

Inhibition of photosynthesis, acidification and stimulation of zeaxanthin formation in leaves by sulfur dioxide and reversal of these effects

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Abstract. Leaves of *Pelargonium zonale* L. and *Spinacia oleracea* L. were fumigated with high concentrations of SO₂ for very short periods of time with the aim of first producing acute symptoms of damage and then observing repair. The response of different photosynthetic parameters to SO₂ was monitored during and after fumigation. The following results were obtained: (1) Inhibition of CO₂ assimilation in the light was accompanied by increased reduction of the quinone acceptor, Q_A, of photosystem II and by increased oxidation of the electron-donor pigment P₇₀₀ of photosystem I. Increased control of photosystem II activity in the SO₂-inhibited state was also indicated by increased light scattering and by increased non-photochemical quenching of chlorophyll fluorescence. Both are indicators of chloroplast energization. Apparently, SO₂ did not decrease but rather increased energization of the chloroplast thylakoid system by light. (2) Accumulation of dihydroxyacetone phosphate, fructose-1,6-phosphate and ribulose-1,5-phosphate and a decrease of 3-phosphoglycerate and hexosephosphate indicated that SO₂ inhibited enzymes of the Calvin cycle. (3) Stimulated postillumination CO₂ evolution suggested that when photosynthesis declined respiration increased to provide energy for repair reactions. (4) Increased leaf absorbance at 505 nm indicated increased stimulation of zeaxanthin formation in thylakoid membranes under the influence of SO₂. A similar increase in 505-nm absorbance could be induced by high

concentrations of CO₂. In darkened leaves, SO₂ did not produce changes in 505-nm absorbance. (5) While zeaxanthin formation was stimulated, changes in the fluorescence of the pH-indicating dye pyranine, which had been fed to the leaves, indicated acidification of the cytoplasm of leaf cells by SO₂. Maximum acid production by SO₂ required light. In contrast, cytoplasmic acidification of leaf cells by CO₂ was similar in the light and in the dark. (6) Since zeaxanthin formation is known to depend on the acidification of the thylakoid lumen, SO₂-dependent zeaxanthin formation indicated SO₂-dependent acidification of the thylakoid lumen as the indirect result of cytoplasmic acidification by SO₂. (7) Inhibition of photosynthesis and other effects of SO₂ were fully reversible in the light. Detoxification of SO₂ and reactivation of the photosynthetic apparatus were slow or absent in the dark. Light had a dual effect on the action of SO₂. Transiently, it first increased the extent of inhibition of assimilation, but, finally, it reversed inhibition. Sulfur dioxide was inhibitory as a consequence of the chemical reactivity of its hydration products rather than as a result of cellular acidification by the produced acid. The initial acidification was followed by an appreciable alkalinisation demonstrating the action of the pH-stat mechanism. (8) The data are discussed in relation to SO₂ toxicity under field conditions when plants are chronically exposed to polluted air.

Key words: Acidification – Chlorophyll fluorescence – Photosynthesis (SO₂ effect) – Xanthophyll cycle – *Spinacia* – *Pelargonium*

Dedicated to Professor O.L. Lange on the occasion of his 65th birthday

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Abbreviations: Chl = chlorophyll; DHAP = dihydroxyacetone phosphate; FBP = fructose-1,6-bisphosphate; F6P = fructose-6-phosphate; F, Fm, Fm', Fo, Fo' = chlorophyll fluorescence levels; PGA = 3-phosphoglycerate; P₇₀₀ = primary donor of photosystem I; Q_A = primary quinone acceptor of photosystem II; q_p = photochemical quenching of chlorophyll fluorescence; NPQ = non-photochemical quenching of chlorophyll fluorescence; RuBP = ribulose-1,5-bisphosphate

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Introduction

Sulfur dioxide has long been known to damage plants when present in the air at excessive concentrations. The gas diffuses into leaves through open stomata. It is highly soluble in aqueous phases where it can react with water

to form sulfurous acid. The acid (pK values 1.79 and 6.99) is trapped in alkaline leaf compartments. Both SO₂ and the anions or radicals formed from it can undergo different reactions with cellular constituents thereby interfering with metabolism. Anionic products formed from SO₂ are toxic. Detoxification must outcompete damaging reactions to be effective, and is possible either by reduction which finally yields sulfur-containing amino acids, or by oxidation which results in sulfate formation. Reductive detoxification does not burden cytoplasmic pH-stat mechanisms which maintain pH within physiological limits, whereas formation of sulfurous or sulfuric acid necessitates either mobilization of bases or export of protons to maintain proper control of the cytoplasmic pH. An extensive literature exists on the effects of SO₂, or the sulfite formed from SO₂, on individual cellular reactions (Winner et al. 1985). Nonetheless, little is known about the integrated response of intact leaves to doses of SO₂ which do not cause permanent damage. In the following, we report on experiments performed with the aim to obtain information on SO₂ effects on leaves and the possible repair of SO₂ damage.

Material and methods

Plant material. Spinach (*Spinacia oleracea* L. cv. Yates) was grown in a greenhouse with additional illumination in 10 h light/14 h dark cycles. Seeds were obtained from Fa. Wagner, Würzburg, FRG. *Pelargonium zonale* L. cv. Kardinal, also from a greenhouse, was propagated from cuttings and grown in normal day/night cycles. Leaves were cut from the plants and were used for experimentation with their petioles immersed in water.

Gas exchange measurements and illumination. During measurements, segments of the leaves were enclosed in a thermostatted sandwich-type cuvette which permitted controlled gas flow over their lower surface. The gas flow rate was 30 l · h⁻¹. The upper surface was usually attached with a transparent glue made from soluble starch and water (1:10, w/v) to the thermostatted window of the cuvette to minimize an increase in leaf temperature during illumination. Leaf temperature was controlled by a thermocouple and kept close to 20° C. The composition of the gas stream passing over the leaf was adjusted by mass-flow controllers (Tylan, Echling, FRG). The relative humidity of the gas entering the cuvette was 60%. Transpirational water loss and CO₂ exchange were recorded by an infra-red gas analyzer (Binos, from Leybold-Heraeus, Hanau, FRG) in the differential mode. Stomatal and boundary-layer resistance to water and SO₂ diffusion were calculated according to Nobel (1983) and Taylor and Tingey (1983). Short-wavelength red actinic illumination from a 250-W halogen lamp was provided by a red cutoff filter (RG 630; Schott, Mainz, FRG) and K 65 and Calflex X filters (Balzers, Liechtenstein). An additional water layer (4 cm) was used as a heat filter. The half-bandwidth of red light was from 625 to 663 nm. To obtain far-red light from the same light source, the filters RG 724 and RG 610 from Schott were combined with Calflex X, and the half-bandwidth was from 720 to 760 nm. The light was brought to the upper surface of the cuvette-enclosed leaf segment by means of fiber optics which simultaneously transmitted optical signals back from the leaf to receiving devices.

Fluorescence and absorbance measurements. Modulated chlorophyll (Chl) fluorescence was measured by a PAM fluorometer (Walz, Effeltrich, FRG). For fluorescence analysis of the energy state of the thylakoid system and of the redox state of the primary quinone acceptor (Q_A) in the reaction center of PSII, the saturation-pulse

method was employed (Bradbury and Baker 1981; Schreiber et al. 1986). In parallel to Chl fluorescence, leaf absorption of modulated 820-nm light was measured by a second PAM fluorometer (Schreiber et al. 1988) to obtain information on the redox state of P₇₀₀, the primary electron donor in the reaction center of PSI. The fast decrease in 820-nm absorption seen when illumination was interrupted by brief darkening periods (6–10 s) was interpreted to indicate full reduction of P₇₀₀ and of plastocyanin which changes its redox state in parallel with P₇₀₀ (Klughammer and Schreiber 1992). Simultaneously, the decrease in Chl fluorescence permitted measuring the so-called Fo' level of Chl fluorescence. This is needed for the evaluation of photochemical quenching (q_p) (van Kooten and Snel 1990), which is defined as (Fm'–F)/(Fm'–Fo'). The term 1–q_p was taken as a measure for the reduction state of the "primary" acceptor Q_A (Dietz et al. 1985). The quantum yield of PSII was estimated from fluorescence parameters, (Fm'–F)/Fm', according to Genty et al. (1989). Non-photochemical fluorescence quenching was expressed as NPQ=(Fm–Fm')/Fm' (Bilger and Björkman 1990). Fm is the maximum modulated fluorescence observed after a light-saturating flash in a leaf which had been predarkened for a prolonged period of time, and Fm' the maximum flash-induced modulated fluorescence of the leaf in the presence of actinic illumination. Fo' is the minimum modulated fluorescence observed immediately after turning actinic illumination off; F is fluorescence in the presence of actinic illumination. Absorption of a non-modulated, weak green measuring beam was measured by a photomultiplier which was equipped with a BG 18 filter from Schott and filters 9782 and 9780 from Corning Glass Works (Corning, N.Y., USA) to protect it against actinic light and the modulated red and far-red beams. The green measuring beam was produced by a monochromator and was directed to the lower surface of the leaf. Transmitted green light was collected from the upper surface by the multi-branched light guide. One of the branches was connected to the multiplier. Wavelengths mainly used were 505 nm to record changes in zeaxanthin concentration and 540 nm to record changes in light scattering by the leaf which indicate changes in thylakoid energization (Heber 1969; Bilger et al. 1988). To record changes in cellular pH, a fluorescent pH-indicating dye, pyranine, which enters mesophyll cells even though it is a trisulfonate anion, was fed to leaves through the petiole and its fluorescence was excited and measured as described by Yin et al. (1990).

Pigment analysis. For the pigment analyses made in conjunction with measurements of Chl fluorescence and ΔA₅₀₅, an illuminated part of a leaf was frozen in liquid nitrogen immediately after opening the cuvette and stored at –80° C. Extraction and high-performance liquid chromatography (HPLC) analysis of pigment composition were as described by Gilmore and Yamamoto (1991).

Metabolite assays. To relate the state of photosynthesis to the levels of Calvin-cycle metabolites in leaves, a sandwich-type cuvette designed so as to permit both gas-exchange and Chl fluorescence measurements and rapid freeze-stopping of metabolism was used (Siebke et al. 1990). Freeze-stopping of metabolism was done by driving a copper piston which had been precooled with liquid nitrogen through the upper window of the cuvette onto the leaf. Illumination was provided through the lower window. The apparatus used was similar to that described by Badger et al. (1984). Metabolites were determined enzymically as described in Dietz and Heber (1984). When metabolites were determined in the non-aqueously isolated chloroplast fraction of freeze-dried leaves, another larger cuvette was used which permitted monitoring of Chl fluorescence and CO₂-exchange measurements. Non-aqueous fractionation of the leaves was performed according to the methods of Heber and Willenbrink (1964) and Dietz and Heber (1984).

Chlorophyll was determined in acetone extracts according to Arnon (1949) or, after its conversion to pheophytin, in acid-treated leaf homogenates according to Vernon (1960).

Results and discussion

Inhibition of CO₂ assimilation and zeaxanthin formation by a leaf during and after fumigation with SO₂. Figure 1 shows simultaneous recordings of transpirational water loss, photosynthetic CO₂ uptake and absorbance at 505 nm of a leaf of *Pelargonium* before, during and after a 4-min exposure period to three different concentrations of SO₂ in air. During fumigation with SO₂, CO₂ uptake declined. The extent of photosynthesis inhibition increased with increasing concentration of SO₂. Inhibition was not related to stomatal closure, although stomata closed partially during fumigation (as shown by decreased transpiration), accounting for part of the observed decline in CO₂ uptake. Direct inhibition of photosynthesis by SO₂ is shown by a progressive increase in the intercellular concentration of CO₂, c_i , when photosynthesis was progressively inhibited by increased levels of SO₂ (Fig. 2). Inhibition of assimilation by stomatal closure should be accompanied by decreased c_i , not by increased c_i . The calculation of c_i was based on the measured transpiration rate (Nobel 1983).

After termination of fumigation, photosynthesis con-

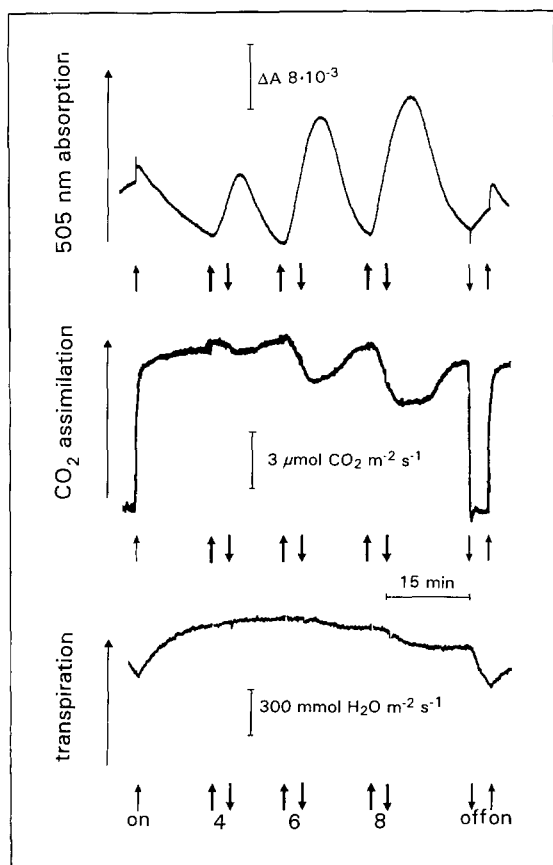


Fig. 1. Effect of three fumigation periods of 4 min each with different concentrations of SO₂ on transpiration, carbon assimilation, and absorption of a weak 505-nm measuring beam by an illuminated *Pelargonium* leaf. Arrows indicate addition of SO₂ to and its removal from the gas phase, or illumination and darkening. Numbers show concentrations of SO₂ in $\mu\text{l} \cdot \text{l}^{-1}$. The CO₂ concentration in air was $520 \mu\text{l} \cdot \text{l}^{-1}$ and the irradiation $120 \text{ W} \cdot \text{m}^{-2}$

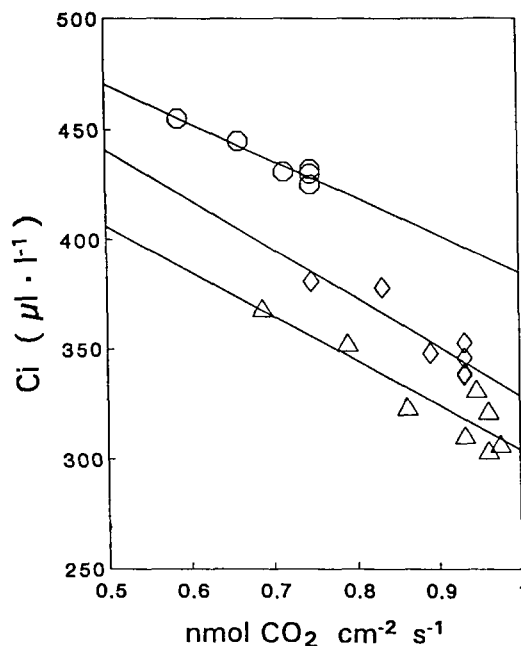


Fig. 2. Measured CO₂ uptake and calculated internal CO₂ concentration (c_i) when inhibition of photosynthesis was maximal (see Fig. 1). Data are from experiments with three different *Pelargonium* leaves (○, Δ, ◇). Each leaf was gassed for 5 min with SO₂. Increasing SO₂ concentrations produced not only a progressive decrease in CO₂ uptake but also an increase in c_i . Between the SO₂ treatments, leaves were illuminated for 40–60 min in the absence of SO₂ to allow recovery from the fumigations. The irradiation was $80 \text{ W} \cdot \text{m}^{-2}$ and the CO₂ concentration in air $600 \mu\text{l} \cdot \text{l}^{-1}$

tinued to decline for a few minutes and then recovered, indicating repair of damage. Simultaneously, zeaxanthin formation, which is indicated by increased absorption of 505-nm light, was stimulated by SO₂ (Fig. 1). After fumigation was terminated, 505-nm absorbance continued to increase, but then declined. Original levels were reestablished while photosynthesis recovered. In parallel with inhibition of CO₂ assimilation, the scattering of a weak green measuring beam (set to 540 nm) by the leaf increased. It decreased as photosynthesis recovered (data not shown). Increased light scattering is known to be an indicator of an increased proton gradient across the thylakoid membrane (Heber 1969; Köster and Heber 1982; Kobayashi et al. 1982; Bilger et al. 1988). An increase in the magnitude of the proton gradient, when light remains unchanged, indicates either SO₂-dependent inhibition of ATP formation by the thylakoid ATPase or decreased consumption of ATP by SO₂-inhibited carbon assimilation. In both cases such a decrease would lead to an increased acidification of the thylakoid lumen. This would stimulate the violaxanthin de-epoxidase, the enzyme responsible for zeaxanthin formation, which is known to be pH-dependent (Hager 1980). It should be noted that, at 505 nm, there is no overlap between the absorption change which indicates zeaxanthin formation, and the light-scattering change which peaks close to 540 nm (Bilger et al. 1989; Bilger and Björkman 1990).

In Fig. 3, Q_A reduction, P₇₀₀ oxidation (plus oxidation of some plastocyanin which donates electrons to P₇₀₀

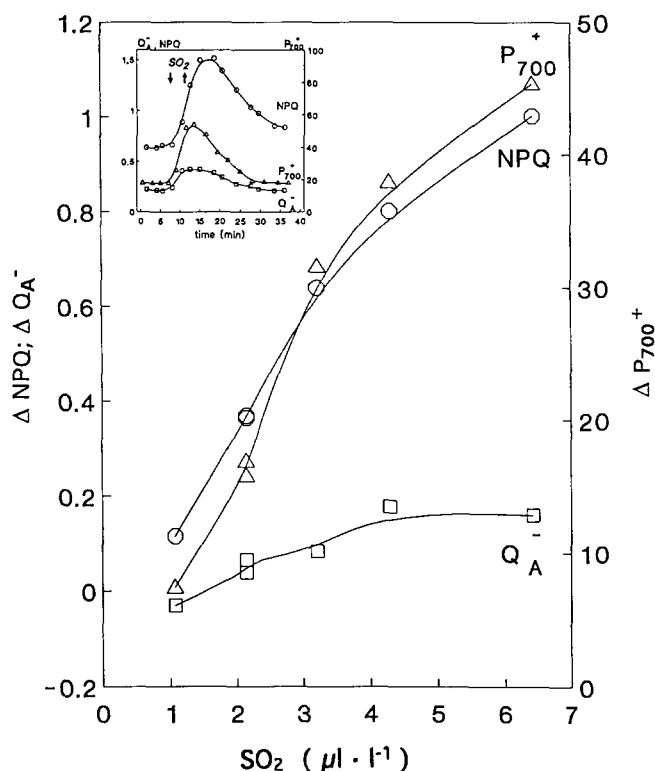


Fig. 3. Increased reduction of Q_A (\square), oxidation of P_{700} (Δ) and amount of NPQ (\circ) of a representative *Pelargonium* leaf as a function of the concentration of SO_2 in air. Values were measured between 3 and 5 min after 3-min fumigation periods when inhibition of photosynthesis was maximal (inset), and are expressed as a difference from the control values measured before fumigation (see inset). The control value for CO_2 assimilation was $10.7 \mu\text{mol } CO_2 \cdot \text{m}^{-2} \cdot \text{s}^{-1}$; P_{700} was 20% oxidized in the control and Q_A was about 40% reduced; NPQ was 0.89 in the control. The irradiation was $80 \text{ W} \cdot \text{m}^{-2}$, the CO_2 concentration $600 \mu\text{l} \cdot \text{l}^{-1}$. Before the experiment, the leaf was kept for 2 h in the light in air with $600 \mu\text{l} \cdot \text{l}^{-1} CO_2$

and changes its redox state together with P_{700} (Klughammer and Schreiber 1992)) and non-photochemical quenching of chlorophyll fluorescence (NPQ) are compared in their response to a 3-min exposure of a *Pelargonium* leaf to SO_2 . The time course of Q_A reduction, of P_{700} oxidation and of NPQ in response to SO_2 is shown in the inset. As photosynthesis inhibition increased due to increased concentrations of SO_2 , P_{700} photooxidation, Q_A reduction and NPQ increased also (Fig 3). This shows that control of electron flow from PSII to PSI is increased under the influence of SO_2 . The control is exerted both at the level of PSII (as shown by increased NPQ which indicates increased radiationless dissipation of excitation energy) and at the level of plastoquinone oxidation. The latter is shown by increased reduction of the quinone acceptor Q_A in the reaction center of PSII, while P_{700} oxidation is increased (Fig. 3). Parallel to the changes in NPQ and Q_A^- , the so-called dark level of chlorophyll fluorescence (F_0') decreased transiently (not shown). This is another indication of increased conversion of light energy into heat. In leaves, light scattering and NPQ are both indicators of the proton gradient

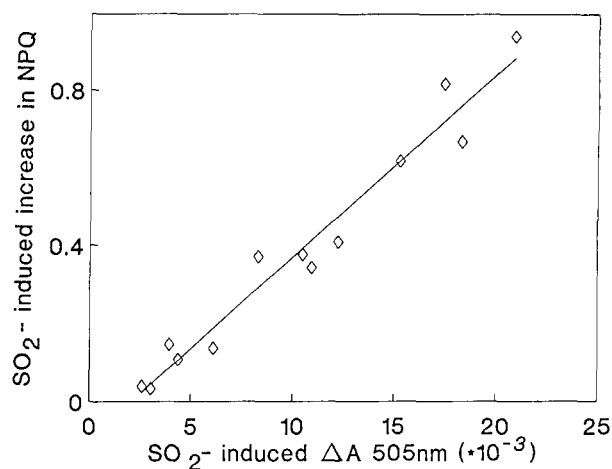


Fig. 4. Representative graph of the correlation between SO_2 -induced increases in NPQ and in 505-nm absorbance. *Pelargonium* leaves were illuminated for at least 1 h before fumigation with SO_2 at concentrations ranging from 2 to $8 \mu\text{l} \cdot \text{l}^{-1}$. Other conditions were as described in the legend of Fig. 3

between the chloroplast stroma and the intrathylakoid space (Krause 1973; Köster and Heber 1982; Krause and Weis 1991). A close correlation between the extent of NPQ and zeaxanthin levels has been demonstrated in leaves by Demmig et al. (1987). Figure 4 shows a linear correlation between the SO_2 -induced increase in 505-nm absorbance and the simultaneously observed increase in NPQ.

The effect of SO_2 on metabolite levels. In order to answer the question of whether the observed SO_2 -dependent inhibition of photosynthesis is caused by inhibition of the thylakoid ATPase (Cerovic et al. 1982), of electron transport (Daniell and Sarojini 1981; Shimazaki et al. 1984) or by inhibition of enzymes of the Calvin cycle (Anderson and Duggan 1977; Tanaka et al. 1983), metabolite levels were measured. A part of a leaf was enclosed in a cuvette which permitted, in addition to measuring gas exchange and Chl fluorescence, fast freeze-stopping of leaf metabolism. After a spinach leaf had reached steady-state photosynthesis in saturating light (after about 40 min), it was fumigated with different concentration of SO_2 for 6 min. When inhibition of photosynthesis was maximal, and before photosynthesis started to recover (see Figs. 1, 3), the part of the leaf which was enclosed in the cuvette was cut out with a precooled (-180°C) copper piston and the metabolites were determined in leaf extracts (Fig. 5). With increasing SO_2 concentration and increasing inhibition of photosynthesis levels of dihydroxyacetone-phosphate (DHAP), fructose-1,6-phosphate (FBP) and ribulose-1,5-bisphosphate (RuBP) increased, whereas levels of 3-phosphoglycerate (PGA) and fructose-6-phosphate (F6P) decreased (Fig. 5).

In Fig. 6, ratios of DHAP to PGA, FBP to F6P and RuBP to PGA from different leaves are compared with the SO_2 -dependent decrease in the quantum yield of electron transport in PSII which was measured by Chl fluorescence (Genty et al. 1989). As the concentration of

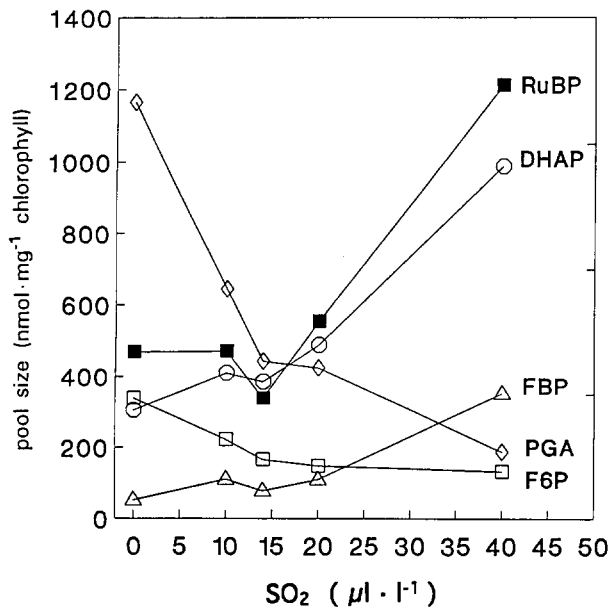


Fig. 5. Representative demonstration of the effect of SO₂ on the levels of DHAP, PGA, FBP, F6P and RuBP in spinach leaves. Leaves were kept for 30 min in the light (200 W · m⁻²) in air with 600 μl · l⁻¹ CO₂ and were then fumigated with SO₂ for 6 min. When inhibition of photosynthesis was maximal (about 7 min after termination of fumigation) metabolism was arrested by freeze-clamping of the leaves

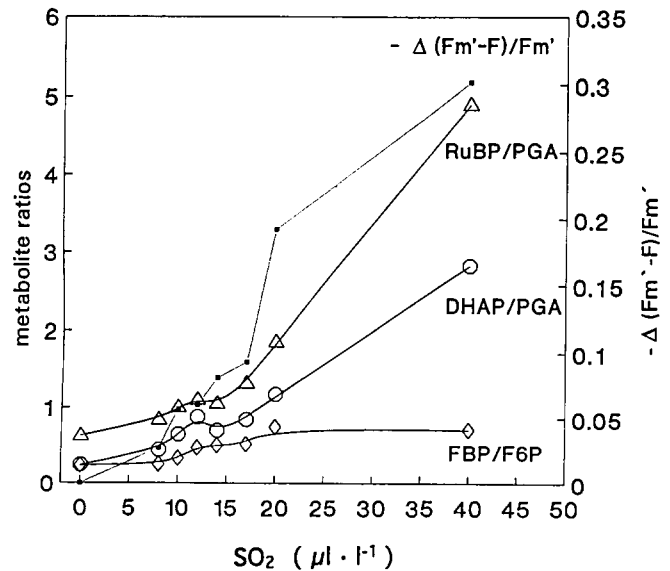


Fig. 6. Effect of SO₂ on ratios of PGA to DHAP (○), FBP to F6P (◇) and RuBP to PGA (△) and on quantum yield of PSII (■). Data are from Fig. 5 and from other experiments. Control values are means of three different leaves. A saturating light pulse had been given 2 min before leaves were frozen. The $\Delta(Fm' - F)/Fm'$ is the loss of quantum yield of electron flow through PSII caused by SO₂ at the time of freezing the leaves

SO₂ in air was increased, photosynthesis became progressively inhibited. The quantum yield of electron transport in PSII decreased accordingly. This is shown in Fig. 6 as difference between the quantum yield of PSII in the leaf before and after fumigation with SO₂ [$-\Delta(Fm' - F)/Fm'$]. All shown metabolite ratios increased while the decreased quantum yield indicated progressive inhibition of photosynthesis.

Metabolites were also measured in chloroplasts which had been isolated non-aqueously from fumigated spinach leaves. Illuminated leaves had been exposed to SO₂ for 5 min. Ten minutes after termination of fumigation, when photosynthesis inhibition was still close to maximum, or 40 min after fumigation, when recovery was extensive, they were frozen in liquid nitrogen and freeze-dried. Chloroplasts were isolated from the dry leaves as described by Dietz and Heber (1984). Fumigation of the leaves had produced a reduction in the quantum yield of PSII by about 50% [$(Fm' - F)/Fm'$ was decreased from 0.6 to 0.3]. As in the leaf extracts, the inhibition of photosynthesis by SO₂ was associated with a marked increase in chloroplast DHAP and a decrease in PGA (Table 1). Chloroplast hexose phosphates decreased and FBP and RuBP increased. The ATP/ADP ratio increased from 1.18 in control chloroplasts to 2.38 in chloroplasts of leaves which were inhibited by SO₂ (Table 1).

The data show that inhibition of photosynthesis by SO₂ cannot be explained by inhibition of the thylakoid ATPase. They are consistent with the inhibition by SO₂ of stromal enzymes such as fructose biphosphatase (Tanaka et al. 1984) and, at higher SO₂ concentrations, ribulose biphosphate carboxylase (Ziegler 1973). Inhibi-

tion of these enzymes should be expected to result in the accumulation of their substrates and in decreased levels of their products. This was observed. In contrast, the level of the substrate of the system phosphoglycerate kinase/glyceraldehyde phosphate dehydrogenase/triose-phosphate isomerase, PGA, declined in the presence of SO₂, whereas its product, DHAP, increased. A decrease of PGA by a factor of about ten has also been observed in SO₂-fumigated spinach leaves by Tanaka et al. (1984). The observations show that the thylakoid system continued to provide ATP and electrons for the reduction of PGA even in the presence of high SO₂ levels, and that the enzymes involved in PGA reduction remained sufficiently active to reduce available PGA.

Importantly, inhibition was reversible, as shown by the recovery of photosynthesis and by the changes in the levels of chloroplast metabolites or in their ratios (Table 1). Of particular interest is the observation that recovery of fructose biphosphatase activity is indicated only by increased carbon assimilation and an increased level of F6P, not by a decreased ratio of FBP to F6P. Apparently, repair reactions proceed at different rates with different enzymes.

Relationship between SO₂-induced stimulation of zeaxanthin and decreased carbon assimilation. Inhibition of Calvin-cycle enzymes by SO₂, as shown by the experiments of Figs. 5 and 6, decreases photosynthetic energy consumption thereby permitting increased thylakoid energization. An increased transthylakoid proton gradient (or, rather, a decreased intrathylakoid pH) could account for increased zeaxanthin formation which in turn would

Table 1. Metabolite levels (*A*) and metabolite ratios (*B*) in non-aqueously isolated chloroplasts. Leaves of spinach (10 for each treatment) were kept in the dark (control dark) or were illuminated with white light (200 W · m⁻²) for 1 h in air with 600 μl · l⁻¹ CO₂ (control light). In the SO₂ experiment, the leaves were first illuminated without SO₂ in air with 600 μl · l⁻¹ CO₂ for 30 min and then fumigated in the light with 20 μl · l⁻¹ SO₂ for 6 min, followed by

a 10-min illumination period without SO₂. The recovery experiment was identical to the SO₂ experiment except that the illumination period after the fumigation was not 10 but 40 min. Leaf metabolism was quenched by pouring liquid nitrogen into the leaf chamber either in the dark (control dark) or in the light. The frozen leaves were freeze-dried and used for non-aqueous fractionation

<i>A</i>	Amounts (nmol · mg ⁻¹ Chl)							
	DHAP	PGA	RuBP	FBP	F6P	G6P	ATP	ADP
Control (dark)	1	100	0	1	12	41	39	33
Control (light)	13	200	64	18	34	72	36	32
+ SO ₂	100	7	200	12	9	28	62	26
Recovery	76	40	100	50	22	44	61	–

<i>B</i>	Metabolite ratios		
	PGA/DHAP	RuBP/PGA	FBP/F6P
Control (light)	14.3	0.32	0.53
+ SO ₂	0.07	28.6	1.33
Recovery	0.53	2.5	2.27

lead to increased control of PSII activity and increased dissipation of light energy as heat (Demmig et al. 1987).

The experiment shown in Fig. 7 is intended to reveal whether decreased ATP consumption due to decreased activity of Calvin cycle is the only factor which contributes to increased zeaxanthin formation. The quantum yield of PSII was chosen as a measure for photosynthetic activity. A *Pelargonium* leaf was permitted to photosynthesize initially in the presence of 600 μl · l⁻¹ CO₂ and both absorption at 505 nm and Chl fluorescence yield were monitored. At various times the gas phase surrounding the leaf was then changed to reduce photosynthesis by either acceptor limitation (e.g. by removal of CO₂ or O₂) or by application of SO₂ or by high CO₂ partial pressures. In the latter case, high concentrations of CO₂ were used to produce acidification of cytoplasmic compartments of leaf cells. Wagner et al. (1990) have shown that high CO₂ concentrations inhibit photosynthesis of isolated chloroplasts and, transiently, also of mesophyll protoplasts, by decreasing cytoplasmic pH values.

At the start of the experiment the leaf was kept in air containing 600 μl · l⁻¹ CO₂ (denoted by 0 at the bottom of Fig. 7). Gassing the leaf with 8 μl · l⁻¹ SO₂ for a period of 5 min (Fig. 7; 1) caused a reduction of PSII quantum yield by 25%. Simultaneously NPQ and ΔA₅₀₅ increased appreciably. Upon returning to an SO₂-free atmosphere, all three parameters recovered almost fully within the following 20 min. A reduction of the CO₂ content of the atmosphere from 600 to 200 μl · l⁻¹ (Fig. 7; 2) caused a drop in PSII quantum yield of similar magnitude to that during the preceding SO₂ fumigation. This time, however, ΔA₅₀₅ responded to a much smaller extent, whereas the response of NPQ even increased.

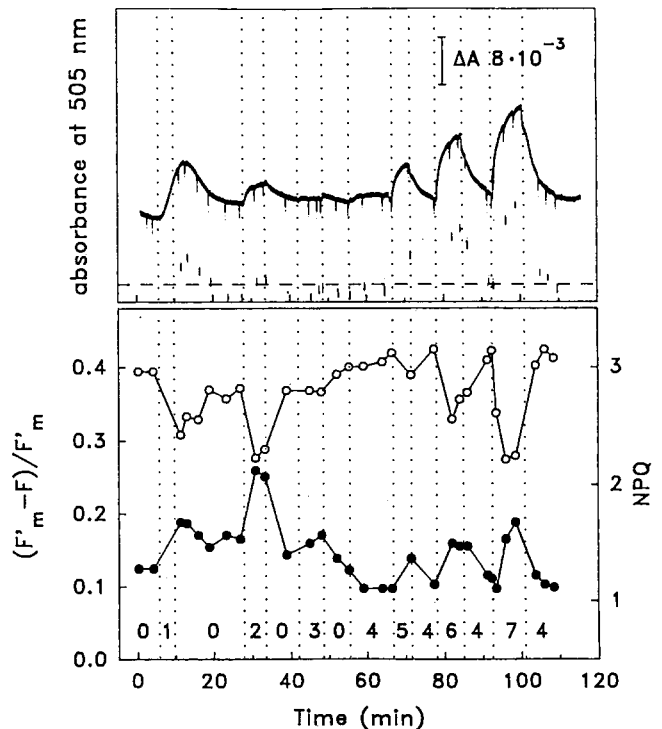
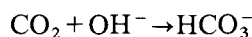


Fig. 7. Absorbance at 505 nm, quantum yield of PSII ($F_m' - F$)/ F_m' (○) and NPQ (●) during inhibition of photosynthesis by either acceptor limitation (removal of CO₂) or acidification (high CO₂, SO₂) in a representative leaf of *Pelargonium*. The leaf was kept initially in 21% O₂ and 600 μl · l⁻¹ CO₂ (=0). The gas phase was then changed as marked by numbers (1–7): (1) 21% O₂, 600 μl · l⁻¹ CO₂ + 8 μl · l⁻¹ SO₂; (2) 21% O₂ + 200 μl · l⁻¹ CO₂; (3) 2.6% O₂ + 600 μl · l⁻¹ CO₂; (4) 21% O₂ + 0.3% CO₂; (5) 21% O₂ + 2% CO₂; (6) 21% O₂ + 4% CO₂; (7) 21% O₂ + 6% CO₂. The irradiation was 150 W · m⁻²

Suppression of photorespiration by either decreasing the O₂ content of the atmosphere to 2.6% (Fig. 7; 3) or by increasing CO₂ to 3 000 $\mu\text{l} \cdot \text{l}^{-1}$ (Fig. 7; 4) caused only marginal changes in all three parameters. During the intermediate period in air with 600 $\mu\text{l} \cdot \text{l}^{-1}$ CO₂, stomata reopened (as monitored by measuring the humidity of the gas stream leaving the cuvette; data not shown) which caused some increase in quantum yield and a concomitant decrease in NPQ. After subjecting the leaf to 3 000 $\mu\text{l} \cdot \text{l}^{-1}$ CO₂, concentrations of CO₂ were increased first to 2% (Fig. 7; 5), then to 4% (Fig. 7; 6), and finally to 6% (Fig. 7; 7). This mainly reduces the stroma pH owing to the consumption of OH⁻ according to



The intrathylakoid pH is not reduced by direct CO₂-dependent acidification, because the first pK of CO₂ is 6.37, whereas the pH of the intrathylakoid space is not far from 5 in illuminated leaves. However, the intrathylakoid pH will be expected to be lowered indirectly if electron transport manages to maintain the magnitude of the transthylakoid proton gradient.

Gassing the leaf with very high CO₂ concentrations caused a strong reduction in the quantum yield of PSII which was approximately proportional to the applied CO₂ concentration. At the same time ΔA_{505} increased strongly, as did NPQ. The change in ΔA_{505} caused by application of high CO₂ contained a component which was rapidly reversible upon returning to 3 000 $\mu\text{l} \cdot \text{l}^{-1}$ CO₂. This component presumably resulted from the electrochromic shift which can contribute to absorbance changes at 505 nm. The reduction of the quantum yield caused by 4% CO₂ was similar in extent to that caused by fumigation with 8 $\mu\text{l} \cdot \text{l}^{-1}$ SO₂ and to that caused by reduction of the CO₂ concentration to 200 $\mu\text{l} \cdot \text{l}^{-1}$. The response of ΔA_{505} , however, to the reduction in quantum yield caused by changing the CO₂ concentration from 600 to 200 $\mu\text{l} \cdot \text{l}^{-1}$ was much smaller than to the treatments with SO₂ or high CO₂ which also reduced the quantum yield.

The relationship between quantum yield and 505-nm absorbance is explored in Fig. 8. A reduction in PSII quantum yield was produced by reducing the CO₂ and/or O₂ concentration in various combinations. This caused an acceptor limitation which was accompanied by a much smaller increase in ΔA_{505} (triangles) than observed when acidifying gases, SO₂ or CO₂, were used to decrease the quantum yield (circles). In a similar experiment performed with spinach leaves, both 505-nm absorbance changes and levels of xanthophyll cycle components were measured. Figure 9 shows the relationship between the epoxidation state (EPS) of the xanthophylls (Bilger et al. 1989) and the quantum yield of electron flow in PSII. The EPS is defined by $[(V + 0.5A)/(V + A + Z)]$; A, antheraxanthin; V, violaxanthin; Z, zeaxanthin). The reduction in PSII quantum yield produced by reducing the CO₂ concentration (Δ) was accompanied by a smaller decrease of the EPS than when SO₂ was used to reduce the quantum yield of PSII (\circ). The inset shows that the relationship between EPS and 505-nm absorbance changes is linear.

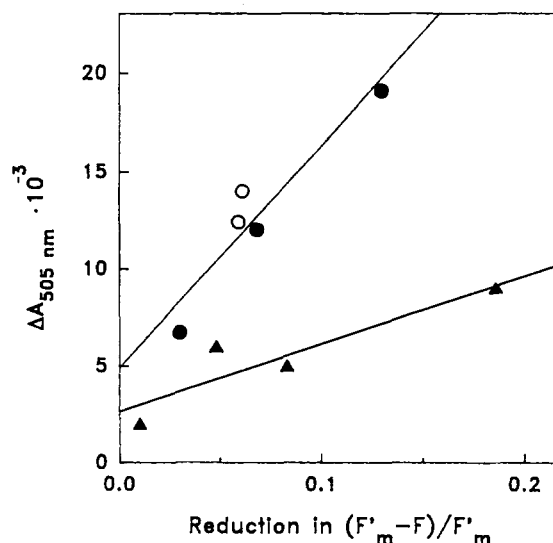


Fig. 8. Comparison between ΔA_{505} and reduction in quantum yield of PSII ($F_m' - F/F_m'$) induced by changes in the gas phase. Data are taken from an experiment similar to that shown in Fig. 5. \blacktriangle , decrease of CO₂ or O₂; \circ , SO₂-dependent acidification; \bullet , CO₂-dependent acidification

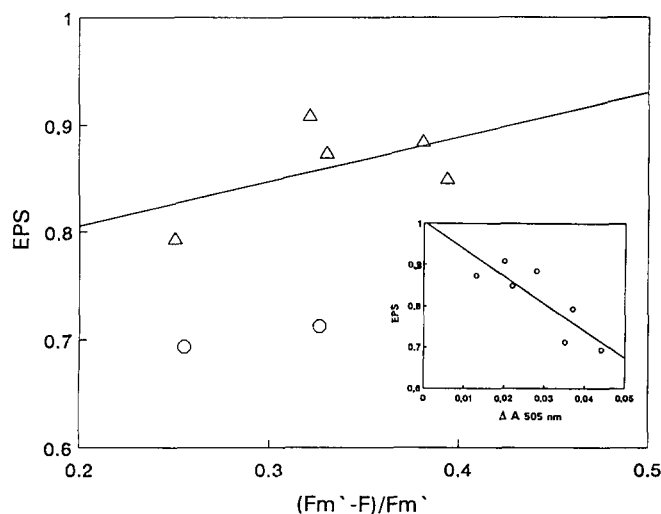


Fig. 9. The relationship between quantum yield of PSII ($F_m' - F/F_m'$) and the epoxidation state, EPS [$\text{EPS} = (V + 0.5A)/(V + A + Z)$; A, antheraxanthin; V, violaxanthin; Z, zeaxanthin], in a representative spinach leaf after imposing an acceptor limitation on electron flow by decreasing the CO₂ level (Δ) or by inhibiting photosynthesis by SO₂ (\circ). The inset shows the relationship between EPS and 505-nm absorbance changes when darkened leaves were illuminated in the presence of different levels of CO₂ or SO₂ in these experiments. Before the experiments, the leaves were kept for 2 h in low light and then 30 min in the dark to minimize initial zeaxanthin levels

From the data of Figs. 7, 8 and 9 it is concluded that increased zeaxanthin formation under the influence of SO₂ is not only the result of the decrease in intrathylakoid pH caused by the decreased ATP consumption of SO₂-inhibited carbon assimilation but also of another, more indirect, acidifying effect of SO₂.

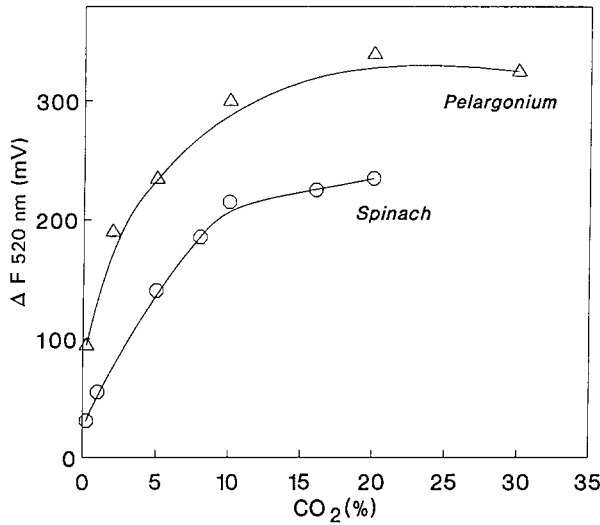


Fig. 10. Demonstration of a typical CO₂-dependent acidification in a spinach leaf and a *Pelargonium* leaf as indicated by the suppression of the fluorescence of pyranine. Leaves were fed for 2 h with pyranine (200 μM in 2 mM Hepes, pH 7.5) in the light. Acidification was caused by 2-min exposure periods to different concentrations of CO₂ in air

Sulfur-dioxide-dependent acidification in leaves and its reversal. Direct measurements of pH in the chloroplasts of intact leaves are difficult. However, it is possible to introduce pH-indicating fluorescent dyes into mesophyll cells of intact leaves by feeding them to the petiole of cut leaves (Yin et al. 1990). After feeding of pyranine which has a pK of 7.3, a light-dependent increase in fluorescence emission at about 520 nm has been shown to originate mainly from the cytosol of green leaf cells, not from the chloroplasts, the apoplast or from the large vacuoles (Yin et al. 1990).

Figure 10 shows the relationship between the concentration of CO₂ in air and the decrease in cytosolic pH values in leaves of *Pelargonium* and spinach in the dark as shown by the CO₂-dependent quenching of pyranine fluorescence. In the light a similar relationship was obtained (data not shown). The pH response is as expected on the basis of the Henderson-Hasselbalch equation, if a weak acid is added to a buffered solution.

In Fig. 11 the time courses of pyranine fluorescence and of CO₂ assimilation in response to gassing with different concentrations of SO₂ are shown. As the concentration of SO₂ in air was increased, inhibition of assimilation increased. Quenching of pyranine fluorescence followed the photosynthesis inhibition, although it was not proportional to the SO₂ concentration (Fig. 11). Characteristically, acidification did not increase immediately when SO₂ was added to the gas stream which passed over the leaf. There was a lag period of about 1 min before slow acidification started. A similar lag phase was usually seen also in the SO₂-dependent increase in ΔA₅₀₅. The acidification of the cytosol caused by a few μl · l⁻¹ SO₂ within a few minutes of fumigation was similar in extent to the corresponding changes produced by 2 500 μl · l⁻¹ CO₂ (Fig. 11). It is known

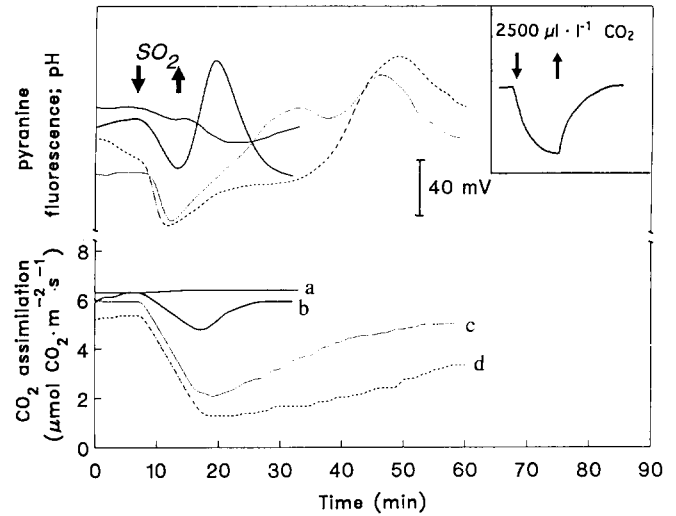


Fig. 11. Changes in the fluorescence of pyranine and CO₂ assimilation of a *Pelargonium* leaf during and after 6-min fumigation periods with SO₂. Concentrations of SO₂ were 1.3 (a), 2.6 (b), 5.14 (c), and 7.7 μl · l⁻¹ (d). The irradiation was 120 W · m⁻² and the CO₂ concentration 500 μl · l⁻¹ in air. The inset shows acidification after gassing the same leaf with 2500 μl · l⁻¹ CO₂. The decrease in cytosolic pH is about 0.05 pH units (Yin et al. 1990)

that oxidative detoxification of SO₂ yields sulfuric acid which, as a strong divalent acid, does not add to cytosolic buffering, whereas CO₂, as a weak acid, increases the buffering capacity of the cytosol.

Some time after termination of fumigation, pyranine fluorescence returned to initial values. Recovery of the original pH was characterized by overshoot or even oscillatory phenomena (Fig. 11). The observations show that fumigation with SO₂ causes the formation of acid which decreases the cytosolic pH in leaves, and that leaves can compensate for acidification even during fumigation either by mobilizing base or by exporting acid to a compartment which does not contain much pyranine, or whose pH is so far from the pK of pyranine that pH changes resulting from acid import do not cause much fluorescence change (Yin et al. 1990). Recovery of pH was always somewhat faster than recovery of photosynthesis.

Role of light and darkness in the effect of SO₂ on the photosynthetic apparatus. When an illuminated leaf of *Pelargonium* photosynthesizing in the absence of SO₂ was darkened, a slow increase in 505-nm absorbance was observed even when no SO₂ was added in the dark (Fig. 12). Its origin is unknown. The typical increase in 505-nm absorbance induced by SO₂ in the light (see Fig. 1) was absent in the dark. However, it appeared when, after short fumigation with SO₂ in the dark, the leaf was illuminated in the absence of atmospheric SO₂ (Fig. 12). Inhibition of CO₂ assimilation by the leaf which had been fumigated with SO₂ during the preceding dark period indicated that SO₂ had already interacted with components of the photosynthetic apparatus in the dark. This interaction proceeded slowly and progressively. Inhibition of the photosynthetic apparatus in the dark

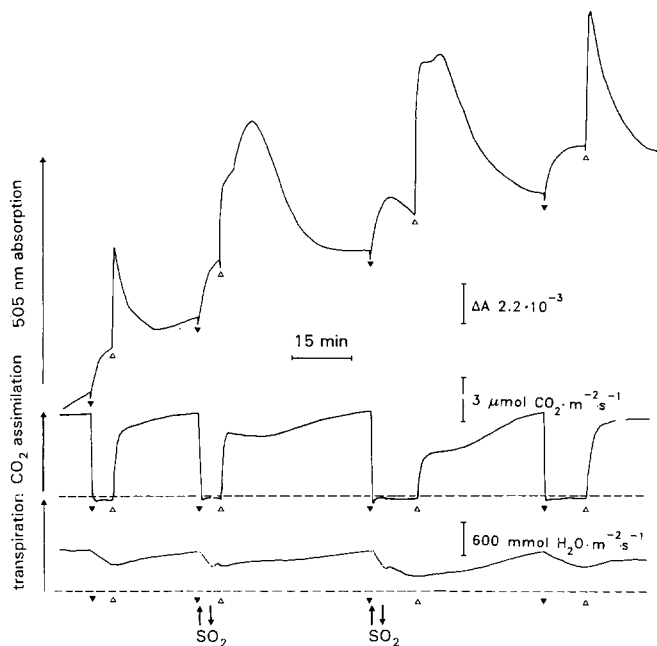


Fig. 12. Effects of fumigation with SO₂ in the dark on a representative *Pelargonium* leaf. Simultaneous recording of CO₂ exchange, transpiration and 505-nm absorbance. An illuminated leaf was first briefly darkened and then reilluminated (control). Closed triangles indicate light off, open triangles light on. During a second darkening period, the leaf was fumigated for 3 min with 7 μl · l⁻¹ SO₂, and the light was turned on after a further 2 min period in the dark. After CO₂ assimilation had recovered, the leaf was again darkened and SO₂ (7 μl · l⁻¹) was added for 3 min to the air stream passing over the leaf. Eight minutes after termination of fumigation, the light was turned on. A fourth dark period without SO₂ served as an additional control. The irradiance was 120 W · m⁻² and the CO₂ concentration in air 600 μl · l⁻¹

becomes visible as inhibition of photosynthesis when a leaf which has been exposed to SO₂ only during a dark period is subsequently illuminated. Interestingly, pyranine fluorescence failed to indicate acidification of leaves when SO₂ was added in the dark, whereas CO₂ caused the same acidification in the dark and in the light (data not shown).

When a *Pelargonium* leaf photosynthesizing in air was fumigated for 3 min with SO₂ and the light was turned off when photosynthesis inhibition and zeaxanthin formation were maximal, a 10-min dark period increased inactivation of the photosynthetic apparatus beyond that observed in the light (Fig. 13). During darkening, the zeaxanthin content remained high. It decreased as a subsequent illumination period resulted in the slow recovery of photosynthesis.

The observations show that SO₂ is particularly inhibitory in the dark, and that light has a dual role. In combination with SO₂, it may increase inhibition of assimilation as shown by the transient decrease of photosynthesis after the initial increase when the light was turned on after a dark period (Figs. 12, 13), but light also helps in reversal of inhibition and detoxification of SO₂.

The different pH responses of leaves to SO₂ in dark-

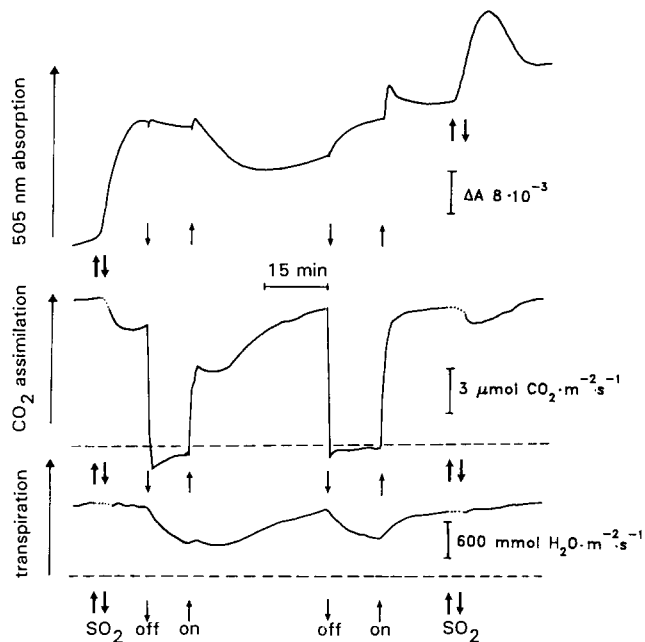


Fig. 13. Effects of illumination and an intermittent dark period (12 min) on inhibition of assimilation by SO₂ and on the 505-nm absorbance change. A leaf of *Pelargonium* was fumigated for two 3-min periods with 3.85 μl · l⁻¹ SO₂. Simultaneous recording of CO₂ and H₂O exchange and of 505-nm absorbance. The irradiance was 120 W · m⁻² and the CO₂ concentration 600 μl · l⁻¹

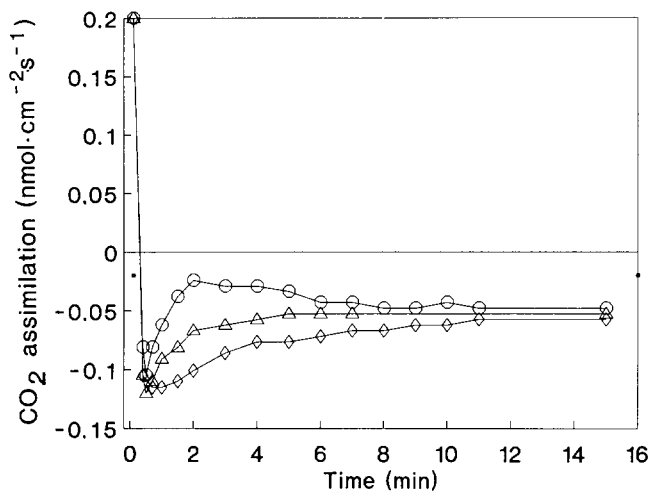


Fig. 14. Stimulation of post-illumination CO₂ production of a *Pelargonium* leaf by SO₂. A control leaf was illuminated for 30 min and then darkened (○). After another 20 min in the light, the same leaf was fumigated with either 4 (Δ) or 6 (◇) μl · l⁻¹ SO₂ for 5 min. Illumination was continued for another 10 min

ness and in light as monitored by pyranine fluorescence closely correspond to the observations made with 505-nm absorption which monitors zeaxanthin formation in the thylakoid membranes. In contrast to pyranine fluorescence which records cytosolic pH events, zeaxanthin formation responds to acidification of the thylakoid interior. Apparently, membrane barriers do not prevent

pH communication between the different cellular compartments (Laisk et al. 1989; Yin et al. 1990; Thaler et al. 1992).

Post-illumination CO₂ evolution after fumigation of an illuminated leaf in air with SO₂. Figure 14 shows post-illumination CO₂ evolution after an assimilating *Pelargonium* leaf had been fumigated for 5 min in air with SO₂ (4 and 6 μl · l⁻¹) and then kept for another 5 min in the light before it was darkened. In the control leaf, a transient CO₂ burst is attributed to CO₂ evolution from the pool of photorespiratory intermediates. After illumination in the presence of SO₂, CO₂ evolution was increased, and the increase was larger after fumigation with 6 μl · l⁻¹ SO₂ than after fumigation with 4 μl · l⁻¹ SO₂.

Discussion

Our results show that SO₂ inhibits photosynthesis reversibly. Loss of photosynthesis in the presence of high concentrations of SO₂ is due to inhibition of Calvin-cycle enzymes such as fructose biphosphatase and ribulose biphosphate carboxylase, not to inhibition of thylakoid reactions (Ziegler 1973; Tanaka et al. 1984; Figs. 2, 5, 6, Table 1). Formation of zeaxanthin by the thylakoid enzyme violaxanthin de-epoxidase is even stimulated by SO₂. After inhibitory reactions have resulted in appreciable loss of photosynthesis during and after a short period of fumigation with high SO₂ concentrations, detoxification is indicated by the recovery of photosynthesis. It was fast in the light and slow or absent in the dark. Obviously, chloroplast reactions are mainly responsible for detoxification. Removal of bisulfite and sulfite by reduction and incorporation of reduced sulfur into amino acids or by oxidation to sulfuric acid, the main detoxification pathways for SO₂ (Dittrich et al. 1992), appear to facilitate dissociation of addition compounds between enzymes and SO₂ or its hydration products. This can explain the observed reactivation of the enzymes. Oxidative detoxification of SO₂ burdens the cells with sulfuric acid. Indeed, during and after fumigation with SO₂, cytoplasmic acidification was indicated in the light by a transiently decreased fluorescence of the pH-indicating dye pyranine (Fig. 10). Decreases in the cytosolic pH were accompanied by increases in the level of zeaxanthin (Figs. 1, 4, 7, 9) which are attributed to a lowering of the intrathylakoid pH (Fig. 8).

Importantly, SO₂ failed to produce an appreciable acidification response in darkened leaves, but was even more inhibitory to photosynthetic enzymes in the dark than in the light (Figs. 12, 13). In darkened leaves, SO₂ neither caused appreciable quenching of pyranine fluorescence nor zeaxanthin formation, whereas high concentrations of CO₂ both quenched pyranine fluorescence in the dark (Yin et al. 1990) and induced zeaxanthin formation (O. Björkman, Department of Plant Biology, Carnegie Institution of Washington, Stanford, USA; personal communication) and increased absorbance at 505 nm (data not shown).

Hydration of SO₂ must result in the formation of

divalent sulfurous acid which liberates at least 1.5 protons per SO₂ at cytosolic pH values. The failure to observe appreciable acidification during short-term fumigation of leaves with SO₂ in the dark then either means that SO₂ can be absorbed in a form which does not act as an acid, or that cytoplasmic buffering and the action of cytoplasmic pH-stat mechanisms effectively prevented acidification in the dark, but not in the light. At any rate inactivation of the photosynthetic apparatus in the dark cannot be attributed to cellular acidification but must result from the chemical reactivity of SO₂ or of the reactive species derived from SO₂.

In isolated spinach chloroplasts, both reductive and oxidative detoxification of SO₂ proceed at comparable rates, but only when the production of oxygen radicals, which promotes sulfite oxidation in a fast chain reaction, is curtailed by the addition of radical scavengers such as superoxide dismutase and ascorbate, and when reduction is stimulated by the addition of O-acetylserine which permits formation of the reduction product cysteine (Ghisi et al. 1990; Dittrich et al. 1992). Cells of intact leaves are usually well equipped with radical-scavenging systems, but may, in the absence of rapid growth, lack sinks for sulfur-containing amino acids. During fumigation of young spruce trees with SO₂ over a period of weeks, essentially all absorbed SO₂ was recovered as sulfate (Kaiser et al. 1991).

Under normal conditions of plant exposure to a polluted atmosphere, SO₂ concentrations are unlikely to ever increase to the concentrations which were used in the present work, and which caused photosynthesis inhibition within a very short time span. However, whereas exposure times were very short in our experiments, plant exposure in the field to a polluted atmosphere may be chronic, even though maximum concentrations of SO₂ will rarely rise beyond 0.1 μl · l⁻¹. Our data suggest that oxidative and, when protein demand is high, reductive detoxification will normally be able to cope with influx of SO₂ in the field so that reactive anions are very unlikely to accumulate under field conditions. However, oxidative detoxification produces sulfuric acid. Sulfate accumulates in the vacuoles. The protons of sulfuric acid must be neutralized. Sulfate concentrations in needles of spruce from the Ore Mountains at the border between Germany and Czechoslovakia may contain more than 100 mM sulfate (Kaiser et al. 1991). The equivalent protons (200 mM) cannot be stored in the cells. If, owing to cation deficiency of the soil or other reasons, effective cation/proton exchange is not possible first at the root/soil interface and then inside the organism, acidification or, in other words, cation deficiency will be a more important cause of eventual damage than reactions between reactive anions and enzymes (Slovik et al. 1992). Cation deficiency is indeed indicated by known needle symptoms such as needle yellowing in Norway spruce (Beyschlag et al. 1987), but its linkage to acidification has not yet been generally recognized.

As a main result of the present work, it has become clear that the observed inhibition of photosynthesis in leaves by high concentrations of SO₂ is not caused by cellular acidification but by the chemical reactivity of

SO₂ or the anions formed during its hydration, and that such inhibition is reversible. Efficient cellular detoxification of SO₂ requires light. It causes cellular acidification. Obviously, neutralization of the acid formed requires a base which must become available if damage is to be avoided in the long term.

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