

Numerical Simulation of Methane Oxidation in Actively-Aerated Landfill Biocover

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Abstract. Methane oxidation in landfill biocover is a complex process involving water, gas and heat transport as well as microbial oxidation. Such technology is useful for dealing with landfill gases. With the aim of finding the most suitable design configuration for actively-aerated methane biocover, an air injection system is proposed in this work to ensure that oxygen availability is no longer a limitation for methane degradation. A numerical model is developed that incorporates water-gas-heat coupled transport in layered landfill biocover with the consideration of methane oxidation and air actively-aerated injection. The model is verified and calibrated using published data from a laboratory soil column test. Moreover, parametric studies are carried out to investigate the influences of air injection amount and location. It is found that injecting air into the biocover through several inlets along the biocover bed is a promising approach to enhance methane oxidation in the cover.

Keywords: Landfill biocover \cdot Aeration \cdot Methane oxidation \cdot Gas transport Numerical simulation

1 Introduction

Landfill is a site stacking the municipal solid waste (MSW) and has wide application worldwide due to its low cost and environmental-friendly characteristics [\[1](#page-7-0)]. The cover system, as the essential structure of landfill, is designed to impede the penetration of rainfall and emission of landfill gas (LFG). One main component of LGF is methane (CH4) which is one of the greenhouse gases and has much higher global warming potential than carbon dioxide (CO₂) [\[2](#page-7-0)]. Therefore, it is necessary to mitigate CH₄ emission from landfill effectively and economically.

Landfill biocover is one type of landfill cover, which is effective in impeding the emission of methane from landfill into atmosphere using biotic methane oxidation. It is quite useful in landfills, especially those lacking landfill gas extraction system [[3\]](#page-7-0). The oxidation that occurs in biocover can consume $CH₄$, generate $CO₂$ and water, and release heat, which can be described by the following equation:

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$$
CH_4 + (2-x)O_2 \rightarrow (1-x)CO_2 + (2-x)H_2O + x - CH_2O \quad + \text{heat} \tag{1}
$$

where $-CH₂O-$ is organics and x is the reaction coefficient.

The oxidation reaction is affected by numerous factors, such as $CH₄$ concentration, $O₂$ concentration, temperature, moisture and other properties. The heat released by reaction can reach 632 kJ/mol [[4\]](#page-7-0). Passive aeration is usually used in traditional biocover. O₂ naturally intrudes into the cover by diffusion and provides the required $O₂$ for oxidation reaction. The diffusion of O_2 depends on the O_2 concentration gradient in cover soils [\[5](#page-7-0)]. Recent researches reveal that the maximum reaction depth of CH_4 is approximately 30 cm, which is limited by the O_2 distribution in cover. Increasing O_2 amount in cover artificially is a possible way to enhance the methane oxidation [[6\]](#page-7-0).

In this study, a new cover structure is proposed. The new cover structure can adjust gas distribution in cover using actively-aerated injection system, which is beneficial for deepening the $O₂$ penetration depth and reducing LFG emission. A numerical model is built to simulate water-gas-heat transport in the new landfill cover structure.

2 Conceptual Model of Actively-Aerated Landfill Cover System

In order to better reduce methane emission, a new landfill cover structure is proposed here. As shown in Fig. 1, the landfill cover consists of three layers with different functions. At the bottom, a gravel layer is used to make gas uniformly distributed. The gravel layer is overlain by a clay layer whose function is impeding methane emission. The silt layer is the reaction layer, which is located at the top. As the major place where the methanobacteria grows, the silt layer also provides the environment for the growth of vegetation. Additionally, the silt layer also helps to avoid cracking of the clay layer under dry condition.

In this study, soil is assumed to be a three-phase porous medium containing liquid, gas and solid particles. The liquid phase is water, while the gas phase is a four-component gas mixture $(CO_2, O_2, CH_4$ and $N_2)$. Gas migration involves advection and diffusion. After the methane migrates from the MSW to the silt layer, methane

Fig. 1. Actively-aerated landfill cover system

oxidation reaction occurs, generating water, carbon dioxide and organic matter. The heat released by the reaction in turn affects the biochemical reaction rate. Thus it is a water-gas-heat coupled process.

In the silt layer, O_2 is required for the oxidation reaction. Therefore, O_2 concentration has significant impact on the oxidation reaction efficiency. In traditional landfill cover, the available O_2 for oxidation reaction in the landfill cover relies on the diffusion of O_2 from atmosphere to the landfill cover, which makes the O_2 concentration becomes the main restriction for the oxidation reaction. To overcome the limitation, horizontal aeration wells are laid in silt layer injecting air under low pressure. In this way, O_2 concentration in landfill cover can be increased artificially to improve oxidation reaction efficiency.

3 Numerical Model of Actively-Aerated Landfill Cover System

In this section, a multi-field coupled model is developed to describe the water-gas-heat transport processes in the landfill cover.

3.1 Governing Equation

In this model, gas is regarded as the ideal gas. Heat balance is assumed to exist between two phases, so each phase has the same temperature at the same location. Thus only one heat transfer governing equation is required [[7\]](#page-7-0).

Based on the conservation of energy and mass, water transfer governing equation can be expressed as

$$
\rho_w \frac{\partial \theta_w}{\partial t} = -\nabla (\rho_w v_w) + \rho_{DB} M_{\text{H}_2 \text{O}} r_w \tag{2}
$$

where ρ_w is the water density; ρ_{DB} is the dry density of soil; v_w is the water flow velocity; θ_w is the volumetric water content; M_{H2O} is the molar mass of water; r_w is the reaction rate of water.

The gas transfer governing equation can be expressed as

$$
\frac{\partial}{\partial t}\left[(1 - S_w)\phi c_g + S_w \phi H_w \right] = -\nabla \left[v g c_g \right] - \nabla \left[v_w H_w \right] - \nabla N_g \pm \rho_{DB} r_g \tag{3}
$$

where ϕ is the soil porosity; N_g is the diffusive flux of gas; S_w is the saturation degree; v_g is the gas mixture advective velocity; c_g is the gas molar concentration; H_w is the gas molar concentration dissolved in water; r_g is the gas reaction rate in methane oxidation; "+" indicates gas generation such as CO_2 ; and "−" denotes gas consumption such as O_2 and $CH₄$.

The heat transfer controlled biochemical equation can be expressed as

$$
\frac{\partial [W(T - T_r)]}{\partial t} = -\nabla(-\lambda_T \nabla T + H_{conv}) + H_{oxi}
$$
\n(4)

where W is the soil heat capacity (J m⁻³°C⁻¹); T_r is the temperature (°C) and its value takes 22 °C in this study; H_{conv} is the heat per unit area (J m⁻² s⁻¹); H_{oxi} is the heat generation rate of the methane oxidation (J m⁻³ s⁻¹); and λ_T is the thermal conductivity $(\text{J m}^{-1} \text{ s}^{-1} {}^{\circ}\text{C}^{-1})$ and the function of water content.

Microbial aerobic methane oxidation (MAMO) is controlled by the concentrations of CH_4 and O_2 , temperature and water content in soil. MAMO can be described by Eq. ([1\)](#page-1-0) mentioned. That $x = 0.5$ is adopted in Eq. (1) gives:

$$
CH_4 + 1.5O_2 \rightarrow 0.5CO_2 + 1.5H_2O + 0.5 - CH_2O - \text{heat.}
$$
 (5)

It is necessary to determine the methane oxidation rate r_g^{CH4} (kg m⁻³ s⁻¹) which can help to investigate the effect of methane oxidation on water, gas and heat transfer:

$$
r_g^{\text{CH}_4} = -\frac{V_{\text{max}}y_{\text{CH}_4}}{K_m + y_{\text{CH}_4}}\tag{6}
$$

where V_{max} is the maximum reaction rate; y_{CH_4} is the CH₄ concentration; and K_m is the half-saturated constant for $CH₄$.

Methane oxidation efficiency is used to evaluate the performance of the biocover (dimensionless), which can be calculated as follows:

$$
\text{Oxidation efficiency}(\%) = \frac{Q_{in}C_{in} - Q_{out}C_{out}}{Q_{in}C_{in}} \times 100\%
$$
\n⁽⁷⁾

where C_{in} and C_{out} are CH₄ inlet and outlet concentrations, respectively; Q_{in} and Q_{out} are the CH4 flow discharges at the inlet and outlet, respectively.

3.2 Numerical Implementation and Model Assumptions

The model is solved based on ANSYS Fluent platform and selected "mixture" model to solve multiphase conservation equation. The coupled pressure-velocity computation used the "phase-coupled SIMPLE" algorithm, and biochemical and kinetic equations are solved with the first-order implicit algorithm.

Since the main mode of gas transfer is convection and diffusion, the gas in the model is considered as laminar flow instead of turbulent flow, which is also determined by the rate of migration. Soil is considered as an ideal porous medium with uniform pore distribution.

4 Numerical Examples

Since researches about actively-aerated landfill cover system are quite limited. In this part, a soil column test [[1\]](#page-7-0) (Fig. [2\)](#page-4-0) is analyzed, which can be viewed as a simple actively-aerated landfill cover.

As shown in Fig. [2](#page-4-0), a 2D finite element model with a height of 90 cm and a width of 7.5 cm is used to simulate the soil column test. The holes are used to simulate the

aeration probes. CH_4 gas is introduced at the bottom of the model. The boundaries can transfer heat, but not gas and liquid. The top of the column is the exit of gas, which can simulate the real atmosphere environment. The initial soil moisture content is 0.236 and the initial methane and water vapor contents are zero, which are the same as the values in the test. The other needed parameters are summarized in Table 1.

First, the experimental test results [\[1](#page-7-0)] are adopted to verify the numerical model. Parametric study is then conducted to investigate the influences of air injection amount and location on CH₄ oxidation efficiency.

Fig. 2. Experimental setup for soil column test and finite element mesh

Parameter	Value	Reference
Porosity	0.73	$\lceil 2 \rceil$
Soil bulk density (kg m^{-3})	980	$\lceil 2 \rceil$
Soil particle density (kg m^{-3})	2280	$\lceil 2 \rceil$
Saturated volumetric water content	0.73	$\lceil 2 \rceil$
Residual volumetric water content	0.02	$\lceil 1 \rceil$
Maximum methane oxidation rate (kg m ⁻³ s ⁻¹)	5.6×10^{-6}	$\lceil 2 \rceil$
Half-saturated constant	0.0066	$\lbrack 3 \rbrack$
Intrinsic permeability (m^2)	5.8×10^{-12}	$\lceil 3 \rceil$
Van Genuchten's parameter (m)	0.33	$\lceil 3 \rceil$
Van Genuchten's parameter, $a \text{ (m}^{-1})$	5	$\lceil 3 \rceil$

Table 1. Parameters used in the simulation

4.1 Model Calibration and Verification

In the test, air aeration is applied through A and B equally (Fig. [2](#page-4-0)). The daily injection flux of CH_4 was 1235.41 g while the total amount of air aeration was 10 times that of CH4. The test lasted for 200 days. The comparison of gas concentration profiles on the $28th$ day is shown in Fig. 3a. It can be found that the calculated results agree with the experimental results reasonably well. $O₂$ concentration increases firstly and then decreases with depth, indicating that the horizontal aeration wells can effectively increase O_2 in the cover. CH_4 concentration keeps decreasing with depth, which reflects that CH_4 is consumed in the whole cover domain. CO_2 , as the product of oxidation reaction, increases from the bottom to the top. N_2 concentration distribution is stable and only affected by the concentration change in the other gases.

Figure 3b shows the comparison of CH_4 oxidation efficiency. After 18 days, the calculated results remain stable. The calculated results are overall enclosed by the observed values. The difference is caused by the change in bacteria quantity during the aeration process, which is not considered in the numerical model.

Fig. 3. Comparison between measured and computed results

4.2 Influences of Air Injection

Figure [4](#page-6-0) shows the CH_4 oxidation efficiency given different injected air amounts while $CH₄$ flux keeps unchanged. The daily injection flux of $CH₄$ is 1235.41 g and air flux varies from 2 to 10 times that of CH_4 . Air is injected through A and B equally and lasts for 200 days. The methane oxidation efficiency gradually increases with the aeration process for all the 5 curves and the results remain stable after about 18 days. The methane oxidation efficiency also increases with the air flux level. It is noteworthy that the methane oxidation efficiencies of the 8 times case and 10 times case are very close, indicating that higher air flux level can significantly enhance the oxidation efficiency but it is not economical to further increase the air flux level when it is high enough.

Figure [5](#page-6-0) shows the influence of the air injection location on methane oxidation efficiency. Three scenarios are simulated; namely, air injected through A (Scenario 1),

Fig. 4. Influence of injected air amount on methane oxidation efficiency

Fig. 5. Influence of air injection location on methane oxidation efficiency

A and B equally (Scenario 2), A, B and C equally (Scenario 3). The gas fluxes for the three scenarios are the same, which is 1235.41 g/d for CH₄ and 5068 g/d for air. The simulation lasts for 200 days. The methane oxidation efficiencies of Scenarios 1 and 2 are 65.8% and 95.8%, respectively, and that for Scenario 3 is slightly higher than that of Scenario 2.

5 Conclusion

A new landfill biocover with horizontal aeration wells is proposed in this study. A numerical model is developed to simulate the water-gas-heat coupled transport in the cover. The numerical model is verified against existing soil column experimental test results. The main conclusions are as follows:

(1) Laying the aeration well at the top of landfill cover system can effectively increase $O₂$ concentration in the landfill cover, which can enhance the methane oxidation reaction and is beneficial for impeding methane emission to the atmosphere.

- (2) Higher air flux level can significantly enhance the oxidation efficiency but it is not economical to further increase the air flux level when it is high enough (e.g., 8 times the methane flux in this study).
- (3) Given the same injected air amount, injecting air though more rows of wells gives higher methane oxidation efficiency. In this study, two rows of wells have similar oxidation efficiency as three rows of wells.

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