Identification of Emerging Contaminants in Drinking Waters

Sana Mokrane, Djamila Harik, and Hubert Cabana

Keywords

Emerging contaminants • Residues • Pharmaceuticals • Pesticides • Detection • Elimination

1 Introduction

The presence of contaminants of emerging concern (CEC) in drinking water was reported in many studies Reddersen et al. (2002), Benotti et al. (2008), Dunn et al. (2014). Organic contaminants such as pharmaceuticals and pesticides residues which result from human activities in different sectors, and escape from traditional treatments by wastewater treatment plants (WWTPs) Haroune et al. (2017), Verlicchi et al. (2012), Boussahel et al. (2003), so they might affect human health Focazio et al. (2008).

After a physico-chemical analysis of three Algerian drinking waters, we presented a contribution to the identification of some emerging pollutants in drinking water of four municipalities in South Central Quebec Canada. This identification is a very important step and a basis in the field of treating drinking water might be applied on Algerian waters to verify the presence of contaminants of emerging concern (CEC) in order to predict the process of their elimination.

D. Harik e-mail: djamila.harik@g.enp.edu.dz

H. Cabana

Université de Sherbrooke, Sherbrooke, Canada e-mail: Hubert.Cabana@USherbrooke.ca

2 Materials and Methods

2.1 Physico-Chemical Analysis of Drinking Water

The quality of drinking water was confirmed by the results of some parameters (by adopting the ISO methods) achieved by Science and Techniques of Environment laboratory (LSTE) of Polytechnic School of Algiers and SEAAL (Algerian Society of Water and Sanitation) in Boudouaou. The parameters were.

The hydrogen potential (pH), conductivity, turbidity, ammonia nitrogen, nitrates, nitrites, phosphates, the organic matter (OM), magnesium and calcium ions, total hardness TH, chloride, alkalinity, chemical oxygen demand (COD), test on marble and active chlorine.

2.2 Liquid-Liquid Extraction Method

Drinking water samples were collected from taps into new amber glass bottles during winter (February 2017). Samples were then divided in two equal volumes to make replicates and were put each in separating funnel, 2% w/v of NaCl were added, then extraction was performed at first by adding 10% v/v of Dichloromethane and then repeated with 10% v/v of Ethyl acetate. The organic phase was collected into glass vial and in order to concentrate it. It was all evaporated to dryness under gentle stream of nitrogen before re-solubilization with 2 ml of methanol. Suspensions were then filtered through 0.2 µm filter before transfer into UPLC vial for quantitative analysis. Samples were stored at 4 °C until analysis by UPLC-MS/MS Ba et al. (2014).

A. Kallel et al. (eds.), Recent Advances in Environmental Science from the Euro-Mediterranean and Surrounding Regions, Advances in Science, Technology & Innovation, https://doi.org/10.1007/978-3-319-70548-4_230

S. Mokrane $(\boxtimes) \cdot D$. Harik

Ecole Nationale Polytechnique d'Alger, El-Harrach, Algeria e-mail: sana.mokrane@g.enp.edu.dz

[©] Springer International Publishing AG 2018

3 Results and Discussion

3.1 Physico-Chemical Analysis of Three Algerian Drinking Waters

The usual water analysis results show that drinking water values of all parameters respect the potability standards required by the Republic of Algeria and the World Health Organization (WHO), and we can conclude that these samples are potable (Table 1).

3.2 Identification of Some Emerging Contaminants in South Center Quebec

Treated drinking water was analyzed for the presence of 70 chemicals, 20 molecules were detected (7 pharmaceuticals and 13 pesticides). The results shown below (Table 2) indicate that the municipality which waters contained high concentrations of pesticides is Sherbrooke while the one with high pharmaceuticals contents is Drummondville; these different amounts of chemical residues are related to many

Mineral water	ENP Water tap	Home Water tap	Potability
(Guedila)			standards
7.52	7.92	8.01	6.5 à 8.5
767	1089	1016	2800
0.144	0.277	0.34	5
<0.015	<0.015	<0.015	0.5
3.62	4.134	4.167	50
<0.02	<0.02	<0.02	0.2
<0.015	<0.015	<0.015	0.5
-	1.84	1.2	5
$[Mg^{2+}] = 45$ $[Ca^{2+}] = 80$	$[Mg^{2+}] = 38$ $[Ca^{2+}] = 76$	$[Mg^{2+}] = 36$ $[Ca^{2+}] = 84$	$[Mg^{2+}] = 150$ $[Ca^{2+}] = 200$
38	34	36	(*)
51.68	148.6	103.37	500
27	22	18	
<30	<30	65	<250
-0.14	0	0.1	(**)
0.06	0.18	0.08	No value detected, but we can notice the taste from 250 mg/l and above
	(Guedila) 7.52 767 0.144 <0.015	(Guedila) 7.52 7.92 767 1089 0.144 0.277 <0.015	(Guedila) -1 -1 -1 7.527.928.01767108910160.1440.2770.34<0.015

 Table 1 Physico-chemical analysis of three Algerian drinking waters

 TH (°f)
 0–7
 7–15
 15–30
 30–40
 +40

 Water
 So pure
 Pure
 Little hard
 Hard
 Very hard

(**): When the saturation index is: Is > 0, water is « incrustante »

Is = 0, water is « inert »

Is < 0, water is « aggressive »

(*): is about (Total Hardness (TH))

(**): is about saturation index

Table 2 Concentration of some pharmaceuticals and pesticides found in drinking water of four different municipalities in South Center Quebec with their treatment method

	Pharmaceuticals (ng/l)	Pesticides (ng/l)	Drinking water treatment method
Sherbrooke 1	0.0656	0.112	Chlorination, ozonation, micro-straining
Sherbrooke 2	0.0336	0.1704	
Richmond 1	0.0696	0.0648	Chlorination, filtration, iron/manganese
Richmond 2	0.0648	0.0288	
Drummond 1	0.1136	0.0944	Chlorination, filtration, charcoal
Drummond 2	0.1288	0.1384	
Windsor 1	0.0696	0.1112	-
Windsor 2	0.0608	0.0376	

factors like demography, treatment methods, size of the sector that provides these products and their consumption and price Kookana et al. (2014).

4 Conclusion

Usual physico-chemical analysis showed that the consumed Algerian drinking waters answer globally the international standards, furthermore, deeper quantitative analysis on Canadian drinking waters highlighted the presence of residues of emerging contaminants such as pharmaceuticals and pesticides. These pollutants may cause several problems to human health and thus their identification is crucial with the view to eliminate them.

References

- Ba S, Haroune L, Cruz-Morató C, Jacquet C, Touahar IE, Bellenger JP, Cabana H. Synthesis and characterization of combined cross-linked laccase and tyrosinase aggregates transforming acetaminophen as a model phenolic compound in wastewaters. Sci Total Env. 2014;487:748–55.
- Benotti MJ, Trenholm RA, Vanderford BJ, Holady JC, Stanford BD, Snyder SA. Pharmaceuticals and endocrine disrupting compounds in US drinking water. Environ Sci Technol. 2008;43(3):597–603.

- Boussahel R, Harik D, Moussaoui KM. Tests on the elimination of present pesticides in Algerian waters. In: Proceedings of the 2nd Scientific days of Oran: ANDRS; 2003. p. 38–40.
- Dunn G, Bakker K, Harris L. Drinking water quality guidelines across Canadian provinces and territories: Jurisdictional variation in the context of decentralized water governance. Int J Environ Res Publ Health. 2014;11(5):4634–51.
- Focazio MJ, Kolpin DW, Barnes KK, Furlong ET, Meyer MT, Zaugg SD, Thurman ME. A national reconnaissance for pharmaceuticals and other organic wastewater contaminants in the United States—II) Untreated drinking water sources. Sci Total Env. 2008;402(2):201–16.
- Haroune L, Saibi S, Cabana H, Bellenger JP. Intracellular enzymes contribution to the biocatalytic removal of pharmaceuticals by Trametes hirsuta. Environ Sci Technol. 2017;51(2):897–904.
- Kookana RS, Williams M, Boxall AB, Larsson DG, Gaw S, Choi K, Yamamoto H, Thatikonda S, Zhu YG, Carriquiriborde P. Potential ecological footprints of active pharmaceutical ingredients: an examination of risk factors in low-, middle- and high-income countries. Philos Trans R Soc London. Ser B, Biol Sci. 2014;369 (1656).
- Reddersen K, Heberer T, Dünnbier U. Identification and significance of phenazone drugs and their metabolites in ground-and drinking water. Chemosphere. 2002;49(6):539–44.
- Verlicchi P, Al Aukidy M, Zambello E. Occurrence of pharmaceutical compounds in urban wastewater: removal, mass load and environmental risk after a secondary treatment-A review. Sci Total Environ. 2012;429:123–55.