

Chapter 10

Constructed Wetlands: Role in Phytoremediation of Heavy Metals

Syed Shakeel Ahmad, Zafar A. Reshi, Manzoor A. Shah, and Irfan Rashid

10.1 Constructed Wetlands: Importance and Types

Constructed wetlands (CWs) are engineered systems that are designed and constructed to utilize the natural processes, involving wetland vegetation, soils, and their associated microbial assemblages to assist in treating wastewater [1]. According to the Interstate Technology Regulatory Council Wetlands Team, USA (ITRC) [2], constructed wetlands (CWs) are “*engineered systems, designed and constructed to utilize the natural functions of wetland vegetation, soils and their microbial populations to treat contaminants in surface water, groundwater or waste streams.*” Synonymous terms for constructed wetlands include man-made, engineered, and artificial wetlands. The first full-scale constructed wetland (CW) for wastewater treatment was built at Petrov near Prague in May 1989. Constructed wetlands are a cost-effective and technically feasible approach to treating wastewater and runoff. The constructed wetland provides a natural environment of warm climate, high water table, and high organic matter for microbes to break down contaminants [3].

The use of constructed wetlands for wastewater treatment is becoming more and more popular in many parts of the world. Today, subsurface flow constructed wetlands are quite common in many developed countries, such as Germany, the UK, France, Denmark, Austria, Poland, and Italy [4]. Constructed wetlands are also appropriate for developing countries but due to lack of awareness their use is not widespread [5–7]. Constructed wetlands can be less expensive to build than other treatment options. Operation and maintenance expenses (energy and supplies) are low and require only periodic, rather than continuous monitoring. Constructed wetlands are primarily used to treat domestic municipal wastewaters, but their use for other types of wastewaters such as agricultural and industrial

S.S. Ahmad (✉) • Z.A. Reshi • M.A. Shah • I. Rashid
Department of Botany, University of Kashmir, Srinagar, J&K 190 006, India
e-mail: ssahmad900@gmail.com

wastewaters, various runoff waters, and landfill leachate have become more frequent [8–10].

Mitsch [11] suggests the following guidelines for creating successful constructed wetlands:

1. Keep the design simple. Complex technological approaches often invite failure.
2. Design for minimal maintenance.
3. Design the system to use natural energies, such as gravity flow.
4. Design for the extremes of weather and climate.
5. Design the wetland with the landscape, not against it.
6. Integrate the design with the natural topography of the site.
7. Avoid over-engineering the design with rectangular basins, rigid structures and channels, and regular morphology. Mimic natural systems.

10.1.1 Types of Constructed Wetlands

The classification of constructed wetlands is based on various factors, such as the vegetation type, hydrology, and flow of direction (vertical or horizontal) [12]. There are mainly three types of constructed wetlands: surface flow wetlands, subsurface flow wetlands, and hybrid systems.

10.1.1.1 Surface Flow Wetlands

In case of surface flow wetlands, water level is above the ground surface; vegetation is rooted and emerges above the water surface; water flow is primarily above ground. The different macrophytes that are used in this type of constructed wetlands include *Phragmites australis*, *Typha unguistifolia*, *Sparganium erectum* etc.

10.1.1.2 Free Floating Macrophyte-Based Wetlands

In this type of constructed wetlands, floating macrophytes are used. The main floating macrophytes used in these systems are *Azolla cristata*, *Salvinia natans*, water hyacinth *Eichhornia crassipes* and duckweeds [13]. The different submerged species used in these constructed wetlands prevent entry of light into the system thereby inhibiting the growth of different algal groups. Hence the macrophytes need to be periodically removed from the wetland. The floating plant mat blocks out sunlight, thereby preventing photosynthesis and inhibiting algae growth hence the macrophytes need to be periodically removed from the system [14]. Duckweeds are extremely invasive and grow in most environments [15].

10.1.1.3 Submerged Macrophyte-Based Wetlands

In these constructed wetlands there are used different kinds of submerged macrophytes. The main submerged species used in these constructed wetlands include *Ceratophyllum demersum*, *Hydrilla* sp., *Potamogeton* sp. etc. They have been proposed as final polishing steps following primary and secondary treatment [16].

10.1.1.4 Emergent Macrophyte-Based Wetlands

In this type of constructed wetlands, emergent macrophytes are used. Emergent macrophyte-based wetlands are the most common type of constructed wetlands. The different emergents used in these constructed wetlands include *Juncus effusus*, *Phragmites australis*, *Typha unguistifolia*, *Sparganium erectum* etc. [17, 18, 19]. A slow flow rate is applied so that a shallow depth is maintained [20].

10.1.1.5 Subsurface Flow Wetlands

They are also called vegetated submerged beds, or plant-rock filter systems. They have below ground water level. The basin mainly consists of sand or gravel. The different macrophytes used in this type of constructed wetlands include *Glyceria maxima*, *Iris pseudacorus*, *Phragmites australis*, *Typha unguistifolia*, *Sparganium erectum* etc.

10.1.1.6 Horizontal Flow Constructed Wetland (HF CWs)

This type of constructed wetland was developed in the 1950s in Germany by Käthe Seidel [12]. This type of constructed wetland consists of rock or gravel beds, impermeable layer and wetland vegetation. Waste water after entering through the inlet passes through the horizontal path before it is discharged through the outlet. Hence the name horizontal flow constructed wetland.

10.1.1.7 Vertical Flow Constructed Wetlands (VF CWs)

In this type of constructed wetland water percolated down through sand medium. Among different kinds of constructed wetlands this type of constructed wetland has very high operational costs.

10.1.1.8 Free Water Surface Constructed Wetland (FWS CW)

This type of constructed wetland consists of a series of impermeable basins about 20–40 cms deep. In this type of constructed wetland the main macrophytes planted include emergents like *Phragmites australis*, *Typha angustifolia*, *Sparganium erectum* etc.

10.1.1.9 Constructed Wetlands with Floating Leaved Macrophytes

Constructed wetlands with floating leaved macrophytes are very rare, and there are no guidelines to design, operate, and maintain these systems [1]. In these systems, the different plants used are *Nelumbo nucifera* [21, 22] and *Nuphur lutea* [23].

10.1.1.10 Hybrid Systems

Different types of constructed wetlands may be combined with each other in order to exploit the specific advantages of the different systems. In hybrid or multistage systems, different cells are designed for different types of reactions. During the 1990s, HF-VF and VF-HF hybrid systems were introduced [24]. Hybrid systems are used especially when removal of ammonia-N and total-N is required [1].

10.1.2 How Constructed Wetlands Work

Constructed wetlands are made up of a series of ponds each designed to perform a particular function. Solids are allowed to settle in the primary storage ponds. The water then enters another pond containing vegetation. Here physical, chemical, and biological reactions reduce contaminants. Nitrogen and phosphorus are used by aquatic vegetation. Heavy metals are also removed by different plants which show tolerance for these metals. Water then enters the tertiary cell. It serves as a habitat for wildlife.

10.1.3 Costs for Creating Constructed Wetlands

The main requirements for establishing constructed wetlands include land, design, vegetation, hydraulic control system and fencing. The total investment costs for establishing constructed wetlands vary from country to country and could be as low as 29 USD per m² in India [25] or 33 USD per m² in Costa Rica [26], or as high as 257 EUR per m² in Belgium [27].

For the development of constructed wetlands, the basic requirements are containers, plant species, sand, and gravel media in certain ratio. In different types of constructed wetlands microbes and other invertebrates develop naturally [28]. The three types of macrophytes that are used in constructed wetlands are floating macrophyte (i.e., *Azolla* sp., *Salvinia natans*, *Lemna* spp. or *Eichhornia crassipes*), submerged macrophyte (i.e., *Ceratophyllum demersum*, *Potamogeton* sp. *Elodea canadensis*) and rooted emergent macrophyte (i.e., *Phragmites australis*, *Typha* spp. *Sparganium erectum* [29]). Plants (free-floating, emergent or submergent vegetation) are the part of constructed ecosystem to remediate contaminants from municipal, industrial wastewater, metals, and acid mine drainage [30]. The different macrophytes that are used in subsurface flow CWs in warm climates are Papyrus sedge (*Cyperus papyrus*), Umbrella sedge (*Cyperus albostrigatus* and *Cyperus alternifolius*), Dwarf papyrus (*Cyperus haspensis*), Bamboo, smaller ornamental species, Broad-leaved cattail (*Typha latifolia*), Species of genus—*Heliconia*: lobster-claws, wild plantains—*Canna*: Canna lily—*Zantedeschia*: Calla lily Napier grass or Elephant grass (*Pennisetum purpureum*).

10.2 Heavy Metals: Sources and Impacts

Constructed wetlands are designed for the removal of different kinds of pollutants including heavy metals from the wastewater. Heavy metals released from different sources enter into the water bodies and pose serious threats to different trophic levels of the food chain including human beings. Heavy metals are metals having a density of 5 g/cm³ [31]. Heavy metals include a category of 53 elements with specific weight higher than 5 g/cm³ [32, 33]. Heavy metals are elements with metallic properties and an atomic number >20. Heavy metals mainly include the **transition metals**, some metalloids, lanthanides, and actinides. The most common heavy metal contaminants are As, Cd, Cr, Cu, Hg, Pb, and Zn.

Over population, industrialization, rapid urbanization, overuse of pesticides, detergent and agricultural chemicals, liquid and solid waste products, and discharge of municipal wastes resulted in heavy metal pollution of natural water resources [34]. Man-made activities such as mining and smelting of metal ores, industrial, commercial, and domestic applications of insecticides and fertilizers have all contributed to elevated levels of heavy metals in the environment [35, 36].

The primary sources of metal pollution are the burning of fossil fuels, mining and smelting of metalliferous ores, municipal wastes, sewage, pesticides, and fertilizers [37] oil, gasoline and coal combustion, smelting, and refuse incineration [38]. In uncontaminated soil, the average concentrations of heavy metals vary in orders of magnitudes, but on average the concentrations are, e.g., Zn: 80 ppm, Cd: 0.1–0.5 ppm and Pb: 15 ppm. However, in polluted soil dramatically higher concentrations are found, e.g., Zn: >20,000 ppm, Cd: >14,000 ppm and Pb: >7000 ppm (<http://www.speclab.com/elements/>). Heavy metals are ubiquitous environmental pollutants that arise from a variety of industrial, commercial, and domestic activities

[36]. Increasing industrial activities have led to an increase in environmental pollution and the degradation of several aquatic ecosystems with the accumulation of metals in biota and flora [39]. According to Phuong et al. [40], most heavy metal contaminants originate from anthropogenic sources such as long-term discharge of untreated domestic and industrial wastewater runoff, accidental spills, and direct soil waste dumping. In addition, heavy metals can enter the water bodies through atmospheric sources [41] and nearby rice fields [42]. Due to innovations in mining and metal-working techniques during ancient times, the close relationship between metals, metal pollution, and human history was formed [43]. Energy intensive and chlor-alkali industries for the manufacture of agrochemicals deteriorate the water quality of lakes and reservoirs due to the discharge of various pollutants, especially a range of heavy metals [44]. Coal mining [45] and its allied/dependent industries (thermal power plants) are major sources of heavy metals in the industrial belts of developing countries such as India [44, 46]. Metals are natural components in soil [47]. Lead is a common pollutant from road runoff. Zinc is a common metal present in variable amounts and if found in appreciable amounts can be an indicator of industrial pollution. While copper is also an indicator of industrial contamination of urban waters [48, 49]. The different macrophytes have a potential to sequester heavy metals from the soils contaminated with these metals. Colonization of macrophytes on the sediments polluted with heavy metals and the role of these plants in transportation of metals in shallow coastal areas are very important [50].

Contamination of aquatic environment by heavy metals is a serious environmental problem, which threatens aquatic ecosystems, agriculture, and human health [51]. Accumulation of metals and their toxic effects through the food chain can lead to serious ecological and health problems [52]. Heavy metals are the most dangerous contaminants since they are persistent and accumulate in water, sediments and in tissues of the living organisms, through two mechanisms, namely “bioconcentration” (uptake from the ambient environment) and “biomagnification” (uptake through the food chain) [53]. Trace elements such as Cu, Fe, Mn, Ni, and Zn are essential for normal growth and development of plants. They are required in numerous enzyme catalyzed or redox reactions, in electron transfer and have structural function in nucleic acid metabolism [54]. Metals like Cd, Pb, Hg, and As are not essential [55]. High levels of Cd, Cu, Pb, and Fe can act as ecological toxins in aquatic and terrestrial ecosystems [56, 57]. Excess metal levels in surface water may pose a health risk to humans and to the environment [50]. Since HM are not biodegradable and may enter the food chain, they are a long-term threat to both the environment and human health [58]. Some of these metals are micronutrients necessary for plant growth, such as Zn, Cu, Mn, Ni, and Co, while others have unknown biological function, such as Cd, Pb, and Hg [59]. Metal pollution has harmful effect on biological systems and does not undergo biodegradation. Toxic heavy metals such as Pb, Co, and Cd can be differentiated from other pollutants, since they cannot be biodegraded but can be accumulated in living organisms, thus causing various diseases and disorders even in relatively lower concentrations [60]. Heavy metals, with soil residence times of thousands of years, pose numerous health dangers to higher organisms. They are also known to have effect to plant growth, ground cover,

and have a negative impact on soil microflora [61]. It is well known that heavy metals cannot be chemically degraded and need to be physically removed or be transformed into nontoxic compounds [59]. Table 10.1 shows the harmful effects of different heavy metals on living organisms.

10.3 Role of Constructed Wetlands in Phytoremediation

The use of wetlands for quality improvement of wastewater, referred to as rhizofiltration, is the best known and most researched application of constructed wetlands. Flooding of wetland sediments leads to rapid denitrification because of anoxic conditions; therefore, wetland soils contain low levels of nitrate [62]. CWs have proven successful for remediating a variety of water quality issues, with advantages over the natural wetland. Constructed wetland (CWs) thus designed to take advantage of natural wetland systems, but do so within a more controlled way. The plants most often used in CWs are persistent emergent plants, such as bulrushes (*Scirpus*), spikerush (*Eleocharis*), and other sedges (*Cyperus*), Rushes (*Juncus*), common reed (*Phragmites*), and cattails (*Typha*). Plants for CWs must be able to tolerate continuous flooding and exposure to waste streams containing relatively high and often variable concentrations of pollutants. The functions of wetland plants make them an important component of CWs. Plants contribute to contaminant removal by altering hydrology, sequestering particulates, and accumulating pollutants [63]. These processes can be utilized to design CWs with a number of treatment approaches, which are mainly phytoextraction, rhizofiltration, and phytostabilization.

Some other macrophytes that are used in wetlands for the removal of heavy metals include *Acorus calamus*, *Carex* spp. (sedges), *Cyperus* (sweet manna grass), *Juncus* sp. (Rushes), *Phalaris arundinacea* (reed canary grass), *Phragmites australis* (common reed), *Sagittaria* (arrow heads), *Scirpus* sp. (Bulrushes), *Sparganium* sp. (bur reeds), *Spartina* spp. (cordgrasses), *Typha* sp. (cattails), *Zizania aquatic* (wild rice), *Ceratophyllum* sp. (coontails), *Eggeria densa* (Brazilian waterweed), *Hydrilla verticillata* (Hydrilla), *Isoetes* sp. (Quillworts), *Myriophyllum* spp. (water milfoils), *Najas* spp. (water nymphs), *Potamogeton* sp. (pond weeds), *Urticularia* spp. (bladderworts), *Lemna* spp. (duckweed), *Azolla* (aquatic fern) and *Hydrocharis* (frog bit). These macrophytes are highly beneficial to aquatic ecosystems because they provide food and shelter for fish and aquatic invertebrates, wildlife also produce oxygen, which helps in overall lake functioning [50]. Macrophytes are considered as important components of the aquatic ecosystems not only as food source for aquatic invertebrates, but they also act as an efficient accumulator of heavy metals [64, 65]. Aquatic plants sequester large quantities of metals [66–68]. Trace element removal by wetland vegetation can be greatly enhanced by the judicious selection of appropriate wetland plant species. Selection is based on the type of elements to be removed, the geographical location, environmental conditions, and the known accumulation capacities of the species.

Table 10.1 Harmful effect of different heavy metals on living organisms

Heavy metal	Harmful effects	References
As	It interferes with oxidative phosphorylation and ATP synthesis	Tripathi et al. [79]
Cd	Inhaling Cd leads to respiratory and renal problems. It also interferes with calcium regulation in biological systems; causes chronic anemia. It is also carcinogenic, mutagenic, and teratogenic; endocrine disruptor	Salem et al. [80] and Awofolu [81]
Cr	It can result in gastritis, nephrotoxicity, and hepatotoxicity. Chromium toxicity causes hair loss	Salem et al. [80] and Paustenbach et al. [82]
Cu	Excessive free copper impairs zinc homeostasis, and vice versa, which in turn impairs antioxidant enzyme function, increasing oxidative stress. It causes brain and kidney damage, liver cirrhosis and chronic anemia, stomach and intestinal irritation	Salem et al. [80], Wuana and Okieimen [83], and Sandstead [84]
Hg	Anxiety, autoimmune diseases, depression, difficulty with balance, drowsiness, fatigue, hair loss, insomnia, irritability, memory loss, recurrent infections, restlessness, vision disturbances, tremors, temper outbursts, ulcers and damage to brain, kidney, and lungs. Toxic effects include damage to the brain, kidneys, and lungs. Mercury poisoning can result in several diseases, including acrodynia (pink disease), Hunter-Russell syndrome, and Minamata disease	Neustadt and Pieczenik [85], Ainza et al. [86], and Gulati et al. [87], Clifton [88], Bjørklund [89], Tokuomi [90], and Davidson [91]
Pb	Exposure to lead produces deleterious effects on the hematopoietic, renal, reproductive, and central nervous system, mainly through increased oxidation. Its poisoning causes problems in children such as impaired development, reduced intelligence, loss of short-term memory, learning disabilities, and coordination problems; causes renal failure; increased risk for development of cardiovascular disease	Flora et al. [92], Salem et al. [80], Padmavathiamma and Li [93], Wuana and Okieimen [83] and Iqbal [94]
Zn	Long-term excessive zinc intakes (ranging from 150 mg/day to 1–2 g/day) have included sideroblastic anemia, hypochromic microcytic anemia, leukopenia, lymphadenopathy, neutropenia, hypocupremia, and hypoferremia. Over dosage can cause dizziness, nausea, vomiting, epigastric pain, lethargy, and fatigue	Hess and Schmid [95] and Fosmire [96]
Mn	Neurological effects in humans and animals and causes disabling syndrome called <i>manganism</i> . It also causes lethargy, increased muscle tonus, tremor, and mental disturbances	USEPA [97] and Kawamura [98]

Macrophytes are unchangeable biological filters and play an important role in the maintenance of the aquatic ecosystem. Aquatic macrophytes are taxonomically closely related to terrestrial plants, but are aquatic phanerogams, which live in a completely different environment. Their characteristics to accumulate metals make them an interesting research objects for testing and modeling ecological theories on evolution and plant succession, as well as on nutrient and metal cycling [69]. Many industrial and mining processes cause heavy metal pollution, which can contaminate natural water systems and become a hazard to human health. Therefore, colonization of macrophytes on the sediments polluted with heavy metals and the role of these plants in transportation of metals in shallow coastal areas are very important. [50]. Despite this, roots of wetland plants may accumulate heavy metals and transport them to aboveground portions of plants [70, 71].

The extent of metal accumulation within aquatic macrophytes is known to vary significantly between species. For example, the emergent aquatic plants usually accumulate lower amount of metals than submerged aquatic vegetation [72]. The emergent macrophytes growing in constructed wetlands designed for wastewater treatment have several properties in relation to the treatment processes that make them an essential component of the design. Several of the submerged, emergent, and free-floating aquatic macrophytes are known to accumulate and bioconcentrate heavy metals [73, 74]. Aquatic macrophytes take up metals from the water, producing an internal concentration several fold greater than their surroundings. Many of the aquatic macrophytes are found to be the potential scavengers of heavy metals from water and wetlands [75]. Yet research has focused mainly on the interaction between biological factors such as competition, coexistence, grazing, life cycles, adaptation, and environmental factors (salinity, depth, wave exposure) of importance for structuring brackish water macrophytes and algal communities [76].

10.4 Heavy Metal Pollution in Kashmir Himalayan Wetlands

Though a number of studies pertaining to the ecology of Kashmir Himalayan wetlands have been carried out, there are only a few attempts related to heavy metal analysis in these ecosystems. Of these studies, worth mentioning are the attempts by Ahmad et al. [77] in recent past. According to Ahmad et al. [77], the main source of heavy metals in the Kashmir Himalayan wetlands is use of pesticides in the rice fields and orchards of Kashmir and use of lead shots for hunting/poaching of birds.

In a series of studies, Ahmad et al. [77, 78] and other unpublished data heavy metal dynamics in different components of the wetland systems including water, sediments, and macrophytes have been worked out. In *Phragmites australis*, the accumulation of the different heavy metals was in order of Al>Mn>Ba>Zn>Cu>Pb>Mo>Co>Cr>Cd>Ni. Translocation factor, i.e., ratio of shoot to root metal concentration revealed that metals were largely retained in the roots of *P. australis*, thus reducing the supply of metals to avifauna and preventing their bioaccumulation. Moreover, the higher retention of heavy metals in the belowground parts of *P.*

australis reduces the supply of metals to avifauna, which mainly feed on above-ground parts of the plant, thereby preventing bioaccumulation of heavy metals in higher trophic levels. This further adds to the desirability of *P. Australis* as a phyto-remediation species [77]. Ahmad et al. [78] also assessed the heavy metal accumulation capability of two dominant species (*Ceratophyllum demersum* and *Potamogeton natans*) in a Kashmir Himalayan Ramsar site. The accumulation of the different metals in *P. natans* was in the order of Al>Mn>Pb>Cu>Zn>Ni>Co>Cr>Cd, while in *C. demersum* it was Al>Mn>Zn>Co>Cu>Pb>Cr>Ni>Cd. In *C. demersum*, the highest bioconcentration factor (BCF) was obtained for Co (3616) and Mn (3589) while in *P. natans* the highest BCF corresponded to Cd (1027). Overall *Potamogeton*–*Ceratophyllum* combination provided a useful mix for Co, Mn, and Cd removal from contaminated sites. Beside *Phragmites australis* some other macrophytes that showed good phyto-remediation potential were *Azolla cristata*, *Hydrocharis dubia*, *Myriophyllum spicatum*, *Nymphaea alba*, *Nymphoides peltata*, *Salvinia natans*, *Typha angustata*, *Sparganium erectum*, and *Trapa natans*. Ahmad et al. [77] reported that Hokersar an important Ramsar site of Kashmir Himalayas filters 73 % of Co, 88.24 % of Cu, 65.13 % of Pb, 51.98 % of Zn, 40.93 % of Mn, 58.36 % of Fe, 41.02 % of Cd, 75.07 % of Cr, and 86.59 % of Ni.

10.5 Knowledge Gaps and Future Directions

Kashmir Himalayas are gifted with a number of wetlands like Hokersar wetland, Haigam wetland, Malangpora wetland, Mirgund wetland, Narkura wetland, etc. These Kashmir Himalayan wetlands are presently subjected to various anthropogenic pressures like encroachment, rapid urbanization and industrialization, dumping of solid waste, sites of gunshots for hunting/poaching, etc. There have been scanty studies of heavy metals in Kashmir Himalayan wetlands except a few attempts in recent past by Ahmad et al. [77, 78]. Constructed wetlands are not common in Kashmir Himalayas. Realizing the important role played by the constructed wetlands, it is expected that constructed wetlands will also become popular in Kashmir Himalayas.

References

1. Vymazal J, Kropfelova L (2008) Is concentration of dissolved oxygen a good indicator of processes in filtration beds of horizontal-flow constructed wetlands? In: Vymazal J (ed) Waste water treatment, plant dynamics and management. Springer, Dordrecht, pp 311–317
2. ITRC (2003) Technical and regulatory guidance document for constructed treatment wetlands. The Interstate Technology Regulatory Council Wetlands Team, Washington, DC
3. TOOLBASE (2001) Constructed wetlands for wastewater treatment. <http://www.toolbase.org/Technology-Inventory/Sitework/constructed-wetlands>. Accessed 30 Jan 2012

4. Hoffmann H, Platzer C, Winker M, Muench EV (2011) Technology review of constructed wetlands: subsurface flow constructed wetlands for greywater and domestic wastewater treatment. Deutsche Gesellschaft für Internationale Zusammenarbeit (GIZ) GmbH, Germany
5. Mohamed A (2004) Planung, Bau und Betrieb einer Pflanzenkläranlage in Syrien (Planning, construction and operation of a constructed wetland in Syria, in German). PhD thesis, University Flensburg, Germany
6. Heers M (2006) Constructed wetlands under different geographic conditions: evaluation of the suitability and criteria for the choice of plants including productive species. Master thesis, Faculty of Life Sciences, Hamburg University of Applied Sciences, Germany
7. Kamau C (2009) Constructed wetlands: potential for their use in treatment of grey water in Kenya. M.Sc. thesis, Christian-Albrechts University, Kiel, Germany
8. Kadlec RH, Knight RL (1996) Treatment wetlands. CRC Press, Boca Raton
9. Kadlec RH, Wallace SD (2008) Treatment wetlands, 2nd edn. CRC Press, Boca Raton
10. Vymazal J, Brix H, Cooper PF, Green MB, Haberl R (eds) (1998) Constructed wetlands for wastewater treatment in Europe. Backhuys Publishers, Leiden
11. Mitsch WJ (1992) Landscape design and the role of created, restored and natural riparian wetlands in controlling nonpoint source pollution. *Ecol Eng* 1:27–47
12. Vymazal J (2010) Constructed wetlands for wastewater treatment. *Water* 2:530–549
13. USEPA (1988) Constructed wetlands and aquatic plant systems for municipal wastewater treatment. EPA/625/1-88/022
14. Lemna Corporation (1994) Innovations in Lagoon-based treatment. Retention Times, Spring/Summer (Promotional literature)
15. USEPA (2000) Manual, Constructed wetlands treatment of municipal wastewaters. EPA/625/R-99/010. United States Environment Protection Agency. Office of Research and Development, Cincinnati
16. Brix H (1994) Use of wetlands in water pollution control: historical development, present status, and future perspectives. *J Water Sci Technol* 30(8):209–223
17. Stottmeister U, Weibner A, Kusch P (2003) Effect of plants and microorganisms on constructed wetlands for waste water treatment. *Biotechnol Adv* 22(1–2):93–117
18. Brix H (2003) Plants used in constructed wetlands and their functions. 11th international seminar on the use of aquatic macrophytes for wastewater treatment in constructed wetlands
19. Shtemenko NI, Shepelenko VN, Richnow H, Kusch P (2005) Surface lipid composition of two emergent water plants used in constructed wetlands, vol 48, Modern tools and methods of water treatment for improving living standards NATO science series. Springer, The Netherlands, pp 325–330
20. Hamilton H, Nix PG, Sobolewski Y (1993) An overview of constructed wetlands as alternatives to conventional waste treatment systems. *Water Pollut Res J Canada* 28(3):529–548
21. Wang J, Cai X, Chen Y, Liang M, Zhang Y, Wang Z, Li Q, Liao X (1994) Analysis of the configuration and treatment effect of constructed wetland waste water treatment system for different waste waters in China. In: Proceedings of fourth international conference on wetland systems for water pollution control. ICWS (\$ Secreteriat Guangzhou, P. R. China, pp 114–120)
22. Yang Z, Goldman N, Friday A (1994) Comparison of models for nucleotide substitution used in maximum likelihood phylogenetic estimation. *Mol Biol Evol* 11:316–324
23. Twilley RR, Brindson MH, Davis GJ (1977) Phosphorus absorption, translocation and secretion in *Nuphar luteum*. *Limnol Oceanogr* 22:1022–1032
24. Vymazal J (2013) The use of hybrid constructed wetlands for wastewater treatment with special attention to nitrogen removal: a review of a recent development. *Water Res* 47(14):4795–4811
25. Billore SK, Singh N, Sharma JK, Dass P, Nelson RM (1999) Horizontal subsurface flow gravel bed constructed wetland with *Phragmites karka* in central India. *Water Sci Technol* 40:171–173
26. Dallas S, Scheffe B, Ho G (2004) Reedbeds for greywater treatment—case study in Santa Elena—Monteverde, Costa Rica, Central America. *Ecol Eng* 23:55–61

27. Rousseau DPL, Vanrolleghem PA, De Pauw N (2004) Constructed wetlands in Flanders: a performance analysis. *Ecol Eng* 23:151–163
28. American Public Health Association (APHA) (1998) Standard methods for examination of water and wastewater. 20th edn. American Public Health Association, Washington, DC
29. Masi F, Martinuzzi N (2007) Constructed wetlands for the Mediterranean countries: hybrid systems for water reuse and sustainable sanitation. *Desalination* 215:44–55
30. Schroder P, Avino JN, Azaizeh H, Goldhirsh AG, Gregorio SD, Komives T, Langergraber G, Lenz A, Maestri E, Memon AR, Ranalli A, Sebastiani L, Smrcek S, Vanek T, Vuilleumier S, Wissing F (2007) Using phytoremediation technologies to upgrade waste water treatment in Europe. *Environ Sci Pollut Res* 14(7):495
31. Nies DH (1999) Microbial heavy-metal resistance. *Appl Microbiol Biotechnol* 51:730–750
32. Holleman D, Wiberg E (1987) *Lehrbuch der anorganischen chemie*. Springer, Berlin
33. Weast RC (1984) *CRC Handbook of chemistry and physics*, 64th edn. CRC Press, Boca Raton
34. Baruah S, Hazarika K, Sarma PK (2011) Uptake and localization of lead in *Eichorniacrassipes* grown within a hydroponic system. *Adv Appl Sci Res* 3(1):51–59
35. Alloway BJ (1997) The mobilisation of trace elements in soils. In: Prost R (ed) Contaminated soils. Proceedings of the third international conference on the biogeochemistry of trace elements, Paris 15–19 May 1995. INRA, Paris, pp 133–145
36. Gorsberg RM, Elkins BV, Igon SR, Goldman CR (1984) The removal of heavy metals by artificial wetlands. In: Proceedings of water reuse symposium III, future of water reuse, San Diego, vol. 2, pp 639–648
37. Kabata-Pendias A, Pendias H (2001) Trace elements in soils and plants. CRC Press, Boca Raton
38. Nriagu JO (1989) A global assessment of natural sources of atmospheric trace metals. *Nature* 338:47–49
39. Espinoza-Quiñones FR, Zacarkim CE, Palacio SM, Obregón CLD, Zenatti C, Galante RM, Rossi N, Rossi FL, Pereira IRA, Welter RA, Rizzutto MA (2005) Removal of heavy metal from polluted river water using aquatic macrophyte *Salvinia* sp. *Brazilian J Phys* 35:44–746
40. Phuong PK, Son CPN, Sauvian JJ, Terradellas J (1998) Contamination of PCB's, DDT', and heavy metals in sediments of Ho Chi Minh city's canals, Vietnam. *Bull Environ Contam Toxicol* 60:115–131
41. Batty J, Pain D, Caurant F (1996) Metal concentrations in eels 471 *Anguilla Anguilla* from the Camargue region of France. *Biol Conserv* 76:17–23
42. Tavecchia G, Pradel R, Lebreton JD, Johnson AR, Mandain-Monval JY (2001) The effect of lead exposure on survival of adult mallards in the Camargue, southern France. *J Appl Ecol* 38:1197–1207
43. Nriagu JO (1996) A history of global metal pollution. *Science* 272:223–224
44. Rai PK, Sharma AP, Tripathi BD (2007) Urban environment status in Singrauli industrial region and its eco-sustainable management: a case study on heavy metal pollution. In: Lakshmi V (ed) Urban planning and environment, strategies and challenges. Macmillan Advanced Research Series, pp 213–217
45. Finkelman RB, Gross PMK (1999) The types of data needed for assessing the environmental and human health impacts of coal. *Int J Coal Geol* 40:91–101
46. Sharma PD (2003) Environmental pollution. In: Ecology and environment, 7th edn. Rastogi Publication, Meerut, India, pp 415–489
47. Lasat MM, Pence NS, Garvin DF, Ebbs SD, Kochian LV (2000) Molecular physiology of zinc transport in the Zn hyperaccumulator (*Thlaspi caerulescens*). *J Exp Bot* 51:71–79
48. Cardwell AJ, Hawker DW, Greenway M (2002) Metal accumulation in aquatic macrophytes from southeast Queensland, Australia. *Chemosphere* 48:653–663
49. Uka UN, Mohammed HA, Aina A (2013) Preliminary studies on the phytoremediation potential of *Phragmites karka* (Retz.) in Asa River. *J Fisheries Aquatic Sci* 8:87–93
50. Vardanyan LG, Ingol BS (2006) Studies on heavy metal accumulation in aquatic macrophytes from Sevan (Armenia) and Carambolim (India) lake systems. *Environ Int* 32:208–218

51. Sasmaza A, Obekb E, Hasarb H (2008) The accumulation of heavy metals in (*Typhalatifolia* L.) grown in a stream carrying secondary effluent. *Ecol Eng* 33:278–284
52. Malik A (2004) Metal bioremediation through growing cells. *Chemosphere* 30:261–278
53. Chaphekar SB (1991) An overview on bioindicators. *J Environ Biol* 12:163–168
54. Zenk MH (1996) Heavy metal detoxification in higher plants—a review. *Gene* 179:21–30
55. Mertz W (1981) The essential trace elements. *Science* 213:1332–1338
56. Balsberg-Pahlsson AM (1989) Toxicity of heavy metals (Zn, Cu, Cd, Pb) to vascular plants. A literature review. *Water Air Soil Pollut* 47:287–319
57. Gullizzoni P (1991) The role of heavy metals and toxic materials in the physiological ecology of submerged macrophytes. *Aquatic Bot* 41:87–109
58. Jarup L (2003) Hazards of heavy metal contamination. *Br Med Bull* 68:167–182
59. Gaur A, Adholeya A (2004) Prospects of arbuscular mycorrhizal fungi in phytoremediation of heavy metal contaminated soils. *Curr Sci* 86(4):528–534
60. Pehlivan E, Ozkan AM, Dinc S, Parlayici S (2009) Adsorption of Cu²⁺ and Pb²⁺ ion on dolomite powder. *J Hazard Mater* 167(1–3):1044–1049
61. Roy S, Labelle S, Mehta P (2005) Phytoremediation of heavy metal and PAH-contaminated brownfield sites. *Plant and Soil* 272(1–2):277–290
62. Otte ML, Jacob DL (2006) Constructed wetlands for phytoremediation. In: Mackova M (ed) *Phytoremediation, rhizoremediation*. Springer, Dordrecht, pp 57–67
63. Kadlec RH, Wallace SD (2009) *Treatment wetlands*. CRC Press, Boca Raton
64. Chung IH, Jeng SS (1974) Heavy metal pollution of Ta-Tu river. *Bullet Inst Zool, Acad Sci* 13:69–73
65. Devlin RMR (1967) *Plant physiology*. Reinhold, New York, p 564
66. Baldantoni D, Alfani A, Tomansi DI, Bartoli G, Santo AVD (2004) Assessment of macro and microelement accumulation capability of two aquatic plants. *Environ Pollut* 130:149–156
67. Mays PA, Edwards GS (2001) Comparison of heavy metal accumulation in natural wetland and constructed wetlands receiving acid mine drainage. *Ecol Eng* 16(4):487–500
68. Stoltz E, Greger M (2002) Accumulation properties of As, Cd, Cu, Pb and Zn by four wetland plant species growing on submerged mine tailings. *Environ Exp Bot* 47:271–280
69. Forstner U, Whittman GTW (1979) *Metal pollution in the aquatic environment*. Springer, Berlin, p 486
70. Dunstan WM, Windom HM (1975) The influence of environmental changes in heavy metal concentrations of *Spartina alterniflora*. In: Cronin LE (ed) *Estuarine research, vol. II. Geology and engineering*. Academic Press, New York, pp 393–404
71. Mudroch A, Capobianco J (1978) Study of selected metals in marshes on Lake St. Clair. *Ontario Arch Hydrobiol* 84:87–108
72. Albers PH, Camardese WN (1993) Effects of acidification on metal accumulation by aquatic plants and invertebrates in Constructed wetlands. *Environ Toxicol Chem* 12:959–967
73. Bryan GW (1971) The effects of heavy metals (other than mercury) on marine and estuarine organisms. *Proc R Soc London B* 177:389–410
74. Chow TJ, Snyder CB, Snyder HG, Earl JL (1976) Lead content of some marine organisms. *J Environ Sci Health A* 11:33–44
75. Gulati KL, Nagpaul KK, Bukhari SS (1979) Uranium, boron, nitrogen, phosphorus and potassium in leaves of mangroves, Mahasagar. *Bullet Natl Inst Oceanogr* 12:183–186
76. Alloway BJ, Davis BE (1971) Heavy metal content of plants growing on soils contaminated by lead mining. *J Agric Sci* 76:321–323
77. Ahmad SS, Reshi ZA, Shah MA, Rashid I, Ara R, Andrabi SMA (2014) Phytoremediation potential of *Phragmites australis* in Hokersar wetland—a Ramsar site of Kashmir Himalaya. *Int J Phytoremed* 16:1183–1191
78. Ahmad SS, Reshi ZA, Shah MA, Rashid I, Ara R, Andrabi SMA (2015) Heavy metal accumulation in the leaves of *Potamogeton natans* and *Ceratophyllum demersum* in a Himalayan RAMSAR site: management implications. *Wetland Ecol Manage* doi:10.1007/s11273-015-9472-9

79. Tripathi RD, Srivastava S, Mishra S, Singh N, Tuli R, Gupta DK, Maathuis FJM (2007) Arsenic hazards: strategies for tolerance and remediation by plants. *Trends Biotechnol* 25:158–165
80. Salem HM, Eweida EA, Farag A (2000) Heavy metals in drinking water and their environmental impact on human health. ICEHM 2000, Cairo University, Egypt, pp 542–556
81. Awofolu O (2005) A survey of trace metals in vegetation, soil and lower animal along some selected major roads in metropolitan city of Lagos. *Environ Monit Assess* 105:431–447
82. Paustenbach DJ, Finley BL, Mowat FS, Kerger BD (2003) Human health risk and exposure assessment of chromium (VI) in tap water. *J Toxicol Environ Health* 66:1295–1339
83. Wuana RA, Okieimen FE (2011) Heavy metals in contaminated soils: a review of sources, chemistry, risks and best available strategies for remediation. *ISRN Ecol*, pp 1–20
84. Sandstead HH (1995) Requirements and toxicity of essential trace elements, illustrated by zinc and copper. *Am J Clin Nutr* 61(3):621–624
85. Neustadt J, Pieczenik S (2007) Toxic-metal contamination: mercury. *Integr Med* 6:36–37
86. Ainza C, Trevors J, Saier M (2010) Environmental mercury rising. *Water Air Soil Pollut* 205:47–48
87. Gulati K, Banerjee B, Bala Lall S, Ray A (2010) Effects of diesel exhaust, heavy metals and pesticides on various organ systems: possible mechanisms and strategies for prevention and treatment. *Indian J Exp Biol* 48:710–721
88. Clifton JC (2007) Mercury exposure and public health. *Pediatr Clin North Am* 54(2): 237–269
89. Bjørklund G (1995) Mercury and Acrodynia (PDF). *J Orthomol Med* 10(3&4):145–146
90. Tokuomi H, Kinoshita Y, Teramoto J, Imanishi K (1977) Hunter-Russell syndrome Nihon rinsho. *Jpn J Clin Med* 35(1):518–519
91. Davidson PW, Myers GJ, Weiss B (2004) Mercury exposure and child development outcomes. *Pediatrics* 113(4):1023–1029
92. Flora G, Gupta D, Tiwari A (2012) Toxicity of lead: a review with recent updates. *Interdiscip Toxicol* 5(2):47–58
93. Padmavathamma PK, Li LY (2007) Phytoremediation technology: hyperaccumulation metals in plants. *Water Air Soil Pollut* 184:105–126
94. Iqbal MP (2012) Lead pollution—a risk factor for cardiovascular disease in Asian developing countries. *Pakistani J Pharmaceut Sci* 25:289–294
95. Hess R, Schmid B (2002) Zinc supplement overdose can have toxic effects. *J Pediatr Hematol Oncol* 24:582–584
96. Fosmire GJ (1990) Zinc toxicity. *Am J Clin Nutr* 51(2):225–227
97. US EPA (2004) Drinking water health advisory for Manganese. EPA-822-R-04-003. US Environmental Protection Agency, Office of Water, Health and Ecological Criteria Division, Washington, DC
98. Kawamura R, Ikuta HS, Fukuzumi S (1941) Intoxication by manganese in well water, Kitasato. *Archiv Exp Med Biol* 18:145–169