



Environmental Impact Measurements: Tool and Techniques

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Abstract

The ecosystem provides various services such as regulation, support, provisions and culture to living beings on the earth. The productivity of a system is greatly affected by the health of the different components and the level of contamination in it. Increasing industrialization and reduced sources of natural resources for safe use by the growing population leads to poor productivity of ecosystems. Environmental impact assessment is a current need for the sustainable survival of human being on earth. The increasing industrialization and population, as well as mismanagement of natural resources, are creating environmental threats. Nowadays people are more worried about natural calamities and the substantial reduction of environmental quality worldwide. Many techniques are available to assess and determine environmental factor intensity and quality at any given time. Therefore, the use of modern technologies in this field can be a viable option to warn of natural calamities and to save or effectively manage human life and natural resources. Most developing countries today need to execute environmental policy and effective guidelines, and provide the infrastructure, to accurately assess environmental effects on natural resources and the ecosystem's different biogeochemical cycles. Spreading awareness among people through governments and nongovernmental organizations also has a valuable place in combating the incidence of natural calamities and the deterioration of environmental health.

Introduction

The environment is a protective cover for living organisms on the earth. In common terms, "environment" refers that all biotic and abiotic things and their interactions with biotic organisms, nonliving things, weather and climatic conditions, and natural resources. The term has been defined differently by different field of studies and their available knowledge and priorities. One field of specialization may be concerned with particular parameters of an environment, whereas another field of science may emphasize others. Natural management researchers focus more on conservation and rational utilization of soil, water, and other available resources on earth, with less emphasis on environmental aspects. Economists, on the other hand, may be more concerned with per-unit investment of credit and less focused on environmental health.

Such ambiguity in environmental perspectives promotes environmental contamination and adversely affects economic and demographic activities within the global market. The International Association for Impact Assessment defines the environmental impact assessment (EIA) as “the process of identifying, predicting, evaluating and mitigating the biophysical, social, and other relevant effects of development proposals prior to major decisions being taken and commitments made” [1]. Some procedures need specialized skills and tools assess real problems in ecosystems [2, 3]. The initial cost of instrumentation is high, and smaller organizations may not be able to afford it [4]. Inventories of precious natural resources and their utilization for the welfare of people are clearly depicted in national and international policies formulated by the governments of some countries. On the basis of environmental impact, these policies may be enacted at regional, national, or international levels; the scale of impact assessments also change with the type of problem and its adverse effect on an ecosystem’s activities. Modern technology and advanced scientific research may lead to identification of the exact impact of environmental calamities [5, 6] contaminants to health, and the availability of natural resources on a global scale [7].

The changing scenario of climate change greatly affects environmental quality and its functions across the globe. Climate change is unequivocal, and its impact on the Indian agricultural production system can already be felt. The climatic changes have mediated soil-forming processes [8, 9], crop productivity [10], and microbial diversity in soil [11]. Increases in atmospheric temperature enhances the respiration rate in plant; and reduces biomass production [12]. This situation is a bit different in low-temperature zones, where increasing temperatures enhance biomass production and crop yields. Increased concentrations of greenhouse gases (GHGs) in the atmosphere enhance the global temperature and affect the depreciation rate of various processes on earth [13]. Numerous agricultural management strategies play a crucial role in minimizing the emission of GHGs into the atmosphere, and, accordingly, impact assessments need to scale up local situations and resource availability [14, 15]. The primary GHGs in the earth’s atmosphere are water vapor, CO₂, CH₄, N₂O, and ozone gas. Agriculture is a potential source of (CO₂, N₂O and CH₄) and sink for (CO₂) GHGs. Adopting agricultural process focused on conservation and dominant cropping systems; adjusting agricultural inputs and land use management practices; and enhancing resource use efficiency, waste recycling, and solid waste management are some of the key options available for developing climate resilience [13]. Management of agricultural land, land use changes, and forest cover profoundly influences the concentration of GHGs in the atmosphere. Regularly and more precisely monitoring changes in climatic events is needed to better use and avoid massive losses of human life as well as economic losses in a country.

Assessment of environmental qualities or parameters is necessary to execute different developmental activities within a region. Public services infrastructure, healthy crop production, industrial growth and development, and wildlife survival must be established [4]. Agricultural crop fields are already categorized per impact analyses of crop productivity with respect to climatic and soil conditions. Soil surveys and land use planning classify agricultural land on the basis of soil characteristics and constraints for agricultural crop production [16]. In a similar way,

forested areas are also studied using environmental impact analyses. Environmental impact analysis is a valuable tool when the climatic conditions within a region abruptly change, as is happening now in some areas. It is also important to take appropriate remediation measures to combat adverse conditions and minimize loss to the ecosystems. This assessment also very complex, with relations between soil types, vegetation, climatic conditions, pollutants and local management options [17, 18]. Selecting the right approach with a high degree of sensitivity to determine environmental contamination is needed today across the globe. Soil is important for the healthy function of most ecosystems, the atmosphere, the lithosphere, and the hydrosphere, and also for the survival of humans on the planet. Most ecosystems are interlinked and mediate biota diversity and function. In this chapter we describe the tools and techniques that are most used for environmental impact analysis with respect to natural resource management.

Role of Climate in Ecosystem Functions

Climate change is a current topic of discussion worldwide. In a nutshell, increasing the concentration of GHGs in the atmosphere increases the global temperature and affects ecosystem functions. Climate fluctuations and increases in GHGs adversely affect the productivity and potential functionality of an ecosystem, and thereby its efficiency. “Ecosystem function” is the technical term used to define the biological, geochemical, and physical processes, and their components, that occur within an ecosystem. It provides various services to humankind: regulatory, supportive, provisional, and cultural. The Ecosystem Service Framework lists 19 types of ecosystem function [19] (Table 1).

Climate Change

Diversity and ecosystem services are the key areas that are vulnerable to climate change. In most scientific discussions, climate change has played a crucial role in agricultural production, management of forest productivity and processes, system efficiency of mechanical services related to directly or indirectly. Changes in the atmospheric concentrations of GHGs, aerosols, land cover and solar radiation alter the energy balance of the climate system and mediates the effects of climate change. In the preindustrial era, the global temperature increased as a result of increases in atmospheric concentrations of GHGs. This increase affects soils, kinetics of nutrient uptake by plants, soil-forming processes, ecosystem services, degradation of contaminants, and climatic events, among others. Rapid industrialization enhances the GHG emission rate, which is higher than what can be absorbed green plants or natural purifiers. The earth reflects solar (shortwave) radiation back to the atmosphere as longwave radiation. Because of the increasing concentration of GHGs in the atmosphere, the rate of reflection is decreasing and thereby increases the temperature. The measurement of global warming potential (GWP) is a primary factor in

Table 1 Ecosystem function

Ecosystem function category	Ecosystem function	Description
Regulating functions	Gas regulation	Balances the gas concentration in the atmosphere in relation to different soils and living processes
	Climate regulation	Regulates the weather and climatic phenomena; adjusts and creates microclimates governing different activities of crop production and plant and animal (including humans) lives and their functions
	Disturbance regulation	Maintains a soil's resilience
	Water regulation	Regulates spatial and temporal distributions of water
	Soil retention	Minimizes soil loss and soil degradation through natural and anthropogenic activities
	Nutrient regulation	Plant nutrient transport, storage, and recycling in different bio-geocycles
	Waste treatment and assimilation	Recycles organic and inorganic wastes
	Pollination	Interactions between plants and (1) biotic vectors (through insects and birds) and (2) abiotic vectors (through wind and water)
	Biological control	Naturally controls of insects and pests
	Barrier effect of vegetation	Reduces the soil and climate degradation by contaminants
Supporting functions	Supporting habitats	Provides suitable breeding and reproduction amenities to species
	Soil formation	Maintains the soil-forming process (e.g., rock to fine soil)
Provisioning functions	Food	Provides food for biota living on and within earth
	Raw materials	Provides raw materials for various industries
	Water supply	Provides water balance through different process (e.g., sediment trapping, infiltration, dissolution, precipitation, and diffusion)
	Genetic resources	Self-maintaining diversity of a system
	Provision of shade and shelter	Provides shade or shelter to plants, animals, and human beings
	Pharmacological resources	Provides pharmacological resources
Cultural functions	Landscape opportunity	Provides different extents and varieties of natural features and landscapes

the exact assessment of global effects of climate change. The Intergovernmental Panel on Climate Change provides generally accepted values for GWP, which changed slightly between 1996 and 2001 [20]. The Intergovernmental Panel on Climate Change's 2001 Third Assessment Report provides an exact definition of how GWP is calculated. It is defined in relative measures of a gas to trap heat from atmosphere in comparison to CO₂:

$$GWP(x) = \frac{\int_0^{TH} a_x \cdot [x(t)] dt}{\int_0^{TH} a_r \cdot [r(t)] dt}$$

where TH is the time horizon over which the calculation is considered; a_x is the radiative efficiency due to a unit increase in the atmospheric abundance of the substance, and $[x(t)]$ is the time-dependent decay in abundance of the substance following an instantaneous release of it at $t = 0$. The denominator contains the corresponding quantities of the reference gas (i.e., CO_2). The radiative efficiencies a_x and a_r are not necessarily constant over time. While the absorption of infrared radiation by many GHGs varies linearly with their abundance, a few important ones display nonlinear behavior for current and likely future abundances (e.g., CO_2 , CH_4 , and N_2O). For those gases, the relative radiative forcing will depend upon abundance and hence upon the future scenario adopted. Carbon dioxide has a GWP of exactly 1; it is the baseline unit to which all other GHGs are compared).

Instrument Used to Measure GHGs Concentration

GHGs are the key component that affects the intensity of global climate change in various ecosystems. GHG concentrations are measured through the use of gas chromatography. It is a method using in analytical chemistry to separate and analyze compounds that can be vaporized without decomposing. This instrument is currently used to separate particular substances or separate compounds within a mixture. It is also used to prepare a pure compound from a heterogeneous mixture. A gas chromatograph has two phases: (1) the mobile phase, an inert gas that acts as the carrier (e.g., helium or nitrogen), and (2) the stationary phase, a microscopic layer of liquid or polymer on an inert solid support inside a piece of glass or metal tubing called a column. Gas samples are collected from the ecosystem into a static chamber (Fig. 1) and run on a gas chromatograph to categorize the compounds within them. In an agricultural crop production system, use efficiency of various inputs, or the production of GHGs by the inputs, is a key research parameter when selecting an input combination to enhance soil and crop productivity and conserve environmental health. As a result of large spatial and temporal variability—for example, GHG production and consumption in soil—a large number of replicating chambers are often required.

Survey for Impact Analysis

Survey plays a crucial role in the assessment of any impact for a particular aspect. For example, in an environmental risk assessment, few sequential steps are needed for monitoring with respect to human health, that is, site characterization, exposure assessment, toxicity assessment, and risk characterization [21]. In all steps, the specific objective is to collect data for a particular region for a particular aspect.



Fig. 1 Static chamber use to collect samples

On the basis of statistical principles, survey is categorized into judgment-based and statistically based survey.

Judgment Based

In this method, samples are taken on the basis of investigator knowledge as well as affected areas within the region. It is a simple task, and contamination in the environment can be quickly identified. It is, however, highly biased and is mostly used for preliminary investigation of a contaminant.

Statistically Based

The statistical principles working in these survey methods are considered science-based procedure across the world [22]. These procedures have low risk of error by investigators. Two types of sampling are used. The first is *random sampling*, in which sampling points are chosen randomly, but not arbitrarily, from a region (Fig. 2a). Legitimate random-number generators are used to collect samples from a particular location. It can show the delineation boundary of contaminants, but has lower efficiency in identifying contaminants “hot spots.” The second is *stratified sampling*, in which regions are divided into subareas based on the geomorphology or the different characteristics of regions (Fig. 2b). Each subarea is analyzed separately and concluded to be an individual site. This type of sampling is laborious, requires high skill, and makes the survey more expensive. It does, however, have higher reliability and avoids the risk of insufficient sampling data from polluted regions.

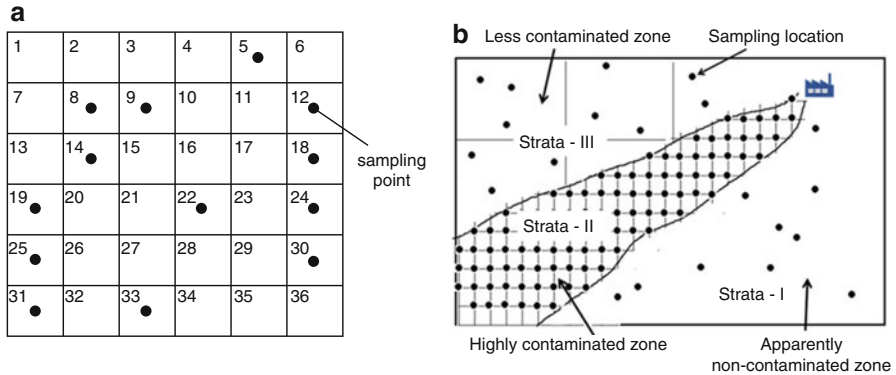


Fig. 2 Schematic layout of sampling patterns: (a) random sampling and (b) stratified sampling

Reasons for Sampling

Sampling is the key function of an impact analysis and determines the accuracy of its results. Sometimes, however, erroneous sampling can lead to misinterpretation of results and may convey the wrong message to investigating agencies.

Sampling Procedure for Soil Fertility Evaluation

Crop production is affected by a soil's fertility and capacity to supply nutrients. Soil fertility and nutrient use efficiency are the key components in the evaluation of new methods of crop cultivation [5, 23] and development, and in an impact analysis of a new chemical molecule for crop production [24, 25]. A soil fertility evaluation requires predictions for targeted crop yields, categorization of soils, and an assessment of the types and degree of soil related constraints (e.g., salinity or alkalinity), and surface soil samples up to 15 cm in depth are needed. The samples should be taken in a zig-zag pattern to avoid human bias. In this procedure, four to six samples are collected from 1 ha of a crop field using an auger; and quartering & removes the two opposite portion of the soil samples. This process must be repeated until a 0.5 kg soil sample remains. When collecting soil samples, the following must be avoided:

- Bunds, heaps of FYM, chemical fertilizers, or agricultural inputs
 - Use of bags or boxes previously used to store fertilizer or agricultural inputs
 - Use of iron *khurpi* to sample micronutrients
 - Mixing of different types of soil samples in a composite sample
- Apply the following guidelines when taking samples:
- Use stainless steel augers.
 - Take a separate sample for each particular type of soil.

- Take a sample from a V-shaped hole after removing the top few centimeters of soil, debris, and crop residue. A depth of 15 cm is appropriate for most agricultural crops, but for plantation or deep-rooted crops it should go down to the depth of the roots.
- Properly label each sample and submit then to the laboratory with complete information: name of the farmer, location, soil type, crop history, date of sampling, agriculture input used, etc.

Environment Impact Measurement Parameters

Assessment of Soil Fertility Levels

Soil fertility and environmental services are interdependent and affect the interrelation of various ecosystems. Soil properties determine crop yield and soil productivity [26]. Soil samples are collected from the surface layer (down 15 cm) for most crops, but soil depth can be extended depending on the purpose and the type of crop. Horticultural crops need soil samples from various depths (15, 30, 60, 90 cm). After processing, soil samples are passed through a 2-mm sieve and used to analyze various properties. Some of the important instruments used to analyze soil properties are listed in Table 2, and associated methods of analysis are listed in Table 3.

pH

Soil pH is the one of the key parameters of soil fertility and the availability of plant nutrients essential for crop growth [103]. Soil pH is a measure of a soil's acidity and alkalinity. For analysis, soil and water at a ratio of 1:2 or 1:2.5 is prepared and the probes of a pH meter are placed into the soil–water mixture. Soil pH ranges from 0 to 14. In acidic soil, the pH measured with this method normally ranges between 4 and 6; a neutral pH is 6.5 to 7.5, and a pH >7.5 indicates an alkaline condition. These limits are further divided into various minor limits.

Electrical Conductivity

The electrical conductivity of soil is measured in a similar way as soil pH except in a clear supernatant and using an electrical conductivity meter. In this method, 0.01 *N* potassium chloride solution is used to standardize the instruments.

Organic Carbon

Organic carbon is greatly affected by climatic conditions as well as the crop patterns of a region. Soil organic carbon acts as an index of soil fertility and microbial activity in soil. On the basis of analysis, it is categorized as low (<0.5%), medium (0.5–0.75%), and high (>0.75%). It is analyzed through the use of the Walkley Black method [29], in which $K_2Cr_2O_7$ is used as an oxidation reagent and back-titrated with 0.5 *N* ferrous ammonium sulphate. Soil organic carbon values are directly related to the availability of plant nutrients in soil. Because of climatic variations,

Table 2 Common instruments for measuring impact assessment related to natural resource management

Instrument	Use in environmental impact assessment activities
pH meter	Soil reaction (acidity, alkalinity)
EC meter	Conductivity in soil
TOC analyzer	Organic carbon in soil
HPLC	Organic compounds in the ecosystem
CHNS	Carbon, nitrogen, sulfur and hydrogen in soil and plants
GC	Gaseous composition of the atmosphere
GM counter	Radioactivity in a system
AAS/ICP-OES	Metal concentration in a system
Flame photometer	Analysis of potassium, calcium and sodium
Flow injection nitrogen analyzer	Nitrogen analysis
Spectrophotometer	Phosphorus determination
MIR spectrophotometer	Soil fertility evaluation
TXRF	Soil fertility evaluation
Pressure plate apparatus	Capacity of a soil to hold water
Cone penetrometer	Soil hardness
Yoder apparatus/permeameter	Soil aggregates
IR gas analyzer	Respiration rate in plants
Mridaparikshak	Soil fertility assessment
COD meter	Chemical oxygen demand
BOD	Biological oxygen demand
Nitrogen auto-analyzer	Nitrogen content in a system
GPS	Identification of location
Munsell color chart	Color of a soil
Bouyoucos hydrometer	Soil texture
Tensiometer	Measurement of moisture in soil
G:C ratio, DNA hybridization, ribotyping, RISA, RAPD, metagenomics	Identification of soil biodiversity

AAS, *BOD* biological oxygen demand, *CHNS*, *COD* chemical oxygen demand, *EC* electrical conductivity, *G:C*, *GC* gas chromatography, *GM* Geiger Muller, *GPS* global positioning system, *HPLC* high-performance liquid chromatography, *ICP-OES* inductively coupled plasma optical emission spectrometry, *IR* infrared, *MIR*, *RAPD* random amplification of polymorphic DNA, *RISA* ribosomal intergenic spacer analysis, *TOC* total organic C, *TXRF* total reflection X-ray fluorescence

plants vary their secretion of root exudates. Carbon accumulated in soil could be measured through the use of this procedure [37].

Available Nitrogen

Easily mineralizable nitrogen is estimated using alkaline 0.32% KMnO_4 . It oxidizes organic matter present in soil and liberates nitrogen in an ammonical form, which is

Table 3 Common techniques/methods for an impact assessment related to natural resource management

Method	Used in environmental impact assessment activities
pH (soil-to-water ratio, 1:2.5)	Jackson [27]
EC (soil-to-water ratio, 1:2.5)	Jackson [27]
Texture	Bouyoucos [28]
Organic carbon	Walkley and Black [29]
Available phosphorus	Olsen et al. [30, 31]
Available nitrogen	Subbiah and Asija [32]
Available potassium	Jackson [27]
DTPA extractable micronutrient	Lindsay and Norvell [33]
Ca + Mg	Singh et al. [26]
DHA	Casida et al. [34]
Alkaline and acidic phosphatases	Tabatabai and Bremner [35]
FDA	Adam and Duncan [36]

DHA dehydrogenase activity, DTPA diethylenetriamine penta-acetic acid, EC electrical conductivity, FDA fluorescein diacetate

absorbed in 2% boric acid solution. The boric acid is back-titrated with 0.02 *N* H₂SO₄. This method was first described by Subbiah and Asija [32].

Available Phosphorus

Worldwide, the method described by Bray and Kurtz [31] is used to measure phosphorus in soil with a lower pH soil (acidic conditions), and that described by Olsen et al. [30] used in soil with a higher pH. In the first method, the soil extraction solution comprises 0.03 *N* NH₄F and 0.025 *N* HCl, whereas the method by Olsen et al. uses a 0.5 *M* NaHCO₃ solution at pH 8.5 during analysis.

Available Potassium

Potassium is also a major plant nutrient that determines the crop yield potential of a cultivar. It also acts against insect pests that attach to crops. It is measured in soil using the method described by Hanway and Heidel [38], using 1 *N* ammonium acetate (pH 7.0) as a extractant.

Micronutrients

Most cationic micronutrients (Cu, Ni, Zn, Mn, Fe) are analyzed using the method described by Lindsay and Norvell [33]; boron, by a hot water method [39]; and molybdenum, based on work by Singh et al. [26].

Assessment of Heavy Metals

Estimation of Available Concentration of Metals

The available concentration of a metal is directly responsible for its toxicity. In terms of plant nutrients, it is an index of soil fertility and affects crop yields. In the case of heavy metals, their concentrations show the negative potential of a metal to effect

crop growth and development [40]. A range of extractants are used to estimate metals in soil. Among these, diethylenetriamine penta-acetic acid is a common extractant used in most soil laboratories to measure available metal in soils. A known quantity of soil is extracted using diethylenetriamine penta-acetic acid (DTPA) with shaking for 2 h. The main precaution during this extraction is that the soil to extractant ratio must be 1:2. After filtration, the concentrations of available metals in the soil are measured through the use of inductively coupled plasma optical emission spectrometry (ICP-OES).

Digestion for Total Metal Estimation

The total concentrations of metals in soil samples are used to compute various indices related to metal accumulation as a result of anthropogenic activities. A known amount of soil is digested with aqua regia, a diacid mixture, or a triacid mixture. During digestion, 5 mL HNO_3 is added to the soil samples for predigestion and kept overnight. Then, 10 mL aqua regia ($\text{HCl}:\text{HNO}_3$ [3:1]), a diacid ($\text{HNO}_3:\text{HClO}_4$ [9:4]), or a triacid ($\text{HNO}_3:\text{H}_2\text{SO}_4:\text{HClO}_4$ [10:1:4]) is added for proper digestion and kept on a hot plate until proper digestion occurs. The white color of the digested samples indicates proper digestion. The sample is removed from the hot plate and the final volume is prepared. The concentrations of metals are measured using ICP-OES.

Heavy Metals Measurement by ICP-OES

A range of instruments and methods are available for determining heavy metal concentrations in ecosystem components. The most common is ICP-OES (Fig. 3). This process uses a fourth state of matter—that is, plasma—with a temperature of 6000–7000 kelvin (K); the inner core, known as the induction zone, can reach a temperature up to 10,000 K. The liquid sample is inserted into the ICP-OES using



Fig. 3 Inductively coupled plasma optical emission spectrometry

peristaltic pump through a nebulizer into a spray chamber. Argon gas plays a crucial role in the generation of plasma, which occurs at the end of a quartz torch. During this process, a high-frequency alternating current and a water-cooled induction coil are properly controlled by the instrument. The injected analyzing solution from a nebulizer and electrons of the metal take thermal energy and reach a higher, “excited” state. After a few moments, electrons reduce their thermal energy and revert to the original ground state and liberate a particular spectrum. Each metal has its own spectrum. Using Echelle grating, a prism and focusing mirror capture the emitted spectrum through a charge-coupled device.

This instrument has very low chemical interference and is reproducible. More than 70 metals and earth metals can be measured with ICP-OES. Simultaneous and sequential analysis of multiple elements is possible, with high sensitivity. The running and maintenance costs of the instruments are high, and skilled manpower is required during the analysis.

Assessment of Soil Biological Activities

Assessment of Soil Enzymatic Activities

Soil enzymatic activities affect soil fertility and the capacity of soil to supply essential plant nutrients. Different types of soil enzymes are produced by microorganisms in the soil and are mediated by the external application of agricultural inputs and contaminants. Soil enzymes take part in initiating and maintaining biogeochemical cycles of plant nutrients and provide the basic support for fertility to achieve the healthy development of plants [41]. They also increase the reactive area to decompose soil organic matter and detoxify trace metals [42]. The addition of organic residue in contaminated soil reduces the soil’s pH and enhances the nutrient availability [43–47]. Decomposition of organic matter releases various types of organic acids and improves soil health [48–51]. During decomposition, microbes take carbon from crop residue and multiply their population, producing a significant amount of organic acids [52, 53]. However, a higher concentration of organic acids enhances soil aggregation and crop productivity. Different tools and techniques are used for impact assessments of organic residues on soil enzyme activities and carbon sequestration; these are key researchable issues in agriculture [54–58]. Some important soil enzymes are listed in Table 4 [59, 60].

Environmental Impact Assessment in Relation to Soil Biodiversity

Soil biodiversity is a main force that affects nutrient dynamics, degradation of various contaminants, and secretion of plant growth-promoting substances in a soil system [61]. Soil biodiversity refers to a variety of species within the soil ecosystem that form a web of biological activities. Increasing the contaminants in a soil affects biodiversity in terms of types and populations of soil biota [62]. Soil biodiversity is mainly a count of fungi, bacteria, actinomycetes, and other small animals in a soil. During an environmental impact analysis, soil diversity is studied to identify precisely the direct and indirect effects of contaminants on soil microbial

Table 4 Soil enzymes involved in nutrient dynamics

Enzyme	Substances acted on in organic matter	End product	Biological significance	Predictor of soil function	
Enzyme for hydrogen transfer to a variety of substrates					
Dehydrogenase	Hydrogen-donating and -accepting species (CO ₂ , organic acids, alcohols, etc.)	Oxidized or reduced products	Microbial electron transport system, proton release	General index of microbial activity	
Enzyme for decomposition of carbon substrates					
α-Amylase	Starch	Dextrins	Release of Carbon compounds required for growth of microbes	OM decomposition	
β-Amylase	Starch	Maltose		OM decomposition	
Maltase	Maltose	Glucose		OM decomposition	
Cellulase					
Endocellulase	Cellulose	Cellodextrins		OM decomposition	
Exocellulase	Cellulose	Cellobiose, celotriose		OM decomposition	
β-Glucosidase	Cellobiose, celotriose	Glucose		General index of microbial activity	
Pectinase	Pectin	Oligosaccharides		OM decomposition	
Lignases					
Lignin peroxidase	Lignin	Partially depolymerized lignin		OM decomposition	
Manganese peroxidase	Lignin	Partially depolymerized lignin	OM decomposition		
Laccase	Lignin	Mixture of aliphatic and aromatic polymers	OM decomposition		
Enzymes for decomposition of nitrogen substrates				Nutrient cycling	
Protease	Protein	Amino acids	Nitrogen source for soil microbes		
Deaminase or ammonia lyase	Amino acids	Ammonia + organic acids	Nitrogen nutrition of plants and soil microbes		
Nitrate reductase	Nitrates	Ammonia			
Amidase	Nitrogen-carbon bonds of	Ammonium (NH ₄)			

(continued)

Table 4 (continued)

Enzyme	Substances acted on in organic matter	End product	Biological significance	Predictor of soil function
	nonproteinaceous organic nitrogen compounds			
Urease	Urea	Ammonia and carbon dioxide		
Enzymes for decomposition of phosphorus substrates				Nutrient cycling
Acid phosphomonoesterase	Organic phosphorus compounds	Orthophosphate	Phosphorus source for plants and soil microbes	
Alkaline phosphomonoesterase	Organic phosphorus compounds	Orthophosphate		
Phosphodiesterases	Nucleic acids and other organic phosphorus compounds with phosphate interlinks	Orthophosphate		
Phytases	Phytin	Orthophosphate		
Enzyme for decomposition of sulfur substrates				Nutrient cycling
Sulfatase	Organic sulfur compounds like sulfur-containing amino acids	Sulphate	Sulfur source for plants and soil microbes	Nutrient cycling

diversity. Two methods are used in practice: (1) viable and (2) culturable. In this era of technological advancement, various molecular tools and techniques are used to extract and measure nucleic acids in soil systems. These save time, are more accurate, and easily identify the species, genera, families, or higher taxonomic classes in complex soils. These tools are important in categorizing microbial variability within species to diversity of communities [63]. Some of the processes involved in the identification of soil biodiversity are molecular tools and techniques.

G:C Ratio

The G:C ratio is the ratio of percentages of guanine (G) and cytosine (C) in DNA. It is measured using chromatographic methods. The G:C ratio of prokaryotes commonly falls between 20% and 80%.

DNA Hybridization

The G:C ratio does not clearly identify the nucleotide sequence of DNA. Hybridization measures the degree of sequence similarity between two DNAs. In this technique, the DNA of an organism is isolated and mixed with a radioactive material

(^{32}P or ^3H), then denatured through heating and mixed with the DNA of another organism, which is not radioactive. The technique is too sensitive to identify the genera of the two organisms [64].

Ribotyping

Ribotyping is useful for measuring the unique sequence generated during digestion of the DNA of a particular organism. Here, different restriction enzymes are used for different patterns during the ribosomal RNA probe.

Multilocus Sequence Typing

Multilocus sequence typing helps to identify the sequence of the same gene set from different strains of a particular organism.

Ribosomal Intergenic Spacer Analysis

Ribosomal intergenic spacer analysis is involved in polymerase chain reaction (PCR) amplification of a particular part of the intergenic space situated between the 16S and 23S ribosomal subunits.

Random Amplified Polymorphic DNA

This is good practice for using molecular markers to identify genetic diversity. In this technique, a random amplified polymorphic DNA band generated by an organism is compared with that of another organism by following a similar procedure of identification.

Length Heterogeneity PCR

This method is similar to terminal restriction fragment-length polymorphism except the latter detects variation in amplicon length during restriction digestion. It also works on the basis of natural polymorphisms that occur as a result of mutation within a gene [65].

Metagenomic Approach

This has helped to identify the collective genomes of an entire community within a particular boundary. It is an emerging field of research known as community genomic or environmental genomics. It was first used by Handelsman et al. [104] at the University of Wisconsin. It is now popular in advanced research laboratories across the world.

Regulation with Respect to Heavy Metal Entry into Agricultural Land

Most developing countries use contaminated waste water for irrigation in periurban areas [66–69]. Waste water has huge water potential and contains organic carbon and plant nutrients for plant growth [70]. The long-term application of contaminated water has accumulated significant amounts of toxic metals and contaminated loads in

Table 5 Maximum allowable concentrations of trace elements in soils of various countries (mg kg^{-1})

Countries	Cd	Cr	Cu	Ni	Pb	Zn	As	Hg
Austria	5	100	100	100	100	300	50	5
Poland	1–3	50–80	30–70	30–75	70–150	100–300	30	5
Germany	1.5	100	60	50	100	200	20	1
Russia	–	0.05 (Cr^{6+})	23 (soluble pool)	35	20	110	2	2.1
United Kingdom	3–15	–	50	20	500–2000	130	10	–
United States	20	1500	750	210	150	1400	–	8
European Commission	1–3	50–150	50–140	30–75	50–300	150–300	–	1–1.5

soils [45], which adversely affect the carbon mineralization rate and enzymatic activity in the soil. Application of chromium at more than 10 mg kg^{-1} through $\text{K}_2\text{Cr}_2\text{O}_7$ reduced the mineralization rate in Vertisol in central India [71]. Chaney et al. [105] reported that application of Cd in soil reduced the microbial respiration rate and carbon mineralization rate in black oak forest soils. Application of Cr in soil reduced the activities of soil enzymes such as DHA, acid and alkaline phosphatases [71]. The cultivation of crops in metal-contaminated soils reduced the percentages of germination and root and shoot growth in wheat [72] and pigeon pea [73]. The concentration of metals reduces the secretion of root exudates from and affects the kinetics of nutrient uptake by plants [74]. Rhizodeposition is a boon for the plant growth, and it enhances the macro- and micronutrient efficiency in soil [75, 76]. These low-molecular organic acids provide a source of carbon for soil microorganisms [77–79]. Some microbes degrade toxicants into nontoxic forms, reducing toxicity [62]. Enhanced soil microbial population and diversity is mostly responsible for nutrient mineralization, contaminant degradation, and plant growth-promoting substance secretion in soil. Kabata-Pendias [80] described the maximum allowable metal concentrations in soil, by country (Table 5).

Assessment of Metal-Contaminated Soils

Metal Enrichment Factor

Heavy metal accumulation in soil causes deterioration of soil fertility and reductions in soil biota. Increasing concentrations in soil can be quantified by the metal enrichment factor (MEF), which identifies and quantifies anthropogenic interference in natural cycles of metal elements. Through the use of the MEF, a ratio is computed between the elevated level of metal in the soil, as caused by human interference, and the geogenic concentration [81–84]. The concentration of a particular metal in the soil and earth's crust, the geogenic concentration acts as a reference metal concentration when calculating the MEF. In most cases, Al, Sc, Ti, Fe, Mn, and Sr are used

Table 6 Metal enrichment factor values with respect to contamination

MEF Class	Interpretation
<2	Deficient mineral enrichment
2–5	Moderate enrichment
>5–20	Significant enrichment
>20–40	Very high enrichment
>40	Extremely high enrichment

to compute the MEF because of their higher concentrations in the earth's crust and because they are not easily altered by human activities [85, 86]. As an example, an MEF can be calculated using aluminum [Al] as the reference, with the following relation:

$$\text{MEF} = \frac{\left(\frac{X}{\text{Al}}\right)_{\text{sample}}}{\left(\frac{X}{\text{Al}}\right)_{\text{crust}}}$$

where $\left(\frac{X}{\text{Al}}\right)_{\text{sample}}$ and $\left(\frac{X}{\text{Al}}\right)_{\text{crust}}$ refer to the ratio of the mean concentration (milligrams per kilogram) of the target metal (X) and that of aluminum in the sediment and continental crust, respectively. On the basis of computed values, MEFs are categorized into five major classes [86] (Table 6).

Geo-accumulation Index

The geo-accumulation index (I_{geo}) depicts metal contamination currently with respect to preindustrial concentrations. To calculate I_{geo} , a control field concentration is used as a reference value, or a preindustrial value can be used. This index is simple and it is easy to calculate metal accumulation over a period of industrial or anthropogenic activities.

$$I_{\text{geo}} = \log_2 \frac{C_n}{1.5B_n}$$

where C_n represents the concentration of a particular metal in the soil, whereas B_n denotes the concentration of that same metal in the earth's crust [87]. The metal concentration accepted as the background is multiplied by the constant 1.5 in order to take into account natural fluctuations of a given substance in the environment and a very small anthropogenic influence (Table 7).

Case Study

Kanpur is a city of leather: more than 500 small and medium-sized tanneries are located in and around the city. Local peoples living in this area are perform tanning processes within their households. A huge volume of tannery effluent is generated and, after undergoing certain treatments, this is used for irrigation to produce crops.

Table 7 Geo-accumulation index values with respect to contamination [84, 88]

I_{geo} value	Class	Interpretation
<0	I	Practically uncontaminated/unpolluted
0–1	II	Uncontaminated/unpolluted to moderately contaminated/polluted
>1–2	III	Moderately contaminated/polluted
>2–3	IV	Moderately to heavily contaminated/polluted
>3–4	V	Heavily contaminated/polluted
>4–5	VI	Heavily to very heavily contaminated/polluted
>5	VII	Very heavily contaminated/polluted

Table 8 Nemerow pollution index (P_n)

Class	Index	Interpretation
I	$P_n < 0.7$	Safety domain
II	$0.7 \leq P_n < 1.0$	Precaution domain
III	$1.0 \leq P_n < 2.0$	Slightly polluted domain
IV	$2.0 \leq P_n < 3.0$	Moderately polluted domain
V	$P_n > 3.0$	Seriously polluted domain

It contains the toxic metal chromium, which is known to be carcinogenic human beings [89]. Long- term application of the effluent accumulated a significant amount of total chromium in the soil and affected soil fertility and crop productivity. A geo-referenced survey was conducted to assess the effluent's impact on soil metal content, and soil and effluent samples were collected. Samples were analyzed using ICP-OES after digestion with aqua regia. The analyzed data fit into the I_{geo} and were categorized into various classes of contamination. The results indicated that chromium accumulation in the soil was 28–30 times more than in soils irrigated with a tube well [90]. The concentration of chromium in the effluent was less, but long- term irrigation deposited a meaningful amount in the soil. The I_{geo} was calculated per the equation developed by Muller [91]:

$$I_{geo} = \log_2 \frac{C_n}{1.5B_n}$$

where C_n is the trace metal concentration in soil and B_n is the geochemical baseline.

In this study, agricultural fields irrigated with tannery effluent was classified as class VII, meaning it was heavily to extremely contaminated/polluted [92].

Nemerow Pollution Index

The Nemerow pollution index is mainly used in cases of single-metal contamination due to mining or industrial activities in a particular area. It provides the degree of soil ecological pollution and a potential assessment of ecosystem function and quality [93–95].

It is categorized into five classes (Table 8).

Table 9 Modified Nemerow pollution index (P_n) (Guan et al. [93])

Class	Index	Interpretation
0	$0 < P_n < 0.5$	Uncontaminated
I	$0.5 < P_n \leq 1.0$	Uncontaminated to moderately contaminated
II	$1.0 < P_n \leq 2.0$	Moderately contaminated
III	$2.0 < P_n \leq 3.0$	Moderately to heavily contaminated
IV	$3.0 < P_n \leq 4.0$	Heavily contaminated
V	$4.0 < P_n \leq 5.0$	Heavily to extremely contaminated
VI	$P_n > 5.0$	Extremely contaminated

Table 10 Limits for contamination indices

Contamination limits	Contamination index (P_i)	Integrated contamination index (P_c)
None	$P_i \leq 1$	$P_c \leq 0$
Low	$1 \leq P_i \leq 2$	$0 < P_c \leq 7$
Moderate	$2 \leq P_i \leq 3$	$7 < P_c \leq 21$
High	$P_i > 3$	$P_c > 21$

Guan et al. [93] later modified the index and replaced single-metal contamination with the I_{geo} , per the formula:

$$P_n = \sqrt{\left(I_{geomax}^2 + I_{geoave}^2\right)/2}$$

where, I_{geomax} is the maximum I_{geo} value and I_{geoave} is the arithmetic mean of I_{geo} . This modification is described in Table 9.

Integrated Contamination Index

The integrate contamination index (P_c) is also an important index for assessing contamination load in natural systems. It defined as the sum of 1 + the difference between the contamination index (P_i) for each metal:

$$P_C = \sum_n^{i=1} (P_i - 1)$$

The mean and standard deviation of P_c were used to establish confidence intervals. Accordingly, a $P_c \times$ distance curve was adjusted using a log scale for real distance. Terminology to describe the contamination index (P_i) and integrated contamination index (P_c) is defined in Table 10.

Assessment of Heavy Metal Content in Plant Parts

For heavy metals analysis, a diacid mixture ($\text{HNO}_3:\text{HClO}_4$ at the ratio 9:4) was used to digest plant samples [96], and metals present in the plant extractant were analyzed using ICP-OES.

For example, crop plant parts contaminated by chromium were harvested and various ratios of metal accumulation were computed. Parameters such as bioconcentration factor (BCF), translocation factor (TF), translocation efficiency (TE), and crop removal capacity could be computed as described below.

- i. BCF defines the capacity of the plant to remove contaminants and is calculated by the formula [97]

$$BCF = \frac{C_{\text{harvested tissue}}}{C_{\text{soil}}}$$

where, $C_{\text{harvested tissue}}$ is the concentration of chromium in the harvested plant parts (roots, shoots), and C_{soil} is the concentration of chromium in the soil after respective treatments.

- ii. TF refers to the transfer of chromium metal ions from root to shoot and is quantified by the formula proposed by Adesodun et al. [98]:

$$TF = \frac{C_{\text{shoot}}}{C_{\text{root}}}$$

where C_{shoot} and C_{root} are the concentrations of chromium in the root and shoot, respectively.

- iii. TE was calculated from the formula described by Meers et al. [99]:

$$TE (\%) = \frac{C_{\text{content in shoot}}}{C_{\text{content in whole plant}}} \times 100$$

- iv. Chromium removal represents the capacity of the crop to remove chromium and was calculated with the formula:

$$Cr \text{ removal } (\%) = \frac{\text{Total Cr uptake by plant}}{\text{Total Cr applied to soil}} \times 100$$

Urban Solid Waste: Use and Management

Because of limited natural resources, a pressing need exists for increased grain production to fill the hungry mouths of the growing population. Soil and water are both important natural resources and affect the quality of the environment and ecological services on earth. Maintaining the quality of natural resources and

producing the necessary quantities of grains are big challenges to researchers and policymakers. The quantity of waste and the amount generated per capita within urban areas are increasing, posing a considerable and quickly growing challenge in their management. Long-term application of municipal/industrial effluents for irrigation or of composts prepared from municipal waste for crop production contaminated the soil and water bodies in agricultural areas [100, 101]. These waste materials contain huge volumes of organic matter and essential plant nutrients [66, 102]. However, they also contain contaminants in the form of heavy metals, microbes, organic and inorganic material, and so on. The introduction of toxic chemicals into soils has disturbed natural ecological processes, threatening the agro-ecosystem.

Collection and Disposal of Municipal Solid Waste

Solid Waste Collection and Disposal

Solid waste collection and disposal are very tough practices in developing countries like India. In ideal solid waste management, one of the cheapest and common waste management practices in many parts of the world is land filling. The majority of Indian cities collect waste through community bins and dump them in landfill areas; a small fraction of uncollectible municipal solid waste (MSW) lies along roadsides.

Compost Preparation

Composting is an environmentally friendly method of MSW disposal and allows further wealth to be generated from the waste. In this process, collected MSW needs to be properly segregated to obtain the biodegradable fraction, which is composted and, after decomposition, disposed of in agricultural land. The biodegradable fractions should be used to prepare the compost. The presence of heavy metals is a big challenge to its use in agricultural crop production systems. A team headed by Dr. J. K. Saha, of the ICAR-Indian Institute of Soil Science in Bhopal, India, prepared MSW composts that had been collected from 34 manufacturers in 29 cities in India. On the basis of fertilizer index and clean index, the team categorized their potential use in agricultural crops and forestry plants.

Categorization of MSW Compost

After preparing the compost from MSW, it should be analyzed for heavy metal content and plant nutrient concentrations. Both parameters are very widely because of the variation in MSW generation from different cities.

Fertilizer Index

The manurial (including plant nutrients) in MSW composts can be calculated on the basis of the fertilizer index using the formula:

Table 11 Criteria for assigning the weighing factor to fertility parameters and the score value to analytical data

Parameters	Score value (S_i)					Weighing factor (W_i)
	5	4	3	2	1	
Total organic carbon (%)	>20.0	15.1–20.0	12.1–15.0	9.1–12.0	<9.1	5
Total nitrogen (%)	>1.25	1.01–1.25	0.81–1.00	0.51–0.80	<0.51	3
Total phosphorus (%)	>0.60	0.41–0.60	0.21–0.40	0.11–0.20	<0.11	3
Total potassium (%)	>1.00	0.76–1.00	0.51–0.75	0.26–0.50	<0.26	1
Carbon-to-nitrogen ratio	<10.1	10.1–15	15.1–20	20.1–25	>25	3
Respiration activity (mg CO ₂ -C/g VS d)	<2.1	2.1–6.0	6.1–10.0	10.1–15	>15	4

$$\text{Fertilizer index} = \frac{\sum_{i=1}^n S_i W_i}{\sum_{i=1}^n W_i}$$

where S_i is the score value of analytical data and W_i is the weighing factor of the i th fertility parameter (Table 11).

Clean Index

The clean index indicates the threat caused by the trace metals in MSW compost with respect to phytotoxicity and mammalian toxicity. The higher the value of the clean index, the lower the potential toxicity of the MSW compost (Table 12). Composts prepared from MSW from various cities are analyzed for heavy metal concentrations and were assessed for toxicity using the formula

$$\text{Clean index} = \frac{\sum_{j=1}^n S_j W_j}{\sum_{j=1}^n W_j}$$

where S_j is the score value of analytical data and W_j is weighing factor of the j th heavy metal.

Practical Utility of MSW Compost on the Basis of Fertilizer Index and Clean Index

While the fertilizer index can be taken as a measure of nutrient supply potential, the clean index can be used by regulatory authorities to restrict the entry of heavy metals into sensitive components of the environment, such as agricultural land and water

Table 12 Criteria for assigning a weighing factor to heavy metal parameters and a score value to analytical data

Heavy metal (mg kg ⁻¹)	Score value (S _j)						Weighing factor (W _j)
	5	4	3	2	1	0	
Zn	<151	151–300	301–500	501–700	701–900	>900	1
Cu	<51	51–100	101–200	201–400	401–600	>600	2
Cd	<0.3	0.3–0.6	0.7–1.0	1.1–2.0	2.0–4.0	>4.0	5
Pb	<51	51–100	101–150	151–250	251–400	>400	3
Ni	<21	21–40	41–80	81–120	121–160	>160	1
Cr	<51	51–100	101–150	151–250	251–350	>350	3

Table 13 Classification of MSW composts for their marketability and use in agricultural crop production systems

Class		Fertilizer index	Clean index	Quality control compliance	Overall quality and area of application
Marketable classes	A	> 3.5	> 4.0	Complying for all heavy metal parameters	Best quality: High manurial value and low heavy metal content Can be used for high-value crops and in organic farming
	B	3.1–3.5	> 4.0		Very good quality: Medium fertilizing potential and low heavy metal content
	C	> 3.5			Good quality: High fertilizing potential and medium heavy metal content
	D	3.1–3.5	3.1–4.0		Medium quality: Medium fertilizing potential and medium heavy metal content
Restricted classes	RU-1	< 3.1	–	Complying for all heavy metal parameters	Low fertilizing potential but safe for the environment Can be used as a soil conditioner
	RU-2	≥ 3.1	> 4.0		Can be used to grow nonfood crops Requires periodic monitoring of soil quality if use repeatedly
	RU-3	≥ 3.1	≤ 4.0		Can be used only to develop lawns/gardens, or tree plantations in forestry (with one-time application)
The samples not falling under the above marketable and restricted use classes are not suitable for agricultural land application but can be used to rehabilitate degraded land.					

bodies. On the basis of fertilizer index and clean index values, the following scheme to classify MSW composts has been proposed for use in different applications and to determine their suitability as a marketable product (Table 13).

The Indian Institute of Soil Science, Bhopal, India, has developed software for use by compost manufacturers and quality control officials that guides analysts using methods of analysis of MSW composts, evaluates their quality, and recommends safe and beneficial areas of application. This software can be downloaded free from the Indian Institute of Soil Science website at <http://www.iiss.nic.in/Software.html>.

Conclusions

Environmental impact assessment is necessary to measure environmental consequences (positive and negative) for the execution, planning, formulation of policy, organization of events, and start of project activities for proposed actions in a particular area. The inventories for the impact assessment of natural resources are key to agricultural production systems. Precise and well-timed environmental impact analysis can minimize losses due to natural calamities. Frequent changes in weather events creates a lot of ambiguity in the execution and proper planning of agricultural production policies and research. A number of tools and techniques are involved in the assessment of soil fertility and contamination and the measurement of GHGs. These require a high initial investment as well as specific skills for particular impact assessments. Presently, environmental impact assessments are more valuable for developing countries because of their limited natural resources with respect to huge population pressure.

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