Chapter 11 Physiological, Biochemical, and Molecular Mechanism of Nitric Oxide-Mediated Abiotic Stress Tolerance

Khushwant Singh, Ila Shukla, Ajay Kumar Tiwari, and Lubna Azmi

11.1 Introduction

Nitric oxide is an omnipresent molecule and an extremely diffusible gas (Siddiqui et al. [2011](#page-20-0)). It has drawn much attention in recent years due to its widespread application in processes linked to normal plant physiology. Due to these properties, it has also been termed as the "molecule of the year." It has been synthesized by plant itself, and it also acts as a gaseous plant growth regulator (Neill et al. [2008\)](#page-18-0). It has been given the name "plant growth regulator" because there are rising evidence reporting its role as mediator of several biological procedures involved in the growth and development of the plant. Seed germination, root organogenesis, photosynthesis, stomatal closure, hypocotyl growth, foral regulation, defense against pathogens, and senescence are the processes included in plant physiology where NO has an extensive role. It has also been reported that nitric oxide is also linked to phytoalexin production and apoptosis (Delledonne et al. [1998\)](#page-16-0). Nitric oxide suppresses growth at higher concentrations, while it provides benefts at lower concentrations. This review puts light on the role of nitric oxide in plants for tolerance of different abiotic stresses (Simontacchi et al. [2013;](#page-20-1) Wendehenne et al. [2001](#page-21-0)). Also, the relation between nitric oxide and other phytohormones has been included here. Many

K. Singh

I. Shukla

A. K. Tiwari Postgraduate Department of Dravyaguna, State Ayurvedic College and Hospital, Lucknow, India

L. Azmi (\boxtimes) Department of Chemistry, University of Lucknow, Lucknow, India

© The Author(s), under exclusive license to Springer Nature 217 Switzerland AG 2021 T. Aftab, K. R. Hakeem (eds.), *Plant Growth Regulators*, [https://doi.org/10.1007/978-3-030-61153-8_11](https://doi.org/10.1007/978-3-030-61153-8_11#DOI)

Department of Biochemistry, University of Lucknow, Lucknow, India

Department of Pharmacology, CSIR-National Botanical Research Institute, Lucknow, Uttar Pradesh, India

recent reports have been found about molecular data, regarding NO-mediated PTMs in animals. This has encouraged botanists to fnd out the effect of these PTMs in plants when exposed to different stresses. Hence, in the last few years, a great deal of effort were put into the identifcation of the role of S-nitrosylated, tyrosine nitrate, and S-glutathionylated proteins in plant function modulation (Wendehenne et al. [2001\)](#page-21-0). Also, current knowledge reports about the role of nitric oxide in cell signaling in plants by involvement of oxidative and reductive pathways (Crawford [2006\)](#page-16-1). This review also attempts to impart knowledge on the N-end rule pathway responsible for protein stability, which alters the steadiness of group VII ethylene response factors under hypoxic conditions.

11.2 Antiquity of Nitric Oxide

In 1722, when Joseph Priestly frst described nitric oxide, it was known as a highly toxic gas and also a component of industrial wastes and exhaust gas. The model concerning the cytotoxicity of free radical substances was changed with the 1980s' discovery about the role of nitric oxide signaling in regulating the cardiovascular system by R.F. Furchgott, L.J. Ignarro, and F. Murad (Nobel Prize winners in Physiology and Medicine 1998) (Delledonne et al. [1998\)](#page-16-0). In 1992, nitric oxide was declared as the Molecule of the Year by the journal *Science*. In that year only, a nitric oxide society was founded, and a scientifc journal devoted to nitric oxide was formed. Nitric oxide is a gaseous free radical, and its emission was reported in soybean many years back. Later sunfower and maize were reported to have in vivo and in vitro nitrate reductase (NR)-dependent nitric oxide producing activity (Klepper [1979\)](#page-18-1). Animals possess nitric oxide synthase, and it acts as chief enzyme that catalyzes the in vivo synthesis of nitric oxide. But there is no report of plants having nitric oxide synthase, though it has been proven to be a functional metabolite in plants (Qi and Dong [1999\)](#page-19-0).

11.3 Nitric Oxide Synthesis in Plants

Involvement of NO in various signaling pathways related to plant's growth and development and biotic and abiotic stress resistance requires quick generation of NO in a site-specifc manner. Production of NO in plants has been known to occur through various routes which are broadly classifed into enzymatic and nonenzymatic mechanism. Enzymatic route of NO production involves three enzymes, including nitrate reductase, nitrate reductase and animals like nitric oxide synthase (NOS), for the production NO in plants. All routes of NO production in plants are shown in Fig. [11.1](#page-2-0).

Fig. 11.1 Various routes of NO production in plants

11.3.1 Nitrate Reductase

Nitrate reductase is a homodimeric protein, containing two subunits of 100 kD. Each subunit constitutes a favin adenine dinucleotide (FAD), a heme protein containing Fe as a metal cofactor, and a Mo-molybdopterin (Mo-MPT). Nitrate reductase facilitates the frst step of nitrogen assimilation in higher plants and is reported to exist dominantly in two forms which are NADH-specifc NR forms (EC 1.6.6.1) and NAD(P)H-bispecifc forms (EC 1.6.6.2). Reaction catalyzes by nitrate reductase is:

$NO₂⁻+NADH \rightarrow NO₂⁻+NAD⁺ + OH⁻$

While catalyzing the reduction of nitrate, FAD, Fe, and Mo present in nitrate reductase are cyclically reduced and oxidized. Due to the ability of FAD and Mo in Mo-molybdopterin to attain 3 oxidation states and 2 oxidation states of Fe in heme in collectively results in approximately 12 to 18 oxidized and reduced states of nitrate reductase which are transient in nature and occur in vivo. Under some circumstance, nitrate reductase may also result in the formation of NO (Dean and Harper [1988](#page-16-2); Rockel et al. [2002](#page-19-1)). In *Arabidopsis thaliana*, nitrate reductase was reported to be present in two isoforms, NIA1 and NIA2. Both isoforms were 83.5% identical to each other in terms of their amino acid constituent, but they had some independent sequence regions including the N-terminal region. Nitrate reductase encoded by NIA1 was demonstrated to control NO emission through stomata guard cells during light to dark transition. In a study on double mutant nia1 and nia2, involvement of nitrate reductase in ABA-induced generation of NO was reported in guard cells (Desikan et al. [2002](#page-16-3)). Involvement of NO in signaling requires its rapid generation during stimulus, and thus a tight control is required on nitrate reductase to show its activity. In a study on *Arabidopsis*, modulation of nitrate reductase activity was demonstrated by phosphorylation of a conserved serine residue (Ser 534)

present in it (Rockel et al. [2002\)](#page-19-1). Therefore, removal of the phospho-regulation through substitution of the Ser 534 can results in increased NO production by. In another study, the role of 14-3-3 proteins was highlighted in controlling the activity of nitrate reductase (Lillo et al. [2004](#page-18-2)) probably through its proteolysis (Weiner and Kaiser [1999](#page-21-1)).

11.3.2 Nitric Oxide Synthase (NOS)

Similar to NOS activity in animals, L-arginine-dependent production of NO has been reported in many plants, which was confrmed by inhibition of NO production with the use of arginine analogues. These inhibitors were also observed to interfere in NO mediated, ABA control over stomata closure. A few examples of the NOStype activity are shown in Table. [11.1](#page-3-0).

Besides the documented NOS-like activity in many plants, no protein homologs to NOS of animals have been yet confrmed in any plant species. In an immunological study, cross-reactivity of anti-mammalian NOS antibodies toward a few plant proteins was demonstrated by Ribeiro et al. [\(1999](#page-19-2)). This made researchers to believe that NOS activity in plants is mediated through proteins orthologous to NOS of animals, until a proteomic study was performed in which cross-reacting proteins were confrmed to be heat shock proteins and glycolytic enzymes (Butt et al. [2003\)](#page-16-4). In the same year, a protein (AtNOS1) was identifed in *Arabidopsis* reported to have 16% similarity with a 60 kD protein present in neuron cells of *Helix pomatia* (snail). This 60 kD protein was demonstrated to have cross-reactivity to anti-human nNOS antibody and was confrmed for its involvement in NO production in snail's neuron cells. Thus, to fnd NOS activity of AtNOS1, a mutant *Arabidopsis* was generated through T-DNA insertion technology, and results of this experiment highlighted the central role of AtNOS1 in NO production in *Arabidopsis*. Unlike in mammals, NOS activity in plants did not required BH4, FAD, FMN, or heme as cofactors. Rather,

Plant species	Tissue	References
Pisum sativum	Leaves	Leshem and Haramaty (1996)
Pisum sativum	Leaf peroxisomes	Barroso et al. (1999)
Pisum sativum	Embryonic axes	Sen and Cheema (1995)
Triticum aestivum	Germ	Kuo et al. (1995)
Nicotiana tabacum	Leaf epidermal cells	Foissner et al., (2000)
Nicotiana tabacum	TMV-infected leaves	Durner et al. (1998)
Mucuna hasjoo	Total extract	Ninnemann and Maier (1996)
Glycine max	Cotyledons	Modolo et al. (2002)
Lupinus albus	Roots and nodules	Cueto et al. (1996)
Zea mays	Root tips and young leaves	Ribeiro et al. (1999)

Table 11.1 Nitric oxide synthase activity observed in different plant species

amino acid sequence analysis revealed NOS of plants were found similar to proteins with GTP-binding or GTPase activity present in bacteria (Guo et al. [2003](#page-17-2)).

11.3.3 Nitrite Reductase

In plants, a plasma membrane-bound nitrate-NO oxidoreductase (Ni-NOR) has been reported in roots (Stöhr and Stremlau [2006\)](#page-20-2). This enzyme was reported to utilize cytochrome-c as an electron donor to produce NO under in vitro conditions. Due to lesser studies on Ni-NOR, it was not yet fully identifed and characterized and needs to be studied thoroughly for its role in the production of nitric oxide.

11.3.4 Non-enzymatic Production of NO in Plants

Apart from the enzymatic production of NO in plants, it is also produced though non-enzymatic mechanisms. For instance, oxidation of $N₂O$ during the nitrificationdenitrifcation cycle results in production of NO as a by-product and is released into the atmosphere (Wojtaszek [2000](#page-21-2)). It was demonstrated that local acidic condition in the chloroplast and apoplastic spaces where ascorbate is present in abundance promotes chemical reduction of nitrate and produces NO and dehydroascorbic acid (Henry et al. [1997\)](#page-17-3). Another example of reduction of nitrite by ascorbate under acidic pH has been demonstrated in aleurone cells of barley (Beligni et al. [2002\)](#page-15-1). Direct dismutation of nitrate into nitrite or NO under acidic environment is also feasible (StÖhr [2002\)](#page-20-3). In 1994, Cooney et al. reported involvement of carotenoids in the production of NO by reduction of $NO₂$ in a light-dependent manner. In addition to this biotic as well as abiotic stress conditions, interaction between nitrogen species and reactive oxygen also results in the formation of NO.

11.4 Signaling Pathways of Nitric Oxide

Since the orientation of the present chapter is toward the plant kingdom, thus all details further provided in this chapter are related to plants. To act as signaling molecule, it should have the ability of rapid generation during a need to relay the signal quickly to the target protein and shall be rapidly removed when there is no need of the signal. Uncharged free radical entity with small half-life led to identifcation of NO as a signaling molecule in plants as well as in animals. Also free radical nature of NO favors its existence in three inter-convertible species which, namely, are the radical (NO[•]), nitrosonium cation (NO⁺), and the nitroxyl radical (NO⁻) (Stamler et al. [1992;](#page-20-4) Wojtaszek [2000\)](#page-21-2). NO is a gaseous molecule with slight solubility in water as well as lipids. This property of NO allows it to diffuse from one cell to

another while travelling through their cell membranes and can also perpetuate signals freely within a cell diffusing through cytoplasm. NO is known to convey the signal through various routes, all of which are displayed diagrammatically in Fig. [11.2.](#page-5-0)

A previous study on *Nicotiana tabacum* demonstrated that NO triggers various cellular events by increasing cytoplasmic calcium ion concentration (Besson-bard et al. [2008\)](#page-15-2). Several other studies displayed similar increase in calcium ion concentration, which was mediated due to NO in the hyperosmotic, abscisic acid, and elicitor transduction pathways (Lamotte et al. [2006](#page-18-6); Garcia-Mata et al. [2003](#page-17-4); Vandelle et al. [2006;](#page-20-5) Gould et al. [2003\)](#page-17-5). Many other studies have also highlighted the vital role of NO in regulating activities of nearly all cellular channels involved in calcium transportation (Clementi [1998\)](#page-16-6). NO is known to regulate activity of calcium channels either directly by causing post-translational modifcations in them, e.g., S-nitrosylation (Stamler et al. [2001](#page-20-6); Ahern et al. [2002](#page-15-3)), or indirectly by activating guanylate cyclase and/or cyclic ADP-ribose (cADPR) resulting in synthesis of cGMP from ADP-ribosyl cyclase and NAD+. cGMP acts as a calcium-mobilizing metabolite and therefore helps in relaying signals (Hanafy et al. [2001](#page-17-6)). In many studies, presence of plasma membrane and intracellular calcium channel inhibitors resulted in inhibition of NO-mediated accumulation of calcium in the cell cytoplasm. This confrmed the involvement of NO in infux of calcium ions from extracellular and intracellular calcium stores (Lamotte et al. [2006;](#page-18-6) Gould et al. [2003;](#page-17-5) Vandelle et al. [2006](#page-20-5); Garcia-Mata et al. [2003\)](#page-17-4). Besides, ryanodine receptors are the only calcium channels identifed to be involved in NO-mediated mobilization of calcium ion from intracellular calcium reserves (Allen et al. [1995](#page-15-4); Fliegert et al. [2007\)](#page-17-7). This increased concentration of calcium ion in the cytoplasm modulates

Fig. 11.2 Diagram displaying various routes of signal transduction by NO in plants

activity of calcium-dependent protein kinases (CDPKs), mitogen-activated protein kinases (MAPKs), and calcium-sensitive channels like $Na⁺$ and $K⁺$ channels. NO-mediated induction of protein kinases has been demonstrated in many plants including *Nicotiana tabacum*, *Vicia faba*, *N. plumbaginifolia*, etc. (Sokolovski et al. [2005;](#page-20-7) Lamotte et al. [2006](#page-18-6)). In a study on *Nicotiana tabacum*, NO was displayed to induce two protein kinases in it, one of which was identifed as NtOSAK (*Nicotiana tabacum* Osmotic-Stress-Activated protein Kinase), a 42 kD protein kinase characterized for its activation under abiotic stress conditions such as hyperosmolarity and high salinity (Lamotte et al. [2006\)](#page-18-6). The second protein kinase which was induced by NO was a 48 kD protein and was identifed to be to MAPK (Zhang and Liu [2001\)](#page-21-3). NO and its derivatives are also known to perform their biological actions by modulating activities of various target proteins through three different types of posttranslational modifcation, which includes metal nitrosylation, S-nitrosylation, and Tyr-nitration.

11.4.1 Metal Nitrosylation

Plants contain many types of metalloproteins, including hemoglobin (Hb) which has been an important choice of researchers to study its interactions with NO. Most of these studies were focused on the three types of hemoglobin proteins such as leghemoglobin (a symbiotic hemoglobin present in root nodules of leguminous crops), two classes of non-symbiotic hemoglobin (differing in their affnity for oxygen), and truncated hemoglobin. Pioneering studies on leghemoglobin present in root nodules of alfalfa, cowpea, and soybean demonstrated formation of NO nitrosyl-Lb complex (LbFeIINO) and suggested interaction between ferrousleghemoglobin and NO produced in nodules (Mathieu et al. [1998](#page-18-7)). In another study, different forms of leghemoglobin, oxy-leghemoglobin and ferryl-leghemoglobin, were demonstrated to scavenge NO and peroxynitrite under in vitro conditions (Herold and Puppo [2005\)](#page-17-8). During the reaction, oxy-hemoglobin gets converted into met-hemoglobin, which is further recycled into oxy-hemoglobin. Similarly, scavenging mechanism was observed to be utilized by class-I hemoglobin in distinct plants, which catalyzed the conversion of NO to nitrate in a NAD(P)-H-dependent manner. In a mutation study on *Arabidopsis*, it was confrmed that class-1 hemoglobins were also regulated through S-nitrosylation. In other mutagenesis experiments, it was demonstrated that interaction between NO and class-I hemoglobin was not only critical for NO scavenging but also for various physiological conditions like abiotic and biotic stresses (Dordas et al. [2003;](#page-16-7) Dordas et al. [2004](#page-16-8); Perazzolli et al. [2004;](#page-19-5) Seregélyes et al. [2004\)](#page-19-6).

11.4.2 S-Nitrosylation

S-nitrosylation is a NO-dependent post-translational modifcation of proteins in which NO is incorporated in the reactive thiol group of cysteine residue to form nitrosothiol. S-nitrosylation is a reversible reaction in which the formation or breakage of the product does not require enzymes, mandatorily. Besides, two enzymes are known to facilitate the forward and backward reactions of S-nitrosylation which are nitrosylase and denitrosylase, respectively. At present thousands of proteins are known in plants which are target for S-nitrosylation, few of these proteins were common in animals which were also characterized for S-nitrosylation. Studies revealed that most of the proteins which were target for S-nitrosylation were involved in major physiological processes including photosynthesis, signaling, genetic information processing, primary and secondary metabolism, cytoskeleton activities, biotic/abiotic stress responses, etc. Few of the proteins which were identifed as potential target for S-nitrosylation in previous studies are summarized in Table [11.2.](#page-7-0)

11.4.3 Tyr-Nitration

Post-translational modifcation of proteins by covalent addition of a nitro group $(-NO₂)$ to tyrosine residue is generally referred to as tyr-nitration. Similar to S-nitrosylation, no enzymes are required for tyr-nitration (Ischiropoulos et al. [1992\)](#page-17-9).

		S-nitrosylation (Cys.	
Function	Target proteins	residue no.)	References
Photosynthesis	Rubisco large subunit	192, 427	Fares et al. (2011) , Abat and Deswal (2009)
	Photosystem I apoprotein A2	559	Fares et al. (2011)
Metabolism	GADPH	155, 159	Abat et al. (2008), Wawer et al. (2010)
	Methionine adenosyltransferase 1	114	Lindermayr et al. (2005), Lindermayr et al. (2006)
Oxidative stress	NADPH oxidase	890	Yun et al. (2011)
	Ascorbate peroxidase	32	Tanou et al. (2009), Bai et al. (2011)
	Catalase	230	Ortega-Galisteo et al. (2012)
Signaling	NPR ₁	156	Tada et al. (2008)
	Transcription factor	53	Serpa et al. (2007)
	Auxin receptor (TIR1)	140	Terrile et al. (2012)
	Calnexin	108	Fares et al. (2011)

Table 11.2 Few of the proteins in plants which are known to undergo S-nitrosylation

Attachment of the nitro group at ortho position of aromatic ring in tyrosine residue is a two-step process. In the frst step, the aromatic ring of the tyrosine residue gets oxidized by one electron, which results in the formation of a tyrosine radical. Electrons required for tyrosine oxidation are provided by either hydroxyl (OH•) or carbonate (CO_3^-) radicals, which are themselves derived from peroxynitrite (ONOO¯) (Yeo et al. [2015](#page-21-6); Radi [2013\)](#page-19-9) through three different pathways. The second step of tyrosine nitration involves interaction of tyrosine radical with nitrogen dioxide radical, which eventually produces 3-nitrotyrosine $(YNO₂)$ (Souza et al. [2000\)](#page-20-11). In this reaction, peroxynitrite is one of the most important molecules required for tyrosine nitration as it produces an oxidant required for the production of tyrosine radical. Peroxynitrite is itself produced through a fast reaction between nitric oxide and super oxide anion radical, and this shows that the peroxynitrite belongs to the reactive nitrogen species (RNS) group which is generated from NO (Patel et al. [1999\)](#page-19-10) and also has its relations with oxidative stress responses. A diagrammatic representation of tyrosine nitrosylation and reactions involved in it is shown in Fig. [11.3.](#page-8-0)

Few of the proteins which were identifed to be the target for tyrosine nitration include methionine synthase (Lozano-Juste et al. [2011](#page-18-10)), photosystem II protein1 (Galetskiy et al. [2011\)](#page-17-11), abscisic acid receptors (Castillo et al. [2015](#page-16-9)), monodehydroascorbate reductase (Begara-Morales et al. [2015\)](#page-15-8), isocitrate dehydrogenase (Begara-Morales et al. [2013](#page-15-9)), mitochondrial Mn superoxide dismutase (Holzmeister et al. [2015\)](#page-17-12), etc.

11.5 Abiotic Stress Tolerance and Its Response

Due to immobile nature of plants, they are not capable of changing their place when encountering unfavorable environmental conditions. Unfavorable conditions may be biotic or abiotic. Biotic factors refer to living organisms that are able to threaten plants' viability via direct physical damage or induce diseases in it, and this category of stress factors include organisms like bacteria, virus, nematodes, fungi,

insects, etc. (Atkinson and Urwin [2012](#page-15-10)). In contrast, abiotic factors are non-living factors like high soil salinity, high/low temperature, high/low light conditions, drought, heavy metals, etc. (Wang et al. [2003;](#page-20-12) Wani et al. [2016](#page-21-7); Nakashima and Yamaguchi-Shinozaki [2006\)](#page-18-11). Among these abiotic factors, high salinity and high temperature are the major role players in decreasing crop yield globally (Grattan et al. [2012](#page-17-13); Pereira et al. [2014;](#page-19-11) Hasegawa et al. [2018;](#page-17-14) Fischer and Edmeades [2010\)](#page-17-15). Regular irrigation, rainfall, and rock weathering are the main reasons why a normal fertile soil is converted into saline soil (commonly known as user soil). Most of the water leaches down the soil and part of it evaporate either directly or indirectly via transpiration by plants into the atmosphere, leaving salt in the upper layer of the soil (Rengasamy [2006](#page-19-12)). Thus, over a long period of time, salt keeps on accumulating in the soil and stays unfavorable for plant development. High soil salinity results in several plant changes at physiological, biochemical, and molecular levels. High concentration of salt in the soil lowers its water potential, disabling plants to absorb water from the soil and getting dehydrated Hoffman and Shalhevet [\(2007](#page-17-16)). As a consequence of dehydration, growth and metabolism of a plant slows down, eventually leading to its death. Due to salinity, water availability to the plant cell may also become limited even if soil is at saturation level, and the phenomenon is known as "physiological drought." High salinity results in reduced leaf area, leaf thickening, and stomatal closure, all of which eventually lead to a drastic reduction in its photo-synthetic efficiency (Delfine et al. [1998\)](#page-16-10). In order to maintain normal growth and metabolism under highly saline conditions, plants activate cell signaling pathways including those that lead to synthesis of osmotically active metabolites (amino acids, glycine betaine, sugars, sugar alcohols, etc.), plant hormones (ABA), specifc proteins (LEA-type proteins, chaperonins), and certain enzymes that control ion and water fux and support scavenging of ROS (Serraj and Sinclair [2002](#page-19-13); Kav et al. [2004;](#page-18-12) Raghavendra et al. [2010](#page-19-14)). Exposure of plants to high temperature also leads to various morpho-anatomical, physiological, and biochemical changes, affecting plant growth and yield (Long and Ort [2010](#page-18-13)). The effect of high temperature on plants growth is complex because plant responses are infuenced by various factors such as diurnal temperature range and water stress (Paulsen [1994\)](#page-19-15). Heat stress is also known to reduce germination and early growth and lead to accumulation of H_2O_2 which is able to oxidize various biomolecules in plant cell and leave them inactive (Bhattacharjee [2008](#page-15-11)). Heat stress is most often associated with drought stress as it impairs physiological processes and plant-water relations of crops (McDonald and Paulsen [1997](#page-18-14)). Also exposure of high temperature to plants results in a higher production of hydroxyl radical (OH·), which is known to initiate peroxidation of thylakoid lipids and thus lead to a decrease in photosynthesis efficiency (Wahid et al. [2007](#page-20-13); Mishra et al. [1993\)](#page-18-15).

Most of the pioneering studies were focused on exposure of different plants to a single abiotic stress condition, but their data could not be validated in feld conditions, since under feld conditions, plants are exposed to different abiotic and biotic stress conditions. This promoted researchers to study plants under combinations of different abiotic stresses, which confrmed that acclimatization done by plants against combinatorial stress conditions was signifcantly different from that of plants exposed to a single abiotic stress condition (Jiang and Huang [2001;](#page-18-16) Rizhsky et al. [2004\)](#page-19-16). Similar results of many such studies changed the fashion of experiments, and now most of the abiotic stress studies on plants are done by exposing them to a combination of different abiotic stresses. In a recent study, it was demonstrated that exposure of those seedling to high temperature which were growing on a medium containing high salt concentration resulted in an enhanced thermostability of catalase present in leaves of two varieties of *Vigna mungo*. In contrast when seedlings of those two varieties were exposed to high temperature ion absence of salt catalase in them was observed to be inactivated (Singh and Mishra [2020\)](#page-20-14).

When plants experience abiotic stress conditions, high-energy electrons which are being transported through electron transport chains of chloroplast and mitochondria spill their energy to nearby oxygen molecules, resulting in an uncontrollable generation of various reactive oxygen species (ROS) in plants (Foyer and Noctor [2003](#page-17-17); Asada [2006](#page-15-12); Foyer and Noctor [2005;](#page-17-18) Navrot et al. [2007\)](#page-18-17). ROS are oxidizing in nature with a short life span and are known to exist in various forms including superoxide anion radical (O_2^-) , hydroxyl radical $(OH[*])$, hydrogen peroxide (H_2O_2) , singlet oxygen $(^1O_2)$, etc. (Wahid et al. [2007\)](#page-20-13). The presence of reactive oxygen species (ROS) generating system in the plants, together with high percentage of polyunsaturated lipids, particularly in the thylakoid membrane, makes them susceptible to oxidative injury. Superoxide anion radical is one of the most commonly generated reactive oxygen species. Although capable of oxidizing various cellular components directly, it exerts most of its damaging effects by initiating the generation of more reactive species including hydroxyl free radicals (Halliwell and Gutteridge [1989](#page-17-19)). Other ROS are also detrimental to plants as they are highly effcient in generating damages to biomolecules such as proteins, lipid molecules in plasma membrane, and genetic material (DNA) present in a plant cell Mazid et al. [\(2011](#page-18-18)). On interaction with ROS, lipid molecules get converted into harmful lipid peroxide and other lipid abducts, and proteins are damaged which results in the reduction of their activity and possibility of DNA molecules getting mutated (Shah et al. [2001\)](#page-20-15).

During the course of evolution, the higher plants have developed an efficient antioxidant defense system, which helps them to cope with environmental stresses. Antioxidant defense system in plants is categorized into two categories. First category of enzymatic defense system is constituted by various enzymes like superoxide dismutase, catalase, peroxidases, glutathione reductase, DHAR, MDHAR, etc. (Asada and Takahashi [1987\)](#page-15-13). These enzymes are involved in detoxifcation of different ROS, e.g., SOD dismutates superoxide anion radical into H_2O_2 (Fridovich [1986\)](#page-17-20), which is further detoxified by catalase and various other peroxidases to H_2O and O_2 (Passardi et al. [2004;](#page-19-17) Asada [2006](#page-15-12)). Various enzymes involved in detoxification of different ROS are displayed in Table [11.3](#page-11-0). It is generally believed that those plant species or their varieties, which contain antioxidant enzymes possessing high tolerance toward different abiotic stresses, are referred to be resistant. In a recent study on seven varieties of *Vigna mungo*, it was demonstrated that the extent of thermostability of different SOD isoforms was not similar among varieties and those varieties with thermotolerant SOD isoforms were considered to be more

resistant toward heat stress (Singh et al. [2019](#page-20-16)). Another category of antioxidant defense system constitutes non-enzymatic molecules like ascorbate, reduced glutathione (GSH), α -tocopherol, carotenoids, flavonoids, etc., which contribute to protection of plants by deactivation/quenching of ROS in multiple ways (Asada and Takahashi [1987](#page-15-13)). In addition to this, proline and glycine betaine are also well-known for their contribution in coping stress conditions such as high temperature and high salinity. Both of these molecules act as an osmolyte, which promotes the entry of water from soil into plant cells. Besides ROS, when plants are exposed to different stress conditions, there occurs a simultaneous increase of nitric oxide and its derivatives in plant cells (Valderrama et al. [2007\)](#page-20-17). Nitric oxide and its derivative are referred to as reactive nitrogen species (RNS) due to the presence of unpaired electron. RNS are known to exist in various inorganic and organic forms which are themselves sub-categorized into radical and non-radical forms (Corpas et al. [2011\)](#page-16-11). Few important examples of RNS are shown in Table [11.3.](#page-11-0)

Among RNS, peroxynitrite (ONOO⁻) has been studied most extensively by researchers due to its involvement in various physiological processes in plants. Peroxynitrite is produced in a very fast reaction between NO and $O_{2} \cdot \cdot$, with a rate constant of approximately $(10^{10} \text{ M}^{\text{T}} \text{ s}^{\text{T}})$ (Estévez and Jordán [2002](#page-17-21)). Due to the coexistence of NO and O_2 •^{$\bar{ }$} in chloroplast, mitochondria, and peroxisomes, these organelles were identifed to be the main site of generation in plants (Blokhina and Fagerstedt [2010](#page-16-12); Corpas and Barroso [2014](#page-16-13)). Peroxynitrite, being highly oxidizing in nature, has been demonstrated to cause oxidative damages by interacting with all biomolecules present in the plant cell. Additionally, as described in the previous section of this chapter, RNS species particularly peroxynitrite has been identifed to mediate post-translational modifcations of various proteins, in turn leading to alterations in the physiological function controlled by them (Szabó et al. [2007;](#page-20-18) Corpas et al. [2009;](#page-16-14) Arasimowicz-Jelonek and Floryszak-Wieczorek [2011](#page-15-14); Calcerrada et al. [2011;](#page-16-15) Berton et al. [2012](#page-15-15); Szuba et al. [2015\)](#page-20-19). Some researchers proposed that increase in protein nitration could be considered as a biomarker for specifc stress conditions in similar to oxidation of proteins, which is utilized as a biomarker for oxidative stress (Corpas et al. [2007;](#page-16-16) Arasimowicz-Jelonek and Floryszak-Wieczorek [2011](#page-15-14)).

Type	Inorganic	Organic
Non-radical	Nitrous acid $(HNO2)$	Nitrotyrosine $(Tyr-NO2)$
forms	Nitrosonium cation $(NO+)$	Nitrosoglutathione (GSNO)
	Nitroxyl anion (NO ⁻)	Nitrosothiols (SNOs)
	Peroxynitrite (ONOO ⁻)	$Nitro-\gamma$ -tocopherol
	Peroxynitrous acid (ONOOH)	Nitro-fatty acids $(FA-NO2)$
	Dinitrogen trioxide (N_2O_3)	
	Dinitrogen tetroxide (N_2O_4)	
Radical	Nitric oxide ("NO)	Lipid peroxyl radicals (LOO ⁻)
Forms	Nitrogen dioxide $(\neg NO_2)$	

Table 11.3 Types of reactive nitrogen (RNS) species generated in plants

There are many previous studies which have demonstrated that the effects of ROS and RNS on plants are tightly regulated by each other. It has been shown that RNS are involved in mediating post-translational modifcations of various enzymes such as catalase, superoxide dismutase (Mn, Fe, and Cu/Zn isoforms), peroxiredoxin II E and F, DHAR, MDHAR, ascorbate peroxidase, etc., which are known to be involved in ROS detoxifcations, and thus RNS regulate their activity during stress conditions (Ortega-Galisteo et al. [2012;](#page-19-7) Chaki et al. [2015](#page-16-17); Holzmeister et al. [2015;](#page-17-12) Romero-Puertas et al. [2007](#page-19-18); Camejo et al. [2015;](#page-16-18) Clark et al. [2000;](#page-16-19) Fares et al. [2014;](#page-17-22) Begara-Morales et al. [2015](#page-15-8)). On the other hand, generation of one of the most primary RNS "peroxynitrite" is dependent on the presence of a reactive oxygen species "superoxide anion radical." On observing these regulatory roles of ROS and RNS over each other, Corpas et al. ([2013\)](#page-16-20) suggested that using the term nitrooxidative stress in place of oxidative stress shall be more appropriate.

11.6 Function of Nitric Oxide in Abiotic Stress Tolerance

11.6.1 Salt Stress

Plant suffers osmotic and ionic stress at high salt concentrations. Due to this many chief metabolic processes get affected which in turn limits the growth of plants. Enzyme activities of plants are badly affected under high salinity which leads to alteration in nitrate and sulfate assimilation pathway and a drop in energy. Further, salinity alters the activities of and sulfur & nitrogen demand increases (Vanlerberghe [2013\)](#page-20-20). High salinity leads to oxidative damage due to ROS and causes cellular injury. To avoid such oxidative damage, plants have developed a wide range of defense methods. However, it is a less known fact that plants become tolerant to salinity in the presence of nitric oxide. Arora et al. reported the impacts of exogenous sodium nitroprusside (SNP), a NO donor. It was found that it remarkably reduced the oxidative damage under highly saline conditions. Its effect was observed in rice, cucumber, and lupin seedlings where it enhanced seedling growth and also maize dry weight even under salt stress (Arora et al. [2016](#page-15-16)). A good stability was observed between carbon and nitrogen metabolism when pretreated with nitric oxide. This effect of nitric oxide was due to increased total soluble protein and enhanced endopeptidase and carboxypeptidase activities in plants under salinity stress (Wojtyla et al. [2016\)](#page-21-8). Salt stress tolerance of roots of red kidney beans was due to the signifcant role of glucose-6-phosphate dehydrogenase enzyme in NR-dependent nitric oxide production. Prior studies have reported that nitric oxide induces ROS scavenging enzyme activities of CAT, peroxidase (POD), SOD, and ascorbate peroxidase (APX), which in turn reduces membrane permeability, rate of ROS production, and intercellular CO2 concentration under high salinity stress. Nitric oxide also induces the expression of transcripts of genes encoding

sucrose-phosphate synthase and Δ′-pyrroline-5-carboxylate synthase, related to stress (Shi and Chan [2014](#page-20-21)).

11.6.2 Drought Stress

Drought is a key factor responsible for limited crop yield. It has been found that nitric oxide synthesis is dependent on intensity of water defcit, mild or severe. When cucumber roots suffer from mild water deficit $(5-10 \text{ h})$, nitric oxide synthesis in root tip cells and the elongation zone surrounding it is slightly increased. But when the duration of this stress increases up to 17 h, it results in rigorous nitric oxide synthesis in the roots of cucumber (Ahmad et al. [2016\)](#page-15-17). Also, pea, wheat, and tobacco have shown enhanced nitric oxide production. Stress due to lack of water induces the production of nitric oxide in mesophyll cells of maize and activity of nitric oxide synthase (NOS) in cytosolic and microsomal fractions of maize leaves. This nitric oxide synthesis was blocked by prior treatment with inhibitors of nitric oxide synthase and nitrate reductase. This suggests that nitric oxide is synthesized from NOS and NR in maize leaves when exposed to water stress. Treatment with nitric oxide synthase and nitrate reductase inhibitors suppressed the activities of chloroplast and antioxidant enzymes present in cytosol, i.e., SOD, and GR. Reduced activities of these enzymes were improved by externally applying NOS and decreasing the buildup of H_2O_2 due to water stress (Delorge et al. [2014](#page-16-21)). Due to induction of subcellular antioxidant defense, the possible ability of nitric oxide to scavenge $H₂O₂$ is in some part only (Laxa et al. [2019\)](#page-18-19).

11.6.3 Temperature Stress

Crop productivity is equally affected by temperature variations. Plant injuries due to high temperature include damage due to oxidative stress, damage to membrane, lipid peroxidation, degradation of protein, inactivation of enzymes, and disruption of DNA strands. Also cold stress retards biochemical and physiological processes and ROS homoeostasis in plants (Roychoudhury et al. [2013\)](#page-19-19). Nitric oxide has a unique role in the management of heat and cold stress. Lucerne cells have been reported for increased nitric oxide synthesis at high temperatures, while tomato, wheat, and maize developed cold tolerance when NO is applied externally (Parankusam et al. [2017](#page-19-20)). NO synthesis was augmented by short-time heat stress in alfalfa. It is likely that this result due to antioxidative property of nitric oxide, which raise adverse effects imposed by the intensifcation of peroxidative metabolism in cold and heat stress. Heat stress stimulated NO synthesis that could play a role in the induction of cell death in *Symbiodinium microadriaticum* by causing a boost in caspase-like activity (Leterrier et al. [2012](#page-18-20)).

11.6.4 Ultraviolet Radiation Stress

The injure of the stratosphere ozone layer amplifed the radiation of UV-B (250–320 nm) responsible for increase in ion leakage, chlorophyll loss, and reducd effciency of PS-II photochemistry (Fv/Fm) and the quantum yield PS-II electron transport (\varnothing PS-II), and augmented H_2O_2 and thylakoid membrane protein oxidation. UV-B radiation leads to a boost in nitric oxide and reactive oxygen species in *Arabidopsis* (Shi and Chan [2014\)](#page-20-21). Nitric oxide generated from NOS-like activity acts synergistically with ROS to stimulate ethyl synthesis in defense response under UV-B radiation in leaves of maize (Arora et al. [2016\)](#page-15-16). UV-B induced augmentation of NOS activity in maize hypocotyls, indicating that NO may act as a second messenger and perform antioxidant response to UV-B radiation, and SNP-exposed maize plants exhibited increased activity of glucosidase and protein synthesis.

11.6.5 Heavy Metals Stress

As we know heavy metals contamination affects the biosphere in many places worldwide. Several studies have been conducted in order to evaluate the effects and remedy of different heavy metals concentration on plants (Schützendübel and Polle [2002\)](#page-19-21). NO also plays a vital role in the enhancement of antioxidant enzyme activities and alleviating the toxicity of heavy metals. Exogenous application of SNP reduced copper (Cu) toxicity and NH4+ accumulation in rice leaves (Manara [2012a\)](#page-18-21). The protective effect of SNP on the toxicity and NH4⁺ accumulation can be reversed by c-PTIO, a NO scavenger, suggesting that the protective effect of SNP is attributable to NO released. These results also suggest that reduction of Cu-induced toxicity and NH4+ accumulation by SNP is most likely mediated through its ability to scavenge active oxygen species. SNP pretreatment significantly reduced O_2 – induced specifc fuorescence in *Lupinus luteus* roots under heavy metal treatment. Results obtained in this study suggest that antioxidant function of NO may be traced by a scavenging of O₂−, resulting in a decrease of superoxide anion (Yadav [2010\)](#page-21-9). The detoxifcation and antioxidative properties of NO are also found in soybean cell cultures under cadmium (Cd) and Cu stress. Moreover, NO decreased the aluminum (Al3+) toxicity in root elongation of *Hibiscus moscheutos* (Manara [2012b\)](#page-18-22).

11.7 Conclusion

Nitric oxide is a ubiquitous bioactive molecule with diverse roles in a huge spectrum of physiological processes in plants. Even then, there are much more rigorous studies required to understand the complete range of activities being performed by nitric oxide. A huge amount of work is already done about functions of nitric oxide as a signaling molecule interacting with plant hormones, nutrients, or metals. But still much more is to be searched upon the activities of nitric oxide.

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