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Low-Energy Au Ion Beam Implantation in the Optical, Structural, and Surface Properties of RF-Sputtered WO₃ Thin Films

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Abstract

 WO_3 is considered to be significant for diverse applications such as gas sensing, photocatalysts, and photovoltaic devices because of its optical wide band gap. Ion beam treatment of various metal oxides produces defects that modify various properties such as topographical, structural, and optical characteristics of the metal oxides. When the energetic ion pass through the target materials, it causes two kinds of energy losses, i.e., nuclear and electronic energy loss. In low-energy ion beam treatment of thin films, nuclear energy loss dominates over electronic energy loss. In our current study, thin films of tungsten oxide were grown on substrates of glass and silicon by the RF-sputtering method. The sputtered WO_3 thin films were exposed to an ion beam of Au iosn with an energy of 80 keV at various fluences of 5E14 ions/cm², 1E15 ions/cm², and 5E15 ions/cm². Atomic force microscopy (AFM) shows that, after implantation, enhancement in grain size was observed from 49.5 nm to 56.6 nm. The optical study displays the alteration in the energy band gap of ion-implanted WO_3 thin films from 2.75 eV to 2.99 eV. From Raman spectroscopy, the phase observed is monoclinic for virgin and implanted samples. PL spectroscopy of virgin and ion beam-implanted WO_3 thin films shows emission spectra at the wavelength of 580 nm at an excitation wavelength of 420 nm. X-ray photoelectron spectroscopy shows the appearance of tungsten and oxygen atoms and shows the electronic structure variation after Au ion beam implantation.

Keywords Thin films · RF-sputtering · ion beam · AFM · Raman · UV-vis · XRD · XPS

Introduction

In the last few years, much focus has been paid to metal oxide nanoparticles or nitride micro-and nanostructured materials due to their physicochemical, structural, optical, and corrosion properties, which show vast potential applications.^{1–3} Metal oxide micro- and nanostructures have been used in several applications, such as solar energy, photocatalysis, wastewater treatment, and microelectronics, because of their unique structural and morphological properties.^{4–7} Of numerous metal oxides, tungsten oxide (WO₃) is an auspicious material utilized by the scientific community because it is cost-effective, and has a huge band gap and

high solubility in aqueous solutions. Tungsten oxide is considered an *n*-type semiconductor having an empty perovskite structure formed by WO₃ nanostructures.^{8,9} Tungsten oxide possesses various structures including monoclinic (17-330 °C), orthorhombic (330-740 °C), and tetragonal (> 740 °C).¹⁰ The most commonly used structure of WO₃ is the monoclinic structure.⁸ WO₃ thin films have been extensively used in discrete application fields like gas-sensing,¹¹ photoelectrochemical,¹² electrochromic devices,^{13,14} and optoelectronic devices.¹⁵ Tungsten oxide has a high melting point and it has better catalytic characteristics for hydro-generation reactions.^{16–18} Tungsten oxide has been used for the decomposition of pollutants and dyes because of its redox potential.¹⁸⁻²⁰ WO₃ nanomaterial has been fabricated in different structures and different phases comprising nanoplates, nanotubes, nanoleaves, nanorods, nanoflakes, etc., and has been synthesized by various techniques, such as solvothermal synthesis,²¹ metal–organic chemical vapor deposition,²² hydrothermal processes,²³ sol-gel technique,^{24,25} magnetron sputtering, spray pyrolysis,²⁶ vacuum evaporation,²⁷ electron

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beam irradiation,²⁸ liquid-phase depositions,²⁹ flame-based solid evaporation,³⁰ electrical arc discharge method³¹ and microwave synthesis.³² Of all these deposition approaches, the sputtering approach is the best method for producing homogenous arrangements and stoichiometrics of WO₃ thin films.^{33,34} Yousif et al.³⁵ prepared WO₃ thin films by pulsed laser deposition and found the alteration in structural and optical characteristics of WO₃ thin films with treatment of heat. Haung et al.³⁶ fabricated WO₃ thin films by a hydrothermal procedure and studied the temperature variation of the structure and photochromic traits of WO₃ powder. Rathika et al.³⁷ synthesized WO₃ thin film by spray pyrolysis and studied modifications in the optical, structural, and morphological properties of WO₃ thin films.

Hence, by modification in tungsten-based materials, they can be used in various applications. Irradiation with an ion beam is a significant approach for alteration in the material at the nanoscale level. Recent research shows that ion beam irradiation modifies the characteristics of nanomaterials with high fidelity.³⁸ When the energetic ions enter the targeted material, energy is lost in two ways: electronic or nuclear energy loss. When the incident ion collides inelastically with the electron cloud of the targeted material, this causes a loss of electronic energy, while nuclear energy loss arises when the ions are incident with energy from keVs to 2 MeV and interact elastically with the targeted material.^{39,40} Electronic energy losses (Se) are overcome at high energy (> 1 MeV/ nucleon) and nuclear energy losses (Sn) are found at low energy (~ keV/nucleon). In the elastic collision, nuclear energy loss was dominated by electronic energy losses which produced point defects, development of dislocation loops, and a cluster of defects in the targeted material.^{41–44} The surface morphology alteration of the sample by irradiation with an ion beam has been used for device applications.⁴⁵⁻⁴⁷ Various research groups have shown modifications in the optical, structural, and morphological properties of the sample after ion beam irradiation. Sivakumar et al. observed the effect of an ion beam on the thin film of WO_3 irradiated by an N⁺ ion beam with MeV energy, and found devaluation in the optical band gap of WO₃ virgin and ionirradiated thin films.⁴⁸ Sharma et al.⁴⁹ irradiated thin films of WO₃ by 100 MeV Ni⁷⁺ ions with varying sputtering pressure from 10 to 50 mTorr, and found that the grain size is enhanced and the band gap decreased after ion beam irradiation. Rathika et al.³⁷ irradiated WO₃ thin film by Ag^{15+} ions with the energy of 200 MeV with varying fluences (5×10^{11} , 1×10^{12} , 5×10^{12} , and 1×10^{13} ions/cm²) and they found that the band gap decreased from 3.45 nm to 3.32 nm and surface modifications were observed after exposure to ion beams. Kumar et al.⁵⁰ grew thin films of TiO₂ by the RFsputtering method and thin films were irradiated with an ion beam with 800 keV with distinct fluences (1E15 ions/cm² to $1E17 \text{ ions/cm}^2$). They observed that the crystal size reduced from 42.09 to 27.36 nm after ion beam irradiation.

Figure 1 shows a schematic of our work. The tungsten oxide thin films were synthesized by RF-sputtering and were then exposed to ion beam irradiation with 80 keV energy and an Au ion beam at different fluences (5E14, 1E15, and 5E15) before we analyzed the variation in different properties, such as optical, morphological, and structural, of WO_3 thin films. Then, various characterizations such as X-ray diffraction (XRD), atomic force microscopy (AFM), UV-visible, Raman spectroscopy, photoluminescence (PL) spectroscopy, and X-ray photoelectron spectroscopy (XPS) were carried out.

Experimental

Thin Film Preparation

WO₃ powder was commercially purchased from Sigma Aldrich with 99.99% purity. Then, the target was formed by using a 2-inch (c. 5-cm) die with 6-tonne pressure. Thin films of WO₃ were grown at MNIT Jaipur by the RF sputtering method because it enables more economical deposition onto substrates of large areas. The thin films were deposited at room temperature by optimizing the parameter with the power of 100 W, working pressure of 3×10^{-2} with 10–12 rpm rotation, and the distance of the target from the substrate of 8 cm. After deposition of the WO₃ thin films, they were annealed at 550°C for 24 h.

Ion Beam Irradiation

After the deposition of the WO₃ thin films, they were irradiated by an Au ion beam with 80 keV energy at varying fluences of 5E14 ions/cm², 1E15 ions/cm², and 5E15 ions/cm². The WO₃ thin films were irradiated by a Pelletron accelerator tandem (15UD) at the Inter-University Accelerator Center, New Delhi. James F. Ziegler and Biersack developed the SRIM software and they also gave us Monte Carlo simulation software known as TRIM.⁵¹ This investigates various parameters such as phonon generation, ionization, losses of energy, and range of the target. SRIM/TRIM simulation software was used for observing the damage originating in the material irradiated with an ion beam. Electronic energy and nuclear energy loss with an 80-keV ion beam are illustrated in Fig. 2. The ionic path, atom displacement, and target ionization of the thin films were calculated by TRIM-2013 simulation software. The ion range was observed to be 20.5 nm and the straggling range was found to be 8.6 nm, as displayed in Fig. 3a-d. In an ion beam irradiation with low energy, nuclear energy loss is greater than the electronic



Fig. 1 Schematics representing the experimental approach: (a) WO_3 thin films synthesized by RF-sputtering; (b) ion beam implantation of WO_3 thin films with various fluences, 5E14, 1E15, and 5E15 ions/



Fig. 2 S_e and S_n plot of WO₃ thin film implanted by Au ion beam.

energy. Therefore, variation in the characteristics of thin films was caused by nuclear energy losses.

Characterization Techniques

The virgin and ion-implanted WO_3 thin films with energy of 80 keV by an Au ion beam at different fluences, 5E14, 1E15, and 5E15, were depicted by several methods such

 cm^2 ; (c) energy loss generated by the effect of irradiation of thin films with low energy; (d) AFM spectra of pristine and ion beam implanted samples with varying fluences.

as PL spectroscopy, AFM), UV-visible spectroscopy, XRD, XPS, and Raman spectroscopy. AFM was used to access the surface roughness and grain size of the thin film in tapping mode. The alteration in the optical characteristics of the WO_3 thin films was investigated by using a Hitachi U-3300 in the wavelength region (200–800 nm). Raman spectra of the thin film were carried out by a Renishaw In-Via Reflex micro-Raman spectrometer with a 488-nm Ar ion laser having a power of 50 mW and diffraction grating of 2400 lines/mm. The PL emission spectra of the WO₃ thin films were studied through the Perkin Elmer LS 55 fluorescence spectrometer having a wavelength range from 300 nm to 650 nm with an excitation wavelength of 420 nm. XPS was used to find the variation in electronic structure with an Omicron ESCA at MNIT Jaipur.

Results and Discussion

X-ray Diffraction

Using XRD, we investigated different factors such as crystal structure, full width half-maxima, dislocation density, and strain of the virgin WO₃ and implanted thin films. The XRD patterns of pristine WO₃ and the 80 keV implanted thin films



Fig. 3 Plots of (a) target ionization (b) ion distribution (c) target vacancies and (d)path along the Y axis of WO₃ thin films.

at fluences, 5E14, 1E15, and 5E15, are displayed in Fig. 4. The XRD patterns of virgin and ion-implanted thin films at different fluences were recorded from 20° to 80°, showing their crystalline behavior. Un-implanted WO₃ thin films were found at peaks 24.36°, 28.9°, 34.06°, 41.7°, 45.8°, 50.1°, 55.7°, and 61.3° with the corresponding (200), (112), (202), (222), (320), (140), (420), and (143) planes. Our experimental data matched previously reported results with JCPDS No. 43-1035.52 After ion beam implantation, no structural changes were observed up to the fluence of 5E15 ions/cm². Miyakawa et al.⁵² showed the reformation of the phases of thin films of WO₃ by helium ions or proton ions. We conclude from the literature that small changes in lattice parameters were observed by the Au ion beam because of distortion in the W-O frame. The intensity of the peaks at a fluences of 5E14, 1E15, and 5E15 sharply decreased, which shows the reduction in the crystalline nature of the samples after implantation and may also be owing to



Fig. 4. XRD spectra of pristine WO_3 and the Au-implanted sample at various fluences of 5E14, 1E15, and 5E15.

damage created during ion beam implantation of the sample by 80 keV.⁵³ The devaluation in the intensity of peaks was observed due to the partial ordering of crystal and point defects and thermal vibration produced in the sample.

The crystallite size of the WO_3 and ion-implanted samples for the plane (200) were measured by using the Debye–Scherrer's equation:

$$D = \frac{0.94\lambda}{\beta\cos\theta} \tag{1}$$

where λ is the wavelength of the X-ray, β is the full width half-maxima and θ is the diffraction angle of the sample. The crystallite size of the WO₃ thin films before and after ion implantation are shown in Table I. The crystallite size of virgin WO₃ for the plane (200) was observed to be 20 nm. After the ion beam implantation of WO₃ thin films at different fluences, 5E14, 1E15, and 5E15, the crystallite size was decreased to 12.4 nm up to the fluence of 1E15 ions/cm². The decrement in crystallite size after implantation by an Au ion beam with 80 keV energy is due to the effect of the Au ion beam which decreases the crystallinity of the sample. The crystallite size depleted after ion beam implantation is attributed to enhancement in the microstrain produced by implantation in the material which generates defects in the thin film.⁵⁴ Ion beam irradiation with an Au ion beam produces defects, mainly oxygen vacancies, in WO₃ thin films which can be due to the sputtering effect which reduces the crystallinity of the sample.³¹ We have demonstrated that oxygen vacancies produced by ion beam irradiation can be used to improve the electrical conductivity. At a higher fluence of 5E15 ions/cm², the peak vanishes, which shows the complete amorphization after ion beam implantation with 80 keV.⁵⁵ Identical observations were recorded by Solanki et al.⁵⁶

The lattice strain and dislocation density of virgin and implanted WO₃ thin films at varying fluences of 5E14, 1E15, and 5E15 ions/cm² for the plane (200) are estimated by Eqs. 2 and $3:^{57}$

$$\epsilon = \beta/4 \tan\theta \tag{2}$$

$$\delta = 1/D^2 \tag{3}$$

where ε is the strain, θ is the diffraction angle, δ represents the dislocation density, and D is the crystallite size of the sample. The lattice strain is generated by vacancies created due to the extra volume of grain boundaries, as reported by Singh et al.⁵⁸ The dislocation density decreases from pristine to 5E14 ions/cm² and fluctuations in the strain were noticed from pristine to irradiated samples at different fluences. The decrement in dislocation density with the fluence is due to the irradiation effect that generates more and more vacancies and enhancement in the demolition rate of the vacancies because of increases in density.

UV Visible Spectroscopy

The optical study of WO₃ thin films in absorbance mode was carried out to study the variation in optical characteristics before and after exposure to irradiation at distinct fluences. The absorption spectra of virgin and irradiated WO₃ thin films at fluences 5E14, 1E15, and 5E15 are shown in Fig. 5a. The absorbance of WO₃ thin films decreases after exposure to ion beam irradiation at distinct fluences. At fluence 1E15 ions/cm², the absorbance sharply decreases, which is attributed to an alteration in the optical band gap by Au ion beam implantation with 80 keV. The decrease in absorbance can be due to defects produced in the material by heat produced during the implantation of the thin films.⁵⁹

The optical band gap of WO_3 thin films was investigated by the spectra of absorption and energy of photons by employing Tauc's plot. The energy band gap of the sample was computed by using Tauc's relationship:^{60,61}

$$(\alpha h\nu)^{n} = B(h\nu - Eg)$$
⁽⁴⁾

where B is the proportionality constant, α is the coefficient of absorption and determined by the relationship $\alpha = 2.303$ (A/t), where A is the absorbance of the sample and t represents the thickness of the sample, ν is the frequency, h is Plank's constant, Eg is the optical band gap energy, and n shows the direct or indirect transition. For the direct band gap, the value of n is 2 and for n = $\frac{1}{2}$, it has an indirect transition band. The alteration in the optical band gap of virgin and implanted WO₃ thin films at varying fluences 5E14, 1E15, and 5E15, is presented in Fig. 5b. The measured values of the direct band gap of virgin and WO₃

Table I Full width half-maxima
(FWHM), crystallite size,
dislocation density and lattice
strain for plane (200) for virgin
and ion implantation with
different fluences

Fluence (ions/cm ²)	FWHM (radians)	Crystallite size (nm)	Dislocation density (δ) (×10 ¹⁶ m ⁻²)	Strain ($\times 10^{-2}$)
Pristine	0.007	20.0	0.25	0.42
5E14	0.104	14.7	0.46	0.77
1E15	0.112	12.4	0.65	0.66
5E15	-	_	-	_



Fig. 5 (a) Absorbance versus wavelength of pristine and ion beam-implanted thin films. (b) Band gap alteration of un-implanted and Au ion beam-implanted WO_3 thin films.

ion-implanted thin films with 80 keV are shown in table II. The band gap values of WO₃ virgin and implanted thin films lie in the region from 2.75–2.99 eV. The optical energy band gap value of WO₃ thin films increases from pristine to 5E15 ions/cm² except for implantation at fluence 1E15 ions/ cm^2 . The enhancement in the band gap was explained by the mechanism that arises during the implantation of the thin films. The energy was transferred by incident ions which produce latent tracks with high temperature compared with the environment. As a result of this, the defects in the forbidden space produce localized states near the conduction and valence bands. During ion beam exposure of the thin films, the heat treatment of defects occurs, which disassociates the conduction band and valence band's energy level and causes an increase in the optical energy band gap.^{62,63} The reduction in band gap at 1E15 ions/cm² was observed and can be explained by the Davis-Mott model.⁶⁴ When the interaction of ions takes place with the sample, it produces local disorder in the system. The Davis-Mott model shows that the increase in ion fluence produces enhancement in local disorder, which increases the localized energy states in the forbidden space, causing the band-gap depletion of the sample.⁶⁵

Urbach energy (E_u) provides significant band gap details and its composition from the absorption spectra of the thin films. In the optical band gap region, an exponential curve with the absorption coefficient curve was observed, called the Urbach tail. The Urbach tail has been detected because of various factors like disordered, impoverished crystalline, and amorphous nature of the material.^{66,67} In the ion beam implantation process, localized energy states arise in the energy band gap causing the broadening of absorption spectra and producing a sub-energy band gap absorption

 Table II The direct band gap and Urbach energy measured value of un-implanted and Au ion beam-implanted thin films at various fluences.

Fluence	Direct band gap (eV)	Urbach energy (eV)	
Pristine	2.75	2.33	
5E14	2.95	1.12	
1E15	2.85	1.40	
5E15	2.99	1.58	

tail known as Urbach energy.⁶⁸ Urbach energy can be estimated by Eqs. 5 and 6^{69} and shown in Fig. 6a:

$$\alpha = \alpha_0 \exp(h\nu / E_u) \tag{5}$$

$$\ln \alpha = \ln \alpha_0 \exp(h\nu / E_u) \tag{6}$$

where Eu is the Urbach energy, α_0 is a constant, and α the absorption coefficient. The plot of ln α versus photon energy is shown in Fig. 6a. The estimated values of Urbach energy of virgin and implanted WO₃ thin films at varying fluences, 5E14 ions/cm², 1E15 ions/cm², and 5E15 ions/cm², are displayed in Table II. The variation in Urbach energy after implantation at different ion fluences was observed because of variation in the defects states in the forbidden energy state.⁷⁰

The thickness at which the optical density value is equal to 1/e at the surface of the sample is known as the penetration depth or skin depth. The skin depth of the thin films leans on conductivity which is based on the energy band gap. The skin depth specifies that, when the electromagnetic waves cover the thickness of the thin films, their amplitude reduces.⁷¹ The penetration depth is computed by the



Fig.6 (a) The plot of Urbach's energy of un-implanted and Au ion beam implanted thin films at different fluences. (b) The modifications in skin depth with photon energy of pristine and implanted WO_3 thin

films. (c) Alteration in extinction coefficient with photon energy of virgin and Au ion beam-implanted WO_3 thin films.

relationship plot of skin depth versus photon energy of virgin and implanted WO₃ thin films with 80 keV, as displayed in Fig. 6b, from which we can see that the increment in skin depth with photon energy was observed until it reached the maximum level and then decreased to zero with increasing photon energy. The value of energy at which skin depth is zero is known as cut-off energy.⁶⁷ The calculated value of cut-off energy was observed to be 4.06 eV for the pristine WO₃ thin film. After ion beam implantation, the variation in height and position of the thin film was observed at fluence 5E14 ions/cm².

The extinction coefficient (*k*) is used for determining some optical measurements. It determines the fraction of the loss of light because of the absorption and scattering process of light. The plot of extinction coefficient versus photon energy of virgin and implanted WO₃ thin films at different fluences, 5E14, 1E15, and 5E15, is displayed in Fig. 6c. The extinction coefficient was calculated by the equation $k = \alpha \lambda/4\pi$.⁷² The value of the extinction coefficient decreased with the enhancement in photon energy and ion fluence. The value of extinction coefficient was a maximum for pristine WO₃ thin film and a minimum at fluence 5E14 ions/cm². It was found that variation in the extinction coefficient arises due to the effect of ion beam implantation.⁷⁰

PL Spectroscopy

To investigate the nature of defects present in the specimen, PL analysis was carried out at ambient temperature. PL spectroscopy is used for getting information about the recombination of electron-hole pairs by inducing photon energy. The electron-hole pair recombination rate is directly related to the PL intensity of the thin film samples.^{73–75} Photoluminescence spectroscopy is also used to determine various defects present in the sample, such as impurities, vacancies, interstitials, and the crystal quality of the thin films and the electronic energy band gap. The PL spectra of the virgin WO₃ and ion-implanted thin films at varying fluences, 5E14, 1E15, and 5E15, were taken in the wavelength



Fig. 7 PL spectra of pristine and implanted WO_3 thin films with 80 keV Au⁻ ions with varying fluence.

range from 380 nm to 700 nm at an excitation wavelength of 420 nm, as illustrated in Fig. 7. The emission band was observed at 580 nm for pristine and low-energy irradiated WO₃ thin films with fluence from pristine to 5E15 ions/cm². The observed emission band was associated with the localized states arising from the singly charged oxygen vacancies in the sample.⁷⁶ Oxygen vacancies have the ability to capture the excitation energy, and light can be released directly by recombination with the valence band. After implantation

by the Au ion beam, the PL intensity of the WO₃ thin film increases at fluence 5E14, and this increase in PL intensity was detected by virtue of the enhancement in the defects concentration, whole oxygen vacancies appear in the sample arising from the Au ion beam implantation. The reduction in PL intensity for the fluences 1E15 and 5E15 was observed because of defects formation which behave like a non-radiative recombination center and, because of increases in the ion fluence of the sample, the centers' concentration increases.⁷⁷ The variation in PL intensity was seen because of enhancement in oxygen vacancies after Au ion beam irradiation of the sample. The PL intensity decrement can be attributed to the demolishing of the defects by heating generated during ion implantation of the thin films.⁷⁸

Atomic Force Microscopy

AFM spectroscopy was employed for the morphological study of the virgin and 80 keV Au ion-implanted WO₃ thin films at different fluences, 5E14, 1E15, and 5E15. The force and potential energy between the sample surface and the small tip were evaluated by a probe identified as a cantilever. The cantilever tip employs repulsive or attractive forces by moving on the shallow part of the thin films. The 3D image of virgin and implanted WO₃ thin films at varying fluence is shown in Fig. 8.



Fig. 8 The AFM images of 3-dimensional virgin and Au ion beam-irradiated WO₃ thin films at different fluence from pristine to 5E15 ions/cm².

Modification in the average surface roughness and root mean square value was observed with distinct fluence. For the pristine WO₃ thin film, the average roughness (B_a) was 16.3 nm, and at higher fluence of 5E15 ions/cm² it became 14.5 nm. The devaluation in the surface roughness was recognized because of the surface diffusion due to material transportation.^{79,80} The roughness depletion may be due to a discontinuous track which produces defects like color centers.⁸¹

The plot of log-normal fitting of virgin and implanted thin films is displayed in the bar graph in Fig. 9. The grain size of the pristine WO₃ was measured at 49.5 nm and, after implantation with Au ion beam at the higher fluence 5E15 ions/ cm^2 , it was 56.6 nm. In ion beam exposure with low energy, point defects were produced by losses created by nuclear energy which enhanced the size of the grains. The variation in the grain size after implantation occurs because, as the energy was transferred by the incident ion, cluster malfunction takes place.⁸² The enhancement in grain size was because the secondary electron transfers their kinetic energy to the lattice atom from the target by the e'-phonon interaction process with the irradiation and this causes an increment in the lattice temperature of the materials resulting in morphological variation in the sample after exposure to ion irradiation. The increase in the grain size after implantation can be ascribed to the addition in the density of defects known as vacancies that act as nucleation centers for the nanocrystallization of the thin films. Liu et al. observed an increment in the grain size when Cu thin films were treated by an Ar⁺ ion beam with an energy of 200 keV.83 The increment in grain size can be due to the quenching effect produced after the bombardment of high-energy ions.

Raman Spectroscopy

Raman spectroscopy was used for structural analysis such as phase formation and Raman shift in the band of the thin films. Raman spectroscopy of virgin and implanted WO₃ thin film with fluences 5E14, 1E15, and 5E15 is displayed in Fig. 10. Raman spectra of unirradiated WO₃ thin film show two bands, one in the lower wavelength region $(100-500 \text{ cm}^{-1})$ and one in the higher wavelength region $(500-900 \text{ cm}^{-1})$. The vibrational peak found lower than 200 cm^{-1} corresponds to lattice modes and the peak above the frequency range 200 cm⁻¹ is related to stretching mode and deformation in the sample. For virgin WO₃ thin films, the bands detected at 268 cm^{-1} and 325 cm^{-1} correspond to the O-W-O bending mode of $\delta(W^{6+}-O)$. The peaks obtained at wavenumbers 714 cm⁻¹ and 807 cm⁻¹ were due to $\delta(W^{6+}-O)$ stretching modes. The peak at 520.5 cm⁻¹ is related to the optical phonon zone of the silicon substrate. A sharp intensity peak was observed at 807 cm⁻¹ which shows the monoclinic nature of the WO₃ thin films.⁶¹ Our results are matched with former reported WO₃ results.⁸⁴ After implantation of WO₃ thin films with Au ions, a small shifting of peaks was observed at fluences 5E14 and 1E15 which signifies the red shift in the thin films. The red shift was observed because of the oxygen vacancies produced when exposed to ion beam irradiation.⁶¹ The alteration in the Raman intensity was observed at fluences 5E14 and 1E15 because of ion implantation defects produced on the surface of the thin films.⁸⁵ At higher fluence 5E15, the Raman peaks completely vanish because of the amorphization of the WO₃ thin films which is ascribed to the large density of defect formation during the implantation of the thin films.⁸⁶ The



Fig. 9 Log-normal fitting of un-implanted and ion beam-irradiated thin films: (a) virgin WO₃ thin films, (b) 5E15 ions/cm².



Fig. 10 Raman spectra of a virgin and implanted sample at different fluence: (a) pristine; (b) $5E14 \text{ ions/cm}^2$; (c) $1E15 \text{ ions/cm}^2$; and (d) $5E15 \text{ ions/cm}^2$.



Fig. 11 Survey spectra of un-implanted and implanted thin films at fluences 5E14, 1E15, and 5E15.

broadening of Raman peaks with intensity shows the reduction in crystallite size matched with the XRD results.

XPS Spectroscopy

The valence state of the W atoms and oxygen vacancies in the pristine and implanted thin films was identified by XPS spectroscopy, which provides elemental composition information in the WO₃ virgin and ion-implanted thin films at fluences 5E14, 1E15, and 5E15. The XPS measurements of the thin films were carried out with 20-mA current, 15-kV voltage, and a chamber temperature fixed at 24 °C. The virgin and implanted samples at different fluences were degassed in vacuum (5.2 E–10 mbar) overnight.

The XPS survey spectra of the unirradiated and irradiated thin films with 80 keV at varying fluences are displayed in Fig. 11. The survey spectra of the RF-sputtered WO₃ thin films depict the existence of W, O, and C atoms. The observed peaks were W4f, W4d, C1s, W4p_{3/2}, W4p_{1/2}, O1s, W4s, and O KLL with the binding energies of 33.3 eV, 244 eV, 284 eV, 426.6 eV, 495.3 eV, 531.6 eV, 597 eV, and 976.1 eV, respectively. At the binding energy 976.1 eV, the O KLL peak shows a series of Auger peaks. The obtained results are consistent with previously reported results.⁸⁷ The XPS survey spectra show no transformation in the spectra before and after implantation of WO₃ thin films at lower or higher fluences.

The high-resolution spectra of tungsten and oxygen are displayed in Fig. 12a and b. The XPS spectra demonstrate two prominent peaks, $W4f_{7/2}$ and $W4f_{5/2}$, for the virgin and implanted WO₃ thin films at distinct fluence. The oxygen peak (O1s) was obtained at 531 eV for pristine WO₃ thin films, while for higher fluence 5E15 ions/cm², it was observed to be 530 eV. The core level with their binding energy for implanted WO₃ thin films at fluences 5E14, 1E15, and 5E15 is displayed in Table III. The shifting in the tungsten peak was observed after implantation with 80 keV at varying fluences, which may be due to variation in the electronic structure of the nanocrystals which also alters the binding energy. From the figure, we can notice that the tungsten atom contains more than one peak, which shows the existence of a more valence state of the W atom. After deconvolution of the pristine WO₃ and higher fluence 5E15 thin films, as illustrated in Fig. 13a and b, the W 4f spectra show two tungsten species, W⁶⁺ and W⁵⁺, with binding energies 35.9 eV, 34 eV, 32.4 eV, and 31.7 eV.⁸⁸ The presence of W⁵⁺ atoms was observed because the deficiency of oxygen occurs during the deposition of the thin films.⁸⁹

Conclusions

RF-sputtered WO₃ thin films were irradiated with fluences of 5E14, 1E15, and 5E15 by 80 keV energy to study the consequences of a low Au ion beam on the structural, optical, and morphological traits of pristine and irradiated WO₃ thin



Fig. 12 High-resolution spectra of (a) W atoms with varying fluence and (b) O atoms of the pristine and implanted WO_3 thin film at fluences 5E14, 1E15, and 5E15.

Table III. The orbital splitting and binding energy of implanted and un-implanted thin films with 80 keV.

Sample	Binding energy (W4f _{7/2})	Binding energy $(W4f_{5/2})$	Spin–orbit splitting (δ4f)
Pristine	34.1 eV	36.1 eV	2.0
5E14 ions/cm ²	34.5 eV	36.4 eV	1.9
1E15 ions/cm ²	35.9 eV	37.9 eV	2.0
5E15 ions/cm ²	33.4 eV	36.3 eV	2.9

films. When the thin films were subjected to ion beams with low energy, the nuclear loss was greater than the electronic energy losses. AFM spectra showed that the roughness of the WO₃ thin films after implantation decreased from 16.3 nm to 14.5 nm. The size of the grains was enhanced after implantation from pristine to 5E15 ions/cm² from 49.5 nm to 56.6 nm. The XRD showed the reduction in crystallite size from pristine to fluence of 1E15 ions/cm² from 20 nm to 12.4 nm, and at 5E15 ions/cm², the peak completely vanishes because of implantation at 80 keV. In PL spectroscopy, an emission band was observed at 580 nm and, after implantation with an Au ion beam, an enhancement in the intensity was observed at fluence 5E14 ions/ cm². From the optical study, several characteristics of the WO₃ thin film were observed, such as absorbance, Urbach energy, skin depth, and extinction coefficient. The absorbance of the WO₃ thin film after implantation by the Au ion beam decreased due to variations in the optical band gap and the generation of defects during ion beam implantation. The



Fig. 13 (a) Deconvoluted spectra of pristine WO₃ thin film. (b) Deconvoluted spectra of WO₃ thin film at fluence 5E15 ions/cm².

band gap alteration from virgin to implanted WO₃ thin film was observed in the range from 2.75 eV to 2.99 eV. Raman spectra of the virgin and ion irradiated sample showed the monoclinic structure with a prominent peak at 807 cm⁻¹. Variation in Raman intensity at fluences 5E14 and 1E15 was observed and it completely vanishes at fluence 5E15 ions/ cm² because the WO₃ thin film was completely amorphized. XPS study showed the shifting in the tungsten peak after implantation with 80 keV at varying fluences, which may be due to variation in the electronic structure of the nanocrystals which also alters the binding energy.

Future Prospectives

The ion beam treatment is an effective progressive method that has a variety of applications in the semiconductor industry. Various defects and dopants in semiconductors have been created by the ion beam irradiation process for the fabrication of nanostructures or microstructures. However, recent research shows that it is a powerful tool for thin films and is also used for a variety of applications, such as superconductivity, piezoelectricity, metal-insulator transition, and gas-sensing properties.⁹⁰ The characteristics of metal oxide semiconductors are tuned by ion beam treatment by modifying their electron density, lattice distortion, and defect states. Various research groups have applied the ion beam technique to increase the performance of metal oxide thin films. Metal oxide semiconductors such as ZnO,⁹¹ SnO₂,⁹² TiO_2 ,⁹³ and In_2O_3 ⁹⁴ have been greatly used in the field of gas sensors, such as nitrogen dioxide sensors,⁹⁵ oxygen sensors,⁹⁶ LPG sensors,⁹⁷ carbon monoxide sensors,⁷⁷ etc. In future, we will study the gas-sensing properties of WO₃ thin films irradiated at different fluences for various gases such as CO₂, NH₃, and NO₂. Sensor properties like selectivity, stability, reversibility, and sensitivity depend upon crystallinity, vacancies, porosity, surface morphology, electronic characteristics, and thin-film thickness characteristics. The ion beam treatment method enhances the performance of the sensor by reducing thin-film resistivity, decreasing grain size, improving gas sensitivity, and so on.⁹⁸

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Ethical Approval Not Applicable

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5949

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