

The Handbook of Environmental Chemistry 119

Series Editors: Damià Barceló · Andrey G. Kostianoy

Yolanda Picó

Julian Campo *Editors*

# Mass Spectrometry in Food and Environmental Chemistry



Springer

# **The Handbook of Environmental Chemistry**

**Volume 119**

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# Mass Spectrometry in Food and Environmental Chemistry

Volume Editors: Yolanda Picó · Julian Campo

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## Series Preface

With remarkable vision, Prof. Otto Hutzinger initiated *The Handbook of Environmental Chemistry* in 1980 and became the founding Editor-in-Chief. At that time, environmental chemistry was an emerging field, aiming at a complete description of the Earth's environment, encompassing the physical, chemical, biological, and geological transformations of chemical substances occurring on a local as well as a global scale. Environmental chemistry was intended to provide an account of the impact of man's activities on the natural environment by describing observed changes.

While a considerable amount of knowledge has been accumulated over the last four decades, as reflected in the more than 150 volumes of *The Handbook of Environmental Chemistry*, there are still many scientific and policy challenges ahead due to the complexity and interdisciplinary nature of the field. The series will therefore continue to provide compilations of current knowledge. Contributions are written by leading experts with practical experience in their fields. *The Handbook of Environmental Chemistry* grows with the increases in our scientific understanding, and provides a valuable source not only for scientists but also for environmental managers and decision-makers. Today, the series covers a broad range of environmental topics from a chemical perspective, including methodological advances in environmental analytical chemistry.

In recent years, there has been a growing tendency to include subject matter of societal relevance in the broad view of environmental chemistry. Topics include life cycle analysis, environmental management, sustainable development, and socio-economic, legal and even political problems, among others. While these topics are of great importance for the development and acceptance of *The Handbook of Environmental Chemistry*, the publisher and Editors-in-Chief have decided to keep the handbook essentially a source of information on "hard sciences" with a particular emphasis on chemistry, but also covering biology, geology, hydrology and engineering as applied to environmental sciences.

The volumes of the series are written at an advanced level, addressing the needs of both researchers and graduate students, as well as of people outside the field of

“pure” chemistry, including those in industry, business, government, research establishments, and public interest groups. It would be very satisfying to see these volumes used as a basis for graduate courses in environmental chemistry. With its high standards of scientific quality and clarity, *The Handbook of Environmental Chemistry* provides a solid basis from which scientists can share their knowledge on the different aspects of environmental problems, presenting a wide spectrum of viewpoints and approaches.

*The Handbook of Environmental Chemistry* is available both in print and online via <https://link.springer.com/bookseries/698>. Articles are published online as soon as they have been approved for publication. Authors, Volume Editors and Editors-in-Chief are rewarded by the broad acceptance of *The Handbook of Environmental Chemistry* by the scientific community, from whom suggestions for new topics to the Editors-in-Chief are always very welcome.

Damià Barceló  
Andrey G. Kostianoy  
Series Editors



# Preface

The science of Environmental Chemistry includes a vast array of disciplines aiming at a complete description of the Earth's environment. It encompasses geological, chemical, physical, and biological transformations of chemical substances occurring on the natural environment, at different scales, providing an overview of the impact of human activities on it. In this regard, the UN have launched the 2030 Agenda for Sustainable Development, which aims to provide a shared blueprint for peace and prosperity for people and the planet, now and in the future. This agenda proposes 17 Sustainable Development Goals that are an urgent call to guarantee food security and promote sustainable use of terrestrial ecosystems. Environmental chemistry is directly related to these SDGs, helping to study and understand the effects of the increasing use of resources to fulfill the consumer's demand for goods and services.

In recent times, great progress has been made in environmental analysis techniques with special attention to mass spectrometry (MS), omics approaches, miniaturization, automatization, nano- and biosensors, and green chemistry. In this book, recent advances in techniques and innovative uses of MS in food safety and environmental quality are summarized and discussed. The latest developments in sample preparation together with improvements in MS platforms have allowed the study of different omics together with the screening of emerging contaminants (ECs) and nanomaterials. *Mass Spectrometry in Food and Environmental Chemistry* offers useful and relevant information on the most recent analytical techniques as well as all major topics of food safety and environmental quality analyzable by them.

The book contains thirteen chapters written by worldwide experts discussing themes such as elemental and isotopic mass spectrometry as well as chromatographic-high resolution and chromatographic-mass spectrometry-based screening methods in food and environmental chemistry analyses. Other approaches such as omics, thermal desorption, and pyrolysis together with chiral mass spectrometry, ambient ionization techniques, ion mobility, and imaging mass spectrometry have also been discussed not forgetting last developments such as chip-based separation

devices coupled to mass spectrometry and their role in food and environmental chemistry. This book is aimed at anyone seeking to learn about new techniques and developments including a broad audience of researchers, from environmental and food chemists, biologists, and food technologists to the food industry and official food and environment control laboratories, together with other professionals responsible for resources management and decision-makers. Important tools for problem-solving in chemical and biological analyses are discussed in detail. The different advances and promising applications described here are state of the art and will be of paramount importance for ensuring food safety and environmental health for current and future generations.

Finally, we would like to thank all the chapters' authors for their continuous and devoted effort and patience. All of them deserve our deepest thanks because without them this book would have been impossible. I also thank the anonymous reviewers of the project for their suggestions. We would also acknowledge with great gratitude Prof. Damià Barceló (Series Editor) who gave us the opportunity of editing this book and the assistance of the Editorial Office of Springer and especially of Dr. Sofia Costa (Editor) and Ms. Ramya Venkitachalam (Project Coordinator for Springer Nature). We hope that this volume lives up to the expectations of the readers starting or with a long experience in food and environmental quality analysis. This book is intended for all of you.

Valencia, Spain

Yolanda Picó  
Julian Campo

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# An Overview of the State-of-the-Art: Mass Spectrometry in Food and Environment



Yolanda Picó and Julian Campo

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**Abstract** Mass spectrometry (MS) has achieved great renown as a specific, selective, sensitive, and rapid technique for the analysis and evaluation of a wide range of food products and environmental matrices. The state of the art of MS in food and environmental safety and quality is presented to show the potential of this technique in the qualification and quantification of chemical characteristics of food and environmental samples, in the evaluation of the quality of meat, fish, fruits, vegetables, and other food products, as well as in the classification of environmental samples, and in the determination of contaminants in all their compartments. The features of each mass spectrometry advance for each category were summarized in the aspects of the investigated quality and safety attributes, and the used systems (low and high resolution mass spectrometry). With its success in different applications of food and environmental quality, and safety analysis and assessment, it is evident that MS can facilitate a variety of tasks. Continued development of methodology and instrumentation will enable more sensitive and timely detection in the coming years.

**Keywords** Adulteration, Authenticity, Chemometrics, Environmental assessment, Food quality, Food safety, High resolution mass spectrometry, Mass spectrometry

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## 1 Introduction

The environment and food are fundamental to ensuring human well-being and to preserve the world for the next generation [1–4]. There are many environmental problems that need to be addressed in the coming years if the planet earth is to remain habitable for humans in the next future [5, 6]. These problems are well-known, climate change, water scarcity and its effects on human health, biodiversity preservation, resilience, waste and wastewater management, and so on [7–10]. There are many challenges ahead that require an important research investment. There are also several food safety and quality issues to consider in the near future. Food safety and quality are two concepts highly related. Food quality can be defined “as a total of traits and criteria, which characterize food for a consumers’ health” [11]. According to the definition of the Food and Agriculture Organization (FAO): “Food security exists when all the people have through physical and economic access to sufficient, safe and nutritious food to meet their food needs” [12]. Thus, food security involves food free of hazards and with appropriate nutritional value. Both can be considered as part of the quality that in addition involves food authentication and adulteration, food characteristics that consist of food nutrients (proteins, lipids, carbohydrates, vitamins and minerals), food additives, contaminants, substance forming during food processing but also values such as sensory characteristics (e.g. taste of food, smell, etc.) and convenience (e.g. easy in preparation, etc.). There is also a strong relation between environment and food quality and safety since intensive food production damages the environment by depleting the soil and drastically reducing marine ecosystems and their biodiversity [10, 11, 13]. In addition, overexploitation of natural resources has put food security and the availability of clean water at risk. Moreover, harmful environmental effects involving climate change and pollution are affecting food safety and quality, as increased temperatures attract greater use of chemicals and change the pattern of biological contaminants in areas and can alter the nutrient composition of food [1, 5, 14].

The third decade of the twenty-first century has started recently, and the challenges related to food and environment that lie ahead, as set out in the UN’s 2030 Agenda for Sustainable Development, are many [6]. Figure 1 shows the 17 Sustainable Development Goals (SDGs). The environmental aspect of development of the SDGs cover natural resource management, climate change, water-related issues, marine issues, biodiversity and ecosystems, circular economy, environmentally sound management of chemicals and waste, and many other topics [15–17]. Three of these goals are primarily food safety and quality related: SDG 3 “Good health and well-being” focuses on health (only achievable through a proper nutrition), while SDG 2 “Zero Hunger” involves the access of the whole world to sufficient food and encompasses eradication of malnutrition and nutrition associated disease and SDG 6 “Clean water and sanitation” is a pre-requisite to favor the achievement of the other two [18–20]. This global action plan approved in 2015 sets out specific measures to achieve a fairer, more prosperous, and environmentally friendly world within 10 years. In this sense, the UN itself is concerned of whether we are or not behind



**Fig. 1** Summary of the Sustainable Development Goals (SDGs) as reported by the United Nations (UNs)

schedule, and the question now is whether there is still time to ensure food safety and quality saving the planet [6].

In the environmental and food fields, mass spectrometry (MS) provides many of the data that underlie policy decisions directly influencing the health of the people and ecosystems [21]. Recent developments and improvements in MS enable it to provide a faster and more sensitive, accurate, and precise analysis of contaminants and components in food and environmental samples [22–25]. Methods combining MS with different techniques, such as inductively couple plasma (ICP), gas and liquid chromatography (GC and LC), pyrolysis (Py), elemental analyzers (EA), alone or several of them altogether, have proven to be very suitable for the analysis of food and environmental samples [26–35].

With the improved performance of techniques and instrumentation within MS, emerging strategies [such as chiral analysis, ion mobility, compound-specific isotopic analysis (SCIA)] and omics applications (such as lipidomics, metabolomics, and proteomics) have been developed [28, 30, 32, 36]. Contemporary MS progresses also include ambient-ionization MS and imaging (both could reach direct food analysis) and the transformation of bench-top mass spectrometry in lab-on-chip [37–40]. MS has also made an effort to be involved in what is coined green analytical chemistry (GAC) [41, 42]. Although this technique cannot be considered as a green technique in the “strict sense” because it has a high consumption of energy, toxic reagents (especially if hyphenated to LC), gases, and time, it has a high identification power and high-analyte throughput and then, can reduce the use of additional techniques for confirmation [43]. The most classical instrumental formats (ICP-MS, GC-MS, LC-MS, IRMS) are at the present time built using less material

(then, they are smaller and lighter) with more energy efficient and fewer consumption of gases and hazardous chemicals to complete the analysis generating less waste without loss of efficiency that is also important [44, 45]. Recent developments such as ambient-ionization MS and imaging are solvent free and reduced importantly the hazardous wastes, and lab-on-chip is clearly included within one of the most relevant principles of GAC that indicates the need to use automated and miniaturized methods [42, 43, 46]. With all these progresses, MS-based techniques are in the state-of-the-art technique and have become essential in both food and environmental analysis, (1) for determining chemicals in environmental and biological compartments and to determine these chemicals as well as endogenous compounds in food analysis [4, 47], (2) for identifying unknown chemicals using high-resolution mass spectrometry and scan-dependent and -independent MS/MS techniques [48, 49], and (3) for helping to fill the knowledge gaps regarding the molecular mechanisms of toxicity and apical effects caused by a certain chemical [11, 47]. Thus, MS plays an important role in both fundamental and applied research in food and environmental analysis, supporting environmental risk assessment and management as well as food safety.

This chapter presents the history and applications of different types of mass detectors, and how these detectors have been combined or hybridized with other separation techniques to achieve a wider range of applications. This history highlights new advances in recently developed mass spectrometry (MS) methods, such as high-resolution mass spectrometry (HRMS), ambient ionization mass spectrometry (AIMS) for direct food analysis, mass spectrometry imaging (MSI), and lab-on-a-chip technologies. It also discusses and compares the advantages and limitations of different MS techniques in their applications to the environment and to food safety and quality (these applications include metals, contaminants, food components, omics, chiral separation, macromolecules, etc.). Finally, this chapter discusses the future perspectives of the different techniques (many of which will be discussed in depth in the various chapters of this book) and the evolution of applications.

## 2 Developments in Mass Spectrometry and Current Applications in Environmental and Food Analysis

Mass spectrometry is an important tool for ensuring environmental and food safety [4, 39, 49–58]. Thanks to this technology, hazards ranging from pesticides to heavy metals and toxins can be identified, even if they are only present in minute quantities [54, 55, 58]. Advances in MS have enhanced the potential for analysis and the important role it has played and will continue to play in determining sample characteristics and ensuring food and environmental safety [49–51, 56, 58]. There have been different improvements in MS along history that have marked advances in the analysis of food and environmental samples [21, 26, 59–61]. Table 1 summarizes these achievements. The enormous structural diversity and huge dynamic range of



**Table 1** Advances in mass spectrometry have been a milestone in environmental and food analysis

Year	Advancement
1987	Determination of mass of the electron by J.J. Thomson
1912	First mass spectrometer (J.J. Thomson) (Faraday cup)
1918	“Directional focusing” mass spectrograph developed
1934	First double-focusing magnetic analyzer
1946	Discovery of the principle of analysis by measuring the time-of-flight of ions Martin and Synge described increasing powerful methods to separate amino acids and other organic compounds
1948	Martin and James developed gas chromatography
1949	Herzog and Viehb established the fundamentals of secondary ion mass spectrometry (SIMS)
1951	Gas chromatography starts to be used in the analytical laboratories
1953	Off-line pyrolysis GC reported Integrated pyrolysis mass spectrometry developed Patented quadrupole and ion-trap analyzers
1955	Introduction of the electron multiplier
1956	First commercial time-of-flight (TOF) mass spectrometry
1959	First coupling of gas chromatography-mass spectrometry (GC-MS)
1961	Liebe and Herzog developed the first SIMS device Pyrolysis gas chromatography off-line mass spectrometry reported
1966	First library of mass spectra published Horváth and Lipsky described the fundamentals and first application of ionic exchange chromatography Chemical ionization discovery First integrated pyrolysis-GC-MS reported
1967	Description of the high performance liquid chromatography (HPLC)
1968	First electrospray (ESI) interface for mass spectrometry
1971	First routine GC-MS with magnetic sector
1972	Introduction of the HPLC commercial devices connected on-line to a detector
1973	Tandem mass spectrometry (MS/MS) described Horning and Coll reported atmospheric pressure chemical ionization (APCI) Advent of reflection time-of-flight mass spectrometers to avoid energy spread
1976	Moving belt interface – First commercial LC-MS interface. Introduction of fused silica capillary columns in GC A combustion chamber between GC and MS incorporate.
1978	First NBS (NIH) database published First commercial MS/MS instrument
1983	First commercial ICP-MS
1984	Vestal developed the first LC-MS with thermospray interface
1985	Coupled laser ablation inductively coupled plasma mass spectrometry (LA-ICP-MS) Hillenkamp and Karas reported first matrix-assisted laser desorption/ionization (MALDI)
1986	First single particle (SP)-ICP-optical emission spectrometry (OES) reported
1987	First GC-triple quadrupole (MS/MS) reported
1988	Tanaka at Shimadzu introduced the first commercial MALDI-instrument

(continued)

**Table 1** (continued)

Year	Advancement
	First commercial GC-C-IRMS was released
1989	First LC-MS commercial instrument with electrospray (designed by Fenn)
1990	Coupling of GC with new low- and high-resolution time-of-flight (TOF)
1991	Frysjnger and Gaines first reported GC x GC
1993	SP implemented in ICP-MS
1995	First commercial LC-orthogonal accelerated (oa)-TOF-MS
1997	Micromass introduced the first commercial LC-QqTOF-MS Caprioli and Vanderbilt published the first papers on MALDI-imaging
1998	Kinghorn and Marriott proposed the first cryogenic modulator for GC x GC
2001	First GC-oaTOF-MS reported
2002	Application of ion mobility mass spectrometry (IMMS)
2003	First GC with atmospheric pressure photoionization (APPI) reported
2004	Cooks introduced desorption electrospray ionization (DESI) Commercialization of liquid chromatography-IRMS (LC-IRMS) Application of electron transfer dissociation (ETD)
2005	Introduction of orbitrap
2005	Commercialization of direct analysis in real-time (DART) interface
2006	Cooks' team developed DESI-MS imaging
2010	First coupling of GC-MS with quadrupole time-of-flight (QqTOF) reported
2012	Petersen and Coll coupled GC-MS with an orbitrap

food and environmental matrices provided analytical challenges for analysis of chemicals present on them [30, 62–64]. The reported achievements in MS technology greatly expand the capability of MS-based approaches for food analysis [21, 50, 65].

The interest in knowing the chemical composition of food is well before the interest in environmental problems [12, 66–68]. The first developments in the field of food analysis date back to the late 1800s and include numerous methods based on solvent extraction, acid digestion, and weight difference determination [21, 68]. From 1900 to 1940 food science was immersed in the chemical characterization of the nutrients –carbohydrates, lipids, proteins, vitamins, minerals – as well as in the discovery of the functions that these components had in foods [67, 68]. From this arose the need to develop analytical methods that could verify the nutritional quality of foods. However, the real revolution in food analysis began in the 1950s with the application of the first instrumental techniques such as chromatography [69]. These in combination with mass spectrometry marked a clear milestone in food analysis [52, 70] enabling to extend food analysis to other, not necessarily nutritional, food components [53]. In addition, chromatography-mass spectrometry allowed the development of methods to ensure food safety, prevent fraud, control the presence of undesirable substances, and ultimately guarantee quality and safety.

Comparatively, interest in environmental preservation and the study of environmental problems is later began in the early 1970s [54]. In fact, the beginning of environmental concern is associated with the publication in 1962 of Rachel Carson's book "Silent Spring," which brought to the forefront the problem of the extensive use of pesticides [71]. A few years later, the Environmental Protection Agency (EPA) was created in the USA and in the early 1970s the first organochlorine synthesis pesticides were banned [72].

Despite this time difference between the two disciplines, the advances in both fields are linked to the development of organic mass spectrometry [47]. Gas chromatograph combined with single quadrupole mass spectrometry (GC-MS) was the first technique that marked a revolution in both fields [32, 33]. Until the early 1990s, GC-MS (commercialized since 1968) was the only technique capable of reaching the required detection limits with the necessary selectivity and specificity to be able to detect minority compounds in very complex matrices, such as environmental and food samples [73, 74]. GC-MS worked especially well in the detection of volatile and thermostable compounds such as many organic molecules [36, 41, 75, 76]. In addition, as there were no robust alternatives, numerous derivatization reactions were developed to transform non-volatile molecules into volatile ones, thus greatly extending the range of determinable molecules [36]. For small organic molecules, the combination of liquid chromatography and mass spectrometry (LC-MS) in the late 1980s was a breakthrough [30, 31, 77]. Overall, the 1980s produced major advances in interface and ionization techniques that greatly helped to improve the applications of mass spectrometry [78–80]. In this respect, it is only necessary to recall that in 2002, Fenn and Tanaka received the Nobel Prize for the development of the electrospray source in the 80s that made it possible to combine liquid chromatography and ionization techniques [58]. Both combinations, LC-MS and GC-MS have allowed the separation of organic molecules, providing an upstream separation that incorporates from the use of chromatographic phases that allow the separation of chiral compounds to the new multidimensional separation techniques such as GC x GC and LC x LC that always facilitates the determination of compounds in complex mixtures [76, 81, 82].

During the 80s decade, the combination of the isotope ratio mass spectrometry (IRMS) to an elemental analyzer (EA) was also developed [74]. The term "continuous-flow isotope ratio mass spectrometry" (CF-IRMS) was introduced in 1988 when Preston and McMillan coupled an elemental analyzer to an isotope ratio mass spectrometer via a variable leak [83]. The gas chromatography-combustion-IRMS (GC-C-IRMS) was developed even before but commercialized in 1988 [84]. In this case, since its commercialization, the systems are well established [85]. However, there are major advances in applications, especially in food analysis (geographical origin, authenticity, etc.) and environmental analysis [85]. Although to a lesser extent, there are also advances in instrumentation such as the combination of IRMS and LC [83].

The ICP source coupled to a quadrupole-based mass analyzer was introduced commercially in 1983 [27]. However, it was not until 2001 that this instrument was equipped with collision/reaction cells that increased the capability of the technique

to process samples with a high matrix content. In the following years, ICPMS has become one of the techniques capable of detecting more than 70% of the elements in the periodic table [86]. Thanks to these instruments, the analysis of metals and metalloids is practically solved except for few cases. One of these cases where analysis is not fully resolved yet is the speciation of the different forms of inorganic species. In this field, ICP is also useful but became an on-line detection tool for common separation techniques, such as GC, LC, and CE. There is also recent instrumental advance within this technique, such as the combination of laser ablation (LA) and ICPMS, which is a particularly useful technique for mapping [13] and the single particle (SP)-ICP-MS [86–88].

The decade of the 90s was marked by the appearance of tandem mass spectrometry, especially triple quadrupole (QqQ)-MS/MS, which increased the specificity, selectivity, and sensitivity of the determination [29, 30]. This has led to an increase in the capacity of these techniques to improve the analysis as they are able to simultaneously determine many compounds based on the use of generic extraction methods and a very specific determination by mass spectrometry coupled or not with a previous separation technique [31, 57, 77].

Since the 2000s, food and environmental applications have experienced a movement from low to high-resolution mass spectrometry [89, 90]. This has allowed to shift workflows from target to non-target analysis and to exponentially increase the number of determinable compounds. The ability of HRMS to provide information on the most probable empirical formula as well as on characteristic fragments of the structure helps not only to increase the number of determinable compounds but also to identify unknown compounds [79]. This high throughput option has also opened the field of “omics” within food and environmental analysis [58]. The various mass analyzers, which were originally designed to be combined with liquid chromatography, have been incorporated into other separation systems such as GC-MS or ICP-MS [78]. In this context, chemometrics plays a key role in resolving some issues related to food authentication or traceability, being a useful tool for food scientists to classify unknown samples with similar characteristics for the determination of the authenticity and to associate a food product to the place of origin in the context of traceability [91].

Also in the 2000s, the development of environmental ionization occurred. This includes ionization techniques capable of achieving desorption and ionization of analytes directly from the surface of an untreated sample [35]. The first desorption source was electrospray ionization (DESI), this was closely followed by direct analysis in real time (DART), a plasma-based environmental ionization technique. Since then, a number of desorption sources with different principles have been developed [54]. AIMS allows direct and rapid analysis with no or minimal sample preparation and is particularly applicable to in situ analysis, enables direct analysis of intact substances, and has been developed through portable instruments.

The determination of macromolecules has also brought other advances to mass spectrometry, including the incorporation of pyrolysis prior to separation and determination, the use of AIMS, ESI, or MALDI sources and finally leading to MSI, which has allowed the evaluation of the distribution of both macro and small

molecules in the matrix. These methods have opened new opportunities for the MS of macromolecules by greatly extending the molecular size and type of samples amenable to MS [2, 92].

### 3 Applications of Mass Spectrometry in Environmental and Food Analysis: Current Status and Future Prospects

The applications of mass spectrometry in environment and food have grown exponentially since its introduction. Table 2 shows the different applications that can be found depending on the type of technique.

The determination of the elements of the periodic table is covered by two techniques: ICP-MS and the IRMS– with all their different advances and combination with other chromatographic techniques. ICP-MS is an inorganic elemental and isotopic analysis technique capable of determining and quantifying most of the elements of the periodic table in a linear dynamic range of six orders of magnitude (ng/L – mg/L) in addition to being able to perform elemental determination in a multi-elemental analysis that provides the composition of the analyzed sample. It can also perform isotopic composition quantification and trace isotope stability studies and most importantly, speciation studies where the technique is able to determine the different species of the metal and to analyze nanomaterials. IRMS is devoted to determining the isotopic analysis of stable isotopes of different molecules. Both cover a wide range of applications, the former to determine the concentration of metals in food and environment and the latter to study the isotope ratio of stable C, O, N, and H. These elemental species techniques are more well-established than those related to organic molecules. However, there are still several pending issues that are not well resolved yet. One of these pending issues is the speciation analysis of the elements present in food and the environment by combining ICP-MS with efficient separation techniques, such as LC, GC, or on-line coupling of flow field-flow fractionation ( $F^4$ ) [93, 104–107]. The relation between safety and Human Health risk and the element form is well-known but still a concern for the scientific community due to the lack of knowledge about some aspects as the relation between food and environment. Interestingly, one of these efforts to contribute new knowledge described the biogeochemical role of Hg in the soil–plant system and associated health risks, demonstrating the need and the interest of these studies [105]. Other is the release of engineered elemental metal nanoparticles (EMNPs) into the environment. EMNPs pose potential risks to living organisms, including human beings. Among the various techniques reported for the characterization of EMNPs, on-line coupling of  $F^4$ -ICPMS – based on size separation – has been well established for the simultaneous separation, identification, and quantification of EMNPs, especially in the complex matrices of environmental samples [56]. However, the real leading technique for the determination of EMNPs in food and environment is single particle (SP)-ICP-MS. In SP-ICPMS each recorded pulse is

**Table 2** Applications of mass spectrometry within food and environmental analysis

	Technique	Application	Reference
Elemental analysis	ICP-MS	<ul style="list-style-type: none"> <li>Determination of elements of the periodical table in gas, solid, or liquid matrices</li> <li>Speciation of the different metal's forms</li> <li>Analysis of EMNPs in food and environment</li> <li>IRMS of non-conventional stable isotopes</li> </ul>	[26, 86–88, 93]
	IRMS	<ul style="list-style-type: none"> <li>Determination of the ratio of stable isotopes (C, H, O, S) in bulk samples</li> <li>CSIA in food and environmental samples' specific compounds (e.g. lipids, proteins, organic matter, water, etc.)</li> </ul>	[74, 83–85, 94]
Small organic molecules	GC-MS	<ul style="list-style-type: none"> <li>Volatile and thermostable compounds [e.g. volatile organic compounds (VOC), contaminants, lipids, etc.]</li> <li>Non-volatile compounds transformable in volatile by chemical reaction → derivatization</li> <li>Separation of chiral compounds</li> <li>“Omic”: metabolomic and lipidomics</li> </ul>	[24, 32, 36, 41, 75, 76, 81]
	LC-MS	<ul style="list-style-type: none"> <li>Determination of any type of organic molecules (nutrients, endogenous compounds, contaminants, toxins, etc.)</li> <li>Separation of chiral compounds and ionic species</li> <li>“Omic”: metabolomic and lipidomics</li> </ul>	[23, 30, 31, 57, 58, 77–80, 89]
	IM-MS	Additional separation step for mobile ions. For compounds that elute together and need an additional separation	[59, 92, 95–99]
	AIMS	<ul style="list-style-type: none"> <li>Determination of organic molecules (nutrients, endogenous compounds, contaminants, toxins, etc.)</li> <li>“Omic”: metabolomic and lipidomics</li> <li>Analysis of sample surface (no extraction)</li> <li>On-site analysis</li> </ul>	[35, 37, 38, 54]
	Imaging	<ul style="list-style-type: none"> <li>Visualization of the molecule in its environment</li> <li>“Omic”: metabolomic and lipidomics</li> <li>Organic molecules</li> </ul>	[92]
Macromolecules	LC-ESI-MS	<ul style="list-style-type: none"> <li>Bottom-up and top-down approaches</li> <li>Omics: proteomics</li> </ul>	[90]
	MALDI-TOF	<ul style="list-style-type: none"> <li>Omics: proteomics</li> <li>Food allergens, bacteria, etc.</li> </ul>	[100, 101]
	Py GC-MS	<ul style="list-style-type: none"> <li>Pyrolysis transforms macromolecules in their units</li> <li>Organic matter, microplastics</li> </ul>	[102, 103, 139]

(continued)

**Table 2** (continued)

	Technique	Application	Reference
	AIMS	<ul style="list-style-type: none"> <li>• Analysis of sample surface (no extraction)</li> <li>• On-site analysis</li> </ul>	[35, 37, 38, 54]
	Imaging	<ul style="list-style-type: none"> <li>• Visualization of the molecule in its environment</li> <li>• “Omic”: proteomics</li> </ul>	[55, 92]

a single NP. The frequency of the pulses is directly related to the NP concentration (number of nanoparticles in solution) and the intensity of each pulse to the mass of the element, in fact to the number of atoms, in each detected NP (size) [64, 86]. Finally, another aspect of interest is the determination of the ratio of stable isotopes of non-conventional elements (Li, B, Mg, Ca, Fe, Mo, etc.). Among the different techniques, thermal ionization mass spectrometry (TIMS), secondary ion mass spectrometry (SIMS), and multiple collector (MC)-ICP-MS are mainly used for isotope ratio measurements. The determination of the ratio of stable isotopes (C, H, O, and N) is performed by IRMS. The most novel aspects involve compound-specific isotope analysis (SCIA), which is mostly performed by GC-C-IRMS. This instrument achieves first separation of the specific compounds and then, determination of the compound's isotope ratio. CSIA can be applied to food and environmental samples generating data on key components such as aroma compounds, sugars, amino acids, carbon dioxide (in carbonated beverages), organic matter (in soil) as well as on contaminants, such as pesticides. The enormous versatility of GC-C-IRMS is evidenced by the numerous applications that are increasing exponentially with the growing versatility of these techniques and it can be expected that it will become a pivotal technique in the detection of fraud, either alone or in combination with other analytical methods [74].

Small organic molecules are determined mostly by GC-MS [75] and LC-MS [69] but also can be done by AIMS [75, 108] or MSI [40]. The technique will depend on the purpose of the analysis because the two latter were designed to provide spatial information on the location of the compound. Advances in MS instruments have provided a substantial enhancement in sensitivity, scan speed, and quantitative accuracy, making it a useful platform for analysis of complex food composition, including nutrients and other endogenous compounds (e.g. amino acids, polyphenols, lipids, vitamins, carotenoids), contaminants (e.g. pesticides, heterocyclic aromatic amines, polycyclic aromatic hydrocarbons, plastic migrants), and toxins (e.g. alkaloids, phycotoxins, phytotoxins, and mycotoxins) [50–53, 70, 80, 89, 109]. In environmental analysis, these techniques have been focused on the determination of micropollutants [3, 4, 47]. LC- and GC-MS(MS) instruments today are continuously improving to increase the range of contaminant classes and physico-chemical properties that can be determined with a single analytical method providing a broad scope of analysis. As a consequence, there is a transition in both, GC and LC, from the triple quadrupole instruments that have been most widely used in analytical laboratories in the last decade, to high-resolution MS systems to avoid the inherent

limitations of QqQ instruments in terms of restricted number of simultaneous transitions due to the scanning speed and inability to detect compounds not selected “a priori” [110]. Analysis with modern HRMS instruments is able to detect and quantify virtually unlimited amount of analytes based on the full scan data, plus retrospective analysis [34]. Recent improvements in sensitivity of HRMS instruments put them on par with modern QqQ-MS/MS but with much better opportunities to detect a wider range of compounds, as to the traditional “target analysis” they add the possibilities of broad screening (against a database that may contain thousands of compounds) and the identification of unknown compounds (much more laborious). In this sense, the different modes of information acquisition: data-dependent and data-independent (DDA and DIA), as well as the versatility of schemas, have greatly increased the amount of information acquired [111]. These increased capabilities to explore the sample composition beyond the traditional analysis along with the explosion of methods based on chromatography hyphenated to HRMS, which provides enough sensitivity and selectivity to cover a wide range of metabolites and other compounds in complex food and environmental samples, are responsible of the boom of metabolomics [25, 112]. In this sense “omics” techniques represent a bridge between small molecule and macromolecule analysis. The introduction of metabolomics in the case of small molecules (molecular weight <1,500 Da) has allowed to work with profiling (target) and fingerprinting (non-target) techniques [25]. Metabolomics identifies differences in small molecules (biomarkers) between a control group and the group of interest (subject to an exposure, a treatment, or any type of difference), in terms of any other endogenous and/or exogenous small molecule whose presence/absence or altered concentration between the two groups may make a difference [112, 113]. The similarities and differences among the samples are established by unsupervised pattern recognition algorithms [principal component analysis (PCA), hierarchical cluster analysis (HCA), etc.] and supervised pattern recognition algorithms [linear discriminant analysis (LDA), partial least squares discriminant analysis (PLS-DA), soft independent modeling of class analogy (SIMCA), the Random Forests (RF), machine learning algorithm, etc.] [112]. The successful implementation of these methods is also related to software improvements that simplify to deal with an enormous amount of generated data. Metabolomics tools are now being used to study volatiles (via GC methods) as a new tool for understanding aroma and flavor chemistry and the mechanisms involved in the formation of these flavor compounds in order to understand better the beneficial and less beneficial aspects of fermentation and other food processing methods [33, 63, 114]. Metabolomics research using GC/MS in food also includes the analysis of non-volatile compounds such as sugars, amino acids, and organic acids that achieved discrimination of food samples based on cultivars and authentication of food samples to prevent food fraud, characterizing the profile of food samples to provide a general overview of the sample, evaluating stress-response, optimizing postharvest processes based on metabolic changes, monitoring changes during growth and food processing, evaluating and predicting food quality, and evaluating food shelf-life [32]. However, untargeted metabolomics by LC-MS is increasingly used to discover metabolic markers for food authentication in routine



control and the assessment of biological identity, geographical origin, agricultural production, processing technology, freshness, cause of animals' death, and so on, through three main steps, data acquisition, biomarker discovery, and biomarker validation [30, 113]. Within environmental analysis, metabolomics is mostly devoted to the identification of different biomarkers of the anthropic influence in the environment. The applications in wastewater-based epidemiology are one of the most relevant examples because these biomarkers can give particular fingerprinting for populations [115]. Difficulties of the implementation of non-target metabolomics in other environmental samples are related to the low concentrations of contaminants present that prevent full development of the technique [10].

The other techniques that can be applied to the analysis of small molecules and macromolecules are AIMS and MSI. These techniques were designed to provide spatial information, that is to offer information on the position of the molecules of interest within the matrix. AIMS techniques were developed to detect the analytes in the surface of the sample and have shown substantial potential in both environmental and food analysis. However, now their spatial dimension is not exploited in most applications and AIMS techniques are just considered as highly suitable for the analysis of large sample sets – especially for qualitative screening, sometimes providing additional (at most) semi-quantitative information due to the increasing implementation of a previous sample preparation (clean-up and/or preconcentration). Food and environmental applications have been mostly focused on the determination of contaminants, such as pesticides, pharmaceuticals, brominated flame retardants, etc. Furthermore, within food quality, these interfaces have been used to study dried fruits used in traditional Chinese herbal medicine, for the authentication of fruit juices, to determine the pungency of peppers, a characteristic that can be caused by capsaicinoids, amino acids in food, etc. [35, 38, 116]. This sample preparation step has become almost mandatory as it decreases sample complexity and concentrates the analyte to meet the sensitivity requirements for analysis but most of the times eliminates the spatial information. Sample preparation is already implemented in most of the applications to food analysis and is even more pronounced in environmental analysis, where (ultra-)trace analysis is the norm. The analysis of the data by chemometric tools that helps to understand and classify the complex information obtained has increased the number of applications related to metabolomics [37]. Although all the limitations reported, AIMS techniques could play an important role in “omics” related analytical chemistry – mainly due to their ease of use and the possibility to process a large number of samples within a short time [117]. AIMS techniques are also well suited to meet the challenges related to on-site analysis. Ideally all parts of the equipment (i.e., the ion-source as well as the MS instrument with all its peripheral devices) should be portable. This would become a clear advantage of these devices for mass spectrometry.

Unlike AIMS, imaging mass spectrometry's (IMS) ability to image molecules in space is its main attraction. MSI is a prevailing tool that enables untargeted investigations into the spatial distribution of molecular species in a variety of samples [118]. Thousands of molecules, such as metabolites, lipids, peptides, proteins, and glycans, can be recorded as imaging in a single experiment without labeling. The

combination of information gained from mass spectrometry (MS) and visualization of spatial distributions in thin sample sections makes this a valuable chemical analysis tool for food and environmental specimen characterization. MSI can study almost anything owing to the wide variety of ionization choices and sample preparation strategies that can be paired depending upon your analyte(s) of interest. From pesticides to peptides and microplastics to metabolites, the future of MSI promises to extend the answerable questions in environmental and food science beyond those that were previously limited to temporal and quantitative data sets [40, 55]. Implementing MSI in environmental and food science research space means that spatial data will become part of routine analysis [92].

The determination of organic macromolecules is another field where MS-based techniques have a growing role. Within this field, omic technologies provide a holistic view of the molecules that make up a cell, tissue, or organism. Omic technologies in the case of macromolecules aimed primarily at the universal detection of genes (genomics), mRNA (transcriptomics), proteins (proteomics), in a specific sample in a non-targeted and non-biased manner. These techniques are named “foodomics” when integrated into food and nutrition and effectively provide analytic information about the effects of food on human health and help to improve food processing, quality control, traceability and authentication. In the field of environmental sciences, omic technologies will provide a better understanding of the impact of complex mixtures of contaminants integrated with other environmental stressors. Nevertheless, “omic” strategies still provide many challenges, such as the bioinformatic tools that are still evolving. There is a lack of standardized methodologies, and it is necessary to establish carefully designed experiments, accompanied by appropriate analytical techniques and statistical analyses.

Within food analysis, LC-ESI-MS techniques with both high-resolution and low-resolution tandem mass analyzers are applied to the identification and detection of qualitative and quantitative information about the proteome, in order to extract new integrative information from the complex multivariable space of omics [2]. The two principal workflows to recognizing and typifying proteins using MS are the “bottom-up,” which analyze peptides after proteolytic digestion, and “top-down,” which analyze intact proteins. Bottom-up indicates that information about the constituent proteins is reconstructed from individually identified fragment peptides. This new information is necessary to achieve a higher level of understanding of processes in food science and technology; consequently, in new functions of food and improved markers of food quality and safety to completely transform the concept of food safety. As an interesting example, Stachniuk et al. [90] presented sets of myofibrillar and sarcoplasmic proteins, which turned out to be the source of 105 heat-stable peptides, detectable in processed meat using LC-MS/MS. A list of heat-stable species-specific peptides was compiled for 11 types of white and red meat including chicken, duck, goose, turkey, pork, beef, lamb, rabbit, buffalo, deer, and horse meat, which can be used as markers for meat authentication. Great efforts are ongoing to clarify the allergenic mechanisms of the diverse allergenic proteins of food origin, aimed to both designing suitable therapies and for a timely and precise diagnosis of the allergic condition [48, 119]. Among the other omics sciences, mass spectrometry

(MS)-based proteomics is gaining a steadily increasing interest by the whole scientific community because of its high versatility. Peptide marker identification is one of the most important steps in the development of a mass spectrometry (MS)-based method for allergen detection, since the robustness and sensitivity of the overall analytical method will strictly depend on the reliability of the proteotypic peptides tracing for each allergen. Pilolli et al. [120] described the results of a comprehensive evaluation of the current literature on MS-based allergen detection aiming at collecting all available information about proteins and peptide markers validated in independent studies for milk, egg, peanut, soybean, hazelnut, and almond. Within the field of environmental analysis, proteomics has been applied to the identification of macromolecules in wastewater analysis including also the proteins that determine the presence of residues of some animal species [115].

The other technique highly used in proteomic and also in the identification of other macromolecules is matrix-assisted laser desorption/ionization (MALDI) mass spectrometry (MS) that has become an indispensable tool in the analysis of macromolecules, such as proteins and peptides. Some biomolecules are too large and can break down when heated, and traditional techniques fragment or destroy the macromolecules. In the case of MALDI-MS, macromolecules of interest are embedded in a matrix of small molecules. Then, MALDI is a soft ionization in which a laser is incident on the matrix that contains analytes so that these macromolecules pass into the gas phase without fragmenting or decomposing them. MALDI-MS is commonly combined with time-of-flight (TOF)-MS and is appropriate to provide the peptide mass fingerprinting (PMF), as well as the selected fragmentation of various precursors using collisional-induced dissociation (CID) or post-source decay (PSD). Recognition of food adulterations, description of food allergens, identification of food spoilers and food-borne pathogens, safety assessment of genetically modified foods and investigation of protein structural modifications induced by various industrial processes that could be an issue in terms of food quality and safety are some of the last key applications of these techniques in food analysis [33, 101]. Furthermore, MALDI-MSI has found within the environmental research various applications in different fields, e.g., animal and plant biology, entomology as well as microbiology. In the latter, despite a simpler cell structure, MALDI-MSI has been applied to unravel the complex microbial communication network and the correlation between the localization of certain molecules and their biological function [55].

The other technique applied to characterize macromolecules is Py-GC-MS, recognized as a valuable technique in food and environmental chemistry [121]. Macromolecules are broken down by pyrolysis into simpler molecules that can be separated by GC and detected by MS [102]. The versatility of Py-GC-MS offers ample possibilities to be exploited in multiple applications related to food industry, from food authentication to the analysis of composition, quality, and additives, including contaminants as well [121]. In environmental analysis, Py-GC-MS has been applied to the chemical characterization of organic matter compounds in soil and sediment but also to assess the possible modifications that disturbances as forest fires or land use changes can cause in such compounds. Evaluation of the contamination by micro and nanoplastics of environmental samples has also become a hot

topic addressed by the application of this technique [102]. Most of the reported studies are focused on the robustness of the pyrolyzer and therefore, latest innovations in GC-MS have not been yet introduced but it is expected that Py-GC/MS takes advantage of the great separation power of GCxGC, the speed of rapid GC, and the identification capability that HRMS and HRMS in tandem can provide [103]. The large number of results obtained and the difficulty in interpreting them are constraints that also need to be considered [102].

There are two global trends related to the advances in the techniques that deserve to be highlighted. One is ion mobility spectrometry (IM-MS) that has grown dramatically over the last decade [95]. Ion mobility spectrometry (IMS) consists of separating ions in an electric field in the presence of a collision gas that separate them according to their shape (three-dimensional arrangement) and electrons [122]. Since both, IMS and MS are performed on ions in the gas phase, they can be easily combined into a single instrument – an ion mobility-mass spectrometer (IM-MS). IM-MS has been implemented in almost all the techniques related to mass spectrometry (e.g., LC-IM-MS, GC-IM-MS, IM-AIMS and IM-MSI, MALDI-IM-MS) [92, 96–99]. This can give an idea of its usefulness. Currently, the food industry utilizes IMS in nutritional analysis, food safety, food fingerprinting, and process control and quality assurance/quality control (QA/QC). Some of these categories are dominated by GC/LC-IM-MS instrument design and development, while others are focused on portable, inexpensive, and fast standalone IMS instrumentation. IM-MS enables the separation of molecules that have an isobaric mass, and in which chromatographic separation fails, becoming indispensable in these cases. Applications of IM-MS within the field of food chemistry involved metabolomics, proteomics, lipidomics including determination of lipopeptides as well as phenolic compounds, terpenes, volatile compounds, biogenic amines, etc. This has provided insights into the study on food control, food processing, authentication, and fraud [123]. In environmental analysis, it is a promising and powerful tool for the (non-) target and suspect analysis of small organic molecules in complex matrices.

The other current trend in mass spectrometry stems from the need to reduce pollution from chemical laboratories and to develop greener methods [41]. Chip-based separation devices coupled to mass spectrometry are a promising platform for food and environmental analysis, which will particularly replace traditional separation systems (LC-MS, GC-MS, and CE-MS) [42]. Compared to traditional systems, they offer several advantages: (1) they can provide quantitative and qualitative information of analytes at molecular level, (2) they have low cost and consume small volumes of samples and reagent, and (3) permit multi-step operations, since sample preparation, separation, and signal analysis can be simultaneously integrated into a single chip-based separation device [46]. Although important technology innovations aiming the fully-integration of chip-based separation devices and mass analyzers have been accomplished, their use for routine analyses is still under evaluation. Applications are still very scarce, but the prospects for the future of this reduction make research and improvement in this field necessary.

## 4 Conclusions

The full possibilities of MS within food and environmental analysis are far away to be reached. Short-term future trends of MS include the wider implementation of (ultra)high-resolution tandem mass analyzers providing high-mass accuracy, IM-MS (considering it as a second MS dimension), AIMS, and MSI among other widely applied techniques within this field. However, in the long term, it is expected that miniaturization, transportable systems that can measure at the sampling site and finally the implementation of the lab-on-chip will make the technique greener. It is expected that in the near future, the number of applications of these techniques will grow exponentially and that the advances in the mass spectrometers used will also develop rapidly.

Within the most classical, LC or GC combined with MS, challenges to analyze more contaminants faster, more efficiently, with low detection limits and to reduce false positive/false negative rates in complex samples place high demands on future improvements in instrumentation and techniques. Improvements in MS detectors' sensitivity, speed, selectivity, specificity, and wider dynamic linear range are expected. Features aimed to reduce instrumental downtime needed for maintenance, and new and improved ionization sources and interfaces, plus combinations of existing ion sources used simultaneously are expected.

Mass spectrometry is ideally suited for use in most areas of environmental and food analysis, including the determination of organic and inorganic contaminants, with special emphasis on emerging ones, characterization of environmental samples, analysis of food components, food processing, food quality/authenticity assessment, and analysis of food additives and contaminants. The actual and potential contributions of mass spectrometry to these areas are unlimited. Recent advances in mass spectrometry have provided new options to ensure food quality and safety, or to determine environmental contaminants, and to establish their fate in the environment as well as the characterization of different components of environmental compartments based mainly on profiling/fingerprinting, and metabolomics approaches. As recent progress demonstrates, chemometrics can be used in parallel with data collected from food samples to test food associations, to explore biological mechanisms, and to determine origin and authenticity. This will be also an area of future research within MS.

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# Food and Environmental Samples Handling and Preparation for Mass Spectrometry



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**Abstract** In current food safety and environmental monitoring, mass spectrometry methodologies are widely used for identification and quantification of various substances. However, studies have shown that various compounds in food and environment samples are found at ultra-trace concentration levels and the complexity of the sample matrix might limit the performance of selective and sensitive analytical method. Therefore, selection of a proper sampling strategy and sample preparation methods plays a critical role in achieving accurate and reliable results. Thus, this chapter is mainly focused on the recent trends in sample handling and preparation of food and environmental samples prior to MS analysis. Methodologies regarding mass spectrometry coupled with chromatographic separation have been reviewed to highlight the improvement in the analytical determinations of various compounds in food and environmental samples.

**Keywords** Environmental monitoring, Food samples, Sample handling, Solid phase extraction, Stir-bar-sorptive extraction

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## 1 Introduction

The developments in food and environmental sample analysis have become more challenging over the past decades [1, 2]. This is due to complexity of the sample matrices as well as presence of mixtures of substances, which are continuously changing. Food samples usually encompass wide range of physical types including biological matrices (such as meat), liquids (beverages, milk) dry powders and solids, among others [3]. Environmental samples, on the other hand, include water, sediments, soil, air, biological materials (plants), and wastes (liquids, solids, or sludges) [4]. The variability of the nature of these sample matrices thus demands rapid, reliable, accurate, and sensible analytical methods. Recently, major trends in food and environmental analysis are focused on the development of rapid and efficient methods for trace analysis of organic compounds (both targeted and nontargeted) [5]. Despite the recent advancements in analytical instrumentation (especially mass spectrometry, MS), sample handling and preparation prior to MS detection are still mandatory.

MS is one of the most powerful analytical techniques that are widely used for qualitative and quantitative analysis of a wide range of analytes in food and environmental matrices [6–8]. This is due to its attractive features such as selectivity, high detectability, and rapidity [6–8]. However, to achieve quality results from an MS instrument, classical analytical workflow including sampling, sample handling, sample preparation (extraction, pre-concentration, clean-up) as well as chromatographic separation is required [8, 9]. It is important to note that every step of the analytical workflow is critical for analytical method performance but bottleneck steps such as sample preparation and separation become the rate determining steps of the overall analysis process [10, 11]. In cases where rapid or urgent decisions need to be made (especially in the area of food safety), the use of fast, sensitive, efficient, and reliable analytical methods is critical [8, 12, 13]. Therefore, there is a call for development of alternative methods that minimize the lengthy analytical workflows. This is achieved by developing improved and efficient sample preparation steps while skipping the chromatographic separation, thus resulting to direct analysis of pretreated samples using MS [12, 14, 15].

Sample preparation prior to the determination of analytes in food and environmental matrices using MS involves critical steps, chiefly because of the complexity of the sample composition as well as low concentrations of many analytes [16, 17]. Therefore, the isolation of target analytes from food and environmental samples requires careful selection of a suitable extraction and pre-concentration procedure [18]. The selection of the sample preparation method is subject to complexity of the sample matrix, chemical structure, and characteristics of the analytes of interests [16, 17]. This chapter gives a summary of recent progress in regard to the development of sample handling and preparation methods in the areas of food and environmental analysis.

## 2 Sample Handling and Preparation

### 2.1 *Sample Handling and Collection*

Chemicals in food and environmental samples exhibit different physical and chemical properties. For instance, diffusion of different pollutants from various sources into water systems is strongly influenced by their characteristics such as partition coefficient, polarity, volatility as well as the properties of the interacting media [19, 20]. As the results, more hydrophilic substances are mostly found in water samples while hydrophobic compounds (those with  $\log K_{ow} > 4$ ) are retained in sediments and suspended matter [19]. Therefore, food and environment sample handling/collection becomes one of the most significant stages because improper collection, processing, and storage could alter the quality of the samples as well as chemical or molecular structures of the target analytes [21, 22]. Among the steps taking place prior to analyte extraction, detection, and quantification, sample collection plays a vital role because a sketchy sampling can lead to incorrect results regardless of flawlessness of the remaining analytical steps [23]. Therefore, the sampling step should be carried out in such way that any aliquot of food and environmental sample taken for analysis must be a good representative of the original bulk sample [23]. For example, it is recommended that for solid food and environmental samples an appropriate crushing must be used to attain smaller representative sample sizes [23, 24]. The amount of sample to be used for further analysis depends on the type of analysis required. For trace analysis in food and environmental samples, larger sample portions are normally required to achieve low detection limits. However, for the analysis of sample composition smaller portions of samples are often used [22, 25].

#### 2.1.1 **Sampling Strategies for Environmental and Food Samples**

Sampling of environmental samples depends on the nature of the sample. As such, several sampling strategies have been reported in the literature. These include passive sampling, grab or discrete sampling at a specific time, time-integrated sampling, and event-based sampling [19, 26]. Grab samples are used to provide a snapshot of the water body or solid food and environmental samples at the specific collecting time and location. Grab sampling method is simple and cost effective but it does not fully represent the status of the water body [19, 26–28]. Once the samples are collected by grab sampling technique, they are usually kept in pre-cleaned bottles (glass amber bottles for organic substances and polypropylene bottles for trace metals analysis) and stored at 4°C until pretreatment. Time-integrated sampling techniques are individual time- or flow-proportional samples collected in individual containers [19, 26]. The collected time-integrated samples are combined to form a composite sample before they are subjected to a specific sample preparation method [19, 26]. Event-based sampling is often conducted at times when high concentration

of pollutants is expected [19, 26]. These events include heavy chemical spills, heavy rainfalls, oil spill, improper industrial and sewage discharges. Event-based sampling can include grab or in-situ time-integrated samples. Passive sampling utilizes samplers that are deployed in water over a set of time to accumulate pollutants in a time-weighted average or equilibrium concentrations [19, 28]. Passive sampling uses one device per selected study area as compared with liters of grab water samples [19, 28]. Passive sampling devices comprise of an adsorbent that is placed between two membranes which allows the analytes to pass through the sorbent and be extracted from the water [19, 28, 29]. Several review articles have extensively discussed the principles of different passive samplers as well as their application in environmental analysis [28–31].

## 2.2 *Sample Preparations*

Sample preparation is the major bottleneck in food and environmental analysis because of the complexity of the sample matrices and ultra-trace concentration of target analytes [32]. The selection of a suitable sample preparation method depends on various factors such as chemical properties of the analytes, sample matrix, and the analytical detection method [23, 33]. Recently, attempts have been made toward the development of various sample preparation methods from extraction and enrichment of numerous analytes with different chemical properties in food and environmental samples. These methods include solid and liquid phase-based extraction methods.

### 2.2.1 **Solid Phase Extraction-Based Methods**

Solid phase extraction (SPE) is one of the most used sample preparation techniques [34–36]. SPE is based on the separation and distribution of analyte(s) between a solid adsorbent and a liquid mobile phase [37]. One of the main advantages of using SPE-based methods is the diverse choice of sorbents and the ease of modification [38]. The main steps of SPE include activation of sorbent, analyte retention, removal of interferences, and elution of analytes [35]. These steps can be varied/modified depending on the nature of the sorbent used [39].

Traditional SPE has drawbacks such as solvent and time consumption [35, 40] and this has led to the development of modified SPE methods [42]. These include techniques such as (1) solid phase microextraction, (2) stir bar-sorptive extraction, (3) matrix solid phase dispersion, and (4) in-tube solid phase microextraction. These sample preparation techniques have been extensively reviewed in recent years [23, 34, 35, 38, 41–44]. The selection of a suitable analysis method depends on the sample matrix nature as well as the type of analytes. For example, in the head space solid phase microextraction (HS-SPME) there is no direct contact between the fiber and the sample [23, 45]. In contrast, in the direct immersion SPME (DI-SPME),

the extracting phase is directly immersed on the sample for direct analyte transfer [46].

### Application of Solid Phase-Based Extraction Methods in Food Samples

Currently, a simple “solid phase extraction” search on web of science returns 15,781 results. The same search modified to show studies from the last 5 years (2017–2021) gives 5,967 while modifying the search with “mass spectrometry (MS)” returns 3,305 results. This shows the vast usage of SPE-based techniques combined with MS for residue analysis. In the past years, solid phase-based extraction methods combined with MS methodologies have received growing interest in food analysis which include food contaminants and components (Table 1). These MS based methodologies include liquid chromatography mass spectrometry (LC-MS) [47], gas chromatography mass spectrometry (GC-MS) [48], high-resolution mass spectrometry (HRMS) [49–51], matrix-assisted laser desorption/ionization mass spectrometry (MALDI-MS) [52], LC coupled to tandem mass spectrometry-electrospray ionization or atmospheric pressure chemical ionization (LC-ESI-MS/MS or LC-APCI-MS/MS) [53], MALDI-time-of-flight tandem mass spectrometry (MALDI-TOF-MS) [54], LC-quadrupole-TOF (LC-Q-TOF-MS/MS) [55, 56], and GC-MS/MS [57, 58]. Table 1 suggests that a large number of MS-based method for analysis of food samples mostly utilizes SPE and other related procedures as preliminary extraction, clean-up and pre-concentration step. In these methods various adsorbent materials are selected based on the nature of the analytes of interest.

A study reported by Zhao et al. developed a magnetic solid phase extraction (MSPE) using polydopamine functionalized magnetic core-shell mesoporous silica nanoparticles ( $\text{Fe}_3\text{O}_4@n\text{SiO}_2@m\text{SiO}_2\text{-PDA}$ ) as an adsorbent. The developed MSPE method was coupled with ultra-high-performance liquid chromatography coupled with Q-Exactive high-resolution mass spectrometry (UHPLC-Q-Exactive HRMS) for extraction, pre-concentration, and determination of fipronil and its metabolites in egg samples. The method showed limits of quantification (LOQs) 0.006–0.010  $\mu\text{g}/\text{kg}$  and limits of detection (LODs) 0.002–0.004  $\mu\text{g}/\text{kg}$ . Repeatability 4.4–10.7% inter-day and 2.2–6.7% intra-day precision were obtained with recoveries in the range of 80.6–95.3% [59]. LC-HRMS coupled with dispersive micro-SPE was used for the determination of carcinogenic 4-methylimidazole (4-MEI) in food from roasting process and cooking. The analyte was extracted from vinegar, tea, coffee, and soy sauce [60]. The determination of Bisphenol A (BPA) carried out by LC-ESI-MS using a microwave assisted extraction showed relative standard deviation (RSD %) of 7–10% with low LODs in the  $\text{ng}/\text{g}$  level. The authors reported that concentrations of BPA in canned food were in the range of 7.3 and 42.3  $\text{ng}/\text{g}$  [61]. Liu et al. reported the use of novel core-shell mesoporous silica (CSMS) microspheres coated by titanium dioxide ( $\text{CSMS}@TiO_2$ ) as a sorbent for extraction of phosphorus-containing amino-acid-like herbicides from baby food. They developed a dispersive SPE coupled with capillary electrophoresis-mass spectrometry (CE-MS). LOQs were in the range of 0.3–1.6  $\text{ng}/\text{mL}$  with excellent repeatability (intra/inter-day,



**Table 1** Selected applications of solid phase-based extraction methods in food samples

Techniques	Samples	Analytes	Performance	References
SPE UPLC-MS/MS	Food	Bisphenol A	LOD:0.01 µg/L LOQ: 0.04 µg/L RSD <1% (n = 5) <3% (n = 5)	[66]
SPE-UHPLC-ESI-MS/MS	Milk, animal tissues and eggs	Gamithromycin	LOQ:0.80–1.0 µg/kg LOD:0.30–0.40 µg/kg	[67]
SPE-UHPLC-MS/MS	Beef, eggs, milk, pork, and lamb	Diuretics	LOQs: 0.53–2.43 µg/kg	[68]
MIP-PE HPLC-MS/MS	Pork	Sulfonamides (SAs)	Concentrations: 3.79–3.57 µg/kg	[69]
d-SPE-UHPLC-MS/MS	Chicken, sheep tissue, bovine & swine	11 macrolides	LOQs:5.0 µg/kg LODs: 0.1–2.0 µg/kg	[70]
d-micro SPE LC-HRMS	Swine body fluids	Fluoroquinolones	LOQs: 0.06–0.1 µg/L LODs: 0.02–0.03 µg/L	[71]
DSPE-LC-APCI-HRMS/MS	Coconut milk	Pesticide residues	LODs: 0.5–5 ng/g LOQs: 1–10 ng/g Concentrations: 5.54–206.99 ng/g.	[72]
DSPE-GC-APCI Q-TOF	Orange, chili pepper, and brown rice	Pesticide multiresidues	LODs: 0.11–0.6 µg/mL LOQs: 0.34–1.9 µg/mL Recoveries: 70–120%	[73]
SPME-GC-MS/MS	Beverages	Parabens, phenolic antioxidants, sulfonamide plasticizer, and flame retardant	LODs: 0.005–0.2 µg/L %RSD: 0.8–5.4% Recoveries: 98–109% Concentrations: 4.75–2220.99 µg/L	[58]
d-micro SPE-LC-HRMS	Milk and honey samples	Sulfonamides	LODs: 0.003–0.2 µg/L; 0.01–1 µg/kg	[74]

(continued)

**Table 1** (continued)

Techniques	Samples	Analytes	Performance	References
			%RSD: 1.3–10.4% Recoveries: 68.8–115.8% Concentrations: 0.3–9.7 µg/L; (3.9 µg/kg)	
µ-SPE-GC-orbitrap-MS	Cereals	Pesticide residues	LODs: 0.005–0.01 mg/kg %RSD: <20% Recoveries: 70–120%	[75]
MIP-µ-SPE/LC-MS/MS	Non-dairy beverages	Aflatoxins (AFB1 and AFB2)	LODs: 0.085–0.207 µg/L %RSD: <18% Recoveries: 91–104% Concentrations: 0.230–0.632 µg/L	[76]
MIP-µ-SPE/HPLC-MS/MS	Cultured fish	Aflatoxins	LODs: 0.29–0.61 µg/kg %RSD: <20% Recoveries: 80–100%	[77]
D-µ-SPE/GC-MS	Honey samples	Polycyclic aromatic hydrocarbons	LODs: 0.14–0.37 ng/g %RSD: <7.5% Recoveries: 72–81%.	[78]
MSPE/HPLC-MS/MS	Milk, pork meat, and fish meat	Sulfonamide antibiotics	LODs: 0.40–1.5 ng/g %RSD: <10% Recoveries: 0.40–1.5 ng/g	[79]
MSPE/HPLC-MS/MS	Milk and meat samples	Fluoroquinolones	LODs: 0.7–6.0 ng/L %RSD: >10% Recoveries: 75.3–104.7%. Concentrations: 0–8.60 µg/kg	[80]
SPE/UPLC-MS/MS	Chicken meat; eggs	Quinolone antibiotics	LODs: 0.1–0.16 µg/kg Recoveries: 66.9–99.0% Concentrations: 6.22–44.4 µg/kg	[81]

(continued)

**Table 1** (continued)

Techniques	Samples	Analytes	Performance	References
MSPE–LC–MS/MS	Milk and pork samples.	$\beta$ -agonists and fluoroquinolones	LODs: 0.1–0.2 ng/g %RSD: <6.8% Recoveries: 94.5–105.8% Concentrations: 0.12–0.42 ng/g	[82]
MI-SPE/LC–MS/MS	Various food samples	Patulin	LOD: 0.05–0.2 ng/g %RSD: <4.5% Recovery: 81.3–106.3%	[83]

*SPE* solid phase extraction, *UPLC-MS/MS* ultra-performance liquid chromatography coupled to tandem mass spectrometry, *SPME-GC-MS/MS* solid phase microextraction – coupled to a gas chromatography–ion trap tandem mass spectrometry, *GC-APCI-Q-TOF* gas chromatography-atmospheric pressure chemical ionization quadrupole time-of-flight, *DSPE/d-SPE* dispersive solid phase extraction, *UHPLC-ESI-MS/MS* ultra-high performance liquid chromatography coupled to electrospray ionization tandem mass spectrometry, *LC-APCI-HRMS/MS* liquid chromatography-atmospheric pressure chemical ionization high-resolution tandem mass spectrometry, *HPLC-MS/MS* high-performance liquid chromatography with tandem mass spectrometry, *LC-HRMS* liquid chromatography-high-resolution mass spectrometry, *MIP-SPE* molecularly imprinted polymer based solid phase extraction, *d-micro SPE* dispersive micro solid phase extraction,  $\mu$ -*SPE* microsolid-phase extraction system, *GC-Orbitrap-MS* gas chromatography-orbitrap mass spectrometry

RSDs of 2.1–8.3%,  $n = 5$ ) [62]. Sun et al. reported a novel sample preparation method based on pipette-tip microsolid phase extraction (PT- $\mu$ SPE) coupled corona discharge ionization-ion mobility spectrometry (CD-IMS) for the determination of benzodiazepines (BZDs) in dietary supplements. The PT- $\mu$ SPE was fabricated using poly (styrene-co-divinylbenzene) (St-co-DVB) monoliths via an in-situ polycondensation reaction [63]. The LODs of the developed method ranged from 5 to 15 ng mL<sup>-1</sup>. Furthermore, the developed PT- $\mu$ SPE-IMS was applied in spiked samples and acceptable recoveries ranging from 84.2 to 112.1%.

Combination of MS based techniques with suitable solid SPE/SPME procedures allows the detection of food adulteration. For instance, Peng et al. conducted a study on determination of maltol, ethyl maltol, vanillin, and ethyl vanillin in foods using isotope dilution headspace SPME coupled with GC-MS. Under optimum conditions, the LODs of the SPME-GC-MS were 0.005–0.50  $\mu$ g/g and concentration of ethyl maltol and vanillin in milk tea were up to 833  $\mu$ g/g and 1,044  $\mu$ g/g, respectively. Moreover, concentrations ranging from 0.286 to 1.65  $\mu$ g/g for maltol, ethyl maltol, and vanillin were detected in blended sesame oil [64]. In another study, Peng et al. investigated the possibility of adulteration of rice using synthetic aromatic flavorings. The synthetic aromatic flavors are used to deceitfully mimic commercially valuable fragrant rice varieties [65]. In their study, the authors applied HS-SPME coupled with GC-MS using selected ion monitoring (GC-MS-SIM) for determination of 2-acetylpyrazine (2-APZ), 2-acetylthiazole (2-ATZ), 2-acetylpyrrole (2-APR), and 2-acetylpyridine (2-APD). The HS-SPME/GC-MS-SIM method had

LODs ranging from 0.5 to 10 ng/mL, thus allowing the analysis of volatiles in rice with minimal pretreatment. The commercially available fragrant rice samples were found to contain 2-ATZ, 2-APD, and 2-APR 2 at the concentrations ranging from 0.018 to 0.035 mg/kg, respectively. Other applications of solid phase-based techniques shown in Table 1 revealed high recoveries, low LOQs and LODs in most of the studies, allowing quantitative analysis of various analytes.

### Application of Solid Phase-Based Extraction Methods in Environment Samples

A web of science search for “solid phase extraction, environment” for the last 5 years returns 7,834 results, similarly showing the extensive use of SPE in environmental monitoring studies. This is because SPE and related procedures have the capability to simultaneously extract large number of compounds in a sample thus warranting them to be simple, rapid, and somehow generic sample preparation methods. By extracting large number of analytes at the same time, the cost, time, and labor are significantly reduced. Table 2 summarizes some of the studies conducted in environmental analysis using SPE-based methods. Similar to food samples, solid phase-based methods allow the analysis of wide range of complex environmental matrices. The application of these methods leads to remarkable analytical performance which achieves better separation by the chromatographic technique and more effective quantification/identification of analytes. Coupling MS methodologies with suitable sample preparation method allows accurate identification of contamination sources. For example, it is suspected that plastic bottles containing bisphenol A (BPA) may release this compound into the drinking water they contain. A study using Bond-Elut-C18 SPE combined with UPLC-MS/MS method was developed and applied in extraction of BPA from bottled water (glass and plastic). The analytical characteristics of the developed method in terms of LOQ, LOD, and intra/inter-day precision RSD were 0.04 µg/L, 0.01 µg/L, and <31% ( $n = 5$ ), respectively. Bond Elut C18 which is a hydrophobic bonded silica sorbent showed better extraction efficiency compared to Oasis<sup>®</sup> Max (mixed-mode polymeric) and Oasis<sup>®</sup> HLB (polymeric reversed phase) sorbents. The results showed that glass bottled samples contain higher concentration (4.34–41.19 µg/L) compared to PET bottled samples (0.29–24.88 µg/L) [66]. A HS-SPME method using metal-organic framework CIM-80(Al) (CIM-80(Al) MOF) as extraction phase in combination with GC-MS was used to determine seven musk fragrances and six methyl siloxanes from water samples. Low LODs in the range of 0.2 and 0.5 µg/L were observed, and RSD lower than 17% and 22%, respectively, [84]. Stir bar sorptive-dispersive microextraction (SBSDE) method using polydopamine-coated cobalt ferrite magnetic nanoparticles (CoFe<sub>2</sub>O<sub>4</sub>@PDA) was employed as an extraction material prior to thermal desorption GC-MS (TD-GC-MS) for the determination of nitro musks from water samples. The method showed low LOQs and LODs (ng/L) and repeatability (inter/intra-day) with RSD <15% [85].

Gao and co-workers [86] reported a method for the determination of androgens and progestogens in environmental water samples with linearity, LODs, and

**Table 2** Selected applications of solid phase-based extraction methods in environment samples

Techniques	Samples	Analytes	Performance	References
Porous thin film-GC-APCI-MS/MS	Seawater and river water	Polycyclic aromatic hydrocarbons	LODs: 1–100 pg/mL Recoveries: 79–115%	[89]
SPE- LC-MS/MS	Surface water and wastewater	Sterols and steroid hormones	LODs: 0.7–15.0 ng/L Recoveries: 70–114% Concentrations: 176–5,162 ng/L)	[90]
SPE-LC-MS/MS	Surface water, groundwater, and drinking water	Per- and polyfluoroalkyl substances (PFASs)	LOQs: 0.6–26 ng/L Recoveries: 83–107%. RSDs: 0.7–15% Concentration: 0.045–17 µg/L	[91]
SPE-LC-MS/MS	Drinking, surface, and effluent wastewaters	Pesticides, hormones, or pharmaceuticals	LODs: 0.1–1.4 ng/L LOQs: 0.3–4.8 ng/L Concentrations: 3–68 ng/L	[92]
SPE-UHPLC-MS/MS	Surface and groundwater	Anthelmintic compounds	LODs:0.05–1.0 ng/L LOQs:0.5–40 ng/L	[93]
SPE-UPLC-MS/MS	Sediments	Carbazole and polyhalogenated carbazoles (PHCs)	LODs: 0.003–0.22 ng/g LOQs: 0.01–0.75 ng/g Recoveries: 63–124.7%	[94]
d-SPE HPLC-MS	Water samples	Bisphenol A, AF, B & tetrabromobisphenol A	LODs: 0.02 µg/L Linear range (0.1–1,000 µg/L)	[95]
SPE-LC-MS	Drinking water	Drugs of abuse and pharmaceuticals	LODs: 0.01–1.09 ng/L %RSD: <15.0% Recoveries: 62: 107% Concentrations: 0.14–2.81 ng/L	[64]
d-SPE/LC-MS	Water samples			[96]

(continued)

**Table 2** (continued)

Techniques	Samples	Analytes	Performance	References
		Phenoxy-carboxylic acids	LODs: 0.55–3.84 /L %RSD: 2.0–9.0% Recoveries: 86.9–101.3% Concentrations: 0.14–2.81 ng/L	
SPE LC-MS/MS	Urban wastewater	Pharmaceuticals	LODs: 0.1–1.5 ng/L %RSD: 2.2–8.6% Recoveries: 84.7–97.2% Concentrations: 11–145,250 ng/L	[97]
MSPE/LC–MS/MS	Water, soil, river, and sediment	Chiral pesticides	LODs: 0.0–0.62 ng/L; 0.02–0.17 ng/g %RSD: 2.1–13.4% Recoveries: 80.3–106.3% Concentrations: 0–45.9 ng/L; 0–79.4 ng/g	[98]

*d*-SPE dispersive solid phase extraction, SPE solid phase extraction, GC-APCI-MS/MS gas chromatography coupled to triple quadrupole mass spectrometry, LC-MS/MS liquid chromatography-tandem mass spectrometry, HPLC-MS high-performance liquid chromatography-tandem mass spectrometry, LC-MS liquid chromatography-tandem mass spectrometry, UPLC-MS/MS ultra-performance liquid coupled to tandem mass spectroscopy, UHPLC-MS/MS ultra-high performance liquid chromatography-tandem mass spectrometry

recoveries ranging 8–2,032 ng/L, 2.0–10.0 ng/L and 80.5–102.4%, respectively, using UiO-66(Zr) MOF as the adsorbent of  $\mu$ -SPE device combined with LC-MS/MS. The  $\mu$ -SPE-LC-MS/MS method had intra-day and inter-day analysis in terms of RSDs of 1.8–6.3% and 2.8–6.5%, respectively. Combining the  $\mu$ -SPE with LC-MS/MS enabled the detection of trace concentrations of progesterone ranging between 11.3 and 72.2 ng/L. Barreca et al. [87] developed an online SPE method coupled with LC-MS/MS for pre-concentration and determination of perfluoroalkyl acids (PFASs) in natural water. The SPE- LC-MS/MS method had remarkable LODs (0.2–5 ng/L) and linearity (0.2–250 ng/L) as well as acceptable accuracy (80–120%), and precision (RSD <15%).

González-Sálamo et al. [88] reported the use of Basolite<sup>®</sup> F300 MOF based DSPE for extraction of phthalic acid esters from water samples prior to LC-MS determination. The matrix-matched calibration, trueness, and LODs of the developed method were 5–500  $\mu$ g/L and 70–118%, 0.022–0.069  $\mu$ g/L, respectively. The analytes were present in all investigated samples at concentrations ranging from 0.21

to 4.04  $\mu\text{g/L}$ . Wang et al. [15] prepared graphene/ $\text{Fe}_3\text{O}_4$  (G- $\text{Fe}_3\text{O}_4$ ) nanocomposites as an adsorbent for MSPE of ketoconazole, econazole, miconazole, butoconazole, sertaconazole, fenticonazole, and isoconazole from environmental water samples. The imidazole antifungals were quantified using LC-MS/MS. The electrostatic, cation- $\pi$ , hydrophobic and  $\pi$ - $\pi$  stacking interactions allowed efficient extraction of the analytes. Satisfactory recoveries (70–92%), acceptable intra-day/inter-day precisions (%RSD 3.0–12.0%), and LOD (0.11–0.32  $\text{ng/L}$ ) were obtained.

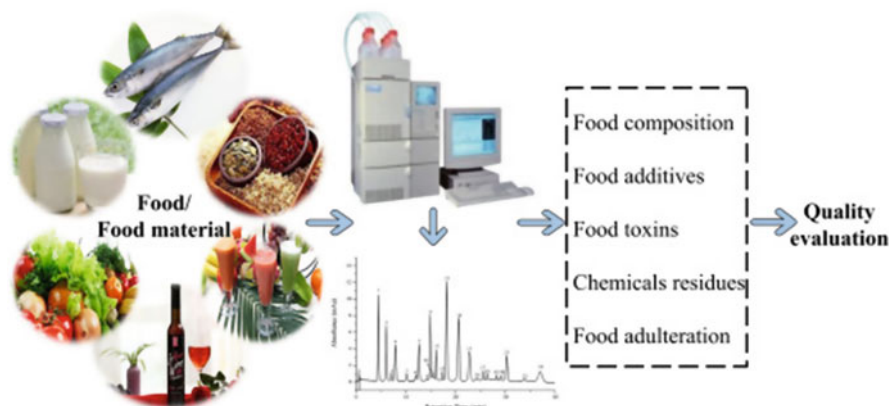
## 2.2.2 Liquid Phase-Based Extraction Methods

Liquid–liquid extraction (LLE) uses organic solvents such as ethyl acetate, hexane, and isooctane among others to partition target analytes from a sample matrix to the organic extractant [99]. LLE is one of the oldest extraction techniques [34, 100]. Due to its ease of application, LLE has been widely applied. Traditional LLE has been modified by the development of miniaturized forms of LLE such as single drop microextraction (SDME), dispersive liquid–liquid microextraction (DLLME), hollow fiber liquid phase microextraction (HF-LPME), and cloud point extraction (CPE) [101, 102]. Various review articles have covered the principles of various liquid phase-based extraction methods [102, 103], and in this chapter their application in food and environmental samples prior to various MS methodologies is highlighted.

### Application of Liquid Phase-Based Extraction Methods in Food Samples

Advances in food analysis has evolved from conventional analysis of the major components of food to more complex and broader aspects. This is because various chemical substances present in food play a major role its production and distribution [104, 105]. Notable the improvement in livestock production and crop yields requires the use of protection, phytosanitary products, and veterinary drugs. In addition, the improvement in food quality requires the use of substances such as colorants, flavors, enzymes, and other additives [106]. Packaging materials (such as paper, polyethylene terephthalate (PET) plastics and paper, among others), on the other hand, are utilized for preservation, presentation improvement, storage, ease of transportation, and maintaining hygienic conditions of food [107]. Even though the above-mentioned efforts increase food production and security, the chemical substances used may lead to their presence in food, thus presenting a potential contamination risk [108]. Figure 1 illustrates that food analysis does not only focus on contaminants, but it also involves analysis of foodstuff nutritional values (such as proteins, amino acids, essential fatty acids, fibers, vitamins, carbohydrates, lipids, minerals, etc.) [109]. Furthermore, food analysis has included the determination of natural contaminants such as mycotoxins and algal biotoxins as well as food authentication, adulteration, and fraud detection [110–113].

As discussed, food analysis encompasses the determination of numerous chemical compounds with different chemical nature. Therefore, sensitive and accurate



**Fig. 1** Typical application of MS based methods in analysis of various chemical substances in food. Adapted from Ref. [104] with Elsevier permission

analytical techniques (GC and LC coupled with MS) combined with various modes of liquid phase extraction procedures have been extensively used in monitoring/quantification of various substances in food samples. Nemati et al. [114] developed a liquid microextraction method based on solidification of floating organic droplet-liquid-liquid microextraction (SFOD-LLME) coupled with HPLC-MS/MS for extraction and determination of four antibiotics in honey. In their findings, they reported low LOQs (1.9–2.6 ng/g) and LODs (0.55–0.79 ng/g) with  $\leq 6.9\%$  RSDs [114]. Previous study reported the development of a micro-LLE coupled with UPLC-MS/MS for the extraction and quantification of herbicides from agricultural products. Method quantification limit (MQL) 0.6–1.0 ng/g and recoveries ranging from 73.7 to 91.4% with RSD of less than 10% were obtained [115].

Zheng et al. have developed an LLE method coupled with LC-MS for quantification of salicylic acid (SA) and acetylsalicylic acid (ASA). These two analytes (SA and ASA) were extracted from flatfish, milk, egg, eel, and porcine muscle. The obtained LOQs were 7 and 10  $\mu\text{g}/\text{kg}$  for SA and ASA, respectively, with RSDs of  $\leq 9.0\%$  (inter-day and intra-day) [116]. Salim et al. [117] developed and validated DLLME combined with LC-MS/MS for analysis of multi-mycotoxin (that is aflatoxins (AFB1, AFB2, AFG1, and AFG2), ochratoxin A (OTA), deoxynivalenol (DON), fumonisins (FB1 and FB2), sterigmatocystin (STG), T-2 toxin, HT-2 toxin, diacetoxyscirpenol (DAS) and zearalenone (ZEA)) in rice bran. The developed method had good linearity (0.5–150 ng/g), LOD (0.5–50 ng/g) and acceptable recoveries (70.2–99.4%) with a %RSD below 2%. The concentrations of mycotoxin in rice bran samples were between 0 and 157 ng/g. Alsharif et al. [118] developed a modified HF-DLLME of multi-aflatoxins and ochratoxin A in commercial apple, grape, orange, and pomegranate juice prior to HPLC-MS/MS detection. The linearity, LOD, recoveries, and %RSD were 0.1–30  $\mu\text{g}/\text{L}$ , 0.04–0.13  $\mu\text{g}/\text{L}$ , 70–120% and  $< 11\%$ , respectively. Table 3 shows that different types of analytes with varying



**Table 3** Selected applications of liquid phase-based extraction methods in food analysis

Techniques	Samples	Analytes	Performance	References
HLLC/LC-orbitrap HRMS	Pork, fish, egg, and milk	19 quinolones	LOQs: 0.028–0.192 µg/kg	[119]
SPE salting-out assisted LLE	Aqueous food simulants	Carcinogenic primary aromatic amines (PAAs)	LOD: 0.05 µg/L LOQ: 0.15 µg/L	[120]
LDDDES-DLLME-GC-MS/MS	Soft drinks	PAHs	LOQ: 0.2 µg/L LOD: 0.01 µg/L RSD: <16–87%	[121]
SALLE-UHPLC-MS/MS	Infant foods	Tetracycline	LOD: 0.05–0.14 µg/kg LOQ: 0.16–0.48 µg/kg %RSD: <7.3%	[122]
Inkjet-based DLLME-UHPLC-MS/MS	Wheat	Aflatoxins	LODs: 0.06–6 µg/kg %RSD: <4.6% Recoveries: 83.2–93.0% Concentrations: 0–2.88 µg/kg	[123]
LLE-LC-MS/MS	Meat-, fish-, seafood-based products, various dairy ingredients, infant formulae, and fats	Tetracyclines	LOD: 10 µg/kg	[124]
VA-DLLME/LC-MS/MS	Tomato fruits	Fipronil and its metabolite fipronil sulfone	LOD: 0.075 µg/kg %RSD: <12% Recoveries: 89.8–96.3%	[125]
IL-based DLLME/LC-MS/MS	Vegetables and fruits	Multi-pesticide residues	LODs: 0.02–0.32 µg/kg %RSDs: 1–16% Recoveries: 70–137%	[126]
DLLME/UPLC-MS/MS	Honey	Pyrrolizidine alkaloids	LODs: 0.009–0.018 µg/kg %RSD: 1–12% Recoveries: 63 to 103% Concentrations: up to 17.5 µg/kg	[127]
DLLME/GC-MS	Milk	Amino acids (alanine, glycine, and glutamic acid)	LODs: 0.31–0.84 µg/mL %RSDs: 7.8–10.3%	[128]

*SS-LPE* Switchable solvent liquid phase extraction, *HLLC* homogeneous liquid–liquid extraction, *SALLE* salting-out assisted liquid–liquid extraction, *VA-DLLME* vortex assisted dispersive liquid–liquid microextraction, *IL-based DLLME* ionic liquid based dispersive liquid–liquid microextraction

physical and chemical properties from complex food samples can be analyzed using MS methodologies via liquid phase-based extraction methods as preliminary extraction and enrichment step. These analytes include pesticides, pharmaceuticals, and PAHs, among others.

### Application of Liquid Phase-Based Extraction Methods in Environmental Samples

Table 4 presents the summary of some of the recent studies describing the application of liquid phase-based extraction methods for enrichment of different analytes in

**Table 4** Application of liquid phase-based extraction methods in environmental analysis

Method	Samples	Analyte	Performance	References
LPE-SPE-GC-MS	Water	PAHs	%RSD < 7.5% and recoveries 87–104% LODs: 4– 110 ng/kg	[136]
DLLME-GC-MS	Surface water	Pesticides	LOQs: 0.003– 3.75 µg/L LODs: 0.001– 1.125 µg/L	[137]
SPE-DLLE UHPLC-MS/MS	Water	Pharmaceuticals and per- sonal care products	LOQ: 3 ng/L LOD: 1 ng/L	[138]
DLLME-SFO- SBE/UHPLC-(+) APCI-MS/MS	Water samples	Nitrated-PAHs(oxy-PAHs)	LOQs: 0.15– 1.10 ng/mL LODs: 0.02– 0.85 ng/mL	[139]
DLLME/GC-MS	Human saliva	Parabens, triclocarban, bisphenols, and benzophenones	LODs: 0.01– 0.15 ng/mL LOQs: 0.05– 0.40 ng/mL	[140]
LLE/GC-MS	Wastewater	40 analytes in refinery wastewater	LODs: 0.3– 1.7 µg/L	[95]
DLLME/GC-MS	Wastewater	Bisphenol A and UV-filters	LOQ:10 ng/L LOD: 2 ng/L	[141]
DLLME /LC-MS/ MS	Water	Pesticides and personal care products	LOQs:0.0125– 1.25 µg/L	[142]
HF-LPME L/C- MS/MS	Water samples	27 emerging contaminants	LOQs: 2.13– 126.50 ng/L LODs:1.09– 98.15 ng/L	[143]
SS-LPE/GC-MS	Well, tap, and lake water samples	Chlorthiamid, ethyl para- thion, penconazole, and fludioxonil pesticides	LODs: 0.28– 0.58 ng/mL LOQs: 0.93– 1.9 ng/mL	[144]

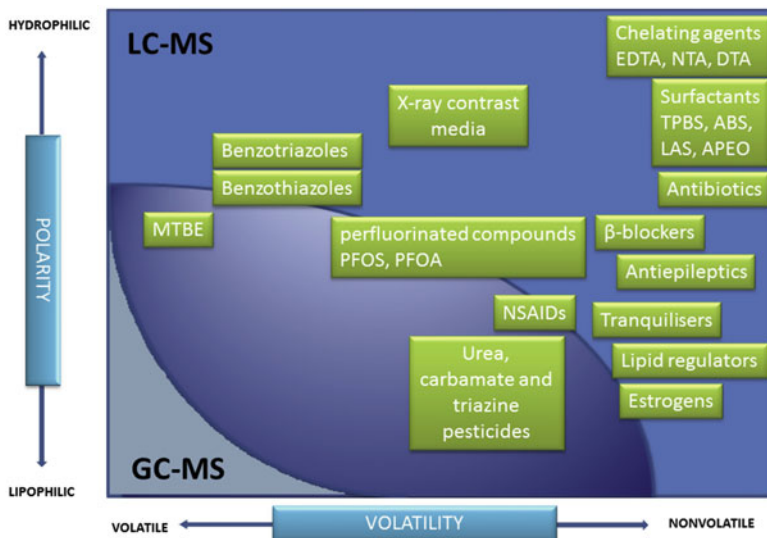
*SS-LPE* switchable solvent liquid phase extraction, *DLLME-SFO-SBE* dispersive liquid–liquid phase microextraction based on the solidification of a floating organic drop coupled with solvent assisted back-extraction, *HF-LPME* Hollow fiber liquid phase microextraction

environmental samples prior to MS analysis. The results obtained by different researchers show that these methods have significant influence on the performance and efficiency of the detection techniques. This supported the differences in the analytical figures of merit (Table 4). For instance, LLE-GC-MS extraction method was developed to extract 9 haloacetic acids from water samples. LOD and LOQ were less than 2.6 and 3.9  $\mu\text{g/L}$ , while the RSDs for inter-day and intra-day precisions ranged from 1.68–19.07 to 0.01–7.31%, respectively. Quantitative recoveries ranging from 79.7 to 113.22% were obtained [129]. These results show that traditional LLE is applicable because it allowed efficient extraction of target analytes prior to GC-MS analysis.

In another study, Erarpat et al. developed a GC-MS method based on liquid phase extraction of estrone, fluoxetine, and pesticides from water. The results obtained demonstrated that the method had relatively low LODs (0.16–8.6  $\text{ng/mL}$ ) and LOQs (0.5–29  $\text{ng/mL}$ ) [130]. Another study reported the application of LPME with GC-MS for extraction of hormones and pesticides [131]. The method exhibited high recovery greater than or equal to 110% with  $\text{RSD} < 10\%$ . The LODs and LOQs ranged from 0.20–13  $\text{ng/mL}$  to 0.90–46  $\text{ng/mL}$ , respectively [131]. This study proved that liquid phase extraction method could be used for extraction of analytes that have different chemical properties. Cacho et al. developed an ionic liquid based dispersive liquid–liquid microextraction (IL-DLLME) coupled with GC-MS for the extraction of nine organophosphorus pesticides from water samples. The obtained RSDs were between 4.1 and 9.7% while the LODs were in the range of 5–16  $\text{ng/L}$  [132]. Ionic liquid dispersive liquid phase microextraction (IL-DLPME) coupled with GC-MS has been used in pre-concentration of chlorobenzene compounds. The developed method was used in both water and soil samples. The LODs were in the range of 8.4–252  $\text{ng/kg}$  for soil and 0.5–7.2  $\text{ng/L}$  for water. The concentrations of the analytes in soil and water samples were 1.5–50  $\text{ng/L}$  and 2.7–256  $\text{ng/L}$ , respectively [133]. A DLLME method coupled with MS for the detection of parathion methyl from water samples was reported elsewhere [134]. The optimized method exhibited high recoveries (up to 99.9%) and acceptable precision 0.32%. A DLLME-GC-MS method was used for the extraction of hormones, pesticides, and endocrine disruptors in sea water, tap water, well water, and municipal wastewater [135]. The method displayed good analytical performance with LODs ranging from 0.3 to 2.0  $\text{ng/mL}$  and acceptable precision.

### 2.3 Separation Techniques

Mass spectrometry-based (MS) techniques exhibit excellent selectivity and high sensitivity, as well as abundant fragments that can be used for structural identification. When coupled with chromatographic techniques, MS makes the separation methods to be suited for both targeted detection and nontargeted [145, 146]. The most commonly used chromatographic separation techniques with MS are gas and liquid chromatography (GC and LC) [42]. GC is used for the separation of non-polar



**Fig. 2** Application of GC-MS and LC-MS in wide range of organic pollutants. Adapted from Ref. [148] with Elsevier permission

and volatile compounds using helium or nitrogen as a mobile phase (Fig. 2). In contrast, LC can separate a wide range of polar, non-volatile, and ionic compounds (Fig. 2) using mainly reversed phase C18 columns as the stationary phase and mixtures of methanol-water or acetonitrile water with volatile additives as mobile phase [147]. The combination of MS with these separation techniques results in a powerful technique with the ability to identify and confirm the identity of compounds, provided structural information needed for analyte characterization as well as identifying coeluting compounds (Fig. 2 and Tables 1, 2, 3, 4).

### 3 Conclusions

Recent advances in the sample preparation of food and environmental samples for mass spectrometric analysis have been discussed in this chapter. Furthermore, sample handling procedures such as passive sampling, grab or discrete sampling at a specific time, time-integrated sampling, and event-based sampling have been briefly discussed. With respect to sample preparation, the use of solid phase- and liquid phase-based extraction techniques, such as SPE, DSPE, DLLME, LLE, and thin films that allow extraction and pre-concentration of analytes as well as sample clean-up prior to MS analysis was also presented. Studies showed that solid phase-based extractions remain the most commonly used sample pretreatment procedures for both food and environmental samples. Additionally, MS based methodologies based on GC and LC/UHPLC for separation of volatile compounds and polar

analytes are still the separation methods of choice prior to MS detection. Furthermore, the interest in use of HRMS, Q-TOF-MS/MS, TOF-MS, and APCI/ESI-MS/MS coupled with GC or LC for food and environmental sample is growing owing to advantages such as high resolution and efficiency. Additionally, the above-mentioned MS approaches provide an improvement in the performance of the analytical methods thus allowing these methodologies to be suitable for control laboratories.

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# Elemental Mass Spectrometry in Food and Environmental Chemistry



Bin Hu, Man He, Beibei Chen, Chi Xu, Qiulin Zhang, Junrong Ma, Yiling Feng, and Zewei Cui

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**Abstract** Analytical challenges of elemental mass spectrometry are discussed, introducing the fundamental principles of inductively coupled plasma mass spectrometry (ICP-MS) and its variants such as chromatographic techniques-ICP-MS, multiple-collector-ICP-MS, laser ablation-ICP-MS, and field flow fractionation-ICP-MS. Applications are discussed and commented. Some examples illustrate the analytical approaches, which are used to address specific issues in various areas of both food and nutrition and environmental research, such as food authentication, metallomics, speciation, toxic and nutritive elements, nanomaterials, and migration studies.

**Keywords** Chromatographic separation, Field flow, Inductively coupled plasma mass spectrometry, Laser ablation, Multiple collector, Solid/liquid phase (micro)extraction

## 1 Introduction

Elemental Mass Spectrometry refers to the analytical techniques for qualitative and quantitative analysis of elements based on mass-to-charge ratio, including inductively coupled plasma mass spectrometry (ICP-MS), thermal ionization mass spectrometry (TIMS), and secondary ion mass spectrometry (SIMS). The latter two techniques are mainly used for isotope ratio measurements, along with multiple collector (MC)-ICP-MS [1]. Comparatively, ICP-MS merits much application potential in trace elemental quantification and speciation analysis. Traditional ICP-MS is limited to the quantification of trace elements in aqueous solution. By combining with efficient separation techniques, the information of specific elemental species can be obtained by ICP-MS. Moreover, single particle (SP) ICP-MS, based on the generation of discrete pulses of ions arising from SPs sequentially introduced into ICP, has been proposed for the measurement of metal-containing nanoparticles, providing the information of particle size and concentration [2]. Likewise, direct elemental analysis for solid samples with ICP-MS can be carried out replacing conventional pneumatic nebulization with laser ablation as the sample introduction system, enabling mapping of elemental distribution with lateral resolution at  $\mu\text{m}$  in bioimaging [3].

This chapter is focused on the development of variants of ICP-MS and their applications in food and environmental chemistry.



## 2 Solid Phase (Micro)Extraction/Liquid Phase (Micro)Extraction-ICP-MS

The quantification of trace elements by ICP-MS is often affected by the complexity of the sample, and sometimes the instrumental quantification limit does not reach the sensitivity levels required for the analysis of food and environmental samples. To achieve accurate elemental concentration at ultra-trace levels, appropriate sample pretreatment is always necessary prior to ICP-MS detection for the removal of sample matrix and simultaneous preconcentration of target elements. Solid phase (micro) extraction and liquid phase (micro) extraction are the two most commonly used pretreatment techniques.

### 2.1 Solid Phase (Micro) Extraction

Solid phase extraction (SPE) is based on the distribution of target analytes between the solid phase and liquid phase. SPE adsorbents determine the analytical performance of the SPE-based method to a great extent in terms of selectivity, enrichment factor, and adsorption capacity. To improve the selectivity of the sorbents, a variety of functionalized or imprinted sorbents have been synthesized. Besides, a series of extraction modes have been proposed to facilitate different analytical purposes, including magnetic solid phase extraction (MSPE), solid phase microextraction (SPME), capillary microextraction (CME), and stir bar sorptive extraction (SBSE).

#### 2.1.1 Trace Elemental Quantification

In conventional SPE, the sorbents are usually packed in a column, which can be easily integrated in a flow injection system for online analysis [4]. In MSPE, magnetic sorbents along with the adsorbed target elements could be easily and rapidly separated from the aqueous solution simply by applying an external magnetic field, while no filtration or centrifugation was necessary. Titania coated magnetic sorbents were synthesized and applied for SPE of trace heavy metal ions from environmental water samples followed by ICP-MS detection [5]. The maximal enrichment factor (EF) of 200-fold could be achieved with the use of 100 mL sample solution. Thiol-grafted magnetic polymer (MPPs-SH) was synthesized by click reaction. MPPs-SH had high S content (5.4 wt.%) and exhibited high adsorption capacity of 254, 64.2, 56.6 mg g<sup>-1</sup> for Hg(II), Cd(II), and Pb(II), respectively. Based on it, a method of MSPE-ICP-MS was developed for the quantification of trace Hg(II), Cd(II), and Pb(II) in environmental water samples [6]. Three heavy metal ions were extracted from 150 mL aqueous solution in 10 min and eluted by 0.2 mol L<sup>-1</sup> HNO<sub>3</sub> containing 2% thiourea in 5 min. The method presented low limits of detection (LODs) (0.21, 0.82, and 2.9 ng L<sup>-1</sup>) for target ions. Erkan et al. [7]

synthesized  $\alpha$ -amylase-modified magnetic carbon multi-walled nanotubes ( $\alpha$ -amylase-Fe<sub>3</sub>O<sub>4</sub>/MWCNTs) and used them for the extraction of arsenic traces from food samples. The composite overcame the problem of inactivation of the free soluble enzyme after extraction, and can be reused at least six times. The microextraction step only takes 15 min. A quantification limit (LOQ) of 47.3  $\mu\text{g kg}^{-1}$  was achieved for the analysis of arsenic in rice and flour samples. With similar strategy, magnetic sorbents doped with different functional groups (such as N, O, S-containing groups) were combined with ICP-MS detection for the analysis of various elements in environmental water samples [8–11].

Besides, imprinting technique is an alternative for the selectivity improvement of the sorbents. Cd(II) imprinted 3-mercaptopropyltrimethoxysilane (MPTS)-silica coated stir bar was prepared by combining sol-gel technique with surface imprinting technique. The imprinted coating exhibited higher selectivity for Cd(II) in the presence of Cu(II) and Ni(II) over non-imprinted coating. The prepared stir bar was employed for SBSE of trace Cd(II) followed by ICP-MS detection. The proposed method was applied for the analysis of trace Cd(II) in rain, lake, and river water samples [12].

### 2.1.2 Elemental Speciation Analysis

With the use of functionalized sorbents, the selectivity of the method could be improved, sometimes enabling speciation analysis for ICP-MS. For example, 3-(2-aminoethylamino) propyltrimethoxysilane (AAPTS) functionalized MWCNTs were synthesized and used for simultaneous speciation analysis of inorganic As, Cr, and Se in environmental water samples prior to their determination by ICP-MS [13]. At pH 2.2, As(V), Cr(VI), and Se(VI) could be quantitatively retained on the prepared sorbent while As(III), Cr(III), and Se(IV) could not be extracted. The retained As(V), Cr(VI), and Se(VI) were eluted by HNO<sub>3</sub> and determined by ICP-MS. With using KMnO<sub>4</sub> as oxidation reagent, the low valence species of As (III), Cr(III), Se (IV) could be transformed into their high valence species (As(V), Cr (VI), and Se(VI)) and the total amount of each element could be determined by the same method. The concentrations of low valence species were then calculated by subtraction. Under the optimized conditions, LODs of 15, 38, and 16 ng L<sup>-1</sup> for As (V), Cr(VI), and Se(VI) were obtained. Moreover, Cr(III)-imprinted AAPTS functionalized silica gel sorbent was prepared for the speciation analysis of Cr in environmental water samples prior to ICP-MS measurement [14]. The Cr(III)-imprinted silica gel shows the selectivity coefficient of more than 700 for Cr(III) in the presence of Mn(II). Cr(VI) could be adsorbed at low pH by the prepared imprinted silica gel, making it feasible to enrich and determine Cr(VI) at low pH without adding reducing reagents. LODs of 4.43 and 8.30  $\text{pg mL}^{-1}$  for total Cr and Cr(VI) were obtained. Compared with most of the analytical methods reported in literature for the speciation of inorganic Cr, the proposed method does not need any reduction/oxidation reagent in the process of separation, which prevented the potential transformation of speciation and contamination.

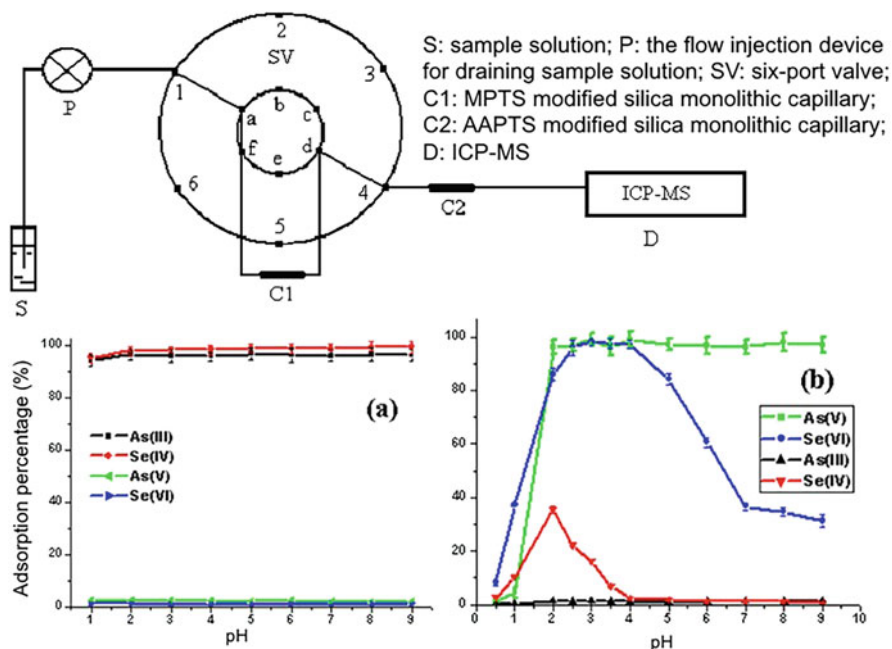
AAPTS-silica monolithic capillary was used for CME of aluminum fractionation, followed by electrothermal vaporization (ETV)-ICP-MS detection with polytetrafluoroethylene (PTFE) slurry as fluorinating agent [15]. In the pH range of 4–7, labile monomeric Al could be retained quantitatively on the monolithic capillary, while non-labile monomeric Al passed through the capillary directly. The total monomeric Al fraction was determined by AAPTS-silica monolithic CME-fluorination-assisted (FETV)-ICP-MS after the sample solution was adjusted to pH 8.8. Non-labile monomeric Al was obtained by subtracting labile monomeric Al from the total monomeric Al. The method was applied to the fraction of Al in rainwater and fruit juice. This system had advantages over the existing 8-hydroxyquinoline (8-HQ) chelating system for Al fractionation such as wider pH range, higher tolerance of interference, and better regeneration. Besides, a method of dual silica monolithic CME online hyphenated with ICP-MS was developed for sequential determination of inorganic As(III)/As(V) and Se(IV)/Se(VI) in natural waters [16]. The dual CME column was composed of MPTS modified silica monolithic capillary (C1) and AAPTS modified silica monolithic capillary (C2) which both were prepared by sol-gel technology. With the aid of valve-switching, As(III)/Se(IV) retained on C1 and As(V)/Se(VI) retained on C2 were sequentially desorbed by 100  $\mu$ L of appropriate eluents for subsequent online ICP-MS determination (Fig. 1). Four species of As(III)/As(V) and Se(IV)/Se(VI) could be accurately determined without oxidation/reduction or subtraction, respectively.

Krishna et al. [17] monitored trace inorganic mercury (iHg) and methyl mercury (MeHg) in environmental water and fish tissues by a microcolumn filled with polyaniline (PANI) and online flow injection-chemical vapor generation (CVG)-ICP-MS. Different Hg species were adsorbed into the microcolumn through the N-containing functional groups on the polyaniline and then eluted. Under optimal conditions, the LOD is 2.52 pg for iHg and 3.24 pg for MeHg (in Hg). The developed method was applied to the direct determination of iHg and MeHg in tap, lake, ground, and sea water. Rodolfo et al. [18] adsorbed different iodophenols through various types of solid phase microextraction PDMS fibers. After elution, they were analyzed by gas chromatography (GC)-ICP-MS. The method can extract iodophenol in a variety of water samples effectively.

### 2.1.3 Nanoparticles Analysis

Metal-containing nanoparticles can be regarded as a specific existing form of the element. Solid phase (micro) extraction combined with ICP-MS detection has been applied for the analysis of metal-containing nanoparticles.

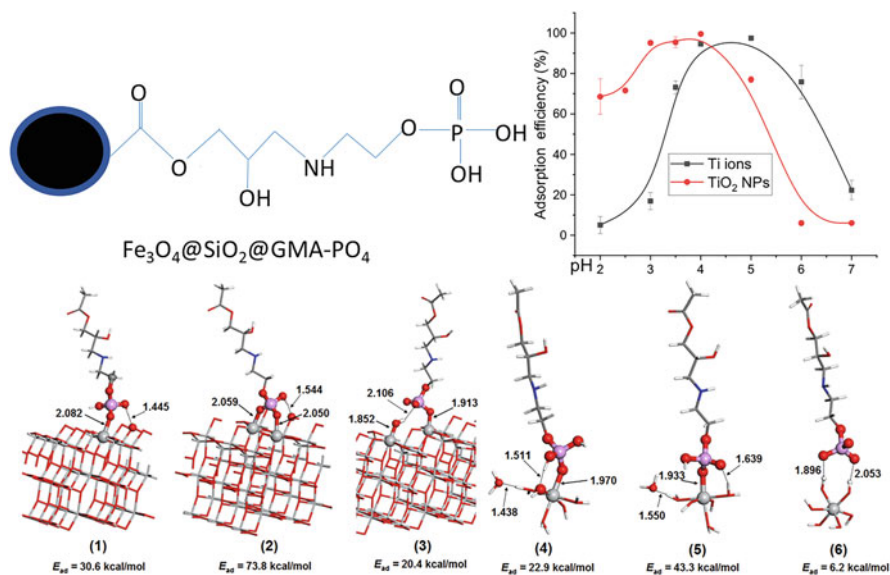
$\text{Al}^{3+}$  immobilized  $\text{Fe}_3\text{O}_4@ \text{SiO}_2$ @iminodiacetic acid (IDA) was prepared and its extraction performance for gold nanoparticles (AuNPs) and gold ions (Au ions) was investigated [19]. AuNPs and Au ions could be simultaneously retained on the adsorbent and their separation was achieved by sequential elution of Au ions and AuNPs with  $\text{Na}_2\text{S}_2\text{O}_3$  and  $\text{NH}_3\cdot\text{H}_2\text{O}$ , respectively. Based on it, a strategy by



**Fig. 1** Effect of pH on the adsorption behavior of inorganic As(III)/As(V) and Se(IV)/Se(VI) on AAPT-silica monolithic capillary (a) and MPTS-silica monolithic capillary (b). Reproduced from *J. Anal. At. Spectrom.*, 2009, 24(8): 1051–1061 with permission from the Royal Society of Chemistry

coupling MSPE with ICP-MS was developed for the speciation of AuNPs and Au ions in environmental water samples. AuNPs in a size range of 14–140 nm and with different coatings could be quantitatively determined. The size and shape of the AuNPs were kept unchanged during the extraction process. The method was applied to the analysis of AuNPs in natural water, sewage water, and even sea water samples. Wang et al. [20] synthesized phosphoric acid-functionalized superparamagnetic iron oxide. LOD of the MSPE-ICP-MS method for  $\text{TiO}_2\text{NPs}$  is  $17 \text{ ng L}^{-1}$  with the enrichment factor of 400. This method was verified to be applicable to the detection of trace  $\text{TiO}_2\text{NPs}$  in river and lake water. Besides, the adsorption mechanism of functionalized adsorbents on  $\text{TiO}_2\text{NPs}$  and titanium ions was investigated by dispersion-corrected density functional theory, which was assumed to be the competitive adsorption of different deprotonated phosphates and OH-anions against  $\text{TiO}_2\text{NPs}$  or aqueous titanium ions (Fig. 2).

Zhang et al. [21] prepared a hydrophilic polymer capillary monolithic column for the separation and enrichment of carboxyl-containing AuNPs and applied this method to the analysis of carboxyl AuNPs in a variety of water samples. The LOD is  $24.2 \text{ fmol L}^{-1}$ , the adsorption is mainly dependent on the affinity of the carboxyl group and the polymerized monomer on the capillary. Based on the difference in the elution behavior of AuNPs with different particle sizes on the monolithic column

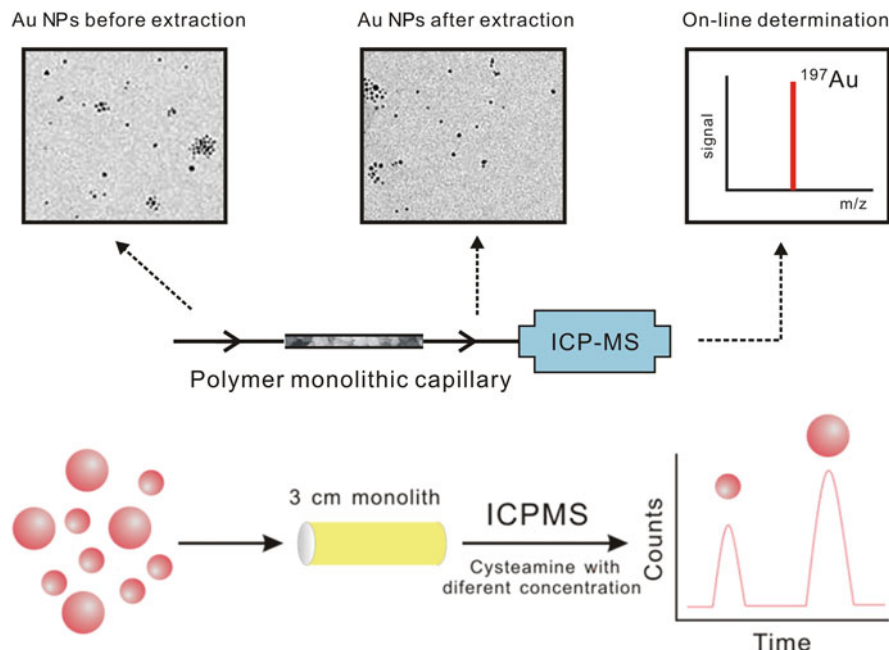


**Fig. 2** Phosphoric acid-functionalized magnetic sorbent used for determination of  $\text{TiO}_2$  NPs. Reproduced from *Sci. Total Environ.*, 2020, 703: 135464 with permission from 2019 Elsevier B.V.

(Fig. 3), Liu et al. achieved the separation of AuNPs with particle size in the range of 3–40 nm and quantification of AuNPs in water sample [22].

## 2.2 Liquid Phase (Micro) Extraction

Liquid phase (micro) extraction is based on the distribution of target elements or species between aqueous and organic phase, including traditional liquid–liquid extraction (LLE) and miniaturized systems. A variety of liquid phase microextraction (LPME) techniques have been developed, such as single drop microextraction (SDME), hollow fiber liquid phase microextraction (HF-LPME), dispersive liquid-liquid microextraction (DLLME), and solidified floating organic drop microextraction (SFODME) [23]. Cloud point extraction (CPE) based on the cloud point phenomenon of surfactant is also a liquid phase (micro) extraction mode. Compared to solid phase (micro)extraction, the application of liquid phase (micro) extraction in trace/ultra-trace elements especially speciation analysis is much less due to the lack of suitable extraction system, and each mode of LPME has its own drawbacks, such as the instability of solvent drop in SDME, the relatively long extraction time in HF-LPME, and the limited applicable organic solvent for DLLME and SFODME.



**Fig. 3** Analysis of AuNPs with different sizes by polymer monolithic capillary. Reprinted (adapted) with permission from *Anal. Chem.*, 2015, 87(3):1789–1796, and *Anal. Chem.*, 2017, 89(1):560–564. Copyright {2015, 2017} American Chemical Society

### 2.2.1 Trace Elemental Quantification

In LLE or LPME system for elemental analysis, a chelating reagent is usually necessary to transform the metal ions with strong polarity into nonpolar species, transferring target elements from aqueous solution into organic phase. To facilitate the introduction of organic phase obtained from LPME or LLE into ICP, conventional pneumatic nebulization can be replaced by ETV, which merits low sample consumption, ability of introducing solid, liquid, or organic solvents, and ability of removing sample matrix or solvents with the use of temperature program or chemical modifier.

Xia et al. [24] applied benzoylacetone as the extractant for Be, Co, Pd, and Cd, and developed a method of SDME-ETV-ICP-MS for trace analysis. Benzoylacetone acted as the chemical modifier in ETV simultaneously to improve the vaporization behavior of target analytes. The LODs of the developed methods are 0.12, 0.99, 1.5, and 0.27  $\mu\text{g mL}^{-1}$ , respectively. Phase transfer (PT)-HF-LPME combined with ETV-ICP-MS was developed for the determination of trace Co, Pd, Cd, and Bi in environmental and biological samples [25]. In PT-HF-LPME, an intermediate solvent (1-butanol) was added into the sample solution to ensure the maximum contact area between the target metal ions and the chelating reagent (8-hydroxyquinoline), which accelerated the formation of 8-hydroxyquinoline-metal complexes and their

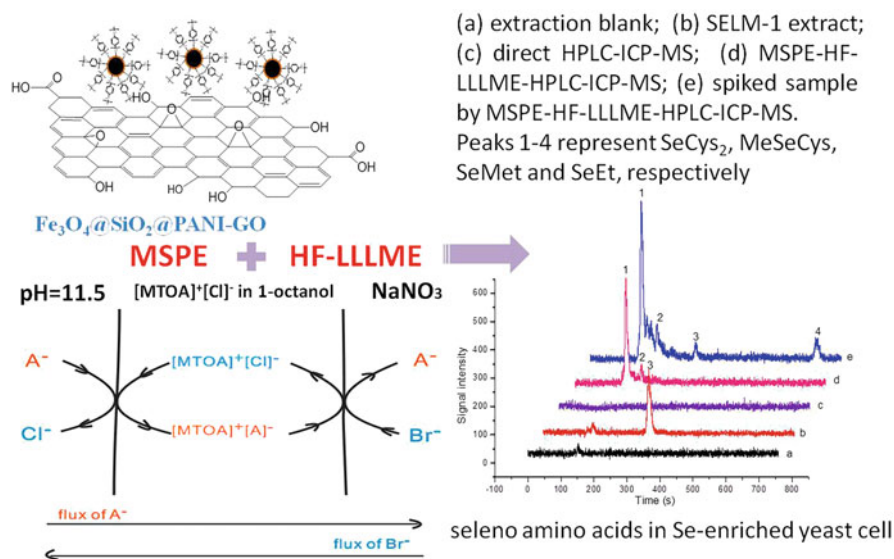
subsequent extraction by extraction solvent (toluene). The EFs for Co, Pd, Cd, and Bi were 110, 393, 121, and 111-fold, respectively, and the LODs ranged from 3.7 to 8.3 ng L<sup>-1</sup>. Moreover, SFODME was combined with ETV-ICP-MS for the determination of trace heavy metals in environmental water samples with sodium diethyldithiocarbamate (DDTC) as both chelating reagent in SFODME and chemical modifier in ETV [26]. Chandrasekaran et al. [27] extracted platinum group elements (Pd, Os, Ir, and Pt) and transition elements (Ag, Cd, Ta, and Re) in the form of chlorine and/or fluorine (anion) complexes through DLLME prior to ICP-MS analysis. During the extraction process, the anion complex formed in the presence of 1 ng L<sup>-1</sup> HCl and 0.5 ng L<sup>-1</sup> F<sup>-</sup> can be extracted into chloroform containing trioctyl methyl ammonium chloride at room temperature, and then the analytes are extracted from the chloroform layer by the mixture of chloric acid and nitric acid for subsequent determination by direct injection in the ICP-MS. The EF for target eight elements was 27–75. This method was applied to the determination of target metal elements in lake water polluted by industrial wastewater and hospital waste.

To further increase the EF, a two-step method involving dispersive solid phase extraction (D-SPE) and DLLME was developed for the preconcentration of 15 rare earth elements (REEs) in environmental water and sediment samples, followed by ETV-ICP-MS measurement [28]. With Chelex 100 as the adsorbent of D-SPE, the target REEs were firstly extracted and then desorbed by 0.1 mol L<sup>-1</sup> HNO<sub>3</sub>. After addition of 125 mmol L<sup>-1</sup> tris and 40 mmol L<sup>-1</sup> 1-phenyl-3-methyl-4-benzoylpyrazolone, target REEs were further preconcentrated into CCl<sub>4</sub> by DLLME. The resultant EFs varied from 234- to 566-fold.

### 2.2.2 Elemental Speciation Analysis

The use of solid phase (micro)extraction or liquid phase (micro) extraction prior to ICP-MS determination can only differentiate a limited number of elemental species, such as Cr(III)/Cr(VI), As(III)/As(V). The speciation of multiple elemental species depends more on chromatographic techniques combined with ICP-MS detection, which will be discussed in Sect. 3. Sometimes, the main contribution of sample pretreatment in speciation analysis is removal of sample matrix and preconcentration of target species, rather than separating interest species.

Ionic liquid-based carrier mediated hollow fiber liquid–liquid–liquid microextraction (IL-carrier mediated HF-LLLME) was combined with high performance liquid chromatography (HPLC)-ICP-MS for the speciation of five phenylarsenic compounds and arsenate in chicken and feed samples [29]. Target arsenic species were extracted from aqueous samples at basic pH into an organic phase (20% (v/v) of methyltrioctylammonium chloride [MTOA]<sup>+</sup>[Cl]<sup>-</sup> in toluene) impregnated in the pores of a hollow fiber, then back extracted into 10 μL acceptor phase of 0.3 mol L<sup>-1</sup> NaBr in the lumen of the hollow fiber. The main driving force for the extraction was the gradient concentration of counter ion from the donor phase to the acceptor phase. The LODs for five target phenylarsenic compounds and arsenate were in the range of 1.4–16 ng L<sup>-1</sup>, with EFs varying from 86 to 372.



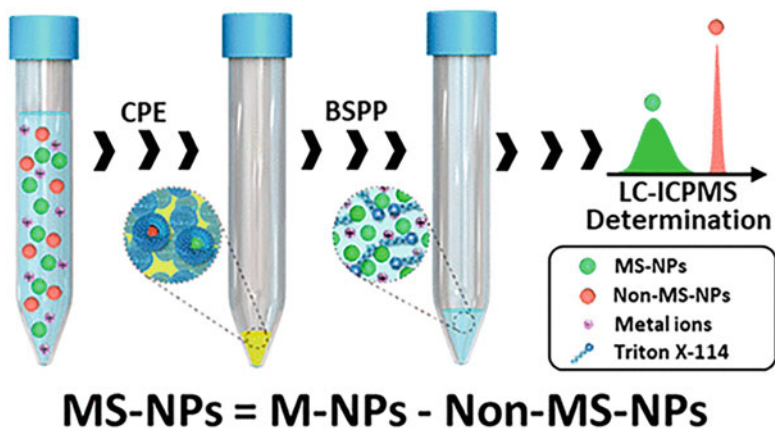
**Fig. 4** Speciation analysis of seleno amino acids by MSPE-HF-LLLME. Reproduced from Ref. [30] with permission from the Royal Society of Chemistry

MSPE was further combined with HF-LLLME for the extraction of seleno amino acids and subsequent determination by HPLC-ICP-MS [30]. Cu<sup>2+</sup> was adsorbed onto the surface of graphene oxide (GO) modified magnetic nanoparticles for the adsorption of seleno amino acids based on the interaction between Cu<sup>2+</sup> and seleno amino acids. Aqueous ethylenediamine was employed as desorption solvent, and the desorption solution was directly employed as the donor solution in the subsequent HF-LLLME (Fig. 4). After preconcentration by MSPE-HF-LLLME, the EFs for target seleno amino acids ranged from 152 to 278, and the LODs for target seleno amino acids were in the range of 0.0075–0.013 μg L<sup>-1</sup>. The method was applied to the speciation of seleno amino acids in rice and Se-enriched yeast cell samples.

### 2.2.3 Nanoparticles Analysis

Surfactant assisted DLLME combined with ETV-ICP-MS was proposed for the analysis of AuNPs in environmental water samples [31]. Effective separation of AuNPs from ionic gold species was achieved by using sodium thiosulfate as a complexing agent. LOD of 2.2 ng L<sup>-1</sup> was achieved for AuNPs with an EF of 152 and the original morphology of the AuNPs could be maintained during the extraction process. TX-114-based CPE was demonstrated by Liu et al. [32] to be an efficient approach for selective extraction and concentration of trace silver nanoparticles (AgNPs) from environmental water samples without disturbing their sizes and shapes. The highest extraction efficiency for AgNPs was obtained at about





**Fig. 5** Separation and determination of different nanoparticles by CPE-LC-ICP-MS. Reprinted (adapted) with permission from *Anal Chem*, 2020, 92(7):4765–4770. Copyright {2020} American Chemical Society

their zero point charge pH ( $\text{pH}_{\text{PZC}}$ ), which was  $\sim 3.0$ – $3.5$  for the studied AgNPs. Addition of salts such as  $\text{NaNO}_3$  or  $\text{Na}_2\text{S}_2\text{O}_3$  enhanced the phase separation and thus increased the extraction efficiency of AgNPs. Furthermore,  $\text{Na}_2\text{S}_2\text{O}_3$  efficiently eliminated the interference of  $\text{Ag}^+$  due to the formation of a complex between  $\text{Ag}^+$  and  $\text{S}_2\text{O}_3^{2-}$  that was not extracted into the TX-114-rich phase. Similar strategy was applied for the determination of  $\text{Ag}_2\text{S}$ - and  $\text{ZnS}$ -NPs in environmental water samples (Fig. 5) [33], with the LOD of  $8$ – $15 \text{ ng L}^{-1}$ .

### 3 ICP-MS Hyphenated with Chromatographic Techniques

The toxicity or bioavailability of a certain element is mainly attributed to its specific existing forms and relevant concentration. Therefore, not only the total amount of the elements, but also the speciation analysis would provide significant information for environmental pollution monitoring and food safety evaluation. Presently, ICP-MS is the most sensitive element-specific detection technique, but cannot provide species information due to the high temperature of ICP ionization source. With the aid of a highly efficient separation technique, this issue can be solved. So far, this strategy is most widely used for elemental speciation analysis. Various separation techniques have been combined with ICP-MS for the analysis of trace elements and their species in complex environmental and food samples. HPLC can be coupled with online ICP-MS detection easily by introducing the post-column effluent into the nebulizer with a connecting tube as short as possible. The interface between GC and ICP should be thermal insulated and the capillary column for chromatographic separation is inserted into ICP torch replacing the central tube. The interface between capillary electrophoresis (CE) and ICP needs compensation

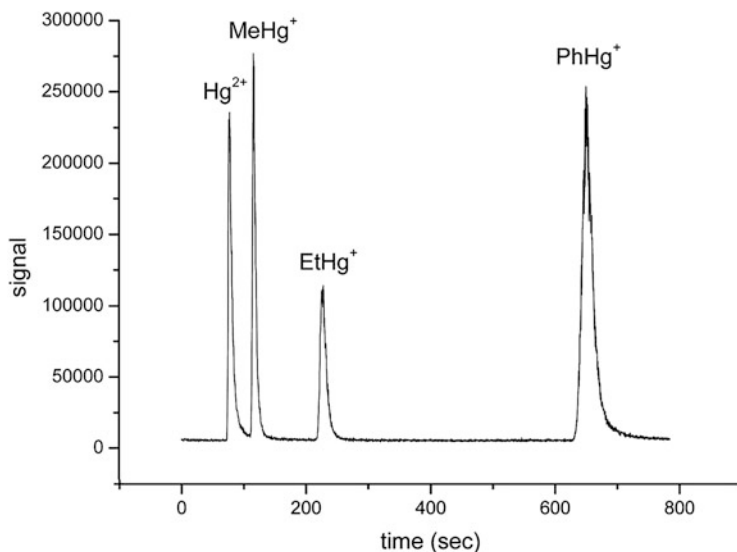
fluid and grounding. Gel electrophoresis is combined with ICP-MS detection often in an offline mode. The application of ICP-MS hyphenated with chromatographic techniques for the analysis of specific elemental species in food and environmental samples is discussed as follows.

### 3.1 Chromium (Cr)

Cr is widely used in chemical and material industry, and vast amounts of Cr-rich wastes are discharged into the environment, causing potential pollution. Inorganic Cr species, Cr(III)/(VI), are the main forms of Cr existing in the environment. Cr(III) is an essential trace element in human body and participates in normal glucose and lipid metabolism, while Cr(VI) is a kind of carcinogen with strong oxidation ability. Anion exchange chromatography is often used to separate them. Gao et al. [34] extracted Cr species from river water samples with ZrO diffusive gradients in thin films (DGT) and desorbed them by EDTA. Then Cr(III) and Cr(VI) existing in the form of  $[\text{Cr(III)-EDTA}]^-$  and  $\text{CrO}_4^{2-}$ , respectively, were subjected to HPLC-ICP-MS analysis. The separation and detection were completed within 4.5 min. It should be noted that the speciation analysis of Cr in solid samples, e.g., soil, requires additional pretreatment process, and the transformation of Cr species should not be ignored. Cr(III) was relatively stable under acidic conditions, and the oxidation of Cr(III) into Cr(VI) may occur under alkaline conditions. Quarles et al. [35] extracted Cr species from soil and sludge by hydrochloric acid and nitric acid; after filtration and dilution, the sample was directly analyzed by HPLC-ICP-MS. Drinčić et al. [36] used NaOH/ $\text{Na}_2\text{CO}_3$  system to extract Cr species from river sediment. To avoid transformation of Cr(III) species, a little  $\text{MgCl}_2$  was used as protective agent. The result for Sava River revealed that Cr(VI) levels are very low even at industrially exposed sites, not representing a considerable environmental burden.

### 3.2 Mercury (Hg)

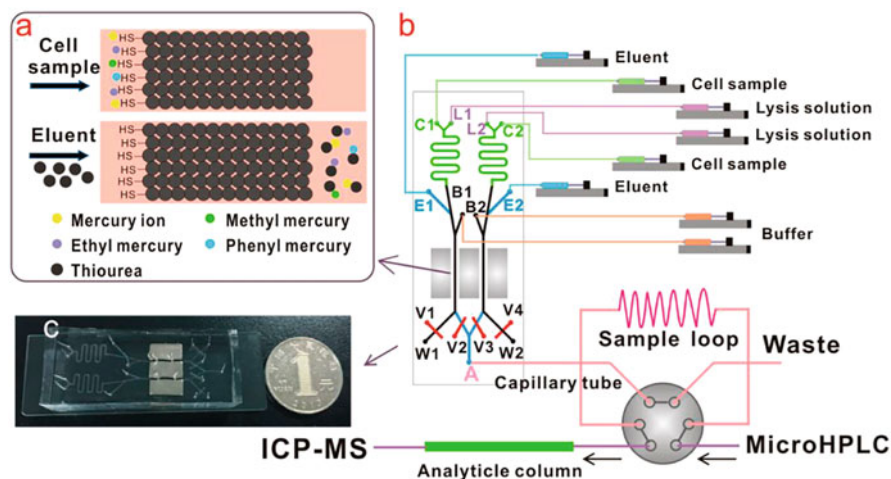
Hg can be easily absorbed by plants, enriched in the food chain, and transformed into various organic forms. Hg exists in the environment mainly in the form of Hg(II), methylmercury ( $\text{MeHg}^+$ ), ethylmercury ( $\text{EtHg}^+$ ), and phenyl mercury ( $\text{PhHg}^+$ ). Among them,  $\text{MeHg}^+$  is highly toxic and humans can be easily exposed to it through seafood. HPLC-ICP-MS is the most commonly used methodology for mercury speciation analysis. With reversed phase C18 column as the stationary phase, various sulfur-containing complexing agents (e.g., 2-mercaptoethanol, L-cysteine) are added into the mobile phase to improve separation and reduce the adsorption of mercury in the HPLC system. Xia et al. [37] applied 2-mercaptoethanol as the complexing reagent in the mobile phase for the separation of three organomercury species. The content of methanol in mobile phase significantly affected the



**Fig. 6** Chromatogram of four mercury species obtained by HPLC-ICP-MS. Each species of  $500 \mu\text{g L}^{-1}$  (as Hg). Column temperature:  $50^\circ\text{C}$ , mobile phase: methanol-buffer solution ( $2.5 \text{ mmol L}^{-1}$  L-cysteine,  $12.5 \text{ mmol L}^{-1}$   $(\text{NH}_4)_2\text{HPO}_4$ , 0.05% triethylamine, pH 7.0) (8:92), flow rate:  $1.5 \text{ mL min}^{-1}$ . Injection volume:  $8 \mu\text{L}$ . Reproduced from *J. Anal. At. Spectrom.*, 2015, 30(4):875–881 with permission from the Royal Society of Chemistry

separation resolution of  $\text{MeHg}^+$  and  $\text{EtHg}^+$ . Under the optimized conditions (mobile phase: methanol/water (7:3, v/v) containing 0.001% 2-mercaptoethanol and  $20 \text{ mmol L}^{-1}$  buffer solution, pH 6.0; isocratic elution with a flow rate of  $1 \text{ mL min}^{-1}$ ), baseline separation of three organomercury species was realized within 5 min. With L-cysteine as the complexing reagent,  $\text{Hg}(\text{II})$ ,  $\text{MeHg}^+$ ,  $\text{EtHg}^+$ , and  $\text{PhHg}^+$  were separated in 12 min (Fig. 6) [38]. With the aid of HF-LLLME, enrichment factors of 120–350 [37] and 221–236-fold [38] were target mercury species, and the quantification limit of the method for target species was significantly improved.

Zhu et al. [39] proposed a MSPE-HPLC-ICP-MS method for the simultaneous speciation of mercury including  $\text{Hg}^{2+}$ ,  $\text{MeHg}^+$ , and  $\text{PhHg}^+$  in environmental water and fish samples. The functionalization of magnetic sorbents was realized by using  $\gamma$ -mercaptopropyltrimethoxysilane ( $\gamma$ -MPTS), and target mercury species were preconcentrated based on their affinity with sulfhydryl groups. Then, rapid separation of three target mercury species was achieved in 8 min by HPLC. By using  $100 \text{ mL}$  sample solution and  $0.5 \text{ mL}$  eluent, enrichment factor of 200-fold was achieved, with LODs of 0.74, 0.67, and  $0.49 \text{ ng L}^{-1}$  for  $\text{Hg}^{2+}$ ,  $\text{MeHg}^+$ , and  $\text{PhHg}^+$ , respectively.  $\text{Fe}_3\text{O}_4@ \text{SiO}_2@ \gamma\text{-MPTS}$  particles were further packed into the channel on microfluidic chip, and chip-based MSPE was fabricated for the analysis of mercury species in cells (Fig. 7) [40]. It is suitable for the analysis of real sample



**Fig. 7** Diagram of the magnetic packed column, microfluidic chip, and online chip-based MSPME-microHPLC-ICP-MS system. **(a)** Microextraction principle in the magnetic packed column. **(b)** Design sketch of online chip-based MSPME-microHPLC-ICP-MS system: the chip consists of a cell lysis unit (green lines), microextraction unit (black lines), and microvalves (red lines), the blue lines stand for the elution channels; C1 and C2 are the inlets of cell sample, L1 and L2 are the inlet of lysate solution, E1 and E2 are the inlets of eluent, B1 and B2 are the inlet of  $\text{Fe}_3\text{O}_4@\text{SiO}_2@\gamma\text{-MPTS}$  and buffer solution, A is the outlet of eluent and is connected with a six-way valve by a capillary tube directly, W1 and W2 are the outlets of waste, and V1, V2, V3, V4 are the gas inlets of microvalves. The width of the microextraction channels is 400  $\mu\text{m}$ , and the microvalves channels are 500  $\mu\text{m}$  wide; all the channels are 50  $\mu\text{m}$  high. **(c)** Photograph of the microfluidic chip. Reprinted (adapted) with permission from *Anal Chem*, 2016. 88(1):796–802. Copyright {2016} American Chemical Society

with limited volume/amount available, and has been used to explore the antagonism of selenium and mercury [41].

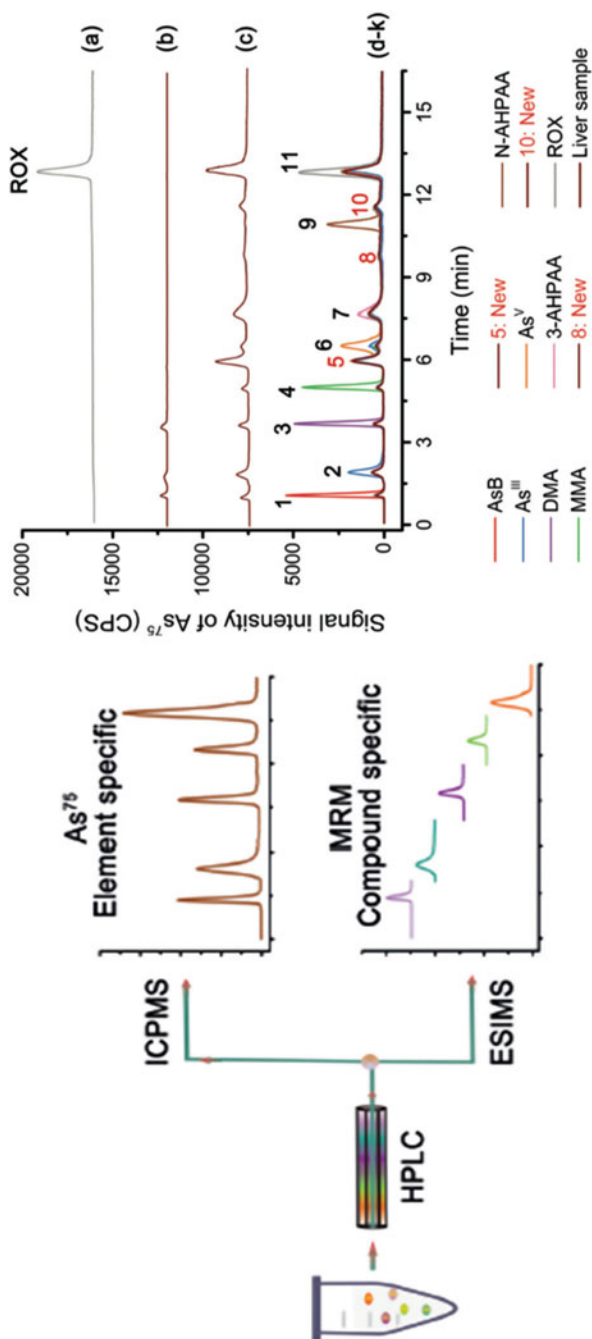
Entwisle et al. [42] extracted Hg species in fish samples using a mild system containing L-cysteine, protease, and esterase. The chromatographic baseline separation of  $\text{MeHg}^+$  and  $\text{iHg}$  was achieved in isocratic mode using reverse phase HPLC with a mobile phase containing L-cysteine as a complexing agent.  $\text{Hg}^{2+}$ ,  $\text{MeHg}^+$  and their isotopic ratio were analyzed by HPLC and CVG-MC-ICP-MS. Song et al. [43] analyzed Pb, Hg species in water by an online SPE-HPLC-ICP-MS method. After treatment with sulfhydryl reagent, target species can be enriched in a  $\text{C}_{18}$  column and desorbed by mobile phase containing L-cysteine and tetrabutyl ammonium hydroxide solution. Six elemental species were simultaneously analyzed with EFs of 459–2485.

### 3.3 Arsenic (As)

Arsenic exists as more than 50 chemical species in the environment and biological systems, which can be mainly divided into inorganic arsenic and organic arsenic species. Generally, the toxicity of inorganic arsenic is much higher than that of organic arsenic, and among inorganic arsenic, the toxicity of As(III) is much higher than that of As(V). As species can easily transform through redox or methylation reactions, making their analysis challenging. Letsoalo et al. [44] extracted As species from river sediments by  $(\text{NH}_4)_2\text{HPO}_4/\text{EDTA}$  assisted with microwave and analyzed them by HPLC-ICP-MS. The analysis of As(III)/(V) and monomethyl arsenic distribution in samples from different areas along Mokolo River showed that As-containing pesticides used in farming activities might be potential sources. Yang et al. [45] optimized the solvent system in DLLME by using solvent with low density and higher compatibility with mobile phase, to extract As species from water. Various As species in river water could be easily and quickly determined by HPLC-ICP-MS. Zhao et al. [46] used protease and hydrochloric acid to digest and extract As species from fish and analyzed them by HPLC-ICP-MS, the matrix interference was effectively reduced. Milani et al. [47] used microwave-assisted extraction with dilute nitric acid to extract As species from herbal tea. Four As species, As(III), As(V), monomethyl arsenic (MMA), and dimethyl arsenic (DMA), were analyzed by HPLC-ICP-MS and it was found that As(V) was the dominate species in Herbal tea from Brazil.

An organic arsenic species, 3-nitro-4-hydroxyphenylarsonic acid (Roxarsone, ROX) has been widely used as a feed additive in poultry industry for decades. However, most metabolites of ROX have not been identified. Peng et al. [48] investigated the arsenic species in chickens fed by ROX-containing food based on HPLC-ICP-MS and ESI-MS (Fig. 8). Methyl-3-nitro-4-hydroxyphenylarsonic acid (methyl-ROX), methyl-3-amino-4-hydroxyphenylarsonic acid (methyl-3-AHPAA), and methyl-3-acetamido-4-hydroxyphenylarsonic acid (or methyl-N-acetyl-ROX, methyl-N-AHPAA) were identified in the chicken livers, and the concentration of methyl-ROX was as high as  $90 \text{ mg kg}^{-1}$ , even after a five-day clearance period. The formation of these newly discovered methylated metabolites from reactions involving trivalent phenylarsonous acid substrates, S-adenosylmethionine, and the arsenic (+3 oxidation state) methyltransferase enzyme As3MT suggests that these compounds are formed by addition of a methyl group to a trivalent phenylarsenical substrate in an enzymatic process. The IC<sub>50</sub> values of the trivalent phenylarsenical compounds were 300–30,000 times lower than those of the pentavalent phenylarsenicals.

Volatile As species play a major role in the transfer of As to the atmosphere, Ilgen et al. [49] developed a cryotrapping-cryofocusing (CT-CF) based method to capture volatile As species in gas samples or in samples after headspace extraction, followed by GC-ICP-MS. Sele et al. [50] used anion exchange column as SPE sorbents to enrich As species in fish oil. GC-ICP-MS analysis showed that As in fat accounted for 13–35% of the total As in fish oil. Comparatively, the application of CE-ICP-MS



**Fig. 8** Chromatograms from HPLC-ICP-MS analyses of ROX and chicken liver samples from a ROX-fed chicken and a control chicken. (a) ROX standard. (b) A liver sample from a control chicken fed the basal diet. (c) A liver sample from a chicken fed the ROX-containing diet. (d-k) Analysis of the same chicken liver sample after replicate aliquots were separately spiked with AsB (d, peak 1), AsIII (e, peak 2), DMA (f, peak 3), MMA (g, peak 4), AsV (h, peak 6), 3-AHPAA (i, peak 7), N-AHPAA (j, peak 9), and ROX (k, peak 11). Peaks 5, 8, and 10 did not correspond to any available arsenic standards. Reproduced from Angew. Chem.-Int. Ed., 2017, 56(24):6773–6777 with permission from John Wiley and Sons (2017)

in trace As speciation analysis is the least although good separation resolution can be allowed [51, 52].

### 3.4 Selenium (Se)

Se is an essential trace element for human, and various compounds containing Se have also been proved to be beneficial to immunity and metabolism. While the excess intake of Se will cause toxic and side effect, and its normal concentration range in human body is relatively narrow. The analysis of Se species in food can provide important nutritional reference. Piñeiro et al. [53] compared several extraction methods to enrich Se species from crops, including acid and enzyme digestions, followed by HPLC-ICP-MS analysis. It was found that both pressure-assisted and microwave-assisted enzymatic hydrolysis showed good extraction efficiency. The author analyzed Se species in Brazilian golden berries and nuts and found that SeMet was the dominant species ( $45.2\text{--}47.6\ \mu\text{g g}^{-1}$  compared to  $0.76\text{--}0.81$ ,  $<0.306$ , and  $<0.119\ \mu\text{g g}^{-1}$  for selenocystine (SeCys<sub>2</sub>), Se(IV), and Se(VI), respectively). Mao et al. [54] combined SBSE with HPLC-ICP-MS for the analysis of seleno amino acids including SeCys<sub>2</sub>, methylseleno-cysteine (MeSeCys), selenomethionine (SeMet) and selenoethionine (SeEt) and seleno-oligopeptides including  $\gamma$ -glutamyl-Se-methyl-selenocysteine ( $\gamma$ -GluMeSeCys) and selenodiglutathione (GS-Se-SG) in yeast and garlic samples. The partially sulfonated polystyrene-titania organic-inorganic hybrid stir bar coating can directly extract high polar seleno amino acids and seleno-oligopeptides.

With the derivatization by ethyl chloroformate (ECF), Duan et al. [55] proposed a method based on HF-LPME coupling with GC-ICP-MS for the speciation of seleno amino acids in the extracts of garlic, cabbage, and mushroom samples. Under the optimal conditions, the LODs obtained for Se-methyl-selenocysteine (SeMeCys), selenomethionine (SeMet), and selenoethionine (SeEth) were 23, 15, and 11 ng Se L<sup>-1</sup>, respectively. Moreover, headspace sorptive extraction (HSSE) was combined with GC-ICP-MS for the determination of volatile organo seleno compounds (dimethylselenide (DMSe) and dimethyldiselenide (DMDS)) in food samples [56]. DMSe and DMDS retained on the sorptive bar were desorbed by methanol and introduced into GC-ICP-MS for separation and determination. LODs of 33 and 7.1 ng L<sup>-1</sup> were obtained for DMSe and DMDS, respectively. The proposed method was applied for the analysis of DMSe and DMDS in garlic, onion and their juice samples, and a good agreement was obtained between the results obtained by standard addition and external calibration methods.

Duan et al. [57] synthesized phenylalanine derivative (L-HP-Phe, L-N-(2-hydroxy-propyl)-phenylalanine) and the chelate of Cu(II)-(L-HP-Phe)<sub>2</sub> was used as the chiral selector for the separation of D, L-SeMet. Based on it, a method of SPE-large volume sample stacking (LVSS)-micelle electrokinetic chromatography (MEKC)-UV detection was proposed for the chiral speciation of D, L-SeMet. With the enrichment factor of 1,400 and 1,378, the LODs of 0.44 and 0.60 ng mL<sup>-1</sup>

for L-SeMet and D-SeMet were obtained, respectively. A certified reference material SELM-1 was analyzed and the results showed that most of SeMet in the selenized yeast sample SELM-1 was L isomer. Zhao et al. [58] used enzyme-assisted extraction method to enrich Se species in rice samples. CE-ICP-MS was used for subsequent analysis with the LOD of 0.1–0.9 ng Se/mL.

### 3.5 Tin (Sn)

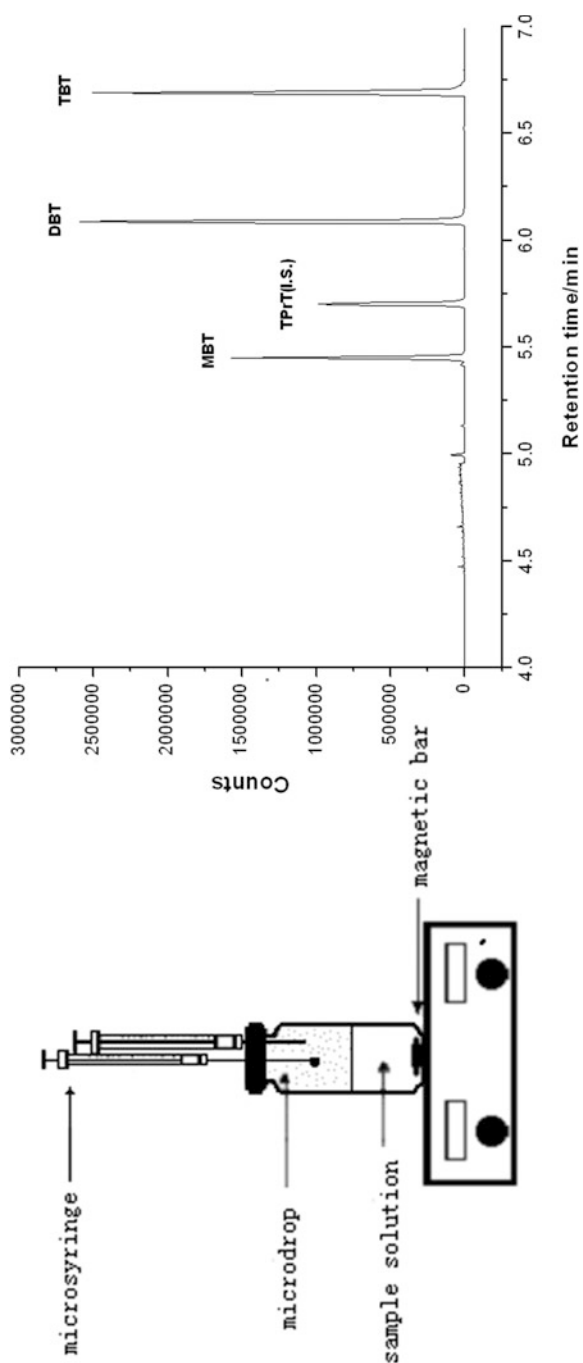
Organotin compounds are now well known as endocrine disruptors and their extensive use as biocides in antifouling paints in the past decades has led to serious environmental problems. Especially in the marine ecosystem, the high toxicity of these compounds toward non-target organisms would cause ecological damage, and influence the human's health through the food chains. Among the organotins, triorganotins are the most biologically active and toxic, which could disturb the functions of mitochondria in tissues and cells; di- and monosubstituted compounds are the main degradation products of triorganotins in the environment. Xiao et al. [59] proposed a method based on headspace single drop microextraction (HS-SDME) in combination with GC-ICP-MS for the speciation analysis of butyltin compounds in seawater and shellfish samples. Sodium tetraethylborate ( $\text{NaBEt}_4$ ) and sodium tetrahydroborate ( $\text{NaBH}_4$ ) were used as the derivatizing reagent for in situ derivatization of the butyltins;  $\text{NaBEt}_4$  derivatization was found to be more sensitive and robust than  $\text{NaBH}_4$ , leading to the LODs of  $1.4 \text{ ng L}^{-1}$  for monobutyltin (MBT),  $1.8 \text{ ng L}^{-1}$  for dibutyltin (DBT), and  $0.8 \text{ ng L}^{-1}$  for tributyltin (TBT). The butyltins found in the real-world samples are  $31 \text{ ng L}^{-1}$  MBT,  $79 \text{ ng L}^{-1}$  DBT, and  $32 \text{ ng L}^{-1}$  TBT for seawater, and  $11.6\text{--}30.4 \text{ ng g}^{-1}$  MBT,  $11.8\text{--}8.9 \text{ ng g}^{-1}$  DBT, and  $12.8\text{--}52.6 \text{ ng g}^{-1}$  TBT for different shellfish samples. The accuracy of the method was validated by analyzing certified reference material (CRM) of PACS-2 sediment (Fig. 9).

Hu et al. [60] tried to analyze Hg and Sn species in sea water by combining dynamic headspace in-tube SPME and GC-ICP-MS. Water sample was treated with  $\text{NaBPr}_4$  [61] to convert Hg and Sn species into gas forms and extract them by a syringe filled with adsorbent material. In this way, TBT, DBT, MBT,  $\text{MeHg}^+$ , and inorganic mercury (iHg) can be analyzed simultaneously.

### 3.6 Other Elemental Species

Lajin et al. [62] used HPLC-ICP-MS to analyze simultaneously five organophosphorus pesticides including glyphosate in water samples. Nelson et al. [63] used GC-ICP-MS to analyze the organic sulfur, organophosphorus, and organochlorine residues on vegetables, tea, and other crops. Faßbender et al. [64] used CE-MC-ICP-





**Fig. 9** Chromatogram (right) of the PACS-2 sediment obtained after  $\text{NaBEt}_4$ -HS-SDME (left) with GC-ICP-MS detection. Conditions:  $0.5 \mu\text{g g}^{-1}$  TPPrT as internal standard (I.S.), 5 min extraction with  $2 \mu\text{L}$  decane at a stirring rate of 600 rpm, injection volume  $1.0 \mu\text{L}$ . Reproduced from J. Chromatogr. A, 2008, 1211(1–2):135–141 with permission from 2008 Elsevier B.V.

MS to detect the S species in river water (mainly sulfates) and compared the difference of S isotope of samples from different rivers.

Polybrominated diphenyl ethers (PBDEs) generally come from the combustion of flame retardants and are exposed to human body in the food or dust. Xiao et al. [65] proposed a sensitive method based on LPME and GC-ICP-MS for the quantification of four PBDEs in human serum, soil, and dust samples. The concentration of PBDEs found in the soil sample was 26.48 ng g<sup>-1</sup> BDE-28, 172.19 ng g<sup>-1</sup> BDE-47, 18.74 ng g<sup>-1</sup> BDE-99, and 87.42 ng g<sup>-1</sup> BDE-100. The concentration of PBDEs found in the dust sample was 0.28 μg g<sup>-1</sup> BDE-28, 1.47 μg g<sup>-1</sup> BDE-47, 1.50 μg g<sup>-1</sup> BDE-99, and 0.45 μg g<sup>-1</sup> BDE-100. Novak et al. [66, 67] detected six kinds of PBDEs in sludge from wastewater treatment plants by GC-ICP-MS. The use of potassium bromate as flour modifier has been banned in many countries, but illegal use still exists. Cheng et al. [68] used microchip electrophoresis (MCE) in combination with ICP-MS to analyze Br<sup>-</sup> and BrO<sub>3</sub><sup>-</sup> in bread. Sodium acetate was used as the electrophoresis buffer, and the separation was completed in 35 s. Circulating volatile methyl siloxane (cVMS) is used in silicon polymer production or some personal care products. It can be easily transported long distances in the atmosphere and finally accumulate in the food chain. The safety of cVMS is under review. River water and fish samples were pretreated by LLE, then the cVMS was analyzed by GC-ICP-MS [69]. Screening results from several rivers in Germany showed that the cVMS content decreased year after year, suggesting a natural degradation. However, the cVMS in fish are still above the safety standard, indicating a risk of cVMS enrichment through the food chain. To a large extent, harmful elements in organism exist in the form of binding with small biological molecules or proteins, and thus continue to accumulate in organisms.

The separation and analysis of metal-binding proteins helps to understand the uptake and transformation pathways of these elements. 2D-gel electrophoresis and LA-ICP-MS were used to investigate the composition of metal-binding proteins in plant roots and stems [70]. Once the protein components were extracted from the homogenate of plant roots and stems, they were separated by 2D-gel electrophoresis. Then LA-ICP-MS was used to imaging Cu, Zn, Na, Mg, K, Ca, Mn, Fe, P, and S, revealing the presence of Cu, Zn, and Mg-binding proteins. Furthermore, these protein components were further digested and analyzed by LC-MS/MS for a discussion about Cu tolerance and accumulation mechanism. Similar methodology was applied to investigate the U-binding proteins in sea shrimp [71]. Eleven kinds of U-containing proteins were found and six of them could correspond to the database, among which ferritin was the main U-containing component. Size exclusion chromatography (SEC) was combined with ICP-MS to analyze the metal components in human whey protein [72]. Mn, Co, Cu, and Se-containing proteins were found. The content of these elements met the reference standard of human milk given by the World Health Organization (WHO).

ZnO NPs are a potential environmental pollutant that has attracted more attention in recent years. Wojcieszek et al. [73] investigated the uptake and transformation of ZnO NPs and ZnCl<sub>2</sub> during the growth cycle of lettuce. Tissue samples were extracted by a three-step extraction (ammonium acetate buffer, pectinase, and

cellulase) reaching an extraction efficiency for Zn higher than 90%. The results obtained by SEC-HILIC-ICP-MS showed that the water-soluble Zn-containing fraction was 16%. Further characterization by ESI-MS showed that the nicotiamine-bound state is the main Zn species.

## 4 Field Flow Fractionation (FFF)-ICP-MS

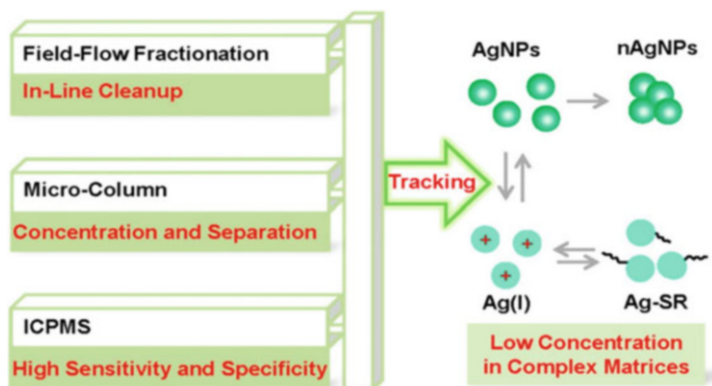
Field flow fractionation (FFF) is a general hydrodynamic separation technology, which is used to separate macromolecules, colloids, nanoparticles, and particles according to the size or mass. FFF is a high-resolution analytical technology that can provide continuous size classification from 1 nm to 100  $\mu\text{m}$ . According to different applied fields, FFF can be divided into flow field flow separation (Fl FFF), settlement field flow separation (Sd FFF), electric field flow separation (El FFF), and thermal field flow separation (Th FFF). The combination of FFF (high separation) and ICP-MS (high sensitivity) has great application potential to separate nanoparticles in environment and food. At present, the combination of FFF and ICP-MS is mainly divided into Sd FFF-ICP-MS and Fl FFF-ICP-MS.

### 4.1 Settlement Field Flow Separation (Sd FFF)

Beckett et al. [74] first proposed the coupling of FFF with ICP-MS, and Sd FFF was first used to be combined with ICP-MS. It was used to separate complex particles or macromolecular mixtures and obtain the complete distribution of element composition within the size range of the sample. Taylor et al. [75] used offline Sd FFF and ICP-MS to determine the soil colloidal components and characterize river suspended solids. This hyphenated technique was then applied to monitor Al, Fe, Mg, Rb, Si, and Sr on clay mineral samples and river suspended particles [76]. By drawing appropriate element atomic ratio distribution diagrams (such as Si/Al, Mg/Al), the change of particle composition was evaluated along with the variation of chemical composition of the mixture. The size distribution of specific components can be obtained from complex mixtures by using appropriate tracer elements. Sd-FFF-ICP-MS was widely used in the analysis of trace elements in particulate matter [77, 78], colloid size determination [79], the distribution of metals in colloid [80], and the influence of natural organic matter (NOM) [81].

### 4.2 Flow Field Flow Separation (Fl FFF)

Fl FFF is the most used separation mode in FFF recently. At present, it is mainly divided into symmetrical flow field flow separation method (F4), asymmetric flow



**Fig. 10** Hollow fiber flow field-flow fractionation (HF5), microcolumn concentration, and inductively coupled plasma mass spectrometry (ICP-MS) online coupling system for studying the aggregation and chemical transformation behavior of silver nanoparticles. Reprinted (adapted) with permission from *Environ. Sci. Technol.*, 2017, 51(21):12369–12376. Copyright {2017} American Chemical Society

field flow separation method (AF4), and hollow fiber flow field flow separation method (HF5). Among them, AF4 is the most commonly used field flow separation mode. Hoque et al. [82] used both online and offline AF4-ICP-MS, to determine the size and concentration of AgNPs in surface water samples and urban sewage samples. The offline mode can be easily implemented in laboratories without any special requirement. Saenmuangchin et al. [83] used self-made polysulfone hollow fiber membrane to establish a method combining HF5 and ICP-MS for the separation of AgNPs. The effects of carrier liquid and stabilizer on the retention behavior of AgNPs were studied. Tannic acid was added to the carrier solution to modify the surface of AgNPs and the surface of membrane to adjust the retention behavior of AgNPs. The volume of HF5 channel is relatively smaller than that of AF4 or F4 channel, resulting in less dilution of NPs in HF5 separation and more sensitive detection. The aggregation of AgNPs under aqueous conditions was also studied by HF5-ICP-MS [84]. It was found that in the presence of various concentrations of  $\text{Na}^+$  and  $\text{Ca}^{2+}$  ions, the attachment efficiency of AgNPs at concentration of  $10 \text{ ng mL}^{-1}$  was much higher than that reported in other studies performed at greater concentrations ( $\mu\text{g mL}^{-1}$ ) (Fig. 10). It proved the sensitivity of HF5-ICP-MS in studying NPs aggregation at low concentrations.

### 4.3 Application of FFF-ICP-MS in Environment and Food Samples

At present, FFF-ICP-MS is mainly used in NPs and colloids analysis. The application of FFF-ICP-MS in environment and food is introduced below.

### 4.3.1 AgNPs

Antonio et al. [85] evaluated the behavior of AgNPs in artificial seawater by AF4 SP ICP-MS, and studied the relationship between the aggregation process of AgNPs in water and environmental parameters (e.g., temperature, dissolved organic matter, and salinity) by measuring the particle size distribution in a long culture time. Chang et al. [86] established online AF4-UV vis -ICP-MS to detect the size and concentration of AgNPs in aqueous matrix. The author performed a combination of online and offline AF4-ICP-MS. Both showed similar response ranges, demonstrating the ability of the method for AgNPs analysis in surface water and urban sewage samples.

Loeschner et al. [87, 88] first used AF4-ICP-MS for the separation of AgNP in food matrices. After the enzymatic sample preparation, AF4-ICP-MS was used to separate the NPs from the remaining meat matrix, and the recovery of AgNPs was about 80%. The determination of AgNPs mass fraction and size exhibited acceptable reproducibility, proving the potential of AF4-ICP-MS in the quantitative analysis of NPs in complex food matrices for food monitoring and control.

### 4.3.2 AuNPs

Meisterjahn et al. [89] established a method of FI FFF-ICP-MS to study the applicability of the detection, quantification, and characterization of AuNPs in the presence of natural nanoparticles (NNP). The results showed that light scattering detection cannot evaluate mixed samples containing AuNP due to the specific interaction between AuNP and NNP, while FFF-ICP-MS was helpful to distinguish the heterogeneous aggregation and homogeneous aggregation of NPs. Lopez Sanz et al. [90] used AF4-ICP-MS to separate gold nanorods (AuNRs) and nanospheres (AuNSs). Generally, AuNSs are the main impurity in the synthesis of AuNRs. Under the optimal separation conditions, sufficient resolution was achieved between commercially AuNSs and AuNRs. This method can be applied for the analysis of products obtained in the synthesis of AuNRs.

### 4.3.3 Other NPs

Soto Alvarado et al. [91] applied online Sd FFF-ICP-MS/MS for the analysis of TiO<sub>2</sub> NPs in lake water samples. The analysis of different water samples (Milli Q water, seawater, and lake samples) showed that the ionic strength had a great impact on the analysis of TiO<sub>2</sub> NPs. Seawater with high salt concentration would lead to obvious aggregation/agglomeration. Li et al. [92] proposed a method of AF4-ICP-MS to characterize the size and quantify the quality of SiO<sub>2</sub> NPs in complex high-fat coffee creamer. By using prechannel calibration with SiO<sub>2</sub> NPs standard and postchannel calibration with ionic silicon standard, good recovery was obtained in the mass content determination of SiO<sub>2</sub> NPs. In addition, AF4-ICP-MS was also used for Cu

NPs in rainwater samples [93], SeNPs in water-based acrylic adhesives for food packaging materials [94], and the behavior of PtNPs in water [95], which proved the application potential of this technology in monitoring NPs in environment and food.

#### 4.3.4 Colloid

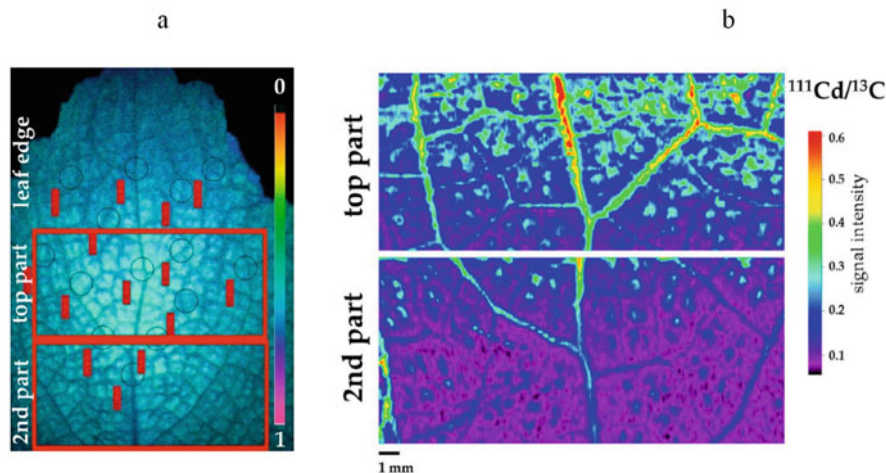
Worms et al. [96] used AF4-ICP-MS to analyze the association between metals and colloidal organics from wastewater treatment plants. It was found that most metals such as Ag, Cd, Cu, Cr, Mn, and Zn are only related to colloidal organics with low molecular weight fractions, while Al, Fe, and Pb are related to colloidal organics with high molecular weight fractions. Saito et al. [97] used FI FFF-ICP-MS to detect the size distribution and elemental composition of nanocolloids in granite groundwater. Geochemical parameters (such as pH value and main element concentration) showed a great influence on the size distribution of organic colloids. Different trace elements revealed different size distributions, reflecting their preferential combination with certain colloidal phases. For example, the distribution of Al, Mg, and Lu was consistent with the particle size distribution of organic colloids, while most other lanthanides and actinides had been proved to be related to organic colloids with small particle size. Moens et al. [98] studied the influence of NOM concentration on the size of Fe-organic carbon colloid and its influence on phosphate absorption by FI FFF-UV-ICP-MS. The results confirmed that the increasing NOM increased specific surface area by limiting crystal growth and outperformed phosphate adsorption in the competition.

## 5 Laser Ablation-ICP-MS

Since 1995, LA-ICP-MS has been applied in elemental imaging with high sensitivity. It is a powerful technique for direct analysis of solid samples, enabling micro-analysis, depth profiling analysis, and 2-D elemental mapping. Comparatively, much more applications of LA-ICP-MS are found in geological field, rather than that in biological and food samples.

### 5.1 *Plant Samples Analysis*

In LA-ICP-MS analysis of the plant samples, most of the current research is focused on (1) the relationship between the biological cycle or physiological function and distribution of some elements and their transporters in plants, (2) transport mechanism and enrichment factors, (3) distribution of some rare or toxic metal elements in plant tissue, and (4) some in-depth studies on the uptake and translocation mechanisms of NPs.

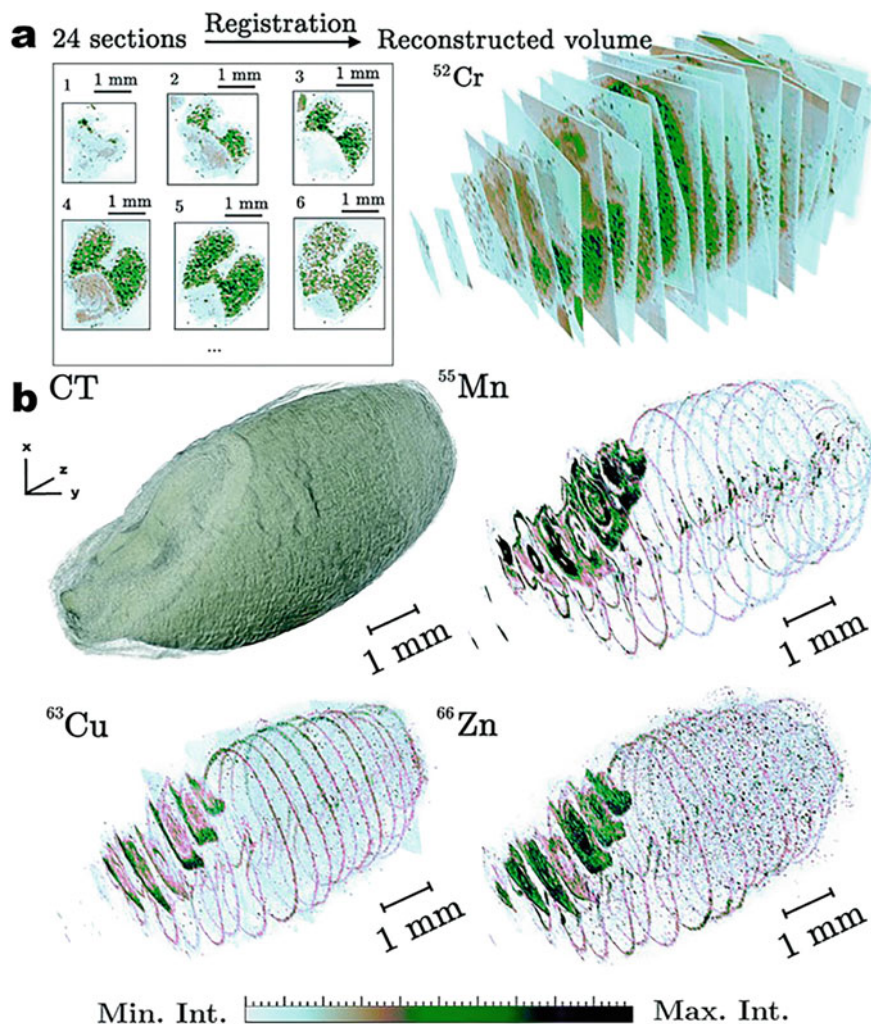


**Fig. 11** CA-IF and LA-ICP-MS Image of *Salvia sclarea* leaves exposed to 100  $\mu\text{M}$  Cd. Reproduced from Materials, 2019. 12(18):2953 (Open Access)

The distribution of elemental Cd in the leaves was observed by LA-ICP-MS with high resolution (Fig. 11a), which helped to better understand its distribution and the relationship between the amount of photosynthetic rate and excess energy dissipation [99]. The quantum yield of photosystem II electron transport ( $\Phi_{PSII}$ ) in the leaves of *Salvia sclarea* after 5 days of  $\text{Cd}^{2+}$  exposure was measured by chlorophyll fluorescence imaging (CF-IA). CF-IA can only vertically divide the leaf into three distinguishable parts and  $\Phi_{PSII}$  is affected by the fluorescence intensity (Fig. 11b). The combination of CF-IA and LA-ICP-MS helped to identify the effects of heavy metals on plants and provide information on tolerance mechanisms. It also facilitated the alleviation of heavy metal toxicity and photosynthetic tolerance in plants.

In addition to 2D imaging of single elements, 3D imaging and quantification of more than 19 micronutrients and toxic trace elements in wheat and rye grains could be obtained by LA-ICP-MS (Fig. 12) [100]. 3D imaging of the grains with high spatial resolution using the strategy of serial sectioning embedded in epoxy resin and LA-ICP-MS has recorded distribution patterns of several elements. The bulk concentrations of these elements were determined by ICP-MS. Longitudinal sections of the grains were imaged using  $\mu\text{-X-ray}$  fluorescence spectrometry ( $\mu\text{-XRT}$ ) to independently validate findings regarding the major element distribution. 3D imaging of the morphology of the grains was performed using X-ray computed tomography ( $\mu\text{-CT}$ ). Based on it, the transport mechanisms of toxic trace elements in the grains and the elemental enrichment in some organs were discussed.

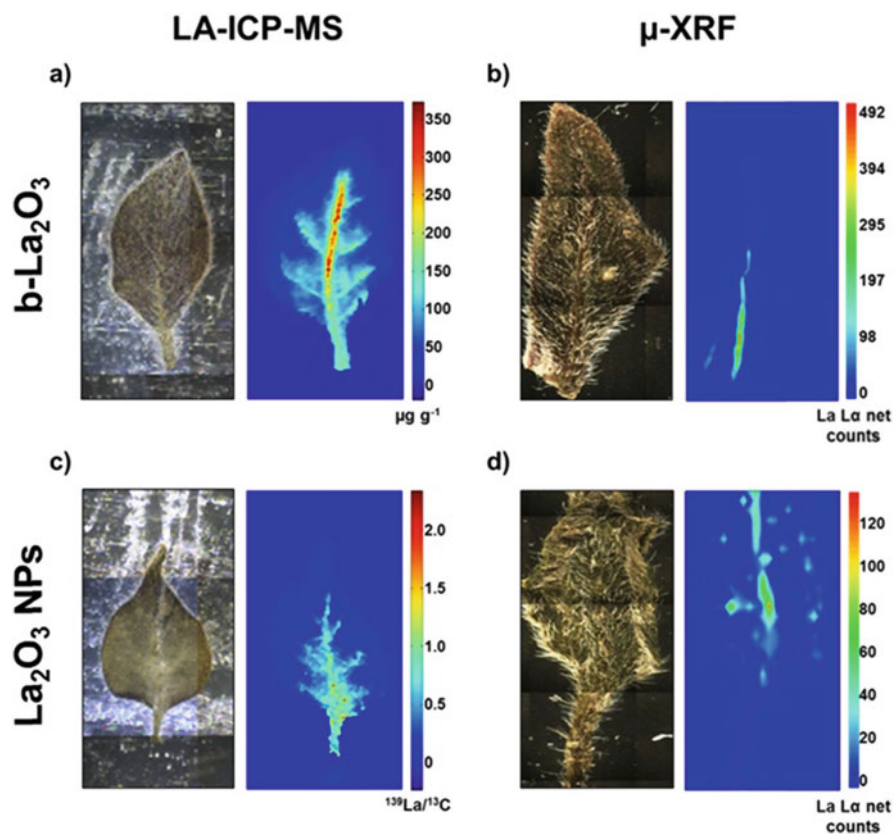
Besides the application in distribution of exposed metals, LA-ICP-MS exhibits good potential in the investigation of the absorption, transformation, and translocation of NPs in plants. For instance, other study of the same research team quantified and imaged the distribution of La in leaves and stems after plant absorption of b-La<sub>2</sub>O<sub>3</sub> and La<sub>2</sub>O<sub>3</sub> NPs by LA-ICP-MS. Since NPs can be observed as sharp peaks,



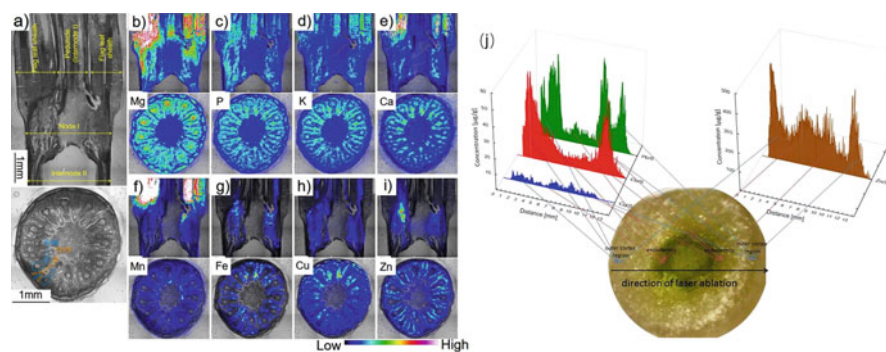
**Fig. 12** 3D element distributions of the grain from LA-ICP-MS. (a) Schematic illustration of the feature-based registration process, showing the transition from a set of 2D images towards a stack of aligned 2D images of the Cr distribution in *S. cereale*. (b) 3D distribution of selected micronutrients in *T. aestivum* L. and a micro-CT image of the grain Reproduced from *J. Anal. At. Spectrom.*, 2017, 32(2):289–298 with permission from the Royal Society of Chemistry

LA-ICP-MS could discern b-La<sub>2</sub>O<sub>3</sub> and La<sub>2</sub>O<sub>3</sub> NPs to some extent [101]. Similar distribution of La in the leaves was observed by  $\mu$ -XRF analysis, but the sensitivity of this technique is lower than that of LA-ICP-MS that showed better detection and visualization of La distribution in the leaf (Fig. 13). LA-ICP-MS has manifestly become a powerful tool to confirm the uptake, translocation mechanism, and distribution of NPs with minimal sample pretreatment.





**Fig. 13** Lanthanum distribution in leaves of *P. glomerata* cultivated in the presence of  $400 \text{ mg L}^{-1}$  of  $\text{b-La}_2\text{O}_3$  and  $400 \text{ mg L}^{-1}$  of  $\text{La}_2\text{O}_3$  NPs analyzed by: (a, c) LA-ICP-MS and (b, d)  $\mu$ -XRF. Reprinted (adapted) with permission from Environ. Sci. Technol., 2019, 53(18):10827–10834. Copyright {2019} American Chemical Society



**Fig. 14** Element distribution of plant roots by LA-ICP-MS. Reproduced from Plant Journal, 2019, 99(6):1254–1263 (Free Access)

Application of LA-ICP-MS was able to show the metals pathway while entering the pea roots [102]. In Fig. 14, linear ablation scanning was performed from the root surface to the center of plant tissues to obtain the quantitative distribution of Cd, Cu, Pb, and Zn after corresponding incubation. (e.g., cell wall, cell membrane, cytoplasm, vacuole and mitochondria). Analysis with EDX enabled imaging of accumulation of four elements in different plant organs, at the cellular and subcellular level, while confocal microscopy allowed us to indirectly detect the presence of metals through imaging of increased ROS generation. The research is to address oxidative stress and apoptosis-inducing mechanism of cells. As can be seen, both LA-ICP-MS and EDX are robust techniques for analyses of plant tissues in both quantification and imaging.

A method combining “soft ablation technique” with cryo-compound containing element standard followed by LA-ICP-MS, succeed in the quantitative and imaging measurement of eight elements in rice nodes (Fig. 14) [103]. Furthermore, differentially specialized vascular bundles around Si, Zn, and  $^{13}\text{C}$  element accumulation gotten the priority mechanism in the Zn distribution in OsHMA2 protein and the transport mechanism of Si transporters. LA-ICP-MS has become a cost-effectiveness tool in linking transporter localization to element distribution at a cellular level.

## 5.2 Environmental Samples

For the analysis of samples in natural environment, LA-ICP-MS has become a powerful method to study climate signals and their preservation, which may help us further study global issues such as global warming and rising sea levels.

Elemental imaging of some impurities, e.g., Na, Mg, Co, and Sr, in glaciers was investigated [104]. With the introduction of low dispersion ablation pool and fluid dynamics image optimization methods, the results obtained by LA-ICP-MS were much richer, with higher resolution and less image distortion than laser camera. The improved LA-ICP-MS method for ice cores offered not only high scan speeds for single line profiles (around  $1\text{ mm s}^{-1}$ ) but can also map the localization of impurities at high spatial resolution ( $35\text{ }\mu\text{m}$ ) and without imaging artifacts. This promises not only new insights into the impurity distribution in glacier ice but also lays the ground for an improved understanding of the LA-ICP-MS signal obtained from ice cores. As a result, LA-ICP-MS can deliver an important contribution for retrieving future paleo-environmental records from highly thinned ice layers.

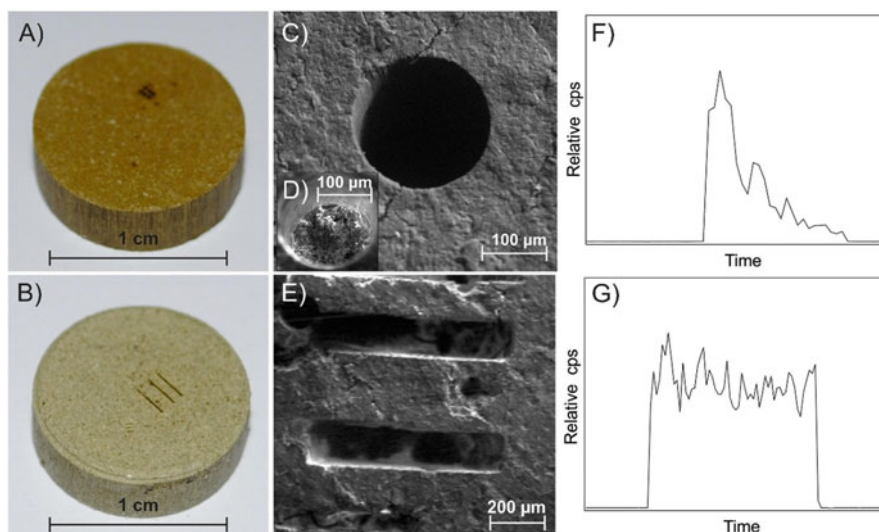
The distribution characteristics of rare earth elements (REEs) have important applications in diagnosis and mineralization, especially in recent years. High-resolution LA-ICP-MS mapping helped to reveal the distributions of major, transition, and REEs at sub- $10\text{ }\mu\text{m}$  scale in cross-sections of micronodules from the western tropical North Pacific seafloor, element distributions across the microlaminated rind, and the conditions of its genesis and metal enrichment [105]. The results reflected the existence of various minerals in micronodules and their diagenetic origin. LA-ICP-MS was combined with acid digestion-ICP-MS detection for the quantitative distribution of Ca, P, Zn, and Mn in 0–15 cm depth

soil layer under different fertilizer and planting methods [106]. The image distribution map in vertical direction was obtained, which is an important contribution to characterize and improve cropping systems, fertilizer and herbicide use, different soil types, and root depth versus root length.

Moreover, chlorine in concrete as building material was investigated by LA-ICP-MS. The imaging and quantitative results for common elements such as C, Al, Si, and Ca were obtained [107]. The effective chloride content determined by the Volhard titration overestimated the chloride content in the cement. Compared with that, LA-ICP-MS measurement delivered correct results and as the above described distinction criterion could be used to selectively quantify chloride in the cement phase. This led to correct results, disregarding the chloride concentration in the aggregates and if they are acid-soluble or not.

### 5.3 Food Samples Analysis

A protocol of grinding and tablet pressings method is imported to analyze the content of nine elements in common food-related Certified Reference Material (CRM) without acid digestion [108]. The photographs of pressed pellets, secondary electron images of laser craters in spot mode and raster mode, and representative time resolved analysis graphs of spot mode and raster mode are shown in Fig. 15.



**Fig. 15** (a, b) Photographs of pressed pellets, (c–e) Secondary electron images of laser craters in spot mode and raster mode, respectively, and (f, g) Representative time resolved analysis (TRA) graphs of spot mode and raster mode, respectively. Reproduced from Food Chemistry, 2019. 274: 726–732 with permissions from 2018 Elsevier Ltd

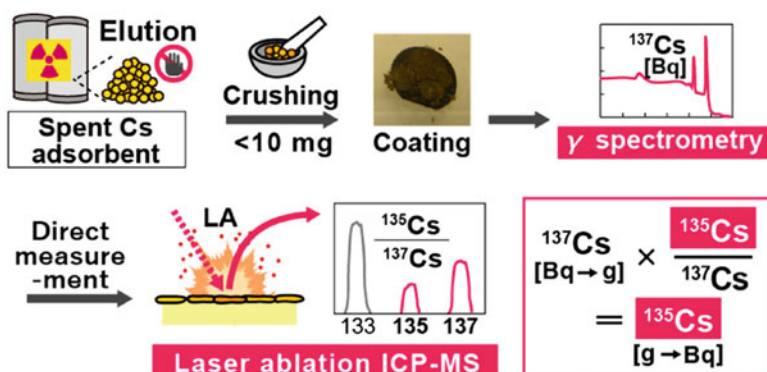
Homogenization of the sample and particle size reduction by milling ensured that there are not signal interferences providing more precise results. Raster mode was recommended over spot mode during ablation, as it provided greater accuracy for a wider variety of samples and elements. The work agrees with the recent hot idea of “green chemistry” and maintains extremely high sensitivity and excellent spatial resolution.

#### 5.4 Other Samples

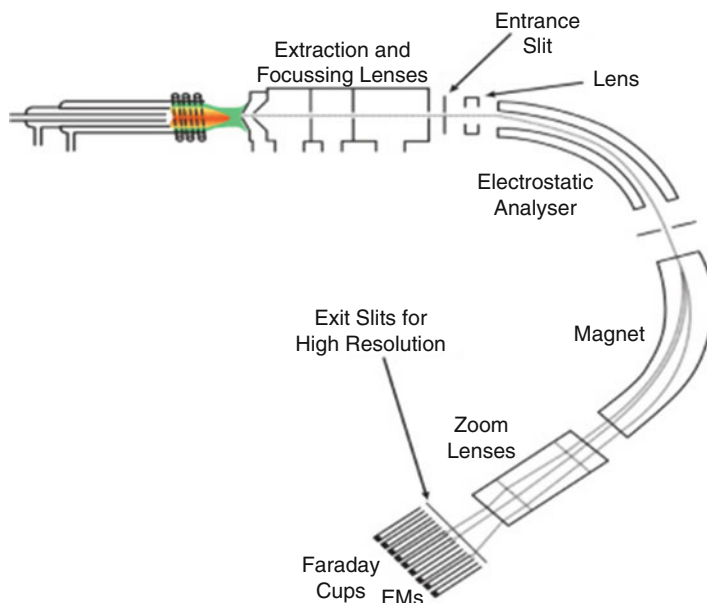
To perform LA-ICP-MS analysis, the sample should be in a form suitable for ablation. A variety of processing methods have been proposed. For example, quantitative analysis of 19 elements was carried out in waste polymer by grinding and laminating [109]. Homogenization of the waste polymers by cryomilling of larger amounts of material and preparation of pressed polymer powder pellets improved the accuracy of measurements and reduced the relative standard deviations.

LA-ICP-MS was used to investigate  $\text{Cs}^{2+}$  in the wastewater of nuclear power station with high radiation and high risk [110]. The ratio of Cs to its isotope in Fukushima nuclear power plant was determined (Fig. 16). This work demonstrates the applicability of LA-ICP-MS for  $^{135}\text{Cs}$  quantitation in a solid sample.

In addition to solving intrinsic problems in environmental and food science, LA-ICP-MS can also serve as a novel crossover technology combining with the booming development of chemometrics [111]. In order to identify counterfeit perfumes, the authors did not start from the perfume itself. They used LA-ICP-MS to quantify 63 elements in a variety of perfume bottles followed by element filtering. The



**Fig. 16** Cs adsorbent pretreatment process followed by LA-ICP-MS analysis. Reprinted (adapted) with permission from *Anal Chem.*, 2020, 92(4):3276–3284. Copyright {2020} American Chemical Society



**Fig. 17** Schematic diagram of a basic multi-collector-ICP-MS with forward Nier-Johnson geometry. Reproduced from *Geosystems and Geoenvironment*, 2022. 1(2):100019 (Open Access)

method of combining principal component discerns fake perfume and traces the origin of the perfume, solving common problems in an innovative way.

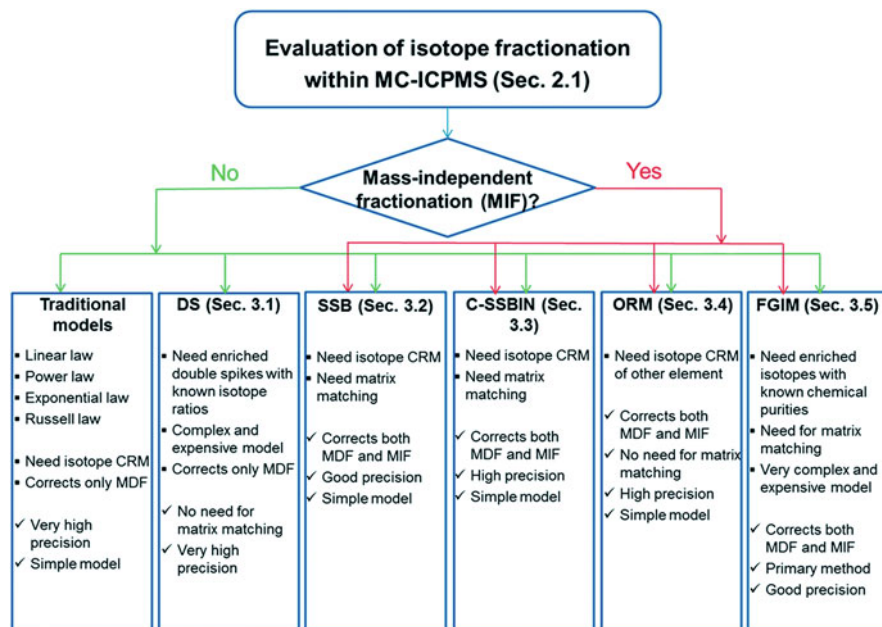
## 6 Multiple Collector (MC)-ICP-MS

MC-ICP-MS, first introduced by Walder et al. in 1993 [112–114], is mainly used for the determination of isotope ratios. It consists of ICP source, magnetic sector analyzer, and multiple sets of Faraday cups (Fig. 17) [115]. The sector-shaped magnetic field mass spectrometer involves an electromagnet as the mass analyzer and a magnetic field to separate ions of different mass-to-charge ratios ( $m/z$ ). The MC part refers to Faraday cups working as receiver, which performs qualitative and quantitative analysis based on the molecular ion peaks of the elements. Sector magnetic field mass spectrometry is the only mass spectrometer which matches multi-receiver detectors. It can achieve static and simultaneous multi-receiving, ensuring high-precision isotope ratio analysis. MC-ICP-MS provides new opportunities for the precise measurement of isotope amount ratios with instrumental precision between 0.1 and  $<0.005\%$ .

## 6.1 Interference Correction for MC-ICP-MS

### 6.1.1 Isotopic Fractionation/Mass Bias Correction

MC-ICP-MS has the advantages of high spatial resolution, high sample throughput, minimal sample preparation, and low contamination. While it suffers from larger isotopic fractionation/mass bias compared to traditional thermal ionization mass spectrometry (TIMS). The isotopic fractionation/mass bias of the instrument is usually attributed to the supersonic diffusion of the ion beam after passing through the sampling cone and the space charge effects after passing through the skimming cone in ICP. To obtain accurate isotope ratios, a variety of conventional isotopic fractionation correction methods (e.g., the power, the exponential and the Russell law) developed originally for TIMS have also been used in MC-ICP-MS. Russell law is the most widely used, while large errors have been reported in relevant research works [116]. It should be noted that traditional isotopic fractionation correction methods were designed to account for mass-dependent fractionation (MDF) as a function of nuclide mass [117]. So far, mass-independent fractionation (MIF) has been recognized to be a genuine and more common phenomenon within MC-ICP-MS. The use of mass-dependent models to correct the MDF-based bias would result in biased isotope amount ratios. Yang et al. [117] proposed a chart (Fig. 18) to inform the selection of isotopic fractionation models for accurate isotope amount ratio results by MC-ICP-MS.



**Fig. 18** Flowchart for an appropriate selection of mass bias correction models for the accurate and precise isotope amount ratio measurements by MC-ICP-MS. Reproduced from *J. Anal. At. Spectrom.*, 2018, 33(11):1849–1861 with permission from the Royal Society of Chemistry

### 6.1.2 Matrix Effect

The presence of large amount of complex matrix in real sample will affect the isotope analysis significantly. In the presence of a high concentration of the matrix, the transmission of ions in the mass spectrometer varies slightly, but the variation is different for different isotopes of the same element. It will result in biased or inaccurate isotopic ratios in MC-ICP-MS measurement. To eliminate or correct matrix effects, traditional solution is the separation of the element from the matrix and the use of a certified isotopic standard of the element for external mass bias correction. Griffiths et al. [118] separated target Pb from seawater by using solid phase extraction with Nobias Chelate PA-1 resin or coprecipitation with  $Mg(OH)_2$ . As an alternative, internal mass bias correction, using another element close in mass, can be applied [119]. For example, silicon internal standardization could correct matrix effects caused by sodium, chlorine, and calcium in waters for sulfur measurements [120]. The natural silicon content of silicon isotopes with masses 29 and 30 in beer was used for internal mass bias correction. The matrix effect was fully corrected by the internal standardization of silicon to the effect on differential ion transmission [119].

## 6.2 Application of MC-ICP-MS in Food Science

In recent years, geographical origin and authenticity of food are topics of interest for both consumers and producers. Isotope analysis is one of the important methods for food traceability. It is generally believed that the isotopic composition of a certain element will not change measurably during the process of entering soil from bedrock, entering plant through water, and finally entering organisms along the food chain. Based on it, isotope analysis provides a valuable tool for tracing the origin of commodities, agricultural products, food and even animals.

### 6.2.1 Geographical Traceability Purposes

Sr is a stable isotope and a mature food tracer. The ratio of  $^{87}Sr/^{86}Sr$  can be accurately determined by MC-ICP-MS. The ratio of  $^{87}Sr/^{86}Sr$  in soil, water, and food can provide reference value for the geographical origin and authenticity of food such as Lambrusco wines, green coffee beans, and Szegedi paprika [121–124]. The combination of multi-element and Sr isotope fingerprints can also be used, with the help of MC-ICP-MS measurement and the use of multivariate statistical data to ensure the accuracy of the results [122, 123]. It is proposed that the combination of statistical and mathematical tools provides a basis for determining the origin of food using special elements and strontium isotope fingerprints.

### 6.2.2 Sources of Pollution in the Food

The isotope ratio in the natural environment is relatively stable. Lead isotopic ratio obtained by MC-ICP-MS is helpful to exploring the source of contamination in food. Ndung'u et al. analyzed 58 brands of balsamic vinegar samples to obtain lead concentration and isotopic composition ( $^{204}\text{Pb}$ ,  $^{206}\text{Pb}$ ,  $^{207}\text{Pb}$ , and  $^{208}\text{Pb}$ ). The lead isotopic fingerprints of all high-lead content vinegars were found to be mainly anthropogenic. It suggested multiple contamination sources including human factors. The results for organically cultivated grape vinegar indicated that the lead content was mainly attributed to industrial processes carried out post-harvest rather than to the pesticides used during the grapes growth period [125]. Wines from three vineyard locations with different lithology and pollution levels exhibited similar lead isotopic signatures ( $^{206}\text{Pb}/^{207}\text{Pb} = 1.16\text{--}1.18$ ), although those of the soils where they grew differed significantly ( $^{206}\text{Pb}/^{207}\text{Pb} = 1.147\text{--}1.22$ ). This fingerprint corresponds to the effect of industrial and urban airborne particulate material both on the soil and on the wine in the studied areas [126].

### 6.2.3 Isotope Composition in Blood of Different Diet Groups

Isotopic observations in animal samples indicate body accumulation of heavy zinc and light copper throughout life. This hypothesis has never been tested for humans, but the existence of a relationship between blood isotopic composition and age could be promising for age assessment methodologies [127]. Dietary habits can also influence the blood zinc isotope composition, being an additional source of isotopic variation. Significant differences in the isotopic composition of Zn in blood from populations with different feeding habits have been reported and the isotopic composition of Zn has been suggested as a potential dietary proxy [128]. Zinc in products of animal origin, such as meat, fish, eggs, and semi-skimmed milk, is more likely to be enriched in lighter zinc isotopes than in human blood samples. On the other hand, vegetables, grains, dairy products, pasta, and chocolate contain heavier zinc isotopes [129]. The lighter zinc isotopic composition in omnivorous people may be related to the lighter zinc isotopic composition in animal-derived foods. In contrast, the large consumption of grain products and vegetables seems to be related to the relative enrichment of heavy zinc isotopes in the vegetarian crowd. Therefore, diet seems to be the main source of changes in the zinc isotopic composition of human blood.

## 6.3 Application of MC-ICP-MS in Nanoparticle Analysis

NPs are widely used in industrial, medical, and biological fields due to their unique physicochemical properties. The NPs present in nature would be mixture of NPs of



various chemical compositions, as reflecting the sources or production process, and thus, rapid and sensitive analytical technique for elemental/isotopic ratio measurements in individual NPs was increasingly desired. MC-ICP-MS is an alternative technique for element/isotopic ratio analysis of a single NP. With MC-ICP-MS, multiple isotopes can be monitored without large-scale scanning. This leads to the time efficiency of data collection close to 100%, avoiding the loss of counts for analysis [130]. In order to obtain reliable isotope ratio data from the transient signals from NPs, a multi-collector setup and a fast response (i.e., high temporal resolution) signal monitoring system are needed. A MC-ICP-MS equipped with a high time resolution data acquisition system (HTR-MC-ICP-MS) was used to measure multiple isotope signals from the NPs. Therefore, accurate signal strength data can be obtained from transient signals derived from NPs, and the size and isotope ratio analysis of a single NP can be performed.

#### ***6.4 Studies on Isotopes in Human Metabolism***

Earlier studies suggested that natural iron isotope ratios in human blood might be a useful tool to assess individual long-term iron absorption efficiency. Compared with the iron in food, human blood is rich in light iron isotopes [131], and the absorption of heavy iron isotopes is lower than the absorption of light iron isotopes when dietary iron is taken in the gastrointestinal tract. MC-ICP-MS was used to screen the tissue samples related to iron metabolism for iron isotope. A lot of isotope fractions were found in the process of intestinal absorption and distributed among body tissues. Compared with feed, the gastrointestinal mucosa is rich in light iron isotopes, and the isotope pattern reflects the main iron-rich sites of intestinal mammals. Studies have shown that when heavy iron isotopes are enriched in the liver, spleen, and bone marrow, light iron isotopes are enriched in red blood cells. Gulson et al. [132] illustrated the use of lead isotopes in a wide spectrum of activities including sources and pathways of lead in diverse environments from urban to mining communities, various applications associated with pregnancy, the contribution of bone lead to blood lead in the elderly, the half-life of lead in blood, and lead in bones and other media.

#### ***6.5 Application of MC-ICP-MS in Environmental Fields***

##### **6.5.1 Dating of Rocks and Minerals**

With the help of MC-ICP-MS, the use of isotopes, especially Pb isotopes, has a wide range of applications in geological research on the dating of the earth's lithospheric rocks and minerals [133, 134]. For example, mean mineral ages with the  $^{204}\text{Pb}$  ion beam measured in the multiplier were as follows: apatite,  $1,715 \pm 23$  m.y.; sphene,

1,789  $\pm$  11 m.y.; and monazite, 1,783 to 1,888 m.y., with relative uncertainties on individual monazite ages of <0.2% but highly reproducible age determinations on single monazite crystals ( $\ll$ 1%).

### 6.5.2 Isotopes as a Tracer in Earth Science

The development of MC-ICP-MS technology enables rapid and stable ionization of samples in high-temperature plasma ion sources, and improves the accuracy of isotope analysis, making Nd isotope widely used as an indispensable tracer in earth science and cosmic chemistry. It can be used to investigate current and past ocean circulation [135]. The determination of Pb isotopes by MC-ICP-MS has also been used to study various ocean processes, such as surface ocean currents, meso-scale vortices, and the time scale and path of ventilation in deep water bodies. For the study of Hg isotopes, detailed isotope fractionation information can be provided to understand the current or ancient environmental cycle of Hg in the global or local/regional, ecosystem or biota [136].

### 6.5.3 Other Applications

In recent years, metal stable isotopes have been critically assessed as a tool to obtain qualitative and quantitative integrated information about sources, pathways, and biogeochemical processes of these elements in the environment. There is a significant difference in the isotope ratio of Zn between anthropogenic emissions and natural rocks or soils, which can be used to determine the source of contamination. Zinc isotopes will be fractionated during major man-made processes such as smelting, resulting in artificial fingerprint isotopes that are different from natural isotopic compositions [137]. The U isotope in the surrounding soil samples after the leakage of the Fukushima nuclear power plant was analyzed. The  $^{235}\text{U}/^{238}\text{U}$  isotope ratio can be used as an indicator of U source in environmental research to assess the radioactive element pollution caused by nuclear leakage [138].

## 7 Future Perspectives and Conclusions

ICP-MS is the most sensitive detection technique for trace elements presently and has been widely used for the quantification of interest elements in food and environmental samples. Due to its inherent characteristics, ICP-MS itself cannot afford information on elemental speciation, distribution, and/or imaging, and cannot meet the requirements in real sample analysis sometimes. So far, ICP-MS involving hyphenated techniques has been developed rapidly to satisfy the needs at different levels.

A variety of sample pretreatment techniques, mainly solid phase (micro)-extraction and liquid phase (micro)extraction, were combined with ICP-MS detection, separating the co-existing interference substances, enriching target element/species, favoring the improvement of the analytical sensitivity and anti-interference ability. Some of the pretreatment can further distinguish different elemental species, helping to achieve speciation analysis. In the following study, simultaneous analysis of species for different elements is expected, and transformation of the species during pretreatment process should be further evaluated and avoided.

Chromatographic separation techniques have been demonstrated with good potential for separating various elemental species, followed by ICP-MS detection. With the aid of ESI-MS, unknown species can be identified along. Besides HPLC, GC, CE, and GE, flow field techniques have been applied for a combination with ICP-MS. These hyphenated techniques provide plenty information on trace elemental species, metal-containing macromolecules, colloids, nanoparticles, and particles.

LA-ICP-MS exhibits good potential in elemental imaging and direct quantification of elements in solid food and environmental samples. It helps to explore the elemental distribution, transportation, pathway, and location in plant and earth.

MC-ICP-MS affords isotope information, which is beneficial to the geological traceability purposes, exploration of the pollution source for the foods, investigation on human metabolism, dating of rocks and minerals, and tracing specific purposes.

Above all, ICP-MS and its variants exhibit significant function in food security, environmental pollution monitoring, and health risk assessment. Relevant methodologies need further improvement to meet the requirements in real sample analysis in terms of anti-interference ability, detection sensitivity, selectivity, sample throughput, and easy operation.

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# Isotopic Mass Spectrometry in Food and Environmental Chemistry



Yolanda Picó and Damià Barceló

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**Abstract** Isotope ratio mass spectrometry (IRMS) measures small differences in the abundances of light-stable isotopes of elements, mostly carbon, hydrogen, oxygen, sulfur, and nitrogen. The ratio of isotopes is dependent on the environment where the samples are located and therefore the measurement of the isotope signature in food and environmental samples can give information to help to distinguish samples that share the same chemical composition but can be under different conditions. This chapter provides an overview of isotope ratio mass spectrometry (IRMS) in the context of food and environmental sample analysis (e.g., food authenticity, food origin, or environmental contamination). The fundamentals and latest developments of the technique, most important aspects of the instrumentation, analysis of isotope ratio information as well as key applications are described to illustrate the impact of

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this rapidly growing field of research especially in terms of the number and type of applications. Finally, applications in nutrition and trophic, environment, soil science, and food authenticity are assessed by giving details on the progresses, advantages, and pitfalls.

**Keywords** Elemental analysis, Gas chromatography, Isotope ratio mass spectrometry (IRMS), Isotopic abundance, Liquid chromatography, Magnetic sector mass analyzer

## 1 Introduction

Isotopes are defined “as atoms with equal number of protons but different number of neutrons” [1]. Thus, the number of protons defines the nature of the chemical element (H, C, O, N, etc.), the number of neutrons the isotope of that element, and the sum of protons and neutrons the atomic mass [2]. Most elements are mixtures of isotopes. These, in turn, can be stable or radioactive. The latter undergo radioactive decay at very different rates, from a fraction of a second to tens of billions of years [3].

Stable isotopes can be produced but are never transformed by radioactive decay. They are only transformed by nuclear synthesis reactions. Both heavy and light-stable isotopes take part in chemical, physical, and biochemical reactions but at different rates. Lighter isotopes react faster than heavier isotopes affecting the relative abundance of isotopes in an established medium or compartment depending on the characteristics of the process [2].

Stable isotope composition of the natural materials is used as indicators of physical or chemical (including biological, aging, transformation, etc.) processes and geochemical cycles [4]. They also act as conservative tracers to identify, for example, food sources, nutrient cycling, environmental conditions, etc. [5, 6]. Typically, light elements (traditional stable isotope systems) are used for the analysis: H, C, N, O, or S, but also non-traditional isotopes (Li, B, Mg, Ca, Fe, Mo, etc.) can be useful in some applications.

The natural fluctuations of isotope ratios are minor. These isotope ratios are therefore always reported relative to an isotopic standard to eliminate bias or systematic error in the determinations [7]. Isotopic ratios of samples are reported relative to universal standards and are expressed using the notation delta ( $\delta$ ) with the unit of parts per thousand (‰).

$$\delta^M X = \left( \frac{R_{\text{sample}}}{R_{\text{standard}}} - 1 \right) \times 1,000$$

where  $R_{\text{sample}}$  is the ratio of the atomic mass of the heavier (commonly the minor) and lighter (commonly the major) isotope for an element in the sample.  $R_{\text{standard}}$

**Table 1** Initial primary reference materials defining the isotope scales according to IAEA [9]

Element ratio	Primary reference standard	Isotope ratio, <i>R</i> Hayes [10] Accepted value $\times 10^6$
Hydrogen $^2\text{H}/^1\text{H}$	SMOW (Standard Mean Ocean Water)	$155.76 \pm 0.10$
Carbon $^{13}\text{C}/^{12}\text{C}$	PDB (Pee Dee Belemnite)	$11,237.2 \pm 2.9$
Nitrogen $^{15}\text{N}/^{14}\text{N}$	Air nitrogen	$3,676.5 \pm 8.1$
Oxygen $^{18}\text{O}/^{16}\text{O}$	SMOW	$2,005.2 \pm 0.43$
Sulfur $^{34}\text{S}/^{32}\text{S}$	(Canyon-Diablo-Triolit)	$45,004.5 \pm 9.3$

is the same ratio but measured in the standard samples with accurately known abundance ratios. The initial standards used for comparison are summarized in Table 1 but many of them are no longer available as their supply has been exhausted. The delta value of a universal standard is zero. For example, carbon isotope ratios were measured against the Pee Dee Belemnite (PDB) standard with  $\delta^{13}\text{C}$  value 0‰ [8]. Nowadays, new primary reference standard materials (including all the NBS isotope standards) are available from the International Atomic Energy Agency (IAEA) in Vienna, the National Institute of Standards and Technology, and the US Geological Survey (USGS) [9]. Furthermore, and for practical purposes, other secondary reference materials have been calibrated during interlaboratory calibrations headed by the Commission on Isotopic Abundances and Atomic Weights (CIAAW) of the International Union of Pure and Applied Chemistry (IUPAC) [10].

The isotopic composition of the samples studied can be enriched or depleted in the heavy isotope with respect to the standard. This means that they can have both, positive or negative delta values. The  $\delta^{13}\text{C}$  value of 10‰ means that material is enriched in the heavy isotope ( $^{13}\text{C}$ ) and value of  $-10\%$  means that material is depleted in the heavy isotope with respect to the standard.

Isotope ratio mass spectrometry (IRMS) is one of the relevant techniques used to measure the ratio of stable isotopes. This technique is applied in fields such as archeology, geology, medicine, biology, and forensic science but also in environmental and food sciences, to investigate human and animal diets in the past, for food authentication, to reconstruct past environmental and climatic conditions, etc. IRMS enables to measure ratios of isotope abundances of an element in a sample with very high accuracy. However, this way of using the technique, measuring the isotopic abundance of the whole sample, which is known as bulk-specific isotopic analysis (BSIA), is not responsible for its increasing use. Considering that complex samples have a very large number of compounds (several thousand), this way of measuring isotopic abundance can hide important information about the distribution of isotopes in the different components. Alternatively, compound-specific isotope analysis (CSIA) refers to the measurement of isotope ratios of individual organic compounds extracted from complex mixtures [11, 12]. CSIA commonly needs preparative steps

(pretreatments, extraction, and derivatization) to get amenable chromatographic analytes from bulk samples. These individual organic compounds can be contaminants or natural sample components, such as proteins, lipids, organic matter, etc. CSIA that requires a previous separation of the compounds to determine their specific isotopic distribution is responsible for the multiplication in the last 30 years of the number of applications of this technique.

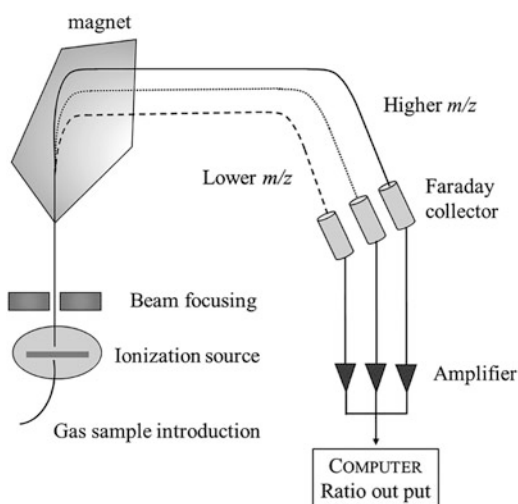
This chapter outlines the methodological development and application of IRMS while also highlighting the main issues of the technique. In addition, the latest studies on environmental issues as well as on the traceability of the origin and/or authenticity of food products are presented.

## 2 Stable Isotope Establishment Using Isotope Ratio Mass Spectrometry (IRMS)

The most common technique for stable isotope ratio analysis is IRMS. Figure 1 shows a scheme of the instrument. An IRMS instrument consists of (1) sample introduction system (injector), (2) analytes ionization, (3) mass analyzer, (4) detector array, and (5) data acquisition system. The instrumental configuration has been widely described in many of the books, monographies, and references [3, 8, 13].

Prior to IRMS determination, the sample components must be transformed into the simpler gases  $\text{CO}_2$ ,  $\text{N}_2$ ,  $\text{H}_2$ , and  $\text{SO}_2$ , but also others such as  $\text{O}_2$ ,  $\text{N}_2\text{O}$ ,  $\text{CO}$ ,  $\text{CH}_3\text{Cl}$ ,  $\text{SF}_6$ ,  $\text{CF}_4$ , and  $\text{SiF}_4$  are produced. Most common systems used for sample introduction included elemental analyzer isotope ratio mass spectrometer (EA-IRMS), gas chromatography-isotope ratio mass spectrometry (GC-IRMS), and liquid chromatography-isotope ratio mass spectrometry (LC-IRMS). This offers

**Fig. 1** Basic scheme of IRMS



a vast instrumental array that has been widely exploited in both, environmental and food analysis. The gases are introduced in the mass spectrometer through the inlet system (sample introduction). This inlet system consists of pneumatic valves and could be dual or continuous flow systems. The dual-system involves the alternate switch of the valves to introduce sample and reference standard in the vacuum chamber and the continuous system considers the injected sample as a transient signal and the reference standard as a gas pulse. Globally, continuous flow injection is easier, quicker, and more economical and has specific advantages such as less sample preparation and sample size. As a counterpart, dual inlet injection is more precise and reliable, especially when samples are prepared off-line [8, 13].

The next step is analyte ionization. Most used ionization sources for IRMS are electron impact (EI), thermal ionization mass spectrometry (TIMS), and multi-collector inductively coupled plasma mass spectrometry (ICP-MS). These three ion sources attain determination of almost all the isotopes in the periodical table. Electron ionization (EI) – also electron impact ionization or electron bombardment ionization – was one of the first ionization techniques developed for mass spectrometry and is still widely used. This technique ionizes the gases by a stream of energetic electrons produced by a filament (W, Rh) and accelerated by a difference of potential that interact with atoms or molecules to produce ions. EI is a strong ionization method that produces a large fragmentation of molecules due to the collision with high energy electrons. Thermal ionization mass spectrometry (TIMS) is another method available for high-precision isotopic analysis, in which elements are thermally ionized. The elements are deposited on a metal filament or filaments through which a current is passed at increasing temperature ( $>1,000^{\circ}\text{C}$ ). The ions created in the filament are accelerated through an electric potential gradient and concentrated into a beam through a series of electrostatically charged slits and plates. Multi-collector ICP-MS (MC-ICP-MS) supplements TIMS with the capacity to determine any type of element that requires higher ionization energies and needs important sample throughput. MC-ICP-MS consists of a robust ICP ion source where the plasma (that provided the ionization) is generated under atmospheric pressure, thus allowing sample introduction by means of continuous nebulization. This ionization source is widely described also in the previous chapter (3. “Elemental Mass Spectrometry”) where many applications are also highlighted. In addition, the combination of laser ablation (LA) —a solid sampling technique that applies a high intensity light to the sample through pulses and is able to transform a solid sample directly into an aerosol—with an MC-ICP-MS has enlarge the possibility to process different types of samples. EI and TIMS present limitations regarding the precise determination of the isotopic composition of elements with high ionization potentials that can hardly be ionized. Instead, MC-ICP-MS achieves determination of almost all compounds of the periodical table [14]. Furthermore, the combination LA-MC-ICP-MS has many advantages mapping compounds [15].

The measurement of the isotope ratios is carried out almost exclusively with magnetic sector type instruments. In a magnetic sector, the mass analyzer uses a magnetic field to separate ions of different  $m/z$  values. Once the sample is ionized, the ions are accelerated toward the magnetic sector by applying a high voltage to



them. The ions are generated at the source and travel continuously to the magnetic sector, where the ions are exposed to the magnetic field. The ions are deflected according to Fleming's left-hand rule. The intensity of the deflection depends on the  $m/z$  of the ions, so that lighter ions (of the same charge) will experience more deflection and heavier ions less. The magnetic sectors used in IRMS applied multiple detectors (multi-collector deep Faraday cups) to simultaneously monitor and integrate the ion currents of interest. This type of instruments is also named "Nier-type" because Alfred Nier was one of the first designers of the current-type of marketed instruments [5]. The different sample introduction systems are very important because these systems condition the information obtained. This is crucial especially in the case of complex samples. The three most important sample introduction systems are EA-IRMS, GC-IRMS, and LC-IRMS [14].

## ***2.1 Elemental Analyzer Isotope Ratio Mass Spectrometer (EA-IRMS)***

The combination of elemental analyzer (EA) with IRMS provides the mean isotopic signal of the bulk sample. In this system, the solid or liquid sample (placed in a tin or silver capsule) is burnt into a combustion or pyrolysis furnace (the AE) at high temperature [13]. This process can take place by combustion under a flow of oxygen (for analysis of C and N) producing different gases ( $\text{NO}_x$ ,  $\text{CO}_2$ ,  $\text{SO}_2$ , or  $\text{H}_2\text{O}$ ) or by thermal conversion at  $1,400^\circ\text{C}$  [high-temperature conversion (HTC) in oxygen-free environment (for analysis of H and O) [16]. One pending issue within these systems is the lack of accurate hydrogen isotopic analysis of halogen- and sulfur-bearing organics with traditional HTC because the formation of hydrogen-bearing reaction products other than molecular hydrogen ( $\text{H}_2$ ) is responsible for non-quantitative  $\text{H}_2$  yields and possible hydrogen isotopic fractionation. Recently, new chromium-based Elemental Analyzer-Chromium/High-Temperature Conversion Isotope Ratio Mass Spectrometry (EA-Cr/HTC-IRMS) has showed very promising results to overcome these drawbacks [17]. However, some technical and analytical issues concerning halogen- and sulfur-bearing samples remained unsolved and needed further improvement of the systems.

Other problem that affects globally EA-IRMS is the presence of interferents. Numerous gases are formed which can also react with each other and their own reaction products can interfere with the determination of the isotopic abundance of a particular element. Therefore, after combustion, a series of purification steps are required, depending on the analyte to be determined. The most common takes place in the combustion chamber and is the oxidation of the formed gases. The gases formed then pass through a He stream to the reduction chamber where the  $\text{NO}_x$  is reduced to  $\text{N}_2$  and the  $\text{O}_2$  is retained. The system then has a trap to retain the water (as  $\text{CO}_2 + \text{H}_2\text{O}$  can form  $\text{HCO}_2^+$  which is an isobaric interference of  $^{13}\text{CO}_2$ ), then there is a gas chromatograph which separates  $\text{CO}_2$  and  $\text{N}_2$  prior to their introduction

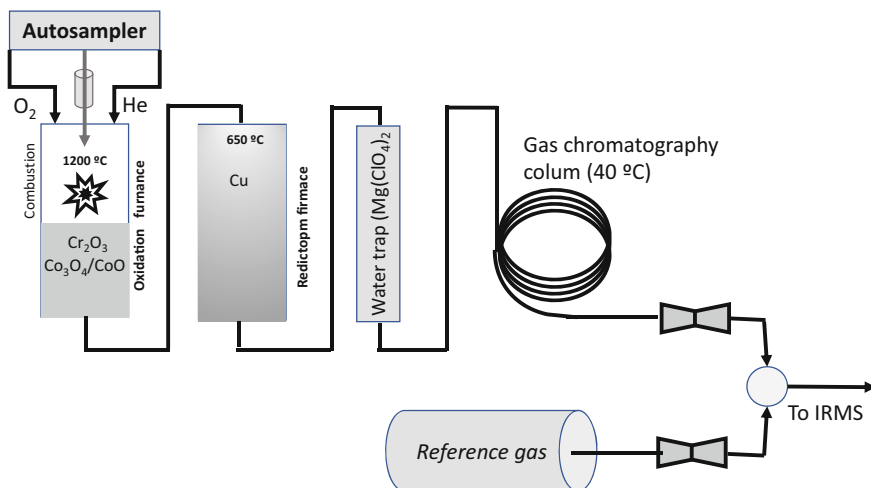


Fig. 2 Basic scheme of the EA sample treatment and purification

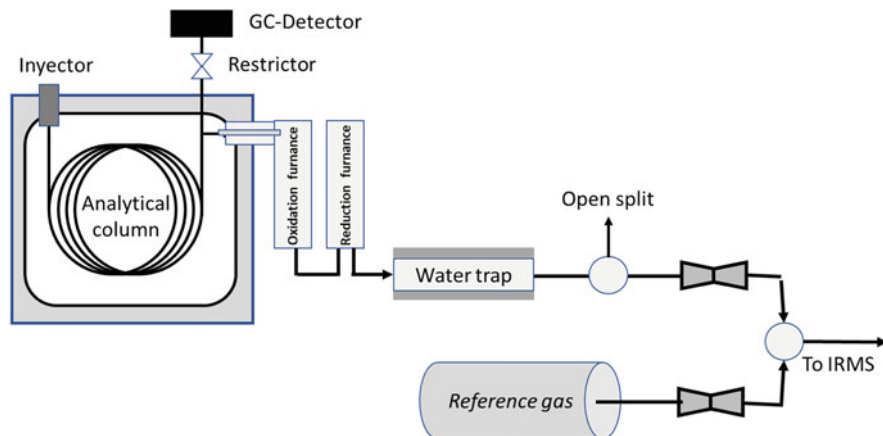
into the IRMS [16]. A basic scheme of this sample introduction that illustrated the working flow and purification system is shown in Fig. 2.

The isotope ratios of the gases from the samples are calculated with respect to a standard reference gas, which to provide optimal results must have a similar signal intensity to that of the samples and be analyzed alternatively to the samples. The flow of each reference gas is regulated by an interface system that rapidly alternates between the reference gas and the sample to ensure closeness in time [13].

## 2.2 Gas Chromatography-Combustion-Isotope Ratio Mass Spectrometry (GC-C-IRMS)

This system enables separation and isolation of specific compounds and then determination of their particular isotopic abundance (certain fatty acids, sterols, proteins, volatile compounds, etc.) [18]. The compounds to be separated must be GC amenable. This means volatile and thermally stable. Very polar compounds (initially non-volatile) can also be separated by GC-C-IRMS after chemical derivatization to transform them in volatile derivatives. The instrument (schematized in Fig. 3) contains three main parts:

1. *Gas chromatograph (GC)* separates the organic compounds of interest using a gas (usually He) as mobile phase and an inert column as stationary phase in an oven at controlled temperature according to their volatility. The compounds are desorbed and volatilized in the analytical column at different times and can be separately analyzed in a combustion chamber. The sample is introduced in the GC using the conventional split/splitless or on-line injector [13, 19]. However,



**Fig. 3** Basic scheme of the GC sample separation and C-IRMS

modifications and latest advances in the hyphenation of several techniques to GC that improve the capabilities of the system can be applied to these instruments. In this sense, pre-concentration techniques, such as solid-phase microextraction (SPME), prior to the GC have also been reported, particularly for volatile compounds, adding versatility to the methods because these techniques increased several hundred times sensitivity of the determination [20–22]. Using this technique, analytes are adsorbed in a fiber that is thermally desorbed in the GC oven. Furthermore, pyrolysis gas chromatography (Py-GC) has also been incorporated as previous separation step for macromolecules (organic matter, lignins, polysaccharides, plastics, etc.) to get the monomers of the compounds and determine their isotopic signature [23].

In most of these systems, a small part of the eluent is diverted to a conventional GC detector such as flame ionization detector (FID) or MS whereas the main part goes to the next part, the combustion chamber [13].

2. In *C* (combustion/conversion) the analytes after elution of the GC pass to the combustion or high-temperature conversion (HTC) furnaces where they are transform in gases. As reported for EA-IRMS the proper determination of isotopic abundances involved an oxidation or HTC step, a reduction step, and the removal of water.
3. Finally, the gases are sent to the IRMS, where they are ionized in an electron ionization (EI) source, separated in the magnetic sector, and then detected with Faraday cups. The different isotopic combinations detected allow to calculate the isotopic ratios of specific elements in the compounds of interest.

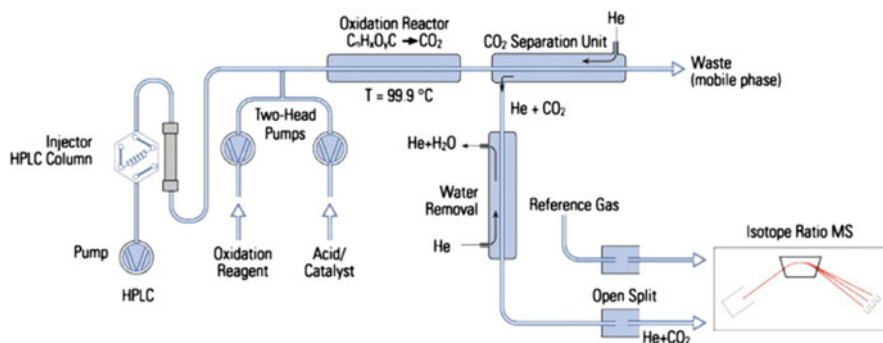
GC-C-IRMS has many fields of use including environmental and food studies. Examples of applications are reported for the analysis of explosives [24], organophosphorus pesticides [25], brominated flame retardants (BFRs) [26],  $\text{NO}_3^-$  and  $\text{NO}_2^-$  [27], lignin monomers [28], vanillin [21, 29, 30], acetic acid, acetoin, and

other organic acids [22, 31], bis(methyl-thio)methane [32], different types of sugars [33–36], alcohol [37], and so on.

### 2.3 Liquid Chromatography-Isotope Ratio Mass Spectrometry (LC-IRMS)

The advantage of LC-IRMS over the GC-C-IRMS is that it is able to separate non-volatile compounds without the need of derivatization. LC separates organic-specific components of the sample mostly according to their polarity. This is a complementary technique to GC. This interface is much more recent than the combination GC-C-IRMS [38]. Up to the moment, LC-IRMS applications are typically dedicated to carbon isotope ratio analyses. Nowadays there are two interfaces able to connect and HPLC to IRMS (1) a moving wire interface (still into the prototype state) and (2) a wet-chemical oxidation interface already in use by Thermo (LC IsoLink) and Elementar Analysensysteme GmbH (LiquiFace) with commercial instruments [39]. The scheme of this latter interface is shown in Fig. 4.

The advantage of wet-chemical oxidation is that it converts the individual compounds eluting from the column into  $\text{CO}_2$  directly in the mobile phase by a post-column stream of an oxidant (ammonium peroxodisulfate) and a catalyst (silver nitrate). The disadvantage is that the HPLC mobile phase (mostly water) cannot contain organic solvents or redox modifiers because they interfere with the results [39]. This is not a minor disadvantage but an important limitation of the technique considering the role that organic solvents and modifiers play in LC separation improving peak shape and separation power. Once the reagents for the wet oxidation are added to the mobile phase, it passes through a capillary oxidation reactor where the organic compounds are oxidized to  $\text{CO}_2$  at a temperature of about  $100^\circ\text{C}$ . A



**Fig. 4** Principles of the LC-IRMS interface according to [40]. Compounds separated by LC are converted to  $\text{CO}_2$  by wet-chemical oxidation. In the separation unit, this  $\text{CO}_2$  is transferred from the liquid eluent into a helium flow. Before the  $\text{CO}_2$  flow is carried into the IRMS the water vapor is removed. Reproduced from reference [39] with permission from Elsevier

membrane exchanger separates the CO<sub>2</sub> from other gases that are mostly present in the mobile phase. The CO<sub>2</sub> is transferred to the IRMS through a permeable membrane.

### 3 Applications of IRMS in Food and Environmental Analysis

#### 3.1 Environmental Application

Some applications in environmental analysis can provide good results when bulk material is analyzed by EA-IRMS.

Several applications impact on the use of IRMS to determine the concentration and isotopic composition of atmospheric CO<sub>2</sub>, essential for the identification of sources and sinks of anthropogenic CO<sub>2</sub> and the monitoring of various emission reduction techniques, such as geological CO<sub>2</sub> storage [41–43]. Atmospheric particulate matter (PM) is also the subject of intensive environmental investigation as it significantly influences the radiative balance of the Earth and, hence, its surface temperature. The thermal-based EA-IRMS technique offers an opportunity, which can substantially expand our current knowledge of carbonaceous particles. For example, the ratio of <sup>13</sup>C and <sup>12</sup>C stable isotopes ( $\delta^{13}\text{C}$ ) can provide important information about the potential sources and atmospheric processes influencing the characteristics of these particles.

Considering their persistence, plastic waste is an increasing environmental problem, especially in marine ecosystems. Interestingly, Berto et al. [44] performed a preliminary characterization of the stable carbon isotopes ( $\delta^{13}\text{C}$ ) of different plastic polymers both conventional (petroleum-derived) and “bio” (plant-derived) in order to increase the data set of isotopic values to characterize the polymers and to apply the technique to marine environmental monitoring. The  $\delta^{13}\text{C}$  values determined in different food packaging reflect the plant origin of the “bio” materials, while recycled plastic materials showed a  $\delta^{13}\text{C}$  signature between the plant and petroleum polymer source. The advantages of IRMS over other analytical methods used to characterize the composition of plastic polymers are sensitivity, reduced material required, rapidity of analysis, and low cost. It is not well-established in the presented method but future application will require separation of plastic from the matrix.

The measurement of the <sup>18</sup>O/<sup>16</sup>O ratio of water is one of the most important applications of IRMS. Very diverse applications, ranging from studies of effective identification and quantification of groundwater infiltration into sewage collection networks [45] to variations in natural abundance in the ecohydrologic cycle of some forest [46, 47] have been widely studied.

Isotopic abundance ratios can also establish dietary patterns and livestock movements as well as determine nutrient flows in aquaculture systems. Gamboa-Delgado [48] reviewed the most recent applications and findings derived from studies that

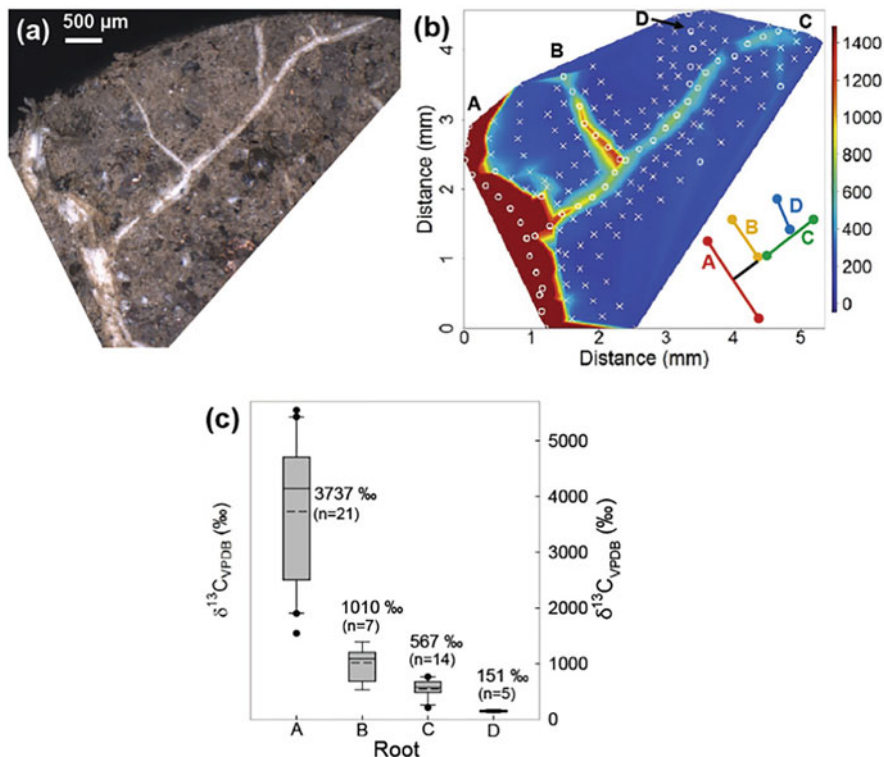
have used stable isotope analyses to (1) estimate nutrient assimilation in larval and juvenile organisms, (2) evaluate trophic dynamics and bioremediation potential of macroalgae and invertebrates in integrated multi-trophic aquaculture systems, and (3) investigate the trophic plasticity of aquatic species. The increasing application of CSIA of amino acids and fatty acids is emphasized as it has allowed exploring the physiological fate of specific organic compounds, while also assisting in the definition of nutritional requirements for aquatic species.

In addition to the examples compiled Gamboa-Delgado's [8] review, stable isotope analyses were used to estimate trophic niche width and overlap for flood and dry seasons in an Amazon floodplain lake. Fish ecology is influenced by the bioavailability of food that can be related to the analysis of the values of stable carbon ( $^{13}\text{C}/^{12}\text{C}$ ) and nitrogen ( $^{15}\text{N}/^{14}\text{N}$ ) through the trophic web. The results indicate that seasonal variation in isotopic niche width is specific to feeding habit. Understanding how fish trophic ecology responds to changes in the hydrological regime during the seasons is crucial for sustainable fishery management [49]. The same concept was used by Pouil et al. [50] in aquaculture to establish the fate of commercial pellets and natural resources in the trophic web structure (i.e., plankton of different sizes, macroinvertebrates, periphyton, fish, sediment and commercial pellets) in Indonesian giant gourami (*Osphronemus goramy*) aquaculture ponds. The results showed that, in a typical omnivorous species reared in semi-intensive pond conditions, the endogenous resources of the pond are barely utilized and the diet is mainly based on commercial pellets.

The integrated stable isotope analysis of nitrogen and carbon also provided information on the trophic position of invasive sally lightfoot crab *Percnon gibbesi* and a comprehensive evaluation of its feeding ecology. Results highlighted significant seasonal variation in *P. gibbesi* diet, improving our understanding of its trophic plasticity and potential dietary overlaps with other herbivore species [51].

Thanks to the possibility to extend the isotopic signature to non-conventional stable elements, IRMS can also contribute to understanding of complex Hg metabolism and HgSe nanoparticles formation of giant petrels [52]. This seabird species is one of the largest avian scavengers in the Southern Ocean, highly exposed to methyl mercury (MeHg) through their diet. Species-specific isotopic measurement ( $^{202}\text{Hg}$ ,  $^{199}\text{Hg}$ ) of HgSe nanoparticles in seabirds showed the largest variations between  $^{202}\text{Hg}_{\text{bulk}}$  and  $^{202}\text{HgSe}$  in muscle and brain tissues. This could be attributed to the higher fraction of Hg present as MeHg in these tissues. Hg-biomolecules screening highlights the importance of the isotopic characterization of these (unknown) complexes.

One of the hot topics in environmental research is to elucidate the intimate relationships between plant roots, the rhizosphere, and the soil. These relationships are favored by the release of organic compounds from the plant to the soil through various forms of rhizodeposition and the simultaneous uptake of nutrients from the soil to the plant. Denis et al. [15] spatially tracked and mapped the migration of plant-derived carbon (C) through roots into the rhizosphere and surrounding soil using LA-IRMS. This technique, in conjunction with a  $^{13}\text{C}$  tracer, allows the spatially resolved quantification of recent photosynthate in plant-soil microcosms at  $>> 100$  mm resolution. Using the measured  $^{13}\text{C}$  values, the fraction of recent



**Fig. 5** LA-IRMS tracks variable carbon flow through different roots. (a) Light image of area of soil pellet ablated. (b) Isotopic heat map of root “o” and rhizosphere-soil “x” ablation spots (50 mm spot size). Dark red indicates ablation spots with C isotope values greater than 1,500‰. Bottom right diagram shows the segment associated with each root letter, where the black line represents root that connects root A to branch point of root B and C. (c) Box plots of isotope values of the four roots with the mean listed (dashed line). Means are significantly different ( $P < 5\%$ ). Reproduced from [15] with permission of Wiley

plant-derived C at a given location in the root, rhizosphere, or bulk soil can be calculated. This work demonstrates the utility of LA-IRMS for mapping the extent of <sup>13</sup>C-label migration and root exudation into rhizosphere and soil. The relatively straightforward sample preparation does not require addition of any embedding medium which may confound isotopic measurements. Photosynthate distribution within the sampled roots displayed significant heterogeneity and distinct branch points where roots receiving high versus low photosynthate allocation diverged (Fig. 5).

SCIA using GC-IRMS has also been widely used in environmental analysis. The isolation of compounds from the high complex mixture that constitute the matrix is extremely important.

The compounds isolated for SCIA have been in an important proportion environmental contaminants. One interesting application is the compound-specific

analysis of stable C, N, and O isotopes in explosives contaminants and characterized biodegradation of 2,4,6-trinitrotoluene (TNT) and dinitrotoluene isomers in the soil of a contaminated site [24]. The technique provides information on the biodegradation of explosives in the environment and helps to assess the risk management of military ranges. Other is the CSIA of organophosphorus pesticides with different chemical properties [25] using  $^{13}\text{C}$  and  $^2\text{H}$  analysis. This study particularly compares HTC and Cr/HTC units to explore the limitations of hydrogen isotope analysis of heteroatom bearing compounds. The method showed potential for investigations regarding the fate of these pesticides. However, the enrichment and clean-up of organophosphorus pesticides from environmental samples with residual levels of micrograms are still an important challenge because only high concentrations are detected by IRMS.

Other contaminants whose degradation mechanism has been studied by SCIA are polybrominated diphenyl ethers (PBDEs). In an interesting study, Wang et al. [26] demonstrated that microbial reduction processes of PBDEs are one of the most important degradation mechanisms in an anaerobic environment such as most of the wetland ecosystems and established that the progressive debromination of BDE-47 and BDE-153 is the mechanism responsible for their degradation [26].

Going a step further, Shan et al. [53] examined biomarkers, along with stable carbon ( $\delta^{13}\text{C}$ ) and hydrogen ( $\delta\text{D}$ ) isotopic compositions of n-alkanes in a set of condensates collected from the East China Sea shelf basin to establish their organic matter input, depositional conditions and assess their thermal maturity. The combination of GC  $\times$  GC-TOFMS and CSIA analyses provided additional information concerning the maturity, environment, and post-depositional processes involved in the production of these condensates and enabled them to be distinguished from each other.

SCIA has also been applied to the natural compounds such as lignin that are difficult to transform in their monomers. Wang et al. [28] developed a novel method to measure the  $^{18}\text{O}$  of the three oxygens (O-3, O-4, and O-5) attached to the aromatic ring of the monomeric units (bearing no oxygen in their side chains) releasable by highly selective tungsten carbide supported by activated carbon-catalyzed hydrogenolysis of lignin. O-4 is obtained by measuring the  $^{18}\text{O}$  of H-type monomeric unit, while O-3 and O-5 can be calculated following isotope mass balance between different types of monomeric units measurable simultaneously with GC-IRMS. This study has application in plant physiological, metabolic, lignin biosynthetic, and climate studies.

The  $^{18}\text{O}/^{16}\text{O}$  (and  $^{15}\text{N}/^{14}\text{N}$ ) ratio of natural nitrate ( $\text{NO}_3^-$ ) and nitrite ( $\text{NO}_2^-$ ) can be used to extract valuable information about their source and fate as environmental contaminants, their metabolism as macronutrients in plants and animals, and their behavior in the N biogeochemical cycle. Wang et al. [27] developed a method to selectively isolate and purify nitrate and nitrite from natural water, soil, air, and plant materials by strong anion exchange for low- to normal-salinity samples or strong cation exchange for high-salinity samples, followed by quantitative conversion to their respective benzyl esters, which can be separated and individually analyzed for  $^{18}\text{O}$  (and potentially  $^{15}\text{N}$ ) by GC-IRMS. The method compares favorably with the



bacterial denitrification and chemical reduction methods, in terms of sensitivity and reliability, and has the potential to simultaneously measure  $\delta^{15}\text{N}$  and  $\delta^{18}\text{O}$  of nitrate and nitrite from natural samples of various origins.

Furthermore, as a link with the next section of food application, there are a number of studies that relate the current situation of water scarcity, drought, and climate change to characteristics of produced food. Recently, Spangenberg and Zufferey [54] reported that the carbon isotope composition of the solid residues obtained by freeze-drying white and red wines ( $\delta^{13}\text{C}$ ) could be used for tracing the water status of the grapes used to produce them. These authors compared different methods using  $\delta^{13}\text{C}$  values of other wine components (particularly those of whole wine obtained by EA-IRMS) and of wine ethanol obtained by GC-IRMS, for their suitability to assess the vine water status. For all wine varieties, both  $\delta^{13}\text{C}$  values were linearly correlated with the predawn leaf water potential ( $\Psi_{pd}$ ) and therefore serve as reliable indicators of vine water status. An evaluation of the advantages and drawbacks of the different methods showed that the  $\delta^{13}\text{C}$  analysis of wine ethanol by GC-C-IRMS is the most appropriate.

Drought stress is a major limiting factor for banana production and the incidence of drought spells is expected to increase with climate change. Climate smart practices and varieties are needed, but evaluation in the field is hampered by a lack of reliable physiological drought stress indicators. Vantuyghem et al. [55] investigated the potential of using stable carbon isotope ratios ( $\delta^{13}\text{C}$ ) and leaf temperature and its derived Degrees above Non-Stressed Canopy index (DANS) as proxies for drought stress in banana. Leaf samples were taken for  $\delta^{13}\text{C}$  analysis and leaf temperature was monitored throughout the day in a field trial with different banana plant stages (mother and daughter plants) under two irrigation treatments (rain fed and irrigated) during the dry season in Arusha, Tanzania. The results showed that  $\delta^{13}\text{C}$ , leaf temperature, and DANS were highly sensitive proxies for drought stress in banana and soil volumetric water content had a significant effect on both  $\delta^{13}\text{C}$  values and DANS.

Similarly, the stable isotope and photosynthesis response of tea (*Camellia sinensis*) were determined under different light and temperature conditions. The results showed that isotopes of young tea leaves were more enriched with increasing light intensity. However, the value of  $\delta^{13}\text{C}$  and  $\delta^{15}\text{N}$  seemed depleted, while  $\delta^2\text{H}$  and  $\delta^{18}\text{O}$  became enriched as temperature increasing from 15 to 35°C. Pearson's correlation showed a negative correlation between isotopes ( $\delta^{13}\text{C}$ ,  $\delta^{15}\text{N}$ , and  $\delta^2\text{H}$ ) and photosynthetic parameters that provided new insights which could be used to predict tea isotope responses arising from subtle seasonal or climate change conditions [56].

### 3.2 Food Applications

Food scientists have long recognized the need for analytical techniques that can provide data on issues such as traceability, authenticity, and origin of foods and beverages. Two recent literature reviews cover globally the state of the art of IRMS

in this area. In this sense, van Leeuwen et al. [57] demonstrated that GC-C-IRMS is one of the most powerful techniques available to detect fraudulent, illegal, or unsafe practices in the food and beverages industries. This technique is able to provide data on key food components such as aroma compounds, sugars, amino acids, and carbon dioxide (in carbonated beverages). Such data can be used to determine synthetic and natural ingredients, substitution of one ingredient for another, the use of synthetic or organic fertilizers, and origin of foods and food ingredients, including carbon dioxide. IRMS is gaining wider acceptance in food control laboratories due to its increasing versatility. This is largely due to the new possibilities that now exist to hybridize various diverse instruments, as the reliability of instrumentation and array of “on-line” techniques. These combinations are consolidating the use of IRMS analyzers as “another detection system”. The IRMS has ultimately become a tool to unequivocally prove food adulteration. This explains its increasing routine use in control laboratories to prosecute fraud, enforce the law and thus ensure food safety [19]. An interesting example investigated the fraudulent addition of sugars to coconut water by carbon isotope ratio ( $^{13}\text{C}/^{12}\text{C}$ ,  $\delta^{13}\text{C}$ , ‰) using EA-IRMS. The  $\delta^{13}\text{C}$  can detect the addition of one or more sugars in percentages higher than 2%. This study shows that of randomly sampled commercial coconut waters in Thailand approximately 50% were adulterated by the addition of sugars [33].

Another example of the usefulness of this technique to detect adulteration is the identification of the addition of acetic acid to rice vinegar by SCIA. The specific vinegar compounds for which the stable carbon isotope ratio is measured by GC-C-IRMS are acetic acid and acetoin. The method detected the adulteration of rice vinegar when it contains more than 10% synthetic acetic acid [31].

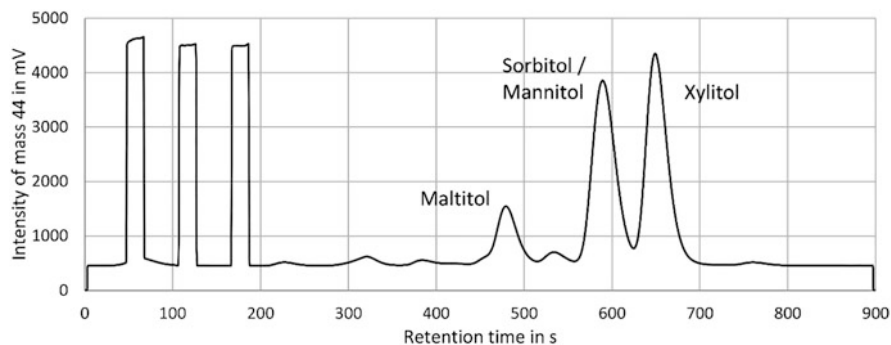
Vanilla flavor is one product also highly susceptible to food fraud since the main component “vanillin” is obtained from vanillin pods (naturally) but can easily be produced by much cheaper synthetic processes. Commonly origin of vanillin is identified because the  $\delta^{13}\text{C}$  ranges for synthetic vanillin from petroleum and C3 plants are depleted in comparison with the reported  $\delta^{13}\text{C}$  range for vanillin from vanilla orchids. Wide et al. [29] reported a preparation procedure that includes a headspace SPME clean-up and GC-IRMS to determine the isotopic carbon ratio of vanillin in food products. The method was highly specific to identify vanillin source but was not always an easy task, especially when very low concentrations are incorporated in complex food matrices. The synthetic vanillin can nowadays be considered as biosynthetic when it is obtained using different precursors (mainly lignin and guaiacol) or by enzymatic bioconversion (biovanillin) from natural precursors (mainly ferulic acid, eugenol and isoeugenol, and turmeric acid). The  $\delta^{13}\text{C}$  value for biovanillin was higher compared to vanillin from vanilla pods [30]. One fraud being perpetrated to obtain the  $\delta^{13}\text{C}$  range of vanillin from vanilla pods is combining vanillin derived from inexpensive sources. Schipilliti et al. [21] proposed also headspace SPME and GC-IRMS to determine the carbon isotope ratio ( $\delta^{13}\text{C}$ ) of vanillin extracted from vanilla pods, commercial flavored foodstuff, and in vanillin samples of different origins. Biovanillin can be easily characterized and distinguished from the synthetic ones. However, with the invention of new biosynthetic pathways, vanillin overlapping with the characteristic  $\delta^{13}\text{C}$  range reported for vanillin from vanilla pods can be produced [21]. Interestingly, a recent study

demonstrated that bulk and site-specific analysis by GC-IRMS of stable carbon and hydrogen isotope ratios of vanillin derived from ex-glucose can be easily identified by 2D plot of the  $\delta^{13}\text{C}_{\text{Bulk}}$  vs.  $\delta^2\text{H}_{\text{Bulk}}$  values of the combined data [30]. This is an important step ahead to detect this fraud.

Similarly, bis(methyl-thio)methane (BMTM), the molecule which provides “white truffle-like” flavor can be synthetic or of natural origin. The molecule can be extracted from various raw materials using anhydrous methanol.  $\delta^{13}\text{C}$  values established by GC-C-IRMS define the range of  $\delta^{13}\text{C}$  values of synthetic BMTM. Measurements by  $^1\text{H}$  NMR made on synthetic BMTM and BMTM extracted from “white truffle-like flavor” confirmed that the method determined the isotopic distribution of  $^{13}\text{C}/^{12}\text{C}$  ratio in two characteristic sites of this molecule [32].

Other important case of adulteration that affects different types of food is the undeclared addition of exogenous sugar products in food and beverages. The isotope analysis of the carbon-bound non-exchangeable (CBNE) hydrogen in mono and disaccharides has demonstrated its capacity to detect these added sugars in fruit juices and honey. The procedure utilizes a simple one-step reaction, with the derivatizing agent N-methyl-bis-trifluoroacetamide, to substitute the exchangeable hydroxyl-hydrogens with trifluoroacetate derivatives that are sufficiently volatile to be separated and measured by a GC-IRMS [34].

The sugar-replacement compound xylitol has also gained increasing attention because of its use in commercial food products, dental-hygiene articles, and pharmaceuticals. The traditional “birch xylitol” is considered a premium product, in contrast to xylitol produced from agricultural by-products such as corn husks or sugarcane straw. BSIA and CSIA of chewing-gum extracts by LC-IRMS were used to determine the  $\delta^{13}\text{C}$  isotope signatures for xylitol. The analyzed chewing gum could be clearly related to the raw-material plant class [35]. The chromatographic separation of a chewing-gum extract by the optimized LC-IRMS method is shown in Fig. 6. This application required the development of an organic-solvent-free



**Fig. 6** Chromatographic separation of xylitol from an aqueous extract of a chewing-gum sample (MSQ) containing xylitol, mannitol, sorbitol, maltitol, aspartame, maltitol syrup, sucralose, and acesulfame potassium. Reproduced from ref. [35] with permission of American Chemical Society

extraction protocol and HPLC method for the separation of xylitol from different artificial sweeteners and sugar-replacement compounds.

Sake is a traditional Japanese alcohol that has been adulterated by the addition of alcohol and sugar obtained as by-products in the brewing industry. Sutto and Kawashima [58] analyzed the  $\delta^{13}\text{C}$  values of the ethanol and glucose in Sake by LC-IRMS to check brewers' alcohol and sugar addition. The method was able to calculate the percentage of brewers' alcohol in commercial sake.

Similarly, the production of "Polish Vodka" is restricted by law to the ethyl alcohol of agricultural origins obtained from rye, wheat, barley, oat, triticale, and potatoes grown on the territory of the Republic of Poland. Unfortunately, genuine "Polish vodka" is counterfeited through the illegal practice of adding cheaper and more accessible corn liquor, which reduces the production costs of the spirit. Quantitative NMR and IRMS of deuterium and  $\delta^{13}\text{C}$  allows to distinguish alcohol from different raw materials. Both approaches led to the same conclusions regarding the percentage of maize-derived ethanol addition [37].

Non-concentrated juice (NFC) is a higher priced product than juice reconstituted from concentrate due to its higher quality and nutritional value. However, a growing problem in the industry is the adulteration of NFCs by the addition of water and sugar. Wu et al. [59] investigated the carbon and oxygen stable isotope ratios ( $\delta^{13}\text{C}$  and  $\delta^{18}\text{O}$  values) of the bulk juice and different juice components, and qualitatively and quantitatively analyzed the addition of water and sugar in NFC juices. Sugar-specific isotope analysis (SSIA) technology was used to determine the  $\delta^{13}\text{C}$  values of different sugars (sucrose, glucose, and fructose) and carbon content to qualitatively infer C3 plant sugar addition. The addition of water to NFC juice can also be determined by comparing  $\delta^{18}\text{O}$  values of extraneous water, pulp, and filtered juice. Thus, the determination of stable isotope ratios allows the detection of adulteration of NFC juice. Related to this fraud, other study compared commercial and directly-pressed orange and apple juices using both, IRMS ( $\delta^{13}\text{C}$  and  $\delta^{18}\text{O}$ ) and ICP-MS (Ni, Cr, Ca, Sr, Fe, K, Mg, and Na). The concentrations of the different metals provide interesting information to distinguish agricultural practices, whether the juice is orange or apple juice or whether it is natural or not. However, only the isotopic abundances demonstrated that most commercial juices come from concentrates and that in some cases they contain fraudulent addition of corn or cane sugars (since they are C4 plants) [36].

The establishment of the geographical origin is also an important potential of these techniques that has been applied to many different foods. The addition of sugars to honey is detected by the AOAC 998.12 that detects the addition of sugar from C4 plants. However, this method is not applicable to honey with non-extractable protein. The authenticity of pure honeys of various botanical and geographical origins with non-extractable proteins was studied on the basis of the stable carbon isotope ratio characterization of total honey, proteins, and the main sugars. Thirty-five honeys from which protein could not be extracted were all identified as adulterated with C-4 or C-3 sugars. The findings obtained supplemented the AOAC 998.12 C-4 sugar method, about honeys from which protein cannot be extracted [60].

Determining the geographical origin of wines is a major challenge in wine authentication, but little information is available regarding non-parametric statistical approaches for wines. Interestingly, a study on Chardonnay wines both vinified on a small scale from grapes cultivated in Japan, and imported from 8 countries, which analyzes the oxygen stable isotope and 18 elements applying non-metric multidimensional scaling (NMDS), kernel principal component analysis (KPCA) and principal component analysis (PCA). The non-parametric methods, NMDS and KPCA, separated domestic from imported Chardonnay wines better than the parametric method, PCA. Of 19 variables, 18 were important for geographical discrimination, with the  $\delta^{18}\text{O}$  value being the most significant in all statistic methods [61]. Similarly, GC-IRMS combined with SPME was able to measure the  $\delta^{13}\text{C}$  values of six typical volatiles commonly occurring in wine (isoamyl acetate, 2-octanone, limonene, 2-phenylethanol, ethyl octanoate, and ethyl decanoate). The  $\delta^{13}\text{C}$  of these volatile compounds is able to distinguish the real imported wines [20]. Recently, Suto and Kawashima [62] developed a method to determine the  $\delta^{13}\text{C}$  values of various organic acids in wine and sake that involves solid-phase separation (SPE) combined with LC-IRMS. This method detected the addition of organic acids (tartaric, malic, lactic, succinic) from different sources to sake and wine. The application of the method to different samples showed that the addition of lactic acid to sake is a common practice while organic acids in wine usually come from grapes.

A curious comparison of the use of conventional IRMS and SPME-GC-MS and non-conventional analytical techniques (Electronic Nose) to characterize and discriminate origin, drying, and age of saffron was reported by Rochi et al. [22]. The IRMS analysis of the stable carbon and nitrogen isotopes was reliable in discriminating the geographical origin of saffron. The SPME-GC-MS was able to discriminate the different origin, drying, and age based on the isolation of volatile compounds. An E-Nose was used as alternative and rapid tool to characterize the complex aroma patterns and to exploit authenticity of saffron samples. Results of this study could contribute to select and identify routine quality control methods for quality and authentication of saffron.

Within the cocoa market (*Theobroma cacao L.*), quality and prices are often determined by geographical origin, making traceability indispensable. Therefore, to investigate possibilities of tracing by analytical methods, 48 carefully selected cocoa samples from 20 countries have been profiled using a combination of stable IRMS and proton nuclear magnetic resonance ( $^1\text{H}$  NMR). Chemometric analysis of combined data sets from both, stable isotope data ( $\delta^{13}\text{C}$ ,  $\delta^{15}\text{N}$ ,  $\delta^{18}\text{O}$ ,  $\delta^2\text{H}$ ) and  $^1\text{H}$  NMR fingerprints, achieved good separation with increased classification rates compared to classification with data of the isolated methods. IRMS contributed primarily to discrimination between countries, while  $^1\text{H}$  NMR significantly contributed to separation of varieties, but also the regions within individual countries. This study thus demonstrates that combination of two analytical methods is an effective tool to enhance both, accuracy and precision, in authenticity testing of cocoa [63].

Another case where geographical identification is important is in the case of shiitake mushrooms to distinguish those grown in other countries (such as Korea or the USA) on imported Chinese inoculated medium from those grown in China.

In this regard, Chung et al. [64] developed a CSIA discriminant model of the  $\delta^{15}\text{N}$  amino acids aspartate and glutamate capable of distinguishing between Korean shiitake, Korean shiitake grown in Chinese inoculated medium, and Chinese shiitake with 100% accuracy.

Flaxseed is an economically important oilseed crop whose geographic origin is of significant interest to producers and consumers because every region may exhibit particular quality characteristics. To improve the discrimination rate and the geographical traceability of this crop, the chemical profiles of the flaxseed samples were characterized via lipids/fatty acids, stable isotopes, and antioxidant capacity. Principal component analysis (PCA) and orthogonal partial least squares discriminant analysis (OPLS-DA) attained a satisfactory discrimination rate after combining fatty acids, stable isotopes, and antioxidant capacity to trace the origin of flaxseed from five regions in northern China [65].

Not all the studies performed have been successful, a study that comprehensively analyzed different kinds of pure honey which was produced in various areas in China according to  $\delta^{13}\text{C}$ -EA-IRMS (AOAC method 998.12) and  $\delta^{13}\text{C}$ -LC-IRMS (proposed by the Intertek laboratory in Europe) methods. As for the  $\delta^{13}\text{C}$ -EA-IRMS method, the study confirmed the C4 sugar of all authentic honey samples. However, about 70% samples had  $\Delta\delta^{13}\text{C}$  (‰) values outside the range, indicating that a large proportion of pure honey in China cannot pass the  $\delta^{13}\text{C}$ -LC-IRMS test, although these honeys were extracted from unadulterated sources. Based on the findings, the  $\delta^{13}\text{C}$ -LC-IRMS method seems not appropriate to reliably detect adulterated honeys with C3 sugars in China [66].

Instead, the stable isotope ratios of carbon, nitrogen, hydrogen, oxygen, and mineral elements and their stoichiometric methods were examined as possible factors that could certify Chinese tea based on its production years. In this sense, a total of 43 multi-element stable isotope ratios of Xiangzhujing Pu'er tea in five production years were determined through ICP-MS and EA-IRMS methods. Two unsupervised learning techniques (PCA and hierarchical clustering analysis) and three supervised learning techniques (partial least squares discriminant analysis [PLS-DA], back-propagation artificial neural network [BP-ANN], and linear discriminant analysis [LDA]) were used on the basis of 18 statistically significant multi-elemental stable isotope ratios to build authentication models for Pu'er tea. The three supervised models correctly separated the corresponding production years of the samples [67].

Reliable identification of the origin of economically important animal species is essential to preventing fraud. Remarkably, Xu et al. [68] investigated the efficacy of stable isotope and multi-element analyses in claw meat, abdominal meat, gills, and the hepatopancreas of swimming crabs (*Portunus trituberculatus*) using specimens from three discrete production areas in China and developed an origin discrimination model using stepwise discriminant analysis. Stable isotope ratios ( $\delta^{13}\text{C}$ ,  $\delta^{15}\text{N}$ ) and the contents of 14 elements (Na, Mg, Al, K, Mn, Fe, Co, Cu, Zn, As, Se, Rb, Ag, and Ba) varied significantly by tissue and by sampling location. Obvious isotopic fractionation effect was observed among different tissues, the muscle tissue and gills isotope ratios were more effective geographical origin indicators than hepatopancreas. Although the accumulation effects of the elements differed between

tissues, the discriminatory correctness was generally consistent across tissues. This method supports the geographical traceability and origin identification of *P. trituberculatus* [68].

Other species such as Octopus play an increasingly important role in ocean ecosystems and global fisheries, but provenance authentication techniques are very scarce. Martino et al. [69] investigated whether chemical profiling can distinguish geographical origins of octopus on international and domestic scales in samples consisted of wild-caught octopus from south-east Asia and southern Australia. To this end, these researchers used a combination of stable carbon ( $\delta^{13}\text{C}$ ) and oxygen ( $\delta^{18}\text{O}$ ) isotope analyses by EA-IRMS of internal calcified structures called statoliths, with elemental analyses (X-Ray Fluorescence using Itrax) of soft-tissue. Multivariate profiles exhibited distinctive regional signatures, even across species, with high classification success back to region of origin. This study validates isotopic and multi-elemental profiling as an effective provenance tool, which could be used to support transparency and accountability of seafood supply chains and thus encourage sustainable use of ocean resources [69].

Growing demand for organically-grown food has resulted in the establishment of multiple standards, certification, and labeling of organic products that were successfully established, thanks to the use of BSIA and CSIA. Mihailova and Kelly [70] reviewed innovative IRMS techniques to ascertain organically-grown food, identifying three growing techniques (1) stable sulfur isotopes (providing additional information about regional agricultural practices and fertilization strategy), (2) CSIA of plant-derived nitrate (distinguishing between crops grown with synthetic and organic fertilizers), and (3) stable magnesium isotope analysis which can be performed using MC-ICP-MS. Independently of these promising techniques, the natural abundance of stable isotope  $^{15}\text{N}$  ( $\delta^{15}\text{N}$ ) in production systems has already become an important technique to distinguish organic products from conventional ones. Trapp et al. [71] evaluated the use of  $\delta^{15}\text{N}$  values recorded for nitrogen fertilizers, soil and plant tissue in order to set the differences between organic and conventional agricultural production systems applied to rice, potatoes, apple, and banana crops. Values recorded for  $\delta^{15}\text{N}$  in food from organic farms were higher than those recorded in conventional farms. Results have allowed for differentiation between production systems.

Similarly, O'Sullivan et al. [72] reviewed the scientific literature on stable isotope analysis for authentication of milk and dairy products. Through numerous examples, these authors demonstrate the usefulness of the technique for identifying geographical origin as well as the influence of processing and seasonality on the results especially using SCIA of caseins. Furthermore, these authors identified the need of complete datasets for comparison and statistics as well as promising areas for the future, such as the analysis of less used elements.

Other field of application of IRMS is animal nutrition that ultimately ensures the quality and safety of food. One of these applications is to obtain time-integrated estimation of assimilated and not just ingested nutrients. This feature turned possible its application to evaluate the effects of dietary nucleotides and glutamate on carbon turnover ( $\delta^{13}\text{C}$ ) in the brain of weanling piglets. The carbon turnover verified for

cerebral tissue was faster for diet containing glutamate in comparison with other diets, supporting the fact that glutamate contributed to develop the piglets' brain, due to the fastest incorporation of  $^{13}\text{C}$ -atoms in this tissue at post-weaning phase, despite the energy deficit experienced by them [73, 74]. Other application of IRMS is to determine whether animals consume animal by-products (ABP), a widely prohibited practice because there is a risk of transmitting scrapie – a fatal prion disease in human beings. Da Silva et al. [75] used IRMS for  $\delta^{13}\text{C}$  and  $\delta^{15}\text{N}$  to trace animal proteins in the serum of sheep. This method is biologically and economically relevant for the veterinary field because it can track protein over time or make a point assessment of animal feed with high sensitivity and resolution, providing a low-cost analysis coupled with fast detection.

The use of SCIA within food analysis has many more fields of application even though the most important have been described. Stimulatingly, Llana-Ruíz-Cabello et al. [23] described the results obtained by hyphenating pyrolysis-gas chromatography (Py-GC) with carbon IRMS for the analysis of a polylactic acid based bio-plastic extruded with variable quantities of a natural plant extract or oregano essential oil. The chemical structural information of pyrolysates was first determined by conventional Py-GC and, then, the measure of  $\delta^{13}\text{C}$  in specific compounds was done by Py-GC-C-IRMS. PyCSIA is shown as a promising tool to study such materials, providing not only a fingerprinting, but also valuable information about the origin of the materials, allowing the traceability of additives and minimizing sample preparation [23].

## 4 Conclusions

IRMS is considered a well-established technique. However, new instrumentation is being developed to solve the existing problems in the technique, such as, isotope analysis in multi-element molecules. Furthermore, new ways of applying it to specific case studies are still being explored and are opening new horizons within the field. Among them, the on-line combination of IRMS with numerous other techniques, a higher degree of automation, and better standardization will improve the reliability of the technique. The future objectives of this technique will also focus on those objectives that dominate the analytical chemistry landscape, which are basically to increase the speed of analysis and reduce its economic cost while simultaneously obtaining higher resolution and precision data and developing methods and protocols that are within the principles of green chemistry, i.e. that are respectful of the environment.

IRMS is now part of a versatile platform with a wide range of possibilities as it can be coupled with various systems for preparation, separation, and isolation of sample components. The most suitable system will depend on the type of sample to be analyzed and the objective of the study. For solid, non-volatile samples, such as food, amino acids, or fatty acids, EA-IRMS is the technique of choice, although the result it provides is the mean value of the isotope ratio for the whole sample, even



though samples can contain several thousands of compounds. However, in complex environmental and food samples, CSIA of some sample components provides a greater amount of information than BSIA. As complementary techniques to determine the isotope ratio in each component of the sample, GC-IRMS or LC-IRMS can be used to separate compounds by GC or LC, respectively, prior to IRMS.

Whichever method of sample introduction is combined with IRMS shows many applications in the environment and foodstuffs. In the environmental field IRMS has shown great versatility to increase the knowledge of carbonaceous materials, identify organic contaminants, establish the water origin, or elucidate relations between different interfaces. In the food field the applications to establish food authenticity, geographical origin, organic food practices, or most common food fraud have grown enormously. A noticeable gap in the market exists for compound-specific isotope ratio standards, although several suppliers are available for bulk isotope ratio standards such as polyethylene, sugar, and flour.

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# Liquid and Gas Chromatography–Mass Spectrometry Methods in Food and Environmental Safety



Yelena Sapozhnikova

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**Abstract** This chapter describes recent advances in applications of gas chromatography (GC) and liquid chromatography (LC) coupled with mass spectrometry (MS) for analysis of contaminants in the field of environmental and food safety during the last decade (2011–2021). Most employed MS analyzers with unit-resolution, different ionization modes, and improvements in liquid and gas chromatography techniques are discussed. Regulatory compliance for GC-MS/(MS) and LC-MS/(MS) identification as outlined by regulatory agencies is presented. Examples of innovative uses of state-of-the-art methods for analysis of diverse contaminants in the last decade are provided, and an opinion on future trends in the field is offered.

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## 1 Introduction

In environmental and food safety, contaminants analyzed by gas chromatography–mass spectrometry (GC-MS/(MS)) methods are non-polar, semi-polar, volatile, and semi-volatile compounds. These include pesticides, polycyclic aromatic hydrocarbons (PAHs), polychlorinated biphenyls (PCBs), polybrominated diphenyl ethers (PBDEs) and other flame retardants, persistent organic pollutants (POPs), i.e. chemicals identified by the Stockholm convention as persistent, bioaccumulative, and toxic, and emerging contaminants, such as chlorinated paraffins, organophosphate flame retardants, plasticizers, and many others. Contaminants analyzed by liquid chromatography–mass spectrometry (LC-MS/(MS)) methods are semi-polar to polar and non-volatile. Examples are some pesticides, pharmaceuticals, personal care products, natural toxins such as mycotoxins, veterinary drugs, polar flame retardants, and many others.

Since pesticides are the most studied contaminants in food and environmental safety, a quick look into the past publication records provides a glance into when mass spectrometry coupled with LC and GC started to be routinely used in analytical laboratories. A search in Web of Science for “LC-MS pesticides” returned 1 paper published in 1989, describing confirmation of pesticides by LC-MS, then 2 hits for 1991 on applications of LC-MS for pesticide analysis. It was not until 2010 that 102 papers reporting LC-MS for pesticide analysis were published. A similar search for “GC-MS pesticides” in the Web of Science engine returned 1 paper published in 1990 on an analysis of 50 pesticides in water by GC-MS, growing to 100 papers per year published in 2003. At that early time of GC- and LC-MS analysis, most MS detectors were single quadrupole or ion traps. Since the 2010s, triple quadrupoles (QqQ) became more common tools in analytical laboratories. Additionally, different ionization techniques associated with both GC- and LC-MS instrumentation were developed to cover a wide range of polarity for analytes of interest (Fig. 1).

This chapter is covering advantageous developments in applications of GC-MS/(MS) and LC-MS/(MS) methods in analysis of small molecule organic contaminants (under 1,000 Da) in environmental and food safety with a focus on the last decade (2011–2021), with the emphasis on unit-resolution MS instrumental applications, with high-resolution MS applications described in a separate chapter.



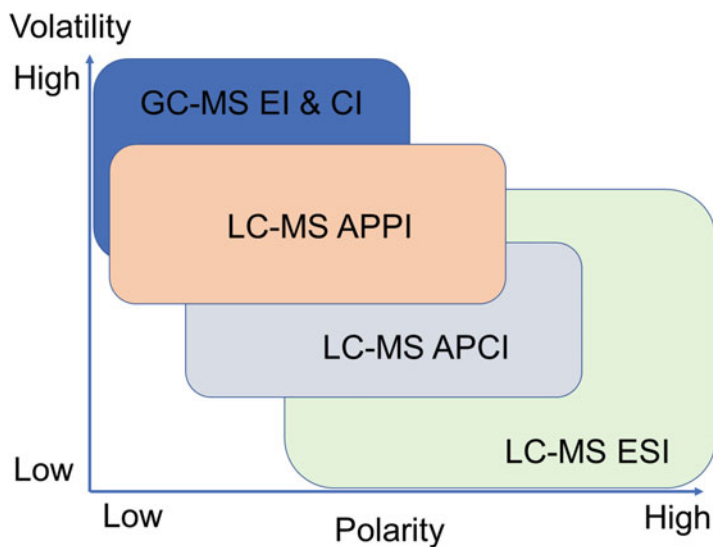


Fig. 1 Instrumental analysis and ionization modes based on analytes' volatility and polarity

## 2 Recent Advances in Gas Chromatography–Mass Spectrometry Analysis

### 2.1 GC-MS Analyzers

GC-MS/(MS) analyzers utilized for analysis of residual levels of contaminants in environmental and food safety encompass single and triple quadrupoles, ion trap, Orbital ion trap (Orbitrap), time of flight (TOF), and hybrid systems: quadrupole (Q)–ion trap (Q-trap), Q-TOF, Q-Orbitrap. In the last 10 years (2011–2021), GC coupled with triple quadrupole (QqQ) analyzers was most often applied for targeted analysis of contaminants and became a gold standard for environmental and food safety testing. Indeed, a Web of Science search for “pesticide analysis GC-MS” with different MS analyzers for 2011–2021 showed 59% of the published articles used GC-MS/MS with QqQ analysis, followed by 15% using GC-MS with single quadrupole MS. The remaining 12% and 15% of the published articles used TOF and ion trap MS, respectively. Starting from the 2010s, many laboratories transitioned from GC-MS in single ion monitoring (SIM) mode to GC-MS/MS analysis operated in multiple/selected reaction monitoring (MRM/SRM) mode, which provided greater sensitivity, selectivity, and specificity compared to GC-MS with single quadrupole analyzers.

Main vendors of GC-MS/(MS) instruments are Agilent Technologies, Thermo Fisher Scientific, LECO, and Shimadzu. The biggest advances in GC-MS instruments in the last decade are improvements in MS analyzers' sensitivity and scan speed (Da/s). The latest GC-MS/MS QqQ instruments provide instrument detection

limit (IDL) sensitivity for octafluoronaphthalene (OFN)  $< 0.4$  fg, MRM speed of up to 800 MRM transition/s, minimum MRM dwell time of 0.5 s, and scan speed  $\leq 20,000$  Da/s. In addition, to reduce down time needed to vent the instrument for ion source cleaning and/or GC column change, Agilent 7010D QqQ offered an automated, self-cleaning ion source, and Thermo Fisher Scientific TSQ 9000 enabled changing ion source and GC column without venting the instrument.

Recent improvements in detectors' scan speed and acquisition rates translate to more acquired MRMs per unit time, resulting in a greater number of contaminants analyzed in a single run, thus increasing the scope of the method. For example, a multi-residue method was developed and validated for 192 pesticides in animal feed by GC-MS/MS [1] with 2 MRM transition for each analyte in a 22 min GC run time. A total of 32 contaminants including PCBs, PBDEs, PAHs, and organochlorine pesticides were analyzed by GC-MS/MS in mussels and clams with 3 MRM transitions/analytes in a 45 min GC run [2]. A multi-class method for pesticides, PCBs, PAHs, PBDEs, and other flame retardants in meats, poultry [3], and catfish [4] covered 232 analytes with 3 MRMs/analytes in a 10 min fast low pressure GC run. In a recent study, 4 MRM transitions/analytes were applied for analysis of 400 pesticides in food samples in a 60 min GC run [5]. These examples demonstrate how modern GC-MS analyzers provide wide scope of analysis by covering hundreds of contaminants and multiple classes of analytes in a single GC-MS/(MS) run, thus increasing throughput and laboratory efficiency.

Taking advantage of improvements in MS detector's speed, 3 MRMs instead of 2 MRMs/analytes are acquired to improve selectivity and provide more confidence in the identification based on additional MRM transitions and their ion ratios (see Sect. 4), thus increasing identification reliability and minimizing false negative findings in complex samples with matrix interferences.

Additional advantage in acquisition of multiple MRMs is software improvements for GC-MS/MS with dynamic/scheduled MRM algorithms, allowing the user to specify retention time and time window for acquiring selected MRMs compared to laborious division of MRMs into time segments.

Modern MS analyzers provide great sensitivity, and each next version is at least 2–3 times more sensitive than the previous version. Agilent 7000 MS/MS QqQ introduced in 2014, for example, has IDL  $\leq 4$  fg while 7010B version, introduced in 2016, has IDL  $\leq 0.5$  fg for 2 fg OFN injected, thus offering an eight-fold sensitivity increase in just 2 years. Increased sensitivity provides greater signal to noise (S/N) ratio and lower limits of detection (LODs), allowing to measure contaminants at lower levels. This is especially valued in the environmental field where a general rule of thumb is to measure the lowest quantities possible. In the food safety arena, most chemical contaminants are regulated and have established tolerances or maximum residue limits (MRL). For example, for most pesticides, MRLs are 10 ng/g, thus the lowest amounts needed to be quantified are  $\frac{1}{2}$  MRL, or 5 ng/g, and the lowest calibration curve point is usually at 0.5–1 ng/g. In this case, increase in MS sensitivity permits injection of more diluted final extracts without resulting in increased LODs, decreased matrix effects, and improved robustness. This advantage can result in fewer sample preparation steps, sample dilution instead of

concentration/evaporation, smaller sample equivalent injected on column, and consequently, less instrument maintenance.

## 2.2 GC-MS/(MS) Ionization Modes

Electron ionization (EI) is most commonly used with GC-MS analyzers. The advantage of EI is its universal applicability. Mass spectra generated in EI mode with 70 eV are highly reproducible with any GC-MS/(MS) instrument and are used for identification and confirmation with existing commercial EI spectral libraries. One of the most utilized spectral databases is the NIST/EPA/NIH EI-MS library containing 306,643 compounds (2020 release), including pesticides, industrial chemicals, petrochemicals, surfactants, drugs and metabolites, toxins, etc. The main disadvantage of EI is in its harsh ionization causing extensive fragmentation, and in most cases lack of a diagnostic molecular ion. Softer ionization techniques, such as chemical ionization (CI) in positive (PCI) or negative (NCI) mode, atmospheric pressure chemical ionization (APCI), and cold EI overcome this challenge and provide highly diagnostic molecular ion.

GC-MS with NCI in selected ion monitoring (SIM) mode using methane as a chemical reagent was used for determination of organochlorine pesticides and PBDE congeners in air particulate matter [6]. In another study, a multi-residue method for 51 pesticides in green coffee beans was developed with GC-MS-NCI [7], and high MS selectivity was achieved, while matrix effects were high for this difficult food matrix. Gonzalez-Gago et al. compared sensitivity of GC-MS-EI with GC-MS-NCI for PBDE congeners and concluded that GC-MS-NCI had lower LODs, especially for higher brominated congeners [8]. Ayala-Cabrera et al. compared GC-MS with positive (PCI) and negative chemical ionization (NCI) modes and EI mode for the determination of fluorotelomer olefins, fluorotelomer alcohols, perfluoroalkyl sulfonamides, and sulfonamido-ethanols in water [9]. For most of the analytes, PCI worked the best generating protonated molecules and low fragmentation compared to high fragmentation observed in EI and NCI modes.

Applications of APCI in GC-MS/(MS) published by 2020 are summarized in three recent reviews [10–12]. These reviews highlighted APCI as an advantageous soft ionization technique for generating spectral data with protonated molecular ions for improved identification and low LODs. Li et al. discussed advantages and drawbacks of GC-MS-APCI [12], while Fang et al. [10] reviewed its applications for the analysis of persistent organic pollutants (POPs) and Niu et al. [11] covered applications in non-targeted analysis and targeted analysis for pesticides, flame retardants, PAHs, PCBs, dioxins and furans, sterols, esters, pharmaceuticals, and cannabinoids. Cherta et al. first studied application of GC-MS/MS-APCI for 25 pesticides selected based on high fragmentation and low/no molecular ion in EI mode [13]. Under APCI conditions with water as a modifier, abundant protonated ions  $[M + H]^+$  were observed for most of the selected pesticides and used as precursors for MS/MS, thus resulting in increased selectivity and sensitivity. Among pesticides,

pyrethroids are especially difficult to analyze with EI due to the lack of molecular ion and extensive fragmentation. GC-MS/MS-APCI was shown to improve analysis of pyrethroids [14] based on the formation of highly abundant protonated molecular ions with low LODs. One disadvantage of APCI is a strong interference of matrix components leading to matrix-induced suppression or enhancement. In the study of pyrethroids in fruits and vegetables, average signal reduction was 55% [14]. GC-QTOF-APCI was used in a non-targeted study to investigate chemicals migrating from food packaging in combination with GC-TOF-EI to increase confidence of identification by using two complementary ionization modes [15]. Other soft ionization techniques with high potential for future applications in GC-MS/MS analysis are atmospheric pressure photo-ionization (APPI) [16], dielectric barrier discharge ionization (DBDI) [17], and supersonic molecular beam (SMB) also known as cold EI [18].

### 2.3 Improvements in Gas Chromatography

Modern analytical laboratories demand fast sample turnaround time and higher throughput to analyze as many samples as fast as possible. Typical GC run with the most often used analytical column: (5%-phenyl)-methyl-polysiloxane, 30 m  $\times$  0.25 mm internal diameter (i.d.)  $\times$  0.25  $\mu$ m film thickness can take 30–60 min depending on the selected analytes and GC conditions. Increasing the speed of GC separation while reducing total run time is important to achieve greater productivity.

Zocalli et al. reviewed GC-MS techniques from the last decade [19] and Pico et al. reviewed recent innovations in GC-MS for pesticide analysis [20], and both reviews presented several fast GC approaches and their applications. One way to reduce GC run time is to alter GC column dimensions, i.e. reduce length, internal diameter (i.d.), and film thickness. To discuss reduction of i.d. here, classification of GC capillary columns was based on the following parameters [21] – microbore: i.d. 0.1–0.2 mm, narrow bore: i.d. 0.2–0.3 mm, wide bore i.d. 0.3–0.5 mm, and megabore: i.d.  $\geq$ 0.5 mm. Use of microbore column (20 m  $\times$  0.18 mm i.d.  $\times$  0.18  $\mu$ m) and pulsed pressure injection was reported for analysis of 356 pesticides by GC-MS/MS [22]. Analysis time was 18 min, plus significantly higher and narrower peaks were observed, resulting in greater signal to noise (S/N) and lower sensitivity. In another study with GC  $\times$  GC-MS, 10 m  $\times$  0.1 mm  $\times$  0.1  $\mu$ m column was used for cryogenic modulation [23] to evaluate rapid two-dimensional comprehensive GC  $\times$  GC analysis with single quadrupole MS for cosmetic allergens. Low polarity 8.9 m  $\times$  0.1 mm  $\times$  0.1  $\mu$ m and medium polarity 1.1 m  $\times$  0.1 mm  $\times$  0.1  $\mu$ m GC columns were used in first and second dimension, respectively, with analysis time of 11.4 min. The well-known drawbacks of microbore columns are reduced sample capacity and method robustness.

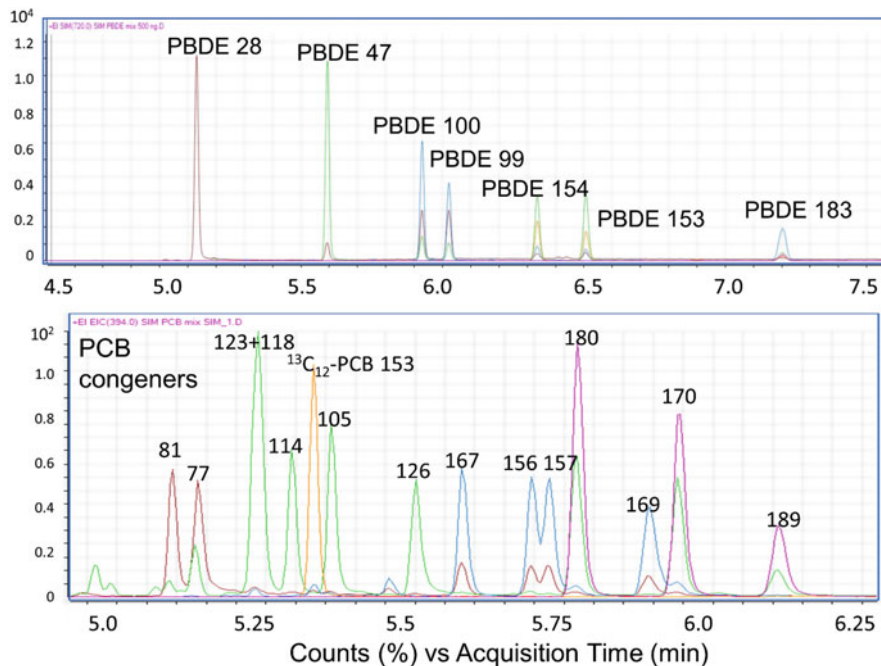
In contrast to microbore column, narrow bore GC columns (i.d. 0.2–0.3 mm) are widely used in environmental and food safety applications. GC runtime of 12 min

was reported for the analysis of PBDEs and their methoxylated metabolites by Cruz et al. [24] with  $10\text{ m} \times 0.25\text{ mm} \times 0.1\text{ }\mu\text{m}$  GC column and triple quadrupole MS/MS using EI and NCI modes. In another study GC separation  $<10$  min for pesticides and PCBs was achieved with  $15\text{ m} \times 0.25\text{ mm} \times 0.25\text{ }\mu\text{m}$  by GC-MS/MS-APCI [25]. Several factors were optimized to reduce GC analysis run time: short GC column (15 m), high starting oven temperature ( $120^\circ\text{C}$ ), fast rate of temperature programming ( $30^\circ\text{C}/\text{min}$ ), and high flow rate (6 mL/min).

High flow rate in addition to increased velocity of carrier gas is the main characteristic of low pressure (LP) vacuum outlet GC-MS/(MS) technique [26]. LPGC-MS/(MS) uses short (5–15 m) megabore (i.d.  $> 0.5\text{ mm}$ ) analytical column connected to a restrictor at the inlet and vacuum at the MS outlet. By extending vacuum conditions through LPGC column, carrier gas (helium) linear velocity becomes 10 times greater compared to under atmospheric pressure, thus speeding up GC separation. In addition, higher flow rates, rapid heating, thicker film thickness provide faster separation, more sensitivity, high sample capacity, greater ruggedness, and less degradation of thermally labile compounds [26, 27]. Khan et al. compared LPGC-MS/MS and conventional GC-MS/MS for analysis of 259 pesticides in tobacco [28]. A fast separation of 14 min was achieved for LPGC-MS/MS analysis with  $15\text{ m} \times 0.53\text{ mm} \times 1\text{ }\mu\text{m}$  column and flow rate of 4 mL/min compared to 42 min using conventional GC-MS/MS with  $30\text{ m} \times 0.25\text{ mm} \times 0.25\text{ }\mu\text{m}$  column. Additionally, low sensitivity with limit of quantitation (LOQs)  $< 2\text{ ng/g}$  was demonstrated for all pesticides. Recent studies using LPGC-MS/MS with  $15\text{ m} \times 0.53\text{ mm} \times 1\text{ }\mu\text{m}$  LPGC column connected to  $5\text{ m} \times 0.18\text{ mm}$  guard column and carrier gas at 2 mL/min for analysis of  $>200$  pesticides, PCBs, PBDEs, and other flame retardants had a total run time of 10 min with  $\text{LOQ} \leq 5\text{ ng/g}$ , and low matrix effects in complex samples of meat, poultry, and fish [3, 4]. Figure 2 illustrates fast separation of PBDE and PCB congeners relevant to environmental and food safety using LPGC-MS/MS in 7 min. On the other hand, disadvantages of LPGC-MS/(MS) are decreased separation efficiency, greater potential for leaks, and typically only MS-based detection since vacuum conditions are required.

Among other recent developments in GC-MS worth mentioning is column backflushing GC-MS/MS method [29] to eliminate unwanted matrix components from GC column by reversing the column flow with a pressure-controlled tee device. Fast analysis of pesticides in dietary supplements with column backflushing resulted in increased sample throughput (50%), decreased instrument maintenance, and greater ruggedness [29]. In another study, column backflushing was used for analysis of nitrosamines in bacon [30] with reporting limits of  $0.1\text{ ng/g}$ , and backflushing was demonstrated to be rugged for long-term use with minimal maintenance.

Another advantageous technique for analysis of challenging complex samples is multi-dimensional GC, which offers improved resolution for analytes with greater separating power, higher peak capacity, improved identification, and lower detection limits compared to one-dimensional GC separation. A recent review outlined the use of comprehensive two-dimensional (2D) GC in environmental analysis in targeted and non-targeted applications [31]. In non-targeted screening of surface water by  $\text{GC} \times \text{GC}$ -TOFMS, over 3,000 chemicals were detected, including pharmaceuticals



**Fig. 2** Example of fast separation of PBDE and PCB congeners using low pressure vacuum outlet GC-MS/MS

and personal care products (PPCPs), sunscreens, pesticides, hormones, fragrances, and emerging endocrine disrupting chemicals [32]. In another study, 327 persistent and bioaccumulative compounds were identified in blubber bottlenose dolphins by GC  $\times$  GC-TOFMS non-targeted analysis, with 280 identified chemicals not typically monitored in environmental surveys [33]. In food safety applications, non-targeted analysis with GC  $\times$  GC-TOFMS was used to screen for chemicals migrating from food packaging with 91 chemicals identified [34].

While most GC  $\times$  GC-MS/(MS) studies to date used EI to utilize common MS databases, some reports have emerged with alternative ionization modes. Pulsed flow modulation GC  $\times$  GC-MS with cold EI (supersonic molecular beam) was reported for pesticide analysis in agricultural products [35]. The identification with this technique improved NIST library identification probabilities. Another study explored milder EI conditions, e.g. 20 eV vs. 70 eV for GC  $\times$  GC – quadrupole MS for analysis of pesticides, sterols, linear alkanes, etc. [36]. Lower energy resulted in increase of the relative abundance of higher-mass diagnostic ions and fragments. GC  $\times$  GC coupled with TOF with APCI was evaluated for analysis of flame retardants and plasticizers with direct probe [37].

GC  $\times$  GC-TOFMS is a powerful tool for targeted and especially non-targeted analysis in complex and difficult matrices which can tolerate little/no sample preparation, however tentative identification of thousands of compounds can be a long

and overwhelming process, and software improvements are expected in the future to streamline and improve data analysis.

### 3 Liquid Chromatography–Mass Spectrometry

#### 3.1 Analyzers

Just as for GC-MS/(MS) analysis, the most common LC-MS/(MS) analyzers are single and triple quadrupoles, ion trap, Orbitrap, TOF, and hybrid instruments combined with a quadrupole: QTRAP, Q-TOF, Q-Orbitrap. Based on Web of Science search for papers on “pesticide analysis LC-MS” with different MS analyzers in the last decade, 57% used QqQ, followed by 19% TOF, 16% ion trap, and 8% single quadrupole. Major vendors of LC-MS/(MS) instruments are Agilent Technologies, Shimadzu, Sciex, Thermo Fisher Scientific, and Waters.

LC-MS/MS with QqQ instruments have become “workhorses” of modern laboratories for analysis of LC-amenable contaminants. Common characteristic of modern LC-MS/MS QqQ (based on Agilent QqQ 6495C) are: MRM speed is 500 MRM transitions/second, minimum MRM dwell time is 0.5 s, polarity switch <25 ms, collision cell clearance time <0.5 s, scan speed ~17,000 Da/s, and IDL sensitivity is <0.6 fg based on reserpine on column. Other advances in the last years, besides improved sensitivity and speed, are improvements in ion source design and diversity of ion source ionization modes.

#### 3.2 LC-MS/(MS) Ionization Sources

The most frequently used ionization sources in LC-MS(MS) analysis are electrospray ionization (ESI), atmospheric pressure chemical ionization (APCI), and atmospheric pressure photo-ionization (APPI). ESI dominates in the published LC-MS/(MS) methods for LC-amenable contaminants due to high selectivity, sensitivity, and efficient ionization for a wide range of diverse analytes. ESI can be operated in positive (+) and negative (–) modes, and ESI (+) is generally preferred mode based on most compounds ionization efficiency producing  $[M + H]^+$  or other adduct ions, resulting in wider scope of analysis. ESI (–), on the other hand, creates deprotonated ions  $[M-H]^-$  and is characterized by lower background noise [38]. Liigand et al. [39] challenged the assumption that most compounds are better suited for ESI (+) by comparing ionization efficiency of 33 compounds ionized in both ESI(+) and ESI (–) modes. Their findings showed that ESI (–) provided better sensitivity for almost half of the selected analytes. To take advantage of both ESI polarities in one LC run, polarity switching is employed. A recent study demonstrated simultaneous analysis of 52 multi-class illegal dyes in food [40] with ESI polarity switching in one LC-MS/MS run of 12 min for acidic, neutral, and basic

analytes. Another interesting investigation of this study was a 100-fold dilution of the extracts then taking advantage of the instrument sensitivity while eliminating matrix effects. In another study, ESI polarity switching was applied to acquire 27 and 710 scheduled MRM transitions in ESI(−) and ESI(+), respectively, in 10 min LC run to determine pesticides, veterinary drugs, and their metabolites in catfish [41].

Four ionization sources: ESI, heated ESI (HESI), APPI with and without dopant, and multi-mode source with ESI and APCI were compared for analysis of 40 pesticides in tomato and garlic [42]. The lowest LODs were achieved with ESI and HESI, and ESI was significantly less affected by matrix effects compared to HESI.

Softer, but currently less utilized ionization sources are APCI and APPI. The applicability of three ionization sources: ESI, APCI, and APPI was tested for analysis of five pharmaceuticals in wastewater samples [43], and ESI provided the best ionization efficiency, lower LODs, greater S/N, and lower matrix effects. Tokumura et al. compared analysis of 14 organophosphorus flame retardants by LC-MS/MS with APCI and ESI, as well as by GC-EI-MS and GC-NCI-MS [44]. LC-MS/MS with APCI afforded the lowest LOQs for 12 analytes. In contrast, Silva et al. compared efficiency of ESI and APCI ionization sources for analysis of 22 pesticides in food and found two- to four-fold lower LOQs for ESI vs. APCI, and greater matrix effects in APCI [45]. These examples highlight the differences between ionization efficiency of various chemicals, depending on their properties and the impact of matrix effects for different ionization mechanisms.

Ramirez et al. reported the determination of 28 parent and 15 alkylated PAHs in environmental waters by LC-MS/MS-APPI with chlorobenzene as a dopant on a polymeric C18 column [46]. Brecht et al. reported the development of a fast-switching dual source operating ESI and APCI simultaneously or in switched mode [47] with potential for routine use in LC-MS/(MS) analysis in the future. Similarly, Galani et al. evaluated a new ionization interface, UniSpray, in comparison with ESI [48] for 81 pesticides in food and water samples. The UniSpray was shown to achieve better sensitivity and improved S/N, but overall, LODs were similar to ESI, while signal suppression from matrix effects was lower with UniSpray compared to ESI.

### 3.3 *Improvements in LC Analysis*

Since ultra-high performance (UHP) LC instruments were introduced in 2004, they became routine LC instruments in analytical laboratories. UHPLC instruments operate at high pressure (up to 1,500 bars) with small column particle size (typically <2–5  $\mu\text{m}$ ) and high flow rates (up to 5 mL/min), which allows short separation time (typically 10–15 min) with high efficiency and resolution [49]. Among LC developments in the last decade, the introduction of sub-2  $\mu\text{m}$  particles, novel monolithic columns, superficially-porous (core-shell) particles, and elevated temperatures are some of the most significant advances to meet demands for efficient separation in multi-class, multi-residue analytical methods.

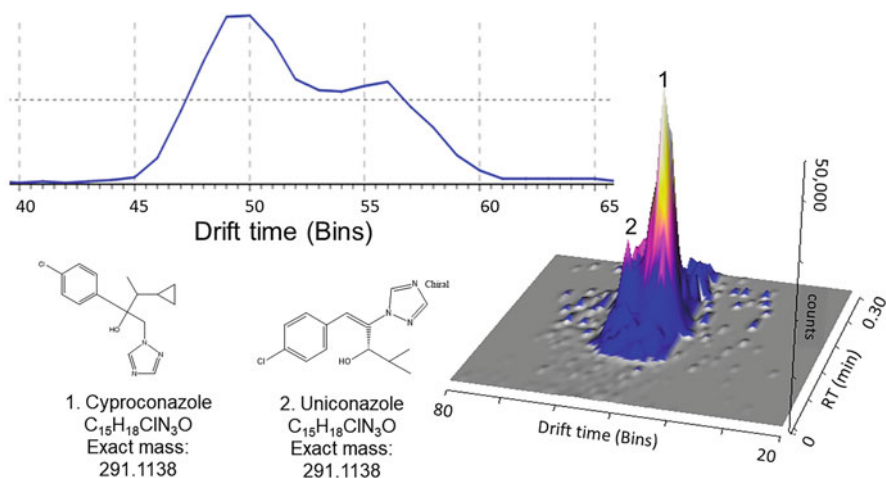


Hundreds of UHPLC-MS(MS) applications have been reported for pharmaceuticals, personal care products, pesticides, drugs, flame retardants, and many other emerging contaminants. Kachhawaha et al. reviewed recently developed LC-MS/(MS) methods for detection and quantitation of PPCPs in environmental waters [50]. UHPLC-MS/MS with ESI was the most extensively used analysis for PPCPs at trace levels (ng/L). UHPLC-MS/MS methods for brominated flame retardants in food [51] and for organophosphorus flame retardants/plasticizers in mussels [52] were reported, achieving low detection limits at sub ng/g levels. Quantitative LC-MS/MS method was reported for 295 bacterial and fungal metabolites, including mycotoxins, in food using “dilute and shoot” approach [53]. The method LOQs for all analytes were below the established regulatory tolerance levels of these contaminants.

To speed up LC analysis, flow injection (FI) coupled to MS was tested in various FI-MS/(MS) applications. FI-MS(MS) eliminates LC separation taking advantage of modern MS detectors used as a “separation” tool. Nanita and co-workers pioneered FI-MS/MS high-throughput analysis of pesticide residues in foods, biological matrices, and water [54], and Mol and van Dam developed an FI-MS/MS method for polar pesticides not amenable to multi-residue methods [55] with run time of 30–60 s. In a recent study, FI-MS/MS was evaluated for simultaneous analysis of selected pesticides and mycotoxins in food and feed samples in 2 min [56]. Despite high complexity of the samples, the method achieved LODs below the established regulatory values. The main advantages of FI coupled with MS are speed, simplicity, high throughput, and low cost, however, it can suffer from high ion suppression caused by matrix interferences in absence of chromatography. FI-MS/MS was utilized for the analysis of organophosphate esters (OPEs), used as flame retardants and plasticizers to avoid background contamination from common LC solvents [57]. Garcia-Ac et al. compared FI-MS/MS with ESI, APCI, and APPI [43] for analysis of five pharmaceuticals in wastewater and found that ESI performed the best among the three in terms of S/N ratio and peak areas.

With increasing complexity of samples and greater demand to analyze more chemicals, comprehensive two-dimensional LC (LC  $\times$  LC) is becoming a more popular and attractive approach to increase separation power with enhanced peak capacities. Selectivity can be dramatically increased when different retention mechanisms are employed for two dimensions. These reviews outlined recent developments and successful applications in LC  $\times$  LC [58, 59], however, no practical applications can be found for environmental or food safety applications, but applications in polymers, peptides, natural medicine, metabolomics [58], and food composition are described [60].

Another interesting approach to meet the challenge for complex samples is LC column backflushing. This approach helps to avoid ghost peaks, reduce matrix effects in high-throughput methods, and reduce instrumental maintenance. Michlig et al. [61] applied LC column backflushing for 3 min followed by 3 min re-equilibration between every injection in the analysis of pesticides in complex samples of hemp and hemp products.



**Fig. 3** Example of partial separation of structurally isomeric pesticides – cyproconazole and uniconazole with flow injection ion mobility TOF-MS in parent trap MS mode for  $[M + H]^+$  ion 292.112

Additionally, ion mobility (IM) spectrometry should be mentioned as an additional dimension to LC-MS, as its hyphenation to LC-MS is drawing more interest and attention recently [62]. Applications of IM for environmental analysis of perfluorocarbons, PAHs, pesticides, terpenes, chlorophenols, etc. in air, water, and soil were reviewed [63]. UHPLC-(Q)TOFMS with traveling-wave ion mobility spectrometry was evaluated for >200 pesticides in fish feed [64] and for screening of multi-class pesticides in fruits and vegetables [65]. In another study, FI-IM-TOFMS was evaluated for separation of structurally isomeric pesticides [56]. Figure 3 shows an example of partial separation of two pesticides used in agriculture, cyproconazole and uniconazole using FI-IM-TOF [56]. Both pesticides have the same formula  $C_{15}H_{18}ClN_3O$  and exact mass 291.1138, which challenges their differentiation. However, their identification and accurate measurement are extremely important in food trade as two pesticides have different regulatory values in different countries. Thus, cyproconazole has MRL = 0.05–0.2 mg/kg in the EU, and MRL = 0.05 mg/kg for corn and wheat grain in the USA, while uniconazole has MRL = 0.01 mg/kg in the EU, and no regulatory value in the USA. False positives and other incorrect results in the analysis of these or other structurally isomeric pesticides can lead to detrimental economic and health consequences, and advanced techniques such as ion mobility in this case can increase confidence of identification and reduce false findings.

## 4 Regulatory Compliance for Identification with GC-MS/(MS) and LC-MS/(MS)

In regulatory testing, compliance with the identification criteria established by governing agencies is required [66, 67]. Table 1 summarizes identification criteria for liquid and gas chromatography with various MS analyzers to meet regulatory compliance as established by some regulatory agencies worldwide in environmental and food safety. For retention time (RT) compliance, most criteria require RT within 6–12 s of RT in midpoint calibration standard in the same sequence, although the International Organization for Standardization (ISO) criteria is different, and RT tolerance highly depends on RT range [68]. Ions selected for MS should be characteristic or structurally significant, and  $\geq 3$  ions should be selected for MS, and  $\geq 2$  product ions with  $S/N \geq 3$  should be selected for MS/MS, and analyte peaks from all product ions in the extracted ion chromatogram must overlap. For ion ratios, most regulators use relative tolerance of  $\pm 30\%$ , while others use absolute tolerance, and ISO uses a formula:  $[\pm(0.1 \times I_{\text{std}} + 10)\%]$  to calculate relative tolerance, where  $I_{\text{std}}$  is a relative ion intensity for calibration standard (Table 1).

Recently, Angeles and Aga pointed out that not all official methods require ion ratio tolerance [69]. For example, the EPA methods 542 and 1694 for PPCPs and method 537.1 for selected per- and polyfluorinated alkyl substances by LC-MS/MS do not have ion ratio requirements, but methods 8270E and 8260B for semi-volatile and volatile compounds by GC-MS/MS do have ion ratio tolerance. The authors attributed this to the lack of the established guidelines, which are still developing and evolving.

In summary, regulatory compliance parameters for identification of analytes by GC- and LC-MS/(MS) are presented in Table 1. In any testing, regulatory or not, the use of these criteria should be implemented to increase the confidence of identification.

## 5 Orthogonal Applications of GC-MS and LC-MS

In an interesting study, Schürmann et al. reported a false positive finding of pesticide sebutylazine in tarragon sample [76] with LC-MS/MS analysis based on retention time, 2 MRM transitions and their ratios. A co-eluting matrix interference resulted in a false positive by producing product ions corresponding to 2 MRM transitions of the compound. However, a 3rd LC-MS/MS MRM transition and a separate analysis by orthogonal GC-MS/MS revealed the false positive findings. Many chemicals are both GC- and LC-amenable (Fig. 1) and can be routinely analyzed by both instrumental techniques. Thus, GC-MS and LC-MS can be not only complementary, but also confirmatory for these compounds. For example, pesticides and veterinary drugs in animal feed were analyzed by GC-MS/MS (192 analytes) and LC-MS/MS (187 analytes) with  $>50$  overlapping analytes analyzed by both techniques

**Table 1** Regulatory compliance with retention times and ion ratios for identification by GC-MS and LC-MS (single quadrupole, ion trap, TOF) and GC- and LC-MS/MS (MS/MS triple quadrupole, ion trap, Q-Trap, Q-TOF) with unit mass resolution

		Criteria	
Regulatory source	Guidelines for	Chromatography: retention time (RT)	Mass spectrometry: # of required ions
US EPA methods	8270E: semi-volatile organic compounds (PAHs, PCBs, pesticides) [70] 8260B: volatile organic compounds [71]	GC: RT within $\pm 10$ s of RT in midpoint calibration standard from the same sequence; or within $\pm 10$ s relative to the shift of the associated internal standard (IS) (delta RT of the IS $\pm 10$ s)	$\geq 2$ ions  MS/MS: $\pm 30\%$ (relative) of expected ion ratio in reference spectrum
EU SANTE 12682/2019 (2020) [72]	Pesticides in food and feed	GC & LC: RT $\pm 12$ s	MS: 3 ions MS/MS: $\geq 2$ product ions with S/N $\geq 3$
US FDA Office of regulatory affairs (ORA) [73]	ORA-LAB.5.4.5. (2020): methods, method verification and validation	Not specified	MS: 3 ions MS/MS: $\geq 2$ product ions with S/N $\geq 3$
USA: FDA guidance for industry 118 confirmation of identity of animal drug [74] residues (2003)	Animal drug residues	LC: RT $\leq 5\%$ GC: RT $\leq 2\%$	MS: $\geq 3$ ions  $3$ ions $\pm 10\%$ (absolute) $\geq 4$ ions: $\pm 15\%$ (absolute) 2 product ions: $\pm 10\%$ (absolute)
USDA Agricultural Marketing Service Science & Technology Pesticide Data Program (2021) [75]	Pesticides in agricultural commodities	GC & LC: RT $\pm 6$ s if an external standard is used	MS/MS: $\geq 2$ product ions  MS: 3 ions MS/MS: $\geq 2$ product ions (absolute) $\geq 3$ product ions: $\pm 20\%$ (absolute) $\pm 30\%$ (relative) when compared to the same relative

The International Organization for Standardization (ISO) 22,892:2006 (confirmed in 2016) [68]	Contaminants in soil	RRT $\pm$ 0.1 min if an internal standard is used GC: RT < 500 s: $\pm$ 1 s RRT = 500–5,000 s, $\pm$ 0.2% RT > 5,000 s, $\pm$ 6 s	product ions with $S/N \geq 3$ MS: 3 ions MS/MS: $\geq 2$ Product ions	abundances observed from a standard solution injection made during the same analytical run $< \pm(0.1 \times I_{std} + 10)\%$ (relative)
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RRT retention time, RRT relative retention time, S/N signal to noise,  $I_{std}$  relative ion intensity for calibration standard

[1]. Similarly, 302 contaminants, including pesticides, PAHs, PCBs, PBDEs were analyzed in catfish, with 128 and 219 by UHPLC-MS/MS and GC-MS/MS, respectively, and 45 overlapping contaminants were analyzed by both. Orthogonal applications of GC-MS and LC-MS provide an additional degree of confidence by employing different chromatographic mechanisms and different ionization modes and are especially important when analyzing complex samples where matrix interferences can result in false negatives or false positives.

## 6 Future Trends

Evolving challenges to analyze more contaminants faster, more efficiently, with low detection limits and to reduce false positive/false negative rates in complex samples place high demands on future improvements in LC- and GC-MS/(MS) instrumentation and techniques.

Just like in the past, improvements in MS detectors' sensitivity, speed, selectivity, specificity, and wider dynamic linear range are expected. We may see new features aimed to reduce instrumental downtime needed for maintenance, and new and improved ionization sources and interfaces, plus combinations of existing ion sources used simultaneously.

With continually improving modern powerful LC- and GC-/MS(MS) instruments, 1,000 contaminants from different classes with various properties may be covered in one analytical method providing a wide scope of analysis. While triple quadrupole instruments have been "workhorses" in analytical laboratories in the last decade, a transition to high-resolution MS systems will likely occur to avoid shortcomings associated with QqQ instruments in terms of developing ion transitions, their optimization, limited amount of transitions due to the scanning speed of QqQ. Kaufmann et al. conducted direct comprehensive comparison of LC-MS/MS and LC-HRMS [77], and found that selectivity of LC-HRMS at 50,000 FWHM was superior to LC-MS/MS. Analysis with modern HRMS instruments allows virtually unlimited amount of analytes based on the full scan data, plus retrospective analysis. Recent improvements in sensitivity of HRMS instruments put them on par with modern MS/MS.

Other trends expected and desired by the analytical community in the future are software improvements that make it easier for analytical chemists to deal with an enormous amount of generated data. Software packages with streamlined and flexible workflows for different types of applications are expected.

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# Chromatography High-Resolution Mass Spectrometry in Food and Environmental Chemistry



Yong-Lai Feng

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**Abstract** In the past 20 years, the evolution of mass spectrometry has undergone tremendous technological advances to improve mass resolution, accuracy, and sensitivity, and brought commercially practical high-resolution mass spectrometers that can measure a mass-to-charge ratio ( $m/z$ ) at the fourth or fifth decimal place (exact mass). Advanced high-resolution mass spectrometry (HRMS) instruments now feature fast scan speeds, sufficient dynamic range, and MS/MS capability in hybrid instruments, which facilitate the exploration of a large number of known and unknown chemicals in modern environmental analysis and food analysis. This chapter introduces the basic concepts of resolution, resolving power, exact mass, accurate mass, mass error, and mass measurement precision. In this chapter, two main modern high-resolution mass spectrometers, namely quadrupole time-of-flight mass spectrometer (Q-TOF) and quadrupole Orbitrap (Q-Orbitrap) mass spectrometer are introduced. The advantages of HRMS combined with gas chromatography and liquid chromatography in modern environmental and food analysis are discussed. This chapter also provides an overview of recent developments in extensive screening and non-targeted analysis (NTA) using state-of-the-art HRMS in food and environmental analysis. The NTA data acquisition modes and data analysis process and applications of HRMS-based non-targeted analysis approaches in the analysis of organic contaminants in foods for food quality and safety, are introduced, including lipidomics, proteomics, and metabolomics, as well as environmental samples for environmental safety.

**Keywords** Chromatography, Environmental analysis, Food analysis, High-resolution mass spectrometry, Non-targeted analysis

## 1 Introduction

Mass spectrometry (MS) is an analytical technique for measuring the mass-to-charge ratio of ions. Finnigan introduced the first commercial quadrupole mass spectrometer to the market in 1968. Although there has been a number of developments in quadrupole mass spectrometry technology, the resolution of quadrupole mass spectrometers is still less than 5,000. Therefore, the quadrupole mass spectrometers are generally considered low-resolution mass spectrometers (LRMS). However, LRMS is limited to measuring single-digit mass units (integer mass), and cannot separate the isobars that have the same integer mass. It is also unable to separate a large number of chemical mixtures in environmental and food samples, which lags far behind the requirements for foodomics and metabolomics analysis and non-targeted analysis (NTA) of environmental and food samples. Although tandem mass spectrometer working in selected reaction-monitoring (SRM) mode improves the mass selectivity and sensitivity and is the most widespread technique in both environmental and food analysis, the use of SRM mode on tandem MS is limited to the number of compounds monitored per analysis, and it cannot screen for unknowns, and needs reference standards.

In the past 20 years, the evolution of mass spectrometry has undergone tremendous technological advances to improve mass resolution, accuracy, and sensitivity, and brought commercially practical high-resolution mass spectrometers that can measure a mass-to-charge ratio ( $m/z$ ) at the fourth or fifth decimal place (exact mass). Advanced high-resolution mass spectrometry (HRMS) instruments now have fast scan speeds, sufficient dynamic range, and the capability of MS/MS in hybrid instruments. The main advantage of HRMS instruments is that they can record a theoretically unlimited number of compounds in full scan mode, which allows the development of screening strategies based on accurate mass database searches to identify non-targeted compounds through retrospective data analysis. Hybrid HRMS instruments (MS/MS) can provide structural information and can detect ultra-trace levels of thousands of chemicals in the NTA approach. The advantages of these aspects combined with time dimension separation (chromatography) greatly benefit the modern environmental and food analysis. This chapter will introduce the basic concepts of resolution, resolving power, exact mass, accurate mass, mass error, and mass measurement precision. This chapter mainly introduces two main modern high-resolution mass spectrometers, namely quadrupole time-of-flight mass spectrometer (Q-TOF) and quadrupole Orbitrap (Q-Orbitrap) mass spectrometer. The possibility of using HRMS for extensive screening and NTA in food and environmental analysis will be outlined, and the NTA data acquisition modes and data analysis process are introduced. This chapter will also introduce some applications of HRMS-based non-targeted analysis approaches in the analysis of organic contaminants and toxins in foods for food quality and safety, including lipidomics, proteomics and metabolomics, as well as environmental samples for environmental safety.

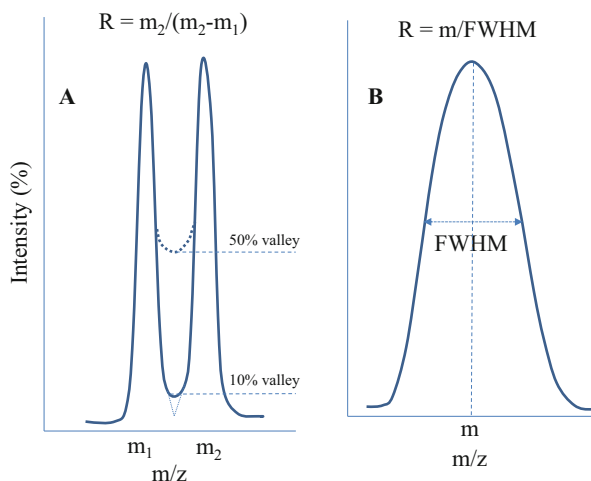
## 2 High-Resolution Mass Spectrometry (HRMS)

In mass spectrometry, the nominal mass is defined as a mass-to-charge ratio ( $m/z$ ) measured or calculated only as an integer. Before modern HRMS appeared on the market, the quadrupole mass spectrometer was considered a nominal mass spectrometer. Due to its low resolution in mass separation, it can only provide information about the nominal mass of molecules or fragments. In contrast, HRMS is a type of mass spectrometry in which the mass-to-charge ratio ( $m/z$ ) of each ion is measured to several decimal places, and the accurate mass is measured instead of the nominal mass on a conventional low-resolution mass spectrometer (LRMS). HRMS is particularly useful for distinguishing molecular formulas with the same nominal mass and is able to provide the more probable empirical formulas.

## 2.1 Mass Resolution and Mass Resolving Power

In mass spectrometry, resolution is a measurement of the ability to distinguish two ions with slightly different  $m/z$  ratios as shown in Fig. 1a. If the valley between two mass spectral peaks with the same height and width is a specified fraction of the height of the smaller peak in a mass spectrum, 10% can obtain the maximum separation, and 50% can get the minimum separation. The IUPAC definition for resolution ( $R$ ) in mass spectrometry is calculated by the ratio of the mass of the second peak ( $m_2$ ) to the difference of two adjacent masses ( $m_2 - m_1$  or  $\Delta M$ ) (Fig. 1a). The mass difference of two adjacent masses ( $m_2 - m_1$  or  $\Delta M$ ) is called the resolving power [1, 2]. However, the mass resolving power at a certain  $m/z$  value is also commonly defined as the ratio of  $m_2/(m_2 - m_1)$  [3, 4]. For two peaks of the same height,  $m_2 - m_1 \approx \Delta m_{50\%}$ , where  $\Delta m_{50\%}$  is the full width at half maximum (FWHM) of a single peak, and the mass resolving power can be calculated by  $m_2/\Delta m_{50\%}$ . For an isolated mass peak  $m$ , the mass resolving power is  $m/\text{FWHM}$  as shown in Fig. 1b. Sometimes, the resolving power ( $R$ ) is simply defined as the value of FWHM at a specific  $m/z$  (usually at  $m/z$  400). Therefore, when reporting values, it is important to clarify the method that is used to determine mass resolution. A higher value for resolution usually indicates a better separation of peaks in mass spectrum. In general, a mass spectrometer with resolving power over 10,000 at  $m/z$  400 and a mass accuracy less than 5 ppm is defined as a high-resolution mass spectrometer. These quadrupole mass filters, triple quadrupole, and quadrupole ion trap mass analyzers, are generally not classified as high-resolution mass analyzers. Therefore, the HRMS is more selective than LRMS.

**Fig. 1** The schematic diagram of mass resolution, (a) The resolution between two adjacent ion  $m/z$  peaks, (b) The resolution for an isolated ion  $m/z$  peak



## 2.2 *Exact Mass, Accurate Mass, Mass Accuracy, and Mass Measurement Precision*

The exact mass, also known as the theoretical  $m/z$  value, is the mass calculated from the molecular formula using the known mass of a specific isotope (usually the most abundant), giving the monoisotopic peak, with the appropriate number of decimal places. For example, the weights of  $^{12}\text{C}$ ,  $^{16}\text{O}$ ,  $^{14}\text{N}$ , and  $^1\text{H}$  are 12.00000 amu, 15.9949 amu, 14.0031 amu, and 1.00783 amu, respectively. As a result, the exact mass or the theoretical mass of 2-octanone,  $\text{C}_8\text{H}_{16}\text{O}$  is 128.12018, and the exact mass or the theoretical mass of naphthalene,  $\text{C}_{10}\text{H}_8$  is 128.06264. In contrast, accurate mass is a measurement of ion mass within a specified error, and mass accuracy is a measurement of mass error. The mass error is defined as the ratio of the difference between the measured and the theoretical  $m/z$  to the theoretical  $m/z$  value, expressing in parts per million (ppm). When measuring the mass of an ion, mass measurement precision is to define the repeatability of the measurement, calculating as  $m/\delta m$ , in which  $\delta m$  is the root-mean-square error from a large number of repeated measurements [5]. In the absence of systematic error, mass accuracy is the same as mass precision.

The accurate mass measured by HRMS is useful for the assignment of candidate molecular formulas of compounds with elemental compositions that are measured as an identical mass by LRMS. For example, on HRMS the accurate masses of the protonated 2-octanone,  $\text{C}_8\text{H}_{16}\text{O}$ , and the protonated naphthalene,  $\text{C}_{10}\text{H}_8$ , are measured as  $m/z$  129.1274, and  $m/z$  129.0699 with a mass error of 5 ppm, respectively. However, the measured masses are the same as  $m/z$  129.1 on a regular low-resolution instrument. Therefore, the accurate mass can provide an exact molecular formula assignment. As the number of atoms in a compound increases, the number of possible elemental compositions in a given experimental mass measurement also increases. As a result, higher resolving power is required to resolve closely spaced mass doublets in complex mixtures for accurate formula assignment, as the resolving power influences the measurement of accurate mass. If two ion peaks are not (or incompletely) resolved, the resulting measured mass profile is the sum of the two individual mass profiles, giving the top of the combined profile locating at somewhere between the exact masses of the two individual peaks thereby impacting the accuracy on the assignment of each mass and thereby impacting the accuracy of formula assignment. In real sample analysis, the sample matrices also affect the accurate mass measurement. Generally, a minimum resolving power of 50,000 and 2 ppm mass accuracy are usually required to provide a consistent and reliable molecular mass assignment of the analyte in complex matrices, but it depends on the complexity of the sample and the concentration levels of analytes and interferences in the sample [6]. It is also worth noting that measuring exact mass can only distinguish molecules with different molecular formulas but the same nominal mass. HRMS alone cannot be used to distinguish isomers.

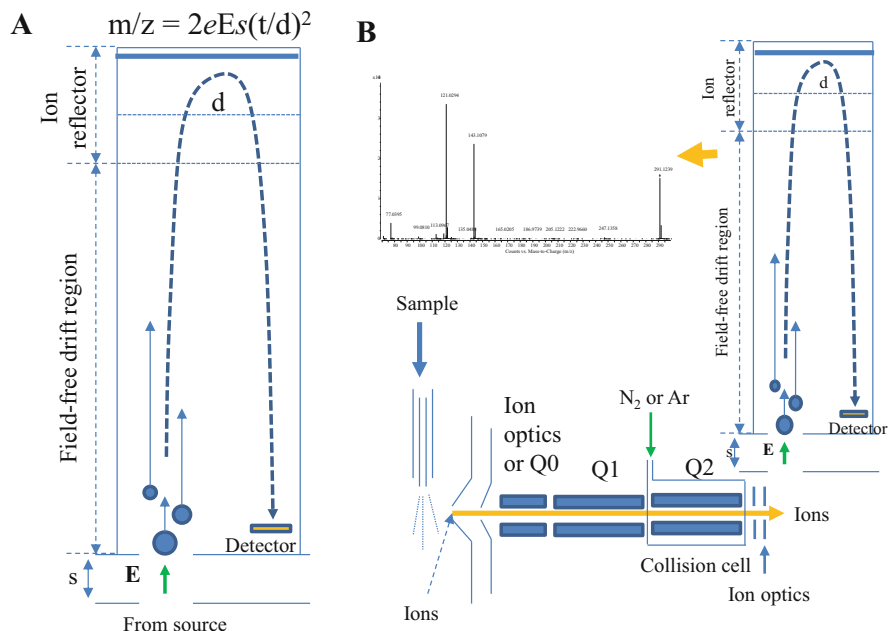


### 3 Types of High-Resolution Mass Spectrometer

The classic high-resolution mass spectrometers include double-focus sector or Fourier-transform ion cyclotron resonance (FT-ICR), which can provide a high resolution of over 500,000 to 1,000,000. However, disadvantages include slow scanning speed, complex processing, and high purchase and maintenance costs. With the advancement of mass spectrometry technology, new modern high-resolution mass spectrometers including time-of-flight (TOF) MS and Orbitrap MS have become more and more common in laboratories. The advantages of these modern mass spectrometers include fast scanning speed, sufficient dynamic mass range, and the possibility of MS/MS when part of a hybrid instrument, as well as greatly improved mass resolution, accuracy and sensitivity, and cost affordable in most laboratories. Meanwhile, because these modern MSs have a better signal-to-noise ratio in scanning mode, they are becoming increasingly popular.

#### 3.1 *Quadrupole Time-of-Flight Mass Analyzers*

Time-of-flight mass spectrometry (TOF-MS) was first proposed by Stephan in 1946 [7]. This is a method of mass spectrometry in which the  $m/z$  of ions is determined by the measurement of their flight times, which is performed by accelerating the ions with an electric field of known strength. In the TOF-MS shown in Fig. 2a, ions having the same charge are given the same kinetic energy. Under the same kinetic energy, the velocity of the ions will depend on the  $m/z$ . Heavier ions with the same charge (larger  $m/z$ ) fly at a lower speed, and lighter ions with the same charge (smaller  $m/z$ ) fly at a higher speed. Since the flight time depends on the velocity of the ion within the same flight distance ( $d$ ), the time required for the ion to reach the detector is measured as its  $m/z$ , as shown in Fig. 2a. In this type of mass spectrometer, masses are separated by the flight time of ions to the detector. Ions exhibiting a lighter mass will have a shorter flight time, while heavier ions will take longer to cross the flight path to the detector. The resolving power of TOF mass analyzer is  $m/\Delta m50\% = (T/2\Delta t)$ , where  $T$  is the total flight time of the ion and  $\Delta t$  is the mass spectral peak width. Modern time-of-flight analyzers also utilize a reflectron device introduced by Mamyrin in 1973 [8], which serves to correct for kinetic energy dispersion and spatial spread of ions with the same  $m/z$  but have varying velocities. In the reflectron system, faster ions penetrate farther into the reflectron than slower ions and travel more distance to reach the detector (Fig. 2a). This reflectron correction allows ions of the same  $m/z$  to reach the detector at the same time. The reflectron device increases the length of the ion flight path without defocusing due to the spread of the initial velocity, and thereby mass resolving power. Recent advances in hybrid mass spectrometry systems combine quadrupole technology (for mass selection and/or ion fragmentation) with a time-of-flight mass analyzer for high-resolution mass measurement and further increase the selectivity, improving the applications of TOF mass analyzer. Figure 2b shows that the Q-TOF-MS instrumentation resembles



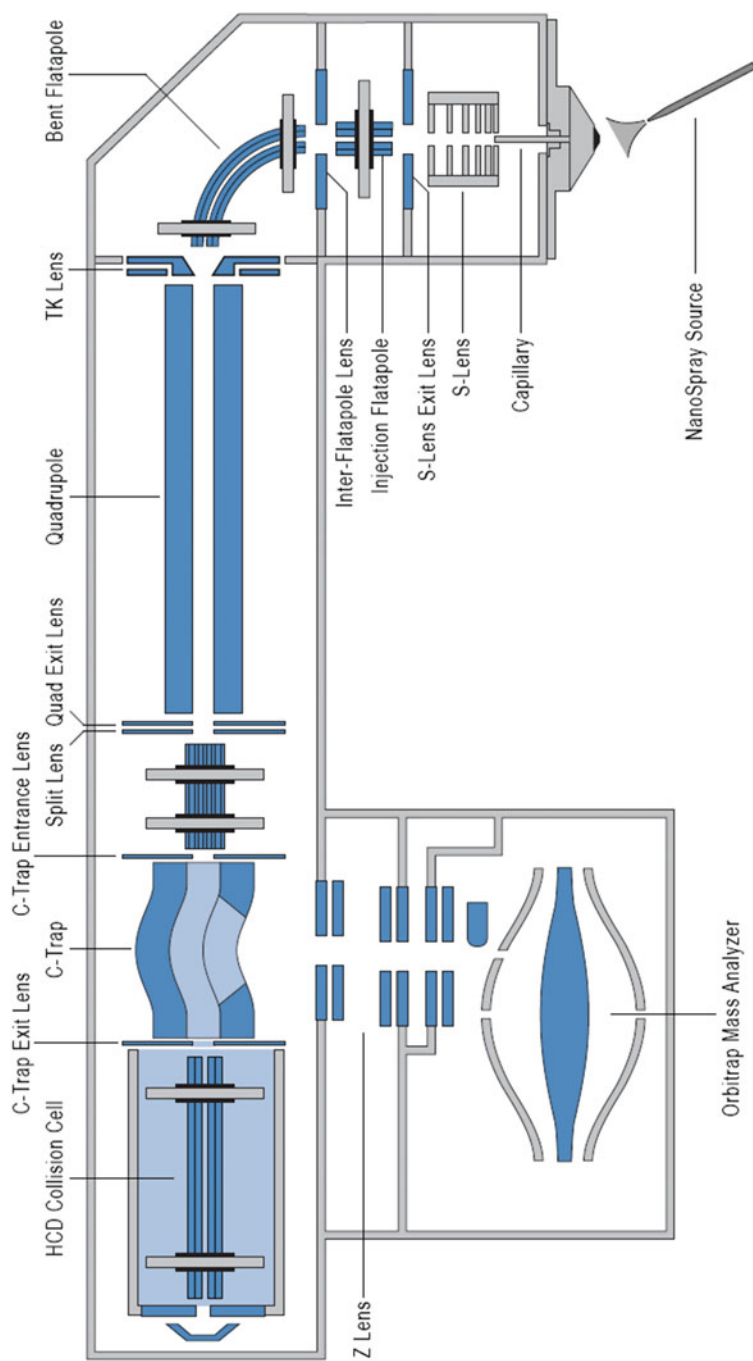
**Fig. 2** Schematic diagram of time-of-flight mass spectrometer (a) and hybrid quadrupole time-of-flight mass spectrometer (b): E, electric field strength; d, distance of flight; t, time of flight; s, the length of the source region; m, mass; z, number of electron charges; e, charge of electron; Q0, quadrupole zero; Q1, quadrupole 1; Q2, quadrupole 2; N<sub>2</sub>, nitrogen gas; Ar, argon gas

two quadrupoles and a time-of-flight tube. The first quadrupole (Q1) can function as a mass filter to select specific ions based on their  $m/z$ . In radio frequency (RF) only mode, the Q1 quadrupole let all ions be transmitted through the quadrupole. The second quadrupole (Q2) acts as a collision cell, where neutral gas molecules such as nitrogen or argon bombard the ions from Q1 to fragment the ions to produce product ions by collision-induced dissociation (CID). In the RF-only mode, all ions from the Q1 pass through the Q2 quadrupole without subsequent fragmentation of the ions. After leaving the Q2, the ions are subjected to an electric field pulse and accelerated orthogonally to their original direction into the ion modulator region of the TOF analyzer for separation and measurement, as shown in Fig. 2b. The resolution of modern hybrid Q-TOF mass analyzers has been significantly improved. For example, the Bruker maXis II™ and maXis II™ ETD QTOF systems can provide a Full-Sensitivity Resolution (FSR) of 80 K, while the Bruker Impact II can provide an FSR of 50 K; Agilent 6546 LC/Q-TOF can provide a resolution of 60 K (for high masses ( $\geq 1,000$  Da)) and a resolution of over 30 K (for low masses). The advantages of TOF mass analyzers include no upper  $m/z$  limit in theory (or larger dynamic range) and faster response/scan rate than FTICR mass analyzers. The former makes it be useful for identifying singly charged ions of high molecular weight, and the later can be used in applications that require shorter acquisition times, such as in combination

with fast chromatographic separation techniques, for example, ultra-high performance liquid chromatography (UHPLC) or gas chromatography (GC).

### 3.2 *Quadrupole Orbitrap Mass Analyzers*

Orbitrap is a type of Fourier transform mass analyzers, which was invented in 1999 [9] and has been widely distributed since it was put into commercial sales in 2004 [10]. Compared with TOF mass analyzers, Orbitrap has much higher mass resolving power and mass measurement accuracy. In terms of working principle, the Orbitrap is an ion trap mass analyzer, which is different from the traditional Fourier transform ion cyclotron resonance (FTICR). The Orbitrap mass analyzer consists of an outer barrel-like electrode and a coaxial inner spindle-like electrode. The latter traps ions in an orbital motion around the spindle and acts as an analyzer and detector at the same time [9, 10]. Since the electrostatic attraction of the ions to the inner electrode is balanced by their inertia, the ions are trapped and circulate around the inner electrode on elliptical trajectory, and move back and forth along the axis of the central electrode, so their trajectories in space resemble helices. Using the Fourier transform of the oscillation frequency signals, the image current from the ions trapped on the outer electrodes is detected and converted into a mass spectrum. With the introduction of the C-trap, different operation modes of Orbitrap have been developed. The C-trap can be used to accept multiple fills and allow calibrant ions and analyte ions to be injected into the Orbitrap analyzer at the same time, and measured in a single spectrum to achieve an internal calibration of each spectrum, which is significantly different from the TOF mass analyzer. In addition, the C-trap also allows multiple injections of ions that are fragmented or selected under different conditions to be stored together and acquired in a single Orbitrap spectrum. The advantages of the C-trap allow the use of higher-energy collision dissociation (HCD) and quadrupole mass filters in combination. Based on this working mechanism, several hybrid Orbitrap mass spectrometers have been developed, including Q Exactive, Orbitrap Fusion, and Orbitrap ID-X Tribrid mass spectrometers. The development of modern hybrid Orbitrap mass spectrometers and their working principles have been comprehensively summarized by Hecht et al. [11]. Figure 3 shows a typical example of a quadrupole Orbitrap hybrid mass spectrometer, Q Exactive MS, which is a combination of a quadrupole mass filter and an Orbitrap analyzer. The Q Exactive instrument incorporates the development of new technologies such as S lens, eFT, extended mass range, and fast polarity switching. In addition to improving selectivity, the quadrupole mass filter also enables multiple scan modes such as a multiple single ion monitoring (SIM) mode to rapidly switch between different narrow mass ranges, which allows ions of selected masses to enter the C-trap and to sequentially accumulate them up to a mixed population of the desired ions, and achieving a joint analysis of ions in a single Orbitrap detection. Similarly, Q Exactive MS also allows fragmentation of different precursor  $m/z$  in rapid succession while capturing and retaining the resulting fragment populations all



**Fig. 3** Schematic layout of quadrupole-Orbitrap hybrid mass spectrometer, the Q Exactive (Reproduced with permission of Thermo Fisher Scientific)

together in the HCD collision cell, followed by their joint analysis in the Orbitrap analyzer. Compared with QTOF, Orbitrap can simultaneously detect ions in a wide  $m/z$  range, and provide higher resolution and more accurate mass measurement without reducing sensitivity when using higher resolutions. The new Orbitrap ID XTM Tribrid™ MS system can provide a resolution up to 500 K, which can provide the fine structure of isotopes and unravel molecular formulas of true unknowns, and completely separate nominally isobaric compounds. In addition to improving mass resolution, the new Orbitrap design can also perform deep scans to automatically generate inclusion and exclusion lists for further MS/MS fragment scan, which is very useful for non-targeted screening analyses and identification of unknowns in metabolomics study. Another advantage of Orbitrap is the capability of reliable quantitative purposes. However, compared with the TOF mass analyzer, the resolution of Orbitrap is proportional to the measurement time, which may impact the data points when using ultrahigh performance liquid chromatography (UHPLC) equipment or attempting polarity-switching experiments.

## **4 Chromatography-High-Resolution Mass Spectrometry for Screening and Non-target Analysis**

### **4.1 Chromatography-High-Resolution Mass Spectrometry**

One of the main advantages of HRMS is that it can provide accurate mass measurement and high resolution for separating two adjacent ions with slightly different  $m/z$  ratios, such as nominal isobaric compounds in a mass spectrum. The accurate mass can better predict elemental compositions in the molecular formula assignment. However, the high resolving power cannot separate molecular isomers or polyatomic ions with the same molecular formula, including constitutional isomers, in which bonds between the atoms differ, and stereoisomers, in which the bonds are the same but the relative positions of the atoms differ. As these isomers have the same number of atoms of each element, their accurate masses are the same in mass spectrum. Although they have the same formula, these isomers may not share similar chemical or physical properties. Therefore, in addition to the  $m/z$  dimension separation, retention time dimensional separation can increase the power to separate isomers. Chromatographic techniques can perform time-dimensional separation to separate these isomers prior to HRMS analysis. Gas chromatography (GC) and liquid chromatography (LC) are the two main time-dimensional separation techniques used in conjunction to HRMS.

Gas chromatography (GC) is a type of chromatography used to separate and analyze compounds that can evaporate without being decomposed. It is also sometimes known as vapor-phase chromatography (VPC), or gas-liquid partition chromatography (GLPC). In gas chromatography, when the compound mixture passes through the stationary phase, the separation is performed by the compound mixture

being distributed on the gas chromatographic column with a gradient temperature between the mobile phase (carrier gas) and stationary phase which is a microscopic layer of viscous liquid on a surface of solid particles. GC is usually applicable to the separation of volatile and semi-volatile small molecular compounds.

Liquid chromatography (LC) is a separation technique in which the mobile phase is a liquid, where sample ions or molecules are dissolved. The sample with the mobile liquid will pass through the column packing with a stationary phase composed of irregularly or spherically shaped particles. LC, namely high-performance liquid chromatography (HPLC) or ultra-high performance liquid chromatography (UHPLC), is the most common chromatographic technique hyphenated to HRMS to be used in environmental and food analysis due to its applicable to a more wide range of chemicals from small to large molecules. Capillary electrophoresis (CE) is another liquid separation technique. In CE, compounds are separated according to their ion mobility and/or partitioning into another phase through non-covalent interactions. Compared to LC, CE theoretically has higher separation ability. However, in the past, it was not commonly hyphenated to HRMS.

Chromatographic technique can not only help separate isomers but also help separate high similar structures or isobars, thereby increasing the separation of analytes in HRMS measurement, thereby achieving accurate molecular assignment. In addition, chromatographic technique can also reduce the influence of matrices in sample on ionization efficiency, especially in electrospray ionization (ESI) mode which is commonly used in LC-HRMS systems for non-targeted analysis.

## ***4.2 Non-targeted Analysis with High-Resolution Mass Spectrometry***

The widespread use of synthetic chemicals in industrial applications has led to food and environmental pollution, as well as public exposure to those chemicals from various sources through foods and the environment. Monitoring the targeted contaminants in food, environmental, and human samples is often used as an approach to assess food and environmental pollution and human exposure to these chemicals. Although the use of targeted monitoring methods can achieve high sensitivity, targeted monitoring only evaluates human exposure to a small portion of the contaminants in food and the environment. In addition to legacy contaminants, hundreds of new chemicals are produced in industry every year. These substances enter the environment where they coexist with hundreds of their transformation products as well as with those of legacy contaminants. This high number of potential compounds pose a challenge to the targeted analysis. Therefore, it is emerging to use non-targeted analysis approaches to explore and elucidate the structures of these unknown chemicals in food and the environment for pollution assessment and human exposure assessment [12, 13].

Compared with conventional low-resolution quadrupole mass spectrometry, HRMS can provide high resolution and accurate mass that can help identify targeted species, narrow down possible chemical formulas, and elucidate unknown structures, which makes it a power tool for non-target screening analysis. A particular advantage of HRMS is the ability to acquire accurate mass spectrometry data of all precursor and fragment ions in the sample in data-independent acquisition (DIA) mode. This enables confident identification of “expected components” through comparison with library data and retrospective investigation of DIA data to detect and identify new unknown compounds that standards are not commercially available or are difficult to obtain. Generally, measurement of the intact exact mass of molecular ions can be obtained from a full scan without collision energy. The full scan with collision energies can be used as the complimentary information of exact mass to prioritize and identify molecular ions through the intensity decrease trend of the candidate ion and intensity increase trend of potential corresponding product ions with increasing of collision energy. A targeted MS/MS scan is usually performed on each prioritized candidate ion to elucidate their structures [14]. Another strength is to conduct semi-targeted analysis through combination of a list of targeted compounds in non-targeted analysis. Non-targeted analysis presents the best coverage of compounds in sample but low reproducibility and substantial data processing. Semi-targeted analysis can represent a mid-way strategy enabling a good compromise between compound coverage, reliability, effort in method construction and data treatments. Although the use of HRMS-based non-targeted analysis (NTA) approaches to identify unknown xenobiotics in food and environmental samples is still a new analytical frontier, the workflow usually includes sample collection, sample preparation, data acquisition, data mining/processing, and reporting. Due to the scope of this chapter, data acquisition, data mining, formula assignment, and unknown identification will be introduced.

### 4.2.1 Data Acquisition

There are two data acquisition modes in NTA: data-independent acquisition (DIA) and data-dependent acquisition (DDA).

The data-independent acquisition (DIA) is a full-scan  $MS^2$  acquisition mode of molecular structure determination. In DIA mode, all ions within a selected  $m/z$  range are fragmented and analyzed in a second stage of tandem mass spectrometry. There are two common DIA modes: all-ion MS/MS ( $MS^{All}$  or  $MS^E$ ) scan mode and sequential window acquisition of all theoretical fragment-ion spectra (SWATH) mode. In the former mode, all ions in the scanned mass window are subjected to fragmentation under various collision energies. The DIA data under zero collision energy provides the intact exact mass information. The change trend of candidate ion intensity with collision energy also helps to identify candidate molecular ions. DIA is non-targeted, does not require any prior knowledge of chemicals, and can be used for truly unknown analysis. Due to non-specific manner without prior knowledge, DIA  $MS^{All}$   $MS^2$  acquisition has higher hit rate than data dependant acquisition

(DDA) mode. However, since all ions in the mass range are fragmented, the product ions from other co-eluting components will reduce the quality of MS<sup>2</sup> spectra. This makes the structure elucidation and assignment challenging. In order to improve the quality of MS<sup>2</sup> spectra in DIA MS<sup>All</sup> mode, the new DIA mode SWATH was introduced for the first time in non-targeted analysis in 2012 [15]. In this mode, a small mass window, for example, a 20 Da window, is used to reduce the product ions from co-eluting components and matrix ions. In this way, the DIA SWATH mode can produce higher quality of MS<sup>2</sup> spectra than the DIA MS<sup>All</sup> mode, while maintaining the same MS<sup>2</sup> acquisition hit rate [16], which is more suitable for a metabolite identification study [17]. The MS<sup>2</sup> spectra in SWATH mode can also perform library search on any precursor and structural analog search based on product fragment by using the chromatographic profile as well. Compared to DDA, this way can identify more compounds at lower concentration levels [18]. The richness of SWATH data allows the use of a variety of data analysis methods, including detection of metabolites through prediction and mass defect filtering, and quantification from high-resolution MS precursor ion chromatograms or fragment chromatograms [19]. However, more effort is needed to define standard operating procedures (SOPs) for DIA MS<sup>All</sup> and DIA SWATH acquisition methods, which can be routinely applied to the identification of unknown compounds in NTA [20, 21].

In contrast, the data-dependent acquisition (DDA) is a product ion scan mode. In this mode, the mass spectrometer automatically selects the precursor ions that meet the predefined peak intensity or ion abundance in the given  $m/z$  range in the first stage of tandem MS (MS1) and then performs MS/MS scan for the selected precursor ions in the second stage of tandem MS (MS2). In DDA mode, data is acquired in a manner of automatically switching between MS full scan and MS/MS product ion scan. Since only these precursor ions with a higher intensity than the predefined intensity are selected, the number of fragmented precursor ions is smaller, which reduces the MS/MS (MS<sup>2</sup>) background and improves the quality of MS<sup>2</sup> spectrum. However, in the analysis of complex mixtures, the ionization suppression of co-eluting endogenous materials and competition with a large number of matrix peaks can interfere with the precursor ions that trigger MS<sup>2</sup> acquisition, resulting in a low MS<sup>2</sup> spectrum hit rate. Therefore, a two-injection workflow is often used to improve the MS<sup>2</sup> acquisition hit rate. In this way, the first injection is used to determine potential compounds, and the second injection is used to conduct targeted MS<sup>2</sup> acquisition to obtain the MS<sup>2</sup> spectra of all these candidates. The advantage of this strategy is that it can increase the MS<sup>2</sup> acquisition hit rate to 100% while maintaining high quality of MS<sup>2</sup> spectra. The disadvantage of this strategy is that the second injection requires a separate method, which includes the target list from the result of first injection, so it is time-consuming. Another disadvantage of DDA is that the stochastic peak selection based on the peak intensity is not reproducible in precursor selection process, which results in quantitation from precursor ions less specific than using fragment ions [22]. Therefore, more effort is needed to prevent the fragmented ions from being reselected into different scans to increase the number of precursors selected for fragmentation and reduce concentration deviations.



### 4.2.2 Data Mining/Processing

After data acquisition, data analysis and data mining usually start with molecular feature extraction to prioritize peaks of interest. This process includes peak detection, peak alignment, and peak selection.

Structure-based information such as specific fragment ions, neutral losses, isotopes, and mass differences can also be used to prioritize molecular ion peaks, identify candidate ions, and assign molecular formulas to certain chemical groups. Generally, specific fragment ions related to the structure of certain chemical groups are first often used to perform extracted ion chromatogram (EIC) or extracted mass chromatogram (EMC) to prioritize molecular ion peaks of the chemical groups. Other molecular features, such as mass defects, isotope patterns, mass accuracy, neutral losses, and ion intensities are usually used to further filter false positive candidates in the first step of EIC or EMC extraction.

The adduct ions, such as  $[M + H]^+$ ,  $[M + Na]^+$ ,  $[M + NH_4]^+$  in a positive ESI mode, and  $[M - H]^-$  and  $[M + HCOOH - H]^-$  in a negative ESI mode, are useful filters for assigning candidate molecular ions. When using adduct ions to assign candidate ions, spectral annotation tools, such as findMAIN, is usually used [23]. Mass defect filtering (MDF) is an effective approach for interpreting and extracting chemical information from large mass spectral data sets. It is a useful tool in non-targeted screening analysis, especially in the screening of unknown compounds in homologs [24, 25]. A defined mass defect window centered at the mass defect value of the precursor compound ions can be used to detect common and uncommon metabolites.

In addition, it is necessary to perform peak alignment in different samples to adjust for minor variations in mass and retention times and to improve identification confidence for tentative candidate ions. If the retention time information in databases is not available, the retention time can be compared with the retention times predicted by quantitative structure-retention relationship (QSRR) based or logP-based models. In case-control studies, the change or difference in mass profile between the control group and treatment group is a very useful criterion for determining the priority of ion peaks of interest for further study [26].

### 4.2.3 Molecular Formula Assignment and Structure Annotation, and Identification Confidence Levels

In data analysis, the formula assignment is considered to be the identification confidence Level 4 according to Schymanski's definition [27]. The high resolution can improve the separation of isotopic fine structures, thereby improving the reliability of elemental composition assignment in assigning formulas. However, it is difficult to answer how much mass resolving power is enough for an accurate formula assignment. When the molecular mass is less than 300 Da and the molecular formula contains only C, H, N, and O, an accurate mass with a resolution of 10 K can

usually provide a clear assignment of elemental composition [28]. However, when the molecular mass exceeds 300 Da, even with a mass error of 0.1 ppm, the assignment of elemental composition becomes more and more difficult [28]. When the number of elements in the composition increases, such as C, H, N, S, O, P, and potentially F, Cl, Br, and Si, more than 1.6 million molecular formulas in the range 0–500 Da can be possibly generated for one accurate mass. Therefore, the high mass accuracy alone is not enough to exclude enough formula candidates in assigning formulas. In general, molecular formula assignment is not a simple process in non-targeted screening analysis. This step usually uses a combination of heuristic rules including the Seven Golden Rules [29],  $MS^n$  fragment interpretation, neutral loss analysis, isotope pattern matching, and rings plus double bonds equivalent (RDBE) for a more accurate assignment [30].

Tandem MS ( $MS/MS$  and  $MS^n$ ) fragmentation is usually used to elucidate the elemental structure information after formula assignment for possible structure(s). In this step, the  $MS/MS$  spectra of the candidates are compared with spectra from databases such as MetFrag or predicted spectra using prediction tools to achieve high match score with the library spectrum to enhance the identification confidence. The retention index that is to best fit the experimental information can also enhance the identification confidence. The probable structures are usually those that have not yet been confirmed by standard or available literature information, but they are named as the best candidate structures using library spectrum matching or predicted spectrum matching. To confirm the probable structures, reference standards are needed to compare MS,  $MS/MS$ , and retention time matching with the probable structures. In the absence of an authentic reference standard, the comparison of the exact masses with online structural information can be the alternative way for confirmation.

## 5 Application of High-Resolution Mass Spectrometry in Food Analysis

Thousands of chemicals, their metabolites, and transformation product from food processing have the potential to contaminate our food. Therefore, monitoring those contaminants in food is important to maintain food safety and protect our health.

### 5.1 Food Lipidomics

Foodomics is a combination of advanced analytical techniques and bioinformatics to address different questions in food science and nutrition, and improve the well-being, health, and confidence of consumers, including all omics in food analysis (genomics, proteomics, lipidomics, glycomics, peptidomics, and metabolomics) [31]. It requires reliable qualitative and quantitative information about food

contamination and food proteome components to extract new integrative information from the complex multivariable space of omics. This new information will improve understanding of processes in food science and technology, consequently new functions of food, markers of food quality and safety, and completely transform the concept of food safety. The use of high-resolution mass spectrometry (HRMS) can greatly improve analytical performance and provide a good combination of selectivity and sensitivity, which can trace the chemical fingerprint of food and define molecular parameters for product qualification, authentication, and safety. HRMS-based analytical methods can ensure the origin of food in terms of food traceability, and discover biomarkers to identify potential food safety issues in food quality.

Lipids are very important for human health. Lipidomics analysis in food science research includes composition analysis of lipids from different biological resources, revealing the mechanism and optimization of reaction conditions for modification of structured lipids, characterizing food safety and quality assurance, and evaluating bio-function and nutrition levels of lipids in living organisms. For example, despite uncontrolled environmental variability and non-standardized diets, the lipidomics analytical method using LC–HRMS was able to successfully profile the difference between canine breeds [32]. Non-targeted lipidomics analysis using an LC–HRMS system in data dependent acquisition mode was used to elucidate the global lipidome of 155 agricultural plant products of 58 species from 23 plant families, including grains, beans, nuts, vegetables, fruits, and tuber. This way was able to comprehensively elucidate the existing lipid structures for each agricultural product. In addition to facilitating quality control of food products, the study has also advanced the understanding of the physiological functions of food on human health [33]. Milk is a rich dietary source of lipids, including many lipids that have important functions in living organisms. Therefore, milk lipidomics analysis can not only provide important information when estimating the nutritional value of milk but also serve as a marker for the characterization of milk types, thereby providing a tool for detecting and preventing milk fraud. In order to provide nutritionists with more comprehensive information about the lipid distribution in different milks, the HRMS-based non-targeted analysis approach was used to profile different milk samples and identify biomarkers for milk fraud identification [34]. Lipidomics can also reveal the global profile of the intact lipid molecules in biological systems and provide qualitative and quantitative information. In this field, the lipidomics analysis using HRMS can also be used to reveal the influence of nitrogen sources on lipid biosynthesis of *Mortierella alpine*. This information provides a basis for further technological developments and the genetic modification of industrial microorganisms to improve the production yield of high-value-added polyunsaturated fatty acids [35]. Overall, HRMS can facilitate lipidomics analysis for many applications in foodomics. There are some reviews on mass spectrometry-based lipidomics to provide more details in this field [36, 37].

From the genome loses its active influence, proteomics is particularly important for when food proteins are subjected to different factors and food processing conditions that may not be genetically regulated. Food allergens are proteins or

peptides that trigger immune-mediated reactions in susceptible people. In recent years, the number of sensitive consumers, especially children with food allergies, has increased. There are some commercially available methods that are mainly based on immunological methods for specific proteins, such as enzyme-linked immunosorbent assays (ELISA) and polymerase chain reactions (PCRs) for specific DNA molecules for allergen detection. However, they have the limitation of distinguishing capability and selectivity. Therefore, there is a need to use MS-based proteomics approaches to explore allergenic ingredients that are not reported on the labels of food commodities. Due to the advantages of high selectivity and sensitivity from accurate mass measurement and resolution of complex mixtures, the HRMS-based NTA method allows identifying and quantifying the undeclared addition of allergenic milk proteins in samples of chicken and turkey sausages labeled as milk-free, thereby protecting consumers with a serious health problem from these food allergens [38]. In order to perform an in-depth comparison of the Whey proteome of different species, the HRMS-based proteomics approach was also applied to explore the milk Whey proteome of Indian *Zebu* cattle. This proteomic information was used as a comprehensive reference database for Indian cattle genetic resources for the production of functional foods and infant products. This assay led to the identification of 6,210 proteins from *in-gel* digestion and *in-sol* digestion [39]. Egg allergy is one of the most common food allergies, especially in children. In order to identify any egg allergenic ingredient, a UHPLC-HRMS based non-targeted analysis approach was developed to identify potential peptide biomarkers to detect and to quantify egg allergens in processed food products. This HRMS-based method is specificity, trueness, and robustness and can also be used as a targeted analysis method to quantify food allergens [40]. Some recent review papers comprehensively reviewed HRMS-based non-targeted analysis approaches used for the characterization and detection of food allergens [41, 42]. Food proteomics analysis can also be applied to adulteration detection. In a recent study, a novel analytical strategy using a state-of-the-art high-resolution Orbitrap mass spectrometer to perform pork gelatin speciation and detect adulteration was developed. In this method, HRMS was used to detect and identify targeted proteotypic peptides using proteomic bottom-up approaches in combination with *in silico* characterization of collagen type I, collagen type II and collagen type III. The MS<sup>2</sup> spectra of targeted proteotypic peptides at high resolution were used to compare with the amino acid sequence of each collagen type targeted proteotypic peptide for identification. The specific fragment ions were used to identify precursor peptide-to-fragment ion transitions for targeted gelatin species determination and adulteration detection [43]. In another recent work, a novel food proteomic approach was developed using HRMS to detect meat adulteration through proteogenomic annotation and carefully selected surrogate tryptic peptides. In this method, four proteotypic peptides (i.e., peptides from myoglobin, myosin-1, myosin-2, and  $\beta$ -hemoglobin) were successfully used for meat species determination and adulteration [44]. HRMS-based food proteomics in methodology and other applications have also been comprehensively reviewed previously [45–48].

Genomics, with its global perspective, is now reshaping the future of human metabolic health. Identification of the genetic components in food can be used as

traceability and to detect the presence of both genetically modified organisms and allergens. Genomics is also widely used in food authentication. DNA modification by exogenous chemicals and endogenous metabolites, also termed as adductomics, can be used as markers of food quality. As the successor of genomics and proteomics, metabolomics shifts the view of biology from genes to proteins to metabolites, which provides the information on processes in cells, tissues, and organisms to each other. Study on biological systems at the level of metabolites can provide in-depth insights into the biochemistry of diets, toxicity, medicine, physiology, and pathology, and therefore ensure food quality and safety. For example, an HRMS-based comparative non-targeted metabolomics analysis approach was utilized to identify marker candidates in microbially inoculated chicken eggs [49]. The metabolite markers discovered in this study may contribute to further industrial implementation of these newer spoilage markers, and ultimately benefit quality assurance and food safety in poultry egg production. In another study, a non-targeted analysis approach using UHPLC-Q-Orbitrap system to profile and identify metabolites in tea to evaluate health benefits of the tea [50]. The assay provided detailed information on the chemical composition of the tea during the fermentation process and provides experimental evidence for quality control and evaluation by observing the occurrence and dynamic changes of metabolites and lipids. Overall, HRMS has become a powerful tool in foodomics analysis due to its high-resolving power in the separation of  $m/z$ . HRMS-based foodomics approaches have been applied to screen and identify metabolites, DNA adducts, lipids, etc. in various food samples to ensure food quality and safety [51–54]. More detailed information about HRMS-based foodomics such as metabolomics and genomics, their applications, and future perspectives have been comprehensively reviewed previously [55–60].

## 5.2 *Food Contaminants and Natural Toxins*

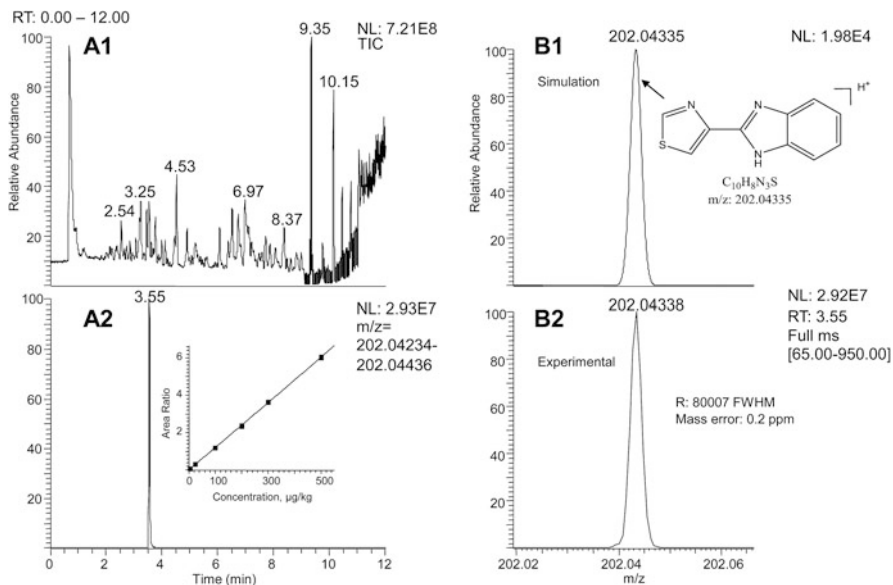
The globalization of the market increases the importance of food quality and safety for consumers. Illegal use of additives and chemical contaminants such as agrochemicals, environmental contaminants, veterinary drugs, allergens, and toxins, is a prominent problem in food safety. Therefore, legislation in the different countries worldwide drives the monitoring of residues of pesticides and veterinary drugs and contaminants in foods and the products traceability to ensure control and management through all stages of food production, processing, and distribution. LC-MS/MS and GC-MS/MS-based methods are commonly applied to such analysis. However, in current monitoring programs, low-resolution mass spectrometry is traditionally used to analyze a limited number of targeted well-known contaminants with MS/MS mode, which is not enough to ensure food safety. Although high sensitivity and the ability to simultaneously monitor multiple contaminants can help meet some urgent requests for measuring low-concentration contaminants in samples, the major requirement for the effective application of such targeted approaches is to know the identity of the compound before it can be measured. In particular, the presence of

undesired molecules either of natural or of synthetic origin is of paramount importance for the assessment of food safety, starting from raw materials up to the final products. Therefore, the development of novel and complementary non-targeted analytical methodologies is highly advisable to screen and identify unknown compounds in foods. Due to the advantages of HRMS, the LC-HRMS-based non-targeted analysis approach has been increasingly used in food residue and contamination analysis. With the development of more data processing tools, LC-HRMS-based non-targeted analysis can now quickly characterize hundreds of never-before-studied chemicals in food.

Compared with traditional LC-MS/MS, Orbitrap and Q-TOF mass spectrometers provide better signal specificity and sensitivity in quantitative multi-residue analyses. The higher mass resolving power of HRMS reduces the error in the assignment of the precursor ion mass of a candidate compound that has the same nominal mass from the co-eluting analytes [55]. Therefore, by monitoring animal urine instead of testing animal products, LC-HRMS has been applied to detect the illegal use of synthetic hormone growth promoters and the illegal use of  $\beta$ -agonists [61, 62]. The combination of liquid chromatography and HRMS was also applied to the comprehensive characterization of contaminants in food, such as polyphenols in beer to improve food safety for consumers [63], abuse of synthetic pesticides in farm and agriculture products [64–66], and non-targeted screening organic trace substances in the resources used for drinking water treatment, and their presence in drinking water for drinking water quality and safety [67, 68].

Veterinary drugs are widely used in the livestock industry, which makes parent drugs as residues, in addition to drug metabolites or transformation products, be present in food products. These drug residues and their metabolites or transformation products are regulated either as maximum limits or with an analytical threshold (minimum required performance limit) for banned substances. Compared with traditional low-resolution mass spectrometry, the application of high-resolution mass spectrometry in residue analysis significantly improves the ability of multi-residue analysis.

Pesticides are used to protect crops from pests, thereby increasing crop yields. However, these chemicals are potentially harmful to human health. Therefore, they are globally regulated by establishing tolerances or maximum residue limits (MRLs) to ensure that they remain at a safe level in foods. Approximately 1,000 pesticides have been produced worldwide for use in crops. Therefore, it is a challenge to analyze the presence of these chemicals, their metabolites, and degradation products when precise knowledge of pesticide application or abuse is lacking. The current common approach is to use GC and LC coupled with tandem mass spectrometry (mostly triple quadrupole with low resolution) to analyze target multi-residue pesticides in agriculture products [69], which can provide high sensitivity and selectivity in quantitation. However, these approaches are limited to analyzing too many potential pesticide targets in one method, thereby reducing selectivity and sensitivity, and they cannot detect the presence of pesticides that are not on the target list. In contrast, hybrid high-resolution mass spectrometers such as Q-TOF and Q-Orbitrap can provide accurate mass measurements and the MS/MS confirmation at the same



**Fig. 4** UHPLC/ESI Q-Orbitrap MS chromatograms and spectra from a full MS/dd-MS<sup>2</sup> experiment: (A1) extracted ion chromatogram (displayed as a stick per scan) of carbendazim  $[\text{M} + \text{H}]^+$   $m/z$  192.07675 with a mass tolerance of 5 ppm; (A2) dd-MS<sup>2</sup> total ion chromatogram of carbendazim  $[\text{M} + \text{H}]^+$   $m/z$  192.07675; (B1) mass spectrum (full MS scan) from chromatogram A1 at 3.25 min; (B2) dd-MS<sup>2</sup> or a product-ion spectrum of carbendazim  $[\text{M} + \text{H}]^+$   $m/z$  92.07675 from chromatogram A2 at 3.25 min. Pesticide spike concentration level (a total of 166 pesticides): 90  $\mu\text{g}/\text{kg}$  in a blank potato sample. (Reproduced with permission of J. Agric. Food Chem.) [72]

time with either targeted analysis for quantitation or non-targeted analysis for exploring more pesticides and new chemicals in crop products [70], vegetables [71], fruits [72–74] and tea [75–77]. As the sensitivity and resolution of modern Q-Orbitrap mass spectrometer have been significantly improved, the LC-Q-Orbitrap mass spectrometer has shown acceptable performance in quantifying pesticide residues in fruits [72]. Fig. 4 shows the extracted ion chromatogram (Fig. 4A1) and dd-MS<sup>2</sup> total ion chromatogram (Fig. 4A2) of carbendazim, respectively. In this NTA method, the DIA data were acquired using Full MS-SIM mode and the DDA data were acquired using Full MS/dd-MS<sup>2</sup> mode. In the later mode, when the ion intensity is higher than the set value, the product-ion spectrum with accurate mass measurement is automatically obtained. In the identification of pesticides, the dd-MS<sup>2</sup> product-ion spectrum (Fig. 4B2) was used to confirm the identity of a pesticide and characterize the structures of unknown pesticides in addition to the full MS spectrum (Fig. 4B1) for online and in-house library accurate mass matching search. In MS/MS mode, the background caused by the presence of isobaric compounds is significantly reduced (Fig. 4A2). Therefore, the use of HRMS for multi-residue pesticide analysis can quantify and identify hundreds of pesticides in one injection [72]. Although HRMS databases containing accurate mass,

chromatographic retention times, and high-resolution mass spectra are more useful in the assignment of unknown compounds, they are different from low-resolution mass databases and libraries. There are still not enough effective HRMS databases for the assignment of unknown compounds using HRMS data. Although HRMS has the advantage of mass-resolving power, compared with triple quadrupole mass spectrometry, it requires different concepts, experiments, and guidance to screen, identify, and quantify pesticides in complex food matrices. New data processing tools need to be developed to create an effective and optimized screening approach to speed up the analysis and identification of compounds, reduce the need for chemical standards, and harmonize analytical procedures. As pollutants are ubiquitous in the environment, it is always a difficult task to analyze organic pollutants from the environment in food and drinking water. It has an extensive history of using GC as the separation technique in food analysis. With the development of HRMS in resolving power and sensitivity to tandem quadrupole mass spectrometry and sufficient full-scan sensitivity, the application of advanced hybrid HRMS in a combination of UHPLC is increasingly used to monitor unexpected contaminants from various sources at relevant concentrations and analysis of chemical spaces [78]. In addition, due to the lack of commercial availability of reference substances and of prior information of unknown degradation products and metabolites, the instrument is required to provide semi-targeted or non-targeted screening analysis, not only detecting suspected peaks but also identifying and confirming underlying chemical structures or any unknown exogenous compounds in a given sample. The liquid chromatography high-resolution mass spectrometry for pesticide screening in foods has been recently reviewed [79].

Natural toxins can be either plant toxins, fungal toxins (mycotoxins) or marine biotoxins (phycotoxins). Since they are naturally generated and commonly contain a number of isomeric analogs, it is difficult to accurately analyze these toxins, which leads to a large number of false-positive and false-negative findings in low-resolution LC-MS/MS and results in the pressure to switch to alternative detection approaches. For example, many mycotoxins have to be detected at a very low level, requiring a good cleanup and high-sensitivity detection. Due to the high-resolution and sensitivity of modern HRMS, many mycotoxins have been identified in green tea [80], nuts [81], feedstuffs [82], beer [83], cereal [84], edible insects [85], and human breast milk [86], indicating the widespread presence in all types of food. Their metabolites identified in breast milk also indicate that these mycotoxins may have potential health effects [86]. Despite the number of regulated mycotoxins is limited, there is still a strong interest in more extensive screening to understand the broader prospects, because there is evidence that a variety of fungal metabolites and conjugates such as glycosides are naturally present in food [87, 88]. The situation of marine biotoxins is similar to that of mycotoxins, and a number of specific toxins have been regulated. In food analysis, there also is interest in monitoring exercises to study the incidence of contamination of such toxins from different structural classes. Compared with conventional LC-MS/MS methods, HRMS can facilitate the monitoring of a broad space of mycotoxins and has become



a preferred technique for analyzing natural toxins without the need of reference materials [55].

## **6 Application of High-Resolution Mass Spectrometry in Environmental Analysis**

Although the modern industry has benefited our lives with more artificial products, the environment has been heavily polluted by various chemicals produced by industry and anthropogenic activity. Due to the diversity of environmental samples, it is always a challenge to analyze contaminants sensitively and reliably with conventional LC-MS/MS methods.

### ***6.1 Analysis of Emerging Contaminants in Environmental Exposure***

In the environmental exposure, organic contaminants in environmental samples are very diverse with different physicochemical properties, which give challenges to analysis of them. Meanwhile, the number of organic pollutants reported in the literature is increasing year by year. Those new pollutants, the so-called “contaminants of emerging concern,” are frequently present in the environment at low concentrations, which makes the analysis of environmental samples even more challenging in terms of sensitivity and identification by using the conventional LRMS. Since HRMS can provide both selectivity and sensitivity, with the recent development of Orbitrap-MS-based instruments and the improvements to TOFMS systems, advanced hybrid HRMS is increasingly used by research laboratories and routine analytical laboratories.

The emerging contaminants in waters raise much more concerns recently, because new and legacy synthetic compounds in industrial and consumer products can enter natural waters via wastewater treatment plant effluents, urban and industrial sewage, erosional runoff, and leaching from agricultural areas. Therefore, the use of LC-HRMS-based methods has been developed to characterize these new and legacy compounds with highly lipophilic characteristics such as polychlorinated biphenyls (PCBs), polyaromatic hydrocarbons (PAH), and pharmaceuticals as emerging contaminants in waters. HRMS can help characterize much more contaminants in sample than LRMS. For example, more than 180 contaminants including pharmaceuticals, personal care products, pesticides, biocides, additives, corrosion inhibitors, musk fragrances, UV light stabilizers, and industrial chemicals in sediments were screened and characterized by an LC-HRMS method [89, 90]. HRMS can also be utilized to monitor and track emerging chemicals and pollution trends in water sources in order to achieve a more holistic water quality assessment, thereby

raising awareness of the multiple potentials for improving water pollution elimination and management efficiency [91]. Compared with tracking source contaminants, the behavior and transformation of organic contaminants in waters is more challenging in analyzing emerging contaminants and protecting waters. Therefore, the information about organic contaminants and their fate and their transformation products (TPs) in the water cycle as well as their loads and input pathways is very important for understanding water quality and contamination. To achieve this goal, the state-of-the-art of LC-HRMS/MS has the advantage of detecting known targets, suspected compounds like TPs, and unknown emerging compounds [92]. It is also important to study the presence of transformation products (TPs) of emerging contaminants that can be potentially found in the environment after biological or chemical degradation. For this purpose, the advanced LC-HRMS systems are perfect and powerful tools for evaluating molecular changes and identifying unknown molecules. In addition, in environmental and exposure science, it is also important to establish experimental evidence of the role of chemical exposure in humans and environmental systems. Therefore, using HRMS data, the global emerging contaminant early warning network can be used to rapidly assess the spatial and temporal distribution of emerging contaminants of concern in environmental samples [93]. The applications of HRMS in a non-targeted analysis of emerging contaminants in the environment have been reviewed previously [94, 95].

## 6.2 *Screening of Halogenated Compounds*

Halogenated compounds are persistent in the environment, including polychlorinated biphenyls (PCBs), brominated flame retardants (BFRs), and per- and poly-fluoroalkyl substances (PFASs). PCBs were studied and banned earlier than BFRs and PFASs. BFRs have been widely used in a variety of commercial products including children's sleepwear, foam cushions in chairs, computers, plastics, and electronics to reduce fire-related deaths, injuries, and property damage. Before it was banned, polybrominated diphenyl ethers (PBDEs) have been a popular ingredient in flame retardants. The widespread use of BFRs has led to their widespread presence in the environment and human and wildlife samples, as well as in places far away from where they were produced or used, which has caused the greatest health concern. BFRs are environmentally persistent, lipophilic, and bioaccumulative in animals and humans, and have many congeners. They can also degrade and transform into various transformation products under environmental conditions, which poses a challenge for screening them in samples. PFASs include fluorotelomer acids, alcohols, and sulfonates. So far, more than 4,000 PFASs have been registered. They have been widely used in making stain repellents (such as polytetrafluoroethylene and Teflon) that are widely applied to fabrics and carpets. PFASs are also used in the manufacture of paints, adhesives, waxes, polishes, metals, electronics, and caulks, as well as grease-proof coatings for food packaging. They are also persistent in the environment and in humans. The long-chain

perfluoroalkyl acids (PFAAs) are of particular concern due to the combination of environmental persistence, high bioaccumulation potentials, and more toxicity. Due to the slow development of global regulatory restrictions or bans on long-chain PFASs, they are being replaced by alternative short-chain PFASs. This increases more unknown PFASs, including new alternatives and legacy substances, and their transformation products in the environment, providing a great source of uncertainty for ascertaining environmental and human health risks. Perfluorooctane sulfonate (PFOS) and perfluorooctanoic acid (PFOA) are two types of PFASs. They are mostly monitored and studied for their presence in the environment and humans due to concerns about widespread global distribution in humans and in wildlife. Mass interferences from different halogenated congeners represent an additional challenge for LRMS-based methods to screen and identify halogenated compounds in the environment and humans to deal with environmental and human exposure. Therefore, HRMS has been becoming a powerful tool for screening a broad range halogenated compounds including organohalogenated compounds in various matrix extract [96–98]. The use of HRMS for the screening and characterization of PFASs in environmental and human samples has been recently summarized [99].

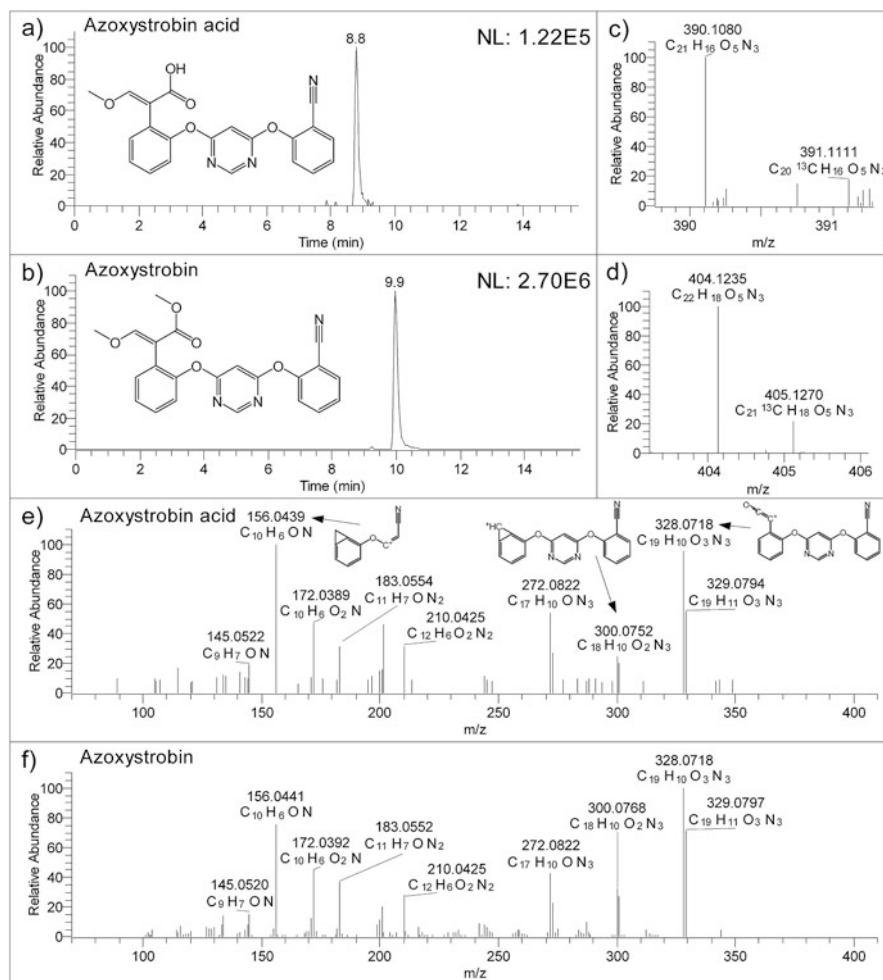
### ***6.3 Identification of Contamination Source***

In the environment, diffuse pollution sources can impact waters to degrade water quality and adversely affect the health of human and ecosystem. This is especially true for the urban receiving waters, rivers, small streams, and nearshore marine waters. Therefore, identifying pollution sources to reduce pollution and exposure risks is very important in environmental monitoring. However, this is challenging and limits the effectiveness of management actions intended to reduce such risk due to the diversity of pollution sources and the lack of fingerprinting of a broad of contaminants in samples. Accurate identification of potential sources of contaminants is essential for effective water quality management. However, the conventional LC-MS/MS methods can only track a small number of target known indicator compounds of chemical sources, so it is not enough to profile source specificity and characterize identity of the pollution source. Compared with LRMS, high-resolution mass spectrometry combined with LC or GC can detect hundreds to thousands of chemicals easily (both known and unidentified). The use of HRMS-based non-targeted analysis methods can establish unique chemical fingerprints and provide more statistical power for source characterization without relying on pre-existing knowledge or assumptions about chemical composition to distinguish and resolve the contributions of anthropogenic contaminant sources to receiving waters. Recently, a non-targeted analysis method based on LC-HRMS as a proof-of-concept has been successfully used to differentiate the chemical fingerprints of municipal wastewater influent, roadway runoff, and baseflow in urban storm drains [100].

Microplastics (MPs) are defined as synthetic polymers with dimensions less than 5 mm. Once microorganisms are generally unable to mineralize and assimilate them, they will show accumulation and persistence in the environment. Because the ingestion of these substances can cause toxicity to both humans and other living organisms, the presence of MPs in the aquatic environment attracts more attention. As they possess different physicochemical properties from chemicals, it is a challenge to characterize these MPs, especially to track their sources to reduce the exposure risk [101]. The traditional method is to use mass spectrometry to measure the total ion chromatogram to reflect the thermal desorption or pyrolysis degradation products of the polymer, as well as additives and leachable substances, which constitute the “chemical fingerprint” of environmental microplastics [102–104]. This approach using LRMS limits the throughput of analysis and reduces the number of high molecular weight compounds that can be included in the chemical fingerprint. In contrast, HRMS can accurately determine thousands of unique elemental compositions to represent more detailed fingerprints, so that it can distinguish subgroups of microplastics, even for the same polymer type, and rapidly identify the source [105].

#### ***6.4 Transformation and Metabolism of Contaminants in the Environment and Human***

In the aquatic environments, organic contaminants, especially compounds with complex structures (such as pesticides, biocides, and pharmaceuticals) that are currently concerned, can be degraded, metabolized, and transformed to form metabolites or transformation products (TPs), which contribute to the overall chemical burden in the environment. Although the TPs have been found in concentrations that are of concern, they are still rarely considered in water and air quality as well as in chemical risk assessments. Due to insufficient understanding of the overall contribution of TPs to the total chemical substances in the environment, there is a gap in knowledge about TPs and risk assessment. Another challenge is the lack of analytical reference standards for most potential TPs, which makes it difficult for researchers to use traditional LRMS, GC-MS/MS, and LC-MS/MS to quantitatively measure TPs in the environment. In addition, the results of laboratory degradation studies may not be representative of actual environmental conditions. In contrast, HRMS-based method has been successfully used to perform overall screening for the presence of potential TPs of 52 highly used and structurally diverse pesticides, biocides, and pharmaceuticals in representative surface water samples from various locations [106]. Figure 5 shows the retention time of the exact mass ( $C_{21}H_{15}N_3O_5$ ,  $[M + H]^+$ ) at  $m/z$  390.1084 for a TP of the fungicide azoxystrobin (Fig. 5a) and the retention time of the parent compound azoxystrobin (Fig. 5b) in the sample from *Le Boiron*. In this study, the identification of the TPs was performed by a combination of fragmentation patterns (Fig. 5c, d for the TP and the parent compound,



**Fig. 5** Structure of the TP azoxystrobin acid and its parent compound, the fungicide azoxystrobin, with the corresponding chromatograms (**a**, **b**), HR-MS spectra of the molecular ions, including isotopic pattern (**c**, **d**), and the HR-MS/MS spectra (**e**, **f**). The drawn fragment structures are those proposed by MassFrontier. All spectra are shown for the sample from *Le Boiron* (Reproduced with permission of Environ. Sci. Technol.) [106]

respectively) and high-resolution MS/MS spectrum (Fig. 5e, f for the TP and the parent compound, respectively). With the aid of MassFrontier fragmentation prediction, the product of ester hydrolysis, azoxystrobin acid, was finally confirmed as a major TP present in the aquatic environment.

Because pharmaceuticals are high-consumption products in the world, they are ubiquitous in the environment. Because of their potential adverse effects on human health, pharmaceuticals as environmental pollutants are receiving more and more attention. Compared with the parent pharmaceuticals, so far, there is little

information about their transformation pathways and their transformation products in the environment. Due to the lack of analytical approaches to monitor these transformation products and metabolites, there is also a lack of information about human occurrence and exposure. Metabolism and transformation are frequently associated with the detoxification of contaminants. However, the transformation products and metabolites can also be toxic or even more toxic than their parent compounds, which can cause health problems by exposure to these transformation products and metabolites. Therefore, it is emerging to screen and identify these pharmaceutical transformation products and metabolites in the environment. Advanced hybrid HRMS (time-of-flight (ToF) or Orbitrap) based non-targeted analysis approaches can provide accurate mass data at satisfactory sensitivity and open up for comprehensive non-targeted screening for pharmaceutical transformation products in environmental samples such as river sediment [107, 108].

## 7 Future Perspectives

In the past decade, high-resolution mass spectrometers have made dramatic advancements in resolution, scan speed, sensitivity, and hybrid capability. Due to the advancement of HRMS, the use of HRMS for food and environmental analysis has become more and more popular. HRMS-based NTA approaches have already shown the feasibility of detecting and identifying emerging contaminants and unknown compounds in food and environmental samples, which can subsequently trigger future target analysis, exposure-mitigating measures, and new priorities for food and environmental safety. However, up to now, HRMS instrument is still generally considered a research tool, and the NTA based on HRMS is still the frontier research of analytical chemistry. Although applications of HRMS in the monitoring of the wide range of potential contaminants and residues in food and environmental samples, including pesticides, veterinary drugs, and natural toxins, are steadily increasing, there is still a gap between the current methodology and routine food and environmental analysis.

In order to support food and environmental chemicals legislation, any information from HRMS-based non-targeted analysis (NTA) needs to be accompanied by clear statements about the quality of NTA data when it enters the regulatory process. Therefore, a harmonized quality assurance and quality control (QA/QC) protocol from sample collection and preparation to data analysis in the NTA process and minimum quality requirements are needed to provide high-quality data for regulatory processes. Given the unknown nature of the unknown chemicals, these same parameters in the targeted analysis might not be directly applied to NTA, which may affect the NTA's performance in terms of data reliability and reproducibility. Therefore, the confidence level for identifying compounds should be defined and standardized to allow effective communication between users through literature and databases. Chemical space is another issue that needs to be well defined and standardized in the NTA protocol. In general, there is no argument to keep as

much content of potential environmental contaminants and their metabolites as possible in the final sample prior to NTA. However, different definitions of chemical spaces will affect sample preparation process and consequently NTA results in communication with data users. Therefore, when applying NTA, it is necessary to harmonize the definition of chemical spaces and corresponding sample preparation in future food and environmental analysis. In addition, different sample matrices may also affect the NTA results in the same defined chemical space, and sample preparation will directly affect the signal-to-noise ratios and the quality of the MS<sup>2</sup> spectrum. However, there is a consensus that the general sample preparation method always starts from targeting certain chemical substances covering the defined chemical spaces, and these related chemical substances will be measured using NTA for method validation. Meanwhile, it is necessary to develop reference standard mixtures and reference matrix materials for HRMS-based NTA method validation to improve the comparability and reproducibility of NTA data. Of course, the number of standard compounds might not be enough to cover the diversity of the physical properties of all types of unknown chemicals to measure the effectiveness of standard mixtures in NTA performance evaluation, but it is still very useful.

The currently available data acquisition methods are not suitable for automatically recording MS/MS spectral data of unknown components in complex food and environmental samples. Improving analytical throughput and productivity in biotransformation study of food and environmental contaminants is another area that requires continuous efforts. MS/MS spectrum interpretation is still one of bottlenecks in metabolite identification and metabolomics profiling in true unknown analysis, although there are some software packages available on the market. Therefore, it is necessary to develop more software to assist data acquisition, processing, and interpretation to improve the data analysis efficiency and the identification confidence. Identification of isomers is still a challenge. The resolution power of HRMS alone cannot distinguish isomers. There are many isomeric compounds in food and environmental samples, especially in metabolomics study. Therefore, more efforts are needed to improve chromatographic separation methods prior to HRMS analysis to separate and identify isomers in food and environmental samples. The molecular weights, mass defects, isotope patterns, and fragmentations of truly unknown xenobiotic metabolites in food and environmental samples are not predictable, which is different from drug metabolites. Currently, there is no method that can universally screen all xenobiotics and metabolites in a single injection. Therefore, more NTA methods based on HRMS are needed to identify unknown xenobiotics and metabolites in food and environmental samples.

More reliable data processing tools are needed to enable to automatically assign molecular ions, elemental compositions and formulas, and even structures. The use of commercial or open-source software tools also needs a harmonized data analysis protocol. More tools are needed to automatically interpret complex data and assign molecule formulas when molecular masses exceeds 300 Da and the molecular formula contains more than elements of C, H, N, and O. In addition, more advances are needed in HRMS instruments to provide the isotopic abundance errors (isotopic pattern accuracy) to that data processing tools can quickly and accurately assign

formulas. Compared to higher mass accuracy without isotopic pattern accuracy, mass accuracy with high isotopic pattern accuracy (for example, 2%) can eliminate most false candidates. The declining trend of ion intensity in  $MS^E$  spectrum with collision energy can be used as a new additional filter. However, more software tools are needed to be able to automatically record, recognize the decrease of ion intensity in the  $MS^E$  spectrum with collision energy, and deconvolution the spectrum mixture to individual chemicals. Furthermore, definition of detectable and identifiable chemical space (chemicals covered in the method) based on an NTA workflow (a sequence of tasks from sample collection and preparation to data analysis) remains challenging and non-standardized. Each step of the NTA workflow, including sample collection, preparation, and NTA data acquisition, processing, and filtering steps will all influence the types of chemicals which are detected and identified. Accordingly, it is challenging to assess whether the non-detection of an analyte in an NTA study indicates true absence in a sample (i.e., blow a detection limit) or is a false negative result caused by limitations of the workflow in which chemicals are missing during the entire analysis procedure from sample preparation to data analysis. Therefore, the definition of criteria for false-positive and false-negative assignments as well as detection limit of NTA method should be standardized in data analysis.

In NTA, sample preparation is usually based on simple extractions, with limited clean-up to minimize the loss of chemicals/metabolites of possible interest from the sample. The sample preparation determines the extractable and detectable chemical space and thereby the identifiable chemical space in NTA results because the preparation step can introduce a bias towards the types of chemicals extracted and detected. Similar to other steps within the NTA workflow, there is currently no consistent approach on how to optimize the sample preparation step, which is critical to obtain reproducible and comparable results.

The current databases, whether they are online databases, commercial databases, or in-house databases, are poorly transferable for their reference spectra. Different libraries define different criteria for determining match scores in data analysis, resulting in that mass annotations based on library matching often inconclusive. This heterogeneity of reference spectra between different platforms hinder the effectiveness of spectral library search in NTA. Therefore, a large platform and server space will be needed in the future to share data by disseminating chromatograms, spectra or suspect lists. In the future, a universal or standardized LC-MS spectral library will be needed for use of HRMS based NTA method in food and environmental analysis.

Considering that there are no standards for unknown compounds in HRMS-based NTA food and environmental analysis, more harmonized prediction tools need to be developed, including retention time (RT) prediction tools and mass fragmentation prediction tools. The predicted RT can help narrow the range of candidate structures by comparing with experimental RTs. The RT prediction models explored so far are limited to current chromatographic methods and cannot be used under all conditions. Further efforts are needed to develop a universal and accurate RT prediction models to increase the confidence of compound identification. Depending on needs,



qualitative information for unknown compounds identified is not enough in exposure assessment and food and environmental safety assessment. To support regulatory activities, there is also a need to develop quantitation prediction tools to provide semi-quantitative level information. So far, there is still a lack of correlation models between the compound concentration and its corresponding mass spectral characteristics, such as ion intensity, peak area, and peak height. The lack of concentration information also hinders the values of NTA data in food quality and safety. Therefore, it is necessary to develop more NTA approaches that can provide qualitative information and quantitative or semi-quantitative information for food quality and safety and environmental safety. Due to the lack of MS/MS spectra for identifying unknown compounds, more in-silico fragmentation prediction tools need to be developed to predict fragmentation patterns and MS/MS spectra for compound identification. The current commercial prediction tools cannot accurately predict fragmentation patterns, nor can they provide suitable predicted MS/MS spectra that match experimental data. There is also a lack of such prediction tools to harmonize the predicted fragments into instrument conditions for the unknown compound.

## 8 Conclusion

Modern HRMS instruments such as TOF and Orbitrap have been dramatically improved to provide higher resolution, an accurate mass, and higher full scan sensitivity and selectivity than their older counterparts. In addition, the improvement of quantification ability in linear ranges and reproducibility makes HRMS very attractive and effective for both target and non-targeted screening of different groups of food and environmental contaminants. Combination with gas and liquid chromatography and coupled with novel software developments have increased the scope of HRMS in food and environmental analysis, which converts HRMS as a true alternative to conventional triple quadrupole MS instruments in food and environmental analysis to ensure food quality and safety, and environmental safety. Considering that HRMS-based NTA approaches can be used to analyze a large number of compounds simultaneously, additional work is needed to optimize and harmonize its terminology together with NTA workflows, validation, and QA/QC procedures from sample collection and preparation to data analysis. Compared with the workflow and sample preparation, data analysis and data mining are more complicated and challenging due to a large amount of MS data and variable MS spectral quality. Therefore, advance software tools and more efforts are required to minimize the data treatment and increase the data processing speed in NTA processes. Sample preparation needs to reduce matrix effects and retain all unknowns in the final sample. In the next decades, NTA's challenges in the extension and automation of NTA workflows for pattern analysis, compound identification, and accurate quantification are expected to be solved by machine learning from computer capabilities and tools. Regarding food quality and safety and environmental safety and monitoring programs, the possibility of retrospective analysis of full scan data will be the

key benefit of HRMS-based NTA data. This retrospective NTA data allows laboratories to evaluate almost all compounds in a sample, so there is no need to rerun samples to search for “new” or “unexpected” contaminants and metabolites. In the future, improvements and advances in hardware UHPLC and HRMS technologies, software tools and approaches, as well as a larger database combining chemical structures, physicochemical properties, standard LC-HRMS MS/MS spectrum libraries, production tonnages, toxic potential combined with RT prediction models will help to significantly accelerate and automate NTA workflows, thereby increasing the confidence in identifying unknowns. The development of high-throughput procedures that simultaneously screen, identify, and quantify residues, contaminants, and other important chemicals in food and environmental samples will be the next ideal accomplishment. Finally, research institutes, regulatory agencies, and policymakers also need to work together to use NTA data to pave the way for the implementation of NTA techniques in the chemical monitoring and management of food and environmental chemistry.

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# Omics Approaches in Food and Environmental Analysis



Marinella Farré

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**Abstract** Environmental-OMICS includes the applications of genomics, transcriptomics, proteomics, and metabolomics to understand better genetics, toxicity mechanisms, and modes of action in response to acute and chronic exposure to chemical pollution on aquatic and terrestrial organisms, and to understand which molecular events initiate these effects. These are essential goals in toxicology to predict adverse responses better or look for more efficient remediation approaches. In parallel and closely related, foodomics is a new discipline (Cifuentes, *J Chromatogr A* 1216:7109, 2009) applying the same omics technologies to study food and nutrition. It is a global discipline that integrates compound profiling assessment in food, food authenticity, and biomarker-detection related to food quality or safety, including contaminants in food, the development of transgenic foods, investigations on food bioactivity, and food effects on human health.

In both cases, the state-of-the-art technologies to assess effects and new mass spectrometry (MS) approaches combined with bioinformatics are crucial to answer

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the main questions driving environmental-omics and foodomics, which are included in the One-Health concept. Human health is cross related to our environment, the organisms in this environment, and the food chain.

The primary aim of the present chapter is to provide an overview of the different strategies that have been used in recent years in the environmental-omics and foodomics with the common driver of the One-Health concept. The advantages and limitations will be discussed as well, and finally, new trends presented.

**Keywords** Foodomics, Genomics, Lipidomics, Metabolomics, Nutrigenomics, Proteomics, Toxicogenomics, Transcriptomics

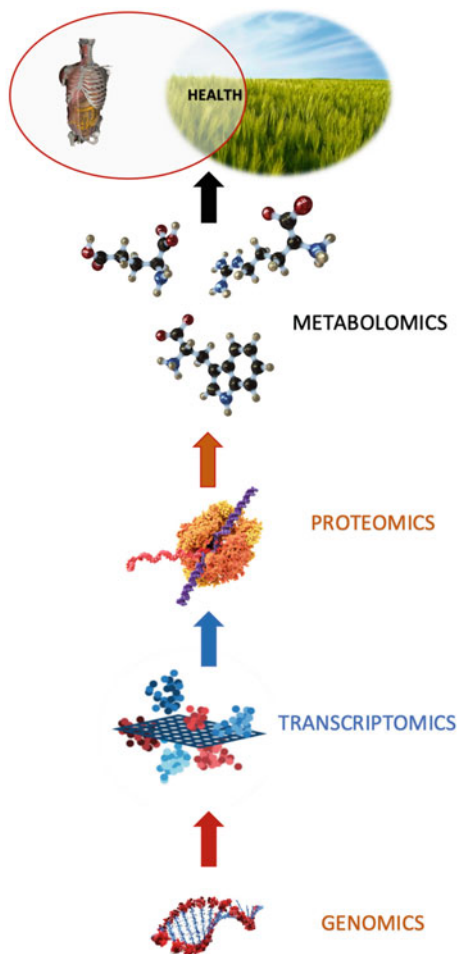
## 1 Genomics, Toxicogenomics, Transcriptomics, and Nutrigenomics

Microarrays are high-throughput analytic devices for simultaneously monitoring quantitative transcription of thousands of genes in parallel, providing comprehensive genome-scale expression. This technology uses synthesized oligonucleotide targets on a solid platform that acts to hybridize with complementary probes generated with fluorescent tags from a biological sample (Fig. 1).

In the last 20 years, impressive advances have been produced in *environmental toxicogenomics* thanks to microarrays and RNA-seq for gene expression analysis. In most cases, model species are used, and then the represented genes are targeted, but for non-model species, uncharacterized targets continue being a challenge, even though the improvement in annotation of microarrays. Gene expression profiling has been used in ecotoxicology to study the effects of different contaminants both organic (pesticides, pharmaceuticals, industrial origin compounds, endocrine disruptors, and nanomaterials, among others) and inorganic such as metals. The platforms using complementary DNA (cDNA) strands were used to study chemicals during the last decade. On the other hand, RNA technologies, such as RNA-seq, entail massive parallel sequencing of transcripts using methods developed by Illumina (NextSeq, TruSeq); and Thermo Fisher Scientific (Ion Torrent); and Pacific Biosystems, among many others. Longer reads or whole genome and targeted sequencing, RNA sequencing, and epigenetics are now possible. In Table 1, there is a summary of the most used Whole Genome Sequencing platforms.

For example, Gómez-Sagasti et al. [1] studied the early gene expression profiles in *Escherichia coli* cells exposed to three concentrations of a mixture of heavy metals by cDNA microarrays. At the lowest metal dose, the effects were an activation of metal-specific transporters and a mobilization of glutathione transporters involved in metal sequestration. At the intermediate dose, an overexpression of ROS scavengers and heat shock proteins were observed. Finally, at the highest dose, *E. coli* cells showed repression of genes related to DNA mutation correctors such as MutY glycopeptidases. In another example, the changes in the genomic expression

**Fig. 1** Summary of main omic fields applied to environmental-omics and foodomics



**Table 1** Summary of commonly used Whole Genome platforms

Platform	Sequencing technology	Read length	Output	Error rate	Type of instrument
Illumina	Sequencing by synthesis	Short reads 1 × 36 bp– 2 × 300 bp	0.3–1000 Gb	Low	Benchtop 2–29 h
Ion torrent	Sequencing by synthesis	Short reads 200–400 bp	0.6–15 Gb	Low	Benchtop 2–4 h
PacBio	Single molecule sequencing by synthesis	Long reads Up to 60 kb	0.5–10 Gb	High	Large scale 0.5–4 h
Oxford nanopore	Single molecule	Long reads Up to 100 kb	0.1–20 Gb	High	Portable 1 min–48 h

profile of marine medaka fish, *Oryzias javanicus* exposed to 4-nonylphenol were studied [2]. The genes differentially expressed were associated with oestrogenic signalling, immune responses, nerve cell maturation, cellular stress and detoxification, tumour transformation, glucose metabolism, RNA processing, lipid biosynthesis, and iron metabolism. These genes could be used as molecular biomarkers of endocrine-disrupting chemicals' biological responses. Sanchez et al. used transcriptomics in male zebrafish after 96-h exposition to study the mode of action of the fungicide chlorothalonil [3]. Gene expression analysis revealed changes in transcriptional sub-networks related to cell division, reproduction, immunity, DNA damage, and xenobiotic clearance.

In addition to laboratory exposures, transcriptomics is becoming more widely applied in field-based experiments of pollutants in the aquatic media. Collf-Dulá et al. used an oligonucleotide microarray to study perfluoroalkyl substances (PFASs) effects on wild largemouth bass [4]. In this study, transcriptomic profiling was conducted in the liver and testis of fish collected from five different lakes to determine molecular signatures. PFASs in tissues were associated with the expression of genes related to lipid metabolism, energy production, RNA processing, protein production/degradation, and contaminant detoxification, all of which were consistent with biomarker responses observed in other studies with PFASs. One of the primary outcomes of this study was how to distinguish the responses due to PFASs compared to other types of pollutants. It should be pointed out that for field studies is necessary the site characterization, with water quality characteristics such as temperature, dissolved oxygen, pH, humic substances, among other variables, to determine better the relative proportion of these factors to the transcriptional responses observed [5].

Transcriptomics can be employed for monitoring pollution over time and to assess biological responses before and after site remediation. Martyniuk et al. [6] used largemouth bass placed into ponds located in muck farms in the Lake Apopka, Florida region [6]. These ponds had a high contaminant load, primarily organochlorine pesticides (OCPs) that were used about 50 years ago on crops. The results revealed that largemouth bass inhabiting the ponds for 4 months showed significantly higher levels of organochlorine pesticides in their muscle, such as DDE, dieldrin, and methoxychlor compared to background levels. Moreover, these high contaminant loads in the muscle corresponded to lower plasma 17 $\beta$ -estradiol but there were no significant changes in gonadosomatic index or plasma vitellogenin levels. On the other hand, transcriptomics profiling suggested that genes involved in reproduction and immune function were significantly changed in fish placed in the mesocosm compared to largemouth bass collected from different reference sites. This study provided new information about the gene regulatory networks affected by OCPs being the most outstanding observation that these contaminants affect the immune system. Recently, Mehinto et al. [7] characterized the toxicological responses induced by persistent organic pollutants in wild-caught horny head turbot (*P. verticalis*). To do so, horny head turbot was intraperitoneally injected with a single dose of polychlorinated biphenyl (PCB) or polybrominated diphenyl ethers (PBDEs) congeners prepared using environmentally realistic mixture proportions.

After 96-h exposure, the livers were excised and analysed using transcriptomic approaches and analytical chemistry. Concentrations of PCBs and PBDEs measured in the livers indicated clear differences across treatments. Distinct gene profiles were characterized for PCBs and PBDEs exposed fish, with significant differences observed in the expression of genes associated with immune responses, endocrine-related functions, and lipid metabolism. This study provided further evidence of the potential of transcriptomics to improve aquatic health assessment and identify causative agents.

During recent years, toxicogenomics has been as well applied to assess plants responses to different exposures. Heavy metal contamination has become one of the important determinants of crop productivity recently. Heavy metal triggers various stimuli in plants, leading to complex mechanisms at the genetic, molecular, biochemical, and physiological levels. Phytoremediation based approaches stand second in bioremediation technologies meant for clean-up of soils/sites contaminated with heavy metals. Recent development in plant genomics, transcriptomics, proteomics, and metabolomics aided in understanding the role of several genes, transcription factors, and metabolites involved in conferring metal tolerance [8]. This branch bridges food safety and environmental analysis.

*Genomics is a powerful tool for food safety* analysis especially in foodborne bacteria outbreaks. In particular, the Next Generation Sequencing (NGS) combined with bioinformatic approaches is a step forward in food microbiology. The whole genome sequencing (WGS) of single isolates is a new approach that allows detailed comparison of individual strains. There are two main approaches:

- single nucleotide polymorphism (SNP) analysis and
- genomic multi-locus sequence typing (MLST)

Metabarcoding and metagenomics, applied to total DNA isolated from either foods or from the production environment, allow the identification of complete microbial populations. Metagenomics identifies the gene content and when these approaches are coupled to transcriptomics or proteomics, allows the identification of the biochemical activity of microbial populations. The technology has shown a rapid development, with continuous improvement in quality and the reduction of costs and nowadays is having a major influence on food microbiology. Therefore, WGS of food pathogens has been introduced into public health surveillance rapidly compared with other methodological advancements [9]. Initially NGS was employed for the retrospective analyses of outbreaks of foodborne illnesses, whilst now has been introduced for prospective surveillance of bacterial foodborne pathogens in Denmark, The United Kingdom, France, and the USA [10–12]. It should be pointed out that after 1 year of implementation in the USA successful results have been shown. The outbreaks were detected earlier and the source more often, and the total number of outbreaks increased [13].

In a recent study, Mahoney et al [14], evaluated the association between *Listeria monocytogenes* genomic characteristics and adhesion to polystyrene at 8°C. *L. monocytogenes* can be adhered to surfaces establishing itself in food production facilities. This study evaluated the adhesion ability of *L. monocytogenes* and

associations between the genetic factors such as internalin A (inIA) genotypes, stress survival islet 1 (SSI-1) genotype, and clonal complex (CC) and the observed phenotypes. The results indicated that premature stop codons in the virulence gene inIA were associated with increased cold adhesion, while invasion enhancing deletion in inIA was associated with the decrease. The results suggested that there is an evolutionary trade-off between virulence and adhesion in *L. monocytogenes*. These results provided understanding of *L. monocytogenes* adhesion in food production surfaces enabling to develop strategies to reduce their occurrence in the food system.

Genomics can be used to assess the origin of some bacterial foodborne transmission. For example, Group B Streptococcus (GBS) outbreak, associated with freshwater fish consumption, occurred in Singapore in 2015. GBS has not been generally recognized as a foodborne pathogen. Therefore, its occurrence in food is not usually tested. A study by Chau et al. [15] was conducted to determine the frequency and distribution of GBS sequence types in local retail food in Singapore. The results in food from retail stores were compared with those from regional fish isolate collection and published sequence types of human isolates. GBS was detected in 3.8% of the prepared seafood samples and 18% of the unprepared food samples. GBS was most frequently detected in fish among the unprepared food sampled, followed by meat, non-fish seafood, and vegetables. GBS sequence type 283 was identified in a sample of uncooked oyster. Seven other known sequence types (1, 3, 7, 10, 23, 24, and 103) and three newly assigned sequence types (1,647, 1,648 and 1,649) were isolated from the studied foods.

Food pathogen genomics is a rapidly developing field and comprises specialities, such as bioinformatics, which have not been subjected to standardization previously. For this reason, there is a need for standardization to assure that data generated is accurate, reliable, and globally comparable regardless of the sequencing platform, the bioinformatic approach, and software used. In that sense, different quality assurance instruments are developing including proficiency tests for microbial WGS analysis. Other initiatives include characterized sets of strains available for evaluating the performance of bioinformatic pipelines. For example, outbreak benchmark dataset including sequence data, metadata, and corresponding known phylogenetic trees for some bacterial species such as *L. monocytogenes*, *Escherichia coli*, and *Campylobacter jejuni* have been launched (<https://github.com/WGS-standards-and-analysis/Datasets>) [16].

Although WGS has revolutionized the molecular typing of pathogens, the main challenges to be addressed are:

- Further work on standardization is requested.
- Investigation into the role of environmental niches on the mutation rates of pathogens should be performed.
- Exploration of the value of mobile genetic element WGS analysis should also be more used.
- Industrial implementation should increase: WGS of bacterial isolates completely changes the way that subtyping has traditionally been performed. This, together

with the significant analytical costs along with the knowledge and competency requirements are currently barriers for its wider use by industry.

Two approaches with NGS technologies are used to study the microbial communities without bacterial culture:

- Amplicon sequencing or *Metabarcoding*, which is based on the amplification and sequencing of specific marker gene families [17]; and
- *Metagenomics*: the random shotgun sequencing of the whole genomic content of communities.

Metabarcoding requires the isolation of DNA directly from samples. Extracted DNA undergoes targeted PCR amplification of phylogenetic marker genes. Massive parallel sequencing of amplicons generates an array of profiling information about the microbiota associated with food products. Finally, the sequencing data is processed with bioinformatic pipelines to structure and annotate the raw information into knowledge.

The potency to follow the succession of microbial populations over time at various taxonomic levels is one of the central benefits of the metabarcoding approach. Thanks to that metabarcoding provides a cost-effective overview of the taxonomic composition of a sample and has already been applied to a variety of food products, e.g., to study the microbiology of fermented food [18, 19], and has also been used for characterizing the microbiota of food spoilage (e.g. the proliferation of lactic acid bacteria in fresh cut lettuce, leading to acidification and loss of structure [20]).

Metagenomics generates sequencing information from the genetic material in a sample, permits identification of individual strains, and can allow the prediction of functions encoded by microbial communities. The use of metagenomics is being explored for the detection, identification, and characterization of pathogens in food [21].

In addition to pathogens detection, metagenomics and metabarcoding hold great potential for the survey of food production, assessing food safety, authenticity [22, 23], and quality. Relevant applications have been published during the last years in food authenticity. Liu et al. [24] presented High-Throughput Sequencing (HTS)-based metabarcoding method successfully applied to pollen/honey identification to determine floral composition that can be used to identify the geographical origins of honeys. In another work, a DNA-barcode assay with fatty acid profile analysis was performed to authenticate the botanical origin of olive oil [25]. Real-time PCR (qPCR) is a central tool in food genomics which can be used to determine product authenticity. In Agrimonti et al. [26], a qPCR for characterization of microbial populations and type of milk in dairy food products was presented. Another important field of application is the improvement of certain food industry processes, for example, improvement in fermentations. Pereira et al. [27] carried out the selection of autochthonous lab strains of unripe green tomato aiming at the production of highly nutritious lacto-fermented ingredients. Genomics has been also applied to data-driven decentralized breeding because increases the accuracy in the prediction of crop productions in difficult environments. Data-driven decentralized

breeding could identify genotypes with enhanced local adaptation providing superior productive performance across seasons.

A branch of foodomics is nutrigenomics that focuses on the study of the effects of foods and their constituents on gene expression. Nutrigenomics assesses the impact of specific nutrients on health through the expression of genetic information [28–33].

## 2 Proteomics

Proteomics is the large-scale study of proteins, the entire proteome, of a cell, tissue, or organism under a specific, defined set of conditions. Proteomics is based on three basic pillars: (1) a method to fractionate complex protein or peptide mixtures, (2) MS to acquire the data necessary to identify individual proteins, and (3) bioinformatics to analyse and assemble the MS data. Thanks to instrumental and technological developments proteomics experienced impressive progress during the last decade.

In recent years, there are three main proteomic approaches established:

- Protein separation in gel (bottom-up approach)
- Shotgun (bottom-up approach)
- Mass spectrometry analysis (top-down approach)

Chromatographic and electrophoretic separations prior protein digestion are called bottom-up approaches, whereas the analysis of intact protein by means of separation with chromatography coupled to mass spectrometry is the top-down approach. The top-down approach in combination with small protein biomarkers (below 30 kDa) can be implemented to identify and quantify protein biomarkers in routine analysis [34].

One of the primary challenges in proteomics is the wide difference of proteins in cell types and individuals and the difference between most and less abundant proteins. Nowadays, different approaches are available to reduce proteome complexity including fractionation, depletion, as well as enrichment of low-abundant proteins. Following the separation and enrichment, MS represents the most-powerful tool in proteomics because it can be used under non-targeted schemes [35]. Current MS techniques allow the analysis of large-scale proteins and peptides. Improved MS with better sensitivity and superior mass accuracy and resolution aims to identify and quantify complex protein (peptides) mixtures in a single experiment. Main mass analysers used in proteomics are time-of-flight (TOF), orbitrap, Fourier transform ion cyclotron resonance (FT-ICR), and ion trap (IT), which are usually combined in one mass spectrometer (triple quadrupole (QqQ), Q-IT, Q-TOF, TOF-TOF, IT-FTMS, etc.).



## 2.1 Environmental Proteomics

It opens a new window for environmental sciences. Currently, environmental proteomics is being developed to provide information about molecular responses to contaminants in living organisms [36], and for the identification of more specific and sensitive markers of pollution for routine monitoring.

In general, for environmental studies proteomics consists in the selection of sentinel organisms on the basis of their susceptibility and extent to exposure, then the proteome study comparing exposed and non-exposed populations to assess cellular and molecular alterations. For example, for the biomonitoring of marine pollution the most abundant candidate species are invertebrates, highlighting among them the bivalve molluscs for their key role in the ecosystems and for their great value to aquaculture [37]. Moreover, bivalves are used as test species in biomonitoring due to ubiquitous distribution, easy accessibility, high filtering capacity as well as increased resistance to a wide range of pollutants, including emerging contaminants such as microplastics [38], nanoparticles [39], and other organic pollutants [40]. For the freshwater environments, the most common organisms studied are fish, crustaceans, and molluscs but there are mainly two organisms that stand out: the bryophyte *Fontinalis antipyretica* [41] and the zebra mussel *Dreissena polymorpha* [42]. It should be mentioned that there are far fewer biomonitoring studies of freshwater pollution than of marine pollution. More recently, native crustacean species have been used as bioindicators of freshwater ecosystems with good results [43].

In general, few studies are based on plants as bioindicator organisms due to technical reasons related to the extraction of complex proteins and to the fact that vegetal tissues have low concentrations of proteins and, in general, a high number of interfering substances are present. Moreover, another limitation is the lack of model plant species, which is limited to maize, rice, and wheat, and the lack of curated protein databases for non-model species.

Environmental proteomics has been specially used to assess contamination in marine environments. The traditional biochemical methods and gel-based proteomics have been particularly used to search protein biomarkers. Among them the two-dimensional gel electrophoresis (2-DE) followed by matrix assisted laser desorption ionization-time of flight mass spectrometry (MALDI-TOF MS) are the more employed techniques. The limitation, however, of these approaches is that only a small subset of proteins is analysed resulting in highly biased results.

In Table 2, a summary of proteomic studies is presented. The proteomic studies carried out in marine environments were mainly based on the use of bivalves. Protein expression was strongly affected by pollutants such as metals, affecting structural proteins such as actin and tubulin. For example, Cu, Zn, Cr, and Ni have been related to impairments in the structural components in oysters [56, 57]. Other organic contaminants such as pharmaceuticals have been shown to cause changes in the same structural proteins in freshwater clam *Corbicula fluminea* after 1 day exposure to wastewater from a psychiatric hospital [58]. In another work, Navon et al. [59]

**Table 2** Summary of recent proteomic studies to assess the impact of contaminants in biota

Pollutants	Organism	Technology	Alterations identified	Reference
As and Cd (1213.9 and 26.9 mg/L, respectively, for 96 h)	Amphipod ( <i>Gammarus sp.</i> )	2D-DiGE	Energetic metabolism	[44]
Au nanoparticles (80 mg/L for 96 h)	Gilt-head bream ( <i>Sparus aurata</i> )	2-DE and MALDI-TOF-TOF MS	Energy metabolism, stress response, cytoskeleton structure	[45]
Bisphenol A	Zebrafish ( <i>Danio rerio</i> )	Shotgun proteomics	Oxidative stress response, metabolic shifts, and degradome perturbations	[46]
Cd (0.1, 0.3, and 1 mg/L for 15 days)	Farrer's scallop ( <i>Chlamys farreri</i> )	iTRAQ	Carbohydrate and lipid metabolism	[47]
Cd (5 and 50 mg/L for 14 days)	Bastard halibut ( <i>Paralichthys olivaceus</i> )	iTRAQ	Oxidative phosphorylation, bile acid biosynthesis Stress resistance Apoptosis	[48]
Co-exposure Cu (II)/Cd (II) and the antibiotics chlortetracycline and florfenicol	Synechocystis sp. PCC 6803, a cyanobacterium	Label-free quantitative (LFQ) proteomics assays UPLC-MS/MS, PRM	Photosynthetic proteins were down-regulated, and ribosome was the primary target	[49]
DDT and benzo (a) pyrene (10 mg/L and 10 mg/L, respectively, for 7 days)	Asian green mussel ( <i>Perna viridis</i> )	iTRAQ	Cytoskeleton structure, gene expression, material and energy metabolism, reproduction and development, stress response, signal transduction, apoptosis	[50]
Microplastics (800 mg/L for 52 days)	Blue mussel ( <i>Mytilus edulis</i> )	Label-free quantification (LFQ)	Immunity and detoxification response	[51]
Microplastics (MPs) and antibiotic resistance genes (ARGs) in contaminated soils	Earthworms	SDS-PAGE electrophoresis and HPLC-HRMS	100 mg/kg of 10 µm MPs changed the microenvironment of earthworm gut, built a new homeostatic process, and thus increased the abundance of key bacterial that carried a variety of ARG	[52]

(continued)

**Table 2** (continued)

Pollutants	Organism	Technology	Alterations identified	Reference
Norfloxacin antibiotic in soil	Rhodococcus biphenylivorans TG9T	TMT kit and LC-MS/MS analysis	Inhibited DNA replication	[53]
Polychlorinated biphenyls (PCBs) (30 mg/L for 3 weeks)	Mediterranean mussel ( <i>Mytilus galloprovincialis</i> )	2-DE and LC-MS analyser	Protein polymerization actin filament severing Myosin II binding protein-containing complex binding	[54]
Zn (50 mg/L and 500 mg/L Zn for 30 days)	Portuguese oyster ( <i>Crassostrea angulata</i> )	Nano-UPLC-MS/MS	Zn and Ca transport, phosphate and energy metabolism, immune regulation, oxidative stress responses, gene expression regulation, fat metabolism	[55]

used ‘shotgun’ proteomics to identify differentially expressed proteins in *Herdmania momus*, after 14-day exposure to the antidepressant carbamazepine in laboratory conditions. Individuals were exposed to 5 or 10 µg/L in addition to a control treatment. Out of 199 identified proteins, 24 were differentially expressed (12%) between the treatment groups, and thus can potentially be developed as biomarkers for carbamazepine contamination. Again, one of the most affected proteins was tubulin followed by myosin [59].

Another significant group of proteins altered by pollution are the proteins involved in the metabolism. These proteins are valuable biomarkers [60]. For example, the proteomic analysis and biochemical alterations in marine mussel gills after exposure to the organophosphate flame retardant, tris(1,3-dichloro-2-propyl) phosphate (TDCPP), showed the inhibition of the acetylcholinesterase (AChE) activity, at both, after 7 and 28 days of exposure. Proteomic analysis revealed subtle effects of TDCPP in mussel gills, since few proteins (<2% of the analysed proteome) were significantly affected. But the epimerase family protein SDR39U1, an enzyme that could be involved in detoxification processes, was up-regulated at both exposure times, and the down-regulation of receptor-type tyrosine-protein phosphatase N2-like (PTPRN2) after 7 days of exposure, which is involved in neurotransmitter secretion and might be related to the neurotoxicity described for this compound [61]. Zhang et al. [62] studied the effect of two concentrations of 6-µm polystyrene beads (0.023 and 0.23 mg/L in seawater) to establish the multigenerational effect of microplastics on the marine copepod *Tigriopus japonicus*. The effects were followed in two-generation exposure (F0-F1) followed by one-generation recovery (F2) in clean seawater. The proteome profiling was conducted for the F1 and F2 copepods under control and 0.23 mg/L microplastics treatment to investigate within-generation proteomic response and transgenerational proteome plasticity. The proteomic analysis demonstrated that microplastics exposure increased several cellular biosynthesis processes and, in turn, reduced energy

storage due to the trade-off, hence compromising survival and reproduction of the treated copepods in F1. Furthermore, the two-generational effect of microplastics in copepods had significant transgenerational proteome plasticity, as shown by the increased energy metabolism and stress-related defence pathway, which accounts for regaining the compromised phenotypic traits during recovery (i.e. F2). In another proteomic study to assess the impact of microplastic pollution in marine organisms, blue mussels (*Mytilus edulis*) [51] were used. The repeated exposure, over a period of 52 days in an outdoor mesocosm setting, showed that microplastics exposure altered the haemolymph proteome and, polyethylene resulted in more changes to protein abundances than the bioplastic based on polylactic acid. Many of the proteins affected are involved in vital biological processes, such as immune regulation, detoxification, metabolism, and structural development [51].

## 2.2 Proteomics in Food Analysis

Nowadays, main applications of proteomics in food analysis under the concept of foodomics are:

- The determination of food allergens
- To trace food quality, composition, and origin
- Nutriproteomics

Different proteins have been identified as food allergens in dairy products, peanuts, soybeans, eggs, cereals, or fruits, for instance. Thus, *proteomics is a useful tool for the identification of allergens in food products* [35]. The bottom-up proteomics strategy has been widely used. Under this approach, the first step consists in the protein digestion in order to produce a set of derived peptides from the proteins in the sample and their identification as a part of a particular protein. MALDI-MS have been very much employed because it is well suited for peptide mass fingerprinting (PMF), as well as selected fragmentation of various precursors using collisional-induced dissociation (CID) or post-source decay (PSD).

Celiac disease, gluten hypersensitivity, wheat-dependent exercise-induced anaphylaxis (WDEIA), and wheat allergy are consequence of wheat, barley, and rye proteins. The most abundant gluten proteins are glutenins and gliadins. These proteins (prolamins), with high contents of proline and glutamine, are responsible for celiac disease. A variety of proteomic studies have been carried out to identify and quantify the gluten proteins from processing products ensuring gluten-free foods. For that reason, special efforts have been paid to elucidate a curated gluten protein sequence database to support proteomic technologies [63]. Wheat quality has been improved due to genetic improvement programmes but by modifications at the genotype level.

Other important groups of food allergens are proteins contained in legumes, nuts, and seeds that can cause anaphylactic reactions. The allergens are mainly proteins of storage, such as 2S albumins, 7S globulins, and 11S globulins. MRM was used to

characterize the allergens present in lentil cultivars determining trace amounts of certain peptides such as vincilin, legumin, lectin and lipid transfer proteins [64]. In a similar manner the contents in allergens, basically storage proteins, from different cultivars of soy were studied by SRM [65]. Shotgun proteomics could be a suitable technique of choice to select the cultivar with lesser allergen proteins in lentil seed. Moreover, the allergenicity of soy protein could be reduced by high hydrostatic pressure. Indeed, significant changes were produced in 7S and 11S globulin, hence reducing the allergenicity in children [66]. However, Ribeiro et al. [67] found minor differences of allergenic potential of 13 hazelnut (*Corylus avellana L.*) varieties.

Other allergens that have been widely studied are those contained in fruits and vegetables. Most adverse reactions are associated with pollen because of 'cross-allergenicity' [68, 69].

Another group of food allergens are present in animal-based foods in general coming from milk, eggs, and fish. Allergies caused by meat are scarce and they have been only described in atopic children. Most cases are related to specific IgE to galactose- $\alpha$ -1,3-galactose, an oligosaccharide of non-primate mammals [70]. Proteomic studies have been carried out to assess red meat allergy against the carbohydrate  $\alpha$ -Gal epitope [71]. On the other hand, most common proteins related to egg allergenicity are ovomucoid (Gal d1), ovalbumin (Gal d2), ovotransferrin (Gal d3), and lysozyme (Gal d4), and in the yolk,  $\alpha$ -livetin (Gal d5) and YGP42 (Gal d6) [72].

Also, seafood includes many allergenic proteins being the more important ones parvalbumin, tropomyosin, and arginine kinase, inducing immunological and clinical cross-reactivity [73]. Most of the allergic reactions are produced by parvalbumins.

Among animal-based allergens also milk allergy is a common abnormal response of the immune system to proteins in milk such as  $\alpha$ -lactalbumin (also called Bos d4),  $\beta$ -lactoglobulin (Bos d5), and casein (Bos d8) based on IgE measurements. However, other minor proteins, such as the bovine serum albumin (BSA), lactoferrin, and immunoglobulins could also produce allergic reactions [74, 75]. Different proteomic studies have been carried out to assess the different allergenicity after milk processed foods. For example, it has been demonstrated that heating treatment together with other food components could reduce the allergenicity of cow milk [76]. In the case of whey proteins, the loss of the allergenic effect is produced after 30 min of heating at 65°C [77]. Additional studies were focused on the search of peptides for the detection of casein and whey milk allergens from baked cookies. Nineteen peptides of four casein proteins and two whey proteins were identified using PRM [78].

In proteomic studies, one important step to assess food allergens is the extraction step. In this sense important amount of research has been carried out during the last years. A simple extraction protocol of allergenic proteins in food commodities such as milk, egg, hazelnut, and lupine seeds was recently presented by Calvano et al. [79]. Bottom-up approaches based on sodium dodecyl sulphate (SDS) gel electrophoresis separation followed by in-gel digestion or direct in-solution digestion of whole samples are described. Then, the analysis was performed by MALDI-MS and MS/MS along with a comparison of data obtained by using the most widespread matrices for proteomic studies, namely,  $\alpha$ -cyano-4-hydroxy-cinnamic acid (CHCA)

and  $\alpha$ -cyano-4-chloro-cinnamic acid (CCICA). Some approaches persecute to improve the extraction without the use of denaturing agents or detergents and optimized protein/enzyme ratio to reduce matrix interference and increase detection capability of target peptides. For example, Stella et al. [80] validated a new approach to assess four allergens on fish or swine processed food combining: (1) an extraction buffer not containing denaturing agents or detergents, (2) a defatting step with n-hexane, (3) an optimized protein/enzyme ratio to reduce matrix interference, and (4) the addition of heavy labelled internal standard peptide to improve method repeatability. The method was proved to be robust enough to detect and quantify proteins from milk, egg, soy, and crustaceans simultaneously with LOQ values below 5  $\mu\text{g/g}$  in complex matrices. The method validation was according to the requirements of Regulation (EC) 333/2007. All information collected for each allergen under consideration was included in a targeted parallel reaction monitoring (PRM) method to analyse four allergens simultaneously [80].

*Another relevant application of proteomics in food is related to its quality, traceability, and authenticity [22, 81].* Adulteration of cheese using less expensive milk is one of the authenticity problems in the dairy industry. The method of reference to assess the presence of cow milk in ewe or goats cheese is based on the detection of the caseins bands for cow  $\gamma$ 2- and  $\gamma$ 3 on an IEF gel [22]. However, this method presents a series of limitations. In the case of percentages lower than 1%, the marker bands for bovine milk are not visible when staining the gel with Coomassie blue [22]. Another limitation has been reported for cheese with concentrations below 5% cow's milk; the marker bands could be not detected using the most common staining methods. Moreover, interfering bands can cause false-positive responses and the need for an excellent electrophoretic separation of  $\gamma$ -caseins [22]. Proteomic approaches based on MS-based protein profiling, using either MALDI-TOF or electrospray mass spectrometry (ESI-MS), have been extensively investigated recently. Protein profiling obtained by MALDI-TOF was used for the investigation of the adulteration of ewe and buffalo milk with bovine milk [82], and in the adulteration of mozzarella cheese. The approach has been used to assess the adulteration of donkey and goat milk with cow, ewe, and buffalo milk at levels down to 0.5% [83].

The addition of UHT milk to raw or pasteurized milk has also been detected by MALDI-TOF MS with LOD of 10% [84]. ESI-MS has been employed to assess  $\beta$ -lactoglobulin to detect bovine milk in ovine milk [85] by means of high performance liquid chromatography, electrospray ionization and mass spectrometry (HPLC-ESI-MS) or capillary electrophoresis-MS. The presence of cow milk was detected at levels not lower than 5%.

Seasonality of buffalo milk production, rapid cheese decay and transport of its preserving liquid have relevant practical/economic consequences for mozzarella production; consequently, a progressive diffusion of cheese products realized with frozen curd or frozen milk. Arena et al. developed a MALDI-TOF-MS polypeptide profiling to assess specific markers of its freshness [86].

Proteomics have been applied to authenticate processed meat foods. For example, a method based on nano-LC-QTOF-MS/MS was applied to quantify highly

processed meat proteins to authenticate processed meat products [87]. Twenty new, heat-stable peptides unique to chicken, duck, and goose were identified as markers, with significant scores. This method permitted the detection of 1% (w/w) of chicken and 1% (w/w) pork in a mixture of the meat of three species. The abundance of proteins was computed by absolute protein expression (APEX) proteomics tool, which is a spectral matching technique [87]. In another work [88], two-dimensional O'Farrell electrophoresis with isoelectrofocusing in ampholine and immobiline pH gradients, the detection of proteins on two-dimensional electropherograms by staining with Coomassie R-250 and silver nitrate, and mass spectrometric identification of proteins by means of MALDI-TOF and MS/MS methods were used to confirm protein and peptide authenticity, and also to detect biomarkers of proteolytic changes in meat after slaughter. A quantitative proteomic distances study between three Spanish bovine breeds (Asturiana de los Valles, AV; Retinta, RE; and Rubia Gallega, RG) was performed from two-dimensional electrophoresis profiles of meat samples of longissimus thoracic muscle at 2-h post-mortem [89]. In total, 18 differentially abundant myofibrillar and sarcoplasmic proteins/isoforms contributing to proteomic distances between breeds were identified by MS/MS [89]. Other applications of proteomics for food authentication have been applied to oilseed [90, 91], and in cereal and vegetable crops [92, 93].

*Nutriproteomics* is a promising tool bridging diet and diseases in nutritional research. Comparative proteomics strategies have been applied to glean insight on the effect of specific natural compounds, nutrients, or diet. Most of these applications are based on bottom-up proteomic approaches; in particular, classical 2-DE separation of proteins and MS detection of the in-gel digested proteins. However, this field is still rather limited. Dietary antioxidants have been studied as chemical compounds candidates against cancer and inhibition of tumour progression. Three combined micronutrients, vitamin E, selenium, and lycopene were studied by [94], of which vitamin E and lycopene have recognized antioxidant activity in mice [94]. In the same field, the antiangiogenic effects of *Lindera obtusiloba* Blume (Hwangmaemok, HMM), which is a plant in the *Lauraceae* family have been studied. The effects of HMM extract on endothelial cell proliferation, migration, and neovascularization in chick chorioallantoic membrane (CAM) were investigated together with the molecular mechanism of antiangiogenesis using a ProteoChip-based proteomics technology [95]. In a more recent study, new strategies for breast cancer prevention are discussed [96]. Nutriproteomics is a promising tool to characterize the nutritional content of food and characterize and search functional ingredients. For example, the sweet chestnut (*Castanea sativa* Miller) is a tree species whose edible fruit was a staple food for many centuries until their consumption was displaced by the introduction of potato and cereal crops. Nowadays, there is a renewed interest in this fruit, partly due to being gluten-free. To describe and evaluate the sweet chestnut germplasm in Portugal a nutriproteomics study was carried out. Two-dimensional electrophoresis and MS were used to further investigate the chestnut proteome. Most of the proteins were assigned as having functions in nutritional storage activity [97]. During the last decade, the interest by isoflavones increased as functional ingredients. For the first time Rowell, Carpenter, & Lamartiniere demonstrated the

usefulness of proteomics for the discovery of novel pathways that might be involved in cancer prevention by isoflavones [98]. Currently a high number of studies have been carried out to assess the proteome and health characteristics of isoflavones. Epidemiological data indicates that regular intake of isoflavones from soy reduces the incidence of oestrogen-dependent and ageing-associated disorders, such as menopause disorders, osteoporosis, cardiovascular diseases, and cancer. Equol, produced from daidzein, is the isoflavone-derived metabolite with the greatest oestrogenic and antioxidant activity [99]. Other studies were more focused on the study of which isoflavones can have higher effects in cancer treatments. For example, the effects of isoflavone ME-344 on the disruption of redox homeostasis and mitochondrial function were characterized [100]. Proteomic studies can help to personalize certain treatments such as for cancer. In this sense, the importance of isoflavone dose and oestrogen receptor status in breast cancer cells were studied by Ilies et al. [101].

### 3 Lipidomics

Lipidomics is defined as the large-scale characterization of pathways and networks of lipids in biological systems. The word lipidome is used to describe the complete lipids profile in an organism, tissue, or cell.

As happens with proteomics, the advances in MS during the last decades increased the resolution and accuracy in untargeted lipidomic analysis. Lipidomics represents a powerful tool for the identification of all detectable lipids and their metabolites, including unknown chemicals [102].

The optimal workflow to understand the lipid profile and their functions by lipidomics is summarized as: the first step is sample collection and preparation, derivatization, LC separation, MS analysis, quality control, data processing, and data interpretation [103].

It is necessary to provide appropriate storage conditions based on the physical and chemical states of the sample to minimize degradation of lipids during sampling and sample preparation, such as storing at low temperatures ( $-20^{\circ}\text{C}$ ,  $-80^{\circ}\text{C}$ , or liquid nitrogen) and avoiding exposure to  $\text{O}_2$  during storage. In addition, to minimize the matrix effect and interference substances, extraction methods are employed before LC/MS approaches such as liquid–liquid extraction (LLE) or solid-phase extraction (SPE). Moreover, contamination problems should be avoided.

Specialized derivatization strategies have been used to enhance the selectivity and sensitivity of MS-based analysis [104]. The combination of derivatization chemistry with MS/MS scan modes (e.g. neutral loss, precursor-ion, or product-ion) and ionization efficiency has greatly improved structural characterization of isobaric lipid molecules and multiplexed absolute/relative quantification in MS/MS-based lipid profiling.



### 3.1 *Environmental Lipidomics*

Lipidomics has been used to assess changes produced by environmental stressors and contaminants in environmental biota [105]. During the recent years, lipidomics has been applied in environmental toxicology. It has been applied to assess oxidative stress produced by xenobiotics and to assess the biota responses to environmental factors, such as ocean acidification [106], climate change [107], and contaminants [108]. For example, the changes in the profile of lipidomics produced in the oysters (*Crassostrea hongkongensis*) exposed to copper (Cu) were studied by Chang et al. [109]. The changes in the lipidome profile of the digestive gland of exposed oysters showed induced phospholipid changes and an increase in polyunsaturated fatty acids that represent precursors of inflammatory mediators [109].

Lipidomics studies provide indication of the perturbation of oxidation pathways, and they have a great promise for a cumulative risk assessment. However, lipidomics is in early steps of developments and only a few untargeted data processing software programs can comprehensively cover the scope of oxidized lipid species for annotation and quantitation [110].

Environmental lipidomics has been applied in wildlife studies to identify lipid biomarkers indicative of environmental stress, and pollution. In aquatic organisms, fatty acid profiling has been used. For example, sewage release has been linked with shifts in the fatty acid profiles of freshwater clams [111]. In another study, the changes in fatty acids profile of caged mussels from a polluted in comparison with those from a clean reference site were used to identify lipid biomarkers of pollution [112].

On the other hand, during the last century, a reduction in communicable diseases has increased the life expectancy, while the mortality due to immune system diseases, Alzheimer's and multiple sclerosis, or inflammatory diseases from environmental exposures has increased, with the 70% of these diseases that can be linked to xenobiotic exposures and the diet [113]. In a similar manner, xenobiotic exposures, infections, and other environmental factors are the major cause of 80–95% of cancers. Lipidomics can play a central role in improving the understanding of the mechanisms ruling these immune disorders. It is important to assess the environmentally induced oxidation of lipids because of their cumulative risk and toxicity. Moreover, a class of biologically induced oxidized lipids, oxylipins, can indicate common responses to environmental xenobiotics that involve immune system dysregulation and inflammation [114]. Oxylipins may therefore serve as general markers of inflammatory-related environmental injuries. Nevertheless, the correlation between an imbalance in lipid homeostasis and the abundance of different molecular species is a poorly explored area of research nowadays [115]. In addition, pollutant effects on immunomodulatory lipids are an intriguing field of investigation. And the effects of contaminant exposure on haemocyte lipidomics related to immune functions remain unexplored to date.

Another application of lipidomics is bioremediation. Lipidomics, in combination with other omic techniques, can be used to characterize both the mechanisms of

toxicity to an organism and the mechanism of tolerance and bioremediation. This new level of understanding may lead to achieve the development of hybrid plants with tolerance to xenobiotics, and favourable characteristics for bioremediation applications. For example, the expression of the mammalian Cytochrome P450 2E1, a gene involved in the metabolism of xenobiotics in the liver, has been expressed in numerous genetically engineered plants, drastically increasing their plants for contaminants degradation [116].

Moreover, lipidomics has been used to characterize biota and their metabolic activity over extended time scales, providing information about the global climate change, and nutrient cycles. Lipids can be preserved in fossilized substrates, sediments, and glaciers, thus providing valuable information on ecosystem processes and their influence at geological timescales not obtainable via visual examination of fossil records [117].

Fatty acids are stable over geological timescales, and they can be used to characterize microbial communities in distinct environments over time and across multiple geographical locations. While most lipidomics-based research studies analyse fatty acid fragments to characterize microbial communities, the literature is currently shifting to the analysis of lipids [118].

Most environmental lipidomics studies have been based on the characterization of fatty acid profiles by GC-MS, but there are many advantages to the study of intact lipids, which possess specific membrane properties and functions, serve as signalling molecules, and can be more representative of certain organisms. Moreover, the expansion to intact lipids studies can be a new route for biomarker discovery. It should be mentioned that the limited number of environmental lipidomics studies is not concordant with their potential, but nowadays, there are a series of bottlenecks to overcome, such as high-level expertise needed for data processing and interpretation, the cost of the analysis using high-resolution mass spectrometers, and the need for characterizing the natural shifts in the baseline lipidome across various species and genotypes over time, and identifying lipid structures unique to specific organisms.

### 3.2 *Lipidomics in Food*

Lipidomics approaches enable large-scale and comprehensive studies of lipids. Therefore, lipidomics has been employed to optimize reaction conditions for the proper modification of structured lipids in the food industry, the characterization of lipids in food safety, and evaluation of biofunctional and nutrition levels of lipids in living organisms. For example, there are great differences in physicochemical, biochemical, and nutritional properties among different edible oils from diverse sources, such as plants, animals, and microorganisms or their blending and modification [84]. The nutritional value and the properties of edible oils are affected not only by the fatty acid composition, but also by the positional distribution of acyl groups bonded to glycerol. A shotgun lipidomic application was used to determine triacylglycerols in soybean seeds. This approach was based on the use of

electrospray ionization (ESI)-MS/MS with ten neutral loss scans performed to quantify 93 major triacylglycerols, but also new species were detected [119]. In another example, a chiral HPLC-atmospheric pressure chemical ionization (APCI)-MS method was employed to determine the composition of triacylglycerols enantiomers and regioisomers in hazelnut oil. In this case two cellulose-tris-(3,5-dimethylphenylcarbamate) columns connected in a series with a gradient of hexane-2-propanol mobile phase were used. Unsaturated fatty acids were preferentially esterified in sn-2 position in hazelnut oil [120]. Using 2D-LC coupled with atmospheric pressure chemical ionization MS (APCI-MS), triacylglycerols of high-oleic and normal peanut oils were profiled and statistically evaluated. The main differences in the content of triacylglycerols between high-oleic and normal peanut oils were found in 18:1–18:1–18:1 (OOO), 18:1–16:0–18:1 (OPO), and 16:0–18:1–18:2 (POL) species [121]. In another example, a comparative analysis of the lipids profiles in three edible bivalve species (clams, oysters, and mussels) permitted the identification of more than 600 different lipids belonging to 14 classes [122]. Despite this study was essentially carried out to establish a relationship between the nutritional value of each species, the application of lipidomics provides the baseline data for further studies on the effects of pollutants on the lipidome. Moreover, lipidomics has been successfully applied in authentication studies. For instance, the authentication of different types of milk [123].

The continued improvement of high-resolution mass spectrometry (HRMS) approaches has boosted the use of lipidomics in different fields of the food industry. Nevertheless, lipidomics is far to be well implemented in production environments. The main needs are more available and easy use of bioinformatic applications and computational methods to characterize the structure and abundance of complex lipids, and to help to interpret bio-function and metabolic pathways. Finally, improvements in the spatial resolution and scan speed of MS imaging equipment will greatly enhance the ability to measure and assess the spatial distribution of lipids in tissues and single cells.

## 4 Metabolomics

Metabolomics is the study of the low molecular weight metabolites that are generally <1,000 Da in cells, tissues, biofluids, or whole organisms by the application of advanced analytical techniques [124]. This includes primary metabolites which are involved in the development and reproduction of an organism, and secondary metabolites which are produced by bacteria, fungi, and plants and have various ecological functions [125].

Metabolomics is considered as the downstream process of genomics, transcriptomics, and proteomics, and the changes of metabolite levels directly relate to biochemical activity and the phenotype [126].

The fundamental metabolic pathways such as those involved in energy cycle, proteins, and lipids are conserved from bacteria to eukaryotes [127].

The primary analytical techniques applied to assess the metabolomic datasets are nuclear magnetic resonance (NMR) spectroscopy and MS because of their ability to detect small molecules. There are two main workflows in metabolomics:

- Target metabolomics which studies a predetermined set of metabolites [128].
- Non-targeted metabolomic analysis is the non-biased study of as many metabolites that can be identified by the analytical approach and databases [128].

Non-targeted metabolomic analysis frequently uses high-resolution MS or NMR spectroscopy, while targeted metabolomics often uses MS as the analytical method of choice [129].

## **4.1 Environmental Metabolomics**

Environmental metabolomics is the study of metabolic response of organisms as a result of their interactions with the environment [130]. Metabolomics studies have been applied to determine the response in front of various environmental stressors such as UV light [131], atmospheric carbon dioxide concentrations [132], drought [133, 134] and contaminants [135–140] among others.

These experiments can be carried out under controlled laboratory experiments with targeted species or in-field studies to evaluate how an organism is impacted when exposed to an ecosystem under environmental stress.

Environmental metabolomics provides for a more accurate biomonitoring and risk assessment. The general workflow to achieve these objectives is:

- study design,
- exposure,
- sample preparation and metabolite extraction,
- data collection,
- data analysis, and finally a biological interpretation of the analysed data.

### **4.1.1 Study Design**

This step involves the selection of the type of experiment to be carried out, at laboratory level or in-field study, and the selection of exposed organisms. Environmental metabolomics studies are done using a variety of organisms that include from microorganisms to plants and animals.

Microorganisms have been the organisms of choice in many metabolomic studies because of their easy manipulation under controlled conditions. The response of microorganisms to contaminants such as polycyclic aromatic hydrocarbons (PAHs) [141] and tetrachloro bisphenol A [142] or also to environmental stressors, for example, acidophilic metal-rich environments [143] have been determined. Earthworms have been studied to assess the metabolomic impact of certain contaminants

in soil. For example, in soils contaminated with arsenic [144], nanomaterials [145, 146], and pesticides [147, 148]. Plants are as well used to investigate the metabolic impact of contaminants in soils. For example, how plants respond to PAHs or metals from remediation efforts [149], the impact of mineral deficiency [150], of UV-B radiation [131], and drought [133]. Crustaceans, bivalves, and fish such as medaka, rainbow trout, salmon, and fathead minnow are the most common used aquatic organisms. Among them, it should be highlighted the use of bivalves because they have a sessile lifestyle and can accumulate contaminants, and therefore an analysis of the metabolic profile of bivalves may reflect the contamination at the site of collection. There are a series of considered model organisms, for example, rat and mouse for mammals, the zebrafish (*Danio rerio*) for aquatic vertebrates, the water flea (*Daphnia magna*) for aquatic invertebrates, the *Arabidopsis thaliana* for plants, the yeast *Saccharomyces cerevisiae* for eukaryotes, and *Escherichia coli* for prokaryotes.

#### 4.1.2 Exposure

In both cases, laboratory and field studies, the number of individuals, type of control, type of exposure, and the reference sites should have careful planning. In addition, since there is natural variation in biological samples an adequate replication is needed for proper statistical analysis [151].

In field studies, the organisms that represent the exposed groups may be collected at contaminated or stressed locations and may be compared to organisms collected at more pristine or less stressed locations, which serve as the group from a reference site. In addition, field studies can be carried out using two different schemes, one involves the sampling of free-living organisms in the environment and the second one involves deploying organisms into environments that are under stress.

Environmental metabolomics studies that sample free-living organisms at a field site are a step forward in environmental monitoring because they better represent the individual variability, which may stem from genetic differences. Moreover, field populations may be exposed to multiple stressors. However, this is also a challenge because it is difficult to distinguish cause and effect, and it may be difficult to distinguish between gender and different life stages. Field-based studies also face the difficulty of locating an ideal reference site that can act as control with minimal stressors but where the natural settings are analogous to the impacted site.

On the other hand, field-deployed organisms have an excellent value for environmental risk assessment. Laboratory populations and the experimental conditions (temperature, light, and diet) are strictly controlled, and they are maintained constant to limit any perturbations to the metabolome. Laboratory studies have the benefit of standard protocols that are available as guidance documents for aquatic toxicity testing and soil toxicity testing that should be followed for laboratory work (OECD, 2002, 2004). For these reasons, controlled conditions are optimal for determining the mode of action of contaminants as the metabolic perturbations can be directly accredited to the stressor [152]. There are standard reporting requirements for

metabolomics experiments of mammalian/in vivo work [153], and for metabolomics experiments of microbial and in vitro work [154]. Although the metabolic disturbances from exposure to stressors in laboratory conditions can be extrapolated to predict the hazard in environmental settings, care must be taken to avoid overestimating or underestimating risk [155].

### **4.1.3 Sample Collection, Sample Preparation, and Metabolite Extraction**

Since the metabolic responses across cells, tissues, organs, and the whole organism may be different, the decision of which biological sample to analyse should be made according to the research carried out. The metabolomics analysis of a certain tissue can give information about the state of an organ and different tissues of an organism containing different baseline metabolic profiles. Sometimes, the collection of a biofluid such as blood or urine is minimally invasive and can provide information about the overall biological state of an organism.

As proper selection of biological samples the proper sample preparation is important to avoid disturbances in the metabolome that may occur during the sample preparation. Sample preparation procedures may need to quickly halt enzymatic activity with freezing in liquid nitrogen and samples must be kept cold throughout any storage. In some cases, water should be removed (e.g. by lyophilization) for analysis by NMR spectroscopy and the solvent extracts of cell cultures need to be dried and reconstituted in an NMR buffer. NMR-based metabolomics most frequently uses a deuterium oxide phosphate-based buffer which extracts polar metabolites. The phosphate buffer maintains a constant pH to minimize variations in chemical shift and a deuterated solvent allows for a lock signal in NMR spectroscopy.

The metabolite extraction procedure for MS analysis typically uses solvents of varying polarity to extract polar and nonpolar metabolites. Also, the metabolite extraction method depends on the biological matrix chosen to be analysed, and there are well-developed protocols for metabolite extractions, for example, from mammalian and bacterial cell cultures [156], and biological fluids such as blood and urine [157].

### **4.1.4 Data Collection and Analytical Techniques**

As mentioned before, MS and NMR spectroscopy are the techniques of choice. However, both NMR spectroscopy and MS analytical techniques encounter some challenges, and there is no single analytical technique that can identify and quantify the whole metabolome. Several studies propose direct infusion mass spectrometry (DIMS) coupled to high-resolution mass analysers such as Orbitrap mass analysers because of the rapid and non-targeted analysis [158–160]. However, ion suppression and the complexity of the mass spectra without sample separation are disadvantages

of DIMS [158]. Therefore, MS is usually coupled to LC, GC, or capillary electrophoresis (CE) for analyte separation prior to mass detection, which can aid in metabolite identification using the retention time. GC-MS of polar metabolites typically requires chemical derivatization. However, the quantitation of derivatized metabolites may be difficult as the derivatized metabolites may not be constant. CE-MS involves the separation of highly polar and ionic metabolites due to different migration speeds of the metabolites [161]. CE-MS based metabolomics studies often couple capillary electrophoresis to a TOF mass analyser [161]. LC-MS analysis includes the selection of the chromatographic column, solvents for the mobile phases, potential additives to the mobile phase, and mobile-phase gradients for separation by LC. Most MS-based metabolomics studies use LC with electrospray ionization (ESI) that is suitable for the ionization of a range of metabolites. Non-targeted metabolomics studies for global profiling frequently use high-resolution and high accuracy mass analysers such as Orbitrap analysers. The main advantage of using MS for environmental metabolomics is that MS can detect and quantify metabolites at very low concentrations in the femtomolar to atto-molar range, and has a high dynamic range and resolution. The limitation is that MS only detects metabolites that can promptly ionize.

NMR spectroscopy can identify unknown metabolites unambiguously, can distinguish isomers, and can be used for structure elucidation of unknown compounds [162]. The most used NMR method in environmental metabolomics is one-dimensional (1D) proton ( $^1\text{H}$ ) NMR spectroscopy. Another relevant advantage of NMR spectroscopy is that it has non-destructive data acquisition, for instance, NMR methods such as  $^1\text{H}$  high-resolution magic-angle spinning NMR ( $^1\text{H}$  HR-MAS NMR) can be performed on intact tissues, although this technique has been mainly used in clinical studies [163]. In vivo NMR spectroscopy using flow-based systems with solution-state NMR [164] and comprehensive multiphase NMR [165] is also emerging for environmental metabolomics applications of living organisms. The main limitation of NMR spectroscopy is the sensitivity, only the most abundant metabolites at concentrations  $>1$  mM can be detected [166]. Also, the resolution in 1D  $^1\text{H}$  NMR spectroscopy is low and although two-dimensional (2D) NMR methods increase resolution, the acquisition time is much. However, nowadays there are technical developments that can increase the sensitivity and resolution of NMR spectroscopy for metabolomics analysis, such as para- $\text{H}_2$ -induced hyperpolarization, cryoprobe technology, and microprobe technology.

#### **4.1.5 Data Analysis, and Finally a Biological Interpretation of the Analysed Data**

Before the statistical analysis and the pattern detection with chemometrics models, and the previous step of pre-processing the raw spectra collected from NMR spectroscopy or MS is required.

After this previous step, the multivariate statistical analysis is performed to simplify and facilitate the interpretation of the large data sets obtained. The most

commonly used method is an unsupervised statistical approach, the principal component analysis (PCA), that has no bias or previous knowledge about the treatment group. PCA is used to determine general patterns in the metabolic profiles of the different treatment groups. The outcome is a scores plot where each point in a two-dimensional or three-dimensional space is the spectrum of a single sample. The samples with similar features in their metabolome are grouped in the same cluster to identify group membership and visualize the data quickly. Finally, the averaged PCA scores plot is made where the scores of all the spectra of one treatment group are averaged and plotted as average values with standard error. PCA loadings plots indicate the metabolites that contribute to the metabolic variation between treatment groups.

On the other hand, to generate predictions about the data supervised statistical methods such as partial least squares (PLS), PLS regression (PLS-r), PLS discriminant analysis (PLS-DA), and orthogonal (O)-PLS-DA models are used. The data from classified samples are considered the training set, which are used to predict the data from unclassified samples, the test set. PLS models require proper validation to avoid overfit of the data and result in the false appearance of group separation. Validation includes leave-one-out cross validation, K-fold cross validation, Monte Carlo cross validation, and double cross validation. In order to test the predictive ability of the PLS model, an independent validation can be carried out with a different data set. To determine the statistical significance of the model, frequently a permutation test is used. This test gives random group assignments to evaluate the prediction accuracy.

Metabolite data can be presented by reporting the absolute concentrations of metabolites across all treatment groups or by reporting the percentage of change of each metabolite in the exposed groups relative to a control. To determine whether the difference of a metabolite concentration between two independent treatment groups is statistically significant, the main univariate statistical tests are the Student's t-test for data with a normal distribution and the Wilcoxon Mann-Whitney test for data that are not normally distributed. When more than two groups with normal distribution are compared, to determine the significance the analysis of variance (ANOVA) test with post-hoc analysis can be used. The result of all these statistical tests is a P-value. If the P-value is below a significance level that is generally 0.05, then the difference between two or more group means is accepted. Also, if the P-value is below the selected significance level, then no significant difference between the two groups is considered.

#### **4.1.6 Data Analysis, and Finally a Biological Interpretation**

The biological interpretation of the metabolic changes is challenging because metabolites can be part of several metabolic pathways. The Online databases of metabolic data include the Kyoto Encyclopedia of Genes and Genomes (KEGG) Pathway Database that shows many metabolic pathways and the individual metabolites in each step of a pathway [167]. The MetaboAnalyst 4.0 has several data



interpretation tools for both MS and NMR spectroscopy data, such as the metabolite pathway analysis (MetPA) where the changes of the individual metabolites are correlated to metabolic pathway information. In this case, the users can enter the list of metabolites with their abundances from two treatment groups, select the model organism among 21 models which MetaboAnalyst 4.0 has the metabolic pathway library from databases as the KEGG. Then, it is necessary to specify the parameters for pathway analysis, including the algorithm for pathway enrichment analysis and the algorithm for topological analysis. The output is presented as a map-style network visualization system and a list of metabolic pathways in order of their statistical significance.

Metabolomics offers a picture of the biochemistry of the organism at any one time. Environmental metabolomics allows characterizing the interaction that occurs between organisms and the surrounding environment. In particular, it has been used to study the interaction between organisms and contaminants that are present in the environment and may have toxic effects. Environmental metabolomics has been as well applied for biomarkers discovery, but this is a relatively new field. It has been in the last 5–10 years when the application of metabolomics for analysing biological samples in environmental monitoring has strongly attracted the attention of research. Consequently, an increasing number of papers are currently being published generating a high amount of data that need to be compiled and harmonized to get relevant information. In Table 3, a summary of recent works in this field is summarized. As can be seen the previous workflows are used, but it should be pointed out that MS analysis together with bioinformatic tools is recently the combination of choice. The main advantage of environmental metabolomics is assess the subacute impact of contamination at realistic concentrations of exposure and the ability to assess complex mixtures and provide integrated responses of these complex mixtures and other environmental stressors.

## ***4.2 Metabolomics in Food Safety and Quality Control***

Metabolomics has already been successfully applied in various fields of food science, showing promise in maintaining food safety and food quality including authenticity and food traceability.

Several food safety issues including the emergence of novel pathogens in food, adulteration, risk assessment of genetically modified foods, and the detection of chemical contaminants have been gaining increased attention. For these reasons, food metabolomics has growing as a powerful tool to comprehensively assess quality and safety aspects of food, providing valuable information on the authenticity of food ingredients [181–183].

For example, a quality classification of Spanish olive oils was proposed by Sales et al. [184], using untargeted GC-QTOF-MS with atmospheric pressure chemical ionization and metabolomics-based statistical approach. A novel integrated non-targeted metabolomic analysis was used to establish significant metabolite

**Table 3** Examples of recent metabolomic studies in the environment

Organism	Contaminant	Exposure route	Timing of exposure	Sample types	Metabolic biomarkers	Reference
Adult zebrafish ( <i>Danio rerio</i> )	Tributyltin	Living in tributyltin solution	Adult	Zebrafish embryos extract	↓ Alanine, 4-Aminobutanoate, L-Valine, Taurine, Creatine, L-Methionine, L-Carnitine, 6-Succino-minopurine, 6-Pyruvoyltetrahydro-p-terin, Hypoxanthine, Adenosine monophosphate, Guanosine monophosphate, 5β-Cyprinolsulfate, Oxidized glutathione, PS (44:12) ↑ L-Glutamine, 5-Oxo-D-proline, L-Tyrosine, L-Acetylcarmitine, 4-Aminohippuric acid, Neopterin, Inosine, Methylhexadecanoic acid, dehydrodiconiferyl alcohol, LysoPC (18:1), LysoPC (18:0), PC (32:1), PC (36:5)	[168]
C57BL/6 J mice	DEHP	Gavage	Postnatal day 56–84	Blood	↑ LTB4 8S-HETE Arachidonic acid, Bishomo-γ-linolenic acid, γ-Linolenic acid, Linolenic acid	[169]
<i>Caenorhabditis elegans</i>	Atrazine	Living in solution	Larvae	Nematodes extract solution	↓ Phosphatidylcholine, CDP-choline, GMP, 7,8-Dihydroneopterin 2',3'-cyclic-p, 2-Oxobutanoate, Neopterin, 7,8-Dihydroneopterin ↑ Salicin 2-Hydroxyethyl-ThPP	[170]
<i>Daphnia magna</i>	Bisphenol A (BPA) Bisphenol F (BPF) and bisphenol S (BPS)	Living in solution	Adult	Organism	↓ Choline, cysteine, glutamic acid, histamine, histidine, isoleucine, leucine, malic acid, and phenylalanine	[171]
Female Wistar rats	di-(2-ethylhexyl) phthalate (DEHP)	Gavage	Postnatal day 28–58	Urine	↓ Kynurenic acid, N6-methyladenosine, L-isoleucyl-L-proline, xanthurenic acid, hippuric acid, tetrahydrocortisol, citric acid, phenylpropionylglycine, cPA(18:2/0:0), LysoPC (14:1)	[172]

Kunming mice	(DEHP)	Injection	Postnatal day 0–21	Plasma	↓ PC, PC-(O), PC-(P), LPC18:0, and Pls ↑ PE, PI, PS, and CE	[173]
<i>Lemma minor</i> L.	Metribuzin, glyphosate	Solution		Leaf	↑ $\gamma$ -aminobutyric acid (GABA), salicylate, caffeate, $\alpha,\alpha$ -rethaloase, and squalene	[174]
Lettuce ( <i>Lactuca sativa</i> )	Imidacloprid and fenvalerate	Water solution		Roots	↑ Arginine, cysteine, homoserine, and 4-hydroxyisoleucine ↓ Glucose, raffinose, maltotetraose, maltopentaose, and stachyose ↓ Organic acids and polyphenolic compounds	[175]
<i>Mytilus edulis</i>	PFOA	Living in PFOA seawater		Gland	↓ LysoPC, taurine, Glutamine, oleic acid ↑ chenoodeoxycholate, MDA, pyruvate, serine, glutamate, cysteine, glycine	[176]
Specific pathogen free grade male Sprague Dawley rats	Cd	Gavage		Blood/kidney	↓ UA, Hypoxanthine (Hyp), Xanthosine, Deoxyinosine, 12-HETE, Guanidinosuccinic acid, Taurocholic acid, Hydroxyphenylacetyl-glycine, ATP ↑ Formiminoglutamic acid, arachidonic acid	[177]
Sprague Dawley rats	PFOS	Gavage	Postnatal day 0–21	Brain section	↓ Phosphatidylcholines, ↑ GABA, taurine, Gly, Met, Pro, Ser, and T4-Hydroxyproline, N-methyl-D-aspartate (NMDA)	[178]
Sprague Dawley rats	Bisphenol A	Gavage	Postnatal day 42–56	Urine	↓ Riboflavin, Biotin ↑ Methylated amino acids	[179]
<i>Triticum aestivum</i> L	Oxytetracycline and microplastic-polyethylene	Water solution	Seedling growth stage	Leaf samples	↑ Carboxylic acids Glyoxylate and dicarboxylate metabolism, amino acid metabolism and isoquinoline alkaloid biosynthesis	[180]

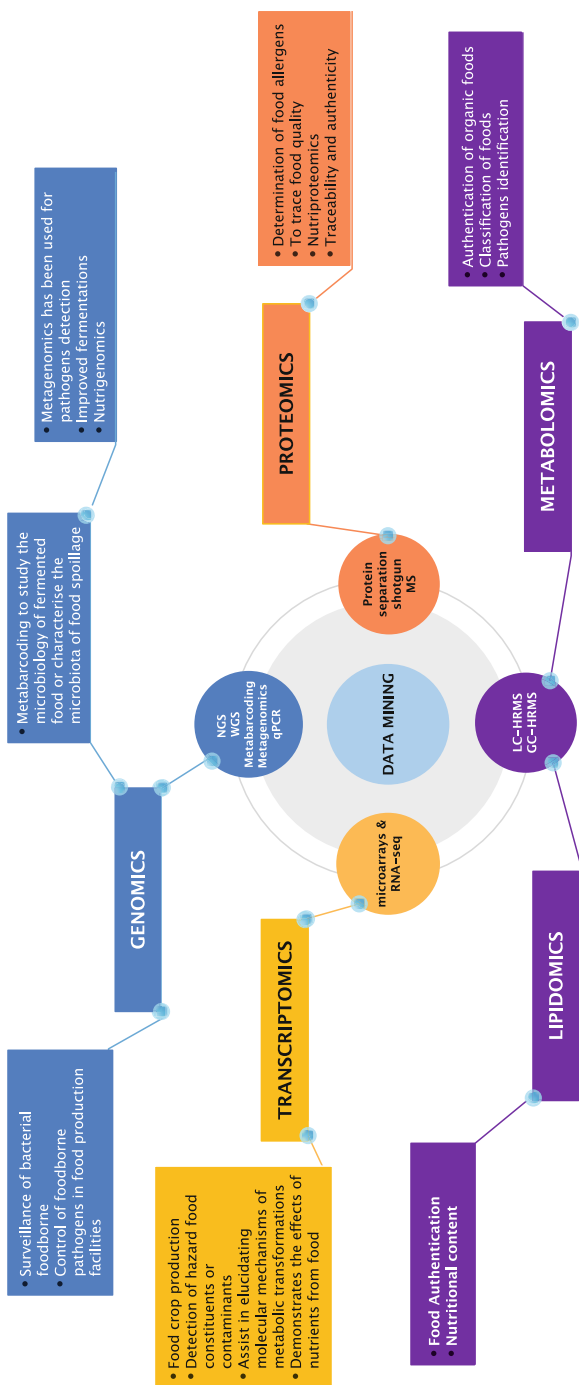
variations between different lettuce varieties [182]. Another metabolomic approach was proposed by Jandrić et al. [181], for the authentication of Indian citrus fruit. The method permits to distinguish adulterated foods.

Metabolomics has been used to authenticate organic foods. For example, the effects of organic food intervention on biomarkers of exposure to pesticides [185–187]. Metabolomics can as well be used as an analytical technique for food processing evaluation [188] such as in fermented food [189–192].

Thanks to the fast development of analysis techniques, metabolomics has achieved significant progress. Promising findings bridging the biological systems information and food analysis have been achieved. Nowadays, it is possible to analyse more than 1,000 metabolites in a single run and integrate various analytical methods. Like any other omics field, the application of metabolomics approaches in food security, quality, and traceability remains far from achieving its maximal potential and suffers from some bottlenecks, including sensitivity and compound identification limitations. Appropriate statistical approaches for the vast amount of data and metabolite coverage and function continue being requested. Moreover, the practical application and their integration on productive schemes are still nowadays low. Therefore, it is essential to enhance the sensitivity and precision of instruments to enlarge the coverage of metabolites and increase data quality with the combination of different analytical methods and platforms.

## 5 Conclusions

Omic technologies provide a holistic view of the molecules that make up a cell, tissue, or organism. Omic technologies aimed primarily at the universal detection of genes (genomics), mRNA (transcriptomics), proteins (proteomics), lipids (lipidomics), and metabolites (metabolomics) in a specific biological sample in a non-targeted and non-biased manner. These systems, primarily when integrated, can provide information on the subacute responses to stressors or offer complete information about processes. Omics can also be referred to as high-dimensional biology; the integration of these techniques is called systems biology. Therefore, when these techniques are integrated into food and nutrition, foodomics effectively provides analytic information about the effects of food on human health, it can help to improve food processing, quality control, traceability, and authentication. In Fig. 2, a summary of main analytical techniques in foodomics is presented. In the field of environmental sciences, omic technologies will provide a better understanding of the impact of complex mixtures of contaminants integrated with other environmental stressors. Moreover, the combination of these disciplines in food and environmental sciences brings together the necessary tools for a more confident risk assessment under the one health initiative. Nevertheless, omic strategies still have many challenges, such as the bioinformatic tools that are still evolving. There is a lack of standardized methodologies. Thus, it is necessary to establish carefully designed experiments, accompanied by appropriate analytical techniques and



**Fig. 2** Summary of main analytical techniques used in food metabolomics. *NGS* next generation sequencing, *WGS* whole genome sequencing, *LC-HRMS* liquid chromatography coupled to high-resolution mass spectrometry, *GC-HRMS* gas chromatography coupled to high-resolution mass spectrometry

statistical analyses, which will assist to compare results tackling many of these challenges, with the potential to generate reliable validated data to answer the biological questions.

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# Thermal Desorption and Pyrolysis Combined with Gas Chromatography–Mass Spectrometry in Food and Environmental Chemistry



Julian Campo and Yolanda Picó

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**Abstract** Pyrolysis–gas chromatography–mass spectrometry (Py–GC–MS) has confirmed to be a versatile technique that benefits food and environmental analyses. It has been used for the chemical characterization of materials and compounds that are not suitable for traditional gas chromatography (GC) because of their large size. The controlled thermal degradation carried out during pyrolysis is able to break down macromolecules into volatile fragments easier to identify because they become separable by GC and detectable by mass spectrometry (MS). A wide array of applications has been reported using Py–GC–MS, from characterization of macromolecules (polymers, paints, lacquers, adhesives, plastic, synthetic fibers, organic matter, etc.) in a variety of disciplines including forensics, history, engineering, and, of course, food and environmental sciences. In recent years, this technique has

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experienced an important increase due to its capability for the chemical fingerprinting of organic matter, and the identification and characterization of nano-, microplastics used for food package and present in environmental and food samples. In this chapter, we describe current Py–GC–MS instrumentation and working modes and summarize recent applications in food and environmental analysis with special emphasis on its strengths and limitations.

**Keywords** Curie-point, Evolved gas analysis, Microfurnace, Reactive pyrolysis, Thermal degradation

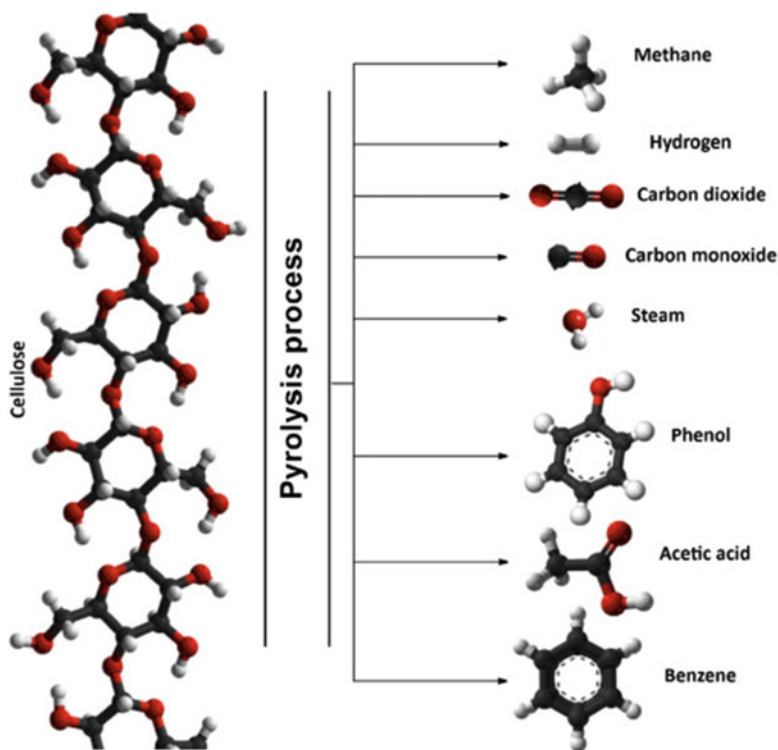
## 1 Introduction

Pyrolysis–gas chromatography–mass spectrometry (Py–GC–MS) can characterize macromolecules [1] because pyrolysis transforms non-volatile macromolecules into their structural units, which are normally small and volatile, while GC provides resolving power and MS identification capability, broadening the range of applications [2]. The International Union of Pure and Applied Chemistry (IUPAC) defines Py–GC as “a version of reaction chromatography in which a sample is thermally decomposed to simpler fragments before entering the column,” and Py–GC–MS as “the characterization in an inert atmosphere of a material or a chemical process in which chemical degradation reaction(s) is/are induced by thermal energy” [3].

Pyrolysis provides the thermal decomposition of the sample at relatively low pressure and at temperatures between 500 and 1,400°C originating characteristic volatile units in the presence of an inert gas (the most common is helium). Chemical changes in the macromolecules are induced by high temperatures with progressive bond breakages going from the weakest to the strongest. Among reactions implicated are depolymerization (resulting in basic units and/or oligomers), random excision (resulting in random small fragments as all the C–C bonds have the same energy), and the elimination of side groups (cleaved of the polymer bond). Other possible but minor reactions include chard formation, oxidation cross-linking, cyclization, isomerization, and hydrogenation [2]. The fragmentation of molecules caused by pyrolysis depends on the relative strength of the bonds and the ability of the formed free radicals to provide steady products [4, 5]. For instance, big complex hydrocarbon molecules can be broken down during pyrolysis into simpler and smaller molecules of char, liquid, and gas (Fig. 1) [6].

The units formed in the pyrolyzers are separated by chromatography, using fused silica capillary columns, and identified by MS (using mass spectral libraries or selecting few characteristic ions). Py–GC–MS is well-matched with most of the treatments used with environmental and food samples but can also eliminate these processes as it is capable to analyze directly macromolecules and/or polymers [7–9]. One important advantage of this technique is the small quantity of sample used (5–200 µg) although samples must be dried.





**Fig. 1** Decomposition of large hydrocarbon molecules into smaller ones during pyrolysis [6]

Py–GC–MS enables the analysis of macromolecules as polymers, copolymers and additives, packaging materials, fibers, coatings, electronic intermediates, paints, lacquers, paper or wood derivatives, bio-oils and biomass, food, drugs, surfactants, fragrances, etc. [9–15]. The applications of Py–GC–MS include research, quality control, and characterization of materials as well as forensic analyses, conservation and restoration of cultural heritage, biotechnology, geology, and agriculture among others. Publications on Py–GC–MS are mostly focused on its applications such as the authentication of extra virgin olive oils [11] or the characterization, for example, of bio-oils [4, 12, 14, 16], polymers [1, 17, 18], lacquer films [9, 15], or lignite tars [19]. There are also a number of publications, including several reviews, dealing with soil organic matter and its chemical fingerprint [8, 20–26], and specially with the occurrence and fate of micro- and nano-plastics (MPs and NPs, respectively) in the environment [27–35], which has gained great attention recently.

In this chapter, we highlight new instrumental advances that combined with the multiple workflows of Py–GC–MS have broad food and environmental applications, mainly in relation to compositional analysis, quality control of authenticity, adulteration, and contamination detection (especially by plastic polymers), and characterization of organic matter.

## 2 Py-GC-MS: Instrumentation and Working Modes

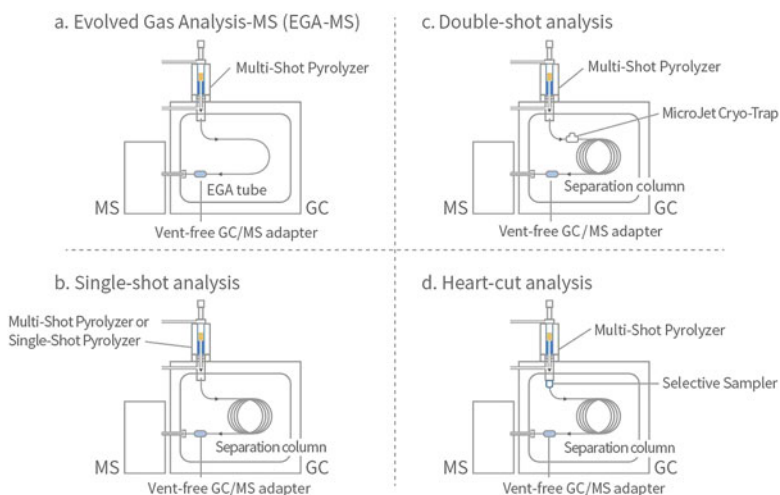
### 2.1 Working Modes in Py-GC-MS

The technological evolution of Py-GC-MS has mostly be focused on devices design, inertness of their components, versatility of the working modes, and reproducibility of the obtained data, and it has already been reported in several review articles [2, 36]. However, there are several important new developments in the pyrolysis devices (microfurnace chambers, quartz liners for the sample, etc.) as well as in the GC-MS process itself (time-of-flight (TOF) MS in the low- or high-resolution mode, comprehensive GCxGC, etc.) that deserve to be highlighted.

Currently, there are different possibilities to work with Py-GC-MS depending on the analyses' objectives summarized as (Fig. 2):

*Evolved gas analysis (EGA-MS)* (Fig. 2a): the degradation products are separated from macromolecules based on the temperature at which they are formed rather than at their volatilization temperature. This is achieved using a slow temperature ramp in the pyrolyzer and a short and narrow (2.5 m, 0.15 mm i.d.) deactivated capillary tube without a stationary phase, which connects directly the GC injector and the MS-detector. During the GC oven is held constant at 300°C, the pyrolyzer is heated by a temperature program from 40 to 500°C. The resulting chart is called an EGA thermogram. EGA-MS is a previous step in which the number of peaks and the temperature zone for each peak are vital information in determining correct temperatures for the other working modes (i.e., single-shot, double-shot, and heart-cut).

*Single-shot analysis* (Fig. 2b): the pyrolysis is performed at a single temperature (normally >500°C depending on the sample analyzed), which increases as fast as



**Fig. 2** System configuration for four analytical techniques (courtesy of Frontier Laboratories Japan Ltd.)

possible (in current instruments <20 ms). The macromolecules are fragmented in the pyrolyzer, and the volatile pyrolysis products, known as pyrolyzates, are separated in the chromatographic column.

*Double-shot analysis (Fig. 2c)*: it includes two stages, the first one involves thermal desorption of the volatile compounds (of low molecular weight analyzed at low temperature, ~80–350°C) examined then by GC–MS. In the most recent instruments, these volatile compounds are cryo-trapped temporarily at the head of a separation column by liquid nitrogen. In the second step, the residual sample left after desorption (in which the non-volatile macromolecules remain) is pyrolyzed at high temperatures (~500–800°C), and the pyrolyzate is again analyzed by GC–MS.

*Heart-cut analysis (HC/EGA-GC-MS) (Fig. 2d)*: it is the two-dimensional way of working in Py–GC. The EGA is used to produce a thermogram, and each temperature zone of interest is analyzed independently by heart cutting evaporating components that are selectively introduced to the GC column (by means of a selective sampler), where they are temporarily trapped prior to the analysis by GC–MS (e.g., cryo-trapped as described above). This method allows searching both the specific components in a highly complex matrix and the whole composition of a complex system. While extremely useful, this technique can add complexity and required-time for the analysis.

*Reactive Pyrolysis–GC–MS*: with this technique the sample undergoes a chemical derivatization reaction in the pyrolysis chamber avoiding extensive sample preparation (the most used derivatizing agent is tetramethylammonium hydroxide). This derivatization may be instead of, or in addition to, heat-induced pyrolysis of the macromolecule into smaller fragments.

In addition to these analytical techniques, modern pyrolyzers can also be used only for thermal desorption of analytes (TD/Py–GC–MS) as in the first step of double-shot analyses. The system configurations for the four analytical techniques can also include a vent-free GC–MS adapter enabling column exchange without venting the MS.

## 2.2 Recent Advances in Py–GC–MS Instrumentation

Py–GC–MS can deliver quantitative results with high accuracy and precision. As in any other analytical method, quantification of determined compounds requires standards, isotopically labeled internal standards, and the selection of the proper ions [12, 26]. The main disadvantage attributed to Py–GC–MS is its poor reproducibility because of factors such as sample heterogeneity, slow transfer of the pyrolyzate to the chromatographic column, and catalytic events in the pyrolyzer that may alter the chemical nature of the sample. Other drawbacks of the technique are its destructive character, and the difficulty of interpreting the pyrograms obtained due to the large number of reactions that might take place and the density of peaks in the pyrograms [36]. Consequently, recent advances have been focused on developing new approaches to overcome these problems by improving the reproducibility of the

pyrolyzer, enhancing the separation of the compounds, and improving the identification of the reactions products.

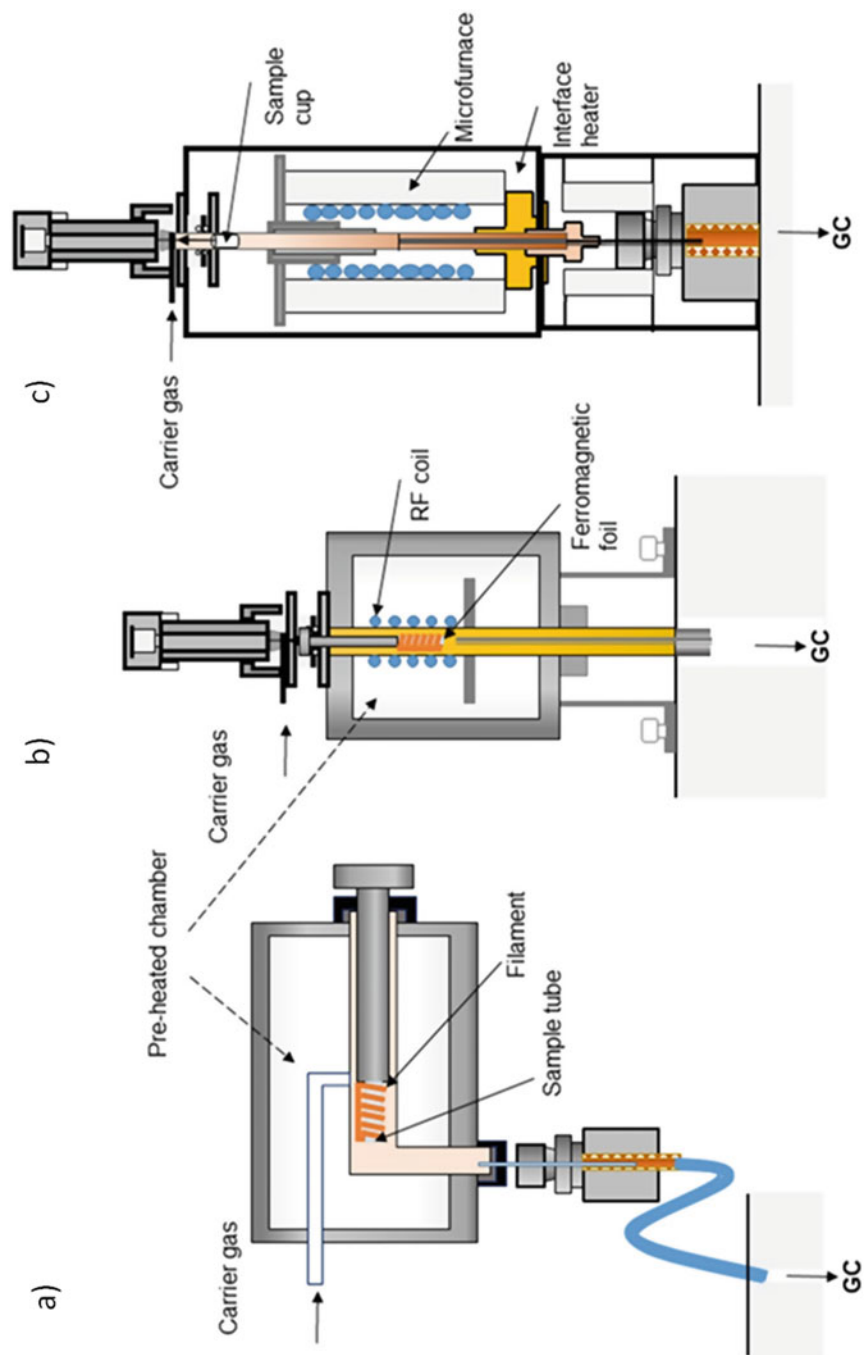
### 2.2.1 Pyrolyzer

Pyrolysis systems have been classified according to their heating mechanisms as: pulse-mode pyrolyzers (filament, Curie-point or induction, and laser) and continuous-mode pyrolyzers (furnace or microfurnace pyrolyzer) (Fig. 3). Current environmental and food applications of Py-GC-MS are mostly focused on microfurnace pyrolyzers because this type of pyrolyzer can work with higher amounts of samples, which is important for the analysis of minority compounds. Improvements in this field have been related to the control of the pyrolysis temperature (rapid rise, reproducibility) and the connection of the pyrolyzer with the GC, which must be as direct as possible (to prevent pyrolyzates back into the injector where they can undergo secondary pyrolysis).

The filament type pyrolyzer heats a sample tube using a filament (Fig. 3a). This technique has some drawbacks in temperature accuracy and reproducibility since some types of samples may be not uniformly heated. Moreover, high boiling compounds may be absorbed in the transfer line carrying the pyrolyzates to GC, which make difficult to introduce them to a GC column. Some progress has been done in developing better and more effective sample holders able to short the analyte diffusion path and to reduce peak broadening. Considerable effort has also been made in the implementation of a technology ensuring accurate monitoring of the filament resistance to guarantee truthful temperature control (heat-transfer variations can affect the temperature of the sample). Modern instruments can perform programmed heating in addition to pulsed pyrolysis, to obtain sequential pyrolysis of a sample combining different temperature ramps, and velocities of heating. Other improvements search to avoid evaporation of the volatile and semi-volatile compounds and denaturation, degradation, or thermofixation of the samples because of sample-preheating (essential in this technique). Consequently, thermal desorption in last instruments is carried out by modular systems or complicated trapping sequences [37, 38].

The Curie-point pyrolyzer (Fig. 3b) is similar to the filament pyrolyzer (preheating of the sample is also necessary) but different temperature steps cannot be programmed as required for EGA-MS analysis, since the temperature ceases to rise when the Curie-point of the metal holding the sample is reached. Then, the EGA analysis is impossible.

The vertical furnace/microfurnace pyrolyzer is a hollow quartz tube covered by a heater device for rapid heating and cooling (Fig. 3c). Modern microfurnaces are equipped with hollow ceramic heaters and powerful cooling fans, able to achieve rapid heating and cooling, together with temperature sensors for the sample. Changes in the sample cup materials for inertness have been made recently such as the use of quartz or chemically bonded quartz thin layers. Microfurnaces avoid the problems caused by the preheating of the filament and Curie-point pyrolyzers by



**Fig. 3** Pyrolysis systems (pyrolyzers) classified according to their heating mechanisms: (a) filament, (b) Curie-point, (c) microfurnace [2]

placing the sample in an inert deactivated cup, held at ambient temperature (in helium) at the same time that the furnace is preheated to the pyrolysis temperature, and then letting drop the cup into the microfurnace. The sample is purged of oxygen before heated avoiding oxidation reactions [5, 39, 40].

The microfurnace pyrolyzer has several advantages over other types of techniques: a) accurate furnace set temperature in a wide-ranging temperature or continuous ramp-up heating, b) direct interface to the GC injector, which allows the complete introduction of all molecular weight compounds (low to high), and c) the sample just needs to be simply placed in a sample cup and it is ready for analysis. Worth mentioning is the Multi-Shot Pyrolyzer EGA/PY-3030D with a low mass ceramic heater allowing pyrolysis temperatures up to 1,050°C, rapid heating (600°C/min) and cooling (100°C/min), very precise temperature control in the microfurnace (40–1,050°C ± 0.1°C), simple handling of samples through weighing in sample cup (approx. 50–200 µg), and no condensation of gases at cold spots (no dead volumes) during transition to GC due to an interface-heating.

In relation to the connection of the pyrolyzer to the GC, the main concern is to transfer the analytes in a tight “plug.” This means heating the sample as fast as possible, in order to have the resulting volatiles in a very short time with narrow and well separated peaks. Cryo-trapping capabilities using liquid nitrogen are available for Py–GC–MS in both pulse- and continuous-mode pyrolyzers [41, 42].

### 2.2.2 Gas Chromatography–Mass Spectrometry

Although most Py–GC–MS applications have been made with conventional GC separation, there is no limitation in terms of their possible modes. There are two major trends in the use of this technique: (ultra-)fast chromatography and comprehensive two-dimensional GC (GCxGC) [4, 12, 14, 16, 19]. Fast GC seeks to reduce the analysis time by using different combinations of narrower and shorter columns, with lower phase thickness and/or higher carrier gas flow, as well as by high temperature heating ramp rates (up to 250°C/min) and oven cool downs (450 to 50°C in <3 min). However and probably due to the complexity of pyrolyzates, there are few publications reporting this combination [4, 12]. On the other hand, GCxGC provides better resolution power and higher peak capacity by adding a second GC column, with different polarity and connected through a modulator (usually a cryo-modulator), which traps the eluting fractions coming from the first column and injects them into the second one achieving true two-dimensional separations [14, 16, 19]. In the future, this technique could add a third dimension of separation if it is used together with heart-cut EGA [2]. Nevertheless, it is important to highlight that the improvement possibly gained by the combination of these techniques will require enhanced data processing tools due to the great amount of information generated and the time needed to interpret it.

The detection technique of pyrolyzates is typically mass spectrometry (MS) often accomplished by a single quadrupole mass analyzer due to its robustness and capacity to detect the fragmentation products. The molecules detected by MS are

identified through their mass spectra using commercial libraries (e.g., NIST/EPA/NIH, Wiley, MPW, Norman Mass Bank, m/z Cloud) or by means of reference substances [4]. Despite triple quadrupole instruments (QqQ) are nowadays the most recommended systems for the analysis of target compounds (because of their higher sensitivity and specificity), it is not commonly used in combination with Py–GC since the compounds formed in the pyrolysis are mostly unknown and the application of the QqQ is less evident [7]. In contrast, high-resolution mass spectrometry (HRMS) and mainly (quadrupole) time-of-flight (TOF or QqTOF) has become essential in Py–GC–MS analyses since these detectors provide information on the most probable empirical formula of both the analyzed molecule and its characteristic fragments. Although HRMS application focuses on using selected ion monitoring, the analysis of the full spectra in the development of the method is mandatory [12, 26].

### **3 Applications of Py–GC–MS in Food and Environmental Analysis**

Py–GC–MS can be used in a wide array of applications from characterization of materials (polymers, lacquer films, etc.) to quality control (of food and daily commodities) including also forensic analyses in environmental sciences among others. In recent years, the most interesting applications of Py–GC–MS in environmental samples are (1) the chemical fingerprinting of organic matter (OM) [8, 20–26] and (2) the identification, characterization, and fate of MPs and NPs [28–32, 34, 35, 43, 44]. Furthermore, Py–GC–MS main application in food are related to (1) the determination of geographical origin, (2) detection of fraudulently added substances, and (3) quality control.

#### ***3.1 Characterization of Food***

Because of its versatility, Py–GC–MS offers ample possibilities to be exploited in multiple applications related to food industry, from food authentication to the analysis of composition, quality, and additives, including contaminants as well. One of the first studies using Py–GC–MS in food analysis was published in 1999 [10]. It presented the results obtained by the European Commissions' Joint Research Centre for the detection of added whey proteins in milk, the ripening stages of cheese, the characterization of vinegar and wine, as well as the assessment of the geographical origin of cocoa butters.

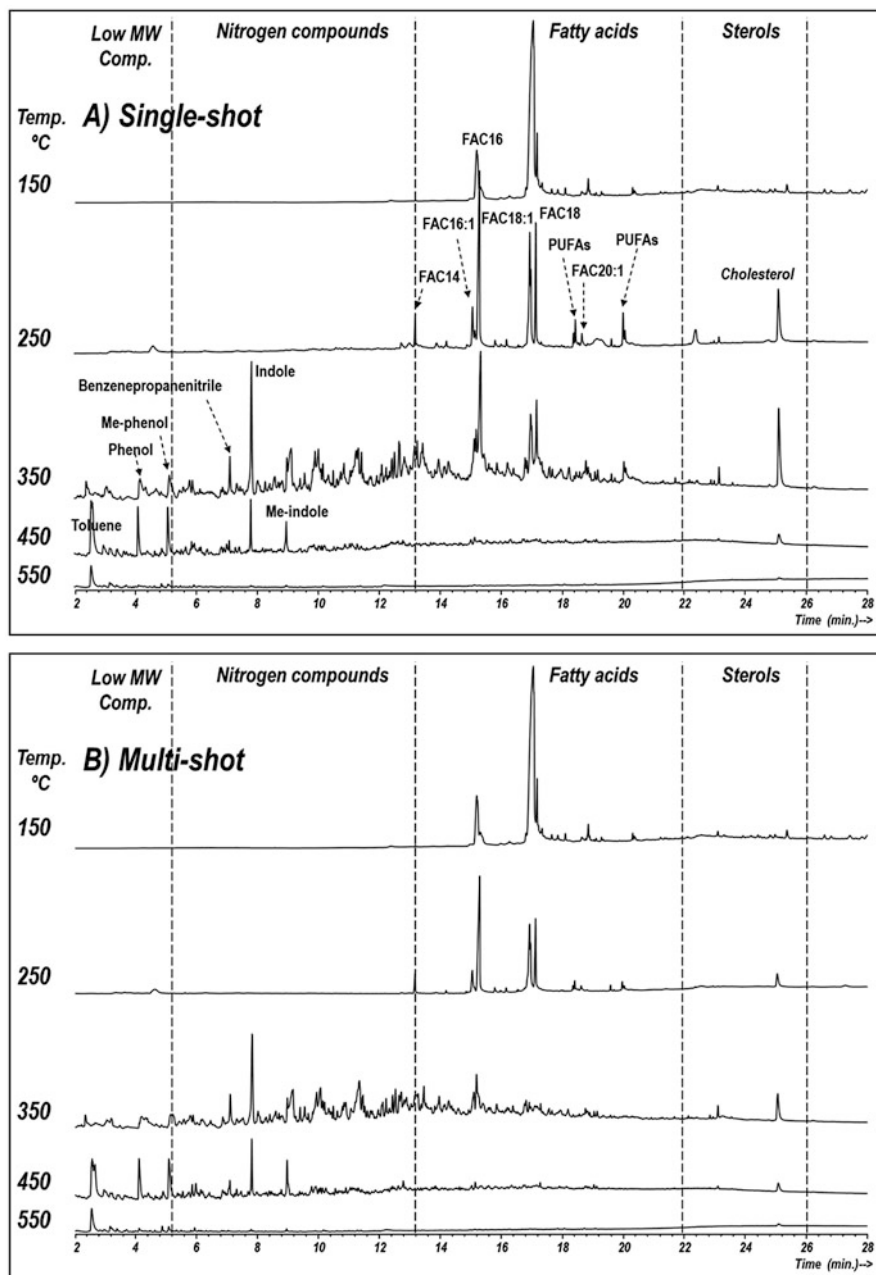
Nowadays, the isotope ratio mass spectroscopy (IRMS) for determination of the stable isotope composition of light elements is an interesting approach to discriminate production areas and to validate the origin of vegetables and fruits. The

analysis of  $\delta^{13}\text{C}$  in discrete oil components using a compound-specific isotope analysis (CSIA) is an accurate method compared to the isotope assessment in bulk samples. Direct analytical Py-CSIA was used for the authentication of extra virgin olive oils avoiding chemical and/or physical pre-treatments [11]. The  $\delta^2\text{H}$  value in nine pyrolysis compounds detected in all extra virgin olive oils ranged from  $-112$  to  $-267$  mUr. These compounds were chosen as likely surrogate descriptors of the olive oil geographic origin as they were significantly correlated with longitude and annual temperature by means of Principal Component Analysis and Multiple Linear Regression. Similarly, and because of its capacity to analyze macromolecules, Curie-point-Py-GC-MS has been used to characterize the solvent extracts (ethyl-acetate, n-butanol, raw theabrownin) from a type of Pu-erh tea and Dian Hong black tea. Results showed substantial different chemical compositions among solvent extracts from both teas, and proved the great characterization capacity of the technique [45].

Py-GC-MS has demonstrated to be a useful tool able to provide precise fingerprinting but also detailed information about quality, composition, and additives in active packages of biogenic polymers employed in food industry. One of these materials is the oregano essential oil, which is used due to its bioactive properties. However, because of oregano essential oils high volatility, characteristics like aging and preparation conditions of films can produce losses of active agent. Py-GC-MS was used to characterize a bioactive polylactic acid with polybutylene succinate (95:5) film extruded with variable amounts of oregano essential oil [13]. Bioplastic pyrolysis enables identifying lactide enantiomers and monomer units from the polylactic acid and unambiguous molecular markers from the polybutylene succinate fraction. Oregano essential oil pyrolysis revealed aromatics and terpenes with terpinene, cymene, and thymol/carvacrol peaks that were identified as diagnostic peaks showing significant linear correlation coefficients with the amount of oregano essential oil included in the bioplastic matrix.

Other applications of Py-GC-MS in food industry are focused on the determination of contaminants. Using this technique, the thermal stability of five pharmaceuticals commonly detected in food (chloramphenicol, florfenicol, lincomycin hydrochloride, diphenhydramine hydrochloride, and carbamazepine) was analyzed [46]. Results showed a number of volatile thermal degradation products (some of them previously unidentified) with a technique simpler than others reported earlier. The test was conducted at  $250^\circ\text{C}$  (i.e., above actual cooking temperatures of  $180$ – $220^\circ\text{C}$ ) for 20 s only, a very short period of time (compared to cooking times). However, it seems to reveal information about the degradation products generated at lower temperatures or longer cooling times. Authors concluded that in future studies, lower pyrolysis temperatures should be explored to simulate the general cooking conditions and provide some insight in the fate of drug residues in food during processing. In this regard, the effect of pyrolysis temperature from  $150$  to  $550^\circ\text{C}$  with  $100^\circ\text{C}$ -increments was studied in raw or cooked fish muscle samples (*Oreochromis niloticus*) to detect cylindrospermopsin and its degradation products by both (1) applying each temperature to a different sample (single-shot pyrolysis) or (2) sequentially applying each temperature to the same sample (multi-shot pyrolysis) (Fig. 4) [47]. Relative abundance of degradation products was found





**Fig. 4** Analytical pyrolysis of tilapia muscle. Total current ion chromatograms (TIC) performed at (a) single-shot mode, i.e. different sample at different temperatures and (b) multi-shot mode, i.e. same sample at increasing temperatures. Chromatograms are divided in sections of preferential compound classes and major compounds detected are labeled on the corresponding peaks. *FA*: fatty acid, *PUFA*: polyunsaturated fatty acid, *Me*: methyl [47]

to be different depending on the cooking process (microwaving, broiling, boiling, and steaming) being boiling the only cooking technique that showed to decrease the relative percentage of cylindrospermopsin compared to the control group.

Thermal desorption using Py–GC–MS was used together with a novel sorbent developed for sorptive enrichment of dilute analytes from liquid samples [48]. The extraction element is made of titanium open tubular tube (30 mm × 1.2 mm i. d. × 1.6 mm o.d.) coated with a chemically bonded layer of polydimethylsiloxane (PDMS 500 µm in thickness). This sorbent was used to develop a method to extract and analyze bethrodine (a herbicide) in water samples over a concentration range from 2.5 to  $2.5 \times 10^4$  ng/L. A detection limit of 0.5 ng/L was achieved with fairly good reproducibility of the measurements (relative standard deviation, R.S.D., below 7.5% at 10 ng/L). This procedure was also applied for the simultaneous determination of five preservatives – benzoic acid, sorbic acid, and methyl, ethyl, and propyl esters of *p*-hydroxybenzoic acid – in soft drinks, yogurts, and sauces [49]. The method attained limits of detection between 0.002 and 0.2 mg/L, and it was applied to real samples spiked at levels of between 20 and 100 mg/L showing good recoveries (92–106%) and precisions (R.S.D.s of 0.9–4.6% ( $n = 5$ )).

### 3.2 Determination of MPs and NPs

Environmental contamination by MPs and NPs is a global problem with a worldwide-recognized dimension as proven by the large number of studies published in recent years [28–35, 43, 50–55]. Py–GC–MS has become one of the most promising techniques to assess MPs and NPs in environmental samples since it is able to detect low MP dimensions, and even NPs (the most difficult ones because of their size). It is more sensitive and less affected by sample impurities than other methods. Py–GC–MS has been used to identify MPs in soil [56, 57], soil amended with solid waste compost [58], plants [27], bio solids [59], river sediments [40, 60], beach sediments [61], coastal sediments [30, 62], tidal flat sediment [63], suspended particulate matter [40], wastewater [64], sea water [65, 66], surface water [28, 52, 67, 68], salt [63, 69], bivalves [66, 70–72], fish [30, 43, 47, 51, 73], and other types of biota [52, 73]. Py–GC–MS has also been applied to determine NPs in water [74–76].

MPs can be directly assessed from the samples, such as soil [58], but they are usually separated and concentrated from the matrix by wet digestion (using oxidants, acids, or enzymes [62, 77]), or by density differences (with solutions of NaCl [61], NaI, ZnCl<sub>2</sub>, NaBr [69], or sodium polytungstate [60]), with a final filtration. In samples with high OM content, as wastewater, sequential filtration can help to isolate the MPs without clogging the pore filters [64]. Very recently, pressurized liquid extraction has been applied to solubilize MPs in solvents such as dichloromethane or tetrahydrofuran [44, 59]. NPs can also be extracted from aqueous samples by filtration working in the range of nanoscale (24 nm–52 nm) [75]. Ultrafiltration has also been used to separate MPs and NPs from water attaining lower sizes (5–50 nm) [74]. Other techniques as Triton X-45-based cloud-point

extraction has been tested to pre-concentrate NPs providing an enrichment factor of 500 without disturbing their actual morphology and sizes [76].

In the determination of MPs and NPs by Py–GC–MS, the pyrolyzer most frequently used is the microfurnace [59–63, 69, 76], because of its higher sample capacity, though filament [57, 58, 75] and Curie-point [57, 58, 75] pyrolyzers have also been applied. A comparison of micro furnace and Curie-point pyrolyzers to assess MPs in the environment proved that the former can process more sample quantity due to its larger sample cups allowing the transfer of pre-concentrated MPs directly from the glass fiber filters [63]. The double-shot pyrolysis is the preferred mode since the first-shot can act as a clean-up step avoiding interferences as those of OM [40, 57, 59, 62, 76], despite none of the OM compounds can interfere with the indicators selected for polypropylene (PP), polyethylene terephthalate (PET), polycarbonate (PC), poly-(methyl methacrylate) (PMMA), and polystyrene (PS).

The GC–MS is commonly used with a quadrupole mass analyzer although TOF analyzer has also been reported in studies of MPs and NPs in water [75] and wastewater samples [34]. In the former study, TOF was operated in nominal mass units not in accurate mass, similar to a quadrupole but offering an increased sensitivity because all the ions accumulated are transmitted to the TOF detector. In most cases, selected ion monitoring (SIM) mode is the only working mode that attains sufficient sensitivity to detect MPs at low concentrations. It is important to keep in mind that other organic molecules (e.g., organic matter) can be present in the extract/sample leading to a number of non-volatile pyrolysis products, which enhance the risk of interferences with specific indicators of compounds, boost the maintenance frequency, and hamper a reliable quantification. As it has been explained, compounds in the pyrograms can be identified by means of mass spectral databases, custom databases containing pre-acquired pyrograms of plastic references, or by comparison with the published scientific literature [61, 68]. In general, to identify and quantify MPs and NPs, specific indicator compounds are selected for each type of plastic as presented extensively by Picó and Barceló [2]. These indicator compounds were chosen after assessing their specificity against a number of natural materials as chitin, wood, pine needles, cellulose, humic acid, etc. [59]. However, this method of data processing is very time consuming. In this sense, the research moves toward the design of software for automatic qualitative analysis of this information as an automated algorithm developed and implemented in the F-Search software [18]. The process is based on the generation of summated mass spectra (SMS) for each polymer, obtained by extracting specific  $m/z$  and retention index coordinates of the characteristic pyrolysis products. The identification of a polymer is performed by comparing its SMS with those of a built-in library. After validation, the algorithm was tested on a reference sample containing 11 types of synthetic polymers providing relative standard deviations around 10%. The lowest estimated amount of polymer detectable in a sample, was lower than 1  $\mu\text{g}$  for most polymers.

Reactive pyrolysis after TMAH derivatization has been able to reduce organic interferences and because this technique improves the detection sensitivity for PET and PC [60, 69, 77]. When pyrolysis is used in combination with TMAH, the

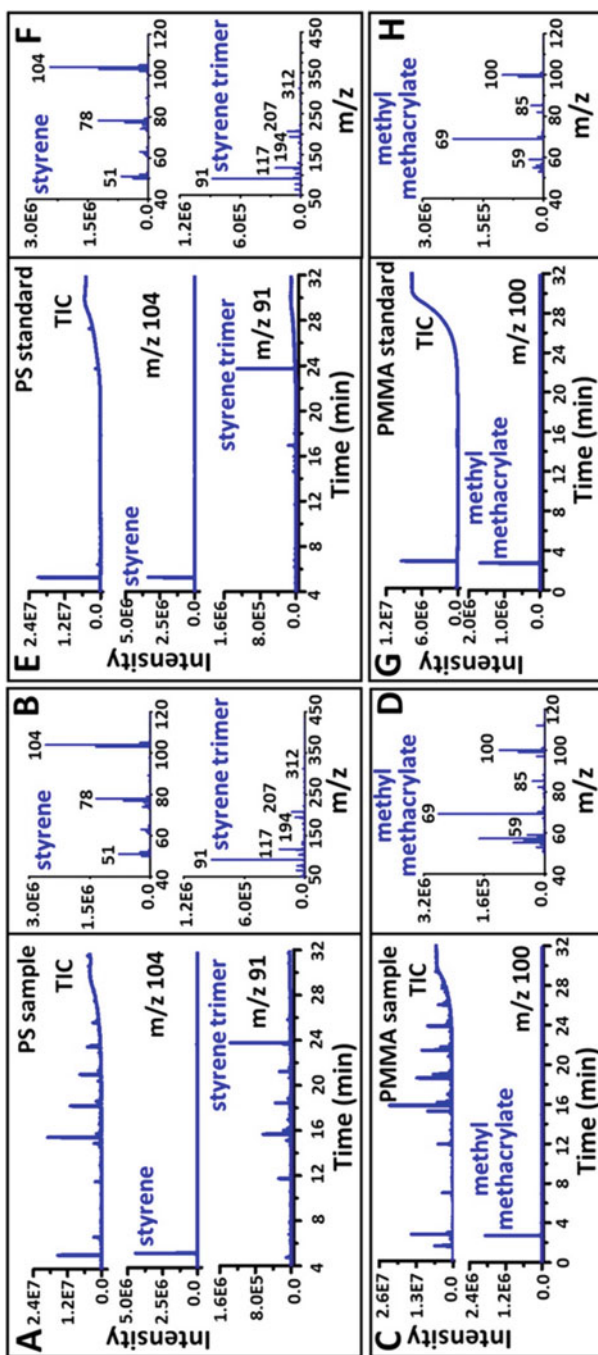
pyrolytic behavior of polyethylene (PE), PP, PS, and polyvinylchloride (PVC) remained unaffected while that of PET, PMMA, PC, and polyamide (PA6) changed. As an example of the complexity of the technique and the problems originated by interferences, determination of PS requires the selection of a characteristic PS trimer ( $m/z$  312), even after derivatization, because chitin (widely present in the natural environment) releases styrene ( $m/z$  104) during pyrolysis (Fig. 5) [76].

The first international attempt (17 laboratories from eight different countries) to compare the suitability of the methods commonly used in MPs analysis showed the performance of Py-GC-MS within the field as well as the main obstacles faced by researchers [78]. Different techniques as microscopy, Fourier-transform infrared microspectroscopy ( $\mu$ -FTIR), Raman microspectroscopy ( $\mu$ -Raman), thermal extraction and desorption or Py-GC-MS, scanning electron microscopy, and particle counter were compared in relation to the total particle number, polymer type, number of particles, and/or the particle mass of each polymer type. For the identification of polymer type,  $\mu$ -Raman and Py-GC-MS performed best. The quantification of polymer mass for identified polymer types was questionable for Py-GC-MS, whereas other methods failed to determine the correct polymer mass. Quantification of particle number per identified polymer type was evaluated successful for  $\mu$ -FTIR and the quantification of total particle numbers was best for microscopy and to a lesser extent for  $\mu$ -FTIR [78].

The performance of Py-GC-MS has also been compared to  $\mu$ -FTIR and  $\mu$ -Raman, as most successful complementary approaches in the identification of MPs [60]. Py-GC-MS identified copolymers as PE-PP or PE-PP-PA6 that could be difficult to distinguish with  $\mu$ -Raman without chemometrics approach leading to results with a finer identification [70]. Chemical information provided by Raman and FTIR spectra is limited compared to the one of Py-GC-MS. This is because the selectivity of vibrational spectroscopy is not enough to identify specific monomers and co-monomers, mixtures, additives, and degradation products [79]. Py-GC-MS is able to characterize MPs at a molecular level identifying synthetic polymers as well as the possible presence of additives in contrast to the second common limitation of Raman spectroscopy and ATR-FTIR that provided spectra relative to a surface portion of the MP fragments [60, 70]. The advantage of Py-GC-MS over FTIR spectroscopy is that both polymer types and organic plastic additives can be studied in a single run [80]. Due to Py-GC-MS important advantages, new statistical approaches to tackle polymer variability and matrix interference are being developed [79]. This augurs that pyrolysis will exponentially increase its applicability for analyzing MPs and NPs in the environment.

### 3.3 *Characterization of Organic Matter*

Py-GC-MS has also demonstrated to be a powerful tool for analyzing the molecular fingerprint of organic matter (OM) in soils and sediments [5, 23, 24, 39, 40, 81–83], as well as to analyze changes in the dissolved organic matter (DOM) fraction [84–



**Fig. 5** Py–GC–MS chromatograms of PS- and PMMA-nanoplastic samples and standards: (a) Total ion chromatogram (TIC) and selected-indicator-ion chromatograms (SIC) of the PS-nanoplastic sample. (b) Mass spectrum of the PS-nanoplastic-sample pyrolytic product. (c) TIC and SIC of PMMA-nanoplastic sample. (d) Mass spectrum of the PMMA-nanoplastic-sample pyrolytic product. (e) TIC and SIC of the PS-nanoplastic standard. (f) Mass spectrum of the PS-nanoplastic-standard pyrolytic product. (g) TIC and SIC of the PMMA-nanoplastic standard. (h) Mass spectrum of the PMMA-standard pyrolytic product [76]

86]. While in water the sample should be filtrated through a glass filter to later isolate and enrich the DOM, by reverse phase mechanisms, as SPE [84] and pre-HPLC [85], in soil and sediment the sample just need to be dried, sieved, and pulverized [5, 39, 40]. DOM can also be extracted from soils using a lysimeter-pump or by shaking soil with water [87].

The study of OM includes the characterization of a diversity of compounds including polysaccharides, amino sugars, proteins, polyhydroxy aromatics, lipids, lignin, etc. Other compounds as humic acids and fulvic acids need prior extraction and clean-up following standard methods (these include elimination of the free OM by flotation with  $H_3PO_4$ , demineralization with  $Na_4P_2O_7$  to release the humic substances complexed with oxides, then, extraction with basic solutions as NaOH, and separation by precipitation in an acidified solution,  $pH < 2$ ). Humic acids are later purified by washing with several acidic solutions [83] and/or dialyzed into cellophane bags [20, 88]. The three types of pyrolyzers described previously have been applied to characterize OM: micro furnace [39, 40, 83, 89, 90], platinum heated filament pyrolyzer [5, 20, 86, 87, 91], and Curie-point [85, 88]. Most of these studies have only used the single-shot at temperatures between  $500^\circ C$  and  $800^\circ C$  [5, 20, 39, 40, 83, 85, 87–91]. The double-shot pyrolysis that makes accessible both free volatiles and high molecular weight structures has been hardly used to characterize OM (probably because OM is formed by non-volatile compounds). The reactive pyrolysis has also been tested in few studies [87, 92] since it simplifies the pyrogram providing less fragments. However, a comparison of the ability of both techniques to characterize DOM in soil from the Three Gorges Reservoir area stated that Py–GC–MS fingerprinting is more helpful to quantify microbial DOM than reactive pyrolysis [86], which is useful to assess the molecular features of polyphenolic (cinnamic acids, lignin, and tannin) and aliphatic (cutin and suberin) fractions.

OM characterization by GC–MS is generally performed using a 30 m capillary column and a simple quadrupole in scan mode. Compounds detected in a pyrogram can range between 100 and 400 and are identified by means of a mass spectral database or library as previously described. Their relative abundance can be calculated by normalizing the peak areas of each individual compound to the total area of all the peaks detected. In general, to avoid the high number of pyrolyzates in a sample, the best is to focus the study on a few good biomarkers giving information about the process of interest. For instance, reactive Py–GC–MS has been used to identify branched-chain fatty acids in humic acids to describe microbial activities occurring during composting processes [92]. Methoxyphenols (12 major guaiacyl- and syringyl-type compounds) have also been detected by Py–GC–MS in topsoil samples since they are distinctive molecules advising on the occurrence and degree of alteration of lignin, and have also been proposed as indicators of the soil's capacity to storage C [82, 89].

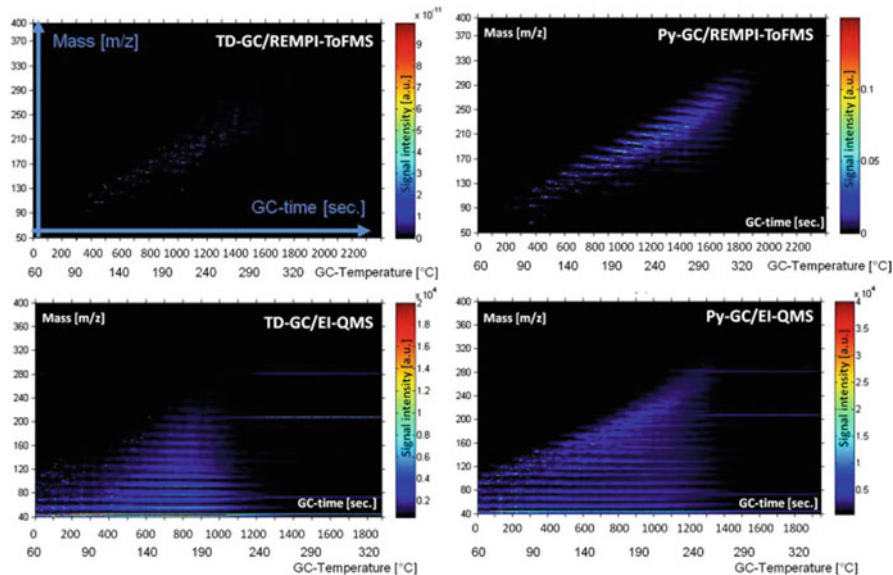
Studies about changes in OM after different disturbances as forest fires [21–23, 81] or land use changes [5, 82] have also been performed. The thermal transformation of soil OM caused the thermal breakdown and cracking of n-alkane compounds, as was revealed by the increase of the ratio of short-to-long chains and the alteration of the typical odd-over-even carbon predominance indexes [21, 22]. Analyses

carried out by using a vertical microfurnace type double-shot pyrolyzer indicated that the fire resulted in an enrichment of aromatic compounds, nitrogen (N) constituents, lignin-derived compounds, and polysaccharides [23, 81]. The effect of land use changes in OM compounds has been studied using different approaches including Py–GC–MS. Zhang et al. [5] used a heated platinum filament pyrolyzer directly connected to a GC–ion trap MS to analyze the effects of freshwater wetlands conversion to sugarcane (*Saccharum officinarum*) on the soil OM cycling and processes. These authors concluded that the land use change not only reduced total OM contents but also significantly increased the aliphatic fractions while lignin fractions were almost completely exhausted. In another study, Campo et al. [82] by means of Curie-point pyrolysis with tetramethylammonium hydroxide assessed the effects of afforestation on the soil lignin content (based on the vanillyl, syringyl, and cinnamyl contents). According to them, afforestation with *Pinus nigra* was the best practice for increasing the soil organic carbon stock and the lignin content in soil.

Py–GC–MS using two ionization systems, electron-ionization (EI) and photon-ionization (PI), with different mass selective analyzers (quadrupole- and a time-of-flight, TOF-MS) has been performed for the analysis of DOM in water [84]. In this research, resonance-enhanced-multi-photon-ionization (REMPI) provided information on the molecular weight and a high sensitivity and selectivity for aromatic hydrocarbons. Consequently, the characterization of natural samples by EI-quadrupole/MS and their aromatic fingerprint (REMPI-TOF-MS) were accomplished (Fig. 6). In general, there are few studies incorporating the recent developments of GC–MS, possibly because these analyses are complex and time consuming besides needing highly specialized technicians to deal with the high amount of data generated. In order to overcome this difficulty, statistical tests have been applied to improve and schematize the visualization/interpretation of the information obtained. As pyrolyzates can fluctuate among samples in their presence and intensity, a lot of effort is devoted nowadays to develop automated identification and quantification software that can help to process all the information obtained from the pyrographs and to visualize the results [20, 93].

## 4 Conclusions

Py–GC–MS is recognized as a valuable technique in food and environmental chemistry as proved by the increasing number of studies using it. Because of its versatility, Py–GC–MS offers ample possibilities to be exploited in multiple applications related to food industry, from food authentication to the analysis of composition, quality, and additives, including contaminants as well. It should be kept in mind that pyrolysis is a time-consuming and complex technique, then, it will be used only in those cases where the other chromatographic techniques failed, such as in the characterization of macromolecules. In recent years, it has been applied to the chemical characterization of organic matter in water, soil, and sediment but also,



**Fig. 6** Direct comparison of the chromatograms resulting from the REMPI-TOFMS (above) and EI-QMS (below) detection for At4 station (Bothnian Bay). The TD step is shown on the left and the pyrolysis step on the right. On the ordinate the GC-time and -temperature are plotted. On the abscissa the mass-to-charge ratio is depicted. Graduated in color are the intensities [84]

to assess the modifications that disturbances as forest fires or land use changes produced. Evaluation of the contamination by MPs and NPs of environmental samples has also become a hot topic addressed by the application of this technique. Py-GC-MS has demonstrated important advantages over others, such as  $\mu$ Raman and  $\mu$ FTIR to evaluate MPs, since it has been the only one able to detect NPs. The use of very small quantities of sample and the lack, in most cases, of pre-treatment have also added value to this technique. However, the large amount of results obtained and the difficulty in interpreting them are constraints that also need to be considered. On the other hand, most of the studies are focused on the robustness of the pyrolyzer and therefore, latest innovations in GC-MS have not been yet introduced in most applications. In this sense, it is expected that Py-GC-MS takes advantage of the great separation power of GCxGC, the speed of rapid GC, and the identification capability that HRMS and HRMS in tandem can provide. Furthermore, environmental analyses can also be enhanced by the application of EGA or shot-cut-EGA that would add a further dimension to multidimensional techniques. Py-GC-MS is a robust, multipurpose, adaptable, and useful technique with a promising future to overcome new challenges as those depicted in this chapter.

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# Chiral Analysis with Mass Spectrometry Detection in Food and Environmental Chemistry



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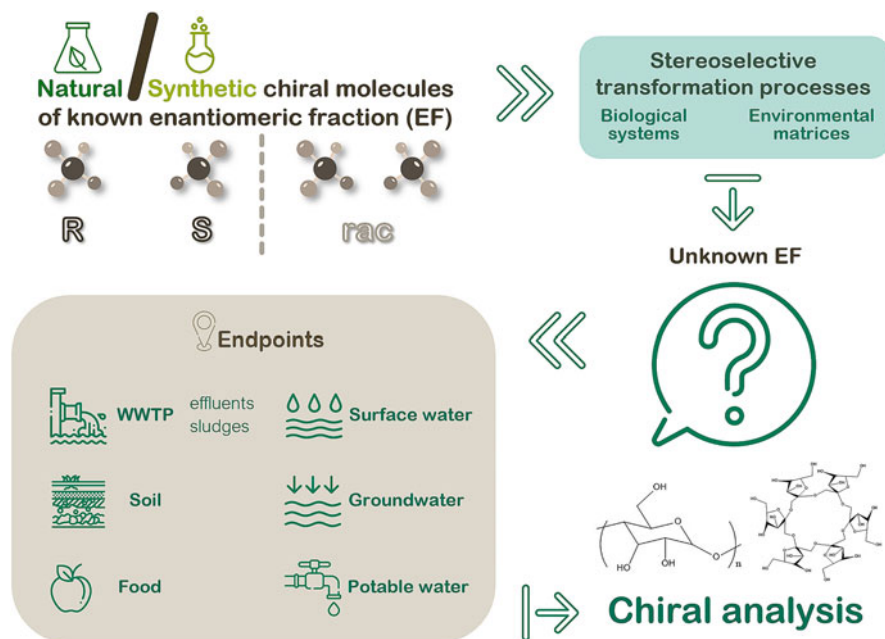
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**Abstract** The challenge to develop enantioselective analytical methods to quantify chiral compounds in food and environmental matrices is an actual and imperative issue. It is well known that enantiomers may differ in their biological activities, but the stereochemistry and the discrimination of enantiomers are frequently ignored in analytical workflows. However, the knowledge about their proportion is crucial to guarantee food and environmental safety. In this sense, chiral analysis is a valuable tool to evaluate food quality and genuineness, to determine the geographical origin of a certain sample aiming to find out fraud or adulteration, to investigate toxicity and bioaccumulation of environmental contaminants (e.g., pharmaceuticals and pesticides). This chapter highlights the importance of chiral analysis in food matrices and its relationship with the environmental contamination. The last advances in chiral analysis by liquid chromatography (LC), gas chromatography (GC), and supercritical fluid chromatography (SFC) coupled to mass spectrometry (MS) are presented, with a special remark in the type of chiral column used for chromatography and the proposed workflow in the method development. Some of the most recent applications in chiral analysis of food and beverages, environmental monitoring of surface water, wastewater, soils, biodegradation studies, and contamination of foodstuff are presented and critically discussed.



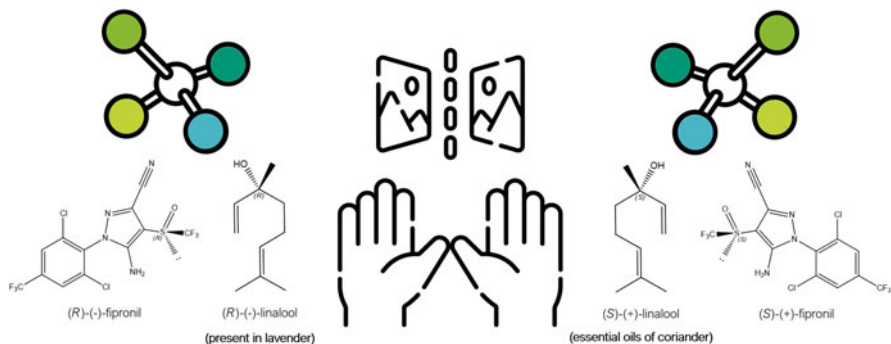
## Graphical Abstract



**Keywords** Chiral chromatography, Chiral stationary phase, Enantiomers, Pesticides, Pharmaceuticals

## 1 Introduction

Chirality is a geometric property associated with the asymmetry and three-dimensionality of molecular structures. Asymmetric three-dimensional molecules with the same atomic connection but with different spatial arrangement generate two non-superposable structures called enantiomers [1]. Chirality is often originated by a tetrahedron carbon with four different substituents, or other atoms such as sulfur, phosphorus, and silicon, which generates a stereogenic center that can confer asymmetry to the molecules (Fig. 1). However, beyond the central chirality, other elements of chirality are described such as planar, axial, and helical, found in asymmetric structures without stereogenic centers [1]. Due to their structural likeness, enantiomers possess similar physical and chemical properties, apart from the rotation of the polarized light. The enantiomers that rotate the polarized light to the right (clockwise) are called dextrorotatory (+), and those that present rotation to the left (counter-clockwise) are levorotatory (−). Regarding their nomenclature, the structure configuration is related to the spatial orientation of the substituents of the

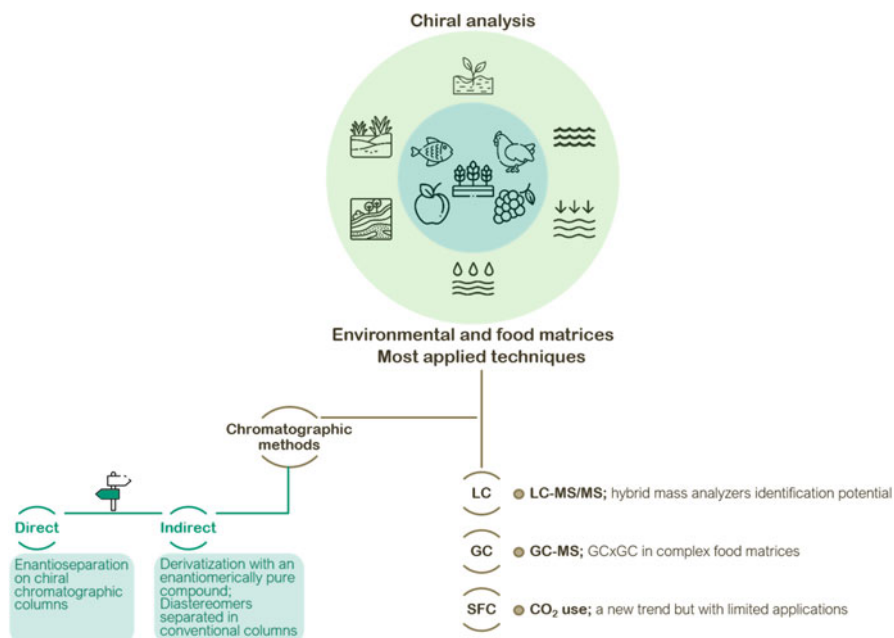


**Fig. 1** Example of enantiomers of linalool and fipronil. The *(R)*-(-)-linalool has a lavender or lily-of-the-valley odor and *(S)*-(+)-linalool has a herbaceous and musty green smell [3]. Fipronil is a widely used insecticide, *(S)*-enantiomer is more active as insecticide and safer to mammals, in comparison with *(R)*-fipronil [4]

stereogenic center according to the specific rules established by the International Union of Applied Chemistry (IUPAC), which can be *(R)* from the Latin *rectus*, or *(S)* from the Latin *sinister* (Fig. 1) or L- or D- for amino acids and sugars [2]. The equimolar mixture of both enantiomers is denominated racemate.

Enantiomers are frequently overlooked in analytical methods and measured as a unique molecular entity. Nevertheless, the importance to assess both enantiomers in analytical methods comes from their different behavior in chiral medium, as occurs in the metabolic and regulatory processes mediated by biological systems that are sensitive to stereochemistry [1, 5, 6]. The essential subunits of the biological systems, namely amino acids, peptides, sugar, and nucleic acids are the chiral building blocks of proteins, glycoproteins, and deoxyribonucleic acid, respectively, resulting in the intrinsic chirality of the final arrangement. Therefore, the enantiomers can interact in different ways with enzymes, receptors, transport systems, or other binding chiral molecules displaying different responses (enantioselectivity/enantiospecificity) such as dissimilar taste, aroma or diverse biological activities. Enantiomers may present similar biological response but differ in potency or their biological activity can be completely distinct, as occurs for many pharmaceuticals, which one enantiomer is pharmacologically active, and the other enantiomer can be inactive or toxic. For instance, the biochemical processes of living organisms are based on D-sugars and L-amino acids, though the presence of D-amino acids in human proteins plays a crucial role in human health and the development of diseases [7]. As examples, D-Asp in elastin,  $\beta$ -amyloid, and  $\alpha$ -crystalline have been associated with arteriosclerosis, Alzheimer's disease, and cataracts, respectively [7].

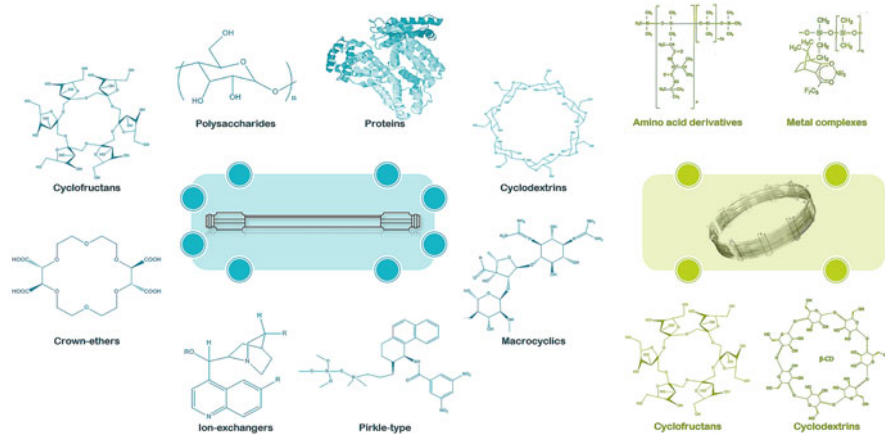
Nowadays, liquid chromatography (LC) or gas chromatography (GC) associated with mass spectrometry (MS) are the most applied methodologies in chiral analysis of food and environmental matrices. Nonetheless, most recently supercritical fluid chromatography (SFC) has also become a new trend. As enantiomers possess the same molecular mass and fragmentation pattern in MS, the enantioseparation by



**Fig. 2** Representation of the main analytical approaches for chiral analysis in environmental and food matrices

chromatographic methods is indispensable for their identification and quantification. There are two ways to separate enantiomers by chromatography: the direct and indirect methods (Fig. 2). The indirect methods require derivatization toward the formation of diastereomers that have different physico-chemical properties and can be separated in conventional chromatography columns. However, this approach is only possible if the racemate has at least one suitable functional group to react. On the other hand, direct methods do not require previous derivatization due to the presence of a chiral selector in the stationary phase or an additive in the mobile phase. The enantioseparation by direct methods occurs by the continuous formation of transient diastereomeric complexes between the target enantiomers and the chiral selector, resulting in different retention times in the chromatographic methods, i.e., enantioseparation. The use of chiral selectors as additives in the mobile phase is a more complex approach because the selectors can interfere with the detection mode. Moreover, the considerable amount of chiral selector required is difficult to recover, resulting in production of waste at large scale.

The chiral selectors in the stationary phase are adsorbed or covalently bonded to a solid chromatographic support. Most reported methods for chiral analysis, especially by LC, are performed by direct methods involving the use of chiral stationary phases (CSPs) (Fig. 2). A variety of CSPs are nowadays commercially available. Figure 3 shows the most applied CSPs used for LC and GC. Regarding LC, the majority of



**Fig. 3** Most applied CSPs for chromatographic methods

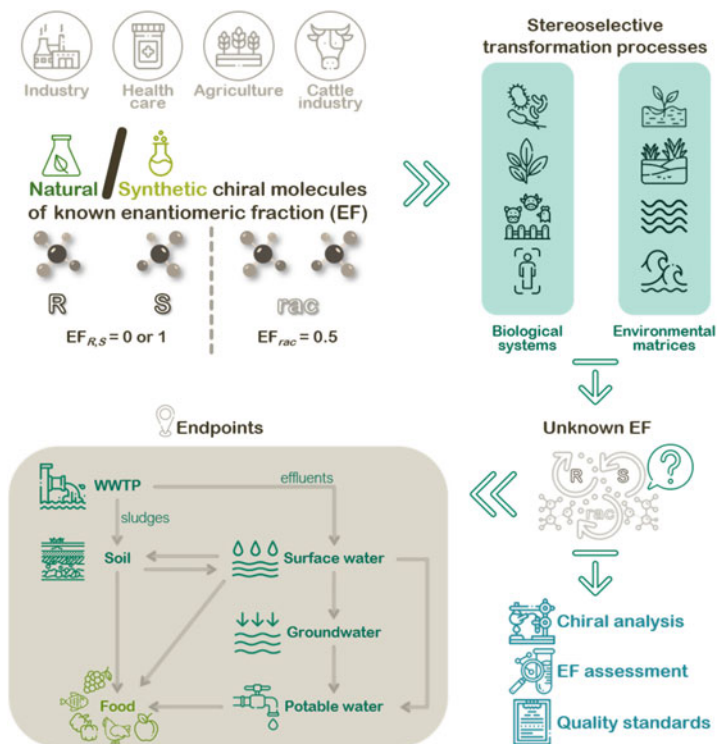
CSPs can be operated in normal, reversed, polar organic, or polar ionic elution modes. The nature of the chiral selector, the mobile phase composition, and the molecular structures of the enantiomers determine the chiral recognition mechanism for the enantiomeric resolution. For more information on models of chiral separations and CSPs, readers can access to some excellent reviews in this field [8–14].

Despite the variety of analytical methodologies to monitor food quality and environmental contamination, only chiral analyses assure a high level of safety and risk assessment since they allow to quantify both enantiomers, which is not possible by conventional analytical methodologies.

The interconnection of environmental pollutants and food safety together with the importance of chiral analysis is schematic represented in Fig. 4. This chapter presents the latest advances in chiral analysis of food and environmental matrices by chromatographic techniques coupled to MS. The proposed workflow in the most innovative methods, including sample preparation, is presented and critically discussed, together with the bottlenecks of the methodologies and some suggestions for further improvements and wider applications.

## 2 Chiral Analysis in Environmental Chemistry

Despite being well known that enantiomers may differ in their biological activities, the stereochemistry is often ignored in analytical methods applied to environmental studies, which can lead to inaccurate data [15]. However, an increasing interest in enantiomers quantification and evaluation of the enantiomeric fraction (EF) has been observed recently [16]. In fact, several studies have been demonstrating the concerning about environmental occurrence of organic pollutants (e.g., pharmaceuticals and personal care products, illicit drugs, pesticides) [17–19]. The constant



**Fig. 4** The interconnection of environmental pollutants and food safety together with the importance of chiral analysis

introduction of an enormous number of pollutants at low concentrations (ng/L to  $\mu\text{g/L}$ ) in water bodies and soils results from different origins, including domestic, hospital, and industrial wastewater effluents, agriculture, livestock, aquaculture, and landfill leachates (Fig. 4) [17]. Wastewater effluents represent a major pathway of entry of many contaminants as wastewater treatment plants (WWTP) are not designed to remove them. Consequently, the non-degradable or partially eliminated compounds, the metabolites resulting from microbial degradation, and other transformation products resulting from abiotic processes (e.g., hydrolysis and photodegradation) can pseudo-persist in the environment. In the case of suspended solids, sewage sludge or animal manure applied to soils, the sorbed pollutants can desorb and be leached into aquifers with subsequent contamination of water, plants, and food chain [20].

These interconnected pathways together with the recalcitrance and toxicity of most pollutants create a favorable scenario for possible impacts in the ecological status and human health (Fig. 4). Although the research in this topic has evolved massively in the last two decades, much less is known about the chiral signatures of

contaminants in environmental studies, which is a valuable tool for many purposes, namely to:

- get information about the environmental occurrence and fate of chiral pollutants and their metabolites (e.g., wastewater, sewage sludge, surface water, sediments, soils, biota) [21];
- use the data for wastewater-based epidemiological studies (WBE), e.g., to get information on the consumption patterns of pharmaceuticals or illicit drugs [22];
- understand their sources (e.g., to identify the use of banned pesticides, clandestine laboratories) and distinguish between licit and illicit production/usage (e.g., to identify the synthesis pathway or to detect illicit sewage discharges) [22];
- evaluate their (bio)degradation and (bio)adsorption [23];
- assess the formation of metabolites and other transformation products [24];
- distinguish between biotic and abiotic processes [23];
- study their ecotoxicological effects [25] and bioaccumulation [26];
- afford an accurate risk assessment [27];
- support future prioritization exercises and regulations [27];
- provide data for future remediation approaches [27];
- evaluate the uptake contamination by food web [28].

In fact, the development of enantioselective analytical methods to quantify residues of chiral drugs in environmental matrices is a challenging task for analytical chemists. The EF is a key parameter to assess the environmental fate and distribution of chiral pollutants [27] since its variation depends on biological processes like biosorption, biodegradation, biotransformation, uptake, bioaccumulation, and/or has impact on the toxicological effects. In this sense, many efforts have been applied in the topic of enantioselective analysis [29–32]. In the field of chirality in the environment, enantioselective monitoring has been the most studied subfield, with most reports focusing pharmaceuticals on aquatic compartments and pesticides in soils [21, 23, 33–35]. This type of study is useful to show the enantiomeric enrichment of certain drugs [36], to verify time variability [37], or to study the enantioselective degradation patterns [24, 32].

The latest advances in MS coupled to chromatographic methods (LC, GC, SFC, and capillary electrophoresis (CE)) allied with the diversity of CSPs commercially available have been boosting the application of enantioselective methods [16, 38–41]. In general, the most recent analytical methods are driven by automation, miniaturization, reduction/elimination of waste and hazardous substances, reduction of sample amount and usage of sustainable solvents (e.g., ethanol, ionic liquids or supercritical fluids) [27].

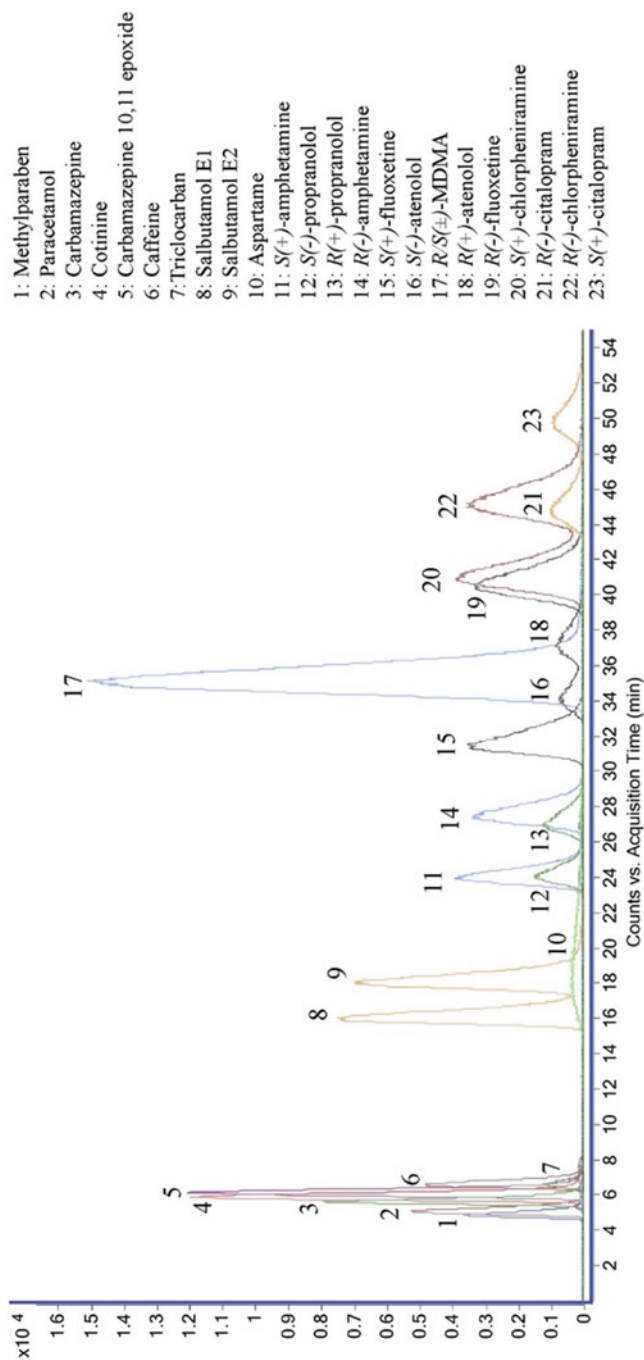
LC using CSPs is the most used technique in environmental studies mainly due to the high number of available commercial CSPs and the versatile multimode elution supported by LC [16, 27]. CSPs currently available that are often used in environmental matrices include protein-based [42], antibiotics-based [31, 43], Pirkle type [31], and polysaccharide derivatives [44, 45]. However, the main bottleneck for a wider application of chiral analysis is the lack of universal or complementary chiral selectors and the difficulty in selecting the appropriate CSP, especially considering

that MS narrows the options available for mobile phase. Moreover, the few available sub-2  $\mu\text{m}$  chiral stationary phases (CSPs) is another shortcoming since this type of CSPs are crucial in reducing the chromatographic run time, to improve efficiency gain, resolution and sensitivity, essential for trace analysis [10, 16].

The versatility, selectivity, and sensitivity of LC-MS/MS are the main contributors for its popularity, with triple quadrupole (QqQ) mass analyzers standing out for trace analysis [27]. In turn, high-resolution mass analyzers are precious for structural elucidation [27]. The latest advances on hybrid mass analyzers as the quadrupole time-of-flight (QTOF) with improved sensitivity, selectivity, and ability for structural elucidation, allow the simultaneous clear confirmation of targets and the structural elucidation of non-targets [16, 27].

The first enantioselective method reported to determine anthropogenic markers in septic tank effluents was carried out by LC-MS/MS using a Chirobiotic V2<sup>®</sup> column that allowed the separation of 16 chiral and achiral anthropogenic indicators under polar organic elution mode [46]. The method reported a high number of chiral compounds with a good enantioresolution, with a relatively short run time for an enantioselective and multi-residue LC-MS/MS method (Fig. 5). In this study, the authors found that amphetamine EF in stream water of rural areas could indicate direct discharge of septic tank wastewater.

Matrix effects may be considered the main drawback of LC-MS/MS for environmental analysis due to the loss of response generated by the co-extracted matrix that can quench the signal of the co-eluting analytes. The first enantioselective method to analyze multi-residue drugs in river sediments reported varying signal suppressions for the 15 target chiral compounds [47], but surprisingly, many selected drugs (e.g., acebutolol, atenolol, citalopram, metoprolol, salbutamol) revealed enantioselective suppression translated by a difference superior to 10% between enantiomers. As example, signal suppression of acebutolol-*E1* and acebutolol-*E2* were -4% and 34%, respectively, showing the need of deuterated surrogates for enantioselective analysis in such complex matrix. However, enantiopure internal standards are scarce and those available are expensive. In turn, another study describing a multi-residue method for the determination of enantiomeric profiles of eight chiral pesticides (diniconazole, metalaxyl, paclobutrazol, epoxiconazole, myclobutanil, hexaconazole, napropamide, and isocarboxiphos) in environmental soils and sediments by LC-MS/MS after a combined sample pretreatment with matrix solid-phase dispersion and dispersive liquid-liquid microextraction (MSPD-DLLME) highlighted that the matrix effect depended on the physico-chemical characteristics of soil and sediment samples, but in this case, the matrix effects were similar for enantiomer pairs in a given matrix [48]. In fact, matrix effect is the reason why sample preparation is a crucial step in environmental analysis. A pioneering approach of sample preparation was reported [31], which was based on the simultaneous extraction of acidic, basic, and neutral analytes (23 chiral pharmaceuticals, stimulants, and illicit drugs). This sample pretreatment skips the cartridge conditioning step and uses an ecofriendly solvent for elution, before LC-MS/MS analysis under reversed elution mode. Chirobiotic<sup>TMV</sup> and a Pirkle type Whelk-O<sup>®</sup>1 CSPs were used to enantioseparate basic and acidic compounds, respectively. As



**Fig. 5** Multiple reaction monitoring LC-MS/MS chromatograms of spiked anthropogenic markers, showing the enantioresolution of salbutamol (8/9), amphetamine (11/14), propranolol (12/13), fluoxetine (15/19), atenolol (16/18), chlorpheniramine (20/22), and citalopram (21/23) – reprinted from *Chemosphere*, 219, Ramage, S., D. Camacho-Muñoz, and B. Petrie, *Enantioselective LC-MS/MS for anthropogenic markers of septic tank discharge*, 191–201, Copyright (2019), with permission from Elsevier [46]



advantages, this method has a reduced environmental impact due to less solvent used and plastic wasted (one cartridge instead of two).

A recent study used an acetonitrile/water extraction combined with graphite carbon black solid-phase extraction (GCB-SPE) to extract penconazole, a triazole fungicide frequently used in agricultural, horticultural, and forestry industries. An enantioselective LC-MS/MS method using a chiral Lux Cellulose-2 column was applied to analyze penconazole enantiomers in complex matrices: grape, tea, soil, lotus root, lotus leaf, lotus seed, and hulls [49]. The method was useful to show that (–)-penconazole degraded faster in grape than the (+)-enantiomer, whereas a modest enantioselectivity was detected in soil, being also (–)-penconazole preferentially eliminated.

A sensitive analytical method was established to determine the enantiomers of the broad-spectrum chiral fungicide pydiflumetofen in food and environmental samples by LC-MS/MS using Quick, Easy, Cheap, Effective, Rugged, and Safe (QuEChERS) as extraction procedure coupled with octadecylsilane-dispersive SPE [50]. This analytical method was assisted by polarimetry and electronic circular dichroism (ECD) to confirm the elution order of the enantiomers on Lux Cellulose-2, being the first eluted peak assigned to *S*-(–)-pydiflumetofen and the second to the *R*-(+)-enantiomer. The figures of merit were consistent with the requirements of trace analysis for 11 types of matrices, including soil where *R*-(+)-pydiflumetofen dissipated faster, which was the opposite trend to the enantioselective dissipation in pak choi. Another study on the enantioselective behavior of triazole pesticide, *cis*-epoxiconazole in the vegetable-soil-earthworm system, was reported [51], using a modified QuEChERS procedure and LC-QTOF-MS, assisted by ECD to assign the absolute configuration of each enantiomer. The method allowed to highlight that distinct enantioselectivities in diverse vegetables might directly or indirectly expose humans to the enriched enantiomer, which brings uncertainty to the environmental risk assessment since the enriched enantiomer might be riskier to human health.

Gas chromatography-mass spectrometry (GC-MS) for enantiomeric profiling in environmental samples has been used under the indirect method using chiral derivatizing reagents (e.g., applications with some pesticides) or under the direct method using CSPs (e.g., applications with polychlorinated biphenyls and polycyclic aromatic hydrocarbons) [21, 39]. Nonetheless, only a few of CSPs are commercially available, being cyclodextrin-based CSPs the most often used [39, 52].

SFC is a multipurpose option that can be used, for instance, to enantioseparate thermolabile analytes or non-volatiles that cannot be determined by GC, being most CSPs available for LC also useful for SFC [16, 27]. The advantages of SFC include the use of CO<sub>2</sub> as mobile phase, its sustainability, high throughput, reduced waste generated, and ability to reuse CO<sub>2</sub> [27]. Recently, chiral analysis by SFC with MS detection has been increasing due to the use of the same LC CSPs, the simplicity of the mobile phase, and the facility of the multi-column screening systems that are useful to achieve the suitable chiral selector for development of multi-residue methods [53, 54].

An interesting study on ultrafast chromatographic enantioseparation of pharmaceutically-related drugs and intermediates showed sub-minute resolutions

for most compounds by both SFC and reversed phase LC, highlighting a better overall selectivity and peak shape within a shorter analysis time by SFC than LC [55]. The first study using ultrahigh performance SFC coupled to a QqQ mass analyzer (SFC-MS/MS) for multi-residue enantioselective analysis of chiral pharmaceutically active compounds in environmental samples used two coated modified 2.5  $\mu\text{m}$ -polysaccharide-based CSPs: amylose tris-3,5-dimethylphenylcarbamate and cellulose tris-3-chloro-4-methylphenylcarbamate [56]. A baseline enantioresolution was obtained for 13 compounds in less than 10 min of running time, whereas for the other 2 only a partial enantioresolution was reported. Another multi-residue SFC-MS/MS method for river and wastewater analysis was reported, in this case encompassing 140 chiral and non-chiral contaminants, with 81 compounds being fully quantitative, 14 semi-quantitative, and 45 qualitatively analyzed. A range of available SFC columns and optimized chromatographic conditions were used for the development of new, combined non-chiral and chiral-SFC methods capable of separating a range of different chiral and non-chiral analytes [57]. With limits of detection in a range between dozens  $\text{pg L}^{-1}$  and hundreds of  $\text{ng L}^{-1}$ , the method proved to give exceptional chromatographic enantioresolution for beta-blockers, benzodiazepines, and antidepressants.

SFC-MS was also used to determine five chiral fungicides in wine and vineyard soils, after sample preparation by SPE and pressurized liquid extraction [58]. The method allowed to reach low limits of quantification (between 0.5 and 2.5  $\text{ng mL}^{-1}$ , and from 1.3 to 6.5  $\text{ng g}^{-1}$ , for wine and soil, respectively), and it was found that soil dissipation and transfer from vines to wines of azole fungicides were non-enantioselective, whereas the application of metalaxyl to vines was confirmed to occur as racemate and as (*R*)-metalaxyl. Another example, a green ultraperformance convergence chromatography with tandem MS method using  $\text{CO}_2$  and ethanol as mobile phase was developed to determine benalaxyl enantiomers in tobacco and soil, after sample preparation by QuEChERS technique [59]. This study reinforced the advantages of SFC as a green, precise (relative standard deviations <5.0%), and rapid methodology, with the baseline separation of benalaxyl enantiomers in less than 5 min.

Another emerging field in environmental analysis is metabolomics. Such studies have been growing in the last years as an effective tool to evaluate biological effects after exposure, to assess the variation of targeted metabolites, or to identify untargeted metabolites and metabolic pathways [25]. However, enantioselectivity has been poorly explored. Recently, LC-TOF/MS-based metabolomics was proposed to understand the enantioselective effects of ibuprofen in adult zebrafish. This untargeted metabolomics study allowed to identify potential biomarkers and related pathways, which was followed by an approach involving targeted metabolomics and oxidative stress analysis [25].

The possible enantioselectivity of biodegradation either in the environmental compartments or in biological treatment of WWTPs has boosted the research on the enantioselective evaluation of biodegradation, with EF being the descriptor most frequently used to depict the relative abundance of enantiomers in biodegradation reports [60]. In the last years, there was a growing application of LC-MS/MS for

environmental monitoring of chiral compounds, which has been also applied to follow enantioselective biodegradation [60]. This is a research area that needs more attention due to the implications of transformation products generated either by biological processes (metabolites) or by abiotic processes. Their potential toxicity and bioaccumulation, either enantioselective or not, urges attention for an accurate environmental risk assessment.

Interestingly, the metabolites detected in the environment can be enriched with the enantiomer with the opposite configuration of the parent compound, as reported elsewhere [61]. Biodegradation can also lead to an enrichment of a certain enantiomer in the environment that is the opposite of that excreted by humans or animals, as reported for fluoxetine (among other compounds) in the same study on WWTPs and receiving waters. Here, the study was monitored by QqQ analyzer assisted by enantioseparation through both protein and antibiotic-based CSPs [61]. Moreover, chiral inversion may also result from microbial metabolism [62]. The enantioselective biodegradation or biotransformation patterns depends greatly on the characteristics of the microbial community [42].

Chiral environmental analysis has also an important role in forensic determinations. LC-MS/MS is the most applied analytical approach for enantioselective determination of chiral compounds in environmental matrices, as occurs in the case of environmental monitoring [22]. Enantiomeric profile is a useful tool to differentiate direct disposal from consumption, to get insights about the synthetic pathway, and to distinguish licit and illicit consumption of certain drugs [63]. For example, WBE and enantioselective analysis were used to estimate consumption patterns of illicit drugs in eight European cities, by using LC-MS/MS with QqQ and a Chiralpak<sup>®</sup> CBH [64]. Another report using a LC-MS/MS with a QTrap and a Chirobiotic V2 CSP for enantioseparation was focused on the enantiomeric profile of amphetamine and methamphetamine in wastewater in regional and urban areas of Queensland (Australia) to find evidence on consumption in each area and the potential synthetic pathways [65]. Although LC-MS/MS is the most frequent analytical methodology used for forensic environmental studies, GC-MS is preferred for some classes of compounds (e.g., amphetamine related substances) [30]. Despite the potential of two-dimensional GC (2D-GC) for forensic applications [66] due to its advantages (robustness even for complex matrices, selectivity, sensitivity), forensic environmental studies using this technique are missing in the literature. The same occurs for analytical methodologies such as SFC and CE, which are mostly used for biological samples in forensic studies [22].

### 3 Chiral Analysis in Food Chemistry

#### 3.1 Evaluation of Product Authenticity and Adulteration

Chirality is inherent to natural products and the evaluation of the enantiomeric ratio (e.r.) is of great importance to assess authenticity of essential oils and fruit flavors, to determine the biosynthetic pathway, to establish food origin (natural or synthetic), and to control the storage conditions and the microbial contamination. In fact, chiral analysis has been used for food quality control in the evaluation of genuineness, to determine the geographical origin for finding out the possible fraud or adulteration [67]. Several recent reviews have reunited the representative applications of chiral analysis in food science. GC, LC, and SFC with MS detection using CSPs for enantioseparation are the most applied methodologies [39, 67–69]. In the case of complex matrices, adequate sample preparation process and enantioselective multidimensional methods such as 2D-GC and 2D-LC are of great relevance. The appropriate selection of an achiral column in the first dimension and a chiral column in the second dimension is an important step. The unresolved target compounds are heart-cut from the primary column and delivered to the chiral column for enantioseparation [68, 69].

Chiral GC plays a key role in analysis of flavor and odor, which are imperative indicators in the evaluation of the quality of many food products. Given the volatility of these substances, many GC methods with MS detection have been developed for the analysis of essential oils, milk, herbal infusions, meat, fish, alcoholic beverages (wine and beer), fruit flavoring, vegetables, and food supplements [39]. For instance, the incorporation of a flow-modulated comprehensive 2D-GC provides a possibility to improve the quality of the results due to the enhancement of separation capacity and resolution. As example, the influence of winemaking technology on specific features of botrytized wines was analyzed by 2D-GC [70]. Interestingly, the pleasant honey-like taste and unique fruit flavor of botrytized (noble rot) wines are the result of a specific winemaking technology, which includes overripe grapes infected by *Botrytis cinerea*. A CSP based on 2,3-di-*O*-ethyl-6-*O*-tert-butyl dimethylsilyl- $\beta$ -cyclodextrin was used in the first dimension and a polar INNOWax column was selected for the second dimension to separate analytes according to their polarity. The results showed a significant difference between wine categories. The prevalence of *S*-(-)-ethyl lactate in the botrytized wines was supposed to be a result of malolactic fermentation supported by *Botrytis cinerea* colonization on the grape berry. It was also found that the enantiomeric distribution of the target analytes changes with the time range of fermentation with grape skins. Increased e.r. of (*R*)-ethyl lactate and a reduction in a *cis/trans* ratio of whiskey lactone were observed with the extension of the fermentation period [70]. Whiskey lactone molecule has four stereoisomers, but only *trans*-(3*S*,4*R*)- and *cis*-(3*S*,4*S*)-whiskey lactones are naturally occurring. The *cis*-stereoisomer has a lower odor threshold, whereas both *cis*- and *trans*-whiskey lactones contribute to fresh wood and coconut aroma of

wines. Moreover, *cis/trans* ratio of whiskey lactone has been suggested as a parameter to distinguish wines aged in America from French oak barrels [71].

Volatile organic compounds in healthy and botrytized grape berries, as well as highly botrytized wines (Tokaj selection essences) were analyzed by GC-MS with solid phase microextraction (SPME) as sample preparation. More than 95 compounds were detected in uninfected grape berries, while the number of detected compounds in Hungarian Tokaj selection essence (206) was doubled when compared to Slovak ones. Limonene,  $\alpha$ -terpineol, hotrienol, diethyl malate, 2,3-butanediol, and whiskey lactones were selected for further chiral separation. Enantiomeric separations were performed by heart-cut 2D-GC system, isothermally, with two independent GC ovens using different chiral columns: Chirasil- $\beta$ -Dex at 50°C (limonene) or 75°C ( $\alpha$ -terpineol and hotrienol), and on the MEGA-DEX DMT-Beta at 50°C (2,3-butanediol) or 100°C (diethyl malate or whiskey lactones). Botrytization influenced the e.r. of studied terpenes in grapes of all varieties. The enantiomer distribution of  $\alpha$ -terpineol was influenced by the wine processing technology used [72].

The terpenes and terpenoid compounds, responsible for the typical and unique smell of the cannabis plant from different hemp cultivars (Kompolti, Futura 75, Carmagnola, Felina 32, and Finola) were analyzed by GC-MS and flame ionization detectors. The compounds were obtained by microwave distillation and GC-MS using a cyclodextrin-based capillary chiral column, beta-DEX<sup>TM</sup>, was used for enantioseparation. The e.r. is an analytical parameter commonly utilized for establishing the authenticity of suspicious samples. The enantiomeric distribution of 10 enantiomeric pairs was well-defined, the method confirmed the predominance of (–)-caryophyllene oxide in *Cannabis sativa* plant and low levels of (+)-caryophyllene oxide [73]. Another recent work described terpenoid enantiomers proportion in essential oils of fifteen cultivars of *Mentha arvensis* and *M. x piperita*, collected from an experimental farm, hydrodistilled using a Clevenger-type apparatus to obtain essential oils which were analyzed by different cyclodextrin coated stationary phases. Among the 14 enantiomeric pairs studied, PM- $\beta$ -CD CSP resolved 12 pairs of enantiomers, 10 pairs were enantioseparated in PM- $\gamma$ -CD and TBDA- $\gamma$ -CD, while TBDA- $\beta$ -CD CSP resolved only eight enantiomeric pairs. TBDE- $\beta$ -CD resolved all 14 enantiomeric pairs present in *M. arvensis*, and *M. x piperita* essential oils. Limonene, menthone, menthol, and menthyl acetate were identified as pure (–)-enantiomers whereas isomenthone, neomenthol, pulegone, and piperitone as pure (+)-enantiomers in menthol mint, and peppermint oils, respectively. The results obtained with this analytical approach may facilitate the origin authentication studies of mint oils as menthol biosynthesis revealed that only (–)-menthol is present in the essential oils of *Mentha* species [74].

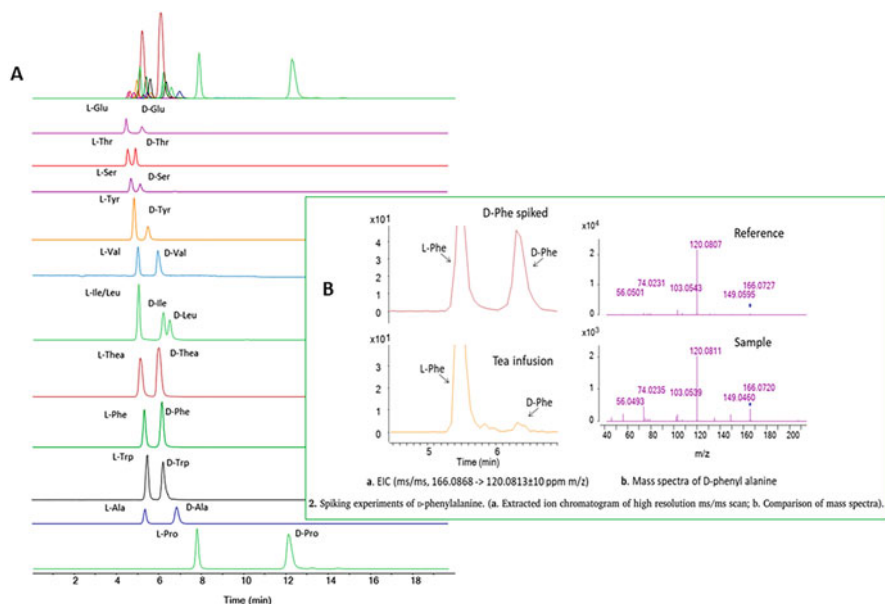
Regarding LC with MS detection, many works report analysis of amino acids. D-amino acids have been identified in various foods, particularly in fermentation systems, such as vinegar, beer, and wine. Food processing under conditions of fermentation, high temperature, strong acid or alkali treatments, or adulteration can also induce the racemization of amino acids [75]. For instance, a sensitive, robust, high-throughput method for D-amino acid profiling was developed with

combination of enantioseparation by chiral columns and TOF/MS. The enantioseparations were performed in CROWNPAK CR-I(+) and CR-I(-) columns, with more than 100 targets, including non-proteinogenic amino acids and amines. The CROWNPAK columns contain a chiral crown ether as a chiral selector, are mainly used to separate chiral amino acids and other small molecules with primary amine groups. The elution order of the enantiomers can be reversed when necessary. Trace of D-amino acids in three kinds of foods (black vinegar, kimchi, and yogurt) were determined [76].

Natural cheese may also contain several D-amino acids originating from microorganisms. A LC-MS/MS high-throughput method combining two types of chiral columns was applied for the analysis of 115 chiral and non-chiral metabolites in cheese in order to know the metabolic profiling of chiral amino acids and related metabolites. CROWNPAK CR-I (+) column was mainly used for the simultaneous separation of targeted metabolites while the identification of D-lysine, L-glutamine, D-threonine, D-allo-threonine and D-homoserine; D-isoleucine, D-allo-isoleucine was possible with CR-I (-). Secondary amines such as D-, and L-proline were analyzed using a CHIRALPAK ZWIX (-). The applicability of the LC-MS/MS method was shown in commercial natural cheeses with different ripening periods (6, 18, and 26 months). The taste and quality of cheese highly depends on metabolites produced during the manufacturing process, and the ripening process especially affects the amount and variety of metabolites. The profiling revealed that the ripening period had a significant impact on the production of each metabolite [77].

Herbal leaves may contain a substantial amount of free amino acids, which contribute to the physiological and pharmacological properties and to the sensory characteristics of many types of herbal infusions. For example, glutamic acid and theanine are the main constituents prompting the umami taste of green tea. The composition of amino acids was used for the differentiation of tea types and the contents of amino acids are indeed positively correlated with the quality and price of certain types of herbal infusions. The herbal infusions manufacturing process and storage period may originate racemization converting their L- to D-forms, this hypothesis was verified with a method developed for analysis of the enantiomers in herbal leaves. The target enantiomers were separated using the CHIROBIOTIC T<sup>®</sup> chiral column. The identification was performed by high-resolution Q-TOF/MS (Fig. 6). The MS and MS/MS acquisitions, quantitative and identification were combined in the same experiment. Only L-forms of amino acids were found in fresh tea leaves. A total of 11 D-amino acids were determined in 19 tea samples, ranging from residual level to 43  $\mu\text{g g}^{-1}$ . Figure 6 shows the results of the spiking experiment, MS and MS/MS scans, and the mass spectra of D-phenylalanine in a tea infusion. The results indicated that the enantioisomerization of amino acids occurred in post-harvest tea leaves affected by process conditions and storage time [78].

Amino acids can also be an important tool to identify the origin of products. Milk contains free amino acids that are crucial for the growth and health of infants. LC coupled to ion-mobility high-resolution MS (IM-HRMS) was applied for the characterization of 18 pairs of amino acids in different types of milk (human, cow, yak,



**Fig. 6** Enantioseparation and Q-TOF/MS identification of amino acids (a); Spiking experiment, MS and MS/MS scans, and the mass spectra of D-phenylalanine in a tea infusion (b) – reprinted from *Food Chemistry* 317 (2020) 126428, Yu Xu, Ziyi Liu, Zhaoye Liu, Zhihui Feng, Liang Zhang, Xiaochun Wan, Xiaogen Yang *Ramage Identification of D-amino acids in tea leaves*, Copyright (2020), with permission from Elsevier

buffalo, goat, and camel). The results demonstrated that the origin of a certain milk can be distinguished based on the D-/L-amino acid ratio-based projection scores. A new approach to investigate amino acids metabolic pathways and to get a comprehensive knowledge on the D/L-amino acid ratio underlying the biological functions of diverse animal milks was provided [79]. Ion mobility spectrometry (IMS) has demonstrated to be a powerful tool for the fast separation of isobaric and isomeric compounds. In drift-tube IMS, ions are separated based on their mobility in an electric field through a neutral gas. As the mobility is a function of the ion collision cross section (CCS), which depends on the size and shape of the ion in the gas phase, compounds may be separated by IMS, even if their  $m/z$  ratio is the same. Although ions of enantiomers have identical charge,  $m/z$ , and CCS, their direct separation by IMS was not feasible and thus, the separation of enantiomers by IMS has been achieved by using gas-phase ion complexation with volatile chiral agents, by forming metal-ion noncovalent complexes with reference compounds or by indirect methods, derivatizing with an enantiomeric pure reagent [80–82].

### 3.2 Evaluation of Chiral Pollutants in Food Web

The evaluation of EF of contaminants such as fungicides, pesticides, and even pharmaceuticals is decisive for studying toxicity and bioaccumulation in food web. LC with MS detection is the most frequently employed technique for the separation of thermolabile and nonvolatile compounds. Methods for food safety, to monitor pesticides and pharmaceuticals enantiomer residues and to assess their risks, are mostly carried out by chiral LC-MS/MS [67], though SFC- MS/MS has been the new trend.

More than 30% of commercial, globally used pesticides, such as pyrethroid insecticides, triazole fungicides and herbicides, are chiral. Due to the challenge and cost of synthesis and chiral purification, most of the pesticides are commercialized as racemates. Nevertheless, the use of enantiopure formulations could reduce pesticide residues released into the environment, by applying the specific enantiomer that is responsible for the efficacy and/or is less toxic for non-target organisms. Triazole pesticides are a class of broad-spectrum fungicides that are useful in preventing several fungal diseases of fruits, vegetables, beans, and food crops. Usually, the inactive enantiomers of this class of fungicides are difficult to be biodegraded by organisms, resulting in the enantiomeric enrichment or accumulation in the ecosystems (water, soil, plants, animals) and even generating chiral carcinogenic intermediates. A LC-MS/MS method for the simultaneous analysis of six chiral triazole (hexaconazole, tebuconazole, triticonazole, flutriafol, diniconazole, paclobutrazol) in six fruits and vegetables was proposed based on a home-made bridged bis( $\beta$ -cyclodextrin)-bonded CSP. Sample preparation was performed by magnetically assisted QuEChERS and 90 samples were analyzed. The tebuconazole EF in strawberry and cucumber was 0.63 and 0.43, respectively, whereas a value of 0.57 was obtained for flutriafol in tomato. This high-throughput method was a suitable strategy for enantiomeric evaluation of chiral pesticides in fruits and vegetables [83].

A novel multi-residue method by LC-MS/MS for the enantioselective analysis of 22 chiral pesticides in cucumber, tomato, cabbage, grape, mulberry, apple, and pear from the local market (Benxi, China) was reported. Pesticides were efficiently extracted by magnetic SPE based on graphene as sorbent. The enantiomeric separation was achieved on a Chiralpak IG<sup>®</sup> column, an immobilized CSP with amylose derivatized with phenyl moieties with chloro- and methyl-groups in meta positions [84].

Five triazole fungicides (penconazole, tebuconazole, triadimefon, myclobutanil, and triadimenol) were analyzed in tobacco by SFC-MS/MS. Optimized separation of the enantiomers was achieved in an ACQUITY UPC2 Trefoil AMY 1–2.5  $\mu$ m amylose *tris*-(3,5-dimethylphenyl)carbamate). The Trefoil columns were designed to use in UPC2 SFC systems. The method was applied to 20 fresh tobacco samples and the results indicated that some samples tested positive for tebuconazole, triadimenol, and penconazole. The EF of tebuconazole, triadimenol, and



penconazole ranged from 0.43 to 0.50, 0.49 to 0.51, and 0.50 to 0.53, respectively [85].

Antibiotics are widely used in veterinary medicine with the purposes of inhibiting the growth of microorganisms, preventing, or treating infections, and promoting growth when used at subtherapeutic doses. The public concern about the presence of antibiotic residues in food-producing animals has increased because they can lead to the growth of antibiotic-resistant bacteria, allergic reactions, or even toxic effects. Flumequine, owing to high antimicrobial activity against a wide range of Gram-negative and Gram-positive pathogens, has become an indispensable part of the treatment of infections in food-producing animals such as cattle, turkey, pig, and poultry. The enantioseparation and determination of flumequine enantiomers in milk, yogurt, chicken, beef, egg, and honey samples by LC-MS/MS on a Chiralpak IC column was reported, providing a reliable method for evaluating the potential risk in animal production [86]. Chiralpak IC is the immobilized *tris*-(3,5-dichlorophenylcarbamate) cellulose column.

Understanding the enantioselective fate of chiral neonicotinoid dinotefuran is of vital importance for accurate dietary exposure assessment and food safety regulation. Dinotefuran enantiomers demonstrated high toxicity to honeybee (*Apis mellifera* Linn.) and (*S*)-dinotefuran was 41.1- to 128.4-fold more toxic than (*R*)-dinotefuran [87]. An analytical method was developed by SFC-MS/MS using an amylose *tris*-(3,5-dimethylphenylcarbamate) chiral column to analyze dinotefuran and main metabolites 1-methyl-3-(tetrahydro-3-furylmethyl) urea (UF) and 1-methyl-3-(tetrahydro-3-furylmethyl)guanidium dihydrogen (DN) in tomato. Estimated half-lives of (*S*)-dinotefuran and (*R*)-dinotefuran were 10.3 and 9.1 days, respectively, by foliage uptake pathway, whereas they were 13.3 and 12.6 days, respectively, by root uptake pathway. Changes in the enantiomeric removal and enantioselectivity orientation of dinotefuran and metabolites were evaluated during home canning of tomato paste, including washing, peeling, homogenization, simmering, and sterilization. Peeling played the key role in reducing (*S*)-dinotefuran by 67.3% and (*R*)-dinotefuran by 69.9% with processing factor of 0.313 and 0.287, respectively. Simmering was the most effective way to remove UF enantiomers by elevated temperature. The study sheds light on the chiral profiles of the fate of dinotefuran from cultivation to processing [28].

## 4 Conclusion and Future Perspectives

Chiral analysis in food and in environmental matrices is clearly interconnected. Regarding the analysis of chiral components of food for authentication and genuineness, multidimensional GC-MS with 2D-GC plays an important role due to the complexity of the sample and the volatility of the substances used as threshold control. LC-MS/MS is the main technique for chiral analysis of contamination in environmental matrices and uptake by food. However, it is noticeable that SFC-MS/MS is gaining a great interest due to the green association with the technique and the

availability of suitable CSPs. 2D-LC and SFC have been showing importance in method development toward metabolic profiling and multi-residue analysis, but there are few reported applications in quality control of foods. In general, innovation in sample preparation protocols with 2D-LC-MS/MS provided a possibility to improve the quality of results, the time of analysis, the size of sample, and the efficiency of the enantioseparation. The applications exemplified that the environmental impact caused by discharged and management of chiral molecules and the relationship between environmental and food analysis. Combined efforts and exchange of experiences are needed to improve the chiral analysis in both fields.

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# Ambient Ionization Techniques in Food and Environmental Analysis



Markus Himmelsbach, Wolfgang Buchberger, and Christian W. Klampfl

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**Abstract** The present chapter gives an overview on the use of ambient ionization techniques in mass spectrometry for analyzing environmental- and food-related samples. Ambient ionization techniques discussed include among others direct analysis in real time, desorption electrospray ionization, and rapid evaporative ionization. Developments within the field displayed in the literature published since 2015 are discussed. Strategies for improving sensitivity and selectivity by implementing sample pre-treatment devices into the ambient ionization source are presented, as are approaches to improve the ability for reliable quantitative analysis and further advancement in instrument design. Relevant applications published over the last 5 years are summarized and discussed.

**Keywords** Desorption electrospray ionization, Direct analysis in real time, Environmental analysis, Food analysis

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## Abbreviations

AIMS	Ambient ionization mass spectrometry
ASAP	Atmospheric solid analysis probe
CBS	Coated blade spray
DAPCI	Desorption atmospheric pressure chemical ionization
DART	Direct analysis in real time
DBDI	Dielectric barrier discharge ionization
DCBI	Desorption corona beam ionization
DESI	Desorption electrospray ionization
EASI	Easy ambient sonic-spray ionization
EESI	Extractive electrospray ionization
FT-ICR	Fourier transform ion-cyclotron resonance
GC	Gas chromatography
HPLC	High performance liquid chromatography
HR-MS	High-resolution mass spectrometry
HRPS	High-resolution product scan
iEESI	Internal extractive electrospray ionization
ISTD	Internal standard
IT-MS	Ion trap mass spectrometer
LDTD	Laser diode thermal desorption
LIT-MS	Linear ion trap mass spectrometer
LTP	Low-temperature plasma ionization
MAII	Matrix-assisted inlet ionization
MIP	Molecularly imprinted polymer
MPT	Microwave plasma torch
MRM	Multiple reaction monitoring
MS	Mass spectrometry
MSI	Mass spectrometry imaging
PC	Principal component
PCA	Principal component analysis
PCR	Polymerase chain reaction
PLS-DA	Partial least squares discriminant analysis
PS	Paper spray
Q-MS	Single quadrupole mass spectrometer
QQQ-MS	Triple quadrupole mass spectrometer
QTOF-MS	Quadrupole time-of-flight mass spectrometer
QuEChERS	Quick easy cheap efficient rugged safe
REIMS	Rapid evaporative ionization mass spectrometry
SBSE	Stir bar sorptive extraction
SPME	Solid-phase micro-extraction
TD-ESI	Thermal desorption electrospray ionization
TLC-MS	Thin-layer chromatography-spray mass spectrometry
TM-DART	Transmission mode direct analysis in real time

TOF-MS	Time-of-flight mass spectrometer
WWTP	Wastewater treatment plant;

## 1 Introduction

The development of desorption electrospray ionization (DESI) and direct analysis in real time (DART) in 2004 and 2005, respectively, substantially boosted the interest in ambient ionization techniques. This is reflected by the impressive number of research as well as review papers published on this topic in the following years (for exemplary review papers, see [1–4]). Additionally to these two techniques, which are also available in commercial instrumentation, paper or tissue spray [5] can be seen as the third important representative in this field, when judging the number of publications. Furthermore, there is a plethora of other ambient ionization techniques such as easy ambient sonic-spray ionization (EASI), dielectric barrier discharge ionization (DBDI), matrix-assisted inlet ionization (MAII), coated blade spray (CBS), or desorption atmospheric pressure chemical ionization (DAPCI) (just to name a few) playing a somewhat less dominant role in environmental as well as food analysis. Focusing on a possible categorization of the different techniques summarized under the term “ambient ionization” one can follow the excellent tutorial presented by group of Shiea in 2011 [6]. Thereby the three distinct groups are defined: (1) direct ionization methods, (2) direct ionization/desorption methods, and (3) methods based on a two-step ionization process. Based on this scheme, paper spray (PS) is part of the first group whereas techniques like DESI with a combined desorption and ionization mechanism or DBDI may be assigned to the second group. DART based on either the direct or the indirect (via ionization of solvent molecules with subsequent proton transfer) interaction of excited gas molecules with the analyte belongs to the third group of this scheme.

When browsing the relevant literature, it can be seen that food is one of the major fields of application of ambient ionization mass spectrometry (AIMS) techniques. This is reflected in the substantial number of review papers and book chapters published within the last decade focusing on this topic (for an exemplary selection, see [7–10]). On the other hand, the number of papers dealing with the employment of AIMS in environmental analysis is much lower (for a recent review, see [11]). An explanation for this asymmetry in the published literature may be found in the typical nature of samples best suited for analysis by ambient ionization techniques. In food analysis a straightforward examination of a sample “as it is” is in many cases sufficient to provide the information desired. Additionally, there is substantial interest in food analysis directly at the site of production – so AIMS seems more appropriate for examining this type of samples. A typical example for such a problem might be the question whether a certain pesticide can be detected on the surface of fruits or vegetables. Typical problems in environmental analysis require analysis at the trace or even ultra-trace level. Therefore, to comply with the

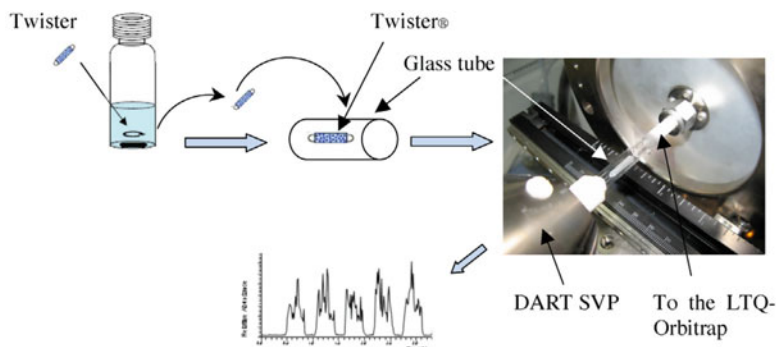
requirements regarding ultra-low limits of detection paired with the necessity to provide reliable and accurate quantitative data, sample pre-treatment procedures cannot be avoided. So, pre-concentration and/or clean-up steps are mandatory. Next to off-line pre-treatment, this challenge in many cases requires the use of separation techniques before MS analysis. This might be a possible explanation, why AIMS techniques are less frequently used in environmental analysis.

When comparing different AIMS techniques with respect to their frequency of use in both environmental and food analysis a clear trend can be detected. Searching the Web of Science for one particular technique, for example, in combination with food analysis, a preponderance of the most successfully commercialized AIMS technique, namely DART can be detected. Focusing on the time span since 2015, twice as many publications report the use of DART than that of DESI. Less prevalent commercially available solutions such as atmospheric solid analysis probe (ASAP) show frequencies comparable with those of techniques based on lab-made instrumentation such as DBDI or easy ambient sonic-spray ionization (EASI) just to name a few.

## **2 Applications of Ambient Ionization Techniques in Environmental Analysis**

### ***2.1 Direct Analysis in Real Time (DART) and Related Techniques***

A major driving force for employing AIMS techniques is the desire to achieve analytical results faster and with less effort than encountered when using conventional approaches, often involving separation with chromatographic techniques, followed by MS detection. Particularly when large numbers of samples need to be categorized (e.g., as contaminated and non-contaminated) this can pose a substantial obstacle. Therefore, AIMS is perfectly suited for screening large sample sets particularly when there is no need for providing highly accurate quantitative analysis. The approach chosen in these cases is to avoid sample pre-treatment and pre-concentration almost completely and analyze samples as they are, resulting in extremely short analysis times. Unfortunately, in the field of environmental analysis, often just the advantage of fast measurement can be exploited as sensitivity issues do not allow to completely sidestep sample clean-up and pre-concentration. An early example for combining conventional sample pre-treatment with fast measurement using AIMS techniques was the analysis of organic UV filters from sunscreens in recreational waters published in 2010 [12]. Here DART-MS was employed in combination with pre-concentration using stir bar sorptive extraction (SBSE). Thereby limits of detection lower than  $40 \text{ ngL}^{-1}$  could be achieved for all eight analytes, making the developed method well suited for the screening of real water samples. On the other hand, when quantitative data were required, the method



**Fig. 1** Coupling of stir bar sorptive extraction (SBSE) with the DART Orbitrap mass analyzer. From [13] with permission

showed its limitations and could only provide semi-quantitative results, although an internal standard (ISTD) (benzyl cinnamate) was used. This was mainly due to fluctuations in signal intensity, something that could in part be attributed to the very basic design of the sample holder used in that work (although the original sample holder had already been replaced by an enhanced custom-made design). On the example of analyzing phosphoric acid esters in water samples Bridoux et al. demonstrated that these issues can be overcome. First, they employed more sophisticated deuterated ISTDs allowing an even better correction of fluctuations in peak intensities during measurement – a promising approach that of course might be hampered by the limited availability of suitable isotopically labeled substances. Second, an improved more reproducible positioning of the stir bars within the DART source (for details on the workflow, see Fig. 1) was achieved by improved instrumental design [13].

The approach just described, namely combining a sample pre-treatment procedure for analyte enrichment, and (to some extent also) clean-up with DART-MS was also followed in a series of more recent papers [14–16]. High-throughput analysis is one of those areas where DART can excel over other techniques – a fact making it well suited for large sample sets. This is also reflected in these papers focusing on the fast analysis of various trace pollutants in water after pre-concentration using sorbent-coated stainless steel bars [14], dispersive magnetic solid-phase extraction [15], and a solid-phase microextraction (SPME) fiber coated with a metal-organic framework [16]. In all these cases, correct positioning of the sample probes in the DART source was accomplished via an automatic sampler ensuring more reproducible results and allowing the consecutive analysis of large sample sets within a short time. Summing up this part it can be stated that a direct combination of DART with devices allowing a sample clean-up and a pre-concentration of the analytes is a promising approach for qualitative and probably also semi-quantitative analysis even at the ultra-trace level. Nevertheless, when accurate quantitative data are required an increased effort must be accepted as demonstrated by several works. This includes highly reproducible (mostly automatic) positioning of the sampling

device in front of the MS orifice and the use of appropriate ISTDs. The latter implies in most cases the need to employ isotopically labeled species of the actual analytes, a fact that might be seen as a substantial obstacle in terms of availability as well as costs.

According to its two-step ionization process, laser diode thermal desorption (LDTD) belongs to the same group of ambient ionization techniques as DART [6]. LDTD was employed by the lab of Roman Grabic for the analysis of pharmaceuticals in soil leachates [17] and in fish brain tissue [18]. Particularly the latter paper is of interest as the analysis of fish brain samples from animals subjected in vivo to environmentally relevant doses of citalopram, a drug commonly found in waste water treatment plant (WWTP) effluents, was reported. Thereby, the biological samples were directly introduced into the AIMS device, a procedure beneficial when only small amounts of sample (as was the cases in this study) are available. The described method allowed a quantitative estimation of the accumulation of this serotonin uptake inhibitor in brain tissue down to the low  $\text{ng g}^{-1}$  level [18]. For an overview of applications of AIMS in environmental analysis published since 2015, please see Table 1.

## 2.2 Other Ambient Ionization Techniques

After DART, DESI is the second most widely employed ambient ionization technique. One “unique selling point” of DESI is its ability to be used in an imaging mode allowing to spatially resolve the distribution of the selected analytes on surfaces. This has been demonstrated by the group of Spengler, who screened plant materials (plant stem, leaves) for the presence of systemic and contact pesticides [19]. Other direct desorption ionization techniques (according to the categorization by Shiea) with applications reported from the field of environmental analysis are DBDI [20, 21] and low-temperature plasma [LTP] ionization [22].

PS or tissue spray ionization is a direct ionization technique characterized by its simplicity and ease of operation. Application of PS for soil analysis was reported in several papers. In the first, Liu et al. described the analysis of tetrabromobisphenol in soil, whereby a strong focus was set on achieving accurate quantitative results [23]. The latter was accomplished by implementation of isotopically labeled standards that were added to the soil extracts allowing to correct for signal fluctuations (a strategy already discussed in the context of DART-MS). This resulted in a low relative standard deviation of 5.3% for five replicates at the  $\mu\text{g L}^{-1}$  level, and excellent spiking recoveries ranging from 90% to 101%, for analyzing ten soil and sediment samples [23].

Dowling et al. described an approach for the straightforward analysis of soil samples, leaving out any sample treatment steps. Target analytes in this study were hydrolysis products from chemical warfare agents [24]. Thereby, as can be seen from Fig. 2, soil was directly placed into a cartridge holding the paper for spraying followed by adding a combined extraction/spraying solvent.

**Table 1** Applications of ambient ionization techniques in environmental analysis

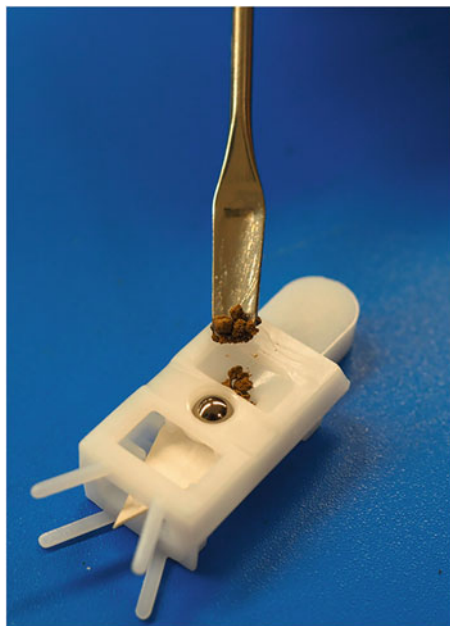
Ambient ionization technique	Mass analyzer	Sample	Analyte(s)/application	Comment	Reference
DART	TOF	Environmental waters	Organic UV filters	Analytes were enriched on the surface of the polydimethylsiloxane-coated stir bars	[12]
DART	Orbitrap	Aqueous samples	Phosphoric acid alkyl esters and mono-butyl phosphate	Stir bar sorptive extraction and quantification using deuterated internal standards	[13]
DART	Orbitrap	Aqueous samples	Phthalic acid esters	Use of sorbent for pre-concentration and solvents for assisting desorption and ionization	[14]
DART	Orbitrap	Environmental waters	Triazine herbicides	Dispersive magnetic solid-phase extraction using magnetic graphene oxide	[15]
DART	Orbitrap	Environmental waters	Triazine herbicides	Use of a laboratory-made solid-phase microextraction device coated with a metal-organic framework fabricated for the on-site enrichment	[16]
LDTD-APCI	QQQ	Soil leachate	Carbamazepine, sulfamethoxazole, irbesartan, fexofenadine	LC-MS/MS and LDTD-APCI-MS/MS method comparison	[17]
LDTD-APCI	QQQ/Orbitrap	Fish brain tissue	Citalopram	LC-HRPS method in comparison with LDTD-APCI-QQ-MS/MS and LDTD-APCI-HRPS	[18]
DESI	Orbitrap	Plant stem and leaves	Pyrethrins, rapeseed oil, imidacloprid, methiocarb, dimethoate	Spatially resolved investigation of pesticides both on the surface of leaves and in cross sections of plant stem and leaves using DESI-MSI	[19]
DBDI	Orbitrap	Aqueous samples	Polycyclic aromatic hydrocarbons and polar trace contaminants	Coupling of an active capillary plasma ionization source based on a dielectric barrier discharge directly to solid-phase microextraction	[20]
DBDI	Orbitrap	Aqueous samples	Acenaphthene	The sample solution was nebulized followed by heating and converting into gas-phase analyte molecules prior to DBDI	[21]
MPT and LTP					[22]

(continued)

Table 1 (continued)

Ambient ionization technique	Mass analyzer	Sample	Analyte(s)/application	Comment	Reference
	LIT-TOF	Apple and grape peel extracts	Atrazine, chlorpyrifos, dimethoate, diphenylamine, imazali, imidacloprid, isoprocarb, and triazophos	Instead of desorption sampling spray-inlet sampling was used	
PSI	LIT	Soils and sediments	Tetrabromobisphenol A	Quantification with a stable isotope internal standard	[23]
PSI	Orbitrap	Soil	Drugs and chemical warfare agent hydrolysis products	To improve quantitative performance an off-line salting out liquid-liquid extraction with internal standardization was also performed	[24]
PSI	LIT	Fish tank water	Non-conjugated steroids	Steroids released in fish water samples were confirmed via tandem mass spectrometry using collision-induced dissociation	[25]
SCWT-ESI	QQQ	Tap and river water	Fluoroquinolone and macrolide antibiotics	The SCWT solid-phase microextraction probe was prepared via silanization and sulfonation	[26]

**Fig. 2** Prosolia paper spray cartridge; solvent well filled with soil. From [24] with permission



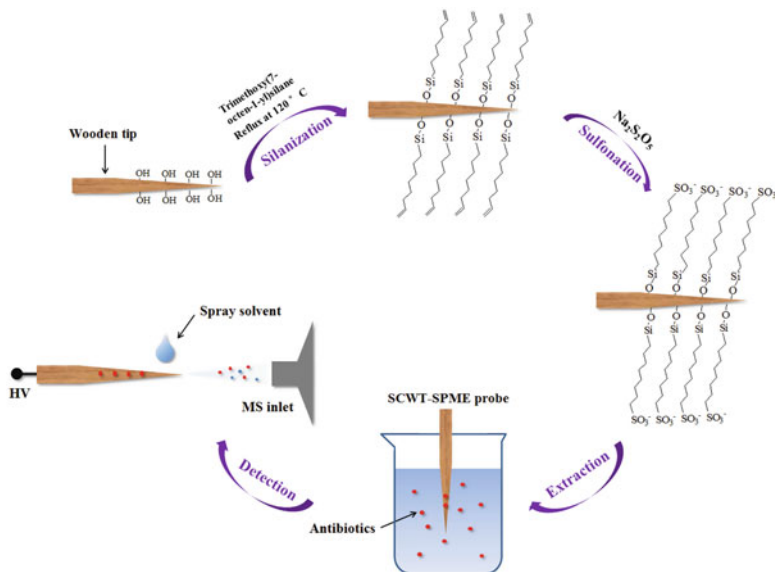
Two further studies presented the use of PS [25], and the quite similar wooden-tip ESI for water analysis [26]. In the latter paper, the strategy of a multi-purpose use of the spraying-tip has been pursued. As can be seen from Fig. 3, wooden sticks were first surface - modified using a multi-step process to allow a selective enrichment of the chosen analyte classes. Subsequently, fluoroquinolones and macrolide antibiotics were extracted from a water sample using this device. Then, by adding a spray solvent and applying high-voltage, an electrospray was induced from the stick and MS analysis was performed. It should be mentioned that such an approach is not limited to wooden sticks but can also be achieved with any other suitable material.

### **3 Applications of Ambient Ionization Techniques in Food Analysis**

#### **3.1 *Direct Analysis in Real Time (DART) and Related Techniques***

Food analysis is one of the primary fields in analytical chemistry, consequently it also plays an important role when looking for applications of AIMS. The development of DART-MS, together with DESI, may be seen as a starting point for the prosperous development of AIMS techniques in a variety of fields. When searching for “desorption electrospray ionization” and “direct analysis in real time” in





**Fig. 3** Schematic diagrams for preparation of a surface-coated wooden-tip electro spray ionization probe and its application to chemical analysis. From [26] with permission

SciFinder, the first shows a distinctly higher number of entries (637 versus 279 – when considering the time span from 2015 till now). The situation is inverted when adding the term “food” to the search algorithm. Here DART-MS definitely plays a more important role (with 177 to 77 entries), revealing the extraordinary success of DART-MS in this field. Food samples have been analyzed with respect to their quality, safety, origin, and their authenticity (just to list some of the most prominent questions asked in the analysis of foods and related products) [27, 28]. For an overview on the use of DART-MS and related techniques in food analysis, see Table 2.

Starting with investigations on the instrumental setup, the importance of reproducibly positioning a sample holder in the DART source in front of the MS orifice was already addressed in the previous section. Now, when coming to food-related applications the question arises how to introduce the sample itself into the DART source – a parameter of utmost relevance when reproducibility of results is an issue. So, throughout the use of DART for a variety of different analytical problems substantial effort towards the development and optimization of sample introduction approaches has been taken. Two papers compared two of the most widely employed modes of sample introduction in DART-MS. This is the classical dip-it glass rod, on the one hand, a device available right from the time when the DART source was launched, and the more recently introduced transmission mode DART, where the sample is applied to a steel mesh [29, 30]. In the first paper, polyphenol profiling for categorization of extra virgin olive oils was done using DART-MS as well as a more conventional HPLC-MS approach. Thereby DART-MS provided comparable results

**Table 2** Applications of ambient ionization techniques in food analysis – DART and related techniques

Ambient ionization technique	Mass analyzer	Sample	Analyte(s)/application	Comment	Reference
DART	Orbitrap	Olive oil	Polyphenols	Olive oil was dissolved in hexane and extracted using methanol–water	[29]
DART	QTRAP	Alcoholic drinks	Artificial sweeteners	Samples were only acidified, diluted, and filtered	[30]
DART	LIT	Ginkgo biloba kernels and leaves	Ginkgolic acids	Methanol extracts of Ginkgo biloba kernels or leaves were analyzed	[31]
DART	QQQ	Grape macerates	Linalool, 3-isobutyl-2-methoxy-pyrazine, 1-hexanol, and $\beta$ -damascenone	Solid-phase mesh-enhanced sorption from headspace is used	[32]
DART	TOF	Powdered milk	Melamine	Using DART operated with argon gas in combination with acetylacetone and pyridine reagent gases allows to selectively ionize melamine and eliminate the interference from 5-Hydroxymethylfurfural	[33]
LDTD	QQQ	Beverages and liquid dietary supplement products	Caffeine	Samples were diluted with a mixture of methanol, water, and deuterated caffeine internal standard	[34]
DART	Orbitrap	Honey	Geographical origin discrimination	DART-MS combined with multivariate statistical analysis was used for the discrimination of two monofloral (chestnut and acacia) honeys for their geographical origins	[35]
DART	Q	Honey	5-Hydroxymethylfurfural	The influence of instrumental parameters on the composition of the DART mass spectra for 5-Hydroxymethylfurfural and partially degraded carbohydrates was investigated	[36]
DART	QTRAP	Honey	Chloramphenicol		[37]

(continued)

Table 2 (continued)

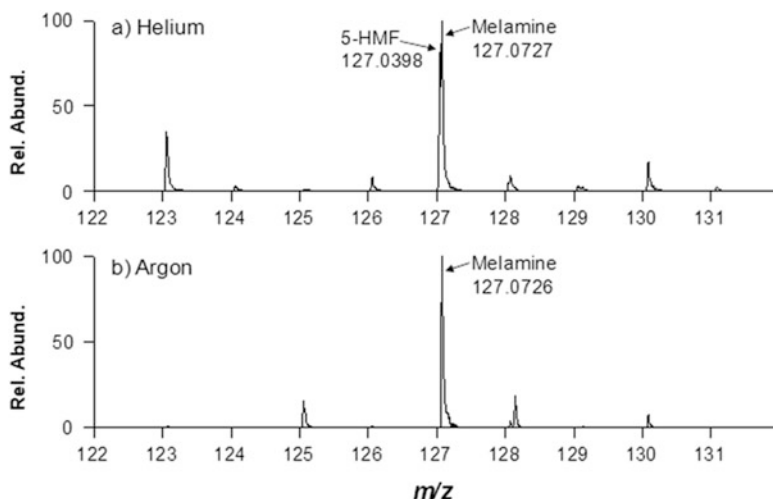
Ambient ionization technique	Mass analyzer	Sample	Analyte(s)/application	Comment	Reference
DART	Orbitrap	Salmon	Assessing fish authenticity	The accuracy was evaluated by comparison with data generated by a validated HPLC-MS/MS method Discrimination between wildtype and farmed salmon using DART and multivariate analysis	[38]
DART	Orbitrap	Wheat varieties	Triticum species authentication	Chemometric evaluation revealed sample clustering according to the wheat species and the presence of 18 significant markers	[39]
DART	Orbitrap	Milk	Authentication of forage-based milk	25 informative biomarkers were identified to authenticate milk samples derived from maize silage, crop silage/hay, and grassland hay based diets	[40]
REIMS/DART	TOF/Orbitrap	Chicken breast meat	Determining poultry production systems	REIMS was compared with DART to assess whether either system was capable of correctly identifying poultry raised in one of six distinct production systems	[41]
DART	TOF	11 herbal teas	Evaluation of the quality of herbal teas	Mass spectra were evaluated by multidimensional chemometric methods, such as cluster analysis, linear discriminate analysis, and principal component analysis	[42]
DART	TOF	Cocoa powders and instant cocoa beverages	Evaluation of cocoa products quality and authenticity	Quantitative analysis of theobromine and caffeine was compared with those obtained from HPLC measurements	[43]
DART/ASAP	Q	Oregano	Authenticity screening of oregano	A set of authentic oregano samples and a wide range of adulterant species were investigated	[44]

DART	Orbitrap	Salmon lipid species	Assessing salmon freshness	The evolution of selected fatty acids and triacylglycerols during the refrigeration of salmon meat was monitored	[45]
DART	Orbitrap	Frying oil	Oxidized triacylglycerols	Various analytical techniques were applied to determine the impact of vacuum frying on quality of potato crisps and frying oil	[46]
MAII/ DAPCI/TM- DART/CBS	TOF/ Orbitrap/ Q/QQQ	Various food samples	Evaluation of ambient ionization techniques for portable mass spectrometry	Four different ambient ionization techniques were investigated for their potential in simplified testing of selected food contaminants such as pesticides, veterinary drugs, and natural toxins	[47]
DART	QQQ	Beef tissue	98 veterinary drugs	Sampling was done with solid-phase microextraction	[48]
DART	TOF	Baby formula	Five drugs of abuse	Comparison of a solid-phase microextraction DART-MS method to a traditional DART-MS method	[49]
DART	QQQ/ Orbitrap	Surface water, milk, grape juice, orange juice	19 pesticides	Sampling was done with the solid-phase micro-extraction – transmission mode technology	[50]
DART	QQQ/ TOF	Wine	31 pesticides	Comparison of different DART methods and different mass analyzers	[51]
DART	Orbitrap/ TOF	Lettuce and celery	Amitrole, cyromazine, propamocarb, melamine, diethanolamine, triethanolamine, and 1,2,4-triazole	A quick polar pesticides extraction (QuPPE) method was used and two accessories for sample introduction were evaluated	[52]
DART	LIT	Herbal and food samples	Six pyrolyzidine alkaloids	Comparison of DART-MS and HPLC-MS methods	[53]
DART	Orbitrap	Foods and surface waters	Cationic dyes	Two temperature steps were employed and results were compared to a HPLC-MS/MS method	[54]

but excelled (as might be expected) through much shorter analysis times [29]. The second study employed DART-MS for the screening of alcoholic drinks with respect to the presence of artificial sweeteners [30]. Coming to the outcome from comparing the two sample introduction modes, the two papers come to quite different results. This finding might be explainable by the quite diverse nature of the samples investigated. In the work on olive oil analysis (whereby oils represent analytes with high viscosities and high boiling points), transmission mode DART provided approximately two times higher signal intensities than the dip-it sampler [29]. In contrast, when analyzing sweeteners in beverages – samples primarily consisting of water – signals obtained with the steel mesh were small and affected by noise. So, in this case the dip-it approach was considered more suitable [30]. These controversial findings might be understood in a way, that (as for many real-world problems) there is no universally valid suggestion for the use of one or the other sample introduction mode. In fact, for finding the best possible solution both the nature of the sample matrix and that of the analytes therein need to be considered.

Focusing on the quantitation of plant ingredients by DART-MS, two recent studies should be highlighted. Food-related plants investigated were *Ginkgo biloba* [31] and grapes [32]. In both cases, quantitative results obtained by DART-MS were compared to data from GC measurements, whereby no perfect agreement between the two methods could be observed. In the first study, the problem might be more associated with the GC measurement as this was burdened by the use of a simple flame-ionization detector combined with a sub-optimal separation of the analytes. So, overlapping peaks for some of the ginkgolic acids (e.g., 17:1 and 17:2) might be responsible for a bias in quantitative GC measurements. In the second study, GC was used in combination with a highly selective MS/MS detector, avoiding problems encountered in the work discussed above. But, plotting the results from the quantitative determination of linalool in grapes obtained by DART-MS versus those by GC-MS/MS led to a slope of 0.16 – a result substantially deviating from the theoretical value of one. Here the most probable explanation for this divergence might be the presence of isobaric interferences, not resolved by the single-stage MS used in combination with DART. This can be regarded as a principal drawback of any direct MS method – as in such cases separation of interfering substances (e.g., by GC or HPLC) before detection is mandatory, particularly when either multi-stage MS is not available or not capable of resolving these issues. Nevertheless, the unique ionization mechanism in DART can also be turned into a powerful tool for distinguishing isobaric (or almost isobaric) compounds, when these substances show a sufficiently different behavior in the DART source. A notable example for such a strategy is the analysis of melamine in the presence of 5-hydroxymethylfurfural (a problem related to the analysis of contaminated dairy products) by DART-MS. Here, as can be seen from Fig. 4, the required selectivity could be achieved by choosing an alternative gas for ionization (argon instead of the more common helium) [33].

Food ingredients were also analyzed by other AIMS techniques with an ionization mechanism belonging to the same category as DART, such as laser diode thermal desorption (LDTD) MS. Andersen et al. reported the determination of

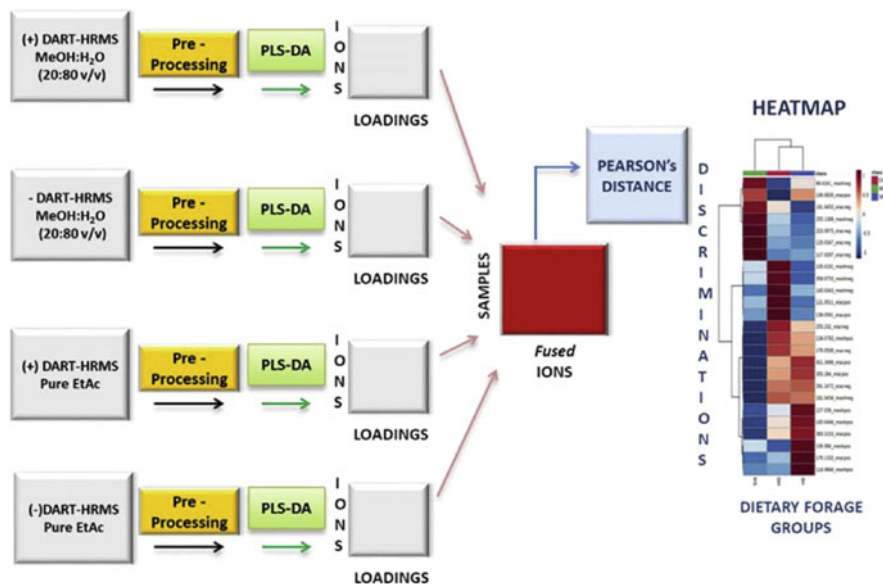


**Fig. 4** The helium DART mass spectrum for 10 ppm of melamine in powdered milk (a) and the argon DART mass spectrum for the same sample (b) over the same sampling time interval. From [33] with permission

caffeine in beverages, whereby the developed procedure allowed the quantitation of caffeine in a range of different samples [34]. A comparison of the results with the official method from AOAC (using HPLC) showed differences of <4%. A potential reason for this good agreement might be the use of caffeine- $d_3$  as internal standard (ISTD) in LDTD MS.

Switching to investigations on food authenticity and origin, DART-MS again plays an important role [27]. A typical example for such an application is the discrimination of monofloral honeys with respect to geographical origin by DART coupled to high-resolution (HR) MS [35]. As distinction between samples in food authenticity and origin studies is rarely possible by evaluating a handful of specific markers, also in this case the obtained MS spectra had to be evaluated using chemometric tools. Thereby testing three supervised statistical approaches (linear discriminant analysis, partial least squares discriminant analysis (PLS-DA), and k-nearest neighbors) combined with the high spectral resolution of HR-MS allowed to determine the origin of chestnut and acacia honey with sufficient certainty. The use of DART for honey analysis was also reported by other groups [36]. One possible reason might be the high sugar content of this natural product, requiring extensive sample preparation when using chromatographic techniques. On the other hand, when using DART for honey analysis, reaction products from sugars formed at the elevated temperature present in the ion source need to be considered as they may substantially complicate the evaluation of the spectra obtained. Furthermore, DART-MS was employed for the analysis of chloramphenicol in honey [37].

Particularly when using direct MS methods in combination with HR-MS instruments, data processing is an important issue due to the complex spectra and the large



**Fig. 5** Flowchart of the detection of the most informative ions from a sample. After pre-processing and Pareto scaling normalization (yellow boxes), preliminary PLS-DA was performed (green boxes); the ions having a loading score  $> 0.3$  (absolute value) were selected (gray boxes) and submitted to mid-level data fusion (red box). Hierarchical cluster analysis (HCA) was performed on the fused ions by the Pearson correlation distance ( $1 - \text{absolute Pearson correlation coefficient}$ ) (blue box). A correlation matrix was generated between the 25 selected ions and the dietary roughage group (heatmap) providing also a clusterization of forage types (among columns) and selected ions (among rows). Red (positive) and blue (negative) color intensities indicate the degree of correlation. From [40] with permission

amount of signals therein. So, also in the case of DART-HR-MS the use of chemometric tools for evaluating the spectra obtained may be beneficial. This combination was employed, for example, for assessing fish authenticity [38], authentication of *Triticum* (a traditional type of wheat) species [39], forage-based milk authentication [40], and for investigating poultry production system history [41]. Fatty acid profiles can be helpful for distinguishing different types of meat. Wild type salmon and animals raised in fish farms were investigated by Fiorino et al. whereby they selected an analyte set comprising of 30 fatty acids (from  $C_8$  to  $C_{22}$ ) [38]. Signals obtained for these compounds in DART-HR-MS spectra were further evaluated and subjected to principal component analysis (PCA). This procedure allowed the unambiguous differentiation between wild type fish from Canada and farmed salmon from Canada, Norway, and Chile.

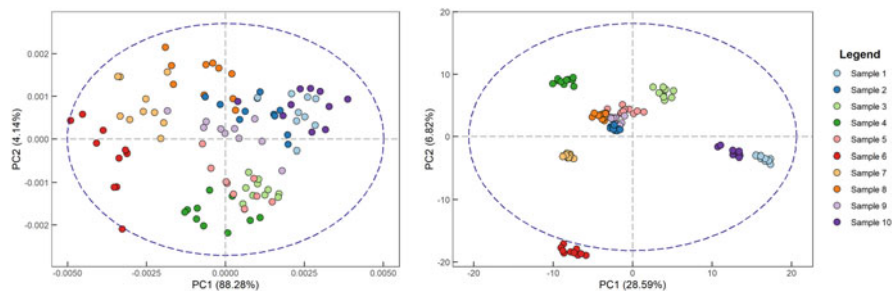
Focusing on the study devoted to forage-based milk authentication, DART-HR-MS spectra were recorded from milk samples [40]. The workflow conducted within this study can be seen from Fig. 5. First, two different solvents, a polar and a non-polar one were employed for extraction of the milk. Subsequently, spectra were recorded from the extracts in the positive and in the negative mode to get a

comprehensive overview of components present in these samples. After pre-processing (see yellow boxes) the preliminary PLS data analysis was performed, and a set of fused ions was obtained. Finally, hierarchical cluster analysis (see heatmap on the right) correlating dietary forage groups with the 25 selected ions (from runs with positive and negative ionization) was performed. This strategy allowed differentiating between cow-milks based on the type of feed provided for the animals.

The groups of Hajslova and Elliott [41] compared DART and rapid evaporative ionization MS (REIMS) (applications of REIMS are discussed in more detail in the next section) for differentiating poultry meat. Chicken were sampled from six distinctly different production systems. The ability of REIMS to categorize birds according to the type of feed employed was between 60% and 100%, which was somewhat less than DART, where 100% of the birds were assigned correctly. But, whereas REIMS allowed a direct analysis of the meat samples, some sample preparation was needed for DART-MS measurements – a fact that substantially influenced sample throughput. Steps in DART-MS analysis included meat homogenization and extraction with polar and non-polar solvents; subsequently the two extracts were investigated separately.

Whereas the studies described above involved the use of DART in combination with state-of-the-art HR-MS (Orbitrap) instruments, authenticity studies can also be performed employing MS instruments with somewhat lower spectral resolution, such as common time-of-flight (TOF) or quadrupole time-of-flight (QTOF) spectrometers [42, 43] or even simple quadrupole MS instruments [44]. Particularly the latter study might trigger curiosity as the combination of ambient ionization with low resolution MS is rarely employed. This is understandable as in direct MS all ionizable substances (all analytes and matrix components) will enter the MS at the same time. So, substance discrimination and subsequently (tentative) identification totally relies on the capability of the MS instrument to resolve a maximum of  $m/z$  ratios, a qualification commonly strongly associated with mass resolution. This is of particular interest, as that study used a non-targeted approach for detecting fraudulent blending of oregano with olive leaves [44]. In more detail, that study presented the critical evaluation of two AIMS techniques, DART-MS and ASAP-MS revealing distinct advantages for the latter one. One main reason was the fact that for the chosen analytical problem ASAP-MS provided reproducible and diagnostic feature-rich spectra, a fact of distinct importance when employing a non-targeted approach for analysis. One main deficit of the DART source, observed in that study, was its basic layout as a source that is open to the environment. Thereby, a substantial influence of the surrounding atmosphere on the DART-MS spectra was detected. In some cases these signals even interfered with analyte-specific features [44]. This is clearly reflected in Fig. 6, depicting scatter plots for PC1 versus PC2 scores for ten technical replicates of ten oregano samples. Without background subtraction, no acceptable clustering of the sample spectra could be achieved (see left plot). Although the situation could be greatly enhanced by measuring background spectra that were subsequently subtracted from each sample spectrum (result, see right plot), ASAP-MS spectra still excelled over DART when considering the differentiation





**Fig. 6** Scatter plot of PC1 vs PC2 scores obtained from ten technical replicates of ten analyzed oregano samples by applying mathematical pre-processing only (left) or mathematical pre-processing and background subtraction (right). Each sample is marked with a different color. Confidence ellipses (95%) are reported as dashed blue lines. From [44] with permission

between pure and fraudulently blended origanum samples. A further reason for the better performance of the ASAP source might be that ASAP, as an ion source with a closed compartment, resulted in a higher percentage of ions entering the MS instrument, leading to spectra with higher signal intensities – a fact further facilitating the categorization of the investigated samples.

Despite not dealing with authenticity in the strict sense, two more papers might also fit to this group. Fiorino et al. employed DART-MS as a rapid tool for assessing fish freshness [45]. Aim of this study was the influence of storage conditions and period on the concentrations of a number of target analytes. The authors revealed that after 4 days of refrigeration the amount of fatty acids had increased while a reduction of triacylglycerols was observed. The second study investigated potato crisps and the respective frying oils, whereby the main focus was set on a comparison of production processes employed [46]. Thereby DART-MS was used for analyzing the processed potatoes with respect to the formation of acrylamide and the corresponding frying oil for oxidation products after a conventional frying process (165° for 105–135 s) and a low-temperature vacuum frying process (125° for 180–360 s). As might be expected, chemical analysis revealed advantages for the vacuum frying technology, but organoleptic analysis by a panel resulted in advantages for the conventionally produced crisps.

On-site testing is a dominant trend in chemical analysis, a fact that is also reflected when it comes to investigating food and feed. Up to now mainly spectroscopic techniques, assays, or biosensors were used for this purpose (for exemplary review papers, see [55, 56]). Ambient ionization, particularly when combined with a portable MS instrument, may pave the way for MS techniques into this field. Blokland et al. investigated a range of AIMS techniques such as transmission DART, matrix-assisted inlet ionization (MAII), coated blade spray (CBS), and desorption atmospheric pressure chemical ionization (DAPCI) combined with a portable MS system for their suitability in food analysis [47]. Thereby DART and CBS excelled over DAPCI and MAII with respect to sensitivity and subsequently robustness at the concentration level of interest for on-site measurements

[47]. Nevertheless, the study also revealed a more fundamental problem (already discussed above) of the combination tested. Up to date, portable MS instruments are mostly relatively simple devices, neither offering multi-stage MS capabilities nor high resolution. As ambient ionization techniques are direct MS techniques, i.e. everything ionized is transferred into the MS, unambiguous identification of substances is often hampered by these shortcomings. Unfortunately, this fact somewhat diminishes the applicability of the proposed combination with respect to actual on-site analysis of food samples unless more powerful portable MS instruments become available.

Advantages and caveats of combining SPME with DART-MS have already been discussed in the section on environmental analysis. This strategy has, to achieve cleaner samples and to enrich the analytes of interest, also been adopted to food analysis. Typical examples are the determination of pharmaceutical drugs in bovine tissue [48] and that of drugs of abuse in infant formula [49]. The group of Pawliszyn presented a further development of SPME fibers used for sample introduction into DART in 2017 [50]. They modified a mesh employed for transmission DART by coating it with adsorbent particles. In this way, SPME extraction could be easily combined with transmission mode DART-MS. The advantages of this approach were demonstrated by analyzing a range of pesticides in food and environmental matrices, whereby the SPME-mesh led to limits of detection in the low  $\mu\text{g L}^{-1}$  range and linearity over almost three orders of magnitude even when complex matrices (such as sugar rich juices) were analyzed. The SPME-mesh allows improving sensitivity and provides cleaner extracts without sacrificing speed. A further advantage of this device is that it can be stored at low temperatures for months allowing retrospective analysis [50].

Yong et al. followed a different approach, namely the direct analysis of wine without any sample pre-treatment [51]. They presented the simultaneous detection of 31 pesticides in wine by DART-MS/MS using a simple customized steel mesh allowing the introduction of a larger sample volume. Thereby sufficient sensitivity was achieved when employing the QQQ in the multiple reaction monitoring (MRM) mode. Repeating the experiment with the same equipment but introducing a preceding QuEChERS extraction step did not improve results while leading to much longer overall analysis times.

Analysis of highly polar pesticides is a quite problematic issue, not only in environmental but also in food samples. Direct (i.e., without derivatization) analysis with GC is often not possible and conventional reversed-phase HPLC methods struggle to resolve the analyte peaks from matrix eluting with the void volume. For this reason DART-MS might be a promising alternative. Based on this assumption, Lara et al. investigated the applicability of combining a quick polar pesticide extraction method with DART coupled either to a time-of-flight (TOF) or an Orbitrap MS for the analysis of these substances in lettuce and celery [52]. By this combination good detection limits (in the  $\mu\text{g/kg}$  range) and recovery rates (70–115%) were achieved. Further work reporting the use of DART-MS includes the determination of pyrrolizidine alkaloids (plant toxins) in herbs and food samples [53] and the detection of illegal cationic dyes in various foods [54].

### 3.2 *Techniques Based on Desorption Atmospheric Pressure Chemical Ionization (DAPCI)*

Desorption atmospheric pressure chemical ionization (DAPCI), involving corona discharge, is an AIMS method whereby molecules emitted from a sample surface upon impact of a heated gas react with reagent species generated by an atmospheric pressure corona discharge [6]. For an overview on the use of techniques based on a DAPCI process in food analysis, see Table 3.

This technique was employed for evaluating the authenticity of scotch whiskies by Smith et al. [57] and for probing the surface of oranges by Zhang et al. [58]. The latter study compared several AIMS techniques for analyzing the tissue of navel oranges [58]. Part of the comparison were DAPCI (heated and without heating), microwave plasma torch (MPT), and extractive electrospray ionization (EESI). In a similar way as expected for the “conventional” ionization techniques, DAPCI excelled for the less polar components, whereas the ESI-like EESI was better suited for more polar analytes. One outcome of the investigation was a list of substances detected at the surface of the fruits providing information on the performance (signal intensities) of the selected AIMS techniques.

Rapid evaporative ionization mass spectrometry (REIMS) was introduced by the group of Takats in 2009 by interfacing an electrosurgical unit (knife) with an MS instrument [67]. This new AIMS technique was primarily designed for use in medicine, but soon its benefit also for food analysis became obvious. One special feature of this technique is that it also offers the possibility to selectively probe certain spots on a sample surface, thereby allowing some sort of an “imaging” approach. Most investigations in the field of food analysis employing REIMS technology are either directed towards categorizing with respect to quality issues [59–62], or more focused on authenticity [63–66]. Starting with the first group, REIMS was mostly employed for testing different meat products. Using an untargeted approach for analysis, Verplanken et al. managed to identify boar taint (an off-flavor typical for non-castrated male pigs) in 150 samples of fat tissue [59]. Closer inspection of the data revealed that fatty acid and phospholipid profiles provided the highest amount of information suitable for categorization of the meat samples. Combining REIMS with machine learning, Gredell et al. were able to predict meat quality attributes such as tenderness, breed type, or carcass type [60]. Addition of cheap offal products to minced beef is a common strategy in food fraud. REIMS allowed the fast and unambiguous identification of adulterated minced meat containing 1–10% of offals [61]. REIMS can also be used for evaluating food processing methods. Song et al. investigated prawn processed using the more and more common air-frying technique [62]. Thereby a clear connection between frying temperature and the amount of lipid oxidation was revealed. Coming to investigations on food authenticity, two recently published research papers on discriminating fish species based on REIMS measurement should be considered [63, 65]. Main goal of both studies was the detection of mislabeling whereby Song et al. focused on salmon and rainbow trout [65]. Black et al. compared REIMS with

**Table 3** Applications of ambient ionization techniques in food analysis – Techniques based on desorption atmospheric pressure chemical ionization

Ambient ionization technique	Mass analyzer	Sample	Analyte(s)/ application	Comment	Reference
DAPCI	QQQ	Scotch whisky	Fingerprinting of authentic and counterfeit samples	Principal component analysis has been applied to dimensionally reduce the data and discriminate between sample groups	[57]
DAPCI/ MPT/ iEESI	LIT	Oranges	Comparative screening analysis	Comparative analysis of navel orange tissue samples with different ambient ionization methods	[58]
REIMS	TOF	Pig neck fat	Boar taint	Classification of tainted and untainted boar samples was achieved	[59]
REIMS	TOF	Beef striploin sections	Quality predictions	Eight different machine learning algorithms were compared to generate predictive models using REIMS data to classify beef quality attributes	[60]
REIMS	TOF	Minced beef samples	Identification of offals within minced beef samples	Offal identifications were obtained for both raw and boiled adulterated samples	[61]
REIMS	TOF	Prawns	Lipid oxidation	The effect of air-frying on the lipid oxidation in prawn at different temperatures was investigated	[62]
REIMS	TOF	Five different white fish species	Metabolomic profiling	478 samples were investigated to detect fish fraud	[63]
REIMS	TOF	Pistachio	Authenticity assessment	Differentiation of pistachio nuts coming from different geographical origins	[64]
REIMS	TOF	Salmon and rainbow trout	Discrimination of fish samples	Lipidomics study for real-time discrimination of salmon and rainbow trout without sample preparation	[65]
REIMS	TOF	Honey	Identification of botanic origin	Classification of honey samples was achieved through principal component analysis-linear discriminant analysis	[66]

polymerase chain reaction (PCR) for fish species identification, testing almost 480 fish samples with five different types of white fish [63]. Thereby REIMS excelled with results comparable to those from PCR, but achieved within a substantially reduced time span (15–20 min versus 24 h). In addition, REIMS has also been successfully employed for authenticity assessment of pistachio [64] and determining the botanical origin of honey [66].

### 3.3 *Paperspray (PS) Ionization and Related Techniques*

The group of Ouyang first reported PS ionization in 2010 [68]. Due to its simplicity, it was soon considered as an additional useful technique within the group of AIMS methods [2]. Throughout the years the basic idea of PS was further modified, thereby implementing additional features and leading to PS-related ionization techniques such as thin-layer chromatography-spray mass spectrometry (TLC-MS) [69], thread spray MS [70], spraying from carbon fibers [71], wooden tips [72], molecularly imprinted polymers [72–74], or steel blades [75, 76] (just to name few). All those techniques were also employed in the analysis of foods and beverages. For an overview on the use of these techniques in food analysis, see Table 4.

Starting with conventional PS-MS, the straightforwardness of this method paired with the fact that (chromatography) paper is a cheap and easily accessible material (that can be disposed after single use), made this technique well suited for the analysis of “dirty” samples or for samples where other approaches would have required at least some basic sample pre-treatment (e.g., in order to protect the instrumentation). A convincing example for that is the high-throughput analysis of vitamin E in olive oil, whereby the oil was simply diluted with acetone and directly applied to the paper triangle for spraying [77]. Furthermore, PS-MS was successfully employed for the analysis of relevant ingredients in a range of food and beverage samples. Pepper was investigated with respect to pungency by an approach called imprint PS-MS [78]. Thereby, for direct analyte transfer, the freshly cut pepper was pressed for a fixed time onto triangular paper that was subsequently used for spraying. This allowed evaluating the capsaicinoid ion intensity for a series of peppers, whereby the results agreed well with values of pungency from the Scoville scale. PS coupled to a triple quadrupole (QQQ) MS for improved selectivity was also employed for analyzing resveratrol in wine [79], tyrosol and hydroxytyrosol in extra virgin olive oil [80] whereby, employment of labeled standards and isotope dilution allowed successful quantitation of the selected analytes. Further examples for PS-MS in food analysis are the detection of methylxanthines in cocoa products [81], and that of ginkgo flavonoids [82]. Besides these works focusing on the analysis of selected food ingredients (or groups thereof), studies have been devoted to the determination of pesticides [83, 84], preservatives [85], illegal dyes [86], and antimicrobial residues [87] in foods and beverages employing PS-MS. Combining chemometric tools with PS-MS analysis is a promising approach for studying the authenticity of food- and beverage-related products. Groups from Brazil published

**Table 4** Applications of ambient ionization techniques in food analysis – Paper spray ionization and related techniques

Ambient ionization technique	Mass analyzer	Sample	Analyte(s)/application	Comment	Reference
TLC spray ionization	TOF	TLC separations/sunscreens	Reaction products/UV filters	A direct MS method to analyze spots from TLC plates with aluminum foil backing without the need of an external ion source or devices for analyte transfer from TLC to MS	[69]
Thread spray ionization	LIT/Orbitrap	Pepper products	Capsaicinoids	Method for sampling the interior of pepper fruits and the analysis of pepper spray residues on clothing	[70]
CFI	TOF	Tomatoes	Pesticides	Qualitative and quantitative analysis of trace pesticide residues from the surface of intact tomatoes	[71]
PSI with MIP	Orbitrap	Apple, banana, and grape methanolic extracts	Diuron and 2,4-dichlorophenoxyacetic acid	MIP was synthesized directly on a cellulose membrane	[73]
MIPCS	QTRAP	Milk	Fluoroquinolone antibiotics	Molecularly imprinted polymer-coated stainless steel sheets were used as ESI emitter	[74]
MIPCWT	FT-ICR	Milk, honey, and water	Macrolide antibiotics	Solid-phase microextraction via the modification of a layer MIP coating on the surface of wooden tips	[72]
CBS	QQQ	Apple, blueberry, grape, and strawberry	Pesticides	Analysis was done through direct-coupling with MS and liquid chromatographic analyses	[75]
CBS	Q/QQQ	Milk	Sulfonamide antibiotics	USB-powered coated blade spray ion source for on-site testing	[76]
PSI	QQQ	Olive oil	Vitamin E	Oil samples were also analyzed with a classical HPLC-FLD method	[77]
PSI	IT	Peppers	Pungency determination	Pepper fruit was pressed onto a triangular paper to imprint and transfer the pepper constituents onto the paper surface	[78]
PSI	QQQ	Red wine	Resveratrol	Quantitative method based on isotope dilution	[79]
PSI	QQQ	Olive oil	Hydroxytyrosol and tyrosol		[80]

(continued)

Table 4 (continued)

Ambient ionization technique	Mass analyzer	Sample	Analyte(s)/application	Comment	Reference
PSI	QQQ	Cocoa products	Theobromine, theophylline, and caffeine	The free form of tyrosol and hydroxytyrosol and their ester conjugates were assessed	[81]
PSI	Q	Powder of ginkgo biloba preparations	Ginkgo flavonoids	Paper spray mass spectrometry assisted by microwave-assisted hydrolysis	[82]
PSI	QQQ	Wine	36 pesticides	An off-line QuEChERS treatment and on-line paper adsorption treatment were used	[83]
PSI	IT	Tomatoes	Acephate, chlorpyrifos, and cyazofamid	Pesticide concentrations during pre-harvest intervals were investigated and the fruits were separated into field and stored groups	[84]
PSI	Q	Beverages	Benzoic acid and vitamin C	Deuterated internal standard was used and results were compared to a HPLC method	[85]
PSI	QQQ	Beverages	Six illegal colorants	Samples were diluted in methanol and isotopically-labeled internal standards were used for quantification	[86]
PSI	LIT	Beef, chicken meat, milk, and egg	Nine antimicrobials	Limits of detections were in the µg/kg range	[87]
PSI	IT	Scottish whiskies	Identification of counterfeit products	PSI combined with partial least square discriminant analysis was applied to discriminate authentic and counterfeit samples	[88]
PSI	FT-ICR	Whisky	Quantification of sugarcane spirit addition	19 binary mixtures were analyzed and evaluated by chemometric models	[89]

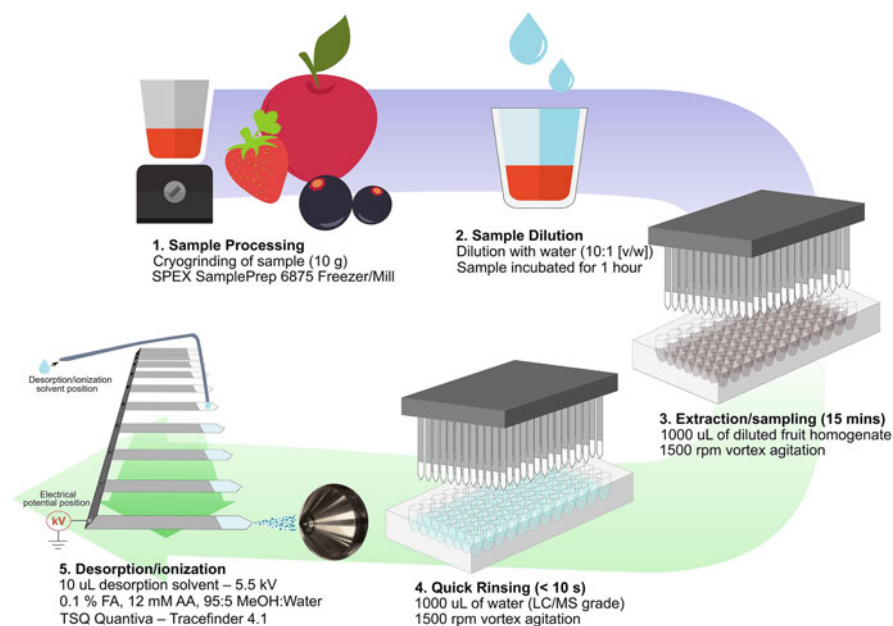
two papers dealing with investigations on the adulteration of whisky by blending [88, 89]. Focusing on the latter paper [89], the proposed approach is rather straightforward, as solely spectra of untreated liquid samples are measured in positive and negative ionization mode, which are subsequently processed using chemometric methods. Coupling ultra-high-resolution MS (employing a Fourier-transform ion-cyclotron MS instrument) with PS allows highly accurate mass determination, and spectral resolution even of ions with small differences in  $m/z$  ratios, a fact that is of substantial significance when measuring spectra from samples directly without any preliminary treatment. This combination and using a multivariate calibration model succeeded in determining the amount of sugar cane spirit added to a series of whisky samples with high accuracy [89].

As already mentioned above, the principle of PS has been further refined by introducing a series of other spraying media, some of them adding further functionalities such as sample pre-concentration, sample cleaning, or rudimental separation of sample constituents [72–76]. An example for such development are spraying devices based on molecularly imprinted polymers (MIPs) [72–74]. Thereby devices including MIPs were first employed to extract agrochemicals from foodstuff [73], fluoroquinolone antibiotics from milk [74], and macrolide antibiotics from milk and honey [72], and subsequently for generating a spray introducing the sample-ions into the MS instrument. In one later paper, coating a wooden tip (subsequently employed for spraying) with a MIP moiety provided enrichment factors between ten (erythromycin from milk) and more than 1,500 (roxithromycin from water) [72].

Coated blade spray (CBS) was first reported by the group of Pawliszyn in 2014 [90]. Coating a metal blade with an appropriate SPME material allows the use of a single device for extracting the analyte from a matrix (ideally providing a cleaning and a pre-concentration effect), and subsequently spraying it into the MS. This was demonstrated on the example of multiresidue pesticide quantitation in a range of fruit matrices by Kasperkiewicz and Pawliszyn [75]. They presented a device particularly suitable for screening large sample sets (one of the major domains of AIMS techniques) allowing the handling of up to 96 samples at a time. Figure 7 provides an overview of the different steps included in the developed workflow.

A major goal pursued by many developments in AIMS is to bring instrumental analysis closer to the sample, i.e. to the field, allowing analyzing food directly at the site of cultivation. An important step towards the realization of this objective has been achieved by Jager et al. when presenting a coated blade spray (CBS) device where the voltage needed for spraying is solely provided by a micro USB connection [76]. Although (unfortunately) not presented in that study, combining USB-powered CBS with a fieldable MS instrument (for possible instruments, see, for example, [91]) would be an excellent solution for real on-site analysis.





**Fig. 7** Workflow for the analysis of pesticides in fruit matrices by coated blade spray MS/MS. From [75] with permission

### 3.4 Desorption Electrospray Ionization (DESI) and Related Techniques

DESI [92, 93] and techniques with a related ionization mechanism such as desorption corona beam ionization (DCBI) [94], EASI [95], LTP [96], and DBDI [97] have successfully been employed in the analysis of food. As already stated in Sect. 2.2, a major asset of DESI is the possibility to scan surfaces and thereby create an image depicting the spatial distribution of selected analytes. For an overview on the use of DESI and related techniques in food analysis, see Table 5.

Mainero Rocca et al. [92] employed DESI for probing the surfaces of vine and olive leaves for the presence of pesticides. Thereby they did not take advantage of the “imaging” abilities of DESI (provided a x,y,z-movable stage for sample positioning is available) but punched out certain areas of the leaves. The group of Spengler investigated the possibility to use DESI for qualitative and (at least semi-) quantitative analysis of pesticides on fruit surfaces [93]. The developed method was seen as a useful tool for fast (qualitative) screening of large sample sets as quantitative analysis (despite using an isotopically labeled ISTD) suffered from a high scatter of calibration curves when compared to routine methods. From the results presented in these works it can be seen that direct quantitation by DESI from real samples is still affected with a series of problems. A fundamental shortcoming of

**Table 5** Applications of ambient ionization techniques in food analysis – DESI and related techniques

Ambient ionization technique	Mass analyzer	Sample	Analyte(s)/ application	Comment	Reference
DESI	QTRAP	Olive and vine leaves	Dimethoate, tebuconazole, and trifloxystrobin	Workers exposure assessment during field re-entry operations since evidence suggests an association between chronic occupational exposure to some agrochemicals and severe adverse effects	[92]
DESI	Orbitrap	Fruit surfaces and extracts	32 pesticides	Quantification of compounds was carried out from QuEChERS extracts and food surfaces using only one isotope-labeled ISTD for normalization	[93]
DCBI	IT	Pork and fish	10 antimicrobials	Analytes were detected in the $\mu\text{g}/\text{kg}$ range	[94]
EASI	Orbitrap	Salmon species	Triacylglycerols	Investigations on the impact of the raising regime of salmon species on their triacylglycerol composition	[95]
LTP	LCQ	Tequila and mezcal	Chemical fingerprinting	Method permits the discrimination between tequila and mezcal products	[96]
DBDI	Orbitrap	Grape juice	24 pesticides	Solid-phase microextraction was used for sample preparation	[97]
LEP	Portable mass spectrometer Mini 11	Pepper peel, fish, and coffee	Pesticides, plasticizers, and drugs	Food authentication via metabolomic fingerprinting and multivariate statistics is demonstrated for the analysis of fish fillets and coffee	[98]
TD-ESI	QQQ	Processed vegetables	Basic colorants	Detecting adulterants in processed vegetables	[99]
TD-CFI	QQQ		Capsaicin		[100]

(continued)

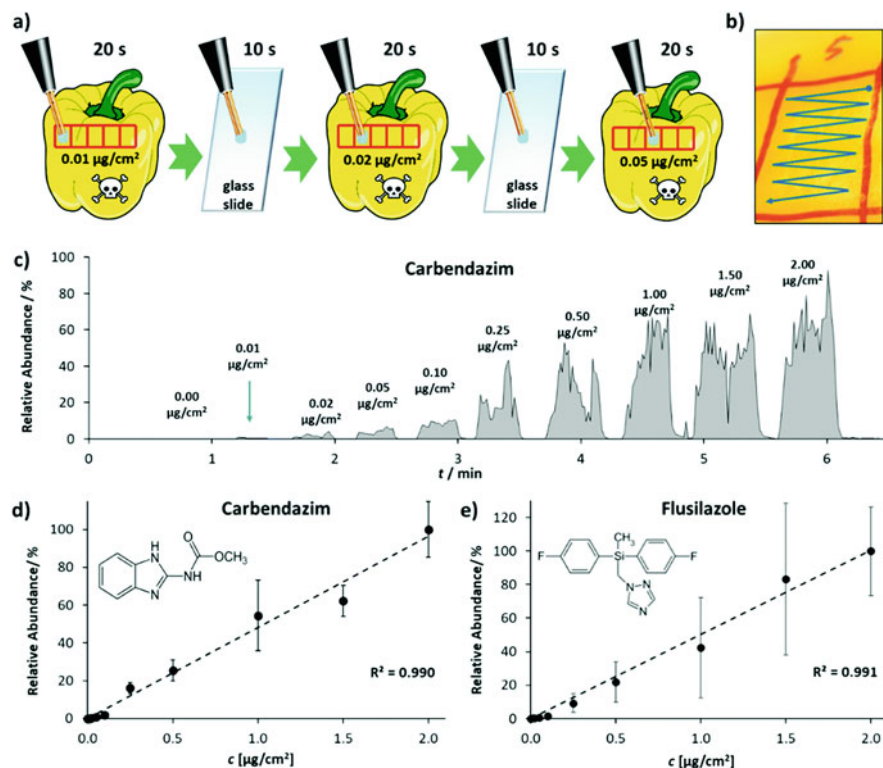
**Table 5** (continued)

Ambient ionization technique	Mass analyzer	Sample	Analyte(s)/ application	Comment	Reference
		Ethanollic extracts of cookies, chips, and oil		A metal ceramics heater was used for analyte desorption	
TD-ESI	QQQ	Fruits and vegetables	308 pesticides	The residual pesticides on fruits and vegetables were collected by sweeping a metallic probe across the sample surface	[101]
EESI	Orbitrap	Nectars and honeys from citrus	12 polyphenols and 8 amino acids	Nectar and its corresponding honey were chemically differentiated using principal component analysis	[102]

all surface-sampling methods is, that the surface texture substantially influences ionization efficiency, a fact making quantitative analysis highly problematic.

DCBI and EASI are both ambient ionization techniques categorized as direct desorption ionization techniques according to Huang et al. [6]. They have been successfully employed for analysis of meat slices (with respect to antimicrobials) [94], and fish with respect to product quality [95]. In the latter study, triacylglycerol profiles were determined in order to draw conclusions on the raising regime of the respective salmon species. Following a thermal imprinting process for extraction, EASI proves a simple, rapid, and effective technique for distinguishing tissue from fish raised upon different diets [95]. Similar as already discussed in the context with DART and PS, SPME has been interfaced to DBDI, an ambient ionization technique belonging to the same group as DESI [97]. This combination allowed a high-throughput analysis of grape juices for the presence of pesticides. The group of Spengler constructed a handheld liquid extraction pen for on-site MS analysis comparable to the iknife employed for REIMS (when looking at the handling of the device) but based on a DESI mechanism for ionization [98]. Figure 8 depicts the measurement procedure when analyzing pepper peel for pesticides using this device. The substantial error bars recognizable in Fig. 8d are typical for surface-sampling methods (as already discussed above) as surface texture severely influences the ionization process. Nevertheless, coupled to a portable MS instrument the technique developed can be regarded as an ideal tool for on-site analysis, allowing the accurate categorization of samples.

Further AIMS techniques based on a two-step ionization mechanism and employed in the field of food analysis are thermal desorption electrospray ionization (TD-ESI) and EESI. In TD-ESI the sample is placed on a heated ceramic part and



**Fig. 8** Quantitative analysis of pesticides from pepper peel, using the self-sustaining and portable LEP source attached to an orbital-trapping mass spectrometer. (a) Sampling procedure for pesticide quantification from pepper peel. (b) Sampling across the spiked sampling spot. (c) Ion intensity chromatogram for different Carbendazim concentrations, detected from pepper peel. (d) and (e) show calibration curves, obtained for the two pesticides carbendazim and flusilazole. Data points represent averages of three technical replicates. Error bars indicate standard deviation. From [98] with permission

evolving vapors are ionized by an ESI process. This technique was used for analyzing food in several cases [99–101], comprising the analysis of basic colorants in vegetables [99], capsaicin in various food samples [100], and pesticides on surfaces of fruit and vegetables [101]. In EESI a sample solution is nebulized and the resulting spray is subsequently mixed with an electrospray generated from a solvent. An example for the use of this AIMS technique in food analysis was provided by Gao et al. who compared the chemical composition of various nectars and honeys [102].

## 4 Conclusions and Perspectives

Over the recent decade, AIMS techniques have shown substantial potential in both environmental and food analysis, whereby the latter distinctly dominates when retrieving the number of publications from the relevant databases. Focusing on trends and developments, one thing that can be observed is that frequently devices for sample preparation (clean-up and/or pre-concentration) are implemented in AIMS techniques (for typical examples, see [26, 72, 74, 75, 89]). Such combinations seem sensible as reducing sample complexity and increasing analyte concentrations to meet the requirements for analysis are of utmost importance in all kinds of AIMS techniques. This is the case for applications in the fields of food analysis but even more pronounced in environmental analysis, where (ultra-) trace analysis is the norm.

Right from the moment of their appearance, AIMS techniques were considered as highly suitable for the analysis of large sample sets – mostly for qualitative screening sometimes providing additional (at most) semi-quantitative information. Particularly the latter issue was addressed in a range of reports over the last decade. Common approaches are the use of isotopically labeled standards, ideally for every analyte of interest – a situation that often cannot be reached, as suitable substances are either too costly or not available at all. In addition, exact positioning of the sample within the device for ionization is mandatory, when reproducible (semi-)quantitative results are required – a fact boosting the use of automated sample positioning systems. What also needs to be mentioned is, that the use of AIMS techniques in combination with complex samples unavoidably requires the application of chemometric tools for data treatment. Also in this field substantial development could be observed over the last decade.

A major trend in environmental analysis but even more in the analysis of food and related products is on-site analysis. Due to their ease of use and the possibility to probe samples “as they are,” AIMS techniques are well suited to meet the challenges related to on-site analysis. Ideally all parts of the equipment (i.e., the ion source as well as the MS instrument with all its peripheral devices) should be portable. A good example for a fieldable ion source can be found in the publication of Jager et al. [76]. Despite this development, the introduction of truly portable MS instruments for combination with ambient ionization is affected with several obstacles. One main deficit can be found in the fact that instrumentation described so far includes mainly very simple devices with low resolution and no MS/MS capabilities. As AIMS techniques tend to deal with the complete sample, a situation affected with a high probability of matrix interferences, HR-MS or multi-stage MS is often required in order to achieve useful results.

Finally, instrumentation primarily developed for use in a medical (clinical) environment has found its way into the field discussed in this chapter. A typical example is the REIMS technology with the iknife that was successfully employed in food analysis [60–66].

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# Ion Mobility-Mass Spectrometry in Food and Environmental Chemistry



Shon P. Neal and Christopher D. Chouinard

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**Abstract** Ion mobility-mass spectrometry (IM-MS) has emerged as a powerful analytical technique currently used in numerous fields including food and environmental chemistry. IM-MS can be employed with several ionization methods and coupled with chromatographic separations to provide multidimensional identification of compounds based on the combination of retention time, collision cross section (CCS), and accurate mass. Experimentally obtained CCS values can be used to populate application-specific libraries that further allow identification and quantification in complex mixtures. This chapter will highlight recent advances in IM-MS methods and technology applied to the food chemistry and environmental chemistry fields. First, instances of IM-MS in nutritional analysis, food safety, food fingerprinting, and process control and quality assurance/quality control (QA/QC) will be discussed. Environmental applications including analysis of per- and

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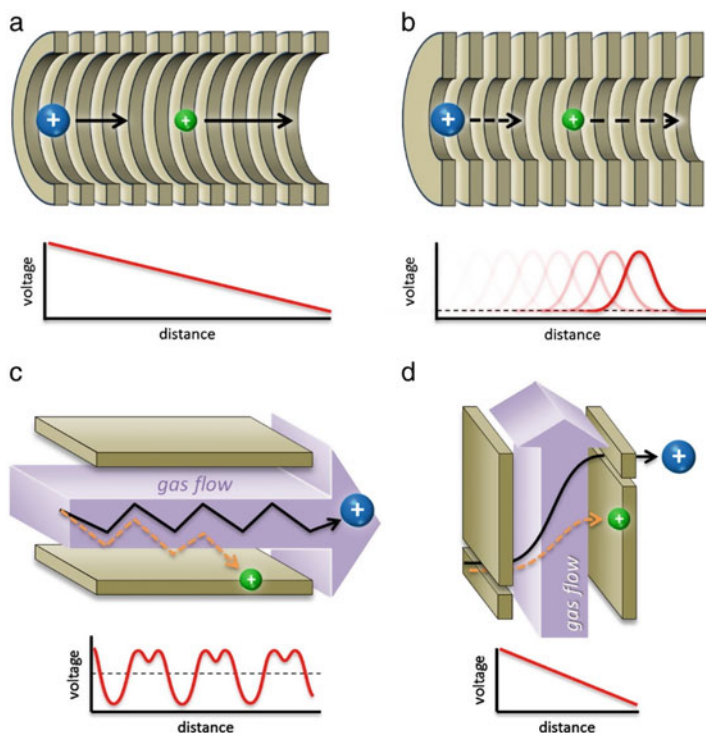
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polyfluoroalkyl substances (PFASs); polycyclic aromatic hydrocarbons (PAHs), benzene/toluene/xylene (BTX), and volatile organic compounds (VOCs); pesticides; and pharmaceutical and personal care products (PPCPs) will also be outlined. Overall, it is expected that IM-MS will continue to grow due to additional information offered in comparison with conventional methods and because of increased commercialization and improved resolution.

**Keywords** Collision cross section, Environmental chemistry, Food chemistry, Ion mobility spectrometry (IMS), Ion mobility-mass spectrometry (IM-MS), Mass spectrometry (MS)

## 1 Introduction to Ion Mobility-Mass Spectrometry (IM-MS)

Ion mobility spectrometry (IMS) is an analytical technique that measures the movement of gas-phase ions through a buffer gas environment under the influence of an electric field [1]. As ions move through the buffer gas (typically helium or nitrogen, maintained at pressures ranging from  $\sim 1$  Torr to atmospheric pressure), they undergo collisions and travel at different velocities based on their size, shape, and charge, such that larger ions experience more collisions and thus travel more slowly [2]. Several different varieties of IMS have been developed which differ in the nature of the electric field applied (Fig. 1) [3], including drift tube (DTIMS), traveling wave IMS (TWIMS) [4], differential mobility spectrometry (DMS)/high field asymmetric waveform IMS (FAIMS) [5], differential ion mobility (DIM)/differential mobility analysis (DMA) [6], and trapped IMS (TIMS) [7]. DTIMS uses a static electric field which causes ions to achieve a constant drift velocity through the separation cell (Fig. 1a), such that the time to traverse the separation cell can be measured as “drift time”; this technique is analogous to the way a time-of-flight mass spectrometer (operated at high vacuum) measures mass-to-charge ratio ( $m/z$ ). TWIMS, also a time-dispersive technique, applies a dynamic voltage profile such that ions will either “ride the wave” or “roll back” based on their mobility (Fig. 1b). DMS/FAIMS applies an alternating high and low electric field (for short and long durations, respectively) perpendicular to the main ion motion. The resulting sawtooth motion is a result of ions’ net motion toward one of the two electrodes. Meanwhile, a compensation field superimposed on one of the electrodes can be scanned to sequentially allow transmission of ions of different mobility (Fig. 1c). DMS/FAIMS is analogous to a quadrupole mass spectrometer in that it operates as an ion filter. DIM/DMA applies a constant electric field which is perpendicular to ion motion propelled by a gas flow. This field can also be scanned to allow for ions of a certain mobility to be transmitted and measured (Fig. 1d). Lastly, TIMS separates ions propelled by a gas flow as they are opposed, or “trapped,” by an electric field. This electric field can be scanned to elute ions based on their mobility.



**Fig. 1** Basic operational principles of contemporary ion mobility techniques. (a) Conventional drift time ion mobility whereby ions are introduced into a gas-filled chamber and separated based upon their differential drift down a continuous declining potential. (b) Traveling wave ion mobility which mobility separates ions in a gas-filled chamber using a dynamic traveling potential hill. (c) Asymmetric field ion mobility whereby ions are separated based upon their differential migration orthogonal to a sweeping gas flow. (d) Differential ion mobility which separates ions based upon their ion mobility-dependent spatial displacement as they traverse an electric field gradient that is applied orthogonal to a constant flow of gas. In all schemes, ion motion is depicted from left to right. Reprinted from “Lipid analysis and lipidomics by structurally selective ion mobility-mass spectrometry,” *Biochimica et Biophysica Acta (BBA) – Molecular and Cell Biology of Lipids*, 2011, Volume 1811, Pages 935–945 with permission from Elsevier [3]

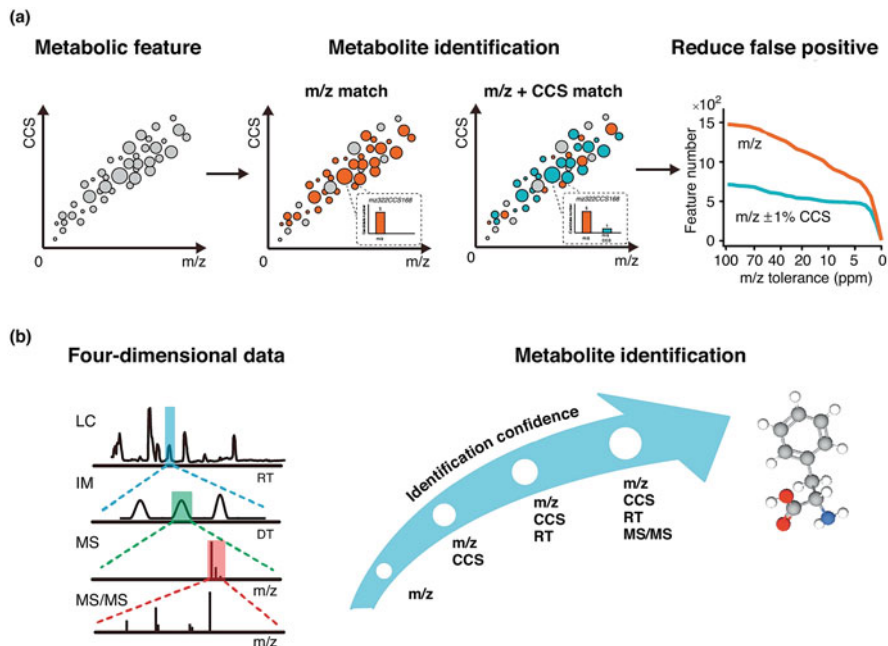
Ions studied by DTIMS can be characterized by their mobility ( $K$ ) which can be calculated by measuring drift velocity ( $v_d = L/t_D$  where  $L$  = drift tube length and  $t_D$  = drift time) as a function of electric field ( $E$ ), or  $K = v_d/E$ . Mobility remains constant, with  $v_d$  proportional to  $E$ , for low electric fields ( $<100$  V/cm). This relationship allows for calculation of an ion’s rotationally averaged collision cross section (CCS), or the average two-dimensional area of an ion that undergoes collisions with a given buffer gas. For consistent experimental conditions, especially drift gas composition, temperature, and pressure, CCS is representative of a given ion and thus can be used for identification purposes. As will be seen in later sections of the chapter, CCS databases allow for improved confidence in determining

components of complex mixtures, especially when compared with chemical standards. Furthermore, measurement of CCS provides a means by which to study molecular structure and changes therein [8]. Modern computational modeling methods allow for fairly accurate determination (<2% difference between experimental and theoretical CCS values) of three-dimensional gas-phase structure when used in conjunction with experimentally obtained CCS [9]; these modeling strategies have also demonstrated some predictive capacity and are currently being implemented with machine-learning algorithms to populate massive theoretical CCS libraries [10].

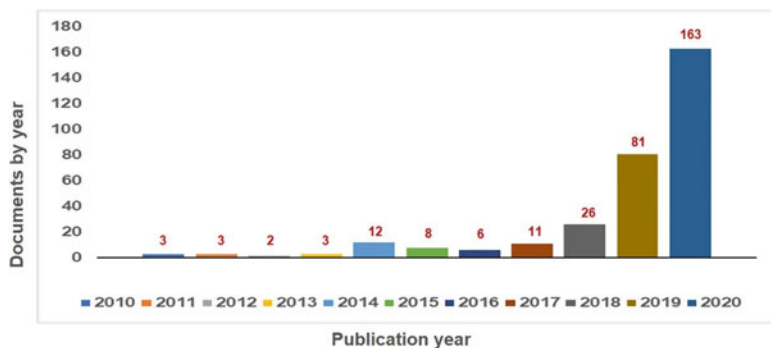
IMS is a fast technique, with time-dispersive methods capable of acquiring a full spectrum in tens of milliseconds or less. Filtering methods such as DMS/FAIMS can operate even faster, with duty cycle approaching 100% for targeted analysis of single or a few analytes. Standalone IMS has seen utility in the defense industry dating back to the 1970s [11] and is still used extensively in modern applications, however its coupling with mass spectrometry (i.e., IM-MS) has expanded its utility for untargeted and biological analyses considerably. Its speed conveniently allows for nesting between chromatographic separations (minutes) and fast acquisition mass spectrometers such as time-of-flight, TOF (hundreds of microseconds). This added separation creates multidimensional techniques (e.g., GC-IM-MS, LC-IM-MS/MS, etc.) that provide multiple measures for a given ion, including retention time, CCS, accurate mass, and fragmentation pattern. Such multidimensional approaches can significantly reduce false positives in analysis (Fig. 2) [9]. Commercialization of IM-MS instrumentation by nearly all major mass spectrometry vendors has caused significant proliferation of its use in structural biology [12], -omics analyses (e.g., metabolomics, lipidomics, proteomics, etc.) [13], and the clinical laboratory [14]. The most recent advances in higher resolution IM-MS instrumentation have included TIMS [7], structures for lossless ion manipulations (SLIM) [15], and cyclic TWIMS [16]; these platforms promise to provide more structural information, better separation of isomers, and more precise CCS measurement. As such, it is expected that the IM-MS footprint in nearly all analytical applications, including in food and environmental chemistry, will continue to grow in coming years.

## 2 Introduction to IM-MS in Food Chemistry

The use of ion mobility spectrometry (IMS) in the food industry has grown dramatically over the last decade, as evidenced by the rise in publications referencing IMS since 2010 (Fig. 3) [17]. This trend is expected to continue as IMS and associated techniques (e.g., LC-IM-MS, GC-IM-MS, etc.) are further developed. Currently, the food industry utilizes IMS in four broad categories: (A) nutritional analysis, (B) food safety, (C) food fingerprinting, and (D) process control and quality assurance/quality control (QA/QC). Some of these categories are dominated by GC/LC-IM-MS instrument design and development, while others are focused on portable, inexpensive, and fast standalone IMS instrumentation. For example, nutritional analysis



**Fig. 2** The use of CCS values to support metabolomics and lipidomics. (a) The CCS value is an important criterion to identify metabolites and reduce false positives; (b) the integration of *m/z* value, CCS value, retention time, and MS/MS spectra improves the confidence of metabolite identification. Reprinted from “Advancing the large-scale CCS database for metabolomics and lipidomics at the machine-learning era,” *Current Opinion in Chemical Biology*, 2018, Volume 42, Pages 34–41 with permission from Elsevier [9]



**Fig. 3** Search of literature related to GC-IMS in analysis from 2010 to 2020 on Scopus database. The terms “gas chromatography-ion mobility spectrometry (GC-IMS)” have been included in the search topic. Reprinted from “Application and development trends of gas chromatography-ion mobility spectrometry for traditional Chinese medicine, clinical, food and environmental analysis,” *Microchemical Journal*, 2021, Volume 168, Page 106526 with permission from Elsevier [17]



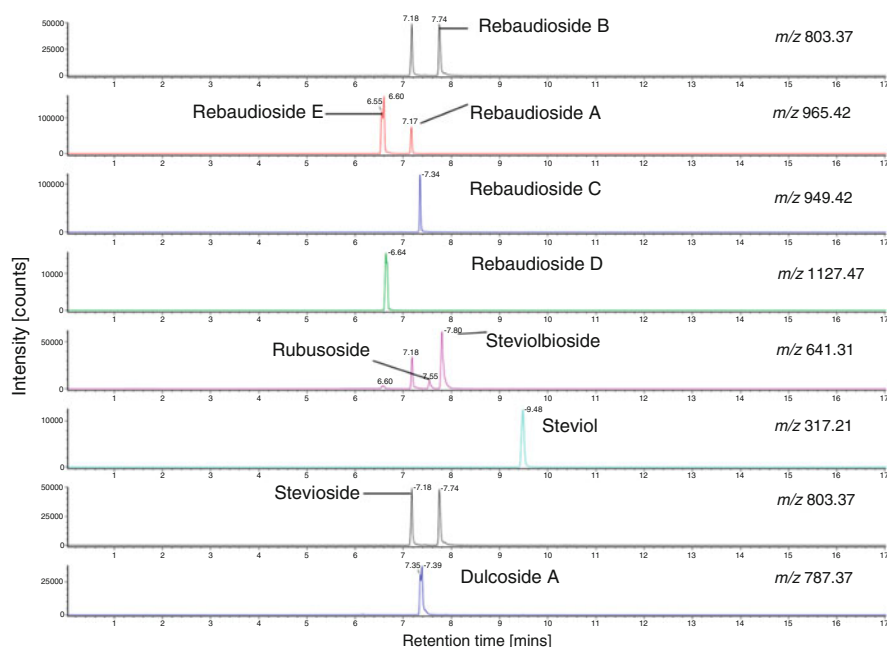
focuses on complete and detailed analysis of complex food matrices, which requires hyphenated systems capable of multiple separation dimensions. Food safety as well should see growth in IM-MS over standalone IMS systems. Governmental regulation drives food safety which generally requires identification capabilities that IMS alone cannot provide [18]. Food fingerprinting and QA/QC, on the other hand, prefer the simplest analytical technique(s) to meet their needs. Lessons learned from more detailed nutritional analysis research (i.e., with more complex hyphenated techniques) eventually translate to more straightforward QA/QC process control methods. This section will discuss the current state of IM-MS in each of the four aforementioned categories.

## 2.1 Nutritional Analysis

Nutritional analysis is the broad categorization of the food matrix into its base nutritional components. Foodstuffs are complex mixtures containing numerous different compounds, many of which are present at very low concentration. However, the resulting requirements for low limits of detection (LODs) and high sensitivity can be complicated by difficulty in separating the myriad of potential interferences. As such, IM-MS has seen increased use in nutritional analysis due to the added separation ability increasing the confidence of identification. The supplemental separation dimension afforded by IM-MS allows for additional filtering of complex datasets and provides cleaner MS spectra. Furthermore, IM measures collision cross section (CCS) values which can be collected to populate nutrition-specific databases that complement existing libraries of chromatographic retention time, accurate mass, and/or MS/MS fragmentation spectra.

Work by Willems et al. is an excellent example of classifying nutritional components by IM-based methods. In this study, ESI-FAIMS-MS was used to separate and identify chlorogenic acid isomers, specifically mono-caffeoylquinic acids which are commonly found in fruits (e.g., apples, pears, etc.) and are among the most abundant phenolics found in the human diet [19]. The three most common (3-, 4-, and 5-caffeoylquinic acid) are isomers and difficult to differentiate with conventional methods. Combining FAIMS and MS/MS provided a method for separation and identification with an analysis time of <1 min, a considerable improvement over HPLC-based methods that often require more than 1 h.

McCullagh et al. used TW-IMS-MS to study steviol glycosides, which are zero-calorie sweeteners used in many manufactured food products [20]. Steviol, a regulated food additive in Europe with a maximum permitted level, currently consists of seven different chemically defined compounds, each with different flavor profile (i.e., sweetness, bitter aftertaste, etc.). The processing industry aims to maximize use of the sweetener while staying under the regulatory limit. McCullagh et al. used TW-IMS-MS to separate and identify isomeric steviol glycoside pairs based on unique CCS values, which eliminated the need to rely on characteristic fragmentation ratios (Fig. 4). This research also has broader implications for food safety and QA/QC



**Fig. 4** Extracted mass chromatograms for steviol and profiled steviosides <100 pg/UL spikes into chocolate. Reprinted from “Exploring the Complexity of Steviol Glycosides Analysis Using Ion Mobility Mass Spectrometry,” *Analytical Chemistry*, 2018, Volume 90, Pages 4585–4595 with permission from American Chemical Society [20]

methods, as it presents potential for development of less expensive and more precise methodologies for such markers.

Groups like Chen et al. combine aspects of food and medical research, such as with their IM-MS studies of the digestive properties associated with food consumption [21]. UPLC-IM-QTOF-MS analysis identified anthocyanins from *Aronia melanocarpa* (AAM), which were involved in decreased lipid content and inflammation in 3 T3-L1 cells without cytotoxicity. Wenjie et al. presented another good example of food/supplement analysis in which they also studied anthocyanins [22]. These compounds, of interest in natural supplements as essential molecules in the so-called super fruits, were analyzed using Hadamard transform ion mobility mass spectrometry (HT-IMMS). The authors were able to separate isomers pelargonidin-3,5-diglucoside (found in pomegranate) from cyanidin-3-rutinoside (found in blackberry). This enabled determination of the anthocyanin ratios in many different fruit types (blueberry, blackberry, raspberry, strawberry, and pomegranate), which offers a possible fingerprinting method.

An extensive database for fatty acid methyl esters (FAMES) was developed using GC with an atmospheric pressure chemical ionization source. The GC-APCI-IMS-TOF-MS method significantly increased selectivity, and thus confidence of identification, of FAMES in this analysis [23]. This research identified trace amounts of

various compounds while also providing a more detailed characterization of the edible oils present. The significantly increased depth and quality of information could be helpful when applied to detecting food fraud or when used in the quality control of manufacturing processes.

Although numerous analytical methodologies are employed in nutritional analysis, ranging from DNA/protein tests to targeted MS/MS, IM-MS techniques will continue to grow and replace other methods in this category due to their precision, speed, and identification capabilities of individual testing. Another noticeable trend is that work completed in this category will provide actionable markers for food safety, QA/QC, and fingerprinting, to be discussed in the following sections.

## 2.2 Food Safety

Governmental regulation drives food safety requirements and thus the analytical methods used to determine food safety. This includes analysis of materials that can cause harm if injected, inhaled, or consumed. However, new methods are adopted slowly, even if they demonstrate tangible benefits. Analytical techniques that increase speed and decrease costs are likely to see growth in this field as screening techniques feeding into the more detailed analysis as required. An important consideration in food safety is governmentally mandated acceptable exposure limits, requiring accuracy and sensitivity in the analytical methods used. As such, MS-based techniques will continue to grow and replace current methods in this category. For instance, steviol derivatives differ significantly in their flavor profile and health safety, but for regulation purposes are often grouped together. It is of interest to the food manufacturers to differentiate and quantitate the different steviols, even though it is not currently required [20]. Despite lack of regulation, IM-MS provides significantly more information than traditional methods and could be of considerable value in the future, allowing manufacturers to stay ahead of the game and comply with potentially new laws, while simultaneously improving food quality by increasing the desired steviol derivative.

Another example of food safety application is analysis of the sealants used in the lids of hermetically sealed food cans. These sealants contain a wide variety of compounds, including associated non-intentionally added substances (NIAS) that can potentially migrate to the food. UHPLC-IM-QTOF-MS has been used to identify such compounds by providing characteristic CCS values and high-resolution  $m/z$ , increasing the confidence in their identification [24]. Another safety application concerns trade organizations, such as the International Olive Council, which have created certain safety regulations for their products [25]. These organizations stand to benefit significantly from the added separation capabilities of IM-MS techniques for monitoring food safety regulations and could adopt the technology more quickly than government-regulated bodies. Despite the widespread use of standalone IMS, its inability to definitively identify molecular composition of potential safety concerns will likely lead to future use of IM-MS and its better identification capabilities.

## 2.3 Food Fingerprinting

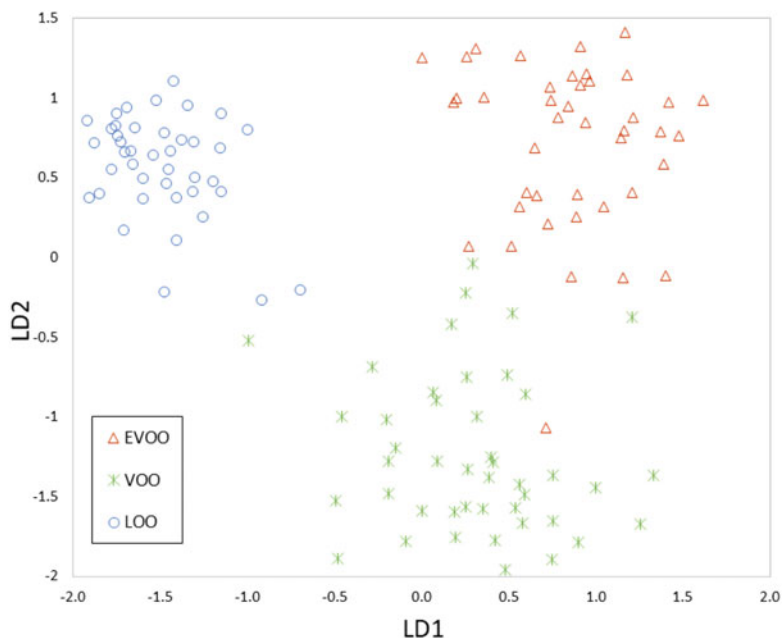
Food fingerprinting is the identification of molecular markers that identify a specific food. Unlike nutritional analysis or food safety, which focus on understanding the food makeup or identifying pesticides and toxins, respectively, food fingerprinting is about determining the origin and purity of the product. Identifying molecular markers, such as a particular analyte or ratio of analytes, representing a characteristic food state or condition provides more effective product discrimination, which is needed because of threats to the world's food supply by fraud, adulteration, and counterfeiting. Many different methods exist to identify and exploit specific molecular markers [26, 27], including MS and IMS techniques which are valuable due to their precision and speed, respectively. Coupling of these methods provides the high sensitivity, selectivity, throughput, and multi-analyte capabilities suited to solve food fingerprinting requirements.

One common product in high demand for fingerprinting is olive oil, a frequent target for alteration or imitation. Manufacturers who market and price their food items based on quality, specific location or origin, or processing techniques can use IM-MS to confirm these properties. For example, García-Nicolás et al. demonstrated both IM-MS and standalone IMS applied to determining the quality level of olive oils (Fig. 5) [28]. Additional studies have been shown capable of identifying specific regions of origin that produced the oil [29].

Arroyo-Manzanares et al. also used headspace (HS) sampling coupled with GC-IMS to differentiate authentic Iberian Ham from fraudulent versions [30]. Their group was able to identify markers based on the different feedstock provided to the pigs. The method improved on prior high-resolution MS fingerprinting because it no longer required extensive sample preparation thus reducing the analysis time. HS-GC-IMS has also been used to differentiate honey according to its purity and level of impurities adulterated from sugarcane or corn syrups (Fig. 6) [32]. These methods determined adulteration and identified counterfeit honey based on the source of the nectar.

The current trend of standalone or GC-IMS techniques to monitor previously identified markers may be giving way to IM-MS-based methods given the economic implications of fraud and counterfeiting and the need to stay ahead of future adulteration schemes. An example is how HPLC-IMS-TOFMS has been used to identify markers that confidently characterize and fingerprint wine using a metabolomics-type workflow [33]. This method distinguished wines by type and location and could help reduce counterfeiting by cheaper wines pretending to be from more desired regions.

Fingerprinting will continue as a growth area for IM-MS methods with expected growth in exploratory sample classification to identify novel markers and/or their ratio(s). Such markers would be used to build new databases that identify unique signatures for food fingerprinting.

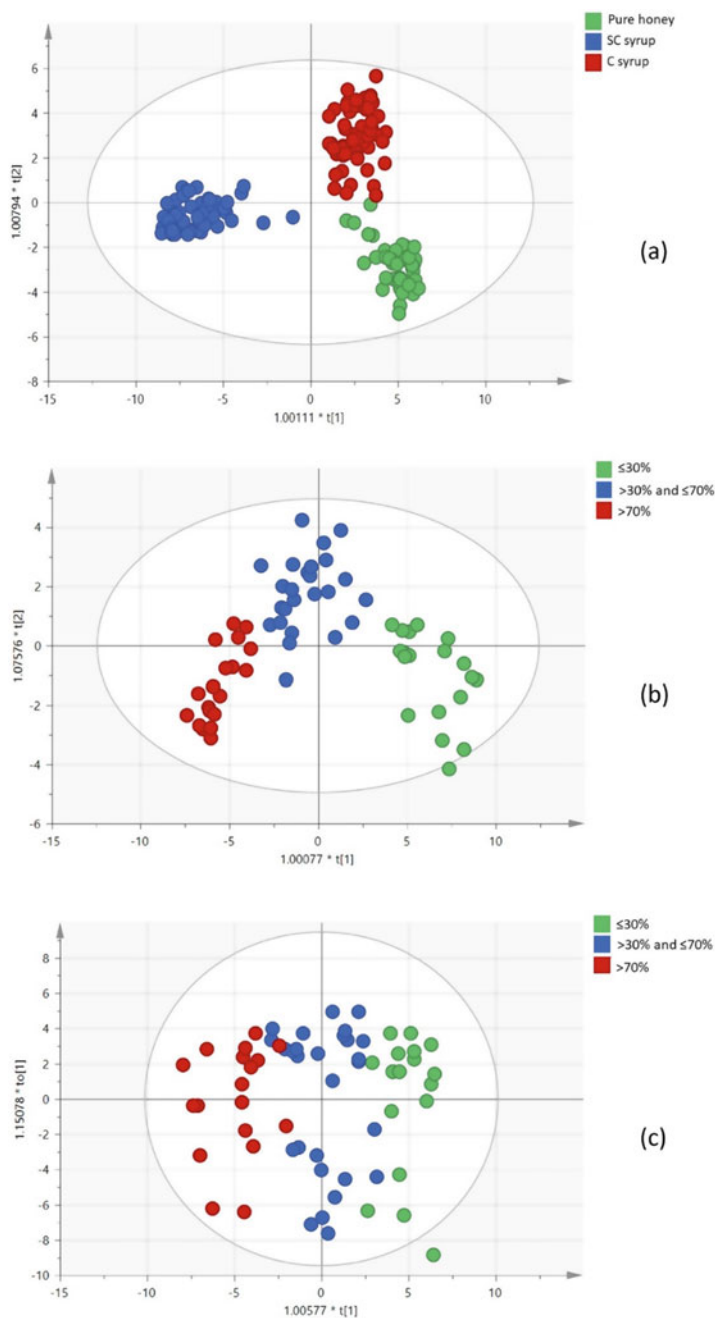


**Fig. 5** Principal component analysis–linear discriminant analysis (PCA-LDA) model constructed using the HS-GC-IMS markers. Reprinted from “Headspace Gas Chromatography Coupled to Mass Spectrometry and Ion Mobility Spectrometry: Classification of Virgin Olive Oils as a Study Case,” *Foods*, 2020, Volume 9, Page 1288 with permission from MDPI [28]

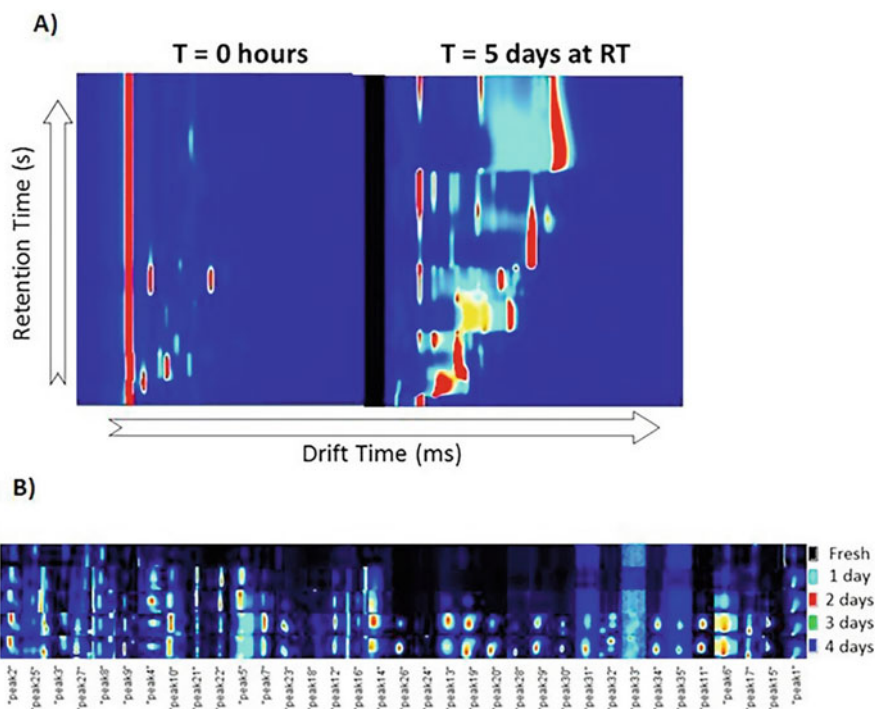
## 2.4 Process Control and Quality Assurance/Quality Control (QA/QC)

Standalone IMS is routinely employed by food companies and researchers for process control due to its low cost, high speed, and ability to test VOCs relating to food quality at the source of production. Such process control and quality assurance/quality control applications, however, can benefit significantly from tools for more sophisticated analysis. A great example of using IMS for process control is how GC-UV-IMS allows for greater precision and shorter hold times for the beer brewing process [34]. Brewing beer requires that diacetyl and 2,3-pentanedione degrade below the odor threshold. Many breweries will allow this process to run multiple days to ensure degradation, but this time-consuming and unnecessary when using an IMS process. IMS can measure diacetyl and 2,3-pentanedione concentration in under 10 min with minimal sample prep. This analysis method decreases current testing costs as well as reducing brewing time while maximizing flavor quality.

QA/QC processes are improved using IMS methods as a rapid and deployable screen tool. A model can discriminate between fresh and spoiled egg products by recording volatile fingerprints (Fig. 7) [35]. The use of IMS offers a low-cost and fast analytical protocol that has also been applied to other food products, such as bread,



**Fig. 6** Ternary OPLS-DA model using normalized data and UV scaling to discriminate between pure and adulterated honey samples. SC syrup: honey samples adulterated with sugarcane syrup; C syrup: honey samples adulterated with corn syrup. Reprinted from “Untargeted headspace gas chromatography – Ion mobility spectrometry analysis for detection of adulterated honey,” *Talanta*, 2019, Volume 205, Page 120123 with permission from Elsevier [31]



**Fig. 7** (a) Ion mobility spectrum of an egg product at  $T = 0$  h (left) and after 5 days at room temperature (right). The red line indicates the Reaction Ion Peak (RIP) position. (b) Global overview of the spots identified in the egg product “Sample\_20” at different time points. Reprinted from “Ion mobility spectrometry coupled to gas chromatography: A rapid tool to assess eggs freshness,” *Food Chemistry*, 2019, Volume 271, Pages 691–696 with permission from Elsevier [35]

in which IMS can be used to measure biogenic amines which affect test as a quality control process. There are still many applications that are expected to benefit from IM-MS technology especially as these methods become less expensive and of higher resolution.

### 3 Introduction to IM-MS in Environmental Analysis

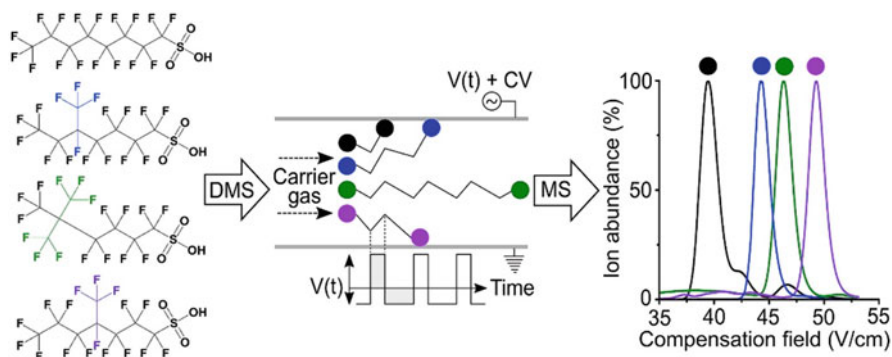
Environmental analysis with MS-based methods primarily focuses on identification and quantification of xenobiotic compounds that pose a toxic threat to ecological health, including both humans and wildlife. These compounds include persistent organic pollutants (POPs) which are often by-products of chemical, manufacturing, and/or agricultural processes. This section will discuss examples of IM-MS in the following environmental applications: (A) per- and polyfluoroalkyl substances (PFASs); (B) polycyclic aromatic hydrocarbons (PAHs), benzene/toluene/xylene

(BTX), and volatile organic compounds (VOCs); (C) pesticides; and (D) pharmaceutical and personal care products (PPCPs).

### 3.1 Per- and Polyfluoroalkyl Substances (PFASs)

Per- and polyfluoroalkyl substances (PFASs) are an environmentally ubiquitous class of persistent organic pollutants. Used in numerous consumer and industrial products including firefighting foam and fire-retardant coatings, and nonstick cookware, PFAS are of significant concern due to their chemically inert properties and toxic effects causing adverse health outcomes (e.g., developmental/hormonal effects, thyroid disease, decreased liver function, etc.). The manufacturing process results in production of many different species including linear and branched isomers with different toxicological properties. The most commonly used analytical approach for identification and quantification is chromatography-mass spectrometry methods, but identification of the many isomers is challenging due to long analysis times and inability to resolve using MS/MS fragmentation pattern.

The first application of ion mobility to PFAS analysis by Ahmed et al. involved high-resolution differential mobility spectrometry (DMS) to separate PFAS isomers [36]. This demonstrated the capability to resolve various linear, secondary- and tertiary-branched species, and even isomers that differed in the position of a single perfluoromethyl group. Figure 8 illustrates the filtering of several isomers by DMS compensation field, allowing resolution of these four compounds. Dodds et al. later used DTIMS to measure CCS for over 50 PFAS species and demonstrated an LC-IM-MS workflow for study of environmental samples [37]. This work also yielded CCS vs.  $m/z$  trends for several subclasses such as fluorotelomer sulfonic acids (FTSAs) and perfluoroalkyl sulfonic acids (PFASAs). Measurement of these



**Fig. 8** Four PFAS isomers, one linear and three branched, are separated based on their transmission through a DMS device at different compensation field. Reprinted from “Rapid separation of isomeric perfluoroalkyl substances by high-resolution differential ion mobility mass spectrometry,” *Analytica Chimica Acta*, 2019, Volume 1058, Pages 127–135 with permission from Elsevier [36]

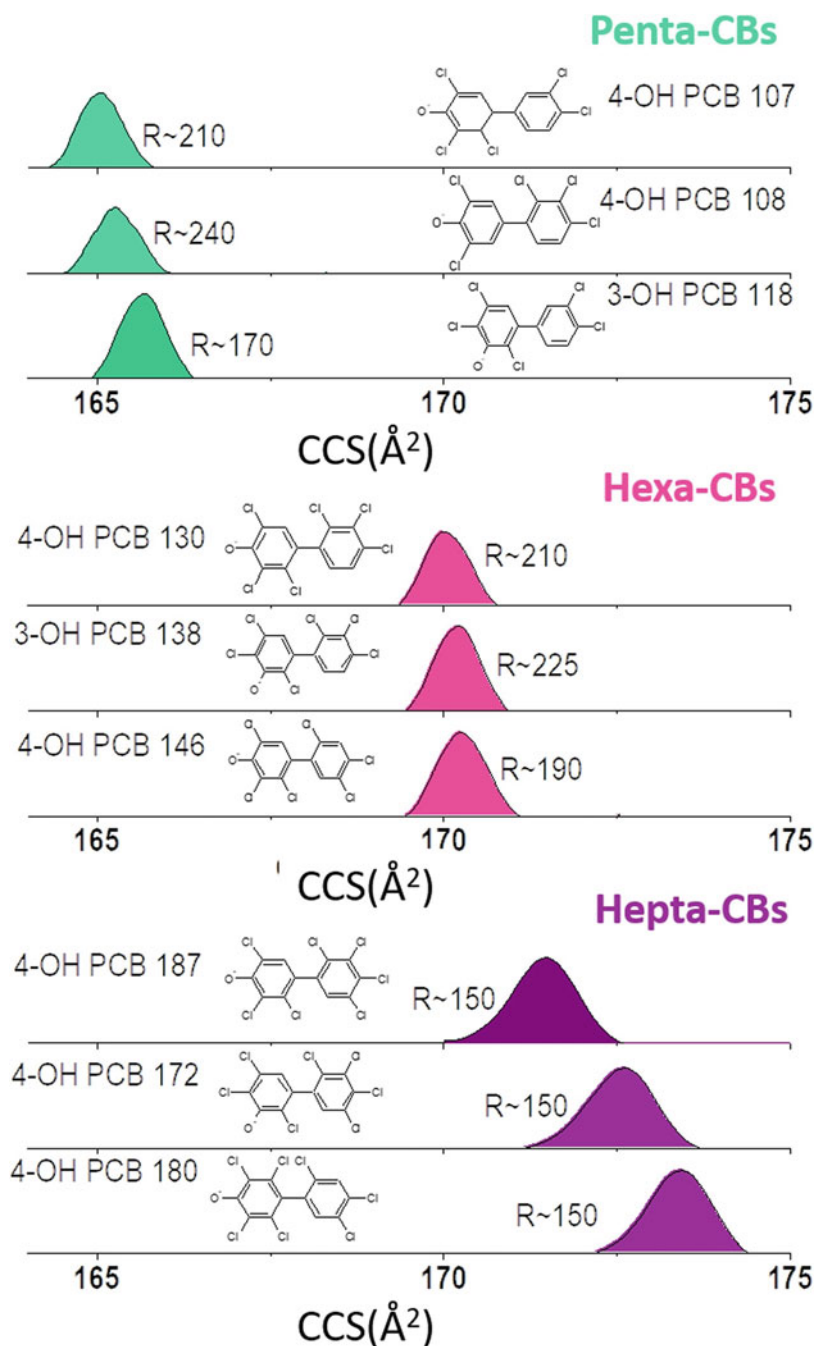


substances in wastewater outfall and lake water samples demonstrated good reproducibility in CCS, differing by  $<0.5\%$  from analysis of pure standards. Luo et al. later applied the developed LC-IM-MS workflow to rapid untargeted characterization of seven different aqueous film-forming foams (AFFFs) [38]. Interestingly, they identified more than 20 homologous series which had not been reported previously, further demonstrating improved confidence in the analysis of complex mixtures containing unknowns. Yukioka et al. applied a novel IM-MS/MS fragmentation flagging approach for identification of several PFAS isomers in a household fire extinguisher liquid [39]. They later applied data-dependent acquisition (DDA) to exclude unrelated chromatographically coeluting species, allowing for improved identification of more than 40 PFAS species in firefighting foam impacted ground-water samples [40].

### **3.2 Polycyclic Aromatic Hydrocarbons (PAHs); Benzene, Toluene, and Xylene (BTX); and Other Volatile Organic Compounds (VOCs)**

Polychlorinated biphenyls (PCBs) are endocrine disruptors still found in the environment, despite their production having been banned in the USA in the 1970s. Adams et al. demonstrated LC-TIMS-TOF for differentiation of the hydroxylated metabolites of isomeric penta-, hexa-, and hepta-CBs (Fig. 9) in diluted human blood plasma [41]. This study showed impressive IM resolving power ( $R_p > 150$ ) with limits of detection ( $\sim 10$  pg/mL) comparable to other traditional methods. Zheng et al. further characterized the hydroxylated metabolites of PCBs, as well as other PAHs and polybrominated diphenyl ethers (PBDEs) using several ionization methods including electrospray ionization (ESI), atmospheric pressure chemical ionization (APCI), and atmospheric pressure photoionization (APPI) [42]. This allowed rapid distinction of isomers and showed characteristic CCS vs.  $m/z$  trends for different molecule types. Zheng et al. also used DTIMS to populate a library of over 500 different chemical species, including xenobiotics [43]. Most recently, Olanrewaju et al. demonstrated a novel analytical workflow combining IM with GC and both atmospheric pressure laser ionization (APLI) and electron ionization (EI) [44]. This approach characterized over 100 PAHs in terms of their retention time, mobility, and accurate mass. Determination of gas-phase candidate structures for each was enabled by complementary computational modeling.

Benzene, toluene, and xylene (BTX) are hazardous volatile aromatic hydrocarbons involved in many chemical manufacturing processes. Environmental monitoring of these compounds has been of particular interest in recent decades due to their toxicity, and several recent advances in IMS-MS instrumentation development have focused on BTX detection in air samples. Sabo et al. coupled corona discharge ionization with a homebuilt 8.2 cm IMS attached to an oa-TOF to investigate n-propanol and isopropanol mixtures [45]. This method produced two-dimensional

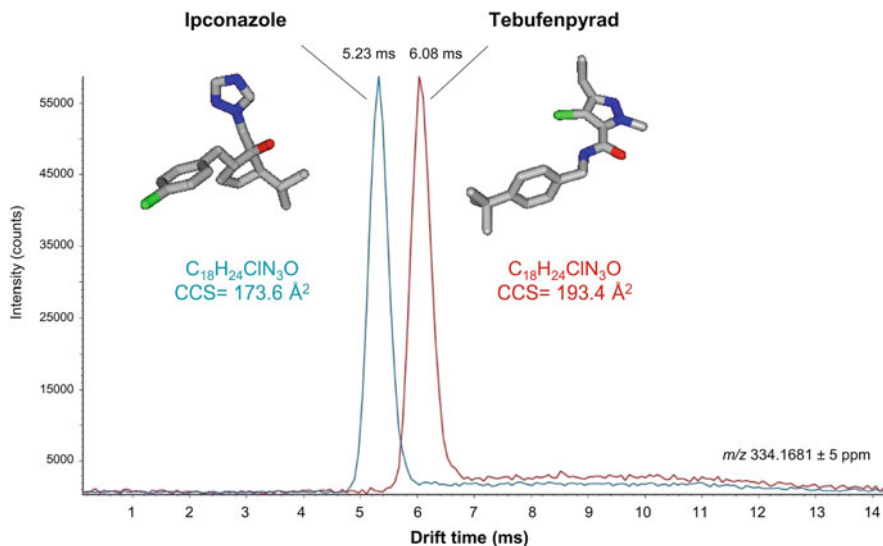


**Fig. 9** Typical mobility profiles of single standards of penta-, hexa-, and hepta-CBs. Reprinted from "Discovery and targeted monitoring of polychlorinated biphenyl metabolites in blood plasma using LC-TIMS-TOF MS," *International Journal of Mass Spectrometry*, 2018, Volume 427, Pages 133–140 with permission from Elsevier [41]

maps of IMS drift time vs. TOFMS flight time, allowing for distinction of benzene, toluene, and xylene, as well as their  $\text{NO}^+$  adduct ions measured in zero air. They further measured mobility trends for the BTX compounds and small molecule alcohols (methanol, ethanol, etc.) as various ion adducts included hydrated species. Heptner et al. incorporated a  $^3\text{H}$  source with atmospheric pressure drift tube to a commercial TOF for measurement of unknown compounds in a gas mixture [46]. The reactive ion products of acetone and BTW compounds could be detected, along with several other small organic molecules such as dimethylmethylphosphonate (DMMP), ammonia, and tetrachloroethylene. Lastly, Allers et al. developed a high-resolution ambient pressure drift tube capable of achieving  $R_p = 100$  and used it for analysis of BTX at low parts per million (ppm) concentrations [47]. This proof-of-concept instrument demonstrated utility in detection of these challenging compounds in air samples. Numerous other volatile organic compounds (VOCs) have also been the focus of IMS-MS studies. For example, Michalczuk et al. analyzed methanol, ethanol, isopropanol, acetone, diethylamino, and ethyl acetate monomers and dimers at different hydration levels using atmospheric pressure chemical ionization (APCI) coupled to a homebuilt IMS-TOF.

### 3.3 Pesticides

Regueiro et al. used UHPLC-TWIMS-QTOFMS to create a CCS database of more than 200 pesticides, with very high intra- and inter-day repeatability ( $\text{RSD} < 1\%$ ) [48]. They applied CCS as an additional measure and demonstrated the ability to resolve isomers ipconazole and tebufenpyrad (Fig. 10). This database was later applied for qualitative pesticide screening of several commercial salmon feed samples, revealing that pirimiphos-methyl and ethoxyquin were frequently found in these feeds [49]. Bauer et al. similarly evaluated an IM-QTOF pesticide screening approach using an in-house database of 280 pesticides applied to 20 different plant-derived foods [50]. With a tolerable CCS error of  $\pm 2\%$ , false assignment of analyte molecules was reduced. Notably, this approach also revealed brodifacoum protomers and spinetoram sodium adducts with unique CCS. The same group then applied this methodology to pesticide metabolites in the *Brassica* plant, identifying phase I and phase II metabolites of eight thiacloprid, 11 azoxystrobin, and three difenoconazole in different plant organs such as leaves, stems, and roots [51]. Gosciny et al. similarly created a pesticide database for 200 compounds, but reported very consistent CCS measurements that differed by no more than 1% from the reference data library [52]. Chen et al. introduced a more rapid method, not involving chromatographic separations to screen for pesticides in various plant samples with total analytical time of only 1.6 min per sample [53]. Limits of detection were reported as  $\leq 20$  ng/mL for all species, and pyrimethanil was confirmed in celery.



**Fig. 10** Extracted ion mobiligrams obtained for the  $[M + H]^+$  ions of the structural isomers ipconazole and tebufenpyrad. Conditions: trap bias, 40 V; IMS wave velocity, 250 m/s; IMS wave height, 45 V. Reprinted from “Ion-Mobility-Derived Collision Cross Section as an Additional Identification Point for Multiresidue Screening of Pesticides in Fish Feed,” *Analytical Chemistry*, 2016, Volume 88, Pages 11169–11177 with permission from American Chemical Society [48]

### 3.4 Pharmaceuticals and Personal Care Products (PPCPs)

Klampfl and coworkers have demonstrated multiple recent instances of IMS-MS for detection of PPCPs. Mlynek et al. used HPLC-IMS-MS to investigate the presence of various xenobiotics and their metabolites in different plant species [54]. Specifically, they studied diclofenac (DCF), a representative non-steroidal anti-inflammatory drug (NSAID), and positively identified 23 previously undetected DCF metabolites using a combination of retention time, drift time, fragment spectra, and accurate mass. Mlynek et al. later investigated phase II metabolites (e.g., glucose, glucuronic acid, and ammonia adducts) of several other NSAIDs (ketoprofen, mefenamic acid, and naproxen) by combining their experimental methods with an in silico database that provided unique identifiers for 41 new metabolites [55]. Similar approaches were undertaken to study uptake of sunscreen ingredients by duckweed [56] and the antihypertensive drug telmisartan [57] and statins and statins by cress [58].

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# Mass Spectrometry Imaging in Food and Environmental Chemistry



Katherine A. Maloof and Kevin R. Tucker

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**Abstract** Mass spectrometry imaging (MSI) is a label-free, nonspecific tool that is effective in characterizing analytes from the surface of samples in situ by creating a two-dimensional image with high spatial resolution. Gaining molecular spatial information produces further insight into cellular processes, allowing for limitless possibilities when it comes to investigating food and environmental substances. MSI can be used exclusively to reveal the localization of material or in conjunction with other techniques to corroborate findings. In the field of food and environmental chemistry, MSI is slowly becoming a more popular technique owing to its countless applications and ease of use. MSI is governed by the ionization technique employed, and each ionization type has its own set of characteristics that control what classes of analytes can be studied. In this chapter, the applications of MSI within food and environmental chemical research are presented focusing on each of the ionization

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techniques and their specific characteristics that make them appropriate for the individual examples highlighted.

**Keywords** Environmental analysis techniques, Environmental sampling, Mass spectrometry imaging, Matrix-assisted laser desorption/ionization, Multidimensional data, Sample preparation

## 1 Introduction

Over the last decade, there have been substantial increases in the amount of research dedicated to food and environmental science. This is due in part to the consumer's growing concern for safer foods and improving the environment. There has been a push for more research into eliminating harmful emerging contaminants from the environment [1], ridding marine environments of evasive microplastics [2], investigating the effects of pesticides on human health [3], and finding non-toxic substitutes for agrochemicals [4]. Safer food is also a concern with research efforts dedicated to detecting pesticides on the surface of food products and preventing adulterated food from entering the market [5]. Mass spectrometry imaging (MSI) has also been used to investigate endogenous molecules found in plant organ tissues to give a detailed look into how changes in the quantities of endogenous molecules affect the growth and development process of plants [6]. In this chapter, both dominant and novel ionization types used for MSI in food and environmental science and the advancements in each with the most recent, notable examples in the field will be explored.

MSI is preferred for specific applications where spatial information is relevant over the quantitative information provided by traditional analysis methods like gas chromatography mass spectrometry (GC-MS) and liquid chromatography mass spectrometry (LS-MS). As a sample is probed by laser beam, ion beam, or stream of charged droplets, a mass spectrum is collected at each individual pixel, ion intensity is plotted as a function of space for every  $m/z$  value, and a two-dimensional image can be generated from the individual mass spectra created for a specific  $m/z$  range. Each picture is generated based on the intensity of signal over a specific  $m/z$  window that the user defines. This picture shows a spatial analyte map where the abundances of analytes are depicted based on a defined color scale.

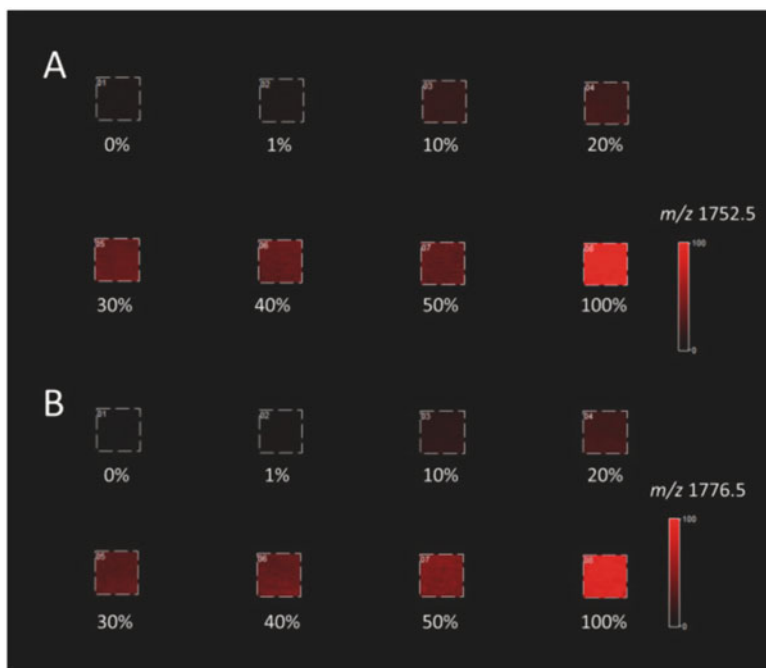
Choosing the best ionization method is critical for the successful analysis of target molecules. There are ongoing research efforts to achieve truly comprehensive analysis, but currently there is no ionization technique that allows for the detection of all compound types simultaneously [7]. In food and environmental science, there are thousands of possible compounds to analyze, each class with its own unique properties. One may want to analyze small molecule plant metabolites or larger molecules like proteins in food stuffs. Secondary ion mass spectrometry (SIMS) is better suited for atomics and very small molecules and matrix-assisted laser desorption/ionization (MALDI) was developed for handling larger molecular species,

while desorption electrospray ionization (DESI) works between those ranges. When ionizing molecules, hard and soft ionization techniques both exist within imaging techniques. Hard ionization techniques, like SIMS, transfer a large amount of energy to the target compound causing a high degree of fragmentation of precursor ions. This is evident within the spectrum as many peaks – most of them fragments – appear, making it difficult to detect the molecular ion, sometimes even in a pure sample. Since the first use of MALDI MSI in 1998 [8], soft ionization MSI has evolved into a highly selective and robust technique for qualitative and quantitative analysis of two- and three-dimensional media. Soft ionization does not impart as much energy, allowing the precursor ion to remain intact and providing information about the intact molecule [9, 10].

Throughput is a consideration for some fields of research. SIMS is slower than MALDI because achieving high resolution takes time. There are two modes to choose from in MSI – microscope and microprobe mode. SIMS microscope mode has been widely commercialized due to its ability to quickly generate whole images in sequence rather than the traditional microprobe mode which creates images via splicing together each individual pixel [11]. It is easier to move an ion beam rapidly, whereas the laser beam is harder to manipulate quickly. However, the scientists at Bruker have developed an advanced optics system that moves a laser beam within the lens without repositioning the stage. Bruker's rapiflex<sup>®</sup> is the highest throughput MALDI-time-of-flight (TOF) MS on the market and promises reliable data within minutes. This drastically improved the time factor, cutting down the time of analysis nearly 90%. This leads to a higher throughput and analysis that would normally take 3 h now can be accomplished in just 20 min. Cao et al. used the rapiflex to investigate the nutritional and physical state of used cooking oil and fresh edible oil to authenticate its quality. Vegetable oil is an edible oil with significant nutritional value that is commonly used commercially for frying foods. Repeated use of edible oil for frying food reduces the oil's quality and its prolonged use can be harmful to public health [12]. The researchers determined a method for checking the quality of fresh cooking oil to confirm that it is pure and not contaminated with used oil. Figure 1 shows samples of olive oil that have been contaminated with various quantities of used deep-fried oil and gutter, or recycled, oil. This figure illustrates vegetable oil that was adulterated with a ratio of 10% or higher of deep-fried oil or gutter oil. This method can be used in the future to help uphold the quality of vegetable oil used for commercial cooking purposes [13].

Dynamic range, in this context, is the range in concentration of compounds an instrument can analyze simultaneously. A single MSI analysis enables the detection of many different compounds simultaneously but can have difficulty detecting analytes that are low in abundance while much more abundant species are present leading to the need for specialized sample preparation to remove abundant classes of molecules (e.g., lipids). The development of TOF-MS mass analyzer drastically improves the limit of detection of the ionization technique and is used in MSI with MALDI and SIMS [14].

There have also been strides made in food science to develop new ionization types that do not include organic matrices and instead employ nanoparticles like



**Fig. 1** Samples analyzed by MALDI-MSI show olive oil contaminated with varying volumes of (a) deep-fried oil and (b) gutter oil [13]

colloidal graphite (GALDI), silicon (DIOS)-MSI, and nanoparticle-enhanced target particles including gold (AuNPET) and silver (AgNPET). This technique called nanoparticle-based surface assisted laser desorption/ionization mass spectrometry imaging (NPs-ALDI-MSI) improves the selectivity and limit of detection but has difficulty ionizing all nanoparticles reasonably due to the complicated ternary interactions of the laser spot with the analytes and nanoparticles. Furthermore, these interactions are inconvenient to study under high vacuum conditions. Despite these complications, the future progress of new nanoparticle applications in MSI will result in advancements in food science [15].

## 2 Ionization Methods

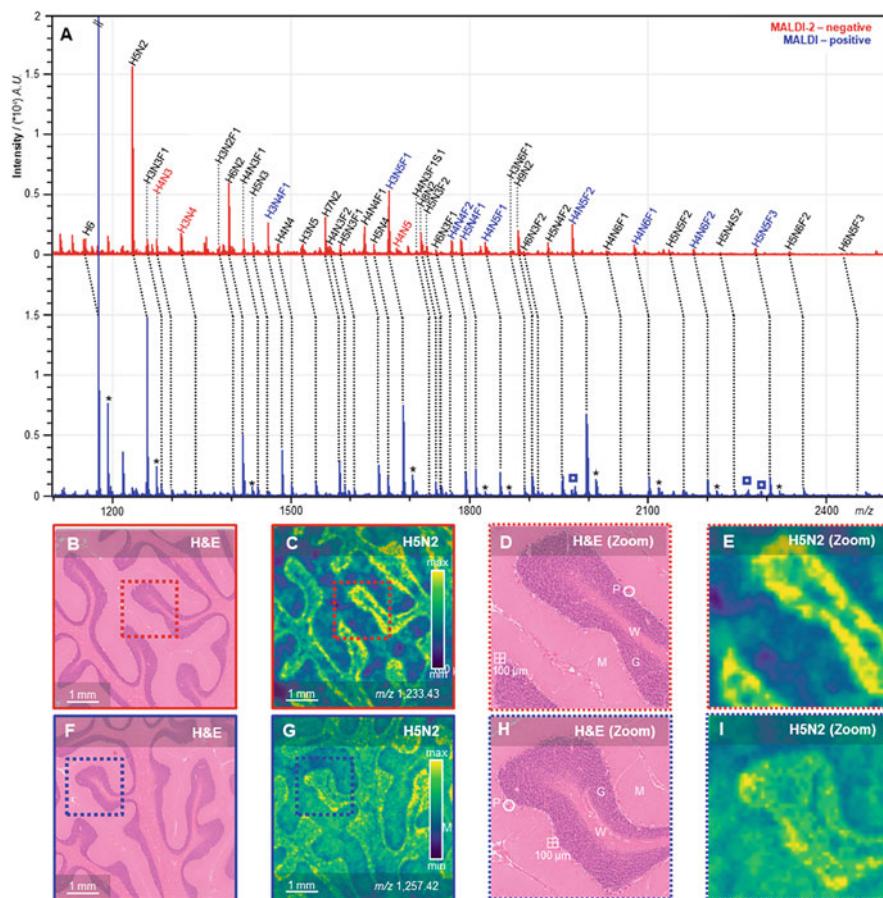
### 2.1 Secondary Ion Mass Spectrometry (SIMS)

The first of the ionization types developed for MSI was secondary ion mass spectrometry (SIMS) in 1962 by Castaing and Slodzian [16]. SIMS is a hard ionization technique involving a high-energy primary ion beam that bombards the sample surface, smashing through the top surface layers and transferring energy

from those primary ions to the surface analytes in a process called secondary ionization. Static SIMS generates images with a spatial resolution between 1 and 5  $\mu\text{m}$ , while TOF-SIMS achieves submicron spatial resolutions with high mass resolution ( $m/\Delta m \sim 10,000$ ) [17], which makes it the preferred ionization technique for small molecule and single cell analysis. The drawback of high-energy ion bombardment is that it causes fragmentation in the larger molecules and it damages the sample surface. The fragmentation limits the mass range of SIMS to less than 1,000  $m/z$  [18]. However, because of the development of cluster ion beams, SIMS MSI has improved mass range up to 10,000 Da while maintaining its spatial resolution and improving the molecular ion yield [19, 20]. Ultimately, SIMS is widely utilized based on its unique capability for submicrometer spatial resolution.

## 2.2 MALDI

Developed in the late 1990s for imaging of larger molecules, MALDI is a soft ionization technique that involves the use of a sample coated in matrix to absorb energy from laser ablation converting analytes to the gas phase to be ionized [21]. When combined with a TOF analyzer, MALDI MSI has a mass range beyond 1,000,000 Da or 1 MDa [22]. In Boughton's review of mass spectrometry imaging for plant biology, MALDI was found to be the most common ionization type used in plant-based MSI research [23], which is likely due to its broad analyte coverage, high spatial resolution, and ease of use of commercial MALDI instruments. However, because MALDI is operated in microprobe mode, performing each scan is time-consuming. The spatial resolution of MALDI MSI is determined by the chosen matrix, matrix application technique, and the laser spot size [24]. Efforts have been made to improve the spatial resolution by reducing the laser spot size, using methods including a pinhole filter to achieve a 5  $\mu\text{m}$  diameter [25] and implementing transition-mode geometry where the laser is projected through a target behind the sample rather than focused directly onto the matrix crystals which achieved a spot size of 1  $\mu\text{m}$  [26]. In combination with transition-mode geometry, MALDI-2 – a post-ionization technique that uses a second laser pulse to generate ions from the original ion plume – can offer improved sensitivity up to three orders of magnitude, ultra-high spatial resolution, and improved ion yields [26, 27]. Heijs et al. used MALDI-2 to image N-glycans to study their role in cancer and auto-immune disease [28]. Parameters for negative-ion mode MALDI-2 MSI were optimized to generate high quality images of glycans in brain tissue as seen in Fig. 2. This same application of MSI could be extended to study foods high in N-glycan, which play an important role in promoting gut health in humans [29, 30]. MALDI-2 MSI was also used by Bien et al. to map sterols in marine flatworms to establish their role in a possible symbiotic relationship with algae [31]. This technique could be applied in the future to monitor environmental contamination by quickly detecting the presence of sterols in surface sediment [32].



**Fig. 2** (a) Mass spectra of MALDI-2-MSI operated in negative-ion mode (red) and MALDI-MSI in positive-ion mode (blue) to generate (b–i) images of *N*-glycan species in human brain tissue. The pink images are negative-ion mode MALDI-2-MSI and the blue-green images are positive-ion mode MALDI-MSI [28]

### 2.3 Desorption Electrospray Ionization (DESI)

Developed in 2004, DESI uses an electrospray ionization (ESI) source to spray the sample surface with charged microdroplets of solvent to desorb and ionize analyte ions simultaneously [33]. This soft ionization technique is performed at ambient conditions without the need for laser or ion beam which allows for in-situ analysis of molecules up to 2,000 Da without any sample preparation requirements [34]. The most distinctive parameter for DESI is in regard to the angle and focus of the solvent sprayer [35]. DESI-MSI analyzes molecules on the very surface of samples and the electrospray jet is only capable of achieving spatial resolution down to 200 μm with DESI or 30 μm using nano-DESI [36]. However, the ease of ambient analysis makes

DESI a popular technique for imaging [33]. Gallage et al. used DESI-MSI to look at the localization of the flavor compound vanillin – and its glucose-conjugated form vanillin glucoside – in *Vanilla planifolia* pods. Vanilla is a globally popular flavor, and these researchers set out to better understand the biosynthesis and accumulation of its flavor compound, vanillin. DESI-MSI was used to confirm the results of  $^{14}\text{C}$ -radiolabeling used to find locations of vanillin and vanillin glucoside biosynthesis activity. The DESI images showed these compounds to be present in the same tissues where the  $^{14}\text{C}$ -radiolabeled precursors suggested vanillin biosynthesis takes place [37].

## 2.4 Other Ambient Ionization Types

While ionization methods involving a vacuum chamber attain better resolutions than ambient ionization methods, there are many advantages with ambient ionization. Since the development of ambient mass spectrometry in the form of DESI, other ambient techniques have gained popularity in MSI [38]. Being able to analyze samples in their native environment reduces the risk of sample changes due to introduction into the vacuum system, preserves volatile compounds, and does not involve time-consuming sample preparation [38, 39]. The following ambient techniques have been applied to MSI and are related to food and environmental science.

### 2.4.1 Laser Ablation Electrospray Ionization (LAESI)

LAESI combines the use of a mid-IR laser beam with a nanoelectrospray source [40]. This ambient ionization technique is matrix free, high throughput and requires no sample preparation [41]. It is used in samples with a high-water content because the IR laser corresponds to the OH vibrations of the water molecule [41]. This makes LAESI perfect for the analysis of living tissue. LAESI can also be used on samples with rough surfaces without sample treatment [42]. Kooijman et al. created a probe that combines ESI with atmospheric pressure chemical ionization (APCI) and is compatible with commercial LAESI instruments. They improved the pixel-to-pixel repeatability for both ionization methods and successfully imaged the fungicide Isotianil and its metabolites from tomato plant leaves [43]. Spatial resolution in LAESI can be achieved down to 30  $\mu\text{m}$  as demonstrated by Shrestha et al. for imaging metabolites in onions associated with their purple pigmentation [44].

### 2.4.2 AP-MALDI

Atmospheric pressure (AP) MALDI is traditional MALDI that still requires sample tissue to be sliced and coated with a matrix but has the advantage of handling samples at atmospheric pressure. While analysis at atmospheric pressure has its

advantages, ion loss is experienced with this technique as ions travel from the ablation plume to the mass analyzer [45]. Li et al. were the first to use AP-MALDI MSI to gain spatial information about the distribution of endogenous metabolites in the endosperm of the coffee plant ripe fruit. The researchers studied eight varieties of coffee beans from regions in Africa, Central and South America, and Asia to investigate the differences in endogenous metabolites of coffee beans from wet and dry processing methods. Images of caffeine, sucrose, and caffeoylquinic distribution were visualized in both positive-ion and negative-ion mode with a spatial resolution of 75  $\mu\text{m}$  as seen in Fig. 3. Colombia and Tanzania were found to have the highest levels of caffeine ( $m/z$  195.09,  $[\text{M} + \text{H}]^+$ ) and Colombia, Tanzania, Nicaragua, and China had the highest levels of sucrose ( $m/z$  381.08,  $[\text{M} + \text{K}]^+$ ). The distribution of caffeoylquinic acid ( $m/z$  393.06,  $[\text{M} + \text{K}]^+$ ) can also be seen [39].

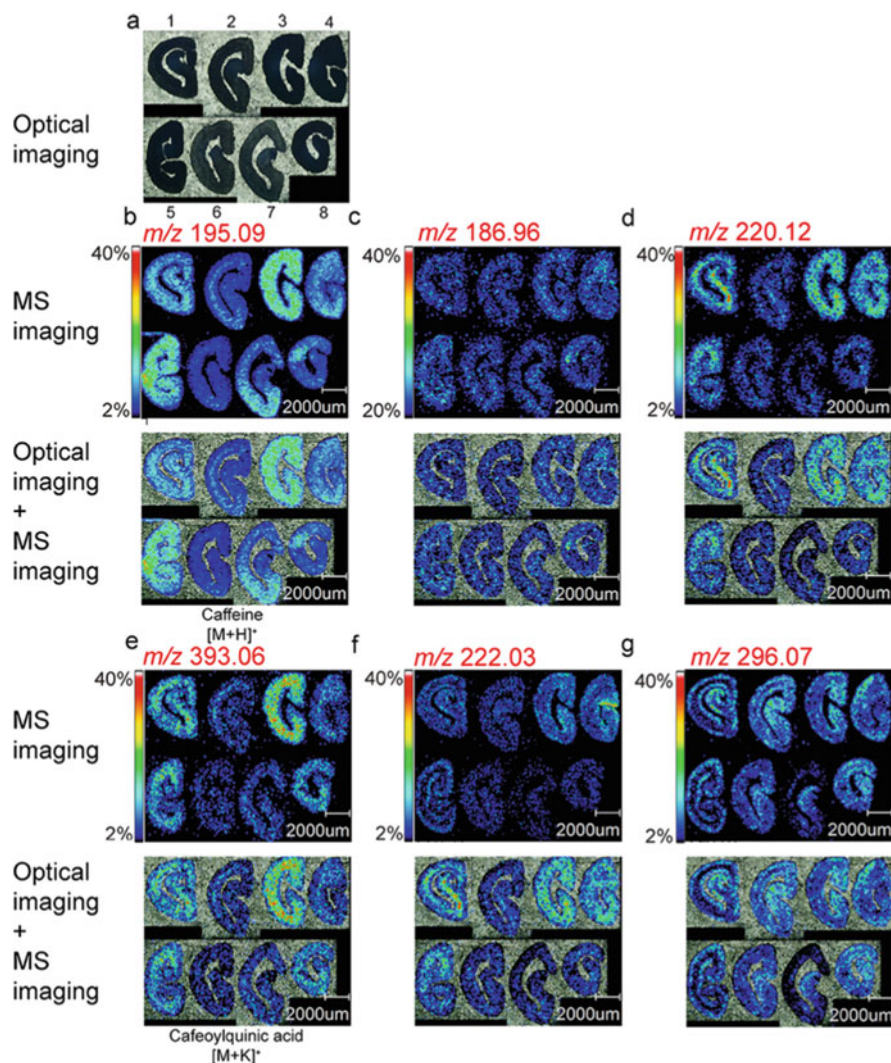
### 2.4.3 Liquid Extraction Surface Analysis (LESA)

In 2011, Eikel et al. developed a fully automated LESA that uses a nanoelectrospray source applied via pipette tip to ionize surface compounds from fruits and vegetables to be analyzed by tandem mass spectrometry (MS/MS) [46]. Ten years later, in a study conducted by Xiao et al. LESA-MSI was used to study the distribution of perfluoroalkyl substances (PFASs) in zebrafish [47]. PFASs are highly stable synthetic chemicals that cause harm to the environment due to their ability to resist degradation. PFASs are known to be detrimental to humans, causing children to have adverse immune responses [48]. Determining the spatial distribution of PFASs throughout fish tissue would help to better understanding the mechanism of action of toxic compounds in fish. Figure 4 shows the distributions of perfluorooctanesulfonic acid (PFOS) and perfluorooctanoic acid (PFOA) in different regions of the zebrafish with the highest accumulation in the organs, roe, pelvic fin, and gills. While the spatial resolution is well below average due to the limitation of the pipette tip diameter (600  $\mu\text{m}$ ), it clearly shows where most of the PFASs accumulated in the body of the zebrafish [47]. This technique is to be used for imaging analysis that does not require high resolution, making it a good use for discovery purposes like guaranteeing that food surfaces do not contain any pesticides or harmful chemicals [48, 49].

### 2.4.4 Thermal Desorption Electrospray Ionization Mass Spectrometry (TD-ESI-MS/MS)

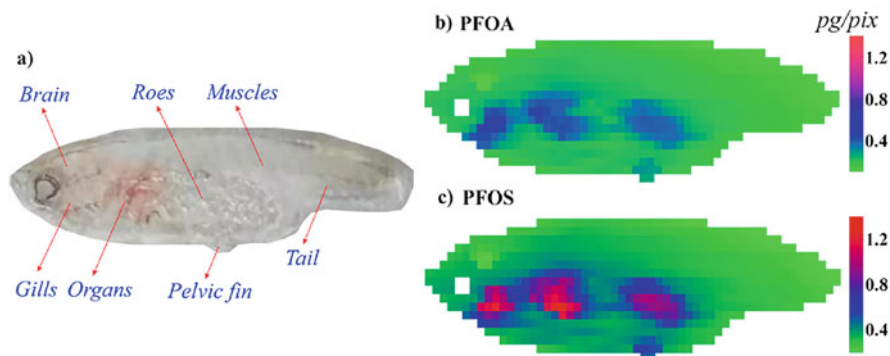
TD-ESI/MS is an ionization technique that uses an oven to desorb analytes from a metal sample probe. The analytes after desorbing are carried via nitrogen gas to the ESI plume. This technique is easy to use and allows for rapid analysis with no sample preparation. Jeng et al. imaged pesticide residues on the surface of strawberries by TD-ESI-MS/MS. The sampling probe – a 2 mm stainless steel inoculating loop attached to a handle – is swept across the surface of the strawberry for 1 cm





**Fig. 3** Coffee bean sections analyzed via MALDI MSI show various distributions of caffeine, sucrose, and caffeoylquinic acid. (a) Eight coffee beans were imaged from 1, Ethiopia; 2, Brazil; 3, Colombia; 4, Nicaragua; 5, Tanzania; 6, Guatemala; 7, Uganda; 8, China. Distributions of (b) caffeine ( $m/z$  195.09,  $[M+H]^+$ ), (c) sucrose ( $m/z$  381.08,  $[M+K]^+$ ), and (e-g) caffeoylquinic acid ( $m/z$  393.06,  $[M+K]^+$ ) can be seen throughout the endosperm and outer edge of the beans [39]

once in each of the 40 predetermined sample locations. The probes were then placed into the TD-ESI unit where they are heated and sprayed with the electrospray source for desorption of analytes. The spatial resolution is not comparable to the other techniques currently being used in MSI, however, this technique could be used for



**Fig. 4** (a) Optical image of a whole-body frozen section of zebrafish. Distributions of (b) perfluorooctanesulfonic acid (PFOS) and (c) perfluorooctanoic acid (PFOA) in different regions of the zebrafish with the highest accumulation seen in the gills, organs, roes, and pelvic fin with a spatial resolution of  $600\ \mu\text{m}$  [47]

many food quality and safety applications to confirm or deny the presence of adulterants [50].

### 3 Sample Preparation

Sample preparation is the act of treating a certain sample to prepare it for analysis. Although some methods require little to no sample preparation in MSI, it is one of the most important steps to guaranteeing accurate results. It is vital that samples are handled with extreme care and processed immediately after collection to ensure that the integrity of the sample is upheld, and no degradation occurs [6]. The goal of sample preparation is to retain all spatial details of the sample without degrading or changing its chemical structure so that the information gathered is a true representation of the analytes found on the surface. If samples are not prepared correctly, it will have negative impacts on analyte detection and can produce inaccurate results [51]. Any inconsistencies in sample preparation can skew analyte abundances making the reproducibility of sample handling techniques crucial components for successful MSI experiments.

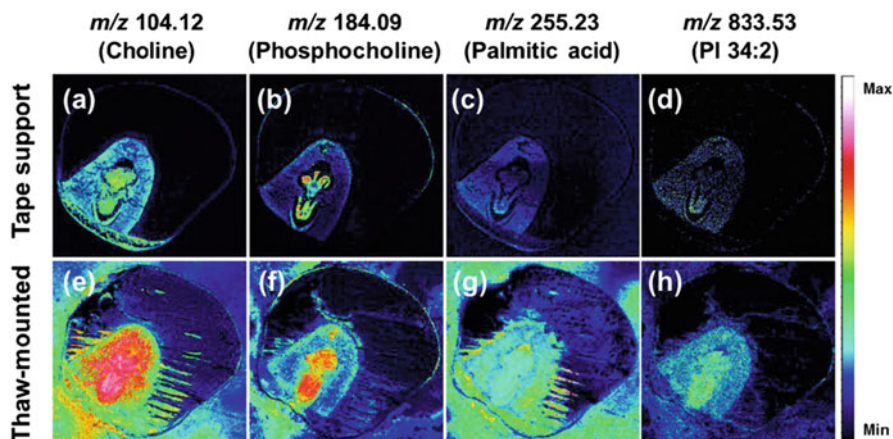
Preparing samples for vacuum techniques in MSI involve freezing the sample, thinly sectioning it, thaw-mounted to a plate or slide, then drying for insertion into the vacuum [52]. When food and environmental science involves working with plant samples, preparation of those samples can be challenging. Plant samples are inherently more difficult to work with than mammalian samples due to the higher water content and lower lipid content of the tissue. While the walls of plant cells are more rigid, the complex cellular structure is inherently complicated and can be easily broken. This challenge has inspired various advances in the sample preparation techniques for plant tissue [53].

Histological sectioning is a popular method used for the preparation of root and stem tissue that can also be applied to fruit and vegetable tissue. Pereira et al. used this technique to detect fungicide 6 mm below the surface of apple tissue [54]. However, when dealing with tissues that are difficult to slice due to fragility or heavy water content, histological sectioning may not be ideal. Imprinting is the sample preparation technique used for more fragile sample types. Gold nanoparticle (AuNP)-immersed paper imprinting MSI can be used on delicate tissue samples to visualize pesticide metabolites on the surface and internally [5, 55].

One limitation in laser based MSI experiments is that samples need to be relatively flat to allow the laser to properly focus and to ensure mass accuracy across the image. Topography of sample surfaces is nonexistent for tissues that have been sectioned via a microtome or cryostat, however, for the analysis of samples with variation in the surface topography like whole leaves or fruits and vegetables, it is harder to achieve acceptable resolution [42]. If the surface topography is not within the range of lateral resolution, with an aspect ratio near 1, false imaging results and signal variation can occur [56]. To combat this, MALDI MSI implements an autofocusing mode to investigate 3D sample surfaces. Using a triangulation laser system integrated with atmospheric pressure scanning microprobe matrix-assisted laser desorption/ionization (AP-SMALDI) MSI, height variations of up to 1.5  $\mu\text{m}$  can be resolved. While this method gains resolution, the measurement time increases by 25% [57].

To help prevent tissue breakage or analyte delocalization, a pre-embedding medium that preserves the sample can be used. Using gelatin as embedding medium prior to mounting cryosectioned leaf tissue results in less diffusion of analytes and better preservation of the sample structure compared to ice and carboxymethyl cellulose [58]. Imprinting and fractioning are two complementary techniques that can also be used when preparing thin plant tissues as first demonstrated by Klein et al. to look at internal metabolites of soybean leaves via MALDI-MSI [59].

With any type of tissue in MSI, one challenge that must be overcome is analyte delocalization on the sample surface. When analytes diffuse on the surface of sample tissue, the image created will not be a true representation and causes results to be skewed. A large portion of food and environmental science studies plant material. The Kawamoto method – originally used to prepare cryosections of hard, fragile animal bone tissue to study the movement of calcium [60] – uses an adhesive film or tape to help preserve sample structure during the adhering process and is an alternative to thaw mounting. The thaw-mounted method uses the heat from the fingertip to adhere the sample to the plate for analysis which causes delocalization of analytes. Kim et al. used adhesive tape on sections of corn seed tissue and compared the results to the thaw-mounted method [61]. They demonstrated that using adhesive tape does not cause surface analyte delocalization, but the thaw-mounted method does which can be seen in Fig. 5. Enomoto successfully used this adhesive tape to keep sections of delicate strawberry tissue intact to study metabolite functions within [55].

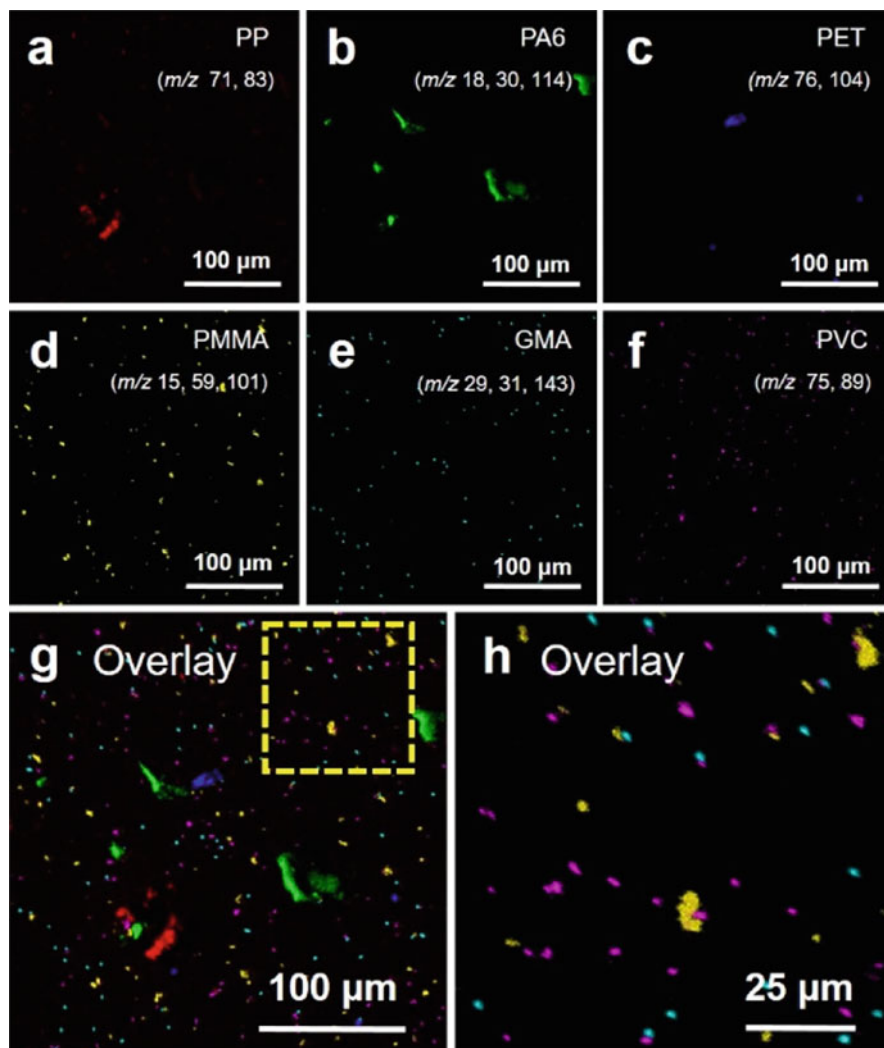


**Fig. 5** Images of corn seed tissue by TOF-SIMS show excessive analyte diffusion in the thaw-mounted samples (e-h) as compared to the tape support method (a-d) for choline (a, e), phosphocholine (b, f), palmitic acid (c, g), and phosphatidylinositol (d, h). [61]

### 3.1 Secondary Ion Mass Spectrometry (SIMS)

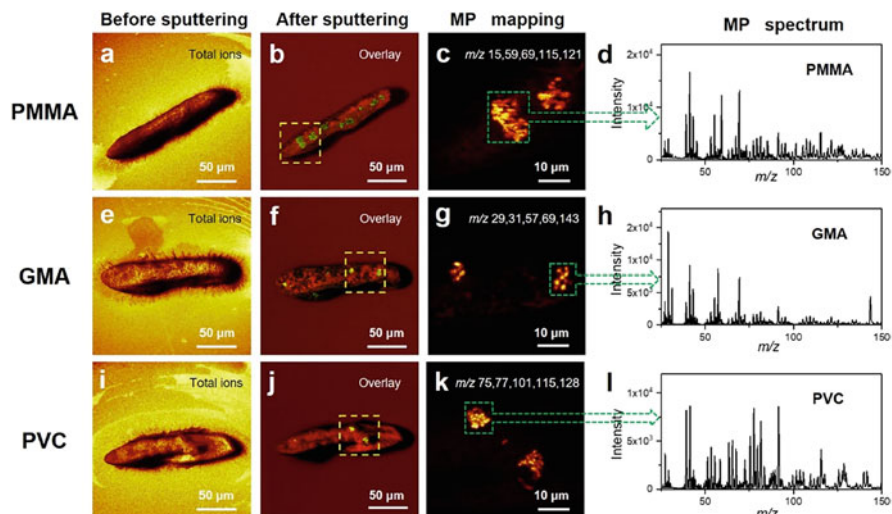
Because SIMS ionization is performed in a vacuum, a big obstacle to overcome with SIMS MSI is the requirement that samples must be dried prior to placement in a high vacuum chamber. This can cause the sample to be damaged or altered in the process. The ultra-high vacuum pressure can also be problematic for samples that have a high-water content like plants, fruits, and vegetables. To date, there have been studies that use MeV-SIMS, a SIMS method that uses heavy, high-energy ions (e.g.,  $^{35}\text{Cl}^{6+}$  and  $^{63}\text{Cu}^{4+}$ ) for ionization, in leaves [62, 63], but none that employ the use of Wet-SIMS. Wet-SIMS was developed to use a low vacuum and MeV-energy ion beams which allows for analysis of samples without drying steps. This technique could be useful to detect heavy metals, like lead, in delicate plant or produce tissue.

In 2021, SIMS was employed for the analysis of microplastics (MPs) in paramecia. MPs pose a large threat to the environment due to their harmful effects on living organisms. This study by Feng et al. offers a simple SIMS method for the in-situ analysis and imaging of six different types of MPs – with a size range between 1 and 50  $\mu\text{m}$  – from paramecia that requires no sample preparation. In situ analysis of MPs via mass spectrometry is not typically performed due to the destructive sample pretreatment that is required to separate the MPs from the paramecia. The authors of this study were able to generate a spatial map of six types of MPs – polypropylene (PP), polyamide 6 (PA6), polyethylene terephthalate (PET), polymethyl methacrylate (PMMA), polyglycidyl methacrylate (GMA), and polyvinyl chloride (PVC) – ingested by paramecia with a high degree of chemical specificity, sensitivity, and spatial resolution of 1  $\mu\text{m}$  (Fig. 6). The cellular structures of contaminated paramecia were also visualized to help investigate the interaction of plastics with living



**Fig. 6** Generated spatial map of six types of identified MPs. (a) PP ( $m/z$  71, 83) (b) PA6 ( $m/z$  18, 30, 114) (c) PET ( $m/z$  76, 104) (d) PMMA ( $m/z$  15, 59, 101) (e) GMA ( $m/z$  29, 31, 143) (f) PVC ( $m/z$  75, 89) (g) Merged image of MPs (a–f) (h) Zoomed in area from (g) [65]

organisms (Fig. 7). These findings will aid future studies to examine MPs influence in the environment [64].



**Fig. 7** The imaging of MPs in a paramecium in situ by TOF-SIMS. (a, e, i) Before SIMS sputtering, the total number of ions was seen. (b, f, j) After SIMS sputtering, the image of MPs inside the paramecium (c, g, k). The high-resolution images of MPs from the highlighted areas in (b, f, j). (d, h, i) PMMA, GMA, and PVC mass spectra collected from the paramecium [64]

### 3.2 MALDI

The focus of MALDI MSI sample preparation is the application of a matrix to aid in the ionization of surface analytes. The technique of application is important for creating quality images of high spatial resolution [45]. Homogenous application of the matrix is crucial which is commonly achieved using automated sprayers [65]. Sublimation is another common method that is a low-cost way to uniformly deposit dry matrix onto sample surfaces [66]. Traditional organic matrices cause spectral interferences for analysis of compounds with low molecular weight which makes application into small molecule investigations more challenging [67]. To avoid using a matrix, Enomoto et al. developed a new ionization method for MSI called desorption ionization using through-hole alumina membrane (DIUTHAME). They developed this method for analyzing large whole strawberry sections too large to fit onto glass slides for MALDI analysis [68].

### 3.3 Desorption Electrospray Ionization (DESI)

DESI-MSI is performed at ambient conditions and therefore does not require any sample preparation [69]. This is an advantage to DESI since sample preparation is time-consuming and expensive [22]. All that is required of DESI MSI is sectioning

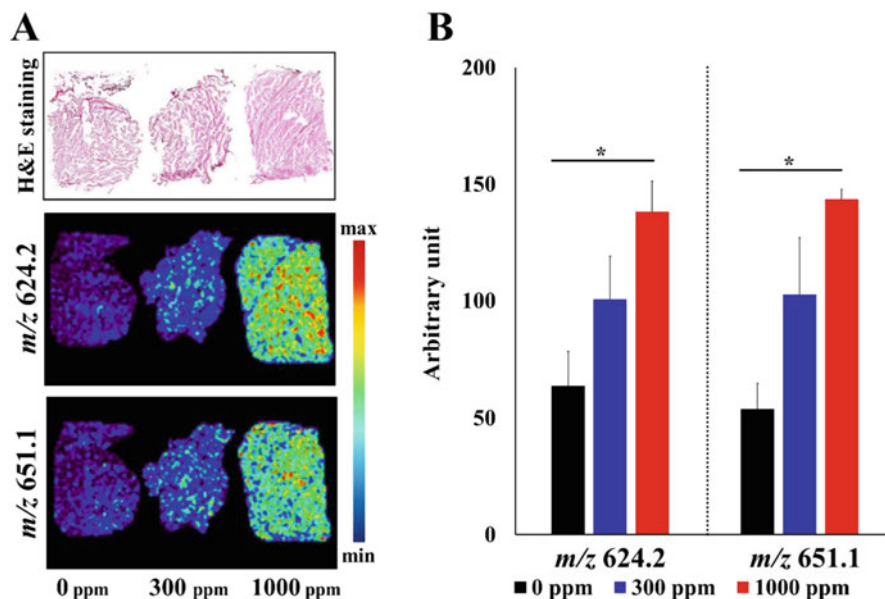
and mounting of sample tissue which is nearly identical to the methods described above.

## 4 Specific Applications of MSI in Food and Environmental Science

Since the beginning of 2020, several related studies assess the quality of food products. Likely due to its popularity in commercial use and accounting for more than 50% of all publications referencing mass spectrometry imaging, MALDI MSI has been employed in these studies. Ambient techniques DESI and AP-SMALDI are also mentioned in this section due to their nondestructive nature even though they account for less than 20% of all similar publications.

Billions of people worldwide consume fish [70]. Since 30% of all fish products are frozen prior to being sold, the quality of frozen fish should be evaluated [71] because its nutritional and physical quality is deteriorated progressively. Morisasa et al. used MALDI MSI to detect formaldehyde (FA) in frozen fish filets to help determine their freshness. Trimethylamine N-oxide (TMAO), a natural compound found in the muscle tissue of sea fish, decomposes to equal parts dimethylamine (DMA) and formaldehyde (FA) in fish post-mortem [72]. FA denatures proteins and causes fish fillets to harden – a process that occurs after their frozen. This study used MSI to determine fish “hardness” by visualizing protein buildup in frozen fish that correlate with increasing concentrations of FA. As seen in Fig. 8, peaks at 624.2  $m/z$  and 651.1  $m/z$  were resolved showing peptides determining FA accumulation [73].

The proper storage of food is important to upholding its quality and to guarantee food safety. Glycoalkaloids (GAs) are toxic secondary metabolites that can form in solanaceous plants from improper storage due to temperature extremes, material damage, or exposure to light [74]. Deng et al. stored potato tubers for different lengths of time and used the iMScope to image the spatially resolved distribution of four GAs – dehydrochaconine,  $\alpha$ -chaconine, dehydrosolanine, and  $\alpha$ -solanine. In 2009, Harada et al. developed a “mass microscope” – which would later be developed and sold by Shimadzu as the iMScope [75] – to analyze samples of freshly cut ginger root. This instrument operates at atmospheric pressure to avoid damage to volatile metabolites or the water-rich structure of ginger rhizome. Coupled with atmospheric pressure quadrupole ion trap (AP-QIT) TOF MSI, the iMScope imaged GAs in the potato tubers with 5  $\mu\text{m}$  resolution and saved the researchers time and money with limited sample preparation. As seen in Fig. 9, potato tubers were imaged at 0, 10, 15, 20, 30, 40, and 60 days showing increased accumulation of four toxic GAs. The images revealed that different parts of the potato tubers have different amount of GAs with the sprout showing the greatest increase over the 60 days. These findings can help procure future guidelines for the proper storage of potato tubers [76].

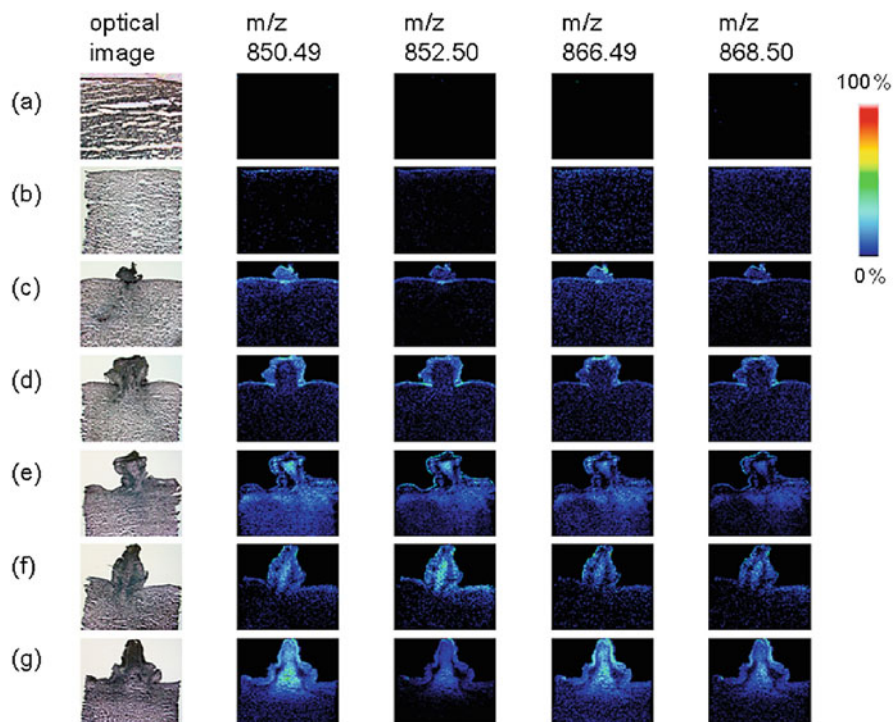


**Fig. 8** (a) Three fish fillets treated with 0, 300, and 1,000 ppm FA were imaged by MALDI MSI and (b) their distinguished molecular ions shown ( $n = 3$ ) [73]

Tea is the second most consumed beverage in the world – only coming in second to water [77]. This popular beverage is not only admired for its taste, but also for its numerous health benefits. Research shows that there are correlations with consumption of tea and higher likelihood for overcoming diseases like cancer [78], obesity [79], and cardiovascular disease [80] and can even lower the risk for type 2 diabetes [77]. There are many kinds of metabolites in tea leaves and studying their localization can offer insight into regulating and improving the quality of tea. Liao et al. used DESI MSI to show the catechin distribution found in tea leaves and roots. Catechins (flavan-3-ols) are specialized metabolites in tea leaves that have been found to have a positive impact on human health, and they also play a role in antistress regulation and pathogen defense in tea plants [81–83]. The nondestructive nature of DESI is what allows for the complete investigation of metabolites found on both sides of the tea leaves. Due to their high wax content, porous polytetrafluoroethylene (PTF) imprints were optimized and used as sample pretreatment. This study successfully imaged the localization of some fatty acids, polyphenols, and catechins within the tea tissue and will allow for future research into determining their function. In Fig. 10, different concentrations of catechins can be seen throughout various structures of the tea leaves showing that distributions of catechins are not uniform throughout the individual features of the tea leaf [84].

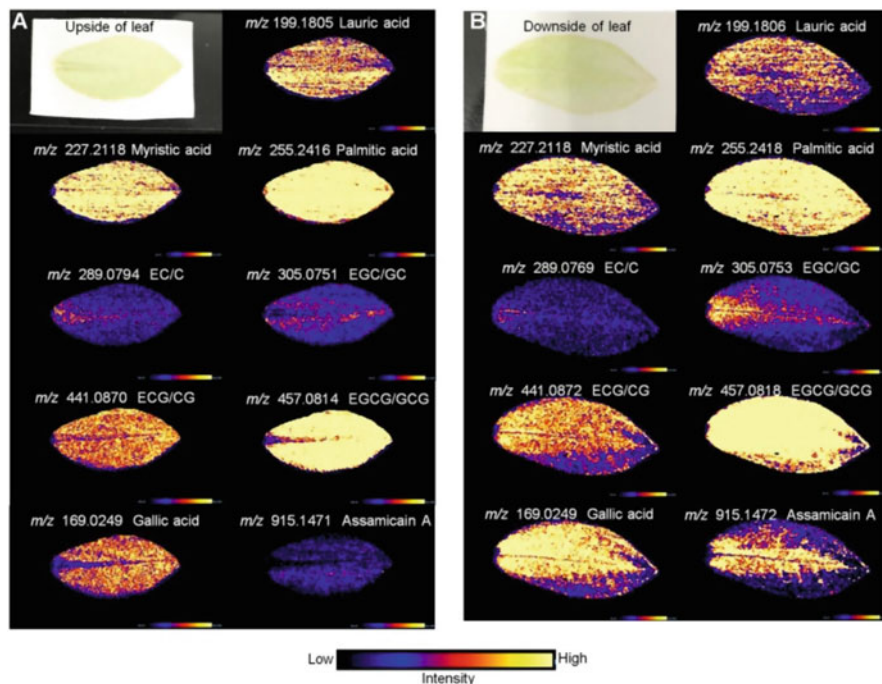
Honeybees play an important role in the environment because they are responsible for the pollination of many plants and crops around the world [85]. Bees are greatly affected by exposure to insecticides. When bees are exposed to them, it





**Fig. 9** Four toxic GAs, dehydrochaconine ( $[M + H]^+$  at  $m/z$  850.49),  $\alpha$ -chaconine ( $[M + H]^+$  at  $m/z$  852.50), dehydrosolanine ( $[M + H]^+$  at  $m/z$  866.49), and  $\alpha$ -solanine ( $[M + H]^+$  at  $m/z$  868.50), in potato tubers were imaged by the iMScope at varying storage times: (a) 0 days (b) 10 days (c) 15 days (d) 20 days (e) 30 days (f) 40 days and (g) 60 days [76]

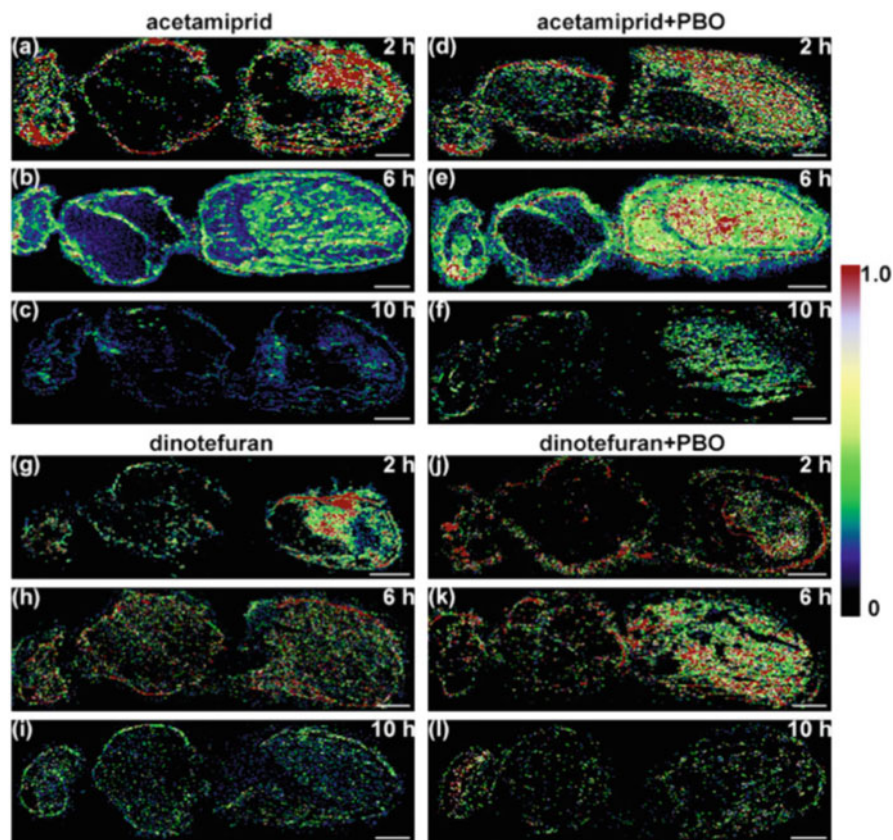
affects the bee's behavior and can threaten their ability to function making them more vulnerable to disease [86] and causes entire colony to disappear [87]. Neonicotinoids are the most used class of insecticides globally due to their success in target pests [88]. In 2017, neonicotinoids were detected in 75% of honey samples collected from all continents and are believed to be the cause of the global decline of honeybees [89]. An increasing number of studies have been addressed to determine their exact mechanism of toxicity in bees, but the spatiotemporal distribution of neonicotinoids has yet to be understood. Zhang et al. employed MALDI MSI to visualize neonicotinoid distribution in honeybees. Healthy adult honeybees were captured and exposed orally and via direct contact to two neonicotinoids, dinotefuran and acetamiprid with co-application of the fungicide tebuconazole. After collection, the bees were frozen at  $-80^\circ\text{C}$ , embedded with optimum cutting temperature (OCT) as per Phan et al. [90] and sliced at  $-20^\circ\text{C}$  to a thickness of  $25\ \mu\text{m}$  using a freezing microtome (CM1950, Leica Microsystems Inc., Wetzlar, Germany) to obtain whole-body, parallel sections were cut from forehead to thorax. After homogenous matrix application, the deposited neonicotinoids were analyzed by MALDI MSI using a TOF-MS that was built and modified by the researchers.



**Fig. 10** Both sides of tea leaves were imaged by DESI-MSI to reveal the spatial distributions of specialized metabolites. (a) The front side of imprinted tea leaf and metabolites. (b) The back side of the imprinted tea leaf and metabolites [84]

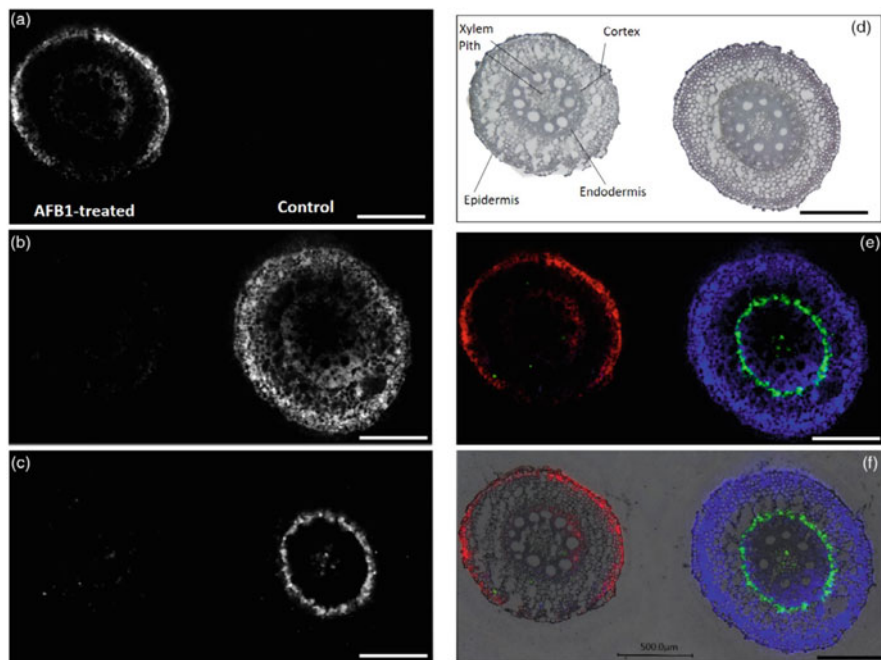
The laser used was focused to 30  $\mu\text{m}$  using an aspheric lens and pulsed three times for 5 ns per pixel to create each image. The images generated showed that both dinotefuran and acetamiprid rapidly penetrate the honeybee's tissues dispersing throughout the whole-body sections in just 2 h as shown in Fig. 11. This figure illustrates that when pretreated with piperonyl butoxide (PBO), an insecticide synergist, acetamiprid becomes 14 times more toxic than dinotefuran. Upon further examination, the degradation rates of dinotefuran and acetamiprid were found to be different. It was determined from the spatiotemporal distributions that these degradation rates cause the toxicity rather than their ability to penetrate through their various barriers. These findings can be used for further research into mechanistic determination of insecticides in honeybees [91].

Plants are incredibly adaptable, able to conform in many ways to enhance their survival. Because they are immobile, plants have developed ways to control the environment around them by communicating with other plant species through airborne signals [92] or by exchanging compounds via the soil-root interface [93]. Plants react to foreign substances by producing metabolites for defense or enzymes to degrade xenobiotics [94, 95]. Current research focuses mainly on the study of compounds that are harmful to plants, but few study the relationship between those compounds and plants in terms of uptake, distribution, and



**Fig. 11** Images of whole-body honeybee sections by MALDI MSI showing acetamiprid and dinotefuran after different exposure times. Acetamiprid ( $m/z$  223.1) after a duration of (a) 2 h, (b) 6 h, and (c) 10 h. Acetamiprid + PBO after (d) 2 h, (e) 6 h, and (f) 10 h. Dinotefuran ( $m/z$  158.2) after (g) 2 h, (h) 6 h, and (i) 10 h. Dinotefuran + PBO after (j) 2 h, (k) 6 h, (l) 10 h. Scale bar is 1 mm [91]

transformation. Righetti et al. used AP-SMALDI MSI to visualize the spatial distribution of aflatoxin B1 (AFB1) and its metabolite aflatoxin M2 (AFM2) to give insight into the defense metabolites produced by maize. Maize roots soaked in a medium containing AFB1 for 7 and 14 days were analyzed via AP-SMALDI. The generated images in Fig. 12 depict sections of AFB1 ( $[M + K]^+$ ,  $m/z$  351.0265) in treated maize root and the lacking distribution of cyanidin glucoside ( $[M]^+$ ,  $m/z$  449.1076), an anthocyanidin compound responsible for stress response in plants, and pheophytin ( $[M + H]^+$ ,  $m/z$  871.5729), a derivative of chlorophyll, in the treated tissue. These results show that AFB1 has significant inhibitory effects on the biosynthesis of these compounds [96].



**Fig. 12** AP-SMALDI images of control and AFB1 treated maize root after 14 days. **(a)** AFB1 accumulation shown in the treated root vs. the control root with no AFB1. **(b)** Cyanidin glucoside and **(c)** pheophytin distribution. **(d)** The optical image of **(a)** with key features labeled. **(e)** Overlay of **(a)** AFB1 (red), **(b)** Cyanidin glucoside (blue) and **(c)** pheophytin (green). **(f)** Red-green-blue image **(e)** merged with optical image **(d)** [96]

## 5 Conclusions and Future Trends in Mass Spectrometry Imaging

MSI is able to study almost anything owing to the wide variety of ionization choices and sample preparation strategies that can be paired depending upon your analyte (s) of interest. From pesticides to peptides and microplastics to metabolites, the future of MSI promises to extend the answerable questions in environmental and food science beyond those that were previously limited to temporal and quantitative data sets. Implementing MSI in environmental and food science research space means that spatial data will become part of routine analysis which will drive these fields forward in the same fashion that MSI has driven biochemical and analytical research since the late 1990s. Providing data that overlaps temporal and spatial information for non-targeted analysis will inform future experiments through identification of unknown adulterants that are needing to be tracked by quantitative analysis in the future. To even further the progress of MSI in this field, there will need to be collaboration between the scientists in traditional MSI fields (e.g., analytical chemistry, human medicine) and newer MSI fields like food and

environmental science. For example, Caprioli determined a method for imaging the whole body of a mouse which could easily be applied in food and environmental MSI to analyze larger sample types to locate toxins [97]. While quantitative MSI (qMSI), a way to absolutely quantify target molecules via MSI, has yet to be used for food or environmental science, it is worth noting here for the significance of its potential future use [98]. Its prevalence in the MSI community, as developed by Tobias and Hummon, has raised the standard that environmental and food science should be looking towards as they become more familiar with the technique of MSI.

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# Chip-Based Separation Devices Coupled to Mass Spectrometry in Food and Environmental Chemistry



Ángel Ríos and Mohammed Zougagh

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**Abstract** Nowadays, chip-based separation devices coupled to mass spectrometry are a leading analytical technique in many science areas of major interest. This interest is based on their economic, environmental, and analytical advantages, such as low volumes consumption of samples and reagent, gain on separation performance, and enhanced sensitivity with MS detectors. The application in food and environmental fields will keep in constant spreading, and new and miniaturized instrumental advances in chip technology will continue to be necessary for the

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coming years. This chapter summarizes the main features of chipLC–MS, chipGC–MS, and chipCE–MS, outlining their main parts. Finally, a description about some diverse applications of chip-based separations mass spectrometry devices is summarized. In this case, showing remarkable examples using chipLC–MS, chipGC–MS, and chipCE–MS to the analysis of food and environmental samples.

**Keywords** Chip-based separation devices, Chip-CE, Chip-GC, Chip-LC, Environmental, Food, Mass spectrometry

## 1 Introduction

The current trend in analytical chemistry is miniaturization using chip technology to improve sensitivity and speed of analysis. Chip-based analyses reduce the sample volume and the consumption of chemical reagents. As a result, chip-based separation devices have been applied to various food and environmental studies. In addition to miniaturized separation devices sensitivity is another key issue. Further confirmation on molecular structures of analytes is generally required for various food and environmental applications. Mass spectrometry (MS) offers additional high-quality structural information for qualitative identification and good sensitivity for quantitative determination of target molecules [1]. Therefore, MS has been widely utilized for sensitive detection in many chip-based separation devices. Coupling of MS to chip-based separation systems has been intensively studied. These developments have been focused nearly exclusively on chip technology. Although chip-based separation devices were first reported in 1976 in gas chromatography (GC) [2], the scientific breakthrough of the idea took until the 1990s when liquid separation techniques such as liquid chromatography (LC) and capillary electrophoresis (CE) [3–7] and their coupling with MS [8, 9] were introduced. Nowadays, areas of the technology have reached a rather mature status which is reflected by the availability of several commercial products like chipCE–MS and chipLC–MS. The crucial point for online coupling of chip-based separation devices with MS instruments is the interface that transfers the dissolved molecules in the chip to gas phase ions which is necessary for subsequent MS analysis. The introduction of miniaturized electrospray ionization, in the middle of the 1990s, immediately revealed outstanding detection becoming the preferential choice to hyphenate with chipLC and chipCE [10]. Wilm and Mann developed the first microelectrospray emitter [10], which by the reduction of the tip internal diameter (from 100  $\mu\text{m}$  to 1–20  $\mu\text{m}$ ) provided much higher sensitivity. Likewise, Emmett et al. [11], Davis et al. [12], and Gatlin et al. [13] also demonstrated similar detectability improvements through miniaturization of the electrospray ionization (ESI) source, even at the nanoscale. Nevertheless, micro-/nano-ESI-MS emitters are not suitable for coupling chipGC instruments. Microchips atmospheric pressure chemical ionization MS (APCI-MS) and atmospheric pressure photoionization MS (APPI-MS) systems [14] which

provided flow rates from tens of nanolitres to several microlitres per minute, efficient ionization, excellent sensitivity, and good reproducibility offer a potential ionization technique for chipGC–MS. In this chapter, the advances of chip-based separation devices, including chipGC, chipLC, and chipCE, all coupled to MS will be discussed. In addition, the usage of these chips to determine molecules in various sample matrices for environmental and food fields will also be presented.

## 2 Chip Chromatography–Mass Spectrometry

The hyphenation of chip-based separation devices like GC or LC to MS is a highly active research area in modern separation techniques. Such methods are particularly attractive for comprehensive food and environmental chemistry. They can handle extremely low sample amounts, with low solvent consumption. Furthermore, they provide unsurpassed analysis speed together with the prospect of integrating several functional elements on a single multifunctional platform. In this part, we describe the latest developments in this emerging field.

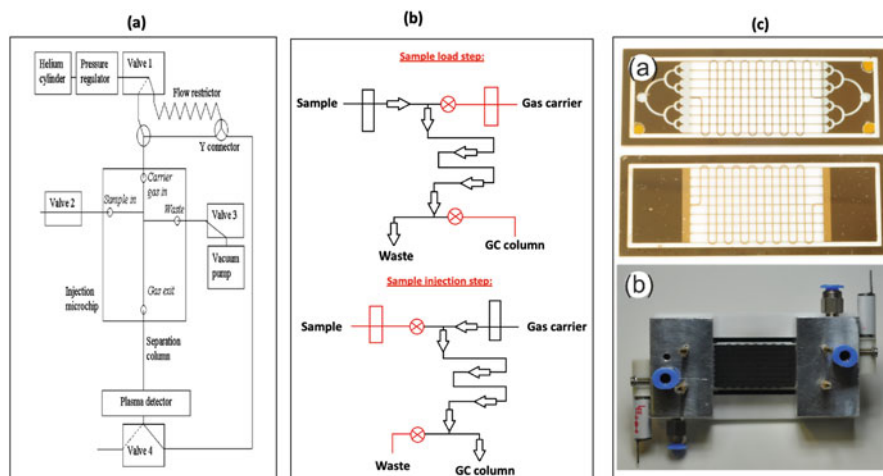
### 2.1 *Chip gas Chromatography–Mass Spectrometry (ChipGC–MS)*

The main components of the ChipGC–MS system are gas injector and preconcentrator, column, MS detector, and electronics for data pre-processing that play an important role in a system integration perspective. Recently, numerous research efforts have benefited from the advances in micro-electromechanical systems (MEMS) and nanotechnology-based solutions to miniaturize the key components of GC instrument. Reducing the dimensions of GC systems enhances the analytical performances and analysis of small-volume samples and increases analysis speed. A chipGC–MS system requires several components to function properly: preconcentrator, micro-valves for injecting the sample into the carrier gas, microfabricated columns well-functionalized for the specific use, heaters, temperature sensors for controlling column temperature, and MS interfaced detector for detecting different types of molecules. Temperature stability is also critical for GC operation, as the adsorption/desorption processes responsible for molecular separation in the column are very sensitive to temperature. The issues of microfluidic integration are therefore critical in chipGC–MS systems. Despite the fact that the first work on chipGC system was a miniaturized GC in 1979 [2] using micro-electromechanical systems (MEMS), this development was hardly pursued afterward, probably because the resolving power was poor, when compared to the conventional ones and the analytical community was not yet ready to embrace this new technology. Unsatisfactory results, due to difficulties in producing

homogeneous stationary phases, may have contributed to this feeling. However, miniaturized GC are of great interest in several application fields. These microsystems could potentially be used for breath analysis, indoor air-quality monitoring, and warfare agent detection, allowing the onsite and portable monitoring of gases in a fast and reliable way. A typical ChipGC–MS system consists of microinjector, micropreconcentrator, column or microcolumn, MS detector, and electronics for data processing in miniaturized form, thus promising an enhanced performance in terms of size, power consumption, resolution, and speed.

### 2.1.1 Microinjector and Preconcentrator

The microinjector is a device used for introducing during milliseconds very small plug of sample vapour (nanolitres) into the carrier gas stream at the head of the capillary column. Terry et al. were the first to report a gas sample microinjector much smaller than that used in GC commercially [2]. It consists of an etched silicon seat and orifice, a nickel diaphragm, and a solenoid actuator and plunger assembly. Both the solenoid assembly and the diaphragm are removable to facilitate cleaning of the internal valve areas. Since this first development, several authors designed new and improved microinjectors [15–18]. Bessoth et al. [19] reported the first injector applied for gas injection having the geometry commonly used in microchip capillary electrophoresis (Fig. 1a). In the sample load step, valve 1 connected flow restrictor and pressure regulator while valve 4 fed the outlet of detector. Valves 2 and 3 were opened and gas was transported from the sample inlet to the waste outlet by the



**Fig. 1** Different methods of introducing sample mixture in separation channel: (a) gas injector having the geometry commonly used in microchip capillary electrophoresis (adapted from [19]), (b) six low-leak valves with 250 nL sample volume, (c) large-capacity preconcentrator device (adapted from [20]: (a) top view with Carbosieve SII as filling material and housing (b))

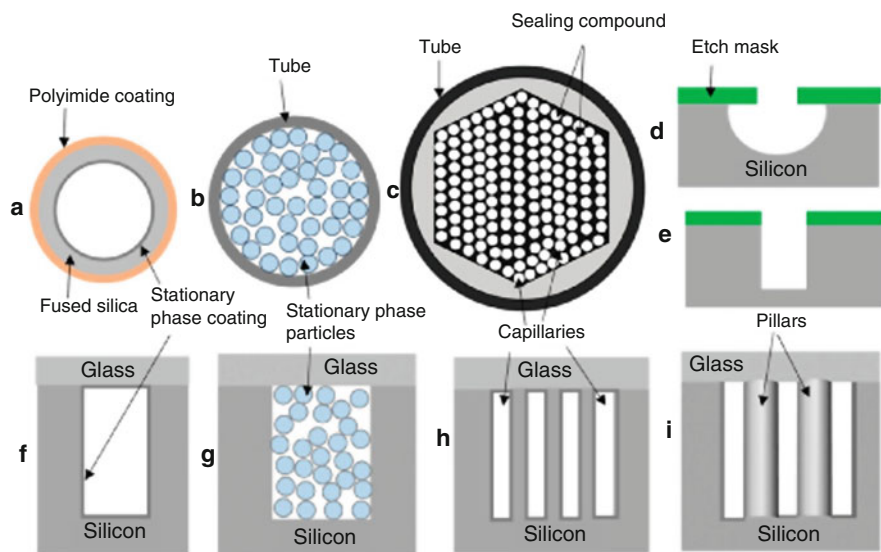
suction of the vacuum pump. The sample is introduced and flushed out into the waste channel. In the sample injection step, valves 1 and 4 are switched to their alternative positions and valves 2 and 3 are closed. As a result, the carrier gas pushes a plug of sample towards the separation column and detector. More recently, Nachef et al. [21] developed a six low-leak microvalve (Fig. 1b) made from polyetheretherketone (PEEK) membranes on silicon substrates, when the inlet and outlet holes are closer to each other, with their centres being distant by 300  $\mu\text{m}$ . The valves are pneumatically actuated, and work in two steps: (1) sample load, which consists of filling the largest microchannel with a known sample volume (250 nL) and (2) sample injection, during which the sample is injected into the separation column.

### 2.1.2 Micropreconcentrators

Micropreconcentrators allow the collection and concentration of analyte(s) followed by their release and injection into the ChipGC system. Since the first micro-machined preconcentrator designed and fabricated by the ChemLab at Sandia National Laboratories in 1999 [21–23], many works have been done with various preconcentrating microstructures of different designs combined with a wide range of adsorbents [20, 23–27]. Alfeeli et al. reported a microthermal preconcentrator (TPC) device with high-aspect-ratio 3D structures which comprises more than 3,500 micropillars and allows large adsorption surface area [23]. This device was used as a GC injector to demonstrate its ability to concentrate a dilute mixture of multicomponent hydrocarbons into a sharp pulse suitable for GC analysis. Tian et al. developed a preconcentrator-focuser (PCF), consisting of a thick micro-machined Si heater packed with graphitized carbon granules (200  $\mu\text{m}$  diameter) with high surface area [24]. The PCF was connected directly to GC detector through a six-way valve using passivated fused-silica capillary tubing. A small aliquot (250  $\mu\text{L}$ ) of sample was extracted by gas-tight syringe and injected into the sample stream which was drawn through the PCF by a downstream diaphragm pump. Subsequent heating of the PCF led to desorption of the captured vapours, which were passed on a positive pressure of carrier gas to GC detector. Jasen et al. designed and fabricated a large-capacity on-chip preconcentrator device with a micro-machined silicon cartridge and an integrated heater [20]. The absorption material was Carbosieve S-II immobilized in PC device channels housed in stainless-steel unit, as represented in Fig. 1c. Zampolli et al. reported a preconcentration column consisting of a micro-machined silicon cartridge (25  $\times$  12  $\times$  1.3 mm), containing quinoxaline-bridged cavitand (QxCav) selective for retention of aromatic compounds. This preconcentrator component has been integrated into a miniaturized gas chromatograph and allowed real-time monitoring of sub-ppb concentrations of aromatic volatiles [25].

### 2.1.3 Column or Microcolumn

Column is the most critical component in the ChipGC system. Two types of columns have been explored for its use, capillary columns and chip-based columns. Conventional capillary columns are circular in cross section, while chip-based columns, known as microcolumns, are typically rectangular in cross section. Figure 2 shows the cross-sectional view of different separation columns of both types used in ChipGC system. Microcolumns offer several advantages. The small size of microcolumns enables high speed and low power heating; large-scale batch production of these columns may result in low manufacturing cost; monolithic integration of the columns with other components potentially minimizes the dead volumes and cold spots; and fabrication of regularly arranged support structures inside the channels enhances the separation performance [28]. Major limitation, however, of these columns is the accumulation of the stationary phase in sharp corners (known as pooling effect) resulting in non-uniform film thickness inside the column leading to band broadening. Nonetheless, in the case of high-aspect-ratio rectangular columns, the efficiency can be increased by decreasing the width of the channel, and the sample capacity (maximum amount of sample that can be injected without overloading). Flow rate can be increased by increasing the height, thereby offering a possibility of getting better efficiencies compared to conventional capillary columns. Considerable efforts are being made to increase the separation efficiency of



**Fig. 2** Cross-sectional view of different GC columns and etched profiles: (a) wall-coated capillary column, (b) packed capillary column, (c) multicapillary column, (d) isotropic etch profile, (e) anisotropic etch profile, (f) wall-coated microcolumn, (g) packed microcolumn, (h) multicapillary microcolumn, and (i) pillar-array column. Adapted from [28]



the columns by exploring different column architectures and stationary phase coating techniques.

Several materials have been used as a substrate to fabricate microcolumns such as silicon wafer [29], metal (nickel, steel, and titanium) [30], polymeric materials and glass [31]. Gaddes et al. [29] microfabricated a 2 m of silicon wafer column using silicon microfabrication techniques and dynamic coating process for PDMS (OV-1) stationary phase. This GC microcolumn along with the high temperature fixture offers a better solution towards the achievement of a ChipGC system for the detection of semi-volatile environmental pollutants and explosives without the thermal limitations reported to date with GC microcolumns using epoxy-based interconnect technology. Lewis and Wheeler developed a nonplanar microcolumn using a planar nickel substrate containing circular through-holes that further apply to separate polar as well as nonpolar analytes [30]. Rankin et al. [31] developed the first moulded GC microcolumn in polymeric material. This microcolumn consists of a single microtextured thermoset polymer composite which acts as both substrate and stationary phase.

Depending on type of microcolumns, several materials have been used as stationary phases. Graphitized carbon black [32], polyethylene glycol (PEG)-coated graphitized carbon black [25, 33], porous polymer beads [25, 34], and carbon molecular sieves have been used for packed columns, particularly for separation of permanent gases and light hydrocarbons. In open and semi-packed columns, PDMS and its derivatives have been predominantly used to coat microcolumns [35, 36]. For separation of polar compounds, polymers of PEGs have been frequently used [37]. There are two types of methods for coating the columns, named static and dynamic methods. Static methods are the more appropriate, because they obtain more homogenous stationary phase films and are easier to implement. First, a known concentration stationary phase solution is prepared in a solvent mixture and the entire column is filled with this dilute solution by means of a pressurized gas. One of the ends of the column is then sealed, and the solvent is evaporated under vacuum through the other end. Static coating produces a more uniform coating, and the coating thickness can be easily calculated by employing the surface area of the column, concentration of the stationary phase in the solution, and the density of the coating solution. One of the disadvantages of this coating method is the formation of bubbles during the evaporation of the solvent. For this reason, extreme precautions are required during this coating process. A 5.9 m long cylindrical fused-silica microcolumn, fabricated in a silicon wafer using deeper reactive ion etching (DRIE) and high temperature fusion bonding, was coated with a 1% vinyl, 5% phenyl, 94% methyl polysiloxane stationary phase using a static method [38]. The authors found that the coating was not uniform, due to the bubbles originated from inefficient degassing of the coating solution in the micro-cavities of the channel walls or at the capillary lead/microchip interfaces. As an alternative, the dynamic method forcing the introduction of the stationary phase through the column by an inert gas and then, evaporating the solvent by continuous gas flow can be used to fabricate columns. This method is fast, but advanced prediction of film thickness is difficult, due to its alteration as the concentration of stationary phase solution and the

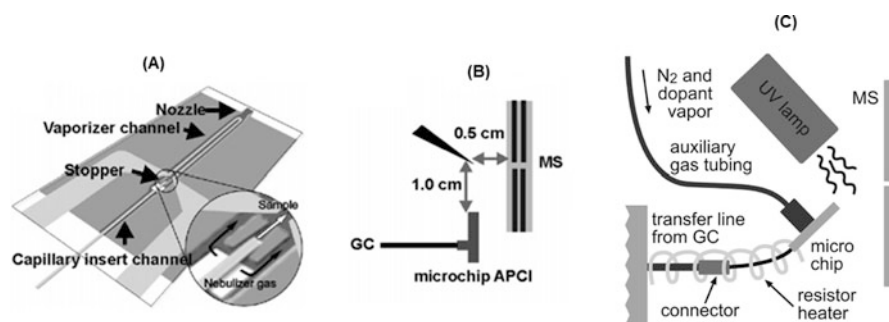
velocity change during the coating process. In both methods (static and dynamic), the limitation is accumulation of the stationary phase at the sharp corners (known as the pooling effect) resulting in the band broadening. To minimize the pooling effect, the walls of microcolumns are rounded using isotropic etching after the DRIE process. In recent years, laser etching technology (LET) has become an important etching method. As it can easily etch deep channels on glass wafers as well as on silicon wafers, the column depth can be substantially increased by bonding the channels on the silicon wafer together with the channels on the glass wafer. Consequently, the column can achieve a channel with a depth of 1.2 mm to 1.5 mm, which greatly enhances the aspect ratio and the sample capacity, improving the separation efficiency of micropacked GC columns. Sun et al. fabricated a micropacked gas chromatograph column integrated with a microheater by using LET for analysing environmental gases [34]. LET is a powerful tool to etch deep well-shaped channels on the glass wafer, and it is the most effective way to increase depth of channel. A number of stationary phases have been developed, with different features depending on the field and analytes to which they are applied. However, research efforts in capillary ChipGC have been focused on achieving high-resolution separations by means of porous selective layers that create new microcolumns. In this sense, the use of nanomaterials (NMs) such as AuNPs, SiO<sub>2</sub>NPs, graphene, carbon nanotubes (CNTs), and metal-organic frameworks (MOFs), constitutes a very interesting alternative to conventional stationary phases because of the extraordinary properties that these kind of materials have, including high surface area, thermal stability, and chemical inertness [39]. Moreover, their ability to be functionalized increases their versatility making the separation of target analytes even possible.

MOFs are the most attractive nanomaterials class for stationary phases because of their chemical functionality and pore size. MOFs are porous materials that consist of organic ligands that bridge inorganic centers (typically metals or metal clusters). The MOF structure is templated by the specific ligand used in the synthesis. MOFs are highly porous, thermally stable, and possess tuneable chemical sorption affinity/selectivity through modification of their chemical functionality and pore structure. Certain MOFs can also be deposited using layer-by-layer techniques directly on selected substrates as conformal thin films [40]. These so-called surface-mounted MOFs (SURMOFs) allow for the direct deposition of a MOF thin film with a known thickness, both attributes being ideal from a chromatography perspective. SURMOFs can be tuned in terms of their chemical functionality and pore size and as such make an attractive class for stationary phases over porous layer open tubular (PLOT) with proprietary formulation, and thus inaccessible for in-house of non-commercial stationary phases. Read and Sillerud from National Laboratories fabricated GC microcolumn using MOF thin films as stationary phases [41]. The authors demonstrated the first reported MOF-stationary-phase, high-speed chromatographic separation of natural gas using GC columns – outperforming commercial Porous Layer Open Tubular (PLOT) stationary phases in terms of both polarity and absolute partition coefficients. Xie et al. developed chiral MOF coated open tubular microcolumns that were used in the GC separation of chiral compounds. The

microcolumns have excellent selectivity and also possess good recognition ability towards a wide range of organic compounds such as alkanes, alcohols, and isomers [42].

### 2.1.4 Mass Spectrometry Detectors

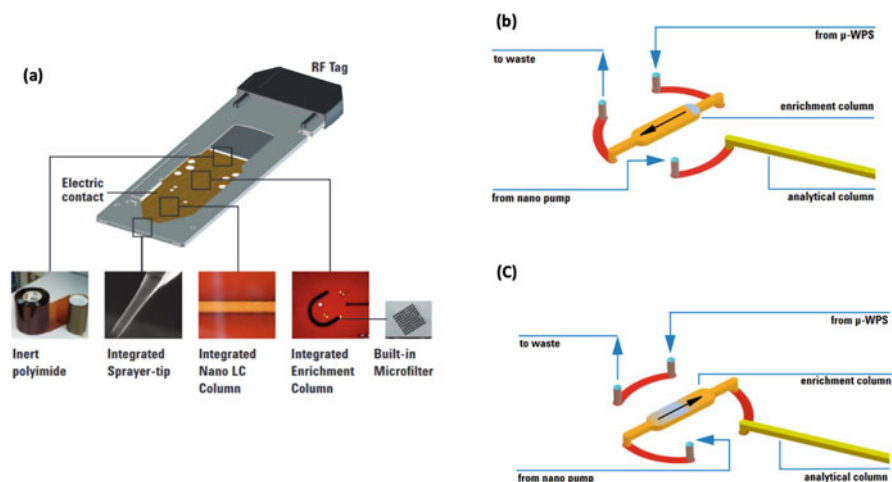
MS detectors provide the determination of ChipGC separated compounds with high sensitivity. However, because of their sizes, weight, and power consumption, MS have rarely accompanied GC in portable instrument. That is why a large part of efforts for the development of MS micro-detectors have been made around the conception of micro-mass sensors [43, 44]. Kostianin research group [44, 45] microfabricated nebulizer chip for combining atmospheric pressure chemical ionization (APCI) and atmospheric pressure photoionization mass spectrometry (APPI-MS) with GC (Fig. 3). The chip consists of a silicon plate and one or two glass plates. The chip includes a sample inlet channel, auxiliary gas and dopant inlet, vaporizer channel, nozzle, and platinum heater. The sample from GC is directed via heated transfer line inserted from the rear edge into a stopper inside the vaporizer channel of the APCI chip. In the ChipGC-APPI-MS, the sample eluted from the GC is mixed with auxiliary gas and dopant (toluene) in the heated vaporizer. The chip forms a confined jet of the sample vapour, which is photoionized as it exits the chip. This ChipGC-APPI-MS provides an efficient method for qualitative and quantitative analysis with detection limits down to the nanomole per litre range and providing the same type of information as classical chemical ionization for structure elucidation.



**Fig. 3** (a) Schematic presentation of the microchipGC. Sample inlet capillary is inserted from the rear edge of the chip to the narrowing. The nebulizer gas flows from behind and along the capillary to the flow channel. (b) Schematic presentation of the positioning of the microchip APCI and the corona discharge needle in front of the mass spectrometer. (c) Schematic view of the ChipGC-APPI-MS. Adapted from [44, 45]

## 2.2 *Chip-Based Liquid Chromatography–Mass Spectrometry (ChipLC–MS)*

Much of the articles reporting the use of chipLC use commercially available systems. ChipLC–MS can be presented as micro-LC–MS or nano-LC–MS systems, which provide extremely sensitive analyses, although some robustness problems can be produced due to dead volumes and leaks in their designs. ChipLC offers a solution for these challenges. In fact, the first chipLC was fabricated by Manz et al. in 1990 [3]. It consists of an open tubular spiral column made on a silicon wafer, interfaced with a conventional valve injector, and a conductometric detector (cell volume of 1.2 pL) fabricated on a Pyrex glass chip. Although actual experiments have not been performed using this chipLC system, this work opened the possibility for LC miniaturization in a chip format, and presented its potential advantages, allowing the commercialization of various formats of chipLC columns and interfaces with various detectors, especially with MS. Based on this model, Agilent chipCL is described [46] (Fig. 4a). It consists of an inert polyimide device integrating a nano-electrospray interface connected to a mass spectrometer, an analytical chromatographic column of appropriate size to the nano-electrospray flow rate, and an enrichment column for online sample concentration. In this device, there are no fittings, adapters, connectors, or any other dispersive flow element negatively affecting the performance in  $\mu$ -LC and nano-LC. Transfer volume between the enrichment column and the analytical column is minimized by installing the chipLC within an LC rotary valve. In this way, the chipLC is interposed between the rotor and the stator with a precise registration. In one position the flow from the



**Fig. 4** (a). View of the commercial Agilent HPLC Chip and scheme of the chip–rotor interface in the LC run mode (b) and in the loading process: the sample-loading configuration (c) and the LC-running configuration. Reproduced from [46]

liquid autosampler is directed to an enrichment area (a short, packed column of RP material for adsorbing the sample components), and then to waste (Fig. 4b). When the rotor travels  $60^\circ$ , the flow from the nanopump enters the enrichment column, sweeping the sample(s) into the analytical column (Fig. 4c).

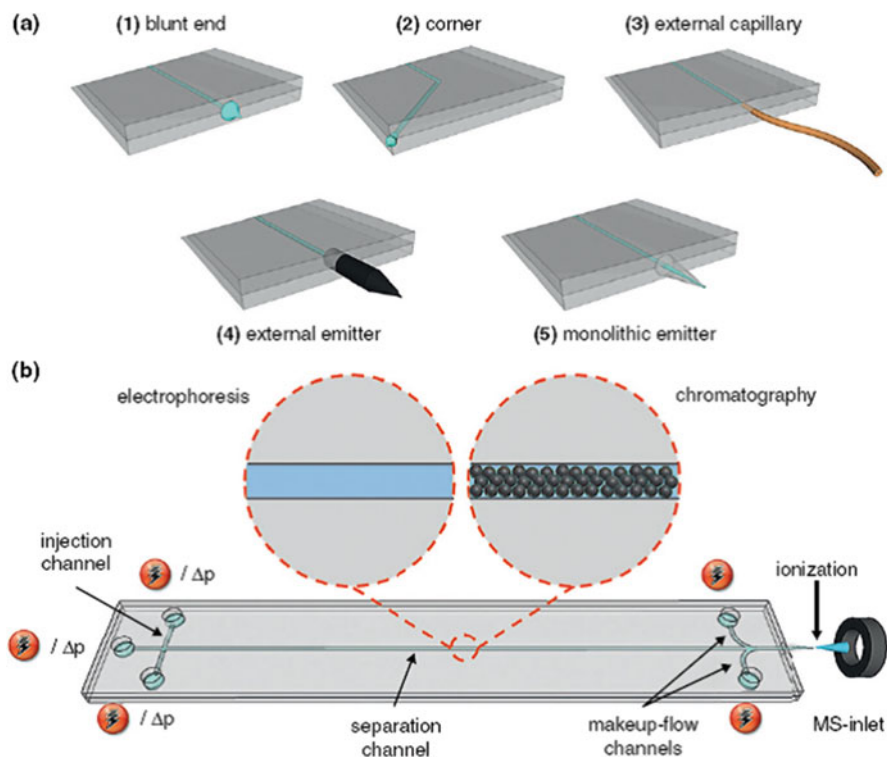
At the end of the column, the flow passes electrical contacts, which allow the biasing of the effluent to the electrospray. The highly stable spray exits through a 2 mm long  $40\ \mu\text{m}$  inner diameter tip and enters to the MS analyser. For coupling chipLC, ESI is the more efficient ionization mode, as described in [47]. The main analytical advantage of chipLC-ESI-MS systems is the enhanced detection capabilities derived from the synergistic compatibility shown by these techniques at a miniaturized scale [48]. Miniaturized flow rates favour the ESI process leading to enhanced MS detectability [49]. The miniaturization of the ESI sources leads to the more efficient transference of the targeted ions to the vapour phase, reducing matrix effects and susceptibility of the ionization process to the effluent composition [10]. On the other hand, nano-ESI sensitivity also directly depends on parameters such as the emitter internal diameter, geometry, material, and position. Hence, nano-emitters architecture has been under continuous evolution, and several works dedicated to improving emitters have been performed in the last decades [50]. Nowadays, two main strategies for coupling chipLC with nanoESI-MS can be distinguished: (a) spraying via microfabrication of integrated emitters and (b) spraying by coupling with external emitters. Matrix-assisted laser desorption ionization (MALDI) has been another versatile ionization source for interfacing chip-based separations with MS. In this case, the coupling has been achieved microfabricating MALDI reservoirs at the end of the separation channel, for on-chip ionization [51], or by dropping off the chip effluent in a MALDI probe, for offline analysis.

### 3 Chip Electrophoresis Mass Spectrometry

Despite the difference in size between a chipCE and a MS, and hence, the a priori difficulties in making them compatible, they are actually well matched due to the similarities in the flow rate generated by the chipCE and those required for the ESI. Various chipCEs have been coupled to MS [52–62] and the most reported chipCE–MS interfaces are ESI [63–66], and MALDI.

A typical chipCE generally included on-chip sample loading and separation without any connectors. The coupling of a chipCE to MS may be achieved using a sheath flow or a liquid junction to provide stable electrospray signals [55–62, 67, 68].

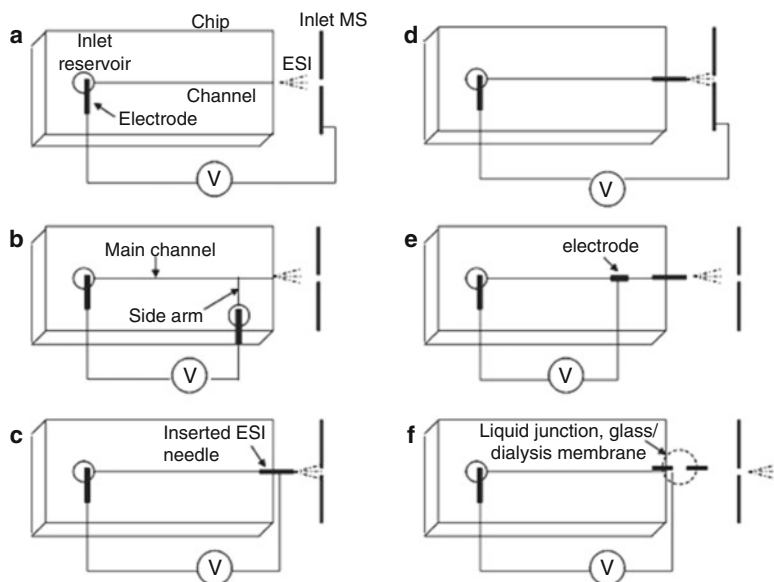
An illustrating possible layout of a chipCE device is shown in Fig. 5. Different approaches for combining ESI with chipCE are schematically shown in Fig. 5a. These are, spraying first, from the blunt end of a chip, second, from the corner of a thin chip, third, from an attached external capillary, fourth, from an attached confined nanospray emitter, and finally from a monolithically integrated confined



**Fig. 5** MS coupling of chipCE. (a) Overview of different approaches for hyphenation of microfluidics with electrospray MS. (b) Schematic of a microfluidic device for chip electrophoresis or chip chromatography. (Reproduced from [69])

nanospray emitter. Such chips can have the footprint of a microscope slide containing crossed microchannels with widths of the size of a human hair or less (Fig. 5b). The short channel is often used for sample injection (especially in electrodriven separations), whereas the actual separation occurs in the long channel, which is filled with a stationary phase in the case of chip chromatography. The pros and cons of these interface types were discussed extensively in literature [70, 71]. To facilitate the electrospray process at the emitter, for example, for the application of an electrical potential, optional makeup-flow channels can be implemented.

Considering spraying, from the blunt end of a chipCE, three basic approaches can be distinguished [66]. The first possibility is to locate the electrospray directly at the end of the microfluidic platform (Fig. 6a and Fig. 6b). Although its simplicity, the position of the electrospray with respect to the chipCE is often difficult to control. To avoid this problem, an approach with the electrospray inserted in the proper chipCE was proposed (Fig. 6c). In another approach, the electrical field is applied between one of the inlet reservoirs on the chipCE and the inlet of the mass spectrometer, as shown in Fig. 6d [72]. The problem here is that a large potential difference between the end of the microfluidic channel and one of the inlets of the microfluidic device



**Fig. 6** Basic coupling configurations for chipCE–MS arrangements (a) Spray directly from the edge of a chip, with the spray potential applied between a reservoir and the inlet of the mass spectrometer; (b) spray from the edge of a chip, with potential applied between inlet reservoir and coated side-arm channel at the end of the main channel which can serve as a separation channel; (c) spray from an inserted fused-silica (FS) needle, with the potential applied between the reservoir and the FS needle; (d) as (c) but with the potential applied between the reservoir and the inlet of the MS; (e) electrode integrated in microfluidic channel for applying the electrical field for ESI; (f) spraying with an integrated liquid junction or dialysis/ porous membrane (Reproduced from [66] with permission of the Royal Society of Chemistry)

would be needed to perform a separation, resulting in the generation of heat. In other approaches, the electrical field is applied to a stainless-steel tube connected to the exit of a chipCE through which the eluent is guided [73], or to electrodes integrated in the microfluidic channel, which may be made of gold [74, 75], conductive carbon or conductive epoxy [76] (Fig. 6e). All these approaches have the advantage that compounds eluting from the column are directly electrosprayed without dilution. A different configuration consists of a liquid junction making use frequently of a sheath flow of make-up buffer. This approach has as advantage to be able to modify the composition of the separation buffer to make it suitable for ESI. Most liquid junctions are made by inserting a spray capillary into the device and coupling it to an existing ion source [77, 78]. Couplings where the potential is applied to a side channel in contact with the separation channel through a porous (glass) membrane [79] or a dialysis membrane made of polysulphone [80, 81] have also been reported (Fig. 6f).

One of the main problems associated to the manual insertion of spray needles into the microchannel is that the approach does not lend itself to mass fabrication [76]. As

a solution, integrated microchip emitters can be fabricated, mainly by using a photolithographic procedure, which allows to simultaneously fabricate many structures in a single step.

The direct integration of chipCE and MALDI targets within a single chip appears to be a particularly natural combination since both technologies generally employ planar substrates. The growing interest in this integration is reflected on the range of approaches developed for delivering analyte to on-chip MALDI target surfaces. From the point of view of microfluidic systems, the offline nature of MALDI–MS analysis represents a key advantage over ESI-MS. For example, the time scales for molecular separations and online MS data acquisition are often incompatible. However, offline MALDI–MS enables on-chip sample processing steps to be fully decoupled from the back-end MS analysis. In addition, offline MALDI–MS analysis provides a powerful method for coupling multiplexed parallel on-chip analyses with MS detection.

This type of detection is particularly suitable for analytical applications. MS is a powerful analytical tool. Its integration to CE in general (CE-MS) provides a valuable contribution, as it combines separation speed, high resolution, minimum sample consumption, selectivity, and structural information, including a highly accurate determination of molecular weights. Thus, this hyphenated technique opens very interesting possibilities for analytical purposes. In consequence, the chipCE–MS integration introduces the additional advantages of the separation in microchips, specially providing a high-throughput for the analysis of food and environmental samples, by using very small sample volumes.

#### **4 Some Applications of Chip-Based Separation Devices and MS in Food and Environmental Chemistry**

Generally, food and environmental samples are considered complex matrices containing numerous macro and micro interferents, which commonly demand highly effective analytical methods to being properly analysed. Within such a context, conventional LC and CE coupled to MS are highly effective techniques to achieve such goals; however, they are also expensive, involving complex operation steps, and time-consuming, in some cases. For these reasons, chipGC–MS, chipLC–MS, and chipCE–MS approaches are emerging to creating new procedures showing similar or even better performance compared to conventional GC–MS, LC–MS, and CE–MS, reducing steps to as few as possible. These platforms (chipGC–MS, chipLC–MS, and chipCE–MS) are excellent for the separation, profiling, and quantification of target analytes in various types of matrixes. In this section, most representative applications of these devices used as separation techniques in food and environmental fields will be addressed. In food field, several applications of chipGC–MS, chipLC–MS, and chipCE–MS have emerged. A great percentage of published papers use these chip systems in food quality and food safety evaluation,



by determining, phenolic compounds [80–88], peptides and amino acids, protein, vitamins [84], antibiotics, pesticides [87], and toxins. These compounds have been determined in different matrices like fruits [84], wine, [85], vegetable oils [84], edible nuts [86], milk and honey [87], and edible nuts [86, 88].

Chip-LC coupled to time-of-flight mass spectrometry (TOF-MS) was applied for accurate and routine characterization and quantification of phenolic compounds such as chlorogenic acid, p-coumaric acid, (+)-catechin, (–)-epicatechin, procyanidin A2, quercetin 3-O-glucoside, quercetin 3-O-rhamnoside, quercetin, and myricetin in cranberry or cranberry based-products [82]. The chromatographic parameters were optimized, and the method was validated using lock-mass calibration obtaining adequate linearity, with  $R^2$  above 0.99, and acceptable values of intra- and inter-day repeatability of the retention time and peak area. The detection limits and quantification were between  $1.0\text{--}15.6\text{ ng mL}^{-1}$  and  $2.0\text{--}62.5\text{ ng mL}^{-1}$ , respectively.

Another challenging topic of interest for using chipLC and chipCE is the separation of peptides and amino acids. Giusti et al. developed a selenopeptide mapping in a selenium–yeast protein digest by parallel chipLC–inductively coupled plasma, ICP–MS, and chipLC–MS after online preconcentration on a C18 capillary precolumn [83]. This method allowed the identification of whole series of Se/S substitutions in individual peptides and, in some cases, sequencing of isomers differing in the position of selenomethionine residues in the amino acid sequence. Other interesting application of chipCE–MS in food samples involves the determination of amino acids by microfluidic chip electrophoresis mass spectrometry method [89]. The described platform enabled the separation of 16 amino acids in  $<2$  min and without the requirement for extensive sample preparation.

Recently, Aydođan et al. [87] used integrated nano-LC–high-resolution orbitrap mass spectrometry (HR Orbitrap MS), for the analysis of antibiotics and pesticides in milk and honey. The stationary phase was prepared by polymerization procedure incorporating multiwall carbon nanotubes (MWCNTs). This monolithic stationary phase achieved the separation of eight antibiotics and six pesticides with high theoretical plate numbers up to 46,000 plates per min. The use of monolith-based nano-LC in combination with HR Orbitrap MS exhibited low limits of detection at the  $\text{ng mL}^{-1}$  level. The improvement was due to the use of a nanoscale monolith ( $75\text{ }\mu\text{m i.d.}$ ), which increased ionization efficiency with respect to that originated by nanospray ionization.

Proteins are very important in wine quality, because they affect taste, clarity, and stability. In this sense, Kwon has determined the profile of soluble proteins in wine by nano-LC–MS [85]. Twenty proteins were identified from a white wine including five proteins derived from the grape, 12 from yeast, two from bacteria, and one from fungi. The proteins profiling in white wine was reported for the first time in this study using one-dimensional sodium dodecyl sulphate–polyacrylamide gel electrophoresis (SDS-PAGE) coupled with nano-LC–MS.

Cerretani et al. [84] reported a very interesting work dealing with the analysis of vitamin E in vegetable oils with different botanical origins by using nano-LC with UV detection. The separation of tocopherols and tocotrienols was carried out in a C18 silica monolithic column ( $150\text{ mm} \times 0.1\text{ mm}$ ) using isocratic elution with

acetonitrile/methanol/water (acidified with 0.2% acetic acid) at a flow rate of  $0.5 \mu\text{L min}^{-1}$ , giving a total analysis time below 18 min. Under the optimum conditions, limits of detection were lower than  $0.16 \mu\text{g mL}^{-1}$ .

Apart from studies involving peptides and other endogenous compounds identification, chipLC–MS was also used for food quality/safe analysis of aflatoxin trace levels in peanut samples and peanut feedstuffs [88]. A remarkable feature is the sub-ng  $\text{g}^{-1}$  limits of detection reported for the five compounds, which underline the great sensitivity related to chipLC–MS approaches. Alcantara et al. [86] determined mycotoxins residues in edible nuts, using a commercial chip-LC system coupled with MS. Chromatographic parameters were optimized and good recoveries (ranging from 75 to 98%) and lowest concentration levels between 0.05 and  $5 \mu\text{g kg}^{-1}$  (lower than the maximum levels established by the current legislation) were obtained. ChipLC coupled to TOF–MS was applied for accurate and routine characterization and quantification of 17 mycotoxins compounds in commercial peanut, almond, and pistachio samples.

When looking at the methods for environmental field in the conventional scale, the significant reduction in analysis time and solvent consumption is a great perspective to support the chipGC–MS, chipLC–MS, and chipCE–MS approaches in this field. In this same way, Haapala et al. [45] developed a chipGC–APPI–MS for determination of polycyclic aromatic hydrocarbons (PAHs) in environmental samples. This chip platform provides high sensitivity down to 0.8 fmol, repeatability (RSD = 7.5–14%), linearity ( $r = 0.9952$ – $0.9987$ ), and short analysis time (14 min).

Another interesting application in environmental field of chip-based approaches also involves analysis of pesticides on a single  $\text{C}_{30}$  bead via a chipLC combined with ESI-Q-TOF-MS [90]. Microfluidic device was introduced to manipulate the single  $\text{C}_{30}$  bead for its miniaturization. The trace level chemicals from the single  $\text{C}_{30}$  bead were detected and identified by mass spectrum, which showed accurate and clear characteristic peaks for aimed compounds. Significant potential on analysis of environmental samples on this combination system was demonstrated. It consists in reduction of analysis time to 5 min and the solvent consumption to  $2.5 \mu\text{L}$ , obtention of high sensitivity detection avoiding the complex sample pretreatment.

## 5 Conclusions

Chip-based separation devices coupled to mass spectrometry are a promising platform for food and environmental analysis, which will particularly replace traditional separation systems (LC–MS, GC–MS, and CE–MS). Compared to these traditional systems, they offer several advantages: (1) they can provide quantitative and qualitative information of analytes at molecular level, (2) they have low cost and consume small volumes of samples and reagent, and (3) permit multi-step operations, since sample preparation, separation, and signal analysis can be simultaneously integrated into a single chip-based separation device. Although important technology innovations aiming the fully-integration of chip-based separation devices

and mass analysers have been accomplished, their use for routine analyses is still under evaluation. Continuing efforts of instrumentation development by manufacturers and researchers will provide more comprehensive chip-based MS in food and environmental fields. Glass chip-based separation devices offer lower background MS signals than polymer-based or polymer-modified chips, due to their good chemical stability. However, the fabrication processes of these chips are relatively more complicated compared with the processes of fabricating polymer-based chips. On the other hand, most chips have noncylindrical (such as quadratic or semi-cylindrical) cross sections of separation channels. In addition to the influence on the packing densities in packed channels, the corner regions are prone to form stagnant flow in channels to reduce separation efficiency. Improvements in fabricating chips, such as exploring new materials and constructing cylindrical cross section, require more attention to make the chip-based devices more rugged and cost-effective. The development of miniaturized MS detectors should also receive more attention from the scientific community. Simultaneously with the chip-based separation devices, the miniaturization and on-chip integration of the mass spectrometer could provide new application opportunities in food and environmental chemistry.

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