

Stimulated Raman scattering in aromatic substances has been studied thoroughly [1, 2].

We report here an attempt to obtain stimulated Raman scattering in one of the aliphatic-aromatic amines - dimethylamine. Substances of this group undergo a rapid oxidation and condensation under the influence of light [3], so we took special measures to seal the vessel and protect the substance from light. We used technical-grade dimethylamine purified by repeated distillation.

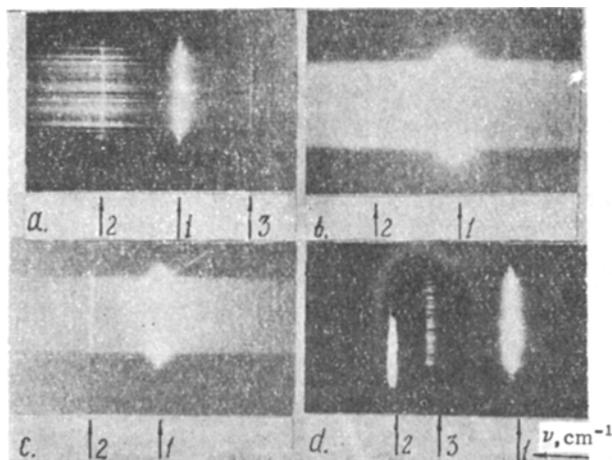


Fig. 1. a: Radiation of ruby laser, $\nu_L = 14,403 \text{ cm}^{-1}$ ($\lambda_L = 6943 \text{ \AA}$); 2) anti-Stokes component, $\nu_a = 15,385 \text{ cm}^{-1}$ ($\lambda_a = 6500 \text{ \AA}$) for $\Omega_{+1}^1 = 982 \text{ cm}^{-1}$; 3) Stokes component, $\nu_s = 13,227 \text{ cm}^{-1}$ ($\lambda_s = 7560 \text{ \AA}$) for $\Omega_{-1}^2 = 1176 \text{ cm}^{-1}$. The intensity ratio was $I_a/I_s = 10$. The pumping energy for the GOR-100m laser was $E = 7.5 \text{ kJ}$. The spectrometer slit was $t = 0.01 \text{ mm}$ wide. The dimethylamine was studied immediately after purification. b: 1) $\nu_L = 14,403 \text{ cm}^{-1}$; 2) anti-Stokes component split into three components [a very clearly defined, intense component at $\nu_{a1} = 15,389 \text{ cm}^{-1}$, a blurred component at $\nu_{a2} = 15,385 \text{ cm}^{-1}$, and a clearly defined but faint component at $\nu_{a3} = 15,372 \text{ cm}^{-1}$]. The pumping energy was $E = 9.0 \text{ kJ}$, and the spectrometer slit width was $t = 0.4 \text{ mm}$. c: one anti-Stokes component, $\nu_a = 15,385 \text{ cm}^{-1}$, obtained after an additional distillation of the dimethylamine. $E = 6.7 \text{ kJ}$, $t = 0.05 \text{ mm}$. d: lines of 1) the ruby laser; 3) a gas laser, $\lambda = 6328 \text{ \AA}$; and 2) the anti-Stokes component, $\lambda = 6500 \text{ \AA}$.

In the experimental apparatus (Fig. 2), a standard GOR-100m ruby laser (2, 4, 5) is used as a master oscillator, KS-19 filters (3, 6) modulate the radiation and suppress the radiation from the laser pumping lamps, and $F = 300 \text{ mm}$ lenses (7, 9) focus the radiation on the center of the cuvette (8) and on the spectrometer slit (10). The pulses are monitored by an FEU-28 photomultiplier (1) and an OK-17M oscilloscope (11); a pulse contains about 10 peaks. After the radiation passes through the ISP-51 spectrometer, its spectrum is recorded on Orwo-27 photographic film, whose high spectral sensitivity in the range $\lambda = 0.4-1.0 \mu$ has been checked in experiments involving stimulated Raman scattering in nitrobenzene [4].

The stimulated Raman scattering was excited at two frequencies: $\Omega^1 = 982$ and $\Omega^2 = 1174 \text{ cm}^{-1}$. The anti-Stokes component was excited at the natural frequency Ω_{+1}^1 , while the Stokes component was excited only at frequencies Ω_{-1}^2 . The anti-Stokes component was always an order of magnitude or more intense than the Stokes component, and in some cases the Stokes component was missing altogether. The accompanying figures in Fig. 1 illustrate the results.

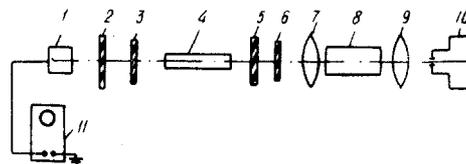


Fig. 2

V. D. Kuznetsov Siberian Physicotechnical Institute, Tomsk State University. Translated from *Izvestiya Vysshikh Uchebnykh Zavedenii, Fizika*, No. 9, pp. 137-138, September, 1970. Original article submitted September 10, 1969.

© 1973 Consultants Bureau, a division of Plenum Publishing Corporation, 227 West 17th Street, New York, N. Y. 10011. All rights reserved. This article cannot be reproduced for any purpose whatsoever without permission of the publisher. A copy of this article is available from the publisher for \$15.00.

The anti-Stokes component disappeared after 4-5 operations, while the Stokes component disappeared after 1-2 operations, so we can conclude that the composition of the substance is affected by the laser radiation. It was shown in [5] that the Stokes component (but not the anti-Stokes component) can split at a power level slightly above the threshold. An interesting feature in addition to the much greater intensity of the anti-Stokes component than that of the Stokes component for the substance in its normal state is the absence of the Stokes component for the transition $\Omega^1 = 982 \text{ cm}^{-1}$ and of the anti-Stokes component for $\Omega^2 = 1176 \text{ cm}^{-1}$.

The authors thank V. E. Zuev for interest in and support for this study.

LITERATURE CITED

1. V. A. Zubov, M. M. Suchinskii, and I. K. Shuvalov, *Usp. Fiz. Nauk*, 83, No. 2 (1964); 89, No. 1 (1966).
2. G. Bret, *Ann. Radioelectr.*, 22, No. 89 (1967).
3. A. N. Terenin, *Photochemistry of Dyes and Related Organic Compounds* [in Russian], Moscow - Leningrad (1947).
4. V. P. Lopasov, M. M. Makogon, and V. I. Mereminskii, *Izv. VUZ, Fizika*, No. 5 (1969).
5. N. V. Zubova, M. M. Suchinskii, and V. A. Zubov, *Zh. Eksperim. i Teor. Fiz., Pis. Red.*, 2, No. 2 (1965).