

# Chapter 15

## Effects of Salt Stress on Photosynthesis Under Ambient and Elevated Atmospheric CO<sub>2</sub> Concentration

Nicole Geissler, Bernd Huchzermeyer, and Hans-Werner Koyro

**Abstract** Soil salinization is one of the most important factors which limit plant productivity. About 3.6 billion of the world's 5.2 billion hectares of dryland used for agriculture have already suffered erosion, soil degradation, and salinization. Global climate change caused by rising atmospheric trace gases such as CO<sub>2</sub> and forced migration add to the urgency of this global problem. Therefore, solutions are desperately needed, such as the improvement of drought and salinity resistance of crops or the use of (xero-) halophytes instead of glycophytic crops. As photosynthesis is a prerequisite for biomass production, this chapter focuses on information related to this essential sequence of reactions, thereby discussing the different levels of photosynthesis. At first, there are primary reactions of photosynthesis, namely absorption of light energy and (1) its conversion to redox energy, conserved in the coenzyme NADPH, and (2) energy of chemical bonds, conserved in the coenzyme ATP. On the second level, we find reactions of the Calvin cycle, nitrate and sulfate reduction as well as sugar, lipid, and amino acid metabolism. Typical reactions on the third level are transmembrane and inter tissues transport of metabolites. The fourth level of photosynthesis relates to physiological aspects of gas exchange and water relations.

Apart from these general effects of salinity on photosynthesis, we will review the probable photosynthetic performance of salt stressed plants under future atmospheric conditions, namely under elevated CO<sub>2</sub> concentration. Special emphasis will be put on gas exchange and photosynthesis of C<sub>3</sub> and C<sub>4</sub> plants because these two photosynthesis types show different responses to elevated CO<sub>2</sub>, leading to different interactions with salinity.

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**Keywords** Na Cl salinity • Global change • Elevated CO<sub>2</sub> • Photosynthesis • Oxidative stress

## Introduction

Feeding the rapidly growing human population is one of the biggest challenges we face on our planet today. Especially in arid and semiarid regions, the increasing population density is collaterally catalysed by the consequences of global climate change which is caused by anthropogenic emissions of trace gases such as CO<sub>2</sub> (IPCC 2007). Thus salinity – already affecting 7% of the global total land area (Glenn et al. 1998; Millennium Ecosystem Assessment 2005) and 20–50% of the global irrigated farmland (Tanji 2002; Hu and Schmidhalter 2005; Koyro et al. 2008) – will pose a more and more serious threat to agriculture and human nutrition in future. Therefore, solutions are desperately needed, such as the improvement of salt resistance of crops or the use of (xero-) halophytes instead of glycophytic crops, which in turn requires a detailed knowledge about salt and drought resistance mechanisms in plants.

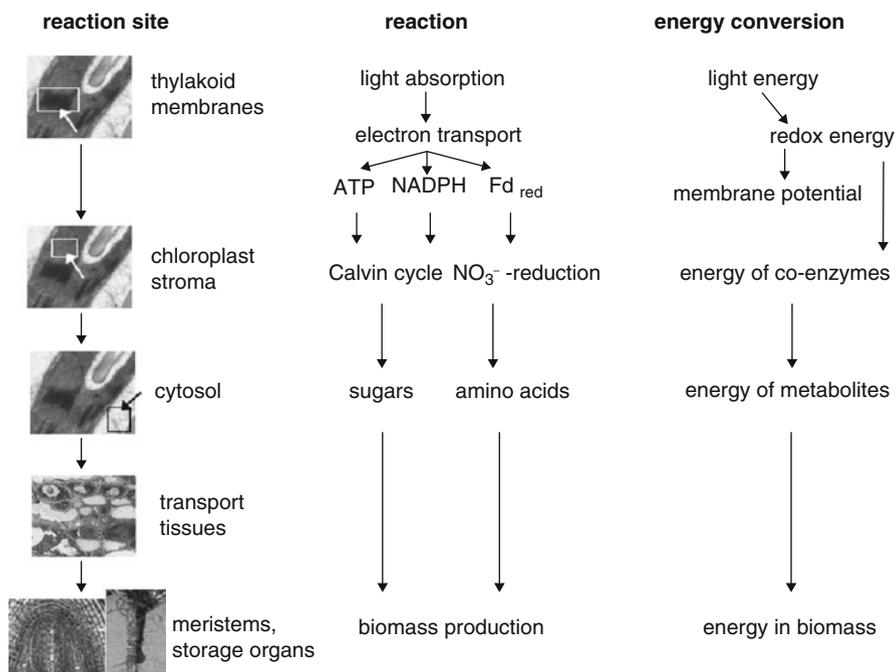
The viability of plants in saline habitats depends on their ability to cope with (1) water deficit due to a low water potential of the soil, (2) restriction of CO<sub>2</sub> uptake, (3) ion toxicity, and (4) nutrient imbalance. The first two constraints are directly related to photosynthesis which is a prerequisite for biomass and crop production, respectively. So this chapter focuses on information related to these constraints, thereby discussing the different levels of photosynthesis as described in the next chapter.

In the face of global climate change, one has also to consider the influence of elevated atmospheric CO<sub>2</sub> concentration on photosynthesis, which we will do in the last part of this chapter. In C<sub>3</sub> plants, CO<sub>2</sub> often improves assimilation while reducing stomatal resistance, thus increasing water use efficiency, but decreasing photorespiration and oxidative stress (Urban 2003; Kirschbaum 2004; Rogers et al. 2004). As these effects are much less pronounced in C<sub>4</sub> plants, it will be differentiated between these two photosynthesis types.

## Effects of Salt Stress on Photosynthesis Under Ambient Atmospheric CO<sub>2</sub> Concentration

### *Levels of Photosynthesis to Be Considered*

Regarding crop production and food and feed supply, limitation of plant growth by environmental factors is a matter of general concern. As photosynthesis is dominating plant growth and production of biomass, the sequence of reactions leading to the phenomenon named photosynthesis is in the focus of interest when breeding for high crop yield. In order to allow a detailed analysis of salt effects, several individual steps have to be distinguished (Fig. 15.1).



**Fig. 15.1** Photosynthetic energy flow. The photosynthetic reaction sequence can be described in terms of reaction sites (*left*), sequence of reactions (*center*), and conversion of energy forms (*right*). *Fd* ferredoxin

- At first, there are primary reactions of photosynthesis, namely absorption of light energy and (1) its conversion to redox energy, conserved in the coenzyme NADPH, and (2) energy of chemical bonds, conserved in the coenzyme ATP.
- On the second level, we find reactions of the Calvin cycle, nitrate and sulfate reduction as well as sugar, lipid, and amino acid metabolism.
- Typical reactions on the third level are transmembrane and inter tissues transport of metabolites.
- The fourth level of photosynthesis relates to physiological aspects of gas exchange and water relations.

Most papers deal on salt effects on physiological aspects of biomass production rather than partial reactions of photosynthesis as defined above. As there are cross reactions as well as cell signaling involved in regulation of photosynthetic metabolic pathways, there are no strict correlations between biomass production and individual gene activities. As a consequence, predicted correlations, though they had been observed in laboratory experiments, could not be shown in subsequent field experiments. In this chapter we try a more differentiated approach and review analysis of salt stress effects at different levels of photosynthesis.

## ***Regulation of Gas Exchange and Water Relations***

According to Munns, plants show a two-phase growth response to salinity (Munns 1993, 2002; Munns et al. 2002). During the first phase of growth reduction water or osmotic stress prevails, and plant responses are presumably regulated by hormonal signals coming from the roots.

Terrestrial plants growing in saline habitats are often surrounded by low water potentials in the soil solution and atmosphere. As a part of resistance mechanisms, water flow through the soil–plant–atmosphere continuum must be ensured, so that a gradient of decreasing water potentials ( $\Psi$ ) must be established. The  $\Psi$  of pure water is defined as 0 MPa; increasing salinity or concentrations of other solutes will decrease  $\Psi$ . Thus, any sharp rise in salinity could effectively hold water osmotically away from plants that lack any physiological or morphological modifications (Larcher 2003). To limit restrictions on water uptake, plants must generate increasingly lower  $\Psi$  to allow continued water flux into belowground structures (Touchette et al. 2009).

If plants cannot escape salinity (such as geophytes or pluviotherophytes), it is also important to prevent water loss by transpiration from being higher than the influx rate. This is necessary to avoid a negative tissue water balance and is only possible if the water potential remains lower in the plant than in the soil. However, experiments demonstrated that the leaf water potential of halophytes does not correlate alone as a single factor with salinity resistance. Plant species with different levels of salt resistance such as *Aster tripolium*, *Avicennia marina*, *Innula critmoides*, and *Sesuvium verrucosum* have a sufficient adjustment mechanism even at high salinities. Clearly, resistance in the form of osmotic adjustment plays an important role in halophytes residing in saline environments (Flowers and Colmer 2008). However, in addition were the osmotic potentials of all four halophytes (and many others) up to sea water salinity level sufficiently low to explain the full turgescence of the leaves (results not shown). Except for differences in concentration and type of osmotica used in plant tissues (in halophytes inorganic ions are more prevalent than organic substances), physiological responses in plants to salt stress were remarkably similar to those employed during drought (Touchette et al. 2009). For plants with limited water availability, physiological adjustments often involve avoidance and tolerance, with most plants using some combination of the two (Yue et al. 2006; Romanello et al. 2008).

Assuming there is no interruption of the water supply, water can flow passively from the root to the shoot and there seems to be no reason for growth reduction by water deficit for any of the studied species. However, by regulating the extent of apoplastic barriers and their chemical composition (long-distance response coordination), plants can effectively regulate the uptake or loss of water and solutes (by structures such as barriers in the hypo- or exodermis). This appears to be an additional or compensatory strategy of plants to acquire water and solutes (Hose et al. 2001). At the extremes of growth under saline or dry conditions even cell layers such as the exodermis can become an absolute barrier for water and ions in the strict sense (Azaizeh and Steudle 1991; North and Nobel 1991; Nublat et al. 2001).

Thus, the rate of water supply to the shoot can be restricted due to the coupling between the flows of water and solutes (Na and Cl) even if the leaf water

potential is low. Therefore, the balance between water flow (sum of water accumulation and transpiration) and the decrease in the amounts of nutrients or unfavorable nutrient ratios (e.g.  $\text{Na}^+/\text{K}^+$ ) are important factors for impaired leaf elongation and plant growth (Lynch et al. 1988; Munns et al. 1989; Neves-Piestun and Bernstein 2001; Cramer 2003).

In any case, plant water loss has to be minimized at low soil water potentials. However, biomass production depends mainly on the ability to keep a high net photosynthesis and a low transpiration simultaneously. In this field of tension, biomass production has always to be seen in connection with  $\text{CO}_2/\text{H}_2\text{O}$ -gas exchange, which can be estimated by the water use efficiency (WUE) of photosynthesis. It needs to be considered that anyway a critical point for the plant is reached when  $\text{CO}_2$  fixation (apparent photosynthesis) falls below  $\text{CO}_2$  production (compensation point).

Several halophytic plants such as *Aster tripolium*, *Beta vulgaris* ssp. *maritima*, *Chenopodium quinoa*, or *Spartina townsendii* reveal a combination of low (but positive) net photosynthesis, minimum transpiration, high stomatal resistance, and minimum internal  $\text{CO}_2$  concentration at their threshold salinity resistance (Koyro 2000; Koyro and Huchzermeyer 2004). However, there is a big bandwidth among halophytes, especially for succulent species such as *Sesuvium portulacastrum* or *Avicennia marina*, which have alternatives if the water balance (water uptake minus water loss) is still positive and not the limiting factor for photosynthesis. In case of *S. portulacastrum*, net photosynthesis and WUE increase but stomatal resistance decreases. These results show that it is quite important to describe the regulation of gas exchange at high salinity in strong reliance with other parameters (such as water relations).

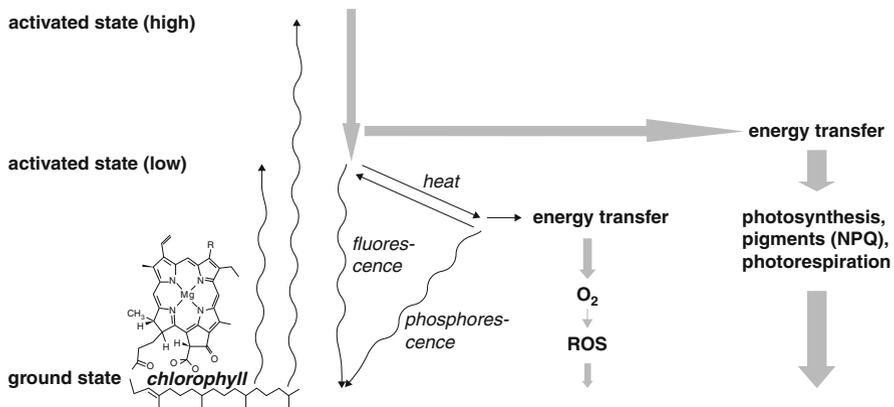
Water deficit is one major constraint at high salinity and can lead to a restriction of  $\text{CO}_2$  uptake and to the development of reactive oxygen species (ROS). The balance between water loss and  $\text{CO}_2$  uptake helps to find weak spots in the mechanism of adjustment (of photosynthesis) to high salinity (Badawi et al. 2004).

## **Primary Reactions of Photosynthesis**

### **Photosynthetic Conversion of Energy**

In plants active in photosynthesis, energy of light quanta is absorbed by chlorophyll. Just like all other pigments, activated chlorophyll can return from its activated state to the stable, non activated state by emitting heat. Other than most pigments, activated chlorophyll is sufficiently stable to allow transfer of energy to acceptor molecules of biological relevance after having absorbed energy of a red light quantum (Strasser et al. 2004; Fig. 15.2). In principle, there are two options of biological relevance in addition to a third one we can use to monitor plant performance:

1. Resonance energy transfer to activate other pigments and
2. Transfer of an electron to an acceptor. The electron acceptor can be a component of the photosynthetic electron transport chain or any other molecule having a less negative potential as compared to activated chlorophyll. This second option requires a “refill” of electrons, and it is well documented that in chloroplasts the



**Fig. 15.2** Competition for absorbed energy of light quanta. Photosynthetic electron transport is competing with “futile reactions” for energy of light-activated chlorophyll. *ROS* reactive oxygen species, *NPQ* non-photochemical quenching

water splitting system fulfills this function, so that chlorophyll is recycled to its ground state (Strasser et al. 2004).

3. Another option, competing for energy with the already mentioned ones, is the emission of fluorescence energy (Schreiber 1997; Schreiber et al. 2002; Strasser et al. 2004). As will be discussed later, any inhibition of an individual pathway will enhance the possibility of the other pathways to occur.

In thylakoid membranes, photosystems I and II (PSI and PSII) can be distinguished from other chlorophyll containing protein complexes, namely the light-harvesting complexes. Only the special pairs of chlorophyll a located in the active center of the two photosystems are involved in electron transport. Energy transfer among other pigments occurs via resonance energy transfer. In this section, we focus on electron transfer reactions.

As known from the literature, half lifetime of activated state of chlorophyll is very short and depends on the environment of the pigment (Rees et al. 1990; Laible et al. 1994; Ma et al. 2009). It is obvious that photosynthetic efficiency depends on the probability that activated chlorophyll will transfer electrons to an acceptor of the photosynthetic electron transport chain rather than “wasting” energy by using one of the other pathways mentioned above (Ruban et al. 1993; Horton et al. 1996; Horton 2000; Allen and Forsberg 2001). In order to meet this requirement, reaction partners are arranged in ideal neighbourhood within protein complexes located in the thylakoid membranes. Moreover, it has been demonstrated that this structure undergoes permanent adjustment to match the requirement of chloroplast metabolism and to adapt to changes in the environment, i.e. changes of light quality and intensity, for instance (Allen and Bennett 1981; Allen 1992; Allen and Forsberg 2001).

By means of photosynthetic electron transport, energy of absorbed light quanta is converted to redox energy stored in the coenzyme NADPH and proton motive



## Salt Effects on Photosynthetic Energy Conversion

In order to understand individual photosynthetic reactions and get the principles of their interaction, thylakoid membranes and protein complexes have been isolated and analyzed with respect to their structure and function. During preparation and subsequent tests of enzyme activities, salt concentrations up to 50 mM NaCl have been applied without any inhibitory effect on individual enzyme activities (Strotmann et al. 1976). Such high salt concentrations are not found inside chloroplasts, neither under physiological conditions nor under salt stress. It therefore can be concluded that primary reactions of photosynthesis are not directly inhibited by ionic or ion specific stress (Richter et al. 2000; Huchzermeyer et al. 2004; Huchzermeyer and Koyro 2005; Koyro and Huchzermeyer 2005). However, this conclusion disagrees with an apparent inhibition of photosynthesis observed in whole plant experiments (Lawlor and Fock 1978; Lawlor 2002a). Therefore, it has to be analyzed in more detail to what extent the observed inhibition may be attributed to salt-dependent changes in thylakoid structure (Hesse et al. 1976) or to physiological effects (e.g. ion transport). One first approach to answer this question could be a detailed analysis of the kinetics of fluorescence light emission subsequent to chlorophyll activation by light pulses. Application of the pulsed amplitude modulation technique (PAM) allowed a detailed analysis of the network of primary reactions in photosynthesis (Strasser et al. 2004). It became quite obvious that salt stress does not directly inhibit primary reactions of photosynthesis, but inhibits product export and fine-tuning of primary reactions by interfering with optimal arrangement of proteins and membranes (Koyro and Huchzermeyer 1999; Huchzermeyer and Heins 2000; Huchzermeyer 2000).

Accordingly, it was observed that maximal photochemical efficiency, indicated by high  $F_v/F_m$  values of chlorophyll fluorescence, remain high under tolerable salt stress, while the growth rate of turf grass, for instance, was reduced under the same salinity level (Lee et al. 2004). This can be seen as an evidence of a reduced real photochemical efficiency.

It will be discussed in subsequent paragraphs how salt can inhibit export of products of photosynthesis and why proper function of the phosphate translocator, located in the inner envelope membrane, is essential for photophosphorylation. At this stage, we can state that inhibition of product export feeds back to primary reactions and finally will inhibit photosynthetic electron transport. One option to release energy from its activated state is blocked and chlorophyll will increase activity of fluorescence light emission, heat production, energy transfer to other pigments, and ROS production. This latter option will be discussed later in this review (see Sect. Dissipation of Surplus Energy and Production of Potential Toxic Intermediates of Photosynthesis).

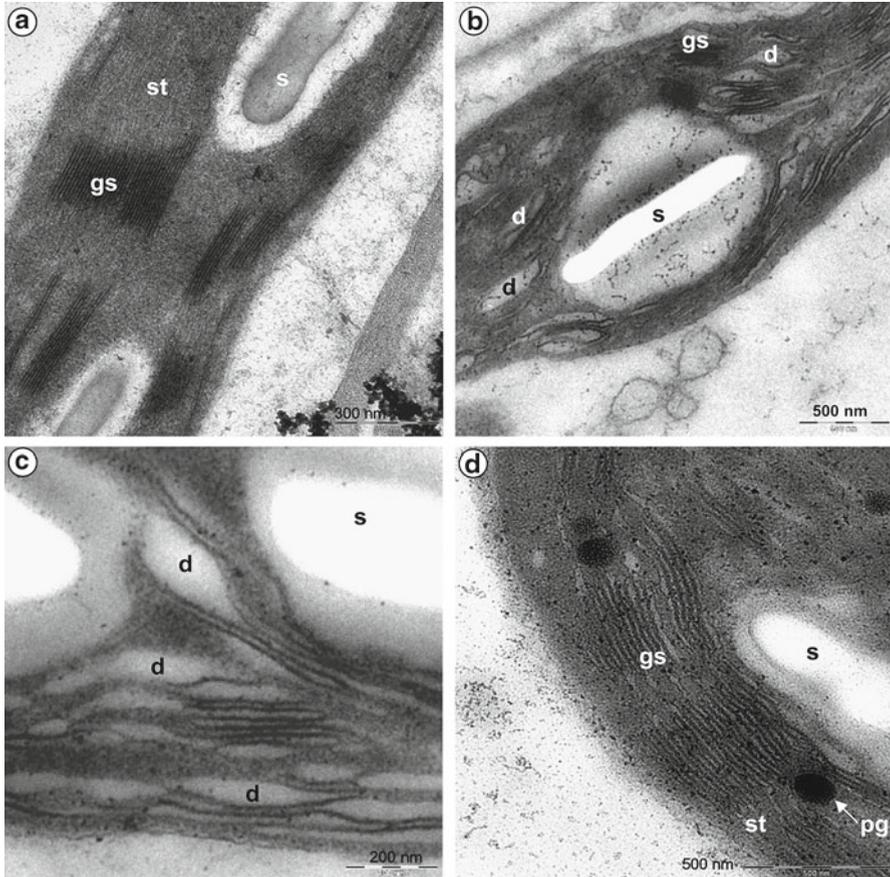
## Salt Effects on Chloroplast Structure and Metabolite Transfer

Chloroplasts exhibit a very high content of proteins which tend to interact and form aggregates inside the chloroplast stroma (Süss et al. 1993). Süss et al. were able to

isolate such “super complexes” and found out that they contain enzymes belonging to individual metabolic pathways. Such an arrangement helps to increase substrate turnover because diffusion distances among enzymes of a pathway are close to zero. For the same reason, formation of such aggregates prevents occurrence of side reactions because intermediates are not available for other enzymes. As will be discussed in Sect. Control of Mechanisms Improving Stress Resistance: Compatible Solutes, on the one hand salt can inhibit the turnover of substrates by destroying enzyme aggregates, while on the other hand intermediates of sugar metabolism can become available for other enzymes and alternative products such as compatible solutes can be formed. Though salt effects on metabolite patterns have been analyzed in several papers, no data linking these findings to the occurrence of protein aggregates have been available up to now.

The occurrence of thylakoid grana stacks has attracted a lot of interest for years (Huchzermeyer et al. 1986; Lam and Malkin 1989; Malkin and Braun 1993; Romanowska and Albertson 1994). It was calculated that about 70% of all PSII complexes are located within stacked regions of the thylakoid membranes, while most of the PSI complexes are found in un-stacked parts of the thylakoid membranes (Schmidt and Malkin 1993; Romanowska and Albertsson 1994). From this, it was concluded that 70% of the PSII complexes must be in an inactive state because the distance between PSII and PSI would be too far to allow sufficient turnover rates of photosynthetic electron transport (Malkin and Braun 1993). Moreover, it was found that grana are unstable, permanently folding and unfolding structures. Apparently CFOCF<sub>1</sub>-complexes are initiating the formation of membrane loops (Boekema et al. 1988), and grana are formed by the interaction of membrane proteins (Staehelin 1975). As active PSII has an extremely short half lifetime in the range of 20–30 min (Kuhn and Böger 1990; Trebst and Soll-Bracht 1996; Keren et al. 1997; Jansen et al. 2001), the function of this permanent modification of membrane structure may be to initiate interactions among membrane proteins and to support the formation of aggregates of protein complexes interacting in the photosynthetic electron transport chain. Such preferred neighbourhood of membrane proteins has been found, indeed (Laszlo et al. 1984; Huchzermeyer and Willms 1985). There are several publications indicating that such neighborhoods of protein complexes control the efficiency of energy conversion in primary reactions of photosynthesis (Laszlo et al. 1984; Löhr and Huchzermeyer 1985; Löhr et al. 1985; Allnutt et al. 1989) and photo-inhibition (Lu and Zhang 1999; Lu et al. 2002).

The facts presented above show that the formation of grana stacks is essential for optimal functioning and permanent repair of the photosynthetic electron transport rate. It was found that grana become destabilized if the ratio of monovalent and divalent cations is impaired (Hesse et al. 1976). Similarly, investigations on *Aster tripolium* showed that NaCl salinity leads to a decreased amount of grana stacks (Geissler et al. 2009b; Fig. 15.4). This was accompanied by an increased chlorophyll *a*/chlorophyll *b* ratio and may be an explanation for a reduced photosynthetic efficiency (Moorthy and Kathiresan 1999) because chlorophyll *b* is mainly located in the light harvesting complex which supplies PSII with energy and is located



**Fig. 15.4** Influence of salinity and elevated  $\text{CO}_2$  concentration on chloroplast structure in *Aster tripolium*. (a) In control plants under ambient  $\text{CO}_2$ , both stroma thylakoids and grana stacks are well developed. (b), (c) In salt treatments (375 mM NaCl) under ambient  $\text{CO}_2$ , dilations of the thylakoid membranes are clearly visible, and well-developed grana stacks are almost lacking. (d) In salt treatments (375 mM NaCl) under elevated  $\text{CO}_2$ , the thylakoid membranes are less damaged. *st* stroma thylakoid, *gs* grana stacks, *d* dilations, *s* starch grain, *pg* plastoglobulus

mainly in the grana stacks. So the disintegration of the latter is probably correlated with the increased chlorophyll a/chlorophyll b ratio and may be an adaptation to prevent an electron surplus in the photosynthetic electron transport chain.

Last but not least it should be mentioned that in various plant species salt stress often leads to dilations of the thylakoid membranes, so that the spaces between the membranes look swollen, and undulated thylakoid areas develop (Kurkova et al. 2002; Rahman et al. 2002; Mitsuya et al. 2003; Fidalgo et al. 2004; Paramanova et al. 2004; Geissler et al. 2009b; Zhen et al. 2011; Fig. 15.4). This structural damage has been discussed by many authors as a consequence of oxidative stress

(Mitsuya et al. 2003; Fidalgo et al. 2004; Oksanen et al. 2005) and is likely to contribute to impaired photosynthetic rates under saline conditions.

## ***Dissipation of Surplus Energy and Production of Potential Toxic Intermediates of Photosynthesis***

### **Photorespiration**

Stomatal closure and a subsequent inhibition of gas exchange is a secondary effect of salt stress, mostly brought about by ABA released from the plant roots. In the presence of light, the  $O_2/CO_2$  ratio will increase inside the leaves and impair  $CO_2$  fixation, especially in  $C_3$  plants. This happens as the enzyme Rubisco can bind  $O_2$  instead of  $CO_2$  to its reaction center, thus catalyzing the synthesis of a  $C_3$  plus a  $C_2$  compound instead of two  $C_3$  compounds in the primary reaction of the Calvin cycle (Fig. 15.3). The  $C_2$  compound 2-phosphoglycolate will be converted to glycolate, a molecule that cannot be metabolized by chloroplasts and the concentration of which eventually would become toxic. Detoxification and recycling of the  $C_3$  compound 3-PGA are understood to be the major function of photorespiration which takes place by metabolite transfer between chloroplasts, peroxisomes, and mitochondria.

The photorespiratory reaction cycle we know from our textbooks is a simplification allowing general estimates. We know that amino acids (glycine, serine, glutamic acid, and glutamine) may be subtracted from or fed into the cycle *in vivo*. Such reactions modify nitrogen flow among cell compartments. But, with respect to equilibrium of carbon flow, another aspect may be more important: It has been shown by Niessen et al. that mitochondrial glycolate oxidation contributes to photorespiration in higher plants as well (Niessen et al. 2007). It has to be kept in mind that photorespiration is a major source of  $H_2O_2$  in illuminated  $C_3$  leaves. On the other hand,  $H_2O_2$  production and interaction with pyridine nucleotide coenzymes make photorespiration an important player in cellular redox homeostasis. Furthermore,  $H_2O_2$  is an important second messenger controlling cell development and tuning effects of hormones like ABA, for instance. Any interference with this reaction will have effects not only on immediate energy status and phosphorylation efficiency, but also on plant hormonal responsiveness, thus on development and plant life cycle (Foyer et al. 2009).

There is evidence that glycolate can inhibit  $Q_A/Q_B$  electron transfer in PSII (Petrouleas et al. 1994). As stated above, this would lead to stimulated ROS production. This interpretation is in line with the observed bleaching of maize in presence of glycolate (s.a.). In several publications, it is suggested that the function of photorespiration is to serve as a sink to dissipate excess redox energy (Kozaki and Takebe 1996; Wingler et al. 2000). In terms of our arguments, this would mean photorespiration is recycling coenzymes functioning as acceptors of the photosynthetic electron transport chain. This would help preventing ROS production as well. But, to our understanding, this interpretation does not sufficiently take into account the compartmentation of coenzymes.

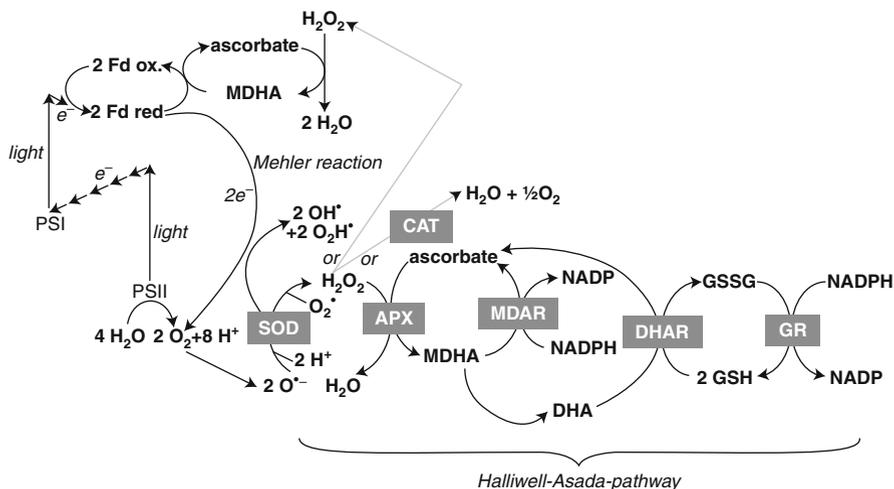
## Non-photochemical Quenching of Energy

Under conditions that limit CO<sub>2</sub> assimilation, the potential rate of NADPH production exceeds the actual rate of consumption of reductive power. In order to be able to grow under stressful conditions, plants have to be equipped with mechanisms preventing excess reducing power. But these futile mechanisms compete with photochemistry for absorbed energy. They lead to a decrease in quantum yield of photosystem II (Genty et al. 1989; Cha-um et al. 2012).

The photosystem II antenna is highly flexible in tuning delivery of excitation energy to the photosystem II reaction center (Horton et al. 1996). The principal adaptation mechanism in photosynthesis is the control of thermal dissipation of excess energy within the photosystem II antenna, thus matching physiological needs (Johnson et al. 2009). In C<sub>3</sub> plants, losses by this mechanism, named non-photochemical energy quenching, may exceed the ones caused by photorespiration. Despite extensive investigations, the reaction mechanism of photochemical quenching is not yet completely understood because the turnover of intermediates is fast and the reaction depends on intact structures of protein complexes and their *in vivo* arrangement inside the thylakoid membranes; i.e. reaction partners may not be extracted and individually analyzed. But some insight was achieved by a combination of molecular biological and biophysical techniques (Johnson et al. 2009).

For experimental approaches investigating salt stress effects, it is important to know that non-photochemical quenching of excitation energy is comprised of a fast and a slow component, qE and qI, respectively. Both reactions are reversible. The trigger of qE is the ΔpH across the thylakoid membrane sensed by the PsbS subunit of the light-harvesting complex (Li et al. 2000, 2004). Full expression of qE is associated with the enzymatic de-epoxidation of violaxanthin to zeaxanthin. This reaction is part of the xanthophylls cycle (Havaux et al. 2007). Enzymes involved are pH controlled and function on the expense of NADPH (Demmig-Adams and Adams 1996). This makes the cycle a futile reversible reaction sequence on its own. The majority of photoactive xanthophylls is bound to the light-harvesting complex. In addition to the ones involved in the xanthophyll cycle, lutein and lutein epoxide are bound there as well and can be turned over in a cycle on their own (Matsubara et al. 2001). Depending on distances among pigments, these two cycles can interact with soluble xanthophylls and control non-photochemical energy quenching in a synergistic, but not well-understood way (Johnson et al. 2009).

Sensitivity of non-photochemical quenching to any experimental approach interfering with membrane structure and protein fine structure indicates that this reaction sequence of outstanding physiological importance will be highly sensitive to any reaction causing imbalance of ionic homeostasis. The threshold ion concentration resulting in significant inhibition of non-photochemical quenching will depend on availability of compatible solutes, for instance. Based on current understanding, it can be expected that such adverse effects can be monitored by measuring pigment shifts due to altered ratios of xanthophylls and by high-resolution chlorophyll fluorescence measurement. Thus, it should be possible to measure salt effects on non-photochemical quenching even in the field using noninvasive techniques.



**Fig. 15.5** ROS production and detoxification. Detoxification of ROS by sequential ascorbate and glutathione cycles will consume NADPH and, thus, result in a relief of NADP<sup>+</sup> shortage in high light. A prerequisite is that (1) enzymes involved are available at ample concentrations and (2) are positioned in ideal neighbourhood to allow high turnover rates. *SOD* superoxide dismutase, *Fd* ferredoxin, *CAT* catalase, *APX* ascorbate peroxidase, *MDAR* monodehydroascorbate reductase, *DHAR* dehydroascorbate reductase, *GR* glutathione reductase, *MDHA* monodehydroascorbate, *DHA* dehydroascorbate, *GSSG* oxidated glutathione, *GSH* reduced glutathione

Especially near the end of a leaf's life cycle, masking of chlorophyll by anthocyanins becomes important. It prevents photooxidative damage and allows an efficient nutrient retrieval from leaves to storage organs (Field et al. 2001). Some plants use such mechanisms regularly under saline conditions. They can be identified because leaf color will vary depending on their growth conditions like it can be observed with *Salicornia* and *Sempervivum*, for instance.

## ROS Production

As described above, any reduction of the electron transport rate, especially under high light conditions, will increase the risk of ROS production (Figs. 15.2 and 15.5). The redox potential of activated chlorophyll is more negative than the one of oxygen. Therefore electron transfer from activated chlorophyll can occur unless energy is abstracted from activated chlorophyll faster than electron transfer to oxygen can occur. There are further options of ROS production as several intermediates of the photosynthetic electron transport are radicals. On the other hand, ROS can spontaneously convert or can be turned over under the control of enzymes (Fig. 15.5). Accordingly, in the literature the occurrence of various forms of ROS is described (Halliwell and Gutteridge 1986; Blokhina et al. 2002). Cytotoxicity may be attributed

to oxidative damage of membrane lipids (Fridovich 1986; Wise and Naylor 1987) as well as oxidation of proteins and nucleic acids (Fridovich 1986; Imlay and Linn 1988).

In the field, salt stress results in severe damage especially in situations when its inhibitory effects occur in the presence of high light intensity. Then PSII activity will result in high oxygen concentrations, especially if stomata are closed under stress. Concomitantly, chlorophyll will remain in its active state for a prolonged period of time, as the electron transport rate is reduced by inhibited off flow of products. Thus the probability for a transfer of electrons from activated chlorophyll to molecular oxygen to form  $O_2^{\cdot-}$  will increase.  $O_2^{\cdot-}$  will rapidly dismutate to yield  $O_2$  and the less reactive ROS  $H_2O_2$ . But in the presence of some cations such as Cu and Fe, highly reactive  $OH^{\cdot-}$  may be formed (Imlay and Linn 1988; Fig. 15.5).

$H_2O_2$  is one of the most important secondary messengers in plant tissues, modulating effects of hormones and involved in developmental control of cells and tissues (van Breusegem and Dat 2006; van Breusegem et al. 2008). It has been shown that mitogen-activated protein kinases (MAP kinases) are involved in transduction of  $H_2O_2$  signaling on cellular level (Pitschke and Hirt 2009). Apparently, MAPK3 and MAPK6 are integrating stress signals that regulate stomatal development (Wang et al. 2008a, b).

Ascorbic acid is a major antioxidant in plants. It detoxifies reactive oxygen species and maintains photosynthetic function (Fig. 15.5). Through its ascorbate recycling function, dehydroascorbate reductase affects the level of foliar reactive oxygen species and photosynthetic activity during leaf development. As a consequence, this enzyme influences the rate of plant growth and leaf aging (Chen and Gallie 2006).

ABA-induced closure of stomata and ABA-mediated inhibition of stomata opening are two ABA effects based on different reaction sequences (Mishra et al. 2006). Nevertheless, both processes are fine-tuned by ROS signaling, and it appears to be clear that ROS signals are transduced via a cascade of MAP kinase reactions (Gudesblat et al. 2007).

## ROS Detoxification

As stated above, ROS production occurs permanently at a certain probability. It is well documented that ROS as well as nitrogen radicals are involved in cell signaling (Wilken and Huchzermeyer 1999) in plant cells like in cells of most other organisms. But as high ROS concentrations are toxic, plants are equipped at varying degrees with several systems to detoxify ROS. In this respect, molecules having antioxidative potential can be discriminated from enzyme-catalyzed reaction sequences.

Alpha-tocopherol is synthesized in the chloroplast (Schultz et al. 1976) and can be found in high concentrations in thylakoid membranes. Alpha-tocopherol disrupts lipid peroxidation cascades, reacts with  $O_2^{\cdot-}$ , and is capable of scavenging hydroxyl-, peroxy-, and alkoxy-radicals (Halliwell 1987). Oxidation of alpha-tocopherol leads to the formation of an alpha-chromoxyl radical, which can be reduced by ascorbic acid.

Ascorbic acid and several redox-active tri-peptides, called glutathiones, can be found in chloroplasts in millimolar concentrations (Halliwell 1982). Several different types are known from plants (Schmidt and Jäger 1992). This finding suggests that they are involved in different pathways controlled by specific enzymes. But there is only little information available to date.

In chloroplasts,  $H_2O_2$  can be detoxified by an ascorbate-specific peroxidase (Chen and Asada 1989) involved in the ascorbate–glutathione cycle (Halliwell and Guteridge 1986) (Fig. 15.5) while in the cytosol  $H_2O_2$  detoxification is catalyzed in a catalase-dependent reaction. Other enzymes involved in detoxification of ROS are superoxide dismutase, which converts  $O_2^{\cdot -}$  to  $H_2O_2$ , and several peroxidases (Chang et al. 1984).

Antioxidants as well as enzymes capable of detoxifying ROS are present in all plants and plant tissues. But their concentrations and catalytic activities, respectively, as well as their patterns differ a lot (Streenivasulu et al. 2000). Therefore plants differ in their capacities to immediately detoxify ROS upon their occurrence and to build up a detoxification potential under stress. If the balance between ROS production and quenching capacity of the respective tissues is upset, oxidative damage will be produced (Dhindsa and Matowe 1981; Wise and Naylor 1987; Spychalla and Desborough 1990). In experimental approaches, it was demonstrated that enzyme activities of antioxidative pathways increase as a salt stress response (Verma and Mishra 2005) and that the maximal level of salt resistance correlates with maximal respective enzyme activities (Kennedy and De Fillippis 1999; Benavides et al. 2000; Lee et al. 2001; Mittova et al. 2002, 2003; Stepien and Klobus 2005; Chang et al. 2012; Mittal et al. 2012).

### Physiological Importance of ROS Production as an Electron Sink

Although  $O_2$  uptake which could not be accounted for Rubisco oxygenation or mitochondrial respiration has been observed in the light (Gerbaud and André 1980; Osmond et al. 1997), evidence of significant extra electron transport was not always found (Genty et al. 1989; Cornic and Briantais 1991). It is thought that  $O_2$  photoreduction increases with increasing reduction of the ferredoxin pool, thus allowing linear electron flow to continue when NADP is scarce. It has been suggested that  $O_2$  photoreduction can assist in maintaining a high trans-thylakoid pH gradient, which in turn enhances nonradiative dissipation of light energy and protects light reactions from photodamage (Lu et al. 2003).

In this context it has to be considered that a slight alkalization of the chloroplast stroma in the light is a prerequisite for the functioning of most plastidic metabolic pathways. The phosphorylation potential of ATP, like the group transfer potential of many other metabolites, depends on pH (and cation concentrations, especially Mg), and reaction equilibria within pathways would not allow metabolism to take place in all other cell compartments because of their lower pH values compared to illuminated chloroplasts. Additionally, the stoichiometries of NADPH and ATP

production and consumption in chloroplasts are different and the alternative sinks are considered necessary to enable ATP production without NADP reduction (Noctor and Foyer 1998).

## ***Dark Reaction of Photosynthesis***

### **Rubisco Limitation**

Rubisco is the most abundant protein in the leaves, contributing to up to 50% of the total protein. It has been discussed that a part of the total Rubisco protein is not catalytically active, but functions as a CO<sub>2</sub> buffer. Different functions of Rubisco, CO<sub>2</sub> binding, and CO<sub>2</sub> turnover, are in accordance with the observation that for full catalytic activity in the light, Rubisco needs to be activated, which in turn requires Rubisco activase and ATP (Robinson and Portis 1988; Salvucci and Ogren 1996). Inhibitors are generally analogues of the enzyme's substrate ribulose-1,5-bis phosphate (RuBP) (Edmonson et al. 1990) and bind to the enzyme in the absence of RuBP. Activase releases tight-binding inhibitors from the Rubisco active sites, thus increasing specific activity. This reaction requires ATP (Robinson and Portis 1988), so a decreased activity and activation state of Rubisco at low relative water content may be related to inadequate ATP supply to the protein complex (Tezara et al. 1999; Parry et al. 2002).

The amount of Rubisco protein is generally little affected by moderate or severe salt stress (Flexas et al. 2002), even if experienced over a period of many days (Tezara et al. 1999; Gunasekera and Berkowitz 1993; Medrano et al. 1997). This means that specific Rubisco activity rather than protein concentration is decreased under saline conditions. Restoration of the assimilation potential by rehydration also suggests that Rubisco is not irreversibly impaired (Medrano et al. 1997; Parry et al. 2002).

### **Calvin Cycle Enzymes**

The reduction of RuBP content at low relative water content could result from a limitation in one or more enzymes of the Calvin cycle. There is little direct evidence regarding the response of the individual enzymes of the regenerative part of the Calvin cycle to increasing cytosolic mineral content subsequent to increased salinity. There do not seem to be significant differences between enzymes from glyco-phytes and halophytes, respectively (Huchzermeyer and Heins 2000; Huchzermeyer 2000). Furthermore, in most cases isolated enzymes did not show a degree of salt inhibition sufficient to explain the extent of inhibition of the respective metabolic reaction in *in vivo* experiments. Therefore it is doubtful that the high salt resistance of halophytes is due to changed properties of their Calvin cycle enzymes. Thus, there is a strong evidence that the Calvin cycle per se is not the cause of the decreased assimilation rate under osmotic stress.

## RuBP Concentration and Rubisco Activity

The net assimilation rate depends on the synthesis of RuBP and Rubisco activity. Therefore, the decrease in leaf Rubisco content at low relative water content is significant (Giménez et al. 1992; Gunasekera and Berkowitz 1993; Tezara et al. 1999). In stressed sunflower plants, the photosynthetic rate correlated with RuBP concentration (Giménez et al. 1992), which suggests that the assimilation potential in these experiments was determined by RuBP content, but not by CO<sub>2</sub>.

Under salt stress, there is a general decline in Calvin cycle intermediates during the phase of stress adaptation. Decreased RuBP concentration might be caused by the general rundown of the Calvin cycle and decrease in assimilation rate. The high ratio of 3PGA/RuBP suggests a limitation in the RuBP regeneration part of the Calvin cycle, either caused by enzyme limitations or inadequate ATP, although it was interpreted by Tezara et al. (1999) as evidence of 3PGA production by mitochondria. In laboratory experiments with halophytes such as *Aster tripolium* however, the metabolite pattern recovered within less than two or three days under constant stress conditions.

Interaction between Rubisco activity and RuBP supply is well illustrated by studies on unstressed tobacco leaves with normal amounts of Rubisco (Laisk and Oja 1974; von Caemmerer 2000). The RuBP pool increased in leaves in bright light when the CO<sub>2</sub> concentration was transiently decreased, so that when returning to normal CO<sub>2</sub> concentrations, assimilation rate was much higher than the steady-state concentrations for a short time until the RuBP was consumed. Thus, under steady-state conditions, RuBP supply limited the CO<sub>2</sub> assimilation rate.

## Limitation of RuBP Regeneration by NADPH and ATP Availability

Synthesis of RuBP depends on ATP and NADPH concentration and on the Calvin cycle activity, or more specifically on PRK activity, and concentration of the substrates ATP and ribose 5-phosphate (Lawlor 2001). In the Calvin cycle, NADPH serves as substrate of the glyceraldehyde 3-phosphate dehydrogenase. If NADPH were limiting at low relative water content, it would decrease glyceraldehyde 3-phosphate and thus ribose 5-phosphate, the same effect as decreasing the activity of an enzyme in that part of the cycle. In salt stressed leaf mesophyll cells NADPH content remained constant (Lawlor and Khanna-Chopra 1984; Tezara et al. 1999), while NADH increased (Lawlor and Khanna-Chopra 1984), indicating that the electron transport capacity is sufficient to maintain and increase the reduction state of these pyridine nucleotides. Thus, with respect to high salinity resistance of the photosynthetic electron transport system in both glycophytes and halophytes, the availability of NADPH to the Calvin cycle is unlikely to limit its capacity to form RuBP (Koyro and Huchzermeyer 1999; Huchzermeyer and Heins 2000).

The rate of ATP synthesis depends on the light reactions, generation of the trans-thylakoid pH gradient ( $\Delta\text{pH}$ ), availability of ADP and P<sub>i</sub>, and activity of the chloroplast coupling factor (CF<sub>1</sub>) (Löhr and Huchzermeyer 1985; Löhr et al. 1985; Huchzermeyer 1988a, b; Richter et al. 2000). Inadequate ATP concentration would

decrease the ability of the Calvin cycle to regenerate RuBP by PRK, so glyceraldehyde-3-phosphate would increase and RuBP decrease. This was the effect of reduced PRK activity (Paul et al. 1995), but ATP increased in the transgenics, in contrast to the case of water stress in some studies (Lawlor 2002a, b; Lawlor and Cornic 2002).

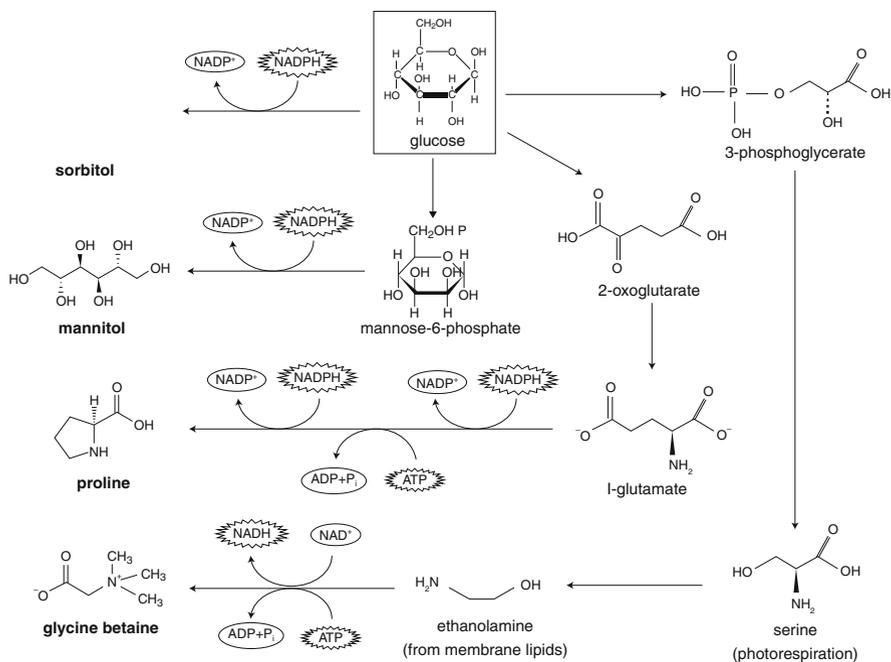
### Triose Export

In the presence of light, sugars are exported as triose phosphates (DHAP and GAP) in exchange of free phosphate via the phosphate translocator (Riesmeier et al. 1993).  $V_{\max}$  of sugar phosphate export is too low to keep up with sugar phosphate synthesis under average light conditions. This would cause shortage of phosphate inside the chloroplast stroma, thus inhibiting the primary reactions of photosynthesis and increasing the risk of ROS production. Starch production inside the chloroplast is a relief, because phosphate will be released and becomes available for the chloroplast coupling factor to produce ATP.  $V_{\max}$  of starch production has not been measured yet as physiological sugar concentrations are too low. The capacity to produce starch is obviously high enough to turn over any sugar concentration that might become available under physiological conditions. There is an equilibrium between starch synthesis and hydrolysis, respectively. Therefore, starch will be degraded at night and allow a permanent supplementation of the cytosol of the host cell. This system allows plant cells to grow permanently on a continuous flow of incoming sugar phosphates.

In the cytosol of green leaf parenchyma cells, sugar phosphates are used to fuel cell metabolism. In a competing metabolic pathway sucrose is formed from glucose and fructose and is exported into the phloem to feed sink organs.

The description above shows that there are several alternatives how incoming salt may affect the functioning of cellular sugar metabolism. In principle, the situations in source and sink tissues are comparable. But under moderate salt stress, salt concentrations have been found to differ a lot between root, leaf, and fruit tissues. Therefore, the observed effect of salt stress also differs among tissues of plant organs.

Salt effects on sugar metabolism and sugar export from the chloroplasts into the cytosol can be explained on the basis of a competition between  $\text{Cl}^-$  and  $\text{H}_2\text{PO}_4^-$  for binding sites on enzymes and receptors, respectively. This competition is due to similar diameters of these two ions when hydrated. This would mean that under salt stress the chloroplast stroma is at risk to run out of free phosphate. As mentioned above, this would inhibit ATP synthesis at the chloroplast coupling factor (Groth et al. 2000). The observed effect would be quite similar to the one observed after energy transfer inhibitors like nitrofen (Huchzermeyer and Löhner 1990). Nitrofen is a herbicide used in rice cultures and it is known to stimulate light-dependent ROS production in plants. By the way, this observation again is a proof for the tight coupling between primary reactions of photosynthesis.



**Fig. 15.6** Overview on metabolism of compatible solutes. Glucose can function as a substrate for synthesis of compatible solutes. Cytosolic synthesis often occurs at the expense of NAD(P)H. Thus, synthesis of compatible solutes is a relief under conditions, when production of redox power and glucose are exceeding consumption of electrons and export of sugar. Cytosolic pattern of compatible solutes varies among plant species depending on respective enzyme activities. In addition to regulation of respective gene expression, synthesis rates are controlled by availability of precursors

## *Protection of Proteins and Biomembranes from Salt Effects*

### **Control of Mechanisms Improving Stress Resistance: Compatible Solutes**

Sequestration of ions to the vacuole is a strategy leading to enhanced salt stress resistance. Some low molecular weight nontoxic compounds have been identified to significantly contribute to ionic and osmotic balance inside cells. They are called “compatible solutes” (Ford 1984; Ashihara et al. 1997; Hasegawa et al. 2000; Zhifang and Loescher 2003; Lee et al. 2008). Chemically they can be described to be poly-amines or poly-hydroxyls. In addition to their function in ionic and osmotic homeostasis, their important function is that they can replace water in its function to stabilize aggregates of soluble proteins and membrane–protein interactions, respectively (Yancey et al. 1982; Crowe et al. 1992; Ashraf and Foolad 2007).

Depending on enzyme patterns and metabolic pathways preferred in individual plants, different compatible solutes have been found in plants (Fig. 15.6).

Among the compounds described in the literature are proline (Singh et al. 2000; Kavi Kishore et al. 2005; Koca et al. 2007; Alla et al. 2012), glycine betaine (Khan et al. 1998; Wang and Nil 2000; Türkan and Demiral 2009; Zhu et al. 2011), sugars (Bohnert and Jensen 1996; Kerpesi and Galiba 2000; Gil et al. 2011), di-, tri-saccharides and other sugar derived compounds (Hagemann and Murata 2003), and polyols (Popp et al. 1985; Orthen et al. 1994; Bohnert et al. 1995; Gil et al. 2011). Overexpression of genes of metabolic pathways leading to production of compatible solutes has been shown to improve salt resistance. (Parida et al. 2002; Parida and Das 2005). Apparently, there are three ways to explain the mechanism how compatible solutes become overproduced under stress. Under optimal growth conditions, these compounds are produced at low rates, so that their concentration is found to be low. Therefore, the question is whether overproduction under stress is due to activation of already existing enzymes or de novo synthesis of such enzymes.

It was observed that many stress conditions like drought and salt stress, for instance, initially inhibit export of products of photosynthesis from their source tissues. This will result in enhanced concentrations of primary products of the Calvin cycle in leaf cells. As shown by Koyro and Huchzermeyer (2005) the most compatible solutes derive from primary products of sugar metabolism. Therefore, increased concentrations of metabolites of photosynthesis will stimulate synthesis rate of compatible solutes. Such an increase will become significant if metabolite concentrations under optimal growth conditions are below the  $K_m$  values of enzymes catalyzing the initial reactions of the pathways leading to compatible solutes. This interpretation agrees with the findings of Soussi et al. (1998), who attributed enhanced concentrations of proline and carbohydrates under salt stress in chick pea to damage of metabolic pathways rather than to protective mechanisms.

A second working hypothesis is based on findings of Süß et al. (1993). They found that enzymes of individual pathways in chloroplasts tend to form clusters. Such conditions would contradict free mobility of intermediates. Thus, substrate concentrations localized to catalytic centers of enzymes may significantly differ from respective bulk phase concentrations. Calvin cycle enzymes, for instance, tend to bind to one another and thus form aggregates of proteins. This allows substrates as well as products formed to be shuttled from one enzyme to the next one of the respective pathway. Süß postulated that modulation of pathways in this model is brought about by modulation of enzyme neighbourhood. From analysis of internal signalling within cells, it is well documented that such modulations can be brought about by protein phosphorylation and de-phosphorylation, for instance. Indeed, such phosphorylation-dependent variations in protein-protein interactions have been observed by Allen and Horton, when analyzing protein localization in thylakoid membranes (Horton and Foyer 1983; Pursiheimo et al. 2001; Allen 2002).

A third model refers to the observation that salt stress response of plants has been found to be under the control of hormones (ABA, for instance) and second messengers (sugar signalling, for instance). This implies that modification of enzyme patterns will result in modified metabolic activity of cells and tissues under stress. Such argumentation would explain the above observations on the level of gene activities.

These latter arguments may suggest not to discuss three different models of regulation of metabolic pathways leading to the synthesis of compatible solutes. It rather appears to us that hormone- and secondary messenger actions explain how formation of protein aggregates may be controlled in plants.

Biochemical reaction sequences leading to improved salt resistance are likely to act synergistically (Iyengar and Reddy 1996). The reactions include the synthesis of compatible solutes, the stimulation of antioxidative enzyme activities, and modifications of the photosynthetic pathway.

### Energy Consumption by N- and S-Pathways

As a rule, in non-woody plants nitrate is reduced in chloroplasts, while it is reduced in root cell plastids of trees. Nitrate competes with chloride (due to similar radii of the hydrated ions) for uptake via specific translocators. These translocators are regulated via the plant's sugar pool and their activity thus matches the photosynthetic activity of source leaves. In this chapter, we focus on coupling of nitrate reduction to photosynthesis in chloroplasts of weeds.

About one-third of the electrons released by PSII finally will be used for nitrate reduction. Therefore, nitrate reduction is a major sink for electrons and sufficient N fertilization contributes to prevention of primary reactions of photosynthesis from over-reduction in high light.

Nitrate reduction and incorporation of reduced nitrogen into glutamate is shown in Fig. 15.3. It is obvious that the first intermediates of nitrate reduction are toxic. Like in primary reactions of photosynthesis, building up of concentrations and occurrence of side reactions that eventually might be toxic for the cell are prohibited by fast turnover of products.

It has been shown in the literature that the affinity for nitrate uptake from the soil can be improved by means of a molecular biological approach (Matt et al. 2001; Wang et al. 2003; Lillo 2004). Furthermore, nitrate fertilization helps to reduce salt stress effects (Syvertsen et al. 1989; Murillo-Amador et al. 2006).

It is obvious that salt stress can have direct as well as indirect effects on nitrate reduction and amino acid biosynthesis. Direct effects mostly are due to competition among nitrate and chloride ions. Indirect effects are based on the dependence of nitrate reduction on electrons released by photosynthetic electron transport, an intact structure of enzymes and enzyme aggregates, and the dependence of amino acid biosynthesis on substrates that derive from sugar metabolism (i.e. sugar supply from photosynthesis). We have to keep in mind that sugars, sugar phosphates, and ROS are functioning as secondary messengers regulating the expression of enzymes. Therefore, there will be another level of regulatory effects. Though a lot information on sugar signaling has been analyzed and a list of participants has been outlined in pretty much detail, the complex pathway structure of second messenger signaling has not been well characterized (Gibson 2000; Sheen 2002; Baena-González and Sheen 2008).

As nitrate reduction and subsequent amino acid biosynthesis depend on photosynthetic activity, there has to be some buffer capacity for intermediates to allow

continuous metabolic activity of plants. In principle, there are two storage compartments in leaf cells: plastids and the vacuole. In storage organs, plastids can differentiate to protein storage compartments. In chloroplasts, only low amounts of storage proteins typical for storage plastids are found. But in some papers it is discussed whether the huge amounts of Rubisco accumulating in chloroplasts may function as storage proteins as well. In vacuoles, free amino acids and other low molecular weight molecules are stored rather than macro molecules like proteins. It has been shown that vacuoles from tissues active in photosynthesis can actively import amino acids at final concentrations several fold exceeding the ones in the cytosol (Homeyer et al. 1989). Amino acid import occurs at the expense of a transmembrane pH gradient. Both v-Type ATPase and PPase of the tonoplast build this gradient at the expense of ATP and pyrophosphate, respectively (Homeyer et al. 1989). It is well known that the activity of these two enzymes is regulated (Taiz 1992; Davies 1997; Han et al. 2005; Park et al. 2005), and that amino acid content is affected by salt stress (Pahlich et al. 1983).

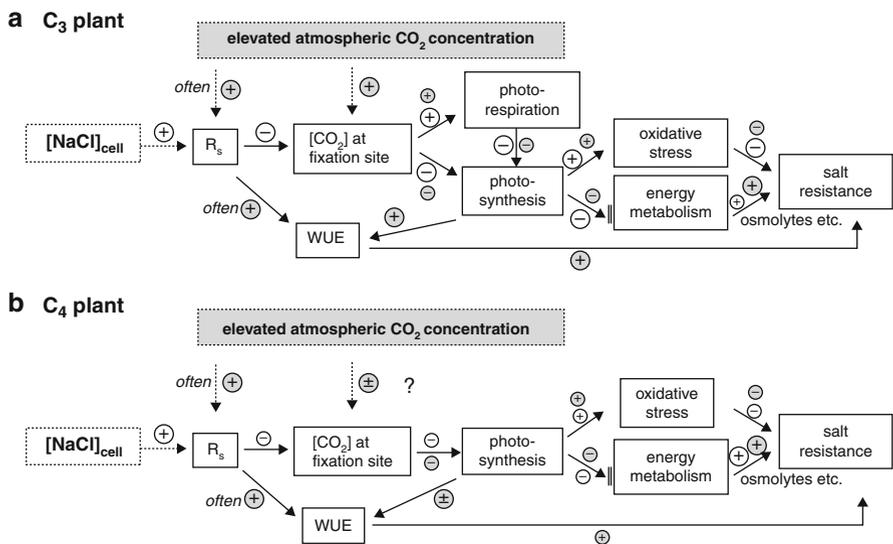
In contrast to nitrogen that is found exclusively in the reduced form in organic molecules, sulfur occurs in metabolites of biological relevance in several redox states. Only a minor portion of the photosynthetic electrons are used for the sulfate reduction pathway. As outlined by Schmidt and Jäger (1992, and citations therein), sulfate reduction pathway in chloroplasts differs from the one in bacteria that had been identified earlier.

Inhibition of sulfur metabolism has major secondary effects, because SH bounds are essential for functioning of catalytic centers of many enzymes and reduced sulfur is found in coenzymes like CoA and liponic acid, for instance. Moreover, a pool of various glutathiones forms the dominant redox buffer of plant cells, and glutathiones are involved in detoxification of heavy metals and ROS. Therefore, enhanced salt stress resistance of sulfur metabolism always goes along with improved metabolic activity and apparent resistance of other essential functions as well.

## **Photosynthesis of Salt Stressed Plants Under Future Elevated Atmospheric CO<sub>2</sub> Concentration**

### ***C<sub>3</sub> Plants***

Elevated atmospheric CO<sub>2</sub> concentration leads to a higher CO<sub>2</sub> concentration gradient between the outside air and the intercellular spaces of the leaves, so that the diffusion of CO<sub>2</sub> into the leaves and the pCO<sub>2</sub>/pO<sub>2</sub> ratio at the sites of photoreduction is increased (Robredo et al. 2007). Therefore, photorespiration and the rates of oxygen activation and ROS formation are usually reduced in C<sub>3</sub> plants due to an increased NADPH utilisation, whereas the net photosynthetic rate and thus the carbon supply is enhanced (Polle 1996; Sgherri et al. 2000; Urban 2003; Kirschbaum 2004; Long et al. 2004; Hikosaka et al. 2005; Ignatova et al. 2005; Fig. 15.7).



**Fig. 15.7** Generalized and simplified model of the influence of NaCl salinity and elevated CO<sub>2</sub> concentration on the photosynthesis of (a) C<sub>3</sub> and (b) C<sub>4</sub> plants. ○= influence of NaCl; ⊖= influence of elevated CO<sub>2</sub> and salinity; ⊕= positive influence; ⊖= negative influence; ⊙= major influence; ⊙= minor influence. The changing quantity of a parameter is taken into account when assessing its influence on the next one (Except for energy metabolism, following ||). R<sub>s</sub> stomatal resistance, WUE water use efficiency of photosynthesis

Furthermore, we often find a higher stomatal resistance (Hsiao and Jackson 1999; Sgherri et al. 2000; Li et al. 2003; Marchi et al. 2004; Rogers et al. 2004), which – together with the higher net assimilation – also leads to a better water use efficiency of photosynthesis (Amthor 1999; Morgan et al. 2001; Urban 2003). Results from free-air CO<sub>2</sub> enrichment (FACE) experiments showed that elevated CO<sub>2</sub> stimulated the light-saturated photosynthesis of C<sub>3</sub> plants by an average of 31% and reduced stomatal conductance by an average of 22% (Ainsworth and Rogers 2007).

However, responses of gas exchange to elevated CO<sub>2</sub> concentration are variable and depend on plant species and abiotic factors such as salinity and humidity (Drake et al. 1997; Ball and Munns 1992; Arp et al. 1993). So stomatal resistance of salt stressed C<sub>3</sub> plants may even decrease under elevated CO<sub>2</sub> if the maximization of photosynthesis and energy gain is given priority over a reduction of water loss in order to reduce oxidative stress. This is the case e.g. in *Aster tripolium*, which increases its WUE only by an improved photosynthesis, but not by a reduced transpiration. This is a reasonable strategy because the salt resistance of this species is mainly limited by an impaired net assimilation rate and accompanying ROS formation (Geissler et al. 2009a, b). Elevated atmospheric CO<sub>2</sub> concentration therefore ameliorates metabolic processes in *A. tripolium* which are typically associated with C<sub>3</sub> metabolism and are disadvantageous on saline habitats compared to C<sub>4</sub> plants (see Sect. C<sub>4</sub> Plants).

As a consequence of the enhanced plant water relations and/or net photosynthetic rate, there might be less need for antioxidants as elevated  $\text{CO}_2$  ameliorates oxidative stress (Schwanz et al. 1996), such as in barley (Pérez-López et al. 2009) and *Solanum lycopersicum* (Takagi et al. 2009). On the other hand elevated  $\text{CO}_2$  can induce carbon partitioning to various pathways associated with stress defence/response (Bokhari et al. 2007; Cseke et al. 2009; Jin et al. 2009), i.e. more energy can be provided for stress resistance mechanisms: The amount and/or activities of antioxidants can be increased, e.g. the activities of catalase (CAT), ascorbate peroxidase (APX) and superoxide dismutase (SOD) in *Quercus* (Marabottini et al. 2001; Schwanz and Polle 2001) or the expression and activities of APX, SOD and glutathione-S-transferase (GST) in *Aster tripolium* (Geissler et al. 2009b, 2010). Some species show also a higher amount of compatible solutes such as soluble carbohydrates (*Carthamus mareoticus*: Abdel-Nasser and Abdel-Aal 2002; *Aster tripolium*: Geissler et al. 2009a) or proline (*Aster tripolium*: Geissler et al. 2009a). Additionally, the larger number of carbon skeletons available due to an improved photosynthesis can increase the thickness of the outer epidermal cell walls and the cuticle (Tipping and Murray 1999; Tingey et al. 2003; Oksanen et al. 2005), which decreases transpiration on the one hand and light reflection and thus ROS production on the other hand (Thomas 2005; Geissler et al. 2009b). Reduced oxidative stress under elevated  $\text{CO}_2$  concentration in turn leads to less structural damage to the thylakoid membranes of the chloroplasts (see Sect. Salt Effects on Chloroplast Structure and Metabolite Transfer) with only minor dilations of the thylakoid membranes (Oksanen et al. 2001; Geissler et al. 2009b; Fig. 15.4). This fact is of major importance because the chloroplasts – as the site of photosynthesis – fulfil a key function in adaptation to salinity, and chloroplast integrity correlates with a higher assimilation rate.

All the factors discussed above can contribute to an increased survival of salt stressed plant under elevated  $\text{CO}_2$ , which has been reported by various authors (Ball and Munns 1992; Rozema 1993; Drake et al. 1997; Fangmeier and Jäger 2001; Wullschleger et al. 2002; Urban 2003; Geissler et al. 2009a; Fig. 15.7).

## *C*<sub>4</sub> Plants

While elevated  $\text{CO}_2$  concentration generally has a positive effect on the water relations as well as on the net photosynthesis of salt stressed *C*<sub>3</sub> plants, the situation in *C*<sub>4</sub> plants is more ambiguous.

In *C*<sub>4</sub> plants,  $\text{CO}_2$  is bound to PEP and pre-fixed by the formation of a *C*<sub>4</sub> compound (that can be reduced to malate, for instance) in the mesophyll cells which do not contain any Rubisco. The final carbon fixation step takes place in the bundle sheath cells surrounding the mesophyll cells coronally. It releases  $\text{CO}_2$  from the *C*<sub>4</sub> compound, and assimilation will be catalyzed by Rubisco in an environment characterized by low oxygen partial pressure. This highly efficient  $\text{CO}_2$  enrichment mechanism at the carboxylation sites in the bundle sheath cells and thus an equally

efficient  $\text{CO}_2$  fixation enable  $\text{C}_4$  plants to maintain photosynthesis with almost closed stomata, which minimizes transpiration. Therefore they can use water more efficiently, which is of high advantage under saline conditions.

While it was proved in various studies that elevated  $\text{CO}_2$  can ameliorate drought stress in  $\text{C}_4$  plants similarly to  $\text{C}_3$  plants due to its effect on stomatal conductance (Mateos-Naranjo et al. 2010; Leakey 2009; Lopes et al. 2011), it has usually only small or no positive direct effects on the photosynthesis of  $\text{C}_4$  plants (Leakey et al. 2006; Ainsworth and Rogers 2007; Leakey 2009; Wang et al. 2012; Fig. 15.7). The latter are adapted to low atmospheric  $\text{CO}_2$  concentrations due to their efficient  $\text{CO}_2$  enrichment mechanism (see above), so their photosynthesis is saturated at the current atmospheric  $\text{CO}_2$  concentration (Fangmeier and Jäger 2001). In consistence with these facts, Polley et al. (1996) predicted that the productivity of various  $\text{C}_4$  species will probably show only a small increase under elevated  $\text{CO}_2$  because the intrinsic water use efficiency was stimulated proportionally more by a given relative increase in  $\text{CO}_2$  over sub-ambient than by elevated concentrations. So we expect the salt resistance of  $\text{C}_4$  species to be less enhanced by elevated  $\text{CO}_2$  concentration than the one of  $\text{C}_3$  species (Fig. 15.7). This could already be confirmed by several studies which have shown that  $\text{C}_4$  plants show less growth stimulation than  $\text{C}_3$  species on saline soils under elevated  $\text{CO}_2$  and/or are likely to be disadvantaged in competition (Schwarz and Gale 1984; Curtis et al. 1989; Rozema et al. 1991; Arp et al. 1993; Lenssen et al. 1993; Erickson et al. 2007; Jaggard et al. 2010).

However, this theory is not supported by Wand et al. (1999) and Ward et al. (1999), so the situation is a bit ambiguous. This is also true for the issue of photorespiration which leads to  $\text{CO}_2$  loss and ROS production in  $\text{C}_3$  plants (Ainsworth and Rogers 2007; Lüttge et al. 2010; Sect. ROS Production), thus limits their salt resistance and can be ameliorated by elevated  $\text{CO}_2$ . In most textbooks, it is stated that  $\text{C}_4$  plants can overcome this problem on the expense of extra energy consumption and that the formation of significant concentrations of glycolate is not a problem of these plants. If salt stress would interfere with aggregate formation of chloroplasts, mitochondria, and peroxisomes, thus inhibiting by glycolate toxicity proper performance of  $\text{C}_3$  plants; such effects should not be observed in  $\text{C}_4$  species. Nevertheless, salt stress might interfere with intermediate transfer among cells and cell compartments in  $\text{C}_4$  plants. This would inhibit photosynthetic activity but would not result in “classical” glycolate toxicity.

But again the situation is not as simple as suggested in textbooks. Several studies have pointed out that a low rate of photorespiration takes place in  $\text{C}_4$  plants as well. In maize leaves grown at normal  $\text{CO}_2$  concentration, photorespiration may reach 5% of the rate found in tobacco grown under identical conditions (Zelitch 1973). Though such a rate may appear to be low, glycolate oxidase activity obviously is essential for  $\text{C}_4$  plants. Maize plants showing less than 10% of glycolate oxidase activity of the wild type could grow at high  $\text{CO}_2$  concentrations, they became necrotic in normal air, and died within 2 weeks (Zelitch et al. 2009). When considering these facts, the salt resistance of  $\text{C}_4$  plants may indeed be increased by elevated  $\text{CO}_2$  concentration.

## Conclusions and Future Perspectives

Abiotic stresses such as soil salinity are responsible for a decrease in yield especially in arid and semiarid regions. It is estimated that 45% of the world's agricultural land experience drought and 19.5% of the irrigated land are affected by salinization. This problem will be further catalyzed by global climate change. Salt stress directly affects the primary and secondary reactions of photosynthesis in numerous ways and consequently has also far reaching consequences on ROS production, on sugar, lipid, and amino acid metabolism, trans membrane and inter tissues transport of metabolites and on water relations and gas exchange.

Elevated atmospheric CO<sub>2</sub> concentration can improve photosynthesis and water relations and reduce oxidative stress in C<sub>3</sub> plants, so that it can enhance their salt resistance and thus their suitability as crops in a future world of climate change. In contrast, the effect of CO<sub>2</sub> on the salt resistance of C<sub>4</sub> plants is less beneficial.

The major challenge to modern plant scientists is to develop stress resistant and high-yield plants. Without doubt halophytic model plants will have to be developed as they are invaluable in precisely assessing the role which e.g. the complex photosynthetic reactions, ROS, antioxidants, or osmolytes play in the functional network which controls stress resistance. They can be of major importance not only for the breeding of halophytes as alternative crop plants, but also for increasing the salt resistance of conventional crops via genetic engineering. Halophytes as naturally salt resistant plants constitute especially valuable genetic models for understanding resistance mechanisms (Flowers and Colmer 2008). They are much more suited than classical glycophytic model plants such as *Arabidopsis thaliana* because genes which are expressed in halophytes but not in glycophytes indicate specific traits which enable survival under saline conditions.

In the face of climate change, it would be very desirable to develop model plants also regarding the interaction of salt stress with other abiotic factors associated with global change, such as elevated atmospheric CO<sub>2</sub> concentration. In this regard, C<sub>3</sub> and C<sub>4</sub> plants should be compared as they respond differently to elevated CO<sub>2</sub>. Suited species for such a comparison would be *Chenopodium quinoa* (C<sub>3</sub>) and *Atriplex spec.* (C<sub>4</sub>) because they are related (both belong to the *Chenopodiaceae*), and show some similarities regarding their salt resistance mechanisms.

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