

# Chapter 10

## Wine, Beer and Cider: Unravelling the Aroma Profile

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**Abstract** The aroma profile of alcoholic beverages is a major factor that distinguishes one product from another, and it is a key attribute that drives consumer preference at points of sale. A longstanding objective has, therefore, been to identify those aromatic compounds that are important to particular olfactory attributes of different styles of wine, beer and cider—whether perceived ortho- or retro-nasally—and to modulate them according to consumer preferences. That this has been achieved only to a relatively small extent to date is partly a reflection on the complexity of the perception of aroma mixtures and also the presence of very low concentrations of potent aroma compounds in these products. It is known, although perhaps not appreciated as widely as it should be, that aroma compounds will interact with each other, with masking or suppressing effects being probably universal for compounds at supra-threshold concentrations, together with additive interactions for compounds at sub-threshold concentrations. Thus it is likely that volatile compounds with marginal aroma impact when isolated, can together provide an influence on aroma. Some of these aroma-active compounds are produced during fermentation. Different yeasts produce differing ranges of aroma-active substances, which may greatly affect the complex flavour of a fermented product such as wine, beer and cider. While these secondary metabolites are often formed only in trace amounts, their concentrations may well determine the distinct aroma of these beverages. This chapter reviews the production of the most important aroma-active compounds produced by yeast at molecular level and seeks to understand how they might be perceived by consumers.

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## 10.1 The Fundamentals of Aroma and Flavour Perception of Fermented Alcoholic Beverages

The aroma of a given foodstuff is formed by the pool of volatile molecules contained in that product with the ability to impact the olfactory receptors located in the olfactory region in the human nose. Since volatile molecules can reach this olfactory region both via orthonasal and retronasal pathways, the information elicited by those receptors affects two different sensory properties of products, i.e. their smell and their flavour. Smell is primarily information originating from the excited olfactory receptors together with, eventually, information produced by trigeminal nerve terminals located in the nostrils, which are also present in the mouth, pharynx and eyes. The information generated by these terminals falls in the category of chemesthesis and is related to semi-tactile properties such as refreshing/cooling effects (menthol), irritation (acids and alkalis) or pungency (chilies' capsaicins) (Bandell et al. 2004; Bautista et al. 2007; Caterina et al. 1997; Macpherson et al. 2006). Many aroma chemicals are able to produce some chemesthesis, although in general the intensity of the response is smaller than the purely olfactory (Prescott 1999a). On the other hand, the flavour of a product is formed by the cerebral integration of the different sensory responses elicited during the consumption of a product. These sensory responses stem from three different chemical sensory systems (olfaction, taste and chemesthesis) and in the tactile and thermal sensory systems which give information about the temperature and rheological properties of the food. From a qualitative point of view, the olfactory system is the one carrying the biggest amount of information and that explains the limited amount of sensory information that can be perceived when the nose is blocked.

Flavour is an integrated cerebral response and it, therefore, not always possible to clearly assign the origin of the stimulus causing a particular flavour perception (Delwiche 2004; Prescott 1999b). For instance, whenever vanillin is present together with sweet tastants, the intensity of its smell increases (Green et al. 2012) and vice versa, the presence of vanillin can increase the perception of sweetness (Sakai et al. 2001). These complex phenomena are broadly included into the concept of perceptual interactions that are responsible for some unexpected and important observations in wine flavour chemistry, such as the prominent role played by fruity aromas on the perception of sweetness, bitterness and astringency (Saenz-Navajas et al. 2010). Although these phenomena are well documented in the general scientific literature, they are not yet well understood in the context of wine and other fermented alcoholic beverages. It is therefore important to note that aroma compounds play a sensory role in fermented beverages that it is not limited to the perception of an odour.

To fully understand the role played by individual chemicals, it is equally important to note that fermented beverages are considered by flavour chemists as 'complex' products. Although there is no definite border between what products are categorised as 'simple' and 'complex', the fact that the aroma of any fermented

beverage is formed by at least 25 different aroma chemicals—all of which are present at concentrations above their corresponding odour thresholds—classes wine, beer and cider as ‘complex’ products. So, given the fact that the sense of olfaction has to be by nature a ‘synthetic’ rather than ‘analytic’ sense (Wilson and Stevenson 2003), the overall odour of fermented beverages has to be a global perception in which the individual chemicals are just poorly identified. In addition, since the odour of fermented beverages has accompanied humankind since the beginning of time, we can postulate the existence of what psychophysicists define as an ‘odour object’ (Ferreira 2012; Stevenson and Wilson 2007; Yeshurun and Sobel 2010). In other words, the human brain transforms the complex signals produced by the interactions of the chemicals present in all alcoholic beverages (alcohol, fusel alcohols, fatty acids, branched acids, ethyl esters, acetates, etc.) into a single unified concept that would be defined as ‘alcoholic’ or ‘vinous’. This is a highly efficient and ‘economic’ of signal processing and this capability has great practical importance in the understanding the chemical basis of aroma perception.

Therefore, although in the following sections we refer to individual compounds or groups of compounds and mention the specific odour properties of a compound or groups of compounds, it should not be concluded that those odour properties are directly responsible for such odour perception in wine, beer or cider.

## 10.2 The Basic Hierarchy of Aroma Compounds in Fermented Alcoholic Beverages

The previous notes, together with many experimental observations obtained in different reconstitution studies performed mainly in wine, make it possible to provide a basic rationale for understanding the contribution of the different chemicals to the aroma of a naturally fermented beverage. At the core of this rationale lies the aforementioned complexity of aroma and flavour perception in relation to fermented beverages and the so-called ‘aroma buffering effect’. The ‘buffering aroma effect’ of a product’s base refers to the demonstrated resistance of a particular aroma mixture to change its aroma both upon the elimination of some of its components or upon the addition of some new aroma compounds (Ferreira et al. 2002; Escudero et al. 2004). This does not mean that such a base always bears the same aroma and has the same composition. The composition of the base depends on some basic factors such as, for example, the concentration of sugar in a grape must, the prevalent yeast strain(s) and the degree of anaerobiosis during fermentation. For instance, the latter factor makes white and rosé wines richer in fatty acids and their ethyl esters while containing less alcohols and isoacids than red wines (Ferreira et al. 1996). Another less well-known factor is that the concentrations of fusel alcohols, fusel alcohol acetates, isoacids and their ethyl esters, all of them related to the yeast amino acid metabolism, are related to the varietal origin of the must (Ferreira et al. 2000; Hernández-Orte et al. 2002). Different

compositions mean slightly different aromas and slightly different buffering abilities.

Notwithstanding of this, there are some compounds or combinations of compounds, that at the ‘natural’ concentrations at which they usually occur in fermented beverages, can break the buffer and transmit to the mixture their specific aroma or a particular feature of their aroma. There are five different possibilities for this to happen:

1. The aroma buffer will be broken whenever a beverage contains an ‘aromatic vector’ with enough odour intensity.
2. Such an aroma vector can be an individual compound (a so-called ‘impact compound’), a family of aroma compounds belonging to the same chemical series, such as ethyl esters of fatty acids, or even an association of aroma compounds sharing a generic descriptor (such as ‘fruity’ or ‘floral’).
3. The vector will express in the beverage an aromatic nuance whose intensity and vicinity to the innate aroma character of the vector will be proportional to its concentration. For instance, isoamyl acetate bearing a characteristic smell of banana, if present at a low concentration does not transmit to the beverage its characteristic aroma. Rather it just transmits its generic ‘fruity’ character.
4. Hence, concentration modulates the role that the vector actually plays in the mixture. The following intensity categories can be identified:
  - a. null
  - b. minor contributor—the elimination of the vector does not bring about any clear aromatic change
  - c. neat contributor—the elimination of the vector brings about a clear decrease in the intensity of the aroma nuance to which the vector contributes
  - d. major contributor—in this case the elimination of such vector will cause a dramatic drop in the intensity of the aroma nuance with even possible qualitative changes in the overall aroma profile
  - e. impact compound—in this case the elimination of the vector will cause a dramatic change on the aroma profile.
5. When several aromatic vectors coexist in the same beverage, they will interact at perceptual level following three different potential patterns of interaction:
  - a. competitive—both aroma nuances are simultaneously and competitively perceived (Campo et al. 2005),
  - b. destructive—only one of the vectors is perceived at a lower intensity,
  - c. creative—a new aroma nuance emerges from the blend (San Juan et al. 2011).

Whenever it comes to fermentative compounds, they are the basic constituents of the base and some of them can also play the role of contributors to different aroma nuances. To the best of our knowledge, they only seldom play the role of

impact compounds. For instance, isoamyl alcohol or  $\beta$ -phenyl alcohols, even if they are present at relatively high concentrations never reach the level at which they would act as real impact compounds. So far, in wine, only isoamyl acetate, ethyl acetate, acetic acid, acetaldehyde and diacetyl can play such a role individually, and leaving aside some specific quirky wine styles, when they are clearly perceived the quality of the wine can be questioned. Fermentative compounds form the following aroma vectors:

1. Alcohols (ethanol, isobutanol, isoamyl alcohol).
2. Methionol.
3. Ethyl esters of fatty acids (ethyl butyrate, hexanoate, octanoate and decanoate).
4. Ethyl acetate.
5. Acetic acid.
6. Fatty acids (butyric, hexanoic, octanoic and decanoic acids).
7. Isoamyl acetate.
8.  $\beta$ -phenyl acetate and  $\beta$ -phenylethanol.
9. Branched fatty acids [isobutyric, isovaleric, 2-methylbutyric and the recently discovered 2-, 3 and 4-methylpentanoic and cyclohexanoic acids (Campo et al. 2007)].
10. Ethyl esters of the previously mentioned branched fatty acids.
11. Diacetyl.
12. Acetaldehyde.

As previously mentioned, these aroma vectors rarely reach the category of impact compounds, but they are important contributors to some key aroma nuances of fermented beverages. The role of two of the vectors, namely fatty acids and branched fatty acids, is quite complex, since apparently they form a kind of 'creative' interaction with the 'fruity' vectors (ethyl esters of fatty acids, branched acids and isoamyl acetate) to form the aroma of 'fresh fruit' (San Juan et al. 2011). Another aspect that must be kept in mind is that compounds of apparently 'bad' aroma, such as isovaleric and 2-methylbutyric acids, are in fact precursors for the strawberry-smelling ethyl isovalerate and ethyl 2-methylbutyrate which are formed by slow esterification of the acids with ethanol during ageing.

Bearing these fundamental aspects of aroma perception of fermented alcoholic beverages and the basic hierarchy of aroma compounds involved in mind, the following sections focus on the contribution of yeast to the aroma and overall quality of fermented beverages.

### 10.3 Wine Yeasts

Winemaking is a complex chemical and biological process in which different genera of yeast and bacteria are involved. During the early stages of spontaneous wine fermentation, different genera of non-*Saccharomyces* yeasts, such as *Candida*, *Cryptococcus*, *Debaryomyces*, *Hanseniaspora* and its asexual counterpart

*Kloeckera*, *Kluyveromyces*, *Metschnikowia*, *Pichia*, *Rhodotorula*, *Saccharomyces*, *Schizosaccharomyces* and *Zygosaccharomyces* play a role (Pretorius et al. 1999). Yeasts of the genera *Kloeckera*, *Hanseniaspora* and *Candida* predominate in the early stages, followed by several species of *Metschnikowia* and *Pichia* in the middle stages when the ethanol concentration rises to 3–4 % (Fleet and Heard 1993). However, some species of *Schizosaccharomyces*, *Zygosaccharomyces*, *Brettanomyces* and its sexual ('perfect') equivalent, *Dekkera*, are more resistant to high concentrations of ethanol and SO<sub>2</sub> and, if present under certain conditions, can adversely affect the sensory quality of wine.

On the other hand, the principal species conducting the alcoholic fermentation in grape wine is *Saccharomyces cerevisiae*, but the closely-related *Saccharomyces uvarum* (*Saccharomyces bayanus* var. *uvarum*) can also participate (Demuyter et al. 2004; Massoutier et al. 1998; Naumov et al. 2000, 2001; Sipiczki 2002, 2008). Both *S. cerevisiae* and *S. uvarum* are able to grow on substrates characterised by high sugar and ethanol content, low pH, high sulphur dioxide concentrations and remains of fungicides, demonstrating that they are genetically well adapted to winemaking conditions (Sipiczki 2008). However, *S. cerevisiae* has higher resistance to high temperature stress (up to 37 °C) and ethanol levels (up to 15 %) than *S. uvarum* (Belloch et al. 2008). From an oenological point of view, these *Saccharomyces* species differ in several properties. Comparison between *S. uvarum* and *S. cerevisiae* reveals that the former is more cryotolerant, produces less acetic acid, lower levels of amyl alcohols, but higher concentrations of glycerol, succinic acid, malic acid, isobutyl alcohol, isoamyl alcohol and numerous secondary compounds (Sipiczki 2002). Wines produced by *S. uvarum* strains have a higher aromatic intensity than those produced by *S. cerevisiae* (Coloretti et al. 2006; Henschke et al. 2000). Specifically, *S. uvarum* produces more of 2-phenylethanol, 2-phenylethyl acetate and ethyl lactate than *S. cerevisiae* (Antonelli et al. 1999; Di Stefano et al. 1981; Gangl et al. 2009). On the other hand, *S. uvarum* is less common and appears mainly in fermentations at low temperatures (Antunovics et al. 2003; Demuyter et al. 2004; Masneuf-Pomarède et al. 2010; Sipiczki et al. 2001).

Other members of the genus *Saccharomyces* (*S. cariocanus*, *S. kudriavzevii*, *S. mikatae*, *S. paradoxus*, *S. arboricolus*, *S. pastorianus*) are not likely to play important roles in wine fermentation (Sipiczki 2008). Nevertheless, *S. paradoxus* has been found in grapes in the north-western region of Croatia and it is currently used to ferment wines (Redzepovic et al. 2002). Likewise, *S. kudriavzevii* has only been isolated in natural environments, like decayed leaves (Naumov et al. 2000) or oak barks (Sampaio and Gonçalves 2008; Lopes et al. 2010). However, there are reports that indicate that *S. kudriavzevii* may participate in hybrid formation with wine-related *S. cerevisiae* and *S. bayanus* species. For example, the genome sequence of a widely used wine yeast strain, VIN7, revealed an allotriploid hybrid genome with *S. cerevisiae* and *S. kudriavzevii* origins (Borneman et al. 2012). Physiological characterization of *S. kudriavzevii* strains has shown that they are able to grow at relatively low (10 °C) and high (up to 30 °C) temperatures;

however, they are not able to tolerate more than 5 % of ethanol (Belloch et al. 2008).

Natural hybrids of *S. cerevisiae*, *S. bayanus* and *S. kudriavzevii* conducting wine fermentations have been recently discovered and characterised by genetic approaches (Belloch et al. 2009; Borneman et al. 2012; Dunn and Sherlock 2008; González et al. 2006, 2008; Horinouchi et al. 2010; Masneuf et al. 1998; Nguyen et al. 2000; Sipiczki 2008). The hybridisation process between *Saccharomyces* species has been proposed as an adaptation mechanism of yeasts to ferment at low temperatures (de Barros Lopes et al. 2002; Barrio et al. 2006; Sipiczki 2008). Physiological data suggest that *Saccharomyces* hybrids might have inherited the ability to grow at high temperatures (30–37 °C) and their ethanol tolerance from their *S. cerevisiae* parent and the ability to grow at low temperatures (10–16 °C) from their *S. bayanus* and *S. kudriavzevii* parents. These physiological characteristics point to *Saccharomyces* hybrids as better adapted to meet the winemakers' trends, such as conducting wine fermentation at low temperatures, which may cause wine aroma improvement (Lambrechts and Pretorius 2000; Torija et al. 2003; Llauradó et al. 2002, 2005; Novo et al. 2003).

Oenological characterization of *S. cerevisiae* × *S. kudriavzevii* hybrid strains has demonstrated that the hybrids are well adapted to ferment at low and intermediate temperatures, producing moderate or higher levels of glycerol and less acetic acid with regard to reference strains of *S. cerevisiae* and *S. kudriavzevii* (Gangl et al. 2009; González et al. 2007). Similar comparative studies, which also included *S. uvarum* and a hybrid between *S. cerevisiae* and *S. uvarum*, in wine and cider (Masneuf et al. 1998; Nguyen et al. 2000), indicated that the highest production of glycerol was produced by *S. uvarum*, *S. kudriavzevii* and the *S. cerevisiae* × *S. uvarum* hybrid (Gamero et al. 2013). Regarding aroma formation, one study indicated that hybrids produced the same quantity of aromatic compounds as *S. cerevisiae* at high temperatures, and the same aromatic intensity as *S. kudriavzevii* at low temperatures (González et al. 2007), whereas in another study this trend was only observed in the case of fusel alcohol production (Gamero et al. 2013). In the latter study, *S. cerevisiae* strains yielded the highest aroma amounts at 28 °C were, whereas *S. uvarum* and some hybrids excelled at 12 °C. Altogether, these studies pointed to the fact that aroma formation is highly dependent on both yeast strain and fermentation temperature (Gamero et al. 2013).

## 10.4 Beer Yeasts

In brewing, a distinction is made between ale yeasts (*top fermentation*) and lager yeasts (*bottom fermentation*). Ale yeasts are classified as *S. cerevisiae* and are mostly used for the production of specialty beers where the fermentation temperatures are relatively high (15–25 °C). Lager yeasts are classified as *Saccharomyces pastorianus*, which include *S. carlsbergensis* and *S. monacensis* isolated by EC Hansen in 1908. Lager yeasts are used for the production of pilsner type beers,

fermented at lower temperatures than ale yeasts (6–14 °C). The genomes of lager yeasts are complex as they are aneuploid and consist of a hybrid of mixed genetic lines of the *Saccharomyces* genus (Kodama et al. 2006). DNA/DNA reassociation studies on the type strains of *S. bayanus* (CBS380), *S. carlsbergensis* (CBS1513) and *S. monacensis* (CBS1503) presented *S. bayanus* as one of the contributors to *S. pastorianus* genome (Vaughan-Martini and Kurtzman 1985). Thereafter many reports agreed with this fact (Tamai et al. 2000; Rainieri et al. 2006; Dunn and Sherlock 2008; Nakao et al. 2009). Later, some evidences pointed to *S. bayanus* to be a hybrid between *S. uvarum* and *S. cerevisiae* (Nguyen et al. 2000; Nguyen and Gaillardin 2005), which was recently confirmed (Nguyen et al. 2011) and being one of its parents *S. uvarum* and the other, a new species isolated from Patagonia and named *S. eubayanus* (Libkind et al. 2011). As *S. eubayanus* carries a ‘pure’ or monogenome it is very likely to be the common contributor of *S. bayanus* and *S. pastorianus*. Lager brewing yeast is now recognised by many authors as *S. eubayanus*/*S. cerevisiae* hybrid (Dunn et al. 2012; Piotrowski et al. 2012; Cousseau et al. 2013; Pengelly and Wheals 2012).

## 10.5 Cider Yeasts

Studies on population dynamics in cider have shown that the composition of yeast flora can vary according to climatic conditions, apple varieties, geographic location and the cider-making technology employed (Cabranes et al. 1990; del Campo et al. 2003; Suárez et al. 2007a). First, an oxidative phase carried out by autochthonous non-*Saccharomyces* yeasts with a low fermentation capacity and with the predominance of *Metschnikowia pulcherrima*, *Hanseniaspora uvarum*, *Hanseniaspora valbyensis* and *Candida* yeasts was observed (Michel et al. 1988; Morrissey et al. 2004; Coton et al. 2006; Suárez et al. 2007a). Furthermore, species of the genera *Pichia*, *Torulaspota*, *Rhodotorula*, *Cryptococcus*, *Zygosaccharomyces* and *Brettanomyces/Dekkera* yeasts, originating from apples or the environment have been also related to cider production (Beech 1993; Michel et al. 1988; Morrissey et al. 2004).

Second, strains with a greater tolerance to ethanol (*Saccharomyces* spp.) completed the cider fermentations. In the aforementioned studies on population dynamics in cider, the *Saccharomyces* species found to be present were *S. cerevisiae* and *S. bayanus*. The data indicated that *S. bayanus* was the predominant species at the beginning and the middle fermentation phases of the fermentation process, reaching a percentage of isolation between 33 and 41 %, whereas *S. cerevisiae* took over the process in the final stages of fermentation (Suárez et al. 2007a). A study that was carried out to examine the dynamics and variability of wild *Saccharomyces* spp. (Suárez et al. 2007b) determined that the number of strains observed was higher than those reported for *Saccharomyces* populations in some wine-growing regions in



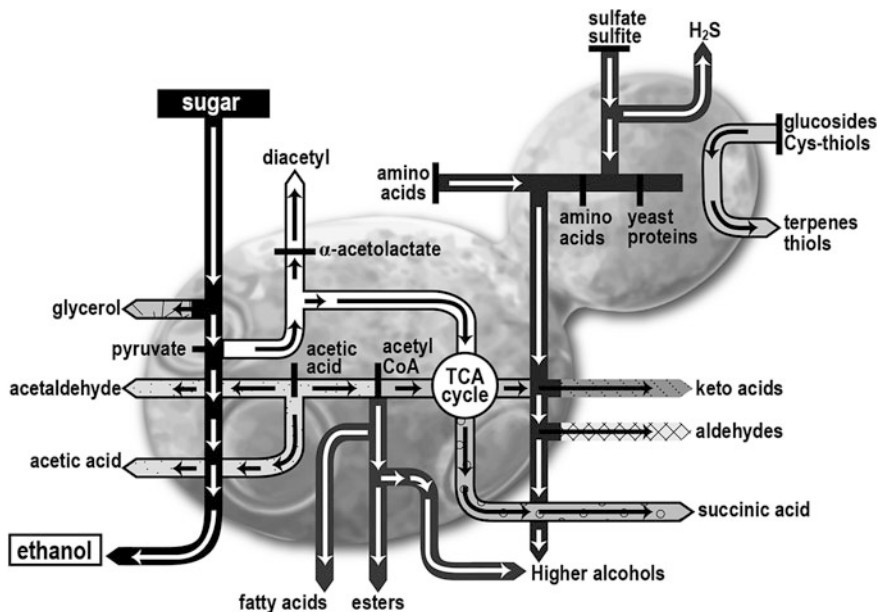


Fig. 10.1 Flavour-related metabolism in yeasts

other studies (Frezier and Dubourdieu 1992; Querol et al. 1994; Gutiérrez et al. 1999; Torija et al. 2001; Schuller et al. 2005).

Finally and as commented before, natural hybrids between *S. bayanus* and *S. cerevisiae* have been described by some authors in wine and cider some years ago (Masneuf et al. 1998; Nguyen et al. 2000).

## 10.6 Fermentative Aroma

The most important compounds within fermentative aroma are higher alcohols, acetate and ethyl esters, aldehydes (acetaldehyde), ketones (diacetyl), organic acids (acetic acid), volatile phenols (4-vinylphenol, 4-vinylguaiacol) and sulphurous compounds (hydrogen sulphide, mercaptans, volatile thiols). A scheme of the synthesis of the main fermentative aroma compounds is shown in Fig. 10.1, while Table 10.1 presents their aroma descriptors and odour thresholds. Finally, Table 10.2 depicts the most important genes involved in flavour-active compound synthesis identified in *S. cerevisiae*.

**Table 10.1** Aroma and odour thresholds of the most representative aroma compounds appearing in alcoholic fermentations

Aromatic compounds	Aroma descriptor	Odour threshold
<i>Higher alcohols (mg/l)</i>		
Methanol	Chemical, medicinal	668 <sup>a</sup>
1-Propanol	Ripe fruit, alcohol	0.830 <sup>a</sup>
2-Methyl-1-propanol (Isobutanol)	Bitter, green, harsh	0.200 <sup>b</sup>
3-Methyl-1-butanol (Isoamyl alcohol)	Alcohol, fusel	30 <sup>b</sup>
3-Methyl-1-pentanol	Vinous, herbaceous, cacao	50 <sup>c</sup>
4-Methyl-1-pentanol	Almond, toasted	50 <sup>c</sup>
1-Butanol	Medicinal, phenolic	150 <sup>a</sup>
2,3-Butanediol	Fruity	150 <sup>a</sup>
1-Pentanol	Almond, synthetic, balsamic	64 <sup>a</sup>
1-Hexanol	Green, grass	8 <sup>b</sup>
(Z)-3-Hexenol	Green, cut grass	0.400 <sup>d</sup>
1-Heptanol	Oily	2.500 <sup>a</sup>
2-Phenylethyl alcohol	Roses, sweet	14 <sup>b</sup>
Benzyl alcohol	Sweet, fruity	200 <sup>a</sup>
<i>Acetate esters (mg/l)</i>		
Ethyl acetate	Fruity, solvent	7.500 <sup>d</sup>
Isoamyl acetate	Banana	0.030 <sup>b</sup>
Hexyl acetate	Green, floral	1.500 <sup>c</sup>
2-Phenylethyl acetate	Rose, flowery	0.250 <sup>b</sup>
<i>Ethyl esters (mg/l)</i>		
Ethyl hexanoate (ethyl caproate)	Green apple, anise	0.014 <sup>b</sup>
Ethyl octanoate (ethyl caprylate)	Sweet, fruity, fresh	0.005 <sup>b</sup>
Ethyl decanoate (ethyl caprate)	Pleasant, soap	0.200 <sup>b</sup>
Ethyl butyrate	Fruity, apple	0.020 <sup>b</sup>
3-Hydroxy ethyl butyrate	Caramel, toasted	20 <sup>a</sup>
Ethyl succinate	Wine	6 <sup>b</sup>
Diethyl succinate	Wine	200 <sup>a</sup>
Ethyl pyruvate	Vegetable, caramel	100 <sup>a</sup>
Ethyl lactate	Acid, medicine	155 <sup>c</sup>
<i>Aldehydes (mg/l)</i>		
Acetaldehyde	Pungent, ripe apple	0.500 <sup>d</sup>
Benzaldehyde	Bitter, cherry	2 <sup>b</sup>
Phenylethanal	Flowery, rose, honey	0.005 <sup>b</sup>
<i>Ketones (mg/l)</i>		
2,3-Butanedione (diacetyl)	Buttery	0.200–2.800 <sup>c</sup>
<i>Organic acids (mg/l)</i>		
Acetic acid	Sour, pungent, vinegar	200 <sup>d</sup>
Propanoic acid	Pungent, rancid, soy	8.100 <sup>c</sup>
Benzoic acid	Chemical	1 <sup>a</sup>
3-Methylbutanoic acid	Cheese, fatty, rancid	0.033 <sup>b</sup>
Butyric acid	Rancid, cheese, sweat	0.173 <sup>a</sup>
Isobutyric acid	Rancid, butter, cheese	2.300 <sup>a</sup>

(continued)

**Table 10.1** (continued)

Aromatic compounds	Aroma descriptor	Odour threshold
Isovaleric acid	Sweet, acid, rancid	0.033 <sup>c</sup>
Hexanoic acid	Fatty acid, cheese	0.420 <sup>b</sup>
Octanoic acid	Fatty acid, rancid	0.500 <sup>b</sup>
Decanoic acid	Fatty, rancid, soap	1 <sup>b</sup>
Phenylacetic acid	Honey, pollen, flowery	2.500 <sup>b</sup>
<i>Volatile phenols (mg/l)</i>		
4-Vinylphenol	Stramonium, almond shell	0.180 <sup>f</sup>
4-Vinylguaiacol	Clove, curry	0.010 <sup>f</sup>
4-Ethyl guaiacol	Phenolic, sweet	0.110 <sup>e</sup>
4-Ethyl phenol	Phenol, spicy	0.440 <sup>f</sup>
2-Methoxyphenol	Medicine, sweet, smoke	0.010 <sup>c</sup>
<i>Sulphur compounds (µg/l)</i>		
Hydrogen sulphide	Rotten egg	10000–80000 <sup>c</sup>
3-(Methylthio)-1-propanol	Cooked vegetable	1000 <sup>d</sup>
Methanethiol (methyl mercaptan)	Cooked cabbage, onion, putrefaction, rubber	0.300 <sup>e</sup>
Ethanethiol (ethyl mercaptan)	Onion, rubber, natural gas	1.100 <sup>e</sup>
Dimethyl sulphide	Asparagus, corn, molasses	25 <sup>e</sup>
Diethyl sulphide	Cooked vegetables, onion, garlic	0.930 <sup>e</sup>
Dimethyl disulfide	Cooked cabbage, intense onion	15, 29 <sup>e</sup>
Diethyl disulfide	Garlic, burnt rubber	4.300 <sup>e</sup>
3-(Methylthio)-1-propanol (methionol)	Cauliflower, cabbage, potato	500 <sup>e</sup>
Benzothiazole	Rubber	50 <sup>e</sup>
Thiazole	Popcorn, peanut	38 <sup>e</sup>
4-Methylthiazole	Green hazelnut	55 <sup>e</sup>
2-Furanmethanethiol	Roasted coffee, burnt rubber	0.001 <sup>e</sup>
Thiophene-2-thiol	Burned, burned rubber, roasted coffee	0.800 <sup>e</sup>
<i>Monoterpenes (µg/l)</i>		
Geraniol	Roses, geranium	30 <sup>d</sup>
Linalool	Floral	15 <sup>d</sup>
α-Terpineol	Pine, lily of the valley	250 <sup>f</sup>
Citronellol	Green lemon	100 <sup>f</sup>
Nerol	Rose, lime	400 <sup>f</sup>
<i>Volatile thiols (ng/l)</i>		
4-Mercapto-4-methylpentan-2-one (4MMP)	Cat urine, box tree/blackcurrant, broom	3 <sup>e</sup>
3-Mercaptohexan-1-ol (3MH)	Passionfruit, grapefruit	60 <sup>e</sup>
3-Mercaptohexyl acetate (3MHA)	Riesling-type note, passionfruit, box tree	4 <sup>e</sup>

<sup>a</sup> Etiévant (1991)<sup>b</sup> Perestrelo et al. (2006)<sup>c</sup> Ferreira et al. (2000)<sup>d</sup> Guth (1997)<sup>e</sup> Swiegers et al. (2005)<sup>f</sup> Vilanova et al. (2010)

**Table 10.2** Most important genes involved in flavour-active compound synthesis

Aromas	Enzymatic activity	Genes identified
Higher alcohols	Branched-chain amino acid transferases	<i>BAT1, BAT2</i>
	Aromatic amino acid transferases	<i>ARO8, ARO9</i>
	Decarboxylases	<i>ARO10, PDC1, PDC5, PDC6, THI3</i>
Esters	Alcohol dehydrogenases	<i>ADH1-7, SFA1</i>
	Alcohol acetyl transferases	<i>ATF1, ATF2</i>
	Acyl transferases	<i>EEB1, EHT1, YMR210 W</i>
	Esterases	<i>EEB1, EHT1, IAH2</i>
Aldehydes	Pyruvate decarboxylases	<i>PDC1-3</i>
Ketones	<i>ILV</i> -encoded enzyme forms and alcohol dehydrogenases	<i>ILV2, ILV3, ILV5, ADH</i>
	Acetaldehyde dehydrogenases	<i>ALD2-6</i>
Volatile phenols	Phenolic acid decarboxylases	<i>PAD1 (=POF1)</i>
Sulphur compounds	Sulphur metabolism related enzymes	<i>CYS4, MET5, MET10, MET14, MET17, MRX1</i>
Monoterpenes	$\beta$ -glucosidases and glucanases	<i>BGL1, BGL2, BEG1, END1, EXG1</i>

### 10.6.1 Higher Alcohols

Higher or fusel alcohols are alcohols with two or more carbon atoms with molecular weights and boiling points higher than those of ethanol (Lambrechts and Pretorius 2000). From a quantitative point of view, higher alcohols are the most important group of volatile compounds produced by yeast during wine fermentation. Higher alcohols are classified in aliphatics like isobutanol, hexanol and isoamyl alcohol and aromatics like 2-phenylethanol and benzyl alcohol. Higher alcohols contribute with an intense aroma to the flavour of wine and other alcoholic beverages. According to Rapp and Versini (1991), concentrations of higher alcohols below 300 mg/l add desirable complexity to wine aroma, whereas higher concentrations (400 mg/l) can be detrimental to wine quality by disguising ester-based fruity aromas and imparting a strong, pungent smell and taste. On the other hand, the concentration of each higher alcohol acting positively or negatively on the aroma is variable. In beer, the flavour of the aliphatic alcohols is distinctly alcoholic (e.g. ethanol) and the aromatic alcohols have a rather sweet, alcoholic or bitter taste (Meilgaard 1975). Nevertheless, in spite of having aroma themselves, the main oenological importance of higher alcohols lies in the fact that they are precursors of acetate esters (Soles et al. 1982).

Many factors affect the levels of higher alcohols in alcoholic beverages. For example, in wine, viticultural conditions, yeast strain and species, initial sugar content of the grape must, pH and composition of the juice, fermentation temperature, assimilable nitrogen and aeration have a strong influence (Fleet and Heard 1993; Houtman et al. 1980a, b; Houtman and du Plessis 1981). In beer, the addition of fatty acids and sterols (Taylor et al. 1979), oxygenation (Quain and Duffield 1985)

or high temperatures (Landaud et al. 2001), cause an increase in higher alcohol content. Those factors also stimulate yeast growth in the fermenting wort. In addition, the production of 2-phenylethyl alcohol appears to be particularly sensitive to temperature, whereas the synthesis of other higher alcohols is relatively unaffected by this factor.

Several studies have demonstrated that *S. bayanus* produces higher amounts of several fusel alcohols (2-phenylethanol, isobutyl alcohol and isoamyl alcohol) than *S. cerevisiae* (Antonelli et al. 1999; Massoutier et al. 1998). Other authors observed that *Saccharomyces* species generally produce higher concentrations of fusels alcohols than non-*Saccharomyces* species (Gil et al. 1996; Herraiz et al. 1990).

Higher alcohols are synthesised by the Ehrlich pathway from branched-chain amino acids, leucine, valine and isoleucine; aromatic amino acids, phenylalanine, tyrosine and tryptophan; and the sulphur-containing amino acid methionine. In this metabolic pathway, the amino acids are transaminated to the corresponding  $\alpha$ -ketoacid, followed by decarboxylation to aldehydes. Finally, these aldehydes are reduced to higher alcohols and NADH becomes NAD<sup>+</sup>. These chemical reactions are carried out by amino acid permeases, transaminases, decarboxylases and dehydrogenases. Amino acid permeases are encoded by the genes *GAP1*, *BAP2*, *BAP3*, *MMP1* and *MUP3* (Didion et al. 1998; Grauslund et al. 1995; Isnard et al. 1996; Jauniaux and Grenson 1990; Mai and Lipp 1994; Rouillon et al. 1999); branched-chain amino acid transaminases by *BAT1* and *BAT2* and aromatic amino acids transaminases by *ARO8* and *ARO9* (Dickinson and Norte 1993; Eden et al. 2001; Hazelwood et al. 2008; Kispal et al. 1996; Lilly et al. 2006b; Ugliano and Henschke 2009). In the valine-degradation pathway, any one of the three isozymes of the pyruvate dehydrogenase complex (PDC), encoded by *PDC1*, *PDC5* and *PDC6*, will decarboxylate  $\alpha$ -ketoisovaleric acid (Dickinson et al. 1998); in isoleucine catabolism, any one of the family of decarboxylases encoded by *PDC1*, *PDC5*, *PDC6*, *KID1* or *ARO10* is sufficient for the decarboxylation reaction (Dickinson et al. 2000); in the leucine-degradation pathway, the major decarboxylase is encoded by *KID1* (Dickinson et al. 1997); in the case of aromatic amino acids, *PDC1*, *PDC5*, *PDC6* or *ARO10* are involved (Dickinson et al. 2003). And finally, ethanol dehydrogenases are codified by *ADH1*, *ADH2*, *ADH3*, *ADH4*, *ADH5*, *ADH6*, *ADH7* and *SFA1* (encoding formaldehyde dehydrogenase) (Delneri et al. 1999; Hazelwood et al. 2008). On the other hand, aryl alcohol dehydrogenases, *AAD10* and *AAD14*, are believed to be responsible for the degradation of aromatic aldehydes into their corresponding higher alcohols (Delneri et al. 1999) and higher alcohols can also be produced de novo through carbohydrate metabolism (Äyräpää 1968, 1971).

Several studies have been carried out in order to understand the complexity of higher alcohol formation and to be able to modulate the aroma of alcoholic beverages by yeasts. Recent screenings based on constructing double- and triple-deletion mutants presented *AAD6*, *BAT2*, *HOM2*, *PAD1*, *PRO2*, *SPE1* and *THI3* as the most important genes affecting higher alcohol production, being *BAT2* the dominant gene in this respect and suggesting that the initial transaminase step of

the Ehrlich pathway is rate-limiting (Styger et al. 2011, 2013). Other studies showed that overexpression of the branched-chain amino acids transaminases *BAT1* or *BAT2* under the control of the constitutive phosphoglycerate kinase I gene (*PGKI*) lead to an increase in the levels of isoamyl alcohol, isoamyl acetate and, to a lesser extent, isobutanol and isobutyric acid or an increase in isobutanol, isobutyric acid and propionic acid, respectively. In both cases, wines presenting higher ‘peach’ and ‘apricot’ notes were obtained (Lilly et al. 2006b). In the case of the wort fermentation, *BAP2* gene was overexpressed under the glyceraldehyde 3-phosphate dehydrogenase promotor (*TDH3*) in a brewer’s yeast (Kodama et al. 2001). As a result, accelerated assimilation rates of branched-chain amino acids resulted in an increased production of isoamyl alcohol derived from leucine, while no increases of isobutyl alcohol derived from valine or of active amyl alcohol derived from isoleucine were observed. These results suggest that the mechanisms for the production of each higher alcohol are, although interconnected, not the same. Finally, a recent study has shown that the synthesis of higher alcohols seems to be influenced by the  $\text{NAD}^+/\text{NADH}$  availability, having the redox balance an important impact (Jain et al. 2012).

### 10.6.2 Acetate Esters

Acetate esters such as ethyl acetate (‘solvent’-like aroma), isoamyl acetate (‘banana’ aroma), ethyl caproate and ethyl caprylate (‘sour apple’ aroma) and 2-phenylethyl acetate (‘flowery’, ‘roses’ and ‘honey’ aromas), give desirable ‘fruity’ and ‘floral’ aromas in the alcoholic beverages (Lambrechts and Pretorius 2000; Swiegers et al. 2005).

The concentration of acetate esters in wines is affected by different factors such as maturity and sugar content (Houtman et al. 1980a, b), yeast species, fermentation temperature (Piendl and Geiger 1980), alcoholic and malolactic fermentation, winemaking method (Herraiz and Ough 1993; Gómez et al. 1994) or the presence of non-soluble material in the must (Edwards et al. 1985). Besides, different factors after the fermentative process, such as time and temperature of ageing and storage, affects ester content in wine (Marais and Pool 1980; Ramey and Ough 1980). Regarding yeast species carrying out the fermentation, acetate ester production depends on each strain (Antonelli et al. 1999; Mateo et al. 1992). Some studies have demonstrated that *S. cerevisiae* produces high amounts of several acetate esters such as isopentyl acetate, phenylethyl acetate, isoamyl acetate, hexyl acetate (Nykänen and Nykänen 1977; Soles et al. 1982; Suomalainen and Lehtonen 1979), whereas *S. bayanus* has demonstrated to be a good 2-phenylethyl acetate producer (Soles et al. 1982). Comparison between *Saccharomyces* and non-*Saccharomyces* regarding acetate esters production showed species dependence in the production of these aromatic compounds (Gil et al. 1996; Lema et al. 1996; Rojas et al. 2001).

Recently, it has been demonstrated that pure and/or mixed cultures of several non-*Saccharomyces* strains are able to increase ester levels in wine: *Hanseniaspora guillermondii* (2-phenyl ethyl acetate) and *H. uvarum* (isoamyl acetate) (Moreira et al. 2008); *Hanseniaspora osmophila*, *H. viniae* and *H. anomala* (2-phenyl ethyl acetate) (Viana et al. 2008, 2009, 2011; Izquierdo-Canas et al. 2011); *Pichia membranifaciens* and *Pichia kluveri* (Viana et al. 2009; Swiegers et al. 2011); *Williopsis saturnus* and *T. delbrueckii* (ethyl and isoamyl acetate) (Erten and Tanguler 2010; Swiegers et al. 2011; Izquierdo-Canas et al. 2011; Tanguler 2012; Azzolini et al. 2012). Some of the strains of the genera *Hanseniaspora* spp., *Torulaspora* spp., *Kluyveromyces* spp., *Pichia* spp., and *Williopsis* spp. have been commercialised.

In lager beers, the only acetate ester that can be sensorially perceived is isoamyl acetate (Dufour and Malcorps 1995). However, the presence of multiple esters can have a synergistic effect, having an impact on the overall flavour (Meilgaard 1975). In addition, it has been demonstrated that small changes in ester concentration can have a significant impact on beer flavour (Hammond 1995). Several fermentation conditions have an important impact on ester formation during brewery fermentations (Verstrepen et al. 2003a): fatty acids (Saerens et al. 2008a), temperature (Saerens et al. 2008a), wort gravity (Saerens et al. 2008b; Pidocke et al. 2009; Lei et al. 2012), pitching rate (Verbelen et al. 2009a) and oxygen (Verbelen et al. 2009b).

Acetate esters are synthesised by a condensation reaction between higher alcohols and acetyl-CoA. This reaction is mediated by acetyltransferases codified by genes *ATF1*, *Lg-ATF1* and *ATF2* (Fujii et al. 1994, 1996; Fujiwara et al. 1999; Lilly et al. 2006a; Saerens et al. 2008b, 2010; Verstrepen et al. 2003c). *ATF1* and *ATF2* are present in both ale and lager strains, but *Lg-ATF1* is found only in lager strains (Yoshimoto et al. 1998). During fermentation, acetate ester production rates are dependent on alcohol acetyltransferases activity (Malcorps et al. 1991). Besides, the effect of esterases encoded by *IAH1* and *TIP1* is also important for the final concentration of acetate esters (Horsted et al. 1998; Lilly et al. 2006a; Saerens et al. 2008b, 2010).

Deletion/overexpression studies indicated that the *ATF2*-encoded enzyme of *S. cerevisiae* plays a minor role as compared with its *ATF1*-encoded enzyme (Lilly et al. 2000, 2006a; Verstrepen et al. 2003b). Additionally, the fact that the double-deletion strain produced considerable amounts of certain esters suggests the existence of additional, as yet unknown, ester synthases in the yeast proteome (Verstrepen et al. 2003b). Interestingly, overexpression of different alleles of *ATF1* and *ATF2* led to different ester-production rates, indicating that differences in the aroma profiles of yeast strains may be partially due to mutations in their *ATF* genes (Verstrepen et al. 2003b).

In addition, it has been recently postulated that the ratio acetyl-CoA/CoA could affect acetate ester synthesis (Cordente et al. 2007). The carnitine acetyltransferases catalyse the reversible reaction between carnitine and acetyl-CoA to form acetylcarnitine and CoA. Overexpression of *CAT2*-encoded mitochondrial and

cytosol carnitine acetyltransferases resulted in lower levels of acetate esters in the fermentation since less acetyl-CoA is available for acetate ester synthesis (Cordente et al. 2007).

### 10.6.3 Ethyl Esters

Ethyl esters such as ethyl propanoate, ethyl butanoate, ethyl hexanoate (ethyl caprylate), ethyl octanoate (ethyl caproate), ethyl decanoate (ethyl caprate) and ethyl lactate give desirable fruity and flowery aroma to the wine. They are produced by condensation between ethanol and acyl-CoA, reaction mediated by acyltransferases. These acyltransferases are encoded by the genes *EHT1* (ethanol hexanoyl transferase 1) and *EEB1* (ethanol hexanoyl transferase) (Rossouw et al. 2008; Saerens et al. 2006, 2008a, 2010), the latter being responsible for the majority of ethyl ester production in *S. cerevisiae* as shown in deletion studies (Saerens et al. 2006). The final concentration of ethyl esters in wine will therefore be influenced by the esterase activity of *EHT1* and *EEB1* encoded-transferases (Saerens et al. 2006), as well as the effect of esterases encoded by *IAH1* and *TIP1* (Horsted et al. 1998; Lilly et al. 2006a; Saerens et al. 2008b, 2010).

Ethyl ester concentrations in alcoholic beverages are affected by the same factors mentioned for acetate esters in the previous section. Regarding yeast species carrying out the fermentation, ester production depends on each strain (Mateo et al. 1992). Several studies have demonstrated that *S. cerevisiae* produced high amounts of several ethyl esters such as ethyl caproate, ethyl caprylate and ethyl caprate (Antonelli et al. 1999; Nykänen and Nykänen 1977; Soles et al. 1982; Suomalainen and Lehtonen 1979), whereas *S. bayanus* has demonstrated to be a good ethyl caprate and ethyl lactate producer (Antonelli et al. 1999; Soles et al. 1982). Comparison between *Saccharomyces* and non-*Saccharomyces* ethyl ester production showed *Saccharomyces* species produced equal or higher ethyl esters amounts (Gil et al. 1996; Lema et al. 1996).

### 10.6.4 Aldehydes

Acetaldehyde is the most important aldehyde present in alcoholic beverages from a quantitative point of view. In beer, acetaldehyde is normally present at close to its flavour threshold (Engan 1981), whereas different levels can be found in wines. The average values are about 80 mg/l for white wine, 30 mg/l for red wine and 300 mg/l for sherries (McCloskey and Mahaney 1981). At low levels, it gives a pleasant, fruity aroma, but at high concentrations it possesses a pungent irritating odour (Miyake and Shibamoto 1993). Excess acetaldehyde produces a 'green', 'grassy' or 'apple-like' off-flavour in beer (Margalith 1981; Adams and Moss 2000), cider (Williams 1974) and wine (Henschke and Jiranek 1993), with the



exception of sherry-type wines, where high acetaldehyde content is a characteristic feature (Sponholz 1993; Cortes et al. 1998).

Acetaldehyde, also called ethanal, is an intermediary of alcoholic fermentation obtained by the decarboxylation of pyruvate. Pyruvate decarboxylase enzymes encoded by *PDC1*, *PDC2* and *PDC3* participate in this process. Later on, acetaldehyde is reduced to ethanol by alcohol dehydrogenase enzymes, primarily the enzyme encoded by the *ADH1* gene (Pronk et al. 1996), although a little quantity always remains in the wine. The conversion of acetaldehyde to ethanol is required for the maintenance of the redox balance of the cell, since it re-oxidises NADH to NAD<sup>+</sup>, which will be available for glycolysis. In this way, sugar is the primary substrate for acetaldehyde formation, but metabolism of amino acids such as alanine also contributes to the synthesis of this compound (Henschke and Jiranek 1993; Boulton et al. 1998).

In alcoholic beverages, acetaldehyde is mainly produced in the first stages of fermentation and its concentration drops at the end of the fermentation and during maturation due to yeast activity. The levels of acetaldehyde in vinification can be considerably affected, from 0.5 to 286 mg/l, depending on the yeast strain (Liu and Pilone 2000), but other factors can affect acetaldehyde level in wines, such as low quantity of zinc, presence of oxygen late in the fermentation, the nature of insoluble material used to clarify the must, increasing fermentation temperature or the excessive use of SO<sub>2</sub> in grape must (Delfini and Costa 1993; Romano et al. 1994; Liu and Pilone 2000). Excessive acetaldehyde levels contribute to a perception of oxidation, although in some Jerez wines such as Fino and Manzanilla high concentrations of this compound are desirable (Zamora 2009). In the case of the beer, high acetaldehyde concentrations reflect premature flocculation or a decrease in yeast viability.

### 10.6.5 Ketones

Vicinal diketones appear normally in beer fermentation and are undesirable compounds affecting lager beer flavour (Inoue 1992; Wainwright 1973), whereas it is a characteristic flavour of some ale beers. The two most important vicinal diketones are diacetyl (2,3-butanedione) and 2,3-pentanedione. Diacetyl confers a 'butterscotch'-like aroma and pentanedione, a 'honey'-like aroma. In the case of the wine, diacetyl can contribute to wine aroma complexity in low concentrations, giving 'nutty' or 'toasty' nuances, but it becomes undesirable at levels between 1 and 4 mg/l (Sponholz 1993).

Diacetyl is synthesised from  $\alpha$ -acetolactate, an intermediate in the valine and leucine biosynthesis pathway, by spontaneous oxidative decarboxylation. Yeasts are also able to reduce diacetyl to acetoin, which may then be further reduced to 2,3-butanediol. Acetoin has a much higher flavour threshold (50 mg/l) than diacetyl, exhibits 'fruity', 'mouldy' and 'woody' flavours (Meilgaard 1975) and does not cause any off-flavours in the beer. The production of diacetyl in beer is

increased by low pH, high temperature, oxygen and the presence of metal ions (Haukeli and Lie 1978) and can be regulated in wort by nitrogen content, valine addition (Krogerus and Gibson 2013) and the enzyme  $\alpha$ -acetolactate decarboxylase (Godtfredsen and Ottesen 1982).

The *ILV*-encoded enzyme forms  $\alpha$ -acetolactate from pyruvate. This enzyme is subject to general amino acid control and very strong feedback inhibition by valine. In the case of lager yeasts, 90–95 % of the diacetyl reductase activity is accounted for alcohol dehydrogenases (Bamforth and Kanauchi 2004), whereas in the case of ale yeasts, enzymes other than alcohol dehydrogenases appear to be more important, these enzymes being only responsible for the 60 % of the reductase activity.

There have been several attempts to try to reduce diacetyl formation in brewing, such as the disruption of the gene *ILV2*. Mutants lacking this gene did not produce diacetyl but because of their inability to synthesise valine and leucine, such yeasts fermented poorly (Ryder and Masschelein 1983). Changing the upstream regulatory sequence of *ILV2* could reduce the level of this enzyme rather than to eliminate it completely (Petersen et al. 1983). An alternative approach was to increase the flux to amino acid synthesis, which has been achieved transforming yeasts with multiple copies of the *ILV5* gene (Villanueva et al. 1990; Goossens et al. 1991). Conversely, transformations with the *ILV3* gene had no effect on diacetyl concentration (Goossens et al. 1987).

### 10.6.6 Organic Acids

Acetic acid is the main responsible for volatile acidity of wines. Other contributors to volatile acidity are propionic acid and hexanoic acid. The optimal concentration in wine is 0.2–0.7 g/l (Corison et al. 1979; Dubois 1983). At high concentrations (0.7–1.1 g/l), acetic acid imparts a ‘vinegar’ flavour to the wine. *S. cerevisiae* wine strains can produce from 100 mg/l to 2 g/l of acetic acid depending on the conditions during fermentation and the type of strain (Radler 1993). *S. bayanus* and *S. uvarum* usually produce less acetic acid than *S. cerevisiae* (Giudici et al. 1995; Eglinton et al. 2000). Furthermore, certain strains of *T. delbrueckii* have been shown to reduce acetic acid production in wine (Bely et al. 2008; Van Breda et al. 2013).

Acetate is produced through acetaldehyde oxidation in a reaction catalysed by acetaldehyde dehydrogenases encoded by *ALD4* and *ALD5* (mitochondrial isoforms) and *ALD6*, *ALD2* and *ALD3* (cytosolic isoforms) (Navarro-Aviño et al. 1999). During winemaking, Ald6p, Ald5p and Ald4p are the main enzymes responsible for acetate formation (Saint-Prix et al. 2004). Deletion of both alleles of *ALD6* in a wine yeast caused a 2-fold reduction in the amount of acetate produced during fermentation, but as a consequence of the redox imbalance

generated, glycerol, succinate and 2,3-butanediol production was slightly increased (Remize et al. 2000).

On the other hand, mutations in the stress response gene *YAPI* (Cordente et al. 2013) or use of non-*Saccharomyces* yeasts such as *T. delbrueckii* (Bely et al. 2008; Van Breda et al. 2013) constitute successful examples of non-GMO approaches to decrease acetic acid formation during fermentation.

In high-gravity brewing, yeast cells are stressed because of high sugar and ethanol concentrations, which can lead to higher production of acetic acid, which can be a problem to beer quality (Mizuno et al. 2003). An alternative to solve this problem could be to employ a mutant overexpressing *ALD4* (Mizuno et al. 2006), which produced half the amount of acetic acid and 1.1 % more ethanol than beer brewed using the wild-type.

### 10.6.7 Volatile Phenols

Volatile phenols can appear in wine as a consequence of a non-oxidative decarboxylation of hydroxycinnamic acids *p*-coumaric and ferulic carried out by yeasts (Chatonnet et al. 1993; Grando et al. 1993) or through decarboxylation of phenolic acids, usually first into 4-vinyl derivatives that are then reduced to 4-ethyl derivatives through enzymes called phenolic acid decarboxylases (Cavin et al. 1993). The genes encoding phenolic acid decarboxylases include *PADI* (also known as *POF1*). However, phenolic acid decarboxylase activity is very low in most *S. cerevisiae* strains (Barthelmebs et al. 2000a, b) and several attempts have been carried out to develop mutant strains to modulate volatile phenol production. Strains overexpressing the *Bacillus subtilis* phenolic acid decarboxylase gene (*padc*), the *Lactobacillus plantarum* *p*-coumaric acid decarboxylase gene (*pdc*) and strains in which *PADI/POF1* gene was disrupted, are examples of successful volatile phenol modulation (Smit et al. 2003). Contrarily, constructed strains overexpressing *S. cerevisiae* phenylacrylic acid decarboxylase gene (*PADI/POF1*) has no significant effect in volatile phenol synthesis (Smit et al. 2003).

Volatile phenols possess low sensory thresholds and, in spite of the fact that they can be desirable in certain wines, normally they appear as off-flavours ('stable', 'barnyard', 'pharmaceutical') (Dubois 1983). Ethyl phenols (4-ethyl guaiacol and 4-ethyl phenol) present a special negative contribution and are derived from the reduction of vinyl phenols (4-vinyl guaiacol and 4-vinyl phenol). Vinyl reductase activity is typically associated with *Brettanomyces* and *Dekkera* spp.

On the other hand, volatile phenols can contribute positively or negatively depending on the beer product. The presence of excessive amounts of vinyl phenols is considered undesirable in bottom-fermented pilsners. Hence the term 'phenolic off-flavour' (POF) is attributed to beers with a strong aroma described as 'pharmaceutical', 'medicinal', 'solvent', 'spicy', 'clove-like', 'smokey' or 'barbeque'. However, these compounds are crucial for the characteristic aroma of Belgian white beers (made with unmalted wheat), German rauch beers and Weizen

beers (made with malted wheat) and in many top-fermented blond and dark specialty beers.

### 10.6.8 Sulphur Compounds

Hydrogen sulphide imparts a 'rotten egg' aroma and has a very low odour threshold of 10–80 µg/l (Swiegers et al. 2005). The concentration of H<sub>2</sub>S produced during wine fermentation depends on the presence of sulphur compounds, wine yeast strain, fermentation conditions, and the nutritional status of the grape juice (Henschke and Jiranek 1991; Rauhut 1993; Spiropoulos and Bisson 2000). However, some strains produce H<sub>2</sub>S constitutively without being affected by environmental conditions (Jiranek et al. 1995; Spiropoulos and Bisson 2000; Mendes-Ferreira et al. 2002).

During wine fermentation, yeast can synthesise hydrogen sulphide from either inorganic sulphur compounds (sulphate and sulfite) or from organic sulphur compounds (cysteine and glutathione) (Henschke and Jiranek 1993; Rauhut 1993; Hallinan et al. 1999; Spiropoulos and Bisson 2000).

The sulphate reduction sequence (SRS) is activated in response to the necessity to produce cysteine and methionine, usually insufficient in wine must (Henschke and Jiranek 1993). The first step involves the transportation of sulphate from the medium into the yeast cell by sulphate permease. Several steps follow to reduce sulphate to sulphide using the enzymes ATP-sulfurylase and sulfite reductase. Subsequently, *O*-acetylserine (from the amino acid serine) combines with sulphide to form cysteine, and *O*-acetylhomoserine (from the amino acid aspartate) combines with sulphide to form homocysteine, which can then be converted to methionine (Thornton and Bunker 1989; Yamagata 1989; Henschke and Jiranek 1993; Rauhut 1993; Jiranek et al. 1995; Spiropoulos and Bisson 2000). Nitrogen limitation leads to insufficient of these precursors and sulphide is accumulated and released to the medium as hydrogen sulphide (Henschke and Jiranek 1993; Rauhut 1993; Jiranek et al. 1995; Spiropoulos and Bisson 2000). Additionally, significant amounts of H<sub>2</sub>S can be produced when the fermentation medium is rich in sulphite since it can diffuse into the cell.

Several attempts have been made to modulate H<sub>2</sub>S production by using certain wine and brewing yeasts that are commercially available. The consequences of overexpression of the *MET17* gene, which encodes *O*-acetylserine and *O*-acetylhomoserine sulfhydrylase in *S. cerevisiae*, seemed to be strain dependent (Omura et al. 1995; Spiropoulos and Bisson 2000). Conversely, the deletion of the *MET14* gene (encoding an adenosylphosphosulfate kinase) or the *MRX1* gene (encoding a methionine sulfoxide reductase), might be the most effective way to prevent wine yeast from producing H<sub>2</sub>S in fermentations (Pretorius 2000, 2003, 2004; Pretorius and Høj 2005). Another attempt to prevent H<sub>2</sub>S formation was carried out through modifying the activity of the sulfite reductase enzyme by engineering one of the enzyme subunits codified by *MET10* (Sutherland et al. 2003).

This strategy has been successfully applied in beer (Hansen and Kielland-Brandt 1996). On the other hand, classical mutagenesis which lead to mutants presenting mutations in *MET5* and *MET10* genes produced 50–99 % less H<sub>2</sub>S than the parental strain (Cordente et al. 2009). Some of these wine strains are now in commercial use. Finally, increased expression of *CYS4* in brewing yeast, encoding the cystathionine  $\beta$ -synthase, has been shown to suppress the formation of H<sub>2</sub>S (Tezuka et al. 1992).

Another sulphur compound that can be detrimental for the flavour of alcoholic beverages is ethanethiol ('onion' aroma), synthesised through the reaction of hydrogen sulphide and ethanol or acetaldehyde (Rauhut 1993). On the other hand, dimethyl sulphide (DMS), which present 'asparagus', 'corn' and 'molasses' notes, might be produced in wine via cleavage of *S*-methyl-L-methionine to homoserine and DMS. In beer production, heat decomposition during malting of *S*-methyl-methionine produces dimethyl sulfoxide (DMSO), which can be reduced to DMS, during storage (Rauhut 1993) or fermentation by yeasts. In *S. cerevisiae*, the *MXR1* gene has been shown to encode a methionine sulfoxide reductase and its disruption prevents DMS production (Hansen 1999). Finally, DMS formation during fermentation has also been linked to cysteine, cystine or glutathione metabolism in yeast (Rauhut 1993; Ribéreau-Gayon et al. 2000).

## 10.7 Yeasts and Its Role in the Development of Varietal Aroma in Wine

In addition to the aroma-active compounds synthesised by yeasts during alcoholic fermentations, some yeasts play a relevant role in the development of the primary or varietal aroma of wines (Gamero et al. 2011a, b). Wine's primary aroma consists of lactones, benzenes, volatile phenols, vanillins, norisoprenoids, terpenes and some polyfunctional mercaptans present at low concentrations in the ng/l– $\mu$ g/l range (Loscos et al. 2007; Mateo-Vivaracho et al. 2010; Tominaga et al. 1998b). Most of these aromas appear in grapes as odourless precursors (glycosides, polyhydroxylated molecules or cysteinyl-derivatives). It has been demonstrated that some yeasts are able to release those aroma compounds by cleavage of the precursor molecules or are even able to synthesise new aroma molecules similar to the ones present in grapes (Darriet et al. 1988; Delcroix et al. 1994; Delfini et al. 2001; Fernández-González et al. 2003; Fernández-González and Di Stefano 2004; Gamero et al. 2011a, b; Hernández et al. 2003; Hernández-Orte et al. 2008; Loscos et al. 2007; Mateo and Di Stefano 1997; Spagna et al. 2002; Ugliano et al. 2006; Ugliano and Moio 2008). In this way, yeast can enhance wine varietal aroma. For instance, *Saccharomyces* species and hybrids are able to release and synthesise de novo vanillins, terpenes, lipid derivatives, volatile phenols and norisoprenoids (Gamero et al. 2011a, b).

In certain wines, varietal aroma compounds play a crucial role. This is the case of some polyfunctional mercaptans in certain white wines (Tominaga et al. 1998b; Mateo-Vivaracho et al. 2010), of linalool and other terpenols in Muscat wines

(Ribéreau-Gayon et al. 2000) or of *cis*-rose oxide in Gewürztraminer (Guth 1997). On the other hand, in most wines, varietal aroma is formed by combinations of many grape- and yeast-derived compounds, none of which play a predominant aroma role, and it is the overall aroma profile the responsible for varietal and origin related difference (Escudero et al. 2007; Loscos et al. 2007, 2010).

### 10.7.1 Monoterpenes

Among the most important key odorants in the so-called ‘aromatic’ grape varieties (e.g. Muscat) are monoterpenes such as linalool, geraniol, nerol, citronellol and  $\alpha$ -terpineol. (Gunata et al. 1985; Loscos et al. 2007; Maicas and Mateo 2005; Strauss et al. 1986; Ugliano and Henschke 2009). The common precursor of all the monoterpenoids is isopentyl pyrophosphate.

During must fermentation the grape-derived glycosidic precursors are hydrolysed by the action of glycosidases and the aromatic volatile compounds released into the wine. Among the most important glycosidases are  $\beta$ -glucosidases,  $\alpha$ -L-arabinofuranosidases,  $\alpha$ -L-rhamnosidases and  $\beta$ -D-xylosidases (Maicas and Mateo 2005; Van Rensburg and Pretorius 2000; Sarry and Gunata 2004). Several research groups the world over have investigated various wine-related *Saccharomyces* and non-*Saccharomyces* (*Brettanomyces/Dekkera*, *Candida*, *Debaryomyces*, *Hanseniaspora* and *Pichia*) yeasts for their ability to produce suitable glycosidases and other enzymes that can release varietal aromas (Charoenchai et al. 1997; Esteve-Zarzoso et al. 1998; Fernández et al. 2000; Fleet 2008; McMahan et al. 1999; Strauss et al. 2001; Ugliano et al. 2006; Zoecklein et al. 1997).

In addition to the search for yeasts that naturally produce aroma-enhancing enzymes, several mutants have been constructed with the aim of enhancing monoterpene-based varietal flavours during wine fermentation. These mutants include a yeast expressing the  $\beta$ -1,4-glucanase gene from *Trichoderma longibratum* (Villanueva et al. 2000) and a wine yeast expressing the *BGL1* and *BGL2*  $\beta$ -glucosidase genes of *Saccharomycopsis fibuligera*, the *ABF2*  $\alpha$ -L-arabinofuranosidase gene of *Aspergillus niger* and a glucanase-encoding gene cassette consisting of several glucanase genes (*BEG1*, *END1* and *EXG1*) (Pretorius 2000, 2003, 2004; Van Rensburg and Pretorius 2000; Pretorius and Bauer 2002; de Barros Lopes et al. 2006).

In addition to studies focused on the release of monoterpenes by yeasts, there were also several studies undertaken concerning the biotransformation of terpenes by *Saccharomyces* species and hybrids, such as the reduction of geraniol to citronellol, translocation of geraniol to linalool, isomerisation of nerol to geraniol and cyclizations of linalool to  $\alpha$ -terpineol (Gamero et al. 2011b; Gramatica et al. 1982; King and Dickinson 2000; Zea et al. 1995; Zoecklein et al. 1997).

In the case of brewing, it has also recently been shown that different hop varieties have different concentrations of monoterpenoids (Takoi et al. 2010).

### 10.7.2 Volatile Thiols

Volatile thiols are sulphur compounds that can appear in wines in very low concentrations, but they can have a profound impact on the aroma of certain wine varieties, such as Sauvignon Blanc, Colombard, Riesling, Semillon, Merlot and Cabernet Sauvignon, since they present very low sensory thresholds (ng/l level) (Tominaga et al. 1995, 1998a, b; Murat et al. 2001b). These compounds are responsible for the 'fruity' or 'tropical' organoleptic flavours. Some examples of volatile thiols are 4-mercapto-4-methylpentan-2-one (4MMP), reminiscent of 'box tree', 'passion-fruit', 'broom' and 'black current' bud; 3-mercaptohexan-1-ol (3MH) and 3-mercaptohexyl acetate (3MHA), responsible for 'passion-fruit', 'grapefruit' and 'citrus' aromas; 4-mercapto-4-methylpentan-2-ol (4MMPOH) that can also contribute to the characters of 'citrus', 'passion-fruit' and 'grapefruit', although its organoleptic role is more limited, due to its concentration in wines seldom exceeding its olfactory threshold of 55 ng/l and 2-furfurylthiol, which can contribute roast coffee aroma to the bouquet of wines aged in oak barrels (Darriet et al. 1995; Tominaga et al. 1996, 1998b, 2000; Tominaga and Dubourdieu 2006).

Most of the thiols that are present in grapes appear as non-volatile, cysteine-bound conjugates and can be released by the action of carbon-sulphur lyases of certain yeasts (Darriet et al. 1995; Tominaga et al. 1995). Deletion and overexpression of the genes encoding these enzymes resulted in a decrease and increase, respectively, in the levels of the corresponding thiols (Howell et al. 2005; Swiegers et al. 2007). The release of thiols occurs during fermentation in a low percentage since it has been detected that only a small fraction of cysteine-bound conjugates (1.6–3.2 %) is released as 3MH (Dubourdieu et al. 2006; Murat et al. 2001b). The efficiency of thiol release is strain dependent (Dubourdieu et al. 2006; Howell et al. 2004); however, some studies reported that *S. bayanus* and *S. bayanus*/*S. cerevisiae* hybrid strains have stronger abilities than *S. cerevisiae* in this sense (Murat et al. 2001a; Swiegers et al. 2006a). In addition, certain non-*Saccharomyces* yeasts can have a significant impact on volatile thiol concentration. Co-inoculation of *Pichia kluyveri*, isolated from a spontaneous fermentation of Chardonnay must, with specific commercial wine strains of *S. cerevisiae* resulted in an increase of the concentration of 3MHA in Sauvignon Blanc wines (Anfang et al. 2009). Recent work also showed that some strains of *Metschnikowia pulcherrima*, *T. delbrueckii* and *K. thermotolerans* have relatively high capacities to release 3MH (Zott et al. 2011).

Thiols can also be synthesised by yeasts. For instance, it has been proposed that cysteine desulphydrase enzyme catalyses the formation of furfurylthiol from furfural (Tominaga et al. 2000) and the formation of H<sub>2</sub>S enhances this process. In this way, the production of furfurylthiol is linked to the production of the HS<sup>-</sup> anion, which is not produced when sufficient ammonium sulphate is present in the medium (Tominaga et al. 2000). During fermentation, 3MH can be converted to 3MHA by the action of alcohol acetyltransferase, encoded by the *ATF1* gene (Swiegers et al. 2006b). There is significant variation in the conversion rates



present by the different yeast strains, which is not correlated with the ability to release 4MMP (Swiegers et al. 2006b).

In addition to the specific yeast strain conducting the fermentation, temperature is also a relevant factor in determining volatile thiol concentration. Concentrations of 4MMP, 3MH and 3MHA were higher when the alcoholic fermentation was conducted at 20 °C compared to 13 °C or a 18 °C compared to 23 °C and 28 °C (Masneuf-Pomarède et al. 2006; Swiegers et al. 2006a). So around 18–20 °C seems to be the optimum.

Regarding beer, there is not much knowledge about the occurrence of volatile thiols. However, Vermeulen et al. (2006) detected more than ten of these compounds in fresh lager beer. Thiols do not appear in wort. The most powerful thiol in beer is 3-methyl-2-buten-1-thiol, and this thiol together with 2-mercapto-3-methylbutanol and 3-mercapto-3-methylbutanol are thought to be derived from hop allylic alcohols (Vermeulen et al. 2006). On the other hand, it is hypothesised that the origin of 2-mercaptoethanol and 3-mercaptoopropanol and their corresponding acetates could be Ehrlich degradation of sulphur amino acids, whereas 2-methyl-3-furanthiol could be produced through Maillard reactions (Vermeulen et al. 2006).

## 10.8 Concluding Remarks

The aroma profile of wine, beer and cider is a defining component of the value proposition to consumers. Producers are therefore keen to understand what the optimal ‘absolute’ and ‘relative’ concentrations of the most important aroma-active compounds are and how they can adapt their practices to gain control over the composition of their products. It is widely accepted that one way to adjust the aroma profile of certain styles of fermented beverages is choice of yeast strain(s) with which the fermentation is conducted. However, further research is required into the range of ‘aroma phenotypes’ that wine yeast exhibit, and how this knowledge can be applied to develop novel aroma-enhancing yeast strains or combinations of yeast strains or mixtures of different yeast species.

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