# Oxyfunctionalization of Linear Alkanes with a Biosynthetic, Self-Sufficient, Selective, and Soluble Hydroxylase

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### Abstract

Selective oxyfunctionalization of C–H bonds carried by hydrophobic molecules is a challenging reaction. Numerous chemical and biological catalysts have been developed during the past decades to perform this difficult task. Chemical methods generally involve toxic or dangerous reagents and are not very selective. Thus, they are far from ideal and not fully compatible with the principles of green chemistry. On the road to biocatalytic and selective oxyfunctionalization of C–H bonds, we have produced and fully characterized a biosynthetic self-sufficient biocatalyst for alkane activation. To do so, we have fused a cytochrome P450 from *Alcanivorax borkumensis* SK2 with a  $Fe_2S_2$ /FMN reductase from *Rhodococcus* sp. NCIMB 9784. With this original enzyme, thereafter referred to as "A13-Red," the highly selective and efficient hydroxylation of alkanes from *n*-hexane to *n*-decane on the terminal methylic position was enabled under mild conditions. This enzyme was the first member of a novel family of biosynthetic omega hydroxylases. Detailed protocols for A13-Red construction, purification, and optimized reaction conditions for the in vitro hydroxylation of its preferred substrate (*n*-octane) are described in detail into this chapter.

Keywords: Biosynthetic self-sufficient hydroxylase, C-H bond activation, Cytochrome P450, Green chemistry, Linear alkanes

### 1 Introduction

Nature provides a large variety of hydrocarbons [1], including paraffins (alkanes, iso-alkanes, and alkenes), cycloparaffins (cycloalkanes, naphthenes), and aromatics, as a result of the metabolic activities of animals, plants, and microorganisms and by the physicochemical action of decay of organic matter. They are most evident in the form of crude oil and gas deposits. There is a body of opinion that petroleum (crude oil) deposits are too valuable to expend as fuel and should be conserved for the production of chemicals alone. Considering their numerous advantages (i.e., versatility, regio-, chemo-, enantioselectivity, environmental-friendly processes, etc.) [2], the role of microorganisms in a scenario of optimal use of such valuable raw materials would be central.

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Indeed, a very large number of microorganisms have the ability to convert hydrocarbons from crude oil into biomass and useful intermediates or products such as polyhydroxyalkanoates, wax esters, or triacylglycerols [3, 4]. These microbes are the foundations of hydrocarbon biotechnology. Understanding the underlying biological mechanisms of these conversions is of fundamental and practical interest. Several groups of researchers have worked to elucidate these biochemical mechanisms for several decades [5, 6], but current economic uncertainties limit the amount of research activity and the implementation of emerging technologies.

One of the most widespread catabolic pathways in hydrocarbon metabolism among microorganisms is initiated by the primary enzymatic attack of the hydrocarbon molecule. This is generally an oxyfunctionalization yielding an alcohol. For alkanes in particular, Nature has developed a great diversity of enzymatic hydroxylation systems [2]. Most of them share common features: they use dioxygen from air as the oxidizing agent, they use reduced nicotinamide dinucleotides (NAD(P)H) as electron suppliers, they are metalloenzymes based on copper and/or iron, and they operate under mild conditions [2, 6].

Among all these enzymatic systems involved into alkane hydroxylation, cytochromes P450 (P450) from the CYP153 family attracted our attention because they are monomeric soluble enzymes that hydroxylate alkanes [7–11]. Despite their attractiveness, every CYP153 enzyme needs at least one supplementary protein (also called a redox partner) to shuttle electrons from NAD(P)H to the active center of the P450 responsible for the oxidation reaction [12–17]. The requirement for these redox partners could be seen as a disadvantage, but in fact, it has been shown through different examples that P450s can be artificially fused with either their natural partner or with a foreign partner [18–20]. The resulting biosynthetic proteins are usually enzymatically active and self-sufficient. Using this strategy, we have fused a cytochrome P450 from A. borkumensis SK2 (CYP153A13a) with a Fe<sub>2</sub>S<sub>2</sub>/FMN reductase from *Rhodococcus* sp. NCIMB 9784 (RhFred) [21, 22]. The resulting polypeptide (thereafter A13-Red) was reliably expressed in Escherichia coli using high-cell-density cultivation (up to 100 g of dry-cell weight per liter of culture medium) [23]. In this chapter, we detail the construction of the gene encoding for A13-Red, our two-step purification strategy to yield A13-Red approximately 90% homogeneous and its use for the regioselective hydroxylation of *n*-octane into optimized conditions.

2 Materials					
2.1 Construction of pA13-Red	1. Authentic s 11573) (wy	ample of <i>Alcanivorax borkumensis</i> SK2 strain (DSM vw.dsmz.de)			
	2. Medium 809, composition for 1 L: NaCl 23.00 g, MgSO <sub>4</sub> × 7 H <sub>2</sub> O 5.80 g, MgCl <sub>2</sub> × 2 H <sub>2</sub> O 6.16 g, CaCl <sub>2</sub> × 2 H <sub>2</sub> O 1.47 g, Na <sub>2</sub> HPO <sub>4</sub> × 7 H <sub>2</sub> O 0.89 g, NaNO <sub>3</sub> 5.00 g, FeSO <sub>4</sub> × 7 H <sub>2</sub> O 0.03 g, sodium pyruvate 10.00 g, pH 7.0–7.5				
	3. Qiagen Gentra PureGene Yeast/Bacteria kit (www.qiagen. com)				
	4. pET28b(+) plasmid, Novagen (www.merckmillipore.com) previously digested by <i>EcoRI</i> and <i>NotI</i> enzymes				
	5. pET28b(+)-pikC-RhFred plasmid [19] (generous gift from Dr D. Sherman, University of Michigan, USA)				
	6. KOD Hot Start Master Mix, Novagen				
	7. Qiagen PC	R Purification kit			
	CYP153A13a, primers Eco887-fw and Eco887-rv for removal of the <i>EcoRI</i> restriction site at position 887 of CYP153A13a, and primers RhFred-fw and RhFred-rv for amplification of RhFred ( <i>see</i> Table 1 for primer sequences)				
	<ul> <li>9. Restriction enzymes: DpnI, NdeI, EcoRI, and NotI, New England Biolabs (www.neb.com)</li> <li>10. T4 DNA ligase, New England Biolabs</li> </ul>				
	12. Kanamycin sigmaaldric	sulfate antibiotic, Sigma-Aldrich (www. h.com)			
	Table 1 Sequences of the RhFred and for t CYP153A13a	e primers used for the amplification of CYP153A13a and he removal of <i>EcoRI</i> restriction site at position 887 in			
	Primer	Sequence			
	153A13a-fw	GTAGCTCATATGTCAACGAGTTCAGT			
	153A13a-rv	TACAGCGAATTCTTTTTTAGCCGTCAA			
	Eco887-fw	ATCAATCGTCCTTTGGAGTTCATTGGTAATCT			
	Eco887-rv	GAGATTACCAATGAACTCCAAAGGACGATTG			

RhFred-fwACGGCTAAAAAAGAATTCGTGCTGCACCGGRhFred-rvTAATGCGGCCGCTCAGAGTCGCA

2.2 Purification of A13-Red on Nickel-NTA Column

- 13. Growth media: Luria Bertani (LB) and Terrific Broth (TB), Sigma-Aldrich
- 14. Qiagen Plasmid Purification kit

1. E. coli strain BL21	Star <sup>™</sup> (DE3)/pA13-Red (fresh or frozen
wet-cell pellet)	

- 2. Lysis buffer: Tris-HCl 25 mM, NaCl 500 mM, glycerol 10% (v/v), triton X-100 1% (v/v), lysozyme 0.25 mg/mL, PMSF 1 mM, DNAse I 5 mg/mL, pH 7.5
- 3. Binding buffer: NaCl 800 mM, Tris-HCl 50 mM, pH 7.5
- 4. Washing buffer: NaCl 800 mM, Tris-HCl 50 mM, glycine 250 mM, pH 7.5
- 5. Elution buffer: NaCl 800 mM, Tris-HCl 50 mM, L-histidine 80 mM, pH 7.5
- 6. HisPrep FF 16/10 prepacked column (GE Healthcare)
- 7. Amicon Centrifugal Ultrafiltration Units (50-kDa cutoff)
- 8. Conservation buffer: Tris-HCl 25 mM, NaCl 500 mM, glycerol 20%, pH 7.5
- 2.3 Desalting of A13-1. Desalting buffer: Tris-HCl 25 mM, NaCl 25 mM, glycerol 20%, pH 7.5 **Red Fractions After** Ni-NTA Column
  - 2. HiTrap desalting 5-mL prepacked columns (GE Healthcare)
  - 1. Buffer A: Tris-HCl 25 mM, NaCl 80-mM glycerol 20%, pH 7.5
  - 2. Buffer B: Tris-HCl 25 mM, NaCl 200 mM, glycerol 20%, pH 7.5
  - 3. Q Sepharose XL prepacked column (1 mL) (GE Healthcare)

2.5 Enzymatic Hydroxylation of n-Octane

2.4 Purification

of A13-Red on Q

Sepharose Column

- 1. Wheaton 5-mL V-Vials with a magnetic stirrer
- 2. Reaction buffer: potassium phosphate buffer (100 mM, pH 7.4), glycerol 10%
- 3. *n*-Octane (Sigma-Aldrich; puriss. p.a.,  $\geq$ 99.0% GC)
- 4. Reduced nicotinamide adenine dinucleotide phosphate (NADPH) (Sigma-Aldrich, 25-mM stock solution into reaction buffer)
- 5. Bovine serum albumin (BSA) (Sigma-Aldrich, 100 g/L stock solution into reaction buffer)
- 6. NADP-dependent isocitrate dehydrogenase from Bacillus sub*tilis* (BS-iDH, Sigma-Aldrich, 186 U/mL)
- 7. D/L-isocitrate (Sigma-Aldrich, 100-mM stock solution into reaction buffer)

8. MgCl<sub>2</sub> (Sigma-Aldrich, 500-mM stock solution into reaction buffer)

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- 9. Catalase from bovine liver (Sigma-Aldrich, stock solution at 50 U/mL into reaction buffer)
- 10. Purified A13-Red (11.6- $\mu$ M solution into Buffer B)
- 11. Dodecane (Sigma-Aldrich) 400-mM solution into absolute ethanol (Sigma-Aldrich)

### 3 Methods

The expression vector was designed for generating a fusion between the respective genes encoding for a  $6 \times$ His-tag (N-Term), CYP153A13a from *A. borkumensis* SK2 and the P450 reductase domain (RhFred) of P450 RhF from *Rhodococcus* sp. NCIMB 9784 (C-Term) (Fig. 1) [22]. This construction enables the expression of A13-Red protein which can be captured by

10	20	30	40	50	60
MGSSHHHHHH	SSGLVPRGSH	MMSTSSSTSN	DIQAKIINAT	SKVVPMHLQI	KALKNLMKVK
70	80	90	100	110	120
RKTIGTSRPQ	VHFVETDLPD	VNDLAIEDID	TSNPFLYRQG	KANAYFKRLR	DEAPVHYQKN
130	140	150	160	170	180
SAFGPFWSVT	RYEDIVFVDK	SHDLFSAEPQ	IILGDPPEGL	SVEMFIAMDP	PKHDVQRRAV
190	200	210	220	230	240
QGVVAPKNLK	EMEGLIRKRT	GDVLDSLPLD	TPFNWVPVVS	KELTGRMLAS	LLDFPYDERE
250	260	270	280	290	300
KLVGWSDRLS	GASSATGGEF	TNEDVFFDDA	ADMAWAFSKL	WRDKEARQKA	GEEPGFDLIS
310	320	330	340	350	360
MLQSNEDTKD	LINRPLEFIG	NLALLIVGGN	DTTRNSMSGG	VLALNQFPEQ	FEKLKANPKL
370	380	390	400	410	420
IPNWSLKYSL	ATPLAYMRRV	AKQDVELNGQ	TIKKGDRVLM	WYASGNQDER	KFENPEQFII
430	440	450	460	470	480
DRKDTRNHVS	FGYGVHRCMG	NRLAELQLRI	LWEELLPRFE	NIEVIGEPER	VQSNFVRGYS
490	500	510	520	530	540
KMMVKLTAKK	EFVLHR <u>HQPV</u>	TIGEPAARAV	SRTVTVERLD	RIADDVLRLV	LRDAGGKTLP
550	560	570	580	590	600
TWTPGAHIDL	DLGALSRQYS	LCGAPDAPSY	EIAVHLDPES	RGGSRYIHEQ	LEVGSPLRMR
610	620	630	640	650	660
GPRNHFALDP	GAEHYVFVAG	GIGITPVLAM	ADHARARGWS	YELHYCGRNR	SGMAYLERVA
670	680	690	700	710	720
GHGDRAALHV	SEEGTRIDLA	ALLAEPAPGV	QIYACGPGRL	LAGLEDASRN	WPDGALHVEH
730	740	750	760	770	780
FTSSLAALDP	DVEHAFDLEL	RDSGLTVRVE	PTQTVLDALR	ANNIDVPSDC	EEGLCGSCEV
790	800	810	820		
AVLDGEVDHR	DTVLTKAERA	ANRQMMTCCS	RACGDRLALR	L	

**Fig. 1** Complete protein sequence of A13-Red. The construction was designed to obtain a  $6 \times$ His-tag (residues from 5 to 10) merged with CYP153A13a from *A. borkumensis* SK2 (residues from 21 to 490) and the P450 reductase domain (RhFred) of P450 RhF from *Rhodococcus* sp. NCIMB 9784 (residues from 509 to 821). P450 and reductase domains are separated by the native peptidic linker (residues from 491 to 508) found in P450 RhF from *Rhodococcus* sp. NCIMB 9784. The *underlined sequence* corresponds to a characteristic peptide identified by nanoLC/MS/MS analysis

immobilized metal affinity chromatography (IMAC). After highcell-density cultivation, cells of BL21 Star<sup>TM</sup> (DE3)/pA13-Red can be harvested by centrifugation and frozen at  $-20^{\circ}$ C until the purification process described below.

**3.1 Construction** of pA13-Red Construction of pA13-Red was performed in three steps. The first step is the cloning of *rhfred* gene (accession number AAM67416) into pET28b(+) between *EcoRI* and *NotI* to yield the intermediate plasmid pETRed. The second step is the amplification of *cyp153a13a* gene (accession number AY505118) from *A. borkumensis* SK2 genomic DNA and its modification in order to remove the *EcoRI* restriction site. Finally, the third step is the cloning of this modified gene into pETRed to yield pA13-Red.

(a) Step 1: Cloning of *rhfred* gene into pET28b(+)

- The RhFred part of the construction should be amplified using RhFred-fw and RhFred-rv primers. The PCR reaction conditions are 95°C for 5 min; 30 cycles of 95°C for 20 s, 60°C for 10 s, and 70°C for 30s; and a final extension at 72°C for 5 min. The final reaction volume is 50 μL.
- 2. Purify the PCR product using the Qiagen cleanup kit.
- Digest the PCR fragment with *EcoRI* and *NotI* enzymes for 2 h at 37°C followed by gel extraction or PCR cleanup.
- 4. Clone the digested RhFred insert into pET28b(+) plasmid previously digested by *EcoRI* and *NotI* enzymes by standard ligation procedure and transform it into *E. coli* DH5α competent cells following manufacturer's protocol.
- 5. The success of the cloning should be verified by colony PCR using RhFred-fw and RhFred-rv primers and sequencing.
- 6. Prepare plasmid pETRed using Qiagen MiniPrep kit and following manufacturer's instructions for cultivation.
- (b) Step 2: Amplification and modification of cyp153a13a gene
  - 1. Freeze-dried sample of *A. borkumensis* SK2 (DSM 11573) is reactivated following provider instructions and cultivated at 28°C into Medium 809 for 48 h.
  - 2. After cultivation, cells are harvested by centrifugation and genomic DNA is extracted using Qiagen Gentra PureGene kit and following manufacturer's instructions.
  - 3. For removal of *EcoRI* site at position 887 of CYP153A13a, overlap extension strategy is performed. Extracted genomic DNA was used as a template for fragment A and fragment B amplifications. Fragment A is amplified using 153A13a-fw and Eco887-rv primers. The PCR reaction conditions are 95°C for 2 min; 30 cycles of 95°C for 20 s, 50°C for 10 s, and 70°C for 20 s; and a final extension at 72°C for 5 min. Fragment B is amplified using Eco887-fw and 153A13a-rv

primers under the same PCR conditions. The final reaction volumes are both 50  $\mu L.$ 

- 4. The crude PCR mixtures are digested by *DpnI* and PCR products purified with Qiagen PCR Purification Cleanup kit.
- 5. Fragments A and B are reassembled using 153A13a-fw and 153A13a-rv primers. The PCR reaction conditions are 95°C for 2 min; 30 cycles of 95°C for 20 s, 50°C for 10 s, and 70°C for 30 s; and a final extension at 72°C for 5 min. The final reaction volume is 50  $\mu$ L.
- 6. Purify the PCR product using the Qiagen cleanup kit.
- (c) Step 3: Cloning of *cyp153A13a* gene into pETRed
  - 1. Digest the PCR fragment with *NdeI* and *EcoRI* enzymes for 2 h at 37°C followed by gel extraction or PCR cleanup.
  - 2. Clone the digested CYPA53A13a insert into pETRed plasmid previously digested by *NdeI* and *EcoRI* enzymes by standard ligation procedure and transform it into *E. coli* DH5 $\alpha$  competent cells.
  - 3. The success of the cloning should be verified by colony PCR using 153A13a-fw and RhFred-rv primers and sequencing.
  - 4. Prepare plasmid pA13-Red using Qiagen MiniPrep kit and following manufacturer's instructions for cultivation (*see* **Note 1**).
  - 5. Transform pA13-Red into BL21 Star<sup>™</sup> (DE3) competent cells following manufacturer's protocol (*see* **Note 2**).
- 1. Thoroughly resuspend the cell pellet in cold (4°C) lysis buffer (use approximately 2 mL for 1 g of pellet).
- 2. Incubate at  $4^{\circ}$ C for 30 min.
- 3. Centrifuge at 10,000 rpm, 30 min, 4°C.
- 4. Filter the supernatant through a 0.45-µm membrane.
- 5. Before loading, equilibrate the HisPrep FF 16/10 column with 10 column volumes (CV) of binding buffer.
- 6. Load the cell-free extract on the column (2.5 mL/min).
- 7. Wash the column with binding buffer (5 mL/min) until optical density at 280 nm (OD280) reaches baseline.
- 8. Wash the column with washing buffer (5 mL/min) until OD280 reaches baseline.
- Elute the protein with elution buffer (2.5 mL/min). Collect 1-mL fractions and pool protein containing fractions.
- 10. Analyze the captured A13-Red by SDS-PAGE (Fig. 2) (see Note 3).
- 11. Elution buffer can be exchanged by ultrafiltration against conservation buffer and samples stored at  $-20^{\circ}$ C.

3.2 Purification of A13-Red on Nickel-NTA Column



**Fig. 2** SDS-PAGE (12% polyacrylamide) analysis of ion metal affinity chromatography fractions. The soluble fraction after lysis (1) was applied to HisPrep 16/10 column at a 2.5 mL/min flow rate. The flowthrough fraction containing unbound proteins (2) was collected. Two steps of washing were performed: the first one is carried out with binding buffer and the second one with washing buffer. Both fractions (respectively, 3 and 4) were collected. Elution of A13-Red was performed using histidine (5). A final elution with 2-M imidazole (6) was performed as a control

3.3 Desalting of A13- Red Fractions After Ni-NTA Column	After NI-NTA capture, A13-Red samples are contained into a salt- rich buffer (elution buffer). Prior to ion exchange chromatography, this salt-rich buffer has to be exchanged against desalting buffer for an optimal interaction between A13-Red and the stationary phase:			
	1. Before loading, two HiTrap desalting columns of 5 mL are connected in series and equilibrated with 10 CV of desalting buffer ( <i>see</i> <b>Note 4</b> ).			
	2. Load 1 mL of Ni-NTA purified A13-Red (1 mL/min).			
	<ol> <li>Elute the protein with desalting buffer (1 mL/min). Collect 0.5-mL fractions; pool those containing the protein and having the desired conductivity.</li> </ol>			
3.4 Purification of A13-Red on Q Sepharose XL Column	A second purification step for removing the 70-kDa impurity was set up based on ion exchange. Since A13-Red has a calculated pI of 6.01, Q Sepharose stationary phase (strong cation) was preferred ( <i>see</i> <b>Note 5</b> ). The optimized protocol is described below:			
	1. Before loading, equilibrate the Q Sepharose XL 1-mL pre- packed column with Buffer A (10 CV).			
	2. Load the desalted A13-Red sample (up to 25 mL, 1 mL/min).			
	3. Wash the column with Buffer A (1 mL/min) until OD280 reaches baseline.			
	4. Elute A13-Red with Buffer B (1 mL/min). Collect 1-mL frac- tions and pool A13-Red containing fractions.			



Fig. 3 SDS-PAGE (12% polyacrylamide) analysis after ion exchange and size exclusion chromatography. Lane 1: A13-Red sample after Q Sepharose XL (strong cation) chromatography. Lanes 2 and 3: A13-Red samples after Superdex 75 16/60 chromatography (not described in this chapter)

- 5. Analyze the purified A13-Red by SDS-PAGE (Fig. 3) (see Notes 6 and 7).
- 6. Store the protein at 4°C. For prolonged storage (more than 1 week), it is preferable to store A13-Red in small aliquots at  $-80^{\circ}$ C.

A13-Red from this two-step purification protocol can be used without further purification in hydroxylation reactions.

Every reaction was performed into Wheaton 5-mL V-Vials at 25°C under gentle magnetic stirring, repeated thrice, and errors were estimated to be 5–10%. The protocol described below is optimized for a 1-mL total reaction volume (aqueous phase 500 µL/organic phase 500  $\mu$ L).

- 1. Add 279 µL of reaction buffer into a Wheaton 5-mL V-Vial.
- 2. Add 50 μL of BSA 100 g/L (see Note 8).
- 3. Add 5 U (25 µL in our case) of BS-iDH stock solution (see Note 9).
- 4. Add 100  $\mu$ L of D/L-isocitrate 100 mM.
- 5. Add 5  $\mu$ L of MgCl<sub>2</sub> 500 mM.
- 6. Add 0.5 U of catalase stock solution (10  $\mu$ L in our case) (see Note **10**).
- 7. Add *n*-octane (500  $\mu$ L) drop wise under gentle magnetic stirring.
- 8. Add 21 µL of A13-Red stock solution (final concentration 500 nM) and incubate for 1 min.

3.5 Enzymatic Hvdroxvlation of n-Octane

- 9. Add 10  $\mu$ L of NADPH 25 mM to trigger the hydroxylation reaction.
- 10. Tightly screw the V-Vial cap and seal it with parafilm.

For analysis, the emulsified solution was transferred into a 1.5-mL tube and dodecane was added (5  $\mu$ L, final concentration into the organic phase 4 mM, neglecting dodecane solubility into the aqueous phase). The sample was vortexed for 1 min and subjected to centrifugation (2 min, 10,000*g*, room temperature), and the organic layer was collected and analyzed by GC–MS (Shimadzu GC-2010 instrument) with an SLB-5MS column (30-m length; 0.25-mm diameter; Supelco). The injection and detection temperatures were both 250°C. For reactions with octane, the method used was an isotherm at 35°C for 5 min, followed by a temperature raise to 190°C at a rate of 10°C/min. The products were quantified by selected ion monitoring (SIM) program, and their concentrations were calculated from the corresponding area using dodecane as the internal standard (*see* **Note 11**).

### 4 Notes

- 1. The resulting plasmid pA13-Red and the strain BL21 Star<sup>™</sup>(DE3)/pA13-Red are freely available for academic and collaborative research under request.
- 2. Cultivation of microorganisms in flasks with a rich medium (LB, TB, or  $2 \times YT$ ) is the most simple approach, but important parameters for reproducibility such as medium composition, aeration, or pH cannot be controlled tightly. Expression of A13-Red in recombinant *E. coli* using this simple approach yielded in average  $35 \pm 25$  mg of functional CYP per liter of culture with TB as described in references [22, 23]. We even witnessed unproductive cultivations. These large variations from batch to batch prompted us to develop a more reliable protocol to express A13-Red using high-cell-density cultivation (HCDC) with instrumented fermenters as described in details in reference [23]. Using this protocol, up to 5,000 nmol of functional A13-Red can be obtained per liter of HCDC.
- 3. IMAC can be sufficient to obtain A13-Red with 95% purity after flask cultivation, but it was not true for A13-Red produced by HCDC where only 70–75% purity was reached by performing IMAC only (Fig. 2). Since A13-Red is irreversibly inhibited by imidazole, elution was performed using histidine. One of the most abundant contaminating proteins (approx. 70 kDa) was identified to be CYP153A13a by acrylamide gel extraction and trypsin digestion of the extract followed by NanoLC/MS/MS analysis (Fig. 1). We assumed that a portion

of A13-Red was subjected to proteolysis into the cytoplasm of the bacteria during expression.

- 4. For scaling up the desalting step on A13-Red samples, up to five columns can be connected in series. For sample volumes up to 15 mL, HiPrep 26/10 Desalting is available. Up to four HiPrep 26/10 Desalting columns can be connected in series without increased backpressure (up to 60-mL sample volume).
- 5. A first series of experiments with a linear gradient of NaCl showed that the 70-kDa impurity was eluted at 80-mM NaCl while A13-Red at 200 mM.
- 6. High molecular weight impurities (above 200 kDa) still remained present after this step (Fig. 3). It is likely that these contaminants are polymers of A13-Red since their concentration progressively increased over time into the conservation buffer at -80°C into pure A13-Red samples. They can be eliminated by size exclusion chromatography using a Superdex 75 16/60 column (Fig. 3), but their presence had no significant impact on enzymatic hydroxylation activity.
- 7. For each purification step, functional P450 concentration is determined from the carbon monoxide difference spectra as previously described [24], using an extinction coefficient of 91 mM<sup>-1</sup> cm<sup>-1</sup> for the 450 minus 490-nm peak. Compared to the Bradford protein assay, CO-binding is a selective assay for P450. Moreover, it allows to determine the ration between functional (450 nm) and nonfunctional P450 (420 nm).
- Stock solutions for the different additives should be prepared with reaction buffer (potassium phosphate buffer (100 mM, pH 7.4), glycerol 10%). For example, to prepare MgCl<sub>2</sub> 500mM stock solution, solid MgCl<sub>2</sub> should be dissolved into reaction buffer.
- 9. By definition, for BS-iDH, one unit (U) corresponds to the amount of enzyme which converts 1- $\mu$ mol D-isocitrate to  $\alpha$ -ketoglutarate per minute at pH 7.5 and 37°C (NADP as cofactor).
- 10. By definition, one unit of catalase will decompose  $1.0 \ \mu$ mole of  $H_2O_2$  per min at pH 7.0 at 25°C, while the  $H_2O_2$  concentration falls from 10.3 to 9.2 mM, measured by the rate of decrease of OD240.
- 11. Under these conditions, A13-Red remains active up to 78 h, and 1.6 mM of *n*-octanol is formed corresponding to a total turnover number about 3,250. No other regio-isomer or overoxidation product was detected confirming the remarkable regio- and chemoselectivity of A13-Red.

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#### References

- Arakawa H, Aresta M, Armor JN, Barteau MA, Beckman EJ, Bell AT, Bercaw JE, Creutz C, Dinjus E, Dixon DA, Domen K, DuBois DL, Eckert J, Fujita E, Gibson DH, Goddard WA, Goodman DW, Keller J, Kubas GJ, Kung HH, Lyons JE, Manzer LE, Marks TJ, Morokuma K, Nicholas KM, Periana R, Que LJ, Rostrup-Nielson J, Sachtler WM, Schmidt LD, Sen A, Somorjai GA, Stair PC, Stults BR, Tumas W (2001) Catalysis research of relevance to carbon management: progress, challenges, and opportunities. Chem Rev 101(4):953–996
- Bordeaux M, Galarneau A, Drone J (2012) Catalytic, mild, and selective oxyfunctionalization of linear alkanes: current challenges. Angew Chem Int Ed 51:10712–10723
- Sabirova JS, Ferrer M, Lünsdorf H, Wray V, Kalscheuer R, Steinbüchel A, Timmis KN, Golyshin PN (2006) Mutation in a 'tesB-like' hydroxyacyl-coenzyme A-specific thioesterase gene causes hyperproduction of extracellular polyhydroxyalkanoates by Alcanivorax borkumensis SK2. J Bacteriol 188(23):8452–8459
- Kalscheuer R, Stöveken T, Malkus U, Reichelt R, Golyshin PN, Sabirova JS, Ferrer M, Timmis KN, Steinbüchel A (2007) Analysis of storage lipid accumulation in Alcanivorax borkumensis: evidence for alternative triacylglycerol biosynthesis routes in bacteria. J Bacteriol 189 (3):918–928
- van Beilen JB, Li Z, Duetz WA, Smits THM, Witholt B (2003) Diversité des systèmes alcane hydroxylase dans l'environnement. Oil Gas Sci Technol 58(4):427–440
- van Beilen JB, Funhoff EG, van Beilen JB (2005) Expanding the alkane oxygenase toolbox: new enzymes and applications. Curr Opin Biotechnol 16(3):308–314
- Maier T, Förster HH, Asperger O, Hahn U (2001) Molecular characterization of the 56kDa CYP153 from Acinetobacter sp. EB104.

Biochem Biophys Res Commun 286 (3):652–658

- Funhoff EG, Bauer U, García-Rubio I, Witholt B, van Beilen JB (2006) CYP153A6, a soluble P450 oxygenase catalyzing terminal-alkane hydroxylation. J Bacteriol 188(14):5220–5227
- Funhoff EG, Salzmann J, Bauer U, Witholt B, van Beilen JB (2007) Hydroxylation and epoxidation reactions catalyzed by CYP153 enzymes. Enzyme Microb Technol 40 (4):806–812
- van Beilen JB, Funhoff EG, Van Loon A, Just A, Kaysser L, Bouza M, Ro M, Li Z, Witholt B, van Loon A, Holtackers R, Rothlisberger M (2006) Cytochrome P450 alkane hydroxylases of the CYP153 family are common in alkanedegrading eubacteria lacking integral membrane alkane hydroxylases. Appl Environ Microbiol 72(1):59–65
- 11. van Beilen JB, Lu D, Bauer U, Witholt B, Duetz WA (2005) Biocatalytic production of perillyl alcohol from limonene by using a novel Mycobacterium sp. Cytochrome P450 alkane hydroxylase expressed in Pseudomonas putida. Appl Environ Microbiol 71(4):1737–1744
- 12. Sono M, Roach MP, Coulter ED, Dawson JH (1996) Heme-containing oxygenases. Chem Rev 96(7):2841–2888
- Groves JT (1985) Key elements of the chemistry of cytochrome P-450: the oxygen rebound mechanisms. J Chem Educ 62:928–931
- 14. Ogliaro F, Harris N, Cohen S, Filatov M, de Visser SP, Shaik S (2000) A model 'rebound' mechanism of hydroxylation by cytochrome P450: stepwise and effectively concerted pathways, and their reactivity patterns. J Am Chem Soc 122:8977–8989
- Rittle J, Green MT (2010) Cytochrome P450 compound I: capture, characterization, and C-H bond activation kinetics. Science 330 (6006):933–937

- 16. Shaik S, Lai W, Chen H, Wang Y (2010) The valence bond way: reactivity patterns of cytochrome P450 enzymes and synthetic analogs. Acc Chem Res 43(8):1154–1165
- 17. Shaik S, Cohen S, Wang Y, Chen H, Kumar D, Thiel W (2010) P450 enzymes: their structure, reactivity, and selectivity-modeled by QM/ MM calculations. Chem Rev 110 (2):949–1017
- Sibbesen O, De Voss JJ, Ortiz de Montellano PR (1996) Putidaredoxin reductaseputidaredoxin-cytochrome P450 cam triple fusion protein. Biochemistry 271(37): 22462–22469
- 19. Li S, Podust LM, Sherman DH (2007) Engineering and analysis of a self-sufficient biosynthetic cytochrome P450 PikC fused to the RhFRED reductase domain. J Am Chem Soc 129(43):12940–12941
- 20. Scheps D, Honda Malca S, Richter SM, Marisch K, Nestl BM, Hauer B (2013) Synthesis of ω-hydroxy dodecanoic acid based on an

engineered CYP153A fusion construct. Microbiol Biotechnol 6(6):694–707

- Nodate M, Kubota M, Misawa N (2006) Functional expression system for cytochrome P450 genes using the reductase domain of selfsufficient P450RhF from Rhodococcus sp. NCIMB 9784. Appl Microbiol Biotechnol 71 (4):455–462
- 22. Bordeaux M, Galarneau A, Fajula F, Drone J (2011) A regioselective biocatalyst for alkane activation under mild conditions. Angew Chem Int Ed 50(9):2075–2079
- Bordeaux M, Girval D, Rullaud R, Subileau M, Dubreucq E, Drone J (2014) High-cell-density cultivation of recombinant Escherichia coli, purification and characterization of a selfsufficient biosynthetic octane ω-hydroxylase. Appl Microbiol Biotechnol 98:6275–6283
- Omura T, Sato R (1964) The carbon monoxide-binding pigment of liver microsomes. I. Evidence for its hemoprotein nature. J Biol Chem 239:2370–2378