

The Effect of Salinity on the Acute Toxicity of Cadmium to the Tropical, Estuarine, Hermaphroditic Fish, *Rivulus marmoratus*: A Comparison of Cd, Cu, and Zn Tolerance with *Fundulus heteroclitus*

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Abstract. The mangrove-dwelling fish, *Rivulus marmoratus*, is the only vertebrate that is a synchronous, internally self-fertilizing hermaphrodite. This unique reproductive mode yields offspring with little genetic variation, which offers significant advantages for the use of this species in bioassays. We conducted acute (96 h) LC₅₀ tests of Cd toxicity under four different water chemistry conditions, representing fresh water (low [Ca + Mg] and low [Na + K]), 14 ppt sea water simulated with Cl salts (high [Ca + Mg] and high [Na + K]) and two artificial conditions (high [Ca + Mg], low [Na + K] and low [Ca + Mg], high [Na + K]). Two replicates were conducted at different times for each of the four treatments and the results were very reproducible. The mean LC₅₀'s as mg total Cd/L were 2.96 (fresh water), 21.12 (high [Ca + Mg], high [Na + K]), 17.86 (high [Ca + Mg], low [Na + K]) and 12.67 (low [Ca + Mg], high [Na + K]). An additional test in 14 ppt sea water (made up from Instant Ocean™ salts) yielded a 96h LC₅₀ of 24.48 mg Cd/L, and was thus similar to the high [Ca + Mg]-high [Na + K] treatment, despite some differences in anion and cation concentrations. The degree to which [Ca + Mg], [Na + K] and [Cl] interact to determine Cd toxicity is still unclear, although the role of [Cl] is likely to be equal to or greater than that of [Ca + Mg]. When all solutes are high, it is likely that the formation of a Cd complex with Cl (248 mM) leads to the observed reduction in Cd toxicity in comparison with hard fresh water, not the increased divalent [Ca + Mg] levels. Further tests are needed in which the various cations and anions are varied as independently as possible in order to determine their roles in affecting free [Cd], or interacting with the gills to reduce Cd toxicity.

Interstrain variation in Cd toxicity was marked; four strains of *R. marmoratus* were exposed to three levels of Cd (LC₂₅, LC₅₀ and LC₇₅ estimated from tests of high-[Ca + Mg], high-[Na + K] conditions). Strain DS was consistently the most sensitive, whereas strain NA was consistently the most resistant to Cd across the three Cd exposure levels. Thus we recommend use of a single strain to conduct toxicity tests, or randomization of strains across treatments.

Two replicate 96h LC₅₀ tests of each of the three heavy metals, Cd, Cu and Zn, for *Rivulus marmoratus* were found to be similar to values for *Fundulus heteroclitus*, a standard estuarine bioassay fish. The mean LC₅₀'s of Cd, Cu and Zn at 14

ppt for *Rivulus* were 21.1, 1.4 and 147.9 mg/L, respectively, whereas those for *Fundulus* were 18.2, 1.7 and 129.5 mg/L. This similarity in toxicological response by two species from different biogeographic regions (tropical vs temperate) may be due to similar physiological and biochemical adaptations to the abiotically stressful estuary.

The atheriniform fish, *Rivulus marmoratus* (Cyprinodontidae) found in mangrove areas ranging from southern Florida throughout the Caribbean to central America is remarkable for its synchronous, internally self-fertilizing hermaphroditism (Harrington 1961). It is a unique vertebrate in that it reproduces sexually and uniparentally. It was completely homozygous for the 31 loci studied by Vrijenhoek (1985) and clonally stable for at least three generations as demonstrated by single-sequence DNA fingerprinting (Turner *et al.* 1990). Early studies (Kallman and Harrington 1964; Harrington and Kallman 1968) have shown the complete acceptance of tissue transplantation between parent and offspring, and between siblings. This highly inbred reproductive mode generates little genetic variation within each parental line and makes this species an ideal organism for use in bioassays (Davis 1986). *Rivulus marmoratus* matures within 2–3 months in small bowls and then produces eggs on a weekly basis throughout the year. Depending on the size and age of the fish and the amounts of food provided, *R. marmoratus* can produce up to 20 eggs per day (Lin and Dunson, unpublished data). *Rivulus marmoratus* has been used as a subject for teratogenic studies (Lindsey and Harrington 1972; Harrington and Crossman 1976; Davis 1988) and is one of a number of fish species that has been used for cancer research (Koenig and Chasar 1984; Park and Kim 1984; Thiyagarajah and Grizzle 1985, 1986). However, it has not been widely used as a bioassay species.

The toxicity of Cd to estuarine animals varies with salinity (Eisler 1971; Westernhagen *et al.* 1974; Jones 1975; Westernhagen and Dethlefsen 1975; Theede 1980; McLusky *et al.* 1986; Reardon and Harrell 1990; Forbes 1991). Usually, the LC₅₀ increases as the salinity increases; one interpretation is that metals are simply less toxic at increased salinities due perhaps to interactive effects with ions such as Ca at the target

Table 1. Water chemistry during Cd tests. Concentrations in mg/L (mM in parentheses) are shown as the average of the two tests of the same treatment. The [Mg]/[Ca] ratios (as mM) of tests 1 to 6 and test 9 were set at 5–7.5, similar to that of sea water. [Cl] was estimated as [Na + K + 2(Ca + Mg)], except for test 9 which was a close simulation of sea water and contains anions other than Cl (Cl_{est} was thus calculated from the Cl/Na ratio of sea water). In the first column, high level of ions was abbreviated as H and low as L

Test no./Treatment	[Na]	[K]	[Ca]	[Mg]	[Cl] _{est}
1,2.H-Ca, H-Na + K	4224 (184)	231 (6)	157 (4)	614 (25)	8792 (248)
3,4.H-Ca, L-Na + K	12 (0.6)	1.2 (0.03)	178 (4)	618 (25)	2140 (60)
5,6.L-Ca, H-Na + K	4789 (203)	202 (5)	35 (0.8)	135 (6)	7842 (221)
7,8.L-Ca, L-Na + K	11 (0.5)	1.0 (0.02)	25 (0.6)	18 (0.8)	114 (3.2)
9. Instant Ocean [®]	4465 (194)	213 (6)	192 (5)	618 (25)	8037 (226)

tissue site. An alternative explanation is that the proportion of the total metal that is toxic (the “free” fraction) diminishes. This generally inverse relation of toxicity with salinity has been described for Zn and Pb (Jones 1975), for Cu and Ag (Coglianese 1982), and for 20 metals in 30 compounds (Dorfman 1977). These papers give very detailed discussions of the effects of varying water hardness and salinity on reducing metal toxicity. However, the interactive effects of non-toxic anions and cations with Cd toxicity over a range of Na/Ca ratios are relatively unexplored.

The purpose of this study was to evaluate the use of *R. marmoratus* as a bioindicator for tropical estuarine pollution. To do so, we (1) conducted acute (96 h) Cd tests under different water chemistry concentrations, (2) examined the role of non-toxic ions (Na + K, Ca + Mg, and Cl) in changing Cd toxicity, (3) examined interstrain variation in response to Cd, and (4) compared the 96h LC₅₀'s of Cd, Cu and Zn between *R. marmoratus* and *Fundulus heteroclitus*, a standard bioassay species for temperate zone estuarine ecotoxicological studies.

Materials and Methods

The Organisms

The fish stocks originated in Florida, USA, and Yucatan, Mexico, and were obtained from several suppliers. Five different inbred strains were obtained from Dr. K. Kallman at the Genetics Laboratory of the New York Aquarium; these came originally from Dr. R. Harrington, the first to describe self-fertilizing hermaphroditism in this species (Harrington 1961). These fish have been kept in captivity for 13–40 generations.

Our colony was established in May 1989, following standard procedures for cultivation (Koenig and Chasar 1984). Fish stocks and all experiments were maintained at 26–27°C and a photoperiod of 14L:10D. Adults were kept individually in 11.4 cm (4.5-in.) diameter stackable glass culture dishes (Carolina Biological) filled with 150–200 ml of approximately 12.5 ppt (≈33%) sea water made from a commercial synthetic mix (Instant Ocean[™] powder). Newly hatched *Artemia* nauplii were fed to the fish at least 5 days a week to ensure good growth and reproduction, and the water was changed every 4–5 weeks. Eggs were incubated in dechlorinated tap water and transferred to 12.5 ppt sea water 5–10 days after hatching. Hatchlings about 4–6 weeks old (wet mass = 0.03–0.1 g; length = 11–18 mm) were used for testing.

Adult *Fundulus heteroclitus* were collected from an impoundment on Assateague Island, Virginia. Eggs and sperm were obtained by stripping the fish (three males and six females), as described by Costello *et al.* (1957). Fertilized eggs were incubated in 12.5 ppt sea water and hatching was stimulated by replacing this sea water with dechlorinated hard water. Hatchlings were maintained in 12.5 ppt sea water

and fed with *Artemia* nauplii for 4–6 weeks before testing (wet mass = 0.02–0.13 g; length = 12–20 mm).

The Flow-Through Test System

This consisted of six bottles of 20 L stock solutions (one control and five levels of toxicant) connected to six 2.4 L plastic test chambers. For each 96 h acute toxicity test, 60 *R. marmoratus* were used; 10 in each container. The flow rate was controlled by a peristaltic pump (approximately 2–3 ml/min), with 90% replacement of solution in the container within 48 h (Sprague 1969).

Water samples taken at the beginning and the end of each test were filtered (0.1 μm pore size) and acidified for Cu and Zn tests only (see below for Cd protocol) with Ultrex[™] (ultrapure reagent nitric acid) and analyzed using a Perkin-Elmer model 2280 atomic absorption spectrophotometer (air acetylene flame). These analyses gave total, non-particulate metal concentration.

Cadmium tests

To determine the effects of water cation concentrations on Cd toxicity, 96 h LC₅₀ tests were conducted under five different water chemistry conditions. Four combinations of two levels of [Ca + Mg] and two levels of [Na + K] were tested and made to simulate 14 ppt sea water (≈40% SW, designated high [Ca + Mg], high [Na + K]), hard fresh water (low [Ca + Mg], low [Na + K]) and two artificial conditions (high [Ca + Mg], low [Na + K], and low [Ca + Mg], high [Na + K]). Note that concentrations of the divalent cations (Ca and Mg) and the univalent cations (Na and K) were each varied simultaneously, although the treatment designation (Table 1) refers only to the former member of each pair; with two replicate tests for each condition. Chloride salts (NaCl, KCl, CaCl₂, and MgCl₂) were used exclusively. One additional test was made, using Instant Ocean[™] salts to make up approximately 40% sea water. Water chemistry parameters are summarized in Table 1. The toxicant used was cadmium sulfate (3CdSO₄ · 8H₂O).

There was no difference in Cd concentrations of filtered and unfiltered samples from several pilot tests and no sign of decline in total Cd concentrations over time for any test conducted. Thus, we consequently acidified and measured Cd in samples without filtration. Several studies have also shown no loss of Cd concentration over time (Jackim *et al.* 1970; Eisler 1971; Eisler and Gardner 1973; Eisler and Hennekey 1977; Pascoe *et al.* 1986).

Interstrain Test

The four strains of *R. marmoratus* represented progeny of a single parental fish derived from Vero Beach, Florida (DS and NA), Miami, Florida (M) and Yucatan, Mexico (E). Fish were assigned to three levels of Cd concentrations, representing the average 96-h LC₂₅, LC₅₀,

and LC_{75} of the three high Ca + Mg, high Na + K tests (tests 1, 2, and 9) estimated by the probit procedure, and a control containing Instant Ocean™ salts at 14 ppt (no cadmium). A total of 256 fish [(8 fish/container) × (2 containers/conc./strain) × (4 conc.) × (4 strains)] were to be used for this 10-day test. However, strain DS only had 14 individuals (instead of 16 as originally planned) for each of the three Cd concentrations. Also, one fish from strain M at the LC_{75} treatment level was lost at the beginning of the experiment. Within each strain, the fish were either from the same parent or from the same grandparent. There was no feeding during the test. Water samples were collected on days 0, 3, 7, and 10. The means ± SD of Cd concentrations were 20.66 ± 0.41 , 23.88 ± 0.32 , and 27.59 ± 0.15 mg total Cd/L for the LC_{25} , LC_{50} , and LC_{75} levels, respectively. Relatively small standard deviations indicated that the Cd concentrations were nearly constant.

Cd, Cu and Zn Tests

The same experimental flow-through device as described for the Cd toxicity tests was used to examine the toxicity of Cd, Cu, and Zn for *F. heteroclitus*, and Cu and Zn for *R. marmoratus*. Zinc sulfate heptahydrate ($ZnSO_4 \cdot 7H_2O$) and cupric sulfate pentahydrate ($CuSO_4 \cdot 5H_2O$) were used. Two tests were made for *R. marmoratus* and one for *F. heteroclitus* for each metal. Except for two Cd tests for *R. marmoratus*, all 96-h LC_{50} tests were performed in 40‰ sea water (≈ 14 ppt) made with Instant Ocean™ salts. Both Zn and Cu precipitated to varying degrees during the experiment. The samples were filtered at 0.1 μm (pore size) to remove particulates and acidified before analysis on an atomic absorption spectrophotometer.

Data Analyses

All LC_{50} 's were derived by the maximum likelihood estimate calculated with a SAS probit procedure (SAS Institute 1985) and are presented as mg total metal concentration per liter. We used the SAS GLM procedure (SAS Institute 1985) to derive a two-way analysis of variance (ANOVA) table and evaluate the effects of cations (Ca and Na) on the LC_{50} 's of Cd for *R. marmoratus*. The concentrations of Mg and K were kept at a fixed ratio with those of Ca and Na, respectively, and these ratios reflect those of seawater and hard fresh water conditions. This two-way ANOVA represents the effects of divalent and univalent cations. The 96-h LC_{50} values were normalized by a log transformation for the analysis of variance.

Life tables and survivor functions of BMDP 1L (BMDP 1990) were used to estimate mean survival time and to analyze interstrain variation. Since there is a predictable negative relationship between [Cd] and survival time, the different survivorship of fish among the three Cd levels was not of primary interest.

To compare the LC_{50} 's between two tests when their 95% fiducial limits did not overlap, the trimmed Spearman-Kärber method (Hamilton et al. 1977) was employed, using the procedures suggested by the Water Quality Standards Handbook (EPA 1983). The required natural logarithm (ln) of LC_{50} and its variance were obtained and transformed from the SAS-probit output, designated as "mu" and "sigma" expressed as a \log_{10} value.

Results

Tests of Cadmium Toxicity Under Various Water Chemistry Conditions

Both [Ca] and [Na] and their interaction had significant effects on Cd toxicity (Table 2). As electrolyte concentrations in-

Table 2. Analysis of variance of effects of electrolyte levels on Cd toxicity. The effect of cation levels were confounded with that of Cl. Analyzed 96 h LC_{50} values were log transformed. df = degree of freedom, SS = Type III Sum of square

Source	df	SS	F value	p-value
Ca	1	0.571	235.97	0.0001
Na	1	0.286	117.91	0.0001
Ca × Na	1	0.158	65.32	0.0005
Error	5	0.012		

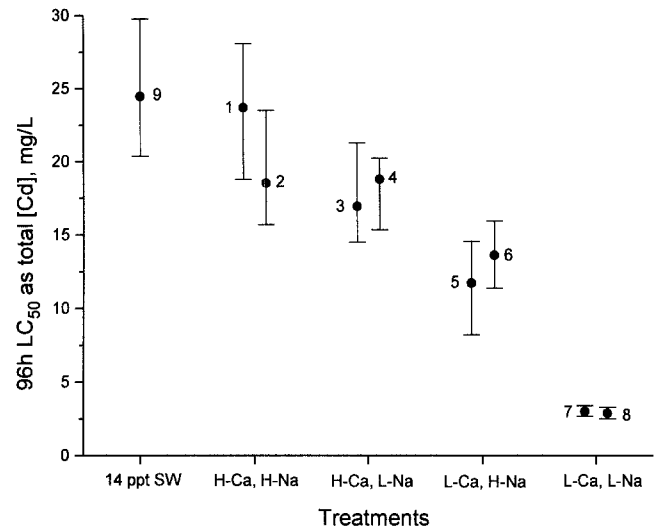


Fig. 1. The acute Cd toxicity for *R. marmoratus* expressed as concentration of total cadmium under various water chemistry conditions. The solid circle is the 96-h LC_{50} of each test, and the bars attached to it are the 95% fiducial limits. The numbers designate each test and correspond to those in the text and Table 1. H = high level, L = low level. SW = made with Instant Ocean™ sea water salts

creased, the LC_{50} values increased; i.e. Cd became less toxic (Figure 1). In tests 1 to 4 [Na + K] was varied, but [Ca + Mg] held at approximately the same high level. The four LC_{50} 's overlapped (Figure 1) and the numerical values of tests 1 to 4 were 23.70, 18.53, 16.93, and 18.79 mg/L, respectively. When [Na + K] was held at high levels and [Ca + Mg] varied (tests 1, 2, 5, and 6), the LC_{50} 's of tests 1 and 5 (the two most extreme of the four) were not statistically different ($p > 0.05$) using a trimmed Spearman-Kärber method (EPA 1983). Tests 5 and 6 had estimated LC_{50} values of 11.73 and 13.60 mg/L, respectively. Tests conducted at low electrolyte concentrations had significantly lower LC_{50} values (3.03 and 2.89 mg/L for tests 7 and 8, respectively).

The overlaps in fiducial limits between tests 9 ($LC_{50} = 24.48$ mg/L) and tests 1 and 2 show the similarity in results when using Instant Ocean™ salts and similar overall concentrations of chloride salts only (Figure 1).

When Ca, Na, and Cl were plotted individually against the 96h LC_{50} values, the patterns were different (Figure 2). Tests 5 and 6 had higher LC_{50} values than tests 7 and 8 probably because of the higher Na (203 vs 0.5 mM in Table 1) and Cl levels (221 vs 3.2 mM). The potassium concentrations of these four tests were very low in comparison with those of Na and Cl. In contrast, tests 3 and 4 had higher LC_{50} values than tests 7 and

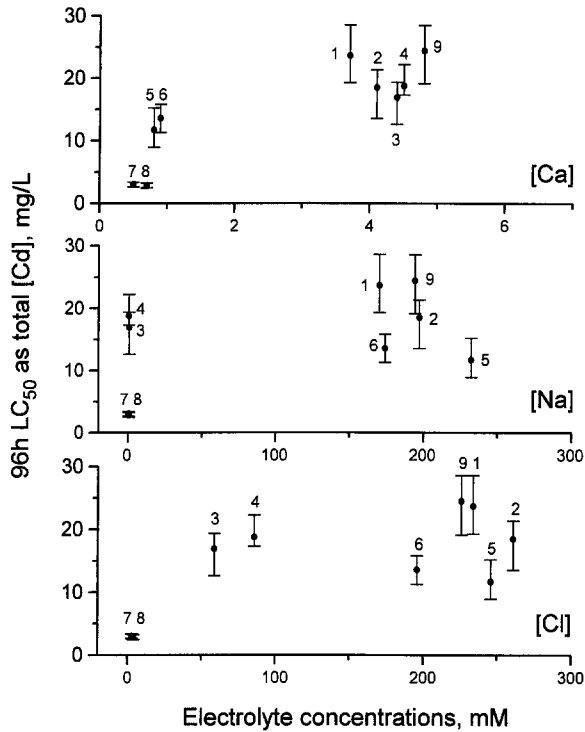


Fig. 2. The relation between the 96-h LC₅₀ of Cd for *R. marmoratus* and concentration of each of the three non-toxic ions (Ca, Na, and Cl). The solid circle and the bars attached to it represent the LC₅₀ and its 95% fiducial limits, respectively. The numbers are test numbers corresponding to Table 1 and Figure 1

Table 3. Variation of tolerance among four strains of *Rivulus marmoratus* held at three levels of Cd for 10 days. The Cd levels were the 96h LC₂₅, LC₅₀, and LC₇₅ estimated from earlier tests using the pooled fish stocks. Values are mean survival time (days) ± SE

Strain	Cd concentrations (mg/L)		
	20.66 (LC ₂₅)	23.88 (LC ₅₀)	27.59 (LC ₇₅)
DS	3.86 ± 0.51	3.50 ± 0.14	2.64 ± 0.13
E	5.75 ± 0.76	5.50 ± 0.80	4.00 ± 0.51
M	9.25 ± 0.48	4.75 ± 0.58	2.93 ± 0.12
NA	9.25 ± 0.56	8.25 ± 0.84	4.75 ± 0.60

8 probably because of the higher Ca + Mg levels (29 vs 1.4 mM) and Cl levels (60 vs 3.2 mM).

Interstrain Variation in Cd Toxicity

Strain DS was the most sensitive, and NA the most resistant across the three Cd levels tested. The estimated mean survival time and its standard error for each strain are shown in Table 3. Based on times to death, strain NA was 1.8–2.4 times more resistant than strain DS. The responses of strains E and M were between these two extremes. Note that strain E was more sensitive than strain M at the LC₂₅, but became more resistant at the LC₇₅.

The variation in response among the four strains were all statistically significant at three different levels of Cd toxicity (LC₂₅, LC₅₀, and LC₇₅). The test statistics of the nonparametric linear rank tests in comparing the survivorship functions across

Table 4. Statistical tests for survivorship among the strains at three levels of Cd. The test statistic was derived from the Mantel-Cox test (Generalized-savage test)

Comparison	Test statistic	df	p-value
At LC ₂₅ (20.66 mg/L)	37.254	3	<0.0001
At LC ₅₀ (23.88 mg/L)	17.184	3	0.0006
At LC ₇₅ (27.59 mg/L)	21.342	3	0.0001

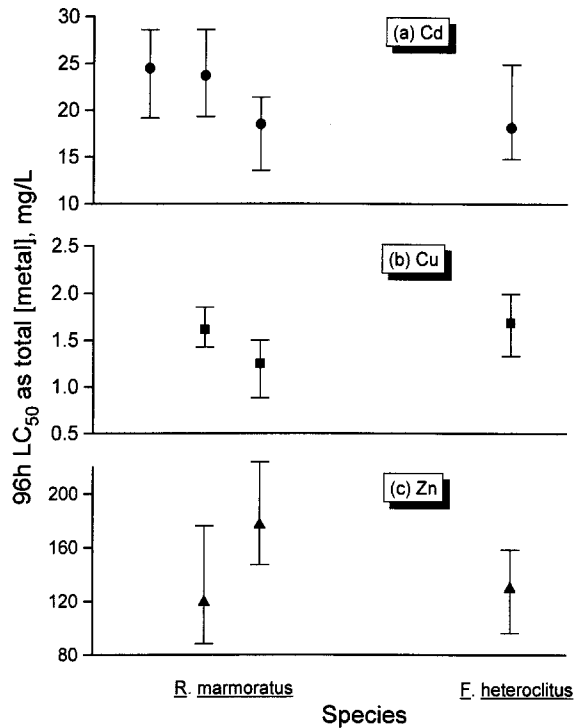


Fig. 3. The similarities of the 96-h LC₅₀ and 95% fiducial limits of (a) Cd, (b) Cu, and (c) Zn for *Rivulus marmoratus* and *Fundulus heteroclitus*. The solid figure and the bars attached to it represent the LC₅₀ and its 95% fiducial limits, respectively. The three Cd LC₅₀ values for *R. marmoratus* (from left to right) were derived from tests 9, 1, and 2 (Table 1, Figure 1). Only tests 1 and 2 were conducted using a simulated sea water made up entirely with Cl salts. Instant Ocean™ salts were used for the rest of the tests. All tests were conducted at a salinity of 14 ppt and a temperature of 26°C. Fish were 1–2 months old (12–17 mm in length)

the strains were all highly significant with a p-value ≤ 0.001, which supported the hypothesis that there was variation among strains in response to Cd toxicity (Table 4).

Comparison of Tolerance to Cd, Cu and Zn Tests with Fundulus

The LC₅₀ values of the two species were similar; 95% fiducial limits overlapped and there was no statistical difference between them (Figure 3a). For *R. marmoratus* we used the LC₅₀'s of tests 1, 2, and 9 to compare with that of *F. heteroclitus*. Our estimate of the LC₅₀ for *F. heteroclitus* was 18.2 mg/L with a 95% fiducial limit of 14.8 and 24.9 mg/L.

Although the LC_{50} of Cu for *F. heteroclitus* (1.69 mg/L) was numerically slightly higher than those of *R. marmoratus* (1.25 and 1.61 mg/L), the fiducial limits overlap with no statistical difference (Figure 3b).

The LC_{50} 's for Zn were also similar between *F. heteroclitus* and *R. marmoratus*. Our estimates of the LC_{50} 's were 129.5 mg/L for *F. heteroclitus* and 119.3 and 176.6 mg/L for *R. marmoratus*. Their fiducial limits overlapped and there was thus no statistical difference (Figure 3c).

Discussion

Cd Toxicity Under Various Water Chemistries

The toxicity of Cd was inversely related to the electrolyte concentrations. This study focused on the roles of non-toxic ions in reducing Cd toxicity by manipulating their concentrations while keeping the remaining electrolytes at defined levels. At low Ca + Mg levels, the LC_{50} 's of tests 5 and 6 were significantly higher than those of tests 7 and 8 (Figure 1), suggesting that [Na + K] has a bit less, if not the same, effect as that of Ca + Mg, on protecting fish from Cd poisoning. This interpretation is confounded by the fact that Cl was also much higher in tests 5 and 6, than in 7 and 8 (Table 1) and free Cd might be less in the former two tests due to complexation. In their study of the chemical speciation of Cd and other metals in 100% sea water (35 ppt), Zirino and Yamamoto (1972) found that the free Cd ion (Cd^{2+}), the most toxic form of Cd, can be as low as 2.5%. Sunda *et al.* (1978) provided an extensive discussion on the importance of free Cd ion concentration in their study of grass shrimp, *Palaemonetes pugio*. As the salinity increased from near 0 to 14 ppt, the percent free Cd decreases approximately from 63% to 10% (supposedly due to complexation with Cl), leading to an apparent decrease in Cd toxicity. There was an almost linear relationship with a negative slope between the log of the fraction of free ion to total Cd (designated as $\log([Cd^{2+}]/Cd_t)$) and salinity in their study. Thus, the general inverse relation between Cd toxicity and salinity is due to greater complexation of metals by Cl at higher salinity (Sunda *et al.* 1978; Engel and Fowler 1979). This can be tested by multiplying the percentage of free Cd times the measured LC_{50} 's at different salinities and observing if this essentially removes the salinity effect on toxicity. For our data, such a calculation indeed indicated that changes in free Cd were the major factor when comparing the decline in LC_{50} between tests 1, 2 vs 7, 8. Thus, the same amount of free metal may be present at the different LC_{50} 's characteristic of low and high salinities, despite a higher total metal concentration in the latter condition.

When other electrolytes were held at relatively low levels, [Ca + Mg] was correlated with reduced Cd toxicity (tests 3, 4, 7, and 8 in Table 1 and Figure 1). With a low stability of divalent cation-protein complex (compared to that of Mg) at the epithelium of fish gills (Szumski and Barton 1983), Ca has been suggested to compete with Cd for binding sites on the gills (Wright 1977, 1980; McCarty *et al.* 1978; Wright and Frain 1981; Wright *et al.* 1985; Pascoe *et al.* 1986). However, the [Cl] of tests 3 and 4 was a confounding variable. The difference in [Cl] (60 vs 3.2 mM) could lead to the formation of a greater degree of Cd complexation when [Cl] was higher. Indeed,

using the values for proportion of free to total Cd at varying salinities provided by Sunda *et al.* (1978), Cl complexation can account for the majority, but not all, of the observed decrease in Cd toxicity.

Interstrain Variation in Cd Toxicity

Striking differences were found in Cd toxicity among strains; strain DS was apparently the most sensitive among the four tested. Strains E and M reversed sensitivity at low (LC_{25}) and high (LC_{75}) levels of Cd, although their responses at the LC_{50} 's were very close (with mean survival times of 5.50 days for E and 4.75 days for M). Thus the mortality curves of these two strains may share a similar inflection point, but with different slopes. This may also imply that the variation in sensitivity at different toxic levels can be a source of variation for estimating the spectrum of lethal concentrations and their 95% fiducial limits from probit analysis when a pool of strains is utilized for the study. Given the consistencies within strains observed here, a test group made up of a single strain is recommended for toxicity testing. Pilot studies should be conducted before a given strain is selected for toxicity tests. However, there is the logistic difficulty of rearing a large number of mature fish to produce hatchlings of the same strain, age and size. Thus, a randomized pool of various strains could be used for acute toxicity tests, especially if an estimate of the LC_{50} is the primary interest.

Comparisons with Fundulus heteroclitus—Cd, Cu and Zn Tests

Based on the 96h LC_{50} 's, *R. marmoratus* and *F. heteroclitus* were quite similar in their tolerance to Cd, Cu and Zn. Eisler and Hennekey (1977) reported the 96-h LC_{50} for *F. heteroclitus* to be 22 mg Cd/L at 20°C and 20 ppt salinity. Dorfman (1977) studied the effects of 20 metals in 30 compounds on *F. heteroclitus* and found the 96-h LC_{50} 's to be 36 and 23 mg Cd/L at 7.2 and 24 ppt salinity, respectively (20°C). Burton and Fisher (1990) reported the 48h LC_{50} for Cd (10 ppt, 20°C) in *F. heteroclitus* to be 44.4 mg Cd/L. Thus, our estimate of the 96-h LC_{50} for *Fundulus* of 18.2 mg Cd/L was analogous to those reported previously.

The estimates of 1.69 mg Cu/L for *F. heteroclitus* and 1.25 and 1.61 mg Cu/L for *R. marmoratus* were very comparable to other measurements of the 96h LC_{50} . Copper was the most toxic among the three metals studied, when comparing the LC_{50} 's expressed in molarity. Jackim *et al.* (1970) reported a 96-h LC_{50} for *F. heteroclitus* of 3.2 mg Cu/L; Dorfman (1977) had an estimate of 0.4–3.1 mg Cu/L at salinities ranging from 24.0 to 5.5 ppt.

The LC_{50} 's for Zn of *F. heteroclitus* and *R. marmoratus* in this study were very similar. However, Eisler and Hennekey (1977) reported a 96-h LC_{50} of 60 mg Zn/L for *F. heteroclitus*, while Dorfman (1977) reported 27.5 and 32.0 mg Zn/L at 23 and 6 ppt salinity respectively, using zinc sulfate. The 48-h LC_{50} of *F. heteroclitus* was estimated to be 96.5 mg/L by Burton and Fisher (1990). The concentrations reported here (129.5 mg/L for *F. heteroclitus* and an average of 147.9 mg/L for *R. marmoratus*) were much higher. The most obvious rea-

son was the problem of precipitation which creates great difficulties in maintaining constant Zn levels. The water samples collected at the beginning and the end of the tests sometimes were very different in our study, especially those of high target concentrations. The differences among persons and/or laboratories in handling and collecting samples become more crucial when precipitation appears. In any case, we can reliably conclude that these two species were quite similar in their tolerance to Zn, regardless of the absolute value of the LC₅₀.

Free ion concentration is likely to be the most direct predictor of metal toxicity (Sunda and Guillard 1976; Sunda and Hanson 1987; Jonas 1989; Hughes and Poole 1991). Sunda and Hanson (1987) provide an extensive discussion of a technique for measuring free cupric ion concentration in sea water, and compare it with other techniques. They also summarize the literature on the proportion of inorganic copper species to total Cu; it was only about 1% (0.03–2%) across all the techniques listed. Among those methods available for estimating free metals in sea water, most of them still involve various assumptions, and calculation from total metal concentration and complexes (Hughes and Poole 1991). Sunda and Guillard (1976) stressed the importance of maintaining a stable pH in keeping free copper ion constant. However, the pH at the gill area tends to be lower due to the local release of carbon dioxide and this, in turn, facilitates the release of free metal from its precipitate or complex (Cusimano *et al.* 1986; Sorensen 1991). Thus, although the free ionic metal concentration in the testing solutions can be carefully controlled by the addition of different levels and kinds of chelators (Sunda *et al.* 1978; Oakden *et al.* 1984), it is not necessarily equivalent to the immediate metal concentration to which the fish gill is exposed (Pagenkopf 1983; Cusimano *et al.* 1986; Lauren and McDonald 1987; Playle *et al.* 1992). At the moment, there is no single method to directly determine the toxicity and bioavailability of the metals in solution. Measurement of total metal concentration is probably the least ambiguous way to present these data. Also, most of the LC₅₀ values from previous studies were presented as total metals. Since only total metal concentrations were measured, the LC₅₀'s are likely to be over-estimates of the true effective levels of each metal.

One should not conclude that *Rivulus marmoratus* and *Fundulus heteroclitus* are necessarily toxicologically alike for other stresses. For example, *F. heteroclitus* was quite tolerant of low pH and has a 96-h LC₅₀ (H₂SO₄) of pH 3.8–3.9 (Gonzalez *et al.* 1989, Dunson *et al.* in press). Other estuarine fish such as *Menidia beryllina* were considerably less tolerant (96h pH₅₀ = 4.6). The estimate of the 96-h pH₅₀ for *R. marmoratus* was 4.21, with 95% fiducial limits of 4.12–4.3 (unpublished data). *Rivulus marmoratus* is a promising species as an estuarine bioindicator. The naturally inbred genetic attributes of this species, its fast growth rate, and its simple husbandry are strong arguments for its adoption as a tropical estuarine bioassay species of choice. At present a much greater effort needs to be extended in screening a wide variety of xenobiotic toxins and natural stresses with the *Rivulus* system.

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