

# Sodium conductance in calcium channels of guinea-pig ventricular cells induced by removal of external calcium ions

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**Abstract.** An inward current characterized by a slow inactivation, was induced when the extracellular  $\text{Ca}^{2+}$  concentration was reduced by EGTA. It was suppressed by replacing external  $\text{Na}^+$  with  $\text{Tris}^+$  or by D-600, increased by epinephrine, and was not affected by TTX. These findings suggest that this current is carried by  $\text{Na}^+$  ions through the Ca channels. The Na current decreased in amplitude as the concentration of external divalent cations was elevated. Blocking the Na current by divalent cations could be approximated by a bimolecular interaction between divalent cation and channel, with a dissociation constant of  $1.2 \mu\text{M}$  for  $\text{Ca}^{2+}$  and  $60 \mu\text{M}$  for  $\text{Mg}^{2+}$ . Single channel currents were recorded in the cell-attached configuration. With a pipette solution of  $\text{pCa} = 7.5$  or  $\text{pCa} > 8$ , the single channel I–V relationship was linear and the slope conductance was  $70\text{--}75 \text{ pS}$ . For  $40 \text{ mV}$  depolarizations from the resting potential, unitary currents were smaller at  $\text{pCa} = 6$  than at  $\text{pCa} = 7.5$ . However, single channel events, which were observed after the repolarizing step to the resting potential, were much the same amplitude. The open time histogram was fitted with a single exponential having a time constant of  $1.9 \text{ ms}$  at around  $-40 \text{ mV}$  ( $\text{pCa} > 8$ , with  $5 \mu\text{M}$  Bay K 8644 in the bath solution), which was decreased with increasing the  $\text{Ca}^{2+}$  concentration in the pipette solution. Noise power spectra of patch currents at  $\text{pCa} = 6$  revealed a high-frequency component at around  $1500 \text{ Hz}$ . These results suggest that Ca binding to the sites with a high affinity for  $\text{Ca}^{2+}$  blocks the Na conductance in Ca channels. Reduction of the unitary current at higher concentrations of  $\text{Ca}^{2+}$  might be attributed to a rapid block by  $\text{Ca}^{2+}$ .

**Key words:** Cardiac cells – Ca channels – Ion selectivity – Na ions

## Introduction

In preceding work, we demonstrated that the Ca channel current was not affected by removal of external  $\text{Na}^+$  when the intracellular  $\text{Ca}^{2+}$  of single ventricular cells was controlled by internal dialysis (Matsuda and Noma 1984). This finding suggests that under physiological conditions, Ca channels are not measurably permeable to  $\text{Na}^+$  (but see

Reuter and Scholz 1977). On the other hand, it has been known that  $\text{Na}^+$  ions can carry inward currents through the Ca channel when the external  $\text{Ca}^{2+}$  is reduced (Rougier et al. 1969; Garnier et al. 1969; Hess and Tsien 1984).

Inward currents carried by monovalent ions through Ca channels have been detected in a variety of tissues other than cardiac fibers and recently several investigators have attempted to clarify how ion selectivity of Ca channels is changed in the absence of external divalent cations: in snail neurones (Kostyuk et al. 1983), in single guinea-pig ventricular cells (Hess and Tsien 1984), in frog skeletal muscle fibers (Almers et al. 1984; Almers and McCleskey 1984) and in mouse lymphocytes (Fukushima and Hagiwara 1985). These results suggest that there are binding sites with a high affinity for  $\text{Ca}^{2+}$  and that the occupation of this site by  $\text{Ca}^{2+}$  prevents the permeation of monovalent cations, albeit the location of this site being controversial.

Relatively little is known of the properties of single Na currents through Ca channels and the effects of  $\text{Ca}^{2+}$  on the Na conductance at the level of single channel (Lansman et al. 1985). Thus, I analyzed the fluctuation of whole cell Na currents and recorded single Na currents in the cell-attached configuration, at different concentrations of  $\text{Ca}^{2+}$ . The results show that  $\text{Ca}^{2+}$  inhibits the Na current through the Ca channels by reducing the mean open time. Decrease of the unitary current amplitude was ascribed to fast blocking and unblocking kinetics by  $\text{Ca}^{2+}$ .

## Methods

Single ventricular cells from adult guinea-pigs were obtained by enzymatic dissociation (Matsuda et al. 1982) and recordings of the currents were performed using a heat polished patch electrode in the whole cell and cell-attached configurations (Hamill et al. 1981).

The methods used for the whole cell current recordings were similar to those described elsewhere (Matsuda and Noma 1984). The electrode resistance ranged between  $1$  and  $2 \text{ M}\Omega$ . The series resistance arising mainly at the patch-electrode tip was compensated by summing a fraction of the converted current signal to the command potential. The compositions of the main external and pipette solutions are listed in Table 1. To minimize currents through the K channels, the pipettes were perfused with a Cs-rich solution and external  $\text{K}^+$  was removed. Under this condition, currents in inwardly rectifying  $\text{K}^+$  channels are abolished (Matsuda and Noma 1984). A standard Ca-free external solution was prepared by omitting  $\text{CaCl}_2$  and adding

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**Table 1.** Composition of solutions (mM)

	External						
	NaCl	NaH <sub>2</sub> PO <sub>4</sub>	KCl	MgCl <sub>2</sub>	CaCl <sub>2</sub>	EGTA	HEPES-NaOH
Normal solution	144	0.33	5.4	0.5	1.8	—	5
0 K solution	144	0.33	—	0.5	1.8	—	5
Standard 0 Ca solution <sup>a</sup>	144	0.33	—	2.3	—	0.1	5
0 Mg, low-Ca solution							
pCa = 4.2	144	0.33	—	—	0.2	0.1	5
pCa = 4.6	144	0.33	—	—	0.13	0.1	5
pCa = 5.0	144	0.33	—	—	0.11	0.1	5
pCa = 5.5	144	0.33	—	—	0.102	0.1	5
pCa = 5.9	144	0.33	—	—	0.1	0.102	5
pCa = 6.4	144	0.33	—	—	0.1	0.11	5
pCa = 7.0	144	0.33	—	—	0.1	0.13	5
pCa = 7.5	144	0.33	—	—	0.1	0.2	5
0 Ca, Mg-containing solution <sup>b</sup>							
pMg = 2.3	144	0.33	—	5.0	—	0.1	5
pMg = 2.6	144	0.33	—	2.3	—	0.1	5
pMg = 3.0	144	0.33	—	1.0	—	0.1	5
pMg = 3.3	144	0.33	—	0.5	—	0.1	5
pMg = 4.0	144	0.33	—	0.1	—	0.1	5
Pipette solution for whole cell current recordings							
	Cs aspartate	CsCl	MgCl <sub>2</sub>	EGTA	ATP <sup>c</sup>	HEPES-CsOH	
Cs-rich solution	110	20	1	5	5	5	
Pipette solution for single channel current recordings							
	NaCl	NaH <sub>2</sub> PO <sub>4</sub>	CaCl <sub>2</sub>	EGTA	HEPES-NaOH		
0 Mg, 0 Ca solution							
pCa > 8	144	0.33	—	2	5		
0 Mg, low-Ca solution							
pCa = 6.0	144	0.33	1	1.03	5		
pCa = 7.5	144	0.33	1	2	5		

External solution contained 5.5 mM glucose and the pH was adjusted to 7.4. In pipette solution for whole cell current recordings, pH was adjusted to 7.3. The liquid junction potential between the internal solution and the normal external solution was corrected (Matsuda and Noma 1984). Pipette solution for single channel current recordings contained 31  $\mu$ M TTX unless otherwise indicated and the pH was adjusted to 7.4.

<sup>a</sup> Containing 0.5  $\mu$ M epinephrine

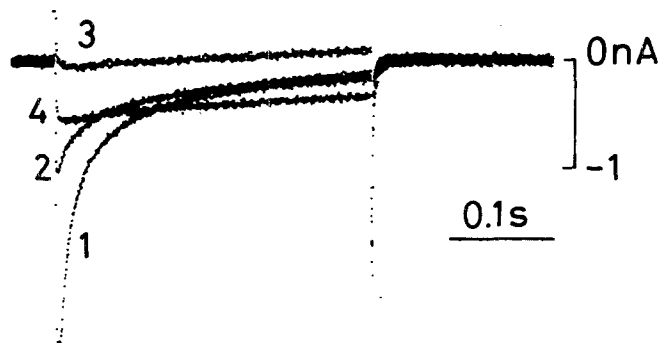
<sup>b</sup> Mg<sup>2+</sup> and Ca<sup>2+</sup> concentration was calculated from the apparent dissociation constants (pCa > 8)

<sup>c</sup> Adenosine 5'-triphosphoric acid as dipotassium salt; Sigma

0.1 mM ethyleneglycol-bis( $\beta$ -aminoethylether)N,N'-tetraacetic acid (EGTA). It should be noted that the cell membrane becomes leaky in solutions without divalent cations (Kostyuk et al. 1983; Fukushima and Hagiwara 1985). An increase in leakage current developing in Ca-free solution was depressed by adding equimolar Mg<sup>2+</sup>, which partially depressed the Na current under investigation (see Results). Thus, to obtain a larger Na current, 0.5  $\mu$ M epinephrine was added in most experiments with 2.3 mM external Mg<sup>2+</sup>. When the current was compared at different external Ca<sup>2+</sup> concentrations, external Mg<sup>2+</sup> was removed. The free Ca<sup>2+</sup> concentration was measured with a Ca<sup>2+</sup>-sensitive electrode (Model 93-20, Orion Research) and expressed by its inverse log<sub>10</sub>. Mg<sup>2+</sup> solution was prepared by adding a desired concentration of MgCl<sub>2</sub> to Ca-free solution containing 0.1 mM EGTA. The Mg<sup>2+</sup> concentration was calculated

from the apparent dissociation constants (Fabiato and Fabiato 1979) and expressed by its inverse log<sub>10</sub>. Whole cell current and voltage signals were stored on magnetic tape (TEAC, R-210) for computer analysis (Hitachi, E-600). Current traces were displayed without subtraction of leak current or capacitive current. The procedure used to estimate the unit amplitude from the variance-to-mean current relationship is described in the Results

Single channel currents were recorded from cell-attached patches on cells superfused with normal external solution, to which Bay K 8644 was added in most experiments to promote extended opening of the channel. The pipettes (5–10 M $\Omega$  resistance) were made from capillaries of high melting temperature glass and were coated with silicone (Shin-etsu Chemical Co.). Records were made with an EPC-5 or EPC-7 patch clamp amplifier (List Electronic).



**Fig. 1.** Change in the Ca channel current after removal of external  $\text{Ca}^{2+}$ . Depolarizing pulses of 300 ms were applied from  $-43$  to  $+7$  mV every 10 s. Membrane currents before (trace 1) and 3 min (trace 2), 5 min (trace 3) and 10 min (trace 4) after exchanging K-free external solution for standard Ca-free solution were superposed

Data were stored on magnetic tape (TEAC, R-210) in earlier experiments (Fig. 6) and on a video cassette (Victor, BR6400) using a PCM converter system (NF, RP-880; reproduce bandwidth 10 kHz) in later experiments (Figs. 7–10). Currents were filtered using a four-pole low-pass Bessel filter (NF Circuit Design Block Co. FV-625A, 48 dB/oct) with a 3-dB corner frequency of 1.4 kHz and sampled every 0.2 ms, unless otherwise indicated. To analyze fast flickering kinetics of single channel current recordings, patch current data were low-pass filtered to 3 kHz by a four-pole Butterworth filter and sampled every 0.1 ms. Sampling was begun 10 ms after the start of the 130 ms clamp pulse. Power spectral densities were calculated on 1024 point data sections via a fast Fourier transform routine, resulting in one-sided spectra. Twenty to forty such spectra were averaged. Spectra of background noise sampled from frames without events were subtracted.

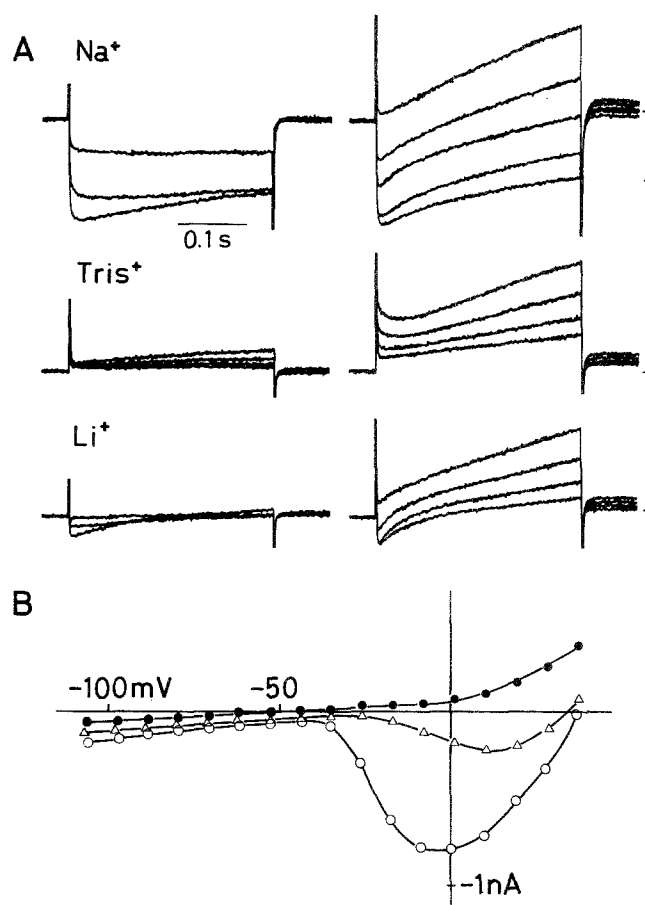
Voltage steps were applied every 6 s in whole cell current recordings and every 3–4 s in single channel current recordings, unless otherwise indicated. All experiments were carried out at  $35$ – $37^\circ\text{C}$ . Average results throughout this paper are given as mean  $\pm$  SD.

## Results

### *Development of the inward current in the absence of $\text{Ca}^{2+}$*

It is known that removal of external  $\text{Ca}^{2+}$  induces a biphasic effect on action potential of cardiac preparations; a shortening of the action potential followed by a progressive lengthening (e.g. Rougier et al. 1969). The change in the Ca channel current underlying such an effect was studied, under conditions where the K currents were minimized by removing  $\text{K}^+$  from both sides of the membrane (Fig. 1). After K-free external solution containing 1.8 mM  $\text{CaCl}_2$  was exchanged for standard Ca-free solution, the Ca current (trace 1) decreased gradually (trace 2) and disappeared 5 min later (trace 3). Further exposure to the Ca-free solution induced an inward current which was sustained during the 300 ms clamp pulse (trace 4).

The proposal that the inward current in the absence of external  $\text{Ca}^{2+}$  is carried through the Ca channels is deduced from the following findings. (1) The inward current was not affected by addition of  $3.1 \mu\text{M}$  TTX to the external solution (not shown). (2) Organic Ca channel blockers such as verapamil or D-600 suppressed this current (Fig. 3). (3) The



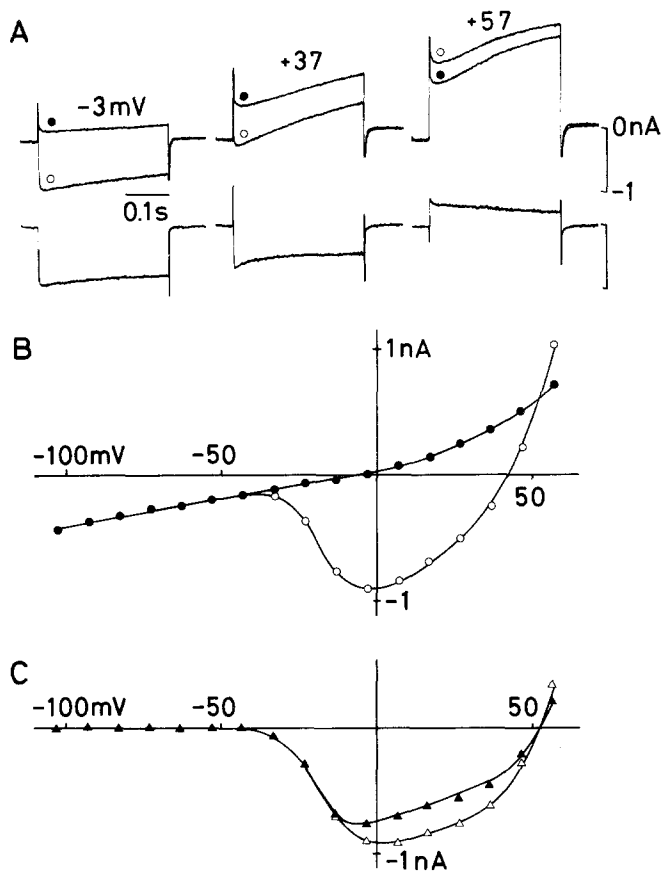
**Fig. 2A, B.** Membrane currents in Ca-free solutions containing  $\text{Na}^+$ ,  $\text{Tris}^+$  and  $\text{Li}^+$  as the major cation. **A** Total NaCl in standard Ca-free solution (upper panel) was replaced with Tris-Cl (middle panel) or LiCl (lower panel). The pH was adjusted by 5 mM LiOH-HEPES in  $\text{Li}^+$  solution and by Tris-Cl buffer (149 mM) without HEPES in  $\text{Tris}^+$  solution. Traces with  $\text{Na}^+$  and  $\text{Tris}^+$  were obtained during voltage steps to between  $-26$  and  $+36$  mV in  $9$ – $10$  mV increments, and traces with  $\text{Li}^+$  during voltage steps to between  $-17$  and  $+37$  mV in  $9$ – $10$  mV increments. Holding potential,  $-43$  mV. The marks to the right of each panel indicate zero current. The current calibration represents  $0.5$  nA. **B** Initial  $I$ – $V$  relations with  $\text{Na}^+$  (open circles),  $\text{Tris}^+$  (filled circles) and  $\text{Li}^+$  (open triangles)

inward current was increased in amplitude in the presence of  $0.5 \mu\text{M}$  epinephrine by  $99 \pm 50\%$  ( $n = 5$ , figure is not shown).

### *Na and Li conductance in Ca channels in the absence of $\text{Ca}^{2+}$*

Figure 2A shows membrane currents in Ca-free solution containing  $\text{Na}^+$ ,  $\text{Tris}^+$ , or  $\text{Li}^+$ , as the major cation. The initial currents are plotted against the membrane potential in Fig. 2B (leakage currents are not subtracted). It is evident that  $\text{Na}^+$  carries a large inward current while the large organic cation  $\text{Tris}^+$  does not. Replacing  $\text{Na}^+$  with  $\text{Tris}^+$  unmasked the outward current, which may be attributed to the Cs current, either through a non-specific voltage-dependent conductance (Byerly and Hagiwara 1982; Matsuda and Noma 1984) or through the Ca channel.

After replacing external  $\text{Na}^+$  with  $\text{Li}^+$ , the inward current peaked at a smaller magnitude and at more positive potential than the Na current. The reversal potential of the



**Fig. 3A–C.** Isolation of the Na current by D-600. **A** The upper panel shows superimposed current records before (*open circles*) and after (*filled circles*) addition of  $1 \mu\text{M}$  D-600 to standard Ca-free solution. 300 ms step depolarizations from  $-43 \text{ mV}$  to the potentials indicated. The lower panel shows differences between currents before and after D-600 application. **B** Initial currents before (*open circles*) and after (*filled circles*) addition of D-600 are plotted against the membrane potential. **C** Difference currents (*open triangles*, initial; *filled triangles*, late) are plotted against the membrane potential

Li current obtained using D-600 ( $1 \mu\text{M}$ ) in other cells was more positive by  $10\text{--}15 \text{ mV}$  than that of the Na current. Thus there may be a difference in screening effects between  $\text{Na}^+$  and  $\text{Li}^+$ . The current amplitude was measured at the potential that gave the largest inward current after subtraction of linear leakage currents (see below). It decreased to  $30 \pm 6\%$  ( $n = 3$ ) of the control Na current after substitution of  $\text{Li}^+$  for total  $\text{Na}^+$ , suggesting that  $\text{Li}^+$  is a weaker charge carrier than  $\text{Na}^+$ . The channels under investigation seem to be less permeable to  $\text{Li}^+$  than the fast Na channel, for which the  $\text{Li}^+:\text{Na}^+$  permeability ratio is reported to be 0.92 in rat ventricular cells (Akaike et al. 1984), 0.93 in frog nerve (Hille 1972) and 1.1 in the squid axon (Chandler and Meves 1965).

#### Isolation of the Ca channel current developed in the absence of external $\text{Ca}^{2+}$

In order to isolate the Ca channel current, the current was recorded before (*open circles*) and after (*filled circles*) application of  $1 \mu\text{M}$  D-600 (upper panel of Fig. 3A) and the Ca blocker-sensitive current (difference current) was obtained by subtracting the current after the drug application from that before the application (lower panel of

Fig. 3A). The resulting current reversed between  $+37 \text{ mV}$  and  $+57 \text{ mV}$ , and currents in both directions were inactivated slowly during the depolarizing pulse. The difference current at  $+37 \text{ mV}$  decayed more slowly than the current before subtraction, indicating that the time-dependent current change during larger depolarizations results partly from activation of a non-specific time- and voltage-dependent outward current.

In Fig. 3B, the initial current before (*open circles*) and after (*filled circles*) application of D-600 was plotted against the membrane potential. The current-voltage ( $I\text{--}V$ ) curves after application of the drug were almost linear up to around  $+10 \text{ mV}$ . This validates assessing the current amplitude as the difference between the point on the  $I\text{--}V$  curve and the point on the straight line extrapolated from the hyperpolarizing region (manual subtraction of linear leakage currents).

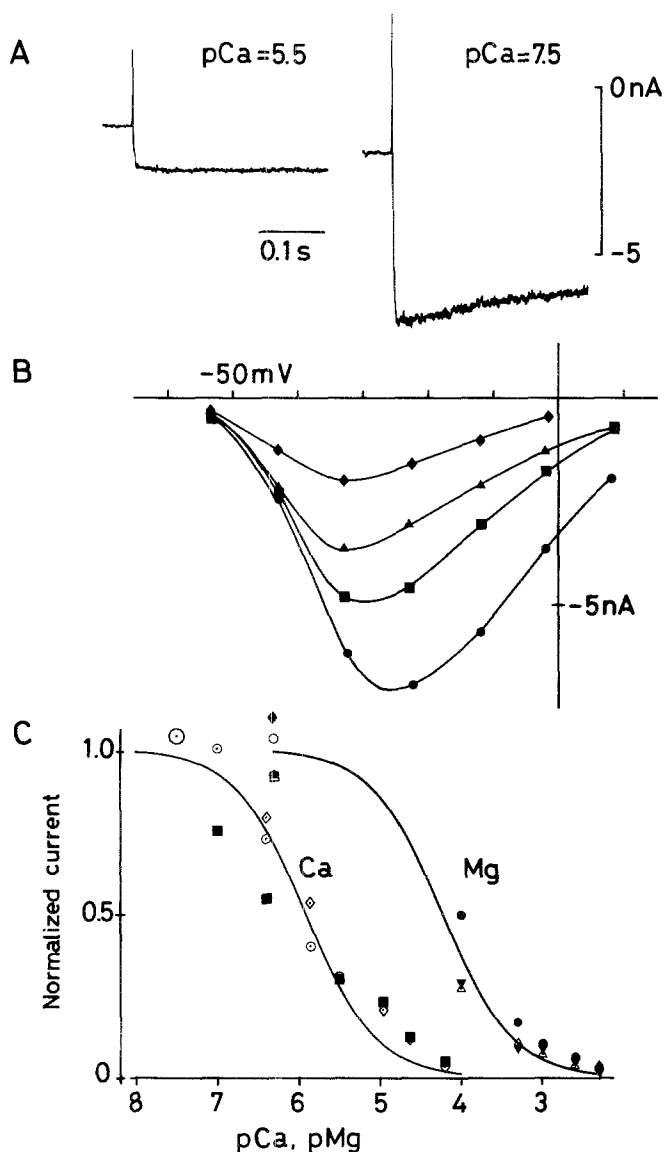
In Fig. 3C the initial amplitude (*open triangles*) and amplitude near the end of the 300 ms clamp pulse (*filled triangles*) of the difference current have been plotted against the membrane potential. Both the initial and late current reverse at around  $+50 \text{ mV}$ . Essentially the same results were obtained in four other experiments and the reversal potential averaged  $46 \pm 6 \text{ mV}$  ( $n = 5$ ). This is in contrast to the case of the Ca current, where the late current reverses at a potential  $10\text{--}20 \text{ mV}$  less positive than the initial current (Matsuda and Noma 1984). (The reversal potential of the initial current was  $76 \pm 5 \text{ mV}$ ,  $n = 10$ .) The above result suggests that the reversal of the D-600 sensitive Na current is a true reversal of the Ca channel current.

If  $\text{Na}^+$  is assumed to be the only charge carrier, the averaged reversal potential of  $+46 \text{ mV}$  would mean that the intracellular  $\text{Na}^+$  concentration is  $25.4 \text{ mM}$ . Such a high intracellular  $\text{Na}^+$  concentration is unlikely because the pipette solution did not contain  $\text{Na}^+$  and the fast Na current reversed around  $+100 \text{ mV}$ . Under the condition where the Ca channels become permeable to  $\text{Na}^+$ ,  $\text{Cs}^+$  is likely to be permeant, and the less positive reversal potential than that of the fast Na current may be attributable to the outward Cs movement through the Ca channels.

As described below, the Na current in Mg-free, low-Ca solution peaked at a potential  $20\text{--}30 \text{ mV}$  more negative than the original Ca current. This is ascribed to surface potential changes caused by reducing divalent cations in the external solution (McLaughlin et al. 1971). The degree of the negative shift of the peak inward current was decreased in standard Ca-free solution (Fig. 3C). The reversal potential of the Na current is more positive in standard Ca-free solution than in Mg-free, low-Ca solution. (The reversal potential in Mg-free solution predicted from the single channel  $I\text{--}V$  curves is  $+10 \text{ mV}$ , see Fig. 7C.) Therefore the positive shift of the potential that gave the largest Na current in standard Ca-free solution is ascribed to surface potential changes caused by increasing  $\text{Mg}^{2+}$ , rather than alteration of the activation kinetics.

#### Depressive effect of divalent cations on the Na current

The findings described above favor the view that the Na current developed in Ca-free solution is carried through Ca channels, the ion selectivity and inactivation process of which are modified by removal of external  $\text{Ca}^{2+}$ . To clarify the mechanism of modification of the Ca channels, I studied the dependence of the amplitude of the Na current on the



**Fig. 4A–C.** Depression of the Na current by external  $\text{Ca}^{2+}$ . **A** The current records at  $\text{pCa} = 5.5$  and  $\text{pCa} = 7.5$  during steps from a holding potential of  $-63$  mV to  $-33$  mV. **B**  $I-V$  relations after leakage currents were subtracted manually.  $\text{pCa} = 5.5$  (diamonds),  $6.4$  (triangles),  $7$  (squares) and  $7.5$  (circles). **C** Dose-response curve for blockade of the Na current by  $\text{Ca}^{2+}$  and  $\text{Mg}^{2+}$ . Currents were measured at the potential that gave the largest inward current at each concentration and expressed as the fraction of the current at  $\text{pCa} = 7.5$  (large open circle) in cases of  $\text{Ca}^{2+}$  and as the fraction of the original Ca current in cases of  $\text{Mg}^{2+}$ . The data from the same cells were presented by the same symbol. The curves which give the best fit to the data points were determined by the least squares method with the assumption that the Na current through the Ca channel is blocked by one  $\text{Ca}^{2+}$  or by one  $\text{Mg}^{2+}$  per channel. The dissociation constant is  $1.2 \mu\text{M}$  for  $\text{Ca}^{2+}$  and  $60 \mu\text{M}$  for  $\text{Mg}^{2+}$ . Current amplitude was normalized as maximum current to be 1. Maximum current in cases of  $\text{Mg}^{2+}$  was obtained in Mg-free, low-Ca solution ( $\text{pCa} = 7.5$ ). In this figure, data were plotted assuming that nominally Mg-free solution contained  $0.5 \mu\text{M}$  free  $\text{Mg}^{2+}$ .

extracellular  $\text{Ca}^{2+}$  or  $\text{Mg}^{2+}$ . In these experiments, the concentration of  $\text{Ca}^{2+}$  or  $\text{Mg}^{2+}$  was lowered stepwise and the  $I-V$  relation was measured at each concentration after the new steady state had been reached. The membrane was held at  $-63$  mV in Mg-free, low-Ca solution and at  $-43$  mV in Ca-free, Mg-containing solution. Figure 4A shows the

current traces recorded at  $\text{pCa} = 5.5$  and  $\text{pCa} = 7.5$  during steps from a holding potential of  $-63$  mV to  $-33$  mV. As the external  $\text{Ca}^{2+}$  was reduced, the Na current was increased in amplitude and was accompanied by fluctuations. The peak amplitude of the Na current at  $\text{pCa} = 7.5$  is greater than that of the original Ca current by a factor of 5 ( $5.03 \pm 0.47$ ,  $n = 4$ ).

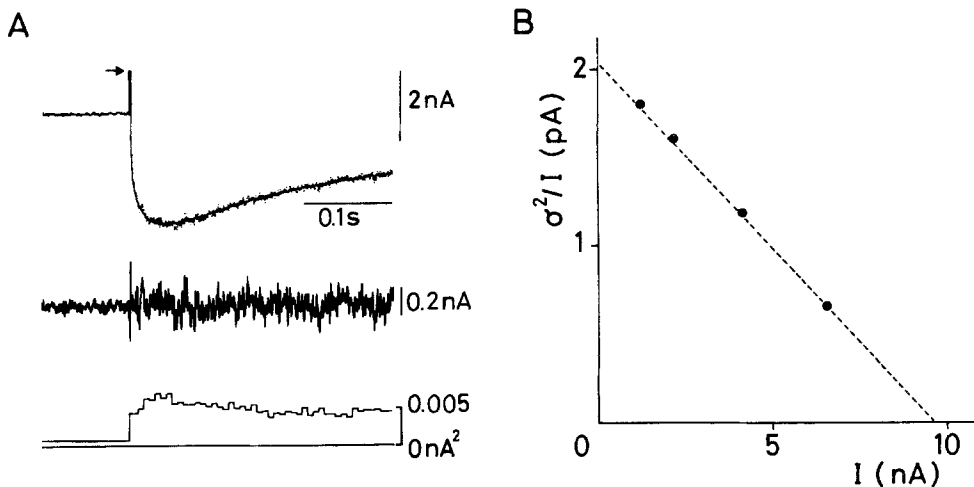
The peak inward current was plotted against the membrane potential after subtraction of the linear leakage current. Figure 4B illustrates the  $I-V$  relations measured in the same cell as in Fig. 4A at  $\text{pCa} = 5.5$  (diamonds),  $\text{pCa} = 6.4$  (triangles),  $\text{pCa} = 7$  (squares) and  $\text{pCa} = 7.5$  (circles). The membrane potential giving the maximum peak inward current shifted to the negative direction by 20–30 mV, when the external  $\text{Mg}^{2+}$  was removed and the external  $\text{Ca}^{2+}$  concentration decreased from  $1.8$  mM to  $\text{pCa} = 5$ . With further decrease of  $\text{Ca}^{2+}$  from  $\text{pCa} = 5$  to  $\text{pCa} = 7.5$ , there was a slight shift of less than 10 mV, the direction of which was not constant. In Ca-free solution containing  $0.1$  mM  $\text{MgCl}_2$  and  $0.1$  mM EGTA, the peak potential of the Na current was more negative by 10–20 mV than that of the original Ca current. The degree of the negative shift was decreased with increasing external  $\text{Mg}^{2+}$  from  $0.1$  to  $5.0$  mM.

To examine the voltage-dependency of the depressive effect of  $\text{Ca}^{2+}$  on the Na current, the ratio of the current amplitude to that with the lowest divalent concentration at the same membrane potential was calculated, but no consistent results could be obtained. For example, in the case illustrated in Fig. 4A and B, the depression increased as the membrane potential became positive. In another case, a U-shaped voltage dependency of relative amplitude was detected at lower  $\text{Ca}^{2+}$  concentrations ( $\text{pCa} > 6$ ) as reported by Fukushima and Hagiwara (1985), and there was no voltage dependency at  $\text{pCa} = 5$ .

Current amplitude was measured at the potential that gave the largest inward current at each concentration and expressed as the fraction of that at  $\text{pCa} = 7.5$  in cases of  $\text{Ca}^{2+}$  and as the fraction of that in K-free,  $1.8$  mM Ca solution in cases of  $\text{Mg}^{2+}$ . The relative amplitude was plotted against the  $\text{Ca}^{2+}$  or  $\text{Mg}^{2+}$  concentration in Fig. 4C. The curves which give the best fit to the data points were determined by the least squares method with the assumption that the Na current is blocked by one or two  $\text{Ca}^{2+}$  ions per channel. The one-to-one binding curve fits the data points better than the one-to-two binding curve, and so only one-to-one binding curves are illustrated in Fig. 4C. The dissociation constant is  $1.2 \mu\text{M}$  for  $\text{Ca}^{2+}$ . Although there are a few data points at  $\text{pMg} > 4$ , the data points for  $\text{Mg}^{2+}$  could be fitted by the one-to-one binding curve, with a dissociation constant of  $60 \mu\text{M}$ . Thus, the depression of the Na current by  $\text{Mg}^{2+}$  was one-fiftieth of that by  $\text{Ca}^{2+}$ .

#### Unit amplitude and total number of channels determined by noise analysis

The amplitude of the macroscopic current depends on three factors: the number of functional channels ( $N$ ), the probability that an individual channel is open ( $p$ ), and the current carried by an open channel ( $i$ ). To know which of these factors decreases with increasing external  $\text{Ca}^{2+}$ , I analyzed current fluctuations developing at lower  $\text{Ca}^{2+}$  concentrations.



**Fig. 5A, B.** Measurement of single channel current. **A** Na current in response to depolarizations from  $-63$  to  $-41$  mV at  $pCa = 6.4$  (plotted by point) shows current fluctuations during clamp pulses around the mean time course of the current (plotted by line; upper trace). The middle trace shows the residual fluctuations obtained by subtracting the mean time course of the current, which were made for individual currents by shifting vertically the average current, derived from all or half of fifty successive records, so that the variance of the residual fluctuations would be minimized. This procedure prevented error caused by slight long-term drifts in the current amplitude. Variance of the residual fluctuations was calculated for each sampling point and averaged for consecutive 15 sampling points during the depolarizing pulse and for consecutive 250 points at the holding potential. The bottom trace is the average variance from 43 sweeps. Membrane currents were fed via a four-pole Butterworth filter with a cut-off frequency of 1 kHz to a computer and digitized with a sampling interval of 0.4 ms. The arrow indicates the leakage current amplitude. **B** The variance-to-mean current relationship. Mean variance was obtained by averaging over about 300 successive time points in the middle of clamp pulses and subtracting the variance associated with the holding current. The peak amplitude of the average current was used for the mean current ( $I$ ) after subtraction of the linear leakage current. Mean variance and mean current obtained at different voltage steps spaced 2–3 mV apart were plotted. From an intersection of Y-axis and from a slope of a straight line, a single channel current amplitude of 2.04 pA (around  $-40$  mV) and channel number of 4700 per cell were calculated

The variance of current fluctuations ( $\sigma^2$ ) is a function of the mean amplitude of the current ( $I$ ), the unit amplitude ( $i$ ) and the total number of channels ( $N$ ), according to the following equation:

$$\sigma^2 = i \cdot I - (I^2/N)$$

where  $I = N \cdot i \cdot p$ .

To measure the unit amplitude of the Na current from the variance-to-mean current relationship (Sigworth 1980), fifty consecutive current records were obtained with a constant 300-ms voltage step from  $-63$  to  $-41$  mV every 6 s. After the mean time course of the current had been subtracted, the variance of the residual fluctuations was calculated (Fig. 5A). The mean variance increased slightly during the initial 50 ms and then decreased corresponding to the slight inactivation of the current (Fig. 3A). However, these changes are so small that variance-mean plots are not useful for estimating the unit amplitude. Thus, the same procedure was repeated at other voltage steps to  $-46$ ,  $-43$ , and  $-38$  mV. Even such a small change in the voltage step size caused fairly large current changes because of sharp voltage dependence of the current activation. The reversal potential of the Na current in Mg-free, low-Ca solution predicted from the single  $I-V$  relations is  $+10$  mV (Fig. 7C). Therefore, difference in the driving force is less than 8% among these test potentials, and so variation in the unit amplitude was neglected. The variance was averaged over about 300 successive time points in the middle of clamp pulses, and the variance associated with the holding current was subtracted. The peak amplitude of the average current was used for the mean current ( $I$ ) after subtraction of the linear leakage current. Then variance-mean plots were applied, as shown in Fig. 5B. The plot of  $\sigma^2/I$  versus  $I$  yields

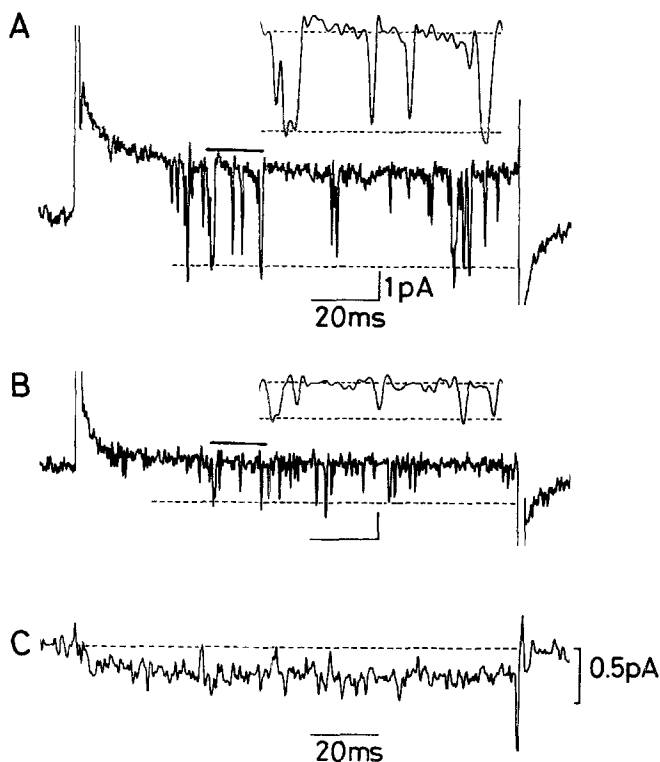
a straight line that intersects the Y-axis at  $i = 2.04$  pA (at  $pCa = 6.4$ , around  $-40$  mV) and that has a slope equal to  $-1/N$  ( $N = 4700$ ).

To repeat the constant clamp pulses at each  $Ca^{2+}$  concentration was so time-consuming that the experiment was frequently interrupted by deterioration of the cell and I could determine the single channel current at only one  $Ca^{2+}$  concentration, in each cell;  $i = 1.15$  pA and  $N = 4400$  at  $pCa = 5.5$ ,  $i = 0.87$  pA and  $N = 5600$  at  $pCa = 5.5$ , and  $i = 1.38$  pA and  $N = 4800$  at  $pCa = 5.9$ . These results suggest that it is not the number of functional channels, but rather the single channel current that is decreased with an increase of  $Ca^{2+}$ .

#### Single Na current through the Ca channel

Single Ca channel currents, with  $Na^+$  as charge carriers, were recorded from cell-attached membrane patches using the pipette filled with the solution containing 144 mM NaCl and buffered at  $pCa = 6$ ,  $pCa = 7.5$  and  $pCa > 8$ . Depolarization of the patch membrane from the resting membrane potential (RP; around  $-80$  mV in normal external solution; see Matsuda and Noma 1984) induced inward-going channel currents characterized by a short life time (less than 2 ms) and clustering of the channel openings, as shown in Fig. 6A ( $pCa = 7.5$ ) and B ( $pCa = 6$ ). That this current is different from the Na channel current is evident from the following findings: (1) the current was induced in the presence of TTX in the pipette solution, but was suppressed by adding 1  $\mu$ M D-600 to the bathing solution (not shown) and (2) the averaged current was inactivated little during 130 ms depolarizing pulse (Fig. 6C).

In accordance with results of noise analysis of the whole cell currents, unitary current amplitude at  $pCa = 7.5$



**Fig. 6 A–C.** Single channel currents recorded in the cell-attached configurations. **A, B** Current responses to 40 mV voltage steps from a resting potential of around  $-80$  mV (leakage and capacitive currents are not subtracted). Pipette solution contained  $63 \mu\text{M}$  TTX and was buffered at  $\text{pCa} = 7.5$  (**A**) and  $\text{pCa} = 6$  (**B**). Currents were low-pass filtered with a cut-off frequency of  $1.2$  kHz ( $-3$  dB) and sampled every  $0.2$  ms. The inset shows an expansion (by four times) of the part marked by the solid line. Unitary current amplitude and baseline level (in inset) are indicated by the dotted line. The cell was superfused with normal external solution. **C** Average current from the data in **B**. Fifty current sweeps were averaged after subtraction of leakage and capacitive currents. Zero current level is indicated by the dotted line

seemed larger than that at  $\text{pCa} = 6$ . The unit amplitude was measured from events which showed clear levels of open channel currents, though such events were not so frequent at  $\text{pCa} = 6$ . Average amplitude during depolarizing clamp steps of  $40$  mV from the resting potential is  $3.30 \pm 0.13$  pA at  $\text{pCa} = 7.5$  (five patches) and  $1.44 \pm 0.06$  pA at  $\text{pCa} = 6$  (six patches). However, if  $\text{Ca}^{2+}$  and  $\text{Na}^+$  compete for permeation through the Ca channel and blocking kinetics by  $\text{Ca}^{2+}$  is too fast to be detected, the unitary current amplitude might be reduced. Expecting that discrete blocking events were revealed by Bay K 8644, which has been reported to prolong the mean open time without affecting the unitary current amplitude (e.g. Hess et al. 1984), I added the drug to the bathing solution at a concentration of  $5 \mu\text{M}$  and studied the effect of  $\text{Ca}^{2+}$  on single Na currents.

The properties of single Na current through Ca channels recorded under the minimum influence of  $\text{Ca}^{2+}$  will be described briefly, and then the effect of  $\text{Ca}^{2+}$  will be described. The Ca-free ( $2$  mM EGTA) solution containing  $144$  mM NaCl ( $\text{pCa} > 8$ ) was used to minimize the effect of  $\text{Ca}^{2+}$ . Figure 7A shows voltage dependence of single Na current through Ca channels. Membrane potential was stepped by the indicated step size from the resting potential. Ca channels switch randomly between open and closed states, and do

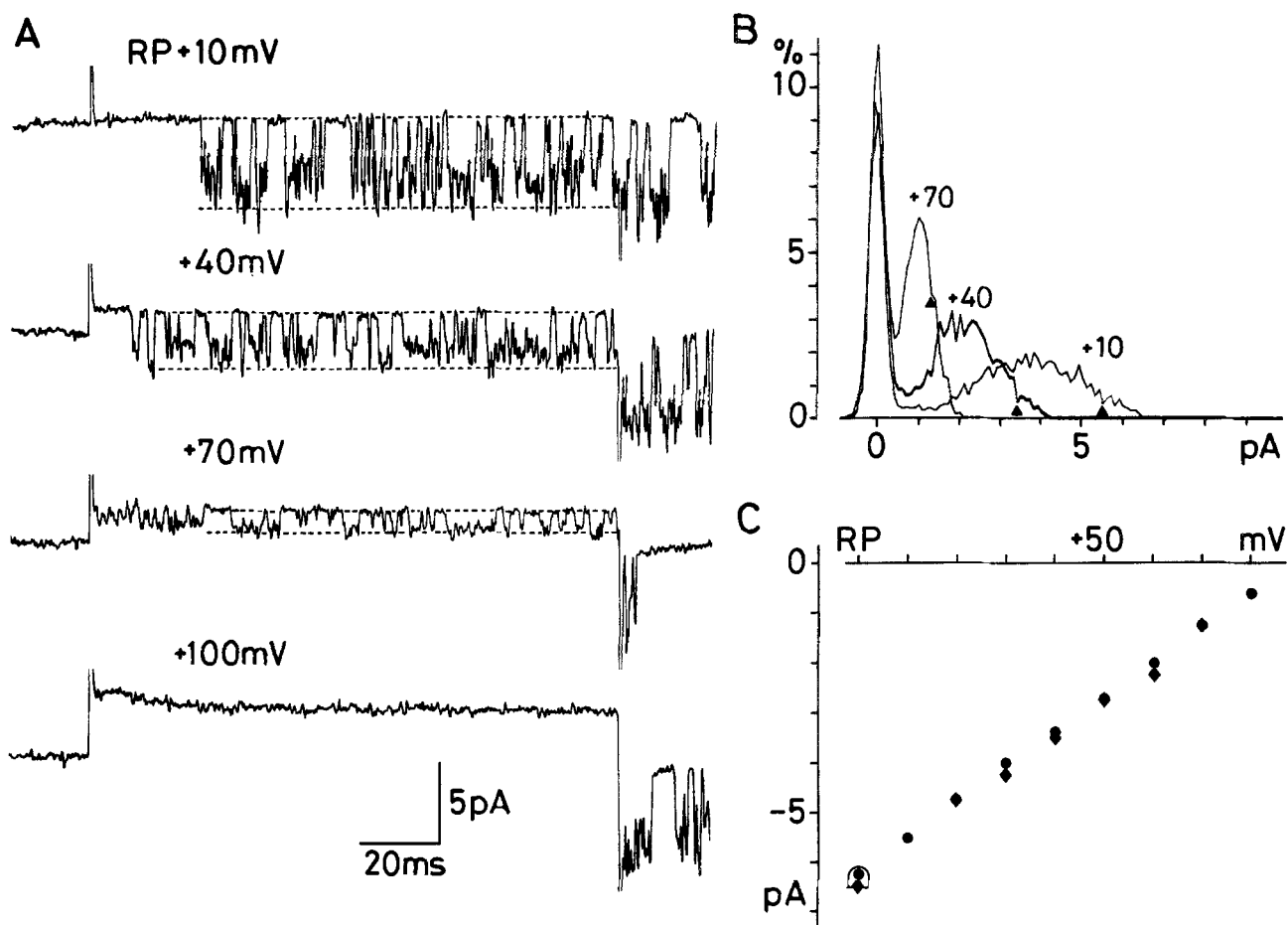
not dwell at a constant level during channel openings. As a result, a probability density histogram of the patch current obtained for the current traces in Fig. 7A showed a rather obtuse peak at the open current level compared to a sharp peak around zero current (Fig. 7B). Thus I decided to estimate the amplitude of unitary current simply by eye. The level of single channel currents is indicated by a dotted line on each trace in Fig. 7A, except for the bottom trace where no obvious events are observed. This level is indicated also in Fig. 7B by an arrow head. The current amplitude was plotted against voltage in Fig. 7C.  $I-V$  relationship is approximately ohmic with a slope conductance of about  $70$  pS. The reversal potential predicted from the  $I-V$  curve is  $+10$  mV (RP  $+90$  mV), though I did not apply depolarizing pulses larger than  $100$  mV and could not observe the outward current.

To analyze the kinetic characteristics, open time and closed time histograms were constructed (Fig. 8). The threshold for openings was set at about one-third the amplitude of the openings studied (inset). The open time histogram (Fig. 8A) was described by a single exponential function with a time constant ( $\tau_o$ ) of  $1.89$  ms (during depolarization of  $40$  mV). Neglecting a few events of the closed time longer than  $10$  ms, the closed time histogram was described by a single exponential function with a time constant ( $\tau_c$ ) of  $1.59$  ms (Fig. 8B). This represents the mean duration of the short gaps during bursts. If a burst is defined as any series of openings interrupted only by gaps shorter than  $4.5$  ms ( $= 4 \times \tau_c$  at  $\text{pCa} = 6$ , see below; Sakmann and Trube 1984), histograms of the burst durations and the gap durations were fitted by a single exponential function. In this case, the time constant fit for the histogram of the burst durations (the mean burst length) was  $28.7$  ms and the time constant fit for the histogram of the gap durations longer than  $4.5$  ms (the mean closed time between bursts) was  $10.1$  ms (not shown).

Figure 9 shows the current traces recorded from four different cells with pipette solution buffered at  $\text{pCa} = 7.5$  (Fig. 9A) and  $\text{pCa} = 6$  (Fig. 9B). Also in the presence of Bay K 8644, single channel currents at  $\text{pCa} = 6$  appeared smaller than that at  $\text{pCa} = 7.5$  during  $40$  mV depolarization from the resting potential. Currents after the repolarizing voltage step to the resting potential, however, had nearly the same amplitude. The current amplitude at the resting potential at  $\text{pCa} = 6$  averaged  $6.30 \pm 0.46$  pA (10 patches), which was plotted by open symbol in Fig. 7C. This result can be explained by assuming that the blocking effect of  $\text{Ca}^{2+}$  is voltage dependent and that repolarization relieves it. The slope conductance measured at  $\text{pCa} = 7.5$ ,  $70-75$  pS, was much the same with that at  $\text{pCa} > 8$  (Fig. 7C).

Kinetic characteristics of single Na currents through Ca channels measured during  $40$  mV depolarization at different  $\text{Ca}^{2+}$  concentrations are summarized in Table 2.  $\tau_o$  is remarkably decreased at  $\text{pCa} = 6$ .  $\tau_c$  and the burst length are also decreased, but to a lesser extent.

To analyze the fast kinetics of single channel current recordings, power density spectra of patch current fluctuations were calculated using a fast Fourier transform routine. Figure 10 shows a typical example obtained at  $\text{pCa} > 8$  and  $\text{pCa} = 6$ , during  $40$  mV depolarization from the resting potential. The data at  $\text{pCa} > 8$  were fitted well with the double Lorentzian function. The time constants associated with the low-frequency corner and the high-frequency corner are  $0.82$  ms and  $0.16$  ms, respectively. The former corresponds



**Fig. 7 A – C.** Voltage dependence of single Na currents through Ca channels. **A** Records in response to 130 ms depolarizing pulse. Membrane potential was stepped by the indicated step size from the resting potential (RP, about  $-80$  mV). The capacitive transient was reduced by analog compensation. Single channel current amplitude and baseline level are indicated by the dotted line. Ca-free pipette solution ( $pCa > 8$ ). Bath solution contained  $5 \mu M$  Bay K 8644. **B** Amplitude histogram obtained for the current traces in **A**. It showed an obtuse peak at the open channel level and the amplitude of single channel current was determined by eye (arrow head). **C** Single channel I–V relationships obtained from the same patch as in **A** and **B** (filled circles,  $70$  pS) and from another patch using the pipette solution buffered at  $pCa = 7.5$  (filled diamonds,  $75$  pS). Open symbol represents the current amplitude observed at  $pCa = 6$  after the repolarizing voltage step to the resting potential ( $6.30 \pm 0.46$  pA, 10 patches)

well with the time constant of  $0.86$  ms calculated from  $\tau_o$  and  $\tau_c$  presented in Fig. 8. The latter may represent the fast flickering kinetics of the single Na current through the Ca channel shown in Fig. 7A and inset of Fig. 8A. Both corner-frequencies showed no obvious potential dependency ( $212$  Hz and  $1080$  Hz during  $20$  mV depolarization,  $164$  Hz and  $864$  Hz during  $30$  mV depolarization and  $200$  Hz and  $952$  Hz during  $50$  mV depolarization). The same result was obtained in two other patches.

A slow component below  $50$  Hz was revealed in the spectra at  $pCa = 6$ . This might be ascribed to gating activities of bursts, because the burst length and closed time between bursts at  $pCa = 6$  were  $18$  ms and  $10$  ms, respectively (Table 2). The solid line in the spectra at  $pCa = 6$  was the best fit obtained by applying single Lorentzian function to the data between  $400$  and  $3000$  Hz. The corner-frequency was  $1584$  Hz in this case and the time constant associated with the averaged corner-frequency ( $1505 \pm 95$  Hz, 5 patches) is  $0.10$  ms.

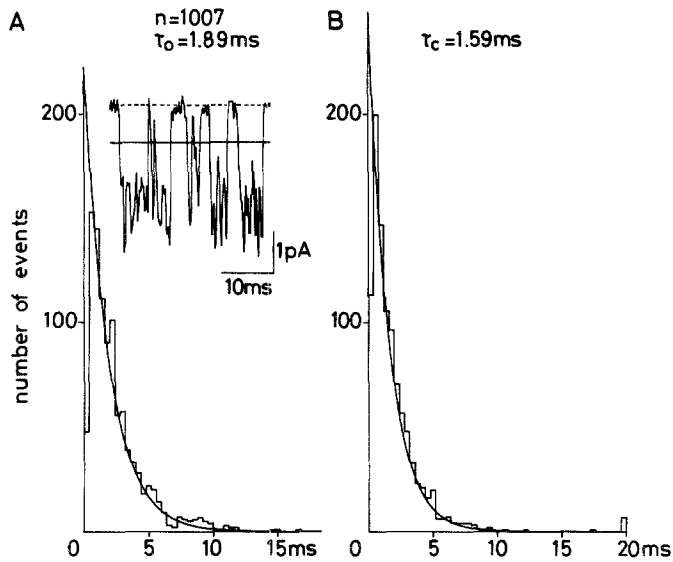
### Discussion

In cardiac cells, the Na current in the absence of external  $Ca^{2+}$  decays much more slowly than the Ca current. This is in contrast to the monovalent currents through Ca channels

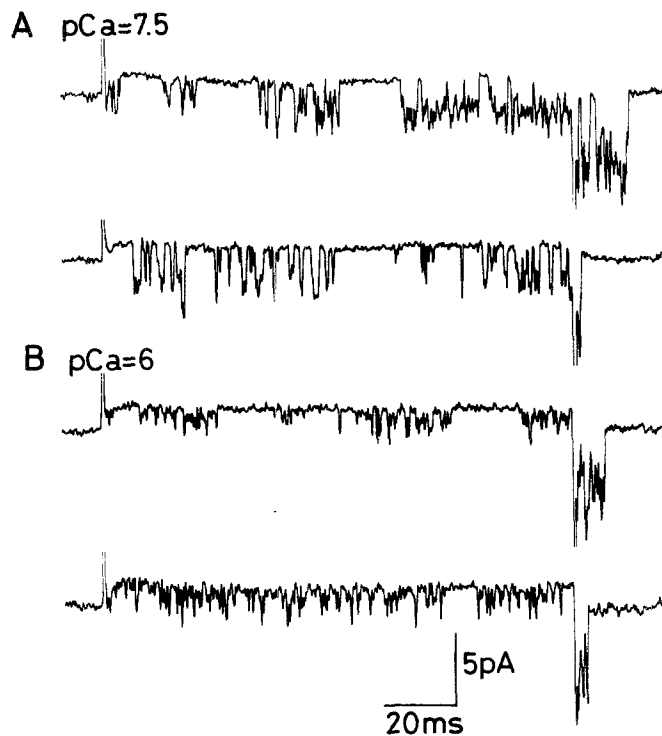
in other excitable cells, which show a similar decay time course to the Ca current (Kostyuk et al. 1983; Almers and McCleskey 1984). Nevertheless, the suppression by Ca channel blockers (Rougier et al. 1969; Garnier et al. 1969) and augmentation by epinephrine (in this work) strongly suggest that the cardiac Na current is also carried through Ca channels. The slow inactivation of Ca channels after Ca chelation suggests that the major inactivation mechanism of the Ca channel is not operating when  $Na^+$  is the charge carrier. This is consistent with the view that inactivation of the Ca channel results from Ca entering upon depolarization rather than from the membrane potential effects alone (Ca-mediated inactivation of the Ca conductance: e.g. Eckert and Tillotson 1981; in the cardiac fibers, Mentrard et al. 1984; Kokubun and Irisawa 1984; Lee et al. 1985).

Another difference is that the cardiac Ca channel is less permeable to  $Li^+$  than to  $Na^+$ . The Ca channels, under conditions of reduced external  $Ca^{2+}$  concentrations, are reported to be equally permeable to  $Li^+$  and  $Na^+$ : in molluscan neurones the  $Li^+ : Na^+$  permeability ratio is  $0.8$  (Kostyuk et al. 1983) and, in frog skeletal muscle, it is  $1$  (Almers et al. 1984). However, in cardiac cells,  $Li^+$  cannot carry such a large inward current as  $Na^+$  (Chesnais et al. 1975; Lansman et al. 1985). In the present study, the





**Fig. 8.** Distribution of open time (A) and closed time (B) at  $pCa > 8$  during 40 mV depolarization from the resting potential. The same patch and series of records as in Fig. 7. A threshold level for discrimination of open and closed states was set at around one-third of the open level, as shown by the solid line in the inset. The dotted line indicates the baseline level. The open time histogram formed in 0.4 ms bins was fitted with a single exponential function with a time constant of 1.89 ms (A). The closed time histogram was constructed with 0.4 ms bins and fitted with a single exponential function with a time constant of 1.59 ms, neglecting a few events of the closed time longer than 10 ms (B). Bath solution contained 5  $\mu M$  Bay K 8644

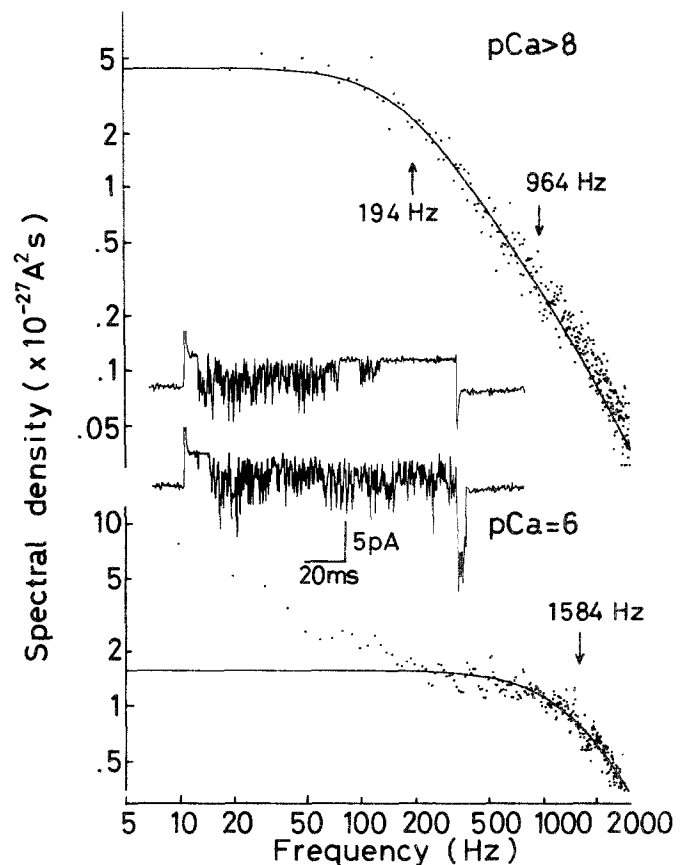


**Fig. 9 A, B.** Voltage-dependent block by  $Ca^{2+}$  of single Na currents through Ca channels. Each record was obtained from different patches, using the pipette solution indicated. Bath solution contained 5  $\mu M$  Bay K 8644. Single channel currents at  $pCa = 6$  appeared at a less magnitude and a shorter duration than at  $pCa = 7.5$  during 40 mV depolarization from the resting potential. However, currents after the repolarizing voltage step to the resting potential are much the same amplitude as that at  $pCa = 7.5$

**Table 2.** Kinetic characteristics of single Na currents through Ca channels during 40 mV depolarization at different  $Ca^{2+}$  concentrations

	$pCa > 8$	$pCa = 7.5$	$pCa = 6$
Open time ( $\tau_o$ ; ms)	$1.94 \pm 0.31$	$1.48 \pm 0.11$	$0.26 \pm 0.05$
Closed time within bursts ( $\tau_c$ ; ms)	$1.52 \pm 0.07$	$1.44 \pm 0.12$	$1.17 \pm 0.17$
Burst length (ms)	$26.5 \pm 3.3$	$24.9 \pm 1.5$	$18.3 \pm 1.2$
Closed time between bursts (ms)	$9.7 \pm 1.3$	$12.3 \pm 6.0$	$10.0 \pm 3.5$

Analysis was done in patches containing one active channel (three patches at  $pCa > 8$  and  $pCa = 7.5$  and six patches at  $pCa = 6$ ). A burst was defined as any series of openings interrupted only gaps shorter than 4.5 ms ( $= 4 \times \tau_c$  at  $pCa = 6$ ; Sakmann and Trube 1984)



**Fig. 10.** Power density spectra of patch current fluctuations measured during 40 mV depolarization from the resting membrane potential. Bath solution contained 5  $\mu M$  Bay K 8644. The spectra at  $pCa > 8$  fitted well with the double Lorentzian function. The corner-frequencies of slow and fast component, 194 and 964 Hz, are indicated by arrows. The zero frequency asymptotes of the slow and fast components were  $4.16 \times 10^{-27}$  and  $2.10 \times 10^{-28}$   $A^2 s$ , respectively. The same patch as in Fig. 7 and 8. The spectra at  $pCa = 6$  was obtained from patch containing several active channels (inset). The solid line was the best fit obtained by applying single Lorentzian function to the data between 400 and 3000 Hz. The corner-frequency was 1584 Hz and the zero-frequency asymptote was  $1.56 \times 10^{-27}$   $A^2 s$ . The slow component below 50 Hz of power density spectra was ascribed to the bursting activities of the channels

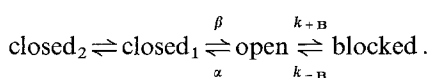
amplitude of the Ca channel current decreased to 30% after total replacement of the external  $\text{Na}^+$  with  $\text{Li}^+$  (Fig. 2). Such a diminished permeability of the Ca channel to  $\text{Li}^+$  than to  $\text{Na}^+$  was observed also in case of gastrointestinal smooth muscle (Prosser et al. 1977) and uterine smooth muscle (Mironneau et al. 1982), where the prolonged action potential induced by Ca chelation disappeared after replacing  $\text{Na}^+$  with  $\text{Li}^+$ .

Despite these differences, the Na current is depressed by  $\text{Ca}^{2+}$  of a micromolar level as is the case in other excitable cells. Hess and Tsien reported that 1.3  $\mu\text{M}$  external  $\text{Ca}^{2+}$  lower the Na influx through the Ca channel to  $33 \pm 4\%$  in single ventricular cells of guinea-pig (1984). In the present work, it was shown that depression of the Na current by external divalent cations can be described by a bimolecular interaction between a single ion and channel. The dissociation constant of 1.2  $\mu\text{M}$  for  $\text{Ca}^{2+}$  and 60  $\mu\text{M}$  for  $\text{Mg}^{2+}$  agrees well with the result of Hess and Tsien (1984) and with those obtained in other excitable cells: 0.3  $\mu\text{M}$  and 60  $\mu\text{M}$  in *Helix* neurones (Kostyuk et al. 1983), 2  $\mu\text{M}$  and 30  $\mu\text{M}$  in mouse lymphocytes (Fukushima and Hagiwara 1985) and 0.7  $\mu\text{M}$  for  $\text{Ca}^{2+}$  in frog skeletal muscle (Almers et al. 1984).

Prevention of Na movements through the Ca channel by micromolar level of  $\text{Ca}^{2+}$  has been attributed to a binding site with a high affinity for  $\text{Ca}^{2+}$  which regulates the ion selectivity of the Ca channels (Kostyuk et al. 1983). When Ca chelation makes this site free, the Ca channel acquires a permeability to  $\text{Na}^+$  that is not present at physiological  $\text{Ca}^{2+}$  concentrations.

There is no direct evidence as to where such a Ca-binding site locates. From the finding that reduction in the Na current by  $\text{Ca}^{2+}$  is not voltage-dependent, Kostyuk et al. (1983) concluded that this high-affinity Ca-binding site is located near the external surface of the membrane and that Ca binding to this site causes conformational changes in the ion-selecting filter. On the other hand, Hess and Tsien (1984) and Almers and McCleskey (1984) assumed two high affinity sites within the channel, thereby applying to the Ca channel the multi-ion single-file pore model described by Hille and Schwarz (1978) for the K channels. Their model can account for competitive interaction between  $\text{Ca}^{2+}$  and  $\text{Ba}^{2+}$  (anomalous mole fraction effect) as well as the prevention of Na permeation by micromolar level of  $\text{Ca}^{2+}$ . In the present work, no consistent results could be obtained on the voltage-dependency of the depressive effect of  $\text{Ca}^{2+}$  in the whole cell clamp experiments. However, unitary currents during 40 mV depolarization were smaller at  $\text{pCa} = 6$  than at  $\text{pCa} = 7.5$  or  $\text{pCa} > 8$  and currents after repolarizing step to the resting potential had nearly the same amplitude. This finding can be interpreted that upon repolarization the Ca bound to the site within the channel enters the interior of the cell to relieve the block, thereby favoring location of a binding site in the channel.

The noise analysis of single channel currents at  $\text{pCa} = 6$  revealed a high-frequency component at around 1500 Hz, indicating fast blocking kinetics. Therefore, a decrease in the single channel current amplitude at  $\text{pCa} = 6$  is most likely to be an apparent decrease attributed to a rapid block (Yellen 1984). If we assume that the block is a first order process, a simple open ion channel block can be described as:



In this situation, there are two ways out of the open state (closing or blocking) with transition rates  $\alpha$  and  $k_{+B} \cdot (\text{Ca}^{2+})_p$ , respectively, where  $(\text{Ca}^{2+})_p$  is the Ca concentration of the pipette solution. Thus, the mean open life time is given  $1/[\alpha + k_{+B} \cdot (\text{Ca}^{2+})_p]$  (Colquhoun and Hawkes 1983). Because the blocking effect of  $\text{Ca}^{2+}$  is not expected at  $\text{pCa} > 8$ , we can obtain the approximate value of  $\alpha$ ,  $526 \text{ s}^{-1}$ , from the relation of  $\tau_o$  (1.9 ms, at  $\text{pCa} > 8$ ) =  $1/\alpha$ . Using this and  $\tau_o$  at  $\text{pCa} = 7.5$  (1.5 ms) and  $\text{pCa} = 6$  (0.26 ms), we can obtain the blocking rate of  $141 \text{ s}^{-1}$  at  $\text{pCa} = 7.5$  and  $3320 \text{ s}^{-1}$  at  $\text{pCa} = 6$ . The latter value is smaller than that expected from the linearly proportional relation to the Ca concentration ( $4460 \text{ s}^{-1}$ , see below). However, this discrepancy might be ascribed to an overestimation of  $\tau_o$  at  $\text{pCa} = 6$ , because the data were low-pass filtered with a cut-off frequency of 1.4 kHz ( $-3 \text{ dB}$  frequency). From the blocking rate at  $\text{pCa} = 7.5$ , we can obtain the value of  $k_{+B}$  at around  $-40 \text{ mV}$ ,  $4.46 \times 10^9 \text{ M}^{-1} \text{ s}^{-1}$ . Using the dissociation constant ( $K_B$ ) at around  $-30 \text{ mV}$  obtained from the whole cell current recordings (Fig. 4),  $k_{-B}$ ,  $5352 \text{ s}^{-1}$  was calculated from the relation  $K_B = k_{-B}/k_{+B}$ . Because the closing rate  $\alpha$  and opening rate  $\beta$ , which are estimated from  $\tau_o$  and  $\tau_c$  at  $\text{pCa} > 8$ , are about one order of magnitude smaller than unblocking rate  $k_{-B}$  and blocking rate at  $\text{pCa} = 6$ , we can calculate an approximate time constant of spontaneous fluctuations from the equilibrium between blocking and unblocking of the Na current as  $1/[k_{-B} + k_{+B} \cdot (\text{Ca}^{2+})_p]$ . The resulting value (0.10 ms) is identical to that obtained from noise analysis.

At  $\text{pCa} > 8$ , the current level of open channels shows flickering. Noise analysis reveals that a fast incomplete closing and opening kinetics with a corner frequency around 1 kHz operates in addition to the dominant slower gating kinetics. The fast flickering kinetics in Ca channels lacking an apparent voltage dependency has been noted also in chick vestibular hair cells (Ohmori 1984). It is interesting that such fluctuations of the single open channel are not obvious with  $\text{Li}^+$  or  $\text{Ba}^{2+}$  as charge carriers (see Lansman et al. 1985 for  $\text{Li}^+$ ). Some species of ions might block their own passage through the Ca channels.

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