Fungal Biology

R. Naraian Editor

Mycodegradation of Lignocelluloses



Fungal Biology

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About the Series

Fungal biology has an integral role to play in the development of the biotechnology and biomedical sectors. It has become a subject of increasing importance as new fungi and their associated biomolecules are identified. The interaction between fungi and their environment is central to many natural processes that occur in the biosphere. The hosts and habitats of these eukaryotic microorganisms are very diverse; fungi are present in every ecosystem on Earth. The fungal kingdom is equally diverse, consisting of seven different known phyla. Yet detailed knowledge is limited to relatively few species. The relationship between fungi and humans has been characterized by the juxtaposed viewpoints of fungi as infectious agents of much dread and their exploitation as highly versatile systems for a range of economically important biotechnological applications. Understanding the biology of different fungi in diverse ecosystems as well as their interactions with living and non-living is essential to underpin effective and innovative technological developments. This series will provide a detailed compendium of methods and information used to investigate different aspects of mycology, including fungal biology and biochemistry, genetics, phylogenetics, genomics, proteomics, molecular enzymology, and biotechnological applications in a manner that reflects the many recent developments of relevance to researchers and scientists investigating the Kingdom Fungi. Rapid screening techniques based on screening specific regions in the DNA of fungi have been used in species comparison and identification, and are now being extended across fungal phyla. The majorities of fungi are multicellular eukaryotic systems and therefore may be excellent model systems by which to answer fundamental biological questions. A greater understanding of the cell biology of these versatile eukaryotes will underpin efforts to engineer certain fungal species to provide novel cell factories for production of proteins for pharmaceutical applications. Renewed interest in all aspects of the biology and biotechnology of fungi may also enable the development of "one pot" microbial cell factories to meet consumer energy needs in the 21st century. To realize this potential and to truly understand the diversity and biology of these eukaryotes, continued development of scientific tools and techniques is essential. As a professional reference, this series will be very helpful to all people who work with fungi and should be useful both to academic institutions and research teams, as well as to teachers, and graduate and postgraduate students with its information on the continuous developments in fungal biology with the publication of each volume.

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R. Naraian Editor

Mycodegradation of Lignocelluloses



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Preface

Lignocellulose is the most abundantly available biomass of complex biopolymers originated from plants growing on earth. This is a biological material that comprises several agricultural residues, woods, grasses, and forest wastes chemically composed of different concentrations of components varying in different plant species from which it is originated. Lignocellulosic residues are basically composed of two major carbohydrate polymers, viz., cellulose and hemicellulose, and a non-carbohydrate closed ring-bearing phenolic polymer lignin. Cellulose is considered as the principal structural polysaccharide constituent of primary plant cell wall by accounting for 30-50% of lignocellulose on dry weight basis. The hemicellulose is considered as the second most significant polysaccharide constituent of lignocellulose, which generally accounts for 15-30% in all plant cell walls. Hemicelluloses are found to be imbedded in the complex structure of plant cell walls with a significant function of cellulose microfibrils binding to strengthen the plant cell wall. Distinct to the cellulose, hemicellulose structurally constitutes a random and amorphous structure made up of several heteropolymer units including xylan, glucomannan, glucuronoxylan, xyloglucan, and the arabinoxylan. However, the third constituent is a highly complex component of lignocellulose known as phenolic lignin accounting the amount of 15–30%. The lignin, as the part of all vascular plants, represents the second top most carbon source after the available cellulose on the earth. Lignin indeed provides strength and rigidity to cell wall and protects it from microbial attacks causing decomposition because of the presence of closed phenolic rings in its complex structure.

Nowadays, huge amount of lignocellulose in rural areas is either burnt or negligently thrown, which creates serious post-disposal problems of health and environment. Thus, it is necessary to implement a method of efficiently degrading lignocelluloses to a non-polluting level. The microbial bioconversion of lignocelluloses into several valuable products of human usages is an efficient technological solution. Among several microbial classes, fungi are the most efficient at this bioconversion because of the potential of their strongest lignocellulolytic enzyme-producing versatile machinery. The degradation of lignocelluloses accomplished by the fungi is termed as "mycodegradation," which can be executed by the huge diversity of fungal species.

The content of this book is comprised of the contributions of chapters from several eminent academicians and researchers of reputed institutions providing recent updates on mycodegradation of lignocelluloses. The chapters of the book cover almost all the aspects of mycodegradation such as the biochemistry of lignocellulose, involved fungal biodiversity and several enzymes playing a role in the process of mycodegradation integrating hemicellulolysis, cellulolysis, and ligninolysis. The book also covers the information in respect to mechanism of lignocellulose mycodegradation both under natural and in vitro conditions. The aspect of biofuel production using the process of lignocellulose degradation and enhancement of lignocelluloses employing nanotechnological approaches has also been incorporated to make the content advanced and informative. Consequently, the book has a comprehensive and up-to-date information on mycodegradation of lignocellulose with clear and easy illustration with simplified elucidations.

I am thankful to many people who helped me to bring this book to its genuine structure. I extremely wish to express my hearty thanks to Dr. Eric Stannard, senior editor, Botany, Springer, Dr. Vijai Kumar Gupta and Maria G. Tuohy, series editors, Fungal Biology, Springer; Ms. Sanjana M. Sundaram, project coordinator, Springer; and Mrs. Cathrine Selvaraj, project manager for their unified generous assistance, constant support, and patience in finalizing the book. I give my special thanks to Prof. (Dr.) Raja Ram Yadav, Hon'ble Vice-Chancellor, V. B. S. Purvanchal University, Jaunpur (UP), India, for constant motivations. My sincere thanks also go to all chapter contributors, without which the structure of the book would not have been possible. From the depth of my heart, I extend my thanks to my entire team of research scholars including Dr. Abhishek Kr. Bharadwaj, Mr. Roshan L. Gautam, Ms. Shweta Singh, Ms. Nahida Arif, Mr. Tripuresh P. Shahee, Ms. Madhumita Singh, and Mr. Prameel Kumar for their substantial support during editing the book in a very prompt manner. I also give special thanks to my esteemed friends, well-wishers, colleagues, and senior faculty members of V. B. S. Purvanchal University, Jaunpur (UP), India.

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Chapter 1 Basic Mechanism of Lignocellulose Mycodegradation



Roshan Lal Gautam, Shweta Singh, Simpal Kumari, Archana Gupta, and R. Naraian

1 Introduction

Lignocellulose refers to a dry plant biomass abundantly available on earth in huge amount, which is annually synthesized and microbially degraded in nature. Tremendous amount of lignocellulosic wastes is also additionally produced by different industries including forestry, pulp and paper, agriculture, food from municipal solid waste, and animal wastes (Champagne 2007; Wen et al. 2004). The woody plant cell wall consists of major structural polymer components, namely, cellulose, hemicellulose, and lignin (Scheller and Ulvskov 2010; Gibson 2012). The lignocellulosic complex in plant cell wall contains approximately 40-60% cellulose, 20-40% hemicellulose, and 10-25% lignin (Kubicek 2012a, b), which can vary from one plant species to another. Sometimes huge biomass of lignocelluloses remains unused, which is either dumped or burned negligently creating big environmental issues. However, lignocellulosic biomass stores tremendous amount of energy that can be converted into another form and be employed for different valuable purposes. The microorganisms are the only predominantly responsible factors for lignocellulose degradation including fungi, and the most rapid degraders in fungi are the members of basidiomycetes (Sanchez 2009). White-rot fungi (WRF) have a versatile potential to secrete a wide range of wood-degrading enzymes involved in the breakdown of carbohydrate components (cellulose and hemicellulose) and aromatic constituent (lignin) (Kameshwar and Qin 2016). There are reports of approximately 10,000 species of WRF, having different aptitude for depolymerization of lignocellulosic materials and further mineralization to carbon and water (Kirk 1984; Halis et al. 2012). WRF such

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as *Phanerochaete chrysosporium*, is highly preferred, which simultaneously degrade all cell wall polymers, while several others selectively decay lignin ahead of cellulose and hemicellulose (Akhtar et al. 1992; Behrendt and Blanchette 1997; Srebotnik and Messner 1994; Hori and Cullen 2016). These abilities of white-rot fungi are because of that they can grow under less restrictive conditions to degrade lignocellulosic wastes (Salvachua et al. 2011).

The cellulose-degrading organisms very luxuriantly use lignocellulose as feedstock for their growth, reproduction, and consequently bioconversions (Chen 2014). The microbial conversion of lignocellulosic complexes to monomers involves the synergistic action of multiple enzymes, such as peroxidases, laccases, xylanases, endoglucanases, exoglucanases, β -xylosidase, and β -glucosidases (Hasunuma et al. 2013). The extracellular enzymes are much crucial for depolymerization of cellulose and degradation of lignin into simple metabolites (Akhtar et al. 1997). Moreover, enzymatic hydrolysis of lignocelluloses is one of the most efficient options for the fruitful conversion of polysaccharide into monosaccharide by avoiding toxic metabolite formation (Pramanik and Sahu 2017).

The efficient biodegradation of lignocellulosic biomass with the use of versatile microbial communities has been regarded as an excellent and efficient approach for multitude of biotechnological applications (Fang et al. 2012; Li et al. 2012). The potential applications of lignocellulolytic enzymes for a variety of uses such as industrial and environmental technologies need huge quantities of these enzymes at lowest costs (Elisashvili et al. 2008). Cellulases and xylanases are characterized as inducible enzymes, usually synthesized by microbes when grown on polysaccharide substrates, with lignocellulosic biomass (Howard et al. 2003; Iqbal et al. 2013; Fatokun et al. 2016).

2 Lignocellulose

Lignocellulosic materials including wood and grass along with forest and agricultural residues are chemically composed of cellulose, hemicellulose, and lignin, with varying ratios of these components according to the plant species from which they belong (Rennie and Scheller 2014; Bosetto et al. 2016). Lignocelluloses are the structural polysaccharides of plants composed of cellulose (~50%), hemicellulose (~30%), and lignin (~20%), which are widely distributed among the vascular plants (Coral et al. 2002; Sajith et al. 2016).

3 Cellulose

Cellulose originated from plants as the major constituent of lignocelluloses is the most abundant organic polymer on earth (Naika et al. 2007). This is the main building material of which plants are made as a very important polymer substance (Lavanya et al. 2011). Cellulose is a high-molecular-weight-containing (Klemm et al. 2005), linear polymer of glucose linked through α -1,4-linkages and is usually arranged in microcrystalline structures (Kuhad et al. 1997; Leschine 1995). Its structure consists of extensive intramolecular hydrogen bonding networks, which tightly binds the glucose units. This is a water-insoluble polymer with a rigid linear structure (Festucci-Buselli et al. 2007). This polymer known to be an important renewable source of bioenergy is involved in several microbial treatments for its significant conversion into soluble and fermentable glucose (Bhalla et al. 2013).

4 Hemicellulose

Hemicelluloses are the second most plentiful constituent of lignocellulosic biomass (Okeke and Lu 2011) after celluloses. Hemicelluloses are generally found associated with cellulose in primary and secondary walls of all plants (Jeffries 1994). In the structure of cell wall, hemicelluloses function as an adhesive between the stiff reinforcing cellulose microfibrils and the matrix lignin (Salmen and Burgert 2009; Kulasinski et al. 2016). The most common form of hemicellulose is the xylan, composed from the backbone of β -(1-4)-linked xylose monomers (Scheller and Ulvskov 2010). Hemicellulose may primarily contain the mannose and galactose but can also contain pentose, xylose, and arabinose. The hemicellulose as a hetero-polysaccharide can be composed of different water-soluble sugars including hexoses, pentoses, and glucuronic acid monomers (Kuhad et al. 1997), interacting with cellulose and lignin through either hydrogen bonding or covalent linkages (Ebringerova et al. 2005). Basically, xylan-type of hemicelluloses are known to be the hemicellulosic constituent of the cell walls of hardwoods and of any herbaceous plants constituting approximately 20-35% weight of the total biomass (Chen et al. 2014). Hemicelluloses are composed of three groups of amorphous polymers such as xylans, mannans, and galactans (Fan et al. 1987).

5 Lignin

Lignin is a major structural component of plants with high-molecular-weight, threedimensional aromatic hetero-polymer, associated to other cell wall components cellulose and hemicellulose (Jeffries 1990). This is the most abundant source of carbon in soil after cellulose (Kubicek 2012a, b). Lignin is highly irregular in structure and is an insoluble polymer composed of phenylpropanoid subunits, namely, p-hydroxyphenyl (H-type), guaiacyl (G-type), and syringyl (S-type) units (Malherbe and Cloete 2002). Lignin is an aromatic polymer constituted of three basic units such as p-coumaryl alcohol (4-hydroxycinnamyl alcohol), coniferyl alcohol, and sinapyl alcohol, which are collectively pronounced as monolignols derived from phenylalanine (amino acid) (Whetten and Sederoff 1995).

6 Biodegradation Mechanism

The biodegradation lignocellulosic complex substrates not only depend on environmental factors alone, but also on the degradative potential of the microbial class and population (Waldrop et al. 2000). The complex polymers of lignocelluloses are degraded by the variety of microorganisms and with the potential synergistic action of various enzymes (Parani and Eyini 2010) (Table 1.1). The cellulose and hemicelluloses are selectively degraded by "soft-rot" fungi (Dix and Webster 1995). The commercially available enzymes can effectively degrade lignocelluloses into their monomers (Okeke and Lu 2011). Most of the fungi have two specific types of extracellular enzymatic systems, viz., the hydrolytic system, which produces hydrolases responsible for polysaccharide degradation, and a unique oxidative and extracellular ligninolytic system that degrades lignin and opens phenyl rings (Sanchez 2009). Several microorganisms including fungi are strongly capable to produce lignocellulolytic enzymes involved in the degradation of lignocellulosic biomass. Mycodegradation of distinct lignocellulosic components (cellulose, hemicellulose, and lignin) occurs variably with diverse group of fungi and with the attack of specific enzymes independently (Fig. 1.1). The breakdown of hemicellulose, cellulose, and lignins can be termed as hemicellulolysis, cellulolysis, and ligninolysis simultaneously.

6.1 Cellulolysis

Cellulolysis is the term which explains the process of cellulose hydrolysis employing efficient cellulosic enzymes. As the cellulose is a polymer of β (1-4)-linked glucose, individually associating with hydrogen bonds (Sjostrom 1993), which are hydrolyzed by enzymes of microbial origin. Fungi are potentially recognized as the strongest group of cellulose-degradable microorganisms producing high level of cellulases (Chen 2014). The efficient cellulolytic fungus genus can be sequentially represented as the species of *Aspergillus, Penicillium, Chaetomium, Trichoderma, Fusarium, Stachybotrys, Cladosporium, Alternaria, Acremonium, Ceratocystis, Myrothecium, Humicola*, etc. (Wood 1985; Mehrotra and Aneja 1990; Sajith et al. 2016). All the above-mentioned genus of fungi have excellent capability of enzymes production involved in cellulolysis (Fig. 1.2).

6.1.1 Enzymes Involved in Cellulolysis

Cellulase as a complex mixture of enzymes is produced by most fungi and other microorganism naturally, which act synergistically to decompose celluloses. Cellulases are able to break β -1, 4-glycosidic bond in the structure of cellulose by attacking as multicomponent enzyme system comprised of three major classes of

S.N.	Fungal genus	Lignocellulosic substrate	References
1.	Aspergillus, Gliocladium	Rice straw	Shruti et al. (2015)
2.	Ganoderma lucidum C7016	Birch wood	Xiao et al. (2013)
3.	Pleurotus sajor-caju	Corncob	Menezes et al. (2017) Naraian et al. (2010)
4.	Hypholoma fasciculare, Stropharia rugosoannulata, Stachybotrys	Wheat straw	Singh et al. (2014)
5.	Trichoderma viride, Ganoderma lucidum	Corn stover	Shahzadi et al. (2014)
6.	Pleurotus pulmonarius	Corncob	Correa et al. (2016)
7.	Agaricus bisporus	Wheat straw	Kabel et al. (2017)
8.	Pleurotus ostreatus	Sorghum and millet chaff	Ryden et al. (2017)
9.	Flammulina velutipes	Ramie stalk	Xie et al. (2017)
10.	Auricularia thailandica	Sawdust, wheat husk, sugarcane bagasse	Bandara et al. (2017)
11.	Inonotus obliquus	Rice straw, wheat straw, bagasse	Xu et al. (2014)
12.	Aspergillus niger	Rice straw	Yang et al. (2018)
13.	Penicellium oxalicum GZ-2	Corn stover	Liao et al. (2014)
14.	Chaetomium globosum BCC5776	Rice straw	Wanmolee et al. (2016)
15.	Ganoderma lucidum	Bean stalks, cotton stalk, maize straw, rice straw, sugarcane bagasse, wheat straw	Rashad et al. (2019)
16.	Lentinula edodes	Birch-wood	Xiong et al. (2019)
17.	Pleurotus tuber-regium	Sugarcane, corncob	Lam et al. (2018)
18.	Pleurotus florida	Mixed leafy biomass	Malayil et al. (2016)
19.	Auricularia auricular	Spent mushroom, rice husks	Meng et al. (2018)
20.	Pleurotus ostreatus	Sawdust, cottonseed hull, corncob	Xiao et al. (2019)
21.	Gymnopilus pampeanus	Populus, eucalyptus sawdust	Colavolpe and Alberto (2014)
22.	Trichoderma reesei	Wheat straw, rice straw, rice husk	Taherzadeh- Ghahfarokhi and Mokhtarani (2019)
23.	Fusarium sp.	Sugarcane bagasse	Bertonha et al. (2018)
24.	Fusarium verticillioides	Gamba grass	de Almeida et al. (2019)
25.	Humicola insolens, Trichoderma reesei	Rice straw	Kogo et al. (2017)
26.	Myrothecium verrucaria	Birch sawdust	Wang et al. (2017)
27.	Acremonium cellulolyticus, Trichoderma reesei	Eucalyptus, Douglas fir, rice straw	Fujii et al. (2009)

 Table 1.1 Different recent studies performed with several fungal species for the degradation of variable lignocellulosic substrates

(continued)

S.N.	Fungal genus	Lignocellulosic substrate	References
28.	Acremonium strictum	Sugarcane bagasse	Goldbeck et al. (2013)
29.	Ceratocystis fimbriata	Coffee husk	Mussatto and Teixeira (2010)
30.	Cladosporium oxysporum GQ-3	Wheat bran	Guan et al. (2016)
31.	Alternaria alternate	Wheat bran	Faten and Abeer (2012)

Table 1.1 (continued)

enzyme: endoglucanases (endo- β -1,4-glucanase, CMCase, EC 3.2.1.4), exoglucanase (cellobiohydrolases, EC 3.2.1.91), and β -glucosidase (cellobiase, EC 3.2.1.21) (Dincer and Telefoncu 2006) (Fig. 1.2). The detailed information of these three enzymes have been accounted as below.

Endoglucanase

Endoglucanase (endo- β 1-4-d-glucan 4-glucanohydrolase, EC 3.2.1.4) acts randomly to cleave beta-linked bonds within the cellulose chain (Saha 2004). Endoglucanase also attacks on cellodextrins, which is an intermediate product after cellulose hydrolysis, and consequently converts them to cellobiose and glucose units (Sajith et al. 2016). These enzymes are generally found active against the acidswollen amorphous cellulose, soluble derivatives of cellulose such as carboxymethylcellulose and cellooligosaccharides (Wood 1989). The endoglucanases are the most economical enzymes employed at large scale for the hydrolysis of celluloses (Henriksson et al. 1999) (Fig. 1.2).

Exoglucanase

Structurally, exoglucanases (exo- β 1-4-d-glucan 4-cellobiohydrolase, EC 3.2.1.91) have long loops and affinity to act with crystalline sites of cellulose chains and primarily result in cellodextrin (Segato et al. 2014). Exoglucanases become usually active on crystalline part of cellulose and cleave disaccharide units one from the both ends either at the non-reducing or reducing end (Mrudula and Murugammal 2011). These are found in two forms; first attack upon the reducing end or working on the non-reducing end; second the cellobiohydrolases acting unidirectionally on the long chain oligomers (Juturu and Wu 2014) (Fig. 1.2).

β-Glucosidase

Beta-glucosidase (EC 3.2.1.21) is the key enzyme, which plays a significant role in the complete conversion of cellobiose to simpler glucose (Berlin et al. 2007). Such enzymes possess a rigid structure with highly active sites present in a large cavity,



Fig. 1.1 The three integral components of lignocellulose (cellulose, hemicellulose, and lignin) and cluster of enzymes involved in hydrolysis of respective component



Fig. 1.2 Major events involved in the process of enzymatic cellulolysis through the catalytic action of several cellulolytic fungal enzymes generating monomers (glucose) by breaking linear polymer of cellulose complex: (1) linear chain of cellulose, (2) attack of exoglucanase I at the non-reducing end of crystalline region, (3) attack of endoglucanase within ends at amorphous region, (4) attack of exoglucanase II at the reducing end of crystalline region, (5 and 6) attack of β -glucosidase on the cellobiose, (7) formation of glucose after breaking of cellobiose

basically termed as the active site pocket, that indeed favours the entry of disaccharides (Nam et al. 2010). It is absolutely inactive against the crystalline or amorphous region of cellulose (Sadhu and Maiti 2013). The saccharification of cellulose seems to be inefficient without the β -glucosidases from any source (Sternberg et al. 1977) (Fig. 1.2). Thus, this enzyme is considered to be a very efficient enzyme without which saccharification of celluloses cannot be accomplished successfully.

6.2 Hemicellulolysis

The term hemicellulolysis can be explained as the breakdown of hemicelluloses into their oligomer units. Hemicellulose is a component of lignocelluloses connected to lignin and cellulose through covalent bonds, but because of the involvement of few hydrogen bonds, it is much promptly broken down in comparison to crystalline cellulose (Jacobsen and Wyman 2000). Hemicellulose comprised of the branching chains of sugars, which consists of (1) aldopentoses (xylose and arabinose) and (2) aldohexoses (glucose, mannose, and galactose). Most biomass of hemicelluloses consists of either xylan or glucomannan backbones with acetyl group, arabinose, galactose, and methyl glucuronic acid on the linked side chains (Zhang et al. 2011). Hemicellulase and xylanase enzymes are used for the hydrolysis of hemicelluloses (El-Zawawy et al. 2011; Hu et al. 2011). In consequence, the hydrolysis of hemicelluloses yields sugars and organic acids as their products (Steinbach et al. 2017) (Fig. 1.3).

6.2.1 Enzymes Involved in Hemicellulolysis

Many bacteria, fungi, and yeast can degrade hemicellulose through the production of a cluster of hydrolytic enzymes, including xylanases, xylosidases, arabinofuranosidases, and glucuronidases (Shallom and Shoham 2003; Maayer et al. 2014). The



Fig. 1.3 Major events involved in enzymatic hemicellulolysis through the catalytic action of several hemicellulolytic fungal enzymes generating monomers (xylose, ferulic, and 4-*o* methyl-D-glucuronic acid) by breaking polymer of hemicellulose structure. Attack of (1) endo 1,4 β -xylanase, (2) acetyl xylan esterase, (3) α -D-glucuronidase, (4) α -L-arabinofuranosidase, (5) α -D-galactosidase, (6) feruloyl esterase, and (7) β -D-xylosidase

quick and complete degradation of hemicellulose needs the synergistic action of a broader group of hydrolytic enzymes (Polizeli et al. 2005). The enzymes selectively remove undesired branches in hemicellulose structure and keep the desired polymer backbone intact (Sedlmeyer 2011). During most of the bioconversion processes, xylan is firstly converted to xylose or into xylooligosaccharides, by the catalytic action of xylanolytic enzymes (Chivero et al. 2001) (Fig. 1.3).

Xylan

Xylan is the group of hemicellulose that represents the most abundant biopolymers present on earth. Xylan is primarily formed in the secondary cell wall together with cellulose (1,4- β -glucan) and lignin as the major constituents of plant cell walls (Kulkarni et al. 1999). This is a complex polymer consisting of 1,4-linked xylosyl residues that can be acetylated or it can also have covalently attached arabinosyl and glucuronic acid as side groups (De Groot et al. 1998). The complete enzymatic hydrolysis of xylans is performed by the action of multiple enzyme such as endo- β -1,4 xylanase and β -xylosidase and additionally through the help of several accessory enzyme, viz., α -L-arabinofuranosidase, acetyl xylan esterase, α -glucuronidase, ferulic acid esterase, and p-coumaric acid esterase (Guais et al. 2010). The detailed information of these enzymes is accounted separately as below.

Xylanase

Xylanases (EC 3.2.1.8) are the enzyme complexes capable for the hydrolysis of xylan (Shallom and Shoham 2003). These enzymes are produced mainly by a variety of microorganisms that take part in breakdown of xylans (Polizeli et al. 2005), fungi being the majority of them. The exploitation of xylanases has been extensively proved in paper industries (Sohpal et al. 2010). Xylanases are classified into glycosyl hydrolase (GH) families 5, 7, 8, 10, 11, and 43 based on their amino acid sequences, structural foldings, and mechanisms of action on substrate (Cantarel et al. 2009).

Endo-1,4-β-Xylanase

The endo-1,4- β -xylanase (EC 3.2.1.8) is a crucial enzyme that is capable of randomly cleaving the β -1,4-glycosidic linkages in the backbone of xylan (Conejo-Saucedo et al. 2017). The attack of endo-1,4- β -xylanase produces oligosaccharides, after the cleavage of xylan (Jeffries 1994), releasing smaller xylooligosaccharides (Collins et al. 2005). Therefore, endo-1, 4- β -xylanases are very promising through direct involvement by cleaving glycosidic bonds and liberating structurally smaller xylooligosaccharides (Verma and Satyanarayana 2012; Anand et al. 2018). On the basis of end product obtained, the working behaviours of endoxylanase are categorized into two major types: (a) non-debranching xylanase and (b) debranching xylanase. The non-debranching xylanase cannot split the 1,3- α -L-arabinofuranosyl branch of arabinoxylans, while debranching xylanase do not hydrolyze side branches to liberate arabinose (Polizeli et al. 2005)

β-Xylosidase

 β -Xylosidases (EC 3.2.1.37) are promising hydrolytic enzymes, which play an important role in degradation (Knob et al. 2010), and they hydrolyze small xylooligosaccharides and xylobiose, releasing β -D-xylopyranosyl residues from the nonreducing terminus (Biely 1985, 1993; Polizeli et al. 2005), and 1,4- β -xylosidase produces xylose from oligosaccharides (Jeffries 1994).

On the basis of variation in amino acid sequences, these enzymes are classified into five GH families: 3, 39, 43, 52, and 54. The β -xylosidases from GH43 family hydrolyze the non-reducing ends of xylooligomers (Graciano et al. 2012).

Arabinofuranosidases

Alpha-L-arabinofuranosidases (α -L-arabinofuranoside arabinofuranohydrolases, EC 3.2.1.55) are exo-enzymes that remove terminal non-reducing end from the arabinose containing poly- and oligosaccharides (Kozlova et al. 2015). Arabinoxylans basically composed of a linear backbone with β -1,4-linked D-xylopyranosyl units, which are partially substituted with arabinofuranosyls (Sorensen et al. 2006). During catalysis, arabinofuranosidases remove L-arabinose residues substituted at positions 2 and 3 of the β -D-xylopyranosyl (Anand et al. 2018). Therefore, α -L-arabinofuranosidases have attracted attention due to their significant role in degradation of lignocelluloses as well as effect on the activity of several other enzymes acting on lignocelluloses (Numan and Bhosle 2006).

α-Glucuronidase

Alpha-1, 2-glucuronidases (EC 3.2.1.131) are enzymes belonging to GH family 67 (Lombard et al. 2014), which act only on short xylan oligosaccharides that carry the (Me) GlcA substitution at their reducing ends (Golan et al. 2004; Nurizzo et al. 2002).

Acetyl Xylan Esterase

Acetyl xylan esterases (AXE, EC 3.1.1.72) are also very important enzymes that cleave attached acetyl side groups in the structure of xylan (Till et al. 2013). These enzymes have the potential to remove the *O*-acetyl groups from both positions 2 and 3 on the β -D-xylopyranosyl residues of acetyl xylan (Bajpai 2014).

Ferulic Acid Esterase and p-Coumaric Acid Esterase

Ferulic acid esterase (EC 3.1.1.73) is an enzymes that catalytically cleaves the feruloyl groups substituted at the 5'-OH group of arabinosyl residues present in arabinoxylans, which is known to modulate their functional properties (Latha et al. 2007). However, the ester bonds are cleaved by ferulic acid and p-coumaric acid esterases (Qi et al. 2011).

6.3 Ligninolysis

The term ligninolysis is defined as the degradation of lignin, that has been extensively researched with wood-rotting organisms, especially white-rot basidiomycetes fungi (Martinez et al. 2004; Wan and Li 2012; Falade et al. 2017). Under the process of ligninolysis, the recalcitrant lignin with the help of lignolytic enzymes is catabolized into simpler metabolites. However, multiple strategies exist in nature for the conversion of lignin, executed by a limited range of microorganisms than cellulose degradation (Cragg et al. 2015). Though lignin degradation is performed by group of microorganisms, WRF have stronger potency than that of other organisms and they are most responsible for the lignin decomposition (Tuomela et al. 2000).

The lignin is a highly insoluble complex branched organic polymer constituted by the phenylpropane units linked by carbon–carbon and ether linkages, which forms an extensive cross-linked network within plant cell wall (Rodrigo de Souza 2013). Lignin is also an amorphic three-dimensional structure composed of phenylpropanoid subunits (Datta et al. 2017). These subunits can be accounted as (1) coniferyl alcohol (guaiacyl propanol), (2) coumaryl alcohol (p-hydroxyphenyl propanol), and (3) sinapyl alcohol (syringyl alcohol) (Bajpai 2016) (Fig. 1.4).



Fig. 1.4 Important events of ligninolysis during depolymerization of complex structure of lignin: (1) lignocellulosic biomass, (2) mycelial biomass, (3) secretion of lignolytic enzymes (LiP, MnP, and lac), (4a) attack of LiP, (4b) attack of MnP, (4c) attack of laccase on lignin polymer (coniferyl alcohol, coumaryl alcohol, and sinapyl alcohol), (5) formation of resultant metabolites after attack of LiP, MnP, and laccase

6.3.1 Enzymes Involved in Ligninolysis

Ligninolytic enzymes secreted from fungal mycelia capable of degrading lignin are collectively termed as ligninases. The extracellular lignin-degrading enzymes such as lignin peroxidase (LiP), manganese-dependent peroxidase (MnP), and laccase are responsible for decomposing a variety of recalcitrant organic compounds ranging from simple phenolic molecules to many xenobiotics (Hofrichter 2002; Steffen et al. 2002; Baldrian 2006). At the time of catalytic action, laccases use molecular oxygen as electron acceptors, while both peroxidases (LiP and MnP) use hydrogen peroxide as a co-substrate (Mai et al. 2004). Both the LiP and MnP were discovered during mid of 1980s in industrially preferred *Phanerochaete chrysosporium* and were described as true lignases because of their high potential redox value (Gold et al. 2000; Martinez 2002). The major fungal members *P. chrysosporium*, *Ceriporiala cerata, Cyathus stercoreus, C. subvermispora, Pycnoporus cinnabarinus*, and *Pleurotus ostreatus* produce enzymes potentially able to degrade lignin (Kumar et al. 2015; Kumar and Wyman 2009; Madai and Abbas 2017).

Lignin Peroxidase

Lignin peroxidase (EC 1.11.1.14) is a biotechnologically very important and potentially delignifying enzyme degrading lignocellulosic materials (Harley et al. 1988; Valaskova et al. 2007). LiP is one of the key enzyme, which acts in the oxidation of non-phenol lignin structures to cation radicals (Gubernatorova and Dolgonosov 2010). Lignin peroxidase resembles horse radish peroxidase, and as highly studied this contains Fe (III) as a cofactor which is structurally pentacoordinated with four heme tetrapyrrole nitrogens and with a histidine (Hammel and Cullen 2008). It is LiP which is thought to directly attack at the C α -C β bonds of propyl side groups attached with aromatic subunits of lignin (MacDonald et al. 2012). LiPs are known to catalyze the H_2O_2 -dependent oxidative depolymerization of several non-phenolic lignin (diarylpropane), β-O-4 non-phenolic lignin model compounds, and many phenolic compounds such as guaiacol, vanillyl alcohol, catechol, syringic acid, and acetosyringone redox (Wong 2009). In consequence of LiP catalysis, formation of intermediate radicals like phenoxy radicals and veratryl alcohol radical cations through oxidizing substrates takes place in multistep electron transfers of lignin peroxidase (Fig. 1.4).

Manganese Peroxidase

The manganese peroxidase (MnP; EC 1.11.1.13) enzyme is catalytically most versatile lignin-modifying peroxidase produced by most wood-colonizing basid-iomycetes causing modification in lignin (Hofrichter 2002). This was first discovered in *P. chrysosporium* white-rot fungus (Glenn and Gold 1985; Paszczynski

et al. 1985). MnP attacks on phenolic and non-phenolic units of lignin through catalyzing lipid peroxidation reactions (Binod et al. 2011; Kuila et al. 2016). MnP contains glycosylated heme that can catalyze the oxidation of Mn^{2+} to Mn^{3+} in a H_2O_2 -dependent reaction (Zhao et al. 2015). During the enzymatic activity of MnP, Mn works like a mediator (Datta et al. 2017). Manganese ions chelated by organic acids are lactate and malonate (Wariishi et al. 1989). This oxidizes the phenolic compounds of lignin to its phenoxy radicals by chelating Mn^{3+} organic acid complex (Aehle 2007). During lignin degradation process, Mn stimulates and enhances the MnP enzymatic activity at higher levels of Mn (Rothschild et al. 1999) (Fig. 1.4).

Laccase

Laccase (EC 1.10.3.2; benzenediol-oxygen oxidoreductases) are multicopper enzymes belonging to the group of blue oxidases that catalyzes oxidation of a wide range of organic and inorganic compounds, such as diphenols, polyphenols, diamines, and aromatic amines (Gianfreda et al. 1999; More et al. 2011). This enzyme was first isolated from the extract of Japanese lacquer tree *Rhus vernicifera* (Yoshida 1883), which is widely distributed in most plants (Mayer and Staples 2002), fungi (Baldrian 2006), bacteria (Claus 2003), and in insects (Hoegger et al. 2006). Laccases have the potential to cleave β -1 and β -O-4 dimer linkages between $C\alpha$ -C β and cause $C\alpha$ oxidation and aryl-alkyl cleavages (Dwivedi et al. 2011). The ability of laccases for oxidizing phenolic compounds and to reduce molecular oxygen into water has led to intensive studies (Jolivalt et al. 1999; Xu 1996; Thurston 1994). The fungal laccases catalyze the polymerization, depolymerization, methylation, and demethylation processes of phenols (Leonowicz et al. 1985). The oxidation reactions catalyzed by laccases result in the formation of free radicals, which act as intermediate substrates for the enzymes (Ferraroni et al. 2007). Laccase also catalyzes the generation of phenoxyl radicals and their unspecific reactions, consequently leading to $C\alpha$ -hydroxyl oxidation to ketone, alkyl-aryl cleavage, demethoxylation, and C α -C β cleavage in phenolic sub-compound (Isroi et al. 2011) (Fig. 1.4). The physiological function of laccase is lignolysis by laccase-mediator system (LMS).

Almost all genus of WRF typically produce this potent enzyme the laccase (Kaarik 1965; Bollag and Leonowicz 1984). In fungi, laccases execute various important functions such as participation in pigmentation (Clutterbuck 1990), fruiting body initiation in mushroom, sporulation, and efficient detoxification (Thurston 1994), and degradation of lignin and pathogenesis (Thurston 1994; Mayer and Staples 2002). It is believed that fungal laccases are involved not only in the degradation of lignin but also in the intoxication of toxic phenols produced during the degradation (Thurston 1994; Solomon et al. 1996; Youn et al. 1995).

7 Conclusion

Based on the findings of the several studies regarding mycodegradation, it can be concluded that the fungi are the very efficient degraders and bioconverters of lignocelluloses. Several groups of fungi are able to secrete variety of lignocellulolytic enzymes under both natural and artificial conditions. The enzymes secreted by fungi attack on different bonds and linkages present in the chemical structure of lignocelluloses and simplify them into oligomeric metabolites. Therefore, the fungi in low-cost biodegradation have established an attractive biological machinery for bioconversion of lignocellulosic substances. The incorporation of this kind of technology with the capitalization of natural lignocellulosic residues into low-cost and environment-friendly high-bearing enzymatic processing of lignocelluloses establishes a sustainable bioconversion into many value-added by-products.

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Chapter 2 Fungal Biodiversity Producing Cellulase Involved in Efficient Cellulolysis



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1 Introduction

Cellulases are a class of lignocellulolytic enzymes involved in converting cellulose into simple sugars. Lignocellulose constitutes 60% of the plant cell wall and is one of its major components. Lignocellulose comprises a family of lignin, hemicelluloses, and cellulose which are sugar biopolymer and its derivative (Galbe and Zacchi 2012). Cellulase is a polymer chain of D-glucose as monomeric units. It is the most widely available organic polymer in the world having 4×10^{10} tonnes produced every year (Coughlan 1985). Cellulases from microbial sources have become the main focus due to varied biotechnological applications. These are synthesized on cellulosic material by various fungal and bacterial sources (aerobic, anaerobic, mesophilic, or thermophilic) during their growth. Fungal cellulases are known to be simpler than their bacterial counterpart having two separate domains. The two separate domains are (1) catalytic domain and (2) cellulose-binding module (CBM), having approximately 35 amino acids (Carvalho et al. 2003).

Cellulase is a 3-enzyme membered family each having a particular function. The members in this family are exo-(1,4)- β -D-glucanase which is responsible for attacking cellulose chain ends, endo-(1,4)- β -D-glucanase which acts on O-glycosidic bonds in a random order which gives different lengths of glucan chain. The third is β -glucosidases which produces cellulose by acting on β -cellobiose (Kuhad et al. 2011). The most widely used and studied microorganisms for cellulase production are *Trichoderma*, *Aspergillus*, *Cellulomonas*, *Penicillium*, *Clostridium*, etc. (Kuhad et al. 2011).

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2 Classification of Cellulases

Cellulases are lignocellulolytic enzymes responsible for converting cellulose into simple sugars. They are complex enzyme system comprises endoglucanase (endo-1,4- β -D-glucanase), exoglucanase (exo-1,4- β -D-glucanase), and β -D-glucoside glucanohydrolase(β -D-glucosidase) (Sajith et al. 2016).

2.1 Endoglucanase

Endoglucanase, which can be endo- β -1,4-D-glucanase or endo- β -1,4-D-glucan-4glucano-hydrolase, is responsible for hydrolyzing carboxymethyl cellulose (CMC) in a random fashion. Hence, they are often referred as CMCase. Accordingly, the decrease in the length of polymer results in the rise of reducing sugar concentration. Endoglucanase converts the intermediate product of cellulose hydrolysis, cellodextrins into cellobiose and glucose. These enzymes are found to be inactive when acted against crystalline cellulose (e.g. Avicel, cotton) (Sajith et al. 2016).

2.2 Exoglucanase

Exoglucanase, which can be $exo-\beta-1,4-D$ glucanase or cellobiohydrolase, degrades cellulose by splitting off the cellobiose units present in the non-reducing end of the chain. These enzymes are unable to hydrolyze hydroxyethyl cellulose and CMC derivates; however, they are found to be active against cellodextrins and amorphous solids which are present in partially degraded form. Exo-1,4-D-glucan-4-glucohydrolase or glucohydrolase are found as minor components in some cellulase derivatives (Sajith et al. 2016).

2.3 β -Glucosidase

This enzyme is responsible for hydrolyzing alkyl and aryl β -glucosides. Also, they are found to cleave cellobiose and shown to remove glucose group from the non-reducing tail of oligosaccharides thereby resulting in hydrolysis of cellulose (Sajith et al. 2016).

3 Fungal Cellulases

The synergetic action of different cellulases like exoglucanases, β -glucosidases, and endoglucanases is necessary for hydrolyzation of insoluble cellulose by microbial species, thereby converting it to simple sugars. Among the various microorganisms,

	Class of
Species of fungi	fungi
Coniophora puteana	Brown-rot
Fomitopsis sp.	
Lanzites trabea	
Poria placenta	
Tyromyces palustris	
Aspergillus species: Aspergillus niger; A. nidulans; A. oryzae; A. terreus	Soft-rot
Chaetomium species: Chaetomium cellulyticum; C. thermophilum	1
Fusarium species: Fusarium solani; F. oxysporum	
Humicola insolens; H. grisea	
Penicillium species: Penicillium brasilianum; P. occitanis; P. decumbans	
Trichoderma species: Trichoderma reesei; T. longibrachiatum; T. harzianum;	
Trichoderma atroviride	
Agaricus arvensis	White-rot
Phanerochaete chrysosporium	
Phlebia gigantea	
Pleurotus ostreatus	
Sporotrichum thermophile	
Trametes versicolor	

 Table 2.1 Fungal species with cellulolytic abilities

fungi bring about 80% of the total cellulose breakdown and are considered as dynamic cellulose decomposers. The members of the Ascomycota, Basidiomycota, Deuteromycota, and Chytrids are some of the cellulose-decomposing fungi.

Industries prefer aerobic fungal cellulases as they are highly adaptive, extracellular, and secreted in bulk during their growth. Whereas cellulase extracted from other variable types of bacteria and fungi present in the form of enzyme complexes, these cellulases are mostly occurred in membrane-bound structure such as cellulosomes. Aerobic fungi such as *Trichoderma* sp., *Penicillium, Aspergillus* sp., and *Sclerotium* are highly cellulolytic in nature. *T. reesei* is a widely studied aerobic fungi as it is capable of hydrolyzing native cellulose. Many thermophilic fungi such *as Talaromyces emersonii, Chaetomium thermophilum, Sporotrichum thermophile,* and *Thermoascus aurantiacus* are being widely studied as the cellulose fibers bulge at high temperatures making them accessible for the action of hydrolytic enzymes (Sajith et al. 2016) (Table 2.1).

4 Characteristics of Cellulases

Cellulases are lignocellulolytic enzymes that hydrolyse the β -1,4 linkages in cellulose chains. Cellulases contain non-catalytic carbohydrate-associated module/ carbohydrate-bound modules (CBMs) and/or other functionally known or unknown units which are situated at the N- or C-terminus in a catalytic segment (Zhang et al. 2018; Zhang and Zhang 2013). The major fungus for industrial cellulase production is *Trichoderma reesei*. Its genome encodes 10 cellulases and 16 hemicellulases (Meenu et al. 2014).

5 Occurrence

Microbes such as aerobic bacteria, fungi, yeast, and actinomycetes are involved in cellulase production. Fungi are the major producers of cellulase among the microbes and accounts for approximately 80% of the cellulose hydrolysis on Earth (Singh et al. 2007). Due to the constant contribution of cellulosic carbon in the form of clutter which acts as a substrate for disintegration and decomposition by the microbes, the residues of mangrove timberlands are found to be appropriate for discovering cellulose-disintegrating microorganisms (Behera et al. 2017).

6 Structure

The chemical assembly of cellulose molecule involves the β -anhydroglucose units with dominant hydroxyl groups. Each monomer component in the polymer involves primary and secondary hydroxyl assemblies. Each of the anhydroglucose units is linked together by β -1-4 glycosidic bonds (Fig. 2.1). Cellulose is a linear homopoly-saccharide and has the capacity to form wide-ranging intra- and intermolecular hydrogen bonds due to its specific chemical arrangements (Sahin and Arslan 2008).

7 Cellulase Extraction from Fungal Sources

7.1 Aspergillus

Both the submerged (SmF) and solid-state fermentation (SSF) conditions are useful in the production of cellulase by using *Aspergillus niger*. The maximal production of cellulose was observed after incubating for 72 h in SSF and 96 h in SmF. The extracellular cellulose productivity is 14.6-fold high in SSF compared to that of SmF. The optimal conditions for maximum production of cellulase by *A. niger* was found to be with pH 6, temperature 30 °C, and additives like lactose, peptone, and coir waste as substrate (Kuhad et al. 2011). In this sense, *Aspergillus fumigatus* produces cellulase by using substrates like lignocellulosic materials under the controlled condition of SSF. In order to produce the maximum cellulase production from *A. fumigatus* Z5, the use of carboxymethyl cellulase (CMCase) in cultivation environments is acknowledged. Other encouraging conditions are identified as tem-



Fig. 2.1 Structure of cellulose (Chawla et al. 2014)

perature condition at 50 °C, initial moisture 80%, pH 4.0, and initial inoculum 7% (Liu et al. 2011).

7.2 Trichoderma

Like *Aspergillus niger*, *Trichoderma reesei* is extensively used for cellulase production. With the help of controlled overexpression of the gene *Trvib-1*, the outcome of cellulase production can be improved (Zhang et al. 2018). It was observed that the recombinant *T. reesei* Vib-1 is helpful in the increasing production of cellulase as compared to similar production from the parent strain. In addition, it has been also noted that the cellulase enzyme is produced by *T. harzianum* if it is supplemented with corn cob. Apart from these, *T. viride, T. asperellum, T. koningii, T. atroviride, T. longibrachiatum*, and *T. virens* are also capable of cellulose production (Pandey et al. 2015).

7.3 Penicillium

The submerged cultures of *Penicillium echinulatum* 9A02S1, with sorbitol and cellulose, can be used for cellulase production (Ritter et al. 2013). *Penicillium funiculosum* ATCC11797 is also capable of cellulase production. It was found that the optimum composition of culture media for increased production uses Avicel, urea, KH₂PO₄, MgSO₄·7H₂O, and yeast extract (Lins et al. 2014). *Penicillium janthinellum* is also capable of cellulase production (Singh et al. 2017).

7.4 Sclerotium

Sclerotium rolfsii is an efficient producer of cellulolytic and hemicellulolytic enzymes (Sachslehner et al. 1997). On ultraviolet mutagenesis of crushed sclerotia, a mutant of *Sclerotium rolfsii* CPC 142 is formed, which is known to secrete a solution that has an effective filter paper-degrading activity in NM-2 growth medium cultures compared to the parent strain (Sadana et al. 1979).
7.5 Thermophilic Fungi

The thermophilic enzyme is gaining a lot of importance in recent times. This is because they produce thermostable cellulases. The best-known sources of cellulases such as that from species *Trichoderma*, *Penicillium*, or *Aspergillus* species are thermostable at temperature 50 °C. Chances of contamination of reactors are high by microorganisms using sugars liberated by the hydrolytic reactions, and at this temperature and the sugar recovery rate is very low. Hence to overcome this, the reaction temperature can be increased (Durand et al. 1984).

Search for new thermophilic having been taken up by many researchers to fill this need. Species like *Thermoascus aurantiacus* (Shepherd et al. 1981), *Sporotrichum thermophile* (Coutts and Smith 1976), and *Chaetomium thermophile* (Eriksen and Goksøyr 1977) are found to produce thermostable cellulase. Other patented species also producing thermostable cellulase are *Thielavia terrestris* and *Sporotrichum cellulophilum*. The species *Sporotrichum pulverulentum* was improved genetically and was compared to the species *T. reesei* with respect to the enzymes produced. The species *Sporotrichum pulverulentum* were found to be not suitable for the process of saccharification as they have low activity on the crystal-line cellulose at the temperature 50 °C.

8 Production of Cellulase

Submerged fed-batch fermentation is used for low-cost (\$10–40/kg of dry protein wt.) production of cellulase by companies (e.g. Iogen, Genencor) which gives them a large amount of crude cellulase production of more than 100 g/L of broth used. Such companies use strains of *Aspergillus* and *Trichoderma* as cellulase derivative (Zhang and Zhang 2013).

Cellulase producing microorganisms use carbohydrate as their energy source as they are not capable of consuming proteins and thus degrade carbohydrates for growth and metabolism. Species like *Cellulomonas* and *Cytophaga* and other fungi have the capability to use other carbohydrates as an energy source with cellulose (Stapleton et al. 2004; Poulsen and Petersen 1988), whereas anaerobic cellulolytic species only stick to cellulose and its derivatives (hydrolytic products). Some fungi have the capability to produce high levels of cellulase and these are extensively studied. Species that can secrete extracellular proteins are Trichoderma reesei, Humicola, Penicillium, and Aspergillus which are responsible for converting cellulase to glucose (Zeikus and Ng 1982). Other common fungal species are Humicola, Trichoderma, Aspergillus, Penicillium, Pseudomonads, Cellulomonas, Actinomycetes, Streptomyces etc. Some fungal strains are able to secrete a cellulase enzyme complex that can hydrolyse cellulose. Species like T. reesei, H. insolens, A. niger, Thermomonospora fusca, and Bacillus are among the major fungal species which are exploited for commercial purpose (Sukumaran et al. 2005) (Table 2.2).

Strains of fungi	Substrates used
Aspergillus niger	Palm kernel cake
Aspergillus nidulans	2 mM of p-nitrophenyl-fl-o-glucoside in 50 mM acetate buffer
Aspergillus oryzae	p-nitrophenyl β -D-cellobioside
Fusarium solani	H ₃ PO ₄ -swollen cellulose, <i>O</i> -(carboxymethyl) cellulose, cotton fiber
Humicola insolens	CMC (Carboxymethylcellulose) for to determining activity profiles of endoglucanases (EG), cellohexaose, and cellotriose for determining the activity profiles of CBH (cellobiohydrolase I) and CBH II
Humicola grisea	<i>p</i> -nitrophenyl (PNP)- β -D-saccharides
Penicillium brasilianum	Azo-arabinoxylan or azo-CMCase was used as substrate for determination of endoglucanase and endoxylanase activity
Penicillium occitanis	Combination of wheat flour and wheat bran remnants with esparto grass paper pulp and cellulose substrates were shown to obtain high cellulase yields

 Table 2.2
 Microorganisms involved in the production of celluloase (Sethi et al. 2013)

9 Biotechnological Applications

Cellulase enzyme finds a wide range of application in agriculture, food, fermentation, etc., industries (Table 2.3).

10 Conclusion

The degradation of lignocellulosic waste by microbes is the coordinated action of several enzymes in which the cellulases are considered to be the most prominent enzyme. The researches on microbial cellulases have not only resulted in significant scientific knowledge about the enzymes but have also revealed their enormous potential in biotechnology. As a result, isolation and screening of novel cellulaseproducing microorganisms that can produce profuse amounts of the enzyme at extreme conditions using cheap substrates are necessary. Although there are various species producing xylanase enzyme, species of Trichoderma, Aspergillus, and Penicillium are exploited for the wide variety of cellulolytic enzymes that they possess. This scenario has however shifted in the recent days to other promising strains producing xylanase that show better increased thermostability, activity, and better stability under alkaline or acidic medium. Recently, strategies are being carried out to convert cellulase into useful products, and to this end cellulase have been identified as key enzymes with diverse industrial applications. The bulk production of cellulase at an economically viable rate is the major bottleneck in commercial applications of cellulase.

Industry	Application
Agriculture	Enzyme combinations of cellulase, pectinase, and hemicellulose improve crop growth. Also, enzymes like β -glucanases give resistance to diseases to the plant species by degrading the cell wall of the pathogenic species (Kuhad et al. 2011). Cellulase extracted from certain fungal strains can degrade the cell wall of species pathogenic to plant and help them control diseases
Bioconversion	Conversion of lignocellulosic material into useful products such as biofuels like methane and ethanol is considered to be a major industrial application of cellulase. Other products including organic acids, lipids, and single-cell proteins are also produced from cellulase. In addition, production of energy- rich animal feed with improved absorption is successfully done with the help of cellulase. Furthermore, cellulase improves performance of ruminants and helps in preservation of high-quality fodder
Detergent	Production of cellulase for manufacturing detergents that can dissolve the cellulose related stains. Cellulase imparts quality cleaning mechanism without causing damage to fibers along with enhanced color brightness and removal of dirt and rough protuberances in fabrics. In addition, it prevents ink particles from redeposition
Fermentation	Cellulase improves malting, mashing, pressing, and primary fermentation due to cellulose activity. It enhances wine aroma, stability, color extraction of grapes, and beer quality. Cellulase reduces wort viscosity which makes filterability more feasible
Food	Cellulase finds a varied application in the food processing industry as well. One such application is the production of juices extracted from fruits and vegetable. They are also used in clarification, stabilisation, and increasing the yield of the juices produced (De Carvalho et al. 2008). Enzymes such as cellulose and hemicellulose also find application in the extraction of olive oil
Pulp and paper	Cellulase acts as a co-additive in processes such as pulp bleaching and biomechanical pulping. It improves the properties in fiber brightness, strength, pulp freeness, and cleanliness as well as improving drainage in pulping and paper milling. Cellulase eases deinking process and reduces energy and chlorine demands. Production of biodegradable products such as cardboard, sanitary napkins, and paper towels is carefully done with the help of cellulase. Moreover, cellulase enzyme with xylanase enzyme is used for the purpose of deinking used paper
Textile	Cellulase is known to increase fabric quality and stability. It helps in biostoning of jeans, biopolishing, and improved absorbance of textile fibers, garment softening, color brightness restoration, and in excess dye removal from fabrics due to cellulose activity. Cellulase enzyme is used in wet processing and it improves appearances. These are used for biological polishing of cotton and other fabrics. These are also used for biological stoning of jeans. In this, the cellulase enzyme is directly placed on the fabric, which breaks off small fibers in the fabric from the surface of the yarn and thereby loosens the dye which can now be removed by washing
Soap and detergent industry	Cellulase is used with lipase and protease enzyme in detergents. The cellulase counterparts may improve the dirt-removing ability of cotton blend garments and gives the garments a more glossy, intense-colored, and brighter appearance

Table 2.3 Industrial application of cellulase (adopted from Kuhad et al. 2011; Singh et al. 2007;Bhat 2000; Dhiman et al. 2002)

(continued)

Industry	Application
Animal feed industry	Pretreatment of stored fodder with cellulase improves the nutritional value of the fodder. They also help in eliminating the anti-nutritional components in the fodder like the non-starch polysaccharides like cellulose into other easily absorbed parts
Others	Cellulase improves extraction of carotenoids and enhancement of its oxidation and color stability properties. It enhanced olive oil extraction with good quality and olive paste malaxation improvement. Cellulase plays a huge role in waste management and risk reduction of cellulosic biomass by converting it to beneficial products such as enzymes, energy sources, and nutrients. Cellulases are helpful in the production of hybrid molecules like cellulosomes for improvement of cell surface properties to utilize insoluble substrates with the help of these cellulosomes. Apart from that, cellulase converts hemicellulose and is used for production of wine. It improves the clarification of must, production of wine, quality, stability, etc. Wines can be made more aromatic by the enzyme β -glucosidases

Table 2.3 (continued)

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Chapter 3 Fungal Diversity and Enzymes Involved in Lignin Degradation



Harit Jha

1 Introduction

The term lignin is derived from the Latin word *lignum* meaning wood. Lignin is an amorphous polymer formed by the free-radical-mediated polymerization of three different phenylpropane units (p-coumaryl, coniferyl and sinapyl alcohol). The amount and chemical composition of lignin vary depending on the source. Lignin in gymnosperms is prominently composed of dehydrogenation polymers of coniferyl alcohols, while angiosperm lignin contains mixed dehydrogenation polymers of coniferyl and sinapyl alcohol. The main purpose of lignin is to give the plant structural support, impermeability and resistance against microbial attack and oxidative stress. It is also not soluble in water and soluble and inactive optically; thus, lignin is difficult to degrade.

2 Structure of Lignin

The basic building unit for all lignins is the phenylpropane unit. These units are linked together by C–O–C and C–C bonds in such a way as to form a threedimensional polymer by cross-linking (Fig. 3.1). The phenylpropanoid monomers of the lignin show similarity; however, the functional group's substitution on the phenyl ring differs. A single hydroxy or a methoxy group is present in the H-structure (4-hydroxy phenyl), two hydroxy or methoxy groups are present in the G-structure (guaiacyl) and three hydroxy or methoxy groups are present in the S-structure (syringyl). The different types of bonds found in lignin structure include carbon–carbon and carbon–oxygen bonds between various monomers. A carbon–oxygen

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Fig. 3.1 Structure of lignin



Fig. 3.2 Lignin monomeric building blocks

link between a p-hydroxy moiety and the β - end of the propenyl group (β -O-4) is the most common bond present in lignin (Fig. 3.2) (Kukkola et al. 2004).

3 Applications of Lignin

3.1 Bioconversion of Lignocellulose into Fuel and Useful Chemicals

Technologies based on the efficient conversion of low-quality or waste lignocellulosic residues into fuel and industrially important chemical present long-term solution to the finite supplies of petroleum. Many efforts have been made to develop economically viable processes derived from the cellulose and hemicellulose and lignin components. Low-molecular-weight chemicals such as vanillin and dimethyl sulfoxide are also currently prepared on an industrial scale from lignosulfonates and kraft lignin, respectively. The range of low-molecular-weight compounds formed during lignin degradation by white-rot fungi and actinomycetes has raised the possibility of using controlled biodegradation to generate some of the high-value chemicals currently derived via the petrochemical industry (Ramasamy 1993).

Selective modification of byproducts of lignin and waste lignocellulose (demethylation, hydroxylation) in order to furnish additional reactive group has been suggested as a means of providing chemical feedstock for the manufacture of high-value polymers. Lignin degradation could improve the nutritive value of the agricultural wastes. It has been showed that presence of lignin in carbohydrate-rich residues restricts its digestibility by various enzymes of rumen fluid. Chemical and physical delignification methods have been used extensively, but with emerging and classified knowledge of lignin biodegradation, more attention is centred on delignification treatment based on lignin-degrading microorganisms particularly the basidiomycetes fungi. Digestibility of wheat straw, wood and sunflower stalk was improved using white-rot fungi pretreatment. The list of uses of lignin for which patent has been granted are as follows:

- 1. Drilling and additive.
- 2. Rubber additive.
- 3. Dyestuff filler.
- 4. Ink with lignin-based colourant.
- 5. Lignin derivative surfactant.
- 6. Cationic lignin amine surface-active agent.
- 7. Lignin-epoxysuccinate adduct as dispersant and metal complexing agent.
- 8. Lignin-reinforced polymer.
- 9. Polyvinyl alcohol and lignin sulfonate-containing adhesive.
- 10. Synthetic lignin-polyisocyanate resin.
- 11. Polyoxyalkylene ether polyols from lignin.
- 12. Isocyanate lignin products.

- 13. Copolymerizable colourant in polyesters.
- 14. Lignin use in pharmaceuticals.
- 15. Acid polymerized lignosulfonate as extender.
- 16. Polymerizable lignin derivatives.
- 17. Binder for glass fibre insulation.
- 18. Lignin-sulfonate-phenol-formaldehyde glue system.
- 19. Controlled released pesticides.
- 20. Lignin-urea composition for flame proofing.

4 Biodegradation of Lignin

Microbial degradation of lignin is an important step in carbon cycle as it enables the microorganisms to utilize the plant carbohydrates, overcoming the obstacle of lignin (Martínez et al. 2005; Kersten and Cullen 2007). Lignin removal is also a central aspect in industrial uses of cellulosic biomass, such as bioethanol production and manufacture of cellulose-based chemicals and materials, including paper cellulose pulp manufacture basically consists in breaking down (chemically or mechanically) the middle lamella in such a way that wood fibres are individualized (Sixta et al. 2006). In industrial applications, biotechnology based on lignin-degrading microbes and their enzymes can contribute to more efficient and environmentally sound use of renewable lignocellulosic feedstocks for sustainable production of materials, chemicals, biofuels and energy.

Lignin is a large polymer and the major challenges faced by a microorganism in its degradation are as follows:

- 1. The degrading system must be extracellular because lignin polymer (600–1000 kDa) is large and its direct uptake by microbial cells is unlikely.
- 2. The degradative system must be oxidative rather than hydrolytic because the structure is comprised of interunit carbon–carbon and ether bonds.
- 3. The ligninolytic agent must be less specific because lignin polymer is stereo-irregular.

Lignin degradation is performed by microorganisms like fungi and bacteria (Voeller et al. 2016). Lignin-degrading enzymes act to break the barrier of the plant cell wall for subsequent access to cellulose complex by cellulases and, eventually, may replace physicochemical processes, if the process becomes sufficiently rapid. Biological extraction of lignin employs wood-degrading microorganisms, including white-, brown-, and soft-rot fungi and bacteria to modify the chemical composition and/or structure of the lignocellulosic biomass so that the modified biomass was more amenable to enzyme digestion. Advantages of biological pretreatments are low energy requirement and mild operation conditions. However, long incubation time is required for enzymatic/biological treatment due to low activity of enzymes (Cardona and Sánchez 2007). Numerous studies have been carried out to find the microbes, which degrade lignin polymer. Lignin is however not biodegraded anaer-

obically, only minor modification in functional groups and nonlignin compounds takes place and some low-molecular-weight materials are degraded (Hatakka 1994).

The white-rot fungi are the main producers of ligninolytic enzymes. Among the several hundred species of white-rot fungi belonging to a variety of fungal families among the *Holobasidiomycetidae* are the *Agaricaceae*, *Hydnaceae*, *Corticiaceae*, *Polyporaceae* and *Thelephoraceae*. There are also a few Ascomycetes in the *Xylariaceae* that causes white -rot (Ma et al. 2016). The most efficient ligninolytic microorganisms are, however, Basidiomycetes (Kent et al. 1987). Various white-rot fungi being used to degrade lignin preferentially over cellulose or hemicelluloses are *Polyporus berkeleyi*, *Pycnoporus cinnabarinus*, *Fome sulmarius*, *Polyporus resinosus*, *Pleurotus ostreatus*, *Phlebia radiata* and *Merulius tremellosus* (Fritsche and Hofrichter 2000). *Phanerochaete chrysosporium* was the first fungus used for pulp delignification under conditions optimized for the mineralization of synthetic lignin.

The facts about lignin degradation derived from studies on white-rot fungi also suggest oxidative, extracellular and non-specific nature like there is substantial increase in carboxyl and carbonyl groups, and a decrease in hydrogen content in lignin residues left after decay oxidation of α -carbon of the propyl side chain to a carbonyl group hydroxylation and oxidative cleavage of the aromatic rings have been also observed. The demonstration of the insertion of an oxygen-derived hydroxyl group at beta-carbon of the side chain utilizing beta-1 lignin substructure compounds has authenticated the oxidative nature of lignin biodegradation (Shoemaker and Leisola 1990).

Bacterial lignin degradation has been most extensively studied in actinomycetes particularly *Streptomyces sp., Streptomyces viridosporus* and *Streptomyces setonii. Nocardia, Pseudomonas* and *Corynebacterium* have also been found to grow on lignin-related phenols. Lignin-related chemicals can be further processed by bacteria. For instance, the aerobic gram-negative soil bacterium *Sphingomonas paucimobilis* is able to degrade lignin-related biphenyl chemical compounds. A *Xanthomonas* strain has been reported to mineralize 30% synthetic [¹⁴C] lignin in 20 days. However, generally bacteria are primarily responsible for the degradation of plant material in certain environments, but mineralization of lignin was minimal.

Wood decay and lignin degradation caused by various species of ascomycetes and fungi imperfecti include *Fusarium solani*, *Trichoderma strains*, *Trichoderma harzianum*, *Chaetomium piluliferum* and *Aspergillus fumigatus*. A few species of ascomycetes, e.g. species of *Xylaria*, *Libertella* and *Hypoxylon*, cause white-rot decay accompanied by substantial lignin loss. Lignin biodegradation by non-whiterotting basidiomycetes has been reported and it includes Gastromycete Cyathus *stercoreus*, *Ectomycorrhizal* fungi *Cenococcum*, *Amanita*, *Tricholoma* and *Rhizopogan*. These fungi mineralize ¹⁴C-labelled lignin, albeit slowly.

The white-rot basidiomycetes degrade lignin rapidly and extensively than any other microbial groups. They invade the lumens of wood cells, where they secrete enzymes that degrade lignin and other wood components. The electron microscopic studies showed that lignin is degraded at some distance from the hyphae and is removed progressively from the lumens towards the middle lamella. During its min-

Class	Microorganism	Enzymes	Reference	
Ascomycota	Magnaporthe grisea, Myrothecium verrucaria	Laccase	Dashtban et al. (2010); Su et al. (2018)	
Ascomycota	Melanocarpus albomyces	Laccase	Hakulinen et al. (2002)	
Ascomycota	Neurospora crassa	Laccase	Znameroski et al. (2012)	
Ascomycota	Xylaria sp. and Hypoxylon sp., xylariaceous	Laccase	Dashtban et al. (2010); Pointing et al. (2005)	
Basidiomycota (white-rot)	Trametes versicolor, Trametes pubescens, Cyathus bulleri	Aryl-alcohol oxidase, laccase	Leonowicz et al. (2001); Rana et al. (2018)	
Basidiomycota (white-rot)	Pleurotus sajor-caju, Pleurotus eryngii, Pleurotus ostreatus	Aryl-alcohol oxidase, VP, laccase	Kalmi et al. (2008); Snajdr and Baldrian (2007)	
Basidiomycota (white-rot)	Phanerochaete chrysosporium	MnP, aryl- alcohol oxidase, VP, laccase	Jager et al. (1985)	
Basidiomycota (white-rot)	Bjerkandera adusta, Bjerkandera <u>fumosa</u>	VP		
Basidiomycota (white-rot)	Fomes annosus	Laccase	Papinutti and Martınez (2006); Leonowicz et al. (2001)	
Basidiomycota	Pholiota mutabilis	Laccase	Leonowicz et al. (2001); Balaes et al. (2017)	
Basidiomycota	Coniophora puteana	MnP, aryl- alcohol oxidase, VPs, laccase	Memić et al. (2017)	
Basidiomycota	Chaetomium thermophile	MnP, aryl- alcohol oxidase, VPs, laccase	Chefetz et al. (1998)	
Basidiomycota	Pycnoporus coccineus, Pycnoporus sanguineus	MnP, aryl- alcohol oxidase, VPs, laccase	Uzan et al. (2010)	
Basidiomycota (white-rot)	Pycnoporus cinnabarinus	Laccase	Eggert et al. (1996)	
Basidiomycota	Phlebia radiata, Phlebia tremellosa	LiP, MnP	Hatakka (1994); Fritsche and Hofrichter (2000); Maciel et al. (2010)	
Basidiomycota	Ceriporiopsis subvermispora	MnP	Hatakka (1994); Fritsche and Hofrichter (2000)	
Basidiomycota	Nematoloma frowardii	MnP	Sack et al. (1997); Fritsche and Hofrichter (2000)	
Basidiomycota	Stropharia rugosoannulata	MnP	Hatakka (1994); Fritsche and Hofrichter (2000)	
Basidiomycota	Armillaria tabescens	Laccase	He et al. (2014)	
Basidiomycota	Ganoderma lucidum	LiP, MnP and laccase	Hariharan and Nambisan (2013)	

 Table 3.1
 Diversity of ligninolytic fungi

(continued)

Table 3.1	(continued)
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Class	Microorganism	Enzymes	Reference
Basidiomycota	Agaricus bisporus	LiP, MnP and laccase	Bonnen et al. (1994)
Basidiomycota	Marasmius quercophilus	Laccase	Farnet et al. (2004); Dedeyan et al. (2000)

eralization by white-rot fungi, lignin undergoes oxidative changes including aromatic ring cleavage and progressive depolymerization with release of low-molecular-weight fragments (Ward et al. 2004) (Table 3.1).

5 Enzymatic System Involved in the Lignin Biodegradation

Lignin peroxidase (LiP) was first reported from the white-rot fungus *Phanerochaete chrysosporium* with a variety of lignin-degrading reactions. There are also several other microbial enzymes that are believed to be involved in lignin biodegradation, such as LiPs (also 'ligninase', EC number 1.14.99), manganese peroxidase (MnP) (EC 1.11.1.14) and laccase (EC 1.11.1.13. Lignin degradation by fungi is an oxida-tive process and phenol oxidases are the key enzymes (Kuhad et al. 1997; Leonowicz et al. 1999). Of these, LiPs, MnPs, versatile peroxidases (VPs), dye-decolourizing peroxidases (DyPs) and laccases from fungi have been best studied. The oxidation of substrates by LiPs and MnPs occurs through two successive one-electron oxidation with intermediate cation radical formation. These enzymes act non-specifically through the generation of lignin free radicals, which are unstable and undergo a variety of spontaneous cleavage reactions.

Laccase has broad substrate specificity and oxidizes phenols and lignin substructures with the formation of oxygen radicals. Bacteria are relatively rich in another type of peroxidase, the so-called DyPs (EC 1.11.1.19) (Van Bloois et al. 2010; Gonzalo et al. 2016). One group of accessory enzymes are oxidases, e.g. arylalcohol oxidase (AAO) and glyoxal oxidase (GIO). They produce peroxide, which is essential for peroxidase activity. Other accessory enzymes include dehydrogenases that reduce lignin-derived compounds (e.g. aryl-alcohol dehydrogenase and quinone reductase) and those that modify lignin by generating hydroxyl radicals (e.g. cellobiose dehydrogenase). H_2O_2 -producing enzymes and oxidoreductases are also involved in the lignin degradation processes.

The fungal peroxidases LiPs, MnPs and VPs are all heme-containing glycoproteins which require hydrogen peroxide as an oxidant and have heme (protoporphyrin IX) as their prosthetic group coordinated by two highly conserved histidine residues. The heme cofactor is located in an internal cavity connected by two access channels. The main channel is used by hydrogen peroxide and the second is the site where MnP and VP oxidize Mn²⁺ to Mn³⁺. Typically for LiP activity, the amino acid residue needed is a tryptophan, Trp171, in the isozyme LiPA (LiP H8) of *P. chrysosporium*. Tryptophan exposed on the LiP protein surface is conserved in LiP sequences and also in VPs. It is assumed that it takes part in long-range electron transfer (LRET) from a protein radical at the surface of the enzyme, which would act as the substrate oxidizer, to the heme cofactor. This could allow the enzyme to oxidize bulky substrates such as polymeric lignin that cannot directly contact the oxidized heme in the active centre of LiP or VP.

5.1 Lignin Peroxidase (EC 1.11.1.14)

LiP (at first designated as 'ligninase') was discovered in cultures of *Phanerochaete chrysosporium* after many years of research (Glenn and Gold 1983). The glycosylated enzyme was identified and described independently by two American teams and a Japanese group in supernatant of nitrogen and carbon limited cultures of the corticoid basidiomycete *Phanerochaete chrysosporium* that were flushed with pure oxygen. It has also been reported in *Phlebia radiata*, *Phlebia tremellosa*, *Trametes versicolor*, etc. (Hatakka 1994). LiP comprises of heme in the active site. Its molecular mass varies between 38 and 47 kDa. Its catalytic cycle resembles that of horse radish peroxidases and it catalyses several oxidations in the alkyl side chains of lignin-related compounds. The major reactions involve C–C cleavages in the side chains of lignin subunit, oxidation of veratryl alcohols and related substrates to aldehydes or ketones, intradiol cleavage of phenyl glycol structures and hydroxylation of benzylic methylene groups. It is also active against highly methylated lignin (Kent et al. 1987).

LiP is also capable of cleaving aromatic rings via one-electron subtraction and subsequent incorporation of oxygen. The optimum activity of LiP is extremely low (between pH 2.5 and 3.0), while the enzyme is acidic (pI 3–5). It is also active against highly methylated lignin. The catalytic cycle of both peroxidases is similar to other peroxidase enzymes, in which the resting state of the enzyme contains ferric heme, which reacts with hydrogen peroxide to form a compound I oxo-ferryl intermediate (two-electron oxidized form), and subsequently a compound II intermediate (one-electron oxidized form). However, two aspects in their molecular structure differentiate ligninolytic peroxidases from other peroxidases: first, a heme environment that confers high redox potential to the oxo-ferryl complex due to the position of N3 of the side chain of the proximal histidine residue, increasing its electron deficiency and increasing the redox potential of theoxo-ferryl complex); secondly, the presence of binding sites for oxidation of their specific substrates, including non-phenolic aromatics in the cases of LiP and Mn²⁺ in the case of MnP.

Ferriperoxidase +
$$H_2O_2$$
 / Compound I + H_2O (3.1)

Compound I + AH2 / Compound II + AH (3.2)

Compound II + AH2 / Ferriperoxidase + AH + H₂O

5.2 Manganese Peroxidase (EC 1.11.1.13)

MnP was also discovered in cultures of *Phanerochaete chrysosporium* by Kuwahara et al. (1984). The enzyme resembles LiPs; it is extracellular, glycosylated and contains heme as the prosthetic group. MnP is also expressed in multiple forms with molecular weights from 40 to 48 kDa and pI is between 2.9 and 7.0. Besides Phanerochaete chrysosporium, MnPs have also been found in many other white-rot and soil litter decomposing basidiomycetes, e.g. Trametes versicolor, Phlebia radiata, Ceriporiopsis subvermispora, Agaricus bisporus, Nematoloma frowardii and Stropharia rugosoannulata. The catalytic cycle of MnP resembles that LiP including native ferric enzyme as well as peroxidases compound I and compound II redox states (Fritsche and Hofrichter 2000). However, in MnP during reductive reactions, Mn²⁺ is oxidized to Mn³⁺. The stabilization of the Mn³⁺ ions to high redox potential is achieved by chelation with organic acids such as lactate, succinate, malonate, oxalate, malate or tartrate. Subsequently, oxidation of phenolic lignin structure occurs due to action of chelated Mn³⁺acting as diffusible redox mediator. The enzyme has been successfully applied to depolymerize ¹⁴C-labelled synthetic and natural lignins in vitro; as a result, low-molecular-mass, water-soluble lignin fragments and significant amounts of CO_2 have formed (Gold et al. 2000).

MnP has been shown to generate H_2O_2 in the oxidation of certain thiols (e.g. glutathione) and NAD(P)H₂. Recently, evidence was provided that the enzyme is capable of acting efficiently in the absence of external H_2O_2 by oxidizing organic acids (e.g. oxalate, malonate, malate) in 'oxidase-like' autocatalytic reactions involving the transient formation of several radical species.

MnP (EC 1.11.1.13), an extracellular glycosylated enzyme, was found to be produced by *Phanerochaete chrysosporium* and a variety of lignin-degrading white-rot fungi. MnPs have been found and purified from various organism, viz. *Dichomitus squalens, Ceriporiopsis subvermispora, Lentinus edodes, Trametes versicolor, Phlebia radiata, Pleurotus ostreatus* and *Panus tigrinus.*

5.3 Fungal Laccase (EC 1.10.3.2)

It is an enzyme secreted into medium by mycelia of basidiomycetes, ascomycetes and deuteromycetes. It could be induced by anilines, methoxy phenolic acids, lignin preparations, heat shock, 2, 5-xylidine and ferulic acid. It is shown in *Trametes versicolor, Fomes annosus, Pholiota mutabilis, Pleurotus ostreatus*, etc. (Leonowicz et al. 2001).

Initially discovered in the Japanese lacquer tree *Rhus vernicifera*, laccases have since been found in many other plants and insects. Laccases have been extensively studied in white-rot fungi, such as *Lentinus tigrinus*, *Pleurotus ostreatus* D1,*Cerrena unicolor* strain 137, *T. versicolor*, *Trametes sp.* strain AH28–2, *Trametes pubescens* and *Cyathus bulleri*, brown-rot fungi, including *Coniophora puteana*, and ascomycetes such as *Melanocarpus albomyces*, *Chaetomium thermophile*, *Magnaporthe*

grisea, Myrothecium verrucaria 24G-4 and Neurospora crassa. Laccase genes have been detected in different fungal species including *Pycnoporus cinnabarinus*, *Cyathus* sp., *Pycnoporus coccineus*, *Pycnoporus sanguineus* and also in *xylariaceous* ascomycetes *Xylaria* sp. and *Hypoxylon* sp. Laccases were also detected in some bacteria such as *Bacillus subtilis* and *Klebsiella aerogenes*, *Bacillus licheniformis* and *B. subtilis*, while other bacteria, like *Streptomyces griseus*, produce a laccase-like phenol oxidase (http://www.brenda-enzymes.org).

Laccase is an enzyme, which oxidizes a variety of aromatic hydrogen donors. Thus, it catalyzes the removal of an electron and a proton from phenolic hydroxyl or aromatic amino groups to form free phenoxy radicals and amino radicals, respectively (Claus 2004). Copper-containing laccase has four copper atoms all in the 2⁺ oxidation state in the active site (i.e. blue oxidase) and does not only oxidize phenolic and methoxyphenolic acids but also decarboxylates them and attack their methoxyl groups through demethylation or demethoxylation. Laccase reacts with polyphenols and other lignin-derived aromatic compounds which can be polymerized and depolymerized or act as low-molecular-weight mediator.

Laccase can be used as a bleaching agent in the pulp and paper industry. It helps in removal of colour by oxidation of lignin and other chromogenic components. Laccase is also used as a stabilizer during wine processing, as a dechlorinating agent and for removal of phenolic and other aromatic pollutants from natural and industrial wastewaters. Lignin degraded by LiPs provides a substrate for laccase; both enzymes can be considered as 'partners' in certain biotransformation routes of lignin (Boominathan and Reddy 1992). The oxidation of phenolics generates phenoxy radicals and guinoid intermediates, which are subsequently transformed into dimers and insoluble polymers, which can be easily removed from water after sedimentation. Laccases structure is composed of four copper atoms termed Cu T1 (where the reducing substrate place) and trinuclear copper cluster T2/T3 (where oxygen binds and is reduced to water). Laccase oxidizes four substrate molecules to obtain sufficient electron (4) to reduce molecular oxygen. Thus, for complete reduction of molecular oxygen to water, oxidation of four reducing substrate molecules is necessary. The oxidation of substrate by laccase is one-electron reaction. The reaction generates free radical. This unstable free radical may either undergo second enzyme catalyzed reaction or resort to non-enzymatic reactions like hydration and polymerization. In general terms, substrate oxidation by laccase is a one-electron reaction generating a free radical. The initial product is typically unstable and may undergo a second enzyme-catalyzed oxidation or otherwise a non-enzymatic reaction such as hydration, disproportionation or polymerization. The bonds of the natural substrate, lignin, that are cleaved by laccase include C α -oxidation, C α -C β cleavage and aryl-alkyl cleavage (Madhavi and Lele 2009). Laccases can even catalyze the breakdown of non-phenolic lignin structures in the presence of redox mediators, including the cleavage of β-O-4 linkages. Mediators typically used in laccase-mediator systems include 1-hydroxybenzotriazole (1-HBT, 2,2'-azinobis(3-ethylbenzothiazoline-6-sulfonic acid) (ABTS)) and the natural mediator (4-hydroxy-3,5-dimethoxyacetophenone) or acetosyringone.

5.4 Versatile Peroxidases (EC 1.11.1.16)

VPs have the properties of both LiPs and MnPs. It can catalyze a wide array of substrates. They can oxidize various phenolic and non-phenolic substrates and can also oxidize Mn²⁺ to Mn³⁺ like MnP. VP oxidizes typical LiP substrates, e.g. VA, methoxybenzenes and non-phenolic model lignin compounds, as well as Mn²⁺. The structure of VP includes 11–12 helices, four disulfide bridges, two structural Ca²⁺ sites, a heme pocket and a Mn²⁺-binding site similar to that of MnP. The substrate specificity of VPs is similar to that of LiPs, including oxidation of high and medium redox potential compounds. Yet, VP also oxidizes azo-dyes and other non-phenolic compounds with high redox potential in the absence of mediators. VP was detected in members of the genera *Pleurotus* [*Pleurotus eryngii*, *P. ostreatus*] and Bjerkandera [*Bjerkandera adusta*, *B. fumosa*]. VPs are glycoproteins secreted as several isoen-zymes with a molecular mass ranging between 40 and 45 kDa with a pI ranging between 3.4 and 3.9. The basic catalytic cycle of VPs is similar to those of other peroxidases with the two intermediary compounds I and II but is more complex due to a more diversified pool of potential substrates.

5.5 Dye-Decolourizing Peroxidase (EC 1.11.1.19)

DyP-type peroxidases were first found in fungi and subsequently in bacteria. They are not included in the classical plant/microbial peroxidase superfamily as they differ in structural and functional aspects. DyPs have a protomer weight of around 40–60 kDa and various oligomeric states have been observed (Colpa et al. 2014). They belong to the peroxidase-chlorite dismutase superfamily of proteins and contain a non-covalently bound heme b cofactor (Zámocký et al. 2015). DyP-type peroxidases can degrade phenolic dyes and lignin-like compounds and act on non-phenolic lignin model compounds containing β -O-4 linkages. Most DyP substrates are too big to enter the active site and are therefore unable to interact directly with the heme cofactor; thus, similar to the typical fungal LiPs, MnPs and VPs, DyPs also seem to be able to promote lignin degradation by oxidizing redox mediators. Redox mediators as veratryl alcohol, monophenolic substrates and Mn²⁺.

Peroxide-Generating Enzymes The peroxidases LiP and MnP require extracellular H_2O_2 for their in vivo catalytic activity. The H_2O_2 -producing enzymes are thus required for the function of peroxidases (Blanchette 1995).

Glucose Oxidase (EC 1.1.3.4) It is a FAD-dependent oxidase that can be found inside and outside fungal cells and many other organisms during the enzymatic catalysis. Glucose is oxidized first to the corresponding gluconolactone with

simultaneous reduction of FAD to FADH₂. In the second step, FADH₂ is oxidized back to FAD by dioxygen (O_2) while H₂O₂ is formed (Buswell and Odier 1987).

 β D Glucose + E FAD \rightarrow D Glucose 1,5 lactone + E FADH₂

E FADH₂ + $O_2 \rightarrow E$ FAD + H_2O_2

In absence of glucose oxidase (GO) activity, *Phanerochaete chrysosporium* did not show ability to degrade lignin.

Glyoxal Oxidase (EC 1.2.3.5) GlO is a FAD-dependent, H_2O_2 -generating enzyme discovered in growing cultures of *Phanerochaete chrysosporium*. Enzyme purification and characterization revealed that among others it oxidizes glyoxalic acid into oxalic acid while generating H_2O_2 . In this way, GlO supports peroxidases activity and supplies chelators for Mn³⁺ (Hatakka 1994).

Aryl-Alcohol Oxidase (EC 1.1.3.7) It is an extracellular, FAD-dependent enzyme which was discovered in *Trametes versicolor* and later also found in *Phanerochaete chrysosporium, Pleurotus sajor-caju, Pleurotus eryngii*, etc. It is a glycoprotein and oxidizes a number of aromatic alcohols (aryl- α and α - β -unsaturated γ -aromatic alcohols) to the corresponding aldehydes while oxygen is reduced to H₂O₂.

6 Superoxide Dismutase (EC 1.15.1.1)

Superoxide dismutase (SOD) is an almost universal enzyme of aerobic microorganisms. It catalyzes the dismutation of the superoxide anion radicals to molecular oxygen and H_2O_2 in all oxygen-metabolizing organisms and even in certain anaerobes. SOD thus protects organisms from toxic effects of the reactive superoxide radical (O_2 SOR). All-known SODs are metal proteins containing iron (FeSOD) or manganese (MnSOD) or copper plus zinc (Cu–ZnSOD) or nickel (NiSOD) as prosthetic group.

$$2O_2^{-} + 2H \rightarrow H_2O_2 + O_2$$

In summary it appears that H_2O_2 required for a LiP and MnP may be supplied by several different oxidases (Kent et al. 1987).

The ligninolytic systems composed of H_2O_2 -generating enzymes and peroxidases MnP, LiP and laccase have been shown to degrade a large number of recalcitrant compounds, xenobiotics and the organopollutants. The ligninolytic enzymes especially laccase and peroxidase are involved in ring cleavage and the breakdown of aromatic and phenolic (Leonowicz et al. 1999).

7 Conclusion

The ligninolytic enzyme system is found in mainly Basidiomycota and Ascomycota fungi. The ligninolytic system comprises of hydrogen peroxide generating system and peroxidases and oxidases. The enzymes act in tandem to degrade phenolic as well as non-phenolic substrates. This broad specificity is utilized for the bioremediation of xenobiotics, dyes and industrial pollutants. Enzymes can also be used to generate value-added products from the lignocellulosic wastes. Ligninolytic enzymes are found in fungi and in some aerobic bacteria; however, most of the reported lignin-degrading fungi are from Ascomycota and Basidiomycota. The enzymes also vary in activity and expression although the ligninolytic system is relatively nonspecific due to oxidative mediator-based system; thus, it has been extensively used for bioremediation of recalcitrant compounds. Lignin-degrading microorganisms and enzymes can be also used for targeted modification of array of products.

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Chapter 4 Fungal Biodiversity Producing Xylanase Enzymes Involved in Efficient Uses of Xylanolysis



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Abbreviations

AGX	Arabino(glucurono)xylan
Araf	Arabinofuranose
AX	Arabinoxylan
DTT	Dithiothreitol
GAX	Glucurono(arabino)xylans
GlcA	Glucuronic acid
HX	Heteroxylan
MeGlcA	4-O-methyl-D-glucuronosyl
MLG	Mixed-linkage glucan
SbmF	Submerged fermentation
SSF	Solid-state fermentation
Wis	Water-insoluble
Х	Homoxylan
Xyl	Xylanase

1 Introduction

Xylan is among the most common hemicellulose which can be found in nature and comes second among the world's most widely available biopolymer in plants (Muchlisyam et al. 2016). Xylanases are hydrolytic enzymes that depolymerize the plant cell component xylan, by cleaving the β -1,4 backbone of xylan. Fungal sources

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are among the widely used producers of xylanase. But xylanases can also be produced by other species like marine algae, bacteria, yeast, snails, protozoans, crustaceans and insects (Bajpai 2014).

Many fungal species like *Aspergillus*, *Trichoderma* and *Fusarium* species are active xylanase producers. Xylanases have various biotechnological and industrial applications such as in the animal feed industry, processing of bread, food industry, cellulose pulp bleaching and textile manufacturing (Bajpai 2014).

Due to the increasing interest in the biotechnological applications of xylan, intensive research is being carried out mainly focussing on xylanolytic enzymes that are derived from microbial sources majorly mesophilic fungi. This review highlights the occurrence, structure, property of xylans and the production and application of xylanases.

2 Structure of Xylan

Hemicellulose residues were known to be first extracted from plant using dilute alkali. They are found to be of lower molecular weight than cellulose. They can have D-xylose, D-mannose, L-arabinose or D-galactose as their principal monomeric unit. Xylan contains close to or more than 90% of D-xylose as monomer and also traces of L-arabinose. Xylan homopolymer consisting of only D-xylose are very difficult to isolate and are normally found to contain 2–4 sugar monomers. In the initial broad classification, polysaccharide such as chains of D-xylose, D-galactose residues and D-mannose (either alone or in association with D-glucose) (Sunna and Antranikian 1977) can be recognized.

Xylan contains the same backbone structure of D-xylose but differences occur because of the sugar substituents present. These contain acetyl, arabinose and glucuronosyl residues. The xylan isolated from esparto grass is found to be made of the chains of (1 3 4)-P-D-xylopyranose residues. Similarly, xylan isolated from land plants are found to contain the same backbone but show the difference in the arrangement of sugar substitutes. Substitutes found were D-glucuronic acid and L-arabinose, and 4-methyl ether was also found attached (Jayme and Sätree 1942).

2.1 Backbone (D-Xylose)

Xylan that was isolated from plants and grass have found to contain the same backbone of β -(1 \rightarrow 4)-linked xylose remnant. The main chain of xylan is composed of fl-(1 \rightarrow 4)-linked β -xylopyranose residues. The presence of β -(I \rightarrow 4) linkages was also found established between two xylose residues adjacent to each other (Biely 1985) (Fig. 4.1).



Fig. 4.1 Xylan structure with side chains (Biely 1985)

2.2 Side Chains

Common substituents like acetyl, glucuronosyl and arabinosyl residues are commonly found in the xylan backbone (Biely 1985).

3 Properties of Xylan

3.1 Solubility

Microbial and fungal sources need to secrete an enzyme complex to hydrolyze heteropolymers such as xylan to simple sugars. These enzymes, having different substrate specificity and catalytic activity, are capable of depolymerization of hemicellulose. Xylan varies insolubility. They can be either found to be soluble or non-soluble. Also, it has been shown that the solubility of the xylan structure directly depends upon the degree of substitution in the xylan structure. Another factor determining the degree of solubility is the degree of polymerization of the xylan molecule (Izydorczyk and Biliaderis 1992).

3.2 Characteristics of Xylan Extracted from Different Sources (Table 4.1)

Table 4.1	Xylanases	from	different	microbial	strains	along	with	their	distinct	characteristics
(Bastawde	1992)									

Microorganism	Characteristics
Aspergillus fumigates	Molecular weight is 19, 8.5 kDa. Optimum temperature and pH is at 55 $^{\circ}$ C and 5.5, respectively. Its stability ranges from pH 5.0 to 9.0 and temperature of 50 $^{\circ}$ C
Aspergillus fischeri	Molecular weight is 31 kDa. Optimum temperature and pH is at 60 $^{\circ}$ C and 6.0, respectively. Its stability ranges from pH 5.0 to 9.5 and temperature of 55 $^{\circ}$ C
Streptomyces sp.	Molecular weight ranges from 42 to 44 kDa. Optimum temperature and pH is at 55 °C and 5.5, respectively. Its stability ranges from pH 4.0 to 9.0 and temperature of 50 °C
Aspergillus sojae	Molecular weight is 33, 36 kDa. Optimum temperature and pH is at 50, 60 °C and 5.0, 5.5, respectively. Its stability ranges from pH 5.0 to 9.0 and temperature of 50 °C
Aspergillus nidulans KK-99	Optimum temperature and pH is at 55 °C and 8.0, respectively. Its stability ranges from pH 4.0 to 9.5
Acrophialophora nainiana	Molecular weight is 22 kDa. Optimum temperature and pH is at 55 °C and 7.0, respectively. It's thermostable up to 55 °C
Bacillus sp	Molecular weight is 99 kDa. Optimum temperature and pH is at 75 °C and 6.0, respectively
Myceliophthora sp	Molecular weight is 53 kDa. Optimum temperature and pH is at 75 °C and 6.0, respectively. It's stable at pH 9.2 and temperature of 60 °C
Penicillium capsulatum	Molecular weight is 22 kDa. Optimum temperature and pH is at 48 $^{\circ}$ C and 3.8, respectively. It's thermostable at pH 5.0 and temperature of 50 $^{\circ}$ C
Thermomyces lanuginosus	Molecular weight is 24.7 kDa. Optimum temperature and pH is at 70 $^{\circ}$ C and in the range of 6.0–6.5, respectively. Its stability ranges from pH 5.5 to 10 and temperature of 60 $^{\circ}$ C

4 Enzyme Complex Involved in the Hydrolysis of Xylan

4.1 Endoxylanase

The enzyme 4-endoxylanase having the structure 1,4-p-o-xylan xylohydrolase is one of the enzymes responsible for xylan degradation. The substrate acts at the specific site and not randomly. The site of attack depends on the nature of the substrate and also on the presence of substituents. They are responsible for cleaving the glycosidic linkages of the xylan backbone. This converts the main products into oligosaccharides and further hydrolysis converts them into xylose, xylobiose and xylotriose (Reilly 1981).

4.2 p-Xylosidase

These are exoglycosidases having the formula p-D-xylosidexylo hydrolase and molecular weight in the range of 60 and 360 kDa. Short xylooligosaccharides and xylobiose are hydrolyzed to liberate xylose by this substrate (Wong et al. 1988). These are reported to be found in bacteria and fungi and also among cell associated with yeast and bacteria. The purified form of P-xylosidases are not capable of hydrolyzing xylan. However, some reports show that they attack xylan slowly producing xylose (Sunna and Antranikian 1977).

4.3 L-Arabinofuranosidases

This substrate plays an important role in hydrolyzing xylan despite the fact that only a few such enzymes have been isolated. The two types are only existence-exo-acting and endo-acting. Exo-acting are found to be active against branched arabinans, whereas endo-acting are found only active towards linear arabinans. Such endoarabinases are found to be present in *Bacillus subtilis, Clostridium felsineum* and various fungi sources. Most enzymes studied that degrade arabinan are exo-acting (Kaji and Saheki 1975; Van der Veen et al. 1994; Kaji 1984).

5 Occurrence and Isolation of Xylan

5.1 Occurrence

Xylanases occur in both prokaryotes and eukaryotes and are found to occur widely. They are found among higher eukaryote-like snails, insects, protozoa, etc. Among prokaryotes, bacteria and cyanobacteria are found to produce xylanase in the marine environment. Among two types of xylanase (extracellular and intracellular), intercellular xylanases are found in bacteria and protozoa from tureen sources (Dekker and Richards 1976).

Terrestrial plants and algae are known to have xylan-type polysaccharides. These are also found in different tissues of the same plant at different sites. The structural diversity in plants can be due to the fact that structural diversity is related to the functionality in the plant and hence can show the relation between the distribution of particular classes of xylan.

Xylan can also be found in the botanically oldest plant families. In green algae, Homoxylans (X) having b-(1e3) glycosidic linkages have been found to function as substitute cellulose in the cell wall architecture. Also, red seaweeds in the Palmariales and Nemaliales have cell wall comprising of homoxylans having mixed b-(1e3 and 1e4) glycosidic linkages (Painter 1983).

6 Naturally Occurring Xylans

6.1 In Plant Material

Different agricultural crops like corn stalks, sorghum, corn cobs, sugar cane, straw, hulls of fruits and dry husks of fruits and vegetable are known to be very rich sources of xylan. Other sources include forest and pulping waste products from hardwoods. Glucuronoxylan, arabinoglucuronoxylan and arabinoxylan are other structural forms of xylan produced from certain plant sources. Wheat flour xylan consists of 30–40% arabinose which are irregularly attached to the xylose backbone. A distinguishing feature of wheat bran xylan is that it has a large number of side groups attached arabinose). The barley husk xylan are found to contain glucuronic acid and xyloarabinose groups attached to it. Hardwood xylan are found to have 4-*O*-methylglucuronic acid on each tenth xylose. Softwood xylan along with arabinose side group contains more side groups as 4-*O*-methylglucuronic acid are found in each sixth xylose (Ebringerova and Heinze 2000) (Table 4.2).

6.2 In Cell Wall

Xylan is a hemicellulose found in the grass cell wall. The type II grass walls are found to be rich in β -(1,3/4) glucan (also called mixed-linkage glucan) and GAXs (glucurono(arabino)xylans). Grass xylans are found to contain similar general

Plant source	Xylan type
Corn	Mature leaves
	GAX I
	GAX II
	Bran
	HX
	Coleoptile
	GAX I
	GAX II
Rice	Cobs
	wis-AGX
	ws-AGX
	Endosperm
	AX
	GAX
	Bran
	AX
	GAX

Table 4.2 Structural features of xylans found in various plant tissues which are found within Zeamays and Oryza sativa (Ebringerova and Heinze 2000)

xylans structure (i.e. substitution with MeGlcA or α -Araf, GlcA residues). In addition to it, it also contains some unique features like the presence of Xylarabinofuranosyl side chains and the presence of feruloyl groups at the C-5 position of Araf residues. The grass xylans are of two main groups-GAX, making up to 35% of vegetative cell wall that is found to contain GlcA/MeGlcA and Araf remnants; also the cell wall of cereals containing AX, lacking GlcA (Faik 2010).

7 Xylanase from Thermophilic Fungal Source

The enzymes produced from extremophilic and thermophilic microorganism show greater stability. Recently, a lot of efforts has been put in the isolation from such sources. The common thermophile fungi involved in xylanase production includes *Chaetomium thermophile, Humicola insolens, Humicola lanuginosus, Humicola grisea, Melanocarpus albomyces, Paecilomyces variotii, Talaromyces byssochlamydoides, Talaromyces emersonii, Thermomyces lanuginosus* and *Thermoascus aurantiacus*. These xylanases are usually glycoproteins and can withstand optimum temperatures between 60 and 80 °C. They show the highest activity at an acidic pH of about 4.5–6.5 and exhibit variable molecular weights in the range of 6–38 kDa (Polizeli et al. 2005).

8 Xylanase from Mesophilic Fungal Source

The genera *Aspergillus* and *Trichoderma*, belonging to the class of mesophilic fungi, are leading in xylanase production.

8.1 Aspergillus Species

The production of xylanase from soybean residues by *Aspergillus foetidus* was optimized with the production level of 13.98 U/mL occurring at fermentation for 168 h, pH 7.0, 28 °C and 120 rpm. The specific xylanolytic activity of *Aspergillus fumigatus* was found to be 1055.6 U/g and 558.3 U/g, after 5 days of solid-state fermentation (SSF) in wheat bran and soybean, respectively, by Delabona et al. (2013). In the same work, he found that *Aspergillus niger* showed a specific xylanolytic activity of 1285.0 U/g, 484.2 U/g and 1050.0 U/g using the residue of wheat bran, soybean and wheat bran with sugarcane bagasse, respectively. De Souza Moreira et al. (2013) found that xylanases could be produced by *A. terreus* under submerged fermentation at an optimum pH and temperature of pH 6.0 and 50 °C at 120 rpm and 5.0 and 45 °C at 120 rpm, respectively. Guimaraes et al. (2013) found a xylanolytic activity of 10.50 and 11.92 U/mL, respectively, for *Aspergillus niger* and *Aspergillus flavus*,

	Production	
Aspergillus species	level	Fermentation details
Aspergillus foetidus	13.98 U/mL	Time 168 h, pH 7, 28 °C and 120 rpm
Aspergillus	1055.6 U/g	After 5 days of SSF
fumigatus		
Aspergillus niger	1285.0 U/g	Using wheat bran residue
Aspergillus terreus	Unknown	Under submerged fermentation at pH 6.0, 50 °C at 120 rpm
Aspergillus flavus	11.92 U/mL	Using 0.5% corn cob
Aspergillus awamori	Unknown	SSF at pH 5 and 50 °C

 Table 4.3
 Details about the production of xylanase from different fungal sources (adopted from Delabona et al. 2013; Souza Moreira et al. 2013)

using wheat bran 0.5% and corn cob 0.5% as residues. Based on the data obtained from different literature, it was found that the xylanase from *A. foetidus* was more effective in comparison to xylanase from *A. niger, A. flavus and A. fumigatus.* (Cunha et al. 2018). The SSF of tomato pomace by *Aspergillus awamori* was known to produce hydrolytic enzymes like xylanase. At pH 5, the enzyme shows optimum activity and temperature of 50 °C. Hg²⁺ and Cu²⁺ could strongly inhibit the enzyme, whereas the enzyme can be activated by Mg²⁺. The enzymatic activity was observed to be quite high when the extract was preserved at a pH of 3–10 and a temperature range of 30–40 °C (Umsza-Guez et al. 2011).

Xylanase can be produced by *Aspergillus* species on fermentation of soybean residues, wheat bran, residues of wheat bran with sugarcane bagasse, corn cob and tomato pomace. Based on the data obtained from different literature, it was found that the xylanase from *A. foetidus* was more effective in comparison to xylanase from *A. niger, A. flavus* and *A. fumigatus* (Cunha et al. 2018). Hg²⁺ and Cu²⁺ could strongly inhibit the enzyme, whereas the enzyme can be activated by Mg²⁺ (Table 4.3).

8.2 Trichoderma Species

Ascomycetes filamentous fungi *Trichoderma reesei* are known to produce alkaline xylanase (Mewada et al. 2017). The cultivation of the crude extracellular extract of the strain *Trichoderma inhamatum*, Xyl I and Xyl II were stable at a temperature of 40 °C and at pH 4.5–6.5 for Xyl I and 4.0–8.0 for Xyl II. These xylanases were stable at a temperature of 40 °C and at pH 4.5–6.5 for Xyl I and 4.0–8.0 for Xyl I and 4.0–8.0 for Xyl II. The activity of enzyme was strongly reduced by ion Hg²⁺ and the detergent SDS, whereas 1,4-dithiothreitol was found to stimulate both Xyl I and Xyl II enzymes (Silva et al. 2015). *Trichoderma viride* VKF-3 produces xylanase using coconut oil cake as substrate with an activity of 3.045 IU/mL (Nathan et al. 2017). Rifai, a strain of *Trichoderma harzianum*, isolated from decaying *Aspidosperma sp.* (peroba) wood was used to produce xylanase using sugarcane bagasse as substrate. The high-

est xylanase activity was observed to be 288 U/mL on the seventh day using this method (Rezende et al. 2002).

8.3 Fusarium Species

Xylanase can be purified from *Fusarium heterosporum*, using barley remnants by SSF. The molecular mass of xylanase obtained by this method was found to be 19.5 kDa. The optimum pH for the xylanase was 5.0. Xylanase activity can be enhanced by Ba²⁺, Ca²⁺, Mg²⁺, dithiothreitol (DTT) and β-mercaptoethanol, whereas Hg²⁺, Pb²⁺ and Zn²⁺ strongly inhibited the enzyme activity (Heinen et al. 2014). High amounts of xylanase can be produced on birchwood xylan and agricultural residues like wheat bran and peptone by using a mutant strain of *Fusarium oxysporum* (NTG-19) (Kuhad et al. 1998). *Fusarium sp.* BVKT R2 are known producers of xylanase. For maximum production, a temperature of 32.5 °C and a pH of 5 are required along with 1.5% of yeast and sorbitol extract. This should also be kept under agitation at 175 rpm (Ramanjaneyulu and Reddy 2016).

8.4 Penicillium Species

Using cheap substances like wood, rice, wheat bran and sesame oil, the production of xylanase from Penicillium oxalicum under SSF was optimized. Xylanase can also be produced by the fungus *Penicillium sclerotiorum* under submerse cultivation. The best inducers of xylanase activity were oat spelts xylan and wheat bran. The conditions for optimum production of xylanase were found to be at a temperature of 30 °C and a pH of 6.5 and under stationary condition for 5 days using liquid Vogel medium. The optimum activity was found at a temperature of 50 °C and at pH 4.5 (Knob and Carmona 2008). Xylanase can also be produced from submerged cultures of Penicillium echinulatum 9A02S1. The highest activity of xylanases was observed in the medium containing 0.25% (w/v) cellulose and 0.75% (w/v) sorbitol added after 36 h of cultivation (Todero Ritter et al. 2013). Xylanase can be characterized and purified by the usage of agro-industrial waste which was proved when Penicillium glabrum was found to produce xylanase enzyme by the use of brewer's spent grain as a substrate. For optimum production of Xylanase by P. glabrum, it was grown at a pH of 5.5 and a temperature of 25 °C in a liquid medium kept under stationary condition for straight 6 days. Xylanase was purified by P. glabrum by using inexpensive procedures like molecular exclusion chromatography and also ammonium sulfate fraction at ion. The optimum activity of xylanase was found at a pH of 3.0 and temperature of 60 °C. The xylanase activity can be enhanced by the use of Mn^{2+} ion and by DTT and β -mercaptoethanol, both of which are reducing agents, whereas detergent SDS, the ions Hg2+, Zn2+ and Cu2+, were found to be strongly inhibiting the enzymes (Knob et al. 2013). Penicillium janthinellum FS22A are known to produce cellulolytic-xylanolytic enzymes (Okeke et al. 2015) (Table 4.4).

Microbial species	Optimal temperature (°C)	Optimal pH
Aspergillus aculeatus	50, 50, 70	4.0, 4.0, 5.0
Aspergillus oryzae	60	5.0
Aspergillus sydowii	50	4.0
Aspergillus kawachii	60, 55, 50	5.5, 4.5, 2.0
Aspergillus fischeri	60	6.0
Aspergillus sojae	60, 50	5.0, 5.5
Acrophialophora nainiana	22 55	7.0
Myceliophthora sp.	75	6.0
Cryptococcus sp.	40	2.0
Chaetomium cellulolyticum	50	2.0
Thermomyces lanuginosus	50	5.0-7.0
Penicillium capsulatum	48	3.8
Penicillium sp.	70	6.0–6.5

Table 4.4 Table showing characteristics of some xylanases produced by different microorganisms(Polizeli et al. 2005)

9 Applications of Xylanases

Xylanases have great potential application in many industrial processes. About 20% of the world enzyme market is occupied by xylanase, cellulase and pectinase. Due to the potential effectiveness of xylanolytic enzymes in bread making, they are used in baking and brewing industries. Xylanases are used as animal feed, where arabinoxylans which are present in feed ingredients are broken down by them reducing the viscosity of the raw material (Harris and Ramalingam 2010). Xylan and xylanases are also used in pharmaceutical industries. Xylanases along with enzymes like proteases and hemicellulase are used for treating indigestion. The hydrolytic products of xylan can be used to make artificial sweeteners, solvents, ethanol etc. Xylanolytic complex also finds varied application in textile industries and is also used for making plant fibres like linen or hessian. Xylanases are also used in bleaching of cellulose pulp. Xylanases are also used to enhance sugar recovery from agricultural residues (Chakdar et al. 2016) (Table 4.5).

10 Comparison of Xylanase Production from Bacterial and Fungal Sources

Xylanases are known to be produced by organisms like bacteria, snails, fungi, yeast, protozoans, fruit and vegetable seeds and insect. Bacterial genera like *Bacillus, Pseudoxanthomonas, Rhodothermus, Paenibacillus, Cellulomonas, Staphylococcus* and *Microbacterium* are known to produce xylanase enzyme. Also, thermostable

		Optimal	
	Optimal	temperature	
Microorganisms	pH	(°C)	Application
Aspergillus niger	5.3	5.3	Used in the improvement of animal feed
Humicola	n.c.	n.c.	
A. niger	n.c.	n.c.	Used in the preparation of feed for pigs and birds
Bacillus sp	9.5	50	Used in paper-making industry
Trichoderma reesei	5.0-6.0	5.0-6.0	Cellulose pulp bleaching
T. longibrachiatum	5.0-5.5	5.0-5.5	Used in food industry
Trichoderma koningii	5	5	Used in the production manufacture of mushrooms and vegetable extracts, peeling of cereals through enzymatic methods, making of bread, preparation of animal feed
Trichoderma sp	n.c.	n.c.	Utilized for structural studies of carbohydrates

 Table 4.5
 Table showing commercial xylanases produced by microorganisms

SbmF submerged fermentation, SSF solid-substrate fermentation, n.c. not cited

xylanases that are active at 60–70 °C are produced by *Thermotoga* sp., *Streptomyces* sp., *Clostridium thermocellum, Bacillus* spp., *Rhodothermus marinus* and *Stenotrophomonas maltophilia. Flavobacterium frigidarium, Penicillium Strain* and *Clostridium* sp. produce cold-adaptive xylanases (Chakdar et al. 2016).

Different organisms produce xylanase in different conditions. For example, bacterial species produce xylanase neutral/alkaline pH and xylanase are produced in acidic range by fungal species. Fungal xylanases are less attractive as the low pH requirement for growth of fungi and production of fungal xylanases necessitates additional steps in the subsequent stages. The secondary structures especially at the loop areas are different in bacterial and fungal xylanases. The nucleophile and proton donor in bacterial and fungal xylanases are always the glutamic acid, though their position may change. Fungal xylanases are produced with cellulose and hence there is increase in the length of downstream processing. In bacteria, xylanases are mostly produced alone and hence reduces the downstream process time (Chakdar et al. 2016).

11 Conclusion

Xylanases are hydrolases which are capable depolymerizing xylan polysaccharide. Xylan being a cell component is also known to be the second most widely available polysaccharide. Xylanases are known to be produced by insect, snails, bacteria, fungi, algae (marine), crustaceans, hulls and husks, seeds, protozoans, etc., but the filamentous fungi are the principal commercial sources of xylanase. These enzymes are synthesized by extremophiles, mesophiles and thermophiles microorganisms as the enzyme produced by them is found to be more stable. In order to consolidate the entire economics of the pre-treatment of lignocellulosic material, strategies are being carried out to convert xylan into useful products, and to this end xylanases have been identified as key enzymes with diverse industrial applications. The bulk production of xylanases at an economically viable rate is the major bottleneck in commercial applications of xylanase-based enzymatic processes. The study conducted by authors of this chapter is focussed on the sources, production and applications of fungal xylanolytic enzymes in the context of their biotechnological potential.

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Chapter 5 Biodegradation of Lignocellulosic Biomass and Production of Ethanol Using Potential Microorganisms



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1 Introduction

The consumption of world petroleum is at a rapid rate. This has led to resurgence in the development of alternative energy sources that can displace fossil transportation fuel (Alper and Stephanopoulos 2009; Gupta et al. 2009). Ethanol as biofuel has several advantages such as low toxicity, biodegradable nature, generation of fewer airborne pollutants than petroleum fuel, and easy integration in the existing vehicle (McCarthy and Tiemenn 1998). Lignocellulosic biomass represents an abundant carbon-neutral renewable resource for the production of bioethanol (Ragauskas et al. 2006). Lignocelluloses are mainly comprised of cellulose, hemicellulose, and lignin (Kuhad et al. 1997). The bioconversion process mainly consists of three major steps, i.e. pretreatment, hydrolysis, and fermentation. Pretreatment is required to remove lignin and hemicellulose fraction, increase the porosity of the materials, and reduce cellulose crystallinity.

Researchers have displayed that biological pretreatment by microorganisms like bacteria and fungus (white-, brown-, and soft-rot fungi) enhances the hydrolysis productivity as the inhibitors generation is low and the energy utilized is also low (Wang et al. 2013; Taha et al. 2015; Hatakka 2005). Biological pretreatment is also considered reasonably cost-effective compared to other pretreatment processes. Such biological pretreatments have been less investigated (Chaturvedi and Verma 2013). The highest efficiency among the pretreatment methods has been achieved by lignin-degrading white-rot fungi (WRF) for the soft and brown fungi only attack cellulose (Kang et al. 2013; Fernandez-Fueyo et al. 2012; Suhara et al. 2012). Moreover, using white rot fungi consumes less environmental damage and less energy consumption (Narayanaswamy et al. 2013; Kumar et al. 2015). The ability of biological pretreat-

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ment to degrade lignocellulosic materials is due to their highly efficient enzymatic system. The effectiveness of the white-rot fungi (WRF) is studied here for the biological pretreatment and further fermentation to produce bioethanol.

2 History of Ethanol as Biofuel

Ethanol as a fuel has been used throughout technologically advanced man's modern history. Even the invention of ignition engines was with bioethanol. Ethanol was one of the most popular lamp illuminants used in the 1850s, and approximately 90 million gallons of ethanol was produced in the United States. The tax imposition on ethanol to assist the finance of the civil war brought changes in its established status. Kerosene became cheaper, and it quickly replaced ethanol as the premier illuminant in the 1860s (Morris 1993). In 1906, alcohol tax was lifted, which renewed the interest in ethanol, and in 1908, Henry Ford designed the automobile car Model T to run on ethanol. By 1914, the production of ethanol had come back and could reach ten million gallons. But in 1919, the emergence of petroleum as fuel made the use of ethanol as fuel decreased again. By the early 1940s, the production of ethanol reversed again when it was used during World War II for fuel and to make synthetic rubber. During this period, the United States produced about 600 million gallons of ethanol annually (Morris 1993). At the end of World War II, the demand for ethanol reduced and continued to decline for the next two decades, mostly due to cheap petroleum imports. But again, the oil embargo by Arab countries in 1973 created petroleum shortages which resulted in significant increase in gasoline price (Campbell and Laherrere 1998). Since the 1970s, the gasoline shortage had accelerated the concerns about the rising prices for the crude oil and increasing political instability which made the use of ethanol as biofuel is again under consideration worldwide.

3 Bioethanol: First and Second Generations

3.1 First-Generation Bioethanol

The bioethanol produced by fermentation of sugar (sugarcane juice, molasses, sugar beet juice, fruit juice) and starchy feedstocks (wheat, corn, potato) is commonly known as first-generation bioethanol (Antoni et al. 2007). The ethanol production methods used are enzymatic digestion (to release sugars from stored starches), fermentation of the sugars, distillation, and drying. According to the US Energy Information Administration, the first-generation bioethanol have played an important role in establishing the infrastructure and policy drivers required to support renewable transport fuels in the international marketplace. However, the International
Energy Agency (IEA 2008) has a number of concerns about the potential drawbacks of first-generation bioethanol, such as:

- (a) Competition between food and fuel: It is clear that the development of bioenergy options, particularly food-based bioethanol, may adversely affect food demands. It is an obvious fact that the higher food prices will have adverse effects on the developing world, where disposable incomes are lower. This alarm to use food resources led to a search for alternative biofuels.
- (b) **Deforestation**: The constant production of first-generation biofuels might lead to major deforestation, and land thus available may be changed from permanent forest cover to agriculture.
- (c) Multi-feedstock flexibility: For commercial viability, technologies and plant designs which are able to process a number of different feedstock in a flexible way are preferable. The storage of crops raw materials may accelerate the cost of biofuel production if many single food crops used for biofuels are seasonal for the plant to operate around the year, and the cost of biofuel production may be accelerated due to the storage of raw materials.

3.2 Second-Generation Bioethanol

Now, it is clearly understood that the production of first-generation bioethanol is an unsustainable approach, and the increasing criticisms have highlighted for the use of non-food crops for the production of second-generation bioethanol. The production of second-generation bioethanol is from lignocellulosic biomass comprised of the residual non-food parts of the food crops and other crops that are not used for food purposes along with municipal, industrial and construction waste. Second-generation biofuels are expected to reduce net carbon emission, increase energy efficiency and reduce energy dependency, potentially overcoming the limitations of first-generation biofuels (Antizar-Ladislao and Turrion-Gomez 2008). In addition to these, the renewable nature of ethanol, its long-term sustainability, low net carbon emission, high energy efficiency, and low energy dependency are the other major benefits of switching to cellulosic ethanol (IEA 2008). However, there is still much work to be done in terms of improving second-generation biofuel technology pathways, to reduce costs and to improve the performance and reliability of the conversion process.

4 Current Status of Bioethanol

4.1 Status of Bioethanol Production Worldwide

Bioethanol production worldwide has increased considerably since the oil crisis in 1970 (Campbell and Laherrere 1998). Its market grew from less than a billion litres in 1975 to more than 65 billion litres in 2008 and is expected to reach 100 billion

litres in 2015 (Licht 2005). According to the IEA (2008), a projection of 1% rise per year has been made for worldwide demand for oil. This is mostly due to increasing demand in energy market of developing countries, especially India (3.9%/year) and China (3.5%/year). With regard to bioethanol, the share of the USA in global production is 50% and Brazil provides 39% of the total global supply. The share of the Organisation for Economic Cooperation and Development (OECD), Europe, is 5% (Gnansounou 2010). Since Brazil is one of the most developed nations in ethanol production, almost all the Brazilian vehicles use either pure ethanol or the blend of gasoline and ethanol (75:25) (Mussatto et al. 2010). The high percentage in which ethanol is added to gasoline in Brazil is also an effort on part of the government to reduce the imports of oil (Prasad et al. 2007). As a result of these efforts, ethanol production in Brazil has substantially risen from 555 million litres (1975/1976) to 16 billion litres (2005/2006) (Orellana and Bonalume 2006), but a major reason for this is sugarcane juice. The innovations introduced by the automobile industry with flex-fuel cars, which may be fuelled with ethanol and/or gasoline in any proportion, increased the market for ethanol.

It is noteworthy that the United States, the largest consumer of petroleum products (2.42 billion litres/day or 20.7 million barrels/day in 2007), meets its demand by importing about 58%, i.e. 1.4 billion litres or 12 million barrels/day. It is predicted that gasoline consumption will rise further along with the rising population, as gasoline is a primary energy source that meets non-commercial transportation demands. Similar to Brazil, the USA is also a big investor in bioethanol research (Solomon et al. 2007) and has increased the ethanol production from 6.16 billion litres or 1.63 billion gallons in 2000 to 39.3 billion litres or 10.4 billion gallons in 2009, representing a sevenfold increase (Petrova and Ivanova 2010). At present, more than 95% of ethanol production in the United States is from corn. The rest is made from wheat, barley, cheese whey and beverage residues (Solomon et al. 2007). However, it is expected that about 1.53 billion litres or 405 million gallons of cellulosic ethanol will be produced by the end of 2012 (Solomon et al. 2007).

In Europe, the maximum amount of ethanol is produced from wheat and sugar beet, and France, Germany and Spain are the European countries more strongly committed to ethanol production (Prieur-Vernat and His 2006). The European Union strategy for biofuels is to decrease their dependence on oil and reduce the negative impact caused to the environment. The share of OECD countries in global oil demand is also expected to decrease from 57% in 2007 to 43% in 2030 (IEA 2008).

China has also invested much in the production of ethanol, and since it is the world's largest auto market, it imported about 52% of the total transportation oil consumed in 2008 (Fang et al. 2010). The production of ethanol for biofuels began in China in 2001, using corn as a raw material, and by 2007 four grain-based ethanol plants with the production of about 1.75 billion litres or 1.4 million metric tons (MMT) ethanol have been developed. However, the Chinese government supported the production of "non-food ethanol" (ethanol made from non-food crops) to restrict the competition for ethanol and food applications. The projects on ethanol based on grains were restricted by the Chinese government (Fang et al. 2010). Many technologies of ethanol production based on non-food crops, such as cassava, sweet

sorghum, sweet potato, Jerusalem artichoke and kudzu vine root, are being developed (Li and Chan-Halbrendt 2009). Till now, the exclusive application of gasoline containing 10% ethanol to motor vehicles has been enforced in all areas of Heilongjiang, Jilin, Liaoning, Henan, Anhui, Guangxi and selected areas of Hebei, Shandong, Jiangsu and Hubei provinces (Fang et al. 2010). Similarly, Thailand has also invested in the production of ethanol. In 2007, there were seven ethanol plants with a total capacity of 955 thousand litres/day, comprising 130 thousand litres/day cassava ethanol and 825 thousand litres/day molasses ethanol, and as a result of government promotions, 12 new plants with a total installed capacity of 1.97 million litres/day are being constructed (Silalertruksa and Gheewala 2009).

Other countries like Japan and Korea are also in the race. An indigenous and affordable source of energy has become a high priority for energy security and sustainability. The production of biofuels in Japan started in 2003, and by 2007 the total amount of bioethanol production reached approximately 30 thousand litres per year (Matsumoto et al. 2009). Similarly, Korea is also very concerned about its high CO_2 emission and dependence on imported crude oil (Kim et al. 2010). Korea consumes about ten billion litres of gasoline annually and three or five million litres of ethanol blended gasoline (E5) countrywide (Kim et al. 2010). Therefore, the Korean government announced its plan to increase the supply of transportation biofuels from 0.2 billion litres (2008) to 5 billion litres by 2030.

4.2 Status of Bioethanol Production in India

In the year 2003, the Planning Commission of the Government of India brought out an extensive report on the development of biofuel, and bioethanol was identified as a principal biofuel. The ethanol blend in gasoline was proposed to 10% by 2011– 2012 in India, and 5% ethanol blend in gasoline was made mandatory in 11 states and 3 union territories of the nation (Sukumaran et al. 2010). In 2006, the demand for ethanol for 5% gasoline doping/blending level was 0.64 billion litres. The estimated current demand for 10% blending is projected to be 2.2 billion litres in 2017 (Sukumaran et al. 2010). According to a 2006 estimate, the actual production of ethanol was only 0.39 billion litres which was not sufficient to meet the fuel demand if the entire gasoline had to be doped at 5% level.

In India, ethanol is mainly produced from sugarcane molasses, but the substrate has to compete with the food demand and therefore cannot supply the required amount of ethanol. Therefore, the nation needs to develop bioethanol technologies, which use biomass feedstock that does not have food or feed value. The most appropriate bioethanol technology for the nation would be to produce it from lignocellulosic biomass, such as rice straw, rice husk, wheat straw, sugarcane tops and bagasse, municipal waste and forest waste (Sukumaran et al. 2010). According to Kim and Dale (2004), the total bioethanol production from plant biomass is estimated to be 491 gigalitres (GL)/year globally.

India alone has the capacity to produce 25%, i.e. 123 GL/year, of the total world ethanol production, if the entire lignocellulosic residues available are used for ethanol production. Hence, to contemplate a bioethanol production plant, the lignocellulosic biomass assessment with geographical distribution and accurate information on availability of biomass in different parts of the country is a prerequisite. With this in view during the Ninth Plan, the Ministry had sponsored 500 taluka-level biomass assessment studies in 23 states to compile data on availability of lignocellulosic biomass.

As an extension of this effort, a project for preparation of "Biomass Resource Atlas of India" has been jointly sponsored to the Indian Institute of Science (IISc), Bangalore, and Regional Remote Sensing Service Centre (RRSSC), Bangalore, which aims at the integration of the data on biomass availability obtained from taluka-level studies and from other reliable sources, with information on crop distribution pattern derived from GIS-based maps provided by RRSSC.

4.3 Plant Biomass

Plant biomass is the most abundantly available and renewable natural resource on earth. Lignocellulose is comprised of three main polymers: cellulose, hemicellulose and lignin (Fig. 5.1), collectively termed as lignocellulose. The chemical properties of the components of lignocellulose make them a substrate for enormous biotechnological products (Kuhad and Singh 1993; Kuhad et al. 1997, 2007).

4.4 Cellulose

Cellulose is a glucan polymer of D-glucopyranose units, which are linked with each other by β -1, 4-glucosidic bonds. The wood cellulose has an average degree of polymerization (DP) of at least 9000–10,000 and possibly as high as 15,000. An average DP of 10,000 corresponds to the linear chain length of approximately 5 μ m in wood. An approximate molecular weight for cellulose ranges from about 10,000 to 150,000 Da (Goring and Timell 1962).

Wood has many types of cellulose: crystalline and non-crystalline and accessible and inaccessible. Most wood-derived cellulose is highly crystalline which may contain as much as 65% crystalline regions. The remaining portion is referred to as amorphous cellulose which has a lower packing density. Accessible and nonaccessible refer to the availability of the cellulose to water, microorganisms, etc. The surfaces of crystalline cellulose are accessible but the rest of the crystalline cellulose is nonaccessible, whereas most of the non-crystalline cellulose is accessible. Part of the non-crystalline cellulose is covered with both hemicelluloses and lignin which makes it nonaccessible (Rowell et al. 2005; Kuhad et al. 2011). Concepts of accessible and nonaccessible cellulose are of significance in moisture adsorption,



Fig. 5.1 Different components of lignocellulosic biomass and structures

pulping, extractions, chemical modification and interactions with microorganisms. Unlike cellulose, hemicelluloses are not chemically homogenous (Eriksson et al. 1990; Kuhad et al. 1997, 2011). The hemicelluloses are comprised of both linear and branched heteropolymers of D-xylose, L-arabinose, D-mannose, D-glucose, D-galactose and D-glucuronic acid (Fig. 5.2a). In general, the hemicellulose fraction of woods consists of a collection of polysaccharide polymers with a lower DP than cellulose (100–200), containing mainly the sugars D-xylopyranose, D-glucopyranose, D-galactopyranose, L-arabinofuranose, D-mannopyranose, D-glucopyranosyluronic acid and D-galactopyranosyluronic acid with lower amounts of other sugars. It usually has one repeating sugar unit linked β -(1 \rightarrow 4) with branch points (1 \rightarrow 2), (1 \rightarrow 3) and/or (1 \rightarrow 6).

4.5 Hemicelluloses

Hemicelluloses usually consist of more than one type of sugar unit and are called according to their components, e.g. galactoglucomannan, arabinoglucuronoxylan, arabinogalactan, glucuronoxylan and glucomannan. The hemicelluloses also contain acetyl- and methyl-substituted groups (Rowell et al. 2005). The hemicelluloses









Fig. 5.2 (a) Structure of monosaccharides commonly present in xylan backbone (where (1) β-D-glucopyranose; (2) α-L-rhamnopyranose; (3) α-L-fucopyranose; (4) β-D-xylopyranose; (5) β-D-mannopyranose; (6) β-D-galactopyranose and (7) α-L-arabinofuranose). (b) Polymeric unit structures of hardwoods (o-acetyl 1-4-*O*-methylglucoronoxylan). (c) Polymeric unit structures of hardwoods (glucomannans). (d) Polymeric unit structure of softwoods (cetyl-galactoglucomannan)

from agricultural residues and hardwood are rich in xylan. Softwood contains less xylan and more mannan (Kuhad et al. 1997, 2011).

4.5.1 Softwood Hemicelluloses

Galactomannans are the principal hemicelluloses in softwood. Their backbone is a linear chain built up by 1, 4-linked β -D-glucopyranose and β -D-mannopyranose units (Fig. 5.2a). The mannose and glucose units in the backbone are partially substituted at C-2 and C-3 positions by acetyl groups, approximately 1 per 3-4 hexose units as reviewed earlier (Kuhad et al. 1997). Arabino-glucuronoxylan is another major hemicellulosic sugar and is composed of 1,4-linked- β -D-xylopyranose units. This chain is substituted at C-2 by 4-*o*-methyl- α -D-glucuronic acid group with approximately two such units per ten xylose units. The xylose backbone is also substituted by α -L-arabinopyranose units, on the average 1.3 residues per ten xylose units (Kuhad et al. 1997, 2007, 2011). Arabinogalactan is a minor component in hardwoods and softwoods. The backbone of this galactan is built up by 1, 3-linked α -D-galactopyranose units, and almost every galactose unit is substituted at the C-6 position.

4.5.2 Hardwood Hemicelluloses

The o-acetyl-4-*o*-methyl-glucurono- β -xylan (commonly known as glucuronoxylan) is the major component of hardwood hemicelluloses (Kuhad et al. 1997) (Fig. 5.2b-d). The xylan content varies between 15 and 30% in different hardwood species. The backbone of xylan consists of β -D-xylopyranose units linked by 1,4-bonds. Seven of ten xylose units are substituted by acetyl group at C-2 or C-3 position and in one of ten xylose units, the 4-*o*-methyl- α -D-glucuronic acid residue unit is linked at C-1 and C-2 positions to the hemicellulose backbone (Kuhad et al. 1997). Glucomannan is another hemicellulose in hardwoods. This polysaccharide comprises 2–5% of the wood and is composed of β -D-glucopyranose and β -D-mannopyranose units by 1,4-bonds. Depending on the wood species, the glucose–mannose ratio varies between 1:1 and 1:2 (Kuhad et al. 1997; Rowell et al. 2005).

4.6 Lignin

Lignin is an amorphous and highly complex polymer of phenylpropanoid units and is considered to be an encrusting substance in the plant biomass (Kuhad et al. 1997). The precursors of lignin biosynthesis are coniferyl alcohol, *p*-coumaryl alcohol and sinapyl alcohol. *p*-Coumaryl alcohol is a minor precursor of hardwood and softwood lignins. The predominate precursor of softwood lignin is coniferyl alcohol.

Coniferyl and sinapyl alcohols are both precursors of hardwood lignin. Softwood lignin has methoxyl content of ~15–16%. Hardwood lignin has a methoxyl content of $\sim 21\%$. Moreover, ecological factors such as age of the wood, climate, plant sustenance and amount of sunlight also affect the chemical structure of lignin (Kuhad and Singh 1993; Kuhad et al. 1997). Lignin does not have a single repeating unit like cellulose does but instead consists of a complex arrangement of substituted phenolic units. Lignins are usually divided according to their structural elements. Wood ligning consist of three basic building blocks which are guaiacyl, syringyl and *p*-hydroxyphenyl moieties; other aromatic units also exist in different types of woods. There is a wide variation of structures in different wood species. The lignin content of softwoods is usually in the range of 18-25%, whereas the lignin content of hardwoods varies between 25 and 35%. The phenylpropane can be substituted at the α , β or γ positions into various combinations linked together both by ether (C-O-C) and carbon to carbon (C-C) linkages. Lignins from softwoods are called guaiacyl lignin which is mainly a polymerization product of coniferyl alcohol. Hardwood lignins are mainly syringyl-guaiacyl lignin, because they are a copolymer of coniferyl and sinapyl alcohols. The ratio of the two types differ from one lignin to others from about 4:1 to 1:2.

4.7 Bioconversion of Lignocellulosic Biomass to Ethanol

The bioconversion of lignocelluloses to ethanol consists of two main processes: hydrolysis of lignocellulosic carbohydrate to fermentable reducing sugars and fermentation of the sugars to ethanol (Fig. 5.3). The hydrolysis is usually catalyzed by cellulase enzymes, and the fermentation is carried out by bacteria or yeasts. The factors which have been identified to affect the hydrolysis of cellulose include porosity (accessible surface area) of the waste materials, cellulose fibre crystallinity and lignin and hemicellulose content (Mosier et al. 2005; Kuhad et al. 2011). The presence of hemicellulose and lignin in lignocellulosic materials makes the access of cellulase enzymes difficult, thus reducing the efficiency of the hydrolysis significantly improves the efficiency of hydrolysis due to the removal of lignin and hemicellulose, reduction of cellulose crystallinity and increase of porosity (McMillan 1994; Mosier et al. 2005; Kuhad et al. 2011).

4.7.1 Pretreatment of Lignocellulosic Biomass

The effect of pretreatment strategies of lignocellulosic materials has been recognized well for a long time (Mosier et al. 2005; Sanchez and Cardona 2008; Kuhad et al. 2011). The pretreatment strategies must meet the following requirements: (1) improve the formation of sugars or the ability to subsequently form sugars by enzymatic hydrolysis; (2) avoid the degradation or loss of carbohydrate; (3) avoid



Fig. 5.3 Bioethanol production from lignocellulosic biomass

the formation of hydrolysis and fermentation inhibitory byproducts; and (4) costeffectiveness of the process. Various physical, biological, physico-chemical and chemical processes have been used for pretreatment of lignocellulosic materials.

4.7.2 Bioconversion of Lignocellulosic Materials to Ethanol

The pretreatment has become a necessity to maximize the hydrolysis of lignocelluloses and eventually the production of ethanol. The advantages of biological delignification of plant material over chemical and mechanical pretreatment methods include (1) mild reaction conditions, (2) avoidance of the use of toxic and corrosive chemicals, (3) higher product yield, (4) fewer side reactions, (5) less energy demand and (6) less reactor resistance to pressure and corrosion (Lee 1997; Kuhar et al. 2008; Sanchez 2009). In situ microbial delignification appears to be a feasible strategy for achieving improved depolymerization of hemicellulose and cellulose.

The white-rot fungi (WRF) are the most effective microorganisms for biological pretreatment as they degrade lignin more extensively and rapidly than any other known group of organisms (Eriksson 1993; Kuhad et al. 1997, 2007; Keller et al. 2003). Some WRF have been reported to degrade lignin selectively and this capability of selected WRF can be exploited for delignification of plant materials without affecting much of cellulose (Kuhar et al. 2008; Gupta et al. 2011). Thus, selected

lignin-degrading WRF with comparatively low cellulase and xylanase activities could be advantageous for efficient delignification and eventually in the reduction of chemical and energy inputs for chemical or enzymatic hydrolysis of the substrate(s).

Fungi have two types of extracellular enzymatic systems: hydrolytic and ligninolytic systems. The hydrolytic system produces hydrolases that are responsible for polysaccharide degradation and production of sugars. The unique oxidative and extracellular ligninolytic system degrades lignin component and opens phenyl rings (Lundell et al. 2010). WRF is known to produce one or more of four such cellular enzymes which are of importance for the degradation of lignin. Laccase contains copper (Lac), heme contains manganese peroxidase (MnP) and lignin peroxidase (LiP) and also aryl alcohol oxidase (AAO). Major hydrolytic enzymes like amylases and xylanase are also produced by them (Rosales et al. 2002; Elisashvili et al. 2006) These enzymes of WRF are very significant for the efficient and productive bioconversion of plant residues (Radhika et al. 2013; Songulashvili et al. 2006).

Few studies have been reported on the pretreatment of plant biomass with WRF for its effect on cellulose hydrolysis. According to Hatakka (1983), 35% of the wheat straw is convertible to reducing sugars when pretreated with *Pleurotus ostreatus* for 5 weeks. Taniguchi and co-workers (2005) also recorded a similar conversion rate in rice straw pretreated with *P. ostreatus* for 60 days. Keller et al. (2003) also observed a three- to fivefold improvement in the enzymatic cellulose digest-ibility in corn stover pretreated with *Coriolus versicolor* in over 30 days. Thus, most of these fungal pretreatments have suffered because of long incubation periods. Therefore, to economize microbial pretreatment of lignocellulosic materials, to improve the hydrolysis of carbohydrates to reducing sugars and to eventually improve ethanol yield, there is a need to test more and more basidiomycetous fungi for their ability to delignify the plant material quickly and efficiently (Kuhad et al. 2011).

Biological pretreatment in combination with other pretreatment technologies has also been studied (Itoh et al. 2003; Balan et al. 2008). Itoh et al. (2003) reported production of ethanol by simultaneous saccharification and fermentation (SSF) from beech wood chips after bio-organosolvation pretreatments by ethanolysis and white-rot fungi, Ceriporiopsis subvermispora, Dichomitus squalens, Pleurotus ostreatus and Coriolus versicolor. The yield of ethanol obtained was 0.294 g/g of ethanolysis pulp and 0.176 g/g of beech wood chips after pretreatment with C. subvermispora for 8 weeks. The yield was 1.6 times higher than when obtained without the fungal treatments. The combined process allowed the separation of lignin, cellulose and hemicelluloses with the use of only water, ethanol and wood-rot fungi. The biological pretreatments saved 15% of the electricity which was needed for ethanolysis. In another approach, Balan et al. (2008) studied the effect of fungal treatment of rice straw followed by AFEX pretreatment and enzymatic hydrolysis. They reported that treating rice straw with white-rot fungus, followed by AFEX gave significantly higher glucan and xylan conversions. Recently, it was found that the SSF of Prosopis juliflora and Lantana camara followed by acid hydrolysis and observed that fungal treatment significantly reduce the number of inhibitors generated, and eventually the requirement of the detoxifying agent was also reduced (Gupta et al. 2011). The pretreated cellulose-rich biomass is then hydrolysed with a consortium of cellulolytic enzymes, i.e. endocellulase, exocellulase and β -glucosidase. These enzymes act in synergy and hydrolyse the cellulosic polymer to simple sugars (glucose) (Zhang et al. 2006; Kuhad et al. 2011). Enzymatic hydrolysis has demonstrated better results for the subsequent fermentation because no degradation components of glucose are formed; however, the process is costlier (Sanchez and Cardona 2008). Various improvements have been attempted in the last three decades that would lower the effective enzyme cost, including enzyme reusage, higher enzyme production or using genetically engineered systems (Zhang et al. 2006; Kuhad et al. 2011); but there is still a need to search for more competent solutions.

Since the lignocellulose contains both hexose and pentose sugar, the efficient fermentation of both the sugars has become a prerequisite for the cost-effective production of bioethanol. A variety of microorganisms are known to ferment hexose sugars, but the microbes for pentose fermentation are limited in number. The most common microbes tried for pentose fermentation are *Candida shehatae*, *Pichia stipitis* and *Pachysolen tannophilus* (Abbi et al. 1996a, b; Kuhad et al. 2011). However, none of these yeasts have been found to be very promising. Intensive efforts have been made towards the utilization of all the sugars present in hydrolysates by employing both genetic manipulations as well as process improvement approaches. Currently, most commercial cellulases are produced from *Phanerochaete* sp., *Aspergillus* sp. and *Trichoderma reesei* (Adney et al. 2003), usually used to describe a mixture of cellulolytic enzymes whose synergistic action is required for effective breakdown of substrate to its monomeric units.

The action of cellulases involves the concerted action of endoglucanases (endo-1, 4- β -glucanases, EGs), which can hydrolyse internal bonds preferably in cellulose amorphous regions releasing new terminal ends, which randomly attacks the internal, β 1,4-linkages cellobiohydrolase (exo-1,4 β -glucanases, Cbhs), which acts on the existing or endoglucanase generated chain ends. Both enzymes can degrade amorphous cellulose but, with some exceptions, Cbhs are the only enzymes that efficiently degrade crystalline cellulose. Cbhs and EGs release cellobiose molecules which cleaves off cellobiose units from the non-reducing ends of the glucan and β 3-glucosidase, which hydrolyses cellobiose to glucose. The pretreated substrate can be then hydrolysed with microorganisms and subsequently be fermented to ethanol, which will serve as biofuel. Overall, the use of biomass to produce bioethanol holds much promise for providing a renewable, indigenously produced liquid energy source that can be a viable alternative to petroleum-based fuels.

5 Conclusion

Plant cells are mainly composed of lignocellulosic material, which includes cellulose, hemicellulose and lignin. Pretreatment of lignocellulosic biomass prior to hydrolysis significantly improves the efficiency of hydrolysis due to the removal of lignin and hemicellulose and reduction of cellulose. Microorganisms, especially fungi, can degrade the plant cell wall through a set of enzymes. Fungi have two types of extracellular enzymatic systems: hydrolytic and ligninolytic systems. The hydrolytic system produces hydrolases and the ligninolytic system degrades lignin component and opens phenyl rings. The white-rot fungi (WRF) are effective for biological pretreatment as they degrade lignin more extensively and rapidly than other microorganisms. The selected lignin-degrading WRF with comparatively low cellulase and xylanase activities could be advantageous for efficient delignification and eventually in the reduction of chemical and energy inputs for chemical or enzymatic hydrolysis of the substrate.

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Chapter 6 Myco-Degradation of Lignocellulose: An Update on the Reaction Mechanism and Production of Lignocellulolytic Enzymes by Fungi



Arvind Kumar

1 Introduction

Lignocellulosic biomass is the most abundant renewable feedstock for secondgeneration biofuel and other bio-based valuable products. It is constantly produced as a by-product of agro-industry, forestry, and solid waste of municipality. Lignocellulosic biomass represents approximately half of the global biomass production. Dry lignocellulosic biomass production was estimated 10–50 tons per year worldwide (Galbe and Zacchi 2002). Lignocellulose is majorly composed of cellulose, hemicellulose, and lignin. Their proportions are dependent on the source of biomass. For example, agricultural residues contain approximately 30–40% cellulose, 20–30% hemicellulose, and 15–30% lignin, whereas angiosperm and gymnosperm woods (hardwood and softwood) contain 40–42% cellulose, 20–30% hemicellulose, and 25–32% lignin (Bornscheuer et al. 2014). In general scenario, it has been proposed that lignocellulosic materials hold 40–60% cellulose, 20–40% hemicellulose, and 10–25% lignin (Jaramillo et al. 2015).

Cellulose is a linear polymer of D-glucopyranose subunits joined together by β -1,4-glycosidic linkage (Bornscheuer et al. 2014). Each glucopyranosyl subunit is rotated 180° in relation to its neighbor molecule forming the repeat unit of cellulose known as cellobiose. The chain length of monomeric subunits in a cellulose molecule depends on the source, and it varies from 250 residues to well over 10,000 residues. These molecules form intra- and intermolecular hydrogen bonds and are compactly packed at the distance of van der Waals' radii due to the absence of side groups which accounts for the rigid and insoluble (crystalline) nature of cellulose microfibrils. Cellulose molecules also contain amorphous regions in addition to

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crystalline regions, and generally a cellulose microfibril is composed of 36 chains (Guerriero et al. 2016).

Hemicelluloses are a group of linear and branched complex carbohydrate structures arranged in different proportions with different substituents. Hemicelluloses are mainly composed of xylose, arabinose, glucose, mannose, and galactose residues. Thus, they are the polymers of pentoses (xylan and araban), hexoses (mannan, glucan, glucomannan, and galactan), and pectin (Bornscheuer et al. 2014; Johansen 2016). Among the hemicelluloses found in the plant cell walls, xylan is characterized as polymer of β -1,4 linked xylosyl residues substituted with acetyl, feruloyl, glucuronic acid, 4-*O*-methylglucuronic acid, arabinose, and galactose side chains (Guerriero et al. 2016). Hemicelluloses of plant origin are also known as acidic and neutral hemicelluloses based on the glucuronic and 4-*O*-methylglucuronic acid content. Similarly, araban, mannan, and galactan are polymers of arabinose, mannose, and galactose, whereas glucomannan is composed of mannose chains interrupted by glucose residues. Hemicelluloses are often joined with lignin which provides additional strength for structural stability.

Lignin is a complex heterogeneous polymer of phenylpropanoid alcohols known as monolignols. These monolignols are *p*-coumaryl, coniferyl, and sinapyl alcohols polymerized by ether and C-C bonds via free radical reaction (Guerriero et al. 2016). Gymnosperm lignin is almost entirely composed of coniferyl alcohols, whereas angiosperm dicot lignin is a mixture of coniferyl and sinapyl monomers. All the three kinds of monolignols are constituents of monocotyledonous lignin (Bornscheuer et al. 2014). The ratios of constituent monomers vary in different plants, woods tissues, and cell wall layers (Johansen 2016).

Plant-derived biomass is a natural resource of carbohydrate polymers. Its utilization for value and revalue has received considerable interest in the recent years with ever-increasing scientific knowledge, worldwide economic and environmental consciousness, demands of legislative authorities, and the manufacture, use, and removal of petrochemical-based by-products (Arevalo-Gallegos et al. 2017). Since a lignocellulosic material consists of 70% sugars, therefore, it is considered one of the valuable biomass for future bio-economy and alternate energy source (Horn et al. 2012; Arevalo-Gallegos et al. 2017). Such biomass offers a great potential for biofuel and many other valuable products without any competing interest for food production and also provides an opportunity for reduced greenhouse gas emission. Thus, biorefinery approach is emerging as a promising sector for the capitalization of lignocellulosic biomass into a variety of bio-based useful products such as biodegradable plastic, biochemicals, and advanced biofuels (Arevalo-Gallegos et al. 2017). The term biorefinery refers to the sustainable conversion of biomass into a variety of valued chemicals and energy in an economical and environment-friendly manner (Arevalo-Gallegos et al. 2017). The important biorefinery-based products/ chemicals selected by the US Department of Energy from lignocellulose are ethanol, furans (furfural and 5-hydroxymethyl furfural), glycerol and its derivatives (propanediol, glycerol carbonate, and epichlorohydrin), hydrocarbons, lactic acid, succinic acid, levulinic acid, sorbitol, and xylitol (Chandel et al. 2018). Excluding

biofuels, the production of bio-based other valued chemicals is estimated to be 50 billion kilos per year with a 3–4% annual growth (de Jong 2011).

Recalcitrant nature is the major obstacle in the saccharification and bioconversion of lignocellulosic material into valued end products. This recalcitrance arises due to an intricate arrangement between cell wall polysaccharides (cellulose and hemicellulose) and lignin (Jaramillo et al. 2015). The paracrystallinity of cellulose, complex hemicellulose coating of cellulose microfibrils, and interpenetration and encapsulation of polysaccharides by lignin impede the accessibility of lignocellulosic materials for carbohydrases, ligninases, and accessory proteins (Cragg et al. 2015). These barriers are overcome by harsh physical, chemical, and physicochemical pretreatments in the industries which aid production cost. High cost of biofuel production is the major constraint of commercialization. It has been estimated that cellulosic ethanol production cost (\$0.5 per liter) is almost double in comparison with starch-based ethanol (Bornscheuer et al. 2014). A considerable cost in the biomass-based production of biofuel is related with enzymes. Thus, it necessitates the development of an efficient and economical lignocellulolytic processing for viable commercialization. Microbial sources of enzymes for lignocellulose biomass degradation offer opportunities to overcome the high cost barrier. Microorganisms degrade lignocellulosic biomass under physiologically tolerable conditions, rendering them to be the suitable host for the production of efficient lignocellulolytic enzymes. Among the microorganisms, saprophytic fungi express higher levels of lignocellulose biomass degrading enzymes and are still considered the most promising hosts to deal with the challenging demand of lignocellulose utilization (Mitrovic and Glieder 2015; Li et al. 2017). There has been intensive progress in the screening of hyper lignocellulolytic fungi and development of efficient cellulolytic platforms in the recent years. Thus, this chapter discusses lignocellulose-degrading enzymes, mechanisms of enzyme actions, and some recent developments in the production of lignocellulolytic enzymes by fungal strains.

2 Lignocellulose-Degrading Enzymes

Enzymes for deconstruction of lignocellulosic material are broadly categorized into two classes known as glycosyl hydrolases and oxidative enzymes. Glycosyl hydrolases degrade the Earth's most abundant cellulose and hemicellulose carbon polymers, whereas oxidative enzymes decompose the most recalcitrant lignin biomolecule as well as cellulose (Hallberg et al. 2003; Mitrovic and Glieder 2015; Beeson et al. 2015). Efficient degradation of lignocellulosic material requires a plethora of enzymes collectively known as enzyme cocktail. Enzymatic cocktail is composed of cellulases, hemicellulases, ligninases, auxiliary activity, and accessory proteins which are discussed in the later sections of the text.

2.1 Cellulases

Cellulases play an important role in the development of an efficient lignocellulolytic process because cellulose content of the lignocellulosic materials is up to 60% (Bilal et al. 2017). Cellulases include a consortium of β -1,4-endoglucanases (EC 3.2.1.4), cellobiohydrolases (EC 3.2.1.91), and β -glucosidases (EC 3.2.1.21) which hydrolyze polymeric cellulose into fermentable sugars. Endoglucanases randomly hydrolyze internal β -1,4-glycosidic linkages in the amorphous regions of the cellulosic microfibrils (Taha et al. 2016; Lopes et al. 2018). Thus, oligosaccharides or smaller polysaccharides are released from long-chain cellulose molecules. Cellobiohydrolases are exo-acting enzymes and catalyze the hydrolysis of glycosidic bonds in the crystalline regions of cellulose. Cellobiohydrolases I and II release glucose or cellobiose molecules from reducing and non-reducing ends, respectively. The released cellobiose and oligosaccharide molecules are further cleaved into monomeric constituents by β -glucosidases (Lopes et al. 2018; Passos et al. 2018; Marques et al. 2018). β-Glucosidases are not directly acting on cellulose and thus are not considered as true cellulases. However, they are important components of the complete saccharification process.

Since cellulose is an insoluble substrate, therefore, endoglucanases have to diffuse, attach, and move a segment of the polymeric substrate to their active sites. Endoglucanases are modular proteins. They consist of catalytic domains connected with carbohydrate binding modules via flexible linker segment. The carbohydrate binding domain recognizes the substrate and adsorbs the enzyme onto the surface of cellulose microfibrils. In this way, they are increasing the effective concentration of the enzyme molecules at the substrate's surface. Additionally, it has also been postulated that carbohydrate binding modules have the ability to disrupt crystalline structure of cellulose (Shoseyov et al. 2006; Araki et al. 2010; Behera et al. 2017). The evidence supporting the above statement has accumulated from the assessment of adsorption and degradation ability of the carbohydrate binding module deleted enzymes. Such truncated enzymes showed a little change in their degradation activity at soluble substrates while their activity and adsorption declined significantly on crystalline cellulose (Behera et al. 2017). Chemical and site-directed mutagenesis studies demonstrated that aromatic amino acids play an important role in the adsorption process. Aromatic amino acids help in the attachment of the enzyme on the insoluble substrate surface by establishing accumulation forces between glucose rings and π -electrons of the aromatic rings (Chen 2014). Therefore, the overall impact of the presence of carbohydrate binding domain is to increase the enzyme's adsorption on crystalline cellulose substrate.

2.2 Hemicellulases

Hemicellulose is a complex and heterogenous polysaccharide mostly composed of xylan and mannan. The full depolymerization of hemicellulose requires synergistic action of an array of enzymes, e.g., xylanases, mannanases, and other side chain

cleaving enzymes depend on the origin of the substrate. Xylanases group includes β -1,4-endo-xylanases (EC 3.2.1.8), β -xylosidases (EC 3.2.1.37), arabinofuranosidase (EC 3.2.2.55), α -glucuronidase (EC 3.2.1.131), acetyl xylan esterase (EC 3.1.1.6), ferulic acid esterase (EC 3.1.1.73), and p-coumaric acid esterase (EC 3.1.1.73). Among a variety of xylanases; β -1,4-endo-xylanases hydrolyze internal β-1,4-glycosidic linkage in the xylan backbone and release short xylooligosaccharides of varying length (Gong et al. 2013; Kim et al. 2014). The released xylooligosaccharides were hydrolyzed by β -xylosidases from the non-reducing end. The side chain acting enzymes; α -L-arabinofuranosidase releases terminal α -L-1, 2-, α -L-1, 3-, and α -L-1, 5-linked arabinofuranosyl residues, α -glucuronidase catalyzes the cleavage of α -1, 2 glycosidic bond between O-4-methyl glucuronic acid and main chain of glucuronoxylan, acetyl xylan esterase cleaves acetyl groups from acetyl xylan, and ferulic acid and coumaric acid esterases hydrolyze the ester linkage between arabinose and monomeric or dimeric ferulic acids, and arabinose and *p*-coumaric acid, respectively (Lopes et al. 2018; Marques et al. 2018). Xyloglucan endotransglycosylase/hydrolase (EC 2.4.1.270) enzyme catalyzes both transglycosylation and hydrolysis reactions. The hydrolytic activity degrades xyloglucans by breaking β -1,4-glucosyl linkages. Other enzymes assisting in the xyloglucan hydrolysis are xyloglucan-specific β -D-galactosidase (EC 3.2.1.23), xyloglucan-specific α -L-fucosidase (EC 3.2.1.51/63), and xyloglucan oligosaccharide-specific α -Dxylosidase (EC 3.2.1.37).

As with xylan, mannan deconstruction also requires a collection of enzymes including β -mannanase (EC 3.2.178), β -mannosidase (EC 3.2.1.25), β -glucosidase (EC 3.2.1.21), acetyl mannan esterase (EC 3.1.1.6), and α -galactosidase (EC 3.2.1.22). β -Mannanase is an endo-active enzyme which catalyzes hydrolysis of internal β -1,4-glycosidic bonds in the mannan backbone in a similar fashion as does endoxylanase. β -Mannosidase and β -glucosidase are exo-acting enzyme and release mannose and glucose, respectively, from the non-reducing ends of mannan and mannooligosaccharides (Inoue et al. 2015; Lopes et al. 2018). In addition to above, debranching enzyme α -galactosidase cleaves α -1,6-linked D-galactopyranoside residues from galactomannan and galactoglucomannan.

2.3 Ligninases

Ligninases are a group of broad substrate specificity enzymes which are capable of oxidizing randomly joined phenylpropanoids units in the lignin polymer. The four major lignin decomposing enzymes are manganese peroxidase (EC 1.11.1.13), heme-containing lignin peroxidase (EC 1.11.1.14), versatile peroxidase (EC 1.11.1.16), and Cu-containing laccase (EC 1.10.3.2) (Jaramillo et al. 2015; Andlar et al. 2018). Lignin peroxidases (LiP) mediate oxidation of phenolic compounds via utilizing hydrogen peroxide, and manganese peroxidases (MnP) oxidize Mn²⁺ to Mn³⁺ which oxidizes phenolic structures into unstable free radicals. Versatile peroxidases (VP) combine the activities of LiP and manganese-dependent peroxidases.

The combinatorial characteristics of versatile peroxidases render them to oxidize Mn²⁺and phenolic/non-phenolic substrates including veratryl alcohol, methoxybenzenes, and lignin model compounds (Andlar et al. 2018; www.cazy.org). Laccases catalyze oxidation of phenolic and aromatic amines by using molecular oxygen as a terminal electron acceptor (Jaramillo et al. 2015; Lopes et al. 2018). The other enzymes showing synergism in the lignin degradation are aryl alcohol oxidase (EC 1.1.3.7), glyoxylate oxidase (EC 1.2.3.5,), pyranose 2-oxidase (EC 1.1.3.4), and cellobiose/quinone oxidoreductases (EC 1.1.5.1).

2.4 Auxiliary Activity Proteins/Oxidative Enzymes

In addition to lignin enzymes, the other proteins classified in the auxiliary activity families also exhibit oxidative characteristic and boost the glycosyl hydrolase activity either by improving the accessibility of lignocellulosic material or by removing inhibitory products. The other most studied oxidative enzymes involved in the lignocellulolytic processes are cellobiose dehydrogenases (EC 1.1.99.18) and lytic polysaccharide monooxygenases (LPMOs). Cellobiose dehydrogenases classified in the subfamily 1 of auxiliary activity 3 family (AA3_1) oxidize cellobiose, lactose, and in some cases glucose with the reduction of cofactor flavin adenine mononucleotide. Thus, relief of cellulase inhibition occurs with the removal of cellobiose, a product inhibitor (Lopes et al. 2018). Initially, LPMOs were believed to act on crystalline substrates chitin and cellulose, but now it has been established that they act on a variety of substrates including starch, xyloglucan, cellodextrins, and glucomannan (Johansen 2016). LPMOs have been placed in the auxiliary activity families 9, 10, 11, 13, and 14 (Levasseur et al. 2013; Couturier et al. 2018). These auxiliary activity family proteins are Cu-dependent LPMOs which cleave crystalline polymeric substrates with oxidation of C-1 and C-4 (www.cazy.org). Therefore, the major role of auxiliary activity proteins is to enhance the amorphogenesis and facilitate the action of lignocellulolytic enzymes (Andlar et al. 2018).

2.5 Accessory Proteins

The accessory proteins have a potential to support the feasibility of lignocellulolytic process by enhancing the degradation of lignocellulosic materials. However, there is no evidence favoring the direct involvement of accessory proteins in the cleavage of glycosidic linkage either by hydrolysis or oxidative roles. It is believed that accessory proteins break intra- and intermolecular hydrogen bonds, thereby reducing the crystallinity and recalcitrance of the substrates, enhancing the substrate access for hydrolytic and oxidative enzymes (Quiroz-Castañeda et al. 2011; Lopes et al. 2018; Andlar et al. 2018). In plants, the loosening of lignocellulosic matrix and the enlargement of plant cell wall have been assigned to the action of expansins which break

down the hydrogen bonds between cellulose microfibrils or between cellulose molecules of neighboring cells (Lopes et al. 2018). Though initially reported in plants, recent studies emphasize the presence of expansins in bacteria and expansin-like proteins in fungi Quiroz-Castañeda et al. 2011. These expansin-like proteins in conjunction with conventional cellulases have greatly improved the hydrolysis of cellulose even though without any proven record of hydrolytic activity, pointing a clue for their synergistic action (Payne et al. 2015; Cosgrove 2017). Similarly, swollenin proteins produced by fungi show sequence similarity to expansins, and thus similar function can be anticipated for them. Congruently, swollenins promoted swelling of the cellulose fibers resulting in the reduction of particle size and crystallinity (Kubicek et al. 2014; Lopes et al. 2018). Likewise, positive synergism has also been observed between swollenins and cellulases (Santos et al. 2017; Andlar et al. 2018). Loosenins are other expansin-type proteins which enhance cellulase complex activity by disrupting intramolecular hydrogen bonds (Quiroz-Castañeda et al. 2011).

Hydrolytic and oxidative enzymes as well as accessory proteins have been separately discussed ascribing to the degradation of different components of lignocellulosic biomass. However, consideration of the interplay between the activities of these enzymes is noteworthy. The actions of one category of enzymes augment the action of another category of enzymes or vice-versa for efficient decomposition of lignocellulosic biomass.

3 Reaction Mechanisms of Enzymes Involved in Lignocellulose Degradation

Based on the type of reaction catalyzed, lignocellulose-degrading enzymes can be put into three major classes, i.e., glycosyl hydrolases which strictly carry out hydrolysis of glycosidic bond via utilizing water molecule, oxidative enzymes break glycosidic linkage by oxidizing one of the carbons of linked monomers, and accessory proteins disrupt intra- and intermolecular hydrogen bonds. Here in the following sections, reaction mechanisms of glycosyl hydrolases and oxidative enzymes are described briefly.

3.1 Reaction Mechanisms of Glycosyl Hydrolases

The reaction mechanism of glycoside hydrolases (cellulases, hemicellulases, and pectinases) is conserved among all the classes. They catalyze the reaction either by retention or inversion mechanism, i.e., either retaining or inverting the configuration of the anomeric carbon of the hydrolyzed bond (Biely et al. 1994; Davies and Henrissat 1995; Collins et al. 2005). Both types of reaction mechanisms utilize general acid-base catalysis with the help of carboxylate pair at the enzyme's active

site. Either in retaining or inverting enzymes, one residue of the carboxylate pair localized at the distance of hydrogen bonding with glycosidic oxygen, acts as a general acid, and protonates oxygen of the O-glycosidic bond. However, another residue of carboxylate pair is located in the close vicinity of the sugar anomeric carbon in retaining enzymes which acts as nucleophile/base. This base is more distantly localized in inverting enzymes, and thus, a water molecule can be accommodated between the sugar anomeric carbon and the base. Briefly, the active site of retaining enzymes is constituted by two glutamic acid residues separated from each other with a distance of approximately 5.5 Å. However, carboxylate pair (glutamate and aspartate) of inverting enzymes' active site is approximately 7.5-10 Å apart from each other (Collins et al. 2005; Bhardwaj et al. 2012; Gupta et al. 2013). This distance creates micro-space to admit a water molecule between two catalytic residues in inverting enzymes. Mechanistically, retaining enzymes follow double displacement mechanism for the cleavage of glycosidic bond. In the process of double displacement, a covalent glycosyl enzyme intermediate is formed which is subsequently hydrolyzed by oxocarbenium-ion-like transition states (Rye and Withers 2000). The first step of the reaction is the protonation of the glycosidic oxygen via one of the carboxylate pair residues. Simultaneously, another residue attacks nucleophilically at the anomeric carbon which results in the removal of leaving group and formation of the α -glycosyl enzyme intermediate. In the second step, the first carboxylate residue (which protonated the glycosidic oxygen in the first step) acts as a general base and withdraws a proton from water molecule, thus activating water molecule for nucleophilic attack at the anomeric carbon. Overall step two events result in a second substitution in which the anomeric carbon forms oxocarbeniumion-like transition state and is subsequently hydrolyzed to a β -configuration end product. Therefore, the ultimate effect is the retention of the anomeric carbon configuration as of the substrate. In contrary to the above, inverting enzymes perform hydrolysis via single displacement mechanism in which one of the carboxylate groups acting as general acid provides proton for the leaving group departure and another accepts a proton from the water molecule, thus activating the nucleophilic attack of the water molecule at anomeric carbon. It produces oxocarbenium-ion-like transition state, which is subsequently hydrolyzed into α -configuration end product (inversion β to α) (Collins et al. 2005).

3.2 Reaction Mechanism of Ligninases

As stated earlier, lignin is the most recalcitrant degraded by the different enzymes with varying specificities. The common feature of all enzymes is the utilization of oxidative reaction mechanism. Lignin decomposing peroxidase (LiP, MnP, and VP) enzymes follow general mechanism of horseradish peroxidases. The only difference is in their high oxidation potential. The large information has accumulated from several studies on the structure, reaction mechanisms, and kinetics of ligninases (Hammel et al. 1985; Glenn et al. 1986; Siddiqui et al. 2014; Lin et al. 2018; Fan

et al. 2019) and reviewed by different authors which serve as the source of compiled information (Kirk and Farrel 1987; Martinez 2002; Kersten and Cullen 2007; Wong 2009; Bouacem et al. 2018). In the following section, we discuss reaction mechanisms of four types of lignin decomposing enzymes.

3.2.1 Reaction Mechanism of Lignin Peroxidases (LiPs)

LiPs are H_2O_2 requiring glycoproteins which contain one mole of iron protoporphyrin IX per protein molecule. The heme is sandwiched between proximal (C-terminal) and distal domains (N-terminal) and accessible from the solvent via two small channels. The heme iron is predominantly pentacoordinated at the distal side with histidine (Edwards et al. 1993; Claus 2004). The catalytic cycle starts with two electron oxidation of the ferric enzyme [Fe(III)] leading to the formation of intermediate compound I (LiP-I, ferry iron porphyrin radical cation). The electrons are accepted by H₂O₂and heterolytic cleavage of O-O of peroxide substrate is facilitated by coordination of heme Fe(III) and protonation by distal histidine 47. The leaving H₂O molecule promotes transient formation of Fe(III)-O⁺, followed by electron transfer from Fe(III) and porphyrin π electron system to form ferry iron porphyrin radical cation. Thereafter, one electron reduction of LiP-I by different phenolic and aromatic non-phenolic substrates takes place in two successive steps (Renganathan and Gold 1986; Wong 2009). The first step of reduction of LiP-I by substrate forms compound II [Fe(IV)=O.porphyrin, LiP-II]. This compound II is converted into resting enzyme at the second successive step of reduction by another molecule of substrate. In some cases, LiP-I may also be directly renewed to the resting state via one equivalent of two electron substrate such as veratryl alcohol. In the reduction of LiP-I, susceptible aromatic nuclei are oxidized by one electron producing unstable cation radicals (Renganathan and Gold 1986; Lundell et al. 1993; Wong 2009). These cation radicals undergo a variety of non-enzymatic reactions leading to side chain cleavage, demethylation, intramolecular addition, rearrangement, and depolymerization of lignin molecules. Thus, the formation of radical cations is enzyme catalyzed, and subsequent steps are non-enzymatic which may take place spontaneously depending on the substitution of aromatic ring.

LiP-II may also be converted into inactive form III [LiP-III, Fe(III) O_2^{-1}] in the presence of excess H_2O_2 and in the absence of reducing substrates. This inactive form of enzyme (LiP-III) is reactivated either spontaneously or by veratryl alcohol radical cation with the displacement of superoxide (Wong 2009). It has been observed that lignin peroxidase producing organism (*Phanerochaete chrysosporium*) also produces veratryl alcohol (Kirk and Farrel 1987). This veratryl alcohol is oxidized to veratryl alcohol radical cation during the reduction steps of LiP-I and LiP-II to the native form enzyme. Therefore, it may be an adaptive strategy of the lignin degrading organisms to overcome enzyme inactivation which has also been supported by the findings of increase in enzyme activity with the extracellular addition of veratryl alcohol(Harvey et al. 1986; Martinez 2002). Another suggested role of synthesized veratryl alcohol radical cation is redox mediator in the oxidative

degradation of lignin (Kersten and Cullen 2007). It has been proposed that instead of direct lignin oxidation by LiP (large molecular weight enzyme, unable to diffuse into the vicinity of lignin molecules), veratryl alcohol radical cation (small molecule) diffuses into the interior of lignin and acts as direct oxidant for lignin degradation.

$$\operatorname{LiP}\left[\operatorname{Fe}(\operatorname{III})\operatorname{.porphyrin}\right] + \operatorname{H}_{2}\operatorname{O}_{2} \rightarrow \operatorname{LiP-I}\left[\operatorname{Fe}(\operatorname{IV}) = \operatorname{O.porphyrin}^{*}\right] + \operatorname{H}_{2}\operatorname{O} \quad (6.1)$$

 $LiP-I[Fe(IV) = O.porphyrin^{+}] + Subtrate \rightarrow LiP-II[Fe(IV) = O.porphyrin]_{(6.2)} + Free radical cation$

$$LiP-II[Fe(IV) = O.porphyrin] + Substrate \rightarrow LiP[Fe(III).porphyrin] + H_2O + Free radical cation$$
(6.3)

3.2.2 Reaction Mechanism of Manganese Peroxidases (MnP)

MnP contain heme and manganese [Mn(II)] ion as their integral structural components, and they are H_2O_2 requiring obligatory manganese oxidizing (Mn²⁺ to Mn³⁺) enzymes (Glenn et al. 1986). MnP is similar to LiP, consisting of heme sandwiched between proximal and distal domains. Likewise, LiP heme iron is mainly pentacoordinated with distal histidine ligand. The Mn-binding site consists of three acidic amino acid residues (Glu35, Glu39, and Asp179) and Mn coordinates with carbonyl oxygen of these residues, heme propionate oxygen, and two other water oxygens (Sundaramoorthy et al. 1997). The catalytic cycle involves two electron oxidation of MnP into MnP-I in the presence of H₂O₂ leading to the heterolytic cleavage of O-O peroxide substrate. These two electrons are transferred from heme-porphyrin component of the enzyme in the same manner as with LiP. Transient-state kinetics has demonstrated that the reduction of MnP-I into MnP-II uses either Mn²⁺ or a variety of phenolic compounds as reducing substrate (Kersten and Cullen 2007; Wong 2009). However, the reduction of MnP-II into the native form enzyme is carried out only by Mn²⁺ demonstrating its unique characteristic of Mn²⁺ dependency (Martinez 2002). The overall process leads to the formation of Mn^{3+} ions which are potent oxidizers and can oxidize different types of phenolic compounds including lignin model compounds. However, Mn³⁺ is quite unstable in the aqueous solution, and its stabilization is achieved with chelating molecules (oxalic and other organic acids) secreted by the fungi. The chelated Mn³⁺ diffuses in the interior of lignocellulosic material and causes one electron oxidation of phenolic lignin constituents to produce unstable radicals (Kuan et al. 1993). These unstable radicals undergo rearrangement, bond cleavage, and non-enzymatic degradation to yield different products. The Mn³⁺ also produces thiol and lipid-derived free radicals in the presence of glutathione and lipids, respectively. The synthesized thiol and lipid-derived free radicals abstract hydrogen from non-phenolic substrates yielding benzylic

radical (Kersten and Cullen 2007). The formed benzyl radicals yield different products by non-enzymatic reactions.

Similar to LiP, MnP-II can also be converted into an inactive Fe(III)-superoxo form (MnP-III) in the presence of excess H_2O_2 . The reactivation uses Mn^{3+} to act either as oxidizing agent of iron coordinated superoxide or catalase type to remove H_2O_2 . Therefore, the reactivation of MnP-III is different from LiP-III where it uses aromatic radical cation.

$$MnP[Fe(III).porphyrin] + H_2O_2 \rightarrow MnP-I[Fe(IV) = O^{+}] + H_2O \quad (6.4)$$

$$MnP-I\left[Fe(IV) = O^{+}\right] + Mn^{2+} \rightarrow MnP-II\left[Fe(IV) = O.porphyrin\right] + Mn^{3+}$$
(6.5)

$$MnP-II\left[Fe(IV) = O.porphyrin\right] + Mn^{2+} \rightarrow MnP\left[Fe(III).porphyrin\right] + Mn^{3+} + H_2O$$
(6.6)

 Mn^{3+} + Phenolic substrate $\rightarrow Mn^{2+}$ + Phenolic substrate radical cation + H₂O (6.7)

3.2.3 Reaction Mechanism of Versatile Peroxidases (VP)

The versatile peroxidases share the characteristics of LiP and MnP. They oxidize Mn^{2+} into Mn^{3+} as well as phenolic and non-phenolic substrates (veratryl alcohol, *p*-dimethoxybenzene, and lignin model dimer veratrylglycerol- β -guaiacyl ether). Owing to its catalytic properties, enzyme molecule contains Mn^{2+} binding site (formed by acidic residues Glu/Glu/Asp) similar to MnP and other amino residues which are involved in the interaction of phenolic and non-phenolic substrates (Kersten and Cullen 2007; Wong 2009; Siddiqui et al. 2014). The reaction of mechanism of versatile peroxidase is similar to that of LiP and MnP. The reaction is triggered in the presence of H_2O_2 involving the formation of two electron oxidized compound I and one electron oxidized compound II intermediates (Martinez 2002; Wong 2009). Compound I intermediate is reduced to compound II by one electron oxidation of either Mn^{2+} or phenolic/nonphenolic substrates depending on the availability; subsequently compound II is converted into native form enzyme in another round of one electron oxidizable substrates.

In the above sections, we had seen that all the lignin decomposing peroxidase enzymes (LiP, MnP, and VP) require H_2O_2 . This necessity has prompted the research for the discovery of H_2O_2 producing enzymes. Soon, the observation of glucose oxidase (glucose-1-oxidase and pyranose 2-oxidase) activity in the ligninolytic mycelia was attributed to the primary source of H_2O_2 (Kelley and Reddy 1986; Daniel et al. 1994). Additionally, several putative aryl alcohol oxidases (structurally related to glucose oxidase) have been implicated in the extracellular production of H_2O_2 to start the redox reaction. The exploration of other possibilities revealed the involvement of fatty acyl-CoA oxidase in the washed and starved

mycelia in the absence of substrates. Glyoxylate oxidase activity observed in the ligninolytic cultures may also be responsible for H_2O_2 production by oxidizing several simple aldehydes, α -hydroxy carbonyl and dicarbonyl compounds (Kersten and Cullen 2007). The most important implication in the start of a catalytic cycle has been evidenced with H_2O_2 production by MnP which oxidizes glutathione, dithiothreitol, and dihydroxymaleic acid with the coupled reduction of O_2 to H_2O_2 (Kirk and Farrel 1987; Wong 2009).

3.2.4 Reaction Mechanism of Laccases

Laccases are blue multicopper oxidases structurally characterized by the presence of three distinct domains, each of which displays Greek key β -barrel architecture known as cupredoxin-like domains (Mot and Silaghi-Dumitrescu 2012). The active site is composed of three types of four copper centers per molecule, i.e., one type 1, one type 2, and two type 3. Type 1 (T1) copper is characterized by paramagnetic behavior and strong electronic absorption at 614 nm due to covalent bond between copper and cysteine and thus imparts the blue color. Type 2 (T2) copper is also paramagnetic but shows no absorbance in the visible range. Type 3 (T3) copper center is a pair of two copper ions designated as $T3\alpha$ and $T3\beta$ which gives a weak absorbance at 330 nm and shows no EPR signal. Thus, the copper pair of T3 center is linked by anti-ferromagnetic coupling and maintained by a hydroxyl bridge (Claus 2004; Sitarz et al. 2016). T2 is strategically positioned close to T3 and forms a T2/T3 trinuclear cluster consisting of three of the four copper ions in three reaction nuclei sites. T1 is located in domain 3 with copper positioned in a shallow depression on the surface of the enzyme, whereas T2/T3 trinuclear cluster is located at the interface between domains 1 and 3. T1 copper coordinates with two histidines and one cysteine as conserved equatorial ligands. Additionally, the axial position contains non-coordinated phenylalanine or leucine and isoleucine residues in the vicinity of T1 copper center. The presence of these non-coordination ligands in fungal laccases has been considered as an advantage in determining the higher redox potential of T1 copper site which is approximately +790 mV (Claus 2004; Wong 2009; Sitarz et al. 2016). Due to high redox potential, T1 copper is the primary electron acceptor site where successively four one electron oxidations of substrate take place. T2 copper coordinates with two histidines and one oxygen ligand (hydroxide or water). Each of the T3 copper ions symmetrically coordinated with three histidines and a bridging oxygen (either hydroxyl or O²⁻). T2/T3 trinuclear cluster is the active center for the reduction of molecular oxygen into water. Therefore, four successive electrons abstracted from substrate at T1 site need to be transported to T2/T3 trinuclear site. However, T1 and T2/T3 trinuclear sites are 13Å distant from each other. Thus, a conserved tripeptide (His-Cys-His) is involved in the electron transfer pathway between these sites (Mot and Silaghi-Dumitrescu 2012). Additionally, two conserved amino acid residues are present in the close

proximity of the catalytic pocket (deprotonated glutamic acid close to T3 and protonated aspartic acid close to T2).

The resting form of the enzyme is fully oxidized, and all the four copper centers are in their (++) oxidation state. The initial phase of the catalysis takes place at T1 center with the abstraction of one electron from the reducing substrate. Successively four electrons are abstracted and transferred to T2/T3 trinuclear cluster forming fully reduced form (active form) enzyme (Solomon et al. 2008; Wong 2009). The reduction of the T3 copper centers results in the weakening of the hydroxide bridge between T3 α and T3 β . Thus, the hydroxide is being protonated from the bulk solvent and released as water molecule (Claus 2004). Simultaneously, catalytic site is vacant to accept molecular oxygen; however, before the binding of the molecular oxygen to T2/T3 site, glutamic acid needs to be protonated from the bulk solution and aspartic acid deprotonated to lose its H⁺ to the hydroxide bound with T2 (forming water molecule). Thereafter, dioxygen docks into the catalytic cavity and preferentially binds with T3BCu⁺ due to its trigonal pyramidal geometry (Augustine et al. 2010; Sitarz et al. 2016). The bound oxygen is reduced by two electrons with one electron transfer from each T3pCu⁺ and T2Cu⁺, forming peroxide intermediate. Still the two electron reduced dioxygen is tightly bound to the trinuclear cluster by coordinating with T3 β Cu²⁺, T3 α Cu⁺, and T2Cu²⁺. The next step is the rapid one electron transfer from T1Cu⁺ to T2Cu²⁺ with the help of T1-Cys-His-T3β pathway (Piontek et al. 2002). Thus, T2Cu²⁺ is reduced to T2Cu⁺ which loosens the bond with peroxide. Thereafter, two electrons, one from T2Cu⁺ and another from $T3\alpha Cu^+$, are transferred to the antibonding orbitals of peroxide. It increases the energy and destabilizes the peroxide structure, leading to the breakage of peroxide (O-O) bond (Sitarz et al. 2016). This step is facilitated by concurrent one proton transfer to peroxide from glutamic acid and protonation of aspartate by abstracting proton from water molecule bound with T2Cu⁺. The overall stoichiometry of this step is the production of fully oxidized native intermediate state of the active site in which two oxides bridge in the catalytic cavity. The native intermediate form is converted to either fully reduced active form or fully oxidized resting form containing one oxide bridge (Solomon et al. 2008; Sitarz et al. 2016). It depends on the availability of reducing substrate. In the presence of enough reducing substrate, the native intermediate is converted to fully reduced form of enzyme by accepting four electrons and four protons. Three of these four protons are directly transferred to oxide bridges releasing two water molecules and one protonated glutamate; at the same time, one proton from aspartic acid is transferred to OH bonded with T2 copper center. Thus, it produces the fully active form of the enzyme ready for another cycle of reaction (Wong 2009). However, in the lack of sufficient substrate, two protons are accepted by one of the oxide bridges of native intermediate, and one water molecule is released leading to the formation of fully oxidized resting form enzyme. The next cycle will start when substrate is available with the abstraction of four successive electrons, protonation of oxide bridge, and release of water.

3.3 Reaction Mechanism of Auxiliary Activity Family LPMO Enzymes

LPMOs are single copper containing enzymes which boost lignocellulose depolymerization by oxidative cleavage of polysaccharide chains. Highly diverse LPMOs acting on different substrates exist in the microbial world. These were classified into four auxiliary activity protein families of carbohydrate active enzyme (CAZy) database: (1) AA9 family contains cellulose and hemicellulose active fungal LPMOs, (2) AA10 contains cellulose and chitin-active bacterial LPMOs, (3) AA11 contains chitin-active fungal LPMOs, (4) AA13 contains starch-active fungal LPMOs, and (5) AA14 is a newly discovered family which contains xylan degrading LPMOs with C-1 oxidation from Pycnoporus coccineus (Levasseur et al. 2013; Couturier et al. 2018). Further, they have been categorized into three groups based on catalyzing the oxidation of specific carbons in the backbone. For example, type 1 and 2 LPMOs oxidize C-1 and C-4, respectively, whereas type 3 LPMOs are mixed nature and can oxidize both C-1 and C-4 (Beeson et al. 2015). Further, low sequence identity in the LPMO family members prompts diverse catalytic mechanisms. Even for the specific enzyme molecule, there may be more than one route of catalysis (Hedegard and Ryde 2018). Therefore, it is difficult to propose a unified mechanism of catalytic action, and mechanistic details of LPMOs are not fully known. However, a considerable research progressed in the understanding of fungal LPMOs reaction mechanisms in recent years. Here, we summarize achievements in the understanding of cellulose-active fungal LPMOs with a few structural considerations.

Initially LPMOs were supposed glycosyl hydrolases and wrongly placed in GH61 family of CAZy database. The first greatest advancement came from X-ray crystallographic structure of cellulose-active TrCel61B which revealed the presence of the active site on flat surface with lack of carboxylate residues (necessary for hydrolase action). This finding provoked scientific community that it may utilize different mechanism for glycosidic bond cleavage (Karkehabadi et al. 2008); thereafter considering other research findings also, these enzymes were placed in the separate family of auxiliary activity proteins (Levasseur et al. 2013). The flat surface exposed active site contains copper which is essential for the catalytic activity (Quinlan et al. 2011; Phillips et al. 2011). Usually this copper is in the Cu(II) oxidation state in the resting enzymes and coordinated with three ligands provided by two histidines (histidine brace moiety) as well as tyrosine and two water molecules. The N-terminal histidine residue coordinates with side chain $N^{\delta 1}$ and backbone amino N atoms while another histidine coordinates with N² of the side chain only (Quinlan et al. 2011). Tyrosine and one water molecule are coordinated at the axial positions opposite to each other, whereas another water molecule is equatorial and trans to amino nitrogen of N-terminal histidine (Kjaergaard et al. 2014). The plane of the active site is made of loops L2, LC, and LS which exhibit plasticity and probably involved in the binding of substrate/reductant (Beeson et al. 2015). Specifically loop regions LC and LS have been predicted to form CDH binding groove (Li et al.

2012). The resting form of the enzyme can bind polysaccharide with the displacement of axial water; however, the catalytic cascade starts with the reduction of

ment of axial water; however, the catalytic cascade starts with the reduction of Cu(II) to Cu(I) via external reducing agent. The reduced and activated LPMO-Cu(I) follows either of the two routes; it may bind molecular oxygen or hydrogen peroxide (Bertini et al. 2018; Hedegard and Ryde 2018). Firstly consider the binding of oxygen, and the bound oxygen is protonated by His147 located in the vicinity of active site. The next step proceeds through one electron reduction and protonation resulting in the elimination of water molecule. Thereafter, the next cycle of reduction and proton transfer from His147 takes place into the formation of Cu(I)-oxyl and Cu(II)-hydroxyl species. These Cu-oxyl and hydroxyl species (another route) can also be generated by binding of H_2O_2 as co-substrate in which His147 proximates H_2O_2 to Cu(I) before cleavage of O-O bond. The Cu(II)-hydroxyl abstracts a hydrogen atom from C-H of glycosidic linkage and with His147 assistance forms Cu(I)-hydroxyl which is converted into product involving transition from triplet spin-state to a closed-shell singlet species (Hedegard and Ryde 2018).

3.4 Reaction Mechanism of Cellobiose Dehydrogenase (CDH)

The diverse roles of CDH have been postulated. It may either directly provide reducing power to LPMOs or reduce oxygen to generate hydroxyl-free radical and hydrogen peroxide (Beeson et al. 2015). The produced hydrogen peroxide may act as co-substrate for LPMOs or boost the action of other H_2O_2 requiring lignocellulose depolymerizing enzymes. On the other hand, hydroxyl-free radicals can diffuse in the internal matrix of recalcitrant lignocellulosic biomass and cause depolymerization (Kersten and Cullen 2007).

CDH is exactly flavocytochrome CDH which consists of a catalytically functional dehydrogenase domain connected to *b*-type cytochrome domain by linker peptide. Dehydrogenase domain is composed of FAD and substrate binding subdomains. The active center of dehydrogenase domain is structurally similar with the active site of cholesterol oxidase and contains conserved His and Asn residue in close proximity of flavin N5 atom (Hallberg et al. 2003). The overall reaction mechanism involves hydride transfer to flavin N5 and lactone substrate formation. It is accomplished with the help of His689 which will act as general base and abstract a proton from C1-OH; it follows a cascade of reactions involving nucleophilic attack at N5, rearrangement, and elimination. The reduced flavin moiety can either donate the reducing equivalence to LPMOs or ferric ions via heme prosthetic group of the CDH. Thus, it participates in the regeneration of ferrous ions which will produce hydroxyl radicals (OH) by Fenton chemistry reaction denoted below (Kersten and Cullen 2007).

$$H_2O_2 + Fe(II) + H + \rightarrow H_2O + Fe(III) + OH$$

4 Recent Advancements in the Production of Lignocellulose-Degrading Enzymes

The major obstacle in the commercialization of lignocellulose-derived biofuel and other products is the cost of enzymes. It necessitates 10–100 times reduction of enzyme costs (Binod et al. 2019). Therefore, researchers continuously take up the challenges to cut down the production costs of lignocellulose-degrading enzymes by searching novel producer organisms as well as manipulating organisms or enzymes employing modern approaches such as genetic engineering, metabolic engineering, and strain improvement. In this section, we will summarize some of these recent developments in the production of lignocellulose deconstructing enzymes.

4.1 Recent Developments in the Production of Cellulose-Degrading Enzymes

Cellulose being the major component of lignocellulosic waste biomass, much attention has been paid for the cost-effective production of cellulases by using different fungal organisms isolated from diverse habitats, cheap substrates, screening of inducers as well as employing genetic modification approaches. Here we discuss some examples of different strategies being used for the cost reduction of the cellulases production. Endophytic fungi are proven rich sources of secondary metabolites, but they are less explored for the production of industrial enzymes. In the recent years, trend has changed towards the utilization of endophytic fungi for the production of lignocellulose decomposing enzymes. In a recent screening of 14 endophytic fungal strains, endoglucanase producing potential of Cladosporium cladosporioides PAJ 03, Phomopsis stipata SC 04, Trichoderma viride PAJ 01, and Botryosphaeria sp. AM 01 was revealed on a mixture of sugarcane bagasse and wheat bran. Higher activities of endoglucanase and β -glucosidase were secreted by Botryosphaeria sp. AM01 and Saccharicola sp. EJC 04 on cottonseed meal and wheat bran mixture. A blend of the enzymatic extracts from these organisms efficiently saccharified pretreated sugarcane bagasse (Marques et al. 2018). Among the 41 endophytic fungal strains screened from mangrove tropical forests, A. niger and A. awamori exhibited higher production of endoglucanase and β-glucosidase with an evidence of twofold more hydrolysis of cellulosic materials than reference A. niger and T. reesei organisms (Maroldi et al. 2018). Screening of common indoor fungi demonstrated the potential of *Cladosporium sphaerospermum* and *Penicillium chrysogenum* to produce cellulase, β -glucosidase, mannase, β-galactanase, and arabinanase. Thus, these organisms can be used as over-producers to supplement industrial enzyme blends for boosting lignocellulose degradation (Andersen et al. 2016). Prajapati et al. (2018) isolated Aspergillus tubingensis from dung compost which emerged high producer of cellulase. The enzyme preparation effectively degraded sugarcane bagasse and rice straw releasing the polymer constituent units. Mohapatra et al. (2018) isolated high cellulase producing Aspergillus fumigatus from the soil containing decomposed lignocellulosic waste. A mixture of equal proportions of partially purified CMCase and FPase enzymes from crude extract of A. fumigatus efficiently hydrolyzed lignocellulose with a release of 396 mg/g reducing sugars from *Pennisetum* biomass. An isolate (T. asperellum) collected from the African jungle soil produced higher activities of cellulase, ß-glucosidase, and CMCase. It has also emerged excellent degrader of natural waste biomass; however, the degradation efficiency was higher for pretreated biomass (Wang et al. 2017). Penicillium subrubescens FBCC1632, a new potential producer of hemicellulase, has been isolated which is capable of degrading crude lignocellulose feedstock, as well as polysaccharides and metabolizing their monomeric components. The cellulolytic activities of P. subrubescens FBCC1632 on different plant biomass substrates were comparable to those of A. niger (Makela et al. 2016). Okereke et al. (2017) has purified and characterized acidophilic cellulase from *P. ostreatus* and suggested it would be a promising renewable source of cellulase for biomass degradation. Han et al. (2017) utilized repeated fed-batch strategy to increase cellulase production by *Penicillium oxali*cum under submerged fermentation condition. Among the different strategies, mutagenesis is also the easiest way of strain improvement. Hyper cellulase producing T. reesei D-7 mutant was obtained by treating T. reesei Rut-C30 strains with ethyl methyl sulfonate (EMS) and plasma irradiation. The obtained mutant strain responds well to corn starch hydrolysate soluble inducer and insoluble cellulose inducer (Zhang et al. 2017a). Textile industries discharge huge amount of waste water rich in cotton fibers and polyester. Utilization of this waste is an alternative of cheap cellulase production by A. niger (Hu et al. 2018). A large amount of domestic waste water is also generated continuously which could serve as a cheap source of medium and utilization of raw domestic waste water as base medium resulted in the 1.4-fold enhancement of filter paper cellulase activity in comparison with synthetic medium (Libardi et al. 2017). A bamboo shoot shell pretreated with Pleurotus ostreatus promoted cellulase synthesis in the cultures of A. niger. Interestingly, analysis of the pretreated shell revealed that the inducer is modified soluble lignin polymer and not a traditional disaccharide (Zhang et al. 2017b). The supplementation of duckweed 50 g L^{-1} in the shake flasks of *T. reesei* culture equally induces cellulase production as avicel; however, the hydrolysis rate of steam-exploded cornstalk by duckweed induced cellulase is 28% higher than avicel. The combined application of both inducers enhanced cellulase production and hydrolysis rates, 39% and 36%, respectively (Li et al. 2019). One of the major challenge in the decomposition of lignocellulosic biomass is the inactivation of enzymes by the phenolic components. The cellulolytic activities of enzyme extracts prepared from Chrysoporthe cubensis and Penicillium pinophilum have much more tolerance to phenolic inhibitors than the enzyme extracts of T. reesei and A. niger. T. reesei and A. niger cellulolytic activities were inactivated at 0.3 and 1.5 mg phenol mg⁻¹ protein, whereas 1:1 mixture of extracts from C. cubensis and P. pinophilum resisted inactivation at 35 mg phenols mg⁻¹ protein (Azar et al. 2017). Thus inclusion of enzyme preparations from these organisms will reduce the enzyme loadings for the hydrolysis of alkali pretreated lignocellulosic substrates. Genetic regulation tunes up the utilization of resources for the synthesis of each and every protein in the cell; cellulases are not exception to this concept. Replacement of transcription repressor gene *ace1* with the coding region of endoglucanase gene egl1 in T. reesei Rut-C30 has enhanced 90 and 132% total cellulolytic and endoglucanase activities, respectively, under flask cultures; however, cellulase activity enhancement remains 74% than wild-type organism in the 5 L fermenter (Meng et al. 2018). A widely used cellulase producer T. reesei has been engineered with an overexpression of *Trivb-1* gene encoding putative transcription factor which has increased 200% cellulase production by the recombinant strain. The crude extract from the recombinant strain released 40% more glucose than wild-type counterpart from the pretreated corn stover (Zhang et al. 2017c). P. oxa*licum* is a promising alternative of *T. reesei*, and its cellulolytic potential has further been enhanced by overexpression of nine genes encoding endoglucanase. All the overexpressing strains displayed higher β -glucosidase activities, and this activity reached 65-fold specifically with the use of inducible promoter of bgl2 gene. The overexpressing mutant has substantially reduced enzyme loads for the saccharification of delignified natural lignocellulose materials (Yao et al. 2016). A bi-functional enzyme produced by the fusion of endoglucanase gene from Teleogryllus emma and xylanase gene from Thermomyces lanuginosus has great potential for lignocellulose degradation. It displayed mild reaction pH (5.5), temperature (50 °C) and long half-life for cellulase and xylanase activities. The supplementation of this bi-functional enzyme with commercial preparation Ctec2 has increased 10-20% release of fermentable sugars from the pretreated rice straw (Chen et al. 2018). The cloning and expression of an endoglucanase gene from Aspergillus terreus in Aspergillus nidulans resulted in the high yield secretion of the enzyme. The purified enzyme showed optimum activity at pH 5.0 and temperature 55 °C, and the best thing is the non-specificity of the substrates. The enzyme preparation displayed high activity against β-glucan and xyloglucan while moderate activity against glucomannan and carboxymethyl cellulose (Segato et al. 2017). The expression and characterization of two endoglucanases from diverging subfamilies of GH45 demonstrated differences in their substrate specificities and thus they can be used as new tools in the enzyme cocktails for cellulose hydrolysis (Berto et al. 2019) (Table 6.1).

4.2 Recent Developments in the Production of Hemicellulolytic Enzymes

Hemicellulose is also a major constituent of plant biomass after cellulose, and efforts are underway to achieve the cost-effective production of enzymes for its degradation into fermentable sugars. Isolation and characterization of new organisms producing hemicellulose degradation enzymes is one of the strategy. A qualitative screening

S.		Producer organism/	Uniqueness of the produced enzyme/ developed	Type of modification/	
no. 1.	Enzyme name Endoglucanase	expression host Cladosporium cladosporioides PAJ 03, Phomopsis stipata SC 04, Trichoderma viride PAJ 01, Botryosphaeria sp. AM 01	system Higher titers of enzyme production on lignocellulosic substrates	new isolate Unexplored endophytic fungi	References Marques et al. (2018)
2.	β-Glucosidase	Saccharicola sp. EJC 04, Paecilomyces sp. SF 021, Ustilaginoidea sp. CV 04, Ustilaginoidea sp. XYA 04	High activity of the enzymes produced in solid state fermentation	Unexplored endophytic fungi assessed	Marques et al. (2018)
3.	Endoglucanase and β-glucosidase	Aspergillus niger and Aspergillus awamori	Higher saccharification performance than reference organisms	Endophytic isolates from mangrove plants	Maroldi et al. (2018)
4.	Cellulase and β-glucosidase	Cladosporium sphaerospermum and Penicillium chrysogenum	_	Screening of novel producer organisms	Andersen et al. (2016)
5.	Cellulase and β-glucosidase	Aspergillus tubingensis NKBP-55	High titers of enzyme production in SSF	Screening of new isolate from dung compost sample	Prajapati et al. (2018)
6.	Cellulase	Aspergillus fumigatus	High producer organism isolated from decomposed lignocellulosic waste soil	New isolate	Mohapatra et al. (2018)
7.	Cellulase and β-glucosidase	T. asperellum T-1	High yield of enzyme activities and excellent waste biomass degrading abilities	The producer organism was isolated from African jungle soil	Wang et al. (2017)
8.	Endoglucanase	Penicillium subrubescens FBCC1632	-	A new potential producer of (hemi)- cellulases	Makela et al. (2016)
9.	Cellulase	Pleurotus ostreatus	Acidophilic cellulase	_	Okereke et al. (2017)

 Table 6.1
 Some examples of recent developments in the production of cellulases

(continued)

			Uniqueness of the produced		
			enzyme/	Type of	
S.		Producer organism/	developed	modification/	
no.	Enzyme name	expression host	system	new isolate	References
10.	Cellulase	Penicillium oxalicum RE-10	Repeated fed-batch fermentation strategy	-	Han et al. (2017)
11.	Cellulase	T. reesei D-7	High cellulose producing strain	Ethyl methyl sulfonate and plasma irradiation mutagenesis of <i>T. reesei</i> Rut-C30	Zhang et al. (2017a)
12.	Cellulase	Aspergillus niger	Utilization of textile waste for enzyme production	-	Hu et al. (2018)
13.	Cellulase	Trichoderma harzianum HBA03	Raw domestic waste water used as base culture medium	-	Libardi et al. (2017)
14.	Cellulase	Aspergillus niger	A new type of soluble inducer of cellulase activity discovered which is a modified lignin polymer	Bamboo shoot shell pretreated with <i>Pleurotus</i> ostreatus	Zhang et al. (2017b)
15.	Cellulase	Trichoderma reesei Rut C30	The cellulase production and hydrolysis of substrates increased	Duckweed combined with avicel as an inducer	Li et al. (2019)
16.	Cellulase	Chrysoporthe cubensis and Penicillium pinophilum	Enzyme mixture is tolerant to high levels of monocomponent phenolic inhibitors	The enzyme extracts prepared from <i>Chrysoporthe</i> <i>cubensis</i> and <i>Penicillium</i> <i>pinophilum</i> were mixed in equal proportions	Azar et al. (2017)

(continued)

Table 6.1 (continued)

			Uniqueness of the produced enzyme/	Type of	
S.	Enzyme name	Producer organism/	developed	modification/	References
17.	Endoglucanase	T. reesei QS305	Endoglucanase and total cellulase activity enhanced 90% and 132%, respectively	Transcription repressor gene <i>ace1</i> of <i>T. reesei</i> Rut-C30 was replaced with <i>egl1</i> gene encoding endoglucanase	Meng et al. (2018)
18.	Cellulase	T. reesei Vib-1	Cellulase production enhanced 200% by the recombinant strain	The gene <i>Trvib-1</i> encoding putative transcription factor was overexpressed in <i>T. reesei</i> Rut-C30	Zhang et al. (2017c)
19.	β-Glucosidase	Penicillium oxalicum	Enhanced saccharification	Overexpression of β-glucosidase encoding genes	Yao et al. (2016)
20.	Bi-functional fusion enzyme EG-M-Xyn	Yarrowia lipolytica	High potential in improving the enzymatic hydrolysis of lignocellulosic materials	Endoglucanase gene from <i>Teleogryllus</i> <i>emma</i> fused with xylanase gene from <i>Thermomyces</i> <i>lanuginosus</i>	Chen et al. (2018)
21.	Endoglucanase	Aspergillus terreus/Aspergillus nidulans	High yield secretion of enzyme	Non-specific endoglucanase from Aspergillus terreus	Segato et al. (2017)
22.	Endoglucanase	Gloeophyllum trabeum/Aspergillus nidulans	Stable at 70 °C for 24 h and have useful properties to be used in enzyme cocktail preparation	Heterologous expression	Berto et al. (2019)
23.	Endoglucanase	Myceliophthora thermophila/Aspergillus nidulans	Stable at 70 °C for 24 h and have useful properties to be used in enzyme cocktail preparation	Heterologous expression	Berto et al. (2019)

Table 6.1 (continued)

showed the potential of edible fungus Tuber maculatum as a new source of xylanase production which may be comparable to xylanases from other sources (Bedade et al. 2017). Among the 14 endophytic fungal strains, P. stipata SC 04 and Botryosphaeria sp. AM 01 stood best producers of xylanase and β -xylosidase, respectively. These organisms produced 69433 and 487 U/g xylanase and β-xylosidase enzymes on 1:1 ratio of mixed sugarcane bagasse and wheat bran substrate (Marques et al. 2018). During the screening of 15 potential xylanase producers, A. niger being the highest producer secreted 456 U/g in SSF on sugarcane bagasse and palm kernel cake supplemented with salt solution. Authors claimed 1.5-fold higher production of xylanase in comparison to the reported literature (Lee et al. 2018). Mangrove endophytes A. niger and A. awamori have also produced higher activities of xylanase in addition to cellulase and released more glucose from the cellulosic materials than the reference organisms (Maroldi et al. 2018). Makela et al. (2016) isolated and screened endoxylanase producing P. subrubescens FBCC1632 as potential producer of hemicellulases. Similarly, T. asperellum T-1 isolated from African jungle soil produced high yield of xylanase which possesses excellent waste biomass degrading abilities (Wang et al. 2017). A. tubingensis NKBP-55 isolated from dung compost produced endoxvlanase, endo-mannanase, and α -galactosidase. The enzyme extract from the organism efficiently degraded sugarcane bagasse and rice straw releasing xylose, glucose, and cellobiose (Prajapati et al. 2018). A new isolate of Penicillium chrysogenum F-15 has also produced high activities of xylanase which showed higher potential of agroindustrial waste biomass degradation than commercial xylan, thus indicating its usefulness in food, feed, pulp, and paper industries (Terrone et al. 2018). A comparison of the biochemical properties of Penicillium roqueforti ATCC xylanase with commercial xylanase (Sigma) produced by Thermomyces lanuginosus revealed superiority of the enzyme from former organism (Souza et al. 2018). Kumar and Shukla (2018) optimized xylanase production from Thermomyces lanuginosus at the pilot scale, and characterization of the purified enzyme demonstrated high thermostability in the presence of polyols. A novel endoxylanase produced A. tamarii Kita has useful characteristics such as maximum activity at 60 °C and pH 5.5 (activity over broad pH 4.0-9.0) for its application in the food, feed, pharmaceutical, and paper industries (Heinen et al. 2018). Likewise, Ye et al. (2017) produced and characterized novel xylose and alkali-tolerant β -xylosidase from *P. oxalicum* with a suggestion that enzyme extract from this organism may promote hemicellulose degradation. Thermophilic fungus isolate *Thermoascus aurantiacus* has produced novel type of acidophilic and thermotolerant xylanase. The purified xylanase exhibited maximum activity at 75 °C and pH 5.0; it was also stable over a broad pH and temperature range (Ping et al. 2018). Purification and characterization of novel endoxylanase from nontoxic A. flavus demonstrated unique enzymatic properties, such as excellent stability at 50 °C, pH 3.5–10.5, and tolerance to various proteases, which may be suitable for biotechnological potential (Chen et al. 2019). Hemicellulose contains different types of side chain groups which may hinder complete degradation. Andersen et al. (2016) isolated common indoor fungi Cladosporium sphaerospermum and Penicillium *chrysogenum* which produces β-galactanase, mannanase, and arabinase. The supplementation of these enzymes may boost the bioconversion of lignocellulosic biomass.
The release of phenolic inhibitors from lignocellulose degradation makes bioconversion process cumbersome, and it necessitates enzymes which are tolerant to these inhibitors. Xylanases produced by Chrysoporthe cubensis and Penicillium pinophilum were tolerant to phenolic inhibitors in the presence of laccase; however, their activity declined twofold in the absence of laccase (Azar et al. 2017). A novel extremophilic xylanase produced by Aureobasidium pullulans NRRL Y-2311-1 has improved dough rheology and bread quality. Remarkably, bread specific volume increased 30% more than the commercial counterpart being used (Yegin et al. 2018). It can also be a good source of xylanase supplementation in the bioconversion of lignocellulose. Genetic modification of the existing organisms is also foreseen as feasible approach for the higher and efficient production of lignocellulose-degrading enzymes. Genetic engineering of hyper cellulase producing T. reesei; replacement of cis-acting promoter region of xyn3 with cis-acting region of xyn1 exhibited improved inductivity against cellulosic carbon over wild-type promoter and acquired inductivity against xylanosic carbon. Further expression of β-glycosidase gene of A. aculeatus under the control of xyn3 chimeric promoter enhanced saccharification ability through increased cellobiase activity. Thus, xyn3 chimeric promoter is very efficient in the expression of carbohydrolase genes and may reduce the production cost of the industrial enzymes (Hirasawa et al. 2018). Heterologous expression and characterization of endoxylanase and β-xylosidase from *P. ostreatus* and *Irpex lacteus*, respectively, revealed novel enzymes with synergistic hydrolysis properties. Both the recombinant enzymes showed synergistic hydrolysis on oat spelts xylan and alkali pretreated cornstalk reaching a value of 2.26-fold degree of synergy (Zhuo et al. 2018). Recently, a novel GH30 family endoxylanase has been expressed in heterologous host, and characterization revealed that it is active on arabinoxylan and glucuronoxylan. Its unique characteristic is the release of xylobiose from xylooligosaccharide while other GH30 enzymes mainly release xylose (Espinoza and Eyzaguirre 2018). The production of xylanase and arabinofuranosidase from genetically engineered A. nidulans has been optimized on soybean fibers as substrate, and subsequently the enzyme extract applied for the hydrolysis of soybean fiber to obtain xylooligosaccharides (Pereira et al. 2018). The xylooligosaccharide yield reached 28% mass fraction of xylan (Table 6.2).

4.3 Recent Developments in the Production of Lignin-Degrading Enzymes

Lignin is the most recalcitrant portion of lignocellulosic biomass, and efficient decomposition of lignin increases the accessibility of enzymes for hydrolysis of other polysaccharides. Therefore, production of lignin-degrading and inhibitory phenolic compound modifying enzymes at low cost will further decrease the bioconversion cost of plant biomass by reduced loading of other carbohydrases. Lignin is chiefly degraded by LiP, MnP, VP, and laccases; hereunder we discuss briefly

S.		Producer organism/	Uniqueness of the produced enzyme/	Type of modification/	
no.	Enzyme name	expression host	developed system	new isolate	References
1.	Xylanase	Tuber maculatum	_	New producer organism	Bedade et al. (2017)
2.	Endoxylanase	P. stipata SC 04	High production of enzyme in solid state fermentation	Unexplored endophytic fungi used	Marques et al. (2018)
3.	β-Xylosidase	Botryosphaeria sp. AM 01	High production of enzyme in solid state fermentation	Unexplored endophytic fungi used	Marques et al. (2018)
4.	Xylanase	Aspergillus niger	High yield of the enzyme in SSF	A new isolate	Lee et al. (2018)
5.	Xylanase	Aspergillus niger and Aspergillus awamori	Higher saccharification performance than reference organisms	Endophytic isolates from mangrove plants	Maroldi et al. (2018)
6.	Endoxylanase	Penicillium subrubescens FBCC1632	_	A new potential producer of (hemi)- cellulases	Makela et al. (2016)
7.	Xylanase	T. asperellum T-1	High yield of enzyme activities and excellent waste biomass degrading abilities	Isolate from African jungle soil	Wang et al. (2017)
8.	Endoxylanase and α -galactosidase	Aspergillus tubingensis NKBP-55	High titers of enzyme production in SSF	Screening of new isolate from dung compost sample	Prajapati et al. (2018)
9.	Xylanase	Penicillium chrysogenum F-15	High activity of the enzyme on agroindustry biomass	New soil isolate	Terrone et al. (2018)
10.	Xylanase	Penicillium roqueforti ATCC 10,110	Produced enzyme is superior than <i>Thermomyces</i> <i>lanuginosus</i> commercial preparation available from Sigma	_	Souza et al. (2018)

 Table 6.2
 Some examples of recent developments in the production of hemicellulases

(continued)

G		Producer	Uniqueness of the	Type of	
S.	Enzyme name	organism/	produced enzyme/	modification/	References
11.	Xylanase	Thermomyces lanuginosus	Enhanced thermostability with supplementation of polyols	A soil isolate	Kumar and Shukla (2018)
12.	Endoxylanase	<i>Aspergillus</i> <i>tamarii</i> Kita	A novel endoxylanase with useful physicochemical properties for application in food, feed, pharmaceutical, and paper industries	_	Heinen et al. (2018)
13.	β-Xylosidase	P. oxalicum	A novel xylose and alkali-tolerant enzyme with high catalytic activity	-	Ye et al. (2017)
14.	Xylanase	Thermoascus aurantiacus M-2	Novel acidophilic and thermostable enzyme	A new isolate	Ping et al. (2018)
15.	Endoxylanase	Aspergillus flavus	A new xylanase with excellent stability	_	Chen et al. (2019)
16.	β-Galactanase, mannanase, and arabinase	Cladosporium sphaerospermum and Penicillium chrysogenum	-	Screening of novel producer organisms	Andersen et al. (2016)
17.	Xylanase	Chrysoporthe cubensis and Penicillium pinophilum	Tolerance to phenolic inhibitors	Presence of laccase enhanced two fold tolerance	Azar et al. (2017)
18.	Xylanase	Aureobasidium pullulans NRRL Y-2311-1	A novel extremophilic xylanase	_	Yegin et al. (2018)
19.	Xylanase	Trichoderma reesei	The produced enzyme cocktail showed high saccharification efficiency	Chimeric promoter designed	Hirasawa et al. (2018)
20.	Endoxylanase	Pleurotus ostreatus	Novel enzyme	Heterologous expression	Zhuo et al. (2018)
21.	β-Xylosidase	Irpex lacteus	Novel enzyme	Heterologous expression	Zhuo et al. (2018)

Table 6.2 (continued)

(continued)

S. no.	Enzyme name	Producer organism/ expression host <i>Penicillium</i>	Uniqueness of the produced enzyme/ developed system	Type of modification/ new isolate	References
	Tryfallase	purpurogenum	family GH30	expression	and Eyzaguirre (2018)
23.	Endoxylanase	A. nidulans	Increased xylolytic activity	Genetic modification of <i>A. nidulans</i> with <i>xynC</i> gene from <i>P.</i> <i>funiculosum</i>	Pereira et al. (2018)
24.	Arabinofuranosidase	A. nidulans	Increased xylolytic activity	Genetic modification of <i>A. nidulans</i> with <i>abfB</i> from <i>A. niger</i>	Pereira et al. (2018)

Table 6.2 (continued)

recent advancements in the production and characterization of these enzymes. A new selectively lignin degrading white rot fungus Echinodontium taxodii produced novel type MnP on natural lignocellulosic material of moso bamboo. The enzyme showed optimum activity at pH 3.5 and 55 °C. The produced MnP oxidizes both phenolic and non-phenolic lignin units, and degradation efficiency increased when combined with laccase from the same producer organism (Kong et al. 2016). Makela et al. (2013) produced lignin-modifying enzymes in the semi-solid cultures with milled grey alder, Norway spruce, and charcoal. Production of LiP and MnP significantly enhanced with the supplementation of wood (milled grey alder) under low nitrogen; however, optimum laccase production has only been obtained in the presence of 1.5 mM Cu²⁺ with different nitrogen sources. Production of lignin-modifying enzymes and removal of phenolic content assessment for seven white rot fungi revealed the high potential of Cerrena unicolor in submerged fermentation of olive mill by-products. C. unicolor produced laccase and MnP in the fermentation of olive pomace while only laccase was produced on olive tree sawdust. Interestingly, both laccase and MnP were produced from 50% diluted olive mill wastewater, and their production was further enhanced with the addition of Cu2+ and Mn2+. The produced laccase has reduced 60-70% total phenols of olive pomace phenolics (Elisashvili et al. 2018). The assessment of different additives to enhance the production of ligninolytic enzymes by Coriolus versicolor on sweet sorghum bagasse demonstrated stimulatory effect of metallic salts and phenolic compounds. Specifically syringic acid increased LiP and laccase, gallic acid increased MnP, and Cu2+ increased laccase production, and the combinations of these supplements synergistically enhanced laccase, MnP, and LiP production. These enzymes had increased 2.43-fold release of fermentable sugars from pretreated sweet sorghum bagasse (Mishra et al. 2017). The optimum production of ligninolytic enzymes by Schizophyllum commune on rice straw was obtained at 35 °C, pH 5.0, C:N ratio 20:1, and in the presence of mediator MnSO₄. The crude extract has removed lignin 47-72% from different agro-industrial residues which enhanced 47% cellulose hydrolysis of untreated biomass by the crude cellulase extract from T. harzianum (Asgher et al. 2016). White rot fungus Inonotus obliquus degraded 72%, 39%, and 47% lignin from wheat straw, rice straw, and corn stover, respectively. MnP and LiP production was very high on wheat straw, and their respective production reached maximum after 2 and 4 days (Xu et al. 2017). A dye decolorizing fungus Bjerkandera adusta produced novel type LiP and MnP which have higher catalytic efficiencies (Bouacem et al. 2018). Consideration of ligninolytic enzyme production from cheap substrate utilized potato peel as sole nutrient source. Trametes hirsuta grown on potato peel produced high yield of laccase which has the ability of 87% anthraquinone dve decolorization (Schalchli et al. 2017). In another study, P. ostreatus produced much higher yield of laccase and MnP on dry potato peel pretreated with distilled water (Ergun and Urek 2017). Although fungi are heterotrophic, different wavelength light radiation can induce cascade of reactions which may influence the production of specific enzyme. White rot fungus P. chrysosporium exposed to various wavelength radiation produced LiP maximally under green light in the submerged fermentation (Ramírez et al. 2010). In addition to the conventional methods, genetic engineering is also an accurate and effective approach of attaining targeted yield. Therefore, a hyper lignin degrading Phanerochaete sordida has been genetically engineered for the expression of MnP under the promoter control of a highly expressed protein of the same organism (Sugiura et al. 2012). Ceriporiopsis subvermispora produces all known types of MnP, and their utilization is hampered due to lack of efficient preparation methods. The soluble expression of C. subvermispora MnP in E. coli may offset the problem. The efforts have resulted in the high yield of MnP under the control of cold shock promoter (Lin et al. 2018). As the name denotes, VPs are much more efficient than any other peroxidases, and their production seems fruitful for the efficient lignin degradation. VP from *Physisporinus vit*reus has been overexpressed in E. coli in inclusion bodies and refolded in vitro which has displayed efficient catalysis (Liu et al. 2019). Free hydroxyl-phenolic compounds are also known inhibitors of lignin peroxidases and hinder the efficient ligninolysis. However, these inhibitory phenolic compounds may be modified into non-inhibitory compounds by enzymes. One of the enzymes, O-methyltransferase, discovered and characterized from P. chrysosporium, methylated free hydroxylphenolic compounds at para and meta positions, rendering them non-toxic to the lignin-degrading enzymes (Mai and Kim 2016) (Table 6.3).

4.4 Recent Developments in the Production and Characterization of LPMOs

LPMOs are a recently discovered class of enzymes which plays an important role in the enhancement of lignocellulolytic processes. In spite of their less proportional contribution in the cleavage of glycosidic linkage, LPMO activity is significant due

		References	Kong et al. (2016)	Makela et al. (2013)	Elisashvili et al. (2018)	Mishra et al. (2017)	Asgher et al. (2016)	Xu et al. (2017)	Bouacem et al. (2018)	Schalchli et al. (2017)	Ergun and Urek (2017)	Ramírez et al. (2010)	Sugiura et al. (2012)	Lin et al. (2018)	Liu et al. (2019)	Mai and Kim (2016)
	Type of modification/new	isolate	New selective lignin degrading white rot fungus			Addition of metallic salts and phenolic compounds	I		Ι	I		1	Promoter of a highly expressed protein used	Expression of MnP genes under control of a cold shock promoter	Expressed in <i>E. coli</i> as inclusion body and then refolded in vitro	Heterologous expression
	Uniqueness of the produced enzyme/developed	system	A novel enzyme oxidizes phenolic and non-phenolic lignin units	Production of the enzymes optimized with the supplementation of copper, nitrogen, and wood as nutrients	Fermentation of olive mill by-products for enzyme production	Synergistic action of the produced enzymes enhanced 2.43-fold lignin degradation	Delignification of agro-industrial residues	High activities of LiP and MnP were observed in wheat straw culture	Novel peroxidases from dye decolorizing fungus	Potato peels used as sole nutrient source	Potato peel waste used as substrate support	Illumination with green light significantly enhanced enzyme production	High yield of MnP activities	Soluble expression of active enzyme in E . <i>coli</i>	Efficient enzyme for biocatalysis and biodegradation	Methylates para and meta positions of lignin peroxidase inhibitory free hydroxyl-phenolic
	Producer organism/	expression host	Echinodontium taxodii 2538	Phlebia radiata	Cerrena unicolor	Coriolus versicolor	Schizophyllum commune	Inonotus obliquus	Bjerkandera adusta	Trametes hirsuta	Pleurotus ostreatus	Phanerochaete chrysosporium	Phanerochaete sordida	Ceriporiopsis subvermispora/E. coli	Physisporinus vitreus PF18 E. coli	Phanerochaete chrysosporium/E. coli
•	Name of lignin-	degrading enzyme	MnP	LiP, MnP, and laccase	LiP and laccase	LiP, MnP, and laccase	MnP, LiP, and laccase	MnP and LiP	LiP and MnP	Laccase	MnP and laccase	LiP	MnP	MnP	VP	O-methyltransferase
		S. no.	1.			5	3.									

Table 6.3 Some examples of recent developments in the production of lignin deconstruction enzymes

to their action on the crystalline regions of the polysaccharides. Presently attention is being paid to these enzymes, and still their production and characterization need to be elucidated in detail. A very few studies have been undertaken for the enhanced production of LPMOs than other enzymes involved in the lignocellulose deconstruction. Most of the studies have either just evaluated the boost of lignocellulose hydrolysis with LPMO augmentation or overexpressed the enzyme in heterologous host for structural determination and understanding of catalytic mechanisms. However, one of the necessary components for feasible application in industry is the cheap availability of the enzyme. Attempts need to be made for optimization of their production on different substrates by various isolates because LPMOs are differentially regulated by multiple substrates. AA9 family LPMOs of A. nidulans are differentially regulated by xylan, xyloglucan, glucan, pectin, and cellulose (Jagadeeswaran et al. 2016). A. nidulans LPMO overexpressed in the same organism showed activity against hemicellulose xyloglucan. Thermophilic fungus Myceliophthora thermophila secretes over a dozen of different LPMOs. One of the enzymes from this organism expressed in A. nidulans and named MtLPMO9J is a single domain and shares 63% sequence similarity with the catalytic domain of NcLPMO9C. The enzyme is stable over pH 5.0 and temperature up to 50 °C and active against cellulose, soluble cello-oligosaccharides, and xyloglucan (Kadowaki et al. 2017). Two LPMOs of mangrove-associated fungus *Pestalotiopsis* sp. were expressed in yeast, and their heterologous production has reached up to 260 mg/L in the bioreactor. They had showed enhanced stability and activity in the presence of sea salt (Patel et al. 2016). White rot fungus *Heterobasidion irregulare* contains ten genes putatively encoding AA9 LPMOs. Two of these (PiLPMO9H and PiLPMO9I) were expressed in P. pastoris and characterized. PiLPMO9H breaks glycosidic linkage of cellulose by C1 oxidation while PiLPMO9I cleaves linkage with strict oxidation at C4 carbon of glucose unit and is also active against glucomannan (Liu et al. 2017). One of the limitations of heterologous production is the N-terminal processing and generation of catalytically active enzyme. Tanghe et al. (2015) tried several secretion signals and found that inclusion of native secretion signal is best. It has enabled correct processing of the N-terminal and high yield of the protein in the heterologous host *P. pastoris*. Recently Bulakhov et al. (2017) expressed β -glucosidase of *A*. niger and LPMO of T. reesei under the control of inducible glucoamylase promoter gla1 in Penicillium verruculosum. The heterologous LPMO expression preparations boosted 10-43% lignocellulose hydrolysis (Table 6.4).

5 Concluding Remarks

The cost reductions in the production of bio-based valuable products are seen on the availability of low priced lignocellulolytic processes, which can only be sustained with the higher yields of efficient enzymes per capita following different strategies such as isolation and screening of high producer organisms, modification of producer organisms by classical and site-directed mutagenesis targeting the regulatory

S. no.	Producer organism/ expression host	Uniqueness of the produced enzyme/ developed system	Type of modification/new isolate	References
1.	Myceliophthora thermophila/ Aspergillus nidulans	Single domain LPMO with 63% similarity to the catalytic domain of <i>Nc</i> LPMO9C	Heterologous overexpression	Kadowaki et al. (2017)
2.	Pestalotiopsis sp. Nci6/Pichia pastoris	Salt tolerant	Heterologous expression	Patel et al. (2016)
3.	Heterobasidion irregulare/P. pastoris	-	Heterologous expression	Liu et al. (2017)
4.	T. reesei/ P. pastoris	Use of native secretion signal	Heterologous expression	Tanghe et al. (2015)
5.	Penicillium verruculosum	The enzyme preparation proved to be more effective in the hydrolysis of lignocellulosic substrates	<i>T. reesei</i> LMPO gene expressed under the control of inducible glucoamylase promoter (gla1)	Bulakhov et al. (2017)

 Table 6.4
 Some examples of recent developments in the production and characterization of LPMOs

components, screening of novel and efficient inducers, and metabolic and genetic engineering. In addition to traditional glycoside hydrolases and lignin-degrading enzymes, inculcation of inhibitory compound modifying enzymes, auxiliary activity oxidative enzymes, and crystallinity reducing accessory proteins in the enzyme cocktails can be helpful to make the process economical. Although recently the reaction mechanism of auxiliary activity family 9 LPMOs has been discovered, they follow more than one route of catalysis. It would be fruitful to decipher the efficient route of catalysis, and thereafter they can be engineered for unified mode of efficient catalysis. Simultaneously, much attention needs to be paid for the optimization of enhanced yields of the enzymes from the cheapest substrates. In this regard, production of LPMOs and accessory proteins can be prioritized.

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Chapter 7 Strategic Role of Fungal Laccases in Biodegradation of Lignin



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1 Introduction

Lignocellulosic biomass is considered as a sustainable and renewable energy resource on earth. It is replenished at the rate of 60,000 million tons every year as organic carbon on earth (Singh 2006; Prasad 2018). Lignocellulosic biomass consists of carbohydrate polymers like cellulose, hemicellulose, and lignin (Singh 2006). The carbohydrate constituents of lignocellulosic biomass carry dissimilar monomers (six and five carbon sugars) which are firmly linked to the lignin. Lignin is highly inflexible polymer which is not easily degraded in nature because of its complex structure characterized by three-dimensional cross-linking (Ruiz-Dueñas and Martínez 2009).

The removal of lignin from biomass makes the hydrolysis of hemicellulose and cellulose easy thereby facilitating the use of its sugar proportion in generation of sustainable fuels like ethanol and other value-added products (Siqueira et al. 2012).

In pulp and paper industries, the removal of lignin from wood is required for the production of paper products. These industries remove lignin chemically from the wood pulp in delignification and bleaching stages and generate huge amount (50–60

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R. Naraian (ed.), *Mycodegradation of Lignocelluloses*, Fungal Biology, https://doi.org/10.1007/978-3-030-23834-6_7

million tons) of lignin compounds in processed water (Zhu et al. 2016; Avanthi and Banerjee 2016; Gosz et al. 2018). Environmental problems caused by chemicals used in pulping and bleaching stages of the paper production have compelled paper industries to consider environmental friendly alternatives for removal of lignin from wood pulp.

It is therefore necessary to develop effective green pretreatment technologies for the removal of lignin from lignocellulosic biomass and waste water originating from pulp and paper industries (Avanthi and Banerjee 2016).

Enzymatic degradation of lignin is considered as green technology for the degradation of lignin compounds. In nature, biodegradation of lignin occurs by ligninolytic enzymes secreted by microbes. Microbes degrade lignin in two stages with the help of ligninolytic enzymes. In first step, de-polymerization of lignin takes place leading to formation of heterogeneous aromatic compounds. In second step, heterogeneous aromatic compounds are mineralized in water and carbon dioxide.

White rot fungi belonging to the class basideomycetous are major lignin degraders in nature. Fungi such as *Phanerochaete chrysosporium*, *Pleurotus ostreatus*, *Phlebia radiata*, *Trametes versicolor*, *Peniophora* sp., *Agaricus bisporus*, *Pycnoporus cinnabarinus*, *Trametes hirusuta*, and *Ceriporiopsis subvermispora* are some well-known lignin-degrading fungal species (Singh 2006; Shankar and Shikha 2012; Afreen et al. 2018). White rot fungi degrade lignin with the help of lignino-lytic enzymes which include laccases, manganese peroxidase, lignin peroxidase, etc. These enzymes depolymerize lignin via formation of lignin phenoxy radicals (Brijwani et al. 2010).

Laccases are regarded as one of the most active enzymes in lignin degradation (Shankar and Shikha 2012; Avanthi and Banerjee 2016; Bilal et al. 2017; Agarwal et al. 2018). The enzyme was first identified in sap of Japanese lacquer tree *Rhus vercinifera* (Thurston 1994). Laccases are multinuclear enzymes with molecular weights ranging from 60 to 80 KDa (Kiiskinen et al. 2004; Khlifi-Slama et al. 2012; Afreen et al. 2018; Hilgers et al. 2018). Laccases effectively degrade phenolic moieties in lignin C α oxidation, C α - C β cleavage and aryl cleavage (Archibald et al. 1997; Shankar and Shikha 2015).

The oxidation of different types of substrates like polyphenols, lignins, aromatic amines, methoxy-substituted monophenols by laccase results in formation of oxygen-centered free radical, which is ultimately transformed into quinone. Quinone and free radical may be polymerized in later stages of the reaction (Thurston 1994).

In the presence of suitable mediators such as ABTS [2, 2'-azinobis (3-ethylbenz thiazoline-6-sylphonate)] or HBT (hydroxybenzotriazole), laccases hold the potential to oxidize non-phenolic proportion of lignin and other substrates (Shankar and Shikha 2015).

Owing to the aforesaid properties, laccases show potential for biological degradation of lignin. In the light of the aforesaid context, this chapter highlights production, characterization and role of laccases and laccase mediator system in degradation of lignin.

2 Lignin

Swiss botanist A. P. de Candolle mentioned lignin in 1813 for the first time as tasteless, fibrous material which was insoluble in water but soluble in weak alkaline solutions. He termed this substance as "lignine," which is derived from the Latin word *lignum*, meaning wood. Lignin in the most abundant organic polymer on earth after cellulose representing 30% non-fossil organic carbon and 20–35% dry biomass of wood on earth.

Lignin occurs in nature as most abundant aromatic polymer followed by cellulose. The elemental composition of lignin differs from plant species to species. In aspen sample, there is 5.9% hydrogen, 63.4% carbon, 0.7% ash (mineral components), and 30% oxygen (by difference) corresponding to the formula approximately $(C_{31}H_{34}O_{11})_n$. Depending on source of biomass, dry weight of a plant cell wall is constituted by 20–35% of lignin (Galbe and Zacchi 2007). Hardwood generally contains 20–25% lignin while softwoods contain 25–35% lignin (Faravelli et al. 2010).

Cellulose–hemicellulose matrix in plant cell is surrounded by lignin which imparts rigidity to the plant cell wall and connects plant cells together firmly. Lignin protects wood constituents from microbial decomposition. Because of hydrophobicity, lignin also acts as a barrier against the penetration of water (Balakshin et al. 2009).

Because of aforesaid reasons, lignin behaves as most recalcitrant biopolymer on earth (Ruiz-Dueñas and Martínez 2009). Different types of functional groups are present in lignin. Phenolic, methoxyl, hydroxyl carbonyl, and alcoholic hydroxyl are some prominent groups present in lignin. Lignin consists of phenylpropanoid building blocks/monolignols linked together by carbon–oxygen and carbon–carbon bonds.

Phenylpropanoid units present in lignin are coniferyl, sinapyl alcohols, and *p*-coumaryl. These units are precursors of guaiacyl (G), syringyl (S), and *p*-hydrophenyl (H) type units in biopolymer lignin (Zhu et al. 2016). In plants, laccases and peroxidases generates radicals which biosynthesize lignin via oxidative coupling reactions leading to synthesis of a growing polymer.

Aryl-glycerol, β -aryl ether structure with the β -O-4 linkages is the most abundant linkage in lignin constituting most (approximately 60%) of the total internal linkage in lignin (Mainka et al. 2015; Murciano Martínez et al. 2016; Agarwal et al. 2018).

3 Fungal Degradation of Lignin

For the past two decades, researchers have investigated role of ligninolytic enzyme system of fungi, bacteria, and actinomycetes in biological degradation of lignin.

Under natural environmental conditions, biological degradation of lignin results in generation of monomeric aromatic compounds which are employed as a source of carbon and energy by microorganisms. Oxidizing potential of enzymes of ligninolytic system regulates degradation of lignin in the presence of oxygen and hydrogen perox-

ide. However, laccases do not require hydrogen peroxide for its activity. This property of laccases makes them appropriate candidate for environmental bioremediation of toxic pollutants (Knezevic et al. 2013; Avanthi and Banerjee 2016; Munk et al. 2018).

Fungal degradation of lignin is catalyzed by highly efficient and unique oxidative extracellular ligninolytic enzymes like laccases, lignin peroxidases, manganese peroxidases, versatile peroxidases, glyoxal peroxidases, aryl alcohol oxidases, quinone reductases, and xylanases (dos Santos et al. 2019).

White rot fungi have drawn significant attention of the researchers in last decade as they are the dominant sources of lignin-degrading enzyme laccase (Saito et al. 2018; Rudakiya and Gupte 2017).

Under adverse environmental conditions, white rot fungi with the help of ligninolytic enzymes completely mineralize lignin into water and carbon dioxide. Brown rot fungi are not considered as efficient degraders of lignin (Wu et al. 1996; Cardoso et al. 2018). Lignin-degrading white rot fungi occupies diverse ecological niches and grows on different habitats, for instance, conifers, deciduous trees, forest litter, crops, grassland soils, and roots of plants. Prominent lignin-degrading white rot fungi include *Coriolus versicolor*, *Pycnoporus cinnabarinus*, *P. ostreatus*, *Peniophora* sp. *Ganoderma*, *Phlebia radiate*, *Lentinula edodes*, *Phanerochaete chrysosporium*, and *Armillaria* spp. (Hatakka 2005; Singh 2006; Sigoillot et al. 2012; Rytioja et al. 2014; Wang et al. 2018).

4 Ligninolytic Enzymes of Fungi

4.1 Laccases

Laccase (benzenediol oxygen oxidoreductase, EC 1.10.3.2) is a copper-containing phenol oxidase. Laccases catalyze the oxidation of a wide range of aliphatic and aromatic compounds with a consequent reduction of molecular oxygen to water with the help of oxygen-centered free radical. Laccases are widely distributed in higher plants, fungi, bacteria, and insects. It has been demonstrated that almost all fungi produce laccase except some physiological groups such as lower fungi, i.e., Chytridiomycetes and Zygomycetes (Morozova et al. 2007). Although brown rot fungi are not known to produce laccase, the oxidation of syringaldazine and ABTS have been detected in *Coniophora puteana* and *Laetiporus sulphureus* (Schlosser and Hofer 2002), respectively. Almost all species of wood rotting basidiomycetes causing white rot are the most widely known species that produce appreciable quantity of laccase to varying degree (Hatakka 1994; Baldrian 2006; Sharma et al. 2007; Kolomytseva et al. 2017; Young et al. 2018).

Laccases from white rot fungi are responsible for degradation of lignin, fructification, detoxification, sporulation, phyto-pathogenicity, etc. Fungal laccases generally have high redox potential as compared to bacterial laccases. Because of this reason, fungal laccases are widely preferred over bacterial laccases for degradation of lignin model compounds. For biotechnological and environmental applications, much preference is given to fungal laccases as compared to bacterial laccases since fungal laccases have high redox potential than bacterial laccases (Surwase et al. 2016).

4.2 Production of Fungal Laccases

For widespread industrial and biotechnological applications, laccases are required in large amounts. Heterologous expression of laccase gene, optimization of physical and chemical conditions (Kiiskinen et al. 2004), or a combination of both strategies is used to cater the need of the laccases for these applications. White rot fungi produce laccases during secondary metabolism under solid as well as submerged culture conditions (Robinson et al. 2001; Rivera-Hoyos et al. 2013). In solid state fermentation, the fungi grow on solid material in the absence or near absence of free flowing water, while in submerged culture conditions, the fungi grow on a continuous liquid phase (Musatti et al. 2017). Laccase production is greatly influenced by different cultivation and nutritional parameters such as concentration of nitrogen source, carbon source, and micronutrients as well.

4.3 Laccase Production Under Submerged Culture Conditions

Production of laccases under submerged culture conditions involves cultivation of fungi under aerobic conditions in aqueous medium containing nutrients required for growth (Shiv Shankar and Shikha 2011). In order to achieve high production, the earlier research attempts have been oriented towards optimization of culture conditions (Bertrand et al. 2013; Elisashvili et al. 2018). The production of laccase at industrial scale is undertaken by submerged culture conditions (Dong et al. 2005; Couto and Toca-Herrera 2007; Piscitelli et al. 2010).

Production of laccase under submerged culture conditions is disadvantageous as abundant growth of fungal mycelium in the culture medium affects enzyme yield due to metabolic rate limitation and mass transfer. In addition, extensively grown mycelia cover impellers, leading to obstruction in the system. This problem can be solved by immobilization of the cells on different solid supports (Couto 2018). The efficiency of laccase production requires a well-defined media composition optimized in terms of physico-chemical and nutritional parameters (carbon and nitrogen sources, inducers, pH and temperature, etc.).

5 Effect of Physico-chemical Factors on Laccase Production

The physico-chemical parameters affecting production of laccases under SmF are temperature, pH of the fermentation medium, inducers, and aeration level during the cultivation (Afreen et al. 2018). Majority of researchers have reported that the

optimum pH for laccase production by fungi is 4.5–4.6. The optimum temperature for laccase production by fungi has been reported to be 25–30 °C (Pointing et al. 2000).

6 Influence of Carbon and Nitrogen Sources on Laccase Production

Easily available carbon sources, such as cellobiose, mannitol, and glucose, generally favor higher laccase production as compared to the carbohydrates like lactose and cellulose (Mikiashvili et al. 2006; Afreen et al. 2018).

Rodriguez Couto et al. (2006) reported that the incorporation of different sources of carbon, such as fructose followed by glycerol, resulted in increase of the rate of laccase production *T. hirsuta* as compared to glucose alone. Bettin et al. (2009) reported that the fructose as a source of carbon optimum production of laccase by *Pleurotus sajor-caju* (Bettin et al. 2009).

Low concentration of glucose in culture medium has been found to increase the rate of laccase production (Thiruchelvam and Ramsay 2007; Kannaiyan et al. 2012). The optimum concentration of nitrogen for fungal laccase production by fungi is more ambiguous than that of carbon (Kannaiyan et al. 2012; Afreen et al. 2018). Generally, nitrogen starved conditions trigger the production of ligninolytic enzymes like laccases. Nitrogen in the concentration of 2–3 mM in culture media significantly increases the production of the laccase.

7 Induction of Laccase

Laccase production under submerged culture conditions can be improved by the incorporation of some substances called inducers to the media. Use of an appropriate inducer has been shown to be a prerequisite to achieve effective production of the enzyme at an industrial scale. Compounds frequently used as inducer of laccase production by white rot fungi include 2, 5-xylidine (a xenobiotic compound), veratryl alcohol (a secondary metabolite), lignin, CuSO₄, etc. (Xavier et al. 2001; Shankar and Shikha 2015). Inducers, like ethanol, has been found to trigger laccase production indirectly.

8 Solid State Fermentation

Laccase production under solid state fermentation is carried out particularly using filamentous fungi. Selection of appropriate feed stock for SSF is essential since the selected substrate determines the process of laccase production by laccase (Reddy

et al. 2003). Factors like cost, chemical composition of substrates, porosity and size of the substrate should be considered while going for laccase production under solid state fermentation. Substrates, which are used for laccase production under solid state fermentation, have been categorized into two classes: Non-inert support which not only provide site for attachment to fungi but also render nutrients to the fungi (minerals, proteins, fibers, and sugars). Inert supports are synthetic compounds without any nutritional value for fungi. The wide varieties of solid materials used in SSF are classified into two main classes: inert. Fluidized bed bioreactors, rotating horizontal drum bioreactors, rotating disc reactors and column tray bioreactors are frequently used for the production of laccases under solid state fermentation (Couto 2018).

9 Structure of Laccases

Fungal laccases exist in different forms viz. multimeric, heteromeric, homotetrameric, and monomeric. Depending upon microorganism, the molecular weight of laccases ranges from 50 to 130 kDa (Jaiswal et al. 2015). Fungal laccases generally contain 10–30% carbohydrate content of molecular weight while plant laccases contain 45% carbohydrate content (Baldrian 2006).

The stability of the protein component is ensured by the carbohydrate proportion of the laccases. In addition, carbohydrate moiety in laccases prevents inactivation of laccases (Morozova et al. 2007).

Greek key β barrel topology is the main structural feature of the laccases which is generally comprised of 500 amino acids arranged in three domains consecutively. In laccases, amino acid moieties are divided into three domains:

- Domain 1 (150 amino acids)
- Domain 2 (150–300 amino acids)
- Domain 3 (300–500 amino acids)

Structure of laccases is consolidated by disulfide linkage between area I and III and area I and II (Plácido and Capareda 2015). Single laccase unit comprised of four dissimilar copper catalytic forms (Fig. 7.1). These four copper ions are grouped into three classes:

9.1 Type 1: Paramagnetic "Blue" Copper Site

Multi-copper proteins in laccases appear blue due to type 1 copper and covalent copper–cysteine bond. In laccases, type 1 site has a rhombohedral co-ordination, with two histidines and one cysteine. Axial position ligand affects the oxidative nature of laccase, responsible for oxidation of different substrates (Kumar et al. 2003; Enguita 2011).



Fig. 7.1 Structure of laccase elaborating different active sites

9.2 Type 2: Paramagnetic "Non-blue" Copper

Non-blue copper site display normal electron paramagnetic resonance (EPR) spectra. Type 2 site dose not absorb intense radiation in visible region and unified by two histidine residues (Hilgers et al. 2018).

9.3 Type 3: Diamagnetic Spin-coupled Copper-Copper Pair

Type 3 site is a double nuclear center controlled by six histidines. This site is characterized spectroscopically as it absorbs electronic radiation at 330 nm. Type 3 site does not display electron paramagnetic resonance signal (Matera et al. 2008).

A try nuclear cluster is constituted by type 3 copper and type 2 copper. A try nuclear cluster reduction of oxygen takes place.

10 Categories of Laccases

10.1 White Laccases

White laccases show absorption peak at 400 nm, but they do not display peak at 330 nm and 605 nm. Abnormal metal content is the main property of white laccases. White laccases are optimally active at neutral pH (Palmieri et al. 1997). White laccase does not display T1 and T2 signals and electron paramagnetic resonance spectra. This happens because valency state of Cu^{2+} is changed (Zhao et al. 2012).

10.2 Yellow Laccase

These are artificially reduced blue laccases which do not absorb visible light and electron paramagnetic resonance spectra (Pozdnyakova et al. 2006a, b). Yellow laccases are formed due to reduction of type I copper site by aromatic products of lignin degradation (Mot and Silaghi-Dumitrescu 2012).

Several aspects of yellow laccases like their catalytic properties and characterization is still to be addressed by the researchers. Few yellow laccase producers include *Trametes* sp. F1635, *Stropharia aeruginosa*, *Leucoagaricus naucinus* LAC-04, *Aureobasidium pullulans* NAC8, *Sclerotinia sclerotiorum*, *Pleurotus ostreatus* D1 (YLPO), and *Sclerotinia sclerotiorum* (Pozdnyakova et al. 2004, 2006a, b; Mot and Silaghi-Dumitrescu 2012; Daroch et al. 2014; Ademakinwa and Agboola 2016; Ning et al. 2016; Wang et al. 2018).

11 Role of Laccases in Lignin Degradation and Their Mode of Action

11.1 Degradation of Phenolic Units in Lignin

Laccase-mediated degradation of phenolic units in lignin results in aryl-alkyl cleavage, $C\alpha$ - $C\beta$ cleavage, and C- α oxidation (Fig. 7.2). Laccases result in reduction of diatomic molecular oxygen to two water molecules. In this process, laccases facilitate one electron oxidation of different types of organic and inorganic substrates in general and aromatic amines, monophenols, and polyphenols in particular (Camarero et al. 1994; Bourbonnais et al. 1995; Moya et al. 2011). The oxidation of these substrates by laccases results in formation of an oxygen-centred free radical which is transformed in quinone ultimately. The quinone and the free radicals can then undergo polymerization (Archibald et al. 1997; Shankar and Shikha 2012).



Fig. 7.2 Oxidation of phenolic subunits of lignin by laccase (Adopted from Archibald et al. 1997)

11.2 Degradation of Non-phenolic Subunits of Lignin in the Presence of Redox Mediators

The substrate range of laccase can be extended to non-phenolic subunits of lignin (Fig. 7.3) by the inclusion of a mediator such as 2, 2'-azinobis-(3-ethylbenzthiazoline-6-sulfonate) (ABTS), 1-hydroxybenzotriazole (HBT), and *N*-hydroxyphthalimide (NHPI) (Bourbonnais et al. 1997).

11.3 Biodegradation of Lignin with Laccase Mediator Systems

By using small molecular weight compounds called redox mediators, the oxidation potential of fungal laccases can be increased from 0.78 to 1.084 V (Bourbonnais et al. 1997; Zille et al. 2003). Redox mediators are low molecular weight compounds which are uninterruptedly oxidized by the laccase enzyme and eventually reduced by the substrate (Wu et al. 1996; Cañas and Camarero 2010).

High molecular weight substrates are difficult to enter into laccase active site. Redox mediators transfer electrons between substrate and laccase enzyme thereby surmounting steric interferences existing between enzyme and substrate (Li et al.



Fig. 7.3 Oxidation of 4-methoxy benzyl alcohol (non-phenolic lignin compound) to 4-methoxy benzaldehyde: (a) electron transfer (ET); (b) hydrogen atom transfer (HAT) (Barreca et al. 2004)

1999). Degradation of high molecular weight non-phenolic compounds is limited by laccases due to reduced enzyme activity and low accessibility.

The problem of limited substrate accessibility can be solved by employing appropriate redox mediators. Redox mediators with high redox potential are oxidized to stable intermediates in initial step by laccases. Thereafter, the diffusion of oxidized mediator from enzyme takes place. Mediators easily enter into plant cell wall because of their small size and act on target substrate lignin in general and nonphenolic content of the lignin in particular. An ideal redox mediator should be effective, atoxic, and cheap with consistent reduced and oxidized forms and should be capable of upholding the cyclic redox conversions (Morozova et al. 2007). The problem of instability and formation of toxic intermediates is the main problem in application of synthetic mediators in degradation of lignin model compounds. Instability of redox mediators sometimes results in unfinished redox cycles leading to inadequate oxidation of the substrates. It is difficult to re-utilize synthetic mediators as they may generate toxic intermediates. In addition they are toxic and expensive as well (Yaohua et al. 2019). At higher concentration of enzyme (more than 1 mM), synthetic redox mediators hamper the activity of laccases. Natural redox mediators can be the better option of synthetic mediators as they are cost-incentive and do not result in the formation of toxic intermediates. Few good examples of natural redox mediators of laccase include: vanillin, acetosyringone, syringaldehyde, p-coumaric acid, acetovanillone, and HBT (Andreu and Vidal 2011; Bibi et al. 2011).

Degradation of non-phenolic units using laccase mediator system in lignin takes place via three different types of mechanisms (a) Transfer of electrons, (b) transfer of radical hydrogen, (c) ionic mechanism. First two mechanisms have been shown in Fig. 7.3.

Bourbonnais and Paice (1990) reported for the first time that laccase oxidizes lignin model compounds such as 1-(3,4-dimethoxyphenyl)-2-(2-methoxyphenoxy)-propane 1,3-diol, veratryl alcohol into derivatives of α -carbonyl.

Bourbonnais et al. (1997) also reported that laccase in the presence of redox mediator ABTS results in demethylation and delignification of kraft pulp. Call and Mücke (1997) reported that the redox mediator 1-Hydroxybenzotriazole (HBT) improves degradation of lignin and bleaching of the kraft pulp (Call and Mücke 1997). Shankar and Shikha (2012) reported that the laccase from the *Peniophora* sp. (NFCCI-2131) results in delignification of kraft pulp effectively. Munk et al. (2018) reported that the laccase from Myceliophthora thermophile and Trametes versicolor in presence of four individual mediators, N-hydroxyphthalimide (HPI), 1-hydroxybenzotriazole 2,2'-azino-bis(3-ethylbenzothiazoline-6-sulfonic acid) (ABTS). (HBT). and 2,2,6,6-tetramethylpiperidin-1-oxyl (TEMPO) enhances degradation of lignin. The use of laccase mediator system has drawn significant attention of the researchers for diverse applications such as treatment of industrial waste water, decolorization of industrial and leather dyes, bioremediation water and soil contaminated with hydrocarbons and bio-deinking of waste paper (Shankar and Shikha 2012).

Redox mediator ABTS results in the oxidation of non-phenolic subunits of lignin model compounds via transfer of electrons while redox mediator 2,2',6,6'-tetramethylpiperidine-1-oxyl (TEMPO) mediates the process of oxidation via ionic mechanism (Bourbonnais et al. 1997; Fabbrini et al. 2002). A coupling intermediate compound is formed by the redox mediator HBT which is subsequently degraded and release benzotriazole, a reduced form of HBT. Benzotriazole facilitates the connection of lignin to HBT (Potthast et al. 1997). The degradation of softwood lignin via HBT-laccase system has been found to result in reduction in G-units and aromatic component of the lignin while increase in *p*-hydroxyphenyl content.

Laccase mediator system results in the formation of degradation products like 2,6-dimethoxy-4-((E)prop-1-enyl)benzaldehyde, 4-ethyl-2,6-dimethoxybenzaldehyde, and 2,6-dimethoxy-4-methylbenzaldehyde during degradation of lignin (dos Santos et al. 2019; Couto 2018).

Laccase mediator system is widely used bio-bleaching step in pulp and paper industries. The system results in reduction in kappa number in the pulp and enhances its ISO brightness (Poppius-Levlin et al. 1999). Use of laccase mediator system during bio-bleaching of the pulp minimizes the use of chlorine compounds (elemental chlorine, sodium hypochlorite, chlorine dioxide, etc.) thereby reducing environmental risks in the process. The degree of degradation of lignin in wood pulp is determined by the type of mediator and pulp as well as dose of the laccase. Use of combination of more than one redox mediator has been found to increase the extent of the degradation of the lignin in soft wood and hard wood.

12 Conclusion

Lignin is an important constituent of lignocellulosic material. Lignin is highly resistant towards degradation because of its complex chemical structure and diverse chemical bonding. The removal of lignin from lignocellulosic material is required for the production of fermentable sugars and other value-added products. In pulp and paper industries, the removal of lignin from wood is required for the production of paper products. Environmental problems caused by chemicals used in pulping and bleaching stages of the paper production have compelled paper industries to consider environmental friendly alternatives for removal of lignin from wood pulp. In addition, the degradation of lignin in waste water originating from pulp and paper industry is required by using green approaches as lignin and its derivative compounds cause adverse biochemical effects on aquatic flora and fauna.

Fungal laccases (benzenediol oxygen oxidoreductase, EC 1.10.3.2) in combination with natural redox mediator compounds degrade lignin in ecofriendly manner and offer sustainable way of pretreatment/disposal of lignocellulosic waste. Fungal laccases to be employed for degradation of lignin should be substrate unspecific, thermostable with high redox potential, cost-incentive, resistant towards toxic metal ions, halides, inhibitors, and free radicals.

Degradation of lignin using fungal laccases is an ecofriendly practice but availability of laccase for commercial applications is limited due to low production of enzyme by reported strains. In this respect, the exploration of novel laccase hypersecretary strains may be helpful. In addition, genetic improvement/mutagenesis in previously isolated strains and heterologous expression of laccase genes in appropriate host may be helpful in increasing the level of laccase production for different applications.

In this context, this chapter was an attempt to put forth the details of chemical structure of lignin, linkages consolidating the structure of lignin, production and characterization of fungal laccases, their mode of action, and role of laccase mediator system in degradation of phenolic and non-phenolic subunits of lignin.

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Chapter 8 Rapid Bioconversion of Lignocellulosic Biomass by Fungi



Adesh Kumar, Divya Srivastva, and Ramesh Chand

1 Introduction

A large amount of lignocellulosic residue (150 billion tons) is produced annually worldwide and the dominant biomass differs depending on the region (Asim et al. 2015). Lignocellulose is derived from unusable portions of plant biomass in the form of agricultural, industrial, domestic, and forest residues (Gupta et al. 2016). India accounts for 770 m tonnes of crop residue production annually. If these residues were not degraded, there would be an accumulation of carbon on Earth which would ultimately break carbon cycle and make life impossible on the biosphere. The recycling of the wastes is both ecological necessity and economic compulsion. The crop residue management in agriculture for sustaining high levels of crop productivity is the biggest challenge for researches. Composting is basically a process for decomposition of organic solid wastes. The decomposition process is accomplished by various microorganisms including fungi, bacteria, and actinomycetes. Primarily, nature has taught us that the breakdown of plant materials is mainly carried out by fungi, by means of potential cellulosic enzymes (Gupta et al. 2016; Solomon et al. 2016; Takasaki et al. 2013). Fungal cellulosic enzymes present efficient and gentle way of converting recalcitrant lignocellulosic materials into precursors for industrial purposes (Bischof and Seiboth 2016; Gupta et al. 2016; Solomon et al. 2016). Currently, several cellulolytic enzymes have been identified and are well characterized, but there may be many more cellulases, hydrolytic and non-hydrolytic enzymes available with a greater range of applications. Some previously uncharacterized

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fungi may also possess unique enzymes or individual cocktails that could supersede commercial cellulase preparations, resulting in faster and more complete biomass hydrolysis at a reduced cost (Berrin et al. 2012; Gupta et al. 2016). Under optimal conditions, bioconversion of lignocelluloses converts at faster rate. Several fungi possess cellulolytic enzymes which can be utilized for bioconversion of agricultural residues. The main problem with bioconversion of agriculture residues is the presence of lignin. As compared to cellulose, lignin is recalcitrant and cellulose fibrils are always embedded in an amorphous matrix of lignin and hemicellulose. These three kinds of polymers bind strongly to each other by non-covalent forces as well as by covalent cross-links making a composite material, lignocelluloses, representing over 90% of dry weight of plant. For effective utilization of lignocelluloses residue, several physical and chemical pre-treatments are required to disrupt the bounds and hydrolysis of lignin. The delignification steps also increase the cost of process as well as generate toxic compounds due to hydrolysis (Batt 1991). Lignocellulolytic enzymes producing fungi are widespread and include species from the ascomycetes, e.g., Trichoderma reesei, basidiomycetes including white-rot fungi, e.g., Phanerochaete. chrysosporium, brown-rot fungi, e.g., Fomitopsis palustris, chytridiomycetes including Caecomyces communis, Neocallimastix frontalis, Cyllamyces aberensis, and mushroom fungi, e.g., Pleurotus ostreatus which can be used for solid state fermentation (SSF) of agricultural residues after delignification and finally a few anaerobic species, e.g., Orpinomyces sp. which degrade cellulose in gastrointestinal tracts of ruminant animals (Kamra and Zadrazil 1986; Ozkose et al. 2001; Yoon et al. 2007; Doi 2008; Ljungdahl 2008; Griffith et al. 2009). Since none of the fungal strains including the best mutants are able to produce high levels of the enzymes at the same time necessary for bioconversion of lignocelluloses, there is a need to identify consortium of efficient cellulolytic and lignolytic microorganisms or fungal co-culturing which might act synergistically for rapid bioconversion of agricultural residues without any chemical pre-treatments. The mesophiles are not very successful for large-scale bioconversions through solid state fermentation process because temperature inside the residue during active fermentation reaches up to 50-70 °C. During this phase, thermophilic fungi play an important role in degradation of lignocellulose (Sharma and Johri 1992; Ahamed and Vermette 2008).

2 Composition of Lignocellulosic Biomass

Lignocellulose is the term used for biomass from woody or fibrous plant materials, being a combination of lignin, cellulose, and hemicellulose polymers interlinked in a heterogeneous but resilient matrix (Chen 2014; Guo et al. 2015). The composition of lignocellulosic biomass is highly complex and dynamic leading to its recalcitrance (Gandarias and Luis 2013; Himmel et al. 2007; Johansen 2016). A typical representation is given in Fig. 8.1. On a dry weight basis, cellulose, hemicellulose, and lignin are the primary constituents of cell walls, whereas proteins (which are highly glycosylated), lipids, soluble sugars, and minerals are minor components

(Chundawat et al. 2011; Guo et al. 2015). While cellulose and hemicellulose are polysaccharides that can be hydrolyzed to fermentable sugars, lignin has a threedimensional phenolic structure that is highly resistant to microbial attack or to chemical hydrolysis (Gupta et al. 2016; Sims et al. 2008). These three compound fibrils and depolymerize cellulose and hemicellulose into simple sugars (Guo et al. 2015).

Cellulose is a complex macromolecule composed of linear β 1-4glucan chains that tightly aggregate into microfibrils held together via strong intra- and intermolecular hydrogen bonds and van der Waals forces resulting from pyranose ring stacking (Kumar et al. 2009). It is an unbranched chain of glucose units linked head to tail (Fig. 8.2), and it is highly resistant to chemical and biological hydrolysis because of its structural arrangement (Gandarias and Luis 2013). The chain conformation of the glucose residues in cellulose forces the hydroxyl groups into radial (equatorial) orientation and the aliphatic hydrogen atoms into axial positions. As a result, there is strong inter-chain hydrogen bonding between adjacent chains in a cellulose sheet and weaker hydrophobic interactions between cellulose sheets. The hydrophobic face of cellulose sheets makes crystalline cellulose resistant to acid hydrolysis because it contributes to the formation of a dense layer of water near the hydrated cellulose surface while the strong inter-chain hydrogen-bonding network makes crystalline cellulose resistant to enzymatic hydrolysis (Beckham et al. 2011; Ding and Liu 2012).



Fig. 8.1 Composition of lignocellulosic materials and their potential hydrolysis products and further degradation compounds. Source: (Guo et al. 2015)



Fig. 8.2 Chemical structure of cellulose (Gandarias and Luis 2013)

Cellulose mainly contains carbon (44.44%), hydrogen (6.17%), and oxygen (49.39%). It is the most abundant material on Earth, being the main constituent of plants serving to maintain their structure and is also present in bacteria, fungi, algae, and even in animals (Chaplin 2016). The content of cellulose in these species except plant is negligible. The variability of cellulose is such that it forms linear crystals of about 3 nm thick in some plants, but which reach widths of more than 20 nm in certain algae. The degree of polymerization of cellulose varies, depending on its source, between 100 and 10,000 (Chundawat et al. 2011). Cellulose with high crystallinity tends to be of shorter polymeric units and is insoluble (Chaplin 2016). Pure cellulose exists in various polymorphic forms viz. cellulose I α , I β , II, III1, III11, IV1, and IV11 (Zugenmaier 2001). The crystalline forms are Iα, Iβ, II, and III (Chaplin 2016). Cellulose I is the form found in nature while other crystal forms of artificial cellulose under artificial processing. Cellulose II, the second most extensively studied form, may be obtained from cellulose I by either of two processes: (a) regeneration, which is the solubilization of cellulose I in a solvent followed by reprecipitation by dilution in water to give cellulose II, or (b) mercerization, which is the process of swelling native cellulose in concentrated sodium hydroxide, to yield cellulose II on removal of the swelling agent. Celluloses III1 and III11 are formed, in a reversible process, from celluloses I and II, respectively, by treatment with liquid ammonia or some amines, and the subsequent evaporation of excess ammonia. Polymorphs IV1 and IV11 may be prepared by heating celluloses III1 and III11, respectively, to 206 °C, in glycerol (Chaplin 2016). These differences in glucan chain packing of these polymorphs have been shown to influence their hydrolysis rates (Chundawat et al. 2011). The crystalline structures of cellulose are aligned with each other, giving a structural bias to the cell wall (Cosgrove 2005).

Hemicellulose is another main component in plant fiber materials. It is the second most common polysaccharides in nature (Johansen 2016). Hemicelluloses are highly heterogeneous. These polysaccharides are very different from each other structurally and in physico-chemical properties (Chen 2014; Scheller and Ulvskov 2010). The necessity of the sugars β -(1 \rightarrow 4)-linked backbones of an equatorial configuration was proposed in (Scheller and Ulvskov 2010). In this regard, polysaccharides such as β -(1 \rightarrow 4)-galactan having axial configuration at C4 are not to be included with hemicelluloses.

On the other hand, some other researchers believe hemicelluloses may contain pentoses (β -D-xylose, α -L-arabinose), hexoses (β -D-mannose, β -D-glucose, α -Dgalactose), and/or uronic acids (α -D-glucuronic, α -D-4-O-methylgalacturonic, and α -D-galacturonic acids). Other sugars such as α -L-rhamnose and α -L-fucose may also be present in small amounts and the hydroxyl groups of sugars can be partially substituted with acetyl groups (Girio et al. 2010). These confusions attest to the heterogeneity of hemicelluloses. However, it is generally believed that hemicellulose is the glucan in the matrix of the cell, and the main components are xylan, xyloglucan, glucomannan, mannan, galactomannan, callose among others. The content and structure of hemicellulose in various plants are different, but on a broad classification, hardwood hemicelluloses contain mostly xylans that are partially acetylized, whereas softwood hemicelluloses contain mostly glucomannans with
almost no acetyl groups (Scheller and Ulvskov 2010). The main chain may consist of one or more types of glycosyls, and the connections between glycosyl are also different. The frequency and composition of branches are dependent on the source of xylan (Saha 2003; Scheller and Ulvskov 2010). In actual fact, raw materials from different producing areas and different parts have different glycan compositions (Chen 2014). Hemicelluloses in the cell wall have the primary role of interacting with other polymers to ensure the proper physical properties of the wall. Other roles ascribed to hemicelluloses include functioning as seed storage carbohydrates galactomannans, source of signal molecules—xyloglucans (Cosgrove 2005; Scheller and Ulvskov 2010). The schematic illustrations of the types of hemicelluloses found in plant cell walls indicating its complexity are shown in Fig. 8.3.

Lignin is a complex large molecular structure containing cross-linked polymers of phenolic monomers. It is present in the primary cell wall, imparting structural support, impermeability, and resistance against microbial attack. Thus making it one of the most significant contributors to biomass (Bugg and Rahmanpour 2015; Yuan et al. 2013). Lignin is the nature's dominant aromatic polymer, is found in most terrestrial plants in the approximate range of 15-40% dry weight to provide structural integrity (Wyman and Ragauskas 2015). Lignin is equally heterogeneous and their heterogeneity is manifested in different species of plants, length of growing season, and different parts of the plants. Even in the different morphologies of cells of the same xylem or different cell wall layers, the structures of lignin are not the same (Chen 2014). Generally, lignin is considered as being formed by the dehydrogenative polymerization of three hydroxycinnamyl alcohols: *p*-coumaryl, coniferyl, and sinapyl alcohols (Fig. 8.4). Alkyl-aryl, alkyl-alkyl, and aryl-aryl ether bonds link these phenolic monomers together (De Lima et al. 2016). The monolignols (monomeric units of lignin)-p-coumaryl alcohol, sinapyl alcohol, and coniferyl alcohol—give rise to the aromatic units p-hydroxyphenyl (H), syringyl (S), and guaiacyl (G), respectively, in the lignin polymer. In addition, it is now widely accepted that other monomers also participate in coupling reactions that give rise to the lignin macromolecule, such as coniferaldehyde, coniferyl alcohol, and acylated monolignols (De Lima et al. 2016; Yuan et al. 2013).

In general, herbaceous plants such as grasses have the lowest contents of lignin, whereas softwoods have the highest lignin contents (Pan et al. 2005). Hardwood lignins are predominantly G and S monolignols with trace amounts of H units. Softwood lignins are composed of mostly G units, whereas monocots incorporate equivalent amounts of G and S units along with significantly higher amounts of H monolignols (Chen 2014; Yuan et al. 2013). The composition of the functional groups of lignin and physical distribution, such as syringyl (S) unit to guaiacyl (G) unit ratio have also been demonstrated to affect the enzymatic hydrolysis of cellulose. In general, the yield of enzymatic hydrolysis is believed to be enhanced by low lignin content and a high S/G ratio of the substrate. Lignin appears to limit cellulose hydrolysis by two distinct mechanisms: first by forming a physical barrier that prevents enzyme access and second by non-productive binding cellulolytic enzymes (Dixon 2013; Yuan et al. 2013)



Galactoglucomannan, typical of conifer wood.

Fig. 8.3 Schematic illustrations of the types of hemicelluloses found in plant cell walls (Scheller and Ulvskov 2010)

3 Lignocellulose-Degrading Fungi

In nature, lignocellulose accounts for the major part of biomass and consequently its degradation is essential for operation of global carbon cycle. During decomposition, microorganisms transform organic matter into CO_2 biomass thermo-energy and humus-like end-product. The organic substrate, bulking agents, and the amend-



Fig. 8.4 Basic structural unit of lignin including three primary lignin monomers, *p*-coumaryl alcohol, coniferyl alcohol, and sinapyl alcohol and lignin structural *p*-hydroxyphenyl, guaiacyl, and syringyl units. Source: (Yuan et al. 2013)

ments used in composting are mostly from plant material. The capacity of microorganisms to assimilate organic matter depends on their ability to produce the enzymes needed for degradation of the substrate components, e.g., cellulose, hemicelluloses, and lignin. The more complex the substrate, the more extensive and comprehensive is the enzyme system required. Through the synergistic action of microorganisms, complex organic compounds are degraded to smaller molecules which can be utilized by microbial cells (Golueke 1992). Fungal strategies for lignocellulose depolymerization are substantially very complex, due to the high complexity of the raw materials. Recalcitrance to saccharification is a major limitation for the enzymatic conversion of lignocellulose-containing raw materials to get valuable end-products. The combination of hemicellulose and lignin forms a protective barrier around the cellulose, which must be modified before the hydrolysis of cellulose. But also crystalline structure of cellulose makes it insoluble and resistant for enzyme decomposition. Nevertheless, the removal of lignin is a key challenge to increase enzyme access to hemicellulose and cellulose (Jaramillo et al. 2015). Plant cell wall-degrading filamentous fungi have an important role in recycling nutrients in forest ecosystem. They are known to produce a broad variety of extracellular enzymes with diverse catalytic activities for the hydrolysis of renewable lignocelluloses containing raw materials. Predominantly responsible for lignocellulose degradation are wood-degrading fungi. They decompose and assimilate the most recalcitrant organic polymers, which is mainly attributed to their highly adaptive lifestyles, reflected by a large phylogenetic and phenotypic diversity (Kracher and Ludwig 2016). If they want to overcome the physical and chemical stability of lignocellulose, the fungi have to employ large sets of enzymes, which they release into the environment during their growth. Fungi have two types of degradation systems: intracellular, together with the outer cell envelope layer, and extracellular, important

for polysaccharide degradation. Furthermore, the extracellular enzymatic system includes two types of enzymes: hydrolytic, responsible for polysaccharide degradation; and oxidative, which degrade lignin and open phenyl rings. Three groups of fungi, with different effects and degradation mechanisms onto the lignocellulose, have been described: soft-rot, brown-rot, and white-rot fungi (Sanches 2009). Softrot fungi are mostly ascomycete fungi that can degrade polysaccharides in the surface layers of plants. Degradation leads to darkening and softening of the wood by the produced laccases and peroxidases involved in lignin modifications. These enzymes are unspecific and more limited in function than those isolated from whiterot and brown-rot fungi. The soft-rot fungi belong to genera Aspergillus and Neurospora. Still, little is known about the degradation mechanisms of lignocellulose by soft-rot fungi (Woiciechowski et al. 2013). Brown-rot fungi are basidiomycetes that rapidly metabolize cellulose and hemicellulose while only slightly modifying lignin. They have no lignin-degrading enzymes except small molecule reactive species to depolymerize lignin. At an advanced stage of degradation, wood residue exhibits cube shape and has a brownish color due to the predominant presence of oxidized lignin. Disruption of the lignocellulose matrix by brown-rot fungi can be demonstrated using iron-dependent Fenton chemistry known as chelatormediated Fenton system (CMF). The CMF system is a unique substrate deconstruction system based on oxygen radical chemistry that allows nonenzymatic deconstruction of the cellulose. Briefly, brown-rot fungal hyphae during the growth in the lumen area of plant cells produce oxalic acid, iron-reducing compounds (RC), and hydrogen peroxide (H_2O_2). The oxalic acid binds to a Fe³⁺ ion formatting the complex that diffuses into cell wall along with H₂O₂ and RC. With the pH change, RC sequesters Fe^{3+} from the Fe-oxalate complex and reduces it to Fe^{2+} . Fe^{2+} then reacts with H₂O₂ (Fenton reaction) and produces hydroxyl radicals (-OH). Upon attack of -OH radicals, lignocellulose matrix is disrupted. Models for the study of brown-rot fungi are Gloeophyllum trabeum, Coniophora puteana, and Postia placenta (Arantes and Goodell 2014). White-rot fungi are able to decompose all lignocellulose constituents: lignin, cellulose, and hemicellulose. Degradation of lignin is more efficient than in the case of brown-rot and soft-rot fungi because they possess a unique ability to its complete mineralization to CO_2 (Couturier and Berrin 2013). Therefore, white-rot basidiomycetes could be an interesting source of lignocellulose-active enzymes to supplement the commercial cocktails of hemicellulases and cellulases originated from ascomycetes such as Aspergillus niger or Trichoderma reesei (Sanches 2009). Some of the white-rot fungi capable of causing selective delignification of wood are Phanerochaete chrysosporium, Phanerochaete carnosa, Pleurotus ostreatus, Pycnoporus cinnabarinus, Botrytis cinerea, Stropharia coronilla, and Trametes versicolor. Several thermophilic fungi were isolated by researches which are able to degrade different agricultural residues. An alkalothemophillic Thermomonospora sp. producing high levels of xylanase was isolated from self-heating compost (George et al. 2001). Malbranchea pulchella and Humicola grisea were collected from wheat and broad bean straw composts 45 °C. Several thermophilic fungal genera represented by Penicillium duponti,

Myriococcum albomyces, Thermomyces lanuginosus and Sporotrichum thermophile appeared when temperature ranged between 51 and 54 (Moubasher et al. 1984). Scytalidium thermophilum is an important thermophilic fungus in the production of mushroom compost (Wiegant 1992). Scytalidium thermophilum was also isolated from fresh streak horse droppings and dominated the fungal biota of compost. Myriococcum themophilum, Stilbella thermophila, Thielavia terrestris, and other thermophilic fungi are considered for preparation of compost (Straatsma et al. 1994). Interaction of thermophilic fungi Chaetomium thermophilum, Humicola insolens, Scytalidium thermophilum, and Talaromyces thermophilus isolated from mushroom compost was studied for mutual influence (Bilay and Elliott 1995). Thermophilic and thermo-tolerant fungi Thermomyces lanuginosus, Scytalidium thermophilum, and Paecilomyces variotii were predominant in a well-defined material consisting of straw and an N source when composted in insulated containers (Klamer et al. 1998). Suyanto et al. (2003) isolated a novel thermophilic fungus *Chaetomium* sp. no. M.S.017 having palm-oil mill fiber-decomposing properties. Kalogeris et al. (2003) produced endoglucanase and β -glucosidase (extracellular cellulolytic enzymes) in solid state fermentation of wheat bran, wheat straw, rice straw, corn cobs, and oat bran by thermophilic fungus Thermoascus aurantiacus.

4 Enzymes Involved in Lignocellulosic Biomass Degradation

4.1 Cellulose Degrading

Once a substrate with high accessibility is generated, the enzymatic hydrolysis is required, involving a cellulolytic complex acting in synergy (De Lima et al. 2016; Johansen 2016; Kim et al. 2014; Payne et al. 2015). Cellulose being a very recalcitrant biopolymer needs consortia of enzymes to expedite it bioconversion after thermo-chemical or biological pre-treatment. In actual facts, that fungal cellulases-which were major representatives of free cellulases-acted approximately five times faster than cellulosomes against either untreated or delignified cell walls and were found to penetrate more into the pore structure of lignocellulose microfibril networks as against the larger cellulosome complexes, which could only penetrate the larger wall lamella gaps. These advantageous lignocellulose degradation properties exhibited by free cellusases on cell walls were compromised when digesting purified forms of crystalline cellulose—cellulosomes were much more active on purified cellulose (Dixon 2013). Interestingly, the two systems tend to complement each other. That is, these systems display dramatic synergistic lignocellulose degradation activities (Brunecky et al. 2013; Resch et al. 2013). In spite the markedly different paradigms, the chemistry of cellulose degradation between the two systems remains the same. All cellulases use one of two mechanisms to degrade cellulose: (1) hydrolysis with retention of the stereochemistry of the anomeric hydroxyl

group, or (2) hydrolysis with inversion of the anomeric hydroxyl group. One important difference between these mechanisms is that most retaining enzymes can catalyze both transglycosylation and hydrolysis since a covalent intermediate is produced on the enzyme, while no known inverting enzyme catalyzes transglycosylation (Wilson 2008). Nevertheless, while multiple microorganisms across the various trees of life produce different categories of CAZymes to decrypt the underlying codes of the plant cell wall assembly, filamentous fungi seem to be better evolved for this ability-that is, they are able to attack and efficiently hydrolyze its constituent polymers to their oligomeric and monomeric building blocks through the collective action of multiple carbohydrateactive enzymes, typically acting together as a cocktail with complementary modes of action. This is their sacrosanct duty in nature as they are responsible for the vast majority the biomass degradation in nature. Fungi have colonized a vast range of terrestrial and marine environments and are thus of vital importance for the recycling of carbon on Earth (Couturier et al. 2016; Pavne et al. 2015).

A two-component cellulolytic system was proposed designated as C1 occurs preliminary to the hydrolysis of the native cellulose straight chain by the second component (Cx). The C1 component helps in the conversion of the native cellulose molecule into linear anhydroglucose chains, as well as might be involved in splitting cross-linkages in native cellulose. On the other hand, the second component (Cx) corresponds to enzymes hydrolyzing the 1,4-β-glucosidic linkage in cellulose, is not produced in the absence of substances containing this particular linkage, and largely unable to hydrolyze starch, pectic acid, alginic acids, or bacterial dextran (Reese 1956). Follow up to the hypothesis, there were several propositions on the identity of the C1 and Cx components involved in the microbial conversion of native cellulose to simpler sugars. For example, the Cx component was designated as endoglucanase acting before C1, a cellobiohydrolase (Ladisch et al. 1983); the catalytic domain corresponds to the hydrolytic Cx components and the cellulose-binding domain corresponds to the nonhydrolytic C1 component (Din et al. 1994). Six decades after the proposition and confusions, the two-component system on the microbial strategy for native cellulose degradation is still widely acceptable. However, the general consensus on the elements of the component is that the C1 are largely accessory proteinslytic polysaccharide monooxygenases, swollenins, expansins, hemicellulasesthat helps to overcome the polysaccharide crystallinity and the Cx corresponds to the core cellulases-endoglucanases, exoglucanases, and betaglucosidases (Horn et al. 2012; Johansen 2016; Kang et al. 2013; Kim et al. 2014; Kracher et al. 2016; X. Liu et al. 2015; Martinez 2016; Qin et al. 2013). These groups of enzymes are generally for the core cellulases (Cx), endoglucanases (EC 3.2.1.4) randomly cleave β-1,4 bonds in amorphous areas of cellulose chains and generate soluble cellodextrins with new reducing and non-reducing ends. Endoglucanases display a two possible catalytic mechanisms with retention of configuration or with inversion of configuration and have a characteristic large cleft containing the catalytic amino acids. Cellobiohydrolases (EC 3.2.1.91) reducing ends of cellodextrins produced from the actions of endoglucanases. The characteristic feature of cellobiohydrolases is the presence of tunnel-bearing structures, which allow the enzyme to slide along the cellulose chain to the next cleavage site as the product is released. β -glucosidases (EC 3.2.1.21) hydrolyze cellobiose or cello-oligosaccharides into glucose units. They are characterized by a pocket-containing topology for optimal detection of the nonreducing extremity and of a single sugar unit. β-Glucosidases are non-processive enzymes because the substrate must be released after each cleavage event to allow the new glucose molecule to exit the pocket. Both exoglucanases and β-glucosidases are strongly inhibited by their reaction products, cellobiose and glucose, respectively (Bischof and Seiboth 2016; Girard et al. 2013; Glass et al. 2013; Gupta et al. 2016; Himmel et al. 2010; Kim et al. 2014; Xu et al. 2011). In contrast to the core cellulases, the elements of the C1 components keep expanding with advancement in technologies and bioprospecting. However, major accessory proteins discovered till date are lytic polysaccharide monooxygenases (LPMOs), swollenins, expansins, and the various hemicellulases (Bischof and Seiboth 2016; Cragg et al. 2015; Espino-Rammer et al. 2013; Horn et al. 2012; Kang et al. 2013; Marjamaa et al. 2013; Martinez 2016; Payne et al. 2015; Oin et al. 2013). LPMOs are classified as fungal AA9 (previously GH61) enzymes and non-fungal members of family CBM33. They are implicated in the oxidative cleavage of cellulose and other plant polymers and have gained lot of tractions in recent times because of their huge synergistic effect on the activity of canonical cellulases in the saccharification of pre-treated lignocellulosic biomass. LPMOs are active on varieties of polysaccharides either crystalline or amorphous. Examples include chitin, cellulose, starch, xyloglucan, cellodextrins among others. LPMOs generally work in tandem with the enzyme cellobiose dehydrogenase (CDH), which serves as its redox partner to bring about its oxidative cleavages of polymeric substrates. Other potential extracellular electron source includes ascorbic acid, lignin, phenols procured from plant biomass or produced by fungi, and LPMO-reducing di- or triphenols among others. They have equally been reported to effect classical Fenton chemistry on varieties of substrates (Courtade et al. 2016; Cragg et al. 2015; Eibinger et al. 2014; Hemsworth et al. 2014; Johansen 2016; Kracher et al. 2016; Lo Leggio et al. 2015; Martinez 2016; Patel et al. 2016). Expansins, on the other hand, are noncatalytic, non-hydrolytic proteins that can play a role in cellulose degradation by the disruption of cellulose fibers and by breaking hydrogen bonds between cellulose microfibrils and cellulose-hemicellulose complexes, leading to polysaccharide slippage, also called creep activity. They are normally part of plant cell wall protein but have also been identified in varieties of bacteria and fungi actuating plant cell wall depolymerization activities. A representation of expansin in fungi is called swollenin (Cosgrove 2000; Cosgrove et al. 2002; Gourlay et al. 2013; Kang et al. 2013; Kim et al. 2014; Rocha et al. 2016; Saloheimo et al. 2002).

4.2 Hemicellulose-Degrading Enzymes

One other major accessory proteins fitting into the C1 modules are the group of proteins that acts on hemicellulose broadly categorized as hemicellulases. Hemicelluloses component of a biomass can be degraded by different kinds of enzymes viz. (1) endoezymes-which randomly cleave the bonds between the building blocks of a polymer (2), exoenzymes-which cleave either a single or monomer from the end of the polysaccharide chain, and (3) glycosidases—hydrolyze the oligomers of disaccharides of hemicellulose polymer (Alexander 1977). Hemicellulose being diverse requires a diverse set of enzymes acting on both terminal and/or internal glycosidic linkages. They include xylanases, β-xylosidases, xyloglucanases, acetyl xylan esterases, ferulic esterases, β-mannanases, β-galactosidases, arabinofuranosidases, and glucuronidases (Arantes and Saddler 2010, 2011; Gourlay et al. 2013; Gupta et al. 2016; Gusakov 2013; Linton and Greenaway 2004; Sun et al. 2015), which are produced by many microorganisms including fungi Aspergillus niger (Conrad 1981; Gokhale et al. 1986), Aspergillus, and Trichoderma reesei (Dekker 1983; Chahal 1985). Biswas et al. (1990) have characterized the xylanase produced by a hyperxylanolytic mutant of Aspergillus ochraceus. Bhat et al. (1987) screened several thermophilic fungi for extracellular xylanase. Of the thermophiles, Thermomyces lanuginosus RM-B performed best producing 154U xylanase. The enzyme was reported to have optimum temperature 60-70 °C at neutral pH (Bakalova et al. 2002).

4.3 Lignin-Degrading Enzyme

Lignin degradation is an oxidative process mainly attributed to the secondary metabolism, or to restricted availability of carbon, nitrogen, or sulfur, and it is normally not degraded as sole carbon and energy sources (Silva et al. 2010). In nature, it is generally versatile peroxidases (VPs; EC 1.11.1.16; CAZy AA2) that combine the substrate-specificity characteristics of the two other ligninolytic peroxidases (LiP and MnP). Unlike these two enzymes, it can oxidize phenolic and nonphenolic substrates including veratryl alcohol, methoxybenzenes, and lignin model compounds (Wong 2009). Peroxidases such as HRPs (EC 1.11.1.7, CAZy AA2), secreted by fungus *Phanerochaete chrysosporium*, are potential biocatalysts for bioremediation of environment polluted by harmful compounds (e.g., endocrine disrupting compounds: 17α -ethinylestradiol (EE2)) (Rathner et al. 2017). HRPs have also been implicated in the cell wall biosynthesis, indole-3-acetic acid (plant growth hormone) catabolism, and oxidation of toxic compounds (Falade et al. 2017). The lignin degradation can be further enhanced by the action of other enzymes such as aryl alcohol oxidases (AAO;EC1.1.3.7.; CAZy AA3) that oxidize many primary alcohols containing an aromatic ring and is described in Pleurotus eryngii (Guillén et al. 1992), glyoxylate oxidase (GOx, EC 1.2.3.5) (Kersten and Cullen 2007), pyranose 2-oxidase (glucose 1-oxidase; EC1.1.3.4, CAZy AA3), and cellobiose dehydrogenase (CDH, EC 1.1.99.18; CAZy AA3). Fungal aryl-alcohol dehydrogenases (AAD; EC 1.1.1.90) and guinone reductases (OR, EC 1.6.99.2), tyrosinases (EC.1.14.18.1) and catechol oxidases (EC 1.10.3.1) are also involved in lignin degradation (Gutiérrez et al. 1994; Frommhagen et al. 2017). Significant amounts of peroxidases have been produced by fungi in the submerged or solid state fermentation. In the submerged culture, the fungus *Mucor racemosus* produced lignin peroxidases (75,376 U/L) and manganese peroxidases (4484 U/L) (Bonugli-Santos et al. 2010) while in a solid state fermentation of steam-exploded wheat straw by Phanerochaete chrysosporium manganese peroxidases (1375 U/L) were observed (Bonugli-Santos et al. 2010). Submerged fermentation of versatile peroxidases (7300 U/L) was realized by genetically modified Pleurotus ostreatus (Tsukihara et al. 2006). Finally to conclude, the lignin degradation is not catalyzed by any particular enzyme but concerted action of oxidative coupling of phenol oxidase, peroxidase, glucose oxidase, and ligninase. Ligninase catalyzes breakdown of the ether linkages with glucose oxidase providing necessary co-substrate peroxide, then peroxidase involved in radical formation, phenol oxidases polymerize and depolymerize phenols often, initial degradation depends on the level of glucose oxidases. Laccases are involved mainly in decarboxylation of side chains and to some extent in the oxidative cleavage.

5 Factors Affecting Lignocellulose Degradation

A wide range of fungi, bacteria, and actinomycetes are present on agricultural produce. The rate of their growth depends on the conditions of storage particularly the water activity of the substrate, aeration, ambient temperature, nutritional factors, pH, and type of substrate. If the energy released during the respiration of the plant and microbial cells cannot escape as quickly as it is produced, the substrate heats up and conditions which may favor the growth of thermophilic organisms (Sharma and Johri, 92).

5.1 Carbon to Nitrogen Ratio

The C/N ratio plays an important role in the decomposition of any organic matter. If a material has wider C/N ratio, then it is advisable to bring down C/N ration by adding nitrogen to enhance decomposition. The addition of nitrogen to materials increases their rate of decomposition. Addition of nitrogen to a substrate whose initial C/N ratio is above a critical ratio (20–30) causes a decrease in the substrates C/N ratio during its decomposition. If the initial C/N ratio is below the critical one, no change in the substrates C/N ratio will occur with time. Net mineralization of organic "N" occurs when the substrate being decomposed has an initial C/N ratio which is below the critical one (Henna 1975). Rajasekaran and Sampatkumar (1981) suggested that with proper C/N ratio adjustments, agricultural waste could be probably recycled with a greater degree of "N" enrichment. The lowering of C/N ratio of saw dust to 12: 1 with inorganic N enhanced decomposition of saw dust at 25 °C (Olayinka and Adebaya 1984). The addition of nitrogen in the form of fertilizer or cakes to nitrogen poor materials accelerated their decomposition. About 290 mg of CO₂ was evolved/100 g soil when wheat straw was supplemented with nitrogen as compared to wheat straw alone (120 mg of COp 00 g soil) (Bhardwaj and Gaur 1985). Nitrogen supplementation in the form of 0.08% NH₄CI was found to be optimum for *Pleurotus ostreatus* and *Sporotrichum pulverulentum* to degrade paddy straw (Kahlon and Dass 1987). Nigam et al. (1988) reported the effect of various nitrogenous substances on bioconversion of whole bagasse using four fungal cultures. Yeast extract was better source of nitrogen than NH₄CI for Streptomyces badius and Streptomyces viridosporus (Giroux et al. 1988). Urea nitrogen favored the degradation of lignin as well as cellulose and hemicellulose up to a certain level (1.5% sterile urea or 3% unsterile urea on dry weight basis) beyond which the degradation of lignin was relatively more adversely affected than cellulose (Tripathi and Yadav 1991). The application of paddy straw degraded with efficient fungi in combination with N₈₀ produced a significant effect on wheat yield due to biodegradation of paddy resulting in nutrient-rich compost with a low C:N ratio and high humus content (Kanotra and Mathur 1994). Soybean meal increased xylanase production in mixed cultures of fungi (Gutierrez-Correa and Tengerdy 1998). The organic nitrogen source of soybean helped only growth but not enzyme production in mixed cultures of Trichoderma reesei and Aspergillus niger when grown on sugarcane bagasse (Gutierrez-Correa et al. 1999). George et al. (2001) reported that the use of yeast extract stimulated xylanase (15 IU/mL) by Thermophilic fungi, Thermomonospora sp. cellulolytic enzyme activities were higher with inorganic nitrogen, when Thermoascus aurantiacus was grown on organic residues. Ammonium sulfate, the least expensive nitrogen source, gave the best activities (Kalogeris et al. 2003). The C:N ratio in Phanerochaete flavido-alba inoculated lignocellulosic substrates decreased during the first month of composting and from then onwards increased because of C loss. The final C:N ratio reached with this fungus was close to 28 (Lopez et al. 2002). The main factor influencing enzymatic hydrolysis is lignin and hemicellulose content.

5.2 pH

For solid state fermentation of agricultural residues, pH adjustment is difficult and most of the agro-residues have pH near neutrality. The optimum pH for lignin decomposition by *Phanerochaete chrysosporium* was 4–4.5 with marked suppression above 5.5 and below 3.5 (Kirk et al. 1978). The optimum pH for *Coriolus versicolor* for lignin biodegradation was 4.5 (Ishihara 1983). Hegarty and Curran (1985) showed that different fungi used in their study produced greatest weight

losses of wood in pH 5.0–8.0 range. Most xylanases known so far have their optimum pH around neutrality. Even xylanases produced by most alkaliphiles reported to date have their optimum pH around neutrality (Rajaram and Varma 1990; Park et al. 1992). Maximum xylanase activity by B. *circulans* AB 16 grown on modified medium containing wheat straw was between pH 6–7 retaining **91%** activity at pH 5 and 85 and 55% at pH 8.0 and 9.0, respectively (Dhillon and Khanna 2000). The xylanase from *Thermomonospora* sp. enzyme was active in a broad pH range (5–9) with maximum activity at pH 7.0, and retained 68% and 44% of its maximum activity at pH 8.0 and 9.0, respectively (George et al. 2001). The optimum pH for cellulase production was found to be resulting in high endoglucanase (1572 U/g) and *p*-glucosidase (106.6 U/g wheat straw) activities during solid state fermentation of wheat straw by *Thermoascus aurantiacus* (Kalogeris et al. 2003).

5.3 Aeration

Aeration greatly affects the composting process, during which organic matter undergoes biological and physiochemical reactions, mainly by oxidation, with the final production of carbon dioxide, water, and partially humified organic matter (He et al. 1992). The most important event during composting is the generation of heat. The requirement of oxygen increases in thermophilic range of temperature but normally remains constant at the temperature of 65 °C. The cause of heat generation above 65 °C is exothermic chemical reaction (Currie and Festenstein 1971). The thermophilic fungi can undergo growth as facultative aerobes with 0.7-1.05% aeration (Deploey and Fergus 1975). Solid state fermentation studies were undertaken to investigate the effect of O₂ supply, aeration rate, and carbon source, and level on the shelf life of *Lagenidium giganteum*. Growth rate as measured by CO₂ evolution rate was greater at higher aeration levels (Vander-Gheynst et al. 2000). Carrasco et al. (1999) evaluated a bioreactor for solid state fermentation of sugarcane bagasse and cellulolytic fungus strain Aspergillus fumigatus at the rate of 18 L/min aeration. Submerged fermentation was carried out in fermenter with controlled aeration (1.5 VVM) and dissolved oxygen level (30%), rifamycin production increased to about 1.5-fold greater than the flasks (Venkateswarlu et al. 2000). Lopez et al. (2002) reported that degradation was enhanced when low aeration (3 L of air per week) was supplied during SSF of cucumber and green bean waste inoculated with whiterot fungi.

5.4 Temperature

The rate of carbon mineralization of saw dust increased with temperature from 25 to 40 °C with the optimum being 30–35 °C (Olayinka and Adebaya 1984). Hegarty and Curran (1985) reported maximum wood decay by the majority of the test fungi

was in the range of 15–25 °C. Heat pre-treatment of cereal straw at 60 and 90 °C was considered sufficient for commercial cultivation process with the added benefit of a fermentative pre-treatment reducing the risk of infection but best growth and highest speed of colonization were on the sterilized straw (Maziero and Zadrazil 1994). Horwath and Elliott (1996) composted rye grass for 45 days at a mesophilic temperature of 25 °C.

5.5 Water Activity/Moisture

Most species of fungi require at least 0.75_{aw} while no fungal growth occurs with less than 0.61_{aw} . With changes in temperature, vapor pressure over the substance did not change very much while with change in temperature, the extent of microbial activity of substrate was determined by the water activity (Christensen and Kaufman 1974). The number of moulds was reduced and *Pleurotus ostreatus* yield was increased (Stanek and Bisko 1982). Eighty per cent of water content seemed to be an important parameter for improvement of feed value for ruminants. Solid state fermentation of banana fruit stalk at 70% moisture content was found suitable for SSF with thermophilic *Bacillus* (Krishna and Chandrasekara 1996). The mixed culture of *Trichoderma reesei* with *Aspergillus niger* for solid state fermentation of sugarcane bagasse produced better results at 80% moisture content (Gutierrez-Correa et al. 1999). Initial moisture content (80%) of coffee pulp under solid state fermentation with *Aspergillus niger* had a significant effect on the total amino acid content.

5.6 Other Factors

Jeffries et al. (1981) reported that phosphorus starvation did not influence lignin degradation. These workers also suggested that the balance of trace metals Mg⁺⁺ and Ca⁺⁺ was important for lignin biodegradation. Shaikh et al. (1997) found that the divalents cations such as Co²⁺, Mn²⁺, Fe²⁺, and Zn²⁺ had strong inhibitory effect on enzyme activity under solid state fermentation. The concentration of veratryl alcohol necessary for full protection of ligninase activity varied according to the concentration of H₂O₂ present in the medium, which in turn depends on carbon source (glucose or glycerol). It is proposed that the carbon source influences the overall ligninase activity, not only directly by affecting the rate and the type of synthesized ligninase but also by affecting the ratio of H₂O₂ production (Tandon and Odier 1988).

6 Fermentation

The limitations of physical and chemical methods of upgrading the nutritive value of straws and agro-residues have been overcome by the application of microbial process which can be used for bioconversion as well as production of lignocellulolytic enzyme by using thermophilic as well as mesophilic fungi. The processes are:

- 1. Composting
- 2. Submerged fermentation
- 3. Solid state fermentation.

7 Composting

The formation of compost from the degradation of organic matter depends on the abilities of microflora to produce and excrete specific degradative enzymes (Hankin et al. 1976). Crawford (1983) compiled the information regarding composting of agricultural residues. Several reports are available where inoculation with mesophilic fungi improved the quality of compost. Gaur et al. (1982) reported that inoculation of four mesophilic fungi, *Aspergillus niger, Aspergillus sp., Trichoderma viride,* and *Penicillium* sp., had increased total nitrogen, available phosphorus and humus content of composted sorghum stalk and wheat straw (5:3) and jamun leaves. Inoculation of cellulolytic fungi and actinomycetes increased the temperature faster and accelerated the process of rice straw composting (Hang Won et al. 1995). The mesophilic microorganisms were used for bioconversion of rice husks cane bagasse, wheat straw, maize cob and corn stalk, and many other agricultural residues listed in Table 8.1. Several workers have dealt with rice straw composting after

Substrate	Fungi	Reference	
Sugarcane bagasse	Trichoderma viride	Sharma et al. (1991)	
	Trichoderma reesei	Gutierrez-Correa et al. (1999)	
	Pleurotus sp.	Zadrazil and Puniya (1995)	
Rice straw	Pleurotus florida, Sporotrichum sp.	Mehta et al. (1990)	
	Trichoderma viride	Kanotra and Mathur (1994)	
Rice husk	P. sajor-caju	Beg et al. (1986)	
Maize straw	Phanerochaete chrysosporium	Breccia et al. (1997)	
	T. reesei	Ming et al. (1999)	
Wheat straw	Stropharia sp.	Kamra and Zadrazil (1986)	
	P. ostreatus, P. flabellatus	Rajarathnam et al. (1987)	
Coir pith	Polyporus sp.	Mani and Marimuthu (1992)	
	T. viride	Muniswaram and Charyulu (1994)	
Saw dust	Polyporus sp.	Nandi et al. (1996)	

 Table 8.1
 Bioconversion of agricultural residues by different fungi

supplementation with rock phosphate, glue waste, basic slag, and inorganic nitrogen sources (Jhorar et al. 1991; Wahyono and Sahwan 1998; Abdel-Azeem 2001). The use of cattle manure rather than relatively expensive microbial catalyst could be recommended as decreasing the composting degradation period of rice straw (Bhumibhamon et al. 1988). The inoculation of *Azotobacter* to 1-month-old decomposed rice straw increased the N-content and composting with rock phosphate increased both citrate and water-soluble phosphorus (Tiwari et al. 1988).

The two-step process of composting of rice straw was effective in accelerating humification, cation exchange capacity of the composted material (Kakezawa et al. 1992). Nandi et al. (1996) proposed a two-step process for composting of rice straw which results in better humus formation. The treatment of shredded rice straw supplemented with chemical accelerator, sheep manure, and EM solution (mixture of microbial culture, commercially known as EM solution) gave the most desirable characteristics of the final product particularly narrow C:N ratio and high WHC (Abdel-Azeem 2001). Two white-rot fungi, Pleurotus ostreatus and Lentinus edodes delignified and increased digestibility of corn straw (Zea mays) under solid state fermentation (Sermanni et al. 1994). Under fed-batch process of saccharification of short fiber waste material from the paper industry by *Penicillium* sp. and *Trichoderma* reesei, combined preparation of T. reesei and Aspergillus showed highest saccharification activity (Castellanos et al. 1995). Cellulolytic fungi such as Trichoderma reesei QM9414, Trichoderma viride, Pleurotus sajor-caju, and Coprinus cinereus lowered the C:N ratio of paddy straw during composting (Kanotra and Mathur 1994). Phospholipid fatty acid (PLFA) analysis study of compost samples collected from bench-scale grape pomace and rice straw composting showed that inoculation had little effect on the microbial community structure of the compost when temperature peaked (Lei-Fei et al. 1998). Thermophilic fungi are generally used for mushroom compost preparation either in tunnels or windrows (Gerrits et al. 1995; Straatsma et al. 1995; Sharma et al. 2000). Ross and Harris (1983) reported potential of Torula thermophila as a rapid composting organism. Natural thermophilic flora played a role during composting of wheat and broad bean straw (Moubasher et al. 1984). Biomass of these fungi was estimated by Wiegant (1992) in mushroom compost. Sharma and Johri (1992) have documented the role of thermophilic fungi in composting. Scytalidium thermophilum is an important thermophilic fungus in the production of mushroom compost (Wiegant 1992; Sharma et al. 2000).

8 Submerged Fermentation

It involves the growth of microorganisms as a suspension in a liquid medium in which various nutrients are either dissolved as in many cases of commercial media or suspended as particulate solid (Frost and Moss 1987). As the water content increases, the gas phase is reduced and gas exchange is thus increasingly impeded, finally conditions become anaerobic in the substrate suspension (Rajarathnam and

Bano 1989); hence, better aeration of the medium is a must, which can be done either by an air sprayer or an agitator. In submerged fermentation process, much care has to be taken in maintaining sterility throughout the period of process, and heat during the process has to be removed. Submerged fermentation process is being used for the production of many industrially important products such as antibiotics, alcoholic beverages, and single cell protein from agricultural residues. Garg and Neelakanthan (1982) fermented bagasse with Aspergillus terreus under submerged culture conditions. The fermented products contained 32.9% crude protein and protein recovery was 22.2%. Submerged fermentation of cellulose of solka floc with Aspergillus terreus for 48 h at 35 °C resulted in maximum reduction of cellulose and 37.8% increased crude protein content in the product (Miller and Srinivasan 1983). Humicola aeruginosa and Sporotrichum thermophile performed best in liquid fermentation; the higher productivity of xylanase (18.72 IU/mL) was obtained in liquid culture of *H. lanuginosa* (Grajek 1987). Dighe and co-workers (1988) utilized enzymatic hydrolysates of various crop wastes for the production of yeast biomass. Holder et al. (1989) cultivated Geotrichum candidum and Candida utilis ATCC 9256 in glucose containing bagasse hemicellulose hydrolysate. The cell yields of G. candidum and C. utilis obtained were 19.5 and 15 g per liter, respectively, in hemicellulose hydrolysates. Gulati (1989) studied various factors that affected the submerged fermentation of sugarcane bagasse by Trichoderma reesei OM 9414. The thermophilic fungus Thermomyces lanuginosus was investigated in semi-solid and liquid fermentation for production of cellulase-free endoxylanase from corn cobs. The best results were obtained with solid to liquid ratio 1:2 and inoculums size 20% (v/v) (Damaso et al. 2000).

9 Solid State Fermentation

Durand et al. (1988) defined solid state fermentation (SSF) as a process which refers to the culturing of microorganisms on and/or within the particles of solid matrix (solid substrate) where the liquid content bound with them is at the level corresponding to the water activity assuring growth and metabolism of cells, but not exceeding to the maximal water holding capacity of the solid matrix. The advantages and disadvantages of solid state fermentation are discussed by many workers (Moo-Young et al. 1983; Rosen and Schugerl 1983; Tengerdy 1985). Baca et al. (1993) studied the gas phase of compost during SSF of sugarcane bagasse. The evolution of CO_2 in the compost atmosphere (0–3.5 mmol L) paralleled temperature changes, whereas the highest O_2 concentrations (10–20%) coincided with the end of thermophilic phase. The accumulation of ethylene (0.01–0.4 ppm) was not markedly affected by changes in temperature. The maximum loss of organic matter of wheat straw was reported after 120 days of inoculation period during solid state fermentation process using two species of *Pleurotus* fungus (Zadrazil 1976). Solid state fermentation of sugarbeet pulp using *Thermoascus aurantiacus* and Sporotrichum thermophile (Myceliophthora thermophila) increased two fold protein content in 48 h (Grajek 1988). Roche et al. (1994) reported that solid state fermentation of sugarbeet pulp with thermophilic fungus, *Thermoascus aurantiacus* ATCC 26904 increased the quantity of alpha-L-arabinofuranosidase. Maximum alpha-L-arabinofuranosidase production occurred when medium was inoculated with 5-day-old spores. Jain et al. (1999) produced xylanase by thermophilic fungus *Melanocarpus albomyces* IIS-68 during solid state fermentation on sugarcane bagasse and purified the enzyme in an aqueous two-phase system comprising polyethylene glycol (PEG 6000) and potassium salts.

10 Conclusion

The preponderance of inducible CAZymes as well as non-hydrolytic accessory proteins in the fungus secretome reiterates the concept of synergism between proteins for effective hydrolysis of lignocellulosic biomass. In addition, the preferential association of certain proteins and the attending differential biomass hydrolysis gives an insight into the concept of protein–protein interaction as well as clues about non-hydrolytic proteins from the fungus with possible roles in the synergistic deconstruction of lignocellulosic biomass.

Fungi involved in complex lignocelluloses containing raw materials degradation express a broad spectrum of enzymes. According to the enzyme composition and degradation mechanisms, three groups of fungi are described: soft-rot, brown-rot, and white-rot fungi. A huge effort has been done and is still ongoing to classify and group these fungi enzymes in a central database called the carbohydrate-active enzyme database or short CAZy. The CAZy classification groups the proteins in families according to amino acid sequence similarity. In the last few years, investigation attention was focused on the synergistic action of the fungal enzymes involved in the lignocelluloses degradation. One of the key players in those actions is LPMOs. These enzymes work in synergy with glycoside hydrolases (GHs) and stimulate their activity by increasing the accessibility to the substrate and enhancing overall polysaccharide conversion efficiency. In addition, LPMOs are able to degrade insoluble polysaccharides such as crystalline cellulose and soluble cellooligomers. Importance of these enzymes is also confirmed by the phylogenetic investigations. Genomes of lignocelluloses-degrading fungi usually encode several LPMOs gens. Many cultivation strategies were studied to enhance efficiency of lignocellulose-containing raw materials degradation whereas delignification was indicated as the main barrier for the successful lignocellulose utilization. To remove lignin and enhance hemicellulose and cellulose degradation, one of the strategies was the solid state fermentation of brown-rot and white-rot fungi. This strategy approved the hypotheses of the synergistic action of the fungal enzymes and enhancement of the delignification efficiency.

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Chapter 9 Recent Advancements in Mycodegradation of Lignocellulosic Biomass for Bioethanol Production



Divya Kumari and Barkha Singhal

Abbreviations

CBH	Cellobiohydrolases
CBP	Consolidated bioprocessing
CELF	Co-solvent enhanced lignocellulosic fermentation
COSLIF	Co-solvent-based lignocellulosic fractionation
CRISPR	Clustered regularly interspaced short palindromic repeats
DMC	Direct microbial conversion
EA	Extractive ammonia
GEMs	Genome-scale metabolic models
GVL	Gamma valerolactone
LPMOs	Lytic polysaccharide monooxygenases
MSW	Municipal solid waste
MTBE	Methyl tertiary butyl ether
NADH	Nicotinamide adenine dinucleotide (NAD) + hydrogen (H)
PGASO	Promoter-based gene assembly and simultaneous overexpression
SHF	Separate hydrolysis and fermentation
SPORL	Sulfite pretreatment to overcome recalcitrance of lignocellulose
SSCF	Simultaneous saccharification and co-fermentation
SSF	Simultaneous saccharification and fermentation
TEF	Translational elongation factor
USA	United States of America

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1 Introduction

The phenomenal rise in the demand of environmental safety and energy security for attaining sustainability strongly favors the development of eco-friendly approaches and solutions for the energy resources at the global context (Wyman 2018). The clear anticipation regarding exhaustion of fossil fuels reserves in upcoming 50 years (Rastogi and Shrivastava 2018) due to their continuous demand and overwhelming use leads to the burgeoning quest for finding sustainable and economical substitutes. Apart from that, the burning of fossil fuels leads to the enhancement of greenhouse gases followed by global warming (Boykoff et al. 2019), loss of biodiversity, rise in sea as well as glacier level becoming a significant threat to the environmental safety and security (Opoku 2019). Therefore, the emergence of bio-based approaches (Dhyani and Bhaskar 2018) owing to their remarkable attributes with respect to ecofriendliness, non-toxicity, biodegradability, the production of biofuels has been considered promising and sustainable alternatives for energy sources. The development of biofuels plays a pivotal role for the building the roadmap of bioenergy-based economy (Demirbas 2008). The biofuels include bioethanol, biodiesel, and biogas (Xiong et al. 2019) and among these the development of bioethanol from various renewable natural sources leads to a paradigmatic shift in the field of bioenergy (Scarlat and Dallemand 2019). Currently, various western countries such as Brazil, China, the USA, Canada as well as EU member states (Kutas et al. 2007) have rigorously adopted the credentials of bioethanol and their commitments for the production of bioethanol has been increasingly realized at a global level. It was estimated that total ethanol production has been increased from 175 million gallons (Zabed et al. 2017) in 1980 to 14,810 million gallons in 2015 in the USA.

Till date, numerous renewable sources have been utilized for the production of bioethanol including sugars, starch, lignocellulosic biomass, and algae. The yield of bioethanol may vary with respect to types of sugars retrieved from the biomass due to diversed biochemical composition. The conversion of sugar and starch into bioethanol (Evcan 2012) has been considered to be first-generation bioethanol while utilization of lignocellulosic biomass and algae has been considered second- and third-generation bioethanol (Robak and Balcerek 2018), respectively. The sugarbased feedstock requires only extraction procedures to get fermentable sugars, while starchy feedstock requires hydrolysis to convert starch into glucose. Currently, the great potential has been envisaged in lignocellulosic biomass for bioethanol production that requires extensive pretreatment (Jönsson and Martín 2016) strategies prior to the hydrolysis in order to loosen cellulosic framework for enzyme accessibility for the bioconversion process. Moreover, the easy availability (Balat 2011) and low cost of lignocellulose biomass as well as proclaimed statement of Nova Institute of Germany regarding lignocellulosic resources for environmental sustainability lead to the enthusiastic drive for the bioethanol production. There are four major steps (Mosier et al. 2005) involved in the production of bioethanol from the lignocellulosic biomass that includes pretreatment, hydrolysis, and fermentation followed by distillation. Till date comprehensive methodologies have been developed for the pretreatment (Hsu 2018) as well as hydrolysis but for the effective bioconversion process (Lee 1997) especially fermentation of sugars greatly requires the involvement of microorganisms. A wide variety of bacterial as well as fungal microbial resources have been utilized (Kumar et al. 2008) for fermentation purpose but more prolific potential has been envisioned by exploring the world of fungi. The fungi possess the promising ability of hyphal penetration power as well as consist of extracellular enzymatic machinery that proves them interesting and excellent choice for better degradation of complex cell wall molecules into simplest form. Even though possessing the remarkable attributes, the field of mycodegradation of lignocellulosic biomass is still impeded by low ethanol yields and depicting lower tolerance towards higher titer of ethanol. Besides that, the recalcitrant structure framework of lignocellulose is costly and labor-intensive process. Therefore, in last decades the intervention of genetic and metabolic engineering methods (Mood et al. 2013) provided significant improvements in all of the four steps of bioethanol production and developed the concept of consolidated bioprocessing (CBP) that makes these resources as an amenable system for the fermentation of different sugars concurrently. Though the intervention of synthetic biology approaches led to a paradigmatic shift in engineering, the microbes for enhancing the process technology for bioethanol production still lagged behind the industrial utilization. Still, various technologies are in different degrees of development and industrial scaling has been reported for the production of bioethanol from lignocellulosic biomass. Therefore, this chapter examines recent advancements in mycodegradation of lignocellulosic biomass for bioethanol production. The chapter also delineates the current bottlenecks and future research priorities that paved the way for economic and sustainable production of bioethanol production at the commercial level.

2 Remarkable Attributes of Bioethanol

The enormous potential has been visualized in the production of bioethanol due to their excellent credentials that makes it the most suitable alternative source of biobased energy. The conventional fuel that is commercialized globally is gasoline or petrol which has been produced through distillation process of petroleum (Bakshi et al. 2007) followed by mixing with different kind of additives to make suitable fuel source. Before the commitments for industrial scale production, scientists have comprehensively compared the properties of gasoline with bioethanol and found superior fuel properties. Bioethanol (Nagdeote and Deshmukh 2012) is biodegradable and oxygenated fuel comprising 34.7% oxygen (Malça and Freire 2006) that leads to higher combustion efficiency with respect to gasoline (C_7H_{17}) which is devoid of oxygen. Bioethanol devoid of sulfur serves as a safer alternative of methyl tertiary butyl ether (MTBE) has been used for the enhancement of octane number of gasoline. The only difficulty has been seen that limits its suitability for complete utilization as fuel is low cetane number that reduces the burning efficiency of bioethanol from compression ignition. The detailed comparative characteristics of

Properties	Bioethanol	Petrol/gasoline
Octane number	106–110	91–96
Higher evaporation enthalpy	1177 kJ/kg at 60 °C	348 kJ/kg at 60 °C
Laminar flame speed @100 kPa, 325 K	Near about 33	Near about 39
Heat of vaporization	840 kJ/kg	305 kJ/kg
Oxygenated fuel	34.7%	Absent
Combustion efficiency	Higher	Lower
Sulfur content	Negligible	Present
Density @15 °C	0.7904	0.7387
Vapor pressure @20 °C (kPa)	5.9	55

Table 9.1 Comparative account of various physicochemical properties of bioethanol

bioethanol versus gasoline that gave the idea of blending of bioethanol versus gasoline are summarized in Table 9.1. Therefore, effective measures have been implemented by mixing of bioethanol with ethylhexyl nitrate (Anderson et al. 2012) or diterbutyl peroxide to enhance the cetane number. The reduced pollution through the emission of the sulfur oxide, particulate matter, and nitrogen oxides has been seen when the blending of gasoline was done with bioethanol. The various countries based on their environmental and climatic condition adopted percentage-based blending of bioethanol with gasoline (Al-Hasan 2003) and the internationally recognized codes have been allocated that depicted the clear indication of blending percentage. The flow chart depicting various codes and their percentage has been depicted in Fig. 9.1.

3 Composition of Lignocelluloses

The potentiality of lignocellulose biomass has been realized as a promising source (Singh et al. 2010) of bioethanol production due to their easy availability, no interference in food and feed crop production as well as abrogating the need of land resources in the past decade. This biomass can be categorized into various groups that includes agricultural residues (wheat rice and other cereals straws, bagasse, and stovers), forest biomass (sawdust, hard wood, softwood, wood chips, pruning, and bark residues), energy crops (C_3 or C_4 plants like perennial grasses and other energy crops such as Lantana camara, Ricinus communis), aquatic plants (water hyacinth), and organic materials of municipal solid wastes (MSW). The major components of lignocellulosic biomass are cellulose (30-35%) (Sánchez 2009), hemicellulose (25–30%), and lignin (10–20%) (Wu et al. 2019) but smaller proportions of protein, pectins, lipids, minerals, and water are also present in these resources. The recalcitrant as well as robust structure (Zhao et al. 2012) of this biomass is due to different covalent and hydrogen bonding pattern among cellulose, hemicellulose, and lignin that confer crystallinity and a high degree of polymerization that leads to high fiber strength of lignocellulose biomass.



Fig. 9.1 The percentage of petrol-ethanol mixture and the international codes for their description and identification

Cellulose is one of the most rigid linear and crystalline homopolymers consists of D-glucospyranose subunits linked through β -1,4-glycosidic linkage (Bauer et al. 1999; Xu et al. 2019). The compactness of cellulose is due to the formation of micro-fibrils (Festucci-Buselli et al. 2007) that is stabilized by van der Waals forces and intra- and intermolecular hydrogen bonds. Hemicellulose is a heteropolymer composed of a variety of hexoses (glucose, mannose, galactose), pentoses (arabinose, xylose) (Kang et al. 2014) as well as sugar/uronic acids (D-galacturonic, D-glucuronic, methylgalacturonic) linked with each other through β -1,4- and β -1,3glycosidic bonds. The major component of hemicellulose is xylan (Hendriks and Zeeman 2009) and its composition may vary within different feedstocks. Due to its varied composition, a wide variety of enzymatic resources has been required for the hydrolysis. Lignin is a complex heteropolymer consists of three structural monomers coumaryl alcohol, sinapyl alcohol, and coniferyl alcohol (Buxton and Russell 1988). It is a major constituent of wood (30-60% for softwoods and 30–55% for hardwoods) and linked with both cellulose and hemicellulose. It has been envisioned that a wide variety of lignocellulose feedstocks are available around the world but the production of bioethanol depends on the efficiency of glucan (for hexose sugars) xylan (for pentose sugars) hydrolysis (García-Aparicio et al. 2007) into soluble sugars, types of pretreatment strategies, and finally microorganisms involved in fermentation process. The various lignocellulose resources and their major advantages have been shown in Fig. 9.2 (Nasir and Ghazi 2015; Kumar et al. 2009; Dodo et al. 2017; Amin et al. 2017; Menegol et al. 2017; Kumar and Sharma 2017).



Fig. 9.2 Types of different lignocellulosic biomass, their composition, and advantages for bioethanol production

4 Basic Mechanism for the Conversion of Lignocellulosic Biomass to Bioethanol

The abundant availability (Gamage et al. 2010) of lignocellulosic biomass represents a long-term and sustainable solution for the rising energy demands and global climatic changes due to exhaustion of fossil resources. The versatile composition and physicochemical properties of lignocellulosic biomass, significantly affects the conversion process that considerably affects the yield of bioethanol; therefore, no universalized methods of bioconversion have been developed till date. In general, the conversion process has been initiated from grinding or pre-processing of the biomass for reduction in particle size and increment in the surface area of the particles. Thus, the conversion of lignocellulosic biomass to bioethanol centers around three broad interrelated steps including: (1) pretreatment of lignocellulose biomass (Moreno et al. 2017) for enhancing the accessibility biomass polysaccharides for hydrolytic breakdown, (2) enzymatic hydrolysis (Puttaswamy et al. 2016) of pretreated biomass to fermentable sugars, (3) microbes mediated fermentation of the sugars to produce ethanol. The detailed description is as follows:

4.1 Pretreatment of Lignocellulosic Biomass

Pretreatment step for the conversion of biomass into ethanol is considered as the most complex (Nasir Iqbal and Kamal 2012) and costly step that has the ability to alter macroscopic and microscopic structures of the biomass. This process has been

significantly utilized for the removal of hemicellulose and lignin from the lignocellulosic feedstock; enhance the porosity and surface area of the biomass to reduce the crystallinity of cellulose. An ideal pretreatment strategy should possess the capability of generating reactive solid biomass with minimal reduction in sugars and fewer requirements of enzymes. There are broadly four methods for the pretreatment of lignocellulose biomass such as physical, chemical, physicochemical, and biological (Rezania et al. 2017) and overall success of the strategies are dependent on mode of reaction conditions and their mechanism of action. The extensive literature has been available for the pros and consequences on the pretreatment technologies that have been used previously but currently novel pretreatment technologies have developed for enhancing the efficiency of bioconversion process. These promising pretreatment methodologies include co-solvent-based lignocellulosic fractionation (COSLIF) (Sathitsuksanoh et al. 2010), co-solvent enhanced lignocellulosic fractionation (CELF) (Nguyen et al. 2016), extractive ammonia (EA) pretreatment (Chundawat et al. 2007), pretreatment applying ionic liquid(s) (Swatloski et al. 2002; Dadi et al. 2006; Singh et al. 2009; Li 2010; Cheng and Timilsina 2011; Perez-Pimienta et al. 2016; Singh and Simmons 2015; Konda et al. 2014), γ -valerolactone (GVL) pretreatment (Shuai et al. 2016; Wu et al. 2019), switchable butadiene sulfone pretreatment (De Frias and Feng 2013), and sulfite pretreatment to overcome recalcitrance of lignocellulose (SPORL) (Zhu et al. 2010). Though the above described methodologies have worked efficiently but require high capital cost, input energy as well as cause pollution. Therefore, an efficient, cost-effective, and eco-friendly pretreatment of lignocellulosic biomass has been developed by employing microorganisms (Sindhu et al. 2016) that do not require harsh conditions and specialized infrastructure. Various bacteria (Wilhelm et al. 2019) and fungi (Liu et al. 2019) have been explored for this purpose but great potential has been envisaged in fungi particularly white-rot, brown-rot, and soft-rot fungi. Brown-rot fungi cleave cellulose, while white- and soft-rot fungi cleave both lignin and cellulose. Various white-rot fungi (Baba et al. 2011), such as Phanerochaete chrysosporium, Cyathus stercolerus, Ceriporia lacerata, Ceriporiopsis subvermispora, Pleurotus ostreatus, Pycnoporus cinnabarinus, Pleurotus chrysosporium, Trametes versicolor, Bjerkandera adusta, and Ganoderma resinaceum, possess ligninolytic enzymatic machinery including lignin peroxidases, laccases, and manganese-dependent peroxidases that showed high delignification efficiency (Singh et al. 2014). Table 9.2 depicting various fungal species involved in pretreatment as well as hydrolysis of lignocellulosic biomass (Hahn-Hägerdal et al. 2006; Gray et al. 2006; Rouches et al. 2016; Banerjee et al. 2010). However, this pretreatment methodology suffers from serious limitation of low hydrolysis rate and requirement of longer incubation time; therefore, more comprehensive investigations must be required for the assessment of more white-rot fungi for enhanced ability for the delignification at shorter time interval. Table 9.3 depicting different types of pretreatment methodologies with their advantages and disadvantages (Harmsen et al. 2010; Mora-Pale et al. 2011; Brandt et al. 2011; Nitsos et al. 2013). Therefore, in a nutshell, till date the 100% conversion efficiency (Rocha-Meneses et al. 2017) has not been found in all pretreatment methodologies; therefore, promising potential have been seen by the

Phylum	Class	Family	Species
Ascomycota	Saccharomycetes	Saccharomycetaceae	Saccharomyces cerevisiae
	Saccharomycetes	Dipodascaceae	Yarrowia lipolytica
	Saccharomycetes	Saccharomycetaceae	Hansenula polymorpha
	Saccharomycetes	Saccharomycetaceae	Pichia pastoris
Neocallimastigomycota	Neocallimastigomycetes	Neocallimastigaceae	Piromyces sp.
			Anaeromyces sp.
	Neocallimastigomycetes	Neocallimastigaceae	
Basidiomycota	Agaricomycetes	Polyporaceae	Trametes versicolor
	Agaricomycetes	Nidulariaceae	Cyathus stercoreus
	Agaricomycetes	Phanerochaetaceae	Phanerochaete chrysosporium
	Agaricomycetes	Ganodermataceae	Ganoderma resinaceum
	Agaricomycetes	Polyporaceae	Pycnoporus cinnabarinus

Table 9.2 Examples of various fungal species utilized for pretreatment and hydrolysis of lignocellulosic biomass

 Table 9.3
 An overview of different pretreatment modalities for lignocellulosic biomass with their merits and demerits

Pretreatment				
types	Mode	Features	Merits	Demerits
Chemical	Acid hydrolysis	Acidic breakdown of lignocellulosic constituents via continuous or batch process	Proper eradication of hemicellulose and lignin Less time-taking process	Generation of toxic compounds Might result in corrosion of reaction vessel
	Alkaline hydrolysis	Lysis of lignin and hemicelluloses by water-soluble bases	Weakens the crystallinity of cellulose Enhances the available surface area	Formation of residual substances Not so effective for softwood
	Organosolvent	At 170–190 °C	Proper hydrolysis of lignocellulosic components	Expensive
Physical	Crushing/ grinding/milling	Done at room temperature conditions	Low capital investment Easy and convenient method	Not so effective Very high energy demand in some samples

(continued)

Pretreatment				
types	Mode	Features	Merits	Demerits
Physicochemical	Wet oxidation	Temp: 150–200 °C for 30 min	Less energy input Promotes lignolysis	Expensive
	Ozonolysis	Done at room temperature for lignolysis	Lignin content is decreased Formation of toxic residues	Not cost-effective
	Carbon dioxide explosion	Takes place 50 °C at for 1 h 30 min	Surface area enhancement Almost negligible formation of inhibitors	Not applicable for those biomass which have lignin as their major constituent
	Ammonia recycle percolation method	160 °C for 15 min	Eradicates hemicellulose and lignin	Imperfect hydrolysis of hemicellulose
	Ammonia fiber explosion method	90 °C for 28 min	Efficient Surface area is exponentially increased	Hydrolysis is not at point
	Steam pretreatment	160–260 °C for 10 min	Cost-effective Automatic hemicellulose hydrolysis	Production of inhibitors Xylan damage
Biological	Some bacteria and fungi (major)	Brown-rot fungi cleave cellulose while white-rot cleaves both lignin (via lignolytic enzymatic machinery) and cellulose	Inexpensive Natural methodology	Low hydrolysis rate Takes energy

(continued)

Pretreatment				
types	Mode	Features	Merits	Demerits
Modern	Co-solvent-based lignocellulosic fractionation	Includes water, cellulosic solvent, and an organic solvent (involves reactions at moderate conditions)	High efficiency	-
	Co-solvent enhanced lignocellulosic fractionation	Sulfuric acid (dilute) and water act on cellulose	Cost-effective	-
	Extractive ammonia Pretreatment	Includes transformation of crystalline cellulose into soluble form	Effective method	_
	Pretreatment applying ionic liquids	1-Butyl-3-methyl imidazolium chloride is used	Inexpensive	High pollution
	Gamma- valerolactone pretreatment	A minimal harsh treatment of lignocellulosic biomass	High delignification rate Highly efficient methodology	Expensive
	Switchable butadiene sulfone pretreatment	Done in aqueous conditions to break cellulose	Efficient cellulose degradation	High water requirement Regular energy inputs needed
	Sulfite pretreatment to overcome recalcitrance of lignocellulose	Done to enhance enzymatic activity	Presence of reducing agents	Mostly done for woody biomass

 Table 9.3 (continued)

synergistic action of two or more pretreatment methods. In spite of that, there is a pressing need for finding a novel efficient methods or modifying existing methodologies for harnessing best results. The recent interventions of artificial intelligence or predictive modeling can open a new avenue for the designing, optimization, and process configuration for pretreatment technologies (Zheng et al. 2009).

4.2 Enzymatic Hydrolysis and Saccharification

The pretreatment of lignocellulose biomass is followed by enzymatic hydrolysis for further de-polymerization of cellulose and hemicellulose (Humbird et al. 2011) to monomeric sugars that will be subsequently used for ethanol fermentation. This

process requires either in situ production or utilization of commercial enzymes (cellulases and other enzymes) that enhance the capital cost for the bioethanol production. Besides that, the enzymes need to possess high activity, stability under reaction conditions including high temperatures, resistant to inhibitory molecules generated during the pretreatment process and finally possess a longer half-life. Extensive research has been done for increasing the stability, activity, and reducing the cost of enzymes in the last decade but availability at the cost of \$1.5-\$2.0 per kg has not reached economic viability. Till date, there is dearth of commercial enzymes that can withstand high-temperature applications for the hydrolysis as well as low accessibility to hemicellulose and lignin and susceptible to the inhibition by toxic compounds produced during pretreatment also stymied the complete and efficient hydrolytic conversion and demands high loading of enzymes. Therefore, various strategies like in situ production of hydrolytic enzymes from various microbial consortium described above section have been implemented as well as production of thermotolerant fungal enzymes has been tried. More recently, the discovery of novel non-hydrolytic enzymes (LPMOs) (Guerriero et al. 2016) polysaccharide monooxygenases add further dynamism in the hydrolysis process by alleviating the need of high loading of cellulase enzymes and reduction in the processing cost. The actual mechanism of action is still under investigation; however, the proposed mechanism is believed to be the oxidation of highly recalcitrant (Dien et al. 2003) crystalline regions of cellulose to aldonic acids, therefore provide more reducing/non-reducing ends for enzymatic hydrolysis. Though it was found that these enzymes are responsible for making less stable cellulase; therefore, more sophisticated studies are needed for their long-term commercial utility. Another major reason for slow hydrolysis rate is due to the mixture of hexoses and pentose sugar released during hydrolysis and most commonly ethalnogenic fungi Saccharomyces cerevisiae or bacteria Zymomonas mobilis cannot ferment pentose sugars (Seo et al. 2005) and the higher amount of sugars released cause substrate inhibition. Though methodologies have been developed for the conversion of pentose sugars into other molecules like furfural through chemical catalysis but that doesn't prove to be economically and technically sustainable process. Therefore, rigorous research is still centered on the co-fermentation of hexose and pentose sugars by utilizing the genetically engineered strains for bioethanol production.

4.3 Fermentation

The major impediment for the production of bioethanol from lignocellulosic biomass is fermentation. The process of conversion of hexose and pentose sugars into ethanol through various microorganisms is known as fermentation. Research studies speculated that the theoretical maxium yield of hexoses and pentoses is 0.51 L/ kg ethanol (Chovau et al. 2013) and 0.489 kg CO_2 /kg sugar and overall yield of ethanol becomes 0.719 and 0.736 L/kg of glucan (for hexoses) and xylan (for pentoses), respectively. A most popular and thoroughly investigated fungus, *S*.
cerevisiae has been explicitly utilized for the production of first-generation ethanol but suffers from the limitation of not having the pathway for pentose conversion. Therefore, the problem has been circumventing though more rigorous efforts for identifying other fungi and microbes that can co-ferment both C_5 and C_6 sugars at elevated operating temperatures or make the strains recombinant through genetic engineering that can metabolize pentose as well as hexose. However, the recent studies also suggested the possibility of the direct fermentation of cellobiose and cellodextrins (Kim et al. 2014) into ethanol also leads to progressive drive for the development of recombinant microbe for introducing cellodextrin transporter system from Neurospora crassa into S. cerevisiae. More recently, research was published for engineered yeast strain that can co-ferment cellobiose, xylose, and glucose for the bioethanol production (Zheng et al. 2018). Though engineered strains hold potential for metabolizing hexose as well as pentose sugars, low ethanol titers, and slow metabolism are the challenges that are still prevailing and need to be circumvented. Therefore, during enzymatic hydrolysis, some integrated methods for hydrolysis and fermentation are proposed.

5 Integrated Processes for Bioethanol Production Using Lignocellulosic Biomass

The basic steps described above for the bioconversion of lignocellulose to ethanol necessitates certain integrative strategies for combating the certain limitations of each process. The major four processes have been developed after pretreatment for the conversion of complex biomass into ethanol that includes (a) separate hydrolysis and fermentation (SHF), (b) simultaneous saccharification and fermentation (SSF) (Wyman 1994), (c) simultaneous saccharification and co-fermentation (SSCF) (Lynd et al. 2005), and (d) consolidated bioprocessing (CBP) (Zhao et al. 2018). The separate hydrolysis and fermentation (SHF) strategy is the most conventional two-stage process in which the pretreated biomass is cleaved into sugars in one reactor and fermentation of these sugars into ethanol has been performed in separate reactor. The method is advantageous in terms of carrying out the reactions at their optimum conditions but suffers from the limitation of high investment cost for maintaining two separate reactors as well as inhibition of fermentative microflora due to higher concentration of glucose. Therefore, to alleviate the inhibition of sugars and reduction in number of steps, two other approaches have been adopted in which the same reactor (Ballesteros et al. 2002) can be used for enzymatic saccharification and fermentation. Therefore, the sugars released from the biomass can be rapidly converted into ethanol that prevents the inhibition of sugars with faster hydrolysis rate and requires low enzyme loadings. Apart from that, the process requires less capital cost, easy operation conditions, and rapid conversion of ethanol leads to reduction in contamination. However, the major bottleneck of this process is the non-compatibility of the optimal conditions for enzymatic hydrolysis as well as fermentation. The higher temperatures usually 50-55 °C has been required for enzymatic hydrolysis while the optimum temperature for the growth of mesophilic fungi and bacteria is 28 °C and 37 °C, respectively, that makes this process less efficient as well as confers low ethanol yields (Sun and Cheng 2002). Though protein engineering approaches has been very well tried to reduce the optimum temperature for the enzymatic hydrolysis, the process has not been found to be economically viable; therefore, there is pressing demand for identifying thermotolerant strains for ethanol production at elevated temperature. The modification (Viñals-Verde et al. 2012) in the simultaneous saccharification and fermentation (SSF) has been more efficiently developed by the co-fermentation of both pentoses and hexoses formed during the pretreatment and hydrolytic processes of lignocellulosic biomass. This integrated technology is known as simultaneous saccharification and co-fermentation (SSCF). This process is accomplished either through addition of mixed cultures of yeast or addition of pure culture of fungi having the capability of utilizing both pentoses and hexoses.

However, the utilization of mixed culture leads to the generation of heterogeneous conditions in which the speedy growth was observed for hexose utilizing microorganisms as compared with the slow growth of pentose-utilizing microorganisms. Therefore, the latter approach of utilizing single microorganism is more viable option possessing the qualities of reduced cost and risk of contamination, less operation time as well as reduced substrate and product inhibition. Furthermore, the rigorous efforts have been pursued for consolidation of all the steps for bioethanol manufacturing in a one bioreactor utilizing the single microorganism that can produce the cellulolytic and ligninolytic enzymes, perform enzymatic saccharification and fermentation with the high yield of bioethanol. This strategy is known as direct microbial conversion (DMC) or consolidated bioprocessing (CBP) (Xu et al. 2009). A flowchart depicting CBP is shown in Fig. 9.3 (Alzate and Toro 2006; Stöcker 2008; Hamelinck et al. 2005; Aden et al. 2002). The success of this process lies on stringent selection of the microorganisms as no microorganisms in its native state possess all the desired properties till date. Therefore, the genetic engineering approaches have been well applied for enhancing the potential of the employed microorganisms. These efforts have been categorized the CBP into two subdivisions CBP I and CBP II. Presently, bacterial as well as fungi especially variety of yeast has been explored for this process. Therefore, CBP I involves the engineering of cellulase producing microbes to become ethanologenic and CBP II involves the engineering of ethanologenic microbes to be cellulolytic. In CBP I, the successful bioethanol productivity has been assessed through anaerobic cellulolytic thermophiles from the genus Clostridium (Roberts et al. 2010); however, the inability to metabolize xylose, low ethanol tolerance as well as production of by-products like lactate and acetate limits its applicability at commercial level. It is worth mentioning that metabolic engineering and directed evolutionary approaches have been rigorously improved ethanol titers by blocking the production of major by-products. At present, the inherent potential of various species of cellulolytic fungi like Aspergillus, (Bhardwaj et al. 2019), Fusarium, Rhizobium, and Trichoderma have been looked at as CBP I candidates. As for the second strategy, the primary concerns are expression and secretion of functional cellulases (Sukumaran et al. 2009) in ethanologenic



Fig. 9.3 The comprehensive strategy for the processing of bioethanol from lignocellulosic biomass

species, particularly *S. cerevisiae*, which has been engineered with genes encoding glycoside hydrolases including cellulases and hemicellulases through cell surface display techniques. Unfortunately, expression of the cellobiohydrolases (Van Tilbeurgh et al. 1986) (CBH I and CBH II) from *Trichoderma reesei* is generally poor (Meng et al. 2018); therefore, it's worth considering that engineering the species with more other enzymes or pathways required by the efficient production of bioethanol.

6 Recent Molecular Interventions for Improvement of Mycodegradation for the Bioethanol Production

The efficiency and abundant availability of lignocellulosic biomass is enhancing their potential utilization of bioethanol production. The scarcity and rising cost of non-renewable fuels and increasing carbon footprint lead to the dramatic increments in utilizing microbial factories for the bioconversion of lignocellulosic



Fig. 9.4 Recent molecular approaches for the engineering of various fungi for improvement in bioethanol production

biomass to bioethanol. The most efficient industrial workforce for the production of bioethanol was documented for the utilization of Saccharomyces cerevisiae as a biological tool. The major limitation of S. cerevisiae (Argueso et al. 2009) is the inability to ferment a wide range of substrates like arabinose, xylose, and glycerol as well as could not grow at higher temperature. The current "omics" and molecular interventions like advent of synthetic biology, system biology, and metabolic engineering revolutionized the field of bioenergy production, and these efforts have already been conferred in S. cerevisiae. The combined strategy of various molecular approaches for the improvements in fungi (Madhavan et al. 2017) is depicted in Fig. 9.4. The successful production of homologous and heterologous products has been mediated through promoter selection and productivity has been enhanced through promoter engineering. In S. cerevisiae, the development of synthetic promoter leads to the dynamic utilization of this organism at commercial level. Thus, the fine and tunable control of metabolic pathways has been done through promoter engineering approaches that can be performed through engineering core promoter elements through random or site-directed mutagenesis and modifications in the upstream regulatory sequences and transcription factors. However, the major C₅ and C_6 sugars utilization pathway is depicted in Fig. 9.5 (Aristidou and Penttilä 2000).

More recently, construction of hybrid promoter may open up the possibilities to explore the promoter sequences of other organisms and their introduction in the desired host for the development of highly active synthetic promoters. The bioethanol productivity has been enhanced by the development of hybrid promoter by combining β -galactosidase (lac4) promoter of *K. lactis* with the core promoter



Fig. 9.5 The depiction of C5 and C6 sugars utilization pathway of microorganisms

sequences of cellobiohydrolase 1 (cbh1) from Trichoderma reesei. The advent of (Borodina and Nielsen 2014) CRISPR-Cas (Clustered Regularly Interspaced Short Palindromic Repeat) has led to paradigmatic shift in regulation and control of the expression of genes inside the yeast. This system utilizes the RNA-guided control of gene through nuclease-deficient Cas9 (dCas9) for gene silencing. The technique seems to be promising in context of generation of rapid and efficient double-stranded breaks and multiple targeted integrations in yeast. The component histone deacetylase (Sin3p homolog) is strongly bound with Mxi1 protein after the attachment of Cas9 that further down-regulate the TEF promoter of S. cerevisiae for enhancing bioethanol production. On the other hand, it is worth mentioning that besides the extensive utilization of conventional yeasts for the bioethanol production, non-conventional yeast systems possess remarkable credentials like thermotolerance (Edgardo et al. 2008), ethanol tolerance, genetic diversity, inhibitor tolerance, etc. The application of system biology and synthetic biology in these biological systems expands their horizons for the efficient production of bioethanol. The current examples of non-yeast systems that have been explored yet include Kluyveromyces lactis, Pichia pastoris (Soccol et al. 2010), Hansenula polymorpha (Ryabova et al. 2003), Yarrowia lipolytica (Tsigie et al. 2013), etc. It was observed that conventional yeast metabolize glucose by fermentative mode (Crabtree positive) (Azhar et al. 2017) and non-conventional yeasts metabolize glucose by oxidative pathway (Crabtree negative) (Souto-Maior et al. 2009). Therefore, research studies were exhaustively tried for the conversion of Crabtree-negative yeasts to Crabtree-positive systems for enhancing the efficiency of bioethanol production. Recently, the metabolic engineering has been successfully applied for shifting the metabolic pathway from respiratory to fermentative in Kluyveromyces lactis. A null mutant was constructed for the gene encoding for mitochondrial internal dehydrogenase that displayed higher ethanol productivity by depicting non-susceptibility towards oxidation of exogenous NADH.

In a similar line, the metabolically engineered strain of Crabtree-positive yeast (Dekkera bruxellensis) (De Souza Liberal et al. 2007) has been shown to produce high yield of ethanol. In this yeast system, the overexpression of key enzyme of fermentative pathway, i.e., alcohol dehydrogenase has been performed under the control of highly active TEF1 promoter that enhanced the ethanol productivity 1.7 times more than wild-type strain. In a similar line, the productivity of bioethanol has been significantly enhanced under various physiological stress conditions like high temperature and osmotic stress. The non-conventional yeast systems like some strains of has been Kluvveromvces marxianus (Ballesteros et al. 2004) depicted high temperature tolerance as compared with conventional yeast systems. The thermotolerant strains of Kluyveromyces marxianus NIRE-K1 and NIRE-K3 (Arora et al. 2015) were found to ferment both xylose and glucose at high temperature approximately above 45 °C and displayed high ethanol productivity of about 39.12 and 43.25 g/L, respectively. Similarly, thermotolerant variety of Kluyveromyces sp. IIPE453 has been screened for enhanced production of 0.43 g/g bioethanol at 50 °C from glucose and xylose-rich bagasse hydrolyzates as a lignocellulosic feedstock. In addition to that, non-conventional yeast system including Ogataea polymorpha (Hasunuma and Kondo 2012) has been found to produce bioethanol by fermenting xylose at 45 °C. The osmotolerant strain Zygosaccharomyces rouxii was also identified and screened for producing high concentration of ethanol in the presence of 90% glucose concentrations and 3M NaCl. The resistance towards osmotic stress in terms of high sugar and salt has been attributed due to the presence of unique transporters in plasma membrane. Furthermore, the substrate utilization capacities of non-conventional yeast have also been modified through system biology and metabolic engineering approaches. The capacity of fermenting xylose and starch in the thermotolerant variety of H. polymorpha has been developed by the heterologous expression ofgenes SWA2 and GAM1 (Voronovsky et al. 2009) encoding amylolytic and xylanolytic enzymes isolated from the yeast Schwanniomyces occidentalis. The heterologous expression has been mediated under constitutive promoter of glyceraldehyde-3-phosphate dehydrogenase gene in the H. polymorpha. In a similar line, the less capacity of xylose utilization of K. marxianus has been enhanced by altering the pathway for the conversion of xylose to xylitol followed by ethanol at elevated temperature by replacement of native xylose reductase gene with foreign gene of xylose reductase of P. stipitis (Nigam 2001). Furthermore, the various promoters like TEF1, ADH1p (Almeida et al. 2008), Xy11p (Ponniah et al. 2017), and ADH2p (Salusjärvi et al. 2003) have been screened and engineered for high ethanol tolerance and reducing the Crabtree effect of Pichia stipitis. Thus, extensive efforts (Shaw et al. 2008) have been done for eliciting the higher production of bioethanol in conventional and non-conventional yeast system; however, there is pressing need to make them suitable industrial workforce for (Sticklen 2008) commercial utilization.

7 Current Bottlenecks

The potentiality of bioethanol production from lignocellulose feedstock has been increasingly realized with respect to attaining sustainability in bioenergy demands as well as environmental safety. Despite possessing the great potential, the largescale production of bioethanol from these magnificent resources has not yet been demonstrated as an economically sustainable option. The various technological hurdles have been realized from processing to pretreatment followed by separation through distillation process. The recalcitrant nature and structural complexity is the major impediment for the efficient production of bioethanol. This recalcitrance comes naturally in plants from the evolutionary perspective for their survival from differential environmental barriers therefore to break this natural mechanism is technologically a cumbersome task for the scientific community. The inefficiency of fermentation of xylose is another major challenge for the efficient production of bioethanol. The toxicity generated due to the production of various inhibitory molecules during various pretreatment strategies has been a technologically challenging aspect of bioethanol production. Therefore, the finding of ideal pretreatment strategies is still the major concern for its commercialization. The bioconversion process through the selection of ideal microorganisms also affects the production efficiency and cost. The recent technological developments in genetic engineering approaches opens up the avenue for engineering of microbes especially in conventional and non-conventional yeast in a desired manner but still suffers from limitations for commercial applicability. Therefore, engineering microbes for the economic production of bioethanol has major concerns that include their utilization on wide substrate ranges, efficient and simple framework, and easy downstream processes. Moreover, the inhibitor tolerance, thermotolerance, fuel tolerance, control of redox balance also posits significant consideration for the bioethanol production. The nonhomologous recombination pathway, as well as longer expression cassettes followed by larger cloning steps and PCR reactions also limits the industrial applicability of non-yeast systems at industrial scale. Therefore, rigorous research efforts are still required for alleviation of these challenges for the viable production of bioethanol.

8 Future Research Priorities

Inspite of the above-mentioned bottlenecks present in bioethanol production and its commercialization, the tremendous potential of bioethanol production from the lignocellulosic biomass still fascinates scientific community for the development of efficient and sustainable process technology for its industrial scale production. The continuous demand around the globe for the biofuels and bioethanol production has been increased (Limayem and Ricke 2012; Alvira et al. 2010; Sarkar et al. 2012) over time that can be represented in (Fig. 9.6a, b).



Fig. 9.6 The global share of bioenergy production (a) Major countries involved in biofuels production. (b) Major countries involved in bioethanol production

There is pressing need for identifying novel fungi that can produce cellullase enzymes of higher efficiency and specific activities. The efforts will be done for enhancing the fermentative capacity of different fungus species that has the ability to degrade biomass efficiently but less production efficiency of ethanol. Among the fungal diversity, yeast serves as the most promising workforce for the production of bioethanol. The extensive metabolic engineering of yeast through "promoter-based gene assembly and simultaneous overexpression" (PGASO) doesn't circumvent the toxicity of products during bioethanol production that leads to disruption of cell membrane due to the inhibition of ATP-producing pumps. Therefore, future studies will focus on the engineering of non-conventional, yeast that imparts better selectivity towards inhibitor tolerance, thermotolerance, and product tolerance. The system biology approaches and synthetic biology tools for improving various fungi will decrease the cost through reduction in the number of expression constructs. The transcriptomic data will help to comprehend the regulatory and rate-limiting steps in biosynthetic pathway for ethanol production. The bioinformatics tools will be helpful for making alteration in metabolic pathway by introducing novel catalytic enzymes or making knockouts for increasing biofuel production. Genome-scale metabolic models (GEMs) can be used for species correlation, metabolic flux design, identification of transcription factors and genes, and metabolic blueprint of the cell. The recent interventions of synthetic biology tools via RNA interference, codon optimization, and CRISPR-Cas technology can be utilized for enhancing the biofuel production. The construction of genetic circuits as well as finding novel metabolic pathways and their synergistic combination will be done by ligation-free assembling techniques and BioBricks softwares. Therefore, finding and constructing a genetic switch that can respond towards various environmental cues will definitely enhance the bioethanol production. The future research will also focus on the application of synergistic microbial communities that can develop the robust synthetic fungal-bacterial consortia for the efficient production of bioethanol.

9 Conclusion

This chapter entails about the importance of bioethanol that has proven as one of the most promising and eco-friendly resources for the bioenergy. The production of bioethanol has been significantly enhanced by utilizing sugarcane and corn as feed-stocks in Brazil and the USA but these resources could not fulfill the accelerating demand of bioethanol. Therefore, the lignocellulosic biomass serves as sustainable resources for bioethanol production and requires four major steps for bioconversion. The remarkable credentials have been manifested by various fungi for the fermentation of both C_5 and C_6 sugars. The high ethanol productivity has been still impeded with various technological barriers. The recent interventions of molecular approaches and their successful implications in various fungi led to the paradigmatic shift in enhancing the bioethanol productivity and still thriving for more efficiency. Though the cellulosic bioethanol production plants are increasing continuously, the continued research efforts are still required for the development of viable process technology for their sustainable utilization at a profitable cost.

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Chapter 10 Multiple Factors Influencing the Strategy of Lignin Mycodegradation



Gautam Anand, Sangeeta Yadav, and Dinesh Yadav

1 Introduction

The term "lignin" is derived from the Latin word "lignum," meaning wood. It is the most abundant aromatic compound on earth and is the second (after cellulose) most abundantly present natural polymer. It accounts for 15–30% of the lignocellulosic biomass (Yao et al. 2017). Of all naturally produced organic chemicals, lignin is probably the most recalcitrant. Together with hemicellulose, it forms strong adhesive for cellulose fibers (Grzegorz et al. 2017). This property of lignin is consistent with its biological functions, i.e., to give vascular plants the rigidity and to protect their structural polysaccharides from attack by other organisms. The conversion isolated cellulose or hemicellulose into monosaccharides is known to be a relatively simple process. Many cellulolytic as well as ligninolytic fungi make use of hydrolases which are capable of producing monosaccharides from polysaccharide components of wood. However, when these components are complexed with lignin, they are resistant to hydrolytic breakdown. Therefore, lignin appears to inhibit hydrolytic activity.

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2 Lignin Structure and Biosynthesis

Lignin is a complex, amorphous, three-dimensional aromatic polymer. Lignin is formed in vascular plant cell walls by the oxidative coupling of several related phenvlpropanoid precursors: conifervl alcohol, sinapyl alcohol, and *p*-coumaryl alcohol (Fig. 10.1). Because of the different monomers, lignin can be divided into three types: syringyl lignin polymerized by syringyl propane, guaiacyl lignin polymerized by guaiacyl propane, and hydroxy-phenyl lignin polymerized by hydroxyphenyl propane. Usually, gymnosperm mainly contains guaiacyl (G) lignin; the dicotyledon mainly contains guaiacyl-syringyl (GS) lignin; the monocotyledon mainly contains guaiacyl-syringyl-hydroxy-phenyl (GSH) lignin (Wei and Song 2001). G lignin is chiefly formed through dehydrated oligomerization of coniferyl alcohol, and its structure is homogeneous. Most lignin in softwood belongs to G lignin, which is copolymerized by guaiacyl. GSH lignin is the result of the dehydrated oligomerization of coniferyl alcohol and sinapyl alcohol; the content of lignin is 17–23%. The ratio of syringyl propane to guaiacyl propane is 0.5:0.1; it also contains 7–12% ester groups. p-Coumaryl alcohol in it is linked to lignin in the form of ester (Gao and Tang 1996).

Peroxidases or laccases in the plant cell wall oxidize these monomers by one electron, yielding transient resonance-stabilized phenoxyradicals that then polymerize in a variety of configurations (Adler 1977; Higuchi 1990). The molecular weight of lignin is difficult to determine because lignins are highly polydisperse materials (Argyropoulos and Menachem 1997). Lignin is covalently associated with hemicelluloses in the cell wall via numerous types of linkage (Fig. 10.2). Among the most important are ether bonds between the benzylic carbon of lignin and the carbohydrate moiety, ester bonds between the benzylic carbon of lignin and uronic acid residues, and lignin-glycosidic bonds.

Recent studies show that lignin can incorporate many more monolignols than the traditional three basic units (Vanholme et al. 2008), e.g., acetylated lignin units have been identified in non-woody plants (Martínez et al. 2005). In graminaceous plants,



Fig. 10.1 Three precursors of lignin



Fig. 10.2 Polymeric structure of lignin

hydroxycinnamic acid residues are frequent in the lignin and are attached to hemicelluloses via ester linkages. The matrix of lignin and hemicellulose encrusts and protects the cellulose of the plant cell wall (Jeffries 1991).

Lignin polymer contains characteristic methoxyl groups, phenolic hydroxyl groups, and some terminal aldehyde group. Most of the phenolic hydroxyl groups are bonded to neighboring phenylpropane units. Carbonyl groups and alcoholic hydroxyl groups are introduced in the final lignin structure during dehydrogenative polymerization process. Also, in some of the wood species substantial alcoholic hydroxyl groups are esterified with *p*-hydroxybenzoic acid or *p*-hydroxycinnamic acid.

Biosynthesis of monolignols is carried through the phenylpropanoid pathway. In the shikimate pathway, glucose obtained by photosynthesis is first converted to shikimic acid. This shikimic acid is converted into the final products of the pathway, i.e., phenylalanine and tyrosine through prephenic acid. Phenylalanine and tyrosine are starting components for the cinnamic acid pathway. After a set of reactions, such as deamination, hydroxylation, methylation, and reduction, three monomers of lignin are finally synthesized (Lin et al. 2003; Geng et al. 2003).

The rate-limiting step, controlling lignin synthesis and the pathway from which carbon can go into lignin biosynthesis is catalyzed by Cinnamoyl CoA reductase (CCR). Reduction of coniferaldehyde, another redox reaction in the lignin synthesis process, is catalyzed by cinnamyl alcohol dehydrogenase (CAD). CAD is located in tissues with precipitation of G lignin. Sinapyl alcohol dehydrogenase (SAD), ferulic acid 5-hydroxylase (F5H), and bispecific caffeic acid/5-hydroxyferulic acid *O*-methyltransferase (COMT) are immunolocalized in cells and tissues that have S lignin. It is believed that the last step of redox reactions of different types of lignin

go through different pathways which are catalyzed by different enzymes (Lin et al. 2003). Two-step methylation reaction at the 3' and 5' site is an important step in lignin monomer synthesis. This step is carried out by COMT and CCoAOMT (caffeoyl CoA 3-Omethyltransferase) which are methylases on two different substrate levels.

3 Fungal Degradation of Lignocellulose

Ligninolytic fungi are not able to use lignin as their sole source of energy and carbon. Instead, they depend on the more digestible polysaccharides in lignocellulosic substrates, and the primary function of ligninolysis is to expose these polysaccharides so that they can be cleaved by fungal cellulases and hemicellulases. Oxidative coupling of lignin aromatic alcohol monomers creates a complex structure in lignin which is highly recalcitrant to degradation (Wong 2009). By linking to both hemicelluloses and cellulose, lignin acts as a barrier to any solutions or enzymes and prevents penetration of lignocellulolytic enzymes to the interior lignocellulosic structure. Not surprisingly, of the components of lignocellulosic material, lignin is the most resistant to degradation (Sánchez 2009; Himmel et al. 2007). Although lignin resists attack by most microorganisms, basidiomycetes white-rot fungi are able to degrade lignin efficiently (Wong 2009; Abbas et al. 2005). In most fungi that have been examined, ligninolysis occurs during secondary metabolism, i.e., under nutrient limitation. With this approach, the fungus avoids synthesizing and secreting metabolically expensive ligninolytic agents when substrates more accessible than lignocellulose are present. The limiting nutrient for fungal growth in most woods and soils is probably nitrogen, and most laboratory studies of ligninolytic fungi have been done in nitrogen-limited culture media (Kirk and Farrell 1987). However, a few ligninolytic fungi, e.g., some species of Bjerkandera, are ligninolytic even when sufficient nitrogen is present (Kaal et al. 1993).

Lignin degrading fungi produce extracellular enzymes to break lignin. These fungi are widespread and include species from the ascomycetes and basidiomycetes phyla such as white-rot and brown-rot fungi. In addition, a few anaerobic species are also found to be able to degrade cellulose in the gastrointestinal tracts of ruminant animals (Ljungdahl 2008; Yoon et al. 2007). Biomass degradation by these fungi is performed by complex mixtures of cellulases (Bayer et al. 1998), hemicellulases (Ljungdahl 2008), and ligninases (Weng et al. 2008; Sánchez 2009), reflecting the complexity of the materials. In nature, degradation of cellulosic biomass is performed by mixtures of hydrolytic enzymes collectively known as cellulases. Many microorganisms, including fungi and bacteria, have been found to be capable of degrading cellulose and other plant cell wall fibers contributing significantly to the decay of lignocellulosic residues in nature. The degradation of lignin by different fungi is briefly discussed:

4 White-Rot Fungi

White-rot fungi are a heterogeneous group of fungi classified into the basidiomycota. They are most efficient organisms capable of degrading lignin and the related litter. More than 90% of all wood-rotting basidiomycetes are of the white-rot type (Gilbertson 1980). Different white-rot fungi vary considerably in the relative rates at which they attack lignin. Some of them simultaneously attack lignin, hemicellulose, and cellulose, whereas some other white-rot fungi preferentially work on lignin in a selective manner. Their purpose is to decompose the lignin in wood so that they can gain access to the cellulose and hemicelluloses that are embedded in the lignin matrix. Usually, syringyl (S) units of lignin are preferentially degraded, whereas guaiacyl (G) units are more resistant to degradation. Many white-rot fungi colonize cell lumina and cause cell wall erosion. Eroded zones coalesce as decay progresses and large voids filled with mycelium are formed. This type of rot is referred as non-selective or simultaneous rot. Calcium oxalate and MnO2 accumulate when the decay proceeds (Blanchette 1995). Some white-rot fungi degrade lignin in woody plant cell walls relatively to a higher extent than cellulose, and they are called selective white-rot fungi. In nature, they may cause white-pocket or white-mottled types of rot, e.g., Phellinus nigrolimitatus (Blanchette 1995). There are also fungi, e.g., the tree pathogen Heterobasidion annosum, that are able to produce both types of attack in the same wood (Eriksson et al. 1990). Typical examples of such fungi are Ganoderma applanatum and Heterobasidion annosum. Because fungi selectively degrading lignin are considered the most promising fungi for applications in the pulp and paper industry, the search among these fungi has attained a considerable interest. However, the ratio lignins, hemicelluloses, and celluloses decayed by a selected fungus can differ enormously, and even different strains of the same species, e.g., of Phanerochaete chrysosporium and Ceriporiopsis subvermispora, may behave differently on the same kind of wood. Several screening studies to find suitable fungi for biopulping of wood or straw have revealed fungi that, under certain conditions, degrade lignin preferentially to cellulose. Such lignin-selective fungi are, e.g., P. chrysosporium, C. subvermispora (Otjen et al. 1987; Eriksson et al. 1990), Pycnoporus cinnabarinus (Ander and Eriksson 1977), Phlebia radiata (Ander and Eriksson 1977), Phlebia tremellosa (syn. Merulius tremellosus) (Ander and Eriksson 1977; Eriksson et al. 1990), Phellinus pini (Eriksson et al. 1990), and Dichomitus squalens (Eriksson et al. 1990).

White-rot fungi produce a number of extracellular enzymes that directly attack lignin, cellulose, and hemicellulose of the plant cell wall to decompose it. These enzymes include laccases and peroxidases, such as lignin peroxidase (LiP), manganese peroxidase (MnP), and versatile peroxidase (VP). Laccases and peroxidase enzymes can also cause lignin degradation through low molecular weight-free radicals such as OH, depolymerize the phenolic and non-phenolic lignin polymer, and mineralize the insoluble lignin. Basic research on lignin degradation, e.g., its

mechanisms, physiology, enzymology, and molecular biology, has been mainly carried out with the corticoid fungus *P. chrysosporium* (Kirk and Farrell 1987; Eriksson et al. 1990). After more and taxonomically different fungi had been studied in more detail, it was revealed that both the physiological conditions for lignin degradation and the enzyme systems expressed are fungus-specific and differ from those found in *P. chrysosporium*. Differences may be connected to the taxonomic position and/or ecology of the fungi, e.g., substrate specialization (hardwood, softwood, or certain wood species, heartwood or sapwood), the stage of degradation, etc.

5 Brown Rot

Brown-rot fungi are basidiomycetes that degrade wood to yield brown, shrunken specimens that typically exhibit a pattern of cubical cracks and easily disintegrate upon handling. Only a small proportion, roughly 70%, of all wood decay basidiomycete species falls into this group, which occurs most frequently on gymnosperm wood. In contrast to white-rot fungi, brown-rot fungi, such as Postia placenta, Laetiporus portentosus, Piptoporus betulinus, and Gloeophyllum trabeum, can degrade wood carbohydrates, but not oxidized lignin. Brown-rot fungi mainly decompose the cellulose and hemicellulose components in wood, but they can also modify the lignin to a limited extent (Eriksson et al. 1990). They have been much less investigated than white-rot fungi in spite of their enormous economic importance in the destruction of wood. Lignin degradation by brown-rot fungi mainly involves non-enzymatic oxidation reactions producing hydroxyl radicals via Fenton chemistry (Kirk et al. 1991; Kerem et al. 1998, 1999). Brown-rot fungi partially oxidize lignin via aromatic ring demethylation. During this process, the phenolic hydroxyl content of the reaction mixture increases due to partial oxidation and partially due to the addition of new carboxyl and carbonyl groups (Hatakka and Hammel 2011).

Brown-rotted wood is dark, shrink, and typically broken into brick-shaped or cubical fragments that easily break down into brown powder (Blanchette 1995). The brown color indicates the presence of modified lignin in wood. During brown rot, the hemicelluloses in wood are degraded most rapidly, after which virtually all of the cellulose is removed, leaving behind a complex, aromatic ring-containing polymer derived from the original lignin. To some extent, brown-rot fungi have similar degradative capabilities and pathways as white-rot fungi. Both wood decay mechanisms rely on radical formation, low pH, and the production of organic acids. They cause increased alkali solubility of lignin, and the decay is enhanced by high oxygen tension, all of which indicate a crucial involvement of radicals, especially in the early stages of decay (Kirk 1975).

6 Factors Affecting Lignin Degradation

Lignin degradation, owing to its complex structure and features, is a complicated process. Apart from type of fungi, many different environmental factors influence lignin degradation. The environmental factors affect growth and metabolism of fungi thereby influencing lignin degradation. Moisture content, temperature, pH, aeration, carbon, and nitrogen sources are important parameters that affect fungal growth (Yang et al. 1980; Leisola et al. 1984). Among different microbes, white-rot fungi are the most efficient organisms for lignin degradation. Basidiomycetes govern the list among different white-rot fungi. Expression of P. chrysosporium genes are strongly influenced by nitrogen and carbon limitation. Regulatory elements present in the promoter regions of genes encoding lignolytic enzymes play an important role in transcriptional activation (Cohen et al. 2001). Under aerobic environment P. chrysosporium can reduce 1 g of different separated lignins in 2 days. The fungus is able to degrade lignin efficiently under 30% CO₂ and 10% O₂ concentrations. Addition of 0.12% nitrogen has been shown to increase lignin degradation. The fungus uses lignin, hemicelluloses, and cellulose as substrate. The lignin reduction happens at the end of primary growth by cooperation of other metabolism-like nitrogen (Kang et al. 2013; Leisola et al. 1984; Yang et al. 1980; Hatakka and Hammel 2011; Hatakka 2005). The fungi use oxidation reactions which decrease methoxy, phenolic, and aliphatic content of lignin. These reactions also cleave aromatic rings and form new carbonyl groups. When it comes to mixed microbial population, soft-rot fungi can efficiently degrade lignin. They are better adapted to variations in temperature, pH, and oxygen content (Hamed 2013; Gupta and Verma 2015; Gupta et al. 2016).

pH plays a prominent role in the growth of fungi and it is very difficult to control an optimum pH when the fungi growing on a solid-state substrate. Production of lignolytic gene is influenced by the initial pH of the medium. At pH 4.0–5.0, most of the white-rot fungi grow optimally and also the substrate acidity decreases their growth (Patel and Gupte 2016).

The optimum temperature during biological pre-treatment varies with the type of fungi used. Most of the white-rot ascomycetes fungi grow best around 39 °C while the white-rot basidiomycetes grow optimally around 25 and 30 °C. The heat generated during fungal growth develops temperature gradients in solid-state media. The accumulated heat can inhibit fungal growth and metabolism. Variations in temperature employed are due to the differences in fungal strain, physiology, and type of substrate (Rouches et al. 2016).

Inoculum concentrations have significant role in lignin degradation. The time required for the colonization of the substrate is affected by the type and amount of inoculums. Fungal spores are the commonly used inoculum. Mostly, it is found that larger quantity of inoculum leads to shorter time for colonization of the substrate (Van Kuijk et al. 2015).

High substrate concentrations have to be used for biological pre-treatment to make the process economically viable. When dry substrate is used, it is found that concentration of inhibitors increases and this may influence the degradation process. Hence, pre-treatment should be carried in conditions to minimize the generation as well as accumulation of inhibitory compounds. Initial moisture content is essential for the establishment of microbial growth in the biomass. Initial moisture content critically affects the fungal growth and enzyme production and significantly affects lignin degradation (Rouches et al. 2016).

Aeration also has significant role in fungal degradation of lignin. Aeration has many functions which are including oxygenation, CO₂ removal, heat dispersion, humidity conservation, and also dispersal of volatile combinations produced during metabolism. It also affects production of enzymes that are required during lignin metabolism. Since lignin degradation is an oxidative process, oxygen availability is important for ligninase activity of white-rot fungi. However, high oxygen content has also been shown to slow down lignin degradation process in *P. chrysosporium*. High aeration could improve delignification rate and hence controlled aeration is essential for improvement of biological pre-treatment (Salvachua et al. 2011; Leisola et al. 1984).

7 Steps in Lignolysis

Lignin degradation by the fungi involves both depolymerization and cleavage of aromatic ring. Extracellular enzyme secreted by fungi oxidizes lignin in the following steps:

- 1. Oxidation of β -O-4 linkages to arylglycerol compounds
- 2. Aromatic rings cleavage by the α -ketoadipate pathway
- 3. Formation of cyclic carbonate structures.

8 Enzymes Involved in Lignin Degradation

Lignin does not contain hydrolyzable linkages, which means that the enzymes must be oxidative. Ligninases can be classified as either phenol oxidases (laccase) or heme peroxidases, lignin peroxidase (LiP), manganese peroxidase (MnP), and versatile peroxidase (VP) (Martínez et al. 2005). In general, laccases use molecular oxygen as electron acceptors while peroxidases use hydrogen peroxide as a cosubstrate. These enzymes can directly attack lignin, cellulose, and hemicellulose of the plant cell wall to decompose it. Some of these enzymes and their isoenzymes gene have been investigated and characterized. For example, the genome of *P. chrysosporium* contains ten LiP and five MnP genes (Martinez et al. 2004). Four laccase genes have been recognized in the plant pathogenic fungus *Rhizoctonia solani* (Wahleithner et al. 1996). Although LiP is able to oxidize the non-phenolic part of lignin (which forms 80–90% of lignin composition), it is absent from many

Enzyme	Prosthetic	Substrate	Mediators	Optimum pH	Reaction
Laccase	Four Cu atoms	Ortho and para- diphenols, aminophenol, polyphenols, polyamines, lignin, aryl diamines	Hydroxyben- zotriazole, syringaldazine, phenols, aniline	2.0–10.0	4 benzenediol + $O_2 = 4$ benzosemiquinone + 2 H_2O
Lignin peroxidase	Heme	Halogenated phenolic compounds, polycyclic aromatic compounds	Veratryl alcohol	1.0–5.0	1,2-bis(3,4- dimethoxyphenyl)propane- 1,3-diol + $H_2O_2 = 3,4-$ dimethoxybenzaldehyde + 1-(3,4-dimethoxyphenyl) ethane-1,2-diol + H_2O
Manganese peroxidase	Heme	Lignin and other phenolic compounds	Thiols, unsaturated fatty acids, organic acids as chelators, Mn ³⁺	2.5-6.8	$2Mn(II) + 2H^+ + H_2O_2 =$ $2Mn(III) + 2H_2O$
Versatile peroxidases	Heme	Phenolic and non-phenolic aromatic compounds	Veratryl alcohol, compounds similar to LiP and MnP mediators	3.0-5.0	Donor + H_2O_2 = oxidized Donor + $2H_2O$

 Table 10.1
 Features of fungal lignolytic enzymes

lignin degrading fungi (Wang et al. 2008). Also, oxidative ligninolytic enzymes are too large to penetrate into the wood cell wall micropores (Srebotnik et al. 1988). Hence, prior to the enzymatic attack, low-molecular weight diffusible reactive oxidative compounds initiate changes to the lignin structure (Srebotnik et al. 1988; Tanaka et al. 1999). A comparative list of features of fungal lignolytic enzymes is given in Table 10.1.

8.1 Laccase

Laccase is a copper-containing oxidase that uses molecular oxygen to oxidize various aromatic and nonaromatic compounds through a free radical catalyzed reaction mechanism. Most white-rot fungi typically produce laccase (Bollag and Leonowicz 1984), and the enzyme is common also in higher plants and in other fungi. Laccases couple the electron reduction of dioxygen into two molecules of water with the oxidation of a vast variety of substrates, such as phenols, arylamines, anilines, thiols, and lignins. Four copper ions in their catalytic center mediate the redox process. The oxidation reactions catalyzed by laccases lead to the formation of free radicals which act as intermediate substrates for the enzymes (Thurston 1994; Messerschmidt and Huber 1990; Ferraroni et al. 2007). These mediators can leave the enzyme site and react with a broad range of high-redox potential substrates and thus create non-enzymatic routes of oxidative polymerizing or depolymerizing reactions. Laccases have been found and studied in white-rot fungi, such as *Pleurotus ostreatus* D1 (Pozdnyakova et al. 2006), *Lentinus tigrinus* (Ferraroni et al. 2007), *Cerrena unicolor* strain 137 (Michniewicz et al. 2006), *Trametes* sp. strain AH28-2 (Xiao et al. 2003), *Trametes pubescens* (Shleev et al. 2007), and *Cyathus bulleri* (Salony et al. 2006). Also, ascomycetes such as *Melanocarpus albomyces* (Hakulinen et al. 2006), *Magnaporthe grisea* (Iyer and Chattoo 2003), *Myrothecium verrucaria* 24G-4 (Sulistyaningdyah et al. 2004), and *Neurospora crassa* (Germann et al. 1988) were also found to produce laccases.

8.2 Lignin Peroxidase

Lignin peroxidase (LiP, EC 1.11.1.14) is a glycosylated enzyme containing heme protein with an iron protoporphyrin prosthetic group that requires hydrogen peroxide (H_2O_2) to catalyze the oxidation of non-phenolic lignin units and mineralize the recalcitrant aromatic compounds. LiPs are heme-containing glycoproteins and play a central role in the biodegradation of the cell wall constituent, lignin (Piontek et al. 2001). LiPs catalyze the H_2O_2 -dependent oxidative depolymerization of a variety of non-phenolic lignin compounds (diarylpropane), β-O-4 non-phenolic lignin model compounds and a wide range of phenolic compounds such as guaiacol, vanillyl alcohol, catechol, syringic acid, and acteosyringone (Wong 2009). Lignin oxidation takes place by electron transfer, non-catalytic cleavages of various bonds and opening up of aromatic ring. Typically for LiP activity, the amino acid residue needed is a tryptophan and is assumed to long-range electron transfer from a protein radical at the surface of the enzyme, which would act as the substrate oxidizer, to the heme cofactor (Ruiz-Dueñas and Martínez 2009). This helps in oxidization of bulky lignin substrate. They have been found in many white-rot fungi such as Phanerochaete chrysosporium, Trametes versicolor, and Bjerkandera sp. (Kirk and Farrell 1987; Kaal et al. 1993; Orth et al. 1993). LiPs resemble other peroxidases such as the classical, extensively studied enzyme from horseradish, in that they contain ferric heme and operate via a typical peroxidase catalytic cycle (Kirk and Farrell 1987). However, LiPs are more powerful oxidants than typical peroxidases are, and consequently oxidize not only the usual peroxidase substrates such as phenols and anilines, but also a variety of non-phenolic lignin structures and other aromatic ethers that resemble the basic structural unit of lignin. Unlike the other peroxidases like MnP, LiP is able to oxidize non-phenolic aromatic substrates and does not require the participation of mediators due to its unusually high redox potential (Wong 2009).

8.3 Manganese Peroxidases

Manganese peroxidases (MnPs) are extracellular glycoproteins and are secreted in multiple isoforms which contain one molecule of heme as iron protoporphyrin (Asgher et al. 2008). MnPs occur in most white-rot fungi, and are similar to conventional peroxidases, except that manganese (Mn) is essential for the catalysis. The enzyme MnP plays an important role during the initial stages of lignin degradation (Perez and Jeffries 1990). Compared to laccase, MnP causes greater degradation of phenolic lignin due to its higher redox potential with the eventual release of carbon dioxide (Ten Have and Teunissen 2001). MnP catalyzes the peroxide-dependent oxidation of Mn(II) to Mn(III), which is then released from the enzyme surface in complex with oxalate or with other chelators. Chelated Mn (III) complex acts as a reactive low molecular weight, diffusible redox-mediator of phenolic substrates including simple phenols, amines, dyes, phenolic lignin substructures and dimers (Wesenberg et al. 2003; Asgher et al. 2008). In addition, organic acids such as lactate and malonate can chelate Mn³⁺ ion. Since their discovery in *P. chrysosporium*, they have been found in many basidiomycetes such as Lenzites betulinus, Phanerochaete flavido-alba, and Panus tigrinus (Hoshino et al. 2002; de la Rubia et al. 2002; Lisov et al. 2003).

8.4 Versatile Peroxidases

Versatile peroxidase (VP), as the name suggests, has catalytic properties of both LiP and MnP. VPs are glycoproteins with hybrid properties capable of oxidizing typical substrates of other basidiomycetes peroxidases including Mn(II) and also veratryl alcohol (VA), MnP and the typical LiP substrate, respectively (Wesenberg et al. 2003; Asgher et al. 2008). VP was first purified from the genera of fungi Bjerkandera and was found to transform lignin even without an external mediator (Moreira et al. 2007). VPs form an attractive ligninolytic enzyme group due to their dual oxidative ability to oxidize Mn(II) and also phenolic and non-phenolic aromatic compounds. Versatile peroxidase possesses a hybrid molecular architecture with several binding sites including Mn²⁺ and is able to oxidize Mn²⁺ like MnP and LiP. However, unlike MnP, VP has the dual ability to oxidize Mn²⁺ in the independent oxidation of simple amines and phenolic monomers. VP can also oxidize a variety of substrates (with high and low redox potentials) including Mn²⁺, phenolic and non-phenolic lignin dimers, and aromatic alcohols (Perez-Boada et al. 2005; Camarero et al. 1999). This makes versatile peroxidase a better catalyst than MnP and LiP. Similar to the MnP mechanism, Mn(III) is released from VPs and acts as a diffusible oxidizer of phenolic lignin and free phenol substrates. Like other members of heme peroxidases, heme is buried in the interior of the protein and has access to the outer medium through two channels (Heinfling et al. 1998).

9 Measurement of Lignin Degradation

The delignification of woody tissues by fungi can be assessed by microscopy (Blanchette 1991) with selective staining (Srebotnik and Messner 1994); however, these techniques are relatively complex and semiquantitative. Microscopy is useful only when the substrate contains lignified cell walls.

The removal of lignin from lignocellulosic material can be measured by the simple procedure of extracting its low molecular weight components, weighing the leftover woody tissues, degrading the remaining polysaccharide component with strong acid, and then reweighing the leftover lignin, which is referred to as Klason lignin. Measurement of Klason lignin is simple and useful if interfering substances are not present. However, this method is not reliable if it is used on plant tissues that contain high molecular weight components that cannot be are not removed by extraction and acid treatment. Interfering substances may be proteins and tannins (Theander and Westerlund 1993). On the other hand, the Klason procedure is also not accurate when woody tissues contain polymers which are acid-soluble and consequently lost during the hydrolysis of polysaccharides.

Another approach to study fungal ligninolysis is to assess whether the fungus degrades a simpler molecule rather than growth substrate itself being delignified. Substrates of this type can be incorporated into the natural lignocellulosic growth medium, e.g., wood or litter, or they can be used as probes in defined liquid growth media (Kirk 1975, 1978; Srebotnik and Messner 1994). C¹⁴-labelled synthetic lignins are most frequently used in this type study. Carbon dioxide produced during degradation of the radiolabelled polymer can be trapped in alkali and determined by scintillation counting. However, this method is expensive and requires expert facilities for radioisotope study.

Several chemical procedures have also been introduced for the estimation of lignolysis. For example, pulverized wood samples can be treated with acetyl bromide in acetic acid, and the absorbance of the resulting solution is measured at 280 nm. Comparison of absorbance with absorbance obtained from known lignin standards can be used to determine the extent of lignolysis. Methods of this type are subject to interference from other components, but can be useful for the comparison of closely related lignocellulosic samples (Theander and Westerlund 1993).

10 Importance of Biodegradation of Lignin

Of all naturally produced organic chemicals, lignin is probably the most recalcitrant. Lignin is the most abundant aromatic compound on earth and is second only to cellulose in its contribution to living terrestrial biomass. Lignin degradation can thus play a major role in improving earth's biofuel resources and also serve as an alternative to harsh technologies used in the paper and pulp industry. When plants die or drop litter, lignified organic carbon is incorporated into the top layer of the soil. This recalcitrant material has to be broken down and recycled by microorganisms to maintain the earth's carbon cycle. If this process fails, imagine the amount of carbon which is irreversibly sequestered as lignocellulose.

Lignin biodegradation has diverse effects on soil quality. Microbial lignolysis facilitates the formation of humus by promoting the release of aromatic humus precursors such as flavonoids, terpenes, and tannins from the litter (Hudson 1992). Undegraded lignocellulose, has adverse effect on soil fertility by supporting high microbial load. Increasing population of microbes in soil may not only produce phytotoxins but also compete with crop plants for soil nitrogen and other nutrient. By breaking down the most refractory component of litter, ligninolysis thus contributes to the removal of conditions that inhibit crop productivity.

In paper and textile industries, wood chips are often treated with chemical or mechanical forces to separate their fibers. In chemical pulping, strong chemicals, such as soda, and high pressure are applied to remove lignin. The resulting pulp has high strength but the process is polluting and the overall yield is relatively low (40–50%). Mechanical pulping uses mechanical forces to separate wood fibers. In this process, yields are higher (up to 95%) and results in paper with good printing properties. However, the process requires a lot of electrical energy for refining, the fibers have poor strength properties, and high content of lignin causes a tendency to yellowing. Mechanical pulping can be combined with biological pulping to save refining energy (Akhtar et al. 1998). Also fungal treatment of fibers as a substitute for chemical treatment can not only make the process eco-friendly but high yielding as well by removing the hemicellulose content of the fibers.

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Chapter 11 Application and Biodegradation of Lignocellulosic Biomass



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1 Introduction

Lignocellulosic wastes are generated in large quantity throughout the world as a result of crop residues and these wastes pose an environmental threat because of their disposal problems. To get rid of this bulk, these wastes are generally burnt or left in the field to rot, leading to unhygienic conditions as well as waste of resources. Burning of these wastes not only results in oxygen-deficient environment but also causes several respiratory diseases like asthma, tuberculosis, etc. as well as poor visibility at night (Singh and Gautam 2004; Agarwal et al. 2016). These wastes are composed of lignin, cellulose, and hemicelluloses (Jeznabadi et al. 2016; Agarwal et al. 2018). Because of the composition of these wastes, they are difficult to degrade and are very slowly degraded by ruminants, as lignin polymers provide physical and chemical barrier thereby preventing the access of hydrolytic enzymes, such as cellulases and hemicellulases to their substrates, which are responsible for the degradation of cellulose and hemicelluloses of the substrate, respectively (Cohen et al. 2002; Chaturvedi et al. 2019). Lignin is the second most abundant plant polymer which provides resistance against microbial attack and also gives strength in combination with cellulosic fibers. Lignin forms a matrix around the cellulose, because of which there is slow microbial depolymerization of cellulose, which otherwise can be used for several industrial purposes; therefore, it is necessary to first degrade lignin for the global carbon cycle (Gold and Alic 1993).

The problem related to safe disposal of lignocellulosic wastes and their strict nature towards degradation can be solved by using these wastes for the cultivation of edible mushroom species, especially white rot fungus (WRF), which with its potential enzymatic system convert these wastes into value-added products like nutrient-rich food, improved animal fodder, plant fertilizer, biofuels, chemicals,

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cheap carbon source for fermentation, etc. (Philippoussis et al. 2000; Poppe 2000; Kuhad et al. 2007). Mushrooms produce ligninolytic and hydrolytic enzymes which help in degrading these wastes which are rich source of carbohydrates and therefore, mushrooms can be cultivated on large variety of these wastes like pasteurized wheat straw, paddy straw, sawdust, etc. (Kabirifard et al. 2012; Kulshreshtha et al. 2013; Mikiashvili et al. 2006; Chaturvedi et al. 2018). The enzymes secreted by WRF, i.e., cellulases, hemicellulases, and ligninases, are nonspecific in nature and therefore apart from playing a role in degradation they are also used for bioremediation (Adebayo and Martinez-Carrera 2015). These enzymes have many industrial applications including chemicals, fuel, food, brewery and wine, animal feed, textile and laundry, pulp and paper, and agriculture (Kuhad et al. 2007).

The chemical composition of lignocelluloses makes them a suitable substrate for other industrial purposes as well. Several technologies are available that are used for the bioconversion of lignocelluloses to value-added products. These technologies are gaining interest with the belief that they will provide us with improved quality products with low price (Murarilal and Pereira 2006). They can be used in different industries like pulp and paper, chemical, textile, food, feed and fertilizer, biosensor, bioconversion, etc. The present chapter focuses on the lignocellulosic production, their composition, their degradation through mushroom cultivation making use of its enzymes and applications of lignocelluloses in various industries.

2 Chemical Composition of Lignocellulosic Biomass

From the last several years, agro-industrial biomass comprising of lignocellulosic waste is an inexpensive, renewable, abundant in quantity and provides a unique natural resource for large-scale and cost-effective bio-energy collection (Anwar et al. 2014). In general, the composition of lignocelluloses highly depends on its source whether it is derived from the hardwood, softwood, or grasses. Table 11.1 shows the typical chemical compositions of components present in various lignocellulosic materials. Lignocellulosic biomass is mainly composed of three components such as cellulose, hemicellulose, and lignin together with small amounts of other components like acetyl groups, minerals, and phenolic substituents (Isikgor and Becer 2015).

3 Lignocellulose Production

Large amounts of lignocellulosic wastes are generated as a result of agriculture, horticulture, forests, food and wood processing industries. About 4664.63 million tonnes of agricultural waste is produced as a part of crop residues (Rai and Ahlawat 2005). With expanding agricultural production, there is an increase in the quantities

S.	Lignocellulosic	Cellulose	Lignin	Hemicellulose	
n.	biomass	(%)	(%)	(%)	Reference
1.	Newspaper	40-55	18-30	25-40	Howard et al. (2003)
2.	Corn stover	37–42	19	26	Zhu et al. (2005)
3.	Banana waste	13.2	14	14.8	John et al. (2006)
4.	Sugarcane bagasse	42	20	25	Kim and Day (2011)
5.	Sponge gourd fibers	66.59	15.46	17.44	Guimaraes et al. (2009)
6.	Nut shells	25-30	30-40	25-30	Howard et al. (2003)
7.	Softwood stems	45–50	25–35	25–35	Malherbe and Cloete (2002)
8.	Grasses	25–40	10–30	25–50	Malherbe and Cloete (2002)
9.	Barley straw	33-40	8–17	20–35	Pedersen and Meyer (2010)
10.	Sweet sorghum	45	21	27	Kim and Day (2011)
11.	Rice straw	32–41	10–18	15–24	Van Dyk and Pletschke (2012)
12.	Corn cobs	45	15	35	Prasad et al. (2007)
13.	Bagasse	54.87	23.33	16.52	Guimaraes et al. (2009)
14.	Energy crops	43-45	19–12	24-31	Prasad et al. (2007)
15.	Wheat straw	31-44	16–24	22–24	Szulczyk et al. (2010)

Table 11.1 Chemical compositions of components present in various lignocellulosic materials

of livestock waste, agricultural crop residues, and agro-industrial by-products (Obi et al. 2016). Annually, there is an average rate of 5-10% increase in agricultural wastes (Wang et al. 2016). It has been found that after the processing of one ton of the main product about 1.5 tons of crop residues are generated. According to a survey by Govt. of India, it is estimated that every year about 500 metric tons of crop residue is generated (Mahawar et al. 2015). About 100 million tons of paddy straw alone contributes to crop residue (Tewari and Pandey 2002), adding to this paddy stubble, the leftover lower portion of paddy crop, being heavier, denser, and more bulky in nature as compared to paddy straw accounts for 5-10% of entire crop residue (Singh and Gautam 2004). Dry biomass net productivity as a result of photosynthesis by plants is around 155.2 billion tons per year of which two third is produced on land and one third ends up in the ocean (Singh and Gautam 2004). About 10¹² metric tonnes of plant waste is produced each year in the form of cellulose (Wilson and Irwin 1999). These wastes generally consist of wood residues, waste paper, grasses, agricultural residues (like straw, stalks and bagasse), domestic wastes (lignocellulosic garbage and sewage), and municipal solid wastes (Rodríguez et al. 2008).

4 Lignocellulose Degradation

The structure of lignocellulose is very complex and resistant to enzymatic hydrolysis. The lignin layer of the lignocellulose makes their degradation tough and generally, the rate of decay is directly proportional to the lignin content in the plant material (Cohen et al. 2002). Earlier methods used for degradation are either costly or not much efficient, like the use of alkaline and acid hydrolysis in which weak acids help to remove the lignin but cause problem for cellulose hydrolysis and on the other hand, in strong acid treatment physiological conditions have to be maintained like temperature and pH which also requires expensive instruments (Kuhad and Singh 2007). Therefore, biological delignification is one of the most promising methods to improve their digestion or degradation (Streeter et al. 1982; Kamra and Zadražil 1986).

Cultivation of edible mushrooms on these wastes is value-added process, as the aim is not just to degrade lignin but also the yield of another product of commercial use like human and animal nutrient-rich food (Zhang et al. 2002). It serves as one of the most efficient biological ways of recycling of wastes with another advantage of using spent compost, rich in lignin content, for different purposes like biofuels production, as a fertilizer, etc. (Madan et al. 1987; Murarilal and Pereira 2006).

4.1 Biodegradation Through Mushroom Cultivation

Mushroom cultivation on lignocellulose waste is carried out from long time with varying degrees of success. The slow rate of degradation of lignocellulose is because of the presence of lignin and therefore white rot fungus is best suitable for degradation with the help of its lignin modifying enzymes (LDEs). They convert the lignin polymer to CO₂ and H₂O, after which the cellulose and hemicellulose can easily be attacked by cellulases and hemicellulases, respectively, of the WRF. They can colonize on a wide variety of lignocellulosic wastes like wheat straw, paddy straw, cotton, peanut shells, poplar sawdust, oak sawdust, corncobs, olive press-cake, sorghum straw, pearl millet stalks, rye straw, maize cobs, banana waste, maize straw, hard wood chips, sugarcane bagasse, rice and wheat bran, etc. (Zervakis et al. 2001; Adebayo and Martinez-Carrera 2015; Davari et al. 2012; Orts et al. 2008). These fungi with their phenol-targeting redox enzymes selectively degrade lignin and related compounds (Ntougias et al. 2012). Higher fungi with its enzyme convert the high molecular weight compound to low molecular weight compound, which can be further used for the fungal nutrition (Elisashvili et al. 2003). The enzymes and the exploitation of substrate vary with species, strains, and cultivation method (Zadrazil and Dube 1992). It has been observed that fungus helps in converting the cellulose and other plant carbohydrates to acetic, propionic, and butyric acid in large amounts, which can further be used by ruminant animals as energy and carbon source. These
WRF also can convert the lignocellulose wastes anaerobically in liquid digesters (Ezeji et al. 2006; Martín et al. 2006; Alborés et al. 2006).

During the cultivation of mushrooms, it was reported that the fiber, hemicellulose, and cellulose content of the spent cultivation substrate decreases, whereas the protein content in the spent substrate increases (Koutrotsios et al. 2014). It has been observed that due to large amount of cellulose, which is the most biodegraded component of the lignocellulose waste, followed by hemicelluloses, which is found in less abundant and lastly lignin which is present in the smallest quantity (Salmones et al. 2005). The cultivation of mushrooms are used for various purposes like biodegradation of hazardous compounds, biological detoxification of toxic agroindustrial wastes, biotransformation of agro-industrial residues to mushroom food and animal feed as well as value-added products like biologically active metabolites, enzymes, flavor compounds, etc. (Philippoussis 2009; Philippoussis and Diamantopoulou 2011). During the cultivation of mushroom, care should be taken to check that the wastes do not contain chemical components and toxic substances, which otherwise can affect the mycelium growth and also can cause human health issue (Stamets 2011). The supplementation of substrate improves the nutrient requirements for mushroom production. The carbon and nitrogen content in the substrate have significant effect on mushroom production and it has been observed that during the mushroom cultivation carbon content decreases whereas, nitrogen and ash content in the substrate increases.

The enzymes present in WRF that help in the degradation of these wastes are cellulases, hemicellulases, and ligninases which degrade the cellulose, hemicellulose, and lignin part of the waste, respectively.

4.2 Enzymatic Studies of Cellulases

Cellulases are enzymes that degrade the cellulose part of the lignocellulosic biomass. Cellulose is a homopolysaccharide consisting of β -D-glucopyranose units, which are linked by β -(1 \rightarrow 4)-glucosidic bonds. Both nonreducing and reducing ends are present in cellulose. The repetitive unit of cellulose is called as cellobiose, which can be converted into glucose residues. Fungi are the most studied microorganisms in relation to cellulose degradation by cellulolytic enzymes. The cellulases are further divided into three groups: endo- β -1,4-glucanases (EC 3.2.1.4), exo- β -1,4 glucanases I and II (EC 2.1.1.91), and β -glucosidase (EC-3.2.1.21) (Adebayo and Martinez-Carrera 2015).

The endo- β -1,4-glucanases part of the enzyme randomly catalyze the internal bonds of cellulose chain, i.e., they split the β -1,4-glucosidic linkages. Exo- β -1,4 glucanases split the cellobiose or glucose from the cellulose chain, in which exo- β -1,4 glucanases I attack at the reducing end of the chain whereas exo- β -1,4 glucanases II attack the nonreducing end of the chain. Moreover, β -glucosidase attack the cellobiose and other water-soluble cellodextrin and release the glucose monomer units from them (Eriksson et al. 1990; Adebayo and Martinez-Carrera 2015). There are two other enzymes as well that are involved in the degradation of cellulose; these are the oxidative enzymes, first one is cellulose: quinone oxidoreductase (cellobiose dehydrogenase) which reduces quinones and phenoxy radicals with the oxidation of cellobiose to cellobiono- δ -lactone, and the second one is cellobiose oxidase, which with the help of molecular oxygen oxidizes cellobiose and higher cellodextrins to their corresponding onic acids (Eriksson et al. 1990).

4.3 Enzymatic Studies of Hemicellulases

The hemicellulases degrade the hemicellulose part of the lignocellulosic biomass. Hemicelluloses are the heterogeneous polymers that are built up by pentoses (D-xylose, L-araninose), hexoses (D-glucose, D-galactose), sugar acids (ferulic acids and 4-O-methyl-D-glucuronic acid) and acetyl group (Adebayo and Martinez-Carrera 2015). Fungi secrete some hemicellulolytic enzymes that degrade the hemicellulose; these are either extracellular cell wall bound or intracellular (Dekker 1985; Kuhad et al. 2007). The hemicellulases include α -D-glucuronidase (EC 3.2.1), endo-β-1,4-xylanase (EC 3.2.1.8), α-L-arabinofuranosidase (EC 3.2.1.55), α-Dgalactosidase, β-D-xylosidase (EC 3.21.37), feruloyl esterase, and acetylxylan esterase (EC 3.1.1.6) (Biely 1985; Adebayo and Martinez-Carrera 2015). The α-D-glucuronidase attack 4-O-methyl-D-glucuronic acid, endo-β-1,4-xylanase degrade the xylan chain, whereas α -L-arabinofuranosidase attack the L-araninose chain end. After this ferulic acid and acetyl groups are removed by feruloyl esterase and acetylxylan esterase, respectively. The α -D-galactosidase reduces xylan to xylobiose and this xylobiose is further degraded to D-xylose, a monomer unit of xylobiose, by β-D-xylosidase (Adebayo and Martinez-Carrera 2015).

4.4 Enzymatic Studies of Ligninases

WRF secrete various ligninases, which very efficiently degrade the lignin component of the lignocellulose biomass. Lignin is a complex oxyphenol propanoid polymer monomeric unit linked together with different types of bonds. It is synthesized by one-electron oxidation of precursors: *p*-coumaryl, coniferyl, and sinapyl alcohol generate phenoxy radicals and then non-enzymatic polymerization occurs which forms high molecular weight, heterogeneous, three-dimensional polymers (Garg and Modi 1999). The main enzymes of ligninases are laccase, lignin peroxidase (LiP), manganese peroxidase (MnP), and versatile peroxidase (VP). The LDEs are oxido-reductive enzymes that degrade the lignin by a series of redox reactions. They oxidize the aromatic compounds until the aromatic ring is cleaved, which is then degraded by other enzymes (Husain 2006). They have broad range of specificity, high redox potential, extracellular in nature, and are expressed in nutrientdeficient environment; therefore, they are suitable for lignin degradation (Reddy

1995). Laccase is a benzenediol: oxygen oxidoreductase multi-copper enzyme that uses molecular oxygen to oxidize various aromatic and nonaromatic compounds by radical catalyzed reaction mechanism (Claus 2004; Baldrian 2006). Laccase is monomeric, dimeric, or tetrameric glycoprotein with four copper atoms classified into three types (T1, T2 and T3) present in its catalytic center, which mediate the redox process. T1 carries out the oxidation of substrate that leads to the formation of free radicals, which further acts as intermediate substrate for the enzymes (Yoshitake et al. 1993; Ferraroni et al. 2007; Adebayo and Martinez-Carrera 2015). Laccase catalyzes demethoxylation reactions of terminal phenolic units. It degrades β -dimers and β -O-4 dimers by α -oxidation of alkyl-aryl cleavage and C_{α}-C_{β} cleavage (Youn et al. 1995). Lignin mediator system carries out various physiological functions like lignolysis, lignin synthesis, morphogenesis, pathogenesis, and detoxification (Mayer and Staples 2002). LiP is a heme containing glycoprotein that catalyzes the oxidation of non-phenolic aromatic lignin compounds (Wong 2009). It catalyzes the H₂O₂-dependent oxidation depolymerization of a wide range of phenolic and non-phenolic compounds, polycyclic aromatic hydrocarbons, etc. by a series of non-enzymatic reactions in a multi-step electron transfer process which then forms intermediate radicals like phenoxy radicals and veratryl alcohol radical cations. These radicals undergo different reactions, which finally results in the oxidation of lignin and cleavage of the aromatic ring structures (Ten Have and Teunissen 2001; Umezawa and Higuchi 1987; Wong 2009). LiP cleaves β -O-4 ether bonds and C_{α} - C_{β} bonds in dimeric lignin model compounds. It also catalyzes decarboxylation of phenylacetic acids, oxidation of aromatic C_{g} -alcohols to C_{g} -oxo compounds, hydroxylation and quinone formation (Haemmerli et al. 1987; Gupta et al. 2017). MnP is a glycosylated glycoprotein containing an iron protoporphyrin IX (heme) prosthetic group is secreted in multiple isoforms (Nie et al. 1999; Glenn and Gold 1985; Asgher et al. 2008). It is involved in the oxidation of phenols and phenolic ring structures (Kuhad et al. 2007). MnP is a H_2O_2 -dependent enzyme that catalyzes the oxidation of Mn (II) to Mn (III) which results in oxidation of various phenolic compounds. MnP oxidizes Mn²⁺ to Mn³⁺, which is then stabilized by chelators like oxalic acid (Glenn et al. 1986; Wariishi et al. 1992). Chelated Mn (III) complex acts as a reactive low molecular weight, diffusible redox-mediator of phenolic substrates. For oxidation of non-phenolic compounds, Mn (III) must form a reactive radical in the presence of other mediators like oxalate and malonate (Wong 2009; Wesenberg et al. 2003). Versatile peroxidase is also a glycoprotein that combines the properties of LiP and MnP (Adebayo and Martinez-Carrera 2015). It is an attractive ligninolytic enzyme because they oxidize Mn (II) as well as phenolic and nonphenolic aromatic compounds (Wesenberg et al. 2003). It is superior to LiP and MnP because of the hybrid molecular structure which provides multiple binding sites for the substrates (Camarero et al. 1999). It contains two channels; the first channel is similar to LiP and in the second channel oxidation of Mn (II) to Mn (III) takes place (Ruiz-Dueñas et al. 1999). Other enzymes apart from these, which are involved in lignin degradation, are aryl-alcohol oxidase (AAO), aryl-alcohol dehydrogenases (AAD), quinone reductases (QR) and cellobiose dehydrogenase (CDH) (Adebayo and Martinez-Carrera 2015).

5 Biotechnological Application of Lignocellulose

5.1 Biofuel

Over the years, indiscriminate use of fossil fuels has resulted not only in concerns about their environmental impact, but also their inevitable exhaustion. One possible solution to these problems is the replacement of materials from these non-renewable sources with renewable ones. The use of lignocellulosic biomass and agro-industrial residues from various sources is of fundamental importance for the construction of various sectors of the bioeconomy based on the innovation of biofuels and bioproducts. The bioconversion of lignocellulosic biomass into fermentable sugars for the production of bioethanol has been considered a promising alternative to increase the production of ethanol to meet global demand (Kuila and Sharma 2017; Maurya et al. 2016).

Currently, the first generation processes for the production of ethanol from corn. Only a small part of the corn plant, i.e., the corn kernels are taken from the corn plant and only the starch, which represents about 50% of the dry kernel mass, is transformed into ethanol. Two types of second-generation processes are under development. The first type uses enzymes and yeast fermentation to convert the plant cellulose into ethanol while the second type uses pyrolysis to convert the whole plant to either a liquid bio-oil. Second-generation processes can also be used with plants such as grasses, wood or agricultural waste material such as straw. The principal advantage of second-generation biofuel over fossil fuel is that biofuels are produced from renewable and sustainable resource. The combustion of biofuel releases almost zero or very little net carbon dioxide emission to the atmosphere and has a positive impact on environmental protection, especially the global climate change. Bioethanol is produced from corn and sugarcane, depending on the climatic conditions whereas the feedstock used for the production of bioethanol mainly produced from sugarcane in the tropical regions such as Brazil and Columbia whereas it is produced from corn in the areas such as the USA, European regions, and China. Ethanol fuel has a "gasoline gallon equivalency" (GGE) value of 1.5, i.e., to replace the energy of 1 volume of gasoline, 1.5 times the volume of ethanol is needed. Bioethanol consists of gasoline ethanol mixtures such as E15 (15% Ethanol and 85% Gasoline) and E85 (85% Ethanol and 15% Gasoline) which was partially replaced by gasoline. One of the major applications of lignocellulosic biomass is the production of second-generation bioethanol. Lignocellulosic biomass such as agricultural residues, woods, and grasses are abundant in most land areas of the world and their generation does not have to compete for arable land against food and feed production. They are abundant almost all over the world and have high contents of cellulose and hemicelluloses; therefore, they are used for the production of bioethanol. According to the International Energy Agency, cellulosic ethanol could allow ethanol fuels to play a much bigger role in the future (www.world energyoutlook. org). There is difficulty in directly converting lignocellulosic materials to ethanol; hence, a three-step process is required for the bioconversion:

5.2 Pretreatment/Delignification

The purpose of pretreatment is the primary step to make biomass accessible by breaking the seal of lignin through the process of delignification, by removing hemicelluloses, or by disrupting the structure of crystalline structure of cellulose and increase the porosity of the material, so the hydrolytic enzymes can access their substrates (cellulose and hemicellulose) in the following enzymatic hydrolysis. This pretreatment including the physical, chemical (alkali and acid hydrolysis), and biological processes for lignocellulosic materials. Among these three pretreatment processes, biological pretreatment is probably the most economical and environmentally friendly pretreatment technology for the lignocellulosic materials because fungi such as brown, white, and soft-rot fungi degrade lignin and hemicellulose in lignocellulosic materials.

Brown rot fungi mainly attack cellulose, while white and soft rots attack both cellulose and lignin. White rot fungi are the most effective basidiomycetes for biological pretreatment of lignocellulosic materials. Biological pretreatment using metabolites of white rot fungi in nature for ethanol production from biomass is a promising technology due to several advantages such as its being an eco-friendly and economically viable strategy for enhancing enzymatic scarification rate. Since no chemicals are used in this process, there is no need for recycling of chemicals and it does not release toxic compounds into the environment (Sindhu et al. 2016). Potumarthi et al. (2013) studied simultaneous pretreatment and saccharification of rice husk by *Phanerochaete chrysosporium*. Effective delignification was carried out by growing the fungus on rice husk and the pretreated biomass was subjected to enzymatic hydrolysis. This method avoids operational costs associated with washing and the removal of inhibitors during conventional pretreatment.

5.3 Fermentation

The three major components of lignocelluloses viz. cellulose have the highest conversion rate to ethanol which is 85–90%, hemicelluloses 30–85%, and lignin 0%. The main products of hemicelluloses hydrolysis include hexoses and pentoses. The cellulose and hemicellulose can be readily fermented to ethanol by microorganism whereas lignin is difficult to be converted to ethanol easily but it can be used as a fuel in the fractional distillation in ethanol purification. Ethanol has lower pressure than gasoline, thereby reducing smog formation by lowering evaporative emissions. It also has a high octane value, high heat of vaporization, and other favorable fuel properties.

5.4 Animal Feed

Due to high lignin and low protein contents, the feed value of lignocelluloses is quite low. Hence, it is desirable to utilize some processes for enhancing the protein content of the biomass. Kamara and Zadrazil (1988) reported that white rot fungi can utilize lignocellulosic wastes to make them palatable animal fodder by increasing the protein level, an increase in the digestibility of the lignocellulosic material and an improvement in the dry product palatability. Solid state fermentation (SSF) of lignocellulosic residues by white rot fungi improves the availability of cellulose to ruminants (Mahesh and Mohini 2013). White rot fungi such as P. chrysosporium, Cyathus sp., Coriolus hirsutus, Dichomitus squalens, and Pleurotus spp. are the extensively studied organisms for solid state bioconversion of lignocellulosic materials. SSF is a process in which solid substrates are decomposed by known cultures of microorganisms (mainly fungi, which can grow on and through the substrate) under standard conditions (Zadrazil and Reiniger 1988) and makes these lignocellulosic substrates to more digestible and nutritious feedstuffs for ruminants. Digestibility of the lignocellulosic materials such as straw is dependent on the depolymerization of its structural carbohydrates (Mahesh and Mohini 2013). Species of Pleurotus are most studied white rot fungi for improving digestibility and nutritional quality of straw (Kundu 2005; Streeter et al. 1982; Kakkar et al. 1990). Up to 75% of wheat straw in vitro digestibility is done by using C. stercoreus or D. squaleris (Agosin and Odier 1985).

5.5 Fertilizers

Generation of the equivalent amount of spent mushroom compost from the metric tonnes of mushroom are produced from the mushroom industry. Fermor et al. (2000) reported that the disposal of these spent composts is bulky, and charges and taxes on landfill are implemented in the developed countries. Spent mushroom compost consisting of degraded cellulose, hemicelluloses, and lignin having high contents of mineral, phosphate with high porosity, is a good soil fertilizer for inducing the seed germination and as a prospective bioremediating agent (Chiu et al. 2000) because this spent substrates can be a source of extracellular and intracellular lignocellulolytic enzymes (Fermor et al. 2000). In addition to providing balanced carbon and nitrogen sources for plant growth and germination, this spent compost undergoes further conversion in the soil to humus. According to Ball and Jackson (1995), the composition, activity, and stability of enzymes found from compost extract suggest that they could further be used for the preparation of animal fodder. This compost actually speeds up emission of methane biogas.

5.6 High Value-Added Bioproducts/Chemicals

A number of high-value bioproducts such as organic acids, amino acids, vitamins, and fungal polysaccharides such as xanthans are produced by fermentation using glucose as the initial substrate could be manufactured from lignocellulosic residues (Sánchez 2009). Ribbons (1987) reported that several potential value-added products could be derived from lignin such as vanillin and gallic acid which can be used as herbicide, antifoaming agents, antimicrobial agents and drugs using white rot fungi *P. chrysosporium*. A method developed for the production of furfural from lignocellulosic materials like birch and bagasse biomass which is also the best raw material for levoglucosan production (Gravitis et al. 2004).

The acid hydrolysis of lignocellulosic materials such as xylose (hemicellulose) solutions and this xylose further utilized for making xylitol which is used as a sweetener and a teeth hardening agent (Parajó et al. 1998).

6 Conclusion

As a result of agriculture, food and wood processing industries, large amount of lignocellulosic waste is generated. These wastes are made up of complex compounds and therefore degrade very slowly thereby causing several environmental problems and unhygienic conditions. One of the best utilization and environmentfriendly disposals of these wastes is the cultivation of mushroom on these wastes as they are rich source of nutrients. The mushrooms secrete many extracellular and intracellular enzymes which help in the breakdown of complex compounds of wastes into low molecular weight compounds, which are then taken up by mushrooms for their growth and nourishment. Apart from this, these wastes have found their importance in different industries as well, where they serve as the main source for production of several bioproducts, chemicals as well as biofuel. The wastes generated after mushroom cultivation, that is called as spent mushroom compost, being rich in nutrients can further be used as animal feed and fertilizers, thereby solving the problem of waste produced after mushroom cultivation which is in large quantities. So, keeping in view these many solutions, one can make best use of the lignocellulosic wastes which are generally burnt in the field leading to oxygen-deficient environment which ultimately causes several respiratory diseases.

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Chapter 12 Myco-Nanotechnological Approach for Improved Degradation of Lignocellulosic Waste: Its Future Aspect



Abhishek K. Bhardwaj, Manish Kumar Gupta, and R. Naraian

1 Introduction

Huge biomass of naturally produced lignocellulosic waste during agricultural practices is a big problem during its disposal in developing country. Lignocellulose is a complex carbohydrate polymer that is bonded to strong bonds to give it a highly stable structure and other hand intricate nature of lignin like polysaccharide of plant cell wall is slow or resistant to enzymatic degradation in normal environmental condition (Bilandzija et al. 2018). The farmers most often dispose huge amount of wastes through the process of negligent burning in open, which release huge amount of air pollutants like dense white or black smoke and ash (Perlack 2005; Alftrén and Hobley 2014). However, the lignocellulosic biomass has enormous potential to contribute global renewable energy, chemical, and materials in a sustainable manner (Kumar et al. 2008; Limayem and Ricke 2012; Saratale and Oh 2012; Fapyane and Ferapontova 2017). The technologies need to develop the value-added product such as (sugar, organic acids, surfactants, glues, solvents or beverage softeners, etc.) from huge amount of agriculture and forests, which are prominent sources of carbohydrates (Mamilla et al. 2019).

Many researchers reported that cellulolytic enzymes are found in insect, bacteria, fungi, and plant which can be utilized into various industrial sectors for conversion of cellulose into free glucose (Goyal et al. 1991; Jahangeer et al. 2005; Sharma et al. 2007). Due to wide demand for hydrolyzed cellulose products; various industrial sectors employing cellulolytic enzymes has attracted greater attention since past decades (Tischer and Wedekind 1999). Cellulases are the combination of

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intricate enzymes which catalyze the hydrolysis of cellulose due to the synergistic actions of the exoglucanase, endoglucanase, and β -glucosidase enzymes. These three important components can break the cross-linked bonds present in cellulose and generate monomers of glucose for further fermentation. Although cellulases have many applications in medicine, energy and industry, most of them have low enzyme stability at different media, P^H, and temperatures (Dutta and Saha 2019). In addition, these applications are limited due to lack of reusability (Woodley 2013; Guzik et al. 2014). Various different methods can be used, such as protein engineering, chemical modification, and immobilization to enhance enzyme stability (Rodrigues et al. 2013; Barbosa et al. 2014; Rueda et al. 2016; Vaghari et al. 2016). In view of enhanced capability and stability of cellulose, a group of researchers introduced the concept of nanosupport immobilized by lignocellulosic enzymes (Blanchette et al. 2012; Hwang and Gu 2013; Nawaz et al. 2016; Wan et al. 2016). Surface immobilization, self-immobilization, and entrapment can be divided into three major categories of immobilization of enzyme onto NPs form a complexes. These complexes are made up of immobilized enzyme on to insoluble support (NPs) which enhances the properties of enzyme like bio-catalysis, thermostability, and reusability. The previous reports are suggesting among all the nanosupports, the magnetic nanosupports having a better choice due to less harmful, high surface area and reusable nature (Bilal et al. 2018). These properties can be employed for a wide range of industrial processes such as paper and cotton manufacturing units, fuel and food industries, clarification and extraction process of wine, juice and brewery, detergents, animal feed additives, agriculture, and research (Fig. 12.1). In the industrial processing or production yield, nanomaterials modification with enzymes has the potential to improve the economic feasibility of the entire process (Khoshnevisan et al. 2017). The chapter discusses recent and up-to-date information regarding to the role of various immobilized nanomaterials and nanocomposites on lignocellulose decomposition as well as on the mechanism of enhancing the lignocelluloses decomposition. The properties of various metal and magnetic nanocomposites with immobilized enzymes such as enzyme stability, activity, reusability, recovery and lignocellulose conversion ability in different fuels, materials, and value-added chemicals are also discussed.

Fig. 12.1 Multiple applications of nanosupport (★) immobilized with lignocellulosic enzymes



2 Complexity and Recalcitrant Nature of Lignocelluloses

The plant cell wall basically composed of three major biopolymers (a) cellulose $(\sim 30-50\%$ by weight), which is the main component, (b) hemicellulose ($\sim 20-45\%$ by weight as xylans), and (c) non-polysaccharide aromatic polymer lignin (~15-35) (Bacic et al. 1988; Metzger and Hüttermann 2009). The biomass fractionation is a very complex and lengthy process of these materials. High amount of sugars can be produced from cellulose and hemicellulose, using acid as the catalyst, or enzymatic hydrolysis and reformed into bio-ethanol by a fermentation procedure (Walker and Wilson 1991; Refaat 2012). Most physical and chemical pretreatment using acid, alkali, processes require special instrument and consume a lot of energy and generate inhibitors, which will affect enzymatic hydrolysis and fermentation (Mood et al. 2013). Many researchers reported that biological pretreatments employing bacteria, fungus (white, brown, and soft-rot fungi), deuteromycetes, and ascomycetes can enhance the hydrolysis productivity because of generating low inhibitors and limited energy utilization (Taha et al. 2015; Zhang et al. 2016). The lignocellulases are the constitutive building block of plant material composed of carbohydrate polymers like majority of cellulose (>50%), hemicelluloses, and aromatic component: lignin (~20%) as principal material. The lignocellulosic material includes not only agriculture and forest residue but also considers animal excreta, paper mill sludge, and municipal solid waste (MSW). Generation of more than 50 billion tons of lignocellulosic biomass produced annually worldwide and creates environmental problems due to its larger volume and mismanagement. Scientific communities are currently focused over such kind of abandoned renewable resources of the earth. But researchers facing various hurdles because recalcitrant nature of lignocelluloses such as: (a) considerable lignin content; (b) cellulose protection by lignin; (c) cellulose covering by hemicelluloses; (d) high degree of polymerization and crystallinity; (e) strong fiber strength (Agbor et al. 2011). The fractional segregation is the major difficulty of complex hierarchy structure of lignocellulosic biomass, in which cellulose, hemicellulose, and lignin are affected by many physical, chemical, structural, and compositional factors (Seemala et al. 2018). Lignin is a complex, amorphous hetero-biopolymer, water insoluble of phenylpropane units joined together by carbon-carbon and aryl-ether a linkage which provides frame support, impermeability, and protect from microbial attack. The chief cause of recalcitrant nature of lignocellulose is due to the presence of lignin (Grabber 2005). Generally, the softness of vegetation or wood reflects the high quantity of lignin in comparison to hardwood. The phenylpropane units of lignin are the principal bottleneck in degradation of lignocellulosic biomass (Mussatto 2016). The accessibility of cellulose is influenced by the presence of hemicelluloses which commonly coated with hemicellulose branches with short lateral chains consisting of different sugars (hexoses, pentoses, and acetylated sugars) (Yang et al. 2018). Hemicellulose can easily be hydrolyzed by diluted acid, alkali, and enzyme in comparison to cellulose under mild conditions. During the degradation, high thermochemical sensitivity of hemicelluloses produces several co-products of furfurals which inhibit the fermentation process (Chen et al. 2018). However, during the synthesis of nanocellulose co-product does not influence (Agbor et al. 2011; Hu et al.

2016). For this ground, diverse kind of pretreatment practices usually compromise to abandoned lignin and removal of hemicelluloses during maximum cellulose recovery for further nanocellulose production. Pretreatment of lignocellulosic biomass strategies are dependent on a variety of parameters like pH, temperature, treatment time, and different catalyst. This diversity of effects is pretreatment severity during degradation of biomass (Pedersen and Meyer 2010). The different kind of pretreatment techniques can be summarized as physical (milling and grinding), chemical (acid, alkaline, oxidizing agents, and organic solvent), biological, and some other combined techniques (wet oxidation, steam/auto hydrolysis, microwave, and hydrothermal) (Kumar et al. 2009; Stern et al. 2014; Bhardwaj et al. 2017a, 2018). Physical pretreatment consisting chipping, milling, grinding, and thermal process are safer than chemical methods, but less efficient, and require high energy (Stern et al. 2014). Chemical pretreatments are efficient, low cost biomass disintegration with less pretreatment severity, which may cause ecological threat. In case of biological methods, enzymes play a significant role in mild biomass pretreatments due to low stability of enzymes against pH and high temperatures (Van Dyk and Pletschke 2012).

3 Cellulases Action

Cellulase is an enzyme produced by different taxonomic groups of microorganism like fungi, bacteria, and protozoans. Cellulase hydrolyzes β –1, 4-D-glucan linkages present in the structure of cellulose and produce glucose, shorter polysaccharides and cello-oligosaccharides. Cellobiohydrolase, endo- β –1,4-glucanase, and β -glucosidase are the three major types of cellulose enzymes. These three enzymes disintegrate cellulose with several other enzymes which attack synergistically to change crystalline cellulose to glucose (Béguin and Aubert 1994). Generally, different cellulases are characterized by two domain structures including catalytic domain and cellulose binding domain or carbohydrate binding module connected through a linker peptide. Catalytic or core domain contains catalytic site while cellulose binding domains assist in binding of enzyme with cellulose, EG are regarded as the break down the crystalline structure of the cellulosic substrate (Husain et al. 2011). Tables 12.1 and 12.2 represents the interaction mode; adsorption and covalently with different nanomaterials. These cellulases immobilized on the surface of the NPs or embedded inside the polymeric nanospheres exhibit high catalytic efficiency and immobilization yield.

4 Nanoparticles (NPs) to Enhance Degradation of Lignocelluloses

Currently, scientists are looking a wide scope in the field of nanobiotechnology due to unique nature of nanoscale materials for lignocellulosic degradation. Nanoparticles (NPs) have exclusive physicochemical properties, large surface to volume ratio,

Name of support	Enzyme	Properties improvement	Reference
Silver NPs	Cellulase	Efficient in cellulose hydrolysis	Salunke et al. (2015)
Silver NPs and gold NPs	Cellulase	Heat stability and reusable up to sixth times	Mishra and Sardar (2015)
MnO ₂ NPs	Cellulase	Stability in a broad range of pH and temperatures, reusability, and cellulose stability	Cherian et al. (2015)
Functionalized MWCNTs	Cellulase	High binding efficiency and loading reusability	Mubarak et al. (2014)
Fe ₃ O ₄ @SiO ₂ NPs	Cellulase	Catalytic efficiency, temperature optima, reusability, half life, and thermal stability	Li et al. (2014); Harmoko et al. (2016); Tao et al. (2016)
Magnetic NPs	Cellulase	Effective bioethenol production using <i>Sesbania aculeate</i> biomass	Baskar et al. (2016)
Magnetic NPs and silica-coated magnetic NPs	Cellulase	Recycle up to ten step, recovery yield up to 75%	Roth et al. (2016)
Zinc-doped magnetic NPs	Cellulase from Trichoderma reesei	Immobilization of enzyme 94%, thermostability up to 80 °C, hydrolysis recorded 93%	Abraham et al. (2014)
Nickel-cobaltite	Cellulase from Aspergillus fumigatus	Thermostable for 7 h at 80 °C, and better hydrolysis performance	Srivastava et al. (2014)
3-(2-aminoethyl aminopropyl)- trimethoxysilane- magnetic Nanosphere	Cellulase	High temperature, broader pH stability, and 87% activity recovery	Zhang et al. (2015)
Fe ₃ O ₄ NPs and Fe ₃ O ₄ Alginate nanocomposites	Cellulase from Aspergillus fumigatus	Cellulase activity increased 35% and 40%, thermally stable up to 8 h at 70 °C, and retained 56% of its relative activity; in comparison to 19% for sample	Srivastava et al. (2015)
Magnetic gold mesoporous silica NPs core shell	Cellulase	Thermal stability and facilitated its long time storage (58% enzyme maintained 9 h after catalytic activity)	Poorakbar et al. (2018)

 Table 12.1
 Nanohybrids (adsorption immobilization mode) employed for minimization of steric hindrance during cellulase activity

high catalytic efficiency, strong adsorption ability as well as high surface reactivity which can be easily harnessed to modify for the purpose of enzyme immobilization and surface capping. Such a nanotechnological approach can play an important role to improve thermostability of cellulose enzymes (Srivastava et al. 2014). Nanoparticles demonstrate the potential of immobilized enzymes to reduce the steric hindrances in comparison to bulk support with enzymes (Ansari et al. 2011; Khan et al. 2011) (Fig. 12.2).

Name of support	Enzyme	Properties improvement	Reference
Chitosan-coated magnetic NPs	Cellulase from Trichoderma reesei	80% activity after 15 cycle CMC hydrolysis	Sánchez- Ramírez et al. (2017)
TiO ₂ NPs	Cellulase from Aspergillus niger	Strongly bound enzyme has better stability	Ahmad and Sardar (2014)
β-Cyclodextrin-coated Fe ₃ O ₄ NPs	Cellulase from Aspergillus niger	Stability, reusability makes immobilized cellulase is 20-fold better than free cellulose	Huang et al. (2015)
Chitosan-coated magnetic NPs	Cellulase	50% cellulose immobilized after 10 cycles and stable at 50 °C	Zang et al. (2014)
Fe ₃ O ₄ magnetic NPs	Cellulase	Storage stability 60% of activity after 6 cycle	Jordan and Theegala (2014)
Fe ₃ O ₄ magnetic NPs	Cellulase	Wide range of pH and temperature resistant	Xu et al. (2011)
Fe ₃ O ₄ @silica core-shell magnetic NPs	Cellulase	Improved heat and operational stability, and reusability	Jafari Khorshidi et al. (2016)
Amine functionalized cobalt ferrite magnetic NPs	Cellulase	The immobilized cellulase retains >64% of the activity after 6 cycles at high pH and temperature	Bohara et al. (2016)
Fe ₃ O ₄ NPs entrapped within cross-linked epoxy polymer	Cellulase	Thermal stability and reusability	Hosseini et al. (2018)
PEGylated graphene oxide	Cellulase Aspergillus oryzae	Ionic liquid stability enhanced 30 times and hydrolysis rate reached 87%	Xu et al. (2016)
Attapulgite@ chitosan nanocomposite	Cellulase	Enhanced heat stability and effective hydrolysis of wheat straw	Yang et al. (2016a)
Magnetic core-shell metal- organic framework	Cellulase	Enhanced pH stability, thermal stability, and catalytic efficiency	Hosseini et al. (2018)

 Table 12.2
 Nanohybrids (covalent immobilization mode) employed for minimization of steric hindrance during cellulase activity

CMC carboxy methyl cellulose; PEG Poly ethylene glycol

5 Immobilization of Lignolytic Enzymes on Nanoparticles

Significant technical hurdles were noted during biomass conversion because most enzymes are easily denatured due to small changes in pH, temperature and lose their activity (Mateo et al. 2007). Cellulases are strongly influenced by hydrolyzed product (Andrić et al. 2010) and a chunk amount of enzyme is consumed by slow hydrolysis before being recovered from the reaction system. Therefore, icebreaking research is needed to solve these problems because it has become a bottleneck pro-



Fig. 12.2 Sketch diagram illustrating fate of lignocellulosic biomass in the form of intermediates and metabolites with the employment of NPs

cess from an economic perspective (Honda et al. 2015). In advanced nanotechnology, nanohybrid materials are used as adsorbent, covalent, embedded, cross-linked, and encapsulated cellulase aggregates as a result of immobilized enzymes on nanosupports (Nerv and Kubota 2016; Iype et al. 2017; Sulaiman et al. 2017; Carli et al. 2019; Khan et al. 2019). These nanocomposites can be technical solutions that increase the stability efficiency and reusability of the desired enzyme. Here, we have collected some advanced results of cellulase immobilized NPs for cellulose hydrolysis. Tables 12.1 and 12.2 show that cellulases from different sources are immobilized on several metal and non-metal nanosupports by different bonding modes, and their improved applications/properties. Most of the studies expressed in Tables 12.1 and 12.2 indicate that nanomaterials have low toxicity, biocompatibility, and high affinity for surface functionalization properties (Bhardwaj et al. 2018). Among all the NPs, the magnetic nanomaterials have better choice due to recycling and reusability as well as high surface area and appropriate enzyme immobilization sites (Cha et al. 2005; Shukla et al. 2018). Because of anisotropic dipole attraction, hydrophobic nature and large surface area; uncoated nanomaterials are commonly unstable, which leads to the formation of aggregates, ultimately NPs may oxidized in presence of oxygen and lost their specific properties (Atacan et al. 2017; Bhardwaj et al. 2017b; Shukla et al. 2017). Therefore, an appropriate surface fictionalization is required with molecule as amino-salines (Zhang et al. 2015), chitosan, or polymers to overcome such limitations.

Khan et al. (2016), reported that β -galactosidase isolated from Aspergillus ory*zae*, immobilized onto graphene-iron oxide ($Gr@Fe_3O_4$) nanocomposites by simple adsorption. The prepared magnetic nanosupports are providing excellent stability against pH, temperature, and the bound enzyme retained 83% activity after eighth successive reuse (Khan et al. 2016). β-galactosidase is immobilized on bulk ZnO and nano-ZnO, and ZnO nanoparticles have significant yield and stability compared to bulk ZnO (Husain et al. 2011). Immobilized enzymes on NPs supports are similar to their use in industrial, environmental, clinical, and analytical fields. Magnetite NPs (Fe₃ O_4) have excellent biocompatibility, reusable magnetic, and cellulose immobilization (Zang et al. 2014; Chiaradia et al. 2016). Magnetic nanomaterials are used as nanosupports that ultimately recover from the reaction mixture by applying magnetic flux after the reaction is complete. Metal NPs (MNPs) have been shown to be potentially compatible in immobilizing various cellulolytic enzymes (Zhang et al. 2015; Khan et al. 2016). Thus, nanobiotechnology and nanocarriers have inherent physicochemical properties such as specific surface area, shape, size, surface adjustability, mass transfer resistance, and high binding, demonstrating useful and efficient nanoscaffolds for immobilizing enzymes (Dincer and Telefoncu 2007). In addition, it can be used to improve the deconstruction of lignocellulosic waste into valuable metabolites. A schematic mechanism of crystalline nanocellulose to product (glucose) in the presence of cellulase-based nanosupport has proposed in Fig. 12.3.



Fig. 12.3 A schematic diagram of enhanced product formation (glucose) from a crystalline cellulose using nanosupports (A) independent magnetic NPs (B) immobilization of cellulolytic enzymes on NPs (C) bioconversion of cellulose into simple sugars (D) magnetic separation of product and NPs (Ea) separated cellulosic product (Eb) separated NPs complex ready to reuse

6 Characterization of NPs

It is important to understand that how enzymatic system will interact with nanomaterials along with nanomaterials formation and stabilization. The particle size is inversely proportional to surface area and volume ratio, which allows tunable interaction between the materials and its surroundings (Reineck et al. 2019). Thus, the size and surface area of nanomaterials are basic characteristics which highly influenced enzymatic stability, yield efficiency, surface immobilization of enzyme and fate in environment. The size, shape, surface, and structural characteristics of the material are important factors because it provides important information for understanding the behavior of the interaction between the support and the enzyme (Campbell et al. 2014). Therefore, the established interaction can be evaluated and effectively adjusted in a large-scale process. To this end, several techniques are used to characterize the nanosupport (nanomaterials) to obtain morphological, structural, quantitative, and qualitative information that can help to further manipulate the appropriate interactions. The major techniques used in the characterization of the characteristic information of the support and enzyme have accounted below.

6.1 UV-Visible Spectroscopy

The electromagnetic interaction of material can be explored about shape, size, density, and concentration. Generally, the UV-VIS spectra measure the wavelength ranging between 200 and 800 nm during different stages of nanomaterials synthesis. The acquired spectra depict the characteristic peaks, comparative analysis of shape and size along with the concentration of samples (Maurya et al. 2016; Bhardwaj et al. 2018).

6.2 Fourier-Transform Infrared Spectroscopy (FTIR)

FTIR spectroscopy is a very potent technique used for the characterization of enzymatic matrix. The infrared radiation interacts with the modified nanosupport surface along with different enzymes. This interaction leads to transmission or absorption which corresponds specifically to the chemical bonds present in enzyme vibration mode. Typically, the vibration frequencies are measured as the wave numbers in between 4000 and 480 cm⁻¹. The characteristic FTIR spectra of NPs and enzyme immobilized NPs can produce characteristic vibration frequency which provides clear-cut information about interaction pattern with each other (Baby et al. 2010).

6.3 X-Ray Diffraction (XRD)

The use of XRD in the field of enzyme loaded and engineered nanomaterials study is extensively increasing due to its broad range of application. XRD is the most powerful technique to determine the crystallinity and phase composition of nanosupport. The Bragg's equation can be described by the reflection of collimated X-rays in case of material phase determination. The beam is incident on the crystal face of the sample to be characterized. XRD-based wide-angle elastic scattering, usually used for ordered materials only. X-ray beam passes through the sample and is scattered or diffracted by atoms in the X-ray path are studied. Interference caused by scattering mutual observation using Bragg's law and properly positioned X-rays the crystal structure characteristics of the detector and material was determined (Epp 2016). In this method, X-ray source is used for getting diffraction pattern, which can plot intensity verses 2θ using inbuilt software in XRD machine, consequently some peaks are observed against different nanosupport (Yang et al. 2016b). Finally, the characteristic peaks can be matched using Joint Committee on Powder Diffraction Standards (JCPDS) software. The particles size can also be calculated using Debye Scherrer's formula given as below

$$D = \frac{K\lambda}{\beta\cos\theta}$$

where the constant K = 0.97, $\lambda = 0.6255$ Å is the wavelength of X-ray line used and β is the full width at half maximum.

6.4 X-Ray Photoelectron Spectroscopy (XPS)

This is one of the highly acceptable techniques used for the surface elemental composition analysis of nanosuport modified by enzymes, known as X-ray photoelectron spectroscopy (XPS). The XPS spectrum provides an elementary analysis of the nanosupports based on the photoelectric effect permits the characterization of the sample up to the 10 nm depth. The characteristic peaks are observed at different region and voltage (electron volts) of sample, which may represent a particular element over the nanosupport surface to determine the information about immobilized enzymes (Wilson and Langell 2014). Almost all elements of the materials can be recognized using XPS spectra excluding the lighter atoms like hydrogen and helium. The low energy X-ray beam bombarded over samples induces photoelectron emission through the excitation of electrons (Wang et al. 2012). The emission is further captured by the detector to analyze on the basis of emission kinetic energies which is the characteristics of the particular atom (Xue et al. 2011). The quantitative and qualitative analysis of function of kinetic energy also provides the ideas referring to the valance and the chemical bonds available in enzyme-modified nanosupports.

6.5 Transmission Electron Microscopy (TEM)

Transmission electron microscopy (TEM) is used for getting clear information characterizing shape, size, and structure of nanosupport. TEM is the preferred method for particle size analysis and particle size distribution of NP samples because it is capable of imaging particles from 1 nm to a few microns. This technique is also used to observe the morphology of NPs. Because of its ultra-high resolution capability, it is possible to study particle surfaces particle size or even lattice structure. The image is formed just like simple light microscope but in this technique the electron source is used to illuminate NPs with higher resolution (Zhu et al. 2017). The colloidal and diluted sample is put over a grid, this thin sample kept between path of focused electron beam and after the transmission of electron fall over the fluorescent screen and gives black and white picture (Shukla et al. 2018). The sample atom containing with high density of electron and low density leads to forming dark contrasts and clear region, respectively. Moreover, the enzymatic immobilization is responsible for change in the characteristic SEAD pattern (Khan et al. 2016). Selected area (electron) diffraction can also be analyzed to elucidate the crystalline structure of nanosupport (He and Gao 2010).

6.6 Scanning Electron Microscopy (SEM)

This technique is use for viewing surface morphology, topography of nanosupport modified with enzyme. This technique differ from TEM, sample atom excited by the electron beam and generate electron from the sample surface known as secondary electron instead of electron beam transmit through the sample. The conducting sample surface is responsible for generating secondary electron, which leads to provide information about material surface information with less than a nanometer resolution. Field emission SEM has much better resolution than normal SEM (Bhardwaj et al. 2017a).

Scanning electron microscopy (SEM) is one of the most widely used instrumental methods for the analysis and study of nano- and micro-scale imaging of solid materials. One of the reasons why SEM is preferred for particle size analysis is that its resolution is 10 nm, or 100 Å (Goldstein 2012). The updated versions of these instruments can achieve resolution about 2.5 nm (25 Å). Electron beam acceleration is performed by a high voltage system (20 kV) and the electron beam narrows after passing through the aperture and the electromagnetic lens. The beam scans the surface of sample with the aid of a scanning coil. After interaction of incoming electron beam with the specimen; it emits beam of secondary electrons, which are detected by the detectors, fitted suitably in vacuum chamber and consequently, image of specimen is generated by the signal from the beam area (Hafner 2007). The instrument can also be combined with other related energy dispersive X-ray microanalysis techniques (EDX, EDS, EDAX), used to determine the composition or orientation of a single crystal or feature. Typically, tungsten or field emission guns are used as a source of electron generation. To keep the tip free from contaminants and oxides, field launch guns require ultra-high vacuum conditions (10^{-10} to 10^{-11} Torr) (Pathak and Thassu 2016). The most appropriate and real environmental sample can be analyzed without taking care of moisture by environmental SEM; however, resolution will decrease. The inbuilt EDS analysis tool is also used for the mapping and elemental composition of composites.

6.7 Vibrating Sample Magnetometer (VSM)

The purpose behind the use of vibrating sample magnetometer (VSM) is to measure the induced magnetization of the magnetic nanosupport with and without enzyme. Nanosupport is mechanically vibrated in a uniform magnetic field. This field magnetize sample by aligning the magnetic domains, or the individual magnetic spins, with the field if the sample is magnetic in nature (Foner 1959; Shukla et al. 2017). The stronger is the constant field, the larger will be the magnetization, which creates a net magnetic dipole moment (Wu et al. 2012). This dipole moment of the nanosupport will create a magnetic field around the sample and sometimes called the magnetic stray field. As the sample is swipe from about -15,000 to 15,000(H) g at room temperature, this magnetic stray field will change as a function of time and can be sensed by a set of pick-up coils (Le Goff and Gallet 2004; Shukla et al. 2018). The alternating magnetic field creates an electric field in pickup coils according to Faraday's law of electromagnetic induction (Su et al. 2015). This current is proportional to the magnetization of the sample. Therefore, the greater the magnetization, the greater would be the induced current (Mahdavi et al. 2013).

6.8 Circular Dichroism Spectroscopy (CDS)

The CSD technique is the most powerful technique widely used in the characterization of surface proteins. The CSD work based on the principle of differential absorption of light circularly polarized by optically active molecules and allows determining its absolute configurations. As light passes through a stereogenic center present in the sample, there will be differential absorbance of two circulating components to right or to left. The technique is non-destructive, easy to operate, fast, and requires low sample quantities and can be used under room temperature conditions. The temperature may change the amino acid or protein structure; this conformational change can be studied using CDS measurements. The characteristic peak of α -helix secondary structure of enzyme gets at 220 nm. This technique provides the information about the quantification of the secondary structure of protein in the form of the α -helical, β -strand, and other structure using the far UV range of the EM spectrum between 170 and 260 nm. The aromatic residues of the protein molecule can also be observed in UV range between 260 and 300 nm. The peptide bond shows two transition found in the far UV n- π * at ~220 nm and a π - π * transition at ~190 nm when ground state electrons to an excited state, in protein molecules during absorption of UV. The sum of all the signals from surface protein is determining the characteristic signal of particular immobilized enzyme. Thus, the technique has wide potential to investigate structural change takes place during the immobilization of the enzyme over the support.

7 Conclusion and Future Prospects

The global need for renewable materials as an alternative to nonrenewable energy sources have best opportunity in the development of a environmentally benign technology. The immobilization cellulolytic enzyme over different kinds of NPs leads to enhance yield and stability of enzymes. The sufficient evidence has been discussed with the help of several findings of the work in this chapter. It is observed that among all the nanomaterials the magnetic NPs immobilized enzymes provide better response than free enzyme and enhance the stability against various chemical and physical denaturants. Several NP conjugates have found remarkable stability, high yields, and better recovery to obtain a range of value-added products, especially different fuels and materials. Their application is not limited to biomass conversion; it can be used in the fields of biodiesel, medicine, sensors, cosmetics, industrial production, and bioremediation. Despite the development of smart nanohybrid materials, it still relies on magnetic materials to improve the efficiency and recovery of nanocarriers during biomass conversion. Of course, when we overcome this goal, it is indeed a real step towards sustainable development, which helps to reduce dependency over nonrenewable resources. Lignocelluloses rich in agricultural waste materials would be frequently use to produce a large number of industrial

products such as organic acids, ethanol, and other industrially important compounds. The application of such kind of materials is not confined and the nanomatrix could be designed for a powerful bio-recognition probe in the field of biosensing when used in different enzyme systems. This nanocomposite with a range of fascinating functionality can also explore field of drug design, drug delivery, and biosensing applications.

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