

## Magic Angle Spinning Nuclear Magnetic Resonance of the Chlorosomes

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### Summary

Chlorosomes, the oblong light-harvesting bodies of green photosynthetic bacteria, are attached to the inner side of the cytoplasmic membrane. Magic Angle Spinning Nuclear Magnetic Resonance (MAS NMR) was used to study isotopically labeled chlorosomes and in vitro model compounds. Using uniform isotope labeling of chlorophyll molecules, 2D and 3D MAS NMR dipolar correlation spectroscopy was performed in high magnetic field. The chemical shifts provided invaluable information about the structure via ring current effects, while long range correlations were generated that lead to intermolecular distance constraints. Novel methodology was developed and implemented using a Chl *a*/H<sub>2</sub>O aggregate, and a structural arrangement of bilayers of Chl sheets with interdigitating tails was resolved. Application of this technology to chlorosomes and BChl *c* aggregates provided unambiguous evidence that self-organization of BChl *c* is the principal structural factor in establishing the rod elements in chlorosomes. This confirms that proteins do not play an essential role in the light harvesting function, which is of fundamental biological interest. Finally, MAS NMR leads to a bilayer model for the tubular supra-structure of sheets of BChl *c* in the chlorosomes of *Chlorobium tepidum*.

Note: Readers are encouraged to visit the website (<http://epub.ub.uni-muenchen.de/archive/00000776/>) for 'supplementary material,' where Figs. S1–S2 referred to in the text are posted.

### I. Introduction

Chlorosomes are found in photosynthetic green bacteria as ellipsoid vesicles of about 100–300 nm in length attached to the inner surface of the

cytoplasmic membrane, where they provide a large cross-section for the absorption of sunlight (Blankenship et al., 1995; Olson, 1998). The chlorosomes contain rod-shaped elements of 5 nm in diameter for *Chloroflexus* (*Cfl.*) and 10 nm for *Chlorobium* (*Chl.*)

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as visualized by electron microscopy (Staelin et al., 1978, 1980). The assumption that the rods are formed by protein-pigment complexes, as found in other photosynthetic elements, has been challenged by increasing evidence that self-organized bacteriochlorophyll (BChl) *c* is responsible for the shape and function of the antennae. The crucial evidence is that the *in vitro* aggregation of BChl *c* leads to similar structures and spectroscopic properties as occur in the natural system (for review, see Tamiaki et al., 2002). Small-angle neutron scattering studies have indicated cylindrical shaped micelles for aggregates prepared in organic media (Worcester et al., 1986). In addition, various spectroscopic methods suggest a highly ordered structure of both aggregated BChl *c* and the chlorosomes (Griebenow et al., 1991; Matsuura et al., 1993; Hildebrandt et al., 1994). Molecular modeling has also been used to explain the structural features of the cylindrical micelles (Holzwarth and Schaffner, 1994). For the chlorosomes, an intermolecular bonding network has been proposed, where the 3<sup>1</sup>-hydroxy group of BChl *c* coordinates to the central metal of a neighboring BChl *c* molecule and also hydrogen bonds with the 13<sup>1</sup> carbonyl-oxygen of yet another BChl molecule (Holzwarth and Schaffner, 1994; Chiefari et al., 1995).

In the past, NMR has contributed significantly to the study of chlorophyll chemistry, which was reviewed extensively (Abraham and Rowan, 1991). Two different routes have been worked out to use NMR data for structure determination in uniformly <sup>13</sup>C-labeled chlorophylls. Firstly, the chemical shift assignment obtained from correlation experiments can provide information about the spatial structure. This is consistent with the early analyses of the stacking in small chlorophyll aggregates from ring current shifts (Abraham and Rowan, 1991). In particular, <sup>1</sup>H aggregation shifts are important for the purpose of structure determination. The chemical shift of <sup>1</sup>H is relatively insensitive to electronic perturbations, while shifts induced by ring currents are large on the <sup>1</sup>H shift scale of 10–15 ppm. In a similar way, the ring current shifts are indispensable for structure elucidation in the solid state. Secondly, structural information in the solid state may be obtained from the measurement of distance restraints revealing close

contacts between molecular moieties.

In this Chapter, recent progress in resolving the nature of the chlorosomal antennae that has been acquired by magic angle spinning (MAS) NMR studies is discussed. The structure elucidation strategy using a chlorophyll (Chl) *a*/H<sub>2</sub>O model system is summarized and the MAS NMR investigations of the chlorosomes are reviewed. It has been found that the chlorosomes consist mainly of aggregated BChl that is indistinguishable from BChl *c* precipitated from hexane. Based on MAS NMR experiments, a bilayer tube model for the chlorosomes has been proposed.

## II. Aggregated Hydrated Chlorophyll (Chl *a*/H<sub>2</sub>O) as a Model for Magic Angle Spinning Nuclear Magnetic Resonance Technology Development

Much of the MAS NMR methodology was first tested using a Chl *a*/H<sub>2</sub>O aggregate and then applied in a second step to study the structure of the chlorosomes. The Chl *a*/H<sub>2</sub>O aggregate is very suitable to develop and evaluate novel pulse sequences. As a moderately-sized molecular system, it represents an intermediate between small model compounds, such as the tyrosine HCl salt, that are frequently used for a demonstration of principles of pulse techniques, and larger systems of genuine biological interest. The chemical structure of Chl *a* is depicted in Fig. 1. The aggregation of Chl *a* and the Chl-water interaction in Chl *a*/H<sub>2</sub>O micelles has been extensively studied in the past (Katz et al., 1991). Chl *a* can only form solid aggregates by incorporation of H<sub>2</sub>O, which is believed to coordinate the central magnesium atom and form hydrogen bonds with the carbonyl functions of surrounding molecules. Recent MAS NMR results suggest a close homology between the structure of Chl *a*/H<sub>2</sub>O and crystalline ethyl-chlorophyllide *a*, with two water molecules forming the bridging network (Boender et al., 1995; de Boer et al., 2002; van Rossum et al., 2002).

Complete <sup>1</sup>H and <sup>13</sup>C assignments were obtained in a field of 14.1 T using the <sup>13</sup>C-<sup>13</sup>C radio frequency-driven dipolar recoupling (RFDR) and <sup>1</sup>H-<sup>13</sup>C Lee Goldberg/cross polarization (LG-CP) methods (van Rossum et al., 2002) (see Table S1 in Supplementary Material). The line widths are small (120–200 Hz) revealing a rigid and well-ordered structure. The <sup>1</sup>H and <sup>13</sup>C shifts of the monomer in solution were used

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*Abbreviations:* BChl – bacteriochlorophyll; *Cfl.* – *Chloroflexus*; Chl – chlorophyll; *Chl.* – *Chlorobium*; CP – cross polarization; CS – chemical shift; LG – Lee Goldberg; MAS – magic angle spinning; NMR – nuclear magnetic resonance; RFDR – radio frequency-driven dipolar recoupling; WISE – wideline separation

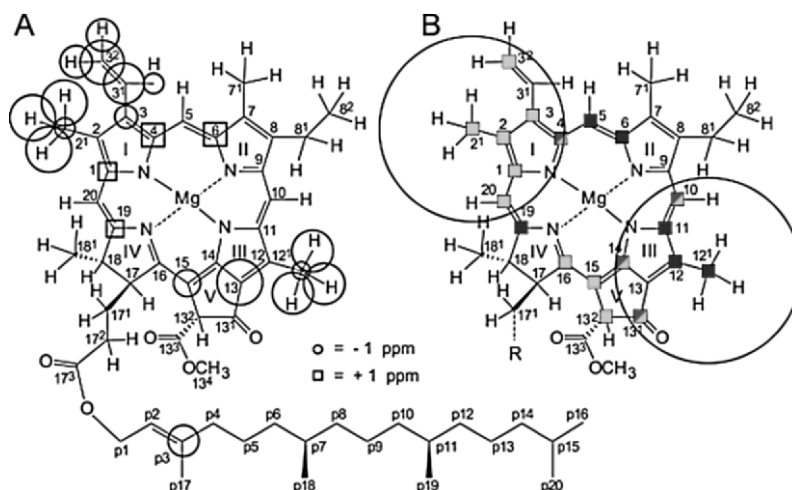


Fig. 1. (A) Visual representation of the carbon and proton aggregation shifts ( $\Delta\sigma$ ) for aggregated Chl *a*. The circles around the carbon and hydrogen atoms represent upfield aggregation shifts, squares represent downfield shifts. The sizes of the circles and squares reflect the magnitude of the aggregation shifts. (B) Schematic representation of the assignment of heteronuclear correlations involving the  $2^1\text{-H}_3$ , which are shown in shaded grey or the  $12^1\text{-H}_3$ , which are solid black. For the 4-C, 10-C,  $13^1\text{-C}$  and 14-C the correlations with proton signals are assigned to transfer from both methyl groups. The ranges for intramolecular transfer for the two methyl groups are indicated with the circles. Intramolecular correlations involving the  $3^1\text{-C}$  and  $13\text{-C}$  could not be assigned. Reprinted from Van Rossum et al. (2002). Copyright (2002), with permission from Elsevier.

to estimate the aggregation shifts. The  $^1\text{H}$  aggregation shifts are consistent with the pattern for  $^{13}\text{C}$  (Fig. 1). Upfield aggregation shifts up to  $\sim 5$  ppm are observed, mostly induced by ring currents. The shift pattern is in agreement with models for the Chl *a*/water aggregates that assume strong overlap between the rings A and the rings C and E of adjacent molecules forming stacks (Worcester et al., 1986). In addition, in a 2D  $^{13}\text{C}$ - $^{13}\text{C}$  RFDR spectrum recorded with a long mixing time of  $\tau_m = 10$  ms, several cross-peaks are observed that can be attributed to intermolecular transfer (Boender et al., 1995).

Heteronuclear  $^1\text{H}$ - $^{13}\text{C}$  spectra were recorded with long LG-CP times to generate intermolecular correlations (van Rossum et al., 2002). For a LG-CP time of 2 ms, a maximum transfer distance ( $d_{\text{max}}$ ) of  $\sim 4.2$  Å was estimated from an analysis of the correlations involving the  $2^1\text{-H}_3$  and  $12^1\text{-H}_3$  protons. Weak transfer was observed from the  $2^1\text{-H}_3$  protons to the 10,  $13^1$ ,  $13^2$ , 14, 15 and 16 carbons at the opposite side of the molecule (Fig. 1B), which was assigned to intermolecular contact over distances  $\sim 4$  Å. Similarly, weak correlations between the  $12^1\text{-H}_3$  protons and the 4, 5, 6 and 19 carbons indicate intermolecular transfer. They confirm the 2D arrangement of the Chl *a* in sheets (Fig. 3A).

In addition, a CP<sup>3</sup> experiment can provide inter-

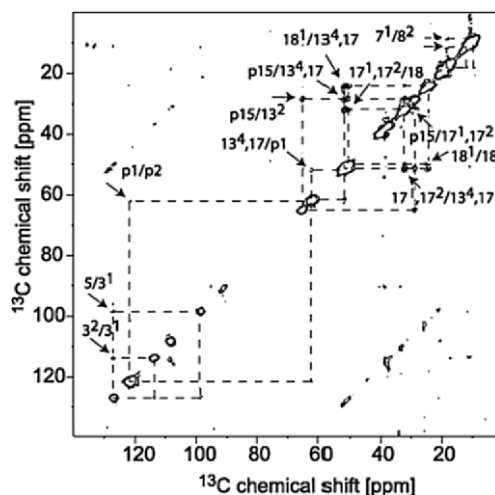


Fig. 2. Contour plot of an absorption mode 2D  $^{13}\text{C}$ - $^{13}\text{C}$  CP<sup>3</sup> MAS NMR correlation spectrum of aggregated Chl *a*/H<sub>2</sub>O recorded with a spinning speed of 14.5 kHz in a field of 17.6 T. Arrows and labels are used to indicate cross-peaks, which are connected to the corresponding diagonal peaks and mirror peaks by dashed lines. Data were acquired with  $\tau_m = 100$   $\mu\text{s}$ . A prescan delay of 1 s was used for a total of 192 scans for each of 200  $t_1$  points.

molecular  $^{13}\text{C}$ - $^{13}\text{C}$  correlations employing mixing by  $^1\text{H}$  spin diffusion. An example of a CP<sup>3</sup> experiment, using a short mixing time of 0.1 ms, is shown in

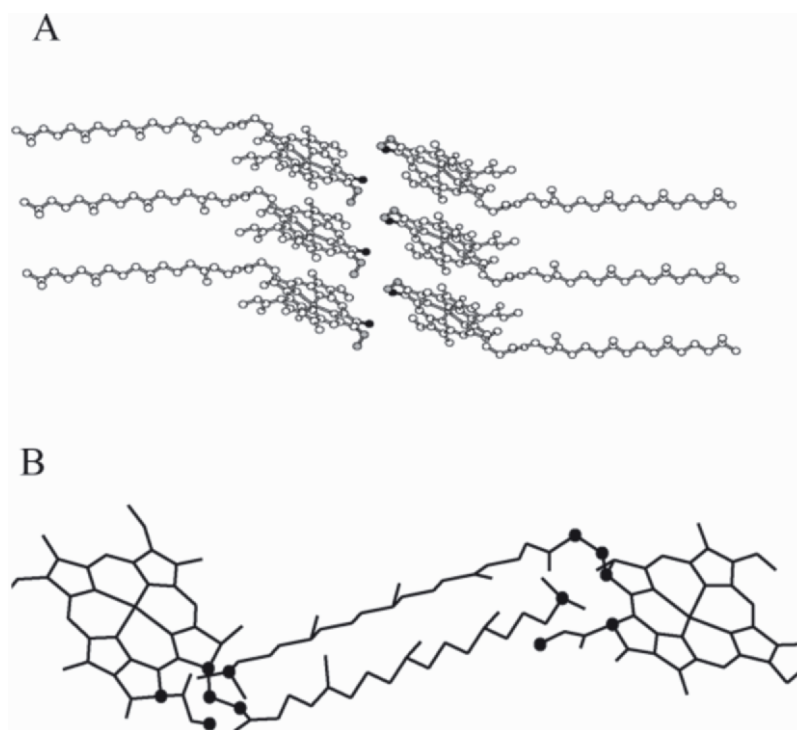


Fig. 3. (A) Schematic representation of a bilayer formed from two sheets of microcrystalline aggregated Chl *a*/H<sub>2</sub>O. The orientation of the 2-D sheets is perpendicular to the plane of the paper and the view is parallel to the stack. The 7-Me and 8-Et moieties are shown in black and grey, respectively. (B) The proposed structural arrangement of the two Chl *a* molecules in the unit cell. Solid circles indicate the carbons involved in the intermolecular correlations.

Fig. 2 (de Boer et al., 2002). Correlations between the P15 carbon at the end of the phytyl chain and the 17<sup>1</sup>, 17<sup>2</sup>, 13<sup>2</sup> and 13<sup>4</sup> carbons near the base of the tail are observed. From a series of experiments using increasing mixing times, it is estimated that  $d_{\max}$  is about  $\sim 4$  Å for a mixing time of 0.1 ms. The length of the phytyl tail is  $\sim 20$  Å, much longer than  $d_{\max}$  and hence the correlations involving P15 are attributed to intermolecular transfer, over distances no larger than  $\sim 4$  Å. This leads to the bilayer model with stretched and inter-digitating tails (Fig 3B). This model is supported by the very modest inhomogeneous line broadening for the phytyl tails and the absence of aggregation shifts for the tails, which provides strong evidence that the tails have essentially the same conformation as in solution.

Finally, <sup>1</sup>H signals, correlating with similar bridging moieties in the heteronuclear MAS NMR spectra, could be assigned to structural water similar to ethylchlorophyllide *a*. In addition, a small doubling of the 7-C, 7<sup>1</sup>-CH<sub>3</sub> and 8<sup>1,2</sup>-C was resolved. This provides evidence for two marginally different well-defined molecular environments at the interface between two bilayers (Fig. 3A).

### III. Self-organization of Bacteriochlorophyll is the Main Structural Feature of the Chlorosomal Antennae

A complete <sup>13</sup>C assignment was made for the NMR response of the chlorosomes and various BChl *c* homologues from the uniformly <sup>13</sup>C labeled chlorosomes of *Chl. tepidum* (Balaban et al., 1995) (see Table S2 in Supplementary Material). Figure 4 shows the chemical structure of the most abundant [8-Et, 12-Et]BChl *c* homologue. 2D <sup>13</sup>C-<sup>13</sup>C RFDR dipolar correlation spectra of solid BChl *c* aggregated in hexane and the chlorosomal antennae are depicted in Fig. 5 (Balaban et al., 1995; Boender, 1996). Many peaks are resolved in the spectra, and the two spectra are almost identical. The predominant component of the MAS NMR signals in the chlorosomes is from BChl *c*. Only a minor fraction is observed in the chlorosomes that is not present in the aggregate. These peaks are attributed to lipids and proteins in the chlorosomes. Correlations between the carbonyl and  $\alpha$  carbons of a small protein component are indicated in Fig 5.

The 2D response of the BChl *c* in intact chlorosomes is virtually indistinguishable from the data

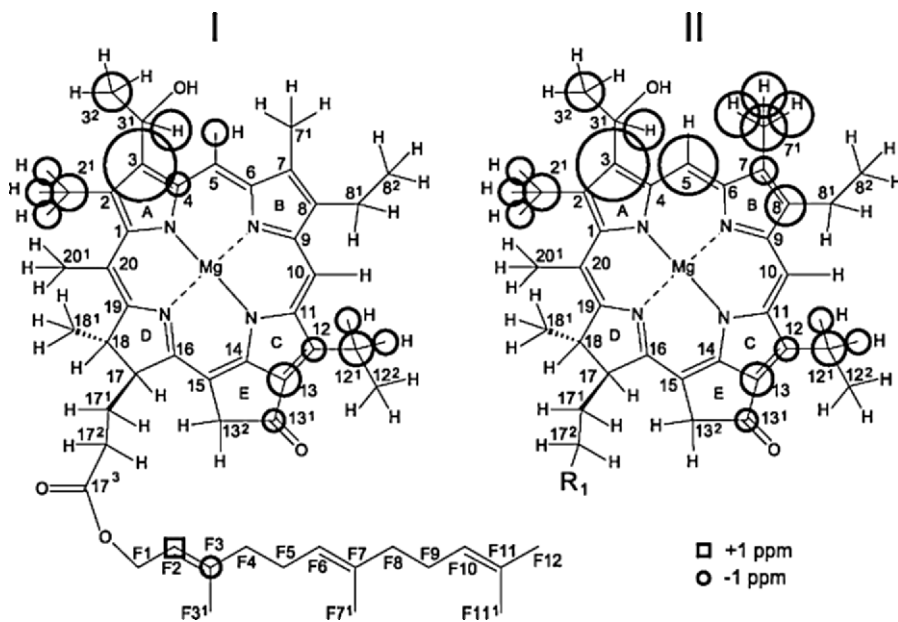


Fig. 4. Carbon and proton aggregation shifts  $|\Delta\sigma_i| \geq 1.5$  ppm of BChl *c* with  $\Delta\sigma_i = \sigma_i - \sigma_{\text{liq}}$ . The NMR response comprises two components, denoted I and II. The circles around the carbon and hydrogen atoms represent upfield aggregation shifts, the squares downfield aggregation shifts. The size of a circle or square is proportional to the magnitude of the aggregation shift. Reprinted with permission from Van Rossum et al. (2001). Copyright (2001) Am. Chem. Soc.

collected from the *in vitro* aggregate with respect to chemical shifts, line widths and relative intensities of the cross-peaks. This demonstrates that the minor fractions of proteins and lipids are not an integral part of the BChl *c* assembly (Balaban et al., 1995; Boender, 1996). In this way, the NMR data provide conclusive evidence that self-assembly of BChl *c* is the structural basis for the BChl *c* organization *in vivo*, without the intervention of proteins.

In earlier 1D MAS NMR experiments, solid BChl *c* was also compared with the chlorosomes, providing the first evidence for the same aggregated forms (Nozawa et al., 1990, 1991, 1994). The  $T_{\text{CH}}$  and  $T_{\text{lp}}^{\text{H}}$  values were also comparable for both samples, which corroborates the structural correspondence between the aggregate and the natural system.

The observation that a biological function can be realized without active participation of protein can be considered anomalous, since the central dogma of molecular biology states that all function originates from the DNA code via protein: DNA  $\rightarrow$  RNA  $\rightarrow$  protein  $\rightarrow$  function. In a chlorosome, the biological function of the antennae is based on self-assembly steered by the physico-chemical properties of the constituting molecules and top-down control from

higher levels in the biological hierarchy. Other examples of this principle have been encountered in the past, such as biomineralization of inorganic matter into a morphology controlled by the organism (Bäuerlein, 2003). For instance, the magnetite crystals in the magnetosomes of magnetobacteria correspond to single magnetic domains of 40–120 nm, and allow the bacteria to orient themselves using the earth magnetic field.

The self-organization of BChl *c* may be of interest for artificial photosynthesis. The elucidation of the functional concepts behind photosynthesis in nature is expected to be of assistance in improving artificial devices like photovoltaic cells. Although the actual application of artificial molecular solar energy converters may still be years away, elucidation of the concepts behind nature's efficient energy converting system will be of use to photovoltaics and molecular electronics research. In particular, the discovery of natural light-harvesting systems without proteins may be of help to indicate novel routes in photovoltaics research.

To produce aggregates resembling the chlorosomes, the solvent used in the process is essential. In particular, the effect of different solvents to form

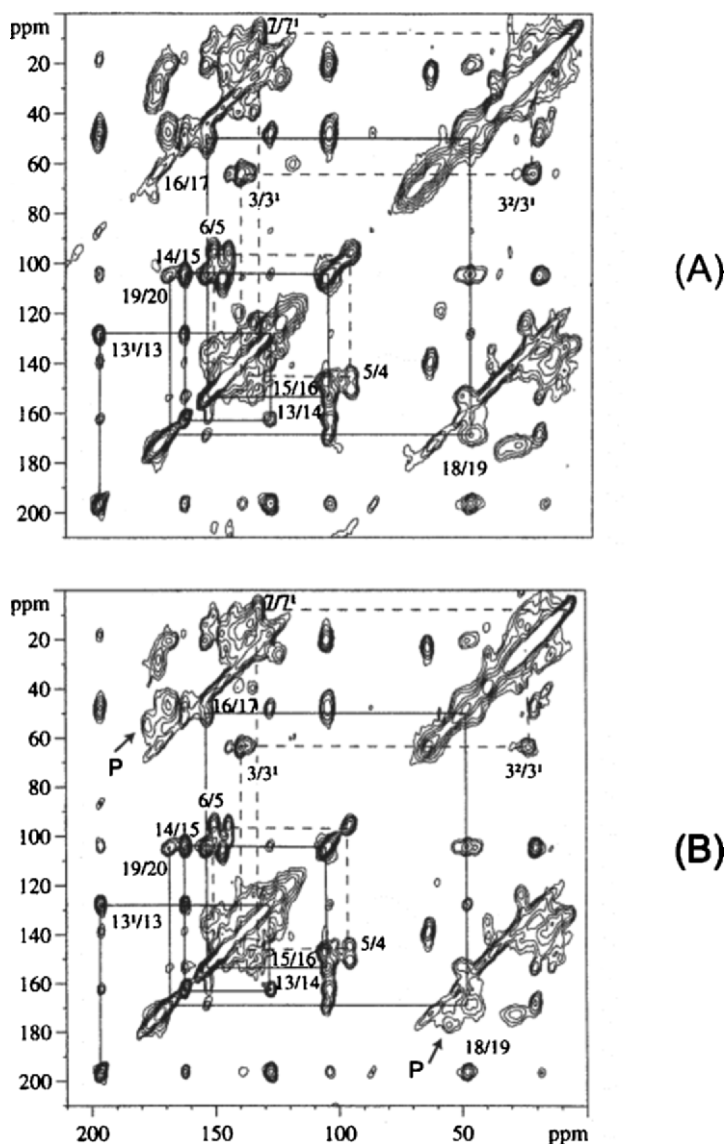


Fig. 5. Contour plots of 2D MAS NMR dipolar correlation NMR spectra of uniformly  $^{13}\text{C}$  labeled BChl *c* aggregates (A) and  $^{13}\text{C}$  labeled chlorosomes (B), recorded in a magnetic field of 9.4 T. The MAS rate was 11 kHz and a CP time of 1 ms was used. The lines indicate sequences of nearest neighbor correlations. A small protein fraction in (B) is indicated with P. Reprinted from Boender (1996).

solid aggregates of ( $3^1 R$ ) BChl *c* was recently studied (Umetsu et al., 1999). Various techniques including  $^{13}\text{C}$  CP/MAS were applied to investigate the size and order of the aggregates formed after drying in diethyl ether,  $\text{CH}_2\text{Cl}_2$ ,  $\text{CCl}_4$  or  $\text{CH}_2\text{Cl}_2$  in an excess of hexane. From the magnetic circular dichroism spectra, the aggregates treated with  $\text{CCl}_4$  and hexane were estimated to be larger than those with diethyl ether and  $\text{CH}_2\text{Cl}_2$ . The circular dichroism spectra show differences between the  $\text{CH}_2\text{Cl}_2$ -treated aggregates and the diethyl ether and  $\text{CCl}_4$ -treated samples, indicating

a different molecular arrangement. In addition, the BChl *c* treated with  $\text{CH}_2\text{Cl}_2$  is weakly diffracting by X-ray similar to methyl-BChlide *c* formed in hexane. This shows that the tails leave the stacking of the BChl *c* intact. In addition, recent MAS NMR experiments revealed two distinct sets of resonances for a  $\text{CH}_2\text{Cl}_2$ -treated uniformly  $^{13}\text{C}$  and  $^{15}\text{N}$  labeled ( $3^1 R$ ) BChl *c* aggregate, indicating the formation of a transient species between the dimer and the stacked form (Umetsu et al., 2004).

#### IV. A 3-Dimensional Model for the Structure of the Chlorosomal Antennae

The structure of the chlorosomal antennae was recently investigated with 2D and 3D MAS NMR. The  $^{13}\text{C}$  shifts assigned from the 2D RFDR experiments provided the first step to a structural model (Balaban et al., 1995). The relatively large  $^{13}\text{C}$  line widths contrast with the microcrystalline Chl  $\alpha/\text{H}_2\text{O}$  aggregates and reveal significant disorder down to the microscopic level. Large ring current shifts are detected in the regions of rings A, -C and E (Fig. 4).  $^{13}\text{C}$ - $^{13}\text{C}$  RFDR experiments were performed, using long mixing times of 5–10 ms (Boender, 1996; van Rossum et al., 1998a). Correlations between the  $^{13}\text{C}$ -carbon and the 19-, 20-, 1-, 2- and 3-carbons were attributed to intermolecular transfer and appear to confirm the layer structure. In addition, rotational resonance experiments show that the  $\text{C}3^1\text{-C}13^1$  distance is similar to the  $\text{C}20\text{-C}3^1$  distance in accordance with hydrogen bonding between the  $3^1$  and  $13^1$  groups (Nozawa et al., 1994).

The  $^1\text{H}$  shifts were determined in a 3D experiment using rapid MAS in combination with a high magnetic field. Under these conditions, it was shown that proton shifts can be determined and intermolecular heteronuclear correlations as well as hydrogen-bonding characteristics can already be determined with simple pulse schemes such as the wideline separation (WISE) technique (van Rossum et al., 1996). Since the chlorosome response is considerably inhomogeneously broadened, a 2D  $^{13}\text{C}$ - $^{13}\text{C}$  RFDR experiment was extended in a straightforward fashion by a third  $^1\text{H}$  WISE dimension to construct a 3D  $^1\text{H}$ - $^{13}\text{C}$ - $^{13}\text{C}$  MAS NMR experiment (van Rossum et al., 2001). As a result, all  $^1\text{H}$  resonances of the chlorosomes could be assigned (see Table S2 in Supplementary Material). The aggregation shifts (Fig. 4) are consistent with the  $^{13}\text{C}$  pattern. Two fractions **I** and **II** are observed in the NMR dataset. In both fractions, two regions with pronounced upfield shifts are visible around ring A and C/E. In addition, in fraction **II**, large upfield shifts for the 5-C and the 7-Me are detected. This suggests the existence of two distinct structural arrangements, related to the rod-like suprastructure in the chlorosomes.

The chemical shift data were used to refine the structural model for the chlorosomes. Firstly, the molecular structure was optimized by quantum chemical calculations for the ( $3^1 R$ ) or ( $3^1 S$ ) stereoisomers using methanol as a fifth ligand. The Mg ion

can be positioned at the same side of the porphyrin plane as the  $17^1\text{-C}$  (*syn*) or at the opposite side (*anti*). The two energetically most favorable forms are the ( $3^1 R$ )-*anti* and the ( $3^1 S$ )-*syn* combinations. Stacks of molecules of the *syn* or *anti* type can be formed by successive coordination of the  $3^1\text{-OH}$  to the Mg of the next molecule (Fig. 6). From model studies, it was concluded that the ratio between the two stereoisomers can vary considerably. Provided a minor fraction of either form is present, the rods can be formed (Steensgaard et al., 2000). This suggests that both species can be accommodated in a stack by rotating the 3-side chain.

The stacks can be combined to form layers by hydrogen bonding between the  $3^1\text{-OH}$  groups of one stack with the  $13^1\text{-C=O}$  of a neighboring stack. Evidence for this has also been found by  $^{13}\text{C}$  CP/MAS experiments. From the CP build-up of the  $^{13}\text{C}$ -carbon signal in the chlorosomes of *Cfl. aurantiacus* and in the hexane treated aggregate, it was concluded that the  $^{13}\text{C}$  carbonyl is hydrogen bonded, probably to the 3-hydroxyethyl group (Nozawa et al., 1990).

Based on the NMR results, a bilayer tube model was proposed for the chlorosome rods in *Chl. tepidum*, where a predominantly *syn* layer forms the inner tube with the tails filling the center, while a layer of *anti* character constitutes the outer tube with the tails pointing away from the surface (Fig. 6) (van Rossum et al., 2001). In the *anti* conformation, the layer of BChls corresponds to the model of Holzwarth et al., where the tails are directed outward (Holzwarth and Schaffner, 1994). The layer formed from *syn* stacks has an opposite curvature with the tails directed inward. The existence of two different parallel chain stacks with *syn* and *anti* configurations has also been detected by NMR in solution (Mizoguchi et al., 1998). The two different layers (Fig. 6) have virtually identical aggregation shifts. Evidently, the overlap in both cases is similar and induces the same ring current shifts. The bilayer model is supported by electron microscopy, since the rod-shaped structures with an outer diameter of 10 nm and an internal hole of 3 nm (Cruden and Stanier, 1970; Staehelin et al., 1980) can be formed easily from the bilayer tube inferred from the MAS NMR data. Calculations of optical spectra based on a single outer layer at least explain the measured spectra satisfactorily, and refinements including the bilayer are expected (Prokhorenko et al., 2000; Psencik et al., 2003).

For the bilayer tubes, an intensity ratio between components **I** and **II** of approx. 3:2 is observed (van

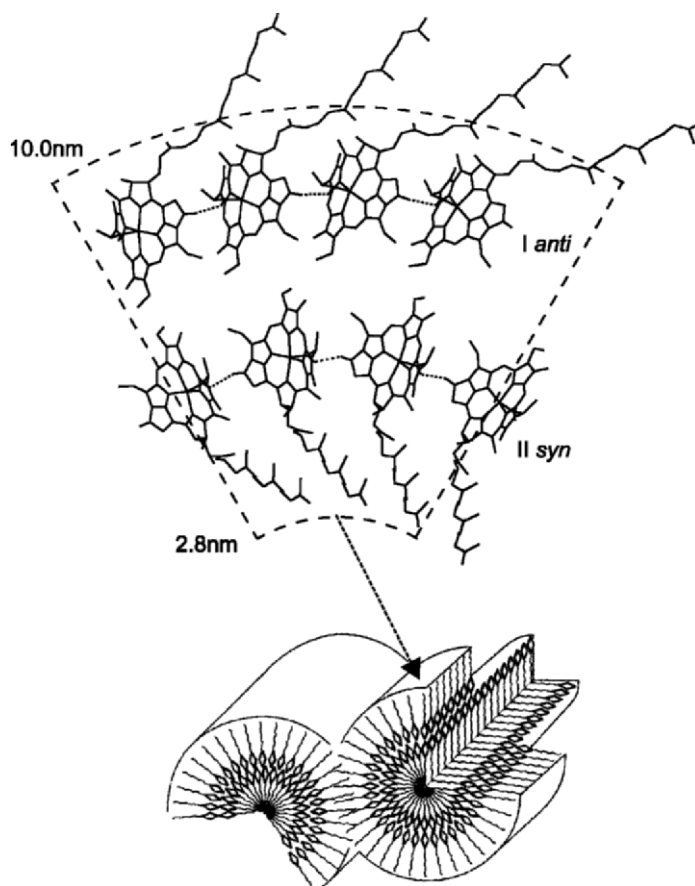


Fig. 6. Schematic representation of a radial wall section of a bilayer tube formed from curved 2-D sheets of *anti* (I) and *syn* (II) stacks. The chlorin rings are completed with the farnesyl tails, which were not included in the *ab initio* calculations. The curvature leads in a natural way to the dimensions determined with electron microscopy. The direction of the stacks is perpendicular to the plane of the paper. The dotted lines indicate hydrogen bonds between  $13^1\text{-C=O}$  and  $3^1\text{-OH}$  of adjacent stacks. In reality, the interface between the outer and inner tube is expected to be more dense than in this schematic representation, to account for the aggregation shifts of the  $7^1$  methyls.

Rossum et al., 2001). The splitting of the chemical shift pattern was explained by an interaction between *syn* and *anti* layers yielding a different electronic environment. The component I in the NMR spectra was attributed to a ( $3^1 R$ ) *anti* structure and the component II to a ( $3^1 S$ ) *syn* structure. The upfield shifts around the  $7^1\text{-CH}_3$  in component II may reflect a different structural arrangement for the 7-Me groups at the interface of the layers leading to an extended overlap for this component only. Finally, comparison of the proton responses of the farnesyl chain of BChl *c* with the phytyl chain of self-aggregated Chl *a*/H<sub>2</sub>O reveals a significant excess line broadening of the BChl *c* proton signals, suggesting that the farnesyl chain may exhibit some random folding. On the other hand, it was demonstrated that the NMR relaxation parameters of the farnesyl chain are highly similar

to those of the rigid ring system, from which it was concluded that at least a substantial fraction of the farnesyl chains should be relatively immobile (van Rossum et al., 1998b). This is consistent with the bilayer tube model of Fig. 6, since it can be expected that at least the fatty tails on the inside will be rigidly held in place.

## V. Conclusions and Future Prospects

With a combination of chemical shift data and intermolecular constraints, detailed information about the structure of chlorosomes and solid chlorophyll aggregates can be obtained. Access to the proton chemical shifts in the solid state yields aggregation shifts that can be related to the stacking of the mac-



rocycles. The intermolecular correlations provide direct structural information in terms of close contacts between molecules. A few contacts strongly reduce the number of possible space filling arrangements for the chlorophyll aggregates.

For the study of the chlorosomes, MAS NMR has proved to be a valuable tool, and the model structure of Fig. 6 is increasingly accepted (Brown and Spiess, 2001; Glaeser et al., 2002; Vassilieva et al., 2002). MAS NMR in conjunction with other spectroscopic methods and microscopy provide converging evidence for the bilayer tube model of Fig. 6. The aggregation shift pattern in the 7-C region of component II in Fig. 4 was not yet explained quantitatively from the model structure. In addition, the large shift for the 5-C is not yet fully clear.

One key issue to be addressed in the near future is to prove that specific side chains can steer the supramolecular structure through the aggregation process. In particular, the role of the ( $3^1 R$ ) and ( $3^1 S$ ) stereoisomers, the long ester tails and their interplay appears to be important. It is expected that MAS NMR will contribute in the near future by resolving the structures of model compounds and other chlorosomes such as those found in *Chloroflexus*.

More material is available on the web, see [WEBSITE]. This includes a section describing the MAS NMR methods for structure determination. In addition, the web section includes tables with the  $^1\text{H}$  and  $^{13}\text{C}$  chemical shifts for solid Chl *a*/ $\text{H}_2\text{O}$  and BChl *c* as well as the aggregation shifts relative to the shifts in solution.

### Note Added in Proof

To investigate the molecular control over the supra-structure, two model cadmium chlorins were studied that were uniformly  $^{13}\text{C}$  and  $^{15}\text{N}$  enriched in the ring moieties (de Boer et al., 2003, 2004). The chlorin models differ from the natural BChl *c* in the central metal and the 3-, 12-, 17-, and 20-side chains. One model system has the farnesyl tail replaced by a methyl, whereas the other has a stearyl tail. The  $^{113}\text{Cd}$  MAS NMR signals reveal a five-coordination of the Cd metal, very similar to the  $\text{HO}\cdot\cdot\cdot\text{Mg}$  coordination in the natural system. Large  $^1\text{H}$  ring-current shifts of up to 10 ppm reveal a dense orderly stacking of the molecules in planar layers, for which a correlation length of at least 24 Å was defined from long-range ring-current shift calculations (de Boer et al., 2004).

The ring current shift calculations are a critical assay in discriminating between tubular and planar structures, since they can be applied to confront models deduced from e.g., electron microscopy with the NMR spectroscopy data from intact chlorosomes. The model structures confirm and validate the essential role of the [ $3^1\text{R}$ ] and [ $3^1\text{S}$ ] stereoisomers in the formation of the chlorosomal antennae. The 3D arrangement of the layers is revealed by intermolecular  $^{13}\text{C}$ - $^{13}\text{C}$  correlations obtained from CHHC experiments. With the tail truncated to methyl, a microcrystalline solid is formed with favorable interactions between the planar sheets in a head-to-tail orientation. The stearyl tails lead to a considerably disordered aggregate consisting of both *syn* and *anti* layers similar to the chlorosomes, as indicated by a doubling of the  $^{15}\text{N}$ -D NMR signal. The MAS results provide converging evidence for a balance between strong local interactions and contributions to the free energy of the system associated with a longer length scale. The metal coordination and hydrogen bonding between stacks lead to a robust layer structure on a mesoscopic length scale, stable against thermodynamic noise. This allows for fine-tuning of the structure by various microscopic functionalities. The chirality of  $3^1$  side chain provides a bias for the two opposite plane curvatures, while bulky side chains like the 20 methyl provide fine tuning of the  $\pi$ - $\pi$  overlap between individual macrocycles (de Boer et al., 2004). Finally the long tails promote an organization into bilayers with a  $180^\circ$  screw axis between the layers (de Boer et al., 2003).

Note: Readers are encouraged to visit the website (<http://epub.ub.uni-muenchen.de/archive/00000776/>) for 'supplementary material,' where Figs. S1–S2 referred to in the text are posted.

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