Proceedings of the International Plant Sulfur Workshop

Luit J. De Kok Malcolm J. Hawkesford Silvia H. Haneklaus Ewald Schnug *Editors* 

Sulfur Metabolism in Higher Plants -Fundamental, Environmental and Agricultural Aspects



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# Sulfur Metabolism in Higher Plants - Fundamental, Environmental and Agricultural Aspects



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Heinz Rennenberg

# Foreword: There Is (Almost) No Way to Escape from Sulfur in Plant Research

This Sulfur Workshop in Plants series was initiated during a sabbatical stay in Groningen in 1988 as a result of discussions between Ineke Stulen, Luit De Kok, and myself. Therefore, it was logical that the first Sulfur Workshop on plants was held in Groningen in 1989. At this time, plant research on sulfur was largely focused on consequences of atmospheric pollution with  $SO_2$  for plant growth and development, and studies on other aspects of sulfur metabolism in plants were not well developed (e.g., Rennenberg 1982) compared to studies in mammals (Meister and Anderson 1983). It is the merit of the highly successful first workshop on "Sulfur Nutrition and Sulfur Assimilation in Higher Plants" (subsequently abbreviated "Sulfur Workshop") held in Groningen that the door became wide open for studies on sulfur metabolism of plants including fundamental and applied aspects. In particular, it was the European plant science community that took advantage of this situation and soon played a leading role in this area of research.

When the first Sulfur Workshop was held in Groningen, I already could look back to more than a decade of studies on sulfur in plants. In my diploma and PhD thesis, I had worked on glutathione production in tobacco suspension cultures, a system that subsequently became recognized as a useful tool for in-depth analyses of glutathione synthesis and degradation in plants (Bergmann and Rennenberg 1993). At this time, it was also established that plants are not only a sink for atmospheric sulfur compounds but are also able to emit volatile sulfur compounds into the atmosphere (Rennenberg 1991). This new view of a bidirectional flux of sulfur between plants and the atmosphere initiated numerous studies on sulfur metabolism in terrestrial and aquatic plants that included volatile products.

Despite the multitude of valuable results obtained by studies with tissue cultures, it became obvious that sulfur compounds undergo long-distance transport (Bonas et al. 1982) and that regulation from the cellular scale-up to the seasonal dynamics of sulfur in plants requires studies at the whole-plant level (Herschbach and Rennenberg 1997; Herschbach et al. 2012). The significance of such studies was

fueled by the generation of transgenic poplar plants with modified glutathione synthesis and reduction capacity (Noctor et al. 1998). With these tools, sulfur in plants entered the era of molecular biology. Molecular research on sulfur became soon extended to the analyses of sulfate transporters (Smith et al. 1997; Hawkesford et al. 2003) and the cross talk of sulfur metabolism with nitrogen and carbon metabolism (Kopriva and Rennenberg 2004) that constitute important areas of plant research until today. The current view on molecular studies has changed dramatically from the initial approaches that were focused on analyses of transcription of a set of enzymes and transporters: today, it is generally accepted that the characterization of metabolic processes and metabolic cross talk requires more than quantification of the transcriptome and largely relies on an integrative view on mRNA, protein, and metabolite abundances, as well as metabolite fluxes (Rennenberg and Herschbach 2014; Kalloniati et al. 2015).

Transgenic poplars with modified glutathione synthesis and reduction capacity became a useful tool to analyze the role of sulfur metabolism in the compensation of abiotic and biotic stress. The multiple stress compensation reactions relying on sulfur metabolism include reactive oxygen species (ROS) scavenging, heavy metal detoxification, and hypersensitive responses to pathogen attack (Noctor et al. 1998; Fover and Rennenberg 2000; Peuke and Rennenberg 2006; Rennenberg and Herschbach 2012; He et al. 2015). Among the stress factors studied, the consequences of SO<sub>2</sub> exposure of plants turned out to be of particular complexity, but surprisingly, this was only recognized in recent years (Hänsch et al. 2007). As a radical, SO<sub>2</sub> can mediate ROS formation that requires reduced sulfur in the form of glutathione for scavenging in the Foyer-Halliwell-Asada cycle (Foyer and Rennenberg 2000); if involved in ROS formation in the apoplastic space, SO<sub>2</sub> can interact with lignin formation and requires scavenging by peroxidase activity for ROS homeostasis (Hamisch et al. 2012); as an essential intermediate, sulfite derived from SO2 interacts with assimilatory sulfate reduction and, as a product of this pathway, also with sulfur nutrition (Rennenberg 1984; Takahashi et al. 2011; Herschbach et al. 2012). Even subsequent to oxidative and reductive SO<sub>2</sub> detoxification (Hamisch et al. 2012), the detoxification products, i.e., sulfate and sulfide, will interact with signaling processes at the cellular and whole-plant level (Leitner et al. 2009; Lisjak et al. 2010; García-Mata and Lamattina 2013; Hancock and Whiteman 2014; Calderwood and Kopriva 2014). This was indicated already by early H<sub>2</sub>S fumigation studies with different plant species (De Kok et al. 1991; Herschbach et al. 1995a, b, 2000) but only recently connected to the consequences of SO<sub>2</sub> exposure (Hamisch et al. 2012). In the research area of sulfur-mediated signaling, the proposed role of sulfate as a root-to-shoot signal controlling stomatal aperture upon drought (Malcheska et al. 2017) provides a new notion, why excess sulfur in the form of sulfate has to be sequestered in the vacuole but also needs to be mobilized from this pool under particular environmental conditions, processes that were already observed in early studies on sulfur in plants (Rennenberg 1984).

Over the years, I made several attempts to escape from sulfur, e.g., by focusing on N and P nutrition, on radiatively active biogenic trace gases in the atmospheres, and recently on plant carnivory. At the end, (almost) all of these attempts ended up with studies on sulfur in plants. Even for plant carnivory, sulfur metabolism turned out to be of pivotal importance. In the Venus flytrap, a plant that actively catches its animal prey with snap traps, the prey is digested in hermetically closed traps by the release of an acidic, sulfur-rich enzyme cocktail from the gland-based secretory system on the inner surface of the traps (see Fasbender et al. 2017 and literature cited therein). The production of the sulfur-rich hydrolytic enzyme cocktail requires stimulation of thiol synthesis from assimilatory sulfate reduction during prey digestion (Scherzer et al. 2017). Thus, whatever process in plants I found interesting to study (almost) always ended up to be connected to sulfur. Therefore, finally, sulfur in plants accompanied me from my diploma, PhD, and habilitation thesis to my position as research associate at the University of Cologne and the DOE Plant Research Laboratory at Michigan State University in East Lansing, and to my professor positions at the University of Cologne, the Technical University of Munich, the Fraunhofer Institute of Atmospheric Environmental Research, and the University of Freiburg up to my retirement in 2017.

The broad range of different aspects of sulfur metabolism could only be studied in more than 40 years of my research activities in collaboration with numerous colleagues and friends, including, among many others (in alphabetical order), Ludwig Bergmann, Christian Brunold, Luit J. De Kok, Manolis Flemetakis, Christine Foyer, Dieter Grill, Robert Hänsch, Rainer Hedrich, Rüdiger Hell, Stanislav Kopriva, Ralf R. Mendel, Andrea Polle, Winfried Rauser, Kazuki Saito, Andreas Weber, and Marcus Wirtz that in several cases stayed in my group for a period of time. In addition, work on sulfur in plants in my group would not have been possible without additional strong partners such as Cornelia Herschbach, Jürgen Kreuzwieser, and Monika Eiblmeier who have accompanied me in my research at the University of Freiburg until today. It is time to thank them all for creating such a fruitful and pleasant working atmosphere.

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### Preface

The International Plant Sulfur Workshop series was initiated in order to bring together scientists from various research disciplines and to discuss all aspects of sulfur metabolism, from molecular biology, biochemistry, and physiology to ecology and agriculture. The first workshop in the series entitled "Sulfur Nutrition and Sulfur Assimilation in Higher Plants: Fundamental Environmental and Agricultural Aspects" was held in Haren, the Netherlands, 1989. The following workshops were held in Garmisch-Partenkirchen, Germany, 1992; Newcastle upon Tyne, United Kingdom, 1996; Wengen, Switzerland, 1999; Montpellier, France, 2002; Kisarazu, Chiba, Japan, 2005; Warsaw, Poland, 2008; Creswick, Victoria, Australia, 2010; and Freiburg-Munzingen, Germany, 2014. Contents of the respective proceedings are included in this volume.

This proceedings volume contains a selection of invited and contributed papers of the 10th Jubilee Plant Sulfur Workshop, which was held in Goslar, Germany, from September 1 to 4, 2015. During this workshop, the outcome of the previous workshops was summarized, and the still existing gaps and prospects for future research were highlighted by a selection of speakers who have significantly contributed to plant sulfur research during the last 25 years.

We are delighted to dedicate this volume to our dear colleague Heinz Rennenberg from the University of Freiburg, Germany, who together with Ineke Stulen, Christian Brunold, and Luit J. De Kok initiated the workshop series and furthermore was involved in the organization and issuing of proceedings volumes of all previous plant sulfur workshops. In addition, he has significantly contributed to the understanding of the regulation of uptake and assimilation of sulfur and the significance of sulfur metabolites in stress tolerance of higher plants over more than four decades thus leaving a durable "sulfur footprint".

Groningen, The Netherlands Braunschweig, Germany Harpenden, UK Braunschweig, Germany Luit J. De Kok Silvia H. Haneklaus Malcolm J. Hawkesford Ewald Schnug

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# Part I Overview Papers

# Sulfate Transport in Plants: A Personal Perspective

#### Malcolm J. Hawkesford

Abstract Early key research milestones for sulfate transport in plants include the first description of kinetics of sulfate uptake into plant roots (Leggett and Epstein, Plant Physiol 31:222–226, 1956), nutritionally regulated sulfate uptake into plants (Clarkson et al., J Exp Bot 34:1463-1483, 1983), and the first gene for a plant sulfate transporter (Smith et al., Proc Natl Acad Sci U S A 92:9373–9377, 1995a). Since then a well-described gene family encoding putative sulfate transporters has been characterized in multiple species, initially most notably in Arabidopsis but subsequently for a number of other models or important crops (examples: Brassica, wheat, rice, poplar and Medicago, see Buchner et al., Genome 47:526-534, 2004a; Buchner et al., Plant Physiol 136:3396–3408, 2004b; Buchner et al., Mol Plant 3:374-389, 2010; Kumar et al., Plant Signal Behav 10:e990843, 2015; Dürr et al., Plant Mol Biol 72:499-517, 2010; Gao et al., Planta 239:79-96, 2014). Regulation of expression has been well documented and this regulation is both a useful marker of sulfur-nutritional status and a model for the elucidation of control pathways. The complexity of the gene family in relation to functional, regulatory and spatial distribution indicates an apparent whole plant management system for sulfur, coordinated with growth and demand and interacting with nutrient availability. In addition to sulfate, there is direct involvement of this transporter family in the uptake and accumulation of both selenate and molybdate, with clear consequences for nutritional quality. Is the story now complete almost 60 years since the first transport description and 20 years since the first sulfate transporter gene isolation, and a plethora of research projects and publications? Do we know how sulfur is acquired and appropriately distributed within the plant? Do we know the critical signals that control these processes? Are we even sure that these processes are coordinated? This review documents research progress and assesses to what extent the key questions have been addressed.

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#### Introduction

All plants require sulfur for growth and for land plants is most is acquired from the external environment as sulfate. For land plants this from the soil via the roots. Typically concentrations are low and often extremely variable. Thus, transport needs to be active to facilitate uptake against a concentration gradient, specific for sulfate and regulated to optimize uptake to growth and ensure optimal energy utilization in this process. For vascular plants transport is not only across a single membrane at the soil-root interface but also across many other plasma membranes to facilitate distribution, across the chloroplast membrane to the site of reduction and also across the tonoplast to allow transport in and out of the vacuole for the transient storage of excess sulfate taken up.

Progress on the understanding of plant sulfate transporters has been substantial and reported in successive volumes of the Sulfur Workshop series, with key landmark papers from a number of groups being published throughout this period. Some early key milestones in the development of the plant sulfate transporter research field are illustrated as a timeline in Fig. 1.

The first suggestion for active absorption was in an analysis of whole plant uptake of sulfate into barley roots. An enzyme based description of affinities and competition by selenate but not nitrate or phosphate unequivocally demonstrated the activity of a transmembrane ion transporter (Leggett and Epstein 1956). It would be 40 years before the molecular components would be identified in plants (Smith et al. 1995b). Prior to this key elements of regulation by de-repression (that is induction upon starvation) were described in a topical legume (Clarkson et al. 1983) and suggestion for involvement of a metabolite linking N and S metabolism, namely *O*-acetylserine (OAS), was described in maize (Clarkson et al. 1999). The importance of OAS as a regulator of gene expression for a cluster of genes has been described, separating S-related and other regulation (Hubberten et al. 2012, 2015). Mechanistic evidence for transport being driven by proton gradients was obtained in a duckweed (Lass and Ullrich-Eberius 1984).

The first substantial progress on the identification of sulfate transporters genes was inevitably for bacteria (Ohta et al. 1971; Sirko et al. 1990), fungi and yeast (Ketter et al. 1991; Smith et al. 1995b; Cherest et al. 1997), mammalian systems (Schweinfest et al. 1993; Hästbacka et al. 1994; Silberg et al. 1995) and finally in plants (Kouchi and Hata 1993; Smith et al. 1995a, 1997; Takahashi et al. 1996). Similarities in the sequence of many of the genes, some not identified as sulfate transporters was first noted by Sandal and Marcker (1994). Much of this work has been reviewed in previous volumes in this series: Kredich 1993 (bacteria); Thomas et al. 1997 (yeast); Davidian et al. 2000, Hawkesford et al. 2003, Buchner et al. 2010, Hawkesford 2012 (plants) and elsewhere (Markovich 2001) for mammalian transporters. These transporters are now recognised to be part of a large family of transmembrane ion transporters known as SulP (see also Price et al. 2004).



Fig. 1 Key early milestones in the development of the understanding of sulfate transport in plants placed in relation to the first 5 Sulfur Workshops

#### A Family of Sulfate Transporters

In a series of papers predominantly from the Takahashi group but with notable contributions from a number of others including the Davidian group it became apparent that a gene family of up to 14 genes encoded a group of related proteins in Arabidopsis (Takahashi et al. 1996, 1997; Vidmar et al. 2000). Similar gene families were subsequently identified in Brassica (Buchner et al. 2004b), in rice (Kumar et al. 2015), poplar (Dürr et al. 2010), Medicago (Gao et al. 2014) and in wheat (Buchner et al. 2004a).

Phylogenetic analysis of plant sulfate transporter sequences indicates discrete clades within the family (Fig. 2) and it has been proposed that these align with discrete functions and that within clades there may be some functional redundancy (Hawkesford 2003). In summary, Group 1 represents high affinity types responsible for up take into the cell, particularly in the roots, and are subject to nutritional regulation. Group 2 are lower affinity, less regulated and distributed throughout the plant. Group 3 are somewhat more enigmatic (see below), Group 4 are uniquely tonoplast located and responsible to vacuolar efflux. Group 5 are the most distantly related to the rest of the family and the 2 members are quite distinct from each other, lack a STAS domain and remain something of a puzzle. They seem to be involved in Mo accumulation, perhaps transport and as such have been name mot1 and mot2 (Tomatsu et al. 2007; Baxter et al. 2008; Gasber et al. 2011).



**Fig. 2** Phylogenetic relationship of the wheat and Arabidopsis sulfate transporter gene families: Neighbour-Joining Tree (Mega 6, Tamura et al. 2013) from Multiple Alignment (ClustalX V.2.1, Larkin et al. 2007) of coding cDNAs of the *Triticum aestivum* cv. Chinese spring D-genome (*white bold – black highlighted*) and *Arabidopsis thaliana (square framed white highlighted*) sulfate transporter gene family. The bootstrap values, expressed as percentage, were obtained from 1000 replicate trees (Courtesy of Peter Buchner)

#### The Transporter Itself: Recent Structural Insights

Early analysis of the amino acid sequences of the transporter was suggestive of 12 transmembrane domains, based on hydrophobicity plots and occurrence of charged amino acids (Clarkson et al. 1993; Smith et al. 1995a; Takahashi et al. 1996). More recent analysis of transporters in the same superfamily (SulP/SLC26 family) combining both topology mapping of for example the BicA transporter (see Price and Howitt 2014) and for prestin, homology modelling, molecular dynamics simulations and cysteine accessibility scanning are strongly supportive of a complex 14 transmembrane model (Gorbunov et al. 2014). In this analysis a

3-dimensional model has been derived which also proposes a central cavity as the substrate-binding site, midway in an anion permeation channel. Features of this cavity are almost certainly involved in substrate specificity and could potentially be modified to further increase selectivity, for example between sulfate and selenate, opening up the potential for designer crops.

An additional feature of members of this family is the STAS domain (Aravind and Koonin 2000; Rouached et al. 2005). Activity is totally dependent upon its presence and it is strongly suggested that is it involved in protein:protein interaction regulating activity, probably involving phosphorylation of a threonine residue.

The question of whether the transport acts as a monomer or oligomer is of interest and it has been suggested that heterodimers are required for activity or may have an import regulatory role. Maximal sulfate uptake and growth were obtained when a Group 3 transporter was co-expressed with a Group 2 transporter from Arabidopsis in yeast complementation approach, suggestive of the activity of a heterodimer (Kataoka et al. 2004a). No activity of the Group 3 when expressed alone was seen in this study. In contrast Group 3 transporter isolated from Lotus root nodules was able to complement a yeast mutant when expressed by itself (Krussell et al. 2005) indicating some variability for this oligomer requirement.

#### Specificity for Sulfate, Selenate and Molybdate

The non-specificity of the transporter was exploited in early studies, particularly with yeast, to obtain sulfate transporter-less mutants by harassing the toxic nature of oxyanion analogues of sulfate, particularly selenate but also chromate (Breton and Surdin-Kerjan 1977; Smith et al. 1995b). Selenate has also been applied as a selection agent for the isolation of Arabidopsis mutants by several groups (see for example, Shibagaki et al. 2002).

As the anions sulfate, selenate and molybdate are all transported by the same transporters, it is not surprising that their respectively accumulations in plant tissues are connected. Analysis of what grain from mildly sulfate deficient plots at Rothamsted showed a remarkable accumulation of Se and Mo (Shinmachi et al. 2010; Stroud et al. 2010). The S-deficient plots had a reduced grain yield and reduced grain S-content, both decreasing by about 10%, but several-fold increases in Se and Mo content. This could be partially explained by the observed induction of sulfate transporters in the roots of these field-grown plants, increasing uptake, and a more favourable ration of selenate and molybdate relative to sulfate in the soil solution. Whilst Se generally flows the distribution of sulfate in term of redistribution during grain filling and in relation to storage protein distribution in the grain, some enriched sub-cellular regions were indicative of specific accumulation on non-protein Se, possibly in vacuoles (Moore et al. 2010). Mo was less efficiently remobilized to the grain than Se during grain filling indicating either a fixation of the mineral in the vegetative tissue or a limitation to its later transportation (Shinmachi et al. 2010; Stroud et al. 2010).

#### Where Now?

Much has been determined about the nature of sulfate transporters in plants, not only in model species but also in crops. A knowledge of the regulation and properties of the transporters helps explain many physiological phenomena and some agronomic responses of crops. The question remains of how may this aid in breeding better genotypes or in informing agronomic treatments.

A previously stated ideotype for optimum S use involves uptake and storage during fluctuating supply, effective remobilization upon demand and appropriate partitioning to ensure healthy and nutritious crops (Hawkesford 2012). Breeding or biotechnology may help deliver such germplasm and the acquired knowledge is an essential prerequisite for such developments. Sulfur will always be required for crop growth so effective capture and utilization are worthy targets.

Acquisition is an important issue. Certainly the adaptation of de-repression will aid scavenging, but only in conjunction with root proliferation. Prospects for improving efficiency of uptake are limited, although constitutive uptake and over-accumulation, followed by storage and effective remobilization remains one key strategy. In relation to this strategy, challenges still exist in the understanding of movement of sulfate within the plant from organ to organ, distribution within specific tissues and finally within individual cells between organelles. It is still unclear as to how S moves into and out of the chloroplast, the key point of entry into the biosynthetic pathway. One reports indicated a chloroplast localizing isoform of the family (Takahashi et al. 1999) but this remains to be corroborated. Clearer is the involvement of Group 3 transporters in release of sulfate from vacuoles, a key storage site (Kataoka et al. 2004b).

David Clarkson proposed the idea of a black box (Fig. 3) in a foreword to the proceedings of the 3rd Workshop (Clarkson 1997). Substantial progress has been made in determining detail within this box since then, but the question of how to improve sulfur nutrient use efficiency remains. Sulfur is required for growth and health, for resistance to biotic and abiotic stresses, and contributes to nutritional properties of food and feed. Decreasing requirements is unlikely to be an option, optimizing agronomic inputs remains the key practical approach, although in the future this may be complemented with plants optimized genetically for specific qualities. Some investigations into natural variation in Arabidopsis have been made



Nutrient in

(Loudet et al. 2007) but there has been little investigation in crop plants and this is a key area for development.

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### **Quo Vadis Sulfur Investigation?: 25 Years of Research into Plant Sulfate Reduction**

#### Stanislav Kopriva

Abstract Sulfate assimilation is an essential pathway of plant primary metabolism providing cysteine for methionine and protein synthesis and as a source of reduced sulfur for synthesis of numerous other essential metabolites. The control of the pathway has long been a focus of plant sulfur research. Significant progress in understanding the pathway and its regulation has been made since the first Sulfur Workshop in 1989. All enzymatic steps have been identified, and the corresponding genes cloned. The investigations of sulfate assimilation were always quickly adopting newly developed approaches and the introduction of molecular biology led to a rapid switch to Arabidopsis as the main model for sulfur research. We learned a lot about the regulation of the pathway and identified genes affected by various environmental conditions or chemical signals. Sulfur research was one of the first areas of plant science to make use of systems biology with several seminal studies of major importance. The increased use of genetic approaches resulted in identification of new regulatory factors and most recently in finding genes responsible for genetic variation of sulfur-related traits in natural populations. Nevertheless, many questions remain open. This overview of the different approaches to study sulfur metabolism will highlight the success stories. The major progress since the first workshop will be summarised, and a new set of open questions will focus on how plant sulfur research will develop over the next 25 years.

#### Introduction

Plants take up the essential nutrient sulfur in the oxidized form of sulfate, reduce and incorporate it into various metabolites and cellular components (Kopriva 2006; Takahashi et al. 2011). The assimilation of sulfate into cysteine is therefore an essential pathway of plant primary metabolism, with the reductive part being key not just for the pathway but for life. Therefore sulfate reduction has been in the focus of plant sulfur research for many years, including being a major topic from

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the beginning of the Plant Sulfur Workshops (Brunold 1990). The final pathway (Fig. 1), however, has been formulated quite recently, after years of controversy (Kopriva and Koprivova 2004). Sulfate is firstly activated by adenylation catalyzed by ATP sulfurvlase to adenosine 5'-phosphosulfate (APS). The activated sulfate is reduced in two steps, first through a two electron reduction to sulfite and subsequently by six more electrons to sulfide. Sulfide is then incorporated into O-acetylserine to form cysteine (Takahashi et al. 2011). The pathway in plants is thus very similar to sulfate assimilation in bacteria and fungi, the original models for dissection of the pathway (Kopriva 2015), which was one of the main reasons for the controversy and discussions regarding the plant pathway. As the model microorganisms Escherichia coli and Saccharomyces cerevisiae require a second activation step for the sulfate to be reduced (phosphorylation of APS to 3'-phosphoadenosine 5'-phosphosulfate (PAPS)), it was believed that plant sulfate assimilation proceeds in the same way with PAPS as an intermediate (Thompson 1967). However, in the beginning of the 1970s it was shown that APS is used directly in plants and algae and the corresponding enzyme was named APS sulfotransferase (Schmidt 1972). The reduction of APS was shown to result in a carrier-bound intermediate and a thiosulfonate reductase was described as a component of plant sulfate assimilation (Schmidt 1973). In the following years, APS sulfotransferase was shown to be highly regulated and a key enzyme for the control of the pathway, however, PAPS sulfotransferases have also been part of the literature (Schmidt and Jäger 1992; Brunold and Suter 1984). Thus, at the time of the first Plant Sulfur Workshop in Groningen, in 1989, many fundamental questions about sulfate reduction remained open (Brunold 1990) which are summarized in the following:

- 1. the physiological significance of the enzyme reactions detected *in vitro*, especially of the APS sulfotransferase versus the PAPS sulfotransferase pathway and of the organic thiosulfate reductase versus the sulfite reductase mechanism,
- 2. the detailed characterization of APS sulfotransferase and PAPS sulfotransferase,
- 3. the molecular basis of the regulatory phenomena observed,
- 4. the contribution of the root system to assimilatory sulfate reduction.

In this review the progress in answering these questions will be summarized, an overview of the different approaches to study sulfur metabolism presented, and new set of open questions formulated on how plant sulfur research will develop in the next 25 years.

#### **Current Understanding of the Sulfate Assimilation Pathway**

As evident from the first question, at the time of the first Sulfur workshop, two alternative enzymes for the first step of sulfate reduction were discussed, APS sulfotransferase and PAPS sulfotransferase (Brunold 1990). The situation however



Fig. 1 Current understanding of plant sulfate assimilation

became even more complicated after the cloning of an Arabidopsis enzyme, which produced sulfite from APS and was named APS reductase (Gutierrez-Marcos et al. 1996; Setya et al. 1996). Three possible routes of sulfate to sulfide have been proposed, APS dependent bound-sulfite way (through APS sulfotransferase and thiosulfonate reductase), APS dependent reduction (APS reductase, sulfite reductase) or PAPS-dependent pathway (Fig. 2). The controversy has been clarified when the enzyme characterized as APS sulfortansferase was purified from *Lemna minor*, its N-terminal sequence was determined and the corresponding cDNA cloned (Suter et al. 2000). The sequence of the APS sulfotransferase was highly similar to the sequence of APS reductase, which together with identification of free sulfite as reaction product resulted in postulating APS reductase as the component of sulfate assimilation. In addition, experiments in oxygen free atmosphere showed that the bound sulfite is the result of reaction of sulfite with oxidized thiols, and therefore probably irrelevant for the sulfate reduction pathway, which proceeds through APS reductase and sulfite reductase (Suter et al. 2000). However, the question of PAPS reductase/sulfotransferase in plants has not been completely resolved. Even though everything suggests that APS reductase is the sole reducing enzyme, it is impossible to completely rule out the existence of PAPS-dependent enzyme. The crucial evidence, which would show that disruption of all APS reductase isoforms is lethal, is not available and thus the plant PAPS reductase is still possibly out there.

Although the question of the first reductive step in sulfate assimilation was resolved, the pathway has changed further. A new enzyme was added to the pathway, which had not been known in plants, sulfite oxidase (Eilers et al. 2001). Sulfite oxidase acts against the flow of sulfur in sulfate assimilation, therefore, to



Fig. 2 Three possible routes of sulfate reduction in plants as discussed 1996–2000

prevent a futile cycle it is spatially separated from the reducing enzymes in peroxisomes. The enzyme protects plants against elevated sulfur dioxide and appears to be important for sulfite homeostasis, but whether this is its main physiological function is still under discussion (Lang et al. 2007; Brychkova et al. 2007, 2013).

In addition, in the last decade it became apparent that a second non-reductive path of sulfate assimilation, into PAPS and further into sulfated compounds, is essential also in plants (Mugford et al. 2010). APS kinase, producing PAPS from APS, has been shown to be important for synthesis of glucosinolates, a group of sulfated secondary compounds with plethora of functions in plant defence (Mugford et al. 2009). However, while the loss of glucosinolates does not compromise Arabidopsis in the absence of pathogens, a mutant disrupted in APK1 and APK2 isoforms of APS kinase showed a clear semi-dwarf phenotype (Mugford et al. 2009). These experiments showed the importance of PAPS for plant performance, however, the exact nature of the essential metabolites is not known.

In conclusion, the first question concerning the pathway of sulfate assimilation now seems to be largely answered (Fig. 1).

#### **Biochemistry of APS Reductase**

The second question, on the detailed characterization of APS and PAPS sulfotransferases, has naturally been limited to the APS dependent enzyme. The analysis of purified APS sulfotransferase revealed that it is identical to recombinant

APS reductase, identified by molecular cloning (Suter et al. 2000). The enzyme uses APS as a substrate with affinity of  $10-20 \,\mu$ M, and glutathione or other thiols as reductant. Sequence analysis has shown that the enzyme is composed from three domains, a plastid targeting peptide, a reductase domain similar to bacterial PAPS reductases, and a C-terminal thioredoxin-like domain (Setva et al. 1996; Gutierrez-Marcos et al. 1996). The enzyme can be split into these two domains and the activity can be reconstituted (Bick et al. 1998). This makes sense because bacterial PAPS reduction is also dependent on thioredoxin or glutaredoxin. The two domains have distinct functions: the reductase domain is responsible for the interaction with APS and the formation of sulfite, whereas the C-terminal domain interacts with the reductant and resumes the function of glutaredoxin (Bick et al. 1998). APS reductase contains an iron-sulfur cluster, which was shown to be diamagnetic and asymmetric, as in the enzyme isoform studied (APR2 from Arabidopsis thaliana) it is most probably bound only by three cysteine residues in the reductase domain (Kopriva et al. 2001). When the enzyme is incubated with APS in the absence of reductants, a stable reaction intermediate can be detected with sulfite covalently bound to a cysteine residue in the reductase domain (Weber et al. 2000). The sulfite can be released by treatment with free thiols or sulfite, or by addition of the recombinant C-domain, confirming that the two domains have different functions in the catalysis, but raising a question on the function of the iron sulfur cluster.

Genes for the  $[Fe_4S_4]$  cluster containing APS reductase have been found in all seed plants, basal plants, and green algae. However, this is not the only form of this enzyme (Fig. 3). Comparison of plant APS reductase with its bacterial counterparts revealed that the bacteria possess two classes of enzymes, similar to the reductase domain. The "classical" PAPS reductases from E. coli and Salmonella (and also yeast and fungi) differ from the plant enzyme mainly due to the absence of two conserved cysteine pairs, which are responsible for binding of the FeS cluster (Kopriva et al. 2002). Correspondingly, the cofactor has never been reported for PAPS reductases. However, another form of bacterial enzyme has been identified, more similar in sequence to the plant APS reductase, including the two cysteine pairs (Bick et al. 2000; Williams et al. 2002). These bacterial enzymes use APS rather than PAPS and contain the FeS cluster (Kopriva et al. 2002). Due to their similarities, the APS and PAPS reductases have been both referred to as sulfonucleotide reductases and shown to have a very similar reaction mechanism (Carroll et al. 2005). The specificity for APS or PAPS seems to be linked to the presence or absence of the cluster, respectively (Bhave et al. 2012). Incidentally this is also true for the dissimilatory APS reductase in sulfur oxidizing bacteria, which, although having a completely different primary structure, possess FeS centre (Frigaard and Dahl 2009).

The variation in APS reductase forms is even greater. Another isoform of APS reductase has been found in the moss *Physcomitrella patens*, which in contrast to the "classical" plant APS reductase does not bind the FeS cluster, and does not possess the thioredoxin-like domain but still reduces APS and not PAPS (Kopriva et al. 2007). This challenges the simple link between FeS cluster and APS reduction, which seems to be valid only in prokaryotes, and reinforces the question



Fig. 3 Diversity of APS reductase forms in different organisms. *Yellow marks* PAPS reductases or APS reductases type B without FeS cluster, the presence of the cluster is marked by *orange colour*. The AprA and AprB isoforms of dissimilatory reductase from *Desulfovibrio vulgaris* are shown for comparison

regarding the function of the FeS cluster. This isoform of APS reductase, named APR-B, is present in several other basal plants, including *Selaginella* and *Marchantia* (Kopriva et al. 2007). A highly similar enzyme is dominating in the Eukaryotic microalgae, such as diatoms and haptophytes. These organisms, e.g. *Thalassiosira pseudonana* or *Emiliania huxleyi* possess genes for APS reductase of the APR-B type, and with the thioredoxin-like domain (Fig. 3) (Patron et al. 2008). Interestingly, the APS reductase activity in the microalgae is approximately two orders of magnitude higher than in higher plants (Bochenek et al. 2013). Thus, although a lot of progress in understanding the biochemistry of APS reductase has been made, the reaction mechanism is not yet completely resolved, since structure is available from the bacterial enzymes, or the APR-B, but not the full plant enzyme with both domains. It is however remarkable, that the same reaction can be catalyzed by enzymes with a similar primary structure but with, and without, the FeS cofactor.

#### **Molecular Mechanisms of Regulation**

At the time of the first Sulfur Workshop, the regulation of sulfate assimilation, particularly the reductive part, was well described at the physiological level. It was shown that sulfate reduction is induced by limiting sulfur availability and cadmium, and decreased by reduced sulfur, both by thiols and atmospheric SO<sub>2</sub> (Brunold 1990). Regulatory interactions with nitrogen nutrition had been described: a decrease of sulfate reduction capacity at low N and induction by ammonium (Brunold and Suter 1984). The developmental regulation of the pathway was described, showing an increase in activity from etiolated seedlings to green tissues and in the development of leaves until maturity was reached (Brunold 1990). In addition, feedback inhibition of individual enzymes of the pathway was described,

such as APS inhibiting ATP sulfurylase and AMP acting negatively on APS sulfotransferase. Generally, it was believed that the regulation was mainly at the level of ATP sulfurylase, with many conditions affecting additionally APS reductase.

The progress in understanding the regulation of sulfate assimilation has been significant over the past 25 years. It is impossible to describe all new findings on compounds and conditions affecting the pathway, the mechanisms, signalling pathways, transcription factors, etc.; these have been described in many reviews (Davidian and Kopriva 2010; Kopriva 2006; Koprivova and Kopriva 2014; Takahashi et al. 2011; Leustek et al. 2000; Hell 1997; Rausch and Wachter 2005; Chan et al. 2013). Instead, a personal selection of the major findings will be given, not only showing the major concepts but also the variety of links between sulfate assimilation and plant metabolism. The progress in uncovering the mechanisms of regulation is highly interconnected with the variety of methods and approaches used and with a shift in focus of the studies towards APS reductase.

The classical physiological studies led to formulation of the major concept in regulation of sulfate assimilation, the demand driven control, which explained most of the known regulatory events (Lappartient et al. 1999; Lappartient and Touraine 1996). The demand driven regulation is an overarching concept that still helps to explain observed regulation of sulfate assimilation by many environmental conditions. Another major impact on studies of the control of the pathway was the identification of signalling function of O-acetylserine (Neuenschwander et al. 1991), leading to some controversies but also to formulation of function on different levels (Hubberten et al. 2012b; Wirtz and Hell 2006). Other signals have been found with a more or less clear contribution to the regulation, such as sugars and phytohormones, particularly jasmonate and cytokinins (Jost et al. 2005; Maruyama-Nakashita et al. 2004). Most important was the finding of sulfur starvation inducible microRNA miR395, which targets three isoforms of ATP sulfurylase and the sulfate transporter SULTR2;1 and contributes strongly not only to response to sulfate starvation but also to regulation of sulfate partitioning (Kawashima et al. 2009, 2011; Liang et al. 2010).

The first major step in understanding the mechanisms of regulation was the introduction of molecular biology methods to complement the physiological and biochemical studies. These showed in the beginning that regulation of the pathway, e.g. by sulfate starvation, light, nitrogen sources, or reduced sulfur compounds, is mostly on the transcriptional level (Kopriva et al. 1999; Koprivova et al. 2000; Vauclare et al. 2002; Takahashi et al. 1997). APS reductase was more highly regulated than other components of the pathway, suggesting its central role in the control of the pathway. The key role of this enzyme was confirmed using flux control analysis, at least for the feedback regulation by thiols (Vauclare et al. 2002). These studies were followed by numerous reports showing transcriptional regulation of one or multiple genes of sulfate assimilation by a large variety of metabolites or environmental conditions (reviewed e.g. in Davidian and Kopriva 2010; Kopriva and Rennenberg 2004).

Numerous expression analyses showed that multiple genes are regulated by a single treatment. The availability of Arabidopsis genome then enabled the design of the first global transcriptomics studies and with three seminal papers on response to sulfate starvation sulfur research entered the systems biology era (Maruvama-Nakashita et al. 2003; Nikiforova et al. 2003; Hirai et al. 2003). The sulfate starvation response is one of the best studied environmental conditions by systems biology approaches, pioneering the combination of metabolite-gene networks and producing some of the best examples of mechanistic understanding derived from omics data (Nikiforova et al. 2005; Hirai et al. 2005; Hubberten et al. 2012b). These studies also delivered first hints to the connection between sulfate assimilation and synthesis of glucosinolates (Hirai et al. 2005; Malitsky et al. 2008). Altogether these studies revealed a core cluster of genes highly regulated by sulfate starvation, some belonging to the pathway, e.g. sulfate transporters SULTR1;1,1;2,4;1,4;2 or the APS reductase and some with unknown function, which is now only starting to be understood. They showed that sulfate starvation response includes modulation of phytohormone synthesis, particularly jasmonate and auxin but, while they identified potential regulatory hubs and factors, the progress in deciphering mechanisms of regulation has been rather limited.

Comparing results of expression analyses with activity measurements or metabolite concentrations it became soon apparent that not all regulation can be explained by transcriptional control and that various post-transcriptional mechanisms contribute to control of sulfate assimilation (Takahashi et al. 2011). Obviously, the miR395 is one of these mechanisms, restricting expression of the SULTR2;1 transporter to the xylem during sulfate deficiency and so increasing efficiency of sulfate root to shoot translocation (Kawashima et al. 2009). Another regulatory mechanism is the fluid dissociation and association of the cysteine synthase complex by changes in concentration of sulfide and O-acetylserine, which is probably an important player in sensing sulfur availability (Wirtz and Hell 2006). However, the most common and fundamental post-transcriptional mechanism of regulation of sulfur metabolism is redox. The first enzyme of the pathway shown to be redox regulated was, not surprisingly, APS reductase. This regulatory mechanism was initially based on *in vitro* observations of activation of recombinant APS reductase by oxidative agents and inactivation by reduction and corroborated in vivo by observed discrepancy between response of APS reductase protein and enzyme activity to oxidative stress (Bick et al. 2001; Kopriva and Koprivova 2004). The next target of redox regulation is the  $\gamma$ -glutamylcysteine synthetase ( $\gamma$ ECS), the first of two enzymes catalyzing glutathione synthesis. The inhibition of this enzyme by glutathione has been shown to be caused by redox regulation rather than product inhibition (Hothorn et al. 2006). The latest acquisition in the list of redox regulating enzymes of sulfur metabolism is APS kinase (Ravilious et al. 2012). Interestingly, the redox regulation has been recognized only after the structure of the enzyme has been solved, whereas previous biochemical characterizations had not found any hints. In contrast to APS reductase, APS kinase is activated in its reduced form and inactivated by oxidation (Ravilious et al. 2012). Since APS kinase is important for control of sulfur partitioning between primary reductive assimilation and PAPS and
secondary products (Mugford et al. 2011), the redox regulation may be important for the mechanism, but its biological relevance needs to be shown *in planta*.

The increasing availability of molecular and genetic resources in Arabidopsis allowed the use of genetic tools to further dissect the regulation of the sulfate assimilation pathway. Reverse genetic approaches helped, e.g., to understand the differences in functions of individual isoforms of the pathway components. The main contributions if this approach were linking the two branches of sulfate assimilation with glucosinolates synthesis through analysis of mutants in APS kinase (Mugford et al. 2009) and evidence that also other components of sulfate assimilation pathway than APS reductase may represent a bottleneck, namely sulfite reductase and serine acetyltransferase (Haas et al. 2008; Khan et al. 2010). To find new regulatory factors, however, other genetic methods have been applied. The biggest success in unravelling the mechanisms of regulation has been the identification of SULFATE LIMITATION1 (SLIM1), a transcription factor of the Ethylene-Insensitive-3-Like family (EIL3), a central regulator of the response to sulfate starvation (Maruyama-Nakashita et al. 2006). SLIM1 is responsible for the induction of transcripts for sulfate transporters, miR395, and many other genes by sulfate deficiency (Maruyama-Nakashita et al. 2006; Wawrzynska and Sirko 2014). The mechanism of its action is unknown yet, as *SLIM1* mRNA is not affected by sulfate starvation, and it is also not the only factor responsible for the response, since upregulation of transcripts for APS reductase is SLIM1 independent (Wawrzynska and Sirko 2014). Other transcription factors have been shown to regulate sulfate assimilation, e.g. the LONG HYPOCOTYL 5 (HY5) (Lee et al. 2011), but the understanding of the transcriptional machinery controlling sulfate assimilation is limited. This is in contrast to the regulation of glucosinolates synthesis, where a complex interplay of six MYB factors and three bHLH (MYC) factors has been described (Schweizer et al. 2013; Sonderby et al. 2010; Frerigmann and Gigolashvili 2014). Interestingly, these MYB factors also regulate genes of primary sulfate assimilation, showing that the glucosinolates biosynthesis should be considered within the core sulfur metabolism, at least in Arabidopsis and other Brassicaceae (Yatusevich et al. 2010).

All the approaches mentioned above gave some insights into the regulation of sulfur metabolism. The next set of experiments, exploiting quantitative genetics, identified further genes that are responsible for the natural variation of sulfur related traits and that can be potentially used for their modification in crops. These approaches profited from the availability of well defined Arabidopsis ecotypes and the progress in their genotyping. Two areas of sulfur research have been focus of the natural variation studies, glucosinolates and sulfate homeostasis. The glucosinolates diversity in Arabidopsis covers both quantitative and qualitative differences, particularly among the aliphatic, methionine derived glucosinolates. Five loci were identified that explain most of the qualitative variation among multiple accessions, primarily in the chain length, and one of them controls up to 75% of the quantitative differences (Kliebenstein et al. 2001b). In a more refined QTL study 20 loci were found to control the variation in glucosinolates between Ler and Cvi accessions (Kliebenstein et al. 2001a). Among the most important loci is

the GS-ELONG locus, containing multiple methylthioalkylmalate synthase genes, which plays a key role in the methionine elongation step of glucosinolate synthesis (Kroymann et al. 2001). The second major QTL is the GS-AOP, which is formed by two genes for 2-oxoglutarate-dependent dioxygenases, responsible for chain modification of the aliphatic glucosinolates and contributing greatly to the diversity (Kliebenstein et al. 2001c). The same loci were identified in a genome wide association study (GWAS), along with other, so far unknown ones (Chan et al. 2011). Therefore, it can be expected that the current view of control of glucosinolates synthesis remains incomplete.

The second theme approached by quantitative genetics is the accumulation of sulfate and sulfur. A QTL analysis of sulfate content in leaves of recombinant inbred lines from Bay-0 and Shahdara ecotypes revealed a non-synonymous single nucleotide polymorphism (SNP) in gene for APR2 isoform of APS reductase, resulting in almost complete inactivation of the corresponding enzyme (Loudet et al. 2007). Loss of APR2 leads to diminishing of total enzyme activity by 75%, consequently the flux through sulfate assimilation is reduced and sulfate accumulates. Interestingly, a second OTL from the same screen has been identified as ATPS1 isoform of ATP sulfurylase, additive to the effect of APR2 (Koprivova et al. 2013). The phenotype is caused by variation in mRNA accumulation, which correlates with a deletion in the first intron of ATPS1 genes, leading to lower enzyme activity, reduced flux, and accumulation of sulfate. The OTL analysis, however, can only reveal variation in the two parent ecotypes, whereas much larger variation exists among the different accessions. When a wider variation in sulfur homeostasis has been assessed in 350 accessions, particularly high total sulfur content was detected in the Hod ecotype. The high sulfur content was caused mainly by high accumulation of sulfate, and exactly as in the Bay-0  $\times$  Shahdara population the causal gene was shown to be APR2 (Chao et al. 2014). A further analysis of links between high sulfate/sulfur accumulation and APR revealed another small group of ecotypes related to Lov-1, with naturally inactive APR2. All three accessions (Shahdara, Hod, and Lov-1) possess different amino acid substitutions that lead to at least 1000-fold reduction of enzyme activity (Chao et al. 2014). Thus, in the Arabidopsis population, APR2 was at least three times independently inactivated, with slightly different consequences for accumulation of sulfate and sulfur.

Three more findings are considered potentially important for understanding the regulation of sulfate assimilation and its integration within general metabolic networks. Firstly, the unexpected finding of coordinated enrichment of expression of sulfate assimilation and glucosinolates synthesis in bundle sheath cells in Arabidopsis (Aubry et al. 2014). This resembles the localization of sulfate assimilation in plants with C4 photosynthesis, which is one of the unsolved questions of sulfur research. Secondly, several reports indicated an important role of the interplay between sulfite reductase and sulfite oxidase in "sulfite network" for sulfite homeostasis and regulation of sulfate assimilation (Brychkova et al. 2013). Finally, linking glucosinolates into core sulfur metabolism brought a connection to an enigmatic enzyme, 3'(2'),5'-bisphosphate nucleotidase, known as SAL1 or Fiery1,

which has been found in many genetic screens targeting mutants in stress response or leaf morphology (Robles et al. 2010). The enzyme metabolizes phosphoadenosine phosphate (PAP), which is a by-product of sulfation reactions, e.g., in synthesis of glucosinolates. PAP accumulates during drought stress and was proposed as a retrograde signal of stress from plastids to the nucleus (Estavillo et al. 2011). The loss of FRY1 also has a direct effect on sulfur assimilation, the glucosinolate content is lower in the mutants and the plants accumulate desulfoglucosinolate precursors (Lee et al. 2012). In addition, fry1 mutants show lower accumulation of sulfate and total sulfur, making the enzyme and PAP interesting targets for further investigations.

Overall, many mechanisms of regulation have been discovered, but since every step in the pathway can become limiting, the control of sulfur metabolism is far from being fully understood.

#### **Sulfate Assimilation in Roots**

The fourth question formulated at the first Sulfur Workshop is one with the least definite answer. Root can reduce sufficient sulfate for their own demand, as shown clearly in root cultures growing in sulfate as the sole sulfur source (Vauclare et al. 2002). However, sulfate reduction in roots cannot complement the reduced APS reductase activity in the shoots, as demonstrated by reciprocal grafting of Col-0 and Hod accessions (Chao et al. 2014). In a split root system, the regulation of sulfate transport seems to be dependent on local sulfate availability and sulfate is translocated to shoots but not to a sulfate deficient part of the root system (Hubberten et al. 2012a). Roots contribute to sulfate deficiency response by increased uptake and translocation of sulfate, and there seems to be a communication between shoots and roots regarding the sulfur status (Hubberten et al. 2012a). The contribution of roots to overall sulfur homeostasis, sensing, and signalling, however, needs to be systematically investigated, e.g., using more grafting experiments with various mutants in sulfate assimilation and signalling pathways.

#### Sulfur Research in the Next 25 Years

The progress in understanding the pathway of sulfur assimilation and the regulation has been overwhelming. However, there are still many open questions, some reformulating the "old" concepts, some derived from the emerged new insights. Thus, in analogy with the first Sulfur Workshop and the most intriguing open questions formulated there, the following questions can be considered key for the next 25 years of plant sulfur research:

*How is the sulfur flux controlled?* This question has not changed much in the past 25 years, but the concept has changed dramatically. Whilst searching for the rate

limiting enzyme, it became obvious that control is shared and may lie on every step of the pathway, dependent on the actual metabolic and developmental status of the plant. Understanding the flux control will require consideration of sulfur fluxes in the context of the whole plant metabolism and employment of advanced mathematical tools and models (Calderwood et al. 2014). Solving this question will obtain a fundamental understanding about the control of the pathway and its integration with other pathways and finding the corresponding mechanisms – transcription factors and signals, and help with applying the knowledge generated in the course of plant sulfur research. Many traits are connected with sulfur, many sulfur compounds are important for various applications, e.g. the health promoting glucosinolates. The knowledge about control of sulfur fluxes will enable smart genetic engineering to create plants with tailored contents of diverse sulfur compounds.

What are the molecular mechanisms of the regulation? In the course of past investigations, much has been learned about the regulation of sulfur metabolism, and this trend will certainly continue. The number of known transcription factors, signalling compounds, and other regulatory mechanisms is limited, and it is crucial to find the full complement of transcription factors regulating the pathway, and to understand their functions. The bigger picture of transcriptional regulation has also to be considered, such as epigenetics and the role of transcriptional complexes, e.g. the Mediator. The redox regulation of APS kinase and APS reductase and possibly other enzymes has to be assessed in plants and the contribution to the control of sulfur fluxes quantified. We expect that further use of quantitative genetics will identify new genes and alleles controlling the variation in sulfur related traits in natural populations.

What are the biochemical properties of the new enzyme isoforms from algae? The new forms of ATP sulfurylase and APS reductase and their numerous fusions await characterization and determination of their biochemical properties. This will enable better understanding of sulfur metabolism in these organisms, that have often very high activity of these enzymes, but also generate sources of new more efficient enzymes for engineering of sulfur assimilation in plants or synthetic microbes.

What is the identity of the unknown sulfur compounds? In an untargeted metabolomics approach, a large number of unknown sulfur containing compounds have been detected in Arabidopsis (Glaser et al. 2014). This shows that the catalogue of sulfur compounds is far from complete even in model plants, and therefore new enzymes have to be connected to the sulfur network, e.g., a large number of sulfotransferase isoforms have unknown substrate specificity. Identification of these compounds will enable a better assessment of sulfur pools and may lead to discovery of new signalling and regulatory compounds.

How is sulfur metabolism integrated in the whole plant metabolism? Although the connections between sulfur and nitrogen nutrition, or sulfur and carbohydrates have frequently been described, the mechanisms and signals are unknown. The sensing of sulfur status and its transduction to the general response is part of this question and is also completely unclear, again a large contribution of mathematical tools will be necessary to answer this question.

What is the role of sulfur metabolism in evolution and adaptation of plants? Large variation in many sulfur related traits have been observed within Arabidopsis or Brassica species. There are also differences in localization of the pathway in C3 and C4 plants, which highlight the links between C4 photosynthesis and sulfur reduction. So, how conserved are the regulatory mechanisms in different plant families? With reduced costs for sequencing and genotyping, this may be answered for some plant species. However, research shows that variation in glucosinolates has a large role in underpinning variation of Arabidopsis accessions in resistance to insects (Kliebenstein et al. 2002). Therefore, the ecological significance of varied sulfate and sulfur levels is of immense interest.

How can we apply the knowledge on sulfur metabolism for improvement of plants and/or added value? Adequate sulfur nutrition is essential for high yields and quality of crops. Due to reduced atmospheric sulfur depositions yields can be sustained only through sulfur fertilization. Therefore, understanding sulfur homeostasis will underpin approaches to breed low input crop varieties allowing reduction of the environmental costs of intensive agriculture. In addition, sulfur compounds are often linked with resistance to pests or abiotic stress, so the improvement of synthesis of these compounds might be beneficial. Many sulfur compounds, however, are important beyond the plant; the best example is the glucosinolate from broccoli, glucoraphanine, which is a precursor of sulforaphane, a plant derived metabolite with a plethora of functions in cancer prevention (Clarke et al. 2008). Broccoli varieties with increased content of glucoraphanine have been created and shown to possess an increased capacity to prevent prostate cancer (Traka and Mithen 2009). A better knowledge of the control of plant sulfur metabolism will enable the generation of plants accumulating other beneficial compounds.

Altogether, we can surely look forward to many new exciting stories about sulfur and hope that the next 25 years of plant sulfur research will be as successful as the past 25 years.

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# Expression Profile of the Serine Acetyltransferase (SERAT) and O-Acetylserine (thiol)lyase (OASTL) Gene Families in Arabidopsis

#### Mutsumi Watanabe and Rainer Hoefgen

**Abstract** Cysteine synthesis in plants constitutes the entry of reduced sulfur from assimilatory sulfate reduction into metabolism. Cysteine synthesis is catalyzed by the sequential action of serine acetyltransferase (SERAT) and *O*-acetylserine (thiol) lyase (OASTL). In the Arabidopsis genome, there are five *SERAT* and three *OASTL* genes. Analysis of the expression data obtained from micro array data in database such as Arabidopsis eFP browser and publications indicates that the *SERAT* genes and *OASTL* genes show distinct expression patterns during development and under diurnal regulation as well as under stress conditions, suggesting the specific function/regulation of SERATs and OASTLs in different subcellular compartments.

#### Introduction

Serine acetyltransferase (SERAT), which catalyzes the formation of *O*-acetylserine (OAS) from serine and acetyl coenzyme A (acetyl-CoA), links serine metabolism to cysteine biosynthesis. OAS reacts with sulfide to yield cysteine catalyzed by *O*-acetylserine(thiol)-lyase (OASTL). In the Arabidopsis genome, there are five *SERAT* genes, *SERAT1;1* (At5g56760), *SERAT2;1* (At1g55920), *SERAT2;2* (At3g13110), *SERAT3;1* (At2g17640) and *SERAT3;2* (At4g35640) and three *OASTL* genes, *OASTL1;1* (At4g14880), *OASTL2;1* (At2g43750) and *OASTL2;2* (At3g59760) (Table 1). OASTLs belong to the beta-substituted alanine synthase (BSAS) family in the large superfamily of pyridoxal 5'-phosphate-dependent enzymes, which comprises beta-cyanoalanine synthase (CAS) (Hatzfeld et al. 2000; Yamaguchi et al. 2000), L-cysteine desulfhydrase (DES) (Alvarez et al. 2010) and *S*-sulfocysteine synthase (SSCS) (Bermudez et al. 2010) in Arabidopsis. The OASTL1;1, OASTL2;1 and OASTL2;2 are also called BSAS1;1, BSAS2;1 and BSAS2;2, respectively (Watanabe et al. 2008a). Decades research has provided

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Family	Group	Gene name	AGI code	Localization
SERAT	Group I	SERAT1;1	At5g56760	Cytosol
	Group II	SERAT2;1	At1g55920	Plastid
		SERAT2;2	At3g13110	Mitochondrion
	Group III	SERAT3;1	At2g17640	Cytosol
		SERAT3;2	At4g35640	Cytosol
OASTL	Group I	OASTL1;1	At4g14880	Cytosol
	Group II	OASTL2;1	At2g43750	Plastid
		OASTL2;2	At3g59760	Mitochondrion

Table 1 SERAT and OASTL gene families

cumulative insights into the function of SERATs and OASTLs by biochemical characterization in vitro, subcellular localization studies and in vivo functionality studies using the T-DNA knockout mutants. SERATs and OASTLs are present in three compartments; cytosol, plastid and mitochondrion in Arabidopsis (Noji et al. 1998; reviewed in Hell et al. 2002; Kawashima et al. 2005) (Table 1). Among five SERAT isoforms, three isoforms (SERAT1;1, SERAT2;1 and SERAT2;2) are biochemically more active than other two isoforms in group III (Noji et al. 1998; Kawashima et al. 2005). Cytosolic SERAT isoforms (SERAT1:1 and SERAT3:2) are feedback sensitive by cysteine (Noji et al. 1998; Kawashima et al. 2005). Among nine BSAS isoforms in Arabidopsis, three isoforms (OASTL1:1, OASTL2;1 and OASTL2;2) are biochemically most active in terms of OASTL activity (Jost et al. 2000). The fact that the single knockout mutants of SERATs and OASTLs did not reveal lethal phenotype indicated that the individual SERAT and OASTL isoforms are functionally redundant and OAS and cysteine are efficiently transported between cytosol and organelles (Heeg et al. 2008; Watanabe et al. 2008a, b). Although they display some functional redundancy, it appears that their contributions of OAS and cysteine synthesis are different in plant tissues and growth conditions. The analysis of serat and oastl mutants revealed that mitochondrial SERAT2;2 and cytosolic OASTL1;1 predominantly contributes to OAS and cysteine synthesis, respectively, in leaves of Arabidopsis grown on agar plate (Watanabe et al. 2008a, b) while the predominant isoforms change in other tissues. For example, in case of SERAT, cytosolic forms SERAT1;1, SERAT3;1 and SERAT3;2 seem responsible for OAS formation rather than mitochondrial SERAT2;2 in the Arabidopsis siliques (Watanabe et al. 2008b). This is consistent with their gene expression patterns in the public transcriptome database such as Arabidopsis eFP browser (Winter et al. 2007; Fig. 1) and gene expression analysis by qRT-PCR (Kawashima et al. 2005); the decrease of gene expression level of SERAT2;2, increase of SERAT3;1 and SERAT3;2 and constant level of SERAT1;1 during silique and seed development. As in this example, the detailed gene expression analysis of SERATs and OASTLs with the experimental knowledge could be more useful for predicting their contributions and specific functions.



**Fig. 1** Developmental and tissue-specific regulation. Gene expression patters in Arabidopsis tissues (Arabidopsis eFP Browser; Winter et al. 2007). The expression data was normalized with the GCOS (Gene Chip Operating Software) method and Target intensity (TGT) value of 100. Error bars represent standard deviation. Most tissues were sampled in triplicate. The specific probe for *SERAT3;2* gene is not found in ATH1 Affymetrix microarray

#### **Developmental and Tissue Specific Regulation**

During plant development, the *SERAT* genes showed tissue-specific expression patterns (Fig. 1). High gene expression levels were observed in mature pollen for *SERAT1;1*, in senescent leaves and late stage of flower for *SERAT2;1* and in silique, especially seed for *SERAT3;1*. *SERAT2;2* showed high expression at the vegetative

stage and in flower, but low expression during seed development in a negative correlation to the gene expression pattern of SERAT3;1. Since the probe of SERAT3;2 was not on microarray, SERAT3;2 was reported to show high expression at the reproductive stage only from qRT-PCR analysis (Kawashima et al. 2005). In contrast to the specific expression patterns of individual SERATs, the three OASTLs showed relatively similar expression patterns and less tissue specificity during plant development. But still the decrease of gene expression level of OASTL2;2 and increase of OASTL1;1 during seed development was observed, which is a similar correlation between SERAT2;2 and SERAT3;1, namely mitochondrial and cytosolic isoforms, suggesting the subcellular specific regulation. On the other hand, although specifically cytosolic SERAT1;1 showed highest expression in pollen, all three OASTLs including cytosolic OASTL1:1 showed low expression, suggesting the specific function of OAS accumulation or strict regulation of cysteine production in pollen. The importance of cysteine production in pollen was supported by the report that at least a functional one of three OASTLs in the pollen is required for the successful fertilization (Birke et al. 2013).

### **Diurnal Regulation**

Sulfur assimilation genes have been reported to be circadian/diurnal-regulated (Harmer et al. 2000). Among the *SERATs*, particularly *SERAT2*;1 and relatively *SERAT2*;2 showed strong diurnal oscillations in expression in Arabidopsis leaves (Espinoza et al. 2010) (Fig. 2). Even though the range of oscillations was small, other *SERATs* and *OASTLs* also showed coordinated diurnal oscillations; slight increase during night for *SERAT3*;1 and *OASTL2*;1, slight decrease during night for *OASTL1*;1 and *OASTL2*;2. Interestingly OAS level also showed diurnal oscillation



**Fig. 2** Diurnal regulation. Gene expression patters in diurnal cycles (Espinoza et al. 2010). Plants were grown under long day condition (16 h light/8 h dark) at 20 °C. Relative expression (log2) is indicated. *White* and *grey bars* indicate the day and night periods, respectively. *ZT* zeitgeber time. The specific probe for *SERAT3*;2 gene is not found in ATH1 Affymetrix microarray

with a peak at middle of night whilst cysteine level was not changed (Espinoza et al. 2010). Although any *SERATs* did not display the peak expression at the same phase of the OAS peak, *SERAT2;2*, which were increased in early onset of night, may contribute to the OAS peak considering the time lag between changes in levels of transcripts, proteins and metabolites. This also fits to the predominant role of SERAT2;2 in Arabidopsis leaves (Watanabe et al. 2008b). However, it cannot be excluded that several regulatory mechanisms such as protein-protein interactions including SERAT-OASTL complex and phosphorylation, which have been described for affecting the activities of SERAT and OASTL and sensitivity of SERAT to the feedback inhibition by cysteine (Liu et al. 2006; reviewed in Wirtz and Hell 2006), might be involved in the diurnal cycles in OAS level. Actually transient increase in OAS levels was also observed in Arabidopsis leaves 5–10 min after transfer to darkness (Caldana et al. 2011), which seems not regulated at the transcriptional level.

### **Regulation Under Stress Conditions**

Nutrient starvation stresses such as sulfate, nitrate and phosphate starvations have been known to share several responsive genes and phenotypes, but specific gene inductions were observed in *SERATs* and *OASTLs* under nutrient starvations. Under sulfur starvation *SERAT3*;2 was highly induced in leaf and root tissues and *SERAT3*;1 was slightly induced in roots and seedling, but the expression levels of the other three *SERATs* and *OASTLs* were not significantly altered except for >2fold change of *SERAT2*;1 and *OASTL2*;1 in seedling (Fig. 3). In contrast to the specific induction of *SERATs* in the group III under sulfur starvation, under other nutrient starvations, for example, under phosphate starvation *SERAT2*;1 and *SERAT2*;2 in the group II were induced (Fig. 3). The transcript levels of the group II *SERATs*, particularly *SERAT2*;1, were also increased under oxidative stress conditions caused by treatments of menadione and H<sub>2</sub>O<sub>2</sub> (Fig. 3). By contrast, the three *OASTLs* and *SERAT3*;1 showed decreases except for slight increase of

Experiment	Tissue	SERAT 1;1	SERAT 2:1	SERAT 2:2	SERAT 3;1	SERAT 3;2	OASTL 1:1	OASTL 2:1	OASTL 2;2	References
-Sulfate	shoot	0.9	0.8	0.9	1.0	40.0	nd	nd	nd	Kawashima et al. (2005)
-Sulfate	root	1.2	0.8	0.9	2.0	45.0	nd	nd	nd	Kawashima et al. (2005)
-Sulfate	root	1.2	1.3	1.1	2.6	nd	1.1	1.3	1.0	Maruyama-Nakashita et al. (2006)
-Sulfate	seedling	1.2	2.1	1.2	2.5	nd	1.0	2.0	0.8	Bielecka et al. (2015)
-Nitrate	seedling	1.1	1.7	1.3	0.7	nd	0.7	0.6	1.2	Scheible et al. (2004)
-Phosphate	seedling	1.6	22	4.5	0.6	nd	1.0	0.7	1.4	Morcuende et al. (2007)
-Potassium	seedling	1.2	1.4	1.3	0.6	nd	1.0	0.6	0.9	Armengaud et al. (2004)
+Menadione	root	2.7	10.3	8.8	0.5	nd	0.7	nd	0.6	Lehmann et al. (2009)
+H2O2	seedling	0.8	17.0	2.3	0.8	nd	0.9	1.0	1.0	Davletova et al. (2005)
							0.125 (	24) 1	8 (21	9

Fig. 3 Response to several stress conditions. The fold change in each stress condition relative to the control condition is shown. *Blue* and *red* colors represent decreases and increases, respectively, as compared with the control plants. *nd* not determined. The specific probe for *SERAT3*;2 gene is not found in ATH1 Affymetrix microarray

OASTL2;2 under nitrate and phosphate starvations. It is noteworthy that the pairs SERAT2;1/SERAT2;2 and SERAT3;1/SERAT3;2 were generated by a gene duplication event in Arabidopsis, respectively (Watanabe et al. 2008b), suggesting that they have kept similar upstream elements and enhancers for the stress inducibility. The high level of SERAT2;1 transcript was also observed in senescing leaves (Fig. 1), where oxidative stress arises with the accumulation of reactive oxygen species (ROS). A key role of SERAT2;1 under oxidative stress was supported by the report that SERAT2;1 interacts with cyclophilin CYP20-3 in plastid, which has peptidyl prolyl isomerase activity for protein folding and assembly (Dominguez-Solis et al. 2009; Park et al. 2013). The formation of SERAT2;1-CYP20-3 complex resulted in activation or stabilization of the SERAT-OASTL complex to produce more OAS, sequentially cysteine and GSH to protect the cell from oxidative stress. Actually the Arabidopsis cyp20-3 mutant with less SERAT activity and thiols was hypersensitive to oxidative stress conditions caused by high light, rose bengal, high salt, and osmotic stress (Dominguez-Solis et al. 2009). The response of SERAT2;1 to oxidative/ROS stress is consistent with the strong diurnal oscillations of SERAT2:1 (Fig. 2) since ROS production is also diurnal-regulated with a peak at midday (Lai et al. 2012). As in the case of the specific role of SERAT2;1 under oxidative stress, the specific induction of SERAT3;1 and SERAT3;2 under sulfate starvations (Fig. 3) and the sensitivity of SERAT3;2 to feedback inhibition by Cys (Kawashima et al. 2005) suggests their distinct roles in sulfur metabolism, especially under sulfur starvation although both SERATs in group III had less capacity to supply OAS in vivo than other isoforms.

The variety of gene expression pattern of *SERATs* and *OASTLs* in plant tissues and environmental conditions, which causes the difference in the ratio of SERAT and OASTL activities, sequentially the balance of OAS and cysteine productions in different subcellular compartments, might be benefit for using substrates such as serine and sulfide efficiently, releasing substrates and products such as sulfide, OAS and cysteine to other compartments and responding to the immediate needs or overcoming transport limitation of OAS and cysteine in specific compartments under specific conditions. Furthermore, the unbalancing in the ratio of SERAT and OASTL (e.g. high SERAT activity and low OASTL activity) might cause a specific OAS accumulation, which induces a specific gene set, the OAS cluster genes (Hubberten et al. 2012).

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# Elucidating the Effects of Higher Expression Level of Cystathionine γ-Synthase on Methionine Contents in Transgenic Arabidopsis, Soybean and Tobacco Seeds

#### Hagai Cohen, Yael Hacham, Ifat Matityahu, and Rachel Amir

**Abstract** Plant seeds accumulate low contents of methionine in their seeds, limiting their nutritional values as a source of proteins. Previous conventional and molecular attempts to increase methionine levels in seeds by classical breeding, selection of mutants or creating 'additional protein sinks' for soluble methionine by expressing methionine-rich seed-storage proteins, have yielded only limited success. Here, we summarize our efforts to increase methionine contents in transgenic Arabidopsis, soybean and tobacco seeds by seed-specific expression of feedback-insensitive mutated forms of the *Arabidopsis thaliana* cystathionine  $\gamma$ -synthase (AtCGS), methionine main regulatory enzyme. Each of these species represents different plant families (Solanaceae, Brassicaceae and Fabaceae). The effects of the manipulations on the levels of soluble and total methionine and on the accumulation of storage compounds in these transgenic seeds are discussed.

#### Introduction

Legumes and cereals supplied as grains and/or forage are among the most important nutritional sources of protein for humans and livestock, but contain limited levels of several essential amino acids. Legume grains are mainly deficient in the sulfur-containing amino acids, methionine and cysteine, while cereals are deficient mainly in lysine, threonine and tryptophan. Combinations of legumes and cereals in diets do not reach the recommended nutritional levels of methionine, which supposed to be 3.5% out of the total consumed amino acids, according to the World Health Organization (WHO 2007). Thus, improving the nutritional quality of seeds by increasing the content of methionine is of critical importance.

The attempts to increase methionine levels in seeds by conventional breeding and selection of mutants have yielded only limited success and were usually

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associated with a significant reduction in yield (Imsande 2001). Numerous molecular approaches were taken to create 'additional protein sinks' for soluble methionine by expressing methionine-rich seed-storage proteins, such as the 2S-albumin proteins from Brazil nut, sunflower and sesame (Altenbach and Simpson 1990; Altenbach et al. 1992; Muntz 1997; Tabe and Higgins 1998; Lee et al. 2003), and the maize zein proteins (Chui and Falco 1995; Hoffman et al. 1987; Kim and Krishnan 2004; Anthony et al. 1997). These transgenic seeds showed no net increases or only slightly higher levels of total methionine indicating that the levels of soluble methionine may limit the synthesis of seed proteins. This assumption encouraged further approaches to elevate the levels of soluble methionine in seeds. One of them is the manipulation of methionine biosynthetic pathway in seeds by expressing cystathionine  $\gamma$ -synthase (CGS), methionine main regulatory enzyme. Mutated forms of the *Arabidopsis thaliana* CGS, which are insensitive to methionine/SAM were used (Hacham et al. 2006; Hanafy et al. 2013).

Here, we summarize our efforts to increase methionine contents in transgenic Arabidopsis, soybean and tobacco seeds expressing one of the mutated forms of AtCGS (Hacham et al. 2006) in a seed-specific manner (Cohen et al. 2014; Song et al. 2013; Matityahu et al. 2013). Each of these species represents different plant families (Solanaceae, Brassicaceae and Fabaceae). The effects of the manipulation on the levels of soluble and total methionine and on the accumulation of storage compounds in these transgenic seeds are discussed.

# Natural Differences in Soluble and Total Methionine Contents in Arabidopsis, Soybean and Tobacco Wild Type Seeds

The analyses highlighted natural differences in soluble and total methionine contents in WT seeds of Soybean (*Glycine max* (L.) Merr. cv Zigongdongdou), tobacco (*Nicotiana tabacum* cv Samsun NN) and Arabidopsis (*Arabidopsis thaliana* ecotype Colombia-0). Soybean and tobacco WT seeds accumulated around 10-fold higher soluble methionine than Arabidopsis seeds (based on nmol per  $g^{-1}$  seed fresh weight; Fig. 1a); possibly since *Glycine* and *Nicotiana* genera belong to the Fabaceae and Solanaceae families, respectively, that accumulate more storage proteins in their seeds, while the *Arabidopsis* genus belongs to the Brassicaceae family that accumulate more oil. Accordingly, the Arabidopsis WT seeds accumulated 56- and 26-fold less total methionine than soybean and tobacco WT seeds, respectively (Fig. 1b), emphasizing natural differences in seed metabolism existing between Arabidopsis and the other two species.

To further understand the natural differences observed in methionine contents in WT seeds from the three species, we also calculated their total/soluble methionine ratios. This highlighted the notion that soybean seeds incorporated most of their methionine into proteins as they accumulate 72-fold higher total methionine than soluble methionine. Tobacco seeds exhibited a 36-fold ratio, while Arabidopsis



Fig. 1 Natural differences in soluble and total methionine contents in Arabidopsis, soybean and tobacco WT seeds. (a) Soluble methionine contents and (b) total methionine contents following protein hydrolysis, measured by GC-MS. (c) Calculated total/soluble methionine ratios. Significance was calculated by the ANOVA one-way test of P < 0.05 and identified by *letters* 

seeds accumulate only 13-fold more total methionine than soluble methionine. These differences might results from different regulation of methionine metabolism, higher usage of soluble methionine into its diverse metabolism, and different incorporation capacity of methionine into seed storage proteins due to different methionine content in their codons (Cohen et al. 2016).

## Seed-Specific Expression of Mutated Forms of AtCGS Leads to Higher Soluble Methionine Content in Transgenic Arabidopsis and Soybean Seeds, but Not in Transgenic Tobacco Seeds

To assess the regulatory roles of CGS in methionine synthesis in seeds of various plant species, we have generated transgenic Arabidopsis, soybean and tobacco seeds, expressing mutated forms of the AtCGS under the control of the seed-specific promoters of phaseolin or legumin B4 (Sundaram et al. 2013; Zakharov et al. 2004). These promoters belong to the most abundant seed-storage protein in the common bean (*Phaseolus vulgaris*) and *Vicia faba* which are stringently turned off during all vegetative stages of plant development (Sundaram et al. 2013). It was also demonstrated that these promoters are induced constitutively during maturation and desiccation stages of seed development (Fait et al. 2011; Zakharov et al. 2004).

Soluble methionine levels were evaluated in the transgenic seeds from the three species, and compared to their respective WT seeds. Transgenic Arabidopsis and soybean seeds accumulated 6- and 1.6-fold higher levels of soluble methionine,



Fig. 2 Seed-specific expression of AtD-CGS leads to differential accumulation of soluble and total methionine contents in transgenic Arabidopsis, soybean and tobacco seeds. (a) Soluble methionine contents and (b) total methionine contents following protein hydrolysis, measured by GC-MS. (c) Calculated fold-changes of soluble and total methionine contents in transgenic seeds compared to their respective WT. *Dashed line* represents the relative levels in the respective WT seeds. Significance was calculated by the Student's *t*-test of P < 0.05 and identified by *asterisks* 

respectively, compared to their respective WT seeds (Fig. 2a). These observations stand in line with previous reports demonstrating that transgenic seeds of *Misuzudaizu* and *Bert* soybean cultivars and azuki beans expressing another mutated form of AtCGS accumulated about two-fold higher levels of soluble methionine (Hanafy et al. 2013).

However, the levels of soluble methionine in the transgenic tobacco seeds did not increase beyond the levels of their respective WT seeds, but was even significantly reduced by about two-fold (Fig. 2a). One possible explanation for the lower soluble methionine found in these seeds is higher catabolism of methionine, as previously reported in transgenic tobacco seeds expressing a key-enzyme of lysine synthesis. These seeds accumulated significantly higher lysine content compared to WT during their development; however, this triggered the expression of lysine catabolic enzyme at later stages of seed development (Karchi et al. 1994). Another possible explanation might be that the manipulation in tobacco seeds triggered the synthesis of additional storage proteins, thus, creating a higher demand to the incorporation of soluble methionine and other amino acids.

### **Transgenic Seeds of All Three Species Accumulate Higher** Levels of Total Methionine in Their Proteins

To reveal whether higher levels of soluble methionine affected total methionine contents in the transgenic seeds, including methionine incorporated into seed proteins, we measured methionine levels after protein hydrolysis. Generally, transgenic seeds from all three species accumulated significantly higher levels of total methionine, where the Arabidopsis and tobacco seeds exhibited 1.6-fold increases and the soybean seeds a 2.3-fold increase (Fig. 2b). The increases in soluble and total methionine contents detected in transgenic soybean seeds were of relatively similar magnitudes (Fig. 2c), suggesting that the excess of soluble methionine due to AtD-CGS expression most probably incorporated into the seed storage proteins. The transgenic Arabidopsis seeds, however, accumulated 6-fold more soluble methionine but only 1.6-fold more total methionine (Fig. 2c). This might be related to the observation that Arabidopsis seeds naturally accumulate less protein, and thus, are characterized by a lower demand for the excess of soluble methionine. Yet, previous observations showed that higher methionine in transgenic Arabidopsis seeds altered the contents of several 12-globulin subunits (Cohen et al. 2016). The transgenic tobacco seeds accumulated 1.6-fold more total methionine even though they exhibited less soluble methionine compared to their respective WT (Fig. 2c). Based on these results, we assume that the levels of soluble methionine in the transgenic tobacco seeds are lower due to demand for protein synthesis, as previously suggested for transgenic seeds with higher levels of soluble lysine, tryptophan and cysteine (Falco et al. 1995; Kita et al. 2010; Kim et al. 2012).

## The Effects of Higher Methionine Levels on the Accumulation of Other Soluble and Total Amino Acids

The total soluble amino acid contents in soybean and tobacco WT seeds are around 10-fold higher than those detected in Arabidopsis WT seeds (Fig. 3a). The highest levels of total amino acid contents were observed in soybean seeds, the tobacco exhibited mid-levels, and the lowest levels were detected in the Arabidopsis seeds (Fig. 3b).

The elevations of soluble methionine contents in transgenic Arabidopsis and soybean seeds were associated with 3.6- and 1.1-fold, respectively significant increases in soluble amino acid contents (Fig. 3c). These findings suggest a yet unknown tight connection between the metabolism of methionine and the accumulation of other amino acids. Such elevations were not observed in transgenic seeds with elevated levels of other amino acids. For example, transgenic soybean seeds with higher levels of total tryptophan and Arabidopsis seeds having higher lysine levels did not show increases in total amino acid contents (Kita et al. 2010; Zhu and



Fig. 3 The effects of higher methionine levels on the accumulation of other soluble and total amino acids. (a) Soluble amino acid contents and (b) total amino acid contents following protein hydrolysis, measured by GC-MS. (c) Calculated fold-changes of soluble and total amino acid contents in transgenic seeds compared to their respective WT. *Dashed line* represents the relative levels in the respective WT seeds. Significance was calculated by the Student's *t*-test of P < 0.05 and identified by *asterisks* 

Galili 2003). This kind of association between the metabolism of methionine and other amino acids is further emphasized in the transgenic tobacco seeds that accumulated less soluble methionine, but also lower soluble amino acid contents compared to their respective WT seeds (Fig. 3a). Higher levels of soluble amino acids can be incorporated into proteins leading to an overall increase in total amino acid contents. Indeed, both transgenic Arabidopsis and soybean seeds exhibited 1.5-and 1.8-fold increases of total amino acid contents, respectively (Fig. 3b, c). The transgenic tobacco seeds also exhibited a significant 1.2-fold increased content of total amino acids, providing a reliable explanation for the lower soluble amino acid contents found in these seeds (Fig. 3b, c).

Altogether, the results suggest that methionine level is a limiting factor of protein synthesis in seeds. Additionally, we observed that when methionine synthesis is enhanced, the levels of most other soluble amino acids increased, enabling them to incorporate into proteins. Yet, further studies are required to define the mechanisms by which methionine metabolism is associated with the accumulation of other amino acids, and to determine if such phenomenon is indeed unique to methionine, or the elevation of other amino acids can also yield higher contents of other soluble and total amino acids.

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# Increased Methionine Levels in Transgenic Seeds Lead to Differential Accumulation of Total Protein and Lipid Contents

The increased amounts of total amino acids incorporated in proteins suggest that the transgenic seeds have higher protein contents. To test this possibility, the levels of total protein contents in seeds were determined according to the Kjeldahl method. As expected, transgenic seeds from all three species exhibited significantly higher protein contents compared to their respective WT seeds (Fig. 4a). Interestingly, even though the transgenic tobacco seeds exhibited the lowest increases total amino acids compared to the transgenic Arabidopsis and soybean seeds, they exhibited the highest net increases in total protein contents, 4.8% (Fig. 4c). The transgenic Arabidopsis and soybean seeds contained 2.2% and 2.1% more protein than their respective WT seeds (Fig. 4c). The observation that no correlation exists between the increases in total amino acid and total protein contents in transgenic tobacco seeds, might be explained by the fact that the first parameter is determined by protein hydrolysis that extracts only the water-soluble protein fractions, while the second by the Kjeldahl method hydrolyses all proteins existing in seeds. Yet the results imply that the regulatory mechanisms underlying the accumulation of protein in tobacco seeds differ than those operate in Arabidopsis and soybean seeds.

Higher protein levels may affect the levels of total lipid contents as another major reserve component in seeds, as reported previously (Hernandez-Sebastia et al. 2005). Indeed, measurements of total lipid contents according to the Soxhlet method revealed a slight significant 2.2% and 1.0% decreases in both transgenic Arabidopsis and soybean seeds, respectively (Fig. 4b, c). The transgenic tobacco seeds, however, exhibited a significant 4% increase in their total lipid contents (Fig. 4b, c), suggesting that the regulatory relationships exist between the metabolism of these reserve compounds in tobacco seeds differ than those in Arabidopsis and soybean seeds.

#### **Conclusions and Future Prospective**

The results presented here suggest a tight link between higher methionine contents during seed development and the accumulation of other soluble amino acids through yet-unknown mechanisms, urging for further research to elucidate the regulation behind these observations. As a result, more soluble amino acids can be incorporated into storage proteins during seed maturation, and the levels of total proteins increase in seeds, affecting the accumulation of total lipids. As the level of soluble methionine in the transgenic seeds remained high, most of it was incorporated into proteins. We assume that other amino acids become rate-limiting for protein synthesis, and thus, methionine cannot be incorporated further into proteins. This might explain the higher soluble methionine levels found in transgenic



Fig. 4 Increased methionine levels in transgenic seeds lead to differential accumulation of total protein and lipid contents. (a) Total protein contents measured according to the Kjeldahl method. (b) Total lipid contents measured according to the Soxhlet method. (c) Net change in total protein and lipid contents in transgenic seeds compared to their respective WT seeds. Significance was calculated by the Student's *t*-test of P < 0.05 and identified by *asterisks* 

Arabidopsis and soybean seeds. Another explanation for this phenomenon is that the capacity of methionine codons in WT seed storage proteins limits the incorporation of the overproduced methionine into proteins. This option was suggested previously for cysteine in lupin plants (Tabe et al. 2010).

WT seeds from the three species exhibit natural differences in the amounts of soluble and total methionine, soluble and total amino acid contents, and the accumulation patterns of total proteins and lipids. This is expected as the three species belong to evolutionary-separated families. For example, soybean seeds as a protein-accumulating legume contain 46% total protein compared to Arabidopsis and tobacco seeds that accumulate 17–22%. On the other hand, these two species contain 36% of total lipids while soybean only 15%. Despite several similarities, when examining the changes occurred due to the genetic manipulation we could clearly distinguish the transgenic tobacco seeds from the Arabidopsis and soybean seeds. It seems that the regulatory mechanisms mediating the effects of AtD-CGS and/or higher methionine in tobacco seeds.

Altogether, the results indicate a way to improve the nutritional quality of crop plants by elevating the levels of methionine and total protein. The data provide new insights into the factors participating in the regulation of methionine and the mechanisms mediating the effects of elevated methionine levels on seed metabolism and composition.

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# **Biosynthesis of S-Alk(en)yl-L-Cysteine Sulfoxides in** *Allium*: Retro Perspective

Naoko Yoshimoto and Kazuki Saito

**Abstract** The biosynthesis of S-alk(en)yl-L-cysteine sulfoxides is one of the most characteristic feature of plants that belong to the genus Allium. Upon tissue damage, these compounds are hydrolyzed by the enzyme alliinase to generate their respective sulfenic acids, which are spontaneously converted to a series of volatile sulfurcontaining compounds with a range of health-beneficial activities. Therefore, the molecular understanding of the mechanism for the biosynthesis of S-alk(en)yl-Lcysteine sulfoxides is important for both basic and applied pharmaceutical researches. Information from chemical analysis and radiolabeling experiments, conducted in the latter half of last century, has suggested that S-alk(en)yl-L-cysteine sulfoxides are biosynthesized from glutathione via  $\gamma$ -glutamyl-S-alk(en)yl-L-cysteines; however, the molecular components that contribute to this biosynthetic pathway and their exact reaction order have long been unclear. Very recently, some genes encoding enzymes involved in the biosynthesis of S-alk(en)yl-L-cysteine sulfoxides have been identified through transcriptome-based approaches, and the characterization of these genes and the encoded enzymes has provided insights into this biosynthetic "black box". Here we briefly summarize the current knowledge on the molecular basis of the generation of bioactive sulfur-containing compounds and the biosynthesis of S-alk(en)yl-L-cysteine sulfoxides in Allium plants.

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#### Introduction

The genus *Allium* is one of the largest plant genera including approximately 700 species (Block 2010). Traditionally, Allium plants, especially garlic, have been cultivated worldwide, not only as flavory crops but also as herbal medicines that are effective in prevention and treatment of various human diseases (Rivlin 2001). The oldest reports of medicinal effects of garlic are found in the Ebers Papyrus, which is an Egyptian medical document of herbal knowledge dating back to the sixteenth century BC. Modern researches have confirmed the medicinal properties of garlic, such as antibacterial, antifungal, antivirus, immunostimulating, antioxidant, cholesterol- and triglyceride-lowering, anti-aggregatory, and hypotensive effects (Iciek et al. 2009). Both the distinctive flavors and the medicinal effects of Allium plants are attributed to their organosulfur compounds. Generally, intact Allium plants are odorless, and it is only when the tissue is damaged that the volatile organosulfur compounds are generated. The origin of these volatile compounds is S-alk(en)yl-L-cysteine sulfoxides, which are non-volatile organosulfur compounds accumulated in the cell. When the tissue is damaged, S-alk(en)yl-L-cysteine sulfoxides are enzymatically hydrolyzed to generate corresponding sulfenic acid, which is spontaneously converted into a series of sulfur-containing compounds with various medicinal properties (Jones et al. 2004; Rose et al. 2005). Because of the chemical diversity and the medicinal properties of sulfur-containing compounds generated from S-alk(en)yl-L-cysteine sulfoxides, Allium plants have attracted considerable interest from both chemists and biologists. In this chapter, we describe the current molecular understanding of the production of bioactive sulfurcontaining compounds from S-alk(en)yl-L-cysteine sulfoxides and the biosynthesis of S-alk(en)yl-L-cysteine sulfoxides in Allium plants.

# S-Alk(en)yl-L-Cysteine Sulfoxides and Their Decomposition by Alliinase

The origin of flavor and medicinal compounds in *Allium* plants has been investigated from the middle of the twentieth century. The first identified flavor precursor from *Allium* plants was *S*-allyl-L-cysteine sulfoxide (alliin) isolated from garlic (Stoll and Seebeck 1948, 1951). Further studies revealed that *Allium* plants generally contain high concentrations of *S*-alk(en)yl-L-cysteine sulfoxides as flavor precursors. To date, four major *S*-alk(en)yl-L-cysteine sulfoxides, alliin, *S*-trans-1propenyl-L-cysteine sulfoxide (isoalliin), *S*-propyl-L-cysteine sulfoxide (propiin), and *S*-methyl-L-cysteine sulfoxide (methiin), have been identified from *Allium* plants (Fig. 1a; Jones et al. 2004; Rose et al. 2005). The abundance ratio of these four sulfoxides is different among *Allium* species (Block 2010; Fritsch and Keusgen 2006). For example, alliin is a major component in garlic (*Allium sativum*), isoalliin



**Fig. 1** Chemical structures of *S*-alk(en)yl-L-cysteine sulfoxides in *Allium* plants. (**a**) The four major *S*-alk(en)yl-L-cysteine sulfoxides. (**b**) The stereochemistry of (+)-alliin [( $R_CS_S$ )-*S*-allylcysteine sulfoxide] and (-)-*allo* alliin [( $R_CR_S$ )-*S*-allylcysteine sulfoxide]

is characteristic of onion (*Allium cepa*), methiin is found in most *Allium* plants, and propiin is a minor component in most *Allium* plants.

In intact Allium plants, S-alk(en)yl-L-cysteine sulfoxides are physically separated from the enzyme alliinase (EC. 4.4.1.4; Stoll and Seebeck 1949a,b), which is a pyridoxal 5'-phosphate-dependent C-S lyase that can hydrolyze S-alk(en)yl-L-cysteine sulfoxides. While S-alk(en)yl-L-cysteine sulfoxides are accumulated in the cytosol of the storage mesophyll cells, alliinase is confined to the vacuole of the vascular bundle sheath cells at least in garlic, onion, and Allium tuberosum, as revealed by the subcellular fractionation and the immunostaining analyses (Ellmore and Feldberg 1994; Lancaster and Collin 1981; Yamazaki et al. 2002). When their tissue is damaged by cutting or crushing, alliinase is released from the bundle sheath cells, and comes into contact with S-alk(en)yl-L-cysteine sulfoxides. Alliinase immediately cleaves the C-S bond of S-alk(en)yl-L-cysteine sulfoxides to generate their respective sulfenic acids, which are rapidly converted to thiosulfinates, such as allicin (diallyl thiosulfinate), by the non-enzymatic selfcondensation. Thiosulfinates, which are chemically unstable, undergo further spontaneous reactions, yielding a variety of sulfur-containing bioactive compounds (Jones et al. 2004; Rose et al. 2005). Representative sulfur-containing compounds produced from alliin and the plausible mechanism of production of them are shown in Fig. 2 (El-Aasr et al. 2011; Nohara et al. 2012, 2013, 2014, 2015).



Fig. 2 Alliinase-mediated hydrolysis of alliin and plausible route for the production of various sulfur-containing compounds

# The Biosynthetic Pathway for S-Alk(en)yl-L-Cysteine Sulfoxides

The biosynthetic pathway for S-alk(en)yl-L-cysteine sulfoxides in Allium plants has been proposed based on chemical analysis and radiotracer studies that fed [<sup>35</sup>S] sulfate or  $[^{14}C]$  amino acids, in the latter part of the twentieth century (Ettala and Virtanen 1962; Lancaster and Shaw 1989; Suzuki et al. 1961, 1962; Turnbull et al. 1980). The origin of sulfur atom in S-alk(en)yl-L-cysteine sulfoxides is mainly sulfate, as [<sup>35</sup>S] was found in S-alk(en)yl-L-cysteine sulfoxides in onion, garlic, and Allium siculum, which were supplied with [<sup>35</sup>S] sulfate (Suzuki et al. 1961; Lancaster and Shaw 1989). This was further supported by the finding that the application of  $CaSO_4$  to soil as a sulfur fertilizer resulted in the increase in the concentration of alliin in garlic (Arnault et al. 2003). As in other plant species, Allium plants take up sulfate from the soil solution by the functions of sulfate transporters, and biosynthesize cysteine through the reductive sulfate assimilation pathway. Cysteine residue in glutathione or  $\gamma$ -glutamylcysteine is S-alk(en)ylated, and the biosynthetic intermediate  $\gamma$ -glutamyl-S-alk(en)yl-L-cysteine undergoes deglutamylation and S-oxygenation to give S-alk(en)yl-L-cysteine sulfoxide (Fig. 3; Lancaster and Shaw 1989). The order of deglutamylation and S-oxygenation might be different among Allium species and/or among four S-alk(en)yl-Lcysteine sulfoxides; the pulse-labeling experiment suggested that S-oxygenation likely occur before deglutamylation in isoalliin biosynthesis in onion (Lancaster and Shaw 1989), while in vitro determination of catalytic activities of deglutamylation enzymes and S-oxygenation enzyme suggested that deglutamylation likely occur prior to S-oxygenation in alliin biosynthesis in garlic, as described later in this chapter (Yoshimoto et al. 2015a, b).

By contrast, the origin of *S*-alk(en)yl groups remains obscure. From garlic and onion, *S*-(2-carboxypropyl)glutathione has been isolated and is suggested as an intermediate for the biosynthesis of alliin, isoalliin, and propiin. Feeding of uniformly labeled [<sup>14</sup>C] valine to garlic resulted in the production of [<sup>14</sup>C]-labeled *S*-(2-carboxypropyl)glutathione and its derivative, *S*-(2-carboxypropyl)cysteine (Suzuki et al. 1962). Similarly, onion that was exposed to [<sup>14</sup>C] valine produced [<sup>14</sup>C]-labeled *S*-(2-carboxypropyl)cysteine (Turnbull et al. 1980). These findings strongly suggested that compound synthesized from valine *in vivo* is the origin of *S*-alk(en)yl group. Based on the chemical structure of *S*-(2-carboxypropyl)glutathione, methacrylyl-CoA, which is known to be synthesized from valine in animal (Shimomura et al. 2004), is suggested to be a plausible candidate compound for the origin of *S*-alk(en)yl group of alliin, isoalliin, and propiin (Suzuki et al. 1962). However, to the best of our knowledge, the presence of methacrylyl-CoA in *Allium* plants has not been confirmed. The origin of *S*-methyl group for methiin is also unclear and needs to be elucidated.



Fig. 3 Putative alliin biosynthetic pathway in garlic

# Enzymes Involved in the Biosynthesis of S-Alk(en)yl-L-Cysteine Sulfoxides

In the proposed pathway for the biosynthesis of *S*-alk(en)yl-L-cysteine sulfoxides from glutathione, enzymes responsible for the following four steps, respectively, are suggested to catalyze: (1) *S*-alk(en)ylation, (2) removal of glycyl group, (3) removal of  $\gamma$ -glutamyl group, and (4) *S*-oxygenation (Fig. 3). Recent studies identified the enzymes catalyzing the removal of  $\gamma$ -glutamyl group and the enzyme catalyzing *S*-oxygenation.

# $\gamma$ -Glutamyl Transpeptidases Catalyze Deglutamylation Reaction

The most extensively investigated enzymatic step among these four steps is the removal of  $\gamma$ -glutamyl group from the biosynthetic intermediates.  $\gamma$ -Glutamyl

transpeptidase (GGT; EC 2.3.2.2) is the enzyme widespread both in prokaryotes and eukaryotes, which catalyzes the transfer of the  $\gamma$ -glutamyl group from  $\gamma$ -glutamyl compounds to the acceptor molecules such as water, amino acids, and small peptides (Tate and Meister 1981). In Arabidopsis, GGT functions in the breakdown of oxidized glutathione in the extracellular space, in the breakdown of glutathione *S*-conjugates of xenobiotics, and presumably in camalexin biosynthesis that requires the removal of  $\gamma$ -glutamyl group from the precursor molecule (Ohkama-Ohtsu et al. 2007a, b, 2011; Su et al. 2011).

Because of its catalytic properties and the potential contribution to the biosynthesis of natural products in non-*Allium* plants, involvement of GGT as the deglutamylation enzyme in the biosynthesis of *S*-alk(en)yl-L-cysteine sulfoxides in *Allium* plants has been proposed. In fact, a GGT partially purified from sprouting onion bulbs showed high substrate specificity for the biosynthetic intermediates of *S*-alk(en)yl-L-cysteine sulfoxides (Lancaster and Shaw 1994), supporting the idea that GGT catalyzes deglutamylation in the biosynthesis of *S*-alk(en)yl-L-cysteine sulfoxides. However, another onion GGT, which was purified to homogeneity from sprouting bulbs, showed a high-affinity to glutathione and glutathione *S*-conjugates, but utilized  $\gamma$ -glutamyl-*S*-propenyl-L-cysteine sulfoxide as a poor substrate (Shaw et al. 2005), indicating that onion has at least two distinct GGT proteins with different *in vivo* functions.

More recently, three genes encoding GGTs, *AsGGT1*, *AsGGT2*, and *AsGGT3*, have been identified from garlic, by homology searching of *Allium* EST database and by using degenerate primers designed based on the conserved regions of known plant GGTs (Yoshimoto et al. 2015a). In the proposed pathway for alliin biosynthesis, a biosynthetic intermediate  $\gamma$ -glutamyl-*S*-allyl-L-cysteine is deglutamylated and *S*-oxygenated to yield alliin (Fig. 3), although the order of deglutamylation and *S*-oxygenation has long been unclear. Recombinant proteins of AsGGT1, AsGGT2, and AsGGT3 catalyzed deglutamylation of  $\gamma$ -glutamyl-*S*-allyl-L-cysteine to yield *S*-allyl-L-cysteine, with the apparent  $K_m$  values for  $\gamma$ -glutamyl-*S*-allyl-L-cysteine of 86  $\mu$ M, 1.1 mM, and 9.4 mM, respectively. By contrast, these GGT proteins showed very low deglutamylation activity toward the corresponding sulfoxide,  $\gamma$ -glutamyl-*S*-allyl-L-cysteine sulfoxide, which is another possible biosynthetic intermediate of alliin, suggesting that deglutamylation occurs prior to *S*-oxygenation in the biosynthesis of alliin in garlic.

# Flavin-Containing Monooxygenase Catalyzes Highly Stereoselective S-Oxygenation Reaction

The other enzymatic step that has been investigated at the molecular level is *S*-oxygenation, particularly in the biosynthesis of alliin in garlic. This step occurs in a highly stereoselective manner, since almost all *S*-alk(en)yl-L-cysteine sulfoxides contained in *Allium* plants have (*S*)-stereochemistry at their sulfur atoms. For

example, in case of alliin in garlic, (+)-alliin  $[(R_CS_S)-S-allylcysteine sulfoxide]$  is the major natural form, while (-)-*allo* alliin  $[(R_CR_S)-S-allylcysteine sulfoxide]$ , one of the diastereomers of (+)-alliin, can be detected only at trace levels (Fig. 1b; Block 2010).

Previous findings that some animal flavin-containing monooxygenase (FMO; EC 1.14.13.8) can catalyze the S-oxygenation of S-allyl-L-cysteine to yield alliin (Krause et al. 2002; Novick and Elfarra 2008; Ripp et al. 1997), and the fact, that five Arabidopsis FMOs classified into plant clade III FMO are responsible for Soxygenation of S-methylthioalkyl glucosinolates to yield their respective sulfoxides (Hansen et al. 2007; Li et al. 2008), suggested the possibility that an FMO is responsible for the S-oxygenation reaction in the biosynthesis of S-alk(en)yl-Lcysteine sulfoxides in Allium plants. Based on homology searching of Allium EST database, AsFMO1, which encodes a clade III FMO, has been identified from garlic (Yoshimoto et al. 2015b). Recombinant AsFMO1 catalyzed highly stereoselective S-oxygenation of S-allyl-L-cysteine to nearly exclusively yield (+)-alliin, with the apparent  $K_m$  value of 0.25 mM. The S-oxygenation activity of AsFMO1 was much higher against S-allyl-L-cysteine than against y-glutamyl-Sallyl-L-cysteine, which is another possible biosynthetic intermediate. Together with the substrate specificity of AsGGT1, AsGGT2, and AsGGT3, the substrate preference of AsFMO1 strongly suggested that, in garlic, the biosynthetic intermediate  $\gamma$ -glutamyl-S-allyl-L-cysteine is deglutamylated by GGTs to yield S-allyl-L-cysteine, which is further S-oxygenated by AsFMO1 to yield (+)-alliin (Fig. 3).

### Expression Profiles of AsGGT1, AsGGT2, AsGGT3, and AsFMO1, and Their Possible In Vivo Roles in Garlic

In growing garlic plants with green foliage leaves, alliin is suggested to be actively biosynthesized from glutathione in green foliage leaves (Bloem et al. 2004; Koch and Lawson 1996; Ueda et al. 1991). During bulb formation, alliin in senescing foliage leaves is presumably translocated to developing bulbs (Bloem et al. 2004). Consequently, mature bulbs contain high amounts of alliin (Ichikawa et al. 2006a). In mature bulbs, the biosynthetic intermediate  $\gamma$ -glutamyl-S-allyl-L-cysteine is also accumulated at high levels (ca. 26 mM), as a storage peptide, especially in storage leaves (Matsuura et al. 1996; Ichikawa et al. 2006a, b; Yoshimoto et al. 2015a).  $\gamma$ -Glutamyl-S-allyl-L-cysteine in storage leaves of bulbs is stored during dormancy, while it is immediately converted to alliin via two enzymatic reactions, deglutamylation and S-oxygenation, during sprouting (Ichikawa et al. 2006b). Alliin produced in storage leaves during sprouting is likely to be translocated to white foliage leaves inside the bulbs (Yoshimoto et al. 2015b). The two-step enzymatic production of alliin from γ-glutamyl-S-allyl-L-cysteine may enable rapid production of high amounts of alliin, which is effective in protecting sprouts from microorganisms and animals.
Gene expression profiles of *AsGGT1*, *AsGGT2*, *AsGGT3*, and *AsFMO1* suggested their possible *in vivo* roles in garlic plants. In pre-emergent nearly sprouting bulbs, the mRNA levels of *AsGGT1*, *AsGGT2*, *AsGGT3*, and *AsFMO1* were higher in storage leaves than in white foliage leaves inside the bulbs (Yoshimoto et al. 2015b). This fact suggested the important roles of AsGGT1, AsGGT2, AsGGT3, and AsFMO1 in synthesizing alliin from  $\gamma$ -glutamyl-*S*-allyl-L-cysteine stored at high levels in storage leaves. Chemical analysis suggested that alliin produced in this process is presumably translocated from storage leaves to white foliage leaves inside the bulbs.

By contrast, in sprouted bulbs, AsGGT1 mRNA accumulated at high levels in green foliage leaves, AsGGT2 mRNA accumulated in roots, AsGGT3 mRNA was abundant in storage leaves, and AsFMO1 mRNA accumulated in various tissues at similar levels (Yoshimoto et al. 2015b). These findings suggested that the contribution of the three GGTs to alliin biosynthesis is different among tissues, while AsFMO1 functions in alliin biosynthesis in various tissues, at this growth stage. Interestingly, the expression patterns of the three GGTs seem to be closely associated with their affinity for  $\gamma$ -glutamyl-S-allyl-L-cysteine and the concentration of  $\gamma$ -glutamyl-S-allyl-L-cysteine in tissues. In green foliage leaves, the concentration of  $\gamma$ -glutamyl-S-allyl-L-cysteine is approximately 38  $\mu$ M (Matsuura et al. 1996; Yoshimoto et al. 2015a). AsGGT1, which has highest affinity for  $\gamma$ -glutamyl-Sallyl-L-cysteine ( $K_{\rm m} = 86 \,\mu {\rm M}$ ) among the three GGTs, is suggested to be the most important GGT responsible for the biosynthesis of alliin from glutathione in green foliage leaves. In storage leaves of sprouted bulbs, AsGGT3, which has lowest affinity for  $\gamma$ -glutamyl-S-allyl-L-cysteine ( $K_{\rm m} = 9.4 \text{ mM}$ ) among the three GGTs, is suggested to be responsible for deglutamylating  $\gamma$ -glutamyl-S-allyl-L-cysteine present at high levels even after sprouting. AsGGT2, which has intermediate affinity for  $\gamma$ -glutamyl-S-allyl-L-cysteine ( $K_{\rm m} = 1.1$  mM), may contribute to both the biosynthesis of alliin from glutathione in green foliage leaves and the deglutamylation of  $\gamma$ -glutamyl-S-allyl-L-cysteine in storage leaves. The different organ-specificity of the three GGTs with different kinetic properties may enable the production of optimized levels of S-allyl-L-cysteine in each tissue, which is further S-oxygenated by constitutively-expressing AsFMO1 to yield alliin. Future investigations of transgenic plants with altered expression levels of AsGGT1, AsGGT2, AsGGT3, and AsFMO1, respectively, will provide a better understanding of their in vivo functions.

#### **Conclusions and Future Perspectives**

In the past few years, significant advances have been made in our understanding of the molecular basis for the biosynthesis of *S*-alk(en)yl-L-cysteine sulfoxides in *Allium* plants, especially for the biosynthesis of alliin in garlic. Three GGT genes encoding deglutamylation enzymes and one FMO gene encoding *S*-oxygenation enzyme have been identified based on the mining of transcriptome data, and the

substrate specificities of their encoded proteins suggested the reaction order of deglutamylation and S-oxygenation in alliin biosynthesis in garlic. However, enzymes catalyzing the removal of glycyl group or S-alk(en)ylation have not been identified and await discovery. Although the quantification analysis of S-alk (en)yl-L-cysteine sulfoxides in garlic and onion suggested that S-alk(en)yl-L-cysteine sulfoxides are translocated among organs (Bloem et al. 2004; Lancaster et al. 1986; Yoshimoto et al. 2015b), the transport proteins mediating these transport processes have not been identified. Regulatory factors controlling the rate of biosynthesis also remain to be determined. Recent advances in omics technologies and computational systems biology will accelerate the identification of additional members contributing to the production and the accumulation of S-alk(en)yl-Lcysteine sulfoxides in Allium plants. Elucidation of the difference in the biosynthesis among Allium species is the area also requiring further investigation. Application of genome editing technologies to Allium plants will facilitate not only a better understanding of *in vivo* functions of genes involved in the biosynthesis of S-alk(en) yl-L-cysteine sulfoxides, but also the molecular breeding of plants beneficial for human health.

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## The Effect of Sulfur Nutrition on Glucosinolate Patterns and Their Breakdown Products in Vegetable Crops

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Abstract Glucosinolates (GSLs) are amino acid derived secondary metabolites naturally occurring in the order of Brassicales. They represent an important class of phytochemicals involved in plant-microbe, plant-insect, plant-animal and planthuman interactions. In Brassica vegetables GSL are known as the bioactive compounds giving the typical flavor and odor, being involved in natural pest control. Still, in high doses GSL remain highly toxic. Even though the GSL content in Brassica species is genetically fixed, breeding programs already aimed for reducing the GSL content, with the engineering of 00-varieties of rapeseed (*Brassica napus*) being the most prominent example. Contrary to their negative effects, GSLs are also discussed to have beneficial nutritional and health effects. But it is more their breakdown products, particularly isothiocyanates (ITCs) and nitriles, formed after hydrolysis within the glucosinolate-myrosinase-system, which the healthpromoting effects can be ascribed to when taken up in low doses. Besides genetic approaches to influence GSL content and pattern and their breakdown products, little is yet known about how agronomic and particularly plant nutritional factors can alter the GSL content and pattern of their different hydrolysis products in the context of improving food quality. Therefore, the influence of the sulfur (S) supply on GSLs, ITCs and nitriles in various Brassica species, such as Indian mustard (Brassica juncea), kohlrabi (Brassica oleracea), and Chinese cabbage (Brassica rapa spp. pekinensis), are exemplarily discussed in relation to nitrogen nutrition.

#### Introduction

Glucosinolates (GSLs) have been part of human life for thousands of years, as being found in edible plants belonging to the order of Capparales, including the family of *Brassicaceae*. Within this family it is mainly the vegetables, such as broccoli, cauliflower, radish, and Brussels sprouts, which play an important role in today's

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diet, not least due to their distinct flavors and tastes (Charron et al. 2005). In 2011, a worldwide production of 100 million tons of *Brassica* vegetables was recorded (FAOSTAT 2013), while the average daily intake of GSLs was estimated to be approx. 14–15 mg per person in Germany (Schonhof et al. 2004; Steinbrecher and Linseisen 2009). Though treated as unfavorable compound in the past because of their putative goitrogenic effect, GSLs have generated considerable pharmacological interest as components of human food not least since epidemiological studies confirmed the positive effect of vegetable consumption on human health.

Besides their effects in human metabolism, GSLs and their breakdown products are known as allelochemicals. Thus, they play an important role in plant defenses, inhibiting, for example, microbial growth (Glenn et al. 1988). Moreover, they are an intermediate in the biosynthesis of indole phytoalexins (Griffiths et al. 1994), or stimulate oviposition and feeding by insects (Mewis et al. 2005). Glucosinolates also act as feeding deterrents for polyphagous herbivores and feeding stimulants for specialist insects in cruciferous crops (Bartlet et al. 1999). The traditional use of crucifers as green manures with phytosanitary effects is partly based upon the toxic nature of the glucosinolate degradation products, which serve to reduce the soil inoculum of pathogens and pests for subsequent agricultural and horticultural crops (Mithen et al. 2000).

Glucosinolates, which belong to the mustard oil glucosides, are a large group of secondary metabolites in plants, containing abundant nitrogen (N) and sulfur (S). Not least because of their enrichment in sulfur, it is contradictorily discussed whether GSLs might play an important role as a S-reservoir in times of sulfur deficiency to maintain ordinary metabolism (Schonhof et al. 2007; Gerendás et al. 2009). Today, more than 120 different GSL have already been identified, which all comprise a B-D-thioglucose group, a sulphonated oxime moiety (glucone) and a variable side-chain (aglycone) derived from one of eight amino acids (Fig. 1). Accordingly, GSLs are classified by their precursor amino acids and further modifications of the side chains. While GSLs derived from alanine, leucine, isoleucine, methionine or valine characterize aliphatic GSL, aromatic GSLs derive from phenylalanine and tyrosine, and tryptophane is part of indole GSLs (Mithen et al. 2000; Fahey et al. 2001). Additionally, most R groups are elongated by methylene moieties and are subjected to further transformations such as hydroxylation, methylation, desaturation or glycosylation (Halkier and Gershenzon 2006). Eventually, it is just this side chain modification that can have a significant impact on the biological activity of the GSLs and their various hydrolysis products (Brown and Morra 1995). However, there is only limited information on the specific effect of each compound and their breakdown products.

#### **Glucosinolates and Their Breakdown Products**

In plant tissues, GSLs remain chemically stable within the cytoplasm until they come into contact with the enzyme myrosinase, which is located in the vacuole. After physical damage of plant tissues, e.g., by cutting, chewing, milling or



Fig. 1 Biosynthesis and general structure of glucosinolates. PAPS 3'-phosphoadenosine-5'-phosphosulfate

penetration, this effective compartmentation is broken down, thus releasing myrosinase, which subsequently leads to the hydrolytic breakdown of GSLs in an ascorbic acid activated manner (e.g., Bones and Rossiter 1996; Kleinwächter and Selmar 2004). This process, also referred to as myrosinase-glucosinolate-system, is regarded as a defense system against any herbivore attack. The myrosinase (EC 3.2.1.147), which belongs to the family of glucoside hydrolases, catalyzes the cleavage d-glucose from GSLs in the presence of water resulting in the formation of thiohydroximate-O-sulfonate (Fig. 2). In a next step, the aglycone undergoes a spontaneous Lossen-like rearrangement, releasing sulfate. Depending on various physiological conditions such as pH or the presence of cofactors, this molecule is further decomposed to form nitriles, epithionitriles, thiocyanates and isothiocyanates (ITCs), which all, but isothiocyanates in particular, contribute to



Fig. 2 Formation of enzymatic breakdown products within the glucosinolates-myrosinase system. *ESP* epithiospecifier protein, *TFP* thiocyanate-forming protein, *ESM* epithiospecifier modifier protein, *NSP* nitrile-specifier proteins

the typical smell and flavor of *Brassica*-vegetables (Kliebenstein et al. 2005; Krumbein et al. 2001).

In human nutrition, GSLs and their hydrolysis products were originally considered as antinutritives and being toxic (Mithen et al. 2000; Vermorel et al. 1988), with nitrile depressing growth and causing liver and kidney lesions, thiocyanates inhibiting iodine uptake by the thyroid, and isothiocyanates having a strong antithyroid effect (Holst and Williamson 2004; Bonnesen et al. 2001). Especially, effects of the latter can be traced back to the formation of oxazolidine-2-thiones (Fig. 2), goitrogenic compounds which are all products from chemically unstable isothiocyanates (Hanschen et al. 2014). However, actual human clinical studies confirm that critical threshold levels for uptake are hardly met in human nutrition (Fahey et al. 2001), which is mainly a result of enzymatic degradation of GSL breakdown products, protein cofactors, and processing conditions such as cooking prior to consumption (McMillan et al. 1986). Only in some parts of the world, where a low iodine supply is a problem, or in animal diets dominated by *Brassica*originated fodder and feed, may these adverse effects still play a role (Tripathi and Mishra 2007). Today, GSLs and their breakdown products are even discussed among other bioactive substances to be linked, for example, with a reduced risk of developing several types of cancer and reduction of the cholesterol level, as well having an antimicrobial effect in humans (Fahey et al. 2001; Mithen et al. 2000; Watzl and Leitzmann 2005). As could be confirmed in several animal and human studies, the anticarcinogenic effect of isothiocyanates is closely related to the suppression and inhibition of phase-I enzymes together with a phase-II enzyme induction in the initiation process of carcinogenesis (Conaway et al. 1996; Boddupalli et al. 2012; James et al. 2012; Hecht 2000). To benefit from such health promoting effects, it is therefore not only desirable to foster an increased consumption of *Brassica*-vegetables, but also to improve the 'health quality' of plants. Increasing the content of bioactive compounds can, for example, be accomplished by including new plant breeding and management strategies, but an even more promising tool seems to be changing and adapting fertilization strategies for *Brassica* species.

# Impact of Nitrogen and Sulfur on Glucosinolates and Their Breakdown Products

In general, the GSL content in plant tissue largely depends on pre-harvest variation (i.e., genotype selection, developmental stage, cultivation period; Schreiner 2005). This is always associated with ecophysiological factors, such as the climate parameters of irradiation and temperature, which exert an enormous impact on GSL accumulation in plants (Mithen et al. 2000; Fahey et al. 2001; Fallovo et al. 2009). Of all environmental factors nutrient management (e.g., nitrogen and sulfur status) has been shown to significantly affect the GSL content in plant tissue (De Pascale et al. 2007; Chen et al. 2006; Zhou et al. 2013). A strong interactive effect between N and S on the GSL content can be expected as their side chains are derived from S-containing or S-free, but in any, case N-containing amino acids, the thiol bridge to glucose from cysteine, and the sulfonate group from phosphoadenosin-phosphosulfate (Fahey et al. 2001). Further GSLs are still discussed as a transient S-reservoir in plants (Bones and Rossiter 1996; Schnug et al. 1995), assuming that under S deficiency GSL concentration may decrease leading to accumulation of their breakdown products (i.e., ITC), presumably resulting from less efficient compartmentalization of GSLs from myrosinase. This is in contrast to other studies, which is nowadays generally disproved (e.g., Gerendás et al. 2008b; 2009;). In addition, sulfate released by myrosinase activity (Fig. 2) is easily consumed by metabolic processes but any further conversion of low molecular weight ITCs may be restricted due to their inefficient compartmentalization. Although energy coupled uptake by protoplasts (Chen and Halkier 2000) and site-directed translocation of intact GSLs have been documented (Brudenell et al. 1999), this seems not to be the case for ITCs. The effect of S limitation may be particularly strong in plants grown with marginal S but ample N supply.

Indeed, while *Brassica nigra* deficient in S but sufficient in N showed lower levels of allyl isothiocyanates, thus leading to increased feeding by *Spodoptera eridania*, plants supplied with high S but low N showed a reduced infestation (Wolfson 1982). This is in line with studies on *Brassica napus*, which showed that high N and low S can minimize the GSL content of seeds (Zhao et al. 1993,

1994). It can be reasoned that the effect of N supply largely depends on the S supply. Whenever the soil was deficient in S, N fertilization decreased the GSL content, whereas N addition to adequately S-supplied soils resulted in either increased GSL concentration or had no effect (Zhao et al. 1993; Bilsborrow et al. 1993). Palaniswamy et al. (1995) also showed that when decreasing the N:S ratio in the growth medium, the phenylethyl isothiocyanate concentrations increased in broccoli florets upon S fertilization (Krumbein et al. 2001). These results are in contrast to that of Vallejo et al. (2003) who reported no effect of S supply on the GSL content of broccoli. However, the soil S status needs to be taken into account since already well S supplied soils will not respond to S fertilization, thus having no effect on GSL synthesis.

#### Effect of S Supply on GSL and ITC in Chinese Cabbage

Up to date, numerous studies investigated the general effect of the N and S supply on GSLs and their breakdown products in *Brassica* vegetables, but only few were designed to show effects of incremental increases of S fertilization (Schnug 1990; Gerendás et al. 2008a, b, 2009; Schnug and Haneklaus 2016). Thus, the minimum level of S supply after which a significant increase in GSL and GSL pattern becomes obvious was investigated. Furthermore, the effect of increasing S on GSL hydrolysis products, especially the ITCs, was investigated. Additionally, it aimed to clarify whether GSLs might serve as S source under limited S conditions.

As Chinese cabbage (*Brassica rapa* spp. *pekinensis*) is one of the most popular leafy-cabbage vegetables in mainland China and is increasingly being used in the western, Mediterranean and American cuisine, it seemed to be an adequate trial plant besides other *Brassica* vegetables. Chinese cabbage has a long history in being used in Traditional Chinese Medicine, not least due to its known high nutritional value and health promoting effects by thiocyanates and isothiocyanates amongst others (USDA 2015).

Being a S indigent plant, Chinese cabbage responded well to increasing levels of S (MgSO<sub>4</sub>; S1: 0 g; S2: 0.01 g; S3: 0.03 g; S4: 0.06 g; S5: 0.1 g; S6: 0.3 g; per 6 kg quartz sand; all other nutrients in optimum), following the classical Law of the Optimum (data not shown, see also Gerendás et al. 2008a, 2009. Surprisingly, increasing S supply did not result in a similar steady incline of total S in plant tissue but more successive increase in the order: S1, S2, S3 < S4, S5 < S6 (Fig. 3a). As there are strong interactive effects of S and N in plant metabolism (e.g., nutrient uptake, protein synthesis), it was expected that N is also responding to increasing S, but which was only the case in the highest S treatments (Fig. 3b). The N/S ratio is frequently used to characterize the nutritional status of crops ((Reuter and Robinson 1988; Bergmann 1993) (Fig. 3c; Ahmad et al. 1999; Schonhof et al. 2007; Gerendás et al. 2009).



**Fig. 3** Effect of sulfur (S) supply on sulfur concentration (**a**), nitrogen concentration (**b**), and N/S ratio (**c**) in *Brassica rapa* spp. *pekinensis* (MgSO<sub>4</sub>; S1 = 0 g, S2 = 0.01 g, S3 = 0.03 g, S4 = 0.06 g, S5 = 0.1 g, S6 = 0.3 g)

Due to the dependence of GSLs on S, it is not surprising that GSL synthesis is attenuated under limited S conditions (Fig. 4a), but responded well to increasing supply (Mithen et al. 2000). Consequently, also the GSL showed the same successive increase as did the S (Fig. 3a) concentration. However, the GSL pattern varied with enhanced S supply. In total, eight GSLs could be detected: aromatic GSL–glucotropaeolin (GTL); aliphatic GSL–progoitrin (PRO), gluconapin (GNA), glucobrassicin (GBC), 4-methoxy glucobrassicin (4MG), and neoglucobrassicin (NGBS), which all increased upon S supply. This is in line with Gerendás et al. (2008a, b, 2009) in which plants were supplied with moderate S, even though here (Fig. 3) plants were grown without S before the initial S-fertilization.

In this context, aromatic and aliphatic GSLs were more responsive to S (Fig. 4b), which might be attributable to the fact that more thiols are provided under optimum S conditions (Falk et al. 2007). In contrast, it is described that indole GSLs are



**Fig. 4** Effect of increasing sulfur supply on concentration of (**a**) total GSL and (**b**) GSL classes in *Brassica rapa* spp. *pekinensis*. Aromatic GSL–glucotropaeolin; Aliphatic GSL–progoitrin + glucobrassicin; Indolic GSL–glucobrassicanapin + neoglucobrassicin + 4-hydroxyglucobrassicin + 4-methoxyglucobrassicin (MgSO<sub>4</sub>; S1 = 0 g, S2 = 0.01 g, S3 = 0.03 g, S4 = 0.06 g, S5 = 0.1 g, S6 = 0.3 g)

positively affected by increasing N (Omirou et al. 2009) due to the nitrogenous origin of tryptophan, the precursor amino acid for indole GSLs (Fig. 1) (Kim et al. 2002; Falk et al. 2007). Thus, GBC, NGBS, and 4-MG dominated in S deficient plants, whereas with increasing S supply the pattern changed with a tendency to GBN, GNA, PRO, and GTL. Nonetheless, indole GSLs dominated even under high S supply, possibly due to the constant and sufficient N supply (Fig. 4b).

Based on the positive effect of S fertilization on the GSL content and with regard to possible health promoting effects, it was further questioned whether the ITC profile corresponds to the GSL profile. Like GSLs, the concentration of their breakdown products showed an increasing trend especially in the high S treatments (S4, S5, S6; Fig. 5a). This is in accordance with the results of Gerendás et al. (2008a) who demonstrated that in kohlrabi a low N/S ratio led to a high concentration of GSL breakdown products. However, total concentration of GSL hydrolysis products was lower than detected by Gerendás et al. (2008b) in cress. After all, differences between species are quite likely. Of all GSL breakdown products, a total of four ITCs and one nitrile could be identified, and the profile was dominated by 2-methylpropyl isothiocyanate (MePrITC), followed by 3-butenyl isothiocyanate (ButITC), 2-phenylethyl isothiocyanate (PEITC), and 4-pentenyl isothiocyanate (PentITC) (Fig. 5b-f) resulting from the hydrolysis of 2-methylpropyl GSL, gluconapin, gluconasturtiin, and glucobrassicanapin. All responded in a similar way, showing concentrations tending upwards upon increasing S supply (Fig. 5bf). Again, relevant concentrations became only obvious under high S conditions (S5, S6).

All identified ITCs have in common that they represent highly volatile GSL breakdown products giving the typical odor of *Brassica* species. Besides antioxidative and herbicidal functions, it is further known that they also have a proven antibacterial activity against, for example, pathogenic bacteria, as well as being cytotoxic (Jang et al. 2010; Khemani et al. 2012). Under health



**Fig. 5** Effect of sulfur supply on concentration of (a) total and (**b–f**) single GSL breakdown products in *Brassica rapa* spp. *pekinensis*. MePrITC–2-methylpropyl isothiocyanate; ButITC–3-butenyl isothiocyanate; PEITC–2-phenylethyl isothiocyanate; PentITC–4-pentenyl isothiocyanate; IAN–indole-3-acetonitrile (MgSO<sub>4</sub>; S1 = 0 g, S2 = 0.01 g, S3 = 0.03 g, S4 = 0.06 g, S5 = 0.1 g, S6 = 0.3 g)

considerations, especially PEITC has been found to inhibit the lung tumorigenesis induced by a potent tobacco-specific carcinogenic nitrosamine in animals (Chung et al. 1992). It was further in humans found that the consumption of watercress led to a dose dependent urinary excretion of the N-acetylcysteine conjugate of PEITC indicative for a pulmonary carcinogen. In animal studies it is suggested that PEITC might inhibit carcinogen activation by CYP enzymes and further decreases the secretion of inflammatory signaling molecules by white blood cells (Hecht 2000; Conaway et al. 2002).

Despite high concentrations of indole GSL concentration (Fig. 4b), concentrations of their breakdown products were low. This is possibly due to the reaction conditions during hydrolysis. The formation of IAN, instead of indolylmethylisothiocyanate as the primary breakdown product, was promoted by ESP (Brown and Morra 1995) and resulted from the breakdown of indole GSLs such as GBC and NGBS, as well as indol-3-ylmethyl GSL, one of the predominant indole GSLs in *Arabidopsis* but also other *Brassica* species (Burow et al. 2008). This reaction usually runs at the expense of generally more toxic isothiocyanates (Lambrix et al. 2001), thus, even as nitrile IAN can be classified as less harmful compared to other ITCs. Further, GBC can be enzymatically hydrolyzed to indole-3-carbinol, which in turn reacts with L-ascorbic acid to ascorbigen, which can induce phase I and II enzymes that are centrally involved in the detoxification of xenobiotics (Wagner and Rimbach 2009).

#### Conclusions

As was shown in various studies, fertilization of *Brassica* species, especially with sulfur, results in a higher GSL and ITC content. But it is not only the total concentration which is affected but rather the GSL and ITC pattern that can be changed by targeted S supply. While indolic GSLs dominated under a low S status, aliphatic GSLs clearly prevailed under high S supply, resulting in a corresponding pattern of ITCs. However, indolic GSLs dominated under all conditions. Especially in Chinese cabbage a clear effect on particularly PEITC, a known ITC of anticarcinogenic activity, could be identified. It can be concluded that biofortification with S seems to be an appropriate tool to improve the S status and thus the health promoting effect of *Brassica* vegetables such as Chinese cabbage.

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## **Prospects for Agricultural Sulfur Research**

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Abstract Only a small number of basic research studies find their way into applied treatments in agricultural production. Advances in plant sulfur research within the past 25 years are documented in the series of the proceedings of the International Sulfur Workshop. Here, key findings in different disciplines such as molecular biology, physiology, agronomy and environment have been published. Based on this substantial knowledge pool the potential significance for future practical applications on production fields, actual knowledge gaps and technology transfer projects for an adaptation to global change and scarcity of fossil fuels has been assessed. An example for each of these contemplations is given. In the first case, the concept of a functional bio-fertilizer, which consists of flexible, tailor-made mixtures of glucosinolate-containing plants against prevailing pathogens and pests is provided. Secondly, it will be shown that research concentrates on sulfur as a minimum factor, but an excessive supply to crops and an imbalanced dietary intake of sulfate by animals and humans unfolds obviously detrimental effects. Last, but not least a conception for the marketing of regional high-quality food products employing organic farming systems is presented which meets future demands of food production.

#### Introduction

Agricultural sulfur research experienced numerous highlights during the past 25 years. After clean air acts came into force in northern Europe during the 1980s the S supply of agricultural crops literally collapsed and caused yield losses in oilseed rape of more than 80% in remote areas. The contribution of soil organic matter to the overall S supply of crops has long been overestimated and finally in the late 1990s it was shown that soil physical and hydrological together with climatic conditions govern the actual sulfate supply in the root zone (Haneklaus et al. 2008). At the same time aspects of crop quality gained increasing interest

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coinciding with a rising awareness and demand for nutritious and constitutional food products. Here, S containing S metabolites as for instance glucosinolates (GSLs) have been attributed a significant role in cancer prevention (Kumar et al. 2015) and S fertilization is a key instrument to increase the GSL content in vegetative and generative plant tissues (Haneklaus et al. 2006a). The significance of the S supply for improving plant health is depicted by the term Sulfur Induced Resistance (SIR) which denotes the reinforcement of the natural resistance of plants against fungal pathogens through triggering the stimulation of metabolic processes involving S by targeted soil-applied fertilizer strategies (Haneklaus et al. 2007). The interested reader may follow up the complete history of S research in the proceedings of the Internal Sulfur Workshop, which has been held every 3 years since 1989 (Anonymous 2015).

In general, the biological know-how is available in order to produce highyielding crops of prime quality. This leaves the question for a justifiable development of agricultural S research that is promising to enhance food and feedstuff quality, and to reduce the input of pesticides.

#### From Cell to Crop or from Crop to Cell?

The usual procedure to implement new agro-technical measures is to quantify dose/ effect data on lab, then greenhouse and finally field scale as pure effects become visible best if no other biotic and abiotic growth factor interferes with the result. Though the working hypothesis might be confirmed under greenhouse conditions, the outcome on production fields is often blurred and irreproducible. In comparison, agro-technical measures may cause distinct effects in the field, which can, however, not be triggered reliably. A prominent example is the phenomenon of biofumigation. The degradation of GSLs by the enzyme myrosinase delivers not only volatile isothiocyanates (ITCs), but also organic cyanides, nitriles, oxazolidinethiones and ionic ITCs all of which have allelopathic potential. GSLs may be released by root exudates of living plants and unfold their allelopathic effects. Another option is their degradation after decomposition of separated plant parts, and harvest residues. Their effect on soil-borne pathogens and pests is resumed by the term biofumigation. Biofumigation might advance to a promising and ecologically sound alternative for crop protection if its efficiency can be controlled.

GSL content and pattern vary in relation to plant species, plant part, growth stage and S supply (Haneklaus et al. 2006b). The efficiency of GSLs and/or ITCs against soil-borne fungal diseases, nematodes and weeds is related to the kind of pathogen and pathotype. ITCs differed in their nematicidal effect by factor 400; their efficacy was usually higher when exposure time was exalted (Lazzeri et al. 2004). Under laboratory conditions, a fungicidal/fungitoxic effect of ITCs lasted not longer than 6 days; only a continuous exposure reduced colony growth efficiently (Smolinska et al. 2003). Only 1–8% of the potential ITC concentration was found after incorporation of plant material into soil (Morra and Kirkegaard 2002).

The potency of ITCs was found to be distinctly higher under laboratory than field conditions if at all. Soils are open systems with a much higher volume than that of sealed containers in the lab, resulting putatively in a lower ITC concentration in the headspace of pathogens. Additional obstacles under field conditions are that the incorporation of the break crop is not homogenous; the GSL content is lower in plant residues than in younger plant material and degradation of GSLs is incomplete as it requires mechanical disruption to destroy cell structures and adequate water for a sufficiently high myrosinase activity. Under field conditions Smith et al. (2004) found no significant relationship between GSL content in roots of oilseed rape and phytosanitary effects. This is supposed to change with the development of a functional bio-fertilizer, which consists of a flexible, tailor-made blend of GSL-containing plants with view to the prevailing pathogen/nematode, a maximization of the GSL content through advanced cultivation, harvesting and preparation procedures and last but not least a mixed formulation of the fertilizer in terms of a short, medium and long-term release of bio-active compounds (Fig. 1; Bloem et al. 2007, 2013; Haneklaus et al. 2006b). A mixture with elemental S as its fungicidal effect on soil-borne pathogens has been shown to be promising (Yang et al. 2006).

#### Maximum Limit Values of S Supply in Plants, Animals and Humans

The classes of S supply are well documented for different crop plants in the deficiency and sufficiency range, while less attention has been paid to an excessive S supply. A similar situation prevails for the impact of excessive intake of sulfate on animal and human performance so that this angle of S research deserves intensified attention in future.

#### Plants

S is commonly evaluated as being highly bio-compliant in such way that excess S neither diminishes productivity, nor impairs quality of the plant products. There are, however, indications that overrated S fertilization may reduce crop yield and that this effect is related to crop type (Haneklaus et al. 2006b).

Important threshold markers for the S supply are: the symptomatological value, which reflects the S concentration below which deficiency symptoms become visible; the critical nutrient value, which stands for the S concentration above which the plant is sufficiently supplied with S for achieving the maximum potential yield or yield reduced by 5%, 10% and 20%; and the toxicological value, which



Fig. 1 Mode of action of a GSL-based biofunctional fertilizer

indicates the S concentration above which toxicity symptoms can be observed (Haneklaus et al. 2006a). In literature, toxicological threshold values for S are rare for the reasons mentioned earlier. An excessive S supply can be expected if plants contain more than 2800 mg/kg d.w. sulfate-S; for fodder crops total S concentrations of only 3.2 mg/g S d.w. may be already disproportionate, while the corresponding value for non-*Brassica* vegetables would equal 10 mg/g S d.w. (Haneklaus et al. 2006b). In general, it can be expected that yield depressions occur at lower S concentrations in plants when green matter is harvested such as forage grasses and cabbage. Upper critical S concentrations in cereals, oilseed rape and sugar beet, which result in yield depressions of 10% have been calculated by Haneklaus et al. (2006b, Table 1).

#### Animals

The dry matter composition of plant products is an important quality parameter of animal feed and foodstuffs. Prominent examples of adverse effects of high S intake on ruminants are polioencephalomalacia (PEM), a neurological disorder and haemolytic anaemia (Gould et al. 2002; Stoewsand 1995). The risk of PEM exists when grass is taken up by animals, which contains more than 0.38% S d.w. (Gould

<b>Table 1</b> The should values for total 5 concentrations (hig/g 5, d.w.) in younger leaves of ons	seeu
rape and sugar beet, and whole above-ground biomass of cereals at start of stem extension	and
canopy closing	

. . .

	Deficiency	Sufficiency		Excess	
Crop	Symptomatological threshold	Lower critical value $(-5\% \text{ yield})$	Maximum yield <sup>a</sup>	Upper critical value (-10% yield)	
Cereals	<1.2	3.2	4.0	>7.5	
Rape	$<2.8^{b}$ and $<3.5^{c}$	5.5	6.5	>14.0	
Sugar	<1.7	3.0	3.5	>4.5	
beet					

Adapted from Haneklaus et al. (2006b)

<sup>a</sup>Seed (oilseed rape), grain (cereals), root and sugar (sugar beet) yield

1.0

<sup>b</sup>Single low

<sup>c</sup>Double low varieties

et al. 2002). According to Kamphues et al. (2015) noxious  $H_2S$  synthesis in the rumen is a possible cause of the disease.

An excessive intake of S and sulfate is associated with loss of appetite, watery feces, secondary trace element deficiency and signs of PEM (Kamphues et al. 2015). In an extended survey the total S content of grass and grass silage did not exceed 8.97 mg/g S d.w., while it was as high as 4.8, 7.6 and 11.8 in extracted soybean meal, oilseed rape meal and dried distillers grains (DDGS; Kamphues et al. 2015). The authors claim the GSL content in rapeseed meal and the sulfuric acid production from DDGS as sources for intolerance. Kamphues et al. (2015) calculated that in extracted rapeseed meal about 60% of the total S content is bound in amino acids, 11% in glucosinolates and 32% as sulfates. Additional S and sulfate sources are drinking water, whey powder, feed additives such as  $(NH_2)_2SO_4$  and methionine hydroxy analogue, and acid salts such as sulfates for the prevention of milk fever. In Table 2 the impact of dietary S levels on animal health is summarized (Kamphues et al. 2015).

From viewpoint of plant nutrition the minimum total S and sulfate concentration for maximum crop production in roughage and corn silage are as follows 2.1 and 1.7 mg/g S d.w., and 500 and 150 mg/kg sulfate d.w., respectively (Haneklaus et al. 2006a). Thus it may be concluded that an adequate S supply of these fodder plants will not interfere with animal health while excessive S fertilization may yield adverse effects on animal health.

#### Humans

The S nutritional status of crops has a significant influence on the nutritive value and sensory features of plant products (Haneklaus et al. 2006b). Methionine is an essential amino acid for monogastric animals and humans; glucosinolates/ITCs have been attributed an anti-carcinogenic potential (Kumar et al. 2015) and S-alk

Sulfur content (g/kg	
d.w.)	Significance
≤1	Deficiency in ruminal flora: degrading activity, synthesis of protein and vitamins is reduced
1.5–2	S requirements is covered; in general, positive S-balance in high pro- ducing cattle
>2	Linear, reduced feed intake in feed lot cattle with increasing S supply up to 4 g/kg S
>2.5	Absorption/utilization of copper and selenium are impaired (secondary deficiency)
>3	Risks for development of PEM at low fiber supply and high levels of concentrates
>4	Risk for PEM, altered metabolism at moderate fiber supply
>5	Risk for PEM and further disturbances/diseases even at high fiber supply

Table 2 Impact of the S content of feedstuff on health of cattle

Extracted from Kamphues et al. (2015)

(en)yl-L-cysteine sulfoxide is active against a broad range of medical disorders. Glutathione (GSH) is another metabolite which content is closely related to the S supply of crop plants and which unfolds a key role in protection against disease in plants, animals and humans because of its protection against oxidative stress and promoting immune functions (Haneklaus et al. 2006a; Grimble 2009). If the dietary intake of S amino acids is insufficient this has serious negative effects on infections and injury (Grimble 2009).

Antinutritives are, for instance, substances, which interfere with the metabolic utilization of minerals (Berdanier 2002). The degradation of GSLs yields for example thiocyanates, isothiocyanates, cyclic S compounds and nitriles, which are goitrogenic. So-called cabbage goiter or struma inhibits the iodine uptake of the thyroid gland in humans (Berdanier 2002).

Less known is presumably the fact that dietary sulfate which is not absorbed in the small intestines is a substrate for sulfate reducing bacteria in the colonic lumen which dissimilate sulfate to H<sub>2</sub>S which again is potentially toxic to the colonic mucosa (Florin et al. 1993). At the moment little is known about the formation of H<sub>2</sub>S and occurrence of ulcerative colitis and colorectal cancer. Florin et al. (1993) identified beer, bread, sausage and cabbage as the main contributors to a sulfate-rich diet. The same authors suggest that excessive beer consumption may result in rectal carcinoma because of an over-proportionally high sulfate intake. A dose/response relationship exists between luminal H<sub>2</sub>S of 0.2 µmol/g perfused into vascularized rat colon and damage to the colonic epithelium (Florin et al. 1993). Such concentrations are found in human feces, too (Florin et al. 1993). Schnug (1988) showed that a close relationship exists between total S and sulfate S content in the vegetative tissue of *Brassica* crops, which increases both, GSL and sulfate content. Fertilizer strategies for the optimization of the nutritional quality of agricultural crops should take possible contrary effects into account.

#### **Regional Organic Products: Tasty and Wholesome**

Peak models for the availability of oil and natural gas indicate that shortages are likely in the near future and, combined with extreme weather events, may threaten the productivity of agricultural systems. The production of mineral fertilizer and pesticides, and fuel for agricultural machinery will be affected as will the transport of agricultural commodities from producer to processor to consumer. Lee et al. (2008) concluded that food production will depend upon more localized farming that can operate effectively with reduced inputs and is adaptable to likely increases in weed, pest and disease problems. Consumer awareness about food quality including that of processed food is increasing and the demand accordingly growing. An expansion of organic farming in the outskirts of bigger cities will fully comply with these demands and is the basis of a holistic concept from cultivation to harvest, storage and consumption.

Favorable energy ratios in organic farming can largely be attributed to lower indirect fossil energy needs, since manufactured fertilizers and synthetic pesticides are omitted (Lee et al. 2008). Based upon an exhaustive understanding of biological and physiological processes, it should be possible to replace chemical aid by a scientific appreciation of these processes. An example of how research can provide substantial advances in crop protection is the concept of SIR (see above). Wider crop rotations and cultivation of catch crops enable the growth of herbal plants which are used as basic material for the production of functional fertilizers (see above).

Fresh organic vegetables had significantly better sensory features (taste, scent) than conventional products (Rembialkowska 2000) and Marckmann (2000) expects a higher intake of organic food because of favorable sensory features, which in return may positively affect diseases such type 2 diabetes and obesity. Sensory features of vegetables are linked to the S supply as for instance the pungency of mustard, radish, onion and garlic and S-containing secondary metabolites exert a health-promoting effect. Field trials have shown that the daily intake of only one onion which was fertilized with S would be sufficient to cover a demand of 12 mg (iso)alliin which corresponds to the recommended daily intake of alliin (Bloem et al. 2004). In comparison, it would be necessary to eat two average bulbs from fresh onions grown at a low S supply for a similar dose. Important will be a nutrient regime that fully satisfies the plant demand, while higher S doses which could further improve qualitative parameters should be set aside in order to avoid nutrient imbalances, enrichment of detrimental components such as sulfate and yield losses (see above).

Besides fresh products, consumers want high-quality products to be processed in restaurants. So, new restaurant chains such as "Hans im Glück" (Hans in Luck, http://hansimglueck-burgergrill.de/) promise fresh regional products of prime quality and address nutrition-conscious people. Here, organic products complete requests for sustainability and adaptation to climate change. The practical implementation of the concept seems affordable if organic farmers are not only paid for

their products but also for their service with respect to precautionary flood protection by maintaining a high infiltration capacity of soils (Rogasik et al. 2007). The spatial extension of organic farming provides an efficient counter-measure against the adverse effects of the anthropogenic sealing of soils resulting from conurbation expansion.

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# Part II Research Papers

# DMSP: Occurrence in Plants and Response to Salinity in *Zea mays*

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Abstract Dimethylsulfoniopropionate (DMSP) is a secondary sulfur compound that is present in high levels in several marine algal species and some higher plant species. DMSP has also been detected in low levels in some other plant species, though its overall occurrence within the plant kingdom remains unclear. The physiological function of DMSP in saline algae and plants is largely unsolved. It is presumed that the compound can function as a constitutive osmolyte and/or antioxidant. In the current study the occurrence of DMSP within the plant kingdom was analyzed. DMSP was present in nanomolar concentrations in all investigated plant species, indicating that it appears to occur widespread within the plant kingdom. Maize (*Zea mays*) appeared also to be a low-DMSP containing species. Shoot DMSP content in this crop increased upon exposure to salinity, also in the presence of atmospheric H<sub>2</sub>S and anoxia. This showed that the DMSP content in maize responded in a similar pattern to salinity exposure which has been observed in DMSP-accumulating plants.

Several marine algal species and some higher plant species contain high levels of the secondary sulfur compound dimethylsulfoniopropionate (DMSP; Stefels 2000, 2007). For instance, in the salt marsh halophyte *Spartina anglica* DMSP accounts for a substantial proportion of the total organic sulfur fraction, ranging from 4 to 50  $\mu$ mol g FW<sup>-1</sup> (Otte et al. 2004). Moreover, low levels of DMSP have been detected in some other plant species (Paquet et al. 1994, 1995), though the overall occurrence of the compound within the plant kingdom remains unclear. DMSP can – amongst others – function as a constitutive osmolyte and/or antioxidant (Stefels 2000; Otte et al. 2004). Similar to observations with marine algae, it has been reported that in *Spartina* species DMSP content may increase upon salinity exposure (Dacey et al. 1987; Stefels 2000; Mulholland and Otte 2001). However, in

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other studies DMSP content of *Spartina* species was not affected by salinity exposure (Van Diggelen et al. 1986; Otte and Morris 1994). In the current study a sensitive method was used to assess the occurrence of DMSP within the plant kingdom. Moreover, the impact of the salt marsh factors salinity, atmospheric H<sub>2</sub>S and anoxia on the DMSP content of the non-DMSP accumulator maize (*Zea mays*) was investigated.

In order to assess the occurrence of the secondary sulfur compound DMSP within the plant kingdom, plant shoots from a wide variety of plant species were collected in the northern part of the Netherlands. In addition, nine-day old seedlings of maize (Zea mays subsp. mays, cv. Ricardinio, Van der Wal, Hoogeveen, The Netherlands) were grown on aerated 25% Hoagland nutrient solutions and exposed to 100 mM NaCl and/or 0.25  $\mu$ l l<sup>-1</sup> H<sub>2</sub>S for 7 days (Ausma et al. 2017). This experiment was later repeated at anoxic conditions (anoxia; solutions were not aerated, but flushed with N<sub>2</sub> for 10 min on each day). For the determination of the DMSP content, fresh plant material was homogenized in a 0.3 M HCl medium using an Ultra Turrax (10 ml g FW<sup>-1</sup>). The HCl medium contained 1  $\mu$ M deuterated DMSP as an internal standard. The homogenate was gently and shortly shaken and 0.3 ml was transferred to a small vial containing 8 ml de-ionized water. Subsequently, one pellet of NaOH was added to the vial, after which the vial was quickly closed: alkalization causes DMSP to convert to the gas dimethylsulfide (DMS). The vials were stored at 4 °C for approximately a week until analysis. The amount of DMS gas in each vial was measured using a proton-transfer-reaction mass spectrometer (PTR-MS; Ionicon, GmbH, Innsbruck, Austria; see Stefels et al. 2012). The recovery of DMSP during the treatment was checked with the internal standard and appeared to be higher than 80%. In plants, DMS can also be formed upon degradation of S-methylmethionine or other sulfonium compounds. However, the content of these compounds and their conversion efficiency into DMS under the experimental conditions (i.e. the pH) appears to be negligible (Paquet et al. 1995) and when no NaOH was added to maize homogenates, no DMS was detected.

The secondary sulfur compound DMSP appeared to be present in foliar tissue of all investigated plant species (Table 1). DMSP was detectable in shoot tissue from monocots and dicots and in species that occur in freshwater and saline environments. Moreover, DMSP was detectable in the evolutionary ancient species *Dryopteris dilatata*, a fern, and *Equisetum arvense*, a horsetail. However, all shoots only contained trace quantities of DMSP, ranging from a half to several nanomoles per gram fresh weight (Table 1).

The crop plant maize (a glycophyte) also contained low levels of DMSP: the shoot and root content were 0.5 and 1 nmol g  $FW^{-1}$ , respectively (Fig. 1). The DMSP content in both shoot and root was not affected by a 7-day exposure to atmospheric H<sub>2</sub>S and/or the absence of oxygen in the root environment (anoxia; see Ausma et al. 2017 for more data). Evidently, in maize DMSP did not act as a metabolic sink of excessive foliarly absorbed and metabolized atmospheric H<sub>2</sub>S. A 7-day exposure of maize to 100 mM NaCl did not affect DMSP content in the root, whilst that in the shoot was enhanced threefold, also in the presence of H<sub>2</sub>S and anoxia (Fig. 1).

Species	cies Family			
Monocots				
Narcissus pseudonarcissus	cissus pseudonarcissus Amaryllidaceae			
Convallaria majalis	Asparagaceae	1.4		
Carex appropinquata	Cyperaceae	1.8		
Carex echinata	Cyperaceae	0.7		
Carex nigra	Cyperaceae	0.5		
Iris pseudacorus	Iridaceae	4.8		
Juncus effusus	Juncaceae	4.2		
Juncus gerardii <sup>a</sup>	Juncaceae	0.7		
Alopecurus pratensis	Poaceae	2.3		
Bromus hordeaceus	Poaceae	2.8		
Dactylis glomerata	Poaceae	9.7		
Elytrigia atherica <sup>a</sup>	Poaceae	0.4		
Festuca rubra <sup>a</sup>	Poaceae	1.2		
Holcus lanatus	Poaceae	0.8		
Lolium perenne	Poaceae	1.7		
Poa annua	Poaceae	2.4		
Puccinellia maritima <sup>a</sup>	Poaceae	1.4		
Dicots				
Anthriscus sylvestris	Apiaceae	2.1		
Artemisia maritima <sup>a</sup>	Asteraceae	0.9		
Leucanthemum vulgare	Asteraceae	0.5		
Taraxacum officinale	Asteraceae	0.6		
Brassica napus	Brassicaceae	1.0		
Cannabis sativa	Cannabaceae	1.0		
Quercus robur	Fagaceae	3.9		
Limonium vulgare <sup>a</sup>	Plumbaginaceae	1.8		
Ranunculus repens	Ranunculaceae	0.9		
Other				
Dryopteris dilatata	Dryopteridaceae	1.5		
Equisetum arvense	Equisetaceae	1.7		

 Table 1
 The occurrence of DMSP in shoots of natural plant species collected in the northern part of the Netherlands

DMSP content is expressed as nmol  $g^{-1}$  fresh weight <sup>a</sup>Saline species

The results show that DMSP appears to occur widespread within the plant kingdom, since the shoots of all investigated species contained detectable levels of this secondary sulfur compound. Moreover, it was evident that similar to observations with halophytic DMSP-accumulating *Spartina* species, exposure of the glycophyte and non-DMSP accumulator maize to salinity resulted in an enhanced DMSP content in the shoot (Dacey et al. 1987; Mulholland and Otte 2001). Whether this enhancement in DMSP content upon salinity exposure has any physiological significance needs to be questioned, since its actual content remained



Fig. 1 The impact of atmospheric H<sub>2</sub>S, NaCl salinity and anoxia on the DMSP content in the shoot and root of maize (*Zea mays* subsp. *mays*, cv. Ricardinio). Plants were exposed to  $0.25 \,\mu$ l l<sup>-1</sup> H<sub>2</sub>S, 100 mM NaCl and anoxia for 7 days (see Ausma et al. 2017 for details). Data represent the mean  $\pm$  SD of 3 measurements with 3 plants in each. Different letters indicate significant differences (P  $\leq 0.01$ , Student's t-test)

rather low. Recently, it was reported that in the low-DMSP containing species tomato (*Solanum lycopersicum*) the leaf DMSP content was strongly enhanced upon drought exposure (Catola et al. 2016). Both salinity and drought exposure can cause osmotic stress. The significance of osmotic stress as a trigger of an enhanced DMSP content in foliar tissue deserves to be investigated further.

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## Impact of Atmospheric H<sub>2</sub>S, Salinity and Anoxia on Sulfur Metabolism in *Zea mays*

Ties Ausma, Saroj Parmar, Malcolm J. Hawkesford, and Luit J. De Kok

Abstract Plants in coastal salt marshes have to deal with salinity, anoxia and excessive reduced sulfur at the same time. Sulfur metabolism is presumed to have significance in plant stress-tolerance. In order to obtain more insight into the physiological significance of sulfur metabolism in plant responses to multiple abiotic stress factors, the glycophyte maize (Zea mays) was exposed to atmospheric H<sub>2</sub>S, salinity and anoxia. Maize seedlings appeared to be rather unsusceptible for the potentially toxic effects of these stressors. A 7-day exposure to 0.25  $\mu$ l l<sup>-1</sup> H<sub>2</sub>S and/or anoxia (anoxic root conditions) slightly enhanced plant biomass production, whereas it was not affected upon exposure to 100 mM NaCl. A simultaneous exposure of plants to salinity with H<sub>2</sub>S and/or anoxia resulted in a decreased biomass production. The total sulfur content of the shoot and root was hardly affected by  $H_2S$  exposure, whereas it was strongly decreased upon anoxia. The total sulfur content of the shoot was decreased upon exposure to salinity. The decreases in total sulfur content could be predominantly ascribed to a decrease in the sulfate content. H<sub>2</sub>S exposure only resulted in an enhanced water-soluble non-protein thiol content in shoots, whereas it was not affected by salinity and anoxia. Only a simultaneous exposure of plants to H<sub>2</sub>S, salinity and/or anoxia resulted in an enhanced water-soluble non-protein thiol content of the root. Anoxia and salinity exposure induced aerenchyma formation in the root, and the increased root thiol contents might be the result of the direct diffusion of atmospheric  $H_2S$  via the stomata through the aerenchyma and subsequent metabolism in the root.

In nature plants are often exposed to multiple abiotic stress factors. For instance, plants in coastal salt marshes not only have to deal with salinity, but also with anoxia and excessive reduced sulfur. Anoxia (*viz.* anoxic root conditions) limits

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root respiration and carbohydrates are then broken down via fermentative pathways to yield at least some ATP, which often is insufficient to support optimal plant growth (Yamauchi et al. 2013). An adaptation to anoxia exposure is the formation of aerenchyma: spongy tissue that consists of air spaces and channels in the leaves, stems and roots (Yamauchi et al. 2013). Aerenchyma facilitates an enhanced  $O_2$ diffusion from the shoot to the root and this enables root respiration at anoxic conditions (Yamauchi et al. 2013). Exposure to salinity may negatively affect metabolism and plant growth by affecting the water balance and by causing an accumulation of the toxic cation sodium in the cytosol (Grattan and Grieve 1999; Parida and Das 2005). Exposure to sulfide in soil and atmosphere may also be harmful (Beauchamp et al. 1984; De Kok et al. 2002). Hydrogen sulfide is a potentially phytotoxic gas, since it may react with metalloenzymes (viz. cytochrome oxidase; Beauchamp et al. 1984; De Kok et al. 2002). However, at low levels foliarly absorbed  $H_2S$  may be metabolized and replace sulfate taken up by the root as sulfur source for growth (De Kok et al. 1997, 1998; Hawkesford and De Kok 2006).

It is presumed that sulfur metabolites may fulfill a role in the tolerance of plants to abiotic and biotic stress (Bloem et al. 2014). A variety of organic sulfur compounds are presumed to have stress-protective functions (Rausch and Wachter 2005). For instance, glutathione and derived metabolites are thought to have diverse functions in stress-protection (Tausz et al. 2004; Noctor et al. 2012). In order to obtain more insight into the physiological significance of sulfur metabolism in plant responses to multiple abiotic stress factors, the glycophyte maize (*Zea mays*) was exposed to atmospheric  $H_2S$ , salinity and anoxia.

Maize (Zea mays subsp. mays, cv. Ricardinio, Van der Wal, Hoogeveen, The Netherlands) was germinated on moistened filter paper at 21 °C in the dark for 2 days. Subsequently, the seedlings were transferred to 15 l containers containing tap water in a climate-controlled room at a day/night temperature of 21 °C/18 °C  $(\pm 1 \text{ °C})$ , a relative humidity of 60–70% and a photoperiod of 14 h at a photon fluence rate of  $300 \pm 20 \text{ }\mu\text{mol m}^{-2} \text{ s}^{-1}$  (within the 400–700 nm range) at plant height, supplied by Philips GreenPower LED (deep red/white 120) production modules. After 7 days the seedlings were transferred to 13 l containers (ten plant sets per container, six plants per set) containing an aerated (oxic) or non-aerated (anoxic) 25% Hoagland nutrient solution (pH 7.0; for composition see Koralewska et al. 2007; the latter nutrient solution was also daily flushed with  $N_2$  for 10 min). Seedlings were exposed to 0.25  $\mu$ l l<sup>-1</sup> H<sub>2</sub>S and/or 100 mM NaCl for 7 days. The containers with seedlings were placed in 150 l cylindrical stainless steel cabinets (0.6 m diameter) with a polymethyl methacrylate top. Air exchange inside the cabinets was 40 1 min<sup>-1</sup> and the air inside the cabinets was stirred continuously by a ventilator. Day/night temperatures were 22 °C/19 °C (±1 °C), relative humidity was 20-40% and the photoperiod was 14 h at a photon fluence rate of  $340 \pm 20 \ \mu mol \ m^{-2} \ s^{-1}$  (within the 400–700 nm range) at plant height, supplied by Philips GreenPower LED (deep red/white 120) production modules. Temperature was controlled by adjusting the temperature of the cabinet wall. Temperature, relative humidity and photon fluence rate at plant height were monitored using data loggers (Hobo type U12, Onset Computer Corporation, Bourne, MA, USA). For atmospheric H<sub>2</sub>S exposure, pressurized H<sub>2</sub>S gas diluted with N<sub>2</sub> gas (1 ml  $l^{-1}$ ) was injected into the incoming air stream and the concentration in the cabinet was adjusted to the desired concentration of 0.25  $\mu$ l l<sup>-1</sup> using mass flow controllers (ASM, Bilthoven, The Netherlands). H<sub>2</sub>S concentrations in the cabinets were monitored by an SO<sub>2</sub> analyzer (model 9850) equipped with a  $H_2S$  converter (model 8770, Monitor Labs, Measurement Controls Corporation, Englewood, CO, USA). The lids of the containers and plant sets were sealed in order to prevent the absorption of atmospheric  $H_2S$  by the solution. On the day before harvest, chlorophyll a fluorescence (F<sub>v</sub>/F<sub>m</sub> ratio) of leaves was measured by using a modulated fluorometer in the morning after a dark-adaptation of at least 1 h (PAM 2000, Walz GmbH, Effeltrich, Germany). Moreover, roots were examined for the presence of aerenchyma under a light microscope. At the day of harvest, shoots and roots of plants were separated and fresh weight was determined. Biomass production was calculated by subtracting final fresh weight from initial fresh weight. For determination of the dry matter content, plants were dried at 80 °C for 24 h. For chlorophyll analysis, pigments were extracted from frozen shoots as described by Shahbaz et al. (2010), and the chlorophyll a and b content were determined according to Lichtenthaler (1987). Water-soluble non-protein thiols were extracted from freshly harvested plants (Shahbaz et al. 2010) and the total water-soluble non-protein thiol content was determined colorimetrically according to De Kok et al. (1988). For determination of total sulfur, sulfate and the mineral nutrient composition, dried shoots and roots were pulverized by using a Retsch Mixer-Mill (type MM2, Haan, Germany). The total sulfur and sulfate content were determined as described by Aghajanzadeh et al. (2016). The mineral nutrient composition was analyzed by inductively coupled plasma mass spectrometry (ICP-MS) and inductively coupled plasma atomic emission spectrometry (ICP-AES) as described by Reich et al. (2016b). Data was statistically analyzed by an unpaired Student's t-test

at  $P \le 0.01$ .

Maize appeared to be not very susceptible to the toxic effects of H<sub>2</sub>S and a 7-day exposure to 0.25  $\mu$ l l<sup>-1</sup> H<sub>2</sub>S resulted in a slight increase in the plant biomass production, accompanied with a slight decrease in the dry matter content of the shoot and root, whereas the shoot to root ratio was not affected (Table 1). A similar increase in biomass production at low levels of atmospheric H<sub>2</sub>S has been observed in some other plant species (Thompson et al. 1979; Durenkamp et al. 2007). The chlorophyll content of the shoot, the chlorophyll a/b ratio and chlorophyll *a* fluorescence, the latter represents the quantum yield of photosystem II, were not affected (Table 1). Moreover, the mineral composition of both shoot and root were hardly affected upon  $H_2S$  exposure (Table 3).  $H_2S$  exposure did also not affect the total sulfur and sulfate content of both shoot and root (Tables 1 and 3), indicating a down-regulation of the sulfate uptake by the root. The latter was supported by the observation that H<sub>2</sub>S exposure resulted in a partial but significant decrease in the sulfate uptake capacity of the root (data not shown).  $H_2S$  exposure resulted in an increase in the content of water-soluble non-protein thiols (presumably cysteine and glutathione) in the shoot, whereas that in the root remained unaffected (Table 1). The latter data were similar to observations with other plant species (De Kok et al. 1997, 1998).

	Control	H <sub>2</sub> S	NaCl	$H_2S + NaCl$				
Plant								
Biomass production	$4.06\pm0.88b$	$5.80 \pm 0.93$ c	$3.72\pm0.76b$	$2.92\pm0.54a$				
Shoot/root ratio	$1.01 \pm 0.13a$	$1.09 \pm 0.13a$	$1.26\pm0.11b$	$1.15 \pm 0.15$ ab				
Shoot								
DMC	$9.6 \pm 0.4b$	$8.8 \pm 0.3a$	$10.9 \pm 0.6c$	$11.2 \pm 0.4c$				
Chl a + b	$0.67 \pm 0.09a$	$0.79\pm0.13a$	$1.11 \pm 0.24b$	$1.26 \pm 0.30b$				
Chl a/b	$2.6 \pm 0.4a$	$2.5\pm0.5a$	$2.7\pm0.2a$	$2.6 \pm 0.1a$				
F <sub>v</sub> /F <sub>m</sub>	$0.75 \pm 0.05a$	$0.74 \pm 0.06a$	$0.77\pm0.04a$	$0.76 \pm 0.04a$				
Thiols	$0.45 \pm 0.10a$	$0.59\pm0.03\mathrm{b}$	$0.48 \pm 0.06a$	$0.75 \pm 0.03c$				
Sulfate	91 ± 6b	$61 \pm 15$ ab	$52 \pm 11a$	$64 \pm 18$ ab				
Total sulfur	$168 \pm 10b$	157 ± 24ab	$131 \pm 12a$	$149 \pm 7ab$				
Root								
DMC	$7.4 \pm 0.3b$	$6.7\pm0.3a$	$7.8 \pm 0.4c$	$8.1 \pm 0.6c$				
Thiols	$0.46 \pm 0.06$ ab	$0.41 \pm 0.04a$	$0.40 \pm 0.05a$	$0.51 \pm 0.03b$				
Sulfate	81 ± 21ab	89 ± 9ab	$108 \pm 6b$	91 ± 6a				
Total sulfur	$152 \pm 18a$	156 ± 22a	166 ± 11a	$172 \pm 9a$				
Aerenchyma	Absent	Absent	Present	Present				

Table 1 The impact of atmospheric  $H_2S$  and NaCl salinity on biomass production, chlorophyll content, chlorophyll *a* fluorescence and sulfur metabolite content of maize

Plants were exposed to  $0.25 \,\mu l^{-1} H_2 S$  and 100 mM NaCl for 7 days. The initial plant fresh weight was  $1.31 \pm 0.17$  g. Data on biomass production (g fresh weight) and shoot to root ratio (on a fresh weight basis) represent the mean of two experiments with 14 measurements with three plants in each ( $\pm SD$ ). Data on dry matter content (DMC; % of fresh weight), chlorophyll content (mg g<sup>-1</sup> fresh weight) and water-soluble non-protein thiol content ( $\mu$ mol g<sup>-1</sup> fresh weight) represent the mean of two experiments with 6, 3 and 3 measurements with three plants in each ( $\pm SD$ ), respectively. Data on chlorophyll *a* fluorescence ( $F_v/F_m$  ratio) represent the mean of two experiments in each ( $\pm SD$ ). Data on the total sulfur and sulfate content ( $\mu$ mol g<sup>-1</sup> dry weight) represent the mean of 3 measurements with three plants in each ( $\pm SD$ ). Different letters indicate significant differences between treatments (P < 0.01, Student's t-test)

A 7-day exposure of maize to 100 mM NaCl hardly affected plant biomass production and the shoot to root ratio, but resulted in a substantial increase in the dry matter content of both shoot and root, even though aerenchyma had developed in the root (Table 1). Similar to observations with other plant species (Grattan and Grieve 1999; Reich et al. 2016a) exposure to NaCl salinity strongly affected the mineral composition of both shoot and root of maize (Table 3). It resulted in a 70-fold and 28-fold increase in the sodium content in the shoot and root, respectively, accompanied by a strong decrease in the content of calcium, potassium and magnesium in both shoot and root (Table 3). Moreover, salinity exposure resulted in a decrease in the total sulfur content, which could for a greater part be ascribed to a decrease in sulfate content (Tables 1 and 3). Exposure to NaCl salinity resulted in an increase in the chlorophyll content of the shoot, whereas the chlorophyll a/b ratio, chlorophyll a fluorescence and the content of the water-soluble non-protein thiols in both shoot and root were unaffected (Tables 1 and 3). Apparently, salinity exposure did not affect the composition and functioning of the photosystems, viz. photosynthetic electron transport.
	Anoxic conditions	8		
	Control	H <sub>2</sub> S	NaCl	$H_2S + NaCl$
Plant				
Biomass production	$5.15 \pm 0.72b$	$7.71 \pm 1.12c$	$2.67\pm0.47a$	$2.74\pm0.33a$
Shoot/root ratio	$1.42\pm0.17b$	$1.77\pm0.18c$	$1.23\pm0.12a$	$1.23 \pm 0.14a$
Shoot				
DMC	$9.3\pm0.3a$	$9.0\pm0.4a$	$11.3 \pm 0.4b$	$11.6 \pm 0.6b$
Chl a + b	$0.68 \pm 0.10a$	$0.89 \pm 0.08a$	$0.95\pm0.10a$	$1.00\pm0.03b$
Chl a/b	$3.1 \pm 0.1b$	$3.0\pm0.0b$	$2.7\pm0.1a$	$2.8\pm0.1a$
F <sub>v</sub> /F <sub>m</sub>	$0.72\pm0.07a$	$0.76 \pm 0.09a$	$0.77\pm0.04a$	$0.78\pm0.06a$
Thiols	$0.42 \pm 0.02a$	$0.54\pm0.04\mathrm{b}$	$0.35\pm0.08a$	$0.99\pm0.02c$
Sulfate	$20 \pm 2a$	$27 \pm 5a$	$36 \pm 2b$	$32 \pm 4b$
Total sulfur	$99 \pm 3a$	$105 \pm 3a$	$108 \pm 5a$	$110 \pm 5a$
Root				
DMC	$6.9 \pm 0.5a$	$6.7 \pm 0.5a$	$7.9\pm0.6b$	$8.1\pm0.7b$
Thiols	$0.34 \pm 0.03a$	$0.62\pm0.07\mathrm{b}$	$0.30\pm0.02a$	$0.81 \pm 0.02c$
Sulfate	$51 \pm 5a$	$67 \pm 12ab$	$68 \pm 3b$	$57 \pm 3a$
Total sulfur	$111 \pm 9a$	$135 \pm 5b$	$134 \pm 5b$	$127 \pm 3ab$
Aerenchyma	Present	Present	Present	Present

**Table 2** The impact of atmospheric  $H_2S$  and NaCl salinity on biomass production, chlorophyll content, chlorophyll *a* fluorescence and sulfur metabolite content of maize at anoxic conditions

Plants were exposed to  $0.25 \ \mu l^{-1} H_2 S$  and 100 mM NaCl at anoxic root conditions for 7 days. The initial plant fresh weight was  $1.23 \pm 0.16 \ g$ . Data on biomass production (g fresh weight) and shoot to root ratio (on a fresh weight basis) represent the mean of 14 measurements with three plants in each ( $\pm SD$ ). Data on dry matter content (DMC; % of fresh weight), chlorophyll content (mg g<sup>-1</sup> fresh weight) and water-soluble non-protein thiol content ( $\mu$ mol g<sup>-1</sup> fresh weight) represent the mean of 6, 3 and 3 measurements with three plants in each ( $\pm SD$ ). Data on the total sulfur and sulfate content ( $\mu$ mol g<sup>-1</sup> dry weight) represent the mean of 3 measurements with three plants in each ( $\pm SD$ ). Data on the total sulfur and sulfate content ( $\mu$ mol g<sup>-1</sup> dry weight) represent the mean of 3 measurements with three plants in each ( $\pm SD$ ). Different letters indicate significant differences between treatments (P  $\leq 0.01$ , Student's t-test)

The observed increased plant biomass production upon  $H_2S$  exposure did not occur upon a simultaneous exposure to NaCl salinity. Biomass production of these plants was even lower than that of unexposed (control) plants (Table 1). Upon a simultaneous exposure to atmospheric  $H_2S$  and NaCl salinity, the chlorophyll content of the shoot, the mineral composition of both shoot and root and the sulfate and total sulfur content of both shoot and root were all quite similar to those observed in the absence of  $H_2S$  (Tables 1 and 3). Moreover, the development of aerenchyma in the root was noticeable. Upon a simultaneous exposure to  $H_2S$  and NaCl salinity there was not only a substantial increase in the water-soluble nonprotein thiol content in the shoot, but also a slight increase in the root (Table 1).

A 7-day exposure of maize to anoxia only slightly affected plant biomass production as compared to oxic conditions and it resulted in an increase in the shoot to root ratio (Tables 1 and 2). Under anoxic conditions, the impact of

	Control	H <sub>2</sub> S	NaCl	$H_2S + NaCl$
Shoot				
Calcium	$198 \pm 28b$	$180 \pm 8b$	$45 \pm 4a$	46 ± 1a
Copper	$0.29 \pm 0.03a$	$0.25 \pm 0.03a$	$0.27 \pm 0.01a$	$0.25 \pm 0.02a$
Iron	$0.74 \pm 0.18a$	$0.96 \pm 0.34a$	$0.87\pm0.02a$	$0.86 \pm 0.05a$
Magnesium	$165 \pm 19b$	$152 \pm 10b$	99 ± 7a	97 ± 1a
Manganese	$1.00 \pm 0.19a$	$0.83 \pm 0.18a$	$0.69 \pm 0.02a$	$0.72 \pm 0.08a$
Molybdenum	$0.015 \pm 0.001a$	$0.017\pm0.003ab$	$0.026\pm0.002c$	$0.020 \pm 0.002b$
Phosphorus	$303 \pm 46a$	357 ± 37a	$309 \pm 4a$	$282 \pm 14a$
Potassium	$1585 \pm 38b$	$1710 \pm 26c$	$1129 \pm 157a$	907 ± 31a
Sodium	$13 \pm 3a$	19 ± 10a	$907\pm202b$	$1114 \pm 28b$
Sulfur	$148 \pm 20$ ab	$154 \pm 7b$	$123 \pm 14a$	$146 \pm 10a$
Zinc	$0.89 \pm 0.13a$	$0.88 \pm 0.24a$	$0.85 \pm 0.13a$	$0.71\pm0.07a$
Root				
Calcium	$257 \pm 13b$	$270 \pm 25b$	$173 \pm 11a$	$140 \pm 14a$
Copper	$0.95\pm0.16ab$	$0.67\pm0.08a$	$1.02\pm0.06\mathrm{b}$	$0.92\pm0.06\mathrm{b}$
Iron	$4.63 \pm 0.87$ ab	$2.88\pm0.37a$	$6.59 \pm 0.41c$	$4.93\pm0.58b$
Magnesium	$217 \pm 3b$	$257 \pm 14c$	$137 \pm 13a$	$112 \pm 9a$
Manganese	$4.09 \pm 0.73$ ab	$3.25 \pm 0.75a$	$6.05\pm0.87\mathrm{b}$	$5.71 \pm 0.73b$
Molybdenum	$0.013 \pm 0.001a$	$0.021 \pm 0.003b$	$0.022 \pm 0.001 \mathrm{b}$	$0.021 \pm 0.002b$
Phosphorus	$217 \pm 2b$	$239 \pm 5c$	$197 \pm 15$ ab	181 ± 7a
Potassium	$1078 \pm 38b$	$1139 \pm 14c$	378 ± 21a	381 ± 6a
Sodium	$72 \pm 18a$	$93 \pm 30a$	$2007 \pm 79c$	$1646 \pm 61b$
Sulfur	162 ± 9a	174 ± 11a	$181 \pm 8a$	$162 \pm 7a$
Zinc	$0.92 \pm 0.12b$	$0.86 \pm 0.20$ ab	$0.65 \pm 0.12$ ab	$0.53 \pm 0.03a$

Table 3 The impact of atmospheric  $H_2S$  and NaCl salinity on the mineral nutrient content of maize

Plants were exposed to 0.25  $\mu$ l l<sup>-1</sup> atmospheric H<sub>2</sub>S and 100 mM NaCl for 7 days. Data on the mineral nutrient content ( $\mu$ mol g<sup>-1</sup> dry weight) represent the mean of 3 measurements with three plants in each ( $\pm$ SD). Different letters indicate significant differences between treatments (P  $\leq$  0.01, Student's t-test)

exposure to atmospheric  $H_2S$ , NaCl salinity and their combination on plant biomass production, chlorophyll content, chlorophyll a/b ratio and chlorophyll *a* fluorescence were quite similar to their impact under oxic conditions (Tables 1 and 2). Again,  $H_2S$  exposure resulted in an increased plant biomass production (Table 2). However, upon exposure to NaCl salinity and upon a simultaneous exposure to NaCl salinity and  $H_2S$ , the plant biomass was reduced and lower than that of unexposed (control) plants (Table 2). Upon exposure to anoxia the development of aerenchyma in the root was observed at all conditions.

Anoxia exposure affected the mineral composition and resulted in a substantial decrease in the total sulfur (Tables 1 and 2), calcium and magnesium content in both shoot and root as compared to oxic conditions (Tables 3 and 4). The decrease in the plant total sulfur content could predominantly be ascribed to a decrease in sulfate

	Anoxic conditions			
	Control	H <sub>2</sub> S	NaCl	$H_2S + NaCl$
Shoot				
Calcium	$124 \pm 24b$	$120 \pm 2b$	$47 \pm 8a$	$42 \pm 6a$
Copper	$0.27\pm0.04a$	$0.22\pm0.01$ a	$0.30\pm0.07a$	$0.22\pm0.02a$
Iron	$0.64\pm0.04a$	$0.89\pm0.06\mathrm{b}$	$0.85\pm0.19ab$	$0.93\pm0.13b$
Magnesium	$107 \pm 11$ ab	$103 \pm 3b$	$101 \pm 20$ ab	$87 \pm 6a$
Manganese	$0.77\pm0.12a$	$0.62 \pm 0.03a$	$0.87\pm0.14a$	$0.62\pm0.03a$
Molybdenum	$0.013\pm0.001a$	$0.022\pm0.001\mathrm{c}$	$0.020\pm0.002\mathrm{bc}$	$0.018\pm0.001\mathrm{b}$
Phosphorus	$246 \pm 24ab$	$290 \pm 14b$	$217 \pm 10$ ab	$208 \pm 6a$
Potassium	$1484 \pm 100 \mathrm{b}$	$1703 \pm 37c$	$642 \pm 76a$	$705 \pm 45a$
Sodium	$7 \pm 1a$	$7 \pm 0a$	$1097 \pm 120 \mathrm{b}$	$945 \pm 121b$
Sulfur	75 ± 5a	$90 \pm 0b$	$86 \pm 7ab$	$98 \pm 8b$
Zinc	$0.56 \pm 0.14a$	$0.49\pm0.06a$	$0.93 \pm 0.26a$	$0.64 \pm 0.12a$
Root				
Calcium	$131 \pm 15a$	$127 \pm 7a$	$135 \pm 39a$	$102 \pm 15a$
Copper	$1.47\pm0.17a$	$1.68\pm0.23a$	$1.93\pm0.52a$	$1.77\pm0.25a$
Iron	$2.25\pm0.20a$	$2.51\pm0.23ab$	$3.03 \pm 0.25$ bc	$3.66 \pm 0.37c$
Magnesium	$135\pm9b$	$155 \pm 9b$	$128 \pm 22ab$	$112 \pm 5a$
Manganese	$4.63\pm0.51a$	$5.37\pm0.65a$	$7.73 \pm 1.49a$	$6.71 \pm 1.00 a$
Molybdenum	$0.013\pm0.002a$	$0.029\pm0.005\mathrm{b}$	$0.018\pm0.005 ab$	$0.018\pm0.002a$
Phosphorus	$212 \pm 6a$	$224 \pm 13a$	$198 \pm 21a$	$195 \pm 20a$
Potassium	$1156\pm67b$	$1264 \pm 25b$	$411 \pm 27a$	$464 \pm 44a$
Sodium	49 ± 8a	$38 \pm 1a$	$1437 \pm 108 \mathrm{b}$	$1463 \pm 103b$
Sulfur	$101 \pm 6a$	$132 \pm 9b$	$118 \pm 12ab$	$129 \pm 7b$
Zinc	$0.78 \pm 0.13$ ab	$0.82 \pm 0.09b$	$0.63 \pm 0.11$ ab	$0.54 \pm 0.04a$

Table 4 The impact of atmospheric  $H_2S$  and NaCl salinity on the mineral nutrient content of maize at anoxic conditions

Plants were exposed to 0.25  $\mu$ l l<sup>-1</sup> H<sub>2</sub>S and 100 mM NaCl at anoxic root conditions for 7 days. Data on the mineral nutrient content ( $\mu$ mol g<sup>-1</sup> dry weight) represent the mean of 3 measurements with three plants in each ( $\pm$ SD). Different letters indicate significant differences between treatments (P  $\leq$  0.01, Student's t-test)

content (Tables 1 and 2). Again,  $H_2S$  exposure hardly further affected the mineral composition of plants under anoxic conditions (Table 4). Similar to observations under oxic conditions, NaCl salinity strongly affected the mineral composition of both shoot and root of maize (Table 4). It resulted in an increase in the sodium content in both shoot and root, accompanied by a strong decrease in the content of calcium in the shoot and potassium in both shoot and root (Table 4). In contrast to oxic conditions,  $H_2S$  exposure resulted not only in an increase in the content of water-soluble non-protein thiols in the shoot, but also in the root (Table 2). Again, exposure to NaCl salinity hardly affected the water-soluble non-protein thiol content in both shoot and root (Table 2). However, a simultaneous exposure of maize to  $H_2S$  and NaCl salinity resulted in a more strongly enhanced water-soluble non-protein thiol content in both shoot and root than that observed in the absence of NaCl salinity (Table 2).

From the current study it was evident that maize seedlings were rather unsusceptible to the potentially toxic effects of exposure to  $H_2S$ , NaCl salinity and anoxia. Only a combination of NaCl salinity with  $H_2S$  and/or anoxia negatively affected plant biomass production. This may indicate that under these conditions, the combination of abiotic stress factors negatively affected the balance between carbon use for structural growth and carbon use for the maintenance respiration required to alleviate the negative effects of the stressors. Furthermore, it was evident that not only anoxia but also NaCl salinity induced the formation of aerenchyma in the roots of maize. It was previously observed that several abiotic stress factors might induce aerenchyma formation in roots (Bouranis et al. 2003; Evans 2003). The enhanced water-soluble non-protein thiol content in the root of maize upon the simultaneous exposure to  $H_2S$ , salinity and/or anoxia might be the result of the direct diffusion of atmospheric  $H_2S$  via the stomata through the aerenchyma and subsequent metabolism in the root.

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## Salinity Influences Single Glucosinolate Content in the Halophyte *Lepidium latifolium*

Christian Boestfleisch, Johann Hornbacher, Annekathrin Rumlow, and Jutta Papenbrock

**Abstract** The influence of salinity on the biosynthesis of secondary metabolites with a focus on single glucosinolates (GSLs) was investigated in *Lepidium latifolium* L., which is a plant species rich in antioxidants. Mature plants were subjected to 0, 15, 22.5, and 35 Practical Salinity Units (PSU) for 1–4 weeks. While phenols, flavonoids, and the oxygen radical absorbance capacity (ORAC) increased with increasing salinity, the ascorbate concentration did not follow a specific pattern. The concentration of single GSLs was influenced by salinity in different ways: While the concentration of aliphatic GSLs like glucoiberin and sinigrin increased, the concentration of aromatic GSLs such as glucobrassicin decreased under salinity stress. Salinity increased the total GSL concentration significantly with sinigrin being the major contributing GSL. The exact molecular role of the different GSLs in abiotic stress defense needs further analysis.

The halophyte *Lepidium latifolium* L. belongs to the Brassicaceae family, known for their high abundance of glucosinolates (GSLs). The role of GSLs and their break-down products under biotic stress, especially their defense function against herbivores, insects and pathogens, is well known: upon tissue damage myrosinase hydrolyzes GSLs, releasing thiocyanates, isothiocyanates and nitriles (Agrawal and Kurashige 2003; Hopkins et al. 2009; Manici et al. 1997; Rask et al. 2000; Tierens et al. 2001). Stress caused by abiotic factors like drought has different effects on GSL composition and content. In several studies (Mewis et al. 2012; Radovich et al. 2005; Schreiner et al. 2009; Tong et al. 2014) either an increase of aliphatic GSLs under drought stress or a decrease, no effect or a less pronounced increase of aromatic including (indolic) GSLs in *Brassica* species and in *Arabidopsis thaliana* L. was shown. Other studies of GSLs in drought-stressed *Brassica* species showed a reduction or insignificant changes in the GSL content (Khan et al. 2010; Robbins et al. 2005). In both studies the content of indolic GSLs was predominant.

Only a few studies exist on the effect of salinity on GSLs. In two *Brassica napus* cultivars the total GSL concentration increased under salinity (Qasim et al. 2003).

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*Brassica oleracea* showed an increased total GSL content under the influence of salt (López-Berenguer et al. 2008, 2009). In these studies measurements of total GSL contents were performed, while other studies also analyzed single GSLs. In radish sprouts, for example, one aliphatic GSL was predominant and its concentration increased by increased salinity (Yuan et al. 2010). Zaghdoud et al. (2012) found that the total GSL content was not altered by salinity in a *B. oleracea* cultivar, while another cultivar showed a decrease in the total GSL content. This cultivar showed a decrease in indolic GSLs but an increase in aliphatic GSLs. Bloem et al. (2014) revealed no significant changes in contents of the aromatic GSL glucotropaeolin in salt stressed *Tropaeolum majus*. All studies previously mentioned applied low salinity to glycophytes, while in this study high salinities to a halophyte species were applied.

Several antioxidants are altered in their accumulation by salinity, often enhanced under stress conditions, like the non-enzymatic metabolites ascorbate, glutathione, carotenoids, tocopherols, and phenolics. These antioxidants serve, besides the enzymatic antioxidants, to scavenge or detoxify reactive oxygen species (ROS) induced by stresses like salinity (Noctor and Foyer 1998; Sharma et al. 2010).

In this study the salt-tolerant species Lepidium latifolium was investigated as we were interested in the GSL concentration in relation to salinity and to identify the time point when the GSL content was highest after beginning of the salt treatment. Sometimes classified as a halophyte, L. latifolium is native to southern Europe and Asia (Kaur et al. 2013; Zhao et al. 2010). Its tolerance to salt allows it to grow along the coastline but it is also found in the cold Himalayan region (Gupta et al. 2013; Kaur et al. 2013). It has been identified as an invasive species in North America (Francis and Warwick 2007). Fortunately, the GSL spectrum of L. latifolium was analyzed previously indicating eight GSLs with sinigrin being the dominant GSL (Kaur et al. 2013). The use as a vegetable and the medicinal utilization were also previously described (Kaur et al. 2013; Navarro et al. 1994). The influence of salinity on biomass, oxygen radical absorbance capacity (ORAC), ascorbate, phenolics and flavonoids, and GSL spectrum and contents of L. latifolium influenced by salt were analyzed in this study. The concentration of aromatic GSLs decreased or showed no clear reaction, whereas indolic GSLs decreased and aliphatic GSLs increased or were indifferent towards salinity stress. All other metabolites except ascorbate that showed varying responses were positively affected by salinity. Thus ORAC increased accordingly. The duration of stress application had an influence on the concentration of some of the metabolites analyzed.

*Lepidium latifolium* seeds (Rühlemann's Kräuter und Duftpflanzen, Horstedt, Germany) were germinated on propagation soil (Einheitserde, Einheitserdewerk Hameln-Tündern, Germany). After a period of 5 weeks, plants were transplanted to sand of 0–2 mm grain size (Hornbach, Hannover, Germany). During the nursing time, the plants were watered with modified Hoagland solution (Epstein 1972). Mature plants were grown under greenhouse conditions at around 22 °C. Sodium vapor lamps (SON-T Agro 400, Philips, Amsterdam, Netherlands) served as an additional light source, providing a photoperiod of 14 h light and a quantum fluence rate of 350 µmol m<sup>-2</sup> s<sup>-2</sup>. Finally, the plants were transferred to aerated containers

with 13.5 L solution containing 3.57 mM NaNO<sub>3</sub>, 316 mM H<sub>2</sub>NaPO<sub>4</sub>  $\times$  H<sub>2</sub>O and 23.5 mM Fe-EDDHA (5.7%) (Duchefa, Haarlem, Netherlands). After 1 week of acclimatization to the hydroponic culture, the sea salt mixture (Seequasal GmbH, Münster, Germany) was added stepwise by an increase of 0.75 PSU every day to the desired concentrations of 0, 15, 22.5, and 30 PSU. Four plants of each salinity treatment were harvested at the time the cultivation solutions reached their final concentration (0 weeks) and 1, 2, 3 and 4 weeks after induction. Whole plants (shoot including leaves) were frozen in liquid nitrogen and stored at -80 °C for further analysis. The metabolite extraction and the determination of total phenols, total flavonoids, oxygen radical absorbance capacity (ORAC) and ascorbic acid were performed as described by Boestfleisch et al. (2014). For the determination of GSLs, frozen, ground leaf material was freeze-dried. One milliliter of 80% methanol was added to 10 mg dried plant material. The sample was placed on a shaker until homogenization and then centrifuged for 5 min at 13,000 g. The pellet was re-extracted in the same way and the supernatants were combined. The supernatant was loaded onto a column (QIAGEN GmbH, Hilden, Germany) containing 2 ml of a 5% (w/v) suspension of DEAE Sephadex A25 (Sigma-Aldrich, Taufkirchen, Germany) in 0.5 M acetic acid (pH 5). The column was then flushed with 10 ml of HPLC-grade H<sub>2</sub>O and 4 ml of 0.02 M acetic acid (pH 5). For desulfating the GSLs overnight at room temperature, 50 µl of sulfatase (Sigma-Aldrich) solution (Thies 1979) was added to 450 µl 0.02 M acetic acid (pH 5) and loaded onto the column as well. Desulfated GSLs were eluted 3 times with 2 ml HPLC-H<sub>2</sub>O. Samples were dried in a vacuum centrifuge overnight and resolved in 300 ml HPLC-H<sub>2</sub>O. Analysis was performed with HPLC system (Knauer, Berlin, Germany) equipped with an Ultra AQ C-18 column (150 x 4.6 mm, 5 µm particle size) (Restek GmbH, Bad Homburg, Germany). For measuring the samples, a volume of 50 µl was injected. A water (solvent A)-acetonitrile (solvent B) gradient at a flow rate of 1 ml min<sup>-1</sup> at room temperature was used. Following gradient was applied: 100% A (6 min), 100-70% A (27 min), 70-40% A (0.1 min), 4.9 min 40% A, 40–100% A (0.1 min), and 19.9 min 100% A. Eluents were monitored at a wavelength of 229 nm. Identification of desulfated GSLs was achieved by comparing the retention time with commercially available GSLs (PhytoLab GmbH and Co. KG, Vestenbergsgreuth, Germany) that were treated the same way as the samples. By means of standard curves of these references, desulfated GSLs were quantified. Integration of peaks and elaboration of data were performed using ChromGate Client/Server Version 3.3.1 (Knauer, Berlin, Germany). GSLs were calculated as the mean of four biological replicates with the standard deviation of these four biological replicates. The total amount of GSLs was calculated as the sum of all individual GSLs. To asses precision and reproducibility of GLS analysis, four technical replicates were prepared by measuring GSL contents in each plant sample four times. The standard deviation relative to the individual GSL content was calculated in these replicates. Values were tested for significance (p = 0.05)with an analysis of variance (ANOVA) using R (version 3.2.2), displaying significant differences between metabolites corresponding to PSU values at different harvest times.

At the starting point there were no significant differences in biomass production of L. latifolium between the different salinities, but after 1 week there was a significant difference between plants grown at 22.5 and 30 PSU and plants grown at 15 PSU, which produced the highest biomass (Fig. 1a). The difference in biomass at the different salinities increased towards week 2 where the highest increase in biomass was observed in plants grown at 0 PSU. This effect was significantly higher than in plants grown at 22.5 and 30 PSU. Plant growth decreased with increasing salinity at this point of experimentation. After 3 and 4 weeks the increase in biomass reached a maximum for plants grown at 15 PSU, followed by 0 PSU. Plants grown at 30 PSU showed the lowest increase in biomass. The ORAC of L. latifolium grown at salt stress (30 PSU) was highest at all sampling dates compared to the other levels of salinity (Fig. 1b). The ORAC maximum was determined at week 0 in plants grown at 30 PSU and decreased slowly towards the 4th week. At week 0 and week 1, ORAC values in plants grown at 30 PSU were significantly higher compared to plants grown at 15 and 0 PSU. This difference decreased, but ORAC values in plants grown at 30 PSU were still significantly higher compared to plants grown at 15 PSU. In the 3rd week plants grown at 30 PSU had again a significantly higher concentration compared to lower levels of salinity. In the 4th week there was no significant difference in the ORAC. The total phenol concentration of plants grown at 30 PSU was the highest during the time of the experiment (Fig. 1c). It was significantly higher than the concentration of plants grown at 0 PSU at all sampling dates with the exception of week 3. In most cases plants grown at 0 PSU had the lowest total phenol concentration followed by 15 PSU. Higher salinity concentrations yielded higher total phenol concentrations. This effect was significant over time except for the 3rd week when plants grown at 0 and 30 PSU had higher total phenol concentrations than plants grown at 15 and 22.5 PSU.

Plants grown at higher salinities of 30 and 22.5 PSU produced higher total flavonoid values than plants grown at 0 and 15 PSU (Fig. 1d). There was a significant difference between plants grown at 30 and 0 PSU at all sampling dates. The differences in total flavonoid values between plants grown in high and low salinities were greatest at the start of experimentation and after 1 week, and became smaller after 3 and 4 weeks.

Small differences in the ascorbate concentration were detected at the beginning of the experiment but plants grown at 0 PSU had a lower concentration compared to plants grown at other salinities (Fig. 1e). In the 1st week plants grown at 0 and 15 PSU had insignificantly lower ascorbate values than plants grown at 22.5 and 30 PSU. Plants grown at 15 and 30 PSU had lower ascorbate values compared to plants grown at 0 and 22.5 PSU in the 2nd week. In the 3rd week there was a significant decline in the ascorbate concentration from plants grown at 0 to plants grown at 30 PSU. In the 4th week plants grown at 0 PSU showed significantly lower values than plants grown at all other salinities. The highest ascorbate concentration in this week was detected in plants grown at 22.5 PSU followed by 15 PSU. The ascorbate content in plants grown at 0 PSU at this point of time.



**Fig. 1** Biomass production (**a**), oxygen radical absorbance capacity (ORAC, **b**) and contents of total phenols (**c**), total flavonoids (**d**), ascorbate (**e**), glucoiberin (**f**), sinigrin (**g**), glucobrassicin (**h**) of *L. latifolium* plants (n = 4). Six-week-old plants were placed into aerated containers and after an acclimatization time of 1 week the salinity was increased by 0.75 PSU every day to the concentrations of 0 PSU, 15 PSU, 22.5 PSU and 30 PSU. Fresh material was harvested upon reaching the targeted salinities (week 0) followed by a weekly interval. Different letters indicate significant differences (p = 0.05) between different PSU values within one point of time

Salinity caused diverging effects on individual GSLs. There are three ways how the GSL concentration was influenced by salinity. Figure 1f shows the first way, an increase in the concentration of glucoiberin. At the beginning of the experiment there was a trend for a higher glucoiberin concentration in plants grown at 22.5 PSU, but this effect was not significant. This trend continued in the 1st week when plants grown at all salinities had a significantly higher concentration compared to plants grown at 0 PSU. In the 2nd week plants grown at 22.5 and 30 PSU had a significantly higher glucoiberin concentration than plants grown at 0 PSU. One week later the significant maximum in glucoiberin concentration was found in plants grown at 0 PSU. This trend continued in the 4th week while all concentrations decreased.

The temporal pattern of the mean sinigrin concentration (Fig. 1g) was similar to the pattern of glucoiberin. A significant difference in the sinigrin concentration in plants grown at 0 and 15 PSU could be detected in week 0. However, after 1 week a trend was visible in such way that plants grown at higher salinities showed higher sinigrin concentrations. There was a significant difference in the sinigrin concentration between plants grown at 30 PSU and plants grown at 0 PSU in week 1 and 4. In week 2, plants grown at 22.5 PSU reflected a higher and significant (towards 0 PSU) sinigrin concentration. In week 3, plants grown at 22.5 and 30 PSU showed both high sinigrin concentrations, which differed not significantly from that of the 0 PSU treatment.

The second way how GSLs were influenced by salt stress can shown exemplary for glucobrassicin (Fig. 1h). There was no significant difference at the beginning of the experiment until the 2nd week, but a trend was observed after 2 weeks. Plants grown at lower salinities accumulated more glucobrassicin, except for plants grown at 15 PSU. After 3 weeks, there was a significant decrease in the glucobrassicin concentration in plants grown at 0 to 30 PSU, which continued towards the 4th week, but plants grown at 15 PSU had the second lowest concentration.

GSL concentrations of other GSLs are shown in Table 1. For the sake of completeness, the values of Fig. 1f-h are also presented in Table 1. The concentration of gluconapin increased with higher salinity towards the 2nd week, and showed the opposite pattern in the 4th week. Salinity decreased the concentration of glucobrassicin and gluconasturtiin. The decline in the GSL concentration from low to high salinity started around the 1st week with gluconasturtiin, but was more distinctive for glucobrassicin. The concentrations of glucocheirolin, glucoraphanin, and glucotropaeolin did not show a clear pattern for a de- or an increase under salinity stress, which represents the third way GSLs were influenced by salt stress. Within week 1 and 2 the maximum of glucotropaeolin was reached in plants grown at 30 PSU but changed to 0 PSU in week 3 and 4. The highest glucocheirolin concentration accumulated in plants grown at 15 PSU in week 2 and 3 and at 22.5 PSU in week 4. With respect to the relatively high standard deviation it is important to keep in mind that each single determination was done with individual plants.

The measurement of GSL contents in technical replicates resulted in standard deviations not higher than 20% with a mean of 8% relative to the individual GSL content (data not shown). In contrast, relative standard deviations of biological

	Total GSL	amount	$5.80 \pm 1.68$	а	$11.6\pm5.83$	þ	$9.60\pm3.14$	ab	$10.9\pm3.49$	p	$9.80\pm2.23$	а	$13.8\pm0.97$	ab	$15.8\pm1.21$	p	$18.0\pm0.95$	þ	$11.9\pm0.89$	а	$15.1\pm2.51$	ab	$18.6\pm1.64$	p	$17.6\pm4.01$	p	(continued)
	Gluco-	nasturtiin	$31.5\pm19.7$	а	$72.5\pm50.3$	а	$69.0\pm50.2$	а	$69.2\pm29.9$	а	$227\pm183$ a		$48.7\pm13.0$	а	$94.8\pm33.7$	а	$83.9\pm12.1$	а	$207 \pm 41.1$ a		$133 \pm 117$ a		$55.3\pm17.6$	а	$73.2\pm19.8$	а	
	Gluco-	brassicin	$1.25\pm0.91$	а	$1.00\pm0.31$	а	$1.28\pm0.99$	а	$1.83\pm1.51$	а	$2.24\pm1.23$	а	$1.30\pm0.49$	а	$2.38\pm0.20$	а	$2.16\pm1.08$	а	$3.00\pm0.77$	а	$1.86\pm0.87$	а	$2.67 \pm 1.35$	а	$1.67\pm0.91$	а	
	Gluco-	tropaeolin	$0.39\pm0.17$	а	$1.14\pm0.45$	а	$0.99\pm0.34$	а	$1.18\pm0.24$	а	$0.99\pm0.58$	а	$0.89\pm0.22$	а	$1.39\pm0.30$	ab	$2.48\pm0.97$	þ	$0.83\pm0.25$	а	$0.80\pm0.35$	а	$1.04\pm0.38$	а	$1.33\pm0.26$	а	
		Gluconapin	$1.45\pm1.58$	а	$2.23\pm2.23$	а	$3.22\pm3.03$	а	$6.12\pm7.46$	a	$8.87\pm12.5$	a	$3.64\pm4.79$	a	$9.41\pm5.91$	a	$16.1\pm13.0$	a	$3.01\pm0.56$	a	$5.81\pm3.37$	ab	$6.21\pm5.49$	ab	$17.2\pm12.4$	þ	
	Gluco-	raphanin	$15.9\pm3.50$	а	$34.4\pm18.7$	а	$19.8\pm9.76$	а	$31.6\pm9.14$	a	$21.4\pm12.5$	a	$21.6\pm3.96$	a	$34.7 \pm 4.88$	ab	$41.2\pm3.53$	þ	$37.9\pm6.13$	a	$27.1\pm11.5$	a	$29.9\pm9.12$	a	$31.6\pm2.51$	a	
e sum of use		Sinigrin	$5.31 \pm 1.45$	а	$10.1\pm5.18$	p	$8.19\pm2.61$	ab	$9.37\pm3.18$	ab	$8.50\pm1.49$	a	$12.4\pm0.70$	ab	$13.6\pm1.31$	þ	$14.8\pm0.33$	þ	$10.7\pm0.84$	a	$13.8\pm2.78$	ab	$16.6\pm1.70$	þ	$15.4\pm3.74$	þ	
opaeoun and m		Glucoiberin	$84.1\pm49.9$	а	$247 \pm 160 \text{ a}$		$297 \pm 148 a$		$236\pm58.9$ a		$73.9\pm30.8$	а	$442\pm118~\mathrm{b}$		$658 \pm 447 \text{ b}$		$507\pm100$ b		$43.5\pm11.0$	а	$321 \pm 216$ a		$871 \pm 412 \text{ b}$		$761 \pm 219 \text{ b}$		
sinigrin, giucou	Gluco-	cheirolin	$0.65\pm0.47$	а	$0.93\pm0.81$	а	$0.70\pm0.57$	а	$0.72\pm0.36$	a	$1.19\pm0.73$	a	$1.02\pm0.47$	a	$1.17\pm0.41$	a	$1.12\pm0.48$	а	$1.15\pm0.22$	a	$1.23\pm0.40$	a	$0.87\pm0.56$	а	$0.51\pm0.29$	а	
FIM TOT S		PSU	0		15		22.5		30		0		15		22.5		30		0		15		22.5		30		
n g Iomu	Time in	weeks	0								1								2								

Time in		Gluco-			Gluco-		Gluco-	Gluco-	Gluco-	Total GSL
weeks	PSU	cheirolin	Glucoiberin	Sinigrin	raphanin	Gluconapin	tropaeolin	brassicin	nasturtiin	amount
e	0	$1.01\pm0.36$	$44.9\pm12.2$	$11.9\pm1.34$	$46.8\pm11.5$	$3.64 \pm 1.94$	$1.58\pm1.51$	$4.82\pm2.32$	$431\pm284~a$	$14.0\pm1.84$
		а	а	а	a	а	а	а		a
	15	$1.13\pm0.70$	$141 \pm 73.4$ a	$14.1\pm3.21$	$24.9\pm12.2$	$8.56 \pm 11.3$	$0.91\pm0.84$	$3.54\pm4.53$	$279\pm391$	$15.4\pm2.23$
		а		а	p	а	þ	ab	ab	a
	22.5	$0.85\pm0.41$	$363\pm82.4~\mathrm{a}$	$14.9\pm2.51$	$21.0\pm 8.20$	$6.8\pm10.4~\mathrm{a}$	$1.08\pm0.47$	$2.42 \pm 1.81$	$72.3\pm29.7$	$16.4\pm2.41$
		а		а	þ		þ	ab	p	a
	30	$0.46\pm0.24$	$701 \pm 170 \text{ b}$	$14.9\pm2.05$	$27.4\pm8.42$	$3.81\pm3.61$	$1.01\pm0.32$	$1.28\pm0.31$	$53.3\pm4.73$	$16.7\pm2.15$
		а		а	p	а	þ	p	p	а
4	0	$0.63\pm0.29$	$27.1\pm8.95$	$8.46 \pm 1.02$	$23.7 \pm 3.98$	$7.63 \pm 3.93$	$1.50\pm0.86$	$5.35\pm4.07$	$247\pm116$ a	$10.3\pm1.32$
		а	а	а	a	а	а	а		а
	15	$0.90\pm0.81$	$78.1\pm6.46$	$14.3\pm1.09$	$21.2\pm9.85$	$4.14\pm3.52$	$0.32\pm0.06$	$1.95\pm0.78$	$159\pm5.97~\mathrm{a}$	$14.8\pm1.14$
		а	а	þ	a	а	а	ab		þ
	22.5	$1.30\pm0.89$	$127 \pm 19.1$ a	$17.6\pm2.10$	$39.8\pm19.9$	$4.80\pm5.81$	$0.47\pm0.16$	$4.20\pm3.31$	$176\pm83.4$ a	$18.5\pm2.31$
		a		þ	а	а	а	ab		þ
	30	$0.68\pm0.60$	$461\pm67.4$	$17.8\pm1.18$	$31.0\pm10.0$	$1.52\pm0.92$	$0.77\pm0.17$	$1.70\pm0.35$	$101\pm13.5\mathrm{a}$	$19.1\pm1.28$
		а	þ	þ	а	а	а	þ		þ
Time		NS	***	***	NS		*	*	*	***
PSU		NS	***	***	NS	NS	*	*	* *	***
Time x		NS	***	NS	*	NS	* *	NS	NS	NS

		0.01, or 0.001, respectively
	lysis of variance for the influence of time and salinity	*, **, *** non-significant or significant at $P \le 0.01, 0.05, 0.0$
•	An	SZ

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 Table 1 (continued)

replicates showed values up to 139% with a mean of 38% (data not shown). Because of the low relative standard deviations of technical replicates compared to the ones of biological replicates, the fluctuations in GSL contents emerging from technical procedures were neglected and the standard deviation of GSL contents in biological replicates was used to calculate significant differences between the treatments.

Lepidium latifolium had an optimal growth at 15 PSU in this experiment, as the gain of biomass was the highest at this salinity condition. This was expected as this plant species is a halophyte. Nevertheless, the salt was added in a short time (4 days) and the plant was still able to survive salinity of 22.5 and 30 PSU. The concentrations of the antioxidants were significantly increased by salinity stress. The ORAC increased at the beginning of the experiment and remained on this level for a week. Total phenols and as part of them the flavonoids were affected strongest 1 week after the start of the experiment. The range of concentrations of the mentioned antioxidants was getting smaller towards the end of the experiment closing the difference between high and low salinity. The ascorbate concentration showed a different pattern in comparison to the other antioxidants. There was no significant difference towards the 3rd week of the experiment. Ascorbate concentration quickly changed in *L. latifolium* within 24 h after the beginning of the salinity treatment (Boestfleisch et al. 2014) and returned to a steady state within 4 days.

If the yield of the antioxidants is calculated (multiplying the antioxidant concentration with the biomass produced), there is only one result for all antioxidants: the increase of biomass exceeded the increase in antioxidants (data not shown).

At the beginning of the experiment only changes in the sinigrin content proved to be significant, but 1 week after the induction of salinity stress the content of four out of eight GSLs reacted significantly to salinity. The maximum of glucoiberin shifted along the timescale with increasing salinity from 15 PSU around week 1 to 30 PSU between week 2 and 3 (Fig. 1e and Table 1). Sinigrin contents showed a larger difference at week 4 compared to the beginning, and gluconapin and glucotropaeolin showed an up- and down-regulation, whereas this was more distinctive in the latter one.

GSLs seemed to react partly different than antioxidants. While ORAC, phenols and flavonoids were positively and highly significant intercorrelated (r > 0.8; p < 0.001), and all of them showed a positive significant though weak correlation with ascorbate (r = 0.27–0.38; p < 0.05), GSLs showed ten positive and two negative correlations (Pearson correlation, data not shown). The negative ones were between glucoiberin and glucobrassicin and between glucoiberin and gluconasturtiin. There were some low correlations of mainly aliphatic GSLs with ORAC (glucoiberin: r = 0.36; p < 0.001, glucotropaeolin: r = 0.32; p < 0.005 and gluconasturtiin: r = -0.28; p < 0.05), phenol (glucoiberin, sinigrin, glucoraphanin and glucotropaeolin: r = 0.31–0.43; p < 0.05), flavonoid (glucoiberin, sinigrin, glucoraphanin and glucotropaeolin, r = 0.23–0.49; p < 0.05) and ascorbate values (glucocheirolin and glucoraphanin r = 0.28–0.3; p < 0.05). From the temporal patterns of the GSLs and the correlations between them, they can be classified into three different groups: an up regulation (aliphatic GSLs), a down regulation (aromatic including indolic GSLs) and an intermediate reaction (aromatic excluding indolic GSLs) under salinity. This up and down, or the different regulation, was previously shown for salinity (Yuan et al. 2010; Zaghdoud et al. 2012) and drought treatments (Mewis et al. 2012; Radovich et al. 2005; Schreiner et al. 2009; Tong et al. 2014). Jensen et al. (1996) gave an explanation for these observations under salinity and drought stress. They showed that GSL synthesis increased when the leaf water potential was less than -1.4 MPa for extended periods. In our study, the VWC (volumetric water content) decreased from 85% to 77% at 30 PSU and from 92 to 83% at 0 PSU (data not shown). However, only aliphatic GSLs had a significant correlation with the VWC. The decrease of glucobrassicin with increasing salinity might also be the reason for its ease of oxidation, as it has a high antioxidant capacity (Cabello-Hurtado et al. 2012). These authors demonstrated a relatively high ORAC value for glucobrassicin, much higher than ascorbic acid. However, in our study the most abundant GSL was sinigrin, which has according to Cabello-Hurtado et al. (2012) a 6-7 times lower ORAC compared to ascorbic acid. Therefore, the total GSL content did not contribute much to the ORAC resulting in a low correlation (r = 0.33; p < 0.01). Furthermore many of the studies had only one point of time for the measurement of GSLs, but we could show that GSL contents in salt-stressed plants changed over time.

AS already mentioned sinigrin was the most abundant GSL in *L. latifolium*. Therefore, an increase in salinity elevated the total GSL yield, which would be beneficial for herbivore protection under abiotic stress conditions (Agrawal and Kurashige 2003; Hopkins et al. 2009), in addition to the enhanced growth and antioxidant production at 15 PSU. It was suggested that transient allocation and redistribution of some GSLs indicate a role in signaling mechanisms under abiotic stress conditions (del Carmen Martínez-Ballesta et al. 2013). However, the determination of the exact functions of GSLs in reaction to abiotic stress needs further investigation.

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# The Application of S<sup>0</sup>-Coated Fertilizer to Durum Wheat Crop

## Styliani N. Chorianopoulou, Georgios I. Saridis, Petros Sigalas, Miltos Margetis, Dimitris Benardos, Haris Mavrogiannis, and Dimitris L. Bouranis

Abstract Elemental sulfur  $(S^0)$  is an ideal slow release S fertilizer with a long history as soil amendment. Recently, S<sup>0</sup> has been attached successfully onto the surface of the beads of a commercial fertilizer (F) via a binder (B) by Sulphur Hellas S.A. under the commercial name "Sulfogrow" (FBS<sup>0</sup>). F beads act as a core effectively covered by an amount of 2% (w/w) of S<sup>0</sup> yellow dust. To assess and evaluate the effectiveness of FBS<sup>0</sup>, we monitored the nutritional dynamics of a durum wheat (Triticum durum, Poaceae) commercial crop. The field was divided into two parts; one subject to control F-treatment, and one with the FBS<sup>0</sup>-treatment. Rhizosphere pH, organic matter and humic substances contents were monitored, along with the dry mass, and sulfate, total sulfur, organic nitrogen and iron concentrations in the aerial plant part. The FBS<sup>0</sup>-treated crop presented denser plantation with more robust plants; the accumulated amounts of iron and organic nitrogen per plant were found to be increased at day 61 after sowing in the aerial part by 120% and by 43% respectively, comprising early effects. After day 100, the accumulated dry mass was twice that of control and all accumulation curves were statistically higher than the control ones. The time-course curve of the relative percentage changes (RC) of iron presented in reverse the pattern of sulfate: in contrast, the time-course curve of organic sulfur RC followed the pattern of organic nitrogen RC precisely.

Elemental sulfur ( $S^0$ ) is an ideal slow release S fertilizer with a long history as soil amendment (Schnug 1982; Schnug and Finck 1982; Germida and Jansen 1993; Somani and Totawat 1998). The commercially available  $S^0$  is provided in the form

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of powder and the minimal guarantee of the product is 95% in total S, which is the highest possible concentration of S in a fertilizer. With regard to  $S^0$  application as fertilizer, two agronomic issues should be taken into account. On the one hand, after application, S<sup>0</sup> needs to be oxidized to sulfate, which is the form of S to be absorbed by the root system; therefore,  $S^0$  is not readily available as nutrient and microbial activity is involved. On the other hand, the application of S<sup>0</sup> in the form of powder is a hindrance due to a number of reasons; it is difficult to apply a fertilizer as powder; in a mix with granular materials segregation is observed; contact with the skin may cause irritation and burning; dust is produced whilst inhalation can cause cough and lung irritation. In order to solve the aforementioned issues, in the last three decades several products have been developed in order to encompass  $S^0$ powder with fertilizers (Vitti and Vale 2014). Taking into account the fact that nowadays fertilizers are considered not only as tools, but also as resources, which require sustainability, stewardship, responsibility and participation (Schnug and Haneklaus 2014), it becomes clear that  $S^0$  is still a challenge for fertilizer production and fertilization, both as a tool and a resource. Recently, S<sup>0</sup> has been attached successfully onto the surface of the beads of a commercial fertilizer (F) via a binder (B) by Sulphur Hellas S.A, under the commercial name "Sulfogrow" (FBS<sup>0</sup>). F beads act as a core effectively covered by an amount of 2% (w/w) of S<sup>0</sup> vellow dust. Thus, a new generation of S<sup>0</sup>-coated fertilizers is now commercially available. Is "Sulfogrow" more effective in relation to its core fertilizer and in positive case how much more?

To assess and evaluate the effectiveness of  $FBS^{0}$ , we monitored the nutritional dynamics of a durum wheat (Triticum durum, cv SIMETO) commercial crop established in Lefktra at Viotia county, Greece, in an area of 2.2 ha with calcareous soil. Sowing day and fertilizer application took place in November 13, 2014 (d0). The field was divided into two parts; one of them was subject of control F-treatment according to the local agricultural practices (control crop), whilst the other one received the corresponding FBS<sup>0</sup>-treatment (FBS<sup>0</sup>-treated crop). The control crop was fertilized with a commercial 20-10-10 fertilizer (where potassium was provided as  $K_2SO_4$ ) at a rate of 300 kg ha<sup>-1</sup>. The FBS<sup>0</sup>-treated crop received the equivalent fertilization with the corresponding "Sulfogrow" 20-10-10 commercial fertilizer at the same rate, carrying 2% S<sup>0</sup> (306 kg ha<sup>-1</sup>). At d146 and d167 after sowing, additional fertilization with commercial ammonium nitrate fertilizer took place at the rates of 270 and 150 kg ha<sup>-1</sup>, respectively. At d161 and d166 after sowing, herbicide application took place at the rates of 70 g ha<sup>-1</sup> (Best) and 1.1 L  $ha^{-1}$  (Foxtrot 6.9 W), respectively. Both crops received no irrigation. Each crop was divided into five plots and sampling took place from the internal three ones. A number of plants were collected with their root system and the surrounding soil by means of a shovel. The pH (in 10 mM CaCl<sub>2</sub>), organic matter and humic substances of the rhizosphere were monitored, along with the dry mass (DM), sulfate  $(SO_4^{2-})$ , total sulfur (Stot), organic nitrogen (Norg) and iron (Fe) concentrations of the aerial plant part. Organic sulfur ( $S_{org}$ ) was calculated by subtracting  $SO_4^{2-}$  content from Stot content.



Days after sowing

**Fig. 1** Time-course of pH (**a**, **b**), organic matter (OM; **c**, **d**) and humic substances (HS; **e**, **f**) in the rhizosphere of the analyzed samples. *Empty circles* and *grey line*: control F-treated crop; *full circles* and *black line*: FBS<sup>0</sup>-treated crop. *Bars* represent standard error. Relative changes (i.e. the differences between the corresponding FBS<sup>0</sup>-values and control F-values divided by the control F-values) out of the box were statistically significant (95% confidence level)

Before the additional nitrogen fertilization, rhizosphere pH of both control F-treated and FBS<sup>0</sup>-treated crops fluctuated around 7.75 and 7.83, respectively. After fertilization, rhizosphere pH of both treatments climbed and stabilized between 7.85 and 7.90. Considering the value of 7.83 as mean level, in overall the fluctuation width was narrower in FBS<sup>0</sup>-treated crop (Fig. 1a). Perhaps, this is indicative of an excess uptake of anions over cations, which results to a net removal of protons, thus rising pH (Neumann and Roemheld 2012). Rhizosphere OM content was around 1% up to d105 and then it rose to 1.5%. At d125 the pattern differentiated: in control F-treated crop it was stabilized there until d197 and then it decreased rapidly to the initial value, whilst in FBS<sup>0</sup>-treated field was colonized by far less weeds, a fact that could explain the higher OM content of the F-treated



**Fig. 2** Time-course of organic sulfur [ $S_{org}$ ; **a**], organic nitrogen [ $N_{org}$ ; **b**], sulfate [ $SO_4^{2-}$ ; **c**], and iron [Fe; **d**] concentrations in the aerial part of wheat plants. *Empty circles* and *grey line*: control F-treated crop, *full circles* and *black line*: FBS<sup>0</sup>-treated crop. *Bars* represent standard error

soil after the herbicide treatment, presumably due to the decomposition of the eliminated weeds. Rhizosphere HS content fluctuated around 0.4%, ranging between 0.2 and 0.56% (Fig. 1e). The rhizosphere HS content of the FBS<sup>0</sup>-treated crop presented a significant decrease (-18.5%) at d60 and a significant increase (17.1%) at d197 (Fig. 1f) relative to that of F-treated crop. The FBS<sup>0</sup>-treated crop presented denser plantation with more robust plants, exhibiting reduced number of tillers. After d100, the accumulated dry mass in the FBS<sup>0</sup>-treated crop plants was twice that of control (Figs. 3a and 4a). At d61, iron concentration in FBS<sup>0</sup>-treated plants was close to twice than control (Fig. 2d), resulting in a 120% relative change in Fe accumulation in the aerial part, a fact that constitutes an early effect (Fig. 4b). Increased Fe (Fig. 2d) and Norg (Fig. 2b) concentrations were early responses of  $FBS^{0}$ -treated crop, whilst the corresponding  $S_{org}$  (Fig. 2a) and  $SO_{4}^{2-}$  (Fig. 2c) concentrations were lower that control. Taking into account that Fe availability is highly restricted in calcareous soils due to Fe<sup>3+</sup> precipitation and immobilization, the fact that the FBS<sup>0</sup>-plants accumulated greater amounts of Fe implies that soil reserves of Fe (1) were in adequate amounts to support the enhanced growth of FBS <sup>0</sup>-plants and (2) they had been mobilized. The late nitrogen fertilization caused a significant increase in both  $N_{org}$  concentration and accumulation in the aerial part. After d100, all accumulation curves were higher than the control ones, with those of Norg and Fe being statistically higher from d61 onward (Fig. 3c, e). The time-course curve of the relative percentage changes (RC) of Fe presented in reverse the pattern of that of  $SO_4^{2-}$  (Fig. 4b); the maximum RC value of  $SO_4^{2-}$  (397%) at d125



Days after sowing

**Fig. 3** Time-course of dry mass (DM; **a**), organic sulfur ( $S_{org}$ ; **b**), organic nitrogen ( $N_{org}$ ; **c**), sulfate ( $SO_4^{2-}$ ; **d**), and iron (Fe; **e**) accumulations in the aerial part of wheat plants. *Empty circles* and *grey line*: control F-treated *crop, full circles* and *black line*: FBS<sup>0</sup>-treated crop. *Bars* represent standard error. *Gray lane* indicates the timing of herbicide application and additional nitrogen fertilization

coincided with the minimum RC value of iron (47.9%). In contrast, the time-course curve of  $S_{org}$  RC followed the pattern of  $N_{org}$  RC precisely (Fig. 4c).  $S_{org}$  accumulation in the aerial part was in strong linear relationship with  $N_{org}$  in both control crop ( $R^2 = 0.9667$ ) and  $S^0$ -treated crop ( $R^2 = 0.8998$ ) (Fig. 4d) with the same slope.

The aforementioned data sketch an emerging scenario for the FBS<sup>0</sup>-treated crop: Fe was mobilized early and in higher amounts, whilst the N nutrition status of the FBS<sup>0</sup>-treated crop at this period was a better one. The metabolic links between S and Fe nutrition are well documented in plants that use the chelation strategy for iron accumulation (Zuchi et al. 2012; Forieri et al. 2013), whilst the link between S and N is well known. It seems that the earlier and higher Fe and N status enhanced the uptake of SO<sub>4</sub><sup>2-</sup> and its conversion to S<sub>org</sub>, as it is shown by the higher



**Fig. 4** Time-course of relative percentage changes (RC, %) of DM (**a**),  $SO_4^{2-}$  vs. Fe (**b**), and  $N_{org}$  vs.  $S_{org}$  (**c**) in FBS<sup>0</sup>-treated crop plants, along with the correlation between  $N_{org}$  and  $S_{org}$  accumulations (**d**) in the aerial part of both the control F-treated crop (*empty circles*) and FBS<sup>0</sup>-treated crop (*full circles*)

concentrations of  $SO_4^{2-}$  and  $S_{org}$  at d100.  $S_{org}$  seems to have covered effectively the needs for the formation of Fe-S carrying compounds and the stoichiometry with  $N_{org}$ . On the other hand it seems that the enhanced accumulation of  $S_{org}$  contributed to plant-plant interactions (less weeds) and plant-microbe interactions (robust plants), a fact that seems to be in accordance with the Sulfur-Induced-Resistance concept (Bloem et al. 2014) and the role of S in these interactions.

In conclusion, these data suggest that the "Sulfogrow" product was more effective in comparison to its core fertilizer, as regards the aforementioned parameters. The added 2% S<sup>0</sup> obviously does not act as soil conditioner but rather as enhancer of soil microbial activity, a hypothesis that needs confirmation. Moreover, another point that needs further research is the fact that higher early mobilization of Fe coincides with the statistically significant decrease in rhizosphere HS content. Is this indeed a decrease in HS content or does this reflect a deactivation of the reactive groups of these molecules that are measured by the Mehlich method used? These groups are known to interact with iron in the soil (Li et al. 2012).

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## **Re-assessing Systems Biology Approaches** on Analyzing Sulfate Metabolism

#### **Rainer Hoefgen and Mutsumi Watanabe**

Abstract Mainly driven by the availability of the first Arabidopsis genome sequence in 2000 a rapid development of high throughput analytical techniques were developed and termed transcriptomics. This development was quickly followed by developing metabolite-profiling technologies, metabolomics. The ever-growing data bases made the development of new biostatistical and bioinformatics tools necessary. These 'omics' approaches were also applied to analyze the response of plants towards sulfate deprivation with the aim to gain a more complete, holistic view on plant metabolism and its control in response to varied nutrient supply. Early results though already providing novel results and fostering new routes of investigation were hampered by the incomplete annotation of the genes of the Arabidopsis genome. In recent years this informational gap was largely filled. Thus, we revisit here one old data set obtained at the infancy of 'omics' research and indicate novel conclusions possible when re-assessing these data as well as indicate new possibilities of continued analyses.

Approaches using high throughput methods were first applied on Arabidopsis seedlings at the beginning of this millennium with the arrival of transcriptomics. Arabidopsis seedlings were germinated on or transferred to culture media containing reduced (deprivation) or even zero (starvation) sulfate levels in the culture medium eventually leading to limitation of internal sulfate resources reducing growth and sulfate starvation over time, eventually leading to death of the plants. Plant tissues were analysed by transcriptomics (Nikiforova et al. 2003; Hirai et al. 2003; Maruyama-Nakashita et al. 2003) and metabolomics (Fiehn et al. 2000; Roessner et al. 2001; Nikiforova et al. 2005a). These high through-put experiments and the conclusions drawn from these holistic approaches are commonly summarized under the term systems biology (Fernie et al. 2004). Approaches were then mainly focusing on the model species *Arabidopsis thaliana* as the respective necessary databases and genome wide gene annotations were emerging. The basic principle was, and is, to obtain in an unbiased way as much data as possible

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from an experimental situation, here usually reduction of available sulfate or resupply of sulfate to deprived plants, or from mutant or transgenic plants. Despite the limited resolution of the first transcriptome studies as e.g. initially 10 k cDNA arrays were used (Nikiforova et al. 2003; Hirai et al. 2003) before commercial arrays became available these approaches yielded entirely new data, revolutionizing our knowledge base and hence the experimental approach to access the response of plants to sulfate availability. Other than targeted approaches which are necessarily focusing e.g. upon the target pathway or selected compound classes, systems biology approaches usually employ high throughput systems and try to obtain more complete and comprehensive data sets. The intention for capturing a holistic data set is also supported by integrating results of targeted analyses. Additional to obtaining a catalogue of responses the eventual aim is to describe a system and its response to perturbations as complete as possible, and on as many levels, *i.e.* transcripts, metabolites, proteins, enzyme activities, fluxes etc., as possible and to identify yet unprecedented links, usually due to co-behaviour, between metabolic and cellular processes. Characteristically, sometimes seemingly unrelated gene or metabolite alterations are identified additional to the expected or known changes within the pathway. In order to fully comprehend plant metabolism and physiology it is necessary to also integrate and link these processes with the main response route. The sheer size of the data sets produced by such holistic approaches made it necessary to develop new biostatistical approaches first developed for protein analysis then applied to all fields of molecular biology, now usually termed bioinformatics (Hagen 2000). A further potential and power of systems biology approaches lies in its principal possibility to fuse diverse data such as transcriptomics, metabolomics, proteomics and other omics or targeted analysis data as long as ratios to controls are provided (Nikiforova et al. 2005b). This then allowed linking processes but also deducing new gene functions (Hirai et al. 2004, 2005).

Experiments were executed in different ways which makes it necessary to define the terms to be used more carefully. Plants were either exposed to a complete withdrawal of sulfate from the culture medium, usually using seedlings on agar, in submerged cultures, or in hydroponic systems which allow exact control of the sulfate supply or re-supply and of the time points (Nikiforova et al. 2003, 2004; Wulff-Zottele et al. 2010; Hubberten et al. 2012b; Bielecka et al. 2015). Still in all cases internal sulfur pools are available and can be utilized for growth and only when the resources are diluted to an extent which prevents further growth the plant enters full starvation. For big seeds these resources might be substantial as e.g. we were not able to provoke starvation symptoms in pumpkin (Cucurbita pepo) seedlings unless we used seeds of mother plants exposed to prolonged sulfate deprivation (unpublished). These resources are either deriving from seeds or from phases where initially sulfate is supplied to foster initial growth to establish uniform plant growth. We would term these conditions constitutive and induced starvation, respectively. Plant responses to induced or constitutive starvation are quite different. When seedlings are exposed to starvation from germination onwards we found that they survive longer and need longer to develop symptoms than seedlings pre-grown on sulfate and then exposed to induced starvation (Nikiforova et al. 2003). Under constitutive starvation plants probably adapt to the adverse condition and constantly economise on their resources while under induced conditions abruptly imposed on the plant seedlings stress responses and accumulation of ROS prevail until adaptation and stress response mechanisms help the plants to achieve a new homeostatic status. Also a controlled minimal supply, which we would term deprivation or depletion of sulfate, is a valid experimental system which can be tuned to prevent e.g. accumulation of excess sulfate in the vacuole and will allow investigating adaptation processes rather than emergency responses induced by starvation of the plants (Forieri et al. 2017). Such a system is likely to be closer to conditions plants might be exposed to in nature but responses might be blurred. Sulfate starvation inevitably leads to senescence and plant death (Watanabe et al. 2010) while plants under sulfate depleted or deprived conditions still manage to survive and enter and finish a reproductive cycle with a major negative effect on yield.

As the response to a condition is dynamic (Whitcomb et al. 2014) a systems approach should at its best aim at describing the entire response space. Usually a set of distinct points along the time line are selected allowing identifying changes in response patterns. When plants are exposed to an either sulfate free (starved) or sulfate reduced (deprived) growth condition on either artificial media or soils this reaches from early responses such as the induction of high affinity sulfate transporters (Maruyama-Nakashita et al. 2004) to adaptation processes and rescue responses for example by downregulating the flux to expendable sulfate pools (Aarabi et al. 2016) to eventually nutrient deprivation induced senescence (Watanabe et al. 2010, 2013) or the response to replenishment of nutrients after insufficient supply (Bielecka et al. 2015). Obviously, initial responses are related to rather targeted changes affecting sulfate assimilation and uptake per se while later responses are more pleiotropic and hence overlapping with general stress response patterns (Watanabe et al. 2010). The questions to be still answered in this case are how a specific stress as sulfate depletion is perceived and which elements such as signal molecules, transcription factors or regulation at the protein level through e.g. phosphorylation or protein-protein interaction contribute to the regulation of the various processes, the response network. Some progress has been made with e.g. identifying SLIM1 (sulfur limitation1; Maruyama-Nakashita et al. 2006) in the ethylene-insensitive-like (EIL) transcription factor family and SDI1 and SDI2 (sulfur deficiency induced 1 and 2; Aarabi et al. 2016) containing tetratricopeptide repeat (TPR)-like domain which mediates protein-protein interactions as regulators and O-acetylserine as a signal (Hubberten et al. 2012a).

One of the first sulfur networks (Fig. 1) was provided in 2005 (Nikiforova et al. 2005b), which was already exploiting joint transcriptomics and metabolomics data. The network was based on similarity of co-behaviour of expression patterns and metabolite concentrations. Essentially, the connectivity was deduced by proof of association. The transcriptomics analysis provided 6454 transcripts changed by at least 1.5-fold and additionally, 81 metabolites were determined. A correlation analysis aiming at reducing the noise component finally resulted in a correlation

matrix containing 541 elements significantly changed and showing stable co-behavior patterns. Within this correlation network genes related to sulfate metabolism formed five distinct groups, three of them related to sulfur (Fig. 1a). When combining this to a hierarchical analysis of elements we find the five groups to span over several layers of the hierarchy as indicated in the table included to Fig. 1b. This enrichment of sulfur related genes and their assorting into connected groups proved the validity of this early network. An important limitation at this stage and in comparable current analyses is the fact that only those elements are identified that show changes between conditions, be it differential transcript or metabolite abundances. Genes such as *SLIM1* (Maruyama-Nakashita et al. 2006) which are hardly regulated but obviously post-translationally modulated upon conditions of reduced sulfate availability will not be picked up and need to be identified by other methods, such as mutagenesis in the case of SLIM1.

A hallmark of such networks is the connectivity, which describes the number of vertices (i.e. genes or metabolites) showing a close co-behavior to a certain gene (Nikiforova et al. 2005b) indicated by a connecting line (edges). Thus, it was possible to derive novel conclusions based on the connectivity of network elements (Nikiforova et al. 2003, 2004, 2005a, b; Sasaki-Sekimoto et al. 2005; Hoefgen and



Fig. 1 Combined transcriptomics and metabolomics co-behavior network. Genes and metabolites of sulfur metabolism form five groups within the network, three of them connected to sulfur

Layer	Metabolite	Gene	ATG ID	
1	sulfur			
2		SULTR4;1	At5g13550	I
		SAMS3	At3g17390	I
		SULTR2;1	At5g1180	п
		BSAS5;1	At3g03630	ш
3		SERAT2;2	At3g13110	I
		CGS	At3g01120	I
		SIR	At5g04590	п
		BSAS1;1	At4g14880	п
		ATPS4	At5g43780	ш
		SULTR3;2	At4g02700	IV
		ATPS2	At1g19920	v
4		SULTR2;2	At1g77990	I
		BSAS3;1	At3g61440	I
		APR2	At1g62180	п
		BSAS4;2	At3g04940	п
		SAMS1	At1g02500	п
		BSAS2;1	At2g43750	ш
		SERAT3;1	At2g17640	IV
		SERAT2;1	At1g55920	v
5		BSAS4;1	At5g28020	I
		MS2	At3g03780	п
		SULTR5;1	At1g80310	v

Fig. 1 (continued)

в

Nikiforova 2008; Falkenberg et al. 2008) such as a connectivity of the sulfur depletion response to calmodulin-controlled responses and a link to auxin (Aux) and jasmonic acid (JA). Some genes displaying very high numbers of connections can be assumed to be of central function. Like in a wheel they form a hub or a centre of connectivity. Such a gene is *IAA28* (At5g25890), which shows under conditions of sulfate depletion 28 links. A detailed analysis of IAA28 showed that it controls to a certain extent thiol levels and might be involved in repressing side root development under sulfate deprived conditions (Falkenberg et al. 2008; Hubberten et al. 2012a, b).

The validity and richness of this early dataset can for example further be seen when pulling out metabolite signatures (Nikiforova et al. 2005b). The above mentioned regulator *IAA28* is directly co-behaving with serine, putrescine, GSH, asparagine and g-aminobutyric acid which in turn is linked to the marker molecules, *O*-acetylserine (OAS) and ornithine, which are known to be altered under



Fig. 2 Reassessing network annotation. Hierarchical clustering within a subfragment of the combined transcriptome and metabolome sulfate response network displaying 39 genes and their connectivities on the basis of co-behaviour. While in the network in 2005 only 7 genes could be annotated, now, all 39 genes can be annotated and functions are at least partially revealed

conditions where sulfate availability is inadequate for sustained plant growth as shown through a detailed metabolomics analysis (Nikiforova et al. 2005a). *IAA28* is co-behaving with GSH, which is in a hierarchical correlation again directly linked to *LSU1* (At3g49580), which is co-behaving directly with sulfur contents under sulfate deprived growth conditions in plants (Sirko et al. 2015). As we could decipher a cause to effect relation in this hierarchy (Nikiforova et al. 2005b) this linear correlation of sulfur-*LSU1*-GSH-*IAA28* would therefore provide a suggestion for an in depth analysis investigating this link. Interestingly LSU has been linked to ethylene metabolism, thus linking auxin and ethylene effects under sulfate depleted conditions (Sirko et al. 2015).

Gene annotation, especially for Arabidopsis has greatly improved over recent years (cf. www.tair.org). When assessing the subnetwork in Fig. 2 we could initially identify only 7 of the 39 displayed gene models (Nikiforova et al. 2005b). However,

by now all 39 genes have been annotated and at least putative functions have been assigned which are displayed in the Table 1. This allows re-assessing the data set of which we will display one example.

IAA28 (At5g25890) is linked to the "downstream" genes At5g42650, At1g16890, At1g48690 in this subnetwork (Fig. 2) which are part of the next hierarchical layer. At5g42650 has already been annotated as allene oxide synthase (AOS) being involved in jasmonic acid (JA) biosynthesis providing a link between auxin and JA metabolism when plants are exposed to a reduced availability of sulfate. JAs trigger multiple stress response pathways and influence developmental processes (Malek et al. 2002). At1g16890 has been newly annotated as ubiquitin conjugating enzyme 13B (UBC13B) which functions in root development. An *ubc13* knockout mutant displays shorter primary roots, less lateral roots and fewer root hairs. Aux/IAA (indole-3-acetic acid) proteins, as for example IAA17, accumulate in the mutant leading to a reduced auxin response and, hence, the observed phenotype (Wen et al. 2014). This perfectly relates to the described function of IAA28 under sulfate deprived conditions (Falkenberg et al. 2008) with which UBC13 is linked in the network (Fig. 2). It has still to be elucidated whether IAA28 is acting upstream of UBC13, which we would assume due to its position in the network hierarchy, or whether IAA28 is a target for UBC13 driven poly-ubiquitination. The third gene, At1g48690, has now been annotated as an auxin-responsive GH3 family protein (Gutierrez et al. 2012), again logically linking to the IAA28 gene. Auxin response factors ARF6, ARF8, and ARF17 form a complex regulatory circuit, which regulates a set of auxin inducible genes, termed Gretchen Hagen3 (GH3) genes, GH3.3, GH3.5, and GH3.6, encoding acyl-acidamido synthetases. These three GH3 genes have been shown by Gutierrez et al. (2012) to be required for fine-tuning adventitious root initiation in Arabidopsis thaliana acting by modulating JA homeostasis. Thus, this subset of genes linked to the auxin responsive IAA28 provides a complex link to root development under sulfate deprivation involving a crosstalk between JA, auxin, and possibly also ethylene. The next steps, thus, will be to analyze the interconnectivity of the mentioned genes to a greater detail to identify their exact position in the regulatory circuit in response to sulfate deprivation.

As such further parts of the network will be worth re-assessing. Likewise, it would be meaningful to perform a network analysis using more complete data sets available now covering more data points. Examples for such kinds of extended investigations are the studies by Watanabe et al. (2010, 2013), Bielecka et al. (2015), Hubberten et al. (2012a) and Aarabi et al. (2016), which are all based on the pioneering work re-assessed in this paper.

In summary, a re-iterative cycle of identification of important elements within a response network, be it genes and metabolites or other parameters such as protein phosphorylation patterns or fluxes, through systems biology approaches is combined to targeted analyses of the identified elements using all available methods, among them again high throughput systems analysis (Fig. 3). It is further important to cross boundaries and determine overlaps between various response modules. One

Lawor	Matabolita	Gana	Annotat	tion
Layer	Metabolite	Gene	2005	2016 (TAIR)
1	sulfur			
2		At3g44320	+	NIT3 (NITRILASE 3)
		At3g27830		RPL12-A (RIBOSOMAL PROTEIN L12-A)
		At3g49580		LSU1 (RESPONSE TO LOW SULFUR 1)
		At5g53350		CLPX (CLP PROTEASE REGULATORY SUBUNIT X)
		At5g19440		CAD (cinnamyl-alcohol dehydrogenase, putative)
		At5g13800		PPH (PHEOPHYTINASE)
		At3g01420		DOX1 (ALPHA-DIOXYGENASE 1)
		At4g37520		Peroxidase superfamily protein
		At4g34720		AVA-P1 (VACUOLAR H+-PUMPING ATPASE C1)
		At4g30470		NAD(P)-binding Rossmann-fold superfamily protein
		At2g47650		UXS4 (UDP-XYLOSE SYNTHASE 4)
		At5g52240		ATMP1 (MEMBRANE STEROID BINDING PROTEIN 1)
		At5g37600		GLN1;1(GLUTAMINE SYNTHASE 1;1)
		At3g14210	+	ESM1 (EPITHIOSPECIFIER MODIFIER 1)
3		At1g28600	+	GDSL-motif esterase/acyltransferase/lipase
		At1g76680	+	OPR1 (12-OXOPHYTODIENOATE REDUCTASE 1)
		At4g02080		SAR2 (SECRETION-ASSOCIATED RAS SUPER FAMILY 2)
		At5g47820		FRA1 (FRAGILE FIBER 1)
		At4g11320		CP2 (CYSTEINE PROTEASE 2)
		At3g14240		Subtilase family protein
		At3g56800	+	CAM3 (CALMODULIN 3)
		At2g27530		PGY1 (PIGGYBACK1)
		At4g14030		SBP1 (SELENIUM-BINDING PROTEIN 1)
		At1g75270		DHAR2 (DEHYDROASCORBATE REDUCTASE 2)
	putrescine			
	asparagine			
	glutathione			
4		At2g30590		WRKY21 (WRKY DNA-BINDING PROTEIN 21)
		At3g57520		Raffinose alpha-galactosidase/SIP2 (SEED IMBIBITION 2)
		At5g21090		Leucine-rich repeat (LRR) family protein
		At5g62190		PRH75 (DEAD/DEAH box RNA helicase)
		At5g25890	+	IAA28 (INDOLE-3-ACETIC ACID INDUCIBLE 28)
5		At5g61410		RPE (D-RIBULOSE-5-PHOSPHATE-3-EPIMERASE)
		At1g07920		GTP binding Elongation factor Tu family protein
		At5g11770		NADH-ubiquinone oxidoreductase 20kDa subunit
		At1g13060		PBE1 (20S PROTEASOME BETA SUBUNIT E1)
		At2g05440		GRP9 (GLYCINE RICH PROTEIN 9)
		At4g35410		Clathrin adaptor complex small chain family protein
		At5g42650	•	AOS (ALLENE OXIDE SYNTHASE)
		At1g16890		UBC13B (UBIQUITIN CONJUGATING ENZYME 13B)
		At1g48690		Auxin-responsive GH3 family protein
		At3g48990		AAE3 (ACYL-ACTIVATING ENZYME 3)



Fig. 3 Process scheme of a systems biology analysis. The process can be repeated (*outer circle*) or alternative analytical approaches can be employed to achieve a consistent answer to the original question

example is that nutrient starvation inevitably culminates into a specific nutrient depletion induced senescence (NuDIS) overlapping, but not identical to developmental senescence. The task will be to determine the regulators controlling how and which parts of certain modules are used to evoke the proper physiological response of a plant to a given environmental or biological challenge.

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# Combining Isotope Labelling with High Resolution Liquid Chromatography-Tandem Mass Spectrometry to Study Sulfur Amino Acid Metabolism in Seeds of Common Bean (*Phaseolus vulgaris*)

#### Jaya Joshi, Justin B. Renaud, Mark W. Sumarah, and Frédéric Marsolais

Abstract Sulfur plays a crucial role in plant metabolism, protein biosynthesis, homeostasis and defense mechanisms. Like in other grain legumes, the protein quality of common bean is limited by the sub-optimal concentration of methionine and cysteine. *S*-methylcysteine is a non-proteinogenic sulfur amino acid, characteristically found in the *Phaseolus* as well as *Vigna* species. In mature seeds it accumulates in the form of its dipeptide  $\gamma$ -glutamyl-*S*-methylcysteine. In human diet, this sulfur containing amino acid and its dipeptide cannot substitute cysteine or methionine. Despite the inverse relationship between the concentration of cysteine and methionine versus *S*-methylcysteine, to date very little is known about the biosynthesis of the latter in common bean. Here, we developed a method, combining stable isotope label tracking with high resolution liquid chromatography-tandem mass spectrometry to investigate the pathways of *S*-methylcysteine and  $\gamma$ -glutamyl-*S*-methylcysteine biosynthesis in the developing seed of common bean.

Sulfur plays a key role in cell metabolism. Sulfur amino acids are essential for protein synthesis and play a crucial role in cell homeostasis through the biosynthesis of glutathione (GSH;  $\gamma$ -glutamyl-cysteinyl-glycine) (Hernández et al. 2015). In plants, sulfur provided in the form of sulfate is assimilated, reduced and incorporated into amino acids for further downstream protein and metabolite synthesis. Sulfate from soil

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is assimilated into adenosine 5'-phosphosulfate, which is reduced to sulfide for incorporation into cysteine (Mugford et al. 2011). Cysteine acts as the first organic compound containing reduced sulfur synthesized by the plant (Takahashi et al. 2011). During cysteine biosynthesis, the amino acid backbone is derived by serine metabolism via *O*-acetylserine (OAS) while sulfur comes from sulfate uptake and assimilatory sulfate reduction (Leustek et al. 2000). Further, cysteine acts as a precursor of methionine and GSH biosynthesis (Hell and Wirtz 2011). GSH plays a critical role in homeostasis and cellular defense, including redox status, signal transduction and detoxification (Noctor et al. 2011). Some other fates of cysteine include phytochelatins, iron-sulfur clusters, vitamin cofactors, and the biosynthesis of multiple secondary metabolites (Bonner et al. 2005). Apart from these metabolites some plant species also synthesize non-proteinogenic S-amino acid derivatives that might act as a storage sink for assimilated sulfur (Table 1). In Allium spp. the S-alk(en)yl-cysteine sulfoxides give rise to their characteristic aroma and flavour. Common bean (*Phaseolus vulgaris*) and several Vigna species accumulate S-methylcysteine (S-methylCys) and its dipeptide,  $\gamma$ -glutamyl-S-methylcysteine ( $\gamma$ -Glu-S-methylCys), which are non-proteinogenic in nature (Zacharius et al. 1959; Giada et al. 1998; Taylor et al. 2008).

Protein quality in legumes is compromised due to suboptimal levels of the sulfur amino acids, cysteine and methionine. Among all legumes produced in the world, common bean is considered as one of the best for human consumption (Broughton et al. 2003). However, major seed proteins present in common bean, such as the 7S globulin phaseolin and lectin phytohaemagglutinin have a low concentration of methionine and cysteine (Sathe 2002; Montoya et al. 2010). Despite having a low concentration of methionine and cysteine, common bean seeds accumulate approximately 0.4 nmol/mg of free S-methylCys and 9.1 nmol/mg of  $\gamma$ -Glu-S-methylCys (Yin et al. 2011), whereas total S-methylCys can reach up to 21.8 nmol/mg (Taylor et al. 2008). These S-amino derivatives accumulate exclusively in the seeds (Watanabe et al. 1971). They cannot substitute methionine or cysteine in the human diet (Padovese et al. 2001). An inverse relationship was previously observed between total cysteine and methionine vs. S-methylCys concentration in common bean. Cysteine was elevated by 70%, and methionine by 10% in SMARC1N-PN1 compared to SARC1, which are genetically related lines that differ in their storage protein composition. Increase in cysteine and methionine was mostly at the expense of the non-protein amino acid S-methylCys, suggesting that S-methylCys and  $\gamma$ -Glu-S-methylCys act as storage forms of excessive sulfur that cannot be accommodated in the protein pool (Taylor et al. 2008).

Although the primary sulfur assimilation pathway in common bean is well studied, the biosynthesis of *S*-methylCys and its dipeptide,  $\gamma$ -Glu-*S*-methylCys is not fully understood. In this study, we developed a method using <sup>13</sup>C and <sup>15</sup>N labelled serine or cysteine in a feeding experiment to elucidate the biosynthesis of *S*-methylCys and  $\gamma$ -Glu-*S*-methylCys in the developing seed of common bean. The fate of the labelled (<sup>13</sup>C<sub>3</sub>, <sup>15</sup>N) serine or cysteine was monitored by a combination of targeted and non-targeted high resolution mass spectrometry (HRMS).

For this study, common bean (*Phaseolus vulgaris*) genotype BAT93 plants were grown in growth cabinets (Environmental Growth Chambers, Chagrin Falls, OH,

Amino acid derivatives	Plant species	References
S-methylCys	Phaseolus vulgaris, Vigna unguiculata, and Vigna radiata	Baldi and Salamini (1973) and Evans and Boulter (1975)
S-methylCys sulfoxides	Brassica species	Fales et al. (1987) and Marks et al. (1992)
γ-Glu-S-methylCys	P. vulgaris, V. unguiculata, and V. radiata	Kasai et al. (1986) and Giada et al. (1998)
γ-Glu-methionine	Vigna mungo	Otoul et al. (1975)
γ-Glu-S-ethenylcysteine	Vicia narbonensis	Arias et al. (2005) and Sanchez- Vioque et al. (2011)
S-Alk(en)yl-cysteine sulfoxides	Allium species	Jones et al. (2004), Rose et al. (2005), and Yoshimoto et al. (2015)
S- Methylhomoglutathione	V. radiata and P. vulgaris	Kasai et al. (1986) and Liao et al. (2013)

 Table 1
 Sulfur amino acid derivatives reported in different plant species

USA) under 16 h light (300–400  $\mu$ mol photons m<sup>-2</sup> s<sup>-1</sup>) at a temperature cycling between 18 and 24 °C as described previously (Pandurangan et al. 2012). Seeds were germinated in vermiculite and 12-day old seedlings were transplanted to pots (17 × 20 cm) containing Pro-Mix BX (Premier Tech, Rivière du Loup, Québec, Canada).

Previously, free amino acids were profiled in BAT93 seeds by HPLC. Free SmethylCys biosynthesis takes place during early seed developmental stages, such as stages III – heart stage, and IV and V – cotyledon stages (Walbot et al. 1972). In early developing stages, S-methylCys concentration was equal to 0.40 nmol per mg seed weight while accumulation of  $\gamma$ -Glu-S-methylCys began later in development, from stage VI – maturation to mature seed (Yin et al. 2011). To track the incorporation of stable isotopes in S-methylCys, we chose to feed the labelled precursors, cysteine and methionine, to developing seeds at stage IV – cotyledon. Developing pods were harvested, surface sterilized with 0.5% bleach and soap water for 5 min. This was followed with three washes for 10 min each and dissection for seed collection. Every seed was weighed and immediately the seed coat was removed for improved uptake of the labelled amino acids. Cotyledon weight was taken before transferring them to 25 ml standard line cell culture flasks (VWR, Mississauga, Ontario, Canada). These flasks have a filtered vent to provide oxygen for the developing seeds. Each flask contained six cotyledons from three developing seeds and 2.5 ml of filter-sterilized culture media. Culture media contained all essential components as described previously (Obendorf et al. 1983) with some modification: 0.1 mM Na<sub>2</sub>EDTA was replaced with 0.1 mM EDTA ferric sodium salt and 0.1 mM CoCl<sub>2</sub>.6H<sub>2</sub>O was also added to the media. Seeds were grown with different concentrations of labelled serine or cysteine to determine the optimal concentrations for the feeding experiments. Amino acid extracts from these seeds were analyzed by targeted liquid chromatography-tandem mass spectrometry (LC-MS/

MS) to determine the uptake of cysteine and serine relative to endogenous levels. Supplementation of growth media with 8 mM serine for 24 h resulted in sufficient uptake of <sup>13</sup>C and <sup>15</sup>N in serine but not in downstream metabolites such as Oacetylserine (OAS). This concentration was used in a time course experiment in order to determine the optimal incubation time. Insufficient incubation time would prevent the incorporation into important intermediates and downstream products while a long incubation time could lead to over dilution of the labelled compound. Seeds collected after 24 and 48 h showed efficient incorporation of isotopologues in endogenous serine and cysteine pools as well as in OAS. Therefore, the optimal time of incubation in culture media supplemented by 8 mM amino acid was determined to be 2 days. Using these optimized conditions, 42 seeds in 14 cell culture flasks and 33 seeds in 11 cell culture flasks were incubated with labelled cysteine or serine, respectively. Three groups of treatments were designed based on culture media: serine supplementation, cysteine supplementation and no supplementation. In each treatment group, six cotyledons were grown in a cell culture flask having either labelled or unlabelled serine or cysteine. One group of treatments was not supplemented with serine or cysteine and acted as treatment control. Culture flasks were kept horizontally at room temperature on a shaker. For better seed development, flasks were kept under continuous light and slow shaking (50 rpm). After completion of incubation, seeds were washed three times for 15 min each with sterile water to remove any traces of amino acids on the surface. Seeds were dried and stored at -80 °C following flash freezing in liquid nitrogen. For free amino acid extraction, the frozen seeds were homogenized using steel beads in 1.5 ml Eppendorf tubes. The ground seeds were extracted in ethanol:water (70:30) which is optimal for sulfur containing  $\gamma$ -glutamyl dipeptides (Kasai et al. 1986). Dried amino acids were reconstituted in a 500 µl methanol:water (50:50) solution and filtered through 0.2 µm filters (Pall Life Sciences, Mississauga, Ontario, Canada) into an amber glass HPLC vial. The samples were prepared immediately prior to MS analysis.

In recent years, pairing stable isotope labelling of metabolites with LC-HRMS analysis has provided information on the biosynthesis of targeted compounds and allowed for the relative quantification of cellular compounds (Allen et al. 2009; Creek et al. 2012; Glaser et al. 2014; You et al. 2014; Allen 2016). Here, targeted LC-MS/MS was used to obtain product ion spectra of the unlabelled compounds that could be involved in the biosynthetic pathway of *S*-methylCys. The formula of the major product ions of these compounds were determined by accurate mass and used to map MS/MS dissociation pathways of the precursor ion. These dissociation pathways were then used to predict the number of stable isotopes on the ions in the products that would be synthesized if  ${}^{13}C_3$ ,  ${}^{15}N$  cysteine or  ${}^{13}C_3$ ,  ${}^{15}N$  serine were incorporated. High resolution LC-MS/MS was then used to monitor both, the unlabelled and predicted isotopically labelled compounds (Table 2).

A challenge in detecting *S*-methylCys in the samples was the presence of homocysteine, which is isobaric to *S*-methylCys. With the use of HRMS/MS we were able to meet this challenge. A secondary product of *S*-methylCys is  $C_3H_5O_2$  (73.02900) that arises following the neutral loss of CH<sub>2</sub>S. Lost CH<sub>2</sub>S carbon is from

Table 2         Product ions tracks	ed for unlabelled and	d labelled (*)	compounds						
Name	Neutral formula	m/z	$m/z^*$	Product ion 1	m/z	$m/z^*$	Product ion 2	m/z	$m/z^*$
Serine	$C_3H_7NO_3$	106.0499	110.0570	C <sub>2</sub> H <sub>6</sub> ON	60.0451	63.0481	$C_3H_6O_2N$	88.0397	92.0464
Homoserine	C <sub>4</sub> H <sub>9</sub> NO <sub>3</sub>	120.0655	124.0726	C <sub>3</sub> H <sub>8</sub> ON	74.0606	77.0638	$C_4H_8O_2N$	102.0552	106.0605
Cysteine	$C_3H_7NO_2S$	122.0270	126.0341	C <sub>2</sub> H <sub>6</sub> NS	76.0222	79.0253	C <sub>3</sub> H <sub>3</sub> OS	86.9904	90.0000
S-MethylCys	$C_4H_9NO_2S$	136.0427	140.0498	$C_4H_7O_2S$	119.0161	122.0262	C <sub>3</sub> H <sub>5</sub> O <sub>2</sub>	73.0290	76.0385
OAS	C <sub>5</sub> H9NO <sub>4</sub>	148.0604	152.0675	$C_3H_8O_3N$	106.0499	110.0570	$C_3H_6O_2N$	88.0396	92.0464
Methionine	C <sub>5</sub> H <sub>11</sub> NO <sub>2</sub> S	150.0583	154.0654	$C_4H_{10}NS$	104.0530	107.0566	$C_5H_9O_2S$	133.0316	136.0418
S-Methylmethionine	C <sub>6</sub> H <sub>13</sub> NO <sub>2</sub> S	164.0740	168.0811	$C_6H_{11}O_2S$	147.0474	150.0575			
Cystathionine	$C_7H_{14}N_2O_4S$	223.0747	227.0818	$C_4H_8O_2NS$	134.0268	137.0304			
$\gamma$ -Glu-S-methylCys	$C_9H_{16}N_2O_5S$	265.0853	269.0924	$C_4H_7O_2S$	119.0160	122.0262	$C_4H_{10}O_2NS$	136.0423	140.0498
GSH	$C_{10}H_{17}N_{3}O_{6}S$	308.0911	312.0982	$C_5O_3N_2SH_{11}$	179.0485	183.0556			
HomoGSH	$C_{11}H_{19}N_3O_6S$	322.1067	326.1138	$C_6H_{13}O_3N_2S$	193.0637	197.0712			
S-Methylhomoglutathione	$C_{12}H_{21}N_{3}O_{6}S$	336.1224	340.1295	$C_7H_{15}O_3N_2S$	207.0793	211.0867			
S-Adenosylmethionine	$C_{15}H_{22}N_6O_5S$	399.1445	403.1516	$C_{10}H_{12}O_3N_5$	250.0928	250.0928	$C_{10}H_{20}O_7NS$	298.0962	298.0962

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S-methyl and is not labelled as expected. This product ion does not occur with homocysteine and was used to distinguish S-methylCys from homocysteine. The secondary product ion of a  ${}^{13}C_3{}^{15}N$  labelled S-methylCys would be  ${}^{13}C_3H_5O_2$  (76.03899) after losing CH<sub>2</sub>S. The product ions observed strongly agree with the expected positions of labelled atoms in S-methylCys and helped to distinguish between two sulfur containing amino acids S-methylCys and homocysteine, which share the same molecular weight.

MS data were acquired with a Q-Exactive Orbitrap mass spectrometer (Thermo Fisher Scientific, Mississauga, Ontario, Canada) coupled to an Agilent 1290 HPLC system (Mississauga, Ontario, Canada). Two µL were injected onto an Agilent Zorbax Eclipse Plus RRHD C18 column (2.1  $\times$  50 mm, 1.8  $\mu$ M) which was maintained at 35 °C. Mobile phase A (0.1% formic acid in LC-MS grade  $H_2O$ . Thermo Fisher Scientific) was maintained at 100% for 1.25 min. Mobile phase B (0.1% formic acid in LC-MS grade acetonitrile, Thermo Fisher Scientific) was increased to 50% over 1.75 min, and 100% over 0.5 min. Mobile phase B was held at 100% for 1.5 min and returned to 0% over 0.5 min). The following heated electrospray ionization (HESI) parameters were optimized for the analysis of SmethylCys: spray voltage, 3.9 kV; capillary temperature, 250 °C; probe heater temperature, 450 °C; sheath gas, 30 arbitrary units; auxiliary gas, 8 arbitrary units; and S-Lens RF level, 60%. MS/MS was performed at 17,500 resolution, automatic gain control (AGC) target of 1e<sup>6</sup>, maximum injection time (IT) of 60 ms and isolation window of 1.0 m/z. A top 5 data-dependent acquisition (DDA) method was comprised of a full MS scan at 35,000 resolution, AGC target of 3e<sup>6</sup>, maximum IT of 125 ms, scan range between m/z 70–450 and intensity threshold of 7.7e<sup>5</sup>. The MS/MS scan conditions were identical to the targeted MS/MS method. Normalized collision energy of 25 was used for both MS/MS and DDA methods. Data were analyzed and all theoretical masses were calculated with Xcalibur<sup>TM</sup> software. The proportion of labelled compound present was calculated by dividing the intensity of the labelled signal over the sum of labelled and unlabelled intensities.

Following amino acid extraction,  $80 \pm 4\%$  of the serine content within the  ${}^{13}C_3{}^{15}$  N serine treatment was labelled. Similarly,  $68 \pm 15\%$  of the cysteine content was labelled in the  ${}^{13}C_3{}^{15}$ N cysteine treatment. These data suggest efficient uptake of serine and cysteine by developing seeds from the growth media. To track cysteine biosynthesis in developing seeds, an initial target was the level of isotope labels in OAS. In the treatment group supplemented with labelled serine,  $78 \pm 6\%$  of the OAS pool was labelled, while labelled cysteine treated seeds showed no incorporation of isotopes in OAS which is in harmony with prior studies where OAS was shown to be a precursor of cysteine biosynthesis (Hell and Wirtz 2011).

Under the proposed pathways, both a  ${}^{13}C_3{}^{15}N$  serine or  ${}^{13}C_3{}^{15}N$  cysteine precursor would yield a  ${}^{13}C_3{}^{15}N$  *S*-methylCys. The major product ion of *S*-methylCys arising following the neutral loss of NH<sub>3</sub> (17.0265 Da) is C<sub>4</sub>H<sub>7</sub>O<sub>2</sub>S (119.0161). The major product ion of a  ${}^{13}C_3{}^{15}N$  labelled *S*-methylCys would be  ${}^{13}C_3CH_7O_2S$ (122.0262) after losing  ${}^{15}NH_3$  (18.0236 Da; Fig. 1). MS/MS of both the labelled and unlabelled *S*-methylCys showed the carbon on the CH<sub>2</sub>S neutral fragment is unlabelled, indicating that the *S*-methyl group is not derived from the labelled



Fig. 1 Representative dissociation of protonated unlabelled S-methylCys and labelled S-methylCys. Upper panel: Product ions of unlabelled S-methylCys. Lower panel: Product ions of labelled S-methylCys

serine. The ability to determine the locations of labelled and unlabelled atoms within a larger compound demonstrates the benefits of isotope tracking by MS/MS over single stage MS.

The dipeptide  $\gamma$ -Glu-*S*-methylCys, which accumulates in maturing common bean seed was also monitored by LC-MS/MS in order to identify its biosynthetic precursors.  $\gamma$ -Glu-*S*-methylCys (C<sub>9</sub>H<sub>16</sub>N<sub>2</sub>O<sub>5</sub>S) has *m*/z of 265.0853 for its protonated product ion. Upon collision induced dissociation, two major product ions were produced including a *S*-methylCys fragment C<sub>4</sub>H<sub>10</sub>O<sub>2</sub>NS (*m*/z 136.0423) and deaminated *S*-methylCys C<sub>4</sub>H<sub>7</sub>O<sub>2</sub>S (*m*/z 119.0160); the incorporation of labelled serine or cysteine into  $\gamma$ -Glu-*S*-methylCys would result in product ions <sup>13</sup>C<sub>3</sub>CH<sub>10</sub>O<sub>2</sub> <sup>15</sup>NS (*m*/z 140.0493) and <sup>13</sup>C<sub>3</sub>CH<sub>7</sub>O<sub>2</sub>S *m*/z 122.0261, respectively (Fig. 2).

Interestingly, results of our feeding experiment with serine or cysteine showed similar percentage of isotope label incorporation in either labelled cysteine or serine treatments in  $\gamma$ -Glu-S-methylCys. These results point out needs for looking at labelling patterns in other sulfur containing compounds such as homoglutathione and its derivatives to investigate the possibility of more than one pathway for  $\gamma$ -Glu-S-methylCys biosynthesis.

Single stage HRMS is commonly used in biosynthetic studies to monitor stable isotope incorporation, however, we determined that using high resolution MS/MS greatly reduced the level of background interferences and furthermore, provided evidence for the location of the labelled isotopes within the larger molecule itself. In addition to targeted LC-MS/MS, samples were also analyzed with a non-targeted LC-DDA method to provide a dataset which could be mined retrospectively for other compounds that were not initially predicted to be involved in biosynthetic pathways.



**Fig. 2** Product ion spectra of protonated unlabelled  $\gamma$ -Glu-*S*-methylCys and labelled  $\gamma$ -Glu-*S*-methylCys. *Upper panel*: Product ions of unlabelled  $\gamma$ -Glu-*S*-methylCys. *Lower panel*: Product ions of labelled  $\gamma$ -Glu-*S*-methylCys

In summary, in the past, several strategies have been developed to understand the sulfur amino acid metabolism in legumes. Use of high resolution MS and isotope labelling techniques will help to overcome limitation with sulfur containing metabolite detection and will lead to better understanding of cysteine and methionine derivatives which play crucial role in protein quality and protein quantity of legumes.

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# Investigations on the Current Sulfur and Sulfate Intake of Cattle in Germany: Are There Any Risks for a Consumption Exceeding Recommended Upper Limits?

#### Josef Kamphues, Anne Dohm, F. von und zur Mühlen, and Petra Wolf

Abstract The primary goal of the investigation was a deeper knowledge on the current sulfur and sulfate intake of dairy cows and beef cattle fed common rations in Germany based on roughages and supplemented with concentrates (as protein and energy sources). Finally, specific feeding practices/conditions should be identified that could result in adverse effects in cattle because upper levels for sulfur intake (>4 g/kg dry matter) were exceeded. In grass silages (especially from the 3rd/4th cut, at using S containing fertilizers) values near to 4 g S/kg dry matter were not rare; unexpectedly high contents of sulfur and sulfate were measured in compound feeds (median: 3.70 g S/kg dry matter; >40% derived from sulfates!). These high values are caused by higher proportions of components that contain higher amounts of sulfur and sulfates physiologically (like rapeseed products) or contain high sulfate levels due to diverse processes in the production (DDGS/sugar beet pulp/corn gluten feed) for example like the use of sulfuric acid or sulfur dioxide. Finally, it is recommended to test more frequently and systematically the sulfate content in grass and grass silages (indicating the S supply of plants) but also in concentrates (with an unexpected high variation of sulfur and sulfate) to identify further routes of sulfur entrance.

Sulfur is an essential element for the ruminal microflora to produce S containing amino acids and further nutrients like vitamins (for example biotin and thiamine). Therefore, the sulfur content should reach values of ~2 g S/kg dry matter in the entire ration (McDowell 2003); on the other hand the S content should not exceed values >4 g/kg dm (NRC 2005) in the whole ration due to predisposing effects

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regarding the polioencephalomalacia (PEM) and further undesired consequences (McAllister 2004). According to Dänicke and Schenkel (2009) the sulfur levels of grass and grass silages (samples from southern parts of Germany) did not exceed upper levels (~4 g S/kg dry matter); in whole plant corn silages the S contents often vary around  $\sim 1$  g/kg dry matter, not meeting the S requirements of high producing cattle. In America distinct concentrates (e.g. DDGS, corn gluten feed) are on debate for their high and risky sulfur contents (Drewnoski et al. 2014; Amat et al. 2013, 2014). Regarding sulfate contents of feedstuffs very few data exist up to now contrary to the sulfate contents in water for drinking with a high variation due to geological conditions (NRC 2005; Kamphues et al. 2007). Kamphues et al. (1999) reported high sulfate levels in distinct milk replacers for calves resulting in diarrhea due to laxative effects when young calves were fed liquid diets with more than 600-800 mg SO<sub>4</sub><sup>2-</sup>/l diet. On the other hand Stemme et al. (2003) published a review on condensed molasses (from sugar beet) sometimes containing excessive levels of sulfates (up to 100 g/kg dm) that were tolerated by fattening bulls without any sign of illness. There are two trends that might affect the intake of sulfur and sulfate under current conditions of feeding dairy cows but also of feeding beef cattle: More and more S containing fertilizers are used in green fodder production, including grassland management. Furthermore, increasing amounts of by-products are used in the feed industry producing the compound/supplementary feeds and concentrates for cattle. Thus, there were two main goals of these investigations: The primary aim was to generate basic data on the sulfur and sulfate contents in feedstuffs representing common feeding practices for dairy cows and for beef cattle in Germany. The second goal was a critical look on the relevance of sulfate as an inorganic sulfur source and its contribution to the total S content of feedstuffs and thereby to the total S intake of cattle.

For analyzing the sulfur and sulfate contents in roughages (green fodder, grass, hay, and silages) samples of different origin were available: The majority of samples were grown in northern parts of Germany, most of them submitted by farmers and veterinary practitioners between 2012 and 2015 for quality control (nutritive value, hygienic status, success of preservation). Secondary, there were samples available from further institutions (LUFA North-West, Oldenburg; Lower Saxony Chamber of Agriculture) sent in as dried and ground samples. The different concentrates (especially protein rich components) also were of diverse origin (submitted for analyzing other nutrients), some of them were taken at farms, further ones were conceded by other institutions engaged in feed control and animal nutrition. Finally, various feedstuffs derived from own sampling in the field or at feed manufacturers. The dried and ground samples of feedstuffs were analyzed on the total S content (as well as on nitrogen content) by the Dumas technique [oxidative combustion and determination of both elements (N, S) by thermal conductivity detection]. The sulfate content was measured gravimetrically (boiling up the dried, ground samples with hydrochloric acid (37%) followed by precipitation of sulfates with BaCl<sub>2</sub>, and weighing as BaSO<sub>4</sub>). To demonstrate the relevance of sulfate as a source for total sulfur content its relative proportion was calculated in general (SO<sub>4</sub>-S in percent of total S). The results are expressed as median value due to the non-normal distribution found in most feedstuffs.

The most important roughages used for cattle feeding in Germany and their contents of sulfur and sulfate are presented in Table 1. At first it has to be underlined that there was a large variation in the sulfur and in the sulfate contents of roughages, especially in grass and grass silages. Furthermore, there was an influence of the date of harvesting (first cut compared to the following ones). In grass 41–51% of the total S derived from sulfates, for grass silages these values were even higher (58–59%). In samples of grass but also of grass silages the total S contents varied on a high level (1st vs. following cuts: median of grass 2.19 vs. 3.78 g/kg dry matter; median of grass silages 2.58 vs. 3.76 g/kg dry matter) when compared to upper recommended levels (max. 4 g/kg dry matter). On the other hand, in corn silage lower values were found in general (median: 1.20 g/kg dry matter). Very low sulfate levels (0.31 up to 1.13 g) were measured in samples of whole plant corn silage, too.

From a field study regarding the impact of fertilization of grassland with sulfur, samples of grass were available for the investigation (Table 2). There was a marked effect of fertilization (N, S) on the total sulfur and sulfate content in grass: depending on the S input (0, 20, 40 kg S/ha) the sulfur and sulfate values increased, but additionally reliant on the intensity of nitrogen fertilization: At moderate N levels (220 kg N/ha) the sulfur content of grass increased by about 45% (0 vs. 20 kg S/ha) and by further 8.3% (20 vs. 40 kg S/ha). At high nitrogen fertilization (300 kg N/ha) there was a similar trend, but on a slightly lower level (+28% vs. +10%). It has to be underlined that S from sulfate related to total sulfur reached values of about 60% (Fig. 1).

As expected, low levels of sulfur without detectable sulfate contents were found in cereals. In dried sugar beet pulp largely moderate levels of sulfur (2.37 g S/kg dry matter) occurred but in some samples also high sulfate levels were proved (up to 12.7 g/kg dry matter). Two thirds of total sulfur stemmed from sulfates. Similar sulfur contents (median 4.4 g/kg dry matter) were measured in soybean meal; in this feed material unexpected sulfate values appeared, too (up to 5.87 g/kg dry matter) and here about 30.9% of total sulfur stemmed from sulfates. In general, high levels of sulfur and of sulfate were measured in rapeseed meal, namely about 7 g sulfur and about 7 g sulfate/kg dry matter. More than 30% of the total sulfur were contributed by sulfates (Table 3).

The highest sulfur and sulfate contents were found in DDGS, but with a huge variation: up to 12 g sulfur and up to 18.9 g sulfates per kg dry matter were determined in this by-product of bioethanol production. Here it is noteworthy that samples of DDGS differed markedly regarding the S content (Table 4).

It has to be emphasized that there was no correlation between the nitrogen and the sulfur content as it might be expected assuming that S containing amino acids (S-AA) are the main source of sulfur. Depending on the available roughages, the entire ration for dairy cows also consists of concentrates up to proportions of 50% of the total dm. Besides the use of cereals, dried sugar beet pulp, and soybean meal it is very common to include supplementary compound feeds in the rations (up to

Feed material	n	Sulfur (g/kg dry m	atter)	Sulfate (g/kg dry n	latter)	Sulfate-S $(\%)^a$
Grass		Median	(min-max)	Median	(min-max)	
1st cut	25	2.19	(1.15-4.44)	3.07	(0.94 - 8.68)	41
Following cuts	17	3.78	(2.08–8.97)	6.52	(2.04 - 18.0)	51
Grass silage <sup>b</sup>						
1st cut	42	2.58	(2.23-4.16)	5.18	(2.37–8.16)	59
3rd cut	12	3.76	(2.35-4.58)	6.08	(4.56 - 9.36)	58
Corn silage <sup>c</sup> (whole plant)	13	1.20	(0.75–1.49)	0.89	(0.31 - 1.13)	25
<sup>a</sup> Culfata C in namant of total C						

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"Sulfate-S, in percent of total S  $^{\rm b}$ LUFA North-West: 14% of all samples (n = 681) of grass silage >3.5 g S/kg dry matter  $^{\rm b}$ LUFA North West: 127 samples of corn silage: all in the range 0.9–1.3 g S/kg dry matter

Fertilization (kg/h	na)	Contents (g/kg dm)			
N	S	Crude protein	Sulfur	Sulfate	
220	0	150	2.90	4.96	$(57.0)^{a}$
300		184	2.67	3.44	$(43.0)^{a}$
220	20	160	4.21	7.55	(59.7) <sup>a</sup>
300		186	3.41	5.29	(51.7) <sup>a</sup>
220	40	156	4.56	8.34	$(60.9)^{a}$
300		194	3.76	6.98	(61.8) <sup>a</sup>

 Table 2
 Sulfur and sulfate contents in grass related to the intensity of fertilization with nitrogen and sulfur

Samples of the 3rd cut from July 2014; Lange (2015)

<sup>a</sup>Sulfate-S, in percent of total S



Fig. 1 Sulfur and sulfate content (g/kg dry matter) as well as sulfate-S related to total S (%) in grass from one single location in Lower Saxony in the course of a year

**Table 3** Sulfur and sulfate contents (g/kg dry matter) of feed materials used in the production of compound/supplementary feeds for cattle

Feed material	n	Sulfur (g/	kg dry n	natter)	Sulfate (g	/kg dry n	natter)	Sulfate-S <sup>a</sup>
		Median	Min	Max	Median	Min	Max	%
Cereals	11	1.51	1.15	1.92	Not detectable		-	
Dried sugar beet pulp	10	2.37	1.58	4.42	4.76	2.34	12.7	66.5
Soybean meal, extr.	9	4.37	3.78	4.84	4.46	1.89	5.87	30.9
Rapeseed meal, extr.	13	7.22	6.63	7.59	6.93	3.97	8.76	31.9
DDGS <sup>b</sup>	22	6.42	3.15	11.8	3.38	0.289	18.9	26.6

<sup>a</sup>Sulfate-S, in percent of total S

<sup>b</sup>Dried distillers grains with solubles

	<5 g S/kg dry matter	>5 g S/kg dry matter
n	9	13
Sulfur (g/kg dry matter)	3.15-4.05	6.11–11.8
Nitrogen (g/kg dry matter)	43.2-60.5	49.4–91.5
N/S ratio	12.3–15.3	5.05-9.73
Sulfate (g/kg dry matter)	0.29–4.68	0.66–18.9
Sulfate-S (%) of total sulfur	2.78-43.8	3.08-65.0
Sulfur, derived from S-AA, % <sup>b</sup>	60.4–75.4	24.8-47.8

Table 4 Characterization of DDGS based on the differences in their total sulfur content<sup>a</sup>

<sup>a</sup>Samples available from the Institute of Animal Production, University of Bonn (Agricultural Faculty)

<sup>b</sup>Calculated, assuming a constant amino acid pattern, independent of protein content

Table 5 Sulfur and sulfate contents of common supplementary compound feeds (n = 20) for dairy cows (Dohm 2015)

	Sulfur (g	/kg dry		Sulfate (g	g/kg dry	/
	matter)			matter)		
	Median	Min	Max	Median	Min	Max
Supplementary compound feeds for dairy	3.70	2.37	6.32	4.84	1.82	8.06
COWS						

10–12 kg per animal per day). Regarding its proportion these concentrates are the most important ones, thus having a marked impact on the total sulfur and sulfate intake (Table 5). Supplementary compound feeds for beef cattle have a similar composition regarding sulfur and sulfate contents but the amounts fed per day are much lower (up to 3 kg). In all samples of supplementary feeds the nitrogen content ( $\rightarrow$  crude protein content) was additionally analysed and compared to the sulfur content: there was a significant weak correlation ( $r^2 = 0.253$ ); but it has to be underlined that in the sample with the lowest protein content (126 g/kg dry matter) the highest level of sulfate (8.06 g/kg dry matter) was found.

Contrary to the majority of published data regarding the sulfur contents in feedstuffs it seems that sulfate belongs to the "forgotten" constituents of feedstuffs and ingredients. Although, Scharrer and Jung (1955) observed that the sulfate content of grass (*Lolium perenne*) increased much more than the content of S containing amino acids when the intensity of fertilization with sulfur was elevated. Up to 79% of the total S content in their experiments were contributed by sulfate and only 20% by the S containing amino acids. Furthermore, it is well known that sulfates serve as stored reserves for plants (or parts of plants), and that sulfates accumulate with increasing maturity (Hell 2002). This might explain the trend for higher S contents during the entire vegetation.

Since decades, by-products from the process of bioethanol production are marketed as DDGS and used in the feed industry as protein source for supplementary feeds for dairy cows and beef cattle. Comparing the low sulfur and sulfate levels (near zero) in cereals to the high values in the by-products of cereal fermentation (like DDGS) indicates that further ways of entrance for sulfur and/or sulfates must exist. Here the technology in the production of bioethanol earns special attention: the differences in the composition of DDGS are mainly caused by the use of sulfuric acid in the process of bioethanol production by yeasts on the basis of diverse cereals. Regarding rapeseed meal there is a different reason for the high values of sulfur: The high contents of cysteine and methionine result in corresponding high sulfur contents (methionine: 21.5% S; cysteine: 26.6% S), but only 60% of the sulfur content are accounted to this fraction. Thus, further constituents that result in higher sulfur values must be taken into consideration. Here the glucosinolates are worth to be mentioned: in spite of a high variation in the composition of the diverse glucosinolates the average proportion of sulfur is about 16% (Henkel and Mosenthin 1989). According to Schnug and Haneklaus (1990) there is a strong correlation between the S content and the glucosinolate

(1990) there is a strong correlation between the S content and the glucosinolate content in rapeseed products (extracted meal, cakes). Based on the current findings on sulfates in rapeseed meal, about 32% of the total S is accounted for sulfates and about 60% for the S-AA. As reported in earlier studies, very low levels of sulfur were found in samples of corn silage (whole plant corn silage) with a mean of 1.2 g/kg dry matter, the minimum value was 0.75 g/kg dm and the highest one reached 1.49 g/kg dry matter

corn silage (whole plant corn silage) with a mean of 1.2 g/kg dry matter, the minimum value was 0.75 g/kg dm and the highest one reached 1.49 g/kg dry matter (Dohm 2015). Also in southern regions of Germany low levels (0.9–1.1 g/kg dry matter, according to Wessels 2002; 0.91–0.93 g S/kg dry matter, Walch 1998) that could result in a marginal S supply of the ruminal flora were found. Regarding the sulfur content of grass silages it seems that there is a trend for some higher values in northern parts of Germany (Müller and Engling 2015), especially with values increasing in the course of the year: in the 3rd and 4th cut the median values reached nearly 4 g/kg dry matter, presumably due to the fact that it is more common to use S containing fertilizers in this region. It has to be specially emphasized that sulfate is a main source for sulfur in grass and grass silages: 40 up to 60% of the total sulfur came from sulfate and not from the protein fraction (S-AA). Such high sulfate levels might indicate a surplus of available sulfur for the plants and maybe an impaired protein synthesis in the plants due to other reasons. Especially in dairy cow rations with high proportions of grass silage the sulfur levels may touch or exceed the recommended safe upper levels (max. 4 g S/kg dry matter).

Regarding the sulfur and sulfate levels in common compound feeds for dairy cows (used as supplementary feeds) it has to be underlined that the values vary near the "upper" level of 4 g/kg dry matter. But these findings are not astonishing because of the use of rapeseed products, DDGS, and by-products of the wet milling industry (like corn gluten feed). In this last group of by-products Myer and Hersom (2008) found a high variation of S contents (3.3–7.3 g S/kg dry matter). Furthermore, dried sugar beet pulp and condensed molasses may contain high levels of sulfate as found in earlier studies (cited by Stemme et al. 2003). Based on current data regarding the sulfur and sulfate contents in diverse feed materials and compound feeds, rations for dairy cows based on grass silage and supplemented with high amounts of compound feeds predispose for an intake of sulfur and sulfate that

is near the recommended "upper" level of 4 g S/kg dm in the total ration. On the other hand, whenever higher proportions of corn silage supplemented with cereals and soybean meal are fed, the intake of sulfur and sulfate should be moderate. Only in cases of high sulfate levels in drinking water (in Germany rather rare!) there could be an unintended higher intake of sulfur.

According to Kamphues et al. (2014) high sulfur and/or sulfate intake of ruminants may be associated with different consequences/undesired effects:

- Lower dry matter intake (due to a loss of palatability or due to effects of H<sub>2</sub>S in the central nervous system).
- Reduced utilization of distinct trace elements (especially of copper, selenium, and zinc).
- Impaired formation/metabolism of S-containing vitamins (especially of thiamine; fat soluble vitamins?)
- Impaired quality/composition of faeces, watery diarrhoea (primarily calves are affected due to "laxative effects of sulfate", adult ruminants are much more tolerant)
- Clinical symptoms due to polioencephalomalacia (as a consequence of the forced  $H_2S$  formation in the rumen and its inhalation from eructated gas  $\rightarrow$  histological alterations in the brain, described as malacia and necrosis)

Summarizing lots of publications related to the sulfur intake Kamphues et al. (2014) stated that S contents per 1 kg dm of the whole ration:

- $\leq 1$  g: S requirements of the ruminal flora are not met
- 2 g: even at high performance of cattle S-requirements are met in any case
- 3 g: increased proneness to secondary effects (trace element utilisation  $\downarrow$ ) and to PEM when low fibre diets are fed (concentrate rich diets!)
- 4 g: higher risk for PEM, even at higher roughage intake
- $\geq$ 5 g: diverse clinically obvious reactions are to expect (diarrhoea, apathy, ...), such rations should not be fed to ruminants in higher proportions.

Although it is known for decades that plants can accumulate sulfate as a physiological constituent it did not find the interest in feed science. According to different textbooks on animal nutrition most authors suppose that S containing amino acids are the predominant source of sulfur, but that assessment does not fit, at least not for various kinds of green fodder and some further concentrates. For the feed industry the fact that diverse by-products commonly used for producing compound/ supplementary feeds are characterized by highly variable contents of sulfur and sulfates is of special interest. Therefore, it is recommended to test the sulfate contents of roughages (need for S containing fertilizers?) but also of distinct concentrates (further way of entrance!) more frequently to avoid secondary effects of sulfur/sulfates exceeding the "normal" values and to detect an unintended high intake of sulfur and/or sulfates.

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# Manganese Toxicity Hardly Affects Sulfur Metabolism in *Brassica rapa*

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Abstract Manganese (Mn) is an essential plant nutrient, though at elevated levels in plant tissues it may become toxic. The physiological basis for phytotoxicity is largely unclear. Exposure of *Brassica rapa* to elevated levels of  $Mn^{2+}$  in the nutrient solution resulted in decreased biomass production at  $\geq 20 \ \mu M$  and chlorosis. The Mn content in the shoot increased with the  $Mn^{2+}$  concentration in the nutrient solution and became toxic when it exceeded a four-fold concentration of the control. In contrast to observations with Cu and Zn, elevated and toxic  $Mn^{2+}$  levels did not affect the water-soluble non-protein thiols in both root and shoot and the expression the sulfate transporters, Sultr1;1 and Sultr1;2, in the root.

Mn is an essential plant nutrient and its availability in soil strongly affects plant growth and development (Kováčik et al. 2014; Sadana et al. 2003). Mn functions in several physiological processes, viz. in photosynthesis, where it is associated with the water-oxidizing complex of photosystem II, which catalyzes the photosynthetic  $O_2$  evolution (Mukhopadhyay and Sharma 1991; Millaleo et al. 2013). Moreover, Mn is an important cofactor of several enzymes, e.g., manganese-dependent superoxide dismutase (MnSOD), catalases, glycotransferases, pyruvate carboxylase, nitrate reductase and is involved in amino acid and lignin synthesis (Marschner 1995; Pedas et al. 2005; Humphries et al. 2006; Pittman 2008). Mn is taken up by the plant root as  $Mn^{2+}$ , which availability is strongly affected by the pH of the soil (Humphries et al. 2006; Socha and Guerinot 2014). In alkaline soils (high pH) Mn availability to plants may be low and deficiency may occur, whereas in in acidic soils (low pH) excessive availability may result in toxicity (Humphries et al. 2006;

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Mundus et al. 2012). Little is known about manganese transporters in plants, though the iron (Fe<sup>2+</sup>), zinc (Zn<sup>2+</sup>) and calcium (Ca<sup>2+</sup>) family transporters seem to be the most probable candidates (Socha and Guerinot 2014). In the xylem, Mn is transported as  $Mn^{2+}$  ion or as complex with citrate or malate; in the shoot high levels of Mn may accumulate in the vacuole (Pittman 2005). The physiological basis for Mn toxicity is largely unclear. Differential tolerance of plants to manganese cannot solely be explained by a restricted Mn<sup>2+</sup> uptake and transport to the shoot but additionally by intrinsic strategies that enhance cellular accumulation capacity (Foy et al. 1978). Sequestration into the vacuole, activity of antioxidant enzymes and formation of chelation complexes in the cytosol are some of the strategies proposed to promote toxic metal tolerance (Pittman 2005). Sulfur metabolites play a role in the detoxification of potential toxic metals (Yaday 2010). Complexation of sulfur compounds (e.g., cysteine, phytochelatins, metallothioneins) with toxic metal ions as a mechanism to overcome their toxicity is widely described for different elements and plant species (Ernst et al. 2008; Yadav 2010; Leitenmaier and Küpper 2013). Other sulfur metabolites, such as glutathione, are also crucial for antioxidant protection against reactive oxygen species, of which levels might be induced upon toxic metal stress (Na and Salt 2011). Some species capable of high manganese accumulation, such as *Phytolacca americana*, show a positive relationship between sulfur and manganese (Peng et al. 2008; Yadav 2010). Moreover, it has been observed that toxic metals, e.g., Cu, Zn may directly induce changes in sulfur uptake by affecting the activity of the sulfate transporters and affect the regulation of enzymes involved in S assimilation and activity of sulfate transporters (Nocito et al. 2002; Sun et al. 2007; Schiavon et al. 2008; Shahbaz et al. 2010, 2013; Na and Salt 2011; Stuiver et al. 2014). All changes induced in sulfur status of the plant could be linked a toxic metal-induced change in activity of the sulfate transporters (Yoshimoto et al. 2002; Sun et al. 2007; Stuiver et al. 2014). Brassica species have high sulfur requirements for growth (Ernst 2000) and are generally considered to be susceptible to Mn toxicity (Foy et al. 1978; Humphries et al. 2006; Lee et al. 2011). In the current paper the interaction between Mn and sulfur metabolism was studied in the Brassica rapa.

*Brassica. rapa* var. perviridis (Komatsuna) seeds were germinated in vermiculite and were subsequently transferred to an aerated 25% Hoagland nutrient solution, containing supplemental concentrations of 0, 10, 20, 50 and 100  $\mu$ M MnCl<sub>2</sub> (pH 5.9) in 30 l plastic containers (20 sets of plants per container, three plants per set) in a climate-controlled room for 10 days. Day and night temperatures were 21 and 18 °C (±1 °C), respectively, relative humidity was 70–80%. The photoperiod was 14 h at a photon fluence rate of 300 ± 20 µmol m<sup>-2</sup> s<sup>-1</sup> (400–700 nm) at plant height, supplied by Philips GreenPower LED (deep red/white 120) production modules. After 10 days of Mn<sup>2+</sup> exposure, plants were harvested 3 h after the start of the light period and shoots and roots separated and weighed. Shoot and root biomass production was calculated by subtracting pre-exposure weight from that after Mn<sup>2+</sup> exposure. Shoot/root ratio was calculated from the shoot and root fresh weight after the exposure. For the determination of pigments and anions, plant material was frozen in liquid N<sub>2</sub> immediately after harvest and stored at -80 °C. For analysis of water-soluble non-protein thiols freshly harvested material was used. Chlorophyll a + b content, chlorophyll a fluorescence and the content of sulfate, nitrate, water-soluble non-protein thiol content, free amino acid and the Mn and S mineral nutrient content were determined as described by Shahbaz et al. (2010) and Stuiver et al. (2014). Total RNA was isolated from shoots and roots of *B. rapa* plants as described by Aghajanzadeh et al. (2014). The full length sequences of sulfur transporter genes are found under the following accession numbers: Sulfur transporter 1.1 (Sultr1;1 XM009128953), Sulfur transporter 1.2 (Sultr1;1 XM009108197, XM009108195 and XM009108196). Transcription was determined by quantitative real-time polymerase chain reaction (qRT-PCR; see Reich et al. 2017). Statistical analysis of the results was performed using unpaired Student's t-test. Different letters indicate significant differences at P < 0.01 between different treatments.

Exposure of *B. rapa* to elevated levels of  $Mn^{2+}$  in the nutrient solution resulted in decreased biomass production at  $\geq 20 \ \mu M$  (Table 1). Shoot growth was relatively slightly more affected upon  $Mn^{2+}$  exposure than root growth resulting in a decrease in shoot to root ratio.  $Mn^{2+}$  exposure also resulted in a substantial increase in dry

	Mn <sup>2+</sup> concentr	ation (µM)			
	0	10	20	50	100
Shoot					
Biomass production (g FW)	$0.97 \pm 0.18c$	$0.85 \pm 0.11c$	$0.73 \pm 0.19$ bc	$0.59 \pm 0.14b$	$0.32 \pm 0.06a$
Dry matter content (%)	7.7 ± 0.3a	$7.9 \pm 0.3a$	8.3 ± 0.3ab	$8.9 \pm 0.7$ bc	$9.4 \pm 0.4c$
Chl a + b (mg $g^{-1}$ FW)	$0.69 \pm 0.04b$	$0.61 \pm 0.05b$	$0.54 \pm 0.13$ ab	$0.52 \pm 0.09a$	$0.47 \pm 0.05a$
Chl a/b	$2.4 \pm 0.2a$	$2.6\pm0.1a$	$2.5\pm0.3a$	$2.2 \pm 0.4a$	$2.7\pm0.3a$
Chl a + b/Car	$3.0\pm0.0a$	$2.9\pm0.1$ ab	$2.9\pm0.2$ ab	$2.9\pm0.2$ ab	$2.8\pm0.1b$
F <sub>v</sub> /F <sub>m</sub>	$0.81\pm0.03a$	$0.81 \pm 0.05a$	$0.82 \pm 0.04a$	$0.80 \pm 0.05a$	$0.82\pm0.05a$
Manganese (µmol g <sup>-1</sup> DW)	3.3 ± 0.2a	$9.8\pm0.6b$	$16.4 \pm 1.1c$	$30.5 \pm 0.4 d$	$53.6 \pm 4.7e$
Sulfur ( $\mu$ mol g <sup>-1</sup> DW)	241 ± 21a	277 ± 7a	$265 \pm 7a$	$289 \pm 25ab$	$300 \pm 10b$
Sulfate (µmol g <sup>-1</sup> FW)	$14 \pm 2a$	$13 \pm 1a$	$15 \pm 4a$	$26 \pm 3b$	$34 \pm 4b$
Thiols (µmol g <sup>-1</sup> FW)	$0.56 \pm 0.01a$	$0.57 \pm 0.09a$	$0.56 \pm 0.05a$	$0.59 \pm 0.04a$	$0.60 \pm 0.06a$
Nitrate (µmol g <sup>-1</sup> FW)	84 ± 7a	$105 \pm 1b$	$102 \pm 4b$	$85 \pm 3a$	$85 \pm 0a$
Amino acids ( $\mu$ mol g <sup>-1</sup> FW)	$14 \pm 1a$	$14 \pm 2a$	$15 \pm 1a$	$15 \pm 2a$	$16 \pm 2a$

**Table 1** Impact of  $Mn^{2+}$  exposure on biomass production, dry matter content, chlorophyll afluorescence and contents of pigments, nitrate and sulfate, amino acids and water-soluble non-proteinthiols of *Brassica rapa* 

(continued)

	Mn <sup>2+</sup> concentr	ration (µM)			
	0	10	20	50	100
Root					
Biomass production (g FW)	$0.17 \pm 0.03 c$	$0.18 \pm 0.03 c$	$0.16 \pm 0.05 bc$	$0.12 \pm 0.04b$	$0.07 \pm 0.02a$
Dry matter content (%)	$6.5 \pm 0.3a$	$7.1 \pm 0.4a$	$7.0 \pm 0.5a$	7.9 ± 1.1ab	$8.6 \pm 0.5b$
Manganese $(\mu mol g^{-1} DW)$	49 ± 5a	$162 \pm 6b$	$168 \pm 75b$	$180 \pm 53b$	154 ± 19b
Sulfur (µmol g <sup>-1</sup> DW)	241 ± 21a	277 ± 7a	265 ± 7a	$289 \pm 25ab$	$300 \pm 10b$
Sulfate ( $\mu$ mol g <sup>-1</sup> FW)	$28 \pm 2a$	$23 \pm 2a$	$27 \pm 2a$	$25 \pm 4a$	$22 \pm 2a$
Thiols (µmol g <sup>-1</sup> FW)	$0.42 \pm 0.03a$	$0.41 \pm 0.08a$	$0.44 \pm 0.02a$	$0.45 \pm 0.06a$	$0.53 \pm 0.06a$
Nitrate ( $\mu$ mol g <sup>-1</sup> FW)	$44 \pm 5a$	$39 \pm 2a$	48 ± 3b	$43 \pm 4ab$	$42 \pm 3a$
Amino acids ( $\mu$ mol g <sup>-1</sup> FW)	$15 \pm 1a$	$15 \pm 2ab$	$18 \pm 2ab$	$18 \pm 1b$	$17 \pm 2ab$
Plant					
Shoot/root ratio	$5.6 \pm 0.4a$	$4.7 \pm 0.3b$	$4.5 \pm 0.7b$	$4.9 \pm 0.9b$	$4.4 \pm 0.8b$

Table 1 (continued)

Ten day-old seedlings were grown on a 25% Hoagland nutrient solution containing supplemental concentrations of 0, 10, 20, 50 and 100  $\mu$ M MnCl<sub>2</sub>. The initial shoot and root fresh weights were 0.100  $\pm$  0.01 g and 0.040  $\pm$  0.01 g, respectively. Data on biomass production, dry matter, pigment and amino acid content represent the mean of two independent experiments, with a total of 12, 6, 6, and 6 measurements with 3 plants in each, respectively ( $\pm$  SD). Data on chlorophyll a fluorescence represents the mean of 10 measurements ( $\pm$  SD). Data on nitrate, sulfate and water-soluble non-protein thiol content represent the mean of 3 measurements with 3 plants in each ( $\pm$  SD). Different letters indicate significant differences at P < 0.01 between different treatments

matter content of both root and shoot at 100  $\mu$ M. Mn<sup>2+</sup> exposure resulted in interveinal chlorosis (especially of the younger leaves) at  $\geq$ 50  $\mu$ M, however the chlorophyll a/b and the chlorophyll/carotenoid ratios were hardly affected (Table 1). Chlorophyll a fluorescence (measured as the quantum yield of the photosynthetic system II photochemistry,  $F_v/F_m$  ratio) was not affected upon Mn<sup>2+</sup> exposure, even at toxic levels (Table 1). The Mn content in the shoot increased with the Mn<sup>2+</sup> concentration in the nutrient solution (Table 1). However, in the root there was a strong increase in the Mn content at 10  $\mu$ M Mn<sup>2+</sup>, which hardly increased further at higher Mn<sup>2+</sup> concentrations. This increase was markedly higher in the shoot than in the root (Table 1). Evidently, Mn became toxic and reduced the biomass production when the content in the shoot was  $\geq$ 16  $\mu$ mol g<sup>-1</sup> dry weight and exceeded four-fold of that of the control. Exposure of plants to toxic Mn<sup>2+</sup> levels hardly affected the content of other essential mineral nutrients in both root and shoot (data not presented); there was only a 30% and 19% decrease at 100  $\mu$ M in the K content of the root and shoot, respectively. Moreover, the Zn content increased



Fig. 1 Impact of  $Mn^{2+}$  exposure on the transcript levels of Sultr1;1 and Sultr1;2 in the root of *B. rapa*. For experimental details, see legends of Table 1. Relative gene expression of these genes was determined by qRT-PCR and the mRNA levels were compared to actin. Data on relative expression represent the mean of 3 measurements with 3 plants in each (±SD). *Different letters* indicate significant differences at P < 0.01% between different treatments

two-fold in both root and shoot; a similar increase in the Cu content was observed in the root.  $Mn^{2+}$  hardly affected the nitrate and free amino acid content of the plants (Table 1). The total sulfur and sulfur metabolite contents of *B. rapa* were only slightly affected at toxic  $Mn^{2+}$  levels. There was only a 1.25-fold increase in sulfur content in the shoot at 100  $\mu$ M  $Mn^{2+}$ , which could be attributed to an increase of the sulfate content.  $Mn^{2+}$  exposure did not affect the total sulfur and sulfate content of the root and the water-soluble non-protein thiols in both root and shoot (Table 1). The sulfate transporters Sultr1;1 and Sultr1;2 are involved in the primary uptake of sulfate by the roots, though the transcript level of Sultr1;2 in the roots of *B. rapa* was 30-fold higher than that of Sultr1;1 (Fig. 1). The expression of these sulfate transporters were not affected upon  $Mn^{2+}$  exposure.

Similar to other essential potentially toxic metals, viz. Cu and Zn, exposure of B. rapa to elevated  $Mn^{2+}$  levels in the nutrient solution resulted in a strong accumulation of the metal in both root and shoot, resulting in decreased plant biomass production and chlororis of the shoot. B. rapa was much less susceptible to Mn than *B. pekinenis* to Cu and Zn toxicity:  $Mn^{2+}$  became toxic at  $\geq 20 \ \mu M$ , whereas  $Cu^{2+}$  and  $Zn^{2+}$  already affected plant biomass production at >2  $\mu$ M (Shahbaz et al. 2010, 2013, 2014; Stuiver et al. 2014). The decrease in biomass production due to toxic metal exposure was accompanied or even preceded by a decrease in pigment content (Foy et al. 1978, Shahbaz et al. 2010, 2013; Stuiver et al. 2014), although chlorophyll a fluorescence upon  $Cu^{2+}$  (Shahbaz et al. 2010) and Mn<sup>2+</sup> exposure remained unaffected, which indicated that development rather than chloroplast functioning was negatively affected. High Mn levels also reduced the pigment content in tobacco (Clairmont et al. 1986), mungbean (Sinha et al. 2002), Chinese cabbage (Lee et al. 2011), spearmint (Asrar et al. 2005), tomato (Shenker et al. 2004) by affecting the chlorophyll, carotenoid and flavonoid biosynthesis (Clairmont et al. 1986; González and López 2013).

Exposure of *B. pekinensis* to elevated levels of  $Cu^{2+}$  and  $Zn^{2+}$  in the nutrient solution substantially affected the uptake, distribution and metabolism of sulfur (Shahbaz et al. 2010, 2013, 2014; Stuiver et al. 2014).  $Cu^{2+}$  and  $Zn^{2+}$  exposure resulted in an up-regulation of the activity of sulfate transporters and expression of the Group 1 sulfate transporters, viz. Sultr1;2, which are involved in the uptake of sulfate by the root in Brassica species (Shahbaz et al. 2010, 2013, 2014; Stuiver et al. 2014). The up-regulation of the sulfate transporters was most likely not due to a higher plant sulfur requirement upon  $Cu^{2+}$  and  $Zn^{2+}$  exposure, since it was accompanied by a substantial increase in the sulfate content of the shoot (Shahbaz et al. 2010, 2013, 2014; Stuiver et al. 2014). It was presumed that the up-regulation of the sulfate transporters was the consequence of a direct interference of these metal ions with the signal transduction pathway resulting in a disturbed regulation of the transporters (Shabaz et al. 2014; Stuiver et al. 2014). However, exposure of B. rapa to elevated  $Mn^{2+}$  levels did not affect the transcript levels of the sulfate transporters Sultr1;1 and Sultr1;2 in the root, despite a slight increase in the sulfate content in the shoot. The impact of elevated Mn<sup>2+</sup> levels was also measured in B. juncea and all results on growth, pigment content and metabolite content were quite similar to that in B. rapa, with the exception that Mn toxicity was not accompanied with higher sulfur and sulfate contents in the shoot, and that it was even slightly decreased (data not presented). Apparently, in contrast to Cu<sup>2+</sup> and Zn  $^{2+}$ , Mn<sup>2+</sup> exposure did not interfere with the signaling of the regulation of the sulfate transporters. In general under normal conditions glutathione is the major watersoluble non-protein thiol compound present in plant tissues. Exposure of B. pekinensis to  $Cu^+$  (Shahbaz et al. 2010, 2013, 2014) and  $Zn^{2+}$  (Stuiver et al. 2014) resulted a strong increase in the water-soluble non-protein thiol content of the root and to a lesser extent in the shoot. This increase could partially be ascribed to an increase in phytochelatins (Shahbaz et al. 2010) and cysteine (Stuiver et al. 2014). An increase in water-soluble non-protein thiols (e.g., cysteine and glutathione) is expected as a defense mechanism against heavy metal toxicity (Leitenmaier and Küpper 2013). Moreover, cysteine and glutathione are the precursors for the synthesis of phytochelatins, which may complex with metals and increase toxic metal tolerance (Ernst et al. 2008). However, Mn<sup>2+</sup> exposure did not affect the water-soluble non-protein content of both root and shoot of *B*. *rapa* (and *B*. *juncea*. data not presented). Apparently, an exposure of B. rapa to elevated and toxic  $Mn^{2+}$ levels did not trigger the synthesis of thiols (e.g. cysteine, glutathione and/or phytochelatins).

In conclusion, in contrast to  $Cu^{2+}$  and  $Zn^{2+}$ , elevated and toxic  $Mn^{2+}$  levels in the root environment hardly affected the uptake and metabolism of sulfate in *Brassica*.

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# Localization of Sulfate Uptake and pH Changes at Sulfur-Deprived Roots of Intact *Brassica pekinensis* Seedlings by Using H<sup>+</sup>-Selective Microelectrodes

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Abstract Proton-selective microelectrodes were used to determine sulfate uptake by roots of intact plant seedlings. The response of  $H^+$  fluxes to sulfate addition showed to be a good proxy for sulfate uptake by the sulfate/ $H^+$  co-transport system.  $H^+$  influx and increase in root surface pH was much higher in sulfate-deprived seedlings than in seedlings grown with sufficient sulfate. The opposite was true for the response of  $H^+$  fluxes to nitrate addition. By using this method sulfate uptake could be mapped along the root axis, which revealed higher uptake rates in mature regions. Sulfate deprived roots showed a lower root surface pH, which correlated strongly with the response to sulfate addition. A possible contribution of this component to a higher sulfate uptake capacity under sulfur deficiency was further tested by using the fungal toxin fusicoccin, which permanently activates the plasma membrane  $H^+$ -pumping ATPase. Application of fusicoccin lowered the pH of sufficient roots to the level of deficient roots, indicating a more activated state of the ATPase under sulfur deficiency rather than a higher abundance.

Sulfate serves as the main source of sulfur to plants and is taken up by plant roots via a H<sup>+</sup>-coupled symporter in the plasma membrane (Hawkesford et al. 1993; Smith et al. 1995; Hawkesford et al. 2003) and. The regulation of sulfate transport is well studied on the molecular, biochemical and whole plant level (Saito 2000; Buchner et al. 2004; Hopkins et al. 2005; Koralewska et al. 2008; Rouached et al. 2008; Reich et al. 2015) but knowledge on the distribution of uptake along the root axis is still scarce. The few studies on localization of sulfate fluxes along roots are rather indirect via localization of the sulfate transporter genes using *in situ* hybridization or GFP fusion (Takahashi et al. 1997; Yoshimoto et al. 2002). Electrophysiological studies applying ion selective microelectrodes offer the possibility to measure ion uptake in real-time and with high temporal and spatial resolution

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(Plassard et al. 1999; Newman 2001). Localization along the root axis of different plant species have been conducted for fluxes of e.g. nitrate, ammonium, potassium and H<sup>+</sup> (Henriksen et al. 1990; Taylor and Bloom 1998; Garnett et al. 2001; Rubinigg et al. 2002; Chen et al. 2005; Staal et al. 2011) and gave important insights into the acquisition of these nutrients in relation to nutrient supply and developmental processes. Application of ion-selective microelectrodes relies on highly sensitive and selective ion exchangers. To date, no ion exchanger with the required selectivity is available for sulfate. Instead H<sup>+</sup>-selective micro-electrodes were used in the present study to determine and localize sulfate uptake activity along roots of intact seedlings of Chinese cabbage, a plant that is known for its high sulfur need and sulfate content (Koralewska et al. 2008). Because sulfate transporters are  $H^+$ /sulfate co-transporters (Hawkesford et al. 1993) sulfate addition should cause an immediate response of H<sup>+</sup> fluxes proportional to the coupled influx of sulfate. To test this assumption we compared fluxes at roots of plants that had been grown with sufficient sulfate in the medium to fluxes of sulfate-deprived plants. As in sulfate-deprived plants the sulfate uptake system is strongly upregulated (Buchner et al. 2004), larger H<sup>+</sup> fluxes in response to sulfate addition were expected. As an additional verification of the method Mg(NO<sub>3</sub>)<sub>2</sub> was added to the roots instead of MgSO<sub>4</sub>. Nitrate uptake is known to be depressed under sulfate deficiency and the response of  $H^+$  fluxes in response to  $Mg(NO_3)_2$  is expected to be higher in sulfate sufficient than in sulfate deficient roots. After these verifications the method was used to localize sulfate uptake along the root axis of sulfatesufficient and sulfate-deprived roots. At the same time, differences in root surface pH were assessed.

Seeds of Chinese cabbage (Brassica pekinensis Rupr., cv. Michico) were germinated on filter paper, moistened with tap water and placed in the dark at 21 °C. After 4 days, seedlings were placed in a climate controlled room with a day/night temperature 21 °C/18 °C (±1 °C), a photoperiod of 14 h and a photon flux of 340 umol m<sup>-2</sup> s<sup>-1</sup>. Seedlings were positioned such that the roots could grow in a layer of water. The seedlings were kept under these conditions for another 3 days, before being transferred from the filter paper to 13 l containers with aerated 25% Hoagland nutrient solution (pH 5.9; for composition see Reich et al. 2016) containing either 0.5 mM (control) or 0 mM MgSO<sub>4</sub>.7H<sub>2</sub>O (sulfate-deprived, -S). The pH was controlled and adapted every 2–3 days to pH 5.9 with diluted HCl. As root excision will affect nutrient absorption (Bloom and Caldwell 1988; Shabala et al. 2009) intact seedlings were used in this study. Prior to performing an ion flux experiment, a plant was taken from the climate room and placed in a 25 ml petri dish and the primary root was mounted carefully in such a way that it was easily accessible with the ion selective electrode. The shoot was stabilized at the edge of the petri dish. Thereafter the root was covered with a low salt measuring solution (MS; containing 200 µM MgCl<sub>2</sub>, 100 µM KCl and 100 µM CaCl<sub>2</sub>). During the entire measurement the shoot of the plant was enclosed in a small cabinet and supplied with LED light with a photon flux of 250  $\mu$ mol m<sup>-2</sup> s<sup>-1</sup>, while the roots were not illuminated (Fig. 1). The MS was continuously exchanged by a perfusion system with a perfusion rate of ca.  $3 \text{ ml min}^{-1}$ . The tip of the electrode was then moved to the root surface by using a threedimensional micromanipulator mounted



**Fig. 1** (a) The measurement took place in a Faraday's cage in which an inverted microscope was installed to position the electrode at the root surface. The amplifiers, the pump for the perfusion system (*left*) and the PC for data read-out (*right*) were located outside of the cage. (b) The plant shoot was enclosed in the lower part of a small chamber and the root was immobilized in the petri dish which was filled with the measuring solution. The electrode was mounted on a 3D-micromanipulator. Inlet and outlet of the perfusion system were placed at the edges of the dish. (c) The plant shoot was fully enclosed and illuminated during the measurement (Visualization by Markus Reich)

on an inverted microscope. Steady fluxes were recorded for at least 10 min before the MS was exchanged with the treatment solution. The total incubation time before a measurement was between 1 and 2 h, depending on the time it took to record a stable flux. Net fluxes of  $H^+$  were measured using  $H^+$ -selective electrodes with the MIFE technique (Microelectrode Ion Flux Estimation; Shabala et al. 1997; Vreeburg et al. 2005; Lanfermeijer et al. 2008).

Microelectrodes were pulled from borosilicate glass capillaries (GC150-10; Harvard Apparatus) and silanized with tributylchlorosilane (Fluka 90974). The H +-selective electrodes were back filled with 15 mM of NaCl and 40 mM of KH<sub>2</sub>PO<sub>4</sub> and front filled with Hydrogen Ionophore II (Cocktail A; Fluka 95297). Only electrodes with a response of >53 mV per pH unit (pH range 5.1–7.8, r<sup>2</sup> > 0.998) were used for measurements. The reference electrode, filled with 300 mM KCl was placed in a separate compartment electrically connected with the measuring chamber via a salt bridge consisting of 300 mM (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> in 2% (w/v) agar. Prior to flux and pH recording the electrode was brought carefully to a distance of 10 µm from the root surface. To ensure a maximum response of H<sup>+</sup>fluxes to sulfate and a minimum response to its accompanying cation sulfate was supplied in the form of MgSO<sub>4</sub>, as the uptake  $Mg^{2+}$  is known to be slower than any of the other cations, especially in the presence of  $Ca^{2+}$  and  $K^+$  (Moore et al. 1961; Schimansky, 1981). For time series recordings, the microelectrode was positioned at the beginning of the differentiated zone, where the first root hairs started to appear (usually 3–5 mm from the root tip). Care was taken not to place the tip of the electrode above an initiating root hair. During the incubation phase before timeseries measurement, the position of the electrode and its distance to the root surface was surveyed and corrected if necessary.

For the time series recording sulfate-sufficient (control) seedlings and seedlings which were sulfate deprived for 1-5 days (-S) were taken. To localize the H<sup>+</sup>-flux and pH changes along the root axis, a root profile was recorded from the root tip until 7 mm towards the base in 0.5 mm increments between 0-5 mm and 1 mm increments between 5 and 7 mm. A control profile was recorded after 1-2 h of incubation in the sulfate-free measuring solution. Subsequently the solution was exchanged for a solution to which  $100 \ \mu M MgSO_4$  was added and after 30 min a second profile was recorded. At each position fluxes and pH were recorded for 3-4 min and averaged. The profiling included the meristematic, the elongation and the beginning of the differentiated zone. For the root profiles sulfate-sufficient seedlings as well as seedlings that had been sulfate deprived for 3 days were taken. For the fusicoccin (FC) experiment excised roots were used instead of seedlings, first because FC is too costly to be applied in the large volume of the perfusion system and second because the response of the H<sup>+</sup>-ATPase to FC is expected to be immediate. Roots were placed in measuring chambers and covered with 1 ml MS. After an incubation time of ca. 1 h 1 µl fusicoccin (Sigma, F0537) was added from a 10 mM stock solution (10  $\mu$ M final concentration) and the solution was vigorously mixed.

A clear response of surface pH, i.e. the concentration of  $H^+s$  at the root surface, could be observed if 100  $\mu$ M MgSO<sub>4</sub> was added to the roots via the perfusion



Fig. 2 The change in response of H<sup>+</sup> fluxes and surface pH at roots of intact seedlings to 100  $\mu$ M sulfate addition (indicated by the *arrow*) over time. Plants were either grown at sulfate-sufficient (control) or sulfate-deprived conditions (–S). Positive values refer to an influx. Mean values  $\pm$  SE are shown; control, n = 12; –S, n = 15



**Fig. 3** Tentative results of the change in response to  $H^+$  fluxes and surface pH to 50  $\mu$ M nitrate addition (indicated by the *arrow*) over time. Plants were either grown at sulfate-sufficient (control) or sulfate-deprived conditions (–S). Positive values refer to an influx (exemplary measurement)

system (Fig. 2). The response was much more pronounced with sulfate-deprived roots with pH increasing about 0.2 units, which corresponds to a decrease in [H<sup>+</sup>] of approximately 0.5  $\mu$ M. Alkalization was also observed if 50  $\mu$ M Mg(NO<sub>3</sub>) was added to the measuring solution, but the sulfur status of the plants had an opposite effect on the NO<sub>3</sub><sup>-</sup>-induced pH response: sulfate-sufficient plants showed a greater response (ca. 0.3 pH units) than sulfate-deprived plants (ca. 0.1 pH units; Fig. 3). While H<sup>+</sup> fluxes were highly variable, the surface pH, as it is the result of net H<sup>+</sup> fluxes over a somewhat longer time period, was more stable. A response to MgSO<sub>4</sub> after 30 min of addition was present at roots of sulfate-deprived seedlings between 3.5 and 7 mm away from the root tip but absent in the first 3 mm which covered the meristematic and the beginning of the elongation zone (Fig. 4). No MgSO<sub>4</sub>-induced changes of sulfate-deprived plants compared to sufficient plants was observed during the measurements prior to any compound addition. There was a linear relationship between the concentration of H<sup>+</sup> before and the change in



**Fig. 4** Root profiles of H<sup>+</sup> fluxes and surface pH in response to sulfate addition at intact roots of sulfate-sufficient (control) and 3 days sulfate-deprived (-S) seedlings. The changes of fluxes and pH in response to sulfate addition at the same position are shown. Positive values refer to an influx. Data represent the mean ( $\pm$  SE, n = 3)



**Fig. 5** Roots of sulfate-deprived seedlings (*open symbols*) showed a lower surface pH than roots of sulfate-sufficient seedlings (*closed symbols*) and the concentration of H<sup>+</sup> correlates with the peak response of [H<sup>+</sup>] to addition of 100  $\mu$ M MgSO<sub>4</sub> The elevation of the two linear regressions significantly differs (p-value shown in graph). The differences in root surface pH were also measured in the growth medium (25% Hoagland). Data presented as boxes with a 5–95 percentile and whiskers (unpaired Student t-test; control, n = 6; -S, n = 5)

concentration of H<sup>+</sup> in response to sulfate addition (Fig. 5). Additional experiments revealed that the difference in surface pH was also apparent if measurements were performed in the growth medium (25% Hoagland) and therefore can not be considered to be an artifact of the incubation in the measuring solution (Fig. 5). Adding fusicoccin (FC) to the roots resulted in an immediate H<sup>+</sup> efflux and a consecutive decrease in pH. Thirty minutes after addition, the pH at roots of sulfate-sufficient seedlings was approximately at the same level as that of the sulfate-deprived ones (Fig. 6). The response of H<sup>+</sup>-flux and surface pH to FC in a typical experiment is presented in Fig. 7.

The results in this study are in agreement with the accepted model of sulfate uptake being depressed under sulfate-sufficient conditions and de-repressed if plants get sulfate-deprived. H<sup>+</sup>-selective microelectrodes appear to be a convenient



Fig. 6 Average of root surface pH and  $[H^+]$  of sulfate-sufficient (control) and sulfate-deprived (-S) seedlings 10 min before (*closed symbols*) and peak values after (*open symbols*) the addition of 10  $\mu$ M FC. Data represent the mean ( $\pm$  SE; \*\* = p < 0.01; One-way-ANOVA; control, n = 3; -S, n = 4)



Fig. 7 Over-time response of  $H^+$  flux and root surface pH following FC (concentration) addition (indicated by the *arrow*). Negative values correspond to an efflux

tool to determine and localize the uptake of sulfate at the root surface. Two main lines of experimental evidence support this. First, the de-repression of the sulfate uptake system under sulfate deprivation is clearly reflected by the pH changes at the root surface in response to sulfate addition (Fig. 2). The relatively weak alkalization at sulfate-sufficient roots disappeared after ca. 10 min while the strong alkalization at sulfate-deprived roots reached a stable plateau. With sulfate being the only ion newly added to the solution during the measurement, the response in H<sup>+</sup>-flux seems to be a good proxy for sulfate uptake. Second, addition of nitrate in preliminary experiments resulted in a converse response of sufficient and deprived roots, if compared to sulfate addition. This was expected, as sulfate-deprived plants usually down-regulate nitrate uptake (Clarkson et al. 1989; Prosser et al. 2001). The use of H<sup>+</sup>-selective microelectrodes appears therefore a reliable method to probe plant roots for differences in sulfate uptake. In a case study, we applied this method to map sulfate uptake along the first millimeters of the roots of seedlings. We found that 30 min after sulfate addition no active sulfate uptake was apparent in both the meristematic zone and the beginning of the elongation zone (0–3 mm) (Fig. 4). Up to date, the only localization studies concerning sulfate transport at roots of intact plants aimed on localizing transcripts of the genes encoding for the sulfate transporters (e.g. Takahashi et al. 1997; Yoshimoto et al. 2002). In these studies, *in situ* hybridization analyses showed gene expression in the root cap, the epidermal layer, in root hairs and along the root cylinder. Abundance of mRNA transcripts is, however, not necessarily reflecting the final localization and activity of transport proteins. Many post-translational modifications determine the activity of membrane transport-proteins. The results of our analysis of the actual fluxes at intact roots suggest that active sulfate uptake takes place in the more differentiated root zones rather than in the meristematic and elongation zone.

Figure 5 shows that sulfate-deprived plants responded stronger to sulfate addition than sulfate-sufficient plants, indicated by the significant difference in elevation (p < 0.001). This is most likely representing the higher abundance of sulfate transporters. Additionally, a strong correlation was found between the H<sup>+</sup> concentration at the root surface prior and after the addition of sulfate (Fig. 5). This dependency of sulfate uptake on the H<sup>+</sup> gradient at the root plasma membrane provides support for the suggested  $SO_4^{2^-}/H^+$ -symport stoichiometry of sulfate transporters (Hawkesford et al. 1993). Additionally, it raises the question whether a lower external pH and a consequently steeper H<sup>+</sup>-gradient could have a physiological function to increase the uptake capacity under sulfate deprivation (Fig. 8). Indeed, a lower root surface pH was also found at seedlings grown in Hoagland solution without sulfate (Fig. 5). While the role of sulfate transporters in sulfate uptake has been studied intensively, the involvement and potential co-regulation of the plasma membrane ATPase has not been investigated. Active rhizosphere acidification has been well described and characterized in plants deprived of phosphate and iron (De Vos et al. 1986; Neumann and Römheld 1999). The uptake of these nutrients is mostly limited by their mobility in the soil solution, which is increased at low pH. Sulfate however, if present, is highly mobile over a wide pH range. A lower root surface pH under sulfate deprivation could also be of passive nature, caused by an altered cation-anion uptake balance (Haynes, 1990): no sulfate and probably less nitrate is taken up and, consequently, also less protons. Our experiments rather suggest an involvement of the activity of the plasma membrane H<sup>+</sup>-ATPase: incubation of roots with fusicoccin, a permanent activator of the plasma membrane H<sup>+</sup>-ATPase (Johansson et al. 1993), led to a decrease of the surface pH at sulfate-sufficient roots to the same level of sulfate-deprived roots (Fig. 6). Thus it appears that the root plasma membrane H<sup>+</sup>-ATPase is not more abundant under sulfate deprivation but in a more active state. Sulfate uptake has been shown to be in a repressed state if sulfate or other sulfur sources are sufficiently abundant, which enables a quick de-repression if sulfate gets limiting (Clarkson and Saker 1989; Herschbach and Rennenberg 1994). The exact cellular and molecular signals involved, still need to be elucidated and some authors



**Fig. 8** Lower rhizosphere pH as an adaptive response to sulfate deprivation (-S)? Isoforms of the 14-3-3 protein were shown to stimulate the H<sup>+</sup>-ATPase and to interact with the sulfate transporter (in Brassica Sultr1;2) in a yet unknown way (Shin et al. 2011) and might act as a coordinator to increase sulfate uptake under sulfur deprivation. Fusicoccin (FC) permanently stabilizes the binding of 14-3-3 to the H<sup>+</sup>-ATPase and can therefore be used to achieve a maximum activation

actually found discrepancies between the sulfate uptake capacity and the expression level of the genes encoding for sulfate transporters (Koralewska et al. 2009). Deeper understanding of the regulation on the protein and membrane level is likely to explain such observed discrepancies. The plasma membrane H<sup>+</sup>-ATPase as driver of H<sup>+</sup>-coupled sulfate uptake (Fig. 8) might be involved in this regulation and increases sulfate uptake in addition to a higher transcript level of the sulfate transporter genes. Because the plasma membrane  $H^+$ -ATPase is building up the  $H^+$ gradient that is utilized also by other transport systems, coordination with the uptake of other nutrients would be needed. A candidate player for such a general regulatory function is the 14-3-3 protein. Isoforms of this protein were shown to interact not only with the plasma membrane H<sup>+</sup>-ATPase (Jahn et al. 1997; Bunney et al. 2002), but also with numerous proteins of the nitrogen, phosphorous and sulfur metabolic pathways, including the sulfate transporter Sultr1;2, which is responsible for primary sulfate uptake at the root plasma membrane (Shin et al. 2011). Being "spiders in a web of phosphorylation" (De Boer et al. 2013), 14-3-3 proteins could be a central coordinator of the complex network of different proteins responsible for ion homeostasis and the uptake of nutrients.
Further clarification is needed whether a lower rhizosphere pH under sulfate deprivation is an active response or rather a passive consequence of reduced anion uptake. However, active or passive, a lower pH has the potential to increase sulfate uptake (Fig. 8). Future studies should aim to further integrate cellular and electrophysiological techniques to elucidate the coordination of H<sup>+</sup>-ATPase activity and nutrient status. Additionally, methods to assess the nutrient status or gene expression level of homogenized plant tissue reached their limit in explaining the processes under nutrient deficiency. Roots are composed of highly diversified tissues, quickly changing their morphology to adapt to changes in nutrient availability with newly formed roots having a different physiology than older ones. Ion-selective microelectrodes are perfectly suited to localize ion uptake on small scales and should be used more often, also in combination with molecular techniques. As long as there are no highly selective sulfate electrodes available, H<sup>+</sup>-electrodes are a reliable alternative to study sulfate uptake with the MIFE technique. Localization of sulfate uptake at different kinds of tissues or the characterization of specific mutants are possible applications. Sulfate-selective microelectrodes will make it possible to investigate the post-translational regulation of sulfate uptake. The putative control by 14-3-3 proteins would be one example.

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### **Evidence for Regulation of the Iron Uptake Pathway by Sulfate Supply in S-Deprived Maize Plants**

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**Abstract** Maize plants follow Strategy II to take up Fe from the rhizosphere. In roots, nicotianamine is used as precursor for the production of deoxymugineic acid (DMA). DMA is secreted to the rhizosphere where it chelates Fe(III) and the Fe(III)-DMA complex is then taken up by the root. In this study, non-mycorrhizal and mycorrhizal maize plants were grown in pots with sterile river sand containing FePO<sub>4</sub>. After a 60-day period of sulfur deficiency, sulfur was provided to the plants in the form of sulfate. The expression profiles of ZmNAS1, ZmDMAS1, ZmTOM1 and ZmYS1, key genes of the Fe uptake pathway in maize, were monitored 24 and 48 h after sulfate addition in both mycorrhizal and non-mycorrhizal roots in order to estimate the impact of sulfate availability on the Fe uptake pathway of two distinct plant systems. Significant differential responses have been recorded between mycorrhizal and non-mycorrhizal plants, both before as well as 48 h after sulfate addition. However, sulfur repletion resulted in a significant downregulation of all studied genes in all plants 24 h after sulfate addition. This finding suggests a strong correlation between the transcriptional regulation of the Fe uptake pathway genes and sulfate availability no matter if the plants are in mycorrhizal association or not. Sulfate is probably a key component of the signal transduction pathway that regulates the expression of the Fe uptake pathway genes in maize plants.

Iron (Fe) is an essential nutrient for all living organisms. In plants, Fe plays a key role in electron transfer in both photosynthetic and respiratory reactions in chloroplasts and mitochondria. In soils, Fe is sparingly soluble under aerobic conditions at

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high soil pH and exists mainly in its oxidized form, as part of insoluble ferric compounds. In addition to the solubility challenge, the chemical properties of iron require cells to place limitations on its accumulation. Fe(II) and Fe(III) act catalytically to generate hydroxyl radicals that can damage cellular constituents such as DNA and lipids (Halliwell and Gutteridge 1992). Higher plants employ two strategies for Fe uptake from their rhizosphere. Dicots and non-grass monocots take up Fe using ferric-chelate reductases to reduce Fe(III) to Fe(II), which is subsequently taken up by Fe transporters (Connolly et al. 2003). This type of Fe uptake is known as Strategy I pathway for Fe acquisition. Alternatively, graminaceous plants take up iron from their rhizospheres using the Strategy II pathway for Fe acquisition. In this pathway, S-adenosyl-methionine (SAM), a sulfur-containing compound, is used for the production of nicotianamine (NA) by the enzyme family of nicotianamine synthases (NAS). NA is then used as precursor for the biosynthesis of deoxymugineic acid (DMA), the first phytosiderophore produced in graminaceous plants and the only one produced in maize. The biosynthesis is conducted in a two-step process by the enzymes of nicotianamine aminotransferase (NAAT) and DMA synthase (DMAS). In maize, the efflux transporter TOM is used for the excretion of the produced DMA to the rhizosphere, where it chelates Fe(III). The complex DMA-Fe(III) is, finally, taken up by the root via a yellow stripe transporter (YS; Nozoye et al. 2013).

Over recent years, sulfur deficiency has become widespread in the world, mainly because of the strong decrease in the inputs of S from atmospheric deposition and the use of S-free fertilizer products such as triple phosphate and urea. Although it is often reported that cereals have a relatively low requirement for S, they can be adversely affected by S deficiency in the field as a result of its effects on growth, grain yield, and quality (Zhao et al. 1999). In plants, cysteine represents the main source of reduced S and is fundamental for the biosynthesis of methionine, Fe–S clusters and glutathione. Thus, there is a strong relationship between S and Fe homeostasis and the role of S in the Fe uptake pathway of graminaceous plants is crucial. It has been demonstrated that sulfur deficiency causes Fe deprivation responses to graminaceous plants which can be reversed, however, by sulfur repletion (Astolfi et al. 2003, 2006, 2010; Bouranis et al. 2003).

In this study, non-mycorrhizal and mycorrhizal maize plants were grown in pots with sterile river sand, in a long-term experiment (Fig. 1). For the mycorrhizal treatment, inoculum of the arbuscular mycorrhizal fungus *Rhizophagus irregularis* was used. Plants were watered with a nutrient solution deprived of Fe and S and containing a minimum P concentration so as to enhance the establishment of the arbuscular mycorrhiza (AM) symbiosis. Iron was provided throughout the experiment in the sparingly soluble form of FePO<sub>4</sub>. After a 60-day period of sulfur deficiency, sulfur was provided to the plants in the form of sulfate, as described previously (Chorianopoulou et al. 2015). In order to estimate the impact of sulfate availability on the Fe uptake pathway, the expression profiles of *ZmNAS1*, *ZmDMAS1*, *ZmTOM1* and *ZmYS1*, key genes of the Fe acquisition pathway in maize (Fig. 2a), were monitored 24 and 48 h after sulfate addition in mycorrhizal and non-mycorrhizal roots. The gene expression analysis was conducted by means



Fig. 1 Depiction of the experimental design of this study. Mycorrhizal and non-mycorrhizal plants were grown under S deficient conditions until day 60 from sowing; all plants were watered with a nutrient solution deprived of Fe and S and containing 10  $\mu$ M of P (P<sub>min</sub>) to trigger the establishment of the mycorrhizal symbiosis. On day 60, sulfate was provided to the plants. Fe was available throughout the experiment in the form of sparingly soluble FePO<sub>4</sub>. Samplings took place on day 60 (before S repletion) as well as on days 61 and 62, 24 and 48 h after sulfate addition, respectively

of Real-Time RT-PCR. The oligonucleotide primers used for RT-qPCR are listed in Table 1. The efficiency of each Real-Time RT-PCR reaction was calculated using the LinRegPCR software (Ruijter et al. 2009). The mathematical formula of Pfaffl (2001) was used for the calculations of the relative expression ratios and ubiquitin was used as reference gene. The results of each sampling were expressed with the use of the previous sampling as control in order to determine the shift in the expression after sulfate supply.

It has been recently shown that S deprived mycorrhizal maize roots suppressed the expression of two key genes of the Fe uptake pathway, ZmNAS1 and ZmYS1, in contrast to non-mycorrhizal ones which enhanced it. In addition, AM symbiosis prevented Fe deprivation responses in the S deprived maize plants suggesting that iron was possibly provided directly to the mycorrhizal plants through the fungal network (Chorianopoulou et al. 2015). Differential response of the four key genes of the Fe uptake pathway in maize was observed between mycorrhizal and non-mycorrhizal roots 48 h after sulfate addition (Fig. 2b, Table 2). Non-mycorrhizal roots upregulated all the respective genes while mycorrhizal roots downregulated only the transporters *ZmTOM1* and *ZmYS1* whilst the relative expression ratios of *ZmNAS1* and *ZmDMAS1* did not alter significantly. These findings revealed that the two systems, mycorrhizal and non-mycorrhizal, respond differently as far as the Fe acquisition pathway is concerned.

Interestingly, despite the observed differential responses between mycorrhizal and non-mycorrhizal plants, a common response of all plants was revealed 24 h after sulfate addition, i.e. an identical decreasing pattern in the expressions of all genes of the Fe uptake pathway. This expression pattern was irrelevant of the symbiotic phenomenon and the short-term common response was related to the sulfate supply itself (Fig. 2b, Table 2).

Thus, our observations suggest a strong relationship between resupply of sulfate and the transcriptional regulation of the Fe uptake pathway genes in S-deprived maize plants, regardless of whether these plants are in mycorrhizal association, or not. This evokes the assumption that sulfate itself is a negative regulator of the Strategy II Fe acquisition pathway and the reduced S compounds (i.e. cysteine,



Fig. 2 (a) Schematic illustration of the Strategy II Fe uptake pathway in maize roots. (b) Expression shift patterns of ZmNASI, ZmDMASI, ZmTOMI and ZmYSI, 24 and 48 h after sulfate addition in mycorrhizal (M) and non-mycorrhizal (NM) maize roots. The expression shift values were calculated, each time, relatively to the previous sampling, revealing the pattern of the expression change the first 24 and 48 h after S supply. *Arrows pointing up* or *down* represent increase or decrease, respectively, in the expressions of the genes

methionine) are catalytically important to trigger the production and secretion of DMA.

Given the fact that most of the metabolically active Fe is bound to S in Fe–S clusters, the interaction between Fe and S homeostasis is of particular importance (Forieri et al. 2013). The biosynthesis of Fe–S clusters requires the supply of reduced S (in form of cysteine) and chelated Fe in a defined stoichiometric ratio, strongly suggesting the development of cross-regulatory mechanisms between the metabolisms of the two nutrients. In this line, the provision of the substrates must be tightly regulated in order to meet the plant's changing demands for Fe–S clusters and to avoid potentially toxic free Fe and sulfide.

In conclusion, our work underlines the importance of sulfate as signaling molecule that regulates the Strategy II Fe acquisition pathway; such a role of sulfate has been previously proposed by Forieri et al. (2013) for Fe homeostasis. In addition, we highlight the significance of sulfate reduction for the enhancement of Fe uptake and provide indications that when sulfate is abundant, the Fe uptake pathway is suppressed.

Table 1	The primers used in the real-time	RT-PCR	
Gene		Forward primer	Reverse primer
T ATA C	-		

Gene	Forward primer	Reverse primer
ZmNASI	GGAACTTTTGAGCACCTATGCG	CACTTCACAATGCATAGCATCGAAT
ZmDMASI	AAGTCCAAGGCCAAGACCG	AGTCCACGATGTCCAGGTTC
ZmTOMI	TGCAGAACTATGCTGTGCCA	GCATCTTGGCGTTTTTGGGT
ZmYS1	GTCTTCCATTCTCGCTCTGG	CAACCAACCACAGTTGATGC

	NM roots		M roots	
	24 h	48 h	24 h	48 h
ZmNAS1	-77**	+1005*	-93**	+50
ZmDMAS1	$-49^{**}$	+44*	-13*	-19
ZmTOM1	$-70^{*}$	+206*	-66*	-41*
ZmYS1	-35**	+263*	-51*	$-30^{*}$

**Table 2** Percentage alterations in the relative expression ratios of ZmNAS1, ZmDMAS1, ZmTOM1 and ZmYS1, 24 and 48 h after sulfate addition in mycorrhizal (M) and non-mycorrhizal (NM) maize roots

Values represent the percentage of increase or decrease (indicated with + or –, respectively) of the relative expression ratios of the genes

 $^{*/^{e*}}$  indicated when the difference between the samplings is statistically significant at p < 0.05/ 0.005, respectively

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# The Sulfur Pathway and Diagnosis of Sulfate Depletion in Grapevine

Sílvia Tavares and Sara Amâncio

**Abstract** Sulfur is an essential nutrient to all plant species. Plants assimilate sulfur in a well-described pathway, which has been taken up by roots. Regulatory mechanism has been the subject of many research papers. However, recent studies highlighted differences between crop plants and the model plant Arabidopsis thaliana. Our work focuses on the identification of genes involved in the sulfur metabolism in the Vitis vinifera genome, and their response to sulfur deficiency and other abiotic stress endured by grapevine in the field, namely water stress. Here, we describe the identification and brief characterization of the first assimilation enzymes involved in the sulfur pathway, the enzyme responsible for sulfur activation, ATP sulfurylase (ATPS), and the two enzymes that reduce sulfate to sulfide, Adenosine 5'-phosphosulate reductase (APR) and Sulfite reductase (SiR). A reduction was observed in the number of ATPS and APR isoforms identified in V. vinifera genome when compared to A. thaliana or Glycine max genomes. Two ATPS isoforms were present in the Vitis genome, of which only ATPS1 transcript was detected in the tested tissues, and one APR isoform, suggesting an absence of redundancy in the role of both enzymes. ATPS1, APR and SiR transcript level was up-regulated in response to 2 days exposure to sulfur deficiency in V. vinifera cell cultures, which was completely reversed by the addition of GSH to the culture medium. Apparently, oxidative stress triggered GSH has a pivotal role in the regulation of ATPS1, APR and SiR transcription level, since their up-regulation was observed in mRNA from field grapevine berries under water stress, which is known to induce oxidative stress.

Grapevine (*Vitis vinifera* L.) is one of the most important crops worldwide for winemaking and also for table grapes. Grapevines can be successfully grown in a range of different climates and management conditions. Global climate changes are associated with water deficit and high evapotranspiration rates that can affect berry development, yield, and wine quality (Hannah et al. 2013).

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In plants sulfur (S) is a major essential plant nutrient required for plant growth and development. Unlike animals, vascular plants use sulfate ( $SO_4^{2-}$ ) taken by the root system as the primary S source for plant growth (Clarkson et al. 1993) and reduce  $SO_4^{2-}$  to sulfide ( $S^{2-}$ ) in plastids of phototrophic organisms, including vascular plants (Shibagaki and Grossman 2008). Sulfide is further assimilated into the amino acids cysteine and methionine. Then most sulfate taken up by plants is incorporated into proteins (Leustek and Saito 1999; Leustek et al. 2000). Organic sulfur can be found in glutathione (GSH), the thiol tripeptide that mediates redox reactions by the interchange of dithiol-disulfide. In addition, several secondary S-metabolites have been suggested to play key roles in defense against pathogens (Hell and Kruse 2007) such as glucosinolates, which are produced mostly by members of the Brassicaceae.

Sulfate assimilation in vascular plants is accomplished in three steps catalyzed by enzymes whose codifying genes were confirmed in *Vitis vinifera* genome: *VvATPS1* and 2 (ATP sulfurylase, EC: 2.7.7.4); *VvAPSR* (APS reductase, EC: 1.8.99.2); VvSiR (sulfite reductase, EC: 1.8.7.1); *VvSERAT1-3* (Serine acetyltransferase, EC 2.2.1.30) and *VvOASTL1-13* (*O*-acetylserine (thiol) lyase, EC 2.5.1.47) (Amâncio et al. 2009).

The Bordeaux mixture, a high-S-content fungicide, has been used since the nineteenth century for the control of downy and powdery mildew in grapevine (Williams and Cooper 2004). Besides its major effect as a fungicide, it was an important source of S. Since sulfur fungicides have been substituted for organic compounds, the S supply to vineyards was reduced. Recent studies of  $SO_4^{2-}$ uptake, assimilation, and symptoms of sulfur deficiency in Vitis species were reported by our group (Tavares et al. 2008, 2013, 2015; Amâncio et al. 2009). In the first study sulfate uptake was correlated with the expression of sulfate transporter genes in cell systems of two Vitis species (V. vinifera and V. rupestris) (Tavares et al. 2008); the second investigation established a link between sulfur deficiency and phenolic compounds (Tavares et al. 2013) and the third work characterized the serine acetyl-transferase protein family revealing major differences to the best described A. thaliana family (Tavares et al. 2015). Altogether the results previously obtained directed our attention to the first steps of sulfate assimilation, namely sulfate activation and reduction, which were explored in different grapevine experimental systems and environmental conditions.

It is largely known that the experimental set up can influence the results. In a study performed to verify differences of plant sensitivity to chemical treatments the sensitivity varied whether species were treated under greenhouse or field conditions (Fletcher et al. 1990). Under water stress different responses were obtained at physiological and transcriptional levels in grapevine plants cultivated in greenhouse and field (Luisa Carvalho, personal communication). Different systems were used to tackle S metabolism in grapevine. Cell suspensions were selected as biological material to obtain a homogeneous experimental system. Studies with maize cells had reported a response to S deficiency following the same trend as intact plants, a de-repression of sulfate uptake (Clarkson et al. 1999). Also cell cultures allow S manipulation in short periods. Cell suspensions of *V. vinifera* var.

Touriga Nacional and V. rupestris were obtained as described in Tavares et al. 2008. Cell suspensions were sourced from liquid culture callus material and were grown in 250 ml flasks on a rotary shaker at 100 rpm, in the dark at 25 °C. After at least two weekly cycles in full sulfate (+S, 1.5 mM) sub-cultures were prepared for sulfate treatments: +S conditions (control) and sulfate depletion (-S, sulfate substituted by chloride). Leaves from V. vinifera Touriga Nacional subjected to abiotic stress were obtained as described in Rocheta et al. (2014) and Coito et al. (2012). Rooted cuttings were transferred to 3 l pots filled with sterilized soil and placed in the growth chamber. The following growth conditions were adjusted: light intensity 200  $\mu$ mol m<sup>-2</sup> s<sup>-1</sup>, 16 h light and 8 h darkness, 25 °C at day/23 °C at night, and watering with nutrient solution when necessary. The potted plants were subjected to drought and sampled when the pre-dawn leaf water potential was -0.9 MPa. V. vinifera cv. Touriga Nacional berries were collected from plants subjected to two irrigation regimes: rainfed (non-irrigation) and full irrigation (water supplied according to evapotranspiration rates) as described in Lopes et al. (2011). Berries were collected at veraison (50% colored berries) and full maturation.

The analysis of the *Vitis vinifera* genome indicates the presence of genes of two ATPS isoforms, one APR isoform with two splicing variants and one SiR isoform. ATPS and APR from plant and algae are encoded by small multigene families. A low number of isoforms is usually found in basal land plants and green algae, the studied species disclosing one or two distinct ATPS isoforms, and a APR unique isoform (Kopriva et al. 2007). In Arabidopsis and soybean genomes, genes encoding four ATPS isoforms and three APR isoforms were identified (Anjum et al. 2015; Hatzfeld et al. 2000; Yi et al. 2010). However, similar to *Vitis vinifera*, in the *Oryza sativa* genome only two ATPS isoforms were identified (Kopriva et al. 2007) and *Selaginella moellendorffii* seems to be the only vascular plant that has a unique ATPS (Kopriva et al. 2009). A single gene encodes SiR in *Vitis vinifera* in accordance with Arabidopsis (Takahashi et al. 2011) while in tobacco and soybean two isoforms were detected (Yi et al. 2010).

In *V. vinifera* the identified genes are located on distinct chromosomes; only ATPS2 shares the same chromosome with two sulfate transporters genes. Both ATPS genes depict five exons and four introns, however ATPS2 is organized in small exons and very long introns, which increased the genomic size from 5.3 to 11 Kbp. SiR is organized in eight exons and seven introns. In contrast the APR gene has only four or five exons. Two variants were identified for the APR gene that differ in a small sequence, variant one interpreted the sequence as an intron, in comparison variant 2 incorporated the sequence in the mRNA (Table 1).

The *Vitis vinifera* sulfate activation and reduction protein sequences present similar traits to other known plant proteins such as protein length and conserved domains (Table 1). *V. vinifera* ATPS coding regions have the N-terminal leader sequences characteristic for plastid-targeting transit peptides and a conserved ATPS catalytic domain (CD00517, Marchler-Bauer et al. 2011), typical of all described plant ATPS. All four ATPS from *Arabidopsis thaliana* (Rotte and Leustek 2000) and *Glycine max* (Yi et al. 2010) have been predicted as chloroplast isoforms

					Locali-
Gene	Chr	Genomic Region <sup>a</sup>	mRNA <sup>a</sup>	Protein <sup>a</sup>	zation
ATPS1	5	NW_003724020.1	XM_002283536.3	XP_002283572.1	С
		5.3 Kbp	1960 bp	467 aa	
ATPS2	18	NW_003724132.1	XM_002276957.2	XP_002276993.1	С
		11Kbp	2253 bp	483 aa	
APSr	12	NW_003724079.	XM_002269703.3	XP_002269739.2	С
		3.9 Kbp	(2) 1928 bp	(2) 467 aa	
			XM_010658907.1 (1)	XP_010657209.1 (1)	
			1885 bp	498 aa	
SiR	6	NW_003724030.1	XM_002285362.2	XP_002285398.1	C
		6.4 Kbp	2690 bp	687 aa	

 Table 1
 Identification of ATP sulfurylase (ATPS), APS reductase (APR) and sulfide reductase (SiR) genes in *Vitis vinifera* genome

*Chr* chromosome number, <sup>a</sup>NCBI reference numbers, *C* putative chloroplastic localization predicted in TargetP site (http://www.cbs.dtu.dk/services/TargetP/)

although ATPS activity is detected in chloroplasts and cytosol in Arabidopsis (Rotte and Leustek 2000) and spinach leaves (Lunn et al. 1990). Recently, Bohrer et al. (2015) showed that in Arabidopsis leaves, ATPS2 has a dual localization, namely in cytosol and chloroplasts, suggesting that a downstream methionine in the transit peptide sequence could act as an additional initiation translation site. In addition, such methionine is not present on the transit peptide sequence of ATPS1, 3 and 4. ATPS2 seems to be a distinct ATPS isoform, which probably confers different physiological roles; this isoform is also the only ATPS from Arabidopsis that is not a target to miRNA395 post-transcriptional control (Kawashima et al. 2009). Likewise, only the ATPS2 sequence from Vitis vinifera has an additional methionine on the transit peptide sequence and, contrary to VvATPS1, it is not a target of Vitis miRNA395 as predicted by the psRNATARGET tool (http:// plantgrn.noble.org/psRNATarget/) (Dai and Zhao 2011). In fact, the unrooted tree constructed with Vitis, Arabidopsis, soybean and Populus ATPS protein sequences (Fig. 1a) also shows a group including the ATPS2 sequences and a second group where the other sequences are not clearly separated. If a certain degree of redundancy in ATPS isoforms is observed among other plant species (Kopriva et al. 2009), apparently in the Vitis vinifera genome such redundancy is reduced or absent. A unique APR isoform was identified in the Vitis genome, with two variants that share 94% sequence homology. Sulfate reduction occurs exclusively in chloroplasts and, similarly to all plant species Vitis APR has a N-terminal transit peptide for plastid-targeting and a conserved multidomain consisting of an reductase domain and a C-terminal thioredoxin-like domain (CD02993, Marchler-Bauer et al. 2011).

Arabidopsis (Kopriva et al. 2009) and soybean (Yi et al. 2010) have three APR isoforms, apparently some level of redundancy exists, since Arabidopsis plants without functional APR1 or APR2 do not show obvious differences when compared with the wild type, however the disruption of *APR2* leads to an 80% decrease in APR activity and an accumulation of sulfate indicating that APR2 is a major



**Fig. 1** Phylogenetic analysis of ATPS sulfurylase (ATPS, **a**) and APS reductase (APR; **b**) protein sequences. Unrooted tree constructed using the T-Cofee and PHYLIP programs, PRODIST and NEIGHBOR (http://tcoffee.crg.cat/ and http://mobyle.pasteur.fr/cgi-bin/portal.py#welcome, respectively). At *Arabidopsis thaliana*, Gm *Glycine max*,, Pt *Populus trichocarpa*, Vv *Vitis vinifera* 

isoform in Arabidopsis (Loudet et al. 2007). The identification of APR in the *Vitis* genome shows a lack of redundancy and the phylogenetic analysis of the unique *Vitis* APR together with APR protein sequences from Arabidopsis, soybean and *Populus* shows that the *Vitis* protein is closer to Arabidopsis and soybean APR2. The APR protein sequences from the selected species seem to group better inside each species, which could be determined by species-specific evolution of APR (Fig. 1b). Together with APR, SiR is strictly plastidic. *Vitis* SiR contains the siroheme and a [4Fe-4S] cluster typical of plant SiR proteins and the transit peptide for chloroplast targeting (Table 1).

The presence of transcripts encoding enzymes involved in sulfate activation and reduction, ATPS1 and 2, APR and SiR, were examined in different grapevine tissues by RT-PCR (Fig. 2). ATPS2 was the only gene not detected in RNA isolated from the sampled tissues, namely leaves (young and mature) and roots of potted plants, berries collected in the field and cells from cell culture (Fig. 2b, in berries). V. vinifera ATPS2 protein sequence was most similar to the protein sequences of Populus, ATPS1 and ATPS2, and AtAPS2 (Fig. 1a) recently proven to have dual subcellular chloroplastic/cytosolic localization, although a physiological role of the cytosolic isoform remains unknown (Bohrer et al. 2015). It has been speculated that cytosolic ATPS may be linked to cytosolic APS kinase in providing PAPS for the secondary metabolism (Rotte and Leustek 2000), namely in the production of glucosinolates. V. vinifera invests a great deal of resources in secondary metabolites, such as phenolic compounds, and its genome is enriched in genes devoted to secondary metabolism (Velasco et al. 2007). V. vinifera synthesizes no glucosinolates, which might be the reason why ATPS2 transcripts were not detected in different plant tissues. RNA isolated from berries at *veraison* showed a very



**Fig. 2** (a) mRNA detection by RT-PCR in different *Vitis vinifera* tissues with specific primers design to ATP sulfurylase1 (ATPS1), APS reductase (APR) and sulfide reductase (SiR) sequences identified in Table 1. (b) detection of *ATPS1* and *2*, *APR* and *SiR* transcripts in grapevine berries collected at veraison and full maturation. RNA extraction and RT-PCR were performed according with standard molecular biology techniques

weak signal in RT-PCR using *ATPS1* primers, contrary to mature berries RNA which produced a strong signal indicating presence of the gene (Fig. 2b). Interestingly, a group three sulfate transporter was up-regulated at berry maturation (Guillaumie et al. 2011) and in seeds, transcripts for sulfur metabolism genes, namely group three sulfate transporters, were over-represented when compared with pulp transcripts (Grimplet et al. 2007). Together these results suggest a stimulus of sulfur metabolism at berry maturation, in particularly in seeds.

The transcript levels of *VvATPS1*, *VvAPR* and *VvSiR* analyzed by qPCR responded equally to S depletion in cell cultures; all genes were up-regulated in cells after 2 days in an S deficient medium (Fig. 3a, b and c), and an increase in mRNA level was observed in *VvAPR* (Fig. 3b). S deficiency is known to be responsible for increasing the transcript levels of group one sulfate transporters and *APR* in plant species, including *Vitis vinifera* (Tavares et al. 2008), the reason why APR activity and sulfate transport are considered to exert the highest control over the S metabolic pathway (Vauclare et al. 2002). Nevertheless ATPS up-regulation was also observed in several plant species, namely *Arabidopsis thaliana* and *Zea mays* (reviewed by Anjum et al. 2015). The addition of sulfate and GSH to the S deficient medium completely reverses the up-regulation of *VvATPS1*, *VvAPR* and *VvSiR* transcripts. Interestingly, in *Vitis* cells the level of GSH as measured by HPLC (Tavares et al. 2015) was the first S compound to



**Fig. 3** Relative expression of ATP sulfurylase1 (*ATPS1*), APS reductase (*APR*) and Sulfide reductase (*SiR*) transcripts in grapevine cell culture ( $\mathbf{a}$ ,  $\mathbf{b}$  and  $\mathbf{c}$ ) and leaves and berries ( $\mathbf{d}$ ). ( $\mathbf{a}$ ,  $\mathbf{b}$  and  $\mathbf{c}$ ). Cells collected after 2 days in sulfur deficiency (-S) and sulfur sufficient (+S) medium, after which sulfur (S), cysteine (cys) and GSH were added to the -S medium and *O*-acetylserine (OAS) and *N*-acetylserine (NAS) to +S. Relative expression was compared to +S cells using RT-qPCR. ( $\mathbf{d}$ ) Water stress grapevine leaves and non-irrigated berries relative transcript level of *ATPS1*, *APR* and *SiR* was compared to control leaves and full irrigated berries by RT-qPCR

significantly decrease in cells after 1 day in S deficient medium. Cysteine added to cells in an S deficient medium also triggered a reversion of the up-regulation observed in the transcript level of *VvAPR* (Fig. 3b); apparently cysteine had a direct effect on the *VvAPR* transcription though we only detected a decrease in the amount of cysteine inside the cells 5 days after S limitation (Tavares et al. 2015). OAS is considered to act as a positive regulator of sulfate transporters, and commonly has similar effects under conditions of S deficiency (Takahashi et al. 2011). Consequently OAS added to *Vitis vinifera* cell culture showed the same up-regulation effect as S deficiency, although the magnitude was not so drastic (Fig. 3a, b and c).

In leaves from potted plants and berries collected from field plants, both grown under conditions of water stress the *VvATPS* and *VvAPR* transcript level increased. In contrast, *VvSiR* was down-regulated in leaves and up-regulated in berries (Fig. 3d). A change in ATPS activity in response to oxidative stress was reported by Kopriva et al. (2007). Similar results were obtained under conditions of abiotic stress (reviewed in Anjum et al. 2015). A high demand for GSH, an important S-compound in the response to oxidative stress, occurs under water stress. This may unfold an up-regulation effect in the mRNA of the first enzymes of sulfur assimilation. Likewise, the serine acetyltransferase (SAT) mRNA level was up-regulated under water stress conditions in *Vitis vinifera* leaves (Tavares et al. 2015).

As *VvSERAT2;1* was up-regulated in leaves of plants under water stress, this result prompted us to test the genes for the first enzymes of the S assimilation pathway (ATPS, APR and SiR) in leaves and berries of plants under water deficit. Although the patterns of transcripts expression were distinctly different between the two plant organs, the results suggest that some enzymes involved in sulfate metabolism are regulated by S-status and by environmental conditions, e.g. water deficit. Our study showed that major characteristics of ATPS, APR and SiR are very well conserved among plant species. However, it would be interesting to highlight differences between these enzymes in order to obtain a complete overview of S assimilation in different plant species.

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## Impact of Sulfate Deprivation and H<sub>2</sub>S Exposure on the Metabolites of the Activated Methyl Cycle in Chinese Cabbage

#### Mei-Hwei Tseng, Chao-Kai Yang, C. Elisabeth E. Stuiver, Chiu-Ching Chang, and Luit J. De Kok

**Abstract** The activated methyl cycle is a central metabolic pathway used to generate (and recycle) several important sulfur-containing metabolites including methionine, S-adenosylmethionine (SAM) and S-adenosylhomocysteine (SAH) and enable methylation. We have developed a precise and sensitive method for the simultaneous measurement of several sulfur metabolites based on liquid chromatography coupled with tandem mass spectrometry (LC-MS/MS) and <sup>34</sup>S-metabolic labeling of sulfurcontaining metabolites including glutathione and the metabolites of the activated methyl cycle. Sulfate deprivation resulted in a decreased biomass production and content of glutathione, methionine, SAH of both shoot and root, and SAM of the root of Chinese cabbage. Foliarly absorbed H<sub>2</sub>S may able to replace sulfate taken up by the root as sulfur source for growth and an atmospheric concentration of 0.2  $\mu$ l 1<sup>-1</sup> alleviated the decrease in the content of sulfur metabolites. The SAM content of the shoot was hardly affected upon sulfate-deprivation, resulting an increase in the SAM/ SAH ratio, indicating a potential higher methylation capacity under this condition.

Seedlings of Brassicacea are characterized by their high growth rate (up to  $0.4 \text{ g g}^{-1} \text{ day}^{-1}$ ) and high sulfur demand; the sulfate uptake rate of some species may exceed 40 µmol g<sup>-1</sup> fresh weight root day<sup>-1</sup> (Shahbaz et al. 2010; Stuiver et al. 2014; Aghajanzadeh et al. 2016). The uptake of sulfate by the root is adjusted to the sulfur demand for growth, even at an external sulfate concentrations close to the K<sub>m</sub> value

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of the high affinity sulfate transporters (approx. 5 µM; Koralewska et al. 2007, 2012). The interaction between atmospheric  $H_2S$  and pedospheric sulfate nutrition of plants has been extensively studied during the last three decades (De Kok 1990; De Kok et al. 2000, 2002, 2007). Foliarly absorbed H<sub>2</sub>S was directly metabolized into cysteine and subsequently into other organic sulfur compounds, and exposure resulted in an increase in the content of water-soluble non-protein thiol content (viz. cysteine and glutathione) of the shoot (De Kok 1990: De Kok et al. 2000, 2002, 2007). H<sub>2</sub>S exposure hardly affected the total sulfur and sulfate contents of Brassica, even not at relatively high atmospheric concentrations (De Kok et al. 2000), but resulted in a down-regulation of the uptake of sulfate by the root and reduction in the shoot (De Kok et al. 2000, 2002, 2007; Koralewska et al. 2008; Shahbaz et al. 2014: Aghaianzadeh et al. 2014, 2016). Upon sulfate-deprivation, the assimilated foliarly absorbed H<sub>2</sub>S may replace sulfate as a sulfur source for growth of *Brassica* (De Kok et al. 2000, 2002, 2007; Buchner et al. 2004; Koralewska et al. 2008; Shahbaz et al. 2014; Aghajanzadeh et al. 2016). H<sub>2</sub>S exposure hardly affected the up-regulated expression and activity of the high affinity sulfate transporters in sulfate-deprived *Brassica* (Koralewska et al. 2008; Shahbaz et al. 2014), and the decrease in shoot to root biomass partitioning upon sulfate deprivation remained largely unaffected upon H<sub>2</sub>S exposure (Koralewska et al. 2008; Shahbaz et al. 2014; Aghaianzadeh et al. 2016). There is apparently a rather poor shoot to root signaling in Brassica of the regulation of both the sulfate transporters in the root and shoot to root biomass partitioning, indicating that both are determined by the sulfate concentration in the root environment rather than by the sulfur status of the plant itself.

Methionine is an essential metabolite in plants and all-living organisms (Ravanel et al. 1998). Apart from a role as a protein constituent, methionine is the precursor of *S*-adenosyl-L-methionine (SAM), the primary biological methyl-group donor (Roje 2006). The highly reactive methylated sulfur of SAM is used by a broad range of methyltransferases (Poel et al. 2013). A by-product of SAM-dependent transmethylation, *S*-adenosylhomocysteine (SAH) is released, which is recycled to methionine via homocysteine through the activated methyl cycle (Bürstenbinder and Sauter 2012). SAH strongly inhibited methyltransferase through competition with the substrate SAM (Barbes et al. 1990; Moffatt and Weretilnyk 2001). The ratio of cellular SAH and SAM is indicative for the methylation capacity of the cell (Fulnecek et al. 2011; Poel et al. 2013). In the current study the impact of sulfate deprivation and  $H_2S$  exposure on the content of the metabolites involved in the active methyl cycle was studied in Chinese cabbage.

Seeds of Chinese cabbage (*Brassica pekinensis* (Lour.) Rupr. cv. Kasumi F1; Nickerson Zwaan, Made, The Netherlands) were germinated in vermiculite in a climate controlled room. Day and night temperatures were 22 °C and 18 °C ( $\pm 1$  °C), respectively, relative humidity of 60–70% and a 14-h photoperiod at a photon fluence rate of 300  $\pm$  20 µmol m<sup>-2</sup> s<sup>-1</sup> (400–700 nm) at plant height, supplied by Philips GreenPower LED (deep red/white 120) production modules. Ten-day-old seedlings were transferred to an aerated 25% Hoagland nutrient solution (pH 5.9) with 0.5 mM sulfate (+S, sulfate sufficient) or 0 mM sulfate (–S, sulfate deprived; all sulfate salts replaced by chloride salts) in 13 l stainless

exposed to 0.2 µl l<sup>-1</sup> H<sub>2</sub>S in 150 l cylindrical stainless steel cabinets (0.6 m diameter) with a poly(methylmethacrylate) top. The lids of the containers and the plant sets were sealed in order to prevent absorption of atmospheric  $H_2S$  by the nutrient solution. Day and night temperatures in the fumigation cabinets were 22 and 19 °C ( $\pm$  2 °C), respectively, and relative humidity was 40–50%. The photoperiod was 14 h at a photon fluence rate of  $300 \pm 20 \text{ }\mu\text{mol}\text{ }m^{-2}\text{ }s^{-1}$  (within the 400–700 nm range) at plant height, supplied by Philips GreenPower LED (deep red/white 120) production modules. The air exchange was 40 l min<sup>-1</sup> whereas a ventilator stirred the air inside the cabinets continuously. Pressurized H<sub>2</sub>S diluted with  $N_2$  (1 ml l<sup>-1</sup>) was injected into the incoming air stream and the concentration in the cabinet was adjusted to the desired level using electronic mass flow controllers (ASM, Bilthoven, The Netherlands). The air exchange was 40  $1 \text{ min}^{-1}$  and a ventilator stirred the air inside the cabinets continuously. The H<sub>2</sub>S concentration in the cabinets was monitored by an SO<sub>2</sub> analyzer (model 9850) equipped with a H<sub>2</sub>S converter (model 8770; Monitor Labs, Measurement Controls Corporation, Englewood, CO, USA). Plants were harvested after H<sub>2</sub>S exposure. Roots were separated from shoots, weighed, and were frozen immediately in liquid N<sub>2</sub> and stored at -80 °C until further use. For determination of the sulfur metabolites, plant tissue was freeze-dried at -60 °C for 48–72 h. Freeze-dried plant tissues were ground to powder with liquid nitrogen in a mortar with pestle. The sulfur metabolites were extracted as described in (Chang et al. 2013). For isotope dilution mass spectrometry analysis, the <sup>34</sup>S-labeled Arabidopsis thaliana tissue were extracted and added to the calibration standards, OC samples, and plant samples in a fixed ratio (Chang et al. 2013). Chromatographic separations of sulfur metabolites were performed on a Thermo Accela LC system using a Thermo Scientific Hypersil Gold aQ C18  $(1.9 \,\mu\text{m}, 2.1 \,\text{mm} \times 10 \,\text{cm})$ . Separations were performed under isocratic condition at a flow rate of 0.25 ml min<sup>-1</sup>. The mobile phase was composed of 0.1% formic acid in water. ESI-MS/MS analysis was performed on a Thermo Scientific TSQ Quantum Ultra. Metabolites were detected in positive ionization mode using multiple reaction monitoring scanning mode. The spray voltage was set to 3.5 kV, the ion-transfer capillary temperature was set to 280 °C, the sheath gas pressure was set to 50 (arbitrary units), and the auxiliary gas pressure was set to 15 (arbitrary units). Collision energy set at 35% was used for each metabolite (Table 1).

An 11-day exposure of Chinese cabbage to sulfate-deprived condition resulted in a decreased plant biomass production (Table 2). The shoot growth was relatively more affected than the root growth resulting in a decrease in shoot/root ratio. Exposure of sulfate-sufficient plants to  $0.2 \,\mu l \, l^{-1} \, H_2 S$  hardly affected plant biomass production, but it alleviated the reduction in biomass production of sulfate-deprived plants. The latter demonstrated that similar to previous observations that at an atmospheric H<sub>2</sub>S concentration  $\geq 0.2 \ \mu l^{-1}$  the foliarly absorbed sulfide fully could replace sulfate taken up by the sulfur source for growth (Koralewska et al. 2008; Shahbaz et al. 2014; Aghajanzadeh et al. 2014, 2016). H<sub>2</sub>S exposure of sulfate-sufficient plants resulted in a slight increase in the total glutathione (expressed as GSH) content of the shoot and not of the root (Table 3). Sulfate

Compound name	Precursorion	Production	Collision energy
Methionine	150	56	15
Methionine S34	152	106	15
GSH	308	162	20
GSH S34	310	181	20
SAH	385	134	20
SAH S34	387	136	20
SAM	399	250	20
SAM S34	401	250	20
<sup>32</sup> S GSSG	613	355	25
<sup>34</sup> S GSSG	617	359	25

Table 1 Mass spectrometry parameters with transition pairs in MRM mode and normalized collision energy (%)

Table 2 Impact of  $H_2S$  and sulfate deprivation on biomass production of shoots and roots of Chinese cabbage

	+Sulfate		-Sulfate	
	$0 \ \mu l \ l^{-1} \ H_2 S$	$0.2 \ \mu l \ l^{-1} \ H_2 S$	$0 \ \mu l \ l^{-1} \ H_2 S$	$0.2 \ \mu l \ l^{-1} \ H_2 S$
Shoot (g FW)	$3.98\pm0.38b$	$5.39 \pm 1.91b$	$1.12\pm0.04a$	$4.69\pm0.66b$
Root (g FW)	$0.78\pm0.05\mathrm{b}$	$1.05 \pm 0.26b$	$0.33 \pm 0.04a$	$1.10 \pm 0.08c$
Shoot/root ratio	$5.13\pm0.27b$	$5.29 \pm 1.75$ ba	$3.46\pm0.44a$	$4.27\pm0.27a$

Ten-day-old seedlings were grown on a 25% Hoagland solution at 0 and 0.5 mM sulfate (+S) and exposed to 0.2  $\mu$ l l<sup>-1</sup> H<sub>2</sub>S for 11 days. The initial fresh biomass of the shoot and root of Chinese cabbage was 0.171  $\pm$  0.001 g and 0.057  $\pm$  0.001 g, respectively. Data on plant yield (g FW) and shoot/root ratio represent the mean of three measurements with three plants in each ( $\pm$ SD) Means with different letters are significant different at p < 0.01 (unpaired Student's t-test)

deprivation resulted in a strong decrease in the glutathione content of both shoot and root but this decrease was largely alleviated upon H<sub>2</sub>S exposure, though its content remained lower than that of the sulfate-sufficient plants. Sulfate deprivation also resulted in a substantial decrease in the methionine and SAH content of both shoot and root and SAM content of the root (Table 3). However, the SAM content of the shoot was hardly affected upon sulfate deprivation, resulting in a fourfold increase in the SAM/SAH ratio (Table 3). The SAM/SAH ratio in plant tissue has often been used as a reporter of the methylation capacity ("methylation index"; Groth et al. 2016). In this view the increase in SAM/SAH ratio in the shoot upon sulfate deprivation would indicate an increased methylation capacity. Exposure of sulfate-deprived plants to H<sub>2</sub>S alleviated the decrease in SAM and SAH content of the shoots, which were comparable to that of shoots of sulfate-sufficient plants (Table 3). However, the methionine content was slightly lower than that of sulfatesufficient plants. The contents of methionine, SAM and SAH sulfur of the roots were decreased upon sulfate deprivation, whereas the content of these sulfur metabolites in the root of sulfate-deprived H<sub>2</sub>S-exposed plants was quite similar to that of sulfate-sufficient plants. The activated methyl cycle is a central metabolic

	+Sulfate		-Sulfate	
	$0 \ \mu l \ l^{-1} \ H_2 S$	$0.2 \ \mu l \ l^{-1} \ H_2 S$	$0 \ \mu l \ l^{-1} \ H_2 S$	$0.2 \ \mu l \ l^{-1} \ H_2 S$
Shoot				
Glutathione	$5.67 \pm 0.22c$	$6.69\pm0.42d$	$0.18\pm0.01a$	$3.87\pm0.16b$
Methionine	$0.224\pm0.004\mathrm{c}$	$0.260 \pm 0.039c$	$0.038\pm0.006a$	$0.179 \pm 0.032b$
SAM	$0.123\pm0.008ab$	$0.134 \pm 0.011b$	$0.106 \pm 0.011a$	$0.113 \pm 0.007a$
SAH	$0.015\pm0.001\mathrm{b}$	$0.014\pm0.001\mathrm{b}$	$0.003 \pm 0.001a$	$0.012 \pm 0.001b$
SAM/SAH ratio	$8.2 \pm 1.1$ a	9.6 ± 1.5a	$35.3 \pm 15.4 \text{b}$	9.4 ± 1.4a
Root				
Glutathione	$4.62\pm0.15c$	$5.17 \pm 0.54c$	$0.55 \pm 0.07a$	$2.79\pm0.10b$
Methionine	$0.136 \pm 0.014b$	$0.135 \pm 0.024b$	$0.024 \pm 0.004a$	$0.121 \pm 0.004b$
SAM	$0.207 \pm 0.002b$	$0.210 \pm 0.024$ bc	$0.076 \pm 0.016a$	$0.231 \pm 0.002c$
SAH	$0.019\pm0.002\mathrm{bc}$	$0.017 \pm 0.002b$	$0.006 \pm 0.003a$	$0.021 \pm 0.002c$
SAM/SAH ratio	$10.9 \pm 1.3a$	$12.3 \pm 2.9a$	$12.3\pm8.6a$	$11.0 \pm 1.1a$

 Table 3
 Impact of  $H_2S$  and sulfate deprivation on the content of glutathione and metabolites of activated methyl cycle in shoots and roots of Chinese cabbage

For experimental details, see legends of Table 2. Data on metabolite content is expressed as  $\mu$ mol g<sup>-1</sup> dry weight (freeze-dried plant material) and represent mean of three measurements with three plants in each ( $\pm$  SD). Means with different letters are significant different at p < 0.01 (unpaired Student's t-test)

pathway responsible for the methylation of cellular components and the recycling of sulfur-containing metabolites. The methylation of essential biological molecules, e.g. nucleic acids, hormones, lipids, proteins, is of crucial importance for many key biochemical processes (Chiang et al. 1996). Apparently the foliarly absorbed sulfide by sulfate-deprived Chinese cabbage upon exposure to an atmospheric level of 0.2  $\mu$ l l<sup>-1</sup> H<sub>2</sub>S was sufficient for reduced sulfur requirement of the plant to support growth and to maintain the levels of the metabolites involved in the activated methyl cycle in both the shoot and the root.

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## Sulfate Transporters Involved in Cd-Induced Changes of Sulfate Uptake and Distribution in *Arabidopsis thaliana*

#### Chisato Yamaguchi and Akiko Maruyama-Nakashita

Abstract Cadmium (Cd) is a highly toxic element for living organisms, hence plants have evolved a variety of detoxification mechanisms. Phytochelatins and glutathione are low-molecular-weight sulfur compounds that function as chelators and play important roles in Cd detoxification. Previous studies have shown that the transcription of the genes involved in sulfate uptake and sulfur assimilation was increased in response to Cd stress. Recently, we reported that Cd-induced sulfate uptake is mainly attributed to the function of SULTR1;2, a high affinity sulfate transporter involved in sulfate uptake from the roots. Another distinct change in sulfate distribution induced by Cd treatment was preferential accumulation of sulfate to the shoots, which is due to the induction of root-to-shoot sulfate transport through xylem. In this study, we compared previous transcriptome data taken with Cd-treated plants to get suggestions about the SULTRs involved in Cd-induced sulfate distribution to shoots. In addition to the induction of SULTR1:1 and SULTR1;2 expressions, we found that the expression of SULTR2;1, a transporter involved in root-to-shoot transport of sulfate, and SULTR4;1 and SULTR4;2, exporters of stored sulfate in vacuole, were increased in roots upon Cd treatment. These SULTRs were suggested as contributors to the increased distribution of sulfate to shoots under Cd exposure.

Cadmium (Cd) is a non-essential and toxic heavy metal for living organisms commonly contaminated in the ecosystems (Nawrot et al. 2006; Järup and Akesson 2009; Clemens et al. 2013; Choppala et al. 2014). In plants, Cd affects many physiological and metabolic processes, including photosynthesis, ion homeostasis, and mineral uptake, partly through the production of reactive oxygen species (Lin and Aarts 2012; Clemens et al. 2013; Choppala et al. 2014). Therefore, plants have evolved a variety of detoxification mechanisms to avoid these toxicities (Verbruggen et al. 2009; Yadav 2010; Lin and Aarts 2012; Chmielowska-Bak et al. 2014; Choppala et al. 2014).

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Low-molecular-weight sulfur compounds, glutathione (GSH) and phytochelatins (PCs) are known to contribute to heavy metal detoxification in plants (Cobbett 2000; Cobbett et al. 2002; Choppala et al. 2014; Seth et al. 2012). GSH or PCs chelate Cd, and these complexes are subsequently compartmentalized into vacuoles, thus lowering Cd levels in the cytosol (Cobbett 2000; Cobbett et al. 2002; Choppala et al. 2014). GSH is synthesized from cysteine, the end product of sulfate assimilation pathway (Leustek et al. 2000; Saito 2004), via two enzymatic reactions catalyzed by  $\gamma$ -glutamylcysteine ( $\gamma$ -Glu-Cys) synthase and glutathione synthase (Lu 2013). Then PCs are synthesized from GSH by the enzymatic conjugation of  $\gamma$ -Glu-Cys to GSH, which is catalyzed by phytochelatin synthase (Cobbett and Goldsbrough 2002). Phytochelatin synthase also catalyzes the sequential conjugation of the  $\gamma$ -Glu-Cys mojety to various PCs, which forms longer PC species with the general structure ( $\gamma$ -Glu-Cys)n-glycine (Gly) (n = 2 to 11) (Cobbett and Goldsbrough 2002).

The sulfur assimilation pathway is started from sulfate uptake. In Arabidopsis, two high-affinity sulfate transporters, SULTR1;1 and SULTR1;2, expressed in the epidermis and cortex of roots facilitate sulfate uptake from the rhizosphere (Takahashi et al. 2000; Vidmar et al. 2000; Shibagaki et al. 2002; Yoshimoto et al. 2002, 2007; Maruyama-Nakashita et al. 2003). Gene expression of both *SULTR1;1* and *SULTR1;2* is dramatically induced by Cd exposure (Herbette et al. 2006; Rouached et al. 2008; Besson-Bard et al. 2009; Jobe et al. 2012). Sulfate uptake activity was also increased by Cd treatment in wild-type Arabidopsis (WT), as reported in maize roots (Nocito et al. 2002, 2006). However, the Cd-induced increase of sulfate uptake activity was diminished in *SULTR1;2* knockout mutant, *sel1-10*, indicating that SULTR1;2 is the main contributor to Cd-induced sulfate uptake (Table 1; Fig. 1; Yamaguchi et al. 2016). Though not significant, Cd treatment tended to induce sulfate uptake activity and induce increased total sulfur and sulfate levels in *sel1-10* shoots, which suggests that the induction of *SULTR1;1* also contributes to the long-term accumulation of sulfate in Cd-treated plants.

Even though the sulfate acquisition was reduced in *sel1-10*, the growth of both WT and *sel1-10* plants were similarly inhibited by Cd treatment, probably because of the similar accumulation of PCs in shoots (Yamaguchi et al. 2016). Sulfate uptake activity, sulfate, and total sulfur content in *sel1-10* plants were less than 50% of those in the WT, both *sel1-10* and WT plants accumulated similar levels of thiols when treated with 20  $\mu$ M CdCl<sub>2</sub>, and upon treatment with 40  $\mu$ M CdCl<sub>2</sub>, *sel1-10* plants accumulated lower levels of thiols in roots, but still maintained similar thiol levels in shoots. In contrast, Cd-induced accumulation of cysteine and GSH observed in the WT was lower or absent in *sel1-10* plants. Such differences in induction of thiol compounds indicates that PC accumulation, especially in shoots, is the first priority for plants suffering from Cd stress.

In addition to the stimulation of sulfate uptake, Cd treatment increased sulfate accumulation in shoots (Yamaguchi et al. 2016). The Cd-induced sulfate accumulation was occurred only in shoots, whereas PC accumulation was enhanced in both root and shoot tissues, and was caused by the increased sulfate transport to shoots as demonstrated with the increased sulfate concentration in xylem sap under Cd

		1												
Reference	(1)		(2)	(3)						(4)	(5)	(9)		(2)
Cd concentration (µM)	15	30	200	5			50			20	15	20	40	200
Treated priod	2 h	2 h	6 h	2 h	6 h	30 h	2 h	6 h	30 h	24 h	24 h	10 days		6 h
SULTR1;1	I	1.03	1	I	1	I	96.0	1.55	1.58	0.89	I	3.30	3.33	1
SULTR1;2	1	1	1	1	1	I	I	1	1	1.73	I	2.15	2.31	3.14
SULTR1;3	I	I	I	I	I	I	I	I	I	n.d.	I	2.13	2.56	
SULTR2;1	0.73	1.48	I	I	1.08	0.94	I	1.40	3.39	n.d.	4.16	2.85	3.44	
SULTR2;2	I	I	I	I	I	I	I	I	I	n.d.	I	I	I	
SULTR3;1	I	I	I	I	I	I	I	I	I	n.d.	I	I	I	
SULTR3;2	I	I	I	I	I	I	I	I	I	n.d.	I	I	I	
SULTR3;3	I	I	I	Ι		Ι	I	Ι	Ι	n.d.	I	I	I	I
SULTR3;4	Ι	I	I	I	0.65	I	I	Ι	I	n.d.	I	1.45	1.43	I
SULTR3;5	I	I	I	I	-0.60	I	I	-0.97	-0.80	n.d.	I	I	I	
SULTR4;1	Ι	I	I	I	I	I	I	0.82	2.05	n.d.	I	I	I	I
SULTR4;2	I	I	I	Ι	I	I	I	I	Ι	n.d.	I	1.27	2.08	
Data previously taken with	microarra	ay and qu	antitative	RT-PCI	R were sun	nmarized.	. Number	s in the "R	eference"	indicate th	ne referen	ces in wh	ich the or	iginal
data were reported; i.e. (1)	Besson-J	Bard et a	.d. (2009)	, (2) Li	et al. (201	0), (3) H	erbette et	al. (2006	), (4) Roui	ached et a	al. (2008)	, (5) Zha	o et al. (	2009),
(6) Yamaguchi et al. (2016	9), (7) Jol	be et al.	(2012). T	The value	es indicate	log ratio	(base 2)	of fold cl	hanges of	gene expi	ession le	vels unde	r Cd trea	utment
compared with those without	ut Cd (p ·	< 0.05 by	y Bonferr	oni corre	ection, n =	= 3, in Re	f. (1); p <	< 0.05 by 5	Student's t-	-test, n =	3, in Refs	s. (2 and :	5); $p < 0$	.05 by
Student's <i>t</i> -test, $n = 2$ , in R <sub>t</sub>	ef. (3); sig	spificant (	difference	es were d	letermined	from SD.	, $n = 4$ to	6, in Ref. (	(4); $p < 0.0$	5 by Tuk	ey-Kram(	er test, n =	= 4, in Re	if. (6);

**Table 1** Expression of several sulfate transporters (*SULTRs*) was influenced by Cd treatment in roots

no statistical analysis, n = 1 in Ref. (7)). Data of the genes with log ratios less than 0.0, between 0.0 and 1.0, and more than 1.0 were shown in italic, plane, and

bold, respectively. "n.d." means not determined. "-" means no significant difference was detected between Cd-treated and non-treated samples



Fig. 1 Cd-induced alterations of sulfate uptake and distribution in *Arabidopsis thaliana*. Arrows indicate the direction of sulfate flow. Cd induces the increase of sulfate uptake mainly depends on SULTR1;2 activity. SULTR2;1, SULTR3;5 and SULTR4 is likely involved in root-to-shoot sulfate transport through xylem under Cd exposure. The mechanism of sulfate transport into xylem is unknown. Enhanced sulfate uptake and the higher distribution rate of sulfate to shoots could contribute to the active thiol synthesis observed in Cd-treated shoots, which results in promoting thiol-mediated Cd detoxification

exposure (Yamaguchi et al. 2016). However, the SULTRs responsible for the Cd-induced increase of root-to-shoot sulfate transport remain to be elucidated.

In Arabidopsis, there are several *SULTRs* involved in root-to-shoot transport of sulfate, i.e. *SULTR2;1, SULTR3;5, SULTR4;1*, and *SULTR4;2* (Fig. 1; Takahashi et al. 2000; Kataoka et al. 2004a, b; Kawashima et al. 2011; Maruyama-Nakashita et al. 2015). *SULTR2;1* and *SULTR3;5* synergistically mediate root-to-shoot transport of sulfate by retrieving apoplasmic sulfate released from xylem in roots (Takahashi et al. 2000; Kataoka et al. 2004a; Kawashima et al. 2011). In addition, increase of *SULTR2;1* expression in roots contributes to increasing root-to-shoot sulfate transport under sulfur deficiency (Kataoka et al. 2004a; Maruyama-Nakashita et al. 2015). SULTR4;1 and SULTR4;2 are the exporters of sulfate from vacuoles in the pericycle and xylem parenchyma cells of roots, and contribute to root-to-shoot sulfate transport by controlling the cytosolic concentration of sulfate in the cells around xylem vessels (Kataoka et al. 2004b).

In order to obtain any suggestions about the molecular machinery of Cd-induced increase of sulfate transport to shoots, the effects of Cd treatments on the expression of *SULTRs* in roots previously detected with microarray and quantitative RT-PCR were summarized in Table 1 (Herbette et al. 2006; Rouached et al. 2008; Besson-Bard et al. 2009; Zhao et al. 2009; Li et al. 2010; Jobe et al. 2012; Yamaguchi et al. 2016). Several studies of Cd-treated Arabidopsis have reported that the expression of *SULTR1*; *1* and/or *SULTR1*; 2 were induced by Cd treatment (Herbette et al. 2006;

Rouached et al. 2008; Besson-Bard et al. 2009; Jobe et al. 2012), in agreement with the stimulation of sulfate uptake upon Cd exposure. In addition, the increase of *SULTR2;1* transcript was often detected under Cd exposure. Transcript levels of *SULTR2;1* in roots were increased by Cd treatment in time- and concentration-dependent manner (Herbette et al. 2006; Besson-Bard et al. 2009; Zhao et al. 2009; Yamaguchi et al. 2016). The transcript levels of *SULTR4;1* and *SULTR4;2* were also increased in response to Cd treatment (Herbette et al. 2006; Yamaguchi et al. 2016). Though the expression of *SULTR3;5* was not increased by Cd treatment, SULTR3;5 also can take part because sulfate uptake activity of SULTR3;5 is controlled by the expression of *SULTR2;1* (Kataoka et al. 2004a).

These results suggest that all SULTRs responsible for sulfate translocation to shoots are involved in Cd-induced increase of sulfate transport to shoots (Fig. 1). We need further evaluation to determine the molecular mechanisms how Cd treatment increases the sulfate transport to shoots and whether these SULTRs are involved in the enhanced sulfate transport, e.g. the analysis of sulfate translocation using the knockout mutants of these SULTRs. Then further exploration of the physiological meaning in enhanced sulfate transport to shoots by Cd treatment would be determined.

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## A Glycine-Rich Protein Encoded by Sulfur-Deficiency Induced Gene Is Involved in the Regulation of Callose Level and Root Elongation

#### Anna Znój, Katarzyna Zientara-Rytter, Paweł Sęktas, Grzegorz Moniuszko, Agnieszka Sirko, and Anna Wawrzyńska

**Abstract** Glycine-rich proteins (GRPs) with characteristic repetitive glycine stretches are ubiquitous in organisms of all Kingdoms and have distinct functions. It is believed that Gly-rich domains serve mainly for interactions with other proteins. Previously, we identified the tobacco UP30 gene as strongly upregulated by sulfur deficiency. It encodes a protein highly similar to cdiGRP which affects tobacco defense response by elevating cell wall callose deposits thus blocking systemic movement of viruses. The closest *Arabidopsis thaliana* homologue of UP30 is GRP-3 (At2g05520). Here we report that GRP-3 is induced in Arabidopsis seedlings in both sulfur and nitrogen deficiency conditions. The transgenic Arabidopsis plants with changed GRP-3 expression (either overexpressing or with silenced GRP-3) tend to have longer roots than the wild type, especially in the conditions of sulfur deficiency. The effect could be alleviated by the addition of auxin to the media. Moreover, we observed the increased callose deposition in both Arabidopsis lines suggesting its negative effects on shoot-to-root movement of auxins in nutrient deficient conditions.

Glycine-rich proteins (GRPs) are characterized by the existence of several domains with little sequence conservation but highly enriched in glycine. Although the first genes encoding GRPs have been isolated from plants, proteins with characteristic repetitive glycine stretches exist in diverse organisms from cyanobacteria to animals (reviewed in Sachetto-Martins et al. 2000). The first reports described plant GRPs as cell wall associated proteins (Showalter 1993), however since then many other GRPs with different domain organizations, sub-cellular localizations as well as tissue specificity of expression were demonstrated. This clearly indicated that these proteins must be implicated in several independent physiological processes, though the specific functions of several so far characterized GRPs remain

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speculative. The diversity of GRPs led even to the concept that they should rather not be considered as a family but as a wide group of proteins sharing a common motif (Sachetto-Martins et al. 2000). The presence of Gly-rich highly flexible domains may act as a molecular glue in the protein-protein interactions. It is therefore quite probable that GRPs are components of different multimolecular complexes, including homo- or hetero-dimerization between family members (Sachetto-Martins et al. 2000).

The *UP30* gene encoding protein belonging to GRP family has been identified as one of the strongly upregulated genes in the leaves of tobacco plants transferred for 2 days to the sulfur deficient medium (Wawrzynska et al. 2005). UP30 is a small protein containing Gly-rich, Tyr-rich and Pro-rich domains and a characteristic Cys-rich motif [-C-X3-CC-X6-C-X2-CC-] (Lewandowska et al. 2005). Subsequent experiments indicated that the protein fusion of UP30 and YFP is located in the cell wall (Fig.1a). The yeast two hybrid system allowed us to exclude the UP30-UP30 homodimerization (Fig. 1b); the interactions with other members of the GRP family were not checked.

UP30 belongs to Class II from the eight classes that the plant GRPs has been divided into based on the domain composition (Mangeon et al. 2010). Class II members have (beside optional signal peptide at the N-terminus) the Gly-rich and Cys-rich domains. Previously identified tobacco protein from the same class (cdiGRP) was proposed to be involved in plant defense. The gene encoding this protein is induced by cadmium and the elevated level of cdiGRP has been suggested to block the systemic spread of the turning vein-clearing tobamovirus (Ueki and Citovsky 2002) due to increasing the callose deposition in the plant cell wall (Ueki and Citovsky 2005). Interestingly, UP30 shares 81% amino acid sequence identity with cdiGRP allowing us to assume both proteins are functionally similar. The closest *Arabidopsis thaliana* homologue of UP30 is GRP-3. Both proteins have very similar domains/regions organization despite relatively low sequence conservation (only 29%; Fig. 2).

The *GRP-3* (At2g05520) gene was previously shown to be expressed in stems and leaves and only poorly in the roots, immature seed pods and flowers (de Oliveira et al. 1990). Such pattern of expression resembles the expression of UP30; the *UP30* mRNA was detected mostly in the leaves (Lewandowska et al. 2005). Likewise, the localization of GRP-3 in the apoplast (Gramegna et al. 2016) was an additional suggestion for the functional homology between UP30 and GRP-3. The GRP-3 protein interacts through its Cys-rich C-terminal motif with the cell wall receptor protein kinase WAK1. The binding of GRP-3 to WAK1 is critical for the structural integrity of the multimeric complex representing the activated signalosome of pathogen response (Park et al. 2001). On the other hand, GRP-3 acts as negative regulator of both elicitor-activated signaling cascade and response to wounding. It was proposed that GRP-3 may function in restoring the plant immune system to the baseline at the appropriate time after the pathogen attack (Gramegna et al. 2016).

To assay the function of GRP-3 during sulfur deficiency we have constructed the transgenic Arabidopsis plants constitutively overexpressing *GRP-3* (OX9 lines) and



Fig. 1 UP30 protein localization in the cell wall of leaf epidermal cells when transiently expressed in N. *benthamiana* (a) and does not form dimers in the yeast two hybrid test (b)

UP30 AtGRP3	MGSKAFLFLGLCLAVFFLISSEVVAALLAETSNSNKSDNENEVH- MASKALVLLGLFAVLLVVSEVAASSATVNSESKETVKPDQRGYGDNOSNYNNO *.**::::*** *:::: ***.** :::***.	- VE GRECNNGVG	
	Signal peptide	Gly-rich region	Cys-rich region [-C-X <sub>3</sub> -C-C-X <sub>6</sub> -C-X <sub>2</sub> -C-C-]

Fig. 2 Amino acid alignment of tobacco UP30 and GRP-3. Characteristic domains are marked

plants with silenced *GRP-3* due to constitutive expression in the antisense orientation (KD10 lines). The level of *GRP-3* mRNA in the transgenic plants was confirmed with semiquantitative RT-PCR. Additionally, the expression level was monitored in plants grown in S- or N-deficient conditions. The level of *GRP-3* transcript was induced in S- and N- conditions in the wild-type seedlings, it was high in the OX9 line and low (or undetected) in KD10 line (Fig. 3).

Interestingly, monitoring the growth of the seedlings in sulfur- deficient (S-), and sufficient (ctrl) media indicated that both overexpressors as well as knock-downs of *GRP-3* have longer roots that the wild-type, especially in the conditions of sulfur deficiency (Fig. 4).

Tobacco cdiGRP represents one of the factors that regulates callose accumulation in the plant vasculature (Ueki and Citovsky 2005). Callose is a 1,3- $\beta$ -d-glucan, synthesized by callose synthase and degraded by 1,3- $\beta$ -d-glucanase, which has been shown to accumulate around the neck region of the plasmodesmata. It is supposed to regulate the traffic through these channels and cell-to-cell communication. Therefore, in the next step we verified the level of callose in the *GRP-3* transformants. Surprisingly, higher level of callose was observed in the tissue of both OX9 and KD10 plants as compared to the wild type (Fig. 5a). It was especially significant for the stem and root vascular tissue, especially at the plasmodesmata region (Fig. 5b).



Fig. 3 The level of *GRP-3* expression in 2-week-old Arabidopsis seedlings grown in  $0.5 \times$  Hoagland optimal (ctrl), or sulfur-deficient (S-) or nitrogen-deficient (N-) medium assayed with semi-quantitative RT-PCR. Tubuline (*TUA3*) was used as quantity control



Fig. 4 The length [in mm] of the roots of the seedlings grown for 2 weeks in  $0.5 \times$  Hoagland medium, either optimal (ctrl) or sulfur deficient (S-)

The plasmodesmata is a specific channel structure in plants that spans the cell wall connecting neighboring cells to allow cytoplasmic exchange and communication. The plasmodesmata contribution to auxins movement, their role as potential sites of receptor signaling and the impact of callose deposition in plasmodesmata on auxins function has been recently discovered (Jackson 2015). Auxins reduce the growth of roots by two types of mechanisms, either by reducing the extend of root growth (IAA, NAA) or the cell production rate (2,4-D) (Rahman et al. 2007). Therefore, we assumed that the OX9 and KD10 transformants have longer roots than the wild type in nutrient deficient conditions because the shoot-to-root movement of auxins in such plants is negatively affected due to extensive callose deposition in their plasmodesmata (Fig. 5). It has been previously shown that Arabidopsis plants grown in S- conditions have lower level of auxins, as indicated by the activity of the auxin response marker DR5::GUS in the roots (Dan et al. 2007). To check this we decided to compare the effect of NAA on the root elongation of the seedlings grown in S- conditions (Fig. 6). In this experiment the OX9 lines were used as plants overexpressing GRP-3 and three T-DNA insertion SALK lines (SALK\_084781, SALK\_084877 and SALK\_012941C) were used as KO grp-3 mutants. These lines had no detectible transcript corresponding to GRP-3 (not shown). The results shown in Fig. 6 indicated that addition of NAA to the S-deficient media negatively affected root length in all lines; however, the


Fig. 5 Callose deposition in the 2-weeks-old Arabidopsis seedlings grown in  $0.5 \times$  Hoagland optimal medium (wt, OX9–2-4 and KD10–1-9 line was used). The callose was stained with aniline blue for 1 h and next the tissues were observed under fluorescent confocal microscopy with 365 nm filter (1:40 magnification)



Fig. 6 Two-weeks-old Arabidopsis seedlings grown for 2 weeks in sulfur deficient  $0.5 \times$  Hoagland medium (S-) with or without the addition of 0.1  $\mu$ M NAA

difference in root lengths observed between the wild-type seedlings and the OX9 transformants and *grp3* insertional mutants was strongly reduced in the presence of auxin.

In conclusion, we have demonstrated that the GRP-3 protein encoded by the sulfur deficiency induced gene modulates the root length of the Arabidopsis seed-lings in nutrient deficient conditions. The longer roots were observed in both, seedlings with increased level of *GRP-3* as well as with the reduced or abolished expression of *GRP-3*. An unbalanced level of GRP-3 results in increased callose deposition in plasmodesmata. Our findings are supported by the reports from other laboratories, where on one hand, the elevated expression of cdiGRP in tobacco resulted in callose accumulation (Ueki and Citovsky 2002, 2005), while on the other hand, the Arabidopsis *grp-3* mutants were reported to have higher level of callose than the wild-type in response to different elicitors (Gramegna et al. 2016). Results of the experiments presented in this work prompt us to hypothesize that GRP-3 is involved in the regulation of root growth in response to nutrient status of the plants through influencing auxin movement due to affecting callose deposits in plasmodesmata.

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