

Chapter 8

Strategies for the Design of Vitamin D Receptor Ligands



Tania R. Mutchie, Daniel A. Webb, Elliot S. Di Milo, and Leggy A. Arnold

Abstract Structure-activity relationship analysis is a powerful tool to elucidate the structural requirements for high-affinity vitamin D receptor (VDR) ligands. This chapter systematically interrogates the structural features of $1\alpha,25(\text{OH})_2\text{D}_3$, the vitamin D metabolite with the highest VDR affinity. It can be concluded that the C1 α and C25 hydroxyl groups of $1\alpha,25(\text{OH})_2\text{D}_3$ are very important for binding. Optimal spatial arrangement of both hydroxyl groups was achieved with either a hydrophobic semi-flexible secosteroid scaffold or a simplified, flexible carbon chain. Y-shaped ligands with high affinity confirmed a highly inducible VDR ligand-binding pocket, which has been visualized by X-ray crystallography. Substitution of the secosteroid scaffold by other hydrophobic spacers such as carboranes or aromatic ring systems has led to many non-secosteroid VDR ligands. Exploration of ligand substitution has led to the development of antagonists that are accommodated by the inducible VDR ligand-binding pocket but alter the overall conformation of VDR in ways that prevent interactions with coactivator proteins from occurring and ultimately result in reduced gene transcription.

Keywords Vitamin D · Vitamin D receptor · 1,25-Dihydroxyvitamin D₃ · Structure-activity relationship · Agonist · Antagonist · Coactivator

8.1 Introduction

The vitamin D receptor (VDR) takes a special place among nuclear receptors because the biosynthesis of its endogenous ligand, $1\alpha,25$ -dihydroxyvitamin D₃ ($1\alpha,25(\text{OH})_2\text{D}_3$), is dependent on sun exposure. Poor living conditions during the industrial revolution, when people were destined to work and live inside with

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minimal light exposure, caused bone deformations and skin diseases with symptoms that were alleviated by light therapy. Once irradiated food was found to have the same medicinal effect [1], isolation of vitamin D was rapidly accomplished [2]. Identification of the corresponding receptor turned out to be difficult; however, vitamin D does not bind VDR at physiological concentrations. Once radiolabeled vitamin D was generated [3], $25(\text{OH})\text{D}_3$ and $1\alpha,25(\text{OH})_2\text{D}_3$ were identified [4, 5], which in turn enabled the identification and cloning of VDR [6, 7]. The genomic function of VDR regulates genes involved in calcium homeostasis, cell proliferation, and cell differentiation. The endocrine receptor is expressed in the epithelia of the endocrine organs, digestive tract, kidneys, and thymus [8] but is also found in leukocytes and bone cells. VDR can be found in the cytosol or membrane-bound [9]. In the nucleus, VDR is liganded and binds DNA and the retinoid X receptor (RXR) [10]. VDR-specific gene promoter sequences have been identified [11]. The transcriptional complex includes, among other proteins, nuclear receptor coactivators and corepressors that interact with RNA polymerase II [12], changing chromatin packing and enabling specific gene transcription [13].

This chapter is not a complete review describing more than 3000 VDR ligands that have been reported since 1970. Therefore, I refer the reader to five chapters within the recent edition of *Vitamin D* and cited references, as well as excellent recent reviews [14–20]. This chapter highlights the relationship between molecular ligand structure and VDR affinity. Other downstream biological effects can be found in the references and the recent two volumes of *Vitamin D*. Most VDR ligands have been characterized by their ability to compete with tritium-labeled $1\alpha,25(\text{OH})_2\text{D}_3$. Transcription assays have been used to distinguish between agonists and antagonists. VDR coactivator recruitment has been studied with two-hybrid assays or biochemically using homogeneous time-resolved fluorescence (HTRF). Tertiary assays employed for the characterization of VDR ligands included, among others, cell differentiation, cell proliferation, cellular calcium uptake, intestinal calcium transport, and serum calcium changes. Herein, we predominately report VDR affinity independent of cell permeability, metabolic stability, plasma binding (vitamin D-binding protein), and other factors that change with the structure of a small molecule.

8.2 Secosteroid VDR Ligands

Calcitriol, or $1\alpha,25(\text{OH})_2\text{D}_3$, has the highest affinity for VDR among all vitamin D metabolites. The competitive VDR binding of $1\alpha,25(\text{OH})_2\text{D}_3$ using [^3H]- $1\alpha,25(\text{OH})_2\text{D}_3$ as a probe has been reported in the range of 0.04–0.16 nM with protein isolated from the tissue and cells or recombinantly expressed VDR as full-length receptor or ligand-binding domain [21, 22]. Other assays such as biochemical coactivator recruitment assays reported EC_{50} of 1.2 nM for $1\alpha,25(\text{OH})_2\text{D}_3$ [23]. For most cases, the affinities of new compounds in comparison to $1\alpha,25(\text{OH})_2\text{D}_3$ are reported as percent affinity in this chapter.

$1\alpha,25(\text{OH})_2\text{D}_3$ has six chiral centers and two trisubstituted double bonds that can adopt an E or Z configuration. First, we will compare VDR binding of $1\alpha,25(\text{OH})_2\text{D}_3$ epimers and stereoisomers depicted in Figs. 8.1, 8.2, 8.3, and 8.4.

$1\alpha,25(\text{OH})_2\text{D}_3$ is the metabolic product of vitamin D_3 , which lacks a hydroxyl group in the C1 and C25 positions (Fig. 8.1). 25-Hydroxyvitamin D-1 α hydroxylase, located primarily in the kidneys but also in other tissues, stereospecifically introduces a C1 α -hydroxyl group [24–27]. The 3-OH group is present in vitamin D_3 and its precursor 7-dehydrocholesterol. The evaluation of A-ring diastereomers of $1\alpha,25(\text{OH})_2\text{D}_3$ demonstrated that binding to VDR is more impacted by the stereochemistry of the C1-position than the C3-position [28]. VDR affinity for **2** was 24% in comparison to $1\alpha,25(\text{OH})_2\text{D}_3$ but only 0.2% and 0.8% for **3** and **4**, respectively. $1\alpha,25(\text{OH})_2$ -3-Epi- D_3 has been identified as a natural metabolite of $1\alpha,25(\text{OH})_2\text{D}_3$ [29] and was intensively studied in vivo. $1\beta,25(\text{OH})_2\text{D}_3$ was first synthesized in 1977 [30] and has also recently been identified as a natural metabolite of vitamin D [31]. $1\beta,25(\text{OH})_2\text{D}_3$ has been reported as a non-genomic antagonist of $1\alpha,25(\text{OH})_2\text{D}_3$ [28].

Secosteroids in contrast to steroids have a “broken” B-ring resulting in a triene system with 5(Z),7(E) configuration (Fig. 8.2). The formation of secosteroids occurs via a retro Diels-Alder reaction in the presence of light followed by a [1,7] sigmatropic rearrangement. In the skin, this conversion occurs with high stereoselectivity.

Isomer **5** retains a good affinity toward VDR, which is 13% in comparison to $1\alpha,25(\text{OH})_2\text{D}_3$ [32]. The E,E stereochemistry can be generated by light in the presence of iodine [33] or by a cheletropic addition-elimination with sulfur dioxide [34]. In contrast, isomers **6** and **7** were not observed for photochemical reactions but synthesized using a chromium(0)-mediated isomerization reaction with a vinylallene precursor [34]. The affinity toward VDR was 0.82% for **6** and 1.6% for **7** in comparison to $1\alpha,25(\text{OH})_2\text{D}_3$. Thus, the position of the A-ring with respect to the B- and C-ring is more important for VDR binding than the location of the terminal alkene.

The stereochemistry of the fused B,C-ring system of $1\alpha,25(\text{OH})_2\text{D}_3$ originates from 7-dehydrocholesterol. Interestingly, the configuration of the fused system has a direct influence on the equilibrium of the thermal [1,7] sigmatropic rearrangement

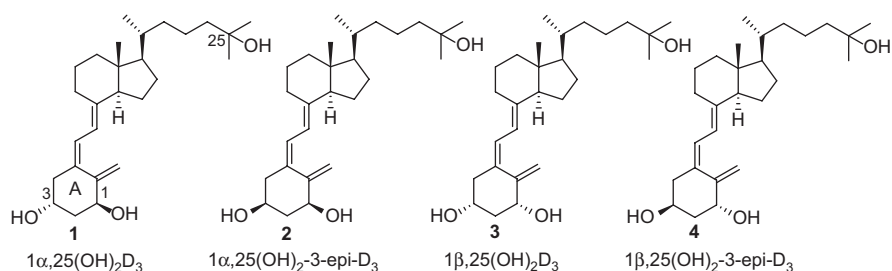


Fig. 8.1 A-ring diastereomers of $1\alpha,25(\text{OH})_2\text{D}_3$

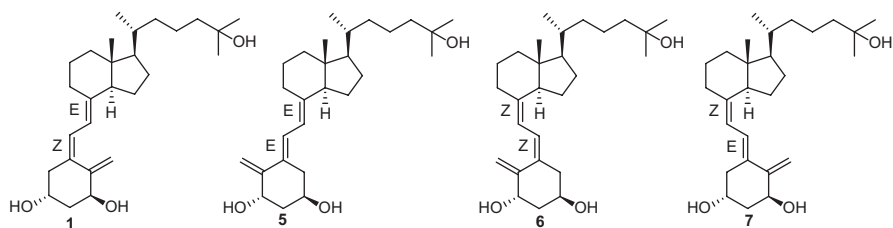


Fig. 8.2 Diene stereochemistry of $1\alpha,25(\text{OH})_2\text{D}_3$

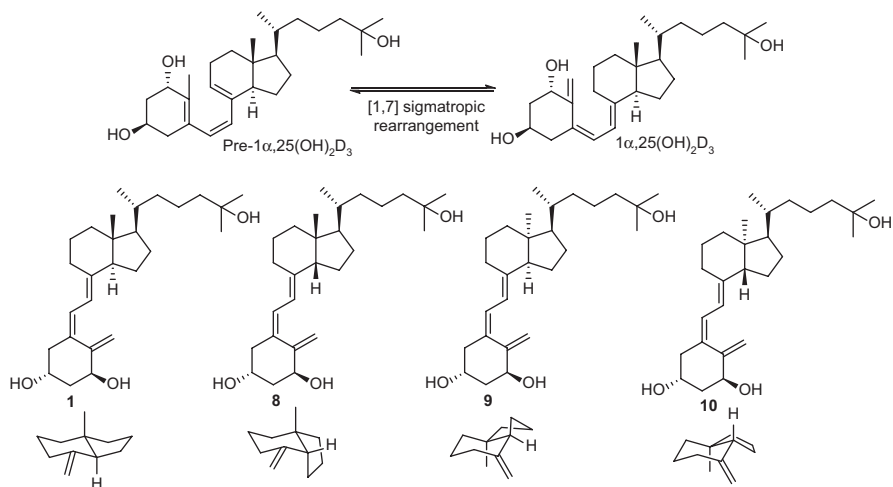
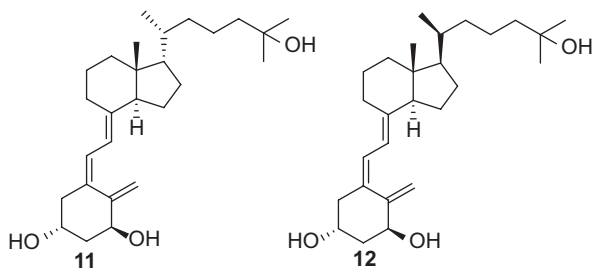


Fig. 8.3 Fused B,C-ring stereochemistry of $1\alpha,25(\text{OH})_2\text{D}_3$

Fig. 8.4 C17 and C20 epimers



reaction (Fig. 8.3). When $1\alpha,25(\text{OH})_2\text{D}_3$ was heated at $80\text{ }^\circ\text{C}$, only 12% of the pre- $1\alpha,25(\text{OH})_2\text{D}_3$ was detected [35]. However, when epimer 8 was heated at $80\text{ }^\circ\text{C}$, a 95:5 ratio in favor of the pre-structure was formed.

Pure 8 was synthesized by epimerization of Grundmann's ketone and retained a VDR affinity of 15% in comparison with $1\alpha,25(\text{OH})_2\text{D}_3$ [35]. Interestingly, no reports were found for compounds 9 and 10.

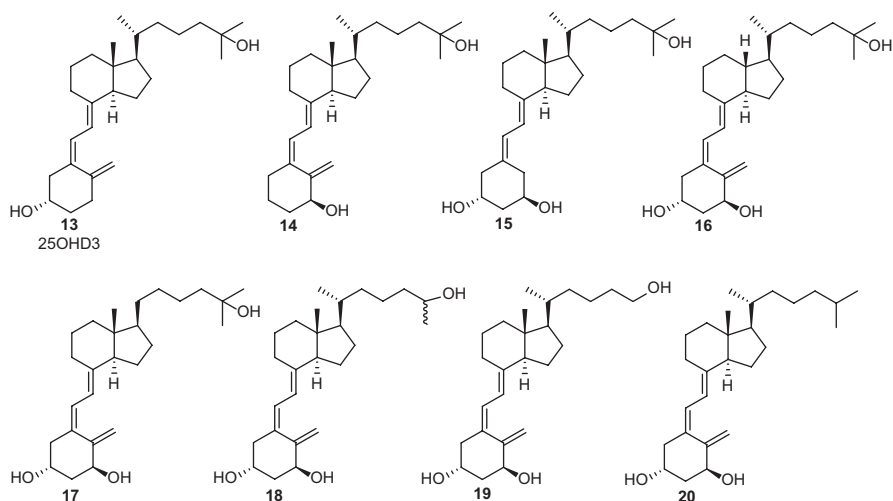


Fig. 8.5 $1\alpha,25(\text{OH})_2\text{D}_3$ analogs that lack certain structural elements

The synthesis of **11** has been reported [36]. Later, an improved route was developed but VDR binding was not reported [37]. However, inhibition of human breast cancer cell (MCF-7) proliferation was more pronounced in the presence of **11** than $1\alpha,25(\text{OH})_2\text{D}_3$. Epimer **12** demonstrated inhibition of T-cell proliferation at picomolar concentrations [38]. The VDR binding was 88% in comparison to $1\alpha,25(\text{OH})_2\text{D}_3$ [39].

Next, the importance of functional groups and substituents with respect to VDR binding is discussed. Analogs that lack certain structural elements are compared to $1\alpha,25(\text{OH})_2\text{D}_3$ (Fig. 8.5).

25OHD_3 is a metabolic product of vitamin D_3 and was identified in 1968 [4]. It is abundant in blood and used to determine the vitamin D status in humans [40]. The binding toward VDR is 900-fold less than $1\alpha,25(\text{OH})_2\text{D}_3$ [41]. In contrast to **13**, the affinity of **14** was only 1/8 less effective than $1\alpha,25(\text{OH})_2\text{D}_3$, making the 1α -OH group significantly more important for VDR binding than the 3-OH group [41]. Compound **15** lacking the methylene group was first synthesized in 1990 and has been shown to induce the differentiation of HL-60 cells at the same concentration as $1\alpha,25(\text{OH})_2\text{D}_3$ [42]. The VDR binding was 30% in comparison to $1\alpha,25(\text{OH})_2\text{D}_3$ [43]. Compound **16** was reported to be three times more potent than $1\alpha,25(\text{OH})_2\text{D}_3$ with respect to porcine VDR binding [44]. Thus the presence of C18 impaired VDR binding, contrasting with compound **17**, in which a lack of the C21 methyl reduced affinity toward chick VDR to 10% that of $1\alpha,25(\text{OH})_2\text{D}_3$ [45]. Interestingly, substitution of C20 by oxygen reduced VDR affinity to 0.1% in comparison to $1\alpha,25(\text{OH})_2\text{D}_3$ [45], emphasizing the importance of hydrophobicity for good receptor binding. Compound **18** with possible (R) and (S) configurations has not been reported. VDR binding of **19** has not been investigated; however, the ability to

differentiate HL-60 cells compared to $1\alpha,25(\text{OH})_2\text{D}_3$ was 1% at the same concentration [46]. For a similar molecule with a terminal alkene in the 2-position, a 1.9% VDR affinity was reported in comparison to the parent compound [47]. Compound **20**, also known as alfacalcidol, was first reported in 1973 [48, 49]. Alfacalcidol is converted to $1\alpha,25(\text{OH})_2\text{D}_3$ in vivo and, therefore, exhibits similar activities. The VDR affinity was 900-fold less than $1\alpha,25(\text{OH})_2\text{D}_3$ [41]. Thus, it can be concluded that hydroxyl groups in the C1 α and C25 positions are the most important substituents to promote VDR binding.

Further investigations into the significance of the bicyclic structure of $1\alpha,25(\text{OH})_2\text{D}_3$ with respect to VDR binding is represented by compounds depicted in Fig. 8.6.

Removal of the five-membered ring of $1\alpha,25(\text{OH})_2\text{D}_3$ was investigated with **21** and analogs thereof [50]. The relative stereochemistry of C17 marginally influenced VDR binding; however, large differences between these epimers were observed for anti-proliferation and calcium homeostasis. VDR binding of **21** was 60% in comparison to $1\alpha,25(\text{OH})_2\text{D}_3$. The VDR affinity of the C20 epimer of **21** was 70%. The same report characterized compounds like **22** with a VDR affinity of 80% in comparison to $1\alpha,25(\text{OH})_2\text{D}_3$. The synthesis of **23** was reported but VDR binding was not determined [51]. However, **24** with the terminal alkene moved from position 10 to position 2, exhibited 80 times lower affinity toward VDR than $1\alpha,25(\text{OH})_2\text{D}_3$ [21].

Overall, it can be concluded that structural changes to the hydrophobic core of $1\alpha,25(\text{OH})_2\text{D}_3$ can still result in high-affinity ligands for VDR. The ligand-binding domain of VDR consists of 12 helices when bound to $1\alpha,25(\text{OH})_2\text{D}_3$. The most essential features of $1\alpha,25(\text{OH})_2\text{D}_3$ are the C1 α and C25 hydroxyl groups, which have been shown to form canonical hydrogen bonds with VDR (Fig. 8.7).

The interaction of 25-OH with His305 (loop H6-H7) and His 397 (H11) is vital to the conformational change VDR undergoes when interacting with coregulator proteins. 1-OH interacts with Ser237 (H3) and Arg274 (H5), anchoring the ligand in the binding pocket. 3-OH interacts with Ser278 (H5) and Tyr143 (loop H1-H2), but compounds like **2** and **14** have shown that these contacts merely provide further stabilization to the complex. $1\alpha,25(\text{OH})_2\text{D}_3$ only fills 56% of the VDR-binding pocket, which helps explain the large variety of high-affinity ligands that have been developed for VDR. Important, however, is the spacing and orientation of the

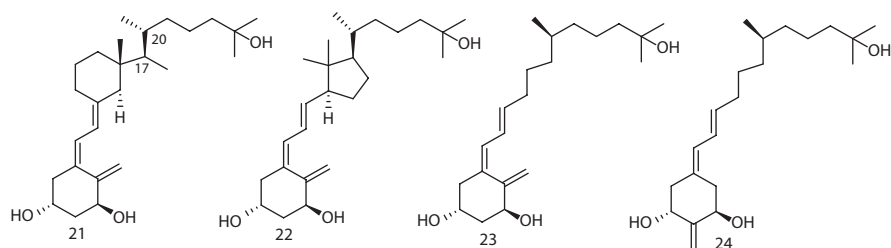


Fig. 8.6 VDR ligands without a fused ring system

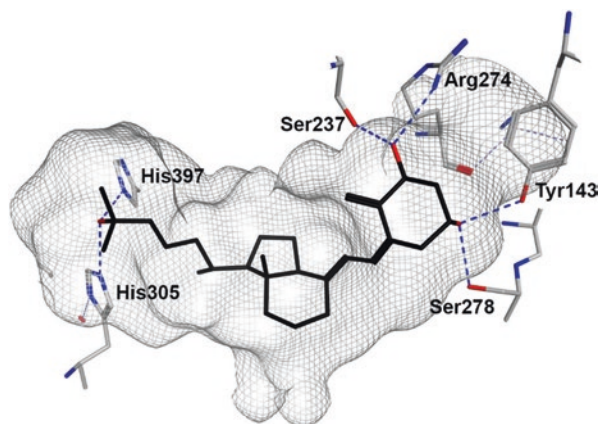


Fig. 8.7 Crystal structure of $1\alpha,25(\text{OH})_2\text{D}_3$ bound to human VDR [PDB ID:1DB1] [52]

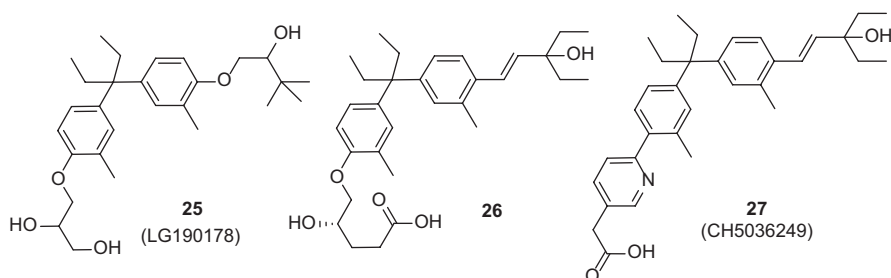


Fig. 8.8 VDR ligands with a diarylmethane moiety

hydroxyl groups, which are supported by ring structures, a diene moiety, and chiral carbon centers. The majority of the central VDR ligand pocket surface is hydrophobic and assembled by leucine, isoleucine, and valine side chains.

8.3 Non-secosteroid VDR Ligands

The first non-secosteroid ligands with a diarylmethane moiety were reported by Ligand Pharmaceuticals (Fig. 8.8). The quaternary carbon center bearing two ethyl substituents was superior to other alkyl substituents, and aligned well with the fused ring system of VDR-bound $1\alpha,25(\text{OH})_2\text{D}_3$ [53].

The racemic mixture of LG190178 exhibited a VDR affinity of 0.3% in comparison to $1\alpha,25(\text{OH})_2\text{D}_3$ [54]. The synthesis of individual LG190178 stereoisomers identified the (2*S*, 2'*R*) isomer as the most active compound with a 28.3% VDR affinity in comparison to $1\alpha,25(\text{OH})_2\text{D}_3$ [55]. The systematic development of these ligands resulted in **26**, which was equally as active as $1\alpha,25(\text{OH})_2\text{D}_3$ in a cell-based transcription assay [56]. An analog of **26**, which replaced the ethyl groups adjacent

to the tertiary alcohol with trifluoromethyl groups, showed a fivefold improvement in the transcription assay [57]. In recent years, many similar compounds were developed with thiophene, pyrrol [58], and other heterocycles, though compound **27**, which contains a pyridine substituent, exhibited the highest affinity toward VDR at 37% of the VDR- $1\alpha,25(\text{OH})_2\text{D}_3$ interaction [59].

Another successful approach for non-secosteroidal VDR ligand design was the incorporation of a dicarba-closo-dodecaborane as a hydrophobic moiety, rather than the fused ring system of natural VDR ligands (Fig. 8.9).

The development of carborane-based VDR ligands with different side chains resulted in **28**, which exhibited an affinity of 640 nM (IC_{50}) [60]. The (R) isomer was one-fifth as active. Subsequent research identified compound **29** being twice as potent as **28** in a HL-60 differentiation assay [61].

Other approaches for VDR ligands included the incorporation of aromatic ring structures, which are absent from natural VDR ligand, $1\alpha,25(\text{OH})_2\text{D}_3$. The earliest examples were identified from a library of bis-aromatic compounds (Fig. 8.10).

CD4528 was characterized with a CYP24A1 reporter assay demonstrating an EC_{50} of 1.7 nM [62]. For the same assay, the EC_{50} of $1\alpha,25(\text{OH})_2\text{D}_3$ was 1.0 nM. Other related VDR ligands exhibited similar low nanomolar activities, for example, CD4849 (0.5 nM) [63]. A recent study employing the A-ring structure of $1\alpha,25(\text{OH})_2\text{D}_3$ resulted in **32**, which displayed a VDR affinity 24% that of $1\alpha,25(\text{OH})_2\text{D}_3$ [64]. Less active was **33**, exhibiting a 0.01% VDR affinity in comparison to $1\alpha,25(\text{OH})_2\text{D}_3$ [65]. Other nutritional ligands with low VDR affinities were reported by Haussler et al. [66].

VDR is highly expressed in the intestine and has been described as a bile acid sensor due to its ability to bind lithocholic acid (Fig. 8.11) [67].

The affinity of VDR for lithocholic acid was less than 0.005% in comparison to $1\alpha,25(\text{OH})_2\text{D}_3$. The corresponding acetate was more potent with a 0.01% VDR affinity in comparison to $1\alpha,25(\text{OH})_2\text{D}_3$ [68]. A later VDR-binding study reported an IC_{50} of 30 μM for both **35** and **36** [69]. Recently, a methylsulfonate analog of lithocholic acid showed an IC_{50} of 1.2 μM [70]. For estrone analog **38**, an EC_{50} of 850 nM was reported in a VDR transactivation assay [71].

The first Y-shaped VDR ligand called Gemini was introduced in 2000 by Norman et al. [72] (Fig. 8.12).

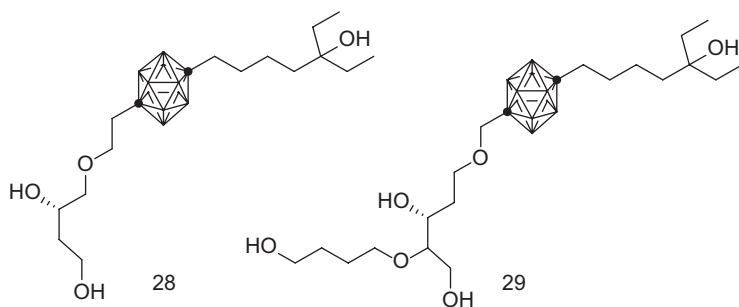


Fig. 8.9 VDR ligands containing a carborane structure

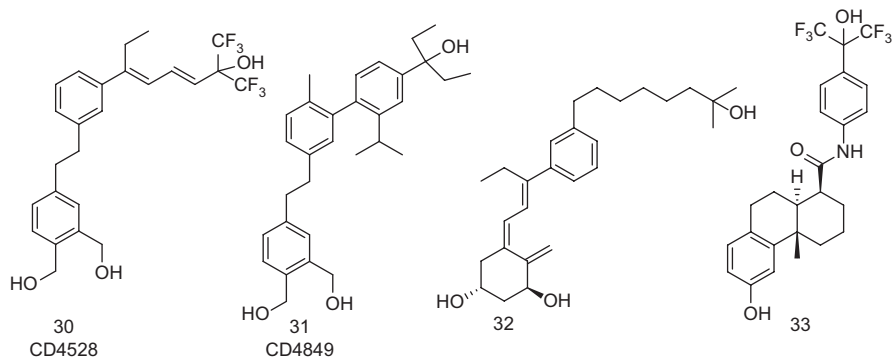


Fig. 8.10 Bis- and tris-aromatic VDR ligands

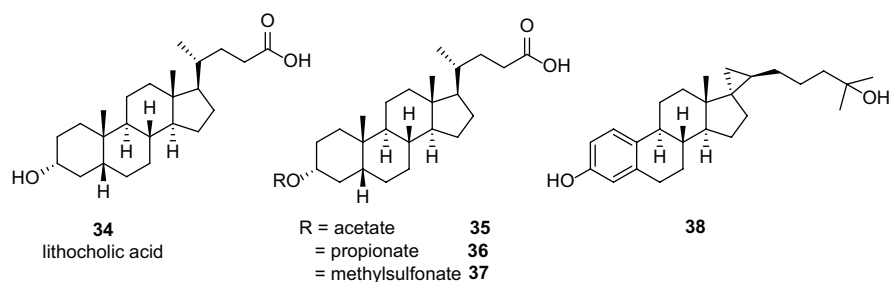


Fig. 8.11 Steroid VDR ligands

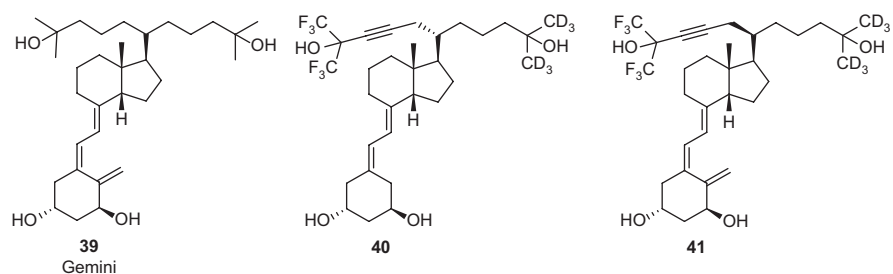


Fig. 8.12 Gemini ligands

Gemini exhibited a VDR affinity of 38% in comparison to $1\alpha,25(\text{OH})_2\text{D}_3$. Based on an available crystal structure of VDR bound to $1\alpha,25(\text{OH})_2\text{D}_3$, it was hypothesized that VDR might accommodate this second side chain in a so-called A pocket (alternative pocket) [73]. The later reported crystal structure of VDR bound to Gemini confirmed the adaptability of VDR to accommodate Y-shaped ligands [74]. Further developments resulted in **40** and its C20 epimer, **41**, which were 36- and 22-fold more active in a gene reporter assay than $1\alpha,25(\text{OH})_2\text{D}_3$ [75].

8.4 VDR Antagonists

Throughout the last two decades, many different antagonists with strong VDR affinities have been reported [76]. The earliest disclosed antagonists were derivatives from the natural-occurring vitamin D metabolite, (23S,25R)- $1\alpha,25(\text{OH})_2\text{D}_3$ -26,23-lactone [77], such as compound TEI-9647 and its C23 epimer TEI-9648 (Fig. 8.13).

The VDR affinity of TEI-9647 was 10% compared to $1\alpha,25(\text{OH})_2\text{D}_3$. Its C23 epimer TEI-9648 bound VDR with a 8% affinity [78]. Further research confirmed that these unsaturated esters underwent a conjugate addition reaction with cysteine residues in the human VDR ligand pocket [79]. Methyl substitutions at positions 2 and 24 (**43**) significantly increased VDR affinity (63% in comparison to $1\alpha,25(\text{OH})_2\text{D}_3$) [80]. The introduction of a cyclopropyl group on the lactone ring resulted in **44**, which surpassed the VDR affinity of $1\alpha,25(\text{OH})_2\text{D}_3$ (166%) [81].

Based on the structure of calcipotriol (**45**), an approved treatment for psoriasis and a high-affinity VDR ligand, other related ligands were able to be synthesized by Schering (Fig. 8.14).

In contrast to agonist calcipotriol, ZK159222 and ZK168281 were poor inducers of VDR transcription and reduced $1\alpha,25(\text{OH})_2\text{D}_3$ -mediated transcription [82]. For ZK191784, a VDR affinity of 33% in comparison to $1,25(\text{OH})_2\text{D}_3$ was reported. [83] Thus, the VDR affinities of these ligands are strong but induce an antagonistic VDR conformation. Additionally, these antagonists were investigated in vivo and demonstrated promising anti-inflammatory properties [76].

Amide-based VDR antagonists that were inspired by calcitriol lactone were introduced in 2004 (Fig. 8.15).

The corresponding lactam **49** exhibited a VDR affinity of 10.2 nM in comparison to $1\alpha,25(\text{OH})_2\text{D}_3$ with 0.5 nM [84]. The measured VDR binding of **50** was 1.9 nM (IC_{50}). Importantly, **50** inhibited VDR-mediated transcription at nanomolar concentrations without showing any agonist activity in the absence of $1\alpha,25(\text{OH})_2\text{D}_3$ [85].

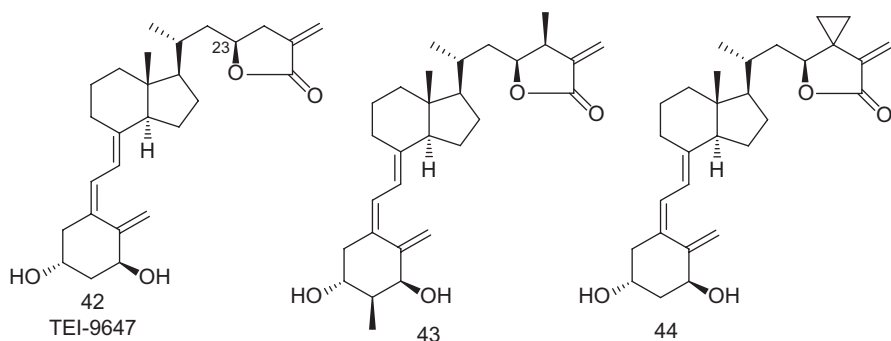


Fig. 8.13 Electrophilic VDR ligands

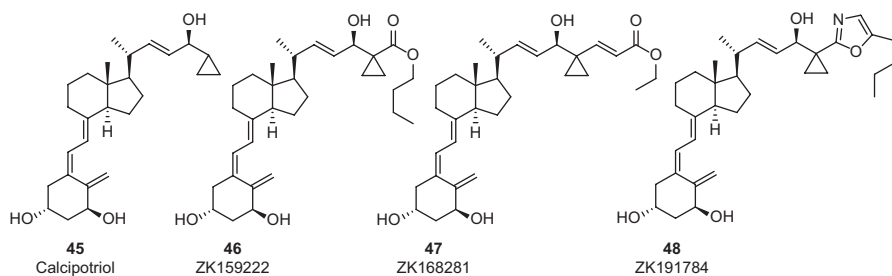


Fig. 8.14 ZK series of VDR antagonists

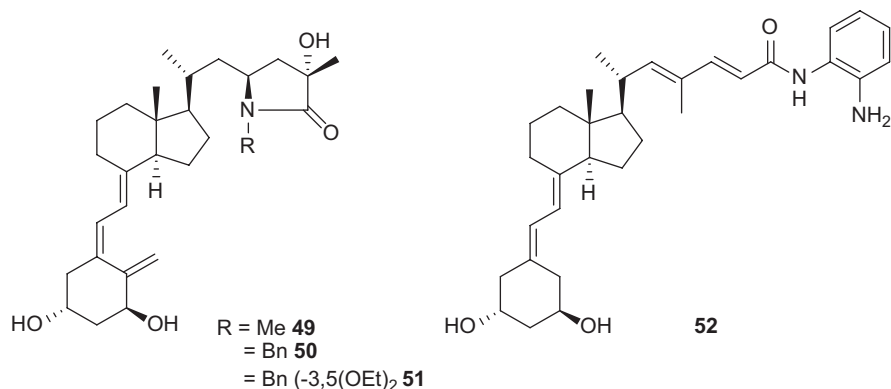


Fig. 8.15 Amide series of VDR antagonists

Further improvement was achieved with **51**, which exhibited 52% VDR-binding affinity compared to $1\alpha,25(\text{OH})_2\text{D}_3$ and inhibited $1\alpha,25(\text{OH})_2\text{D}_3$ -mediated transcription with an IC_{50} of 90 nM [86]. For ortho-aniline compound **52**, an IC_{50} of 107 nM was reported [87].

Introduction of a bulky adamantane group was conceived as another approach to change the conformation of VDR (Fig. 8.16).

Compound **53** exhibited 2% VDR affinity compared to $1\alpha,25(\text{OH})_2\text{D}_3$ [88]. In the presence of $1\alpha,25(\text{OH})_2\text{D}_3$, VDR-mediated transcription was inhibited at 100 nM. Among a series of diastereomeric analogs, **54** exhibited the highest affinity toward VDR with 17% affinity in comparison to $1\alpha,25(\text{OH})_2\text{D}_3$ [89]. Further improvements resulted in ligands with an internal alkyne named ADTK1-4 [90]. The compound with the highest VDR affinity among this group was **55** reaching 90% of the VDR- $1\alpha,25(\text{OH})_2\text{D}_3$ interaction. This compound behaved as a partial agonist. Finally, a library of VDR ligands with a diyne system were synthesized and evaluated, achieving a 7% VDR affinity in comparison to $1\alpha,25(\text{OH})_2\text{D}_3$ [91].

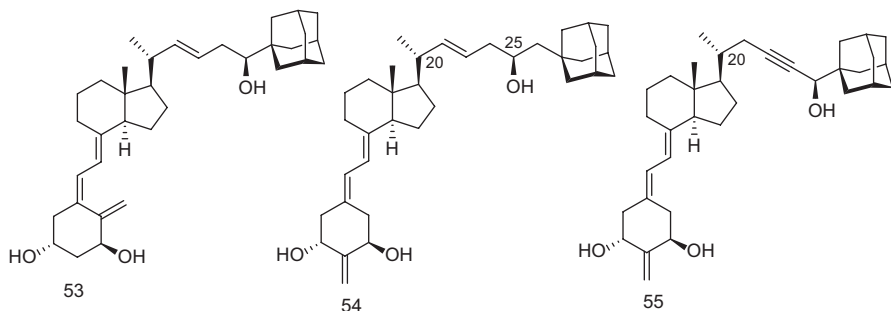


Fig. 8.16 Adamantanyl-derived VDR antagonists

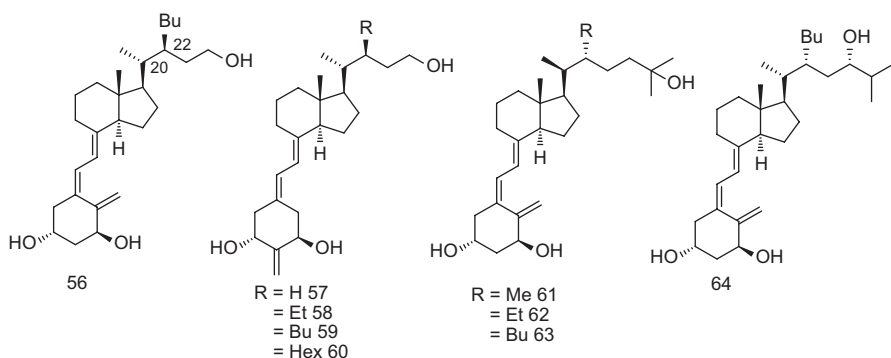


Fig. 8.17 VDR ligands with C22 substitution

A series of Y-shaped VDR ligands that were inspired by Gemini were developed and, dependent upon their substitution pattern, were found to be antagonists, partial agonists, or superagonists (Fig. 8.17).

Among a series of diastereomers with a butyl substituent in the C22 position, **56** was identified as a VDR antagonist [92]. The VDR affinity was 4.1% in comparison to $1\alpha,25\text{-(OH)}_2\text{D}_3$ [92]. Interestingly, the C20 epimer of **56** was identified as an agonist with a 2.5% affinity toward VDR in comparison to $1\alpha,25\text{(OH)}_2\text{D}_3$. The influence of the alkyl chain length with respect to VDR binding was investigated for very similar compounds containing a methylene group in the C2 position (**57–60**) [47, 93]. The presence of a butyl substituent resulted in antagonist **59**, which had a VDR affinity of 61% in comparison to $1\alpha,25\text{(OH)}_2\text{D}_3$. Partial agonists **57**, **58**, and **60** exhibited lower VDR affinity. Further structural changes to compound **59** included the introduction of two methyl substituents in the C24 position or elongation of the hydroxyl-bearing carbon chain by one carbon. Both of these structural changes also resulted in antagonistic ligands; however, implementing the carbon chain inherent to $1\alpha,25\text{(OH)}_2\text{D}_3$ resulted in a superagonist with higher VDR affinity than $1\alpha,25\text{(OH)}_2\text{D}_3$ [47]. Other superagonists were produced with the introduction of C22 substituents for 20-epi- $1\alpha,25\text{(OH)}_2\text{D}_3$ (**12**) [94]. Compound **63** exhibited

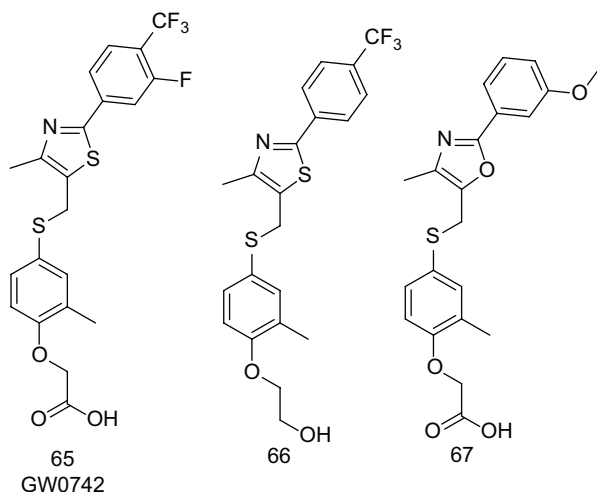


Fig. 8.18 GW0742-based VDR ligands

the highest affinity for VDR (797%) followed by **62** and **61**. Finally, antagonists were produced with an isopropyl group at C24 (**64**) [92]. Among different diastereomers, **64** exhibited the highest VDR affinity of 1.4% in comparison to $1\alpha,25(\text{OH})_2\text{D}_3$.

A high-throughput screen of 390,000 compounds identified PPAR δ agonist GW0742 as a novel VDR antagonist (Fig. 8.18) [95].

The VDR affinity of **65** was 8.7 μM . The structural change of the acid function into an hydroxyl group resulted in partial VDR agonist **66** with an EC_{50} of 120 nM [96]. The evaluation of a library of compounds related to **65** demonstrated that CF_3 substituents in the meta- and ortho-position reduced the affinity toward PPAR δ without influencing the affinity for VDR [97]. Recently, VDR antagonist **67** was reported with activity of 660 nM [98]. This compound did not bind PPAR δ . Virtual screening of known nuclear receptor ligands for application in VDR binding identified several compounds as possible candidates. VDR binding was demonstrated for several compounds, including H6036 [99].

8.5 Concluding Remarks and Future Directions

Novel VDR ligand design and synthesis is still a very active research area, with many international research groups working together to develop new drug candidates for disorders caused by vitamin D deficiency, cancer, and inflammatory diseases. Recently discovered VDR ligands are being investigated in clinical trials, reflecting the need for new medications in the respective disease areas. A great number of ligands have been elucidated in complex with VDR using X-ray crystallography. The structural information has guided new ligand design while also

demonstrating that the VDR ligand pocket is amendable to very different ligand shapes. However, as highlighted in this chapter, hydrogen bonding on opposite ends of the ligand pocket is essential for high VDR affinity. Another important feature is distinct spacing of these groups by a flexible hydrophobic spacer. Recently, endogenous ligands for VDR such as lithocholic acid and fatty acids have been identified, although their biological function is still unclear. Furthermore, new vitamin D metabolites have been identified in the last few decades, offering new areas of research in the field of vitamin D.

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