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# **Use of homoarginine for measuring true ileal digestibility of amino acids in food protein**

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**Abstract** A useful application of homoarginine in animal nutrition is the determination of the true ileal digestibility (TID) of amino acids (AA) in swine complete diets and feed ingredients. The homoarginine method involves the conversion of dietary lysine to homoarginine in a guanidination reaction with methylisourea. Accurate determination of TID of AA, especially in heat-treated feed ingredients, is a key prerequisite for accurate diet formulation with respect to the provision of dietary AA. Thus, the aim of this review is to highlight the homoarginine methodology and its application in animal nutrition. Based on the data from published studies, the

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homoarginine method can be used to accurately determine the digestibility of lysine and the majority of other acid-stable AA in complete diets and feed ingredients fed to animals.

**Keywords** Homoarginine · Guanidination · Amino acids · True ileal digestibility

# **Abbreviations**

- AA Amino acids
- AID Apparent ileal AA digestibility
- TID True ileal digestibilities

# **Introduction**

Protein and amino acids (AA) are important nutrients that must be supplied in sufficient quantities and proportions to support optimal animal performance (Wu [2013a](#page-7-0), [b,](#page-7-1) [2014](#page-7-2)). As protein is an expensive nutrient, balancing its dietary supply with requirements is critical for controlling feed cost and for minimizing nitrogen excretion in manure, which is implicated in environmental pollution (Wu et al. [2014](#page-7-3)). To this end, it is important to accurately determine the availability of dietary AA in feeds and feed ingredients so as to allow accurate diet formulation with respect to optimal protein nutrition (Yin et al. [1993](#page-8-0), [2000](#page-8-1)).

For non-ruminant animals, it is now accepted that AA digestibility coefficients determined at the end of the small intestine provide a better estimate of the amounts of AA that are available to the animals (Yin et al. [1991,](#page-8-2) [1994,](#page-8-3) [2002](#page-8-4); Wu et al. [2014](#page-7-3)). For a long time, apparent ileal AA digestibility (AID) coefficients were determined and widely used to evaluate the availability of dietary AA in broiler chicks (Perryman and Dozier [2012;](#page-7-4) Rochell et al. [2012](#page-7-5)), pigs (Stein et al. [2005;](#page-7-6) Xue et al. [2014\)](#page-8-5), ducks (Kong and <span id="page-1-0"></span>**Fig. 1** Conversion of lysine to homoarginine via guanidination reaction

$$
H_2N-CH_2-CH_2-CH_2-CH_2-CH_2
$$
  
|  
NH<sub>2</sub>

Lysine



<span id="page-1-1"></span>**Fig. 2** Preparation of the *O*-methylisourea solution and guanidination reaction

converted to homoarginine, which is normally absent from diets fed to animals (Schmitz et al. [1991\)](#page-7-19). It is assumed that, compared with diet-derived homoarginine, there is little or no secretion of endogenous homoarginine into the lumen of the small intestine (Nyachoti). Thus, the apparent digestibility of homoarginine represents the TID of lysine (Nyachoti et al. [1997a\)](#page-7-16). The process of producing homoarginine involves primarily two steps: preparation of methylisourea solution and the guanidination reaction. Although detailed procedures for each one of these steps have been modified by various research groups (e.g. Nyachoti et al. [2002](#page-7-20); Boucher et al. [2009\)](#page-6-3), the basic procedures are shown in Fig. [2.](#page-1-1)

#### **Preparation of the methylisourea solution**

The methylisourea solution is prepared by reacting methylisourea with barium hydroxide  $[Ba(OH)_2]$  followed by

Adeola [2010\)](#page-7-7), and dogs (Hendriks and Sritharan [2002](#page-7-8)). However, due to the confounding effects of endogenous AA contribution to ileal digesta, which are not accounted for in the determination of AID coefficients, utilization of such coefficients in diet formulation increases the risk for inaccurate diet formulation. This is because AID values underestimate the true ileal digestibilities (TID) of AA in complete diets and feed ingredients (de Lange et al. [1990\)](#page-6-0).

The endogenous AA originate from various digestive secretions, mucoproteins and desquamated epithelial cells that enter the gastrointestinal tract at various segments during the normal processes of digestion and absorption (Moughan et al. [1998;](#page-7-9) Ravindran et al. [2009\)](#page-7-10). To overcome the shortcomings associated with AID coefficients, it is now recommended that TID coefficients be used in formulating poultry and swine diets with respect to dietary AA supply (NRC [2012\)](#page-7-11). The TID values are obtained by correcting AID values for basal (minimum) endogenous AA (Stein et al. [2007](#page-7-12)). Determination of TID of AA in diets provides accurate data on the release of AA from dietary protein into the lumen of the small intestine for absorption (Fan and Sauer [2002;](#page-6-1) NRC [2012;](#page-7-11) Rutherfurd et al. [2006;](#page-7-13) Wu [2013b](#page-7-1)).

For determining TID of AA, the undigested dietary AA present in the ileal digesta and the endogenous losses of AA should be accurately evaluated. Currently, there are several methods to differentiate between the undigested dietary AA from endogenous AA losses in animals. These methods include feeding N-free diets (Adedokun et al. [2007\)](#page-6-2), regression analysis (Fan and Sauer [2002](#page-6-1)), as well as the use of enzyme-hydrolysed proteins (e.g., casein) coupled with ultrafiltration, the homoarginine technique, and <sup>15</sup>N-labeled proteins (Kluess et al. [2010\)](#page-7-14). In the homoarginine method, the test feed ingredient is subjected to a guanidination treatment in which dietary lysine is converted to homoarginine (Fig. [1](#page-1-0)). As the absorbed guanidinated protein cannot reappear in the digestive tract, homoarginine concentrations in chime have been used to distinguish between exogenous AA and ileal endogenous AA in animal nutrition (Huang et al. [2003](#page-7-15); Nyachoti et al. [1997a](#page-7-16), [b](#page-7-17); Pomar et al. [2008;](#page-7-18) Stein et al. [2007](#page-7-12)). This review focuses mainly on the use of the homoarginine method to determine endogenous AA flow and the TID of AA in complete diets and feed ingredients fed to animals.

## **The homoarginine technique**

The basic principle underlying the homoarginine technique is that, upon treatment with methylisourea, dietary lysine is centrifugation at  $5000 \times g$  to remove the precipitated barium sulfate. In general, BaOH is mixed with degassed water (distilled water boiled for 30 min) and then the solution is heated to 95 °C before adding a known amount of methylisourea (0.3–0.6 M) to the solution. The resulting mixture is centrifuged at  $5000 \times g$  for 15 min, and the supernatant fluid is recovered and its pH is adjusted to pH 11–12 with 1 M HCl before use (Nyachoti et al. [1997a,](#page-7-16) [b,](#page-7-17) [2002;](#page-7-20) Pahm et al. [2008a](#page-7-21), [b](#page-7-22)).

#### **Guanidination reaction**

After preparation of the methylisourea solution, an amount of the test material calculated to contain 200 g of crude protein is thoroughly mixed with 1 L of the solution and adjusted to pH 10.5 using 1 M NaOH. The mixture is then incubated in a refrigerator at 4 °C for 4–6 days (Nyachoti et al. [1997a\)](#page-7-16). Incubation at the low temperature of  $4^{\circ}$ C is critical to minimize AA damage due to the alkaline conditions. Of note, some investigators incubated the mixture for the guanidination reaction at room temperature (20–25 °C) for 1–3 days (Fontaine et al. [2007\)](#page-6-4). During the incubation, the pH is monitored daily and adjusted accordingly to ensure uniform conditions in the mixture. At the end of the incubation period, the guanidination reaction is stopped by lowering the pH of the mixture to the isoelectric point of the test protein (e.g., casein, 4.6; barley, 5.6; canola meal, 4.6) using 1 M HCl so as to maximize the precipitation and, therefore, the recovery of the guanidinated protein. The mixture is then centrifuged at  $4000 \times g$  and  $4 \degree C$ , and the supernatant fluid is discarded. The guanidinated protein is washed three times with distilled water whose pH is adjusted to the isoelectric point of the test protein, and the protein sample is freeze-dried before use for diet preparation. Finally, the contents of lysine and homoarginine in the guanidinated protein are measured to calculate the extent of lysine conversion to homoarginine. This is generally calculated according to the following equation:

Lysine conversion  $(\% )$ 

 $=$   $\left[ \text{MC}_{\text{homoarginine}} / \left( \text{MC}_{\text{homoarginine}} + \text{ MC}_{\text{lysine}} \right) \right] \times 100$ 

where  $MC_{homoarginine}$  and  $MC_{lysine}$  are the molar contents (mol/kg of DM) of homoarginine and lysine in the guanidinated protein, respectively.

# **Evaluation of the guanidination reaction**

#### **Rates of lysine conversion into homoarginine**

The extent of lysine conversion into homoarginine varies with the type of materials. As shown in Table [1,](#page-3-0) for ingredients in which the lysine residues are easily accessible to the methylisourea solution (e.g., casein), the extent of lysine guanidination ranges from 83.0 to 99.6 % (Nyachoti et al. [2002\)](#page-7-20). Similar lysine conversion rates have been observed for canola meal, soybean meal, wheat shorts, seed meal, lupin and fish meal. However, the conversion of lysine in cottonseed protein to homoarginine is very low (36.1 %) according to Ravindran et al. [\(1996](#page-7-23)). These researchers investigated different incubation times (24–144 h), lysine:methylisourea ratios (1:8–1:32), and pH (9.5–13.0) on the guanidination of lysine in cottonseed protein, and recommended that the optimum conditions for the maximum guanidination of cottonseed protein are lysine:methylisourea ratio, 1:12; pH, 12.5; and incubation time, 72 h (Ravindran et al. [1996](#page-7-23)). Meanwhile, the data from Ravindran's group also indicated that different reaction conditions (i.e., chemical reagents, reaction period, pH, etc.) can lead to differences in lysine guanidination in the same type of feed material. Nyachoti et al. ([2002\)](#page-7-20) further demonstrated that there are significant interactive effects of methylisourea concentration and incubation time on the extent of lysine conversion into homoarginine in barley and canola meal (Fig. [3](#page-3-1)) (Nyachoti et al. [2002](#page-7-20)). An increase in the extent of lysine conversion in both the barley and canola meal samples was observed when the concentration of methylisourea was increased from 0.4 to 0.5 M, but no further increase was observed when the methylisourea concentration was increased beyond 0.5 M irrespective of incubation period (Nyachoti et al. [2002](#page-7-20)). Meanwhile, incubation for 6 days has a higher rate of conversion of lysine into homoarginine in both the barley and canola meal samples, compared with incubation for 4 days. Based on these findings, it was recommended that a methylisourea concentration of 0.5 M and a 6-day incubation period can be used to convert lysine to homoarginine for measuring endogenous AA flow and digestibility of barley and canola meal, and perhaps other plant protein sources in the small intestine (Nyachoti et al. [2002\)](#page-7-20). Indeed, in the study of Friesen et al. [\(2006](#page-7-24)) based on the method of Nyachoti et al. [\(2002](#page-7-20)), the rate of the conversion of lysine into homoarginine in different pea cultivars was 96 %.

Fontaine et al. ([2007\)](#page-6-4) analyzed the effects of the guanidination reaction on total lysine and homoarginine in soy products and in corn distillers dried grain with solubles. These ingredients were subjected to deliberate heat damage for up to 30 min in an autoclave with 135 °C hot steam. In this case of heat treatment, both total lysine and homoarginine were decreased in a time-dependent manner, but homoarginine became a more sensitive indicator of lysine damage than was total lysine. Rutherfurd and Moughan [\(1997](#page-7-25)) treated skim milk powder and peas with various temperatures (110, 121, 135, 150, and 165 °C) and found that total lysine decreased by 33 % after 10 min of heating,

<span id="page-3-0"></span>**Table 1** The extent of lysine conversion to homoarginine in different materials

Materials	Rates of conversion (%)	L:M	pH	Incubation time	References
Casein sample	83.0-99.6	0.5 <sub>M</sub>	10.5	4 days at $4^{\circ}$ C	Libao-Mercado et al. (2006), Nyachoti et al. (1997b, 2002)
Barley	$72.5 - 88.0$	1:14	10.5	6 days at $4^{\circ}$ C	Nyachoti et al. (1997b, 2002)
Canola meal	$72.3 - 86.6$	1:4	10.5	6 days at $4^{\circ}$ C	Nyachoti et al. (1997b, 2002)
Soybean meal	$68.0 - 74.5$	0.5 <sub>M</sub>	10.5	1 day at room temperature or 4 days at 4 $^{\circ}$ C	Pomar et al. (2008), Ravindran et al. $(1996)$ , Siriwan et al. (1994)
Wheat shorts	$62.0 - 62.2$	0.5 <sub>M</sub>	10.5	6 days at $4^{\circ}$ C	Libao-Mercado et al. (2006), Siriwan et al. (1994)
Oilseed meal	68.8-79.8	0.5 <sub>M</sub>	10.5		Nyachoti et al. (2002), Siriwan et al. (1994)
Lupin and fish meal	$69.0 - 74.5$		10.5	1 day at room temperature	Ravindran et al. (1996)
Maize	57.3	0.6 <sub>M</sub>	10.5	4 days at $4^{\circ}$ C	Siriwan et al. (1994)
Meat meal	50	0.6 <sub>M</sub>	10.5	4 days at $4^{\circ}$ C	Siriwan et al. (1994)
Sunflower meal	49.3	0.6 <sub>M</sub>	10.5	4 days at $4^{\circ}$ C	Siriwan et al. (1994)
Cottonseed protein	$36.1 - 36.7$	1:12	12.5	3–4 days at $4^{\circ}$ C	Ravindran et al. (1996), Siriwan et al. (1994)
Soy products, $CP = 36.6$ , 42.6, or 46.8 $%$	$60 - 82$	0.6 <sub>M</sub>	11.5	2 days at room temperature	Fontaine et al. (2007)
DDGS, $CP = 23.8$ or 27.0 %	$38.1 - 78$	0.6 <sub>M</sub>	12	2.5 days at room temperature Fontaine et al. (2007)	
Milk-based products	$81.1 - 100$	0.6 <sub>M</sub>	$10.6 - 11.0$	1 or 7 days at 21 $^{\circ}$ C	Rutherfurd and Moughan (2005)

*L:M* lysine:methylisourea or methylisourea concentration, *DDGS* distillers dried grain with solubles



<span id="page-3-1"></span>**Fig. 3** Effects of incubation days and methylisourea concentration on the conversion of lysine into homoarginine in barley and canola (Nyachoti et al. [2002\)](#page-7-20)

whereas homoarginine decreased by 82 %. Although heating of feed ingredients at high temperatures may not affect the guanidination reaction, this processing condition results in damage to lysine and homoarginine. Thus, heat processing should be consistent for feed ingredients when the homoarginine method is used for measuring TID of lysine.

#### **Guanidination on chemical composition**

The guanidination process is associated with incubation, freeze-drying, and washing the test material. Thus, it has been speculated to influence the chemical composition of the guanidinated material or diet. Nyachoti et al. ([2002\)](#page-7-20) have investigated the effects of guanidination treatment on dry matter, crude protein, crude fiber, crude fat, ash, and AA in the barley and canola meal (Nyachoti et al. [2002](#page-7-20)). The results showed that a 6-day incubation period in 0.5 M methylisourea solution and subsequent washings reduced dry matter and ash content in the guanidinated samples, but increased crude protein and crude fiber contents.

The loss of dry matter and ash content may result from the homogenization of the sample between multiple washings. In this process, the sample is solubilized in the washing solution and the supernatant fluid is discarded after centrifugation. The reason for the increased crude protein and crude fiber contents may be a selective removal of nonprotein material during the guanidination process or to an incomplete removal of excess methylisourea solution during the washing procedure, which also removes soluble proteins and carbohydrates (Caine et al. [1998](#page-6-5); Nyachoti et al. [2002](#page-7-20)).

#### **Nutrient digestibility after guanidination**

A previous study has indicated that feeding a diet with guanidinated protein may have detrimental effects on feed intake and weight gain in chicks (Aoyagi and Baker [1994](#page-6-6)). However, Nyachoti et al. [\(2002](#page-7-20)) have compared the apparent ileal digestibilities of dry matter, crude protein, and AA

in the unguanidinated and guanidinated diets and found that there is no difference in the ileal digestibilities of these nutrients after the guanidination procedure (Nyachoti et al. [2002](#page-7-20)). In addition, Pomar et al. ([2008\)](#page-7-18) also demonstrated that feeding a guanidinated diet has no effect on feed intake, while the apparent ileal digestibility of N in diets containing guanidinated soybean meal protein is reduced by 4.5 %. In that study, feeding guanidinated soybean meal protein also decreased the digestibilities of several AA, including alanine, lysine, and isoleucine in pigs (Pomar et al. [2008](#page-7-18)). However, after a careful analysis of the data from the study by Pomar et al. [\(2008](#page-7-18)), it is apparent that the main reason for the decreased digestibilities of AA may be associated with the losses of material during the guanidination process as the contents of these AA are largely reduced by the guanidination procedure. Of note, this observation is not universal (e.g., Nyachoti et al. [2002\)](#page-7-20). Thus, these investigations support the use of the homoarginine method for determining TID of AA in commercial diets fed to animals.

#### **Effects of guanidination on AA racemization**

Compelling evidence has indicated that various conditions (e.g., alkaline and high temperature) can potentially cause L-amino acids racemization into D-amino acids (Chang et al. [1999](#page-6-7); Steen et al. [2013](#page-7-29)). Thus, a particular worry during guanidination procedure is that guanidinated proteins at alkaline pH may cause formation of p-amino acids, which would underestimate AA digestibilities when the homoarginine labeling is used for investigation of the endogenous AA flow as the TID of AA. de Vrese et al. [\(1994](#page-6-8)) evaluated protein racemization during guanidination of casein at pH values between pH 9 and 11 and temperatures between 4 and 65 °C (de Vrese et al. [1994](#page-6-8)). In this study, the optimal conditions for the guanidination reaction were determined to be 4  $\degree$ C and pH 10.5–11.0 or 22  $\degree$ C and pH 10. Higher pH values, and a temperature of 22 °C or temperatures above 22 °C at each pH, lead to the formation of appreciable amounts of p-amino acids.

# **Application of the homoarginine technique in animal nutrition**

#### **Selection of digestibility markers**

The homoarginine technique relies on the use of digestibility markers to make the necessary calculations. Because of the high cost of the methylisourea, it is too costly to feed test diets in which all the protein is guanidinated for an extended period of time, especially when working with large animals such as pigs. Consequently, animals are adapted to the experimental diets containing the test protein that has not been guanidinated and followed by a meal or two of the diet containing guanidinated protein. Thus, different digestibility markers (e.g. chromic oxide, dysprosium chloride, and titanium dioxide) are used to allow calculations of apparent and true ileal AA digestibilities based on the digesta from the non-guanidinated and guanidinated proteins, respectively.

# **TID of lysine**

As stated previously, the apparent digestibility of homoarginine is taken to represent the TID of lysine. The flow of homoarginine at the distal ileum can be calculated based on its concentration and the concentration of the marker associated with the diet containing guanidinated protein in the diet and digesta (Marty et al. [1994\)](#page-7-30). Calculation of TID of lysine is calculated as follows (Pomar et al. [2008\)](#page-7-18):

$$
\text{TID}_{\text{lysine}}(\%)
$$

= [(homoarginine<sub>diet</sub> – homoarginine<sub>flow</sub>)/homoarginine<sub>diet</sub>] × 100,

where homoarginine $_{\text{diet}}$  and homoarginine $_{\text{flow}}$  are dietary and digesta homoarginine concentrations, respectively.

#### **Endogenous lysine loss**

Endogenous lysine loss ( $\text{Endo}_{\text{lvsine}}$ ) at the terminal ileum (g/kg of DMI) can be calculated using the formula (Libao-Mercado et al. [2006](#page-7-26)):

 $\text{Endo}_{\text{lysine}}(\%) = \text{dict}_{\text{lysine}} \times (\text{TID}_{\text{lysine}} - \text{AID}_{\text{lysine}})/100,$ 

where diet<sub>lysine</sub> and  $\text{AID}_{\text{lysine}}$  are lysine content (g/kg of DM) and the apparent ileal digestibility (%) of lysine in the diet, respectively.  $\text{AID}_{\text{lvsine}}$  can be calculated based on concentration of lysine in the diet and digesta, as well as the digestibility marker added to the diet containing unguanidinated protein (Stein et al. [2007](#page-7-12)).

# **TID of AA other than lysine**

As pointed out by Friesen et al. ([2006\)](#page-7-24), in the homoarginine technique, only the endogenous losses of lysine and, therefore, the TID of lysine are directly determined. The endogenous gut losses of other AA are derived based on the ratio of homoarginine to the contents of other AA in the guanidinated diet and ileal digesta (de Lange et al. [1990](#page-6-0); Siriwan et al. [1994](#page-7-27)). Once the endogenous flow of each AA has been established, the TID  $(TID_{AA})$  values for AA other than lysine are calculated according to Nyachoti et al. [\(1997b](#page-7-17)):

 $TID_{AA}(\%) = AID (\%) + (100 \times [Endo_{AA}/AA_{diet}]),$ 

where AID, Endo<sub>AA</sub>, and  $AA<sub>diet</sub>$  are the apparent ileal digestibility (%) of AA, endogenous ileal AA loss (g of

<span id="page-5-0"></span>**Table 2** Application of homoarginine method in animal nutrition



DMI/kg), and AA content in the diet (g of DM/kg), respectively. Alternatively, the  $TID<sub>AA</sub>$  value also can be calculated as follows:

$$
TID_{AA}(\%) = \{ [AA_{\text{dict}} - (AA_{\text{flow}} - \text{EndoAA}_{\text{flow}})] / AA_{\text{dict}} \} \times 100,
$$

where  $AA_{flow}$  and Endo $AA_{flow}$  are the flow and the endogenous flow of AA at the distal ileum, respectively (Pomar et al.  $2008$ ). Again, the EndoA $A_{flow}$  of AA rather than lysine is calculated based on the observed endogenous lysine flow and the composition of AA in endogenous secretions relative to lysine (de Lange et al. [1990;](#page-6-0) Pomar et al. [2008\)](#page-7-18).

The homoarginine method has been used to study the digestibility of AA in feed ingredients for non-ruminant animals (especially pigs) and in investigating the effects of various factors that influence dietary AA utilization for protein deposition in animals. In this review, we have highlighted several investigations on the application of the homoarginine technique to quantify endogenous AA flow at the distal ileum to allow for the determination of the TID of AA in feed and feed ingredients (Table [2](#page-5-0)). However, some investigators have criticized the suitability of this technique in studying AA digestibility and utilization in animal nutrition. For instance, Hodgkinson et al. [\(2003](#page-7-31)) indicated that the homoarginine technique underestimates ileal endogenous lysine and nitrogen flows, compared with other methods. On the contrary, in the study by Friesen et al. [\(2006](#page-7-24)), the TID of some AA exceeded 100 %, which was attributed to the possibility that the homoarginine technique may overestimate the endogenous flow of some AA. Furthermore, it has been argued that the guanidination procedure may change the capacity of digestive enzymes to hydrolyze the guanidinated protein (Hodgkinson et al. [2003](#page-7-31)), which can potentially affect the availability of AA and peptides for absorption (Pomar et al. [2008](#page-7-18)). However, in the study by Nyachoti et al. ([2002\)](#page-7-20), it was demonstrated that the process of guanidination per se does not influence the digestibility of protein in growing pigs. Similarly, Boucher et al.

[\(2009](#page-6-3)) reported a high degree of consistency in nitrogen and AA composition of ileal endogenous secretions. These authors surmised that it is unlikely that the homoarginine technique underestimates ileal endogenous lysine followed by overestimating homoarginine concentration because the formation of homoarginine is specific to the ε-amino group of lysine (Boucher et al. [2009\)](#page-6-3). Based on the findings of these studies and digestive physiology, it can be argued that dietary composition can influence endogenous gut losses of protein and AA (Bergen and Wu [2009\)](#page-6-9). Accurate analysis of AA in both diets and intestinal digesta should be performed to correctly determine dietary AA intake, endogenous AA flow in the small intestine, and digestibilities of AA in feedstuffs (Wu et al. [2013\)](#page-7-32).

Comparison of homoarginine labeling with other methods further supports the use of the homoarginine method to measure the TID of AA and endogenous flow of AA (Boucher et al. [2009;](#page-6-3) Hodgkinson et al. [2003](#page-7-31)). Rutherfurd and Moughan ([1997,](#page-7-25) [2005](#page-7-28)) concluded that the homoarginine technique provides more accurate estimates of available lysine in soy products and some purified dairy-based protein sources (including whole milk protein, whey protein concentrate, evaporated milk, sports formula, and lactose-hydrolyzed milk powder) than do other available methods. According to Souffrant et al. [\(1997](#page-7-33)), the endogenous AA flows determined with the homoarginine technique are quantitatively greater than those determined with the  $15N$ -perfusion technique, and the homoarginine technique provides more consistent values in comparison to data obtained with the  $\rm{^{15}N\text{-}perfusion}$  technique. Roos et al. [\(1994](#page-7-34)) reported similar findings and also suggested that incorporation of  $15N$  into endogenous proteins and re-entry into the intestinal lumen via secretions and cell sloughing are the major cause for the lower digestibility values obtained with the  $15N$  perfusion technique than with the homoarginine technique. Compared with the furosine method for determining reactive lysine, the homoarginine method is positively correlated with standardized lysine

digestibility and, therefore, can be used to predict lysine digestibility from soybean meal, distillers dried grains with solubles, and fish meal samples (Boucher et al. [2009](#page-6-3)). Furthermore, the effect of microbial phytase on the AID and TID of AA in pigs determined with the homoarginine technique gives similar values as the casein-based method (Pomar et al. [2008;](#page-7-18) Traylor et al. [2001](#page-7-36)).

Various studies indicated that the homoarginine technique can be used to estimate TID of AA in addition to lysine in diets on the basis of endogenous lysine flow and the composition of AA in endogenous secretions relative to lysine (de Lange et al. [1990](#page-6-0); Siriwan et al. [1994\)](#page-7-27). Rutherfurd and Moughan [\(2005\)](#page-7-28) reported that the differences in the TID of 80 % acid-stable AA (including lysine) between the homoarginine technique and the conventional methods were less than 3 % and argued that such a small difference was not meaningful in practical terms. However, there were larger differences in the TID of the other 20 % acid-stable AA (including histidine) between the homoarginine technique and the conventional methods (Rutherfurd and Moughan [2005\)](#page-7-28). Taken together, the homoarginine technique can be used to determine the TID of lysine and the majority of other acid-stable AA in diets and ingredients fed to animals.

#### **Conclusion and perspectives**

Homoarginine is virtually absent from conventional foods of plant and animal origin, and is present at exceedingly low concentrations in animal tissues (May et al. [2015;](#page-7-37) Kayacelebi et al. [2014](#page-7-38), [2015\)](#page-7-39). Homoarginine is formed from the lysine residue in food protein after its treatment with methylisourea before feeding and the subsequent hydrolysis of the guanidinated protein by proteases in the lumen of the small intestine. In the field of animal nutrition, the homoarginine technique has been used to measure the TID of lysine and other AA in swine diets and feed ingredients. In this method, the test feed ingredient is subjected to a guanidination treatment in which dietary lysine is converted to homoarginine. The absorbed homoarginine does not reappear in the digestive tract during the period of measurement of endogenous AA flows, and, therefore, the apparent digestibility of homoarginine is considered to represent the TID of lysine (Nyachoti et al. [1997a](#page-7-16), [b](#page-7-17)). Although much is known about the application of the homoarginine technique in animal nutrition, some concerns should be addressed. For example, the uniform guanidination of lysine in a protein source is required and the optimum conditions of guanidination for a certain protein source should be investigated if the homoarginine is employed widely to determine true digestibilities of AA in new feed ingredients. Furthermore, the TID of AA varies with different feed ingredients and physiological status in animals. In our studies, we found that various stresses

(e.g., birth, weaning, and mycotoxins contaminated feed) and dietary supplementation with certain AA can affect AA absorption and metabolism as well as animal health (Yin et al. [2013](#page-8-6), [2014a](#page-8-7), [b,](#page-8-8) [2015\)](#page-8-9). Based on the available evidence, the homoarginine technique can be used to determine the TID of lysine and the majority of other acid-stable AA in swine diets and ingredients. Further studies are needed to validate its utility for studying AA digestibility in diets fed to ruminants, poultry, fish, and other animal species.

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