio exists within the ironstones (KRAUSKOPF, 1957) and a complete separation of manganese and iron has been recorded in numerous deposits (EVANS, 1980).

(iii) the high proportion of P₂O₅ is strictly related to apatite occurrence which is likely to be of biogenic origin. This mineral is responsible for the high phosphorus content of the majority of the Phanerozoic ironstones.

(iv) The fluorine content, similar to that seen in crustal materials, (because of abundance of fluorapatite within igneous rock-forming minerals) is linked to the apatite. In some fish bones the fluorine content can reach 22,100ppm (WEDEPOHL, 1974), whereas within sedimentary apatites it ranges from 2 to 5% (SLANSKY, 1980);

(v) MgO, Na₂O, and K₂O, are mobile elements, and are low in concentration in all types of ironstones. The relative high average value of CaO is only induced from the Ca-rich ironstones;

(vi) concentrations of Co, Ni, Cr, Pb and Zn are higher than crustal material. The Co range is similar to values of other ironstones. Co, Ni and Cr, usually very mobile elements, are abnormally high when compared to other Saharan ironstones (GUERRAK, unpublished data). Zinc, primarily occurring in the

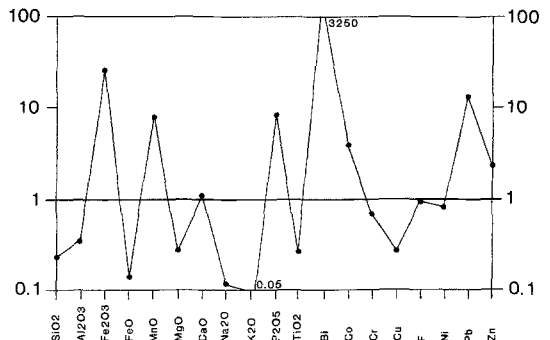


Fig. 11. Comparison of major and minor elements' average versus crustal abundance.

structure of silicates and oxides (WEDEPOHL, 1974) goes into solution during chemical weathering, but its low concentrations in surface waters, indicates a restricted mobility. However it is mainly absorbed by clay minerals such as ferruginous silicates (particularly chamosite) and iron oxides or hydroxides (hematite and goethite);

(vii) lead is weakly linked to iron, but mostly accumulated in clay minerals (WEDEPOHL, 1974) such as chamosite. High Pb values found within Banded Ironstones of Mauritania (BESNUS *et al.* 1969) were related to meteoric sources.

(viii) Copper concentrations was lower than that it is in crustal material: this was also reported by JAMES (1966).

The distribution of minor elements seems parallel to those found in shales, and particularly chloritic shales (JAMES, 1966). This may confirm the suggested importance of chamosite in the genesis of oolitic ironstone.

Conclusion

The discussion of iron transport and oolitic accretion mechanisms have been developed in other papers (GUERRAK & CHAUVEL, 1985; CHAUVEL & GUERRAK, 1986; GUERRAK, 1987). We have here discussed some aspects specific to the Ahnet s. l. area.

The data recorded in this study corroborate our previous reported observations and conclusions on Saharan ironstones. The ironstone was primarily developed in tranquil conditions by intrasedimentary processes within an iron-rich silicated mud (JAMES & VAN HOUTEN, 1979; VAN HOUTEN & PURUCKER, 1985; CHAUVEL & GUERRAK, 1986).

Probably in the same way, ferriferous coated grains are formed in Recent deposits such as Lake Chad (LEMOALLE & DUPONT, 1973), Ogooue delta in Gabon (GIRESSE, 1969), Orinoco and Niger deltas (PORRENGA, 1965) and Mahakam delta in Indonesia (ALLEN *et al.*, 1979). After accretionary periods, detrital ooliths appeared in the sediments, before their final incorporation in the ironstone beds.

The oolitic ironstone sedimentation seems to be induced by several pulses, disturbing the major sedimentary cycles. Therefore the oolitic ironstone deposition can serve as a useful indicator of minor cycles (sub-cycles).

Ironstones formed in cratonic sedimentation environments, on the borders of large shallow epicontinental seas, disappeared when the depth increased.

The ironstone deposits are a branch of the widespread Oolitic Ironstone Belt (GUERRAK, 1987),

which extends from the Rio de Oro (SOUGY, 1964) and Morocco (DESTOMBES, 1977), to Libya (NAKHLA *et al.*, 1978; CHAUVEL & MASSA, 1981; VAN HOUTEN & KARAZEK, 1981). This belt, closely linked to the Gondwanan evolution, contains numerous Paleozoic ironstone occurrences which are Ordovician, Silurian and Devonian in age and are related to alternating cold and temperate climates. The problem of the distribution of oolitic ironstones and their paleo-latitudinal location cannot easily be solved. Indeed, according to the more common continental drift reconstructions (MOREL & IRVING, 1978; SCOTESI *et al.*, 1979; BAMBACH *et al.*, 1980; SPJELDNAES, 1981; BOUCOT & GRAY, 1983; HARGRAVES & VAN HOUTEN, 1985), one would locate Saharan areas within assembled Gondwana, particularly during Devonian times. However, there is a lack of consensus for the location of the Devonian pole. The Msissi norite (Morocco) which was one of the paleomagnetic reference for Devonian pole in Africa (HAILWOOD, 1974; LIVERMORE *et al.*, 1985) is in fact of Upper Jurassic age (136–139 Ma; SALMON *et al.*, 1986). The location of North-Africa would be latitudes 30° and 60° south at this time and corresponds to the present day temperate zone. This relatively mild climate is not incompatible with the deposition of oolitic ironstone in this area. On the other hand, the source of iron would appear to be associated with a remote southeastern continent probably related to the Pan-African mobile Belt of Nigeria and the Congo Shield (Fig. 12). This is suggested by the paleocurrent directions (BEUF *et al.*, 1971), connected with ice-flow (BIJU-DUVAL *et al.*, 1981), the northwestward slope of the Basement and of the Cambro-Ordovician rocks, and the great maturity of the detrital material. In any case, the source of iron cannot be related to a hot and humid climate. Its origin some 2000–3000 km south (60–90° S latitude) of the present deposits would place the original sedimentary environments in a cold area. For explaining some oolitic ironstone occurrences, HECKEL & WITZKE (1979) and VAN HOUTEN & BHATTACHARYA (1982) invoked a supplementary influence of warm currents flooding the Gondwana margin. The process of ironstone formation does not necessarily involve an intertropical warm climate as stated by WOPFNER & SCHWARZBACH (1976) with important biorhexistasy phenomena (ERHART, 1961) for iron concentration, but at most, stages of continental preconcentration. The iron content of the rivers (which are supposed to have been the main iron transport agent), as indicated by LIVINGSTONE (1963), BERNER (1970) or TAYLOR & MC LENNAN (1985), seems to be independant of the climate. DICKKEY (1968) estimated 4.7 million tons per year the pre-

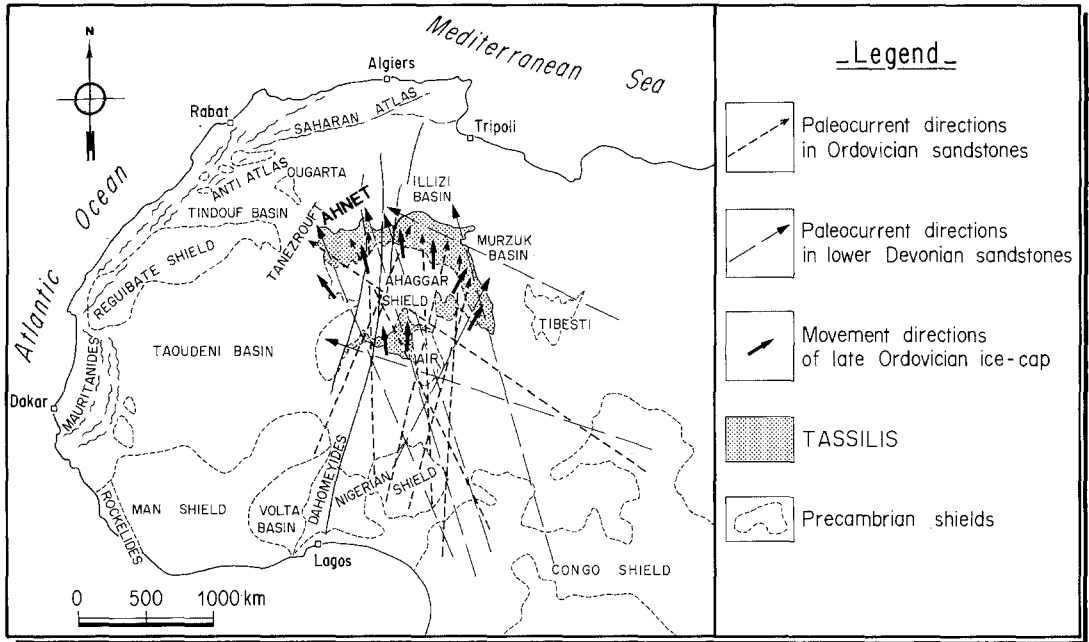


Fig. 12. Supposed iron flow directions and related source (data after BEUF et al., 1971).

precipitation of iron into the Aral sea (Soviet Central Asia); 97% of iron carried by rivers being in the form of suspended matter and 3% in solution. In other respects, transportation of iron (mostly in surface oxide films linked with suspended clays, CAROLL, 1958; JAMES, 1966) by meandering to anastomosing fluvial networks on a relatively flat continent, coincided with a period of general tectonic stability. The variations of detrital influx corresponding to the sedimentary cycles was accentuated by epeirogenic movements. The latter probably influenced the paleotopographies and thus influenced too the bulk solid discharge of the rivers.

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