

Review

Relationships between micro-organisms and formation of aroma compounds in fermented dairy products

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Beziehungen zwischen Mikroorganismen und der Bildung von Aromakomponenten in fermentierten Milchprodukten

Zusammenfassung. Der vorliegende Artikel enthält eine Zusammenstellung von Publikationen über Analyse und Bildung von Aromakomponenten, die durch milchwirtschaftlich genutzte Mikroorganismen produziert werden. Zwischen 1970 und 1993 erschienene Artikel werden besprochen, während die vor 1970 erschienene Literatur durch eine Bibliographie von Übersichtsartikeln abgedeckt wird. Schwerpunkt ist der Beitrag von Lactobacillen, Lactococcen, Streptococcen und Bifidobakterien zur Aromabildung bei fermentierten Milchprodukten. Tabellen mit vergleichbaren quantitativen Daten und den zugehörigen Literaturzitaten gewährleisten mit Hilfe von Schlüsselbegriffen einen schnellen Zugang zu einem Interessengebiet.

Abstract. A compilation of publications on the analysis and formation of aroma compounds produced by micro-organisms used in dairy production is presented. Literature published between 1970 and 1993 is reviewed whereas literature published before 1970 is covered by a bibliography of review articles. Special attention was given to the contribution of lactobacilli, streptococci, lactococci, and bifidobacteria on the formation of volatile aroma compounds in dairy products. Comparative quantitative data including complete citation of the references are listed in tables thus providing a quick access to an area of special interest with the help of key words.

Introduction

There exists a widespread spectrum of fermented dairy products, from fresh products such as yoghurt or cottage cheese via soft and semi hard cheese to hard cheese with

a maturation time of several months. The experimental knowledge on the production of fermented milk products is enormous. Yet the scientific knowledge of the detailed mechanisms and pathways followed during formation of the characteristic aroma of the respective product, despite all the scientific research done over the past 50 years, is only just beginning.

In fresh products the main stage of aroma formation occurs during the first hours of production and is dominated by the starter cultures used. During this time the microorganisms principally metabolise lactate from lactose. Thus the resulting aroma is limited to the main product and the by-products of glycolysis. The formation of the characteristic aroma of semi hard and hard cheeses on the other hand is essentially determined by proteolytic and lipolytic processes as well as by the development of a secondary microbial flora. For these products it is therefore extremely difficult to appraise the direct influence of the starter cultures on the resulting aroma.

The aim of this study is to give an overview of the scientific work that has been done in the field of analysis and formation of aroma compounds by micro-organisms used in dairy production. Special attention is given to the literature published between 1970 and 1993. Literature published before 1970 is covered by a bibliography of review articles. The main focus lies in the contribution of micro-organisms to the formation of aroma. Consequently, aroma compounds produced by starter cultures are listed in tables giving comparative quantitative data and complete citation of the references.

Bibliography of review articles

The literature concerning scientific aroma research published before 1970 is considered only in form of a listing of review articles published between 1970 and 1993 [1–59]. The contents of the articles listed in Table 1 are described through the key words of such categories, such as products, micro-organisms, maturation, aroma, and some further key words.

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Table 1. Review articles concerning aroma production in fermented dairy products

| Ref. | Products | | | Micro-organisms | | | | Maturation | | | Aroma | | Key words | Number of References cited in the article |
|------|----------|----------------|----------------|-----------------|--------------|------------|-------------|----------------|---------|-------------|-----------|-----------|-----------|---|
| | Cheese | Cultured milks | Dairy products | Lactobacilli | Streptococci | Lactococci | Leuconostoc | Bifidobacteria | Enzymes | Proteolysis | Lipolysis | Volatiles | | |
| [1] | x | | | | | | | x | x | | x | | | 80 |
| [2] | | x | | | x | x | | | x | | x | | | 84 |
| [3] | x | | | x | x | x | x | | x | x | x | | | Mould ripening 258 |
| [4] | x | x | | x | x | x | x | | x | x | x | | | Dairy products 111 |
| [5] | x | x | x | x | x | x | x | | | | x | | | Dairy products 143 |
| [6] | x | | x | | | | | x | x | x | x | | | Metabolic pathways 58 |
| [7] | x | x | | x | x | x | x | x | x | x | x | | | Metabolic pathways 90 |
| [8] | x | | | | | | | x | x | | x | | | Metabolic pathways 102 |
| [9] | x | | | | | x | x | x | x | | x | | | Metabolic pathways 73 |
| [10] | x | | | | | | | x | x | x | x | | | Bacterial metabolism, pH, redox potential 85 |
| [11] | x | | | | | | | | | | x | | | Redox reactions 11 |
| [12] | x | | | x | x | x | x | x | x | x | | | | 107 |
| [13] | x | | | | | | | x | | | x | | | Metabolic pathways, off-flavour 26 |
| [14] | x | | | | | | | | | | x | | | Metabolic pathways 63 |
| [15] | x | | | | | | | | | | x | | | Off-flavours 70 |
| [16] | x | | | x | x | x | | x | x | x | x | x | | Free amino acids, analytical methods 55 |
| [17] | x | | | | | | | | x | x | x | x | | 45 |
| [18] | x | | | x | x | x | x | | x | x | x | | | Flavour defects 111 |
| [19] | | x | | x | x | x | x | | | | x | | | Metabolic pathways 138 |
| [20] | x | | | x | x | x | x | x | x | x | x | | | 35 |
| [21] | | | x | | | x | x | | | | x | | | Metabolic pathways 22 |
| [22] | x | | | | | | | x | x | x | | | | Pediococci, micrococci 87 |
| [23] | x | | | | | | | | | | x | | | Mould-ripened blue cheeses 55 |
| [24] | | | | x | x | x | x | x | x | | | | | 97 |
| [25] | x | | | x | | | | x | x | | | | | Non starter lactic acid bacteria 123 |
| [26] | x | | | x | x | x | | x | x | x | | | | Synergisms 114 |
| [27] | x | x | | x | x | x | x | | | | x | | | Genetics 17 |
| [28] | x | | | x | x | x | | | x | x | | | | Cheese defects 230 |
| [29] | | | | x | x | x | x | x | x | | | | | Metabolic pathways 149 |
| [30] | x | | | x | | x | | x | x | x | | | | Accelerated ripening 58 |
| [31] | x | | | | | | | | | | x | | | Free fatty acids, free amino acids 107 |
| [32] | x | | | x | x | x | | | | | | | | Milk quality |
| [33] | x | | | x | x | x | | | | | | | | Starters, propionibacteria |
| [34] | x | | | x | x | x | | x | x | x | x | | | Free amino acids |
| [35] | x | | | x | x | | | | | | x | | | Cheese defects 470 |
| [36] | x | | | | | | | | | | x | x | | Swiss cheese flavour 66 |
| [37] | x | | | x | x | | | | | | x | | | Manufacture 146 |
| [38] | | x | | x | x | | | | x | x | x | | | Manufacture |
| [39] | x | x | | | | | | x | | | x | | | Redox potential, accelerated ripening 68 |
| [40] | | | x | x | x | x | x | | | | x | | | Metabolic pathways 66 |
| [41] | x | x | x | x | x | x | x | x | | | x | | | 43 |
| [42] | x | x | x | x | x | x | x | x | | | x | | | Flavour defects 86 |
| [43] | x | x | x | x | | x | x | x | x | x | x | x | | Accelerated ripening 62 |
| [44] | x | x | x | | | | | | | | x | x | | Lactones, thresholds 159 |
| [45] | | | | x | x | x | x | x | | | x | | | Metabolic pathways 18 |
| [46] | | | x | x | x | x | x | x | | | x | | | Metabolic pathways 135 |
| [47] | | x | x | x | x | x | x | x | | | x | | | Metabolic pathways 76 |
| [48] | x | | | | | | | | | | x | | | Flavour formation 39 |
| [49] | x | | | x | | | | x | x | x | x | | | Cheese defects 159 |
| [50] | x | | | | | x | | x | x | x | x | | | 181 |
| [51] | | x | | x | x | | | | | | | | | Synergetic effects 21 |
| [52] | | x | | x | x | | | | | | x | | | 51 |
| [53] | | x | | x | x | | | x | x | | x | | | Biochemistry, technology 538 |
| [54] | | x | | x | x | | | | | | x | | | Flavour variation 43 |
| [55] | x | x | x | x | x | x | x | x | | | x | | | Metabolism of starters 95 |
| [56] | | x | | x | x | x | x | x | | | x | | | Fermentation types 135 |
| [57] | | x | | x | x | | | | | | | | | Manufacture, storage 132 |
| [58] | x | | | | x | x | | x | | | x | | | Flavour development, non starter bacteria 39 |
| [59] | | | x | x | x | x | x | | x | | x | x | | Quarg, selection criteria 17 |

The products are grouped into three subgroups concerning the main subject of the articles: cheese, cultured milks, and articles that deal with dairy products in general. The micro-organisms considered in this study are lactobacilli, streptococci (namely *Streptococcus salivarius subsp. thermophilus*), lactococci, leuconostoc, and bifidobacteria. Key words in the category maturation give information as to whether the article deals with enzymes in general or with proteolysis or lipolysis in particular. The category aroma gives hints about instrumental analyses of (mostly volatile) organic compounds or about sensory evaluation of food aromas. Supplementary key words as to the contents describe the review articles even further. Finally the number of reference citations found in the respective article is noted. The aim of Table 1 is to provide rapid access to a special field of interest with the help of key words.

There are different approaches possible to the complex mechanisms and processes occurring during aroma formation. As a consequence of the rapid technical development of the early 1960s, the instrumental aroma analysis was then intensified in the field of glycolysis and proteolysis [1]. Some authors, e.g. Forss [1], Adda [4], Adda et al. [10, 15], Behnke [8] or Collins [21] preferred a purely chemical and analytical approach. First the product itself and its different aroma compounds were analysed. In a

second step the evaluated data were used to postulate and identify metabolic pathways for each of the compounds analysed. Other workers like Law [3, 7, 9, 18], Law et al. [24, 28, 30, 58], Marshall [19], Marshall et al. [29, 55] or Sandine et al. [41, 42] characterised the micro-organisms in connection with analyses of the end product. Finally a third group of authors, e.g. Kandler [45] or Keenan and Bills [40] have characterised the micro-organisms of starter cultures. This approach reveals the potential of aroma-producing capacities by breeding the micro-organisms under ideal conditions in artificial growth media.

Bibliography of research articles concerning micro-organisms

In this section references are cited that contain data on micro-organisms and their ability to produce volatile aroma compounds [60–85]. Information on the compound classes produced by lactic acid bacteria are shown in Table 2 (lactobacilli and bifidobacteria) and Table 3 (*S. salivarius subsp. thermophilus* and lactococci).

The compounds analysed belonged to the following compound classes: alcohols, aldehydes (mostly as acetaldehyde), diacetyl, acetoin, ketones, volatile organic

Table 2. References providing data on the production of volatile aroma compounds by *Lactobacillus subsp.* and *Bifidobacterium*

| Volatile aroma compounds | <i>L. delbrückii</i> | | <i>L. helveticus</i> | <i>L. acidophilus</i> | <i>L. casei</i> | <i>Bifidobacterium</i> | | |
|--------------------------|--------------------------|----------------------|----------------------|-----------------------|-----------------|------------------------|---------------|----------------|
| | <i>subsp. bulgaricus</i> | <i>subsp. lactis</i> | | | | <i>subsp. casei</i> | <i>longum</i> | <i>bifidum</i> |
| Alcohols | [60–64] | [64] | [63] | | [64, 65] | [60] | [60] | [60] |
| Aldehydes | [60–62, 66–70] | | | [66] | [65] | [60] | [60] | [60] |
| Diacetyl | [60, 62, 67, 68, 71] | | | [72–75] | [76] | [60] | [60] | [60] |
| Acetoin | [63, 64] | [64] | [63] | [72, 74, 75] | [64, 65, 76] | | | |
| Ketones | [60, 62, 63, 67] | | [63] | | | [60] | [60] | [60] |
| Esters | | | | | [65] | | | |
| Free fatty acids | [60] | | | | [65] | [60] | [60] | [60] |
| Organic acids | [60, 64] | [64] | | | [64, 65] | [60] | [60] | [60] |
| Sulfur c. c. | [60] | | | | [65] | [60] | [60] | [60] |
| Pyrazines | | | [77] | | | | | |
| Amines | | | [77] | | | | | |
| Phenols | | | [77] | | | | | |

c. c. = Containing compounds

Table 3. References providing data on the production of volatile aroma compounds by *Streptococcus salivarius subsp. thermophilus* and *Lactococcus lactis subsp.*

| Volatile aroma compounds | <i>Streptococcus salivarius</i> | <i>Lactococcus lactis</i> | | |
|--------------------------|---------------------------------|---------------------------|-----------------------------|------------------------|
| | <i>subsp. thermophilus</i> | <i>subsp. lactis</i> | <i>subsp. diacetylactis</i> | <i>subsp. cremoris</i> |
| Alcohols | [60, 62, 64] | [61, 62, 64, 81] | [61, 62] | [61, 64] |
| Aldehydes | [60, 62, 67, 70, 82, 83] | [61, 62, 81, 83] | [61, 62, 83] | [61, 83] |
| Diacetyl | [60, 62, 67, 71, 83] | [62, 81, 83, 84] | [62, 71, 83] | [76, 83] |
| Acetoin | [64] | [64, 81, 84] | | [64, 76] |
| Ketones | [60, 62, 67] | [62] | [62] | |
| Free fatty acids | [60] | | | |
| Organic acids | [60, 64, 85] | [64] | | [64] |
| Sulfur c. c. | [60] | | | |

c. c. = containing compounds

acids and compounds containing sulphur or nitrogen. There was only one reference reporting nitrogen-containing compounds such as pyrazines or amines, which would indicate a substantial proteolytic activity.

References providing quantitative data on the production of aroma compounds are discussed below. The listing includes only articles containing results of single-strain investigations either in fermented milks or in model systems. If the fermentation medium is not explicitly specified, all experiments were carried out with milk. The best defined species in this field are *Lactobacillus delbrückii subsp. bulgaricus* and *S. salivarius subsp. thermophilus*, the classical yoghurt starters.

Lb. delbrückii subsp. bulgaricus

Of all secondary metabolites of *L. delbrückii subsp. bulgaricus* listed in Table 4, acetaldehyde is of primary interest because of its key role in yoghurt aroma.

Lees and Jago [61] investigated the role of acetaldehyde as an intermediate in the formation of ethanol from glucose by lactic acid bacteria. They found a weak enzyme activity of aldehyde dehydrogenase but none of alcohol dehydrogenase. These findings are confirmed by the research of Marshall and Cole [66], who also substantiated threonine metabolism as a pathway leading to acetaldehyde production. The lack of alcohol dehydrogenase is essential to account for the relatively high acetaldehyde concentration in single strain cultures of *L. delbrückii subsp. bulgaricus*. There exist at least two pathways leading to acetaldehyde as intermediate but there is no way of reducing it to ethanol. The result is an accumulation of acetaldehyde in milks fermented with this organism.

Hamadan et al. [70] reported that single strains of *L. delbrückii subsp. bulgaricus* and *S. salivarius subsp. thermophilus* produced significantly less acetaldehyde than combined mixtures of it, a fact that could not be confirmed by the work of Yuguchi et al. [60] and Zouari et al. [69]. Scolari et al. [62] and Zouari et al. [69] characterised several strains for their ability to produce volatile aroma compounds and their respective amount of titrable acidity with respect to the preparation of new starter cultures. Principal component analysis, including further parameters such as viscosity, maximum acidification rate and corresponding time and pH value, permitted additional definition of some groups of strains showing similar behaviour [69]. Acetaldehyde production was

considerably reduced after a storage for 21 days at 4° C [69].

Dutta et al. [71] reported the effects of different heat treatments of milk on changes in acid and flavour production by single-strain cultures of *L. delbrückii subsp. bulgaricus*. The organisms did not produce diacetyl in steam-sterilised milk, whereas equal amounts were found in milk exposed to different pasteurisation treatments. Thornhill and Cogan [64] presented a new gas chromatographic method for the analysis of volatile metabolites of lactic acid bacteria. All the strains tested were grown in MRS medium at 30° C (mesophilic species) and 37° C (thermophilic species), respectively. Besides the quantified compounds ethanol, acetoin and acetic acid, further compounds such as propanone, isobutanol, n-butanol, 2,3-buteneglycol and propionic and butyric acid, were traced qualitatively.

L. acidophilus

References providing quantitative data on the ability of *L. acidophilus* to produce volatile aroma compounds are listed in Table 5. This organism is used in the manufacture of acidophilus milk, non-fermented acidophilus milk, as a complement to normal yoghurt cultures or in cultured bifidus-acidophilus milk. It is of interest for its capacity to produce acetaldehyde and diacetyl.

Marshall and Cole [66] examined threonine metabolism as a pathway for producing acetaldehyde. Addition of threonine to milk fortified with whey proteins caused *L. acidophilus* to produce a well flavoured product with a taste and aroma similar to that of yoghurt. This is not the case in milk without threonine addition. Enzyme assays of threonine aldolase and alcohol dehydrogenase re-

Table 4. References providing quantitative data on *Lactobacillus delbrückii subsp. bulgaricus*

| Volatile aroma compounds | References | | | | | | |
|--------------------------|-------------|------------|-------------|------------|------------|------------|------------|
| | [60] (µg/l) | [62] (ppm) | [64] (mg/l) | [66] (ppm) | [69] (ppm) | [70] (ppm) | [71] (ppm) |
| Ethanol | 13.05 | 0-38 | 5.8 | | | | |
| Acetaldehyde | 44.28 | 0.5-30 | | 42-45 | 6.5-15 | 10 | |
| Diacetyl | 0.84 | 0-1 | | | | | 12-13 |
| Propanone | 12.58 | 0.2-1.5 | | | | | |
| 2-Butanone | 0.53 | | | | | | |
| Acetic acid | | | 1.2 | | | | |
| Dimethyl sulphide | 22.31 | | | | | | |

Table 5. References providing quantitative data on *L. acidophilus*

| Volatile aroma compounds | References | | | | |
|--------------------------|------------|-------------|-------------|------------------------------|-------------|
| | [66] (ppm) | [72] (mg/l) | [73] (mg/l) | [74] (µg/mg cell suspension) | [75] (mg/l) |
| Acetaldehyde | 3-41 | | | | |
| Diacetyl | | | 14-62 | 1.0-1.5 | 1-18 |
| Acetoin | | | | 10-20 | 4-794 |
| Diacetyl + Acetoin | | 100-7150 | | | |

vealed modest activities of both enzymes in the organism. In fermentations without excess threonine the acetaldehyde is continuously reduced to ethanol due to the alcohol dehydrogenase activity. Compared to acetaldehyde, ethanol has a very high flavour threshold and its production will reduce the flavour quality and intensity of the product. This may explain the lack of flavour in milks fermented with *L. acidophilus*. As the activity of the alcohol dehydrogenase is rather low, a slight surplus of threonine easily saturates the enzyme with substrate, and acetaldehyde may be expected to accumulate.

The second metabolite of interest in *L. acidophilus* is diacetyl. Benito de Cardenas et al. [72–75] tested production of acetoin and diacetyl by lactic acid bacteria including the species *L. acidophilus*. All the tests were carried out in complex or synthetic media. *L. acidophilus* produces only small amounts of acetoin + diacetyl (about 100 mg/l medium) grown in basal medium containing glucose (BMG). The addition of 1.1% sodium pyruvate to BMG (=BMGP) induced the diacetyl + acetoin forming system and the production increased up to 7150 mg/l medium. Additional increments of lactic acid to the BMGP medium slightly decreased the production. A pH value below 4.25 due to the addition of more lactic acid, inhibited production of aroma compounds and growth [72].

L. acidophilus grown in complex or synthetic media produced significantly more diacetyl and acetoin in the presence of pyruvate, although pyruvate caused distinct reduction in growth [73, 74], and also produced significantly more diacetyl and acetoin at elevated temperatures, reaching an optimum at 45° C [75].

Further lactobacilli and bifidobacteria

References containing quantitative data for *L. casei ssp casei* and *L. delbrückii ssp lactis* are listed in Table 6.

Hegazi and Abo-Elnaga [76] investigated the effects due to changes in pyruvate concentration, incubation period and associative growth on the production of acetoin and diacetyl by various lactic acid bacteria including *L. casei subsp. casei*. The higher the concentrations of pyruvate added to the medium, the higher the amounts of diacetyl and acetoin produced. The time to reach the maximum concentration of both aroma compounds was dependent on the added pyruvate concentration. There

Table 6. References (Ref) providing quantitative data about *L. casei subsp. casei* and *L. delbrückii subsp. lactis*

| Volatile aroma compounds | <i>L. casei subsp. casei</i> | | <i>L. delbrückii subsp. lactis</i> |
|--------------------------|------------------------------|-----------------|------------------------------------|
| | Ref [64] (mg/l) | Ref [76] (mg/l) | Ref [64] (mg/l) |
| Ethanol | 4.2–8.4 | | 13.9 |
| Acetoin | 0–43.1 | | |
| Diacetyl | | 34–57 | |
| Diacetyl + Acetoin | | 127–2884 | |
| Acetic acid | 28.8–540 | | 81.6 |

Table 7. References providing quantitative data on *Bifidobacterium subsp.*

| Volatile aroma compounds | <i>B. longum</i> | <i>B. bifidum</i> | <i>B. breve</i> |
|--------------------------|------------------|-------------------|-----------------|
| | Ref [60] (µg/l) | Ref [60] (µg/l) | Ref [60] (µg/l) |
| Ethanol | 13.4–71 | 54.6 | 62.10 |
| Acetyldehyde | 31.0–45.3 | 12.5 | 77.5 |
| Diacetyl | 0.8–2.1 | 0.6 | 0.7 |
| Propanone | 11.0–11.4 | 12.2 | 11.6 |
| 2-Butanone | 0.5–0.6 | 0.6 | 0.6 |
| Dimethyl sulphide | 19.5–27.6 | 23.6 | 27.4 |

was no diacetyl reductase activity detected for *L. casei subsp. casei*. The amounts of diacetyl produced in media with high concentrations of pyruvate either remained constant or increased slightly at the end of the observation period.

Thornhill and Cogan [64] presented a gas-chromatographic method for the analysis of volatile metabolites of lactic acid bacteria. All the strains tested were grown in MRS medium at 30° C (mesophilic species) and 37° C (thermophilic species), respectively. The compounds quantified were ethanol, acetoin and acetic acid; further compounds could not be detected.

Quantitative data on the volatile compounds produced by bifidobacteria were found in only one publication and are listed in Table 7.

Yuguchi et al. [60] fermented sterilised milk either with a single culture consisting of one of five strains of bifidobacterium, a strain of *L. delbrückii ssp bulgaricus*, a strain of *S. salivarius subsp. thermophilus* or with mixed cultures consisting of one strain of each species. The highest production of acetaldehyde was detected with *B. breve*. In addition some organic acids were traced: citric, malic, succinic, lactic, and acetic acid were determined by HPLC.

S. salivarius subsp. thermophilus

S. salivarius subsp. thermophilus together with *L. delbrückii subsp. bulgaricus* are the classical yoghurt starter organisms, and therefore the production of acetaldehyde is of primary interest. References providing quantitative data about production of acetaldehyde and other volatile flavour compounds are listed in Table 8.

Comparisons of strains of both species showed that single-strain cultures of *S. salivarius subsp. thermophilus* produce smaller amounts of acetaldehyde than strains of *L. delbrückii subsp. bulgaricus* [53]. Lees and Jago [61] confirmed the existence of aldehyde dehydrogenase and in two of four strains of alcohol dehydrogenase in *S. salivarius subsp. thermophilus*, and established that glucose was metabolised to acetaldehyde and ethanol.

Alternative pathways for the production of acetaldehyde were published by Raya et al. [78] although reports of enzyme activities of aldehyde dehydrogenase and alcohol dehydrogenase in this organism [61] could not be confirmed in their experiments. Of the total amount of acetaldehyde produced, only a small portion resulted from

Table 8. References providing quantitative data on *S. salivarius subsp. thermophilus*

| Volatile aroma compounds | References | | | | | | | | |
|--------------------------|----------------|---------------|----------------|---------------|---------------|----------------|---------------|----------------|----------------|
| | [60] (µg/l) | [62] (ppm) | [64] (mg/l) | [70] (ppm) | [71] (ppm) | [76] (mg/l) | [82] (ppm) | [83] (mg/l) | [85] (mg/l) |
| Ethanol | 6.9 | 0-4.9 | 10.1 | | | | | | |
| Acetaldehyde | 25.0 | 3.5-5.1 | | 3.0 | | | 2.5-6.5 | 1.6-7.5 | |
| Diacetyl | 1.8 | 0-19.6 | | | 12-13 | 39-61 | | 0.1-3.8 | |
| Diacetyl + acetoin | | | | | | 181-1536 | | | |
| Propanone | 9.2 | 1.2-1.8 | | | | | | | |
| 2-Butanone | 0.5 | | | | | | | | |
| Formic acid | | | | | | | | | 18-32 |
| Acetic acid | | | 54 | | | | | | |
| Dimethyl sulphide | 16.9 | | | | | | | | |

the decarboxylation of pyruvate, since the α -carboxylase activity was rather low. Production via the hexose monophosphate shunt was excluded because phosphoketolase activity was not detected. In contrast to Lees and Jago [61], deoxyribonuclease activity was found, whereas activity of threonine aldolase could not be detected in strains tested [78].

To evaluate acetaldehyde synthesis from threonine, Wilkins et al. [79] worked with a model system using radioactive labelled threonine as a precursor of acetaldehyde. As a result, only 2% of the total amount of produced acetaldehyde showed radioactive incorporation deriving from threonine. Compared to *L. delbrückii subsp. bulgaricus* the strains of *S. salivarius subsp. thermophilus* revealed significantly higher activities of threonine aldolase.

Dutta et al. [71] reported the effects of different heat treatments of milk on acid and flavour production by single-strain cultures of various lactic acid bacteria including *S. salivarius subsp. thermophilus*. The organism produced only half the amount of acetaldehyde in steam-sterilised milk compared to the different pasteurisation treatments evaluated.

Hegazi and Abo-Elnaga [76] investigated effects due to changes in pyruvate concentrations, incubation period and associative growth on the production of acetoin and diacetyl by various lactic acid bacteria including *S. salivarius subsp. bulgaricus*. The higher the concentrations of pyruvate added to the medium, the higher the amounts of diacetyl and acetoin produced. There was no diacetyl reductase activity detected as the amounts of diacetyl and acetoin produced in media with high concentrations of pyruvate either remained constant or increased slightly at the end of the observation period.

Zouari et al. [82] characterised 20 strains of *S. salivarius subsp. thermophilus* isolated from Greek yoghurt. The strains were compared on the basis of titrable acidity, viscosity, amount of acetaldehyde produced, corresponding time to acidification rate and pH value and urease activity. Principal component analysis was used for strain grouping. Acetaldehyde was reduced by about 30% after a storage of 21 days at 4°C, apparently without alcohol dehydrogenase activity, since an increase in ethanol was not reported.

Production of formate was evaluated on three strains of *S. salivarius subsp. thermophilus* in whole milk and

complex media by Perez et al. [85]. Formate production was dependent on strain, culture medium and temperature. The maximum formate concentration was detected during the late period of exponential growth. Formate is a key component in the synergistic effect between *S. salivarius subsp. thermophilus* and *L. delbrückii spp. bulgaricus*, acting as a stimulus for the latter organism.

Lactococci

The production of lactic acid from lactose is the primary function of starter cultures. Besides this function some mesophilic starter organisms show the capacity to form diacetyl and acetoin by metabolising citrate. Many research articles concerned with the metabolism of citrate by lactococcus and leuconostoc sp. have been discussed

Table 9. References providing quantitative data on *Lactococcus lactis subsp. lactis*

| Volatile aroma compounds | References | | |
|--------------------------|---------------|----------------|----------------|
| | [62] (ppm) | [64] (mg/l) | [83] (mg/l) |
| Ethanol | 26.5-40.0 | 16.4 | |
| Acetaldehyde | 1.7-8.4 | | 0.7-6.5 |
| Diacetyl | 2.4-4.4 | | 0.05-0.6 |
| Acetoin | | | |
| Propanone | 1.3-1.6 | | |
| Acetic acid | | 12.0 | |

Table 10. References providing quantitative data about *L. lactis subsp. cremoris*

| Volatile aroma compounds | References | | |
|--------------------------|----------------|----------------|----------------|
| | [64] (mg/l) | [76] (mg/l) | [83] (mg/l) |
| Ethanol | 16.8 | | |
| Acetaldehyde | | | 0.3-8.0 |
| Diacetyl | | 39.0-42.0 | 0.05-2.3 |
| Acetoin | 16.2 | | |
| Diacetyl + acetoin | | 52-3404 | |

Table 11. References providing quantitative data on *L. lactis subsp. lactis var. diacetylactis*

| Volatile aroma compounds | References | | | | |
|--------------------------|----------------|---------------|----------------|----------------|-----------------|
| | [64] (mg/l) | [71] (ppm) | [81] (mg/l) | [83] (mg/l) | [84] (mg/kg) |
| Ethanol | 25.2–183.1 | | 13.1–26.4 | | |
| Acetaldehyde | | | 2.5–7.1 | 2.2–11.4 | |
| Diacetyl | | 12.0–15.0 | 1.0–1.9 | 0.6–55.0 | 1.5–2.5 |
| Acetoin | 0–200 | 90–100 | 212–311 | | 250–270 |
| Acetic acid | 444–840 | | | | |

Table 12. References to the compound classes of substances analysed in fermented dairy products

| Compound classes | Cheese | | | | | | | Cultured milks | |
|------------------|-------------------------------|-------------------------------|---------------------------|---------------------|---------------------------|-----------------|-----------------------------------|-----------------------------------|------------|
| | Cheddar | Swiss | Gruyère | Semi hard | Soft | Blue | Other | Yogurt | Other |
| Alcohols | [87–100] | [34, 101] | [102–108] | [105, 108–110] | [105, 111] | [105, 111] | [98, 105, 108, 112–116] | [60, 117–129] | [130, 131] |
| Aldehydes | [88, 89, 94, 95, 98–100, 132] | [34, 101, 133, 134] | [102, 103, 104, 106–108] | [108, 109, 134] | [134, 135] | [134] | [98, 108, 113, 114, 134, 136] | [38, 60, 67, 117–129] | [130, 131] |
| Diacetyl | [88, 92, 95, 100, 132] | [34, 101, 133, 137] | [108] | [108] | [111, 135] | | | [38, 60, 67, 117–129] | [130] |
| Acetoin | [93, 95, 99, 100, 132] | | [102] | | | | [113] | [38, 60, 119–123] | |
| Ketones | [87–100, 132] | [34, 101, 133, 134] | [102–108] | [105, 108–110, 134] | [105, 111, 134, 135, 138] | [105, 111, 134] | [98, 105, 108, 112–116, 134, 136] | [38, 60, 67, 117–122, 124–129] | [130, 131] |
| Esters | [88, 91, 93–95, 98, 100] | [34, 101] | [102, 103, 106–108] | [108–110] | [135, 138] | | [98, 108, 112–116] | [120, 121, 124–126, 128] | [131] |
| Lactones | [88, 89, 94, 99, 139] | [34, 101, 134, 133] | [102, 103, 106, 107] | [134] | [134, 135] | | [112–115] | [119] | [131] |
| Free fatty acids | [89, 90, 92–94, 99, 100, 140] | [34, 133, 134, 137, 141, 142] | [143, 144] | [109, 134] | [134, 135] | [134] | [112, 114–116, 134] | [38, 60, 117, 120, 121, 126, 145] | [130, 131] |
| Organic acids | [89, 90, 140, 146] | [34] | | | [146] | [146] | [112, 114, 147] | [60, 146] | [130, 146] |
| Sulfur c. c. | [91–94, 96–98, 100] | [34, 101] | [102–104, 106–108] | [108, 109] | [135, 138] | | [98, 108, 113, 114] | [60, 119, 125] | |
| Hydrocarbons | [91, 93, 94, 100] | [34] | [102, 106–108] | [108] | | | [108, 112–115] | [125] | [131] |
| Pyrazines | [77] | [77, 133, 148] | [102, 104, 106, 107, 149] | | | | [77, 113, 114, 150] | | [131] |
| Amines | [77, 140] | [77, 142] | | [109] | | | [77, 147] | | |
| Nitrogen c. c. | | [34] | [102] | [109] | [138] | | [113] | | |
| Phenols | [77, 89, 99] | [77] | [102, 108] | [108] | | | [77, 108, 113, 114] | | |

c. c. = containing compounds

by Cogan [80]. References providing quantitative data about the production of volatile aroma compounds by lactococci are listed in Tables 9 to 11.

Bottazzi and Dellaglio [83] evaluated the acetaldehyde and diacetyl production in single-strain cultures of lactococci and *S. salivarius subsp. thermophilus*. They found

considerably higher amounts of both compounds in *Lactococcus lactis subsp. lactis var. diacetylactis* than in those of *L. lactis subsp. lactis* and *subsp. cremoris* respectively, either in complex medium or in autoclaved skim milk. The diacetyl: acetaldehyde ratios under the different experimental conditions in most strains were < 1, an unfa-

vourable ratio because of the danger to develop "green flavour" in the product with ratios <2.

Dutta et al. [71] reported effects of different heat treatments of milk in acid and flavour production in selected lactococci. Neither in *L. lactis subsp. lactis* nor in *subsp. cremoris* could production of diacetyl and acetoin be detected. The biovariety *diacetyllactis* produced considerably smaller amounts of diacetyl in steam-sterilised milk than in pasteurised milks whereas for acetoin higher amounts were produced in milks treated at higher temperatures.

Clementi [84] evaluated the production of a frozen dessert from an ice cream mix with *L. lactis subsp. lactis var. diacetyllactis*. Various culture conditions were investigated in order to obtain a product with high diacetyl content together with pleasant acidic taste. Under standard conditions diacetyl accumulated to a maximum level in up to 24 h. After 36 h production then reached a steady state, probably due to an equilibrium between diacetyl synthesis and its reduction to acetoin. Accumulation of diacetyl and acetoin were promoted by pH values below 5.5. A considerable increase was achieved when milk was supplemented with additional citric acid. The temperature optimum was observed at 30° C and growth under stirred conditions resulted in a significant increase of diacetyl and acetoin amounts.

Libudzisz and Galewska [81] investigated the formation of diacetyl, acetoin, 2,3-butylene glycol, acetaldehyde, ethanol and acetic acid during 24 h of cultivation in milk or milk with supplemented citric acid. An increase in the citrate concentration in milk to 0.5% resulted in an increase in the production of diacetyl from 58 to 74% and of acetoin by 2.8 to 3.7 times. Increased concentrations of citrate in milk stimulated the production of diacetyl and acetaldehyde to a similar extent without deteriorating the organoleptic qualities of starters and product.

Bibliography of research articles about analysis of aroma compounds

During this literature search many research articles dealing with analysing aroma compounds or compilations of references on chemical compounds found in milk and dairy products, like that published by van Straten et al. [86], were found. But none of them provided special investigation into a possible contribution of either starter cultures or secondary flora to the aroma. As a result Table 12 lists references that report the analysis of volatile organic compounds. Again the table format provides rapid access to either different classes of cheeses and cultured milks or various chemical compound classes [87–150].

Obviously most of the research done in the field of aroma compounds in fermented milks has been undertaken with cheese. Most of the reports deal with only three types of cheese: Cheddar, Swiss Emmental, and Gruyère. A similar tendency was observed with cultured milks. In this field about nine out of ten publications are concerned with yoghurt, whereas for other products such as quarg or cottage cheese no research reports could be

found. On the other hand, when it comes to the chemical compound classes, the research on fermented dairy products was almost complete.

Future outlook

Research work on the influence of secondary metabolites from micro-organisms on the formation and quality of the aroma of fermented dairy products should be consolidated. More knowledge about the synergistic effects between species as well as single- and mixed-strain cultures would be of enormous interest. Gaining more insight into these mechanisms and effects would also provide desirable progress in the field of quality influencing and quality control. Qualitative and quantitative analysis of the volatile aroma compounds can only be the first step towards achieving this goal. In a second step the threshold of perception of every single compound in the matrix used should be known in order to assess its "aromatic influence" on the resulting aroma. Unfortunately, most of the available data about perception thresholds are determined with water or oil and only few with milk as matrix. Much work remains to be done until a more complete understanding of the formation and factors influencing the formation of the aroma of fermented dairy products are reached.

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