

Chapter 16

Engineering of Extremophilic Phosphotriesterase-Like Lactonases for Biotechnological Applications

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1 Introduction

Organophosphate compounds (OPs) constitute the core of pesticides (i.e. paraoxon, parathion, coumaphos, diazinon, dimethoate and chlorpyrifos) and many nerve-based warfare agents (i.e. sarin, tabun, soman and VX) (Table 16.1).

Pesticides are commonly used in agriculture to obtain high productivity, thanks to the control of plants or animals life (pests). However, although they certainly improve the quality and quantity of the agricultural production, their extensive application has caused pollution of aquatic systems, currently posing risks also to human health.

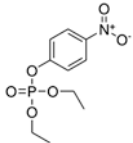
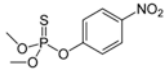
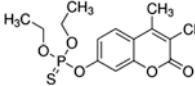
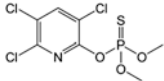
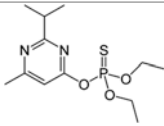
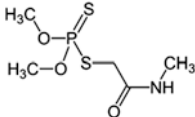
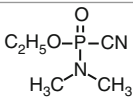
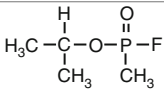
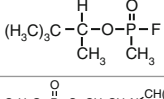
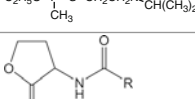
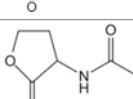
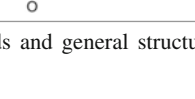
OPs were introduced only in the last century but their use has increased rapidly and now they represent about 38 % of the total pesticides applied worldwide (Singh 2009). In 2008, China produced about 1.74 million tons of 300 different types of pesticides, which has made this country the largest producer and user of such compounds worldwide (Jin et al. 2010).

The latest estimate of the Environmental Protection Agency (EPA) on the annual application of pesticides is more than five million pounds worldwide and one million pounds in the USA (Gavrilescu 2005; Grube et al. 2011). Currently, it is estimated that 200,000 t of OPs are stocked in the world (Singh 2009) and that self-poisoning of workers is very frequent in agriculture settings (Aardema et al. 2008; Calvert et al. 2008; Eddleston et al. 2005).

The fate of the pesticides in the environment depends on several aspects, including soil physico-chemical properties, topography, weather, agricultural management practices and, ultimately, chemical properties of each pesticide (water solubility, tendency to adsorb to the soil, and pesticide persistence) (Tiryaki and Temur 2010).

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Table 16.1 PLL substrates

	Name	Structure	Use
OP pesticides	Paraoxon		Insecticide
	Parathion		Insecticide
	Coumaphos		Acaricide
	Chlorpyrifos		Insecticide
	Diazinon		Insecticide
	Dimethoate		Insecticide
OP nerve gases	Tabun		CWA ^a
	Sarin		CWA ^a
	Soman		CWA ^a
	VX		CWA ^a
Lactones	AHL ^b		—
	3-oxo-AHL ^b		—

Commonly employed organophosphorus compounds and general structure of acyl-homoserine lactone (AHL)

^aChemical warfare agents

^bR corresponds to different size of acyl chain

Usually, the OPs have low persistence in soil because sunlight exposure, water and microbial hydrolysis cause their rapid degradation (Caceres et al. 2010; Ragnarsdottir 2000). However, their over-use, together with the storage of more than half a million tons of obsolete, prohibited or outdated pesticides (Ortiz-Hernández et al. 2013), as well as the limitation or absence in many developing countries of exposure-control programs, led to their anomalous accumulation in soil and water runoff and their magnification through the food chain. The high toxicity of these compounds led the Food and Agriculture Organization of the United Nations (FAO) and World Health Organization (WHO) to establish international food standards and guidelines codes for food safety (Codex Alimentarius) (FAO 2014).

In addition to their use as pesticides, in the early twentieth century extremely poisonous OPs were synthesized and used as chemical warfare agents in World War II (Raushel 2002) and, recently, in terroristic attacks (Tokio subway 1995) and poisoning of civilians (Syria 2013).

OPs are esters of phosphorus with various combinations of oxygen, carbon, sulphur and nitrogen attached, resulting in six different subclasses: phosphates, phosphonates, phosphorothioates, phosphorodithioates, phosphorothiolates and phosphoramidates (Can 2014). They have a general structure in which the phosphorus is linked by a double bond to either an oxygen atom (oxon-OPs), or to a sulphur atom (thion-OPs), and by ester linkages to alkoxy or amino groups, and to another group (halogen, aliphatic, aromatic, or heterocyclic groups).

Based on how the subgroups are oriented around the central phosphorus atom, they have either an R_p or S_p chiral configuration, where the S_p one usually represents the more toxic isomer.

The most studied toxic effect of OPs is the irreversible inhibition of a key enzyme in cholinergic transmission, acetylcholinesterase (AChE), by the covalent phosphorylation of the serine residue within the active site. Such inhibition causes the accumulation of the neurotransmitter acetylcholine in the neuron-neuron or neuron-muscle junctions, determining lacrimation, hypersalivation, bronchial hypersecretion and bronchoconstriction, skeletal muscle fasciculation and twitching, ataxia, respiratory failure, convulsions, hypothermia and, ultimately, death (Carey et al. 2013).

Recent studies on the toxicological properties of OPs also reported gene mutations, chromosomal aberrations, DNA damage (Ojha and Gupta 2014), alteration of semen quality and sperm chromatin (Salazar-Arredondo et al. 2008), and insulin resistance (Lasram et al. 2014). Moreover, the involvement of OPs in cancerogenesis and endocrine disorders has been reported (Gupta 2005).

Since OPs pose hazards to human health and wildlife, it is necessary to realize strategies for the assessment of the respective risk, for their environmental monitoring, and for the treatment of waste and remediation of contaminated sites.

The traditional chromatography and spectroscopy analytical techniques, usually employed in environmental monitoring, show high sensitivity and selectivity, but require a lot of expertise and time, and are very expensive. Detection via biosensors represents a good complement or alternative to the classical methods by simplifying

or eliminating some limiting steps, such as sample preparation and making field-testing easier and faster with significant decrease in cost per analysis (Marrazza 2014).

Many of the classical methods used in the OP decontamination field, including chemical treatment, electrochemical oxidation and reduction, volatilization, incineration or photodecomposition (Kiss and Viråg 2009), are expensive, not eco-friendly, and inefficient to remove low contaminant concentrations (LeJeune et al. 1998). The presence of a robust complement of metabolic pathways and enzymes necessary to use different xenobiotic compounds makes the microbial organisms possible candidate in several promising biotechnology platforms (biomonitoring and bioremediation, destruction of nerve agent stockpiles, use as bioscavengers in medical prophylaxis). However, different drawbacks, including strict dependence of growth on temperature, oxygen, and supply of nutrients, stability under harsh environmental conditions, inhibition of growth in the presence of different chemicals, have limited their application (Singh and Walker 2006).

Therefore enzymes, which hydrolyze phosphoester or phosphothiol bonds in organophosphates, are ideal candidate to be used in the above-discussed applications; indeed, they show stable half-lives, broad substrate ranges, high specificity, work very rapidly and are effective against racemic mixtures and individual stereoisomers (Tsai et al. 2010a, b). These enzymes, generally named phosphotriesterases (PTEs; EC 3.1.8.1), were isolated and characterized from different microorganisms (Islam et al. 2010; Singh 2009; Zhang et al. 2005).

To improve upon desirable traits (stability, activity, stereoselectivity and substrate specificity) of several well-known organophosphatases, many researchers have combined different mutational strategies (directed evolution and rational design) (Goldsmith et al. 2012; Iyer and Iken 2015; Tsai et al. 2012). In particular, rational design methods have provided numerous successful solutions but, over the last years, directed evolution, especially when combined with rational approaches (protein structure analysis and bioinformatics tools) has become the best method to design focused libraries (Cobb et al. 2013a, b; Denard et al. 2015). Directed evolution consists of iterative cycles of random mutagenesis and/or DNA recombination to generate diversity in a target gene sequence and allows the identification of the desired protein variants by high-throughput screening (Cobb et al. 2012). Over the years, directed evolution has enabled the engineering and improvement of known protein features, leading to the generation of optimized biocatalysts to employ in several industrial applications. Moreover, it has also helped the design and optimization of biocatalysts with activities not previously encountered in Nature (Denard et al. 2015).

This review will describe some of the recent achievements in the engineering of OP-hydrolyzing biocatalysts, with a special focus on the evolution of the promiscuous activities of Phosphotriesterase-like Lactonases, which represent the new appealing candidates in the OP remediation.

Promiscuous functions are of great importance because they represent a repertoire of enzymatic activities that can be recruited when the environment changes and a new activity becomes important for fitness. Moreover, they provide excellent opportunities for *in vitro* evolution of new functions to employ in biotechnology and drug development.

2 Organophosphorus Hydrolases

To date, many enzymes able to hydrolyze OP compounds have been identified, such as organophosphorus hydrolases (OPHs), methyl parathion hydrolases (MPHs), serum paraoxonases (PONs) and microbial prolidases/OP acid anhydrolases.

2.1 OPH Hydrolases

A variety of genes different in terms of form, location or host organism involved in OP degradation has been identified so far. The first and best characterized *opd* ('OP degradation') genes and their associated protein products (PTEs) were identified in *Sphingobium fuliginis* ATCC 27551 (previously, *Flavobacterium* sp.) (Kawahara et al. 2010; Singh and Walker 2006) and *Brevundimonas diminuta* GM (previously, *Pseudomonas diminuta* GM) (Harper et al. 1988; Segers et al. 1994); these enzymes showed a strong hydrolytic activity towards OP compounds, in particular, paraoxon (Dumas et al. 1989a).

The protein products of the first identified *opd* genes were initially designated as parathion hydrolases and then classified as OPHs.

OPH from *B. diminuta* (named *bdPTE* in this review) belongs to the amidohydrolase superfamily (subtype I) (Seibert and Raushel 2005) and derives from a gene included in a transposable element of a large plasmid. Structural data showed that *bdPTE* is a homodimer with (β/α)₈-barrel structural fold including a catalytic binuclear centre at the C-terminal end (Benning et al. 1994, 1995, 2001; Vanhooke et al. 1996), containing two metal ions with zinc or cobalt as the favorite cofactors (Dumas et al. 1989a, b; Benning et al. 2001). The residues involved in the catalytic site are: four histidines (H55, H57, H201, and H230) that directly interact with the two divalent cations, an aspartate (D301) and a carboxylated lysine (K169) (Benning et al. 2000).

In *bdPTE* structure, as well as in the structure of other members of the amidohydrolase superfamily, two loops, one (loop 7) involved in the substrate binding and hydrolytic mechanism and the other (loop 8) responsible for the substrate specificity, have been identified (Seibert and Raushel 2005).

bdPTE hydrolyzes different organophosphate nerve agents, such as sarin, soman and VX that were used in recent years as chemical warfare agents (Caldwell and Raushel 1991). Moreover, to date the enzyme is the most efficient in the paraoxon hydrolysis ($k_{cat}/K_M = 4 \times 10^7 \text{ M}^{-1}\text{s}^{-1}$) (Caldwell et al. 1991) among the characterized OP degrading genes. Interestingly, different promiscuous activities (carboxylesterase and lactonase activities) have been detected for *bdPTE* (Roodveldt and Tawfik 2005), but the natural substrate is not yet identified; this suggests that its catalytic activity could have recently evolved from a preexisting enzyme, due to the widespread use of high levels of organophosphate insecticides throughout the world (Raushel and Holden 2000).

In a recent review of Jackson et al., the structural determinants of the high catalytic efficiency towards paraoxon of *bdPTE* were deeply investigated. Two dominant conformational sub-states ('closed' and 'open') with a low-energy structural transition were described for *bdPTE*. The 'closed' state seems to facilitate the substrate recognition and the lowering of the reaction activation energy but not the fast diffusion of the substrate/product; conversely, the 'open' state seems to facilitate mainly the enzyme/product complex dissociation rather than the rate of hydrolysis reaction (Jackson et al. 2008).

This study demonstrated the importance of the conformational changes in the optimization of enzyme catalysis, a concept that has also been revealed for GkaP in the work of Zhang et al. (2015).

A close homolog of *bdPTE* showing organophosphatase activity was isolated from *Agrobacterium radiobacter* P230 and designated as OpdA (Horne et al. 2002, 2003). OpdA was able to hydrolyze diethyl OPs (paraoxon, coumaphos, parathion, diazinon, coroxon) and, at a higher rate than *bdPTE*, dimethyl substrates (methylparathion, phosmet, fenthion, dMUP) (Horne et al. 2002).

The presence of *opd*-like genes has also been demonstrated in several bacterial genome sequencing projects, although these sequences are much more distant (30–40 % identity at the amino acid level) from those found in *Flavobacterium*, *B. diminuta* and *Agrobacterium radiobacter*.

In particular, an OPD-like protein, designated as PHP (phosphotriesterase homology protein), has been isolated from *Escherichia coli*, purified and characterized (Buchbinder et al. 1998). Biochemical analysis showed that PHP of *E. coli* (ePHP) is a monomer with two zinc ions; its structure is similar to that of *bdPTE* and the residues that coordinate the metal ions are conserved (excluding K169) (Buchbinder et al. 1998). The native function of this OPD homolog is not known. As for an ePHP mutant, only weak esterase and phosphotriesterase activities have been reported (Roodveldt and Tawfik 2005).

A protein with features similar to PHP is the product of gene *mll7664* from *Mesorhizobium loti*. Similarly to ePHP, this enzyme has a glutamate bridging the two metals (by similarity) and the main activity observed is the carboxylesterase activity, with promiscuous phosphotriesterase and phosphodiesterase activity. Accordingly this enzyme has been dubbed PLC, phosphotriesterase-like carboxylesterase. By mutation of the glutamate to a lysine, the carboxylesterase activity was completely abolished whereas the phosphodiesterase activity became the main activity (Mandrach and Manco 2009).

In these years many efforts have focused on the improvement of the catalytic efficiency of the known OPH enzymes. For example, several directed evolution strategies have been employed to improve *bdPTE* hydrolytic activity against phosphorothioate substrates, such as methyl parathion and chlorpyrifos, which are hydrolyzed at rates 30 and 1000-fold slower respect to paraoxon (Cho et al. 2002, 2004; Dumas et al. 1989a). By a mutagenesis strategy based on DNA shuffling, a variant of *bdPTE* (22A11) showing an improvement of the methyl parathion hydrolytic activity (25-fold) was identified (Cho et al. 2002).

By using the evolved variant (22A11) as template, a new mutagenesis approach was combined with a chlorpyrifos screening and led to the identification of a variant (B3561) able to hydrolyze chlorpyrifos at a rate 725-fold higher than the wild-type (k_{cat}/K_M value of $2.2 \times 10^8 \text{ M}^{-1}\text{s}^{-1}$) (Cho et al. 2004). Some substitutions (A80V, I274N, and K185R), found in both variants (B3561 and 22A11), cause conformational changes resulting in a widening of the active site and in a stabilization of the enzymatic metal-free state (Cho et al. 2004, 2006).

Many reports have focused on modifying the stereoselectivity of *bdPTE*, in order to make the enzyme more active on the toxic S_p isomers of OP compounds. Structural studies identified three binding pockets (small, large, and leaving group) of *bdPTE* as responsible for the interaction with the substituents of the phosphorus center of the substrates (Chen-Goodspeed et al. 2001a, b). A deep study on the determinants of the *bdPTE* stereoselectivity has demonstrated that the preference for chiral substrates is determined by the small sub-site size. Indeed, the substitution of a glycine residue with an alanine causes an increase of the S_p enantiomer preference by reducing the small subsite size (Chen-Goodspeed et al. 2001b). On the contrary, mutations (I106A, F132A and S308A) that cause a widening of the small sub-site, decrease the S_p preference (Chen-Goodspeed et al. 2001b).

Preferential mutations collected from several studies (K185R, H254Q, H254G, H257F, H257W, I274N and L303T) were chosen to design mutagenic libraries and to create highly effective *bdPTE* variants. From the screening of these libraries, GWT-F5 variant has emerged, which showed catalytic efficiencies for the S_p analogs of nerve agents (soman, tabun, sarin, cyclosarin, VX) ranging between $10^2 \text{ M}^{-1} \text{ s}^{-1}$ and $10^6 \text{ M}^{-1} \text{ s}^{-1}$ (Benning et al. 1995; Madej et al. 2012; Tsai et al. 2012).

The power of the *in vitro* evolution was also highlighted by a recent work in which nine substitutions in *Flavobacterium* OPH (A80V, I106V, F132D, K185R, D208G, H257W, I274N, S308L, and R319S) were used as scaffold for the creation of a synthetic construct (M9), which showed an increase of the rate of catalysis on VX of 35-fold respect to wild type (Jeong et al. 2014). A prediction of M9 structure was obtained and used in a docking simulation with VX substrate. From this analysis two additional critical residues (L271 and Y309) were identified, which, if substituted with alanines, showed an activity fivefold higher than M9 one (Jeong et al. 2014).

In vitro evolution efforts have been also focused on other OP-degrading enzymes. For example, in order to increase the catalytic activity of *Agrobacterium* OpdA on malathion, a combinatorial active site saturation testing (CASTing) was used (Naqvi et al. 2014). In this mutagenesis approach the protein tridimensional structure is analyzed in order to choose groups of few amino acids in the binding pocket which can be randomized simultaneously to generate small mutagenic libraries easy to screen. The CASTing strategy, used for the evolution of the malathion hydrolytic activity of OpdA, led to the identification of a double mutant (S308L/Y309A) which showed a widening of the active site and an increase of 5000-fold in catalytic efficiency towards malathion, reaching the highest value reported for this substrate (Naqvi et al. 2014).

2.2 MPH Hydrolases

A distinct OP-degrading pathway is represented by *mpd* (methyl parathion degradation) genes which confer hydrolytic activity towards methyl paraoxon, methyl parathion, and chlorpyrifos. Based on their similar metallo- β -lactamase domains, the protein products of *mpd* genes were designated as methyl parathion hydrolases (MPHs). The *mpd* genes were found in *Achromobacter*, *Ochrobactrum*, *Stenotrophomonas* and *Pseudomonas* but none of these shows a significant homology to *opd* or other OP-degrading gene (Cui et al. 2001; Yang et al. 2006; Zhang et al. 2005).

Recently, thanks to whole genome sequencing projects, many putative *mpd* homologues were discovered and cloned; a phylogenetic analysis revealed a separated evolution of these genes from *opd* genes.

The isolation of many *mpd* genes from soil bacteria of China (Liu et al. 2005; Zhang et al. 2005, 2006; Zhongli et al. 2001), suggested there is an environmental influence on *mpd* evolution. An AHL lactonase from *Bacillus thuringiensis*, belonging to the β -lactamase superfamily, showed some promiscuous organophosphatase activities, suggesting that it is possible an evolution of OPH and MPH from different lactonase enzymes (Afriat et al. 2006).

MPH enzyme from *Pseudomonas sp.* WBC-3 is a dimer containing a mixed binuclear zinc centre per subunit, where one of the zinc ions is usually replaced by cadmium (Dong et al. 2005). It has been demonstrated that three aromatic residues at the entry of the active site have a crucial role in the determination of affinity towards methyl parathion; indeed, any substitution in these positions results in a significant loss of catalytic activity on the tested substrate (Dong et al. 2005).

Moreover, a newly identified MPH from *Pseudomonas pseudoalcaligenes* (named OPHC2) showed an unexpected thermal resistance and a broad substrate activity spectrum. OPHC2 catalyzes the hydrolysis of a lactone and different phosphotriesters and esters (Gotthard et al. 2013). Its high T_m (97.8 °C) is probably due to the presence of an extended dimerization surface and an intramolecular disulfide bridge (typical of thermostable proteins) and makes OPHC2 a good candidate for OP decontamination (Gotthard et al. 2013).

2.3 Serum Paraoxonases (PONs)

Another interesting group that showed a critical role in OP metabolism is represented by mammalian lactonase/arylesterase enzymes, usually named as serum paraoxonases (Draganov 2010). These enzymes are components of the high-density lipoproteins, which seem to be implicated in the inactivation of toxic by-products of lipid oxidation (Blum et al. 2006; Mackness et al. 2000). The structure of the PON proteins is similar to two lactonases exhibiting promiscuous organophosphatase activity: squid diisopropyl fluorophosphatase (DFPase) and the human senescence marker protein-30 (Belinskaya et al. 2012).

In humans the following PON variants were found: PON1, PON2 and PON3; extensive reports have demonstrated that only PON1 is able to hydrolyze the P–O bond in insecticides, such as paraoxon or chlorpyrifos oxon, and G-series nerve agents (Draganov 2010; Josse et al. 2001; Rochu et al. 2007).

Given its importance as bioscavenger in pharmacological applications, PON1 enzyme was used as template of many evolution attempts. PON proteins are not soluble when produced in recombinant expression systems, therefore, human, mouse and rabbit PON1 genes were mutagenized by DNA shuffling to obtain highly soluble, recombinant enzymes (Aharoni et al. 2004). An interesting variant, named rePON1, was identified and employed as start point for protein engineering studies (Aharoni et al. 2004; Harel et al. 2004; Madej et al. 2012). Functional analysis on rePON1 identified L69, V346, and H115 in the active site as key residues in the catalysis (Harel et al. 2004). Substitutions of these residues were used as target to change the enzymatic stereoselectivity. L69V and V346A substitutions enhance S_p catalysis of soman and diisopropylfluorophosphate (DFP), and H115W substitution increases the specificity for the P–S bond (Amitai et al. 2006; Rochu et al. 2007).

In another study a combinatorial saturation mutagenesis of well-characterized PON1 catalytic sites was used to obtain variants with higher catalytic efficiency on cyclosarin ($10^7 \text{ M}^{-1} \text{ min}^{-1}$). This value is consistent with an appropriate prophylactic protection in vivo; indeed, protection assays in mice demonstrated that, in the presence of these variants, the survivability of mice increases up to 6 h after acute exposure to cyclosarin (Gupta et al. 2011). A chimeric PON1 mutant with high catalytic efficiency towards soman and cyclosarin was also obtained by protein engineering but, however, its activity is still very low on the phosphoramidate tabun and absent on VX (Worek et al. 2014). Other achievements in the evolution of the interesting properties of PON1 are well-described in a recent review (Iyer and Iken 2015).

2.4 Microbial Prolidases/OP Acid Anhydrolases

Despite their main function, many enzymes able to hydrolyze P–O and P–F bonds in G-type nerve agents, were identified and designated as OP acid anhydrolases (OPAA) (EC 3.1.8.2). In particular, prolidase/OPAA enzymes with hydrolytic activity towards G-series nerve agents were isolated from *Alteromonas* species (Theriot et al. 2011; Vyas et al. 2010).

These enzymes, belonging to the prolidase family (Theriot et al. 2011), were found in all domains of life (from bacteria to humans) and represent novel thermostable templates for directed evolution studies. Despite the success achieved in *bdPTE* and PON1 evolution, mutagenesis strategies addressed to engineer efficient prolidases for organophosphorus degradation were underutilized.

OPAA proteins are similar to prolidase enzymes; indeed, they are able to hydrolyze dipeptides that have a proline at the carboxyl-terminal position but are also active on many substrates of OPH and MPH enzymes. They show a high similarity of their structure and their catalytic and stereoselective mechanisms with OPH or

MPH (Cheng and DeFrank 2000) but not significant sequence homology; it is now suggested that OPAA could have evolved from an ancestral prolidase (Theriot and Grunden 2011; Vyas et al. 2010).

Two prolidases were identified in hyperthermophilic archaeons *Pyrococcus furiosus* (Pfprol) and *Pyrococcus horikoshii* (Ph1prol); they were characterized and used as templates for directed evolution (Theriot et al. 2010a, b, 2011; Theriot and Grunden 2011).

In order to obtain thermophilic enzymes with high activity at lower temperatures, many mutagenesis efforts were employed; indeed, the first studies on prolidases from *P. furiosus* (Pfprol) have focused on the increasing of its activity at 35 °C, 50 °C, and 70 °C (Theriot et al. 2010a). By random mutagenesis of *Ph1prol* from *P. horikoshii*, four mutants (A195T/G306S, Y301C/K342N, E127G/E252D, and E36V) with high thermostability and increased prolidase and phosphotriesterase activity in a wider range of temperatures were obtained (Theriot et al. 2011). The mutations responsible for the catalytic improvement were located mainly in the loops and linker regions of the enzyme and seem to cause structural changes in regions away from the active site (Madej et al. 2012; Theriot et al. 2010b, 2011).

To date, *bdPTE* from *Pseudomonas/Brevundiminas*, *OpdA* from *Agrobacterium* and *PON1* from mammalian are the best studied organophosphate-degrading enzymes. These enzymes and their evolved mutants showed significant activity against OPs and nerve gases but their low stability in solution limits their application in OP biosensing and remediation.

At this purpose, in recent years more researches were focused on a new family of OP-degrading enzymes (Phosphotriesterase-Like Lactonases) of which many extremophilic members were identified.

3 Phosphotriesterase-Like Lactonases (PLLs)

Phosphotriesterase-Like Lactonase (PLL) family includes a group of enzymes that have main lactonase activity on lactones and acyl-homoserin lactones (AHLs) and, in addition, low promiscuous phosphotriesterase activity towards organophosphate compounds (OPs) (Table 16.1).

PLLs have first been identified in the hyperthermophilic crenearchaeon *S. solfataricus* (*SsoPox*) and *S. acidocaldarius* (*SacPox*) (Afriat et al. 2006; Merone et al. 2005; Porzio et al. 2007) and also in mesophilic organisms such as *Mycobacterium tuberculosis* (PPH), *Rhodococcus erythropolis* (AhlA), *Deinococcus radiodurans* (*DrOPH/Dr0930*) (Afriat et al. 2006; Hawwa et al. 2009b) and *Mycobacterium avium* subsp. *Paratuberculosis* K-10 (MCP) (Chow et al. 2009). Then new thermostable PLLs from *Geobacillus stearothermophilus* (*GsP*) (Hawwa et al. 2009a) and from *Geobacillus kaustophilus* (*GKL/GkaP*) (Chow et al. 2010) have been reported. Very recently other PLLs have been characterized from *Sulfolobus islandicus* (*SisLac*) (Hiblot et al. 2012b), and from *Vulcanisaeta moutnovskia* (*VmoLac/VmutPLL*) (Kallnik et al. 2014) (Table 16.2).

Table 16.2 Classification of PLLs and main structural and kinetic information

Organism	Enzyme name	Structure (PDB code)	Sequence (UniProtKB code)	Stoichiometry and metal content	Catalytic efficiency for <i>ethyl</i> -paraoxon ^a k_{cat}/K_M (s ⁻¹ M ⁻¹)	Main references ^b
<i>Brevundimonas diminuta</i>	<i>bdPTE</i> ^c	1HZY	P0A434	Homodimer Zn ²⁺ -Zn ²⁺	4 × 10 ⁷	Benning et al. (2001)
Hyperthermophilic archaea						
<i>Sulfolobus solfataricus</i>	<i>SsoPox</i>	2YC5	Q97VT7	Homodimer Fe ³⁺ -Co ²⁺	3.75 × 10 ³	Afriati et al. (2006), Merone et al. (2005)
<i>Sulfolobus acidocaldarius</i>	<i>SacPox</i>	Not solved	Q4I6Z8	-	2.66 × 10 ⁴	Afriati et al. (2006), Porzio et al. (2007), (2013), Bzdrenga et al. (2014)
<i>Sulfolobus islandicus</i>	<i>SisLac</i>	4G2D	C4KKZ9	Homodimer Fe ²⁺ -Co ²⁺	6.98 × 10 ²	Hiblot et al. (2012b)
<i>Vulcanisaeta moutnovskia</i>	<i>VmoLac</i>	4RDY	F0QXN6	Homodimer Co ²⁺ -Co ²⁺	1.86	Kallnik et al. (2014), Hiblot et al. (2015)
Thermophilic/extremophilic bacteria						
<i>Deinococcus radiodurans</i>	<i>Dro0930/</i> <i>DroPH</i>	3FDK	Q9RVU2	Homodimer Zn ²⁺ -Zn ²⁺	1.39	Afriati et al. (2006), Hawwa et al. (2009b), Xiang et al. (2009)
<i>Geobacillus stearothermophilus</i>	GsP	3F4D	D0VX06	Homodimer Co ²⁺ -Co ²⁺	3.28 × 10 ³	Hawwa et al. (2009a)
<i>Geobacillus kaustophilus</i>	GKL/ <i>GkaP</i>	3ORW	Q5KZU5	Homodimer Fe ³⁺ -Zn ²⁺	4.5	Chow et al. (2010), Zeng et al. (2011) to be published
Mesophilic bacteria						
<i>Mycobacterium tuberculosis</i>	PPH	4IF2	P9WHN9	Monomer Zn ²⁺ -Zn ²⁺	8.65	Afriati et al. (2006), Zhang et al. to be published

(continued)

Table 16.2 (continued)

Organism	Enzyme name	Structure (PDB code)	Sequence (UniProtKB code)	Stoichiometry and metal content	Catalytic efficiency for ethyl-paraoxon ^a k_{cat}/K_M ($s^{-1} M^{-1}$)	Main references ^b
<i>Rhodococcus erythropolis</i>	Ah1A	2R1N	Q93LD7	Monomer Fe ²⁺ -C ²⁺	0.50	Afriat et al. (2006), Jackson et al. (2008)
<i>Mycobacterium avium subsp. paratuberculosis</i> K-10	MCP	Not solved	GI:41409766 ^d	–	4.1	Chow et al. (2009)

^aThe reported values of specificity constant on ethyl-paraoxon have been done at the optimal temperature for each group of enzymes: *bdPTE* at 25 °C; PLLs from hyperthermophilic archaea at 70 °C; PLLs from extremophilic bacteria at 35 °C; PLLs from mesophilic bacteria at 25 °C

^bMain references for each enzyme are reported, where it is possible to find main information about structure and catalytic properties summarized in the table

^c*bdPTE* in this table is only to compare its catalytic activity with that of PLL enzymes

^dNCBI code

In fact, thanks to the first solved 3D structure of a PLL (*SsoPox*), very important structural differences emerged (Elias et al. 2008). The structure of the archaeal PLL shows an atypical active site topology and a single hydrophobic channel that perfectly accommodates lactone substrates such as AHLs (Elias et al. 2008). In Fig. 16.2, the superimposition of the structure of *bdPTE* with those of six PLLs (two from hyperthermophilic archaea, *SsoPox* and *VmoLac* two from extremophilic bacteria, *GsP* and *GkaP*; two from mesophilic bacteria, *PPH* and *Ahla*) indicates that the proteins are homologous. It's worth noting that the length of loop 7 in all PLLs, except for *Ahla* (pink), is shorter than *bdPTE* (gray). Loop 8 of all PLLs has almost the same length as in *bdPTE*, but its topology is different for *GkaP* (green) and *GsP* (blue), where it is more detached from the protein core than that of *bdPTE*, and for *VmoPLL*; in the latter case loop 8 is rigid and structured to form an α -helix, differently in comparison with the thermostable PLL *SsoPox* (Fig. 16.2). Previously, some studies have supposed that the elongation of loop 7 in *bdPTE* forms a short α -helix, which may provide an active

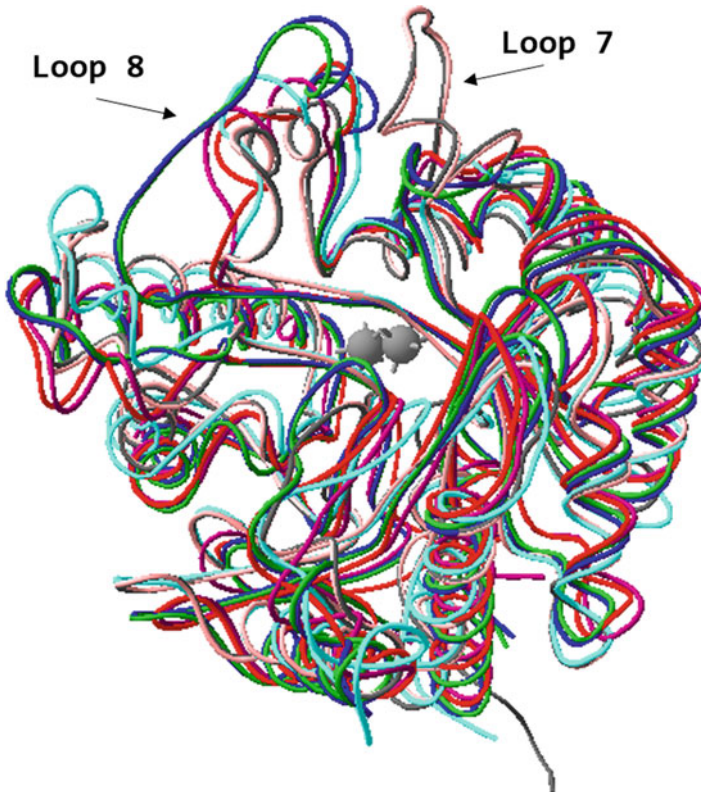


Fig. 16.2 Structural superposition of *bdPTE* with six PLLs. The structural superposition was performed by using Swiss-PdbViewer 4.1.0. *bdPTE* from *B. diminuta* (gray), *SsoPox* from *S. solfataricus* (red); *VmoLac* from *V. moutnovskia* (magenta), *GsP* from *G. stearothermophilus* (blue), *GkaP* from *G. kaustophilus* (green), *PPH* from *M. tuberculosis* (cyan), *Ahla* from *R. erythropolis* (pink); metal cations are represented as gray spheres

“cap” that narrows the active site mouth and may thereby increase phosphotriesterase activity toward the substrate paraoxon (Afriat et al. 2006).

All the PLLs possess a homo- or hetero-binuclear metal center which generates a catalytic hydroxide ion involved in the hydrolysis of both OPs and lactones. The main difference between the two hydrolytic activities is the transition state geometry: sp^3 coordination for phosphotriesters and pentacoordinated state for lactones (Elias et al. 2008). The coexistence of the two activities in the same active site suggests the use of the same catalytic machinery with the optimization for substrate specificity linked to environmental adaptation.

The active site of PLLs contains three sub-sites that are very well adapted for the lactone binding: a small sub-site, a large sub-site and a hydrophobic channel (Elias et al. 2008). The aliphatic chain of the lactones binds within the hydrophobic channel, the large sub-site accommodates the amide group of the N-acyl chain, and the small sub-site positions the lactone ring. Moreover, biochemical analyses confirmed that PLL enzymes are natural lactonases because they have a high activity on acyl-homoserine lactones. Thus it was supposed that PTE has evolved from a yet unknown PLL, whose primary activity was the hydrolysis of quorum sensing homoserine lactones (HSLs) (Afriat et al. (2006). Recently it has been reported that a nine-amino-acid deletion alongside an adjacent point mutation of *bdPTE* gave an intermediate (PTE Δ 7-2/254R) that exhibits both the ancestral acyl homoserine lactonase activity (k_{cat}/K_M values of up to 2×10^4), (undetectable in *bpPTE*), and low paraoxonase activity (Afriat et al. 2012). Bifunctional intermediates with changed specificity have a powerful role in the divergence of new enzymatic functions that are also due to a very important structural changes of loop remodeling.

The bifunctional activity profile of PTE Δ 7-2/254R is compatible with the expected properties of an evolutionary intermediate along a trajectory leading from an HSL with a promiscuous phosphotriesterase activity to a highly efficient phosphotriesterase with a weak, promiscuous lactonase activity (Afriat et al. 2012). The *SsoPox* lactonase activity was tested with synthetic substrates (thiolactones) (Fig. 16.3) but also with real acyl-homoserine lactones (Afriat et al. 2006). However, docking studies and structural analyses demonstrated that the mode of binding is likely to be different (Merone et al. 2010). Starting from this observation, we found a trend of activity on thiolactones quite different from that on AHL lactones, because the specificity constant decreases with the increase of the lateral acyl chain (Fig. 16.3).

AHLs, also named as autoinducer 1 (AI-1), are small signal molecules, produced mainly by gram-negative bacteria, and mediate cell to cell communication known as quorum sensing (QS). When these molecules accumulate in the extracellular environment until a critical threshold, the transcriptional profile of the bacteria is altered (Hentzer et al. 2003).

It is reported that the virulence and the biofilm formation of some pathogens are regulated by QS (Costerton et al. 1999; Dickschat 2010; Jones et al. 2010; Popat et al. 2008), suggesting that the quenching of this mechanism (quorum quenching) could be an interesting strategy against multi-resistant pathogen bacteria which use AHL based QS like *P. aeruginosa* (Amara et al. 2011; Dong et al. 2000, 2001; Ma et al. 2009; Reimann et al. 2002). Therefore, as lactonases, PLLs can hydrolyze AHLs and quench the QS mechanism, as seen for AiiA, human paraoxonases (Dong et al. 2001;

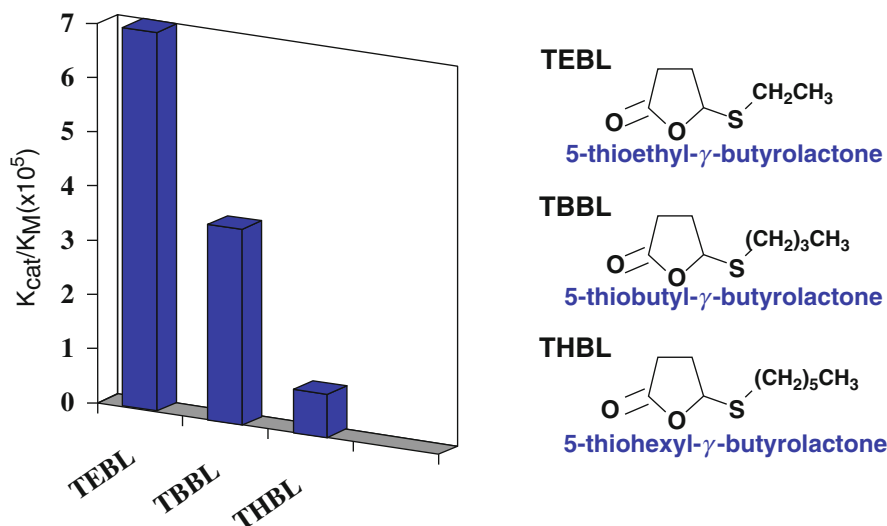


Fig. 16.3 Lactonase activity of *SsoPox* on thiolactones with different acyl chain length. The assays were performed at 70 °C as reported in Afriat et al. (2006). *SsoPox* shows a k_{cat}/K_M of 7×10^5 , 3×10^5 and 0.8×10^5 $M^{-1} s^{-1}$ on TEBL, TBBL and THBL respectively

Ma et al. 2009), and *SsoPox*, *SacPox* and *Dr0930* that, as reported, are able to reduce swarming PAO motility (Mandrigh et al. 2011). Moreover *SsoPox* has also recently been shown to decrease virulence factors expression and biofilm formation in vitro and to reduce rat mortality in a rat pulmonary infection model (Hraiech et al. 2014).

By considering they have both lactonase and phosphotriesterase activity, PLLs represent a very interesting candidate for biotechnological application as quorum quenching agent (Dong et al. 2000) or OPs biodecontaminant (Singh 2009).

However, at present it is still unclear which is (are) their biological function(s), also because in many cases PLLs were found in microorganisms that do not produce or are regulated by AHLs; therefore, probably they could act on the mechanism of communication of other microorganisms. Furthermore, some members of the PLL family efficiently hydrolyze gamma and/or delta oxo-lactones, but not AHLs (Chow et al. 2010; Xiang et al. 2009). Anyway the interest in these enzymes, in particular for biotechnological applications in decontamination field, lies in exploring phosphotriesterase activity. Most of the extremophilic PLLs are able to hydrolyze different OP pesticides, even if at different extent. As representative compound it is reported the catalytic efficiency on ethyl-paraoxon of the described PLLs (Table 16.2). Among the archaeal PLLs, the *SsoPox* promiscuous catalytic activity against paraoxon is 3.75×10^3 $s^{-1}M^{-1}$, tenfold lower than that of *SacPox* (2.66×10^4 $s^{-1}M^{-1}$) at 70 °C (Merone et al. 2010; Porzio et al. 2013). These values are higher than *SisLac* catalytic efficiency (6.98×10^2 $s^{-1}M^{-1}$) at 70 °C (Hiblot et al. 2012b).

SsoPox and *SacPox* are both in dimeric form, which is consistent with the crystal structures of *SisLac*, *Dr0930*, *GsP* and *GkaP* (Table 16.2).

The recent discovered hyperthermophilic PLL from *V. mountronski* (*VmoLac*) shows a low catalytic efficiency against paraoxon of 1.86 $s^{-1}M^{-1}$ (Hiblot et al. 2015), comparable to those of *Dr0930* (1.39 $s^{-1}M^{-1}$) (Hawwa et al. 2009b; Mandrigh et al. 2013) and *GkaP*

($4.5 \text{ s}^{-1} \text{ M}^{-1}$ at $35 \text{ }^\circ\text{C}$; 1.1×10^2 at $75 \text{ }^\circ\text{C}$), even if all of them are highly stable enzymes. Among the others extremophilic bacterial PLLs, GsP from *G. stearothermophilus* has the highest value of $k_{\text{cat}}/K_{\text{M}}$ on paraoxon that is $3.28 \times 10^3 \text{ s}^{-1} \text{ M}^{-1}$ at the same comparable temperature ($35 \text{ }^\circ\text{C}$) (Hawwa et al. 2009a). All the mesophilic PLLs, PPH, AhlA and MCP, exhibit a weak phosphotriesterase activity. For hydrolysis of paraoxon by AhlA, only the $k_{\text{cat}}/K_{\text{M}}$ ratio was estimated ($0.5 \text{ s}^{-1} \text{ M}^{-1}$), while MCP and PPH exhibited a higher value of 4.1 and $8.6 \text{ M}^{-1} \text{ s}^{-1}$, respectively (Afriat et al. 2006; Chow et al. 2009).

4 In Vitro Evolution of Extremophilic PLLs

The promiscuous phosphotriesterase activity of PLL enzymes on OP compounds suggests that they may constitute a “generalist” intermediate from which PTE may have arisen. However, the possibility that *M. loti* PLC (more similar to ePHP) represents such starting point could not be discarded (Fig. 16.4).

Recently, various mutagenesis strategies, such as site-directed mutagenesis and directed evolution, have been used to enhance the promiscuous OP-degrading activ-

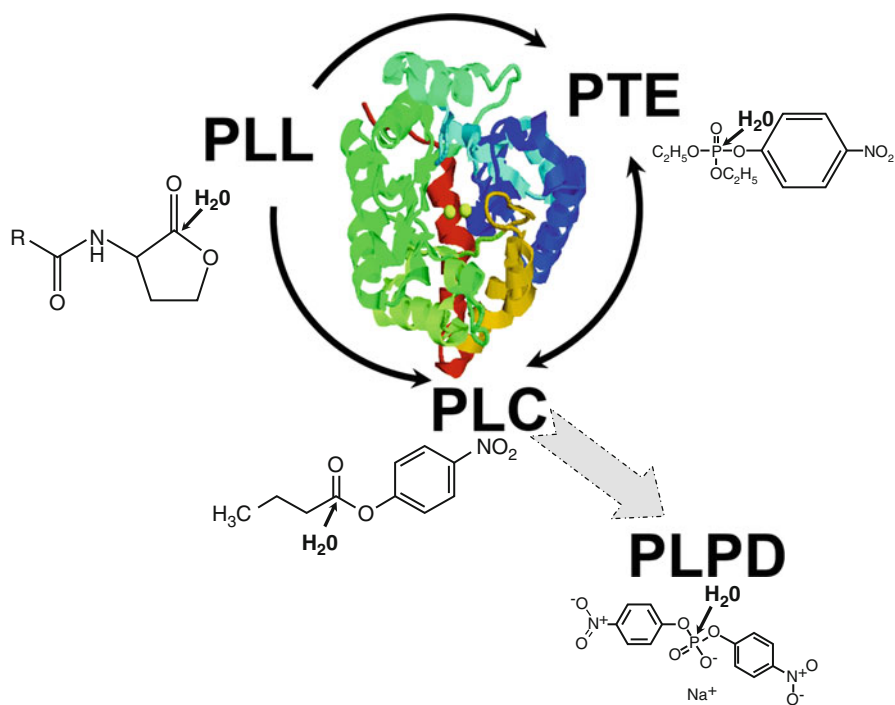


Fig. 16.4 Representation of likely evolution paths among members of the Amidohydrolase Superfamily. Schematic representation of the possible evolution of three different members of the amidohydrolase superfamily: *PLL* Phosphotriesterase-Like Lactonase, *PTE* PhosphoTriestErase, *PLC* Phosphotriesterase-Like Carboxylesterase, enzymes. *PLPD* Phosphotriesterase-Like PhosphoDiesterase was generated from PLC by a single mutation

ity of several PLLs; this approach efficiently reshapes enzymatic proficiency and specificity, and provides a powerful evidence of the evolvability of PLL enzymes for OP decontamination.

An important difference between PLL and PTE enzymes is that loop 7 and loop 8 have a different length and topology. As reported above, both the loops in the amidohydrolase superfamily are often found to interact with the substrates and to play important role in determination of substrate specificity (Seibert and Raushel 2005). The association of protein and solvent engineering for *SsoPox* has led to kinetics parameters (k_{cat} and specificity constant) more similar to those of the PTE, but it was exceptionally more stable and so it represent an interesting detoxification device (Merone et al. 2010). The active site structure of *SsoPox* shows a low elasticity to changes afforded by temperature because various mutations have led to different activity/temperature profiles (Merone et al. 2010).

This means that in the active site a compromise is required between the rigidity necessary to withstand high temperatures assuring at the same time specificity, and the flexibility which is a prerequisite for catalytic activity (Tehei et al. 2005). In detail, variants Y97W, I98F, Y97W/I98F showed profiles very similar to wild type even though with different degrees of activation. As already reported in Elias et al. (2008), single mutant Y97W suggests an involvement of position 97 in the correct orientation of the substrate for catalysis. The single mutant W263F emerged as the best variant, but activating effects were observed for all mutants (Table 16.3).

Table 16.3 Mutants of PLLs obtained by in vitro evolution of paraoxonase activity

PLL	Important identified residues and relevant location/role in the structure	Relevant evolved variants	Effect on catalytic efficiency for ethyl-paraoxon (k_{cat}/K_M)	References
<i>SsoPox</i>	M79 Y97 (<i>interaction with the substrate</i>) I98 I261 (<i>Loop 8</i>) W263 (<i>Loop 8</i>)	M79V Y97W I98F W263Q W263F I261F M79V/W263Q I98F/I261F Y97W/I98F Y97W/I261F Y97W/W263Q Y97W/W263F Y97W/I98F/ I261F	Increase from 1.7- to 6-fold the wt activity	Merone et al. (2010)
		W263L,M, F, D, I, P, Y, V, A, C, E, G, Q, N, S, H, T, R, K	Increase from 1.7- to 23-fold the wt activity	Hiblot et al. (2013)

(continued)

Table 16.3 (continued)

PLL	Important identified residues and relevant location/role in the structure	Relevant evolved variants	Effect on catalytic efficiency for ethyl-paraoxon (k_{cat}/K_M)	References
GkaP	E15 F28 (<i>small binding pocket</i>) D73 Y99 (<i>in the leaving group subsite</i>) T171 F228 (<i>at the base of loop 7</i>) N269 (<i>inside loop 8</i>) V270 (<i>inside loop 8</i>) W271 (<i>inside loop 8</i>) G273 (<i>inside loop 8</i>) F276 I299	Y99L D73Y/Y99L E15V/Y99L/ G273D/F276V/ I299S Y99L/T171S/ V270G/G273D Y99L/W271C F28I/Y99L/ T171S/F228L/ N269S/ V270G/ G273D F28I/D73Y/ Y99L/T171S/ F228L/N269S/ V270G/G273D F28I/Y99L/ T171S/F228L/ N269S/ V270G/ W271C/G273D F28I/D73Y/ Y99L/T171S/ F228L/ N269S/ V270G/W271C/ G273D	Increase from 3- to 230-fold the wt activity	Zhang et al. (2012)

W263F is sixfold more efficient towards ethyl-paraoxon and it also shows a little improvement of rate of hydrolysis towards the warfare agent cyclosarin (Merone et al. 2010).

It has also been reported that the residue W263 plays an important role in substrate binding in the *SsoPox* active site, at the dimer interface in loop 8, and influence the protein stability, enzymatic activity and promiscuity (Hiblot et al. 2013). Recent structural analyses of *SsoPox* show that all the 19 different substitutions of the W263 residue increase the *SsoPox* paraoxonase activity, by inducing very subtle changes in the loop 8 positioning with no effect on the very short loop 7 (Hiblot et al. 2013) (Table 16.3).

Hiblot et al. (2012b) reported that, on the basis of *SisLac* and *SsoPox* structural analysis, variations at the positions 14 and 34 seem to have a big impact between the two structures, with a possible effect on dimerization changes (position 34 of *SisLac*) or stability (positions 14, 34 of *SisLac*). The analysis of T_m of the mutants E14K and Y34Q shows that single substitution destabilize *SisLac*, but the combination of both of them helps to restore partially the stability (Hiblot et al. 2012b). Additionally, the

rational-designed mutants activity have been tested on different substrates, and in particular it was observed an increase of until 17-fold (E14K) the wild type *SisLac* (Hiblot et al. 2012b) (Table 16.3).

To improve the organophosphatase activity of *Dr0930*, in particular on ethyl and methyl paraoxon, site-directed mutagenesis, random mutagenesis, and site-saturation mutagenesis were also used. More than 30,000 potential mutants were screened, and a total of 26 mutated enzymes were purified and characterized kinetically (Hawwa et al. 2009a). In total, all the mutagenesis efforts raised the specificity for paraoxon until 559-fold respect to wt (Table 16.3).

This study underline that by using an iterative approach to mutagenesis, it's possible to achieve a large rate enhancements when mutations are made in already active mutants. Moreover, the effect of new mutations on kinetic parameters allows to have important structural and functional information of the enzyme (Hawwa et al. 2009b). In addition other authors described rational protein design and random mutagenesis methodologies that were combined with an efficient in vitro screening to identify highly efficient hydrolase variants with enhanced degradation ability (Meier et al. 2013). The best mutant displayed a catalytic activity more than two orders of magnitude higher than the wild-type *Dr0930* (Table 16.3).

As regards the *Geobacillus* sp, by combining rational and random mutagenesis strategies, new variants were successfully obtained after four rounds of screening for *GkaP* from *G. kaustophilus*. The catalytic efficiency (k_{cat}/K_M) of the best variant against ethyl-paraoxon was about 40-fold higher than that of the D71G/E101G/V235L *Dr0930* variant, and comparable to that of the W263FS*soPox* variant (Zhang et al. 2012) (Table 16.3).

Eight amino acid changes in *GkaP* have enhanced the catalytic efficiency for ethyl-paraoxon by nearly two orders of magnitude. Among the eight mutations (F28I, Y99L, T171S, F228L, N269S, V270G, W271C, and G273D), some of these substitutions were located far from the catalytic β -metal, whereas others were within the vicinity of the catalytic site. Tyrosine 99, which is located at the entrance of the binuclear metal center and is thought to reside in the leaving group sub-site, is a conserved residue in the PLL family that appears to play a critical role in stabilizing the lactone ring in a favorable state (Elias et al. 2008). The *GkaP* variants showed a strong inverse relationship between thermostability and catalytic activity, even if they were still remarkably thermostable (Zhang et al. 2012).

Beside the in vitro evolution approach, it is worth noting that metals in the active site have an important role in influencing the kinetic parameters, as reported in *bdPTE* (Rochu et al. 2004). Recently we reported that promiscuous paraoxonase activity of *SacPox* seems to be more sensitive than the main lactonase activity to the metal species in the binuclear metal centre (Porzio et al. 2013). In fact the enzyme prepared in presence on Mn^{2+} cations (*SacPox*- Mn^{2+}) shows about 30- and 19-fold increase in paraoxon or me-paraoxon efficiency respectively, with respect to the enzyme prepared with Cd^{2+} (*SacPox*- Cd^{2+}). This finding suggest a new interesting approach to be used to increase phosphotriesterase promiscuous activity for biotechnological application.

Finally, the functional and structural homologies suggest that PTE evolved from an as yet unknown PLL, using its promiscuous paraoxonase activity as an essential starting point. As demonstrated by the properties of *SsoPox*, the paraoxonase activity of PLLs can be considerably high, even without compromising the lactonase activity, and perhaps while acquiring other activities such as aryl esterase. In that respect, *SsoPox* resembles a “generalist” intermediate from which PTE may have emerged (Aharoni et al. 2005; Afriat et al. 2006).

The recent specialization as a phosphotriesterase, as seen in *bdPTE*, probably through an insertion into loop 7, did not completely eliminate the ancestor’s lactonase activity (Afriat et al. 2006). The promiscuous carboxylesterase and phosphodiesterase activities observed in *bdPTE* and PLLs in turn paved the way for the prediction of the existence in Nature of PTE-like enzymes with main carboxylesterase or phosphodiesterase activities. It has been discovered that an homologue of ePHP, from *Mesorhizobium loti*, is an efficient carboxylesterase with promiscuous phosphotriesterase activity, and therefore it was called *MloPLC* (Phosphotriesterase-Like Carboxylesterase) (Mandrigh and Manco 2009). It shows also a low promiscuous phosphodiesterase and lactonase activity, and we suppose it may be originated from either an ancestral PLL-like or PTE-like enzyme, considering that it maintains both these promiscuous activities (Fig. 16.4). Moreover the discovery of *MloPLC* supports the theory on the enzyme evolution starting from promiscuous activities in the amidohydrolase superfamily (Afriat et al. 2006; Aharoni et al. 2005; Khersonsky et al. 2006). However, the conversion of *MloPLC* into an excellent phosphodiesterase by means of a single mutation, in parallel with the complete loss of carboxylesterase activity, is a rare case of substrate assisted gain-of-function via stabilization of the binuclear metal center (Mandrigh and Manco 2009). This scenery let us to speculate the existence of a new intermediate, PLC-like, between PTE and PLL, from which, during evolution by sequence/structural rearrangements, phosphodiesterase activity could have been arising (Fig. 16.4). The mutant obtained shows a complete loss of carboxylesterase and lactonase activities (Mandrigh and Manco 2009), suggesting that a Phosphotriesterase-like Phosphodiesterase enzyme (PLPD) may represent a final point in this evolutionary route, supposing probably other interesting intermediates after PLC to be discovered and explored.

5 Potentiality of Evolved OP-Degrading Enzymes in Biotechnological Applications

The rapid accumulation of data on evolved OP-degrading enzymes led to renewed interest in developing new strategies for several biotechnological applications, such as therapy and prophylaxis of OP poisoning, OP biomonitoring and OP remediation.

5.1 Therapy and Prophylaxis of OP Poisoning

The human PON1 provides the organism with a natural protection from low-dose intake of some organophosphates, such as paraoxon, but is not efficient in the case of assumption of high doses of deadly organophosphates, such as chemical warfare agents. Therefore, mutants of *bdPTE*, *OpdA* and PON1 have been obtained and assayed for in vivo prophylaxis of OP pesticides and nerve agents (Masson and Rochu 2009). Despite the achievements of these studies, their promising therapeutic application requires a deep study of efficient transport and delivery methods in order to reduce the decrease or the full loss of the enzymatic activity. Furthermore, others important issues regard their expression, pharmacokinetics and humanization.

An improvement of the thermal stability and an extension of the mean residence time in mice (from minutes to days) was reached by PEGylation of these enzymes (Trovaslet-Leroy et al. 2011) (for a recent review, see Nachon et al. 2013).

5.2 OP Biomonitoring

While ELISA kits are commercially available for the monitoring of few OPs, in the recent years, enzymatic sensors were deeply investigated for the rapid OP detection (Carullo et al. 2015).

OP biosensors used today are inhibition-based rather than catalysis-based. Enzyme inhibition requires regular replacement of the enzyme part and is measured via amperometric detection of thiocholine, hydrogen peroxide, or p-aminophenol (Wanekaya et al. 2008). These compounds are produced by hydrolysis of acetylcholine (thiocholine) or aminophenyl acetate (p-aminophenol) by AChE, or choline oxidation (hydrogen peroxide) by choline oxidase.

Other OP biosensors use a potentiometric transduction based on the pH variation caused by acetic acid reduction.

AChE-based biosensors are sensitive but suffer of many drawbacks, such as the limited selectivity since AChE is inhibited by several neurotoxins (carbamates, heavy metals), the impossibility to be reused without regeneration with reactivators such as pyridine 2-aldoxime, and the unsuitability for real-time monitoring. Also stability over time is a problem. Recently we investigated the opportunity to use the thermostable EST2 from *Alicyclobacillus acidocaldarius* as substitute of AChE (Febbraio et al. 2011).

The use of catalysis-based systems (typically using OPHs) is extremely attractive for OP biosensing because they have broad substrate recognition, and the hydrolysis product (p-nitrophenol) can be also quantitatively detected by electrochemical and optical methods (Wanekaya et al. 2008).

The use of higher ionic strength buffers (in which the enzymes show the maximum activity for all time course of the measures) represents the main advantage of the use of optical biosensors rather than the potentiometric ones (Wanekaya et al. 2008).

An example of the discussed application is represented by the combination of a mutant of *bdPTE* (H254R/H257L), showing an increased catalytic rate for P–S bond hydrolysis, with carbon nanotubes for the sensitive, rapid and selective amperometric detection of V-type nerve agents and demeton-S insecticide (Joshi et al. 2006).

5.3 OP Remediation

In recent years, a strong public opposition to the use of polluting chemical methods has led to the research and development of alternative eco-friendly devices to remediate OPs. The use of microorganisms has limited applicability because poses many issues (dependence of growth on temperature, presence of growth inhibitors such as toxic compounds and some metabolites, and so forth). Therefore enzymes represent a feasible and good alternative.

Many examples of the application of OP-degrading enzymes in the remediation were reported.

The presence of a permeability barrier represented by the outer membrane hinders the OP/OP degrading enzymes interaction in the cell (Richins et al. 1997); to overcome this problem, different surface anchoring motifs, such as Lpp-OmpA chimera, ice nucleation protein (INP), or autotransporter have been used to address these enzymes to the cell surface (Li et al. 2008; Richins et al. 1997; Shimazu et al. 2001).

By using an INP anchoring motif, Yang et al. (2013) reported the functional display of the PTE-GFP fusion on the cell surface of *Sphingobium japonicum* UT26. The strain containing the fusion showed the ability to hydrolyze OPs and allowed to track the fluorescence during the time course of bioremediation.

The same anchoring motif was successfully used to display an improved mutant of OPH from *B. diminuta* on the cell surface of *E. coli* (Tanga et al. 2014). This engineered biocatalysts showed the highest paraoxonase activity among the already reported OPH-displayed strains, an optimal temperature of 55 °C and an high stability (100 % activity over 1 month at room temperature) (Tanga et al. 2014).

Pinjari et al. (2013) have heterologously expressed *bdPTE* on the membrane of *Pseudomonas* sp. Ind01, a strain isolated from the activated sludge of a pesticide-manufacturing sites (Pinjari et al. 2012). The engineered strain was very efficient in the remediation of different commercially-available OP pesticides (Pinjari et al. 2013).

Many strategies have been used to immobilize, encapsulate or entrap OP-degrading enzymes to create materials with preserved organophosphatase activity. In a recent paper, mesoporous thin films carried over glass slides were used as host matrices for enzymatic remediators. These bio-catalysts were active and very sensitive (down to 15 µM concentration for paraoxon) and, moreover, they were easily prepared, and reused several times with no significant loss in catalytic activity (Francic et al. 2014).

Gao et al. (2014) reported the covalent immobilization of OpdA enzyme on highly porous nonwoven polyester fabrics for OP degradation. First, these materials were activated with ethylenediamine; the enzyme was, then, immobilized by glutaraldehyde, a bifunctional crosslinker. Following the enzymatic immobilization, the K_M for methyl parathion was increased, the pH profile was widened, and the enzymatic stability was enhanced. In batch mode, this remediation system could hydrolyze 20 μM methyl parathion in un-buffered water (Gao et al. 2014).

Despite these applications, many data report a gradual inactivation of *bdPTE* at temperatures up to 35 °C until a full loss of catalytic activity at 60 °C (Caldwell and Raushel 1991) and, moreover, all attempts to immobilize the enzyme were linked to short-term stability, high costs and possible changes of the kinetic properties (Braatz 1994; Caldwell et al. 1991; Gill and Ballesteros 2000; Grimsley et al. 2001; LeJeune et al. 1997).

Therefore it is crucial to find alternative highly resistant and stable enzymes. As already described, phosphotriesterase activity has been found as promiscuous component in several hydrolases such as prolidase, PLL and MBL enzymes isolated from different extremophilic hosts. The exceptional thermal stability of the extremophilic bacterial and archaeal enzymes offers many biotechnological advantages in industrial processes such as high resistance to harsh conditions (presence of solvents or detergents), minimized contamination potential, extreme stability and increased enzymatic longevity. Furthermore, the purification costs of the enzymes expressed in mesophilic hosts are affordable, indeed, the contaminating proteins deriving from the host can be precipitated by simple thermal fractioning.

Thanks to the link between protein stability and evolvability (Bloom et al. 2006), many reports used highly stable scaffold for next-generation engineering technologies.

For this purpose, many efforts have been performed to use extremophilic PLL enzymes as remediation devices. To date, only two examples of practical applications of PLLs have been reported, even if one of them was not properly applied in OP remediation. In fact, the first report deals with the employment of the lactonase activity of *SsoPox* on acyl-homoserine lactones (Ng et al. 2011). The enzyme was absorbed onto nanoalumina membranes; in order to interfere with quorum sensing signaling, bacterial cultures were treated with the immobilized enzyme. The results open up new scenarios of biotechnological applications, for example, it could be possible to use lactonases immobilized on filtration membranes for the control of undesirable microbial activities in water purification systems (Ng et al. 2011).

In the second report, the advantage of a mutant of *SsoPox* (W263F) as detoxification tool was assessed. The enzyme was dissolved in different buffered aqueous solvents (30 % ethanol, 30 % or 50 % methanol and 0.1 % sodium-dodecyl-sulphate) to analyze its activity under stressing denaturing environment, typical, for example, of the conditions usually employed in the toxin extraction from contaminated soils (Merone et al. 2010; Hiblot et al. 2012a). In one case the results were compared with those obtained with *bdPTE* from *B. diminuta*. W263F outperforms *bdPTE* under most of the tested conditions; in 15 min at room temperature about 99.5 % of paraoxon was hydrolyzed in 30 % methanol and ethanol and 0.1 % sodium-dodecyl-sulphate (Merone et al. 2010).

6 Concluding Remarks and Future Perspectives

It is generally accepted the theory that thermostable enzymes are the best candidate for the *in vitro* evolution since they well-accept destabilizing mutations, which can lead to new or improved enzymatic functions. The investigation reported here outlines the *in vitro* evolution potential of PLL enzymes and their successful employment in OP detoxification. It is well-known that also *p*-nitrophenol produced by paraoxon/parathion hydrolysis is toxic (120-fold less than the original OP) (Munnecke 1979), but, in the last years, different soil bacteria able to use *p*-nitrophenol as carbon source were studied offering the opportunity to completely remove the toxicity of such compounds (Spain and Gibson 1991).

We have described many examples of OP-degrading enzymes with considerably improved activities that can be used in combination with PNP-utilizing bacteria for a full degradation of the OP compounds (Shimazu et al. 2001).

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