

Chapter 4

Phosphate-Solubilizing Microorganisms

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4.1 Introduction

Phosphorus (P) is second major nutrient in crop productivity for its involvement in many essential processes such as cell division, development, photosynthesis, breakdown of sugar, nutrient transport within the plant, transfer of genetic characteristics from one generation to another and regulation of metabolic pathways (Tandon 1987; Armstrong 1988; Theodorou and Plaxton 1993). The maintenance of high level of soil phosphorus has been a major challenge to agricultural scientists, ecologists and farm managers because in most of the soils, phosphate is present in unavailable form due to complex formation with Ca^{2+} , Al^{3+} , Fe^{2+} or Mn^{2+} depending on soil pH and organic matter. The main problem of phosphorus in soil is its rapid fixation and the efficiency of P solubilization rarely exceeding 10–20%. The fixed forms of P in acidic soils are aluminium and iron phosphates while in neutral to alkaline soils as calcium phosphates. The manufacture of phosphatic fertilizers requires high-grade rock phosphate (RP) and sulphur which are getting depleted progressively and becoming costlier. The total world reserves of RP are estimated to be around 2,700 billion tons of which 80% are located in the USA, Russia, and Morocco. In India, RP deposits are estimated to be about 145 million tons but bulk of this is of poor quality and is unsuitable for manufacture of phosphatic fertilizers. Only 25% of the total P requirement is met through indigenous sources; hence, about 1.5 million tons of high grade RP is imported annually.

The concentration of total P in soil ranges from 0.01 to 0.2%, with an average approximately 0.05% (Barker 1984). But out of this only 20% of total P is available

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to plants. It makes about 0.2% of plant dry weight (Sahachman et al. 1998), which indicates the importance of P availability in soil. Because of extreme reactivity and fixation of P in soil, it is unavailable for plant growth. Moreover, the applied phosphorus also gets fixed chemically with metal ions in the soil (Bagyaraj and Varma 1995; Holford 1997). Even if the total P is high, there is always a need to apply phosphatic fertilizers regularly, part of which gets fixed in soil and is unavailable to plants depending on soil pH and climatic conditions. Although various chemical transformations help in release of P immobilization, chemical fixation depletes soil of the available P. Therefore, it is important to develop technology for P solubilization for plant growth. Level of soluble P in soil can be increased either by using phosphate-solubilizing microorganisms (PSM) as bio-inoculants for solubilization of fixed soil P which can improve crop yields or application of phosphorus (P) rich compost. The use of PSM as bio-inoculants plays a vital role in maintaining soil nutrient status, structure and sustains the production base. It also reduces reliance on expensive imported phosphate; thereby a significant increase in yield was subsequently reported with inoculation of PSM in different crops.

Uptake of phosphorus by the plants is in the form of soluble orthophosphorus. In general, the availability of these ions to the plants is in the order of $\text{H}_2\text{PO}_4^{-1} > \text{H}_2\text{PO}_4^{-2} > \text{PO}_4^{-3}$. The availability also depends mainly on soil pH (Nath and Borah 1983). The soil microorganisms that are of diverse type (fungi, bacteria and actinomycetes) are responsible for solubilization of inorganic phosphates (Kapoor et al. 1989; Kucey et al. 1989; Holvorson et al. 1990; Illmer and Schinner 1992; Kapoor 1995) causing changes in pH of the soil microenvironment and by producing chelating substances that lead to the solubilization of inorganic phosphates. Iron (Fe) and aluminium (Al) at low pH and calcium (Ca) at high pH fix the available form of P into insoluble forms in soil (Dala 1973; Sanyal and De Datta 1991; Johnson and Loepper 2006; Rengel and Marschner 2005) which are not easily available.

4.2 Phosphate-Solubilizing Microorganisms

A large number of autotrophic and heterotrophic soil microorganisms have capacity to solubilize mineral phosphates. PSM are present in almost all the soils, although their number varies depending upon the soil and climatic conditions (Kucey et al. 1989). First time in 1903, tricalcium phosphate (TCP) solubilization was demonstrated by soil bacteria in liquid and in solid medium (Stalstrom 1903; Sackett et al. 1908). However, the extent of solubilization varies with the source of inorganic P and the microorganisms involved (Banik and Dey 1982). Phosphate-solubilizing fungi (PSF) and bacteria are known as effective organisms for phosphate solubilization (Reyes et al. 1999). Recent report suggests that PSM can be used in union with RP so that phosphorus in the RP can be made available in the soil for plant uptake (Jisha and Alagawadi 1996). Thus, PSM cause the release of nutrients into

soil in naturally balanced proportion and exerts beneficial effects on plant development (Glick 1995).

The use of PSM as bio-inoculants plays a vital role in maintaining soil nutrient status and structure as it reduces reliance on expensive imported phosphatic fertilizers. Application of these bacteria along with RP resulted in increased availability of inorganic phosphate for plant utilization (Hebbara and Devi 1990; Rachewad et al. 1992; Jisha and Alagawadi 1996; Chen et al. 2006). Thus, the beneficial effect of inoculation on the availability of P to crops has led to the development of inoculum which is popularly known as phosphobacterin. The first evidence to show that inoculation of seedling with PSM increase uptake of P and crop yield was by Gerretsen (1948) on oat crop. Sundara Rao and Paul (1959) reported significant increase in the yield of barseem after inoculation with phosphobacterin. Since then, beneficial effects of inoculation with different PSM have been reported with different crops.

4.2.1 Major Groups of PSM

An extensive range of soil bacteria, actinomycetes, cyanobacteria and fungi belonging to the genera *Pseudomonas*, *Enterobacter*, *Bacillus*, *Rhizobium*, *Agrobacterium*, *Micrococcus*, *Aereobacter*, *Erwinia*, *Streptomyces*, *Nocardia*, *Aspergillus*, *Penicillium*, *Trichoderma*, *Anabaena*, *Nostoc*, *Calothrix* and *Scytonema* that are able to solubilize various forms of precipitated P have been reported (Kucey et al. 1989; Roychoudhary and Kaushik 1989; Rodriguez and Fraga 1999; Whitelaw 2000; Tran Thi Ngoc Son et al. 2006; Gulati et al. 2008; Sulbarán et al. 2009) and are generally considered to contribute a significant component of the total soil phosphatase activity (Richardson 1994).

In soil, phosphate-solubilizing bacteria (PSB) constitute 1–50% and fungi 0.5–0.1% of the total respective population (Banik and Dey 1982; Kucey 1983; Kucey et al. 1989; Chen et al. 2006). In general, fungal isolates exhibit greater P-solubilizing ability than bacteria in both liquid and solid media (Banik and Dey 1982; Gaur et al. 1973; Kucey 1983; Venkateswarlu et al. 1984). Fungi in soils are able to penetrate deep into soil more easily than bacteria, and hence may be more important to P solubilization in soils (Kucey 1983).

In addition, actinomycetes *Micromonospora*, *Nocardia* and *Streptomyces* have also been reported to solubilize phosphates. Yeast such as *Torula* sp. which is usually not present in soils has also been isolated from compost and characterized for solubilization of TCP and RP (Singh et al. 1980).

4.2.2 Phosphate-Solubilizing Bacteria

An extensive range of soil bacteria including actinomycetes both aerobic and anaerobic are able to solubilize various forms of insoluble inorganic phosphate

compounds, such as TCP, dicalcium phosphate, hydroxyapatite and RP (Goldstein 1986; Rodriguez et al. 2000; Gulati et al. 2008; Sulbarán et al. 2009). The prominent genera involved in mineral phosphate solubilization are *Pseudomonas*, *Bacillus*, *Rhizobium*, *Burkholderia*, *Achromobacter*, *Agrobacterium*, *Micrococcus*, *Aerobacter*, *Flavobacterium*, *Erwinia*, *Alcaligenes*, *Escherichia*, *Serratia* and *Xanthomonas* sp. (Li 1981; Datta et al. 1982; Venkateswarlu et al. 1984; Fernandez et al. 1984; Thomas and Shantaram 1986; Surange and Kumar 1993; Farhat et al. 2009).

The *Pseudomonas* sp. includes *P. striata*, *P. aeruginosa*, *P. putida*, *P. vermiculosa* and *P. fluorescens* (Sen and Paul 1957; Gaiind and Gaur 1990a, b; Kumar et al. 2002; Buch et al. 2008). Population of PSB mainly depends on different soil properties such as physical, chemical properties, organic matter and P content (Kim et al. 1998).

Major *Bacillus* sp. known for their mineral phosphate-solubilizing abilities are *B. polymyxa*, *B. subtilis*, *B. brevis*, *B. circulans*, *B. megaterium*, *B. mesentericus*, *B. mycoides*, *B. pulvifaciens*, etc. (Sen and Paul 1958; Sundara Rao and Sinha 1963; Paul and Sundara Rao 1971; Rodriguez and Fraga 1999; Swain and Ray 2009).

Most PSB were isolated from the rhizosphere of various plants and are known to be metabolically more active than those isolated from sources other than rhizosphere (Baya et al. 1981; Katznelson and Bose 1959; Vazquez et al. 2000). The P-solubilizing ability in bacteria is generally reduced or lost upon repeated subculturing but no such loss has been observed in the case of PSF (Kucey 1983; Sperber 1958).

P-solubilizing ability of PSB is affected by many physiological factors some of which are C and N source in medium, mineral source, pH, incubation temperature, aeration and incubation period (Gaiind and Gaur 1990a, b; Vassilev and Vassileva 2003; Taiwo and Ogundiya 2008). Singh (1992) showed that PSB could solubilize 240 µg/ml and 5,300 µg/ml of TCP and mussoorie rock phosphate (MRP), respectively, after 9 weeks of incubation at 30°C under stationary conditions. Dave and Patel (1999) showed that *Pseudomonas* sp. release 27–163 mg of P₂O₅ at 37°C in 3 weeks under static conditions at cell concentrations of 6.6×10^7 /ml in 100 ml of Pikovskaya's (PVK) medium containing TCP equivalent to 225 mg P₂O₅ (Dave and Patel 1999). Strains of *Pseudomonas* sp. are capable of releasing 162 µg/ml inorganic phosphate in the medium containing TCP (Santhi 1998). Strains of *Enterobacter* can release inorganic phosphate ranging from 83 to 551 µg/ml in medium containing hydroxyapatite (Kim et al. 1997). Kumar and his coworker (1999) observed that strains of *Acetobacter* released 142–431 µg/ml of inorganic phosphate from TCP. Illmer and Schinner (1992) reported that temperature 25–30°C is optimum for P solubilization for *Pseudomonas* sp. whereas fungi solubilize more P at a slightly higher temperature, i.e. 30–35°C (Gaur 1990). *B. polymyxa* was found to have a higher temperature range of 35–40°C for optimum solubilization (Gaur 1990). The maximum decrease in pH was observed on third and seventh day by *B. polymyxa* and *P. striata*, respectively, which increased on further incubation (Illmer and Schinner 1992). Recent report by Farhat et al. (2009) showed that the soluble phosphorus (P) concentration in optimized medium reached 967, 500, 595 and 326 mg/l from CaHPO₄, Ca₃(PO₄)₂, hydroxyapatite and RP within 72 h of incubation after inoculation with *Serratia marcescens*.

Actinomycetes as PSM are of special interest since these gram-positive filamentous sporulating bacteria can flourish in extreme environments (Jiang et al. 2005; Pathom-Aree et al. 2006) and also produce various plant growth-promoting substances (Fabre et al. 1988; Manulis et al. 1994; Ikeda 2003; Jain and Jain 2007). Many workers have isolated actinomycetes strains with phosphate-solubilizing ability (Ahmad and Jha 1968; Mahmoud et al. 1973; Ibrahim and Abdel-Aziz 1977; Banik and Dey 1983; Mba 1997; Hamdali et al. 2008). Mba (1994) isolated four actinomycetes from earthworm cast of *Pontoscolex corethrurus* which were able to solubilize up to 7 mg soluble P/g RP within one week of incubation. Interestingly, the RP-solubilizing power of these isolates was decreased with acidification of the medium and most of the isolates were able to hydrolyse carboxymethyl cellulose (CMC). Hamdali et al. (2008) isolated 55 actinomycete cultures, which were able to solubilize RP in synthetic minimum medium. Most of these isolates were from genera *Streptomyces* and *Micromonospora* and did not produce any organic acids but found to produce siderophores rendering the P available for plants.

Many workers have used a composite culture of bacteria and fungi to enhance the P solubilization but the efficiency of such mixed inoculants depends on the compatibility among microorganisms. Kundu and Gaur (1981) found that inoculation of *P. striata* and *Aspergillus awamori* together solubilized more phosphates than dual inoculation of *B. polymyxa* and *A. awamori*.

4.2.3 Phosphate-Solubilizing Fungi

A large number of fungi have also been reported to solubilize insoluble forms of phosphorus. These include *Aspergillus*, *Candida*, *Penicillium*, *Rhizopus*, *Cladosporium* and *Paecilomyces*, etc. (Agnihotri 1970; Gaind and Gaur 1991; Banik and Dey 1982; Singh et al. 1984; Venkateswarlu et al. 1984; Darmwal et al. 1989; Sheshadari et al. 2000). These PSF are well known for their ability to solubilize mineral phosphates owing to their ability to produce organic acids (Mattey 1992; Reddy et al. 2002; Bojinova et al. 2008; Richa et al. 2007; Xiao et al. 2009; Khan et al. 2010), and particularly some *Aspergillus* and *Penicillium* species have been tested by inoculating directly into the soil in order to solubilize RP (Kucey 1987; Vassilev et al. 1997). Inoculation of PSF and mycorrhizal fungi also improves the physico-chemical, biochemical and biological properties of RP-amended soil (Caravaca et al. 2004).

In another study, *Aspergillus tubingensis* and two isolates of *Aspergillus niger* have also shown the highest solubilization of RP under in vitro conditions (Reddy et al. 2002). Richa et al. (2007) tested *Aspergillus tubingensis* and *A. niger* for their efficacy to solubilize RP in RP-amended soils and they observed that available P along with organic carbon was significantly increased when compared to initial soil. Also the soil pH was lowered compared to initial pH of the soil. The improvement of physico-chemical and biochemical properties of RP-amended soil with the

inoculation of *A. niger* and mycorrhizal fungi was also reported by Caravaca et al. (2004).

Similarly, El-Azouni (2008) tested the efficacy of *Aspergillus niger* and *Penicillium italicum* to solubilize TCP in vitro and they found the release of inorganic P was 490 and 275 $\mu\text{g P ml}^{-1}$, respectively, after one week incubation. Change in soluble P with time with respect to pH has been observed with *Paecilomyces fusisporus* and other fungi (Goyal et al. 1982; Mishra et al. 1983). A nematofungus *Arthrobotrys oligospora* also has shown the ability to solubilize the phosphate rocks (Duponnois et al. 2006).

Arbuscular Mycorrhizae (AM) fungi are known to enhance P nutrition of plants especially in P-deficient soils by scavenging the available P due to the large surface area of their hyphae and by their high-affinity P uptake mechanisms (Hayman 1974; Moose 1980; Sanders and Tinker 1973). There are also reports of organic acid production by AM (Paul and Sundara Rao 1971) that could solubilize the insoluble mineral phosphates. As the soil phosphorus levels available to the plants increase, the amount of phosphorus also increases in the plant's tissues and thus the concomitant carbon drain from the plant by the AM fungi making symbiosis nonbeneficial to the plant (Grant et al. 2005).

4.2.4 Interaction of PSB with Other Microorganisms

The PSM when used with other plant growth-promoting rhizobacteria (PGPR) act synergistically to enhance crop yields (Saxena and Tilak 1994, 1997). Co-inoculation of *P. striata* and AM significantly increased the soybean yield and P uptake by plants over control. Dual inoculation of *Rhizobium* with PSM (Perveen et al. 2002) or arbuscular mycorrhizae (AM) fungi (Zaidi et al. 2003) has been shown to improve plant growth more than with their sole inoculation in P-deficient soils. Synergistic interactions on plant growth have been observed by co-inoculation of PSB with N_2 fixers such as *Azotobacter* (Kundu and Gaur 1984) and *Azospirillum* (Belimov et al. 1995) or with vesicular AM (Kim et al. 1998). Son et al. (2003, 2006) showed that co-inoculation of *Bradyrhizobium japonicum* and *Pseudomonas* sp. enhanced the number of nodules, dry weight of nodules, yield components, soil nutrient availability and uptake in soybean crop. Research workers have reported similar results with legume crops when PSBs are coinoculated with various N_2 fixing bacteria (Table 4.1).

4.3 Mechanisms of Phosphate Solubilization

Release of organic acids by PSM has been reported as a primary mechanism of phosphate solubilization (Hilda and Fraga 2000; Khiari and Parent 2005). Besides organic acids, the production of chelating substances (2-ketogluconic acid), humic

Table 4.1 Effect of co-inoculation of legumes with phosphate solubilizers and N₂ fixers

Crop	Rhizobia	Co-inoculating PS solubilizers	Plant responses to inoculation	References
Alfalfa	<i>R. meliloti</i>	<i>Pseudomonas</i>	Plant growth, nitrogenase activity, nodule number, total nodule weight and total plant nitrogen showed significant increase	Knight and Langston-Unkefer (1988)
Chickpea	<i>Mesorhizobium</i>	<i>Pseudomonas</i>	Marked increase in nodule weight and shoot biomass when coinoculated with <i>Mesorhizobium</i> and <i>Pseudomonas</i> in sterilized chillum jar conditions. In pot experiments, co-inoculation significantly increased root and shoot biomass	Sindhu et al. (2002a)
	<i>Mesorhizobium</i>	<i>Bacillus</i>	Dual inoculation significantly increased plant dry weight, nodulation, N content, protein content and seed yield, compared to single inoculation	Wani et al. (2007)
	<i>Rhizobium</i>	<i>Pseudomonas</i> , <i>Bacillus</i>	Significantly increased nodule weight, root and shoot biomass and total plant nitrogen	Parmar and Dadarwal (1999)
Clover	<i>R. leguminosarum</i> bv. <i>trifolii</i> 24	<i>Pseudomonas</i> sp.	Co-inoculation significantly increased shoot and nodule weight in comparison to plants inoculated with <i>R. leguminosarum</i> bv. <i>Trifolii</i>	Derylo and Skorupska (1993)
Common bean	<i>Rhizobium</i>	<i>A. brasilense</i>	Co-inoculation promoted root hair formation and an increase in secretion of the nod gene induced flavonoids resulting in greater number of nodules	Burdman et al. (1996)
Soybean	<i>B. japonicum</i>	<i>P. fluorescens</i>	Co-inoculation increased colonization of <i>B. japonicum</i> on soybean roots, nodule number and the acetylene reduction assay	Chebotar et al. (2001), Son et al. (2006)
Greengram	<i>Bradyrhizobium</i> sp. (<i>Vigna</i>)	<i>Bacillus</i>	Co-inoculation enhanced nodulation and growth of greengram	Sindhu et al. (2002b)
Wheat	<i>R. leguminisamarum</i>	<i>Pseudomonas</i> sp.	Dual inoculation along with P fertilizer increase yield by 30–40%	Afzal and Asghari (2008)

substances, mineral acids (sulphuric acids), siderophores and proton extrusion mechanisms also play an important role (Kapoor et al. 1989; Kucey et al. 1989; Illmer et al. 1995).

Growth of PSM is generally accompanied by decrease in pH of the medium and soil (Singh et al. 1980; Mishra and Banger 1985; Darmwal et al. 1989; Stevenson 2005). Reduction in pH is due to the production of organic acids, which include citric, gluconic, fumaric, malic, oxalic, lactic, 2-ketogluconic, malonic acids, etc. (Banik and Dey 1981; Venkateswarlu et al. 1984; Illmer and Schinner 1992; Vassilev et al. 1996; Hwangbo et al. 2003; Patel et al. 2008). However, quantity

and quality of organic acid produced is fully dependent on type of PSM. The highest solubilization of TCP and RP was reported by citric and fumaric acid (Gaur 1990). Calcium and magnesium in RP binds to the solubilized P and carbonate thereby neutralizing the organic acids, which play a vital role in solubilization.

The release of 2-ketogluconic acid by PSM was correlated with P solubilization due to its calcium chelating ability (Firsching 1969). Moreover, these compounds form stable organometallic complexes with Fe and Al and decrease precipitation of solubilized phosphates. Chelation of calcium, especially oxalic acid helps in solubilization of insoluble phosphates (Illmer and Schinner 1992). Besides acids production and chelators, bacterial exo-polysaccharides (EPS) also play an important role in mineral phosphate solubilization. Recently, Yi et al. (2008) observed the synergistic effects of EPS and organic acid on TCP solubilization, which varied with the origin and the concentration of EPS in medium. The increase of P solubilization brought by EPS is mainly attributed to the precipitation of EPS consequently resulting in greater phosphorus released from insoluble phosphate.

Humic substances are produced in soil as a result of organic matter decomposition and contain humic and fulvic acids as their main components. Humic acids are strong chelating agents which chelate Ca^{2+} and release H^+ , and thus help in dissolution of insoluble phosphates (Gaur 1969; Pareek and Gaur 1973; Mishra et al. 1983; Banger et al. 1985; Singh and Amberger 1990). The functional group of these compounds, such as carboxylic, phenolic, hydroxylic and phenolic hydroxyl, forms stable complex with Ca, Fe and Al (Banger and Mishra 1990). They also check reprecipitation of the solubilized phosphates (Singh and Amberger 1990).

The sulphate-reducing bacteria and other heterotrophic organisms under anaerobic conditions release H_2S that solubilize ferric phosphate by forming insoluble sulphides of Fe and render soluble phosphate (Gaur 1990). Rhizospheric microorganisms release CO_2 forming carbonic acid (HCO_3^-), which lowers the pH, and thus help in P solubilization (Hayman 1975). The oxidation of inorganic sulphur and pyrite to sulphuric acid by sulphur oxidizing bacteria (*Thiobacillus*) lowers the soil pH and hence improves the availability of phosphates (Wainwright 1984; Kapoor et al. 1991) due to formation of CaNO_3 and CaSO_4 . Certain rhizospheric microorganisms (*Rhizobium*, *Azotobacter*, *Pseudomonas*) synthesize iron-chelating compounds (siderophores), which in acidic soil remove iron from ferric phosphates and render phosphate for plants (Bossier et al. 1988; Suneja and Lakshminarayana 1999; Hamdali et al. 2008).

Proton extrusion mechanism is a probable mechanism of phosphate solubilization in some microorganisms, which can solubilize phosphates without the release of organic acids in the environment (Illmer et al. 1995). Release of H^+ accompanying respiration and assimilation has been observed in *Penicillium aurantiogriseum* and *Pseudomonas* sp. solubilizing hydroxyapatite (Illmer et al. 1995; Lin et al. 2006). The insoluble phosphates are solubilized at the cell surface of the microorganisms. The NH_4^+/H^+ exchange mechanism depends on the presence of NH_4^+ ions in the medium as uptake of NH_4^+ occurs with the concomitant release of H^+ in the environment (Roos and Luchner 1984). Quality and quantity of root exudates also alter the concentration of P in the soil solution due to presence of organic ligands in the root exudates (Hinsinger 2001).

4.4 Estimation and Enumeration of Phosphate-Solubilizing Microorganisms

PSMs can be isolated from different sources such as soil (Khanna et al. 1979; Gupta et al. 1986; Kapoor et al. 1989; Roychoudhary and Kaushik 1989), rhizosphere (Sardina et al. 1986; Thakkar et al. 1993; Singh and Kapoor 1994), root nodules (Chhonkar and Subba Rao 1967; Surange and Kumar 1993), compost (Kapoor et al. 1989; Thakkar et al. 1993), RPs (Gaur et al. 1973) and earthworm casts (Mba 1994).

Solubilization of tricalcium phosphate (TCP) in agar medium has been used as the initial criterion for isolation and enumeration of PSM. Stalstrom (1903) first demonstrated the solubilization of TCP by soil bacteria in solid and liquid medium. Organisms growing on such media are able to solubilize P and produce clear zone around colonies due to dissolution of calcium phosphate. Mineral P solubilizers have been routinely isolated and screened using PVK medium (Pikovskaya 1948) or Sperber medium (Sperber 1957) by plate assay method, looking for halo/clear zone around colonies of potential P solubilizers (Katznelson et al. 1962; Das 1963; Bardiya and Gaur 1974; Darmwal et al. 1989). The zone of clearance is due to solubilization of inorganic phosphate mainly due to the production of organic acids in the surrounding medium (Johnston 1952; Das 1963; Sethi and Subba Rao 1968; Pareek and Gaur 1973; Kundu and Gaur 1981; Kapoor et al. 1989; Halder et al. 1990; Singal et al. 1991; Yadav and Dadarwal 1997).

Although, microorganism capable of producing zone of clearance around its colonies in plate assay method is selected as a potential P solubilizer, however, several workers have reported that many isolates which did not produce any visible zone of clearance on agar plates but could solubilize various types of insoluble inorganic phosphate in liquid medium (Louw and Webley 1959; Das 1963). Thus, the existing plate assay fails where the halo zone is inconspicuous or absent. This may be because of the varying diffusion rates of different organic acids secreted by different microorganisms (Johnston 1952). On the other hand microorganisms, which showed P solubilization in laboratory medium, do not give any solubilization in soil. For example, Whitelaw et al. (1999) showed that, even though *Penicillium radicum* produced GA when grown in laboratory culture, the organic acid was not the primary mechanism for solubilization of precipitated P. Hence, there is need of suitable medium for quick and reliable plate assay for screening PSM. However, the plate assay can be regarded as generally suitable for isolation and preliminary characterization of PSM.

Gupta et al. (1994) gave a modified PVK medium formulated using different concentrations of bromophenol blue (BPB) dye for screening PSM. Results revealed that as the concentration of dye increased, the clarity and visibility of the yellow coloured halo/zone improved, and most appropriate dye concentration was found to be 2.4 mg/ml. There is no correlation between halo zone formation and quantity of inorganic phosphate solubilized (Ostwal and Bhide 1972; Arora and Gaur 1979). Indicator medium containing bromothymol blue (BTB) has also been developed to enhance chances of picking up efficient PSB (Krishnaraj 1996).

Advantage of using modified medium is that the incubation period required prior to selection of PSB is significantly reduced, i.e. minimum incubation period for PSB was 24 h than in original PVK medium that exceeds 7 days.

National Botanical Research Institute's phosphate growth medium (NBRIP), which is more efficient than PVK medium was developed for screening of PSM (Nautiyal 1999). NBRIP medium was comparable to PVK agar medium; however, in broth assay, NBRIP medium consistently resulted in a 3-fold increase in P solubilization. The rate of phosphate solubilization was increased with increased concentrations of glucose (Nautiyal 1999). Also, glucose, xylose and sucrose are reported to be good carbon sources for PSF (Ahmad and Jha 1968), while for bacteria glucose, galactose and sucrose were found to be effective. However, the solubilization potential of microorganisms varies with different carbon sources depending on the type of insoluble phosphate (Ahmad and Jha 1968; Thakkar et al. 1993). In anaerobic conditions, P release is affected by the physiological state of cells and carbon sources (acetic, propionic and butyric acid) as observed under stationary conditions (Rustrain et al. 1997). The utilization of glucose directly correlates with drop in pH, which was reflected as mineral phosphate solubilization due to conversion of glucose to GA (Goldstein and Liu 1987; Liu et al. 1992). It was observed that 60 mM GA produced results in release of 0.1 mM inorganic phosphate. The amount of acid liberated by PSB roughly is more than 5 per cent of the carbohydrate consumed (Banik and Dey 1982). Phosphate solubilization ability was improved with $(\text{NH}_4)_2\text{SO}_4$ at a lower concentration and was 27.1 percent more effective than KNO_3 as observed by Nautiyal (1999). *P. striata* (PS-9) has been found to utilize urea, asparagine, ammonium sulphate, ammonium nitrate, potassium nitrate and calcium nitrate for solubilization of RP (Gaur 1990). Sodium nitrate and peptone were found to be better substrate for P solubilization by *B. megaterium* var. phosphaticum. *P. striata* solubilized the higher amount of insoluble phosphates among bacteria, and quantities of RP solubilized were much less than TCP and hydroxyapatite (Arora and Gaur 1979). However, incorporation of bacteria with RP resulted in increased availability of inorganic phosphate for plant utilization (Hebbara and Devi 1990; Rachewad et al. 1992; Jisha and Alagawadi 1996). But, in plate assay, for efficient P solubilizers, tricalcium is used instead of MRP. As TCP gets metabolized faster, it is normally used for screening studies compared to RPs (Singh and Kapoor 1994). Also, TCP solubilization is higher as compared to aluminium phosphate and ferric phosphate (Narsian et al. 1993). Microbial solubilization of RP is influenced by the physical and chemical properties of the RP and the microorganisms involved (Garg et al. 1989; Kapoor et al. 1989).

Bacteria prefer to grow in neutral to alkaline (pH 7 to 8) reaction for maximum solubilization (Wani et al. 1979) and fungi do better at slightly acidic to neutral pH (Gaur 1990). A direct correlation was obtained between decrease in pH and increase in available P of the culture media in certain cases (Sperber 1958; Agnihotri 1970; Liu et al. 1992) but others have contradictory reports that solubilization is not always proportional to the decline in pH (Mehta and Bhide 1970; Krishnaraj 1996; Asea et al. 1988; Parks et al. 1990). The form in which inorganic

phosphate exists also changes according to the soil pH. Below pH 6.0, most inorganic phosphate is present as monovalent H_2PO_4 species. The plant uptake is also high at the pH range of 5.0–6.0, which indicates that P is primarily taken up as monovalent form (Furihata et al. 1992). It has been noted that PSB could solubilize both RP and dicalcium phosphate in unbuffered media but failed to solubilize RP in buffered media. The organic acid secreted by these bacteria was 20–50 times less than that required to solubilize P from alkaline soil (Gyaneshwar et al. 1998).

4.5 PSM and Yield of Crops

PSM inoculation can increase crop yields up to 70% (Verma 1993). The increase in yield is mainly attributed by the increased availability of soluble phosphorus on inoculation with PSMs, which enhance the plant growth by improving biological nitrogen fixation (Kucey et al. 1989).

Sundara Rao et al. (1963) reported increase in yields and phosphate uptake in tomato and wheat with phosphobacterin and a strain of *Bacillus megaterium*. Kundu and Gaur (1980) reported increased potato yield with phosphobacteria in potato. Also, increase in wheat yield with inoculation of PSM-*Azotobacter chroococcum* is reported. Similarly, increase in yield was observed in various crops such as chickpea, rice, etc. (Alagawadi and Gaur 1988; Kavimadan and Gaur 1971; Monod et al. 1989).

Similarly, several authors have reported increased yield of wheat (Whitelaw et al. 1997; Omar 1998), onion (Vassilev et al. 1997), alfalfa (Rodríguez et al. 1999), rice (Khan et al. 2008), maize (Richa et al. 2007) and soybean (Abd-Alla and Omar 2001) through simple inoculation of PSF. Recently, El-Azouni (2008) observed that the dual inoculation of PSF (*A. niger* and *P. italicum*) significantly increased dry matter and yield of soybean plants compared to the control TCP-amended soil in pot experiment along with significant increase in the percentage of N and P content of the plant.

4.6 Future Prospects

PSMs are an integral component of soil microbial community and play an important role in P cycle in soil rendering the unavailable P to plants. These PSM have enormous potential for making use of fixed P in the soil particularly in soils with low P availability in tropical and subtropical developing countries. The mechanism of phosphate solubilization by microorganisms has been studied in detail but the P solubilization is a complex phenomenon affected by many factors, such as PSM used, nutritional status of soil and environmental factors. Moreover, the stability of the PSMs after inoculation in soil is also important in P solubilization to benefit crop growth. Therefore, it needs further studies to understand the characteristics

and mechanisms of phosphate solubilization by PSM. To conclude, the efforts should be made to identify, screen and characterize more PSM for their ultimate application under field conditions.

Acknowledgement The authors acknowledge the help from Mr. Harsh Kuhad, Amity University, Noida, India, during the preparation of this chapter.

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