

Did Primitive Microorganisms Use Nonhem Iron Proteins in Place of NAD/P?

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Abstract. Nicotinamide adenine dinucleotide (NAD) and nicotinamide adenine dinucleotide phosphate (NADP) are of universal occurrence in living organisms and play a central role in coupling oxidative with reductive reactions. However, the evidence that the origin and early evolution of life occurred at high temperatures (>95°C) is now strong, and at these temperatures some modern metabolites, including both the reduced and oxidized forms of these coenzymes, are unstable. We believe there is good evidence that indicates that in the most primitive organisms nonhem iron proteins carried out many or all of the functions of NAD/P(H). This has important implications for the way in which investigations of archaeobacterial metabolism are conducted.

Key words: Archaeobacteria — Evolution of metabolism — Nicotinamide coenzymes — Nonhem iron proteins — Redox coenzymes — Thermophiles

Introduction

Of the three kingdoms of living organisms (Woese and Fox 1977a; Fox et al. 1980) (Fig. 1), there is now a

significant body of evidence suggesting that the common ancestor of the archaeobacteria is likely to be most closely related to the earliest form of life. Woese (1987; Woese et al. 1990; Fischer et al. 1983) and others have argued that the ancestral archaeobacterium was an extremely thermophilic anaerobe, dependent on sulphur reduction. Extreme thermophily (optimum growth temperature >70°C) is widespread among the archaeobacteria and the most extreme thermophiles of the group also appear to be the most slowly evolving (Woese 1987). Furthermore, extreme thermophily is also most strongly represented toward the root of the archaeobacterial tree (Woese et al. 1990). Indeed, the archaeobacteria most closely related to the ancestral phenotype of the archaeobacteria seem to be those with the highest optimum growth temperatures, over 100°C in some cases. Additional evidence for a high temperature origin of life is available when we consider the eubacterial evolutionary tree. The most deeply rooted branch in this tree is that leading to the thermotogales (Achenbach-Richter et al. 1987) (optimum growth temperatures up to 85°C), the most extremely thermophilic of the eubacteria (Huser et al. 1986; Huber et al. 1986). The next-most-deeply-rooted branch is that leading to the green nonsulphur bacteria containing a number of thermophiles such as *Chloroflexus* and *Thermomicrobium*. The third-most-deeply-rooted group, the deinococci, includes one of the most widely occurring eubacterial extreme thermophiles, *Thermus* (Woese 1987). Thermophiles are thus very strongly represented at the root of the eubacterial phylogenetic tree, and all these groups are also relatively slowly evolving, strongly sug-

Abbreviations: NAD/P(H): Oxidised and reduced forms of nicotinamide adenine dinucleotide and nicotinamide adenine dinucleotide phosphate

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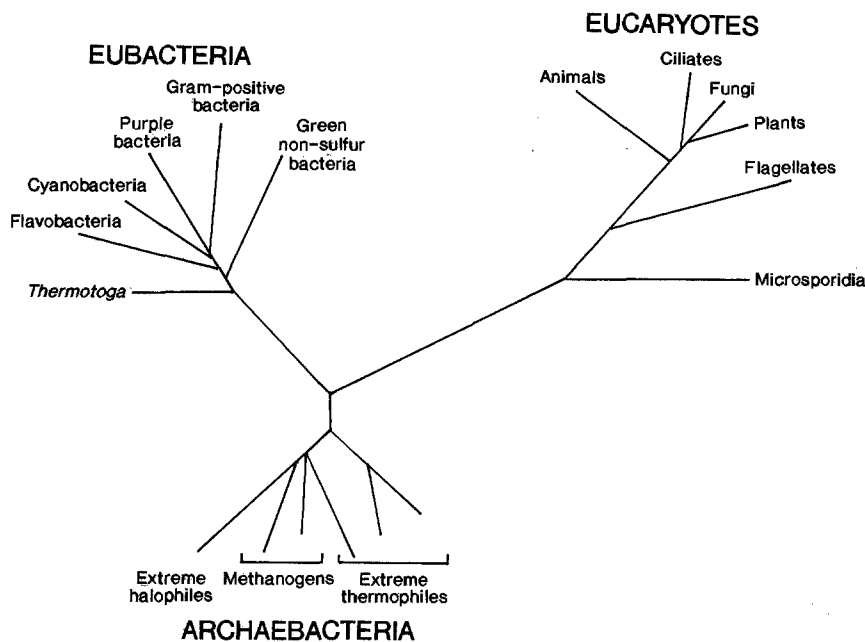


Fig. 1. Universal phylogenetic tree determined from rRNA sequences.

gesting that the eubacteria have also arisen from a thermophilic ancestor (Achenbach-Richter et al. 1987).

On this basis, it now seems likely that life arose at high temperatures, possibly in marine hydrothermal systems (Corliss et al. 1981; Russell et al. 1989; Holm 1992), and there is good reason to suspect that the earliest organisms (Woese and Fox 1977b) may have evolved and grown at temperatures at or above 100°C. Although proteins can be stable at these temperatures, some low-molecular-weight compounds are not (Daniel 1992; Coolbear et al. 1992). The early development of metabolic pathways may have been somewhat constrained by metabolite stability, but in the case of some key metabolites, including widely used coenzymes such as NAD/P(H), such a constraint could have posed serious problems. Even below 100°C NADH (Walsh et al. 1983) and NADPH (R. Hudson, personal communication) are quite unstable, having half-lives of only a few minutes at neutral pH at 95°C. The rate of degradation is unaffected by the presence of a dehydrogenase or of reducing agents and, while the end products of the breakdown are not known, they cannot be re-reduced by a dehydrogenase and the appropriate cosubstrate (Hudson, personal communication). NMR spectra of NADP solutions show significant changes after 30 min at 90°C (Consalvi, personal communication). Given the fundamental importance of NAD/P(H) in metabolism, and the apparent lack of any stabilizing mechanism, did primitive organisms have a means of circumventing the potential constraints posed by the instability of NAD/P(H)?

Proposal

Both nonhem iron proteins and NAD/P(H) are redox coenzymes. Recent research shows that in some archae-

bacteria nonhem iron proteins fill the role played by NAD/P(H) in more modern (in the sense of less closely related to the progenote) organisms. We propose that in primitive microorganisms, notably primitive archaeobacteria, more of these functions were carried out by nonhem iron proteins, i.e., that nonhem iron proteins partially or fully replaced the use of NAD/P(H). During evolution nonhem iron proteins were largely replaced by NAD/P(H), possibly during the adaptation of organisms to lower (<90°C) temperatures.

Evidence

Nonhem iron proteins seem likely to have occurred early in evolution. Hall et al. (1971) have proposed that a ferredoxin may have been among the earliest proteins formed, a view based partly on their simplicity (in anaerobic bacteria), and on the observation that their amino acid composition is dominated by amino acids that are readily synthesized abiogenically and that have been found to occur extraterrestrially. Wächtershäuser (1992) also considers nonhem iron proteins to be the earliest of the redox catalysts. Given the important role some authors (Wächtershäuser 1988, 1992; Russell et al. 1988, 1989, 1990, 1992; Cairns-Smith et al. 1992) have proposed for iron sulphides in the origin of life, it would not be surprising if iron-sulphur-containing proteins played an important role in primitive metabolism.

The upper temperature limit for protein stability is not known, but a variety of proteins have been found with half-lives of tens of minutes at temperatures above 100°C (Coolbear et al. 1992). Primitive organisms probably grew at ~100°C, and there is no reason to doubt the feasibility of a nonhem iron redox protein stable and

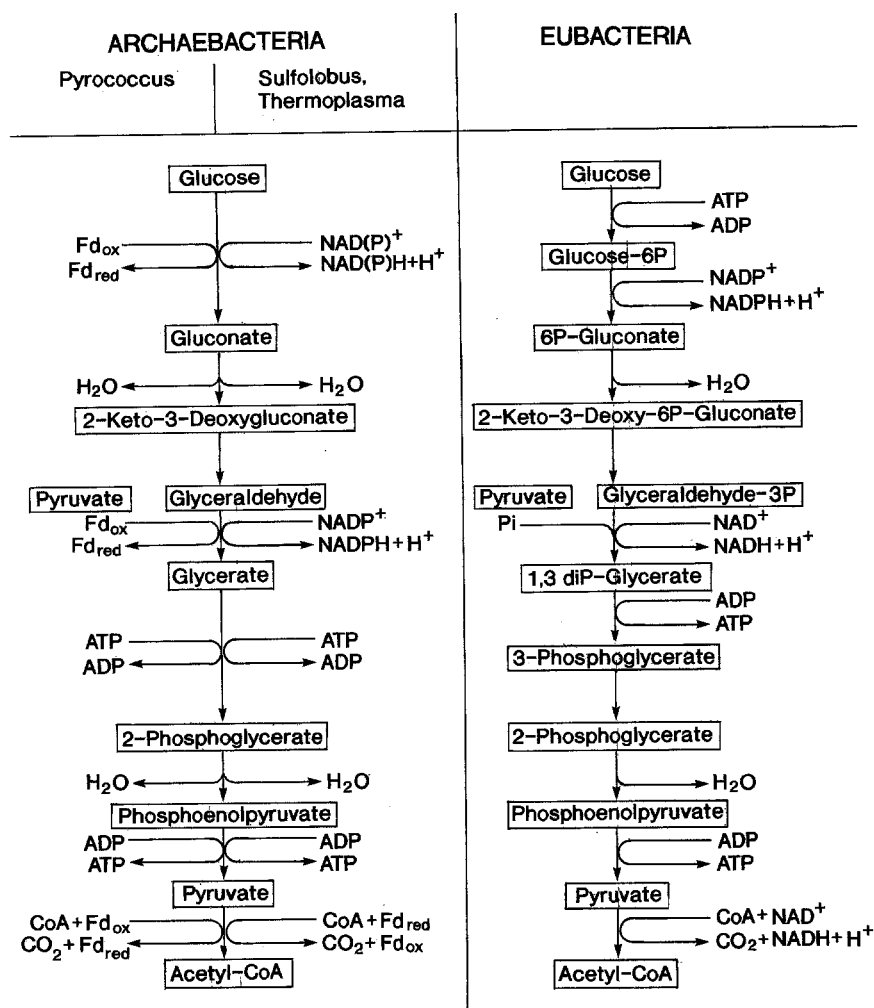


Fig. 2. The Entner-Doudoroff pathways of glucose oxidation in thermophilic archaeobacteria and in eubacteria. The coenzymes for primitive archaeobacteria (*Pyrococcus* [Makund and Adams 1991]) and less-primitive (Kjems et al. 1992) archaeobacteria (*Sulfolobus* [De Rosa et al. 1984] and *Thermoplasma* [Budgen and Danson 1986]) are shown on the same pathway.

functional at 100°C. Indeed, the aldehyde oxidoreductase from the hyperthermophile *Pyrococcus furiosus* belongs in this category (Makund and Adams 1991; Park et al. 1991).

There are a number of enzymes in the central metabolic pathways of thermophilic archaeobacteria that use nonhem iron protein for oxidation-reduction reactions, where NAD and NADP are used in their nonarchaeobacterial (and less primitive) counterparts (Danson 1988). For example, in eukaryotic and the majority of eubacterial organisms, pyruvate is oxidatively decarboxylated to acetyl-CoA via the pyruvate dehydrogenase multienzyme complex (Perham 1991). This enzyme system uses NAD as the oxidant, with lipoic acid as the protein-bound carrier of both acyl and reducing moieties. A similar oxidative decarboxylation of pyruvate is carried out by the archaeobacteria, but the enzyme catalyst is an iron-sulphur pyruvate oxidoreductase, having no dependency on NAD/P (Kerscher and Oesterhelt 1982). The iron-sulphur cluster abstracts the electrons from the oxo-acid and passes them to ferredoxin (Fig. 2). In the hyperthermophile *P. furiosus*, oxidized ferredoxin is regenerated through a hydrogenase with the production of molecular

hydrogen (Makund and Adams 1991). The pyruvate oxidoreductase is found in all archaeobacterial phenotypes (thermophiles, halophiles, and methanogens) and also in anaerobic eubacteria; therefore, Kerscher and Oesterhelt (1982) have proposed that this enzyme existed before the divergence of the three Kingdoms and that the NAD-linked dehydrogenase complex evolved after the development of oxidative phosphorylation.

A further example of the use of nonhem iron proteins in place of NAD/P(H) is provided by the pathway of glucose catabolism in *P. furiosus*. It has been proposed (Makund and Adams 1991) that this hyperthermophilic archaeobacterium oxidizes glucose to pyruvate via a non-phosphorylated Entner-Doudoroff pathway in which the redox reactions are catalyzed by ferredoxin-linked oxidoreductases (Fig. 2). In the less-primitive thermophilic archaeobacteria, *Sulfolobus* and *Thermoplasma* (Kjems et al. 1992), glucose is also catabolised via this nonphosphorylated pathway, but the equivalent redox enzymes (glucose dehydrogenase and glyceraldehyde dehydrogenase) are NAD/P-linked enzymes (De Rosa et al. 1984; Budgen and Danson 1986) (Fig. 2). Interestingly, in those eubacteria that possess an Entner-Doudoroff path-

way, the redox enzymes are also NAD/P-linked but the pathway comprises phosphorylated intermediates (Fig. 2).

A nonhem iron protein from *P. furiosus* has been shown to transfer electrons to a metal electrode without the need for a mediator (Park et al. 1991). It seems likely that this is due to a reduced distance between the electrode and the redox center of the enzyme, implying a relatively exposed site for the iron-sulphur cluster. This exposure should facilitate the exchange of electrons between different nonhem iron proteins to have enabled them to function as a "redox currency" in the same way as NAD/P(H) couples do in modern organisms. It might also confer on such nonhem iron proteins an enhanced ability to carry out coenzyme-type redox reactions in solution, e.g., the transfer of electrons to a separate hydrogenase.

In any event, the wider redox potential range over which nonhem iron proteins act compared with NAD/P(H) may have favored their early use as redox coenzymes.

Some of the properties of the NAD/P(H)-linked dehydrogenases in archaeobacterial enzymes may be interpreted as being due to less well-developed enzyme-coenzyme interactions than in the eubacterial and eukaryotic equivalents. The dehydrogenases from thermophilic archaeobacteria can often use both NAD and NADP, and in some cases with equal efficiency, whereas most dehydrogenases from eubacteria and eukaryotes are specific for one or the other (Perham 1991). Examples of dual-cofactor specificity in thermophilic archaeobacteria include isocitrate dehydrogenase from *Sulfolobus acidocaldarius* (Danson and Wood 1984), malate dehydrogenase from *S. acidocaldarius*, *Thermoplasma acidophilum* (Grossebuter et al. 1986), and *Methanothermobacter feravidus* (Honka et al. 1990), glyceraldehyde dehydrogenase from *M. feravidus* (Fabry and Hensel 1987) and *Pyrococcus woesei* (Zwickl et al. 1990), and glutamate dehydrogenase from *P. furiosus* (Robb et al. 1992) and *S. solfataricus* (Consalvi et al. 1991).

Finally, there are some data (Makund and Adams 1991) to suggest that free NAD/P(H) levels are lower in thermophilic archaeobacteria, although it is not clear whether this reflects lower total intracellular NAD/P(H) or whether these are more tightly bound to dehydrogenases.

Predictions

If our suggestion that the use of nonhem iron proteins as redox coenzymes preceded the use of NAD/P(H) is correct, then there will be fewer NAD/P-linked enzymes and more nonhem iron protein-coupled enzymes in primitive archaeobacteria than in more modern organisms. This possibility should be borne in mind when assaying for dehydrogenases in the archaeobacteria; that is, the absence

of an NAD/P-dependent activity may not imply that the dehydrogenase in question is absent, but rather that the nicotinamide coenzymes are not the redox agents. The pathway of glucose catabolism in *P. furiosus* is one such example (Makund and Adams 1991). To detect dehydrogenases in primitive organisms, monitoring the change of levels of substrate or product in relatively undiluted cell extracts, using an external oxidant or reductant and in the presence and absence of an iron chelator, should supplement the monitoring of the redox change of NAD/P(H).

Although nonhem iron proteins from "modern" organisms transfer electrons between proteins, those from "primitive" archaeobacteria may show properties that especially fit them to this role. We take the ability of a *P. furiosus* nonhem iron protein to transfer electrons directly to a metal electrode (Park et al. 1991) to be an example of this.

The structures of the dual-specific NAD/P-linked dehydrogenases in primitive archaeobacteria will be less well adapted in terms of their protein-coenzyme interaction than the equivalent enzymes in nonarchaeobacterial species.

Concluding Remarks

Clearly, further data are required to test these possibilities, especially needed is an increased knowledge of the metabolic pathways in the hyperthermophilic archaeobacteria and of detailed 3D structures of their NAD/P-linked dehydrogenase and nonhem iron protein-linked oxidoreductase enzymes. Furthermore, it will be particularly interesting to investigate, at the primary sequence level, any evolutionary relationship within the archaeobacteria between an NAD/P-linked dehydrogenase and a nonhem iron-linked oxidoreductase catalyzing the oxidation of the same substrate (e.g., glucose dehydrogenase [De Rosa et al. 1984; Budgen and Danson 1986] and glucose oxidoreductase [Makund and Adams 1991]).

It is tempting to speculate that some of the metabolic pathways observed in modern organisms and not in some primitive archaeobacteria—for example, the phosphorylated Entner-Doudoroff pathway, may have developed only when NAD/P(H) became available. The removal of the limitations imposed by the use of a high-molecular-weight redox coenzyme, such as steric hindrance and relatively low diffusion rates compared with NAD/P(H), may have been a factor. However, it is not clear to us how this effect may be distinguished from other reasons for the development of alternative metabolic pathways.

Finally, it is of course clear that NAD/P(H) is not the only metabolite whose instability may have posed constraints for primitive organisms growing >90°C. However, as we have indicated above, we believe that in the case of NAD/P(H) there is good evidence how this might have been overcome. It will be interesting to see whether, as our knowledge of more primitive and extreme ther-

mophilic metabolism grows, we can see similar evidence in respect to other unstable metabolites.

Note Added in Proof

Kengen et al. [(1994) *J Biol Chem* 269:17537–17541] and Schafer et al. [(1994) *FEMS Microbiol Lett* 121:107–114] have recently presented evidence for glucose catabolism by an Embden-Meyerhof pathway in *P. furiosus*. Kengen et al. (1994) found that the glyceraldehyde 3-phosphate dehydrogenase would accept either NADP or benzyl viologen (and therefore probably ferredoxin) as electron acceptor, providing evidence for the predictions made above. The discovery of ADP-kinases in the same pathway may, given the relatively higher stability of ADP, be another example of how the instability of a low molecular weight metabolite (ATP) can be overcome in extreme thermophiles.

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